

## RESEARCH ARTICLE

# Importance of sample volume to the measurement and interpretation of plasma osmolality

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**Background:** Small sample volumes may artificially elevate plasma osmolality (Posm) measured by freezing point depression. The purpose of this study was to compare two widely different sample volumes of measured Posm (mmol/kg) to each other, and to calculated osmolality (mmol/L), across a physiological Posm range (~50 mmol/kg).

**Methods:** Posm was measured using freezing point depression and osmolality calculated from measures of sodium, glucose, and blood urea nitrogen. The influence of sample volume was investigated by comparing 20 and 250  $\mu$ L Posm samples ( $n = 126$  pairs). Thirty-two volunteers were tested multiple times while EUH ( $n = 115$ ) or DEH ( $n = 11$ ) by  $-4.0\%$  body mass. Protinol™ (240, 280, and 320 mmol/kg) and Clinitol™ (290 mmol/kg) reference solutions were compared similarly ( $n = 282$  pairs).

**Results:** The 20  $\mu$ L samples of plasma showed a 7 mmol/kg positive bias compared to 250  $\mu$ L samples and displayed a nearly constant proportional error across the range tested (slope = 0.929). Calculated osmolality was lower than 20  $\mu$ L Posm by the same negative bias ( $-6.9$  mmol/kg) but not different from 250  $\mu$ L Posm (0.1 mmol/kg). The differences between 20 and 250  $\mu$ L samples of Protinol™ were significantly higher than Clinitol™.

**Conclusions:** These results demonstrate that Posm measured by freezing point depression will be  $\sim 7$  mmol/kg higher when using 20  $\mu$ L vs 250  $\mu$ L sample volumes. Approximately half of this effect may be due to plasma proteins. Posm sample volume should be carefully considered when calculating the osmole gap or assessing hydration status.

**KEYWORDS**

calculated osmolality, freezing point depression, measured osmolality, osmole gap

