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## Suppression of hemin-mediated oxidation of low-density lipoprotein and subsequent endothelial reactions by hydrogen sulfide (H<sub>2</sub>S)

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### Abstract

Heme-mediated oxidative modification of low-density lipoprotein (LDL) plays a crucial role in early atherogenesis. It has been shown that hydrogen sulfide (H<sub>2</sub>S) produced by vascular smooth muscle cells is present in plasma at a concentration of about 50 μmol/L. H<sub>2</sub>S is a strong reductant which can react with reactive oxygen species like superoxide anion and hydrogen peroxide. The current study investigated the effect of H<sub>2</sub>S on hemin-mediated oxidation of LDL and oxidized LDL (oxLDL)-induced endothelial reactions. H<sub>2</sub>S dose dependently delayed the accumulation of lipid peroxidation products—conjugated dienes, lipid hydroperoxides (LOOH), and thiobarbituric acid reactive substances—during hemin-mediated oxidation. Moreover, H<sub>2</sub>S decreased the LOOH content of both oxidized LDL and lipid extracts derived from soft atherosclerotic plaque, which was accompanied by reduced cytotoxicity. OxLDL-mediated induction of the oxidative stress responsive gene, heme oxygenase-1, was also abolished by H<sub>2</sub>S. Finally we have shown that H<sub>2</sub>S can directly protect endothelium against hydrogen peroxide and oxLDL-mediated endothelial cytotoxicity. These results demonstrate novel functions of H<sub>2</sub>S in preventing hemin-mediated oxidative modification of LDL, and consequent deleterious effects, suggesting a possible antiatherogenic action of H<sub>2</sub>S.

### Keywords

Hydrogen sulfide; Low-density lipoprotein; Lipid hydroperoxide; Hemin; Heme oxygenase-1

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## Introduction

Oxidative modification of LDL is currently recognized as one of the early events in atherogenesis [1,2]. Oxidized LDL contributes to atherogenesis in the following ways: (i) it is readily ingested by macrophages through the scavenger receptor that is distinct from the LDL receptor, (ii) it is chemotactic for the circulating monocytes, (iii) it increases monocyte adhesion and it inhibits the motility of macrophages already present in the lesion, and (iv) it stimulates release of cytokines and growth factors and it is cytotoxic to endothelial and smooth muscle cells. However, despite extensive evidence implicating oxidized LDL in the development of cardiovascular disease, the exact mechanisms whereby LDL is oxidized in vivo are still being debated.

We previously suggested that heme—which may be derived from oxidation and decomposition of hemoglobin during intravascular hemolysis [3]—mediates the oxidative modification of LDL [4]. In vitro exposure of LDL to heme promotes extensive oxidative modification of both lipid and apolipoprotein components of LDL, resulting in the formation of an oxidized form of LDL which is cytotoxic to the vasculature. In response to challenge with this material, cells up-regulate the stress responsive genes heme oxygenase-1 (HO-1) and ferritin [5–8]. Up-regulation of HO-1 and ferritin genes in the endothelium occurs in the early phase of progression of atherosclerotic lesions [9], possibly reflecting a cellular response to heme and/or heme-iron-generated lipid peroxidation products. There is growing evidence that induction of HO-1 and ferritin is protective against atherosclerosis. Induction of the HO-1-ferritin system in LDL-receptor knockout mice decreases the formation of atherosclerotic lesions, whereas inhibition of HO-1 enzyme activity by tin protoporphyrin leads to accelerated atherosclerosis in these mice [10].

Hydrogen sulfide (H<sub>2</sub>S) has been traditionally considered as a toxic gas; however, recently it has been identified as the third endogenous gasotransmitter beside CO and NO [11]. Studies reveal that H<sub>2</sub>S, which is generated enzymatically by many types of mammalian cells, is detectable in serum and most tissues at a concentration of about 50 μmol/L [12]. In the vasculature, H<sub>2</sub>S is produced by vascular smooth muscle cells by the pyridoxal-5-phosphate (vitamin B6)-dependent enzyme cystathionine--lyase that uses L-cysteine as a substrate [13]. Hydrogen sulfide exerts a number of physiological actions in the cardiovascular system: (i) it dilates blood vessels mostly, if not exclusively, by a mechanism that involves opening of ATP-sensitive K<sup>+</sup> channels of smooth muscle cells [14], (ii) it is cardioprotective against ischemic reperfusion damage and myocardial inflammation [15], and (iii) it preserves both mitochondrial structure and function after the injury [16]. Mounting evidence suggests that H<sub>2</sub>S has a direct effect on the development of atherosclerosis. Indeed, H<sub>2</sub>S induces apoptosis [17,18] and suppresses endothelin-induced proliferation of vascular smooth muscle cells [19], which in turn might reduce the growth of atherosclerotic lesions. It influences vascular inflammatory reactions [20], and inhibits vascular calcification, a process involving transformation of vascular smooth muscle cells into an osteoblast-like phenotype [21]. It has been shown that progression of atherosclerosis is significantly slower in patients with Down syndrome, a state of H<sub>2</sub>S overproduction [22].

It has been previously shown that plasma thiols inhibit hemin-mediated oxidation of LDL [23], and that exogenous administration of an H<sub>2</sub>S donor reduced lipid peroxidation—including levels of myocardial malondialdehyde, plasma malondialdehyde, and conjugated diene—in a myocardial injury model [23,24]. Therefore we hypothesized that H<sub>2</sub>S might inhibit hemin-mediated oxidative modification of LDL, thereby contributing to the antiatherogenic effect of H<sub>2</sub>S. Indeed, our results show that H<sub>2</sub>S dose dependently prolongs accumulation of oxidation products during hemin-mediated oxidative modification of LDL and decreases the cytotoxic effect of oxidized LDL and of lipid extracted from atheromatous lesions (types IV and Va) [25] on human endothelial cells. Examination of the underlying mechanisms revealed that H<sub>2</sub>S readily reacts with, and reduces the level of, lipid hydroperoxides (LOOH), which is the main contributor to the cytotoxic effect of both oxidized LDL [26] and lipid derived from atheroma. Moreover, H<sub>2</sub>S can directly suppress endothelial cell cytotoxicity caused by either hydrogen peroxide or LOOH. We suggest that both protective mechanisms might be important in the antiatherogenic effects of this gas.

## Materials and methods

### Materials

Medium 199, fetal calf serum, and HankTs balanced salt solution (HBSS) were purchased from Life Technologies, and dispase II was from Boehringer Mannheim (Vienna, Austria). All other reagents utilized were purchased from Sigma unless otherwise specified. NaHS was used to generate H<sub>2</sub>S in solution.

### Cell culture

As in our previous studies [8], human umbilical vein endothelial cells (HUVECs) were removed from human umbilical veins by exposure to dispase and cultured in medium 199 containing 15k fetal calf serum, penicillin (100 U/ml), streptomycin (100 U/ml), and heparin (5 U/ml) supplemented with L-glutamine, sodium pyruvate, and endothelial cell growth factor. Endothelial cells were identified by their morphology and by the presence of von Willebrand factor.

