



# All-Organic Semiconductors for Electrochemical Biosensors: An Overview of Recent Progress in Material Design

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Organic semiconductors remain of major interest in the field of bioelectrochemistry for their versatility in chemical and electrochemical behavior. These materials have been tailored using organic synthesis for use in cell stimulation, sustainable energy production, and in biosensors. Recent progress in the field of fully organic semiconductor biosensors is outlined in this review, with a particular emphasis on the synthetic tailoring of these semiconductors for their intended application. Biosensors ultimately function on the basis of a physical, optical or electrochemical change which occurs in the active material when it encounters the target analyte. Electrochemical biosensors are becoming increasingly popular among organic semiconductor biosensors, owing to their good detection performances, and simple operation. The analyte either interacts directly with the semiconductor material in a redox process or undergoes a redox process with a moiety such as an enzyme attached to the semiconductor material. The electrochemical signal is then transduced through the semiconductor material. The most recent examples of organic semiconductor biosensors are discussed here with reference to the material design of polymers with semiconducting backbones, specifically conjugated polymers, and polymer semiconducting dyes. We conclude that direct interaction between the analyte and the semiconducting material is generally more sensitive and cost effective, despite being currently limited by the need to identify, and synthesize selective sensing functionalities. It is also worth noting the potential roles of highly-sensitive, organic transistor devices and small molecule semiconductors, such as the photochromic and redox active molecule spiropyran, as polymer pendant groups in future biosensor designs.

**Keywords:** conjugated polymer, organic semiconductor, electrochemical, biosensor, material design

## INTRODUCTION

Bioelectrochemistry is the study of naturally-occurring, reduction/oxidation (redox) processes in living systems, encompassing electron transfer in biomolecules, enzyme redox behavior at an electrode, and interactions between synthetic, electro-responsive materials, and biological systems (Wu et al., 2017; Cervera et al., 2018; Oliveira-Brett et al., 2019). Recently, these

natural electrochemical processes have been harnessed in sensors to detect molecules, including biomolecules relevant to the diagnosis, and management of human disease (Naveen et al., 2017). Biosensor materials change their physical, optical, or electrochemical properties in the presence of analyte molecules, thus “sensing” the analyte. For example, optical biosensors undergo changes in optical properties, including optical band gap and absorption and emission spectra, upon interaction with an analyte (Alvarez et al., 2011; Wang J. et al., 2018). Inorganic piezoelectric biosensors have been used as immunosensors and pesticide chemosensors (Skládal, 2016; Pohanka, 2017), while calorimetric (thermal) biosensors detect temperature changes from analyte reactions, and include purely-enzymatic thermistors (Antonelli et al., 2008; Bhand et al., 2010), and piezoelectric quartz oscillators (Gaddes et al., 2017). Electrochemical biosensors, which detect redox reactions of analyte molecules as electrical signals, are especially promising with their low cost, high sensitivity and selectivity, and simple apparatus (Aydin et al., 2018; Moon et al., 2018; Chai and Kan, 2019); they are therefore the focus of this review.

To this end, organic semiconductors are of major interest since their chemical and electrochemical properties can be tailored using organic synthesis to the targeted application. They are the core component of organic bioelectronic devices for biomolecule sensing (Park et al., 2008; Wang et al., 2019), cell stimulation (Fidanovski and Mawad, 2019; Hopkins et al., 2019), and sustainable energy generation (Chen et al., 2004; Wallace et al., 2005; Li et al., 2017). In electrochemical biosensors, small molecule semiconductors are used individually or as polymer pendant groups; for example, spiropyran derivatives are photochromic and undergo ring-opening isomerization under various stimuli (Miyagishi et al., 2019), while organometallic “redox polymers” bearing pendant ferrocene, and osmium complexes have detected numerous neurotransmitter molecules (Casado et al., 2016). However, conjugated polymers (CPs)—polymers with  $\pi$ -conjugated, semiconducting backbones—exhibit improved sensitivity in biosensing due to their high electrical conductivity and efficient, tailorable charge transport characteristics, permitting rapid signal transduction (Park et al., 2008), and their biocompatibility allows their biological application (Cevik et al., 2019). Importantly, their redox-active backbone and propensity for flexible modification with numerous chemical functionalities allow them to mediate electrochemical reactions. Similarly, commercially-available, organic dyes such as methylene blue are readily electropolymerized, producing polymers with semiconducting backbones which efficiently mediate charge transport and redox reactions, with demonstrated utility in biosensing (Barsan et al., 2015). Given these polymeric materials’ advantages, this review examines the material design of all-organic polymers with semiconducting backbones for electrochemical biosensors. We first discuss the operating principles and mechanisms of electrochemical biosensors. We then review the recent syntheses of novel, organic polymers with semiconducting backbones, including functional CPs and polymers of organic dyes, and outline their electrochemical detection of significant biomolecules.

