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The protective role of CsNPs and CurNPs against DNA damage, oxidative stress, and histopathological and immunohistochemical alterations induced by hydroxyapatite nanoparticles in male rat kidney

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Hydroxyapatite nanoparticles (HAP-NPs) are an inorganic component of natural bone and are mainly used in the tissue engineering field due to their bioactivity, osteoconductivity, biocompatibility, non-inflammatory, and non-toxicity properties. However, the current toxicity data for HAP-NPs regarding human health are limited, and only a few results from basic studies have been published. Therefore, the present study was designed to investigate the beneficial role of chitosan nanoparticles (CsNPs) and curcumin nanoparticles (CurNPs) in alleviating nephrotoxicity induced by HAP-NPs in male rats. The results showed that HAP-NPs caused a reduction in antioxidant enzymes and induced lipid peroxidation, nitric oxide production and DNA oxidation. Moreover, HAP-NP administration was associated with intense histologic changes in kidney architecture and immunoreactivity to proliferating cell nuclear antigen (PCNA). However, the presence of CsNPs and/or CurNPs along with HAP-NPs reduced the levels of oxidative stress through improving the activities of antioxidant enzymes. Also, the rats administered the nanoparticles showed a moderate improvement in glomerular damage which matched that of the control group and showed mild positive reactions to PCNA-ir in glomeruli and renal tubules in the cortical and medullary portions. These novel insights confirm that the presence of chitosan and curcumin in nanoforms has powerful biological effects with enhanced bioactivity and bioavailability phenomena compared to their microphase counterparts. Also, they were able to ameliorate the nephrotoxicity induced by HAP-NPs.

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1. Introduction

Hydroxyapatite (HAP) is an inorganic component of natural bone¹ and is extensively used in the tissue engineering (TE) field due to its excellent bioactivity, biocompatibility, non-toxicity, osteoconductivity, and non-inflammatory properties.² Hydroxyapatite has been used as a resin for plasmid DNA and

protein regulation, since ions on its surface are negatively and positively charged (PO_4^{3-} and Ca^{2+}), which can bind electrostatically with basic and acidic bio-macromolecules, respectively.³ Due to HAP similarity to the inorganic component of bone matrix, it has been involved in the development of new preparation methods in which it can be biomedically applied to repair hard tissues or as a drug delivery vehicle. Its use as a bone substitute or replacement comprises filling bones and teeth, partial or complete bone augmentation, and coating in dental implants and orthopedics.⁴

Even though hydroxyapatite on the nanoscale is documented to be biocompatible,⁵ the safety and toxicity concerns of HAP-NPs are growing regardless of their favorable prospect in various biomedical applications.⁶ Due to their nanoscale particle size, it is unknown whether they are safe when used in medical applications, as they can be internalized by the cells and interact with biological molecules, affecting cells in a harmful manner through altering the cell response and

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leading to toxicological response.⁷ Moreover, acute toxicity studies of HAP-NPs *in vivo*,⁸ interaction with different cell sources,⁹ and *in vitro* sensitization studies¹⁰ have been documented in several references.

Antioxidant activity is supposed to be one of the most known functions of chitosan as many studies have addressed chitosan's ability to prevent lipid oxidation and inhibit reactive oxygen species (ROS) in biological and food systems.¹¹ Chitosan can scavenge free radicals or chelate metal ions through donation of lone pairs of electrons or a hydrogen atom.¹² The amino groups (NH₂) and hydroxyl group (OH) in chitosan are the main functional groups for its antioxidant activity, and it is difficult to detach due to its strong hydrogen bonding and semi-crystalline structure.¹³ Chitosan nanoparticles are a natural polymer that is used in nanomedicine production, because they have attractive features for drug delivery and have proven to be very functional when used in the nanoscale form. Properties such as the cationic character and the solubility of chitosan in aqueous medium have been reported as an evidence of this polysaccharide success.¹⁴

Also, curcumin is a well-known antioxidant and a highly pleiotropic molecule¹² that is capable of exerting an extensive range of pharmacological activities like anti-cancer, antioxidant, antibacterial, anti-inflammatory, hypoglycaemic,¹⁵ anti-microbial, anti-atherosclerotic,¹⁶ and wound healing properties.¹⁷ Moreover, curcumin has been found to instantly interact with various intracellular signaling molecules.¹⁸ Its beneficial effects have been specified to be intermediated through the modification of multiple cell signaling molecules.¹⁹ Also, it has been reported to have the capability to directly scavenge reactive oxygen species (ROS).²⁰ Curcumin's poor bioavailability is a major disadvantage related to its use as a therapeutic agent. However, curcumin in the nanoform is considered as the improved form of curcumin that shows better bioavailability and solubility. Nanoparticles such as micelles, nanogels, liposomes, and polymeric nanoparticles can be used to deliver therapeutic concentrations of curcumin, which in turn reinforces the therapeutic efficacy of curcumin.²¹

The goals of this study were (i) to demonstrate hydroxyapatite nanoparticles' involvement in oxidative stress and DNA fragmentation; (ii) to investigate the histopathological and immunohistochemical alterations induced by HAP-NPs; and (iii) to explore whether the nanoforms of chitosan and curcumin are capable of ameliorating the nephrotoxicity induced by HAP-NPs.

2. Materials and methods

2.1. Tested compounds and doses

Hydroxyapatite nanoparticles (HAP-NPs) were prepared using the following reagents: sodium carbonate, Riedel-de-Haën, Germany, and sodium hydroxide, El-Nasr pharmaceutical chemicals Co., Egypt. Chitosan nanoparticles (CsNPs) and curcumin nanoparticles (CurNPs) were purchased from Nanotech Egypt for Photo Electronics. HAP-NPs were dissolved in dis-

tilled water, and the dose (300 mg per kg bw) was chosen according to Sabry.²² Also, CsNPs were dissolved in acetic acid (M_w 310–375 kDa), and the dose (280 mg per kg bw) was chosen according to Tang and Abdel-Wahhab.^{23–25} CurNPs were dissolved in distilled water, and the dose (15 mg per kg bw) was chosen according to Yadav.²⁶

2.2. Characterization of nanoparticles of hydroxyapatite, chitosan, and curcumin

All samples were examined for morphology and characterized using a high resolution transmission electron microscope (HR-TEM).

2.3.1. Transmission electron microscopy analysis. HAP-NPs, CsNPs, and CurNPs were analyzed using high resolution transmission electron microscopy (HR-TEM), which is an imaging mode of specialized transmission electron microscopes that allows for direct imaging of the atomic structure of the sample. The HR-TEM JEOL JEM 2100Plus is a versatile TEM for both large-scale 2D screening and tomography. For TEM analysis, the samples were placed on carbon-coated copper grids and left to dry for 5 min; the excess solution was removed using blotting paper at room temperature. The HR-TEM runs Serial EM and thus is capable of automated multi-position acquisitions. Depending on the type and thickness of the specimen, the acceleration voltage can be chosen from 80 to 200 kV.^{27,28}

2.3. Animals and experimental design

This study was performed in strict accordance with the guidelines of the Institutional Animal Care and Use Committee (IACUC) and approved by the Animal Care and Use Committee of Alexandria University (ethics approval no. 1468-104, revised 2018), provided that animals will not suffer at any stage of the experiment. The study was conducted on 80 male Wistar rats weighing 170 ± 175 g obtained from the Faculty of Medicine, Alexandria University, Egypt. The animals were housed in a comfortable environment on basal diet and tap water which were provided *ad libitum*. They were monitored during the period of treatment. Food and water intake, and body weights were weekly recorded throughout the whole experimental period. After acclimation for 2 weeks, they were randomly divided equally into 8 groups, and each cage housed a maximum of 10 rats. Group 1 was used as control; group 2 was orally treated with CsNPs alone (280 mg per kg bw); group 3 was treated orally with CurNPs alone (15 mg per kg bw); group 4 was orally treated with both CsNPs and CurNPs; group 5 was orally treated with HAP-NPs alone (300 mg per kg bw); group 6 was orally treated with both CsNPs and HAP-NPs; group 7 was orally treated with CurNPs and HAP-NPs; and group 8 was the combination group as it received CsNPs, CurNPs and HAP-NPs. The animals were orally treated with respective doses daily for 45 days.

2.4. Blood sample collection and kidney tissue preparation

At the end of the 45th day, all animals were anesthetized and then sacrificed, and blood samples were collected in heparin

collection tubes. For plasma separation, blood was centrifuged at 860g for 20 minutes, and plasma was kept at -80°C until analysis. Kidneys were removed and washed with saline solution (0.9%), and then connective tissues and adhering fats were immediately removed. Kidneys were minced and then homogenized in a Potter–Elvehjem type homogenizer. The homogenates were then centrifuged at 10 000g for 20 minutes at 4°C to pellet the cell debris, and the supernatant was collected and stored for the analysis of parameters under study.