### Preparation and oxidation of human low-density lipoprotein

LDL was isolated from plasma derived from EDTA (1 mg/ml)-anticoagulated venous blood of healthy volunteers who had fasted overnight. Density of plasma was adjusted to 1.3 g/ml with KBr, and a two-layer gradient was made in a Quick-Seal polyallomer ultracentrifuge tube (Beckman Instruments) by layering 0.9k NaCl on 10 ml of density adjusted plasma, which was then centrifuged at 302,000 g for 3 h at 4°C (VTi 50.2 rotor, Beckman Instruments) [27]. The LDL samples were kept at 4°C and protected from light, and the protein content was determined by the BCA protein assay (Pierce, Rockford, IL). Oxidized LDL was prepared by incubating LDL (200 µg/ml protein content diluted in saline) with hemin (5 µmol/L) at 37°C overnight, buffered with Hepes (10 mmol/L, pH 7.4).

### Tissue sample lipid extraction and oxidation

Specimens of human atherosclerotic lesions (types IV and Va) [25] were obtained from aorta or its primary branches of deceased heart-beating donors for organ transplantation. Removal

of tissue samples from these donors was approved by the Scientific and Research Ethic Comity of Scientific Council of Health of the Hungarian Government. Entire vessels and complete lesions were removed during donation, immediately washed with saline, dried, weighed, frozen in liquid nitrogen, and stored at  $-70^{\circ}\text{C}$  until assay. Blood vessel sections were classified by their morphology. Lesions with thickened intima and large lipid core but no sign of disruption were used to extract lipids from by chloroform:methanol (2:1 v/v) according to the method of Bligh and Dyer [28]. Oxidized lipid extract was prepared by incubating lipid extract (1 mg lipid/ml diluted in saline) with hemin (5  $\mu\text{mol/L}$ ) at  $37^{\circ}\text{C}$  overnight, buffered with Hepes (10 mmol/L, pH 7.4).

### Detection of oxidative modification of LDL and lipid extract

Conjugated diene formation and generation of thiobarbituric acidreactive substances (TBARS) in LDL were measured as in our previous studies [4]. Briefly, for conjugated diene measurement LDL was diluted to 50  $\mu\text{g/ml}$  in saline, and optical density was determined at 234 nm. For TBARS determination 600  $\mu\text{l}$  of thiobarbituric-acid reagent (0.375 w/v% 2-thiobarbituric acid, 2 v/v% HCl, 15 v/v% trichloroacetic acid) was added to 300  $\mu\text{l}$  of LDL (200  $\mu\text{g/ml}$ ) and then boiled for 15 min. Following centrifugation (10,000 g, 15 min) the optical density of supernatant was determined at 532 nm. LOOH content of LDL was detected using the Ferrous Oxidation in Xylenol orange (FOX) assay [29]. LOOH content of lipid extracts was determined by an iodometric method. Briefly, 100  $\mu\text{l}$  of lipid suspension was extracted in 110  $\mu\text{l}$  of chloroform and then acetic acid (120  $\mu\text{l}$ ) was added to 80  $\mu\text{l}$  of the organic phase. After incubating this mixture with 40  $\mu\text{l}$  KI (1.2 mg/ml in distilled water) for 5 min in the dark, 600  $\mu\text{l}$  of 50 mmol/L cadmium acetate was added to stop the reaction. After a centrifugation (2000 g, 10 min), the optical density of the aqueous phase was measured at 353 nm. Results were calculated by using the molar extinction coefficient for  $I_3^-$  of  $2.19 \cdot 10^4 \text{ M}^{-1} \text{ cm}^{-1}$  and are expressed as nanomole lipid hydroperoxides per milligram tissue.

### Hemin determination

Total heme content of the reaction mixtures was measured by a colorimetric assay (QuantiChrom Heme Assay Kit, BioAssay System, Hayward, CA).

### Endothelial cell cytotoxicity assay

Confluent HUVEC, grown in 96-well tissue culture plates, were washed twice with HBSS and then exposed to oxidized LDL (200  $\mu\text{g/ml}$ ) or oxidized lipid extracts derived from atheroma (2 mg/ml) alone, or exposed to oxidized LDL or lipid extracts pretreated with  $\text{H}_2\text{S}$  (25, 50, 100, or 200  $\mu\text{mol/L}$ ) at  $37^{\circ}\text{C}$  for 30 min. After a 4-h incubation, the test solutions were replaced with 100  $\mu\text{l}$  of 3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide (MTT) (0.5 mg/ml) solution in HBSS, and cells were incubated for an additional 6 h. Then the MTT solution was removed, 100  $\mu\text{l}$  of DMSO was added, and the optical density at 570 nm was measured. In other experiments endothelial cells were pretreated with  $\text{H}_2\text{S}$  (50  $\mu\text{mol/L}$ ) at  $37^{\circ}\text{C}$  for 4 h and then challenged with hemin (5  $\mu\text{mol/L}$  at  $37^{\circ}\text{C}$  for 1 h), followed by oxidative stress generated by hydrogen peroxide (100 and 200  $\mu\text{mol/L}$ ) or

oxidized LDL (100 and 200 µg protein/ml). After 4 h of incubation the MTT assay was performed.

### Heme oxygenase-1 mRNA analysis

Oxidized LDL (200 µg/ml) was pretreated with H<sub>2</sub>S (25, 50, 100, or 200 µmol/L) at 37 °C for 30 min and then diluted 4 times with HBSS. Confluent endothelial cells were treated with LDL samples (50 µg/ml) for 1 h, followed by a 4-h incubation in complete media. Total RNA was isolated and HO-1 and 18S rRNA levels were measured by real-time PCR. RNA was reverse-transcribed using Superscript II reverse transcriptase (Invitrogen). Real-time PCR was carried out using the iCycler iQ Real Time PCR System (Bio-Rad Laboratories, Hercules, CA). The 25-1 reaction mixtures contained 0.3 nmol/L of forward and reverse primers and 0.13 nmol/L of fluorescent probe for HO-1 (F, 5'-TTCTTGTTCTACGGCTTGCTAC-3'; R, 5'-CTCCAT CCAGATCTCCAGCACT-3'; P, 5'-FAM-CTCAACATCCAGCTCTTTGAGGAGTGCAG-3') or those for 18S rRNA (F, 5'-TTCACCACCATGGAGAAGGC-3'; R, 5'-GGCATGGACTGTGGTCATGA-3'; P, 5'-FAM-TACCACATCCAAGGAAGGCAGCAGG-5'), 3 mmol/L MgCl<sub>2</sub>, 0.2 mmol/L dNTPs, and 0.05 U/ml Taq DNA polymerase (Invitrogen, Carlsbad, CA). Results were normalized by 18S rRNA levels and are presented as fold increase compared to vehicle-treated control.