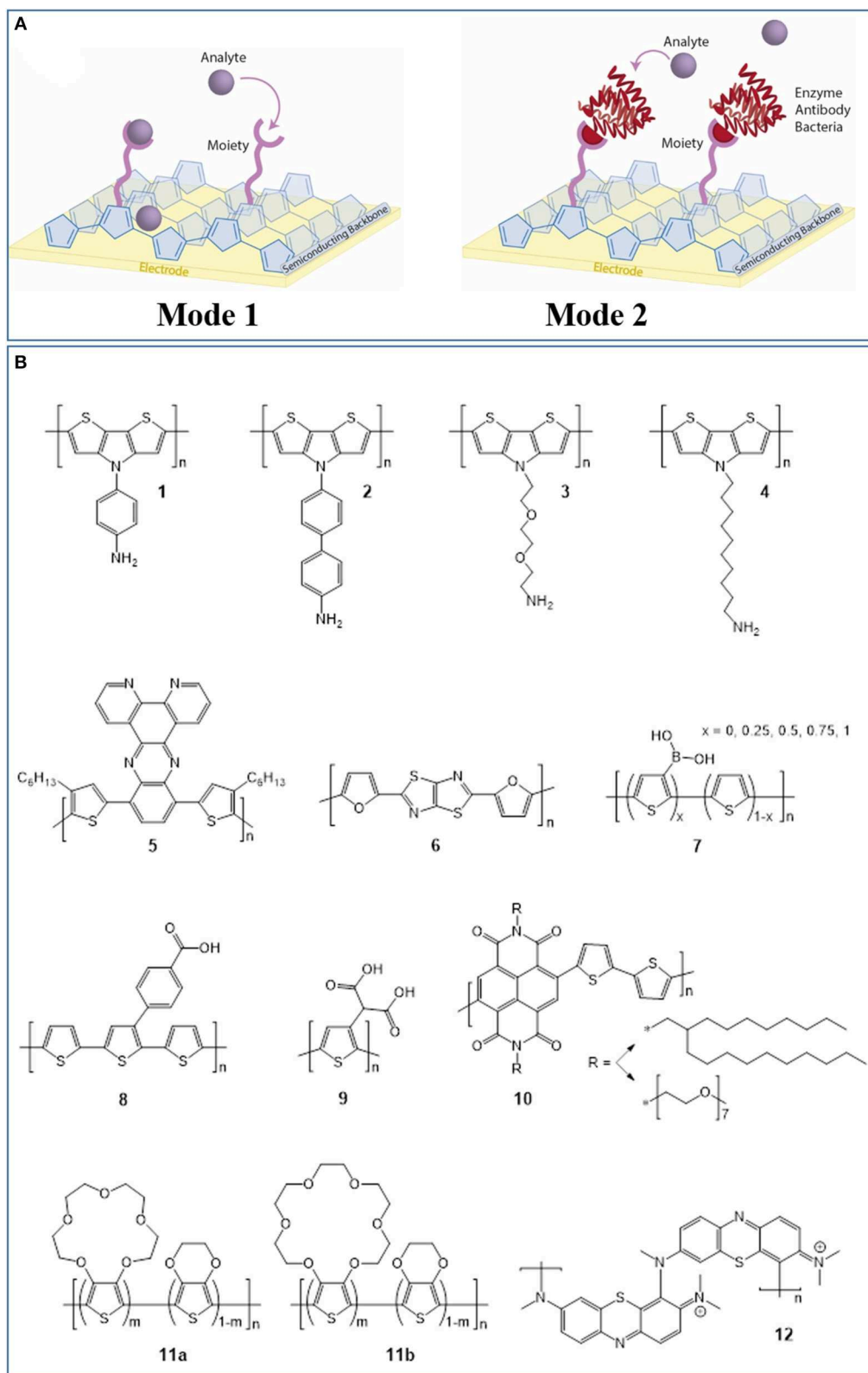
## ORGANIC SEMICONDUCTING POLYMERS FOR ELECTROCHEMICAL BIOSENSORS

### Operating Principles of Electrochemical Biosensors

Electrochemical sensors operate as transducers in an electrochemical cell, facilitating analyte binding or electrochemical reaction at the surface under an applied potential (Park et al., 2008; Moon et al., 2018). This potential is applied using: amperometry, which monitors changes in current at constant potential; potentiometry, which measures potential with no current; impedometry, which measures the steady-state current response to a small alternating potential, typically using electrochemical impedance spectroscopy (EIS); and cyclic voltammetry (CV), which measures the current under a cyclic potential, generating current peaks from redox reactions. Sensor-analyte electron transfer results in electrochemical signals which are transduced through the semiconductor into an underlying electrode, and detected using a potentiostat. Transistor configurations, including organic electrochemical transistors (OECTs) introduced by White et al. (1984), utilize an additional gate voltage to amplify signals and improve biosensor sensitivity, as recently reviewed (Bai et al., 2019; Wang et al., 2019). Electrochemical methods in general offer inexpensive, simple operation, high sensitivity and selectivity, low limits of detection (LOD), and broad linear detection ranges—all important criteria for biosensors (Aydin et al., 2018; Moon et al., 2018; Chai and Kan, 2019). Organic electrochemical biosensors have two modes of detection mechanism (**Figure 1A**). The organic semiconductor can be chemically modified with functionalities that directly facilitate analyte redox reactions. Alternatively, the organic