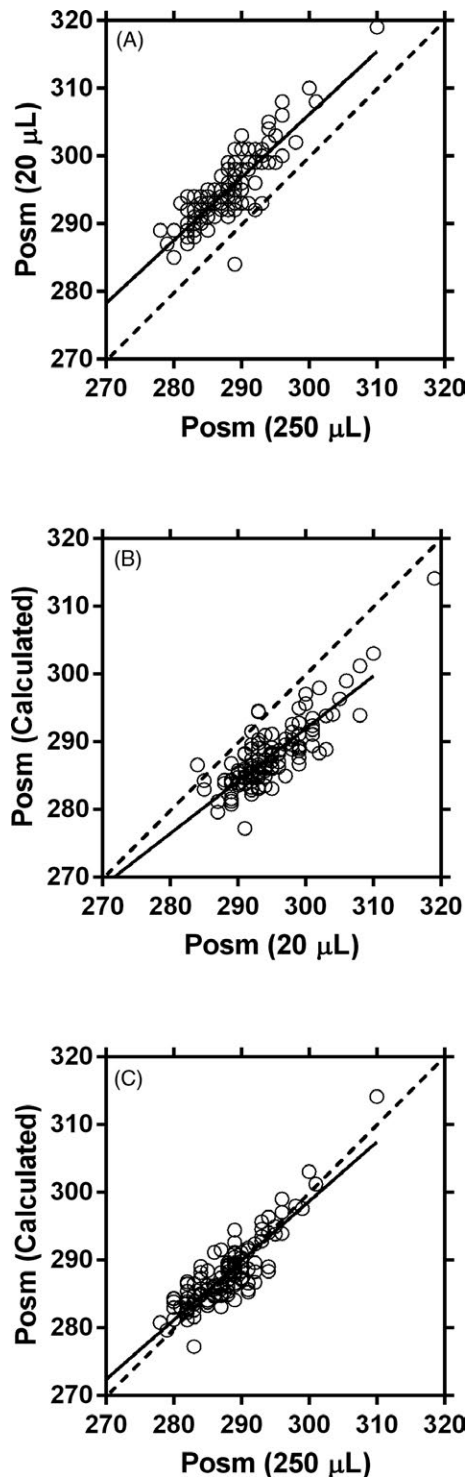
## 1 | INTRODUCTION

The measurement of plasma osmolality (Posm) is used in clinical settings to assess body fluid balance disturbance.<sup>1-8</sup> Moreover, Posm is used for the calculation of the osmole gap (measured – calculated) to help elucidate the presence of undiagnosed substances or metabolic disorders within patient populations.<sup>9,10</sup> In the field of sports medicine, Posm is routinely used

for hydration assessment and has demonstrated superiority over other measures when specifically assessing exercise-induced dehydration (ie, hypertonic-hypovolemia).<sup>11,12</sup> Important details regarding Posm measurement techniques have been reported,<sup>13-15</sup> and specifically, investigations have observed that timing of fluid,<sup>16</sup> method of blood draw,<sup>17</sup> and processing of blood (ie, whole blood vs plasma)<sup>18</sup> can alter the values obtained and consequently the diagnostic interpretation.<sup>19</sup>

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**FIGURE 1** Relationship between plasma osmolality (Posm; mmol/kg) samples when using a 250  $\mu\text{L}$  sample (X-axis) and a 20  $\mu\text{L}$  sample (Y-axis) (A), 20  $\mu\text{L}$  sample (X-axis) and the calculated plasma osmolality (mmol/L; Y-axis) (B), and 250  $\mu\text{L}$  sample (X-axis) and the calculated plasma osmolality (mmol/L; Y-axis) (C). The dashed line represents the line of identity where  $Y = X$ . The solid line represents the line of best fit

Previous work from our laboratory demonstrated that sample volume may impact Posm values.<sup>18</sup> Our previous work and other investigations examining the impact of sample volume on Posm are

not without limitations, as studies were comprised of limited data sets and/or narrow Posm ranges.<sup>17,20</sup> Using freezing point depression osmometry, Posm sample volumes range in the literature from 10 to 250  $\mu\text{L}$ .<sup>17,18</sup> Posm measurement via micro-osmometers has improved the use of osmolality measures in situations where limited biological sample volumes can be obtained.<sup>21,22</sup>

To our knowledge, no prior investigations have systematically examined the independent impact of sample volume, through a significant sample volume and Posm range. Given the widespread use of Posm in research and clinical settings,<sup>23</sup> the potential impact of sample volume on Posm measurement could be an important consideration. Small changes in Posm values can lead to differing diagnoses in both clinical and sports medicine settings.<sup>19,24-26</sup> Thus, knowledge of how sample volume may impact Posm interpretations is clearly warranted. Therefore, the purpose of the investigation was to determine whether sample volume independently impacts Posm values across a range representing euhydrated and dehydrated values.<sup>19</sup> Furthermore, we sought to examine the role of biological fluid complexity in explaining our observations. Our hypothesis, based on previous work,<sup>18</sup> was that osmometry using small sample volumes would result in meaningfully higher Posm measurements ( $>4$  mmol/kg)<sup>19</sup> and that sample fluid complexity would help explain the phenomenon.

## 2 | MATERIALS AND METHODS

Thirty-two physically active soldier and civilian volunteers took part in this study (25 males, 7 females; mean  $\pm$  SD age,  $24 \pm 8$  years; body mass (BM),  $81.85 \pm 17.66$  kg; height,  $177 \pm 9$  cm). All volunteers had passed the Army Physical Fitness Test (or equivalent fitness) within the previous 6 months and received a general medical clearance prior to participation; thus, all volunteers were considered physically fit and healthy. The use of alcohol, dietary supplements, and any medication other than an oral contraceptive was prohibited. Volunteers were provided informational briefings and gave voluntary, informed written consent to participate. Investigators adhered to AR 70-25 and US Army Medical Research and Materiel Command Regulation 70-25 on the use of volunteers in research. The US Army Research Institute of Environmental Medicine Human Use Review Committee approved this study.