2.5. Measured parameters

2.5.1. Oxidative stress markers and kidney antioxidant parameters. Thiobarbituric acid-reactive substances (TBARS) were measured by the method of Tappel and Zalkin.²⁹ Total antioxidant capacity (TAC) in kidney homogenates was assayed according to the Koracevic method.³⁰ Nitric oxide (NO) level was assayed according to Montgomery and Dymock.³¹ Superoxide dismutase (SOD) activity was measured using the method of Misra and Fridovich.³² Glutathione peroxidase (GPx) activity was determined following the Chiu method.³³ Glutathione S-transferase (GST) activity was analyzed according to the Habig method.³⁴ Catalase (CAT) activity was assayed following the Luck method.³⁵ Glutathione (GSH) levels were assayed using the Jollow method.³⁶ The abovementioned assays were conducted in compliance with the instruction manual of Biodiagnostic Kit, Egypt.

2.5.2. Assay of DNA fragmentation. 8-OH-2-deoxyguanosine (8-OHdG) was measured in the samples of DNA using the 8-OHdG commercial ELISA kit (ab201734, Abcam, Cambridge, UK) following the protocol of the manufacturer.

2.5.3. Histopathological examination. Tissues taken from the rats' kidneys were cut and immediately fixed in 10% formalin solution and dehydrated using ascending grades of alcohol and xylene. They were then placed in molten wax and xylene for about 10 minutes, embedded in paraffin wax, and then sectioned by a rotary microtome to obtain sections (thickness,

4–6 μm). After that, they were stained with H&E for investigating histopathological changes following the Drury method.³⁷

2.5.4. Proliferating cell nuclear antigen immunoreactivity (PCNA-ir) measurement. Kidney distribution of PCNA receptor subunits was determined in deparaffinized sections (thickness, 5 μm) using an avidin–biotin–peroxidase (IHC) method (Elite–ABC; Vector Laboratories, CA, USA), and the anti-PCNA monoclonal antibody (dilution 1 : 100; DAKO Japan Co, Tokyo, Japan) was used.

2.6. Statistical analysis

Data were reported as mean \pm SE. Statistical analysis of the parameters was performed using the general linear model (GLM) produced by Statistical Analysis Systems Institute.³⁸ Duncan's new multiple range test was followed to test the significant differences between means according to the Duncan method.³⁹

3. Results

3.1. Characterization of nanoparticles of hydroxyapatite, chitosan, and curcumin

3.1.1. Transmission electron microscopy analysis. The structure and morphology of the samples were further confirmed using high resolution transmission electron microscopy (HR-TEM) images of HAP-NPs, CsNPs, and CurNPs as shown in Fig. 1–3. The TEM analysis confirmed the presence of needle-like crystal morphology of HAP-NPs at different scale bars 50–100 nm (Fig. 1). TEM images of chitosan nanoparticles (CsNPs) are shown in Fig. 2. These nanoparticles have a solid and consistent structure and are also spherical in shape at different scale bars (Fig. 2(A and B)). TEM images of curcumin nanoparticles shown in Fig. 3 show the well-defined crystalline morphology of nano-curcumin. The average particle size of curcumin nanocrystals is 0.04 μm as shown in Fig. 3(A).

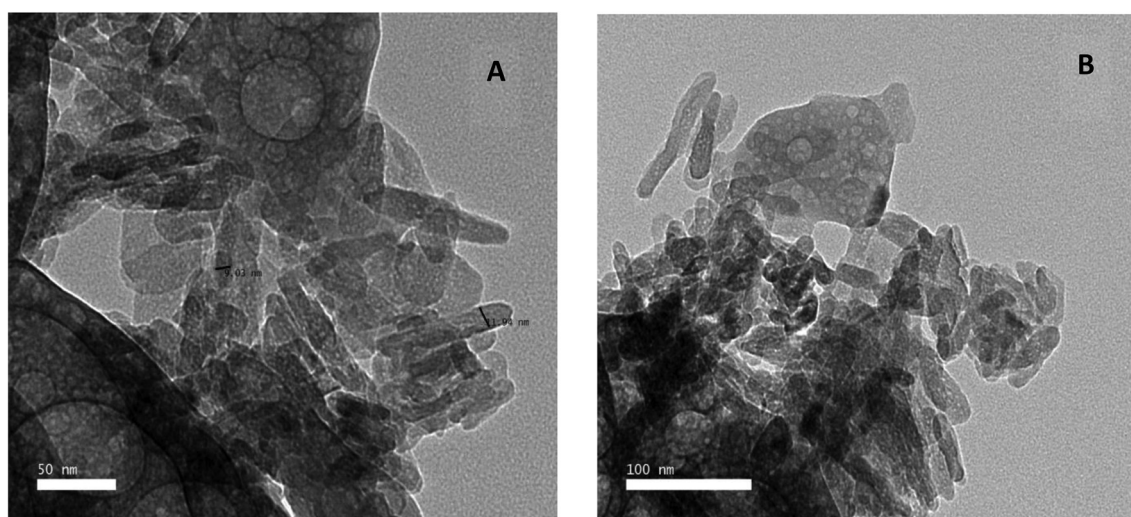


Fig. 1 TEM micrographs demonstrating the appearance of n-HAP needle-like crystals.

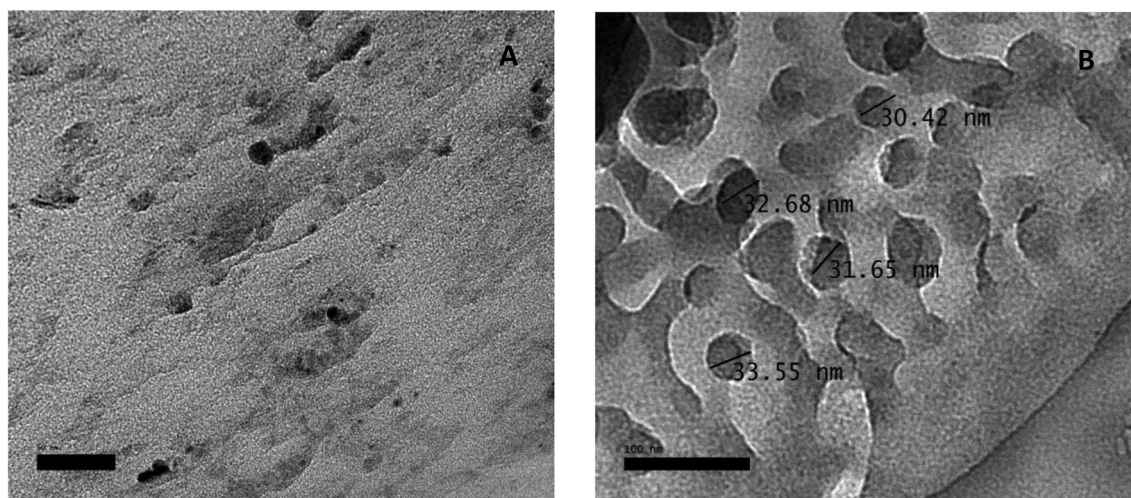


Fig. 2 TEM images and size distribution of chitosan nanoparticles.

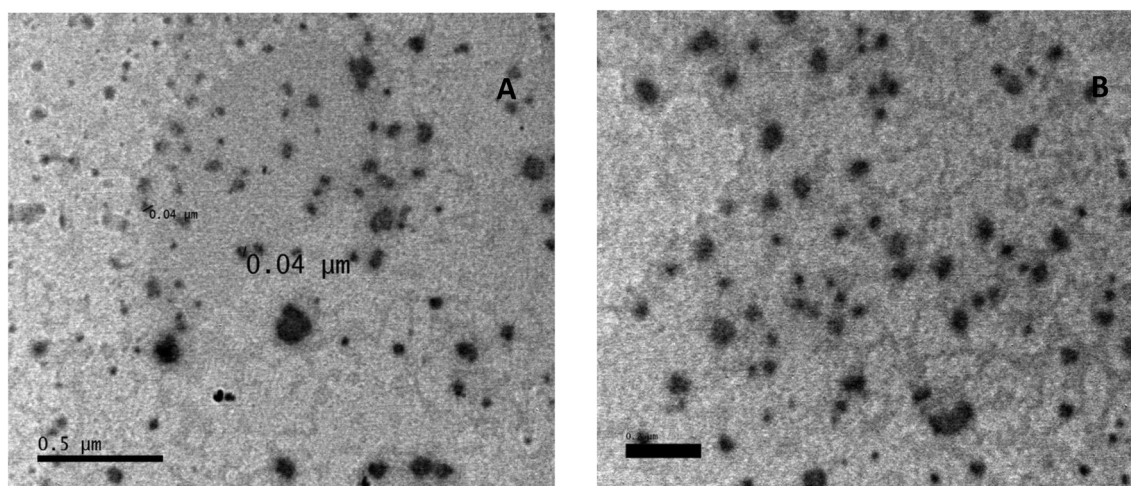


Fig. 3 TEM images and size distribution of curcumin nanoparticles.