semiconductor can be functionalized with complex moieties, including enzymes (enzymatic sensors), antibodies (immunosensors), and bacteria (bacterial sensors), which bind to the analyte, and mediate its reduction/oxidation. In this review, we refer to these mechanisms as “mode-1” and “mode-2,” respectively. Both mechanisms require semiconductor materials combining efficient charge carrier transport within the material with novel functionalities for either sensing mode. Consequently, for organic semiconductor-based biosensors, chemical synthesis is important to access novel material properties and functionalities tailored to detecting specific biomolecules. Therefore, we next discuss the recent syntheses of organic polymers with semiconducting backbones for electrochemical biosensors, focusing on conjugated polymers, and polymers of semiconducting organic dyes.

### Material Design for Electrochemical Biosensors

The tailored synthesis of CPs for biosensors has received significant attention in recent literature. In particular, electrochemical glucose biosensors have been prominent since their invention by Clark and Lyons (1962) because glucose is relevant to many diseases, including endocrine disorders, and diabetes (Cevik et al., 2019). For example, the CP poly(dithieno(3,2-b:2',3'-d)pyrrole) (PDTP) and its



**FIGURE 1 | (A)** The two modes of detection mechanism in organic electrochemical biosensors. Mode-1 detection involves direct interaction between polymer functionalities and the analyte, while mode-2 detection requires additional, biosensing moieties such as enzymes, antibodies, or bacteria to be chemically attached to the polymer. **(B)** Chemical structures of organic semiconductors used in electrochemical biosensors in recent literature.