### Western blot analysis

HUVECs grown in 6-well plates were washed and exposed to oxidized LDL which was pretreated with H<sub>2</sub>S as described above. After 1 h exposure, LDL was replaced by complete media, and cells were further incubated for 8 h. Cells were lysed in Tris-HCl (10 mmol/L) containing Triton X-100 (1k), Nonidet P-40 (0.5k), and protease inhibitors (Complete Mini, Roche, Basel, Switzerland). Protein content was determined, and 20 µg of cell lysate was electrophoresed on 12.5k SDS-PAGE, blotted onto nitrocellulose membrane (Hybond-ECL, Amersham Biosciences, Buckinghamshire, UK). After blocking, the membrane was incubated with polyclonal anti-HO-1 antibody (Calbiochem, Darmstadt, Germany). Antigen-antibody complex was visualized with the horseradish peroxidase chemiluminescence system according to the manufacturer's instructions (Amersham Biosciences, Buckinghamshire, UK). Quantification of HO-1 induction was performed using computer-assisted videodensitometry (AlphaDigiDoc RT, Alpha Innotech Corporation, San Leandro, CA).

### Statistics

Data are shown as mean SD. Statistical analysis was performed by ANOVA test followed by post hoc, Newmann-Keuls test for multiple comparisons. Significance was set at a  $P < 0.05$  level.

## Results

### Effect of H<sub>2</sub>S on oxidative modification of LDL induced by hemin

Sodium hydrogen sulfide, which was used to generate H<sub>2</sub>S, dose dependently prolonged hemin-mediated oxidative modification of LDL. Oxidation of LDL was mediated by the addition of hemin (5 µmol/L) followed by measuring the accumulation of lipid peroxidation

products: conjugated dienes (Fig. 1A, empty triangles), lipid hydroperoxides (Fig. 1B, empty triangles), and TBARs (Fig. 1C, empty triangles). The presence of subphysiologic concentrations of H<sub>2</sub>S (5–20 μmol/L, black triangles) during hemin-mediated LDL oxidation significantly increased the length of the initiation phase of the lipid peroxidation reaction. Possible explanations of the observed effect are: (i) H<sub>2</sub>S can scavenge hemin [30], and reduce the availability of free hemin which acts as a catalyst in the oxidation reaction or (ii) H<sub>2</sub>S acts as a reductant and similarly to its reaction with H<sub>2</sub>O<sub>2</sub> can reduce lipid hydroperoxides which are the ultimate source of lipid alkoxy radicals, the main radical species that propagate transition metal-induced lipid peroxidation. In our experiments a high concentration (100 μM) of H<sub>2</sub>S in aqueous solution reduced hemin content by about 50k (4.92 μmol/L starting concentration vs 2.4 μmol/L) within 20 min at 37°C. In contrast, in the presence of LDL hemin content decreased much slower; 87k of the starting hemin was still present after 20 min and 70k after 1 h of incubation (Fig. 2A). Low concentrations (2–10 μmol/L) of H<sub>2</sub>S did not decrease the hemin content of reaction mixtures containing LDL (200 μg/ml) over a period of 6 h (Fig. 2B).

### **H<sub>2</sub>S dose dependently decreases LOOH content of oxidized LDL**

To examine the possible effect of H<sub>2</sub>S on lipid peroxidation products, first LDL (200 μg/ml protein) was oxidized with hemin (5 μmol/L, 37°C, overnight), and then the oxidized LDL was treated with H<sub>2</sub>S for 30 min, and the concentration of lipid peroxidation products was measured. Conjugated diene and TBAR contents of oxidized LDL were unaffected by H<sub>2</sub>S (Figs. 3A and C). In contrast, H<sub>2</sub>S dose dependently decreased LOOH content of previously oxidized LDL (Fig. 3B), resulting in a 15, 21, 50, and 77k decrease of LOOH content by 25, 50, 100, and 200 μmol/L of H<sub>2</sub>S, respectively, compared to untreated oxidized LDL. H<sub>2</sub>S did not interfere with the assays used to estimate the concentrations of lipid peroxidation products as shown by control measurements using native LDL (Figs. 3A–C, empty bars).

### **Cytotoxic effect of oxidatively modified LDL is dose dependently decreased by H<sub>2</sub>S**

LDL oxidized with hemin was found to be markedly cytotoxic to endothelial cells. Moreover, we have previously shown that oxidized LDL is cytotoxic to vascular endothelium mainly due to its LOOH content [26,31]. Therefore we asked whether the decreased LOOH content of H<sub>2</sub>S-treated oxidized LDL might parallel its reduced cytotoxicity. To answer this question, LDL was oxidized with hemin (5 μmol/L) overnight, followed by a 30-min treatment with H<sub>2</sub>S at the labeled concentrations. As shown in Fig. 4, pretreatment of oxidized LDL with H<sub>2</sub>S prior to the exposure of HUVECs markedly and dose dependently suppressed endothelial cell cytotoxicity.

### **H<sub>2</sub>S decreases the redox-sensitive heme oxygenase-1 gene induction provoked by oxidized LDL**

As a defense mechanism against oxidized LDL-mediated cytotoxicity, endothelial cells up-regulate heme oxygenase-1 when cells are exposed to oxidized LDL at a sublethal dose. Similar to the previous experiment, LDL was first oxidized with heme overnight, and then treated with H<sub>2</sub>S for 30 min prior to treatment of the cells. As shown on Fig. 5, pretreatment of the oxidized LDL with H<sub>2</sub>S dose dependently abolished HO-1 induction provoked by oxidized LDL. Treatment of oxidized LDL with 25 μM H<sub>2</sub>S resulted in a decrease of about

50k in HO-1 mRNA induction (Fig. 5A) provoked by oxidized LDL which was accompanied by an approximately 40k decrease of HO-1 protein induction as shown by Western blot (Fig. 5B). The decreased HO-1 protein expression went along with decreased HO activity of endothelial cells as shown in panel C. In contrast, native LDL in neither the presence nor the absence of H<sub>2</sub>S provokes the induction of HO-1 (Figs. 5A–C, empty bars).

### **H<sub>2</sub>S decreases lipid hydroperoxide level and cytotoxicity of lipid extracted from atherosclerotic soft plaque**

Hemin exposure of lipid extracted from atherosclerotic lesions leads to the rapid amplification of lipid peroxidation products in the lipid extract, which in turn induces endothelial cytotoxicity because of the high lipid hydroperoxide content. We found that H<sub>2</sub>S dose dependently decreased the LOOH content of oxidized lipid extract derived from atherosclerotic lesion (Fig. 6A), which was accompanied by decreased endothelial cytotoxicity (Fig. 6B).