Volunteer participation ranged between 1 and 8 days of experimental testing (1 day,  $n = 2$ ; 3 days,  $n = 4$ ; 4 days,  $n = 24$ ; 8 days,  $n = 2$ ) and provided a total of 126 independent blood samples for analysis. Volunteers arrived to the laboratory at 06:30 after  $\geq 8$  hour overnight fast and having consumed  $\geq 3$  L of water in the previous 24 hours. All volunteers were considered normally hydrated (euhydrated). After 20 minutes of controlled posture (sitting) in the laboratory, a venous blood sample was drawn without stasis from an antecubital vein. A subset of 11 volunteers underwent  $\leq 3$  hours of exercise-heat stress with fluid restriction<sup>19</sup> until they lost  $\sim 3.0$  L of body water, which resulted in dehydration by  $3.9 \pm 0.6\%$  of body mass. A second blood sample was then taken as described above

**TABLE 1** Linear regression parameters and statistical results

y	x	m	b	Sy	Bias	MAE	CCC
Posm 20 $\mu$ L	Posm 200 $\mu$ L	0.929	27.29	2.74	-7.0	7.1	0.439
Posm calc	Posm 20 $\mu$ L	0.775	59.49	2.72	6.9	6.9	0.436
Posm calc	Posm 200 $\mu$ L	0.877	35.57	2.42	0.1	1.9	0.876

$n = 126$  paired comparisons:  $m =$  slope,  $b =$  y-intercept,  $S_y =$  standard error of the estimate, bias = difference ( $x - y$ ), MAE = mean absolute error, CCC = concordance correlation coefficient.

using controlled posture, etc. The goal was to increase Posm by  $\sim 10$  mmol/kg.<sup>26</sup>

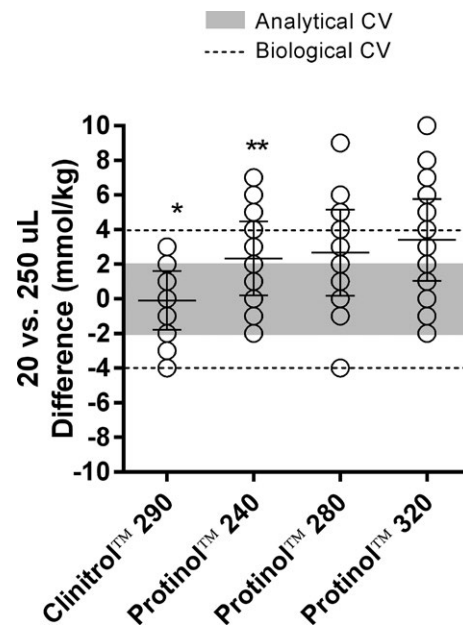
Blood samples were collected into 2.7 mL lithium-heparin tubes (Sarstedt Monovette®, Newton, NC). Samples were centrifuged at  $1500 \times g$  for 15 minutes at  $5^\circ\text{C}$  to acquire plasma and immediately transferred to sample cuvettes for osmolality determination. Note, plasma samples were vortexed prior to analysis. Using the same subjects' plasma, osmometry was performed on two separate freezing point depression devices allowing for direct comparison of sample volume. A 20  $\mu$ L aliquot sample of plasma was performed in triplicate via freezing point depression using a micro-osmometer (Fiske® Micro-osmometer, Model 210; Norwood, MA). Concurrently, samples were measured in triplicate on a larger 250  $\mu$ L single sample osmometer (Advanced® Model 3250; Norwood, MA). When the triplicate intrasample measures differed by  $\leq 3$  mmol/kg ( $\sim 1\%$ ), the median value was used. If the triplicate intrasample measures differed by  $> 3$  mmol/kg, two additional samples were measured and the median value was used.<sup>14</sup> Both osmometers were calibrated before any plasma samples were run using Clinitol™ 290 and Protinol™ 240, 280, and 320 reference solutions (Advanced Instruments, Norwood, MA).