derivatives have previously attracted interest in organic field-effect transistors (OFETs), and more recently in electrochemical glucose biosensors, since their planar structures and fused ring systems yield high hole mobilities reaching  $0.41 \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$  and efficient, enzyme-to-electrode charge transport (Parameswaran et al., 2009; Rasmussen and Evenson, 2013). Azak et al. (2016) synthesized PDTP derivatives bearing *N*-substituted aniline (**Figure 1B**, semiconductor 1) and biphenylamine (2) functionalities for “mode-2” glucose detection. The dithieno(3,2-b:2',3'-d)pyrrole (DTP)-based monomers were synthesized using either Cu-catalyzed, Ullmann-type coupling or Pd-catalyzed, Buchwald-Hartwig coupling of 3,3'-dibromo-2,2'-bithiophene with aryl amines. Monomers underwent CV electropolymerization onto a gold substrate, then glucose oxidase (GOx) enzyme molecules, and gold nanoparticles (AuNPs) modified with amine functionalities were immobilized on the polymer surface. AuNPs are commonly incorporated into biosensors to improve sensor sensitivity and selectivity (Naveen et al., 2017; Moon et al., 2018). The GOx catalyzed glucose oxidation in phosphate-buffered solution (PBS, pH = 7.4) during chronoamperometry. Both PDTP derivatives afforded wide linear detection ranges, and polymer 2 gave an especially low LOD of  $0.0986 \mu\text{M}$  (**Table 1**). The biosensors reliably measured glucose concentrations in spiked human blood samples (<1% relative standard deviation, RSD), emphasizing their potential in diabetes treatment.

The authors subsequently developed PDTP biosensors without AuNPs (Azak et al., 2018). They synthesized alkoxyamine-functionalized PDTP (**3**) *via* Buchwald-Hartwig coupling and performed CV electropolymerization onto an indium tin oxide (ITO)-glass substrate. The electron-donating alkoxyamine side groups were chosen over alkyl amine groups to reduce the energy requirement for monomer oxidation during electropolymerization. GOx was immobilized by covalently bonding to polymer amine groups, assisted by glutaraldehyde (GA) crosslinking. Glucose was detected in PBS (pH = 7.0) with a low LOD of  $0.348 \mu\text{M}$ , superior to many electrochemical glucose biosensors (**Table 1**). The bound enzymes exhibited impressive storage stability, retaining 80% enzymatic activity after 20 days—a significant step toward improving biosensor longevity. The same group synthesized *N*-decylamine-substituted DTP monomer by Buchwald-Hartwig coupling, as reported previously (Udum et al., 2014), followed by electrochemical polymerization onto a glassy carbon electrode giving polymer **4** (Cevik et al., 2019). They then fabricated enzymatic and bacterial biosensors for glucose by utilizing the decylamine functionalities to immobilize GOx and *Gluconobacter oxydans* (*G. oxydans*) onto the surface. *G. oxydans* has demonstrated utility in electrochemical biosensors for numerous functionalities (Katrlik et al., 2007). Both biosensors detected glucose *via* amperometry in PBS at optimized pH. While the bacterial biosensor exhibited more reliable detection in the presence of other molecules, the GOx biosensor was superior overall with a lower LOD (22 vs.  $81 \mu\text{M}$ ) and broader linear dynamic range (0.045–50.0 vs. 0.19–50.0 mM).

Buber et al. (2018) explored an alternative glucose sensor involving a novel, bithiophene-phenazine-based CP (**5**) without

amine functionalities as a GOx-immobilization substrate. The monomer, 10,13-bis(4-hexylthiophen-2-yl)dipyridol[3,2-a:2',3'-c]phenazine (HTPP), was synthesized by Stille coupling of thiophene and phenazine precursors, phenazine reduction to amines, and condensation with an aromatic dione (Esmer et al., 2011). The polymer was deposited using CV onto a graphite substrate. Structurally, the polymer was designed to improve GOx adhesion *via* hydrophobic interactions with hexyl side chains and intermolecular  $\pi$ - $\pi$  interactions with the numerous aromatic rings. Through GA crosslinking, GOx was immobilized on the surface to oxidize and detect  $\beta$ -D-glucose in PBS by measuring molecular oxygen consumption using amperometry. After optimization, the sensor exhibited superior sensitivity ( $105.12 \mu\text{A mM}^{-1} \text{ cm}^{-2}$ ), and LOD ( $2.88 \mu\text{M}$ ) to comparable published systems, and accurately measured glucose levels in commercial beverages (<10% deviation from product label).