### **Direct antioxidant effect of H<sub>2</sub>S against H<sub>2</sub>O<sub>2</sub> or oxidized LDL-mediated endothelial cytotoxicity**

We further tested the possibility that H<sub>2</sub>S might provide protection for endothelial cells against oxidized LDL or hydrogen peroxide-mediated cell cytotoxicity. Endothelial cells were pretreated with H<sub>2</sub>S (50 μM) for 4 h and then challenged with hemin (5 μmol/L for 1 h), followed by oxidative stress generated by hydrogen peroxide or oxidized LDL. After a 4-h incubation MTT assay was performed. HUVECs pretreated with H<sub>2</sub>S were more resistant to oxidative damage provoked by both hydrogen peroxide and oxidized LDL, and decreased the cytotoxic effect of H<sub>2</sub>O<sub>2</sub> (100 and 200 μmol/L by 31.8 and 37.5k, respectively, and cytotoxic effect of oxidized LDL (100 and 200 μg protein/ml) by 64.2 and 20.0k, respectively (Fig. 7).

## **Discussion**

Hydrogen sulfide is increasingly being recognized as an important signaling molecule in the cardiovascular and nervous systems [11]. In the CNS it functions as not only a neuromodulator [32,33], but also a neuroprotectant against oxidative stress [34]. H<sub>2</sub>S is involved in the regulation of vascular tone [14], myocardial contractility [24], neurotransmission, and insulin secretion [35]. Deficiency of H<sub>2</sub>S was observed in various animal models of arterial [36] and pulmonary hypertension [37], Alzheimer's disease [38], gastric mucosal injury [39], and liver cirrhosis [40]. Exogenous H<sub>2</sub>S ameliorates myocardial dysfunction associated with the ischemia/reperfusion injury [41,42] and reduces the damage of gastric mucosa induced by antiinflammatory drugs [39].

Hemin has been shown to cause endothelial cell damage directly or by promoting the conversion of LDL to cytotoxic oxidized products [4]. In fact there are several types of defenses, which can protect against the deleterious effects of heme. The first line of defense is haptoglobin, an acute phase plasma protein that can bind cell-free hemoglobin dimers and shelter the prosthetic heme groups of hemoglobin [43]. When circulating haptoglobin is depleted, free heme in plasma can be scavenged by hemopexin, a soluble protein that binds

free heme with the highest affinity of any other protein described so far [44]. Once the scavenging capacity of hemopexin is exhausted free heme can be scavenged by plasma albumin. Despite the potential for heme binding by plasma proteins, LDL particles successfully compete for heme released from ferrihemoglobin. In fact it has been estimated that approximately 80k of heme added to whole plasma binds immediately to LDL and high-density lipoprotein [45]. Once heme is inserted into the LDL particle it promotes extensive oxidative modification of LDL, which can be amplified by preformed lipid hydroperoxides within the LDL or other oxidizing agents. This will lead to oxidative scission of the heme group and the release of heme iron, which catalyzes further oxidative breakdown of heme and oxidation of the LDL.

Heme and heme-catalyzed oxidation of LDL are cytotoxic to vascular endothelium. In response, cells up-regulate the last line of defense against the deleterious effects of free heme—the stress responsive genes heme oxygenase-1 and ferritin [4]. Evidence that oxidation of free hemoglobin in plasma could threaten vascular endothelial integrity via oxidative modification of LDL and that toxic species of LDL accumulate in vivo [31] derived from experiments involving LDL isolated from the plasma of the heme oxygenase-1-deficient child reported earlier [46].

Recent studies reveal that H<sub>2</sub>S is generated by many types of mammalian cells, including vascular smooth muscle cells, and plasma concentration of hydrogen sulfide is around 50–100 μmol/L in humans [12]. H<sub>2</sub>S is a strong reductant and may easily react with other compounds, especially with reactive oxygen and nitrogen species (ROS and RNS). It has been shown that H<sub>2</sub>S reacts with superoxide anion [47], hydrogen peroxide [24], peroxyxynitrite [48], and hypochlorite [49], leading, in all cases, to the protection of proteins and lipids from ROS/RNS-mediated damage.

In our study we have examined the role that H<sub>2</sub>S may play in the heme-mediated oxidative modification of LDL and consequent endothelial cell reactions. We found that H<sub>2</sub>S dose dependently delays the accumulation of lipid peroxidation products during heme-mediated oxidative modification of LDL. Although we have previously shown that heme at concentrations between 3 and 5 μmol/L does not strictly influence the kinetics of heme-mediated LDL oxidation [50], to understand the nature of this inhibition we first tested the ability of H<sub>2</sub>S to reduce the availability of heme in our system. In fact, high concentrations of H<sub>2</sub>S in an aqueous solution can readily reduce heme content. In contrast, low concentrations of H<sub>2</sub>S (up to 10 μmol/L) in the presence of LDL did not reduce heme content of the reaction mixture, perhaps because heme intercalated into the LDL was unavailable for reaction. H<sub>2</sub>S is a strong reductant, which can readily react with H<sub>2</sub>O<sub>2</sub> resulting in H<sub>2</sub>O and sulfur. Based on the analogy between chemical reactions of inorganic and organic peroxides, we hypothesized that H<sub>2</sub>S can slow down heme-mediated lipid peroxidation by reacting with LOOH. Radical species propagate transition metal-induced lipid peroxidation, lipid peroxy- and lipid alkoxy radicals, and as previously demonstrated [53] epoxy-allylic peroxy radicals (OLOO·). As LOOH is the ultimate source of alkoxy radicals and epoxy-allylic peroxy radicals, its elimination with H<sub>2</sub>S would explain the observed effect (Fig. 8). Nevertheless, the possibility that H<sub>2</sub>S oxidation products may influence LDL oxidation remains as another mechanism that is untested at this time.



Treating oxidized LDL with H<sub>2</sub>S resulted in a dose-dependent decrease of the lipid hydroperoxide content of the oxidized LDL, while levels of other lipid peroxidation products—conjugated dienes and TBARS—remained unaffected. In previous studies we have shown that there is a strong connection between the lipid hydroperoxide content and the toxicity of the LDL oxidized by hemin, and that the enzymatic reduction of LOOH to LOH yields LDL with minimal toxicity [26]. This observation and the fact that the cytotoxic effect of oxidized LDL could be mimicked by an organic hydroperoxide, cumene hydroperoxide, on an equimolar basis prompted us to hypothesize that the reduced lipid hydroperoxide content of H<sub>2</sub>S-treated oxidized LDL is consonant with its reduced endothelial cytotoxicity.

The present results agree with these earlier observations and show that the reduction of lipid hydroperoxide content by H<sub>2</sub>S attenuates cytotoxic effects of oxidized LDL. We have previously shown that oxidized LDL at sublethal doses up-regulates the HO-1 gene. Agarwal et al. recently investigated the mechanism by which oxidized LDL regulates the expression of HO-1. They found that among the components of oxLDL, the most potent inducer of HO-1 is a lipid hydroperoxide—the 13-hydroperoxyoctadecadienoic acid (13-HPODE), which transcriptionally regulates the HO-1 through a 13-HPODE-specific regulatory element in the human HO-1 promoter [51]. In our previous study we have demonstrated that the HO-1 inducing ability of oxidatively modified LDL strongly depends on its total LOOH content; hence we then examined the effect of H<sub>2</sub>S on the induction of HO-1 provoked by oxidized LDL. Our results demonstrate that pretreatment of the oxidized LDL with H<sub>2</sub>S dose dependently decreases HO-1 mRNA and protein induction and HO-1 activity of the cells.