3.2. Oxidative stress parameters and antioxidant enzymes

The activities of GPx, GST, CAT, SOD and TAC, and also the levels of TBARS, NO and GSH were measured in the kidneys of the adult male rats orally treated daily with CsNPs, CurNPs, and HAP-NPs alone or in combination, as shown in Tables 1 and 2 and Fig. 4 and 5. The data showed that there were no significant differences between the effect of CsNPs and CurNPs treated alone on GPx, GST, CAT, SOD, TAC, TBARS, NO, and GSH. However, the presence of CsNPs and CurNPs in combination significantly increased the activities of all these enzymes and decreased the levels of TBARS and NO in the kidney compared to the control group. Treatment with HAP-NPs alone showed a significant decrease in GPx, GST, CAT, SOD, GSH and TAC levels, but showed a significant increase in the levels of TBARS and NO compared to the control group. However, the presence of CsNPs and CurNPs alone or in combination with HAP-NPs was capable of increasing the activities of GPx, GST, CAT, SOD,

TAC and GSH and decreasing the activities of TBARS and NO compared to the HAP-NP treated group but did not reach the control group values.

3.3. Oxidative DNA damage

The results of 8-OH-2-deoxyguanosine (8-OHdG) content in the renal tissues of the studied groups are listed in Table 3 and shown in Fig. 6. The rats treated with CsNPs or CurNPs alone or in combination showed no significant effect on the level of 8-OHdG compared to the control group. On the other hand, the group treated with the hydroxyapatite nanoparticles alone showed a significant increase in the level of renal 8-OHdG to about 2.6-fold compared to the control group values, while treatment with CsNPs or CurNPs alone or in combination with HAP-NPs significantly ameliorated the renal content of 8-OHdG; however, the levels were still significantly higher than the control group values. It is apparent that CsNPs showed a better effect than CurNPs, while the treatment with both of

Table 1 Kidney levels of glutathione peroxidase, glutathione S-transferase, catalase, superoxide dismutase and total antioxidant capacity of male rats treated orally with nanoparticles of chitosan, curcumin, and hydroxyapatite alone or in combination

Experimental groups	Parameter				
	GPx (mU mg ⁻¹ protein)	GST (mU mg ⁻¹ protein)	CAT (mU mg ⁻¹ protein)	SOD (mU mg ⁻¹ protein)	TAC (mU mg ⁻¹ protein)
Control	43.7 ± 1.95 ^b	39.6 ± 2.59 ^a	38.0 ± 1.84 ^b	50.2 ± 2.86 ^{ab}	4.22 ± 0.20 ^c
CsNPs	47.5 ± 1.43 ^b	38.5 ± 1.69 ^a	38.0 ± 1.99 ^b	47.9 ± 2.12 ^{bc}	4.8 ± 0.14 ^{ab}
CurNPs	51.4 ± 1.03 ^a	38.2 ± 1.19 ^a	36.6 ± 1.47 ^b	47 ± 2.20 ^{bc}	4.3 ± 0.14 ^{bc}
(Cs + Cur) NPs	54.4 ± 2.07 ^a	41.0 ± 1.19 ^a	43.2 ± 1.45 ^a	52.4 ± 1.79 ^a	5.0 ± 0.28 ^a
HAP-NPs	19.9 ± 0.77 ^f	13.5 ± 1.02 ^d	12.6 ± 0.33 ^d	25.8 ± 0.80 ^e	2.1 ± 0.12 ^f
(HAP + Cs) NPs	26.2 ± 0.86 ^e	23.2 ± 0.54 ^c	25.9 ± 1.10 ^c	30 ± 1.99 ^{ed}	2.9 ± 0.19 ^{ed}
(HAP + Cur) NPs	30.3 ± 0.92 ^d	27.3 ± 0.48 ^b	27.8 ± 1.29 ^c	33.0 ± 1.08 ^d	2.6 ± 0.21 ^{ef}
(HAP + Cs + Cur) NPs	36.2 ± 0.70 ^c	29.6 ± 1.58 ^b	29.3 ± 0.66 ^c	42.8 ± 1.07 ^c	3.4 ± 0.19 ^d

Mean values within a column not sharing a common superscript letter (a, b, c) were significantly different, $p < 0.05$. GPx = glutathione peroxidase, GST = glutathione S-transferase, CAT = catalase, SOD = superoxide dismutase, TAC = total antioxidant capacity.

Table 2 Kidney levels of thiobarbituric acid-reactive substances, nitric oxide and glutathione of male rats treated orally with nanoparticles of chitosan, curcumin, and hydroxyapatite alone or in combination

Experimental groups	Parameter		
	TBARS (mg ml ⁻¹ protein)	NO (mU mg ⁻¹ protein)	GSH (mU mg ⁻¹ protein)
Control	7.0 ± 0.32 ^e	1.7 ± 0.04 ^{bc}	36.1 ± 1.08 ^a
CsNPs	6.9 ± 0.30 ^e	1.6 ± 0.07 ^{bc}	37.6 ± 2.16 ^a
CurNPs	6.5 ± 0.25 ^e	1.6 ± 0.08 ^{cd}	35.5 ± 0.95 ^a
(Cs + Cur) NPs	5.8 ± 0.30 ^e	1.3 ± 0.06 ^d	35.0 ± 1.71 ^a
HAP-NPs	24.7 ± 1.11 ^a	2.8 ± 0.18 ^a	10.2 ± 0.40 ^d
(HAP + Cs) NPs	17.0 ± 0.82 ^b	2.5 ± 0.20 ^{ab}	24.7 ± 0.67 ^c
(HAP + Cur) NPs	14.2 ± 0.30 ^c	2.5 ± 0.15 ^a	25.7 ± 1.12 ^c
(HAP + Cs + Cur) NPs	12.4 ± 0.41 ^d	2.0 ± 0.08 ^b	29.5 ± 0.77 ^b

Mean values within a column not sharing a common superscript letter (a, b, c) were significantly different, $p < 0.05$. TBARS = thiobarbituric acid-reactive substances, NO = nitric oxide, GSH = reduced glutathione concentration.

them resulted in a significant reduction in and complete normalization of renal 8-OHdG levels, as shown in Table 3 and Fig. 6.

3.4. Histopathological examination

As shown in Fig. 7A, the histological examination of the H&E stained kidney sections in the control rats revealed normal histological structures of the glomeruli and renal tubules. Cells of the Bowman's capsule's outer layer formed a simple squamous epithelium, while cells of the inner layer were very complex in shape and could not be detected by histological staining. The glomeruli were round to oval in shape. No histopathological abnormalities were found in the kidney sections of the rats treated with CsNPs as in G2, or with CurNPs as in G3, or with both CsNPs and CurNPs as in G4, as shown in Fig. 7B–D, respectively. The sections of the kidney treated with HAP-NPs revealed inconstant pathological changes in several parts of the urinary tubules and glomeruli, as shown in Fig. 7E. The most intense changes were in the Malpighian as corpuscles lost their configuration and

the renal tubules appeared with a wide lumen, marked medullary tubular epithelial degeneration, focal tubular epithelial necrosis, moderate hemorrhage, mild to moderate atrophic glomerulus with degenerated epithelium and marked congestion in the renal blood vessels, as shown in Fig. 7E. The kidney sections of the group treated with HAP-NPs and CsNPs together revealed a moderate degree of improvement in glomerular damage, which resembled that of the control group, with minimal vacuolization of tubular cells, as shown in Fig. 7F. The sections of the group treated with HAP-NPs and CurNPs together revealed a mild degree of improvement in glomerular damage with a wide lumen in the renal tubules, but some of the renal tubules were still degenerated (Fig. 7G). The group treated with HAP-NPs along with CsNPs and CurNPs revealed a very good improvement with normal structure as in the control group (Fig. 7H).

3.5. Proliferating cell nuclear antigen immunoreactivity

PCNA immunoreactivity (PCNA-ir) detection of kidney sections in different experimental groups was performed as shown in Figures 8A–H. The kidney sections of the control rats showed faint or negative reaction for PCNA-ir (grade 0) in glomeruli and renal tubules, as shown in Fig. 8A. The kidney sections of rats treated with CsNPs (G2), with CurNPs (G3), and with both CsNPs and CurNPs (G4) showed mild reaction for PCNA-ir (grade 1), as shown in Fig. 8B–D. Strong positive reactions for PCNA-ir (grade 5) were detected in the kidney sections of the rats treated with HAP-NPs alone, as shown in Fig. 8E. The intensity of PCNA-ir in the HAP-NP treated group showed a significant increase when compared to the control group. However, moderate positive reactions for PCNA-ir (grade 3) were observed in the kidney sections of the rats treated with HAP-NPs and CsNPs together, while moderate to marked positive reactions for PCNA-ir (grade 4) were observed in the kidney sections of the rats treated with HAP-NPs and CurNPs together, as shown in Fig. 8F–G. The kidney sections of the group treated with HAP-NPs along with CsNPs and CurNPs revealed mild positive reactions for PCNA-ir (grade 1), as shown in Fig. 8H.