Donor-acceptor CPs, containing electron donor and acceptor repeat units, exhibit efficient electrochromic switching, and applications in OFETs and solar cells (Soylemez et al., 2019), and also possess biosensing capabilities. Soylemez et al. (2019) synthesized a donor-acceptor CP (**6**) containing furan (donor) and thiazole (acceptor) moieties in a “proof-of-concept” glucose biosensor. The monomer, 2,5-di(furan-2-yl)thiazolo[5,4-d]thiazole, was synthesized in a mild, single-step reaction and underwent CV electropolymerization onto a graphite electrode. The polymer exhibited reversible electrochromic behavior with fast redox switching times (0.3 and 0.4 s). After immobilizing GOx with GA crosslinking, glucose was detected in beverages with good sensitivity ( $65.44 \mu\text{A mM}^{-1} \text{ cm}^{-2}$ ) and reasonable LOD ( $12.8 \mu\text{M}$ ). Although these values are inferior compared to Buber et al. (2018), this work emphasizes the wide range of CP designs applicable to biosensing. CPs have also been used to detect neurotransmitters including dopamine, a neurotransmitter involved in several neurological conditions including Alzheimer's disease. Dopamine binds strongly and selectively to boronic acids, permitting mode-1 dopamine detection. However, the low physiological concentration of dopamine necessitates high biosensor sensitivity and selectivity (Jiang et al., 2017). As such, Dervisevic et al. (2017a) copolymerized thiophene and 3-thienylboronic acid by CV onto pencil graphite, producing a PT derivative (**7**) with boronic acid groups. In dopamine solution, these groups immobilized dopamine molecules at physiological pH, altering the devices' impedimetric response. This permitted selective dopamine detection in human urine with a wide linear range ( $7.8$ – $125 \mu\text{M}$ ) and very low LOD ( $0.3 \mu\text{M}$ ). This CP also selectively detected tumor cells (Dervisevic et al., 2017b): electrodes coated with electropolymerized CP were submerged in a medium containing human Caucasian gastric adenocarcinoma (AGS) cancer cells, which generate abnormally large quantities of sialic acid ( $1,000\times$  greater). Through EIS, the boronic acid-functionalized CP detected sialic acid with high selectivity and a low cell LOD ( $10 \text{ cells mL}^{-1}$ ), highlighting its potential in reliable, early cancer diagnosis.

Akhtar et al. (2017) developed an innovative biosensor for the neurotransmitter acetylcholine associated with various