Oxidative modification of LDL is a key event in the development of atherosclerosis which disorder is characterized by thickening and hardening of the vessel wall due to accumulation of lipids, fibrous tissue, blood components, and calcium in the subendothelial space. Lipid peroxidation products are present in lipid extracts derived from atherosclerotic lesions, and hemin exposure further amplifies LOOH content and cytotoxicity of lipid extracts. In this study we have shown that H<sub>2</sub>S dose dependently decreases LOOH content and cytotoxic effects of oxidized lipid extract derived from atherosclerotic lesion. Interestingly, recent studies have shown that H<sub>2</sub>S protects neurons from glutamate-induced [34] and vascular smooth muscle cells from homocysteine-induced oxidative stress [52]. Our present results indicate that H<sub>2</sub>S pretreatment of cells inhibited cytotoxicity provoked by both H<sub>2</sub>O<sub>2</sub> and oxidized LDL, indicating that H<sub>2</sub>S may play a vital role as an antioxidant.

We also investigated the pH dependency of this reaction. Our results indicate that at lower pH (5.0) formation of lipid hydroperoxides, TBARS, and conjugated dienes are all increased (data not shown). On the other hand, the protective effects of H<sub>2</sub>S are significantly decreased in such an acidic environment. We offer two hypotheses: (i) Since the effects of H<sub>2</sub>S are likely to be evanescent given that it is a gas with a vapor pressure of 18.75 10<sup>5</sup> Pa, we hypothesize that this acidic pH facilitates the evaporation of H<sub>2</sub>S, hence decreasing its availability in the solution. (ii) Considering an in vivo situation a local acidic pH is a common consequence of local inflammation. Vascular smooth muscle cells are able to synthesize up to 100 μM H<sub>2</sub>S under physiological conditions. Hence one may argue that inflammation, which is accompanied with acidic pH, can decrease the protective effects of

such physiological concentrations that may translate to increased susceptibility of LDL to be oxidized.

Our findings indicate a novel role for H<sub>2</sub>S in inhibiting hemin-induced oxidative modification of LDL through a mechanism involving a reduction of LOOH content. Downstream processes induced by LOOH—like endothelial cytotoxicity or induction of HO-1—are also abolished by H<sub>2</sub>S. These results may shed new light on the relationship between H<sub>2</sub>S and atherosclerosis and perhaps may lead to new approaches for combating complications related to atherosclerosis.

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## Abbreviations:

<b>HBSS</b>	HankTs balanced salt solution
<b>HO-1</b>	heme oxygenase-1
<b>H<sub>2</sub>S</b>	hydrogen sulfide
<b>HUVECs</b>	human umbilical vein endothelial cells
<b>LDL</b>	low-density lipoprotein
<b>LOOH</b>	lipid hydroperoxides
<b>MTT</b>	3-[4,5-dimethylthiazol-2-yl]-2,5-diphenyltetrazolium bromide
<b>oxLDL</b>	oxidized LDL
<b>BARS</b>	thiobarbituric acid-reactive substances

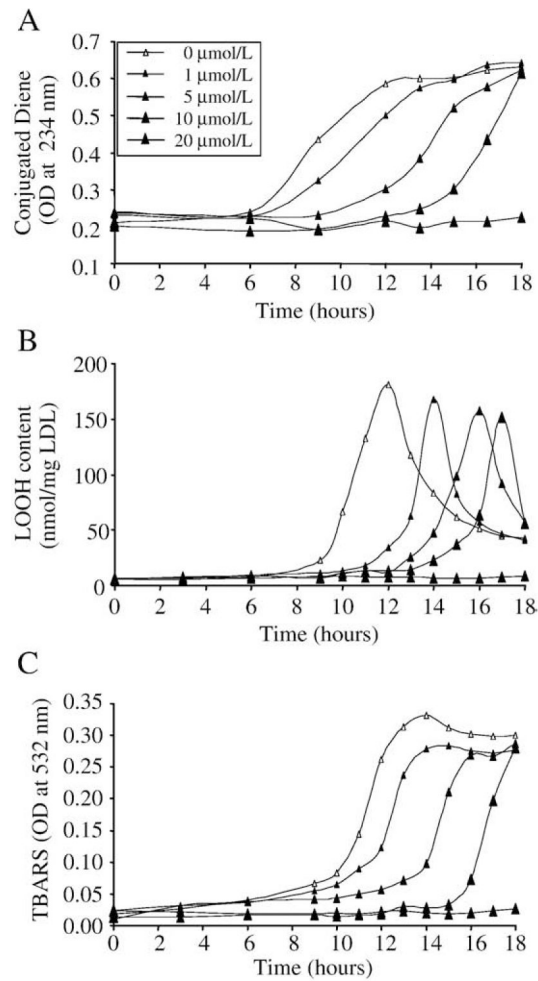
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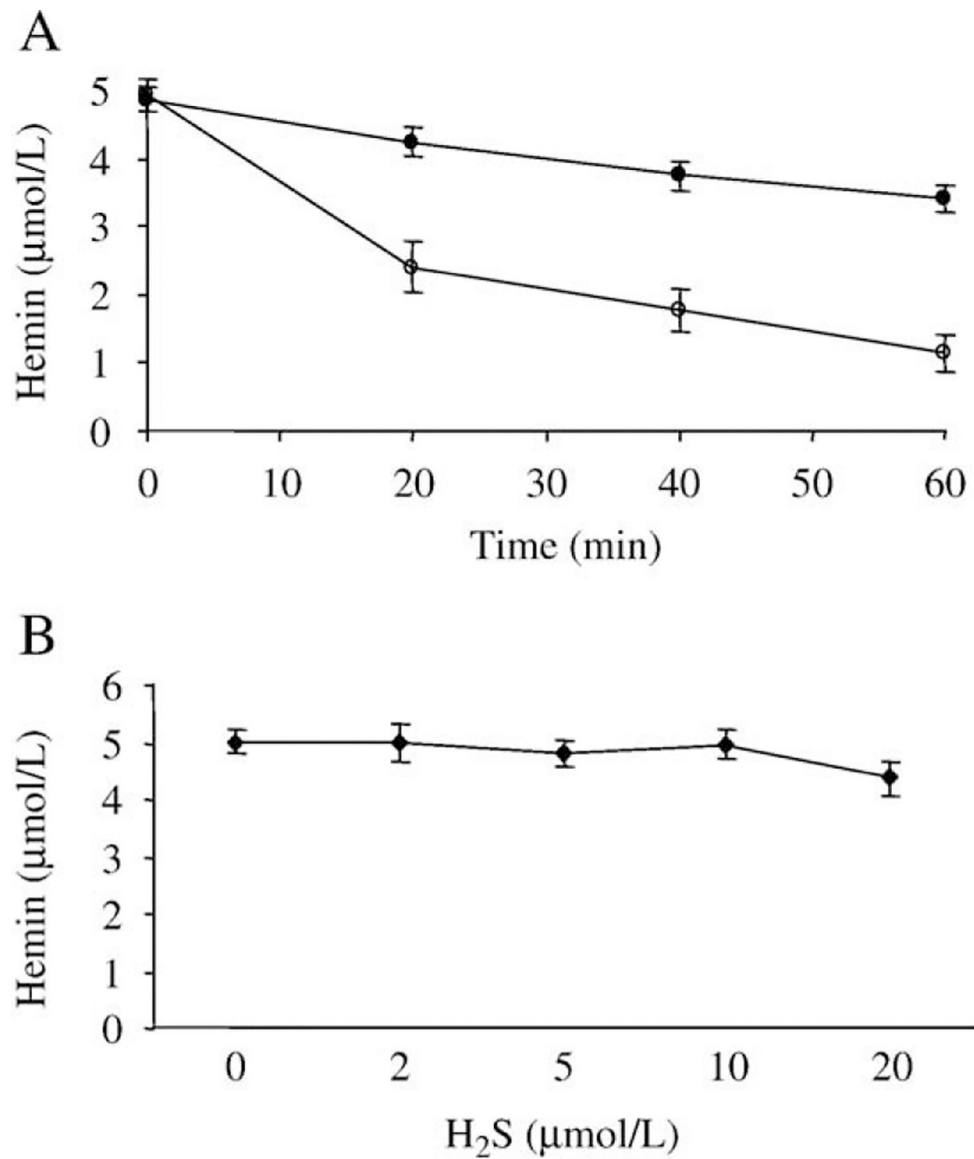
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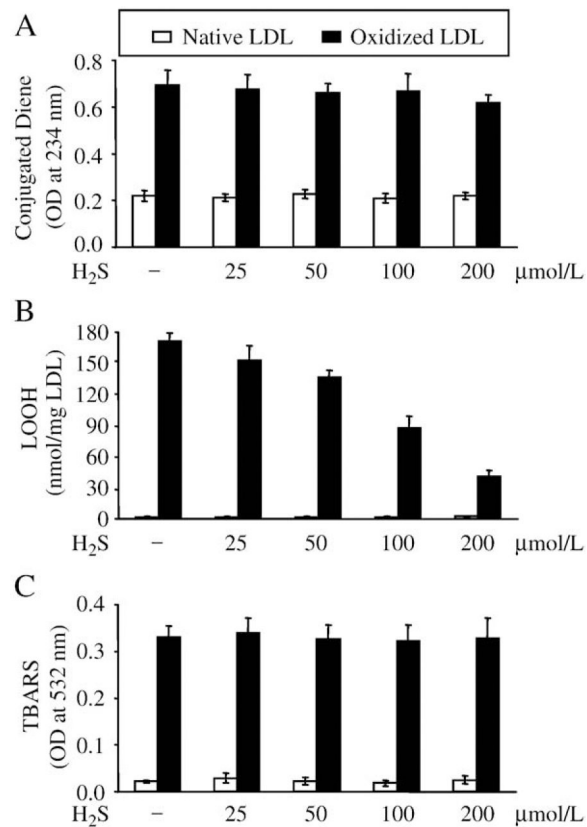
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**Fig. 1.** H<sub>2</sub>S inhibits formation of lipid peroxidation products during hemin-mediated oxidation of LDL in a dose-dependent manner. LDL (200 mg/mL) was incubated with hemin (5 mol/L) alone (empty triangle) or in the presence of H<sub>2</sub>S at a concentration of 1, 5, 10, or 20 mol/L (closed triangles) at 37°C. Lipid peroxidation was monitored by measuring the formation of conjugated dienes (A), LOOH (B), and TBARS (C) for 18 h. Figure shows a representative of three separate experiments.