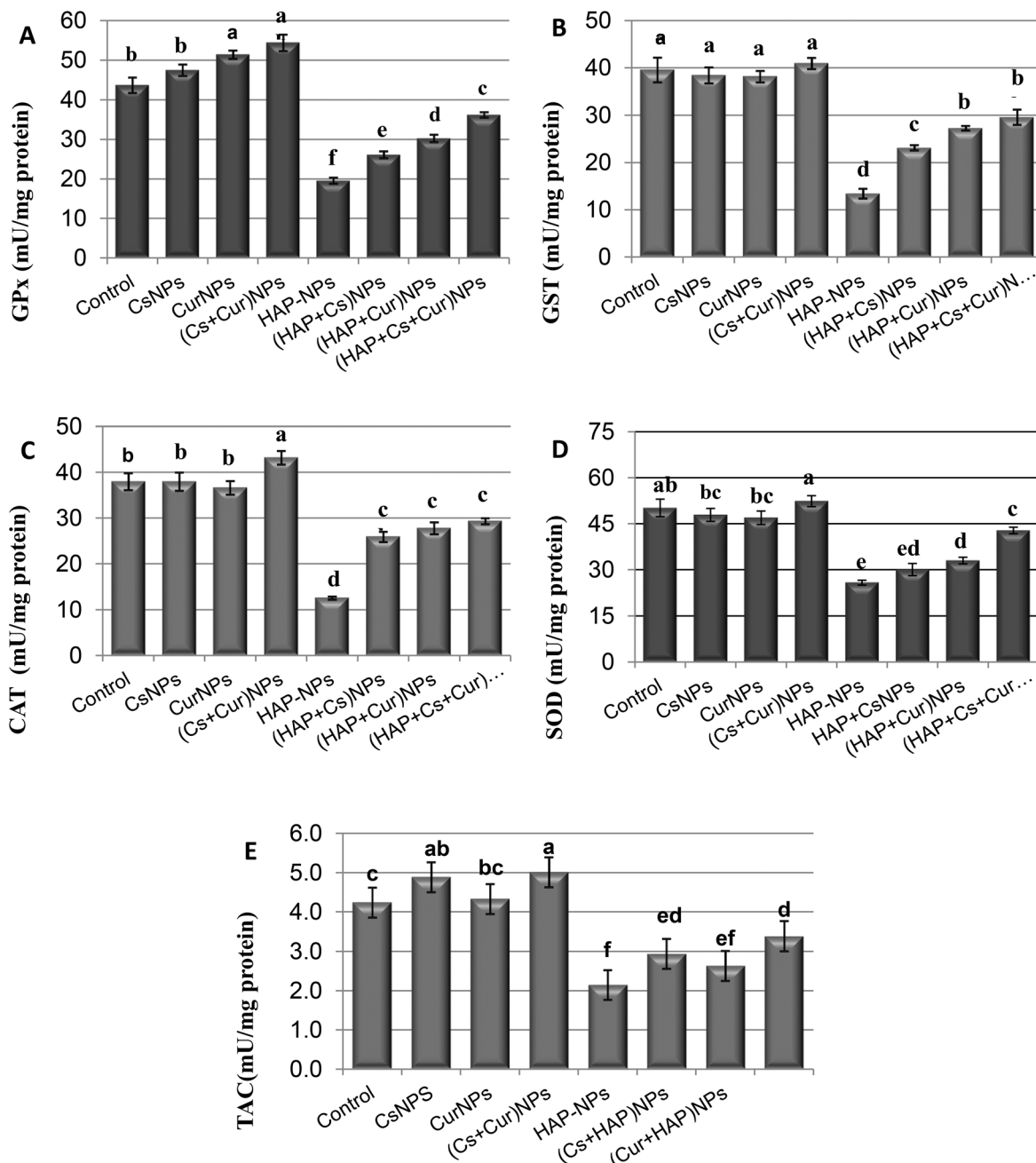


Fig. 4 Levels of GPX, GST, CAT, SOD and TAC in rat kidneys treated orally with CsNPs, CurNPs and HAP-NPs alone or in combination.

4. Discussion

Nano-sized particles have the ability to internalize cells, where they can interact with biological molecules, alter cell response, and affect the cell in deleterious behavior, leading to toxicological response.⁷ Comprehensive and adequate estimate of toxicological effects induced by exposure to nanoparticles should take into consideration the different points through which these particles enter the body and translocate into organs, tissues, and biological systems after being distributed in the body. Despite the promising potential of hydroxyapatite

nanoparticles in various applications, their toxicity is of growing concern. So, in order to indicate the suitable protective intervention against these NPs, it is important to understand its toxicity mechanism.

Kidney is considered as one of the main sites inducing xenobiotic toxicity.⁴⁰ Its specific toxicity is assigned to the high flow rate of blood, which in turn delivers elevated concentrations of xenobiotic to the kidney. In particular, the proximal tubule epithelium is more vulnerable to nephrotoxicity; these cells express various transporters, which enable active intake and intracellular accumulation of metabolites or toxic com-

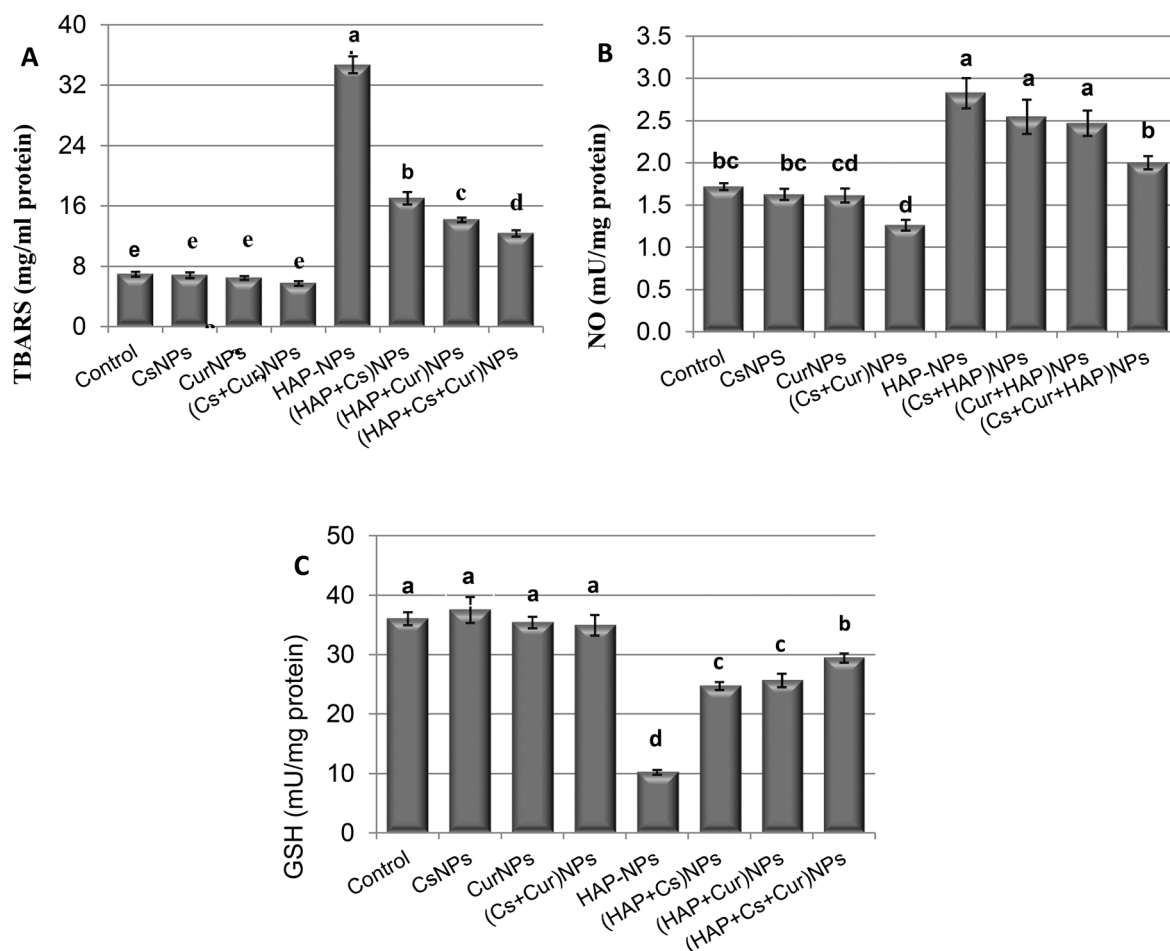


Fig. 5 Levels of TBARS, NO and GSH of male rats treated orally with CsNPs, CurNPs, and HAP-NPs alone or in combination.