**TABLE 1** | Summary of biosensor configurations, mechanisms, and performances.

SC	Additional species	Analyte	Detection mechanism	Detection method	Sensitivity ( $\mu\text{A mM}^{-1} \text{cm}^{-2}$ )	LOD/Linear range ( $\mu\text{M}$ )	References
1	GOx, AuNPs	Glucose	Mode-2, enzymatic	CA	NR	50.0/100–2,500	Azak et al., 2016
2	GOx, AuNPs	Glucose	Mode-2, enzymatic	CA	NR	0.0986/50–1,000	Azak et al., 2016
3	GOx	Glucose	Mode-2, enzymatic	CV	NR	0.348/50–900	Azak et al., 2018
4	GOx	Glucose	Mode-2, enzymatic	AM	NR	22/45–50,000	Cevik et al., 2019
4	<i>G. oxydans</i>	Glucose	Mode-2, bacterial	AM	NR	81/190–50,000	Cevik et al., 2019
5	GOx	Glucose	Mode-2, enzymatic	AM	105.12	2.88/25–1,000	Buber et al., 2018
6	GOx	Glucose	Mode-2, enzymatic	AM	65.44	12.8/5–700	Soylomez et al., 2019
7	—	Dopamine	Mode-1	EIS	NR	0.3/7.8–125	Dervisevic et al., 2017a
7	—	Sialic acid (AGS cells)	Mode-1	EIS	NR	10/10–10 <sup>6</sup> d	Dervisevic et al., 2017b
8	Acetylcholinesterase/ choline oxidase, AuNPs	Acetylcholine	Mode-2, enzymatic	CA	NR	0.6/0.7–1,500	Akhtar et al., 2017
9	Anti-IL-1 $\beta$ antibodies	Interleukin 1 $\beta$	Mode-2, immunosensor	EIS	NR	3 $\times$ 10 <sup>-6</sup> /1 $\times$ 10 <sup>-5</sup> –0.003 <sup>c</sup>	Aydin et al., 2018
10	GOx	Glucose	Mode-2, enzymatic	CA/EIS	NR	10/10–10,000	Savva et al., 2019
10	Lactate oxidase	Lactate	Mode-2, enzymatic	CA	NR	10/10–1,000	Pappa et al., 2018
11a	—	Sodium ions	Mode-1	CA/CV/SS	37 <sup>a</sup>	20/10–10 <sup>6</sup>	Wustoni et al., 2019
11b	—	Potassium ions	Mode-1	CA/CV/SS	49 <sup>a</sup>	100/100–10 <sup>6</sup>	Wustoni et al., 2019
12	GDH, NAD <sup>+</sup>	Glucose	Mode-2, enzymatic	AM	NR	4.0/10–1,000	Dilgin et al., 2018
12	—	Creatine	Mode-1	DPV	0.133 <sup>b</sup>	0.2/0.5–900 <sup>c</sup>	Pandey et al., 2018

SC, Semiconductor; GOx, glucose oxidase; AuNPs, gold nanoparticles; GDH, glucose dehydrogenase; NAD<sup>+</sup>, nicotinamide adenine dinucleotide; SCNs, single-walled carbon nanotubes; CA, chronoamperometry; CV, cyclic voltammetry; AM, amperometry; EIS, electrochemical impedance spectroscopy; DPV, differential pulse voltammetry; SS, steady-state OECT measurements; NR, not reported.

<sup>a</sup>Units are  $\mu\text{A decade}^{-1}$ .

<sup>b</sup>Units are  $\mu\text{A ng mL}^{-1}$ .

<sup>c</sup>Units are  $\text{ng mL}^{-1}$ .

<sup>d</sup>Units are cells  $\text{mL}^{-1}$ .

neurological and physiological conditions. They used a dual-electrode, microfluidic device to improve enzyme loading, and minimize signal interference. The monomer, 2,2':5',2''-terthiophene-3-(p-benzoic acid), was synthesized by boronic acid functionalization of 3'-bromo-2,2':5',2''-terthiophene, Suzuki coupling with 4-bromobenzonitrile, and alkaline hydrolysis of nitrile groups to carboxylic acid groups (Kim et al., 2012). This monomer underwent CV electropolymerization, depositing polymer **8** onto an AuNP-coated "reaction electrode" and a porous, Au-coated "detection electrode." Acetylcholinesterase was immobilized on the reaction electrode and choline oxidase was immobilized on the detection electrode. Detection involved successive conversion of acetylcholine into choline at the reaction electrode, then into betaine and hydrogen peroxide at the detection electrode. Reduction of hydrogen peroxide by hydrazine released electrons which were detected by chronoamperometry. The sensor exhibited a wide dynamic range (0.7 nM–1,500  $\mu\text{M}$ ), very low average LOD (0.6 nM), and high selectivity to acetylcholine from the multi-step reaction sequence. This sensor monitored the *in-vitro* extracellular release of acetylcholine by leukemic T-cells triggered by calcium ions. Aydin et al. (2018) utilized self-assembled films of a densely carboxylated PT derivative, poly(3-thiophene malonic acid) (P3-TMA, **9**), as electrochemical immunosensors for the protein Interleukin 1 $\beta$  (IL-1 $\beta$ ) involved in human immune response. The