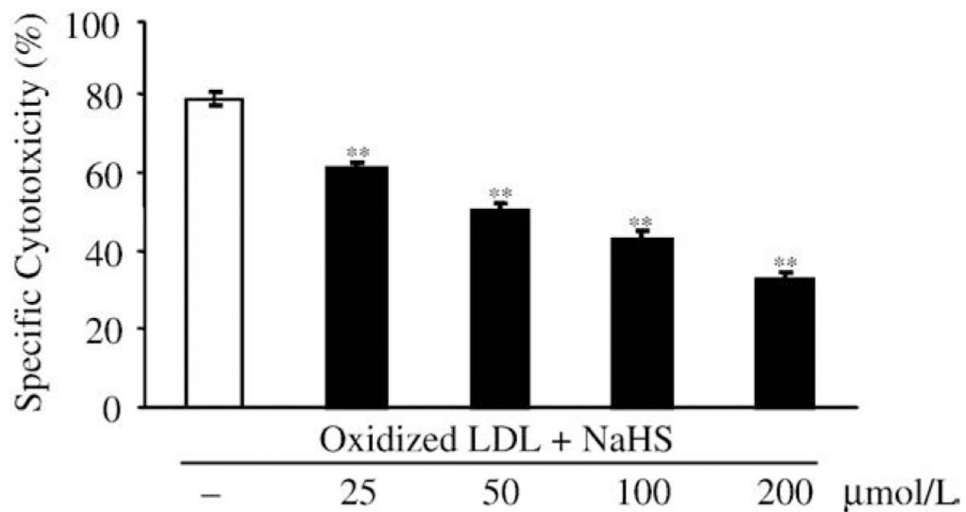


**Fig. 2.** Time and dose dependency of hemin scavenging by  $\text{H}_2\text{S}$ . (A) Hemin (5  $\text{mmol/L}$ ) was incubated with  $\text{H}_2\text{S}$  (100  $\mu\text{mol/L}$ ) in the absence (open circles) or in the presence of LDL (200  $\mu\text{g/ml}$ ) (closed circles) for the indicated time at  $37^\circ\text{C}$ , and then hemin content was determined as described under Materials and methods. (B) Hemin (5  $\mu\text{mol/L}$ ) was incubated with  $\text{H}_2\text{S}$  (2–10  $\mu\text{mol/L}$ ) in the presence of LDL (200  $\mu\text{g/ml}$ ) for 6 h at  $37^\circ\text{C}$ , and then hemin content was determined. Figure shows mean of three independent experiments. Error bars denote standard deviations.