Table 3 Renal tissue content of 8-OH-2-deoxyguanosine of male rats treated orally with nanoparticles of chitosan, curcumin, and hydroxyapatite alone or in combination

Experimental groups	8-OHdG ($\text{pg } \mu\text{g}^{-1} \text{ DNA}$)
Control	4.97 ± 1.5^c
CsNPs	4.30 ± 0.9^c
CurNPs	4.70 ± 1.3^c
(Cs + Cur) NPs	4.30 ± 0.79^c
HAP-NPs	13.13 ± 1.21^a
(HAP + Cs) NPs	5.63 ± 0.99^c
(HAP + Cur) NPs	8.47 ± 0.70^b
(HAP + Cs + Cur) NPs	5.40 ± 0.92^c

Mean values within a column not sharing a common superscript letter (a, b, c) were significantly different, $p < 0.05$. 8-OHdG = 8-OH-2-deoxyguanosine.

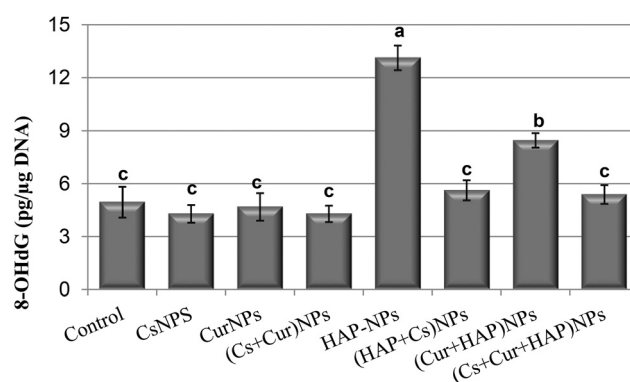


Fig. 6 Renal tissue content of 8-OH-2-deoxyguanosine of male rats treated orally with CsNPs, CurNPs, and HAP-NPs alone or in combination.

pounds. Proximal tubular epithelial cells can bio-activate non-toxic compounds into reactive intermediates which might cause toxicity through damaging cellular macromolecules, and also they are highly metabolically active.⁴⁰

Nemenqani indicated that oxidative stress induction is the main mechanism of nanoparticles' toxicity. When nanoparticles enter the cell, they disturb the redox balance between

oxidants and antioxidants, which in turn induce the intracellular oxidative stress. Excessive oxidative stress may also modify lipids, proteins, and nucleic acids, which in turn deteriorates the antioxidant defense systems or leads to cell death.⁴¹ Meantime, nanoparticles can increase gene expression of the death receptor and cause DNA damage through increased pro-

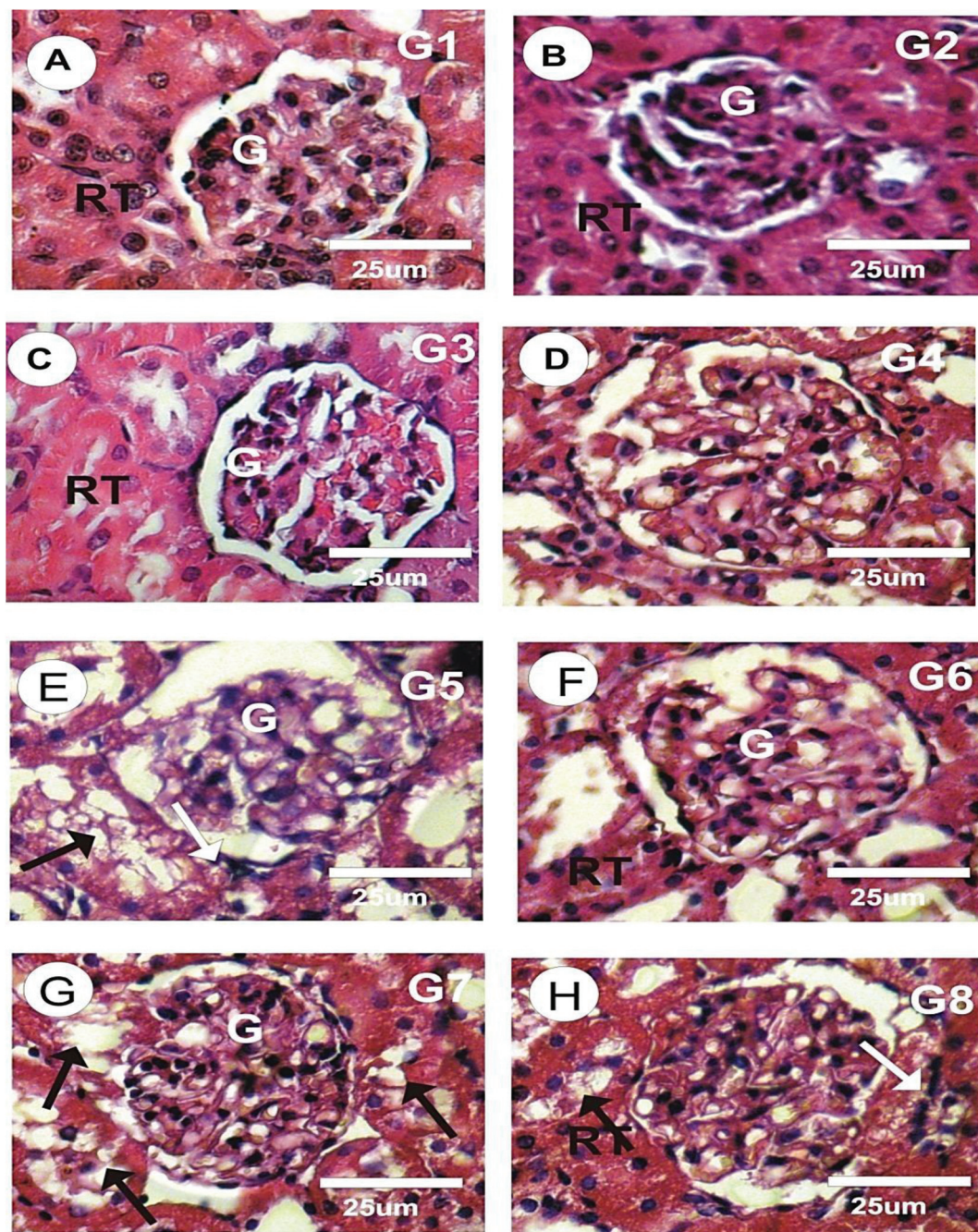


Fig. 7 Photomicrographs of kidney sections of different experimental groups stained by H&E.

duction of ROS.⁴² Also, it was documented that the increased production of ROS induced by nanoparticles in lysosomes can cause DNA mutations or induce DNA single- or double-strand breaks.⁴³

At the level of redox status of kidney tissues, HAP-NPs caused significant reduction in different antioxidant enzymes, including SOD, GPx, CAT, GST, and GSH systems. This deterioration in the kidney antioxidant defense system is linked to the enhanced production of nitric oxide and ROS, which exacerbates oxidative stress in renal tissues due to exposure to HAP-NPs. The depletion of GSH could explain the inhibited

activities of GSH-dependent enzymes (GST and GPx), while the main route to the depletion of GSH levels is the protection against oxidative stress that was confirmed by the elevated level of 8-OHdG due to ROS overproduction induced by HAP-NPs. This ROS overload was faced by the low molecular weight antioxidant, *e.g.* glutathione, that results in GSH depletion.