polymer was synthesized by chemical oxidation of the methyl ester monomer, then hydrolysis yielding ionizable, carboxylic acid groups. These groups both bound to a treated ITO substrate with hydroxyl surface functionalities, forming a self-assembled P3-TMA monolayer, and immobilized anti-IL-1 $\beta$  antibodies on the monolayer permitting biorecognition. Selective immunodetection of IL-1 $\beta$  antigen in human serum and saliva was demonstrated with a very low LOD (3 fg/mL), two orders of magnitude lower than the next-best, CP-free system (300 fg/mL).

CP-based OECT configurations are gaining popularity as sensitive electrochemical biosensors. OECT biosensors commonly utilize water-processable poly(ethylenedioxythiophene):poly(styrene sulfonate) (PEDOT:PSS) (Liao et al., 2019), for example in conjunction with lipid bilayers to detect the ion pore  $\alpha$ -hemolysin (Zhang et al., 2016). Recent OECT biosensors have explored alternative CPs with tailored structures. Giovannitti et al. (2018) synthesized a naphthalene-bithiophene-based donor-acceptor copolymer (**10**) and introduced glycol side chains *via* amine-anhydride coupling to improve ion transport in OECTs. The CP was coated with GOx or lactate oxidase without enzyme immobilization and used for glucose (Savva et al., 2019), or lactate (Pappa et al., 2018) detection, respectively, with good LODs (both 10  $\mu\text{M}$ ). Wustoni et al. (2019) synthesized thiophene derivatives functionalized with 15-crown-5 and 18-crown-6 ethers to selectively trap

sodium and potassium ions, respectively, two important ions in cell signaling. Each monomer was copolymerized with ethylenedioxythiophene (EDOT) onto OECTs using various electropolymerization methods. Under an optimized OECT gate potential, the copolymers (11a/11b) performed selective, real-time detection of sodium and potassium ions with good LODs (20/100  $\mu\text{M}$ ), broad linear detection ranges (10–10<sup>6</sup>/100–10<sup>6</sup>  $\mu\text{M}$ ), and comparable sensitivities to the “gold-standard” PEDOT:PSS.

Alternatively to CPs, semiconducting dyes are commercially available and readily undergo electropolymerization forming redox-active polymers. The earlier use of organic dye-based polymers in electrochemical biosensors has been reviewed (Barsan et al., 2015). However, here we summarize the many recent biosensing applications of poly(methylene blue) (PMB, **12**), a prominent redox-active polymer, with one mode-1 and one mode-2 electrochemical biosensor. Dilgin et al. (2018) electropolymerized methylene blue onto poly(amidoamine)-coated, disposable graphite electrodes using CV, and immobilized glucose dehydrogenase (GDH) on the PMB with GA crosslinking. Aided by nicotinamide adenine dinucleotide (NAD<sup>+</sup>), this PMB-GDH system facilitated mode-2, amperometric glucose detection *via* several reactions. First, GDH oxidized glucose and reduced NAD<sup>+</sup> to NADH; the underlying PMB then re-oxidized NADH; finally, an electrochemical potential re-oxidized PMB generating an electrical signal. This sensor selectively detected glucose in artificial blood serum and commercial dextrose solutions under flow conditions with a reasonable LOD (4.0  $\mu\text{M}$ ). Considering the disadvantages of enzymatic biosensors, including poor storage stability and high cost, Pandey et al. (2018) fabricated a non-enzymatic, mode-1 PMB biosensor for creatinine, an indicator of renal dysfunction. Dendritic PMB nanofibers were synthesized using CV onto a Cu-doped carbon nanofiber substrate, producing a polymer/metal/carbon nanocomposite sensor. Through PMB and creatine coordination to Cu centers, this sensor showed excellent selectivity, sensitivity (0.133  $\mu\text{A ng mL}^{-1}$ ), and LOD (0.2  $\text{ng mL}^{-1}$ ), with consistent measurements in clinical human saliva samples (1–2% RSD). Examples abound of recent electrochemical biosensors with polymers including PMB (Koyun and Sahin, 2018; Li et al., 2018; Wang and Ma, 2018; Bollella et al., 2019a,b), poly(alizarin yellow R) (Amini et al., 2019), poly(azure A) (Agrisuelas et al., 2018), poly(azure B) (Porfireva et al., 2019; Stoikov et al., 2019), poly(azure C) (Liu et al., 2019), poly(brilliant cresyl blue) (da Silva et al., 2019), and poly(thionine) (Shamspur et al., 2018; Wang Y. et al., 2018; Zhao and Ma, 2018; Chai and Kan, 2019; Stoikov et al., 2019), demonstrating the versatility of organic dyes in designing novel, semiconducting polymers for biosensors.