**Fig. 3.**

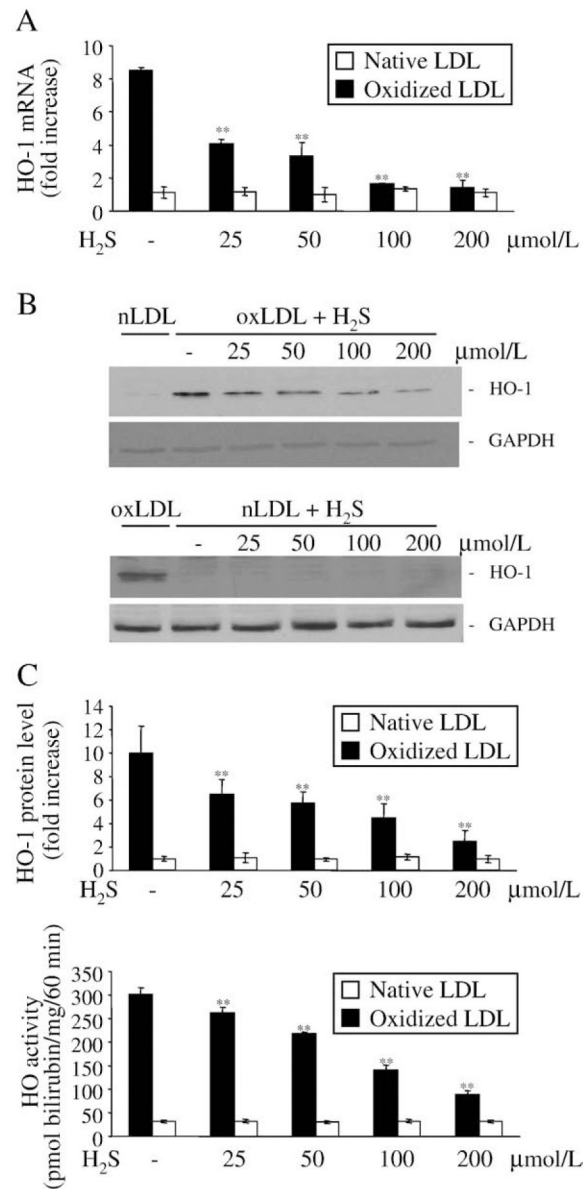
H<sub>2</sub>S dose dependently decreases LOOH content of oxidized LDL. LDL (200 mg/ml) was oxidized with hemin (5 μmol/L) for 12 h at 37°C, and then treated with H<sub>2</sub>S at the indicated concentrations for 30 min at 37°C. Following treatment levels of conjugated dienes (A), LOOH (B), and TBARS (C) were measured. Data are derived from five separate experiments. Error bars denote standard deviations. Statistical significance is indicated by one (Pb0.05) or two (Pb0.01) asterisks.





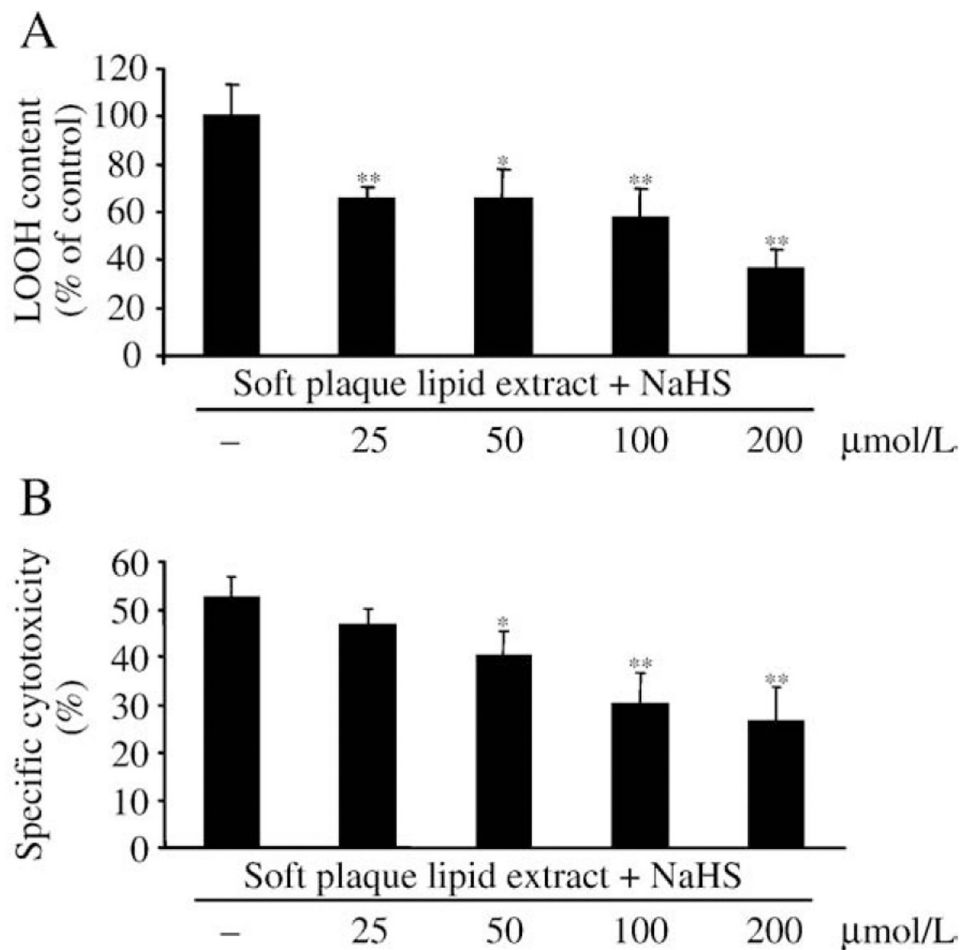
**Fig. 4.**

H<sub>2</sub>S dose dependently abolishes cytotoxic effects of oxidized LDL on HUVECs. LDL (200 μg/ml) was oxidized with hemin (5 μmol/L) for 12 h at 37°C, and then treated with H<sub>2</sub>S at the indicated concentrations for 30 min at 37°C. HUVEC grown on a 96-well plate were washed twice with HBSS and then challenged by the LDL samples for 4 h. MTT assay was performed to measure cytotoxicity. Data are derived from three separate experiments performed in triplicates. Error bars denote standard deviations. Statistical significance is indicated by one (Pb0.05) or two (Pb0.01) asterisks.