The present results are in agreement with those of Chen, who demonstrated that the exposure to HAP-NPs caused oxidative stress with the elevation of the levels of H₂O₂ and MDA and a decrease in SOD and GSH levels.⁴⁴ Also, Xu

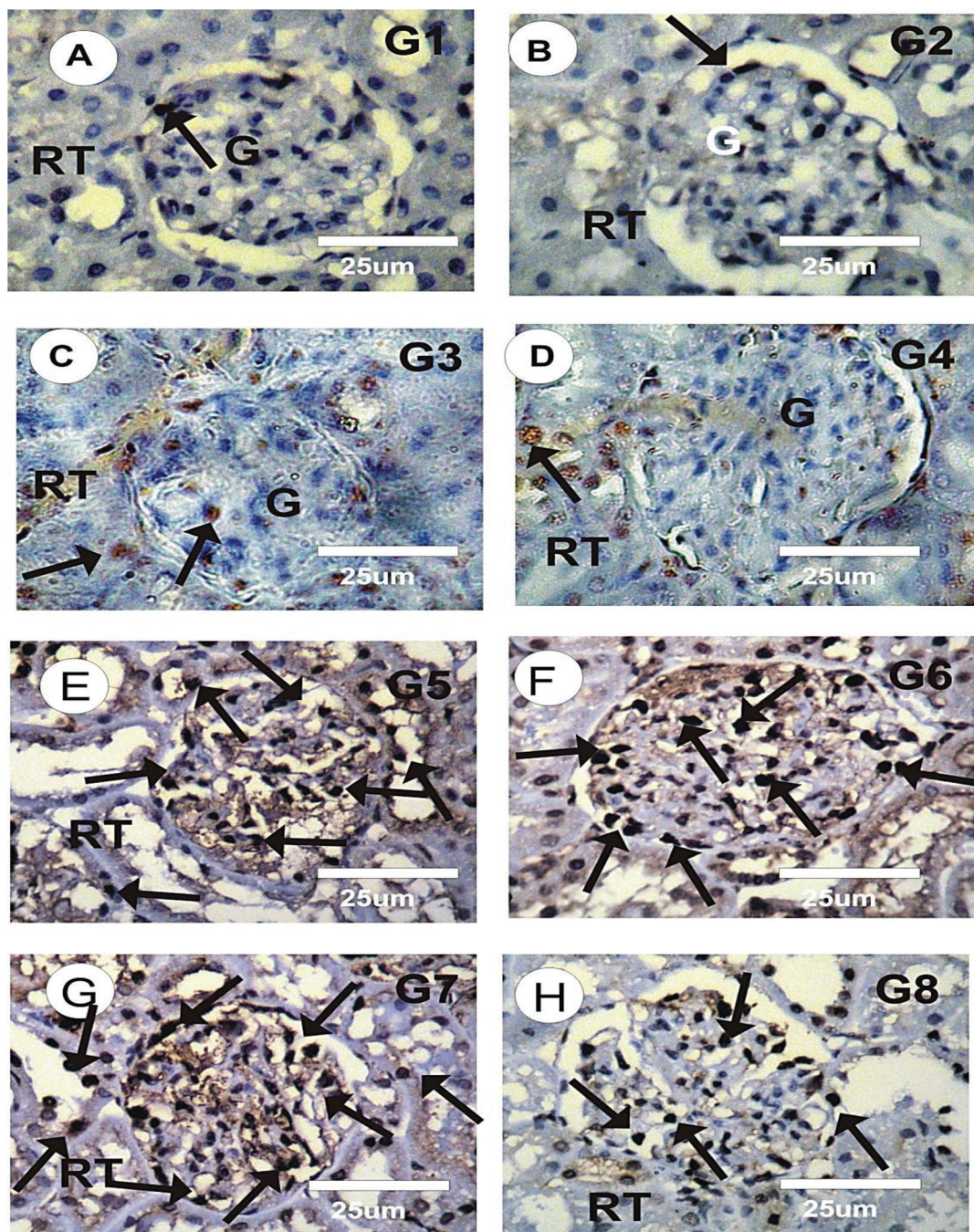


Fig. 8 Photomicrographs of kidney sections of different experimental groups stained with PCNA-ir.

found that HAP-NPs induced ROS and SOD intracellular accumulation in cell lines.⁴⁵ Turkez demonstrated that exposure to HAP-NPs had dose-dependent effects on inducing genotoxicity, cytotoxicity, and oxidative damage in human blood cells.⁴⁶ Also, HAP-NP exposure resulted in significant increase of micronuclei, sister chromatid exchange, 8-OH-dG levels, and chromosome aberration rates as compared to untreated culture.

Moreover, Sies and Akerboom (1984) reported that after administration of nanoparticles, GSH can act as a conjugating

agent in their metabolism.⁴⁷ When these nanoparticles induce oxidative stress by generating hydroperoxides or H_2O_2 , GSH can also be oxidized in a reaction catalyzed by GPx. It was also reported that GSH depletion in tissues leads to deterioration of the cellular defense against ROS and may result in peroxidative injury.⁴⁷ As GSH plays an important role in cell's protection against induced oxidative stress, it also acts as a vital antioxidant and a major cofactor for antioxidant enzymes participating in cellular redox reactions. Likewise, HAP-NPs decreased the total SOD activity in C6 glioma cells.⁴⁸ Also, an

in vivo study documented a dose-dependent decrease in the activity of GPx in dermally HAP-NP exposed rats.⁴⁹

However, at the molecular level, oxidative stress induced by HAP-NPs in kidney tissues caused significant elevation in the oxidative DNA marker, 8-OHdG. Reactive oxygen species are highly reactive with a set of biological molecules, among which DNA is the most important one.⁵⁰ 8-Oxoguanine (8-oxo-G) is one of the distinguished forms of oxidatively generated DNA base modifications and also a sensitive oxidative DNA damage marker.⁵¹ If this base modification (8-OH-dG) was not repaired by the repair mechanism, guanine (G) to thymine (T) substitution may happen upon replication. Alternatively, 8-Oxo-2'-deoxyguanosine-5'-Triphosphate (8-Oxo-dGTP) (in the nucleotide pool) may be mis-integrated opposite adenine (A) producing A-C substitution.⁵² These mutagenic effects of 8-OH-dG formation may denote genotoxicity hazards induced by HAP-NPs.

The antioxidant effects of CsNPs were confirmed in many previous studies; for example, Wen reported that CsNPs' protective role against H₂O₂-induced cell injury was through boosting the endogenous antioxidants' activities (SOD, GPx and CAT) and enhancing their gene expressions.⁵³ Furthermore, El-Denshary documented that treatment with CsNPs succeeded in ameliorating the antioxidant capacity of the body and reducing the oxidative stress as indicated by the increase in antioxidant enzymes (CAT, SOD, GPx) and decrease of MDA levels.⁵⁴ Also, the present results compare with those of Jeon, who reported that the treatment with chitosan significantly decreased MDA levels and increased activities of the antioxidant enzymes, CAT and SOD, against carbon tetrachloride chronic (CCl₄)-induced toxicity.⁵⁵ Also, Xie showed that CsNPs inhibited lipid peroxidation by scavenging the hydroxyl radicals.¹³ It was found that chitosan administration with a dose of 100 mg per kg bw showed a significant increase in antioxidant enzymes (SOD, GPx, GR, CAT) and non-enzymatic antioxidants (GSH, vitamin E and vitamin C) but showed a significant decrease in lipid peroxidation levels in the benzidine induced bladder cancer cell line.^{56,57} Also, Prashanth reported that chitosan was able to scavenge O₂^{•-} and •OH radicals and offered protection against calf thymus DNA damage.⁵⁸

Flora found that CurNPs at the dose of 15 mg kg⁻¹ reduced free radical generation and also restored the antioxidant enzymes' activity in rats treated with lead.⁵⁹ The antioxidant activity of CurNPs along with their enhanced bioavailability makes them more vulnerable to be an effective agent against lead poisoning. Nehra found that treatment with CurNPs significantly decreased redundant leakage of ROS and lipid peroxidation in cardiomyocytes under hypoxia.⁶⁰ Yadav found that CurNPs ameliorated toxic effects induced by fluoride and arsenic in rat tissues and blood through reducing the levels of ROS and restoring the blood glutathione level.⁶¹

These results clearly demonstrated the boosting effects of CsNPs and CurNPs on the antioxidant status of renal tissues through the activation of antioxidant enzymes (SOD, GPx, GST and CAT), increasing total antioxidant capacity (TAC) and correction of glutathione. Also, CsNPs and CurNPs have significantly decreased the production of nitric oxide as indicated by

the lower nitric oxide level. These ameliorative effects on anti-oxidants and free radicals result in correction of the oxidative stress status of renal tissues as evidenced in the present study by the lower levels of MDA; the markers of lipid peroxidation and 8-OH-dG; and the marker of oxidative DNA damage. The obtained results agree with those of Ahmed-Farid, who reported that CurNPs increased the tissue energy rate in addition to a significant decrease in 8-OHdG levels and decreased nitrosative and oxidative stresses.⁶² CurNPs are also considered as a powerful free radical oxidant scavenger *via* electron transfer and hydrogen atom donation.⁶³

It is obvious from the previous results that CsNP and/or CurNP administration along with HAP-NPs has significantly corrected the impairments of most studied parameters related to renal toxicity that has been induced by the treatment with HAP-NPs alone. However, at the histological level, HAP-NPs caused severe changes in the Malpighian corpuscles, which lost their characteristic configuration, and the renal tubules appeared with a wide lumen, marked cortical and medullary tubular epithelial degeneration, focal tubular epithelial necrosis, moderate hemorrhage, mild to moderate atrophic glomerulus and degenerated epithelium, and marked congestion in the renal blood vessels. Also, strong positive reactions for PCNA-ir (grade 5) were detected in kidney sections of rats treated with HAP-NPs. These data may indicate the entry of HAP-NPs into the renal tissues, cells and cellular organelles. However, the administration of CsNPs and/or CurNPs showed a moderate improvement in glomerular damage, which matched that of the control group with only minimal vacuolization in tubular cells. Kidney sections of the rats administered HAP-NPs along with either CsNPs or CurNPs or both revealed very good improvements with normal structure as in the control group.