## CONCLUSIONS

Organic semiconducting polymers are a highly versatile, promising class of materials for biosensors since their organic synthesis can be tailored to achieve various functionalities for different applications. This versatility is evidenced by the numerous, innovative techniques in recent literature to synthesize organic biosensor materials. Electrochemical

biosensors have especially great potential for widespread, *in-vivo* application, since recent biosensor materials exhibit high sensitivity, LODs as low as 0.0986  $\mu\text{M}$ , and broad linear ranges spanning five orders of magnitude. Novel electrochemical transistor configurations including OECTs enhance sensitivity through signal amplification and represent a popular new direction in biosensor design. Mode-2 biosensors utilizing complex moieties are most common and historically well-established; here the organic semiconductor only transduces the electrochemical signal, while additional moieties including enzymes and bacteria are primarily responsible for detection. Consequently, numerous semiconductor functionalities have been developed to bind to these moieties, including amine groups for enzymes, and fused-ring backbones which also improve charge transfer. Developing chemical functionalities with improved enzyme binding may also reduce the current overdependence on GA crosslinking for enzyme adhesion.

Recent mode-1 biosensors, which directly mediate analyte redox, exhibit competitive LODs, and detection ranges compared to mode-2 biosensors (**Table 1**). Both mechanisms can achieve selectivity by introducing specific functionalities, including GOx for glucose, and boronic acids for dopamine. However, Pandey et al. (2018) note that mode-2 biosensors often suffer from reduced storage stability and higher cost from the additional, detecting moiety. Conversely, mode-1 biosensors are uncommon since they require the identification and synthesis of selective biosensing functionalities replacing naturally-occurring moieties. While both mechanisms are worth pursuing, future syntheses should also investigate novel semiconductors for mode-1 detection to complement established, “mode-2” technologies. Continued development of mode-1 biosensors would likely increase their impact in multiplying opportunities for low-cost, disposable devices for clinical use, especially when combined with “disposable” electrode materials including pencil graphite (Dilgin et al., 2018).

One material design strategy to expand the scope of mode-1 biosensors would involve introducing small molecule organic semiconductors such as spiropyran as pendant groups on aliphatic or conjugated polymers. As part of a growing trend toward small molecule semiconductors in biosensors, spiropyran derivatives are becoming increasingly common in optical biosensors due to their reversible photochromism under various stimuli. Recently, Li et al. (2013) utilized silyl-modified spiropyran for optical fluoride detection in biological media, while Shao et al. (2018) used siloxane polymers with spiropyran pendant groups for optical silver and iron(III) ion detection. Tao et al. (2016) extended the use of spiropyran derivatives to mode-1 electrochemical detection of fluoride. Future developments in synthesizing small molecule semiconductors as polymer pendant functionalities would help establish their viability in mode-1 electrochemical detection, expanding our ever-growing “library” of organic semiconductors for electrochemical biosensors.

## AUTHOR CONTRIBUTIONS

All authors listed have made a substantial, direct and intellectual contribution to the work, and approved it for publication.

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**Conflict of Interest:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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