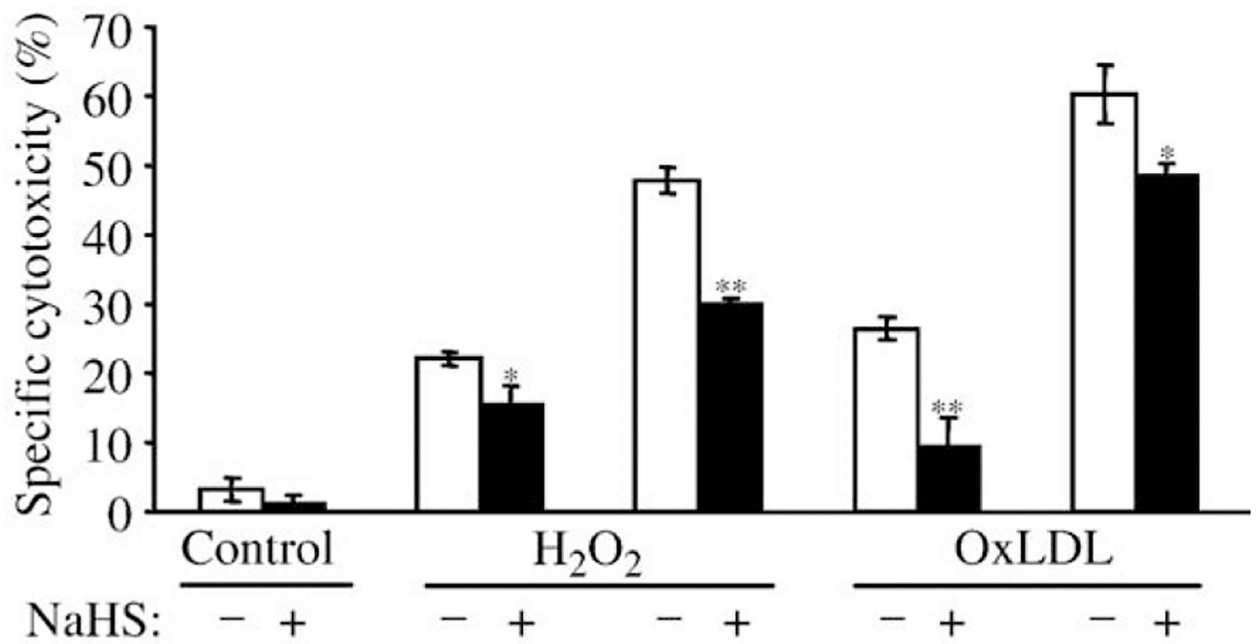
**Fig. 5.**

Heme oxygenase-1 induction mediated by oxLDL is diminished by H<sub>2</sub>S in a dose-responsive manner. Oxidized or native LDL (200 μg/ml) was pretreated with H<sub>2</sub>S (25, 50, 100, or 200 μmol/L) at 37°C for 30 min, and then diluted 4 times with HBSS. (A) For HO-1 mRNA detection confluent endothelial cells were treated with LDL samples (50 μg/mL) for 1 h, followed by a 4-h incubation in complete media. HO-1 mRNA copy numbers were determined by real-time RT-PCR as described under Materials and methods and were normalized by 18S rRNA copies. Fold increase was calculated using vehicle-treated cells as control (B). To determine HO-1 protein levels HUVECs grown on 6-well plates were treated with LDL samples (50 μg/ml) for 1 h, followed by a 8-h incubation in complete media. Cells were lysed and Western blot was performed as described under Materials and methods. After detection of HO-1 membrane was striped and reprobed for GAPDH to prove equal loading.

Quantification of HO-1 induction was performed using computer-assisted videodensitometry. (C) HO-1 activity measured at 8 h after exposure of HUVECs to the same LDL as above. Data for panels A and C are derived from three separate experiments. Error bars denote standard deviations. Statistical significance is indicated by one ( $P<0.05$ ) or two ( $P<0.01$ ) asterisks. Western zblot (B) is a representative of three separate experiments.

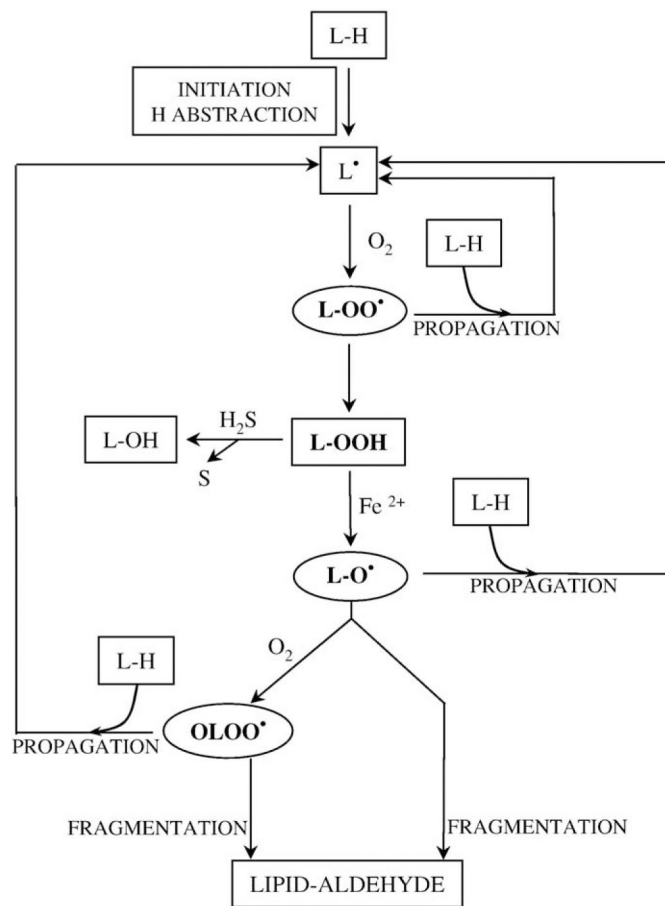


**Fig. 6.** H<sub>2</sub>S dose dependently decreases LOOH content and cytotoxicity of oxidized lipid extract derived from soft plaque. Lipid was extracted from human atherosclerotic lesions as described under Materials and methods. Oxidized extracted lipid (1 mg lipid/ml) was oxidized with hemin (5 μmol/L) for 16 h at 37°C and then treated with H<sub>2</sub>S at the indicated concentrations for 30 min at 37°C. Following H<sub>2</sub>S treatment LOOH levels of samples were determined (A). To measure the effect of H<sub>2</sub>S on the cytotoxicity of oxidized lipid extract, HUVECs grown in 96-well plates were treated with samples generated as above for 4 h, and then MTT assay was performed (B). Data are derived from three separate experiments performed in duplicates. Error bars denote standard deviations. Statistical significance is indicated by one (Pb0.05) or two (Pb0.01) asterisks.



**Fig. 7.**

H<sub>2</sub>S protects HUVECs from cytotoxicity mediated by both H<sub>2</sub>O<sub>2</sub> and oxidized LDL. HUVECs grown in 96-well plates were pretreated with H<sub>2</sub>S (50 μmol/L) at 37°C for 4 h and then challenged with hemin (5 μmol/L at 37°C for 1 h), followed by oxidative stress generated by hydrogen peroxide (100 and 200 μmol/L) or oxidized LDL (100 and 200 μg/ml). After a 4-h incubation MTT assay was performed as described under Materials and methods. Data are derived from five separate experiments performed in triplicates. Error bars denote standard deviations. Statistical significance is indicated by one (Pb0.05) or two (Pb0.01) asterisks.



**Fig. 8.** Proposed mechanism of the protective effect of H<sub>2</sub>S. Lipid peroxidation starts with a H atom abstraction which leads to the formation of a lipid radical (L•). Lipid radical reacts with molecular oxygen forming lipid peroxy radical, one of the radical species which can further propagate lipid peroxidation by reacting with an adjacent polyunsaturated fatty acid producing a new lipid radical and lipid hydroperoxide (LOOH). In the presence of transition metals, e.g., iron, LOOH can form alkoxy radicals (LO•) and epoxy-allylic peroxy radicals (OLOO•) which can propagate the reaction. As LOOH is the ultimate source of alkoxy radicals and epoxy-allylic peroxy radicals its decomposition by H<sub>2</sub>S to lipid alcohols (L-OH) slows down lipid peroxidation.