5. Conclusion

Collectively, the present results concluded that oral intake of HAP-NPs induced renal toxicity at different levels, including kidney antioxidant system, oxidative stress, DNA damage and renal tissue content of 8-OHdG, histological structure and immunohistochemical reactivity towards PCNA. Given the substantial use of HAP-NPs in the field of medicine, the search for protective interceptions against its toxicity is of great medical prominence. The presented results obviously indicated that CsNPs and CurNPs have alleviated the nephrotoxicity induced by HAP-NPs with prominent ameliorative effects observed with CsNPs either administered alone or in combination with CurNPs. Our results also confirm that the nanoforms of these two compounds have powerful biological effects with enhanced bioactivity and bioavailability phenomena compared to their microphase counterparts.

Conflicts of interest

The authors gratefully acknowledge that there are no conflicts of interests.

References

- 1 K. Azzaoui, A. Lamhamdi, E. M. Mejdoubi, M. Berrabah, B. Hammouti, A. Elidrissi, M. M. G. Fouda and S. S. Al-Deyab, Synthesis and characterization of composite based on cellulose acetate and hydroxyapatite application to the absorption of harmful substances, *Carbohydr. Polym.*, 2014, **111**, 41–46.
- 2 R. Murugan and S. Ramakrishna, Bioresorbable composite bone paste using polysaccharide based nano hydroxyapatite, *Biomaterials*, 2004, **25**, 3829–3835.
- 3 E. Schröder, J. Jönsson and T. L. Poole, Hydroxyapatite chromatography: altering the phosphate-dependent elution profile of protein as a function of pH, *Anal. Biochem.*, 2003, **313**, 176–178.
- 4 H. Zhou and J. Lee, Nanoscale hydroxyapatite particles for bone tissue engineering, *Acta Biomater.*, 2011, **7**, 2769–2781.
- 5 N. S. Remya, S. Syama, V. Gayathri, H. K. Varma and P. V. Mohanan, An in vitro study on the interaction of hydroxyapatite nanoparticles and bone marrow mesenchymal stem cells for assessing the toxicological behavior, *Colloids Surf., B*, 2011, **117**, 389–397.
- 6 X. Xiao, W. Wang, D. Liu, H. Zhang, P. Gao, L. Geng and Z. Wang, The promotion of angiogenesis induced by three-dimensional porous beta-tricalcium phosphate scaffold with different interconnection sizes via activation of PI3K/Akt pathways, *Sci. Rep.*, 2015, **5**, 9409.
- 7 M. A. Shahbazi, B. Herranz and H. A. Santos, Nanostructured porous Si-based nanoparticles for targeted drug delivery, *Biomater.*, 2012, **2**, 296–312.
- 8 L. Liu, Z. Xiao, Y. Xiao, Z. Wang, F. Li, M. Li and X. Peng, Potential enhancement of intravenous nano-hydroxyapatite in high-intensity focused ultrasound ablation for treating hepatocellular carcinoma in a rabbit model, *Oncol. Lett.*, 2014, **7**, 1485–1492.
- 9 S. C. Rodrigues, C. L. Salgado, A. Sahu, M. P. Garcia, M. H. Fernandes and F. J. Monteiro, Preparation and characterization of collagen-nanohydroxyapatite biocomposite scaffolds by cryogelation method for bone tissue engineering applications, *J. Biomed. Mater. Res., Part A*, 2013, **101**, 1080–1094.
- 10 Y. Xiong, C. Ren, B. Zhang, H. Yang, Y. Lang, L. Min and A. Mo, Analyzing the behavior of a porous nano-hydroxyapatite/polyamide 66 (n-HA/PA66) composite for healing of bone defects, *Int. J. Nanomed.*, 2014, **9**, 485.
- 11 K. W. Kim and R. L. Thomas, Antioxidative activity of chitosans with varying molecular weights, *Food Chem.*, 2007, **101**, 308–313.
- 12 S. B. Lin, S. H. Chen and K. C. Peng, Preparation of antibacterial chito-oligosaccharide by altering the degree of deacetylation of β -chitosan in a *Trichoderma harzianum* chitinase-hydrolysing process, *J. Sci. Food Agric.*, 2009, **89**, 238–244.
- 13 W. Xie, P. Xu and Q. Liu, Antioxidant activity of water-soluble chitosan derivatives, *Bioorg. Med. Chem. Lett.*, 2001, **11**, 1699–1701.
- 14 A. Grenha, S. Al-Qadi, B. Seijo and C. Remuñán-Lopez, The potential of chitosan for pulmonary drug delivery, *J. Drug Delivery Sci. Technol.*, 2010, **20**, 33–43.
- 15 S. Sharma, S. K. Kulkarni and K. Chopra, Curcumin, the active principle of turmeric (*Curcuma longa*), ameliorates diabetic nephropathy in rats, *Clin. Exp. Pharmacol. Physiol.*, 2006, **33**, 940–945.
- 16 T. Nishiyama, T. Mae, H. Kishida, M. Tsukagawa, Y. Mimaki, M. Kuroda and M. Kitahara, Curcuminoids and sesquiterpenoids in turmeric (*Curcuma longa* L.) suppress an increase in blood glucose level in type 2 diabetic KK-Ay mice, *J. Agric. Food Chem.*, 2005, **53**, 959–963.
- 17 B. B. Aggarwal and B. Sung, Pharmacological basis for the role of curcumin in chronic diseases: an age-old spice with modern targets, *Trends Pharmacol. Sci.*, 2009, **30**, 85–94.
- 18 S. C. Gupta, S. Prasad, J. H. Kim, S. Patchva, L. J. Webb, I. K. Priyadarsini and B. B. Aggarwal, Multitargeting by curcumin as revealed by molecular interaction studies, *Nat. Prod. Rep.*, 2011, **28**, 1937–1955.
- 19 N. Dhillon, B. B. Aggarwal, R. A. Newman, R. A. Wolff, A. B. Kunnumakkara, J. L. Abbruzzese and R. Kurzrock, Phase II trial of curcumin in patients with advanced pancreatic cancer, *Clin. Cancer Res.*, 2008, **14**, 4491–4499.
- 20 S. C. Gupta, S. Patchva and B. B. Aggarwal, Therapeutic roles of curcumin: lessons learned from clinical trials, *AAPS J.*, 2013, **15**, 195–218.
- 21 S. Shome, A. D. Talukdar, M. D. Choudhury, M. K. Bhattacharya and H. Upadhyaya, Curcumin as potential therapeutic natural product: a nanobiotechnological perspective, *J. Pharm. Pharmacol.*, 2016, **68**, 1481–1500.
- 22 R. Sabry, *Preparation of hydroxyapatite nanoparticles by using emulsion liquid membrane [PhD Thesis]*, Al-Azhar University, Egypt, 2012.
- 23 Z. X. Tang, J. Q. Qian and L. E. Shi, Preparation of chitosan nanoparticles as carrier for immobilized enzyme, *Appl. Biochem. Biotechnol.*, 2007a, **136**, 77–96.
- 24 Z. X. Tang, L. E. Shi and J. Q. Qian, Neutral lipase from aqueous solutions on Chitosan nano-particles, *Biochem. Eng. J.*, 2007b, **34**, 217–223.
- 25 M. A. Abdel-Wahhab, A. Aljawish, A. A. El-Nekeety, S. H. Abdel-Aiezm, H. A. Abdel-Kader, B. H. Rihn and O. Joubert, Chitosan nanoparticles and quercetin modulate gene expression and prevent the genotoxicity of aflatoxin B1 in rat liver, *Toxicol. Rep.*, 2015, **2**, 737–747.
- 26 A. Yadav, V. Lomash, M. Samim and S. J. Flora, Curcumin encapsulated in chitosan nanoparticles: a novel strategy for the treatment of arsenic toxicity, *Chem.-Biol. Interact.*, 2012, **199**, 49–61.
- 27 J. C. H. Spence, *High-Resolution Electron Microscopy*, Oxford Univ. Press, Oxford, UK, 2003.
- 28 J. C. H. Spence, H. R. Kolar, G. Hembree, C. J. Humphreys, J. Barnard, R. Datta and J. F. Justo, Imaging dislocation cores—the way forward, *Philos. Mag.*, 2006, **86**, 4781–4796.
- 29 A. L. Tappel and H. Zalkin, Inhibition of lipide peroxidation in mitochondria by vitamin E, *Arch. Biochem. Biophys.*, 1959, **80**, 333–336.

- 30 D. Koracevic, G. Koracevic, V. Djordjevic, S. Andrejevic and V. Cosic, Method for the measurement of antioxidant activity in human fluids, *J. Clin. Pathol.*, 2001, **54**, 356–361.
- 31 H. A. C. Montgomery and J. F. Dymock, The determination of nitrate in water, *Analyst*, 1961, **86**, 414–416.
- 32 H. P. Misra and I. Fridovich, The role of superoxide anion in the autoxidation of epinephrine and a simple assay for superoxide dismutase, *J. Biol. Chem.*, 1972, **247**, 3170–3175.
- 33 D. T. Chiu, F. H. Stults and A. L. Tappel, Purification and properties of rat lung soluble glutathione peroxidase, *Biochim. Biophys. Acta, Protein Struct. Mol. Enzymol.*, 1976, **445**, 558–566.
- 34 W. H. Habig, M. J. Pabst and W. B. Jakoby, Glutathione S-transferases the first enzymatic step in mercapturic acid formation, *J. Biol. Chem.*, 1974, **249**, 7130–7139.
- 35 H. Luck, Catalase, in *Method of Enzymatic Analysis*, ed. M. V. Bergmayer, Verlag. Chemic., Academic Press, New, 1974, p. 885.
- 36 D. J. Jollow, J. R. Mitchell, N. A. Zampaglione and J. R. Gillette, Bromobenzene-induced liver necrosis. Protective role of glutathione and evidence for 3, 4-bromobenzene oxide as the hepatotoxic metabolite, *Pharmacology*, 1974, **11**, 151–169.
- 37 R. A. Drury, E. A. Wallington and S. Carleton, in *Histological Techniques*, Oxford University Press, London, New York, Toronto, 5th edn, 1980, pp. 241–242.
- 38 SAS, *Statistical Analysis System. SAS Procedure Guide. Release 6.03 Edition*, SAS Institute Inc., Cary, Nc, U.S.A., 1998.
- 39 D. B. Duncan, Multiple Range and Multiple (F-test), *Biometrics*, 1955, **11**, 1–42.
- 40 S. E. Hart and L. B. Kinter, Assessing renal effects of toxicants in vivo, in *Toxicology of the Kidney*, ed. J. B. Tarloff and L. H. Lash, 2005, pp. 81–147.
- 41 D. Nemenqani, O. El-Gharib, A. M. Ahmed and A. R. Baiuomy, The Protective Effects of Antioxidant (Vitamin C) Against Hepatic Oxidative Damage Induced by Zinc Oxide Nanoparticles, *Int. Res. J. Appl. Basic Sci.*, 2015, **9**, 502–509.
- 42 Y. Wang, W. G. Aker, H. M. Hwang, C. G. Yedjou, H. Yu and P. B. Tchounwou, A study of the mechanism of in vitro cytotoxicity of metal oxide nanoparticles using catfish primary hepatocytes and human HepG2 cells, *Sci. Total Environ.*, 2011, **409**, 4753–4762.
- 43 H. Yang, C. Liu, D. Yang, H. Zhang and Z. Xi, Comparative study of cytotoxicity, oxidative stress and genotoxicity induced by four typical nanomaterials: the role of particle size, shape and composition, *J. Appl. Toxicol.*, 2009, **29**, 69–78.
- 44 Q. Chen, Y. Xue and J. Sun, Hepatotoxicity and liver injury induced by hydroxyapatite nanoparticles, *J. Appl. Toxicol.*, 2014, **34**, 1256–1264.
- 45 Z. Xu, C. Liu, J. Wei and J. Sun, Effects of four types of hydroxyapatite nanoparticles with different nanocrystal morphologies and sizes on apoptosis in rat osteoblasts, *J. Appl. Toxicol.*, 2012b, **32**, 429–435.
- 46 H. Turkez, M. I. Yousef, E. Sönmez, B. Togar, F. Bakan, P. Sozio and A. D. Stefano, Evaluation of cytotoxic, oxidative stress and genotoxic responses of hydroxyapatite nanoparticles on human blood cells, *J. Appl. Toxicol.*, 2014, **34**, 373–379.
- 47 H. Sies and T. P. Akerboom, Glutathione disulfide (GSSG) efflux from cells and tissues, *Methods Enzymol.*, 1984, **105**, 445–451.
- 48 J. Xu, P. Xu, Z. Li, J. Huang and Z. Yang, Oxidative stress and apoptosis induced by hydroxyapatite nanoparticles in C6 cells, *J. Biomed. Mater. Res., Part A*, 2012a, **100**, 738–745.
- 49 M. Parayanthala-Valappil, S. Santhakumar and S. Arumugam, Determination of oxidative stress related toxicity on repeated dermal exposure of hydroxyapatite nanoparticles in rats, *Int. J. Biomater.*, 2014, 2014.
- 50 A. Mangerich, C. G. Knutson, N. M. Parry, S. Muthupalani, W. Ye, E. Prestwich, L. Cui, J. L. McFaline, M. Mobley, Z. Ge and K. Taghizadeh, Infection-induced colitis in mice causes dynamic and tissue-specific changes in stress response and DNA damage leading to colon cancer, *Proc. Natl. Acad. Sci. U. S. A.*, 2012, **109**, 1820–1829.
- 51 M. S. Cooke, M. D. Evans, M. Dizdaroglu and J. Lunec, Oxidative DNA damage: mechanisms, mutation, and disease, *FASEB J.*, 2003, **17**, 195–1214.
- 52 K. C. Cheng, D. S. Cahill, H. Kasai, S. Nishimura and L. A. Loeb, 8-Hydroxyguanine, an abundant form of oxidative DNA damage, causes G→T and A→C substitutions, *J. Biol. Chem.*, 1992, **267**, 166–172.
- 53 Z. S. Wen, L. J. Liu, Y. L. Qu, X. K. OuYang, L. Y. Yang and Z. R. Xu, Chitosan nanoparticles attenuate hydrogen peroxide-induced stress injury in mouse macrophage RAW264.7 cells, *Mar. Drugs*, 2013, **11**, 3582–3600.
- 54 E. S. El-Denshary, A. Aljawish, A. A. El-Nekeety, N. S. Hassan, R. H. Saleh, B. H. Rihn and M. A. Abdel-Wahhab, Possible synergistic effect and antioxidant properties of chitosan nanoparticles and quercetin against carbon tetrachloride-induced hepatotoxicity in rats, *Soft Nanosci. Lett.*, 2015, **5**, 36–50.
- 55 T. L. Jeon, S. G. Hwang, N. G. Park, Y. R. Jung, S. I. Shin, S. D. Choi and D. K. Park, Antioxidative Effect of Chitosan on Chronic Carbon Tetrachloride Induced Hepatic Injury in Rats, *Toxicology*, 2003, **187**, 67–73.
- 56 N. E. Mohamed, Effect of chitosan on oxidative stress and metabolic disorders induced in rats exposed to radiation, *Am. J. Sci.*, 2011, **7**, 406–417.
- 57 S. Kuppusamy and J. Karuppaiah, Antioxidant and cytotoxic efficacy of chitosan on bladder cancer, *Asian Pac. J. Trop. Dis.*, 2012, **2**, 769–773.
- 58 K. V. H. Prashanth, S. M. Dharmesh, K. S. J. Rao and R. N. Tharanathan, Free radical-induced chitosan depolymerized products protect calf thymus DNA from oxidative damage, *Carbohydr. Res.*, 2007, **342**, 190–195.
- 59 G. Flora, D. Gupta and A. Tiwari, Preventive efficacy of bulk and nanocurcumin against lead-induced oxidative stress in mice, *Biol. Trace Elem. Res.*, 2013, **152**, 31–40.
- 60 S. Nehra, V. Bhardwaj, L. Ganju and D. Saraswat, Nanocurcumin prevents hypoxia induced stress in primary

- human ventricular cardiomyocytes by maintaining mitochondrial homeostasis, *PLoS One*, 2015, **10**, 0139121.
- 61 A. Yadav, S. J. S. Flora and P. Kushwaha, Nanocurcumin Prevents Oxidative Stress Induced Following Arsenic and Fluoride Co-exposure in Rats, *Def. Life Sci. J.*, 2016, **1**, 69–77.
- 62 O. A. Ahmed-Farid, M. Nasr, R. F. Ahmed and R. M. Bakeer, Beneficial effects of Curcumin nano-emulsion on spermatogenesis and reproductive performance in male rats under protein deficient diet model: enhancement of sperm motility, conservancy of testicular tissue integrity, cell energy and seminal plasma amino acids content, *J. Biomed. Sci.*, 2017, **24**, 66.
- 63 A. Barzegar and A. A. Moosavi-Movahedi, Intracellular ROS protection efficiency and free radical-scavenging activity of curcumin, *PLoS One*, 2011, **6**, 26012.