Plasma samples (150  $\mu$ L) were also analyzed for sodium and blood urea nitrogen by direct ion selective electrode (ISE), and glucose by enzymatic determination, all being done using a Stat Profile Critical Care Xpress (Nova Biomedical, Waltham, MA). In an effort to minimize user complexity wherever possible, plasma osmolality was calculated from the simplest of the most accurate equations available: sodium, glucose, and blood urea nitrogen (BUN) ( $2 \times$  sodium [mmol/L] + glucose [mmol/L] + BUN [mmol/L]).<sup>27,28</sup>

Sample volume comparisons were made using ordinary least squares regression (OLS), whereby proportional (slope), constant (y-intercept), and random (SEE) errors were assessed in accordance with Westgard and Hunt.<sup>29</sup> Bias and mean absolute error were also compared using conventional calculations. Lin's concordance correlation coefficient<sup>30</sup> was also computed by hand to uniquely examine agreement between sample volumes with respect to the line of identity (concordance line, perfect agreement), rather than the line of best fit (OLS). The difference between sample volumes using reference standards was compared using analysis of variance and Tukey's post hoc procedure with a conventional  $P < 0.05$  threshold. Differences of magnitude  $> 4$  mmol/kg were considered of practical importance, a priori.<sup>18,19</sup> Smaller differences that were statistically significant were considered for additional explanatory power of the hypotheses tested. All statistical analyses were completed with the use of computerized statistical software packages (GraphPad Prism® version 7 for Windows; Microsoft Excel, Microsoft Corporation, Redmond, WA).

### 3 | RESULTS

A total of 126 paired plasma samples were compared. Figure 1A compares 20 and 250  $\mu$ L samples. The 20  $\mu$ L samples of plasma showed a 7 mmol/kg positive bias compared to 250  $\mu$ L samples and displayed a nearly constant proportional error across the range tested (slope = 0.929). Bias and the MAE were nearly identical because errors were uniformly positive. Calculated osmolality, when compared to 20  $\mu$ L Posm, displayed negative bias ( $-6.9$  mmol/kg; Figure 1B; Table 1) that was also uniform (MAE = 6.9 mmol/kg). However, calculated values were not different from 250  $\mu$ L Posm (0.1 mmol/kg; Figure 1C; Table 1), and the MAE was also small (1.9 mmol/kg). The random error about the best fit line was small in all comparisons (2.4-2.7 mmol/kg), but the concordance correlation coefficient (Table 1) showed substantial agreement<sup>31</sup> with respect to the line of identity for the comparison in Figure 1C only.



**FIGURE 2** Differences in osmolality (mmol/kg) between 20 and 250  $\mu$ L standards of Clinitol™ 290 ( $n = 138$  pairs) and Protinol™ 240 ( $n = 48$  pairs), 280 (42 pairs), and 320 (54 pairs). Shaded region represents the typical analytical variation; dashed lines represent the typical day-to-day biological variation.<sup>19</sup> Lines with error bars represent mean  $\pm$  SD. Identical differences appear as single overlapping values within a column. \*Significantly different from all Protinol™ trials ( $P < 0.05$ ); \*\*significantly different from Protinol™ 320

A total of 282 pairs of standard reference solution comparisons were made. The differences between 20 and 250  $\mu\text{L}$  samples of Protinol™ (240, 280, and 320 mmol) were significantly higher than Clinitol™ 290 (Figure 2). In addition, Protinol™ 320 was significantly higher than Protinol™ 240. Mean differences for all Protinol™ standards were higher than the typical analytical variation for Clinitol™ standards. While the mean difference fell below the typical day-to-day biological variation for Posm,<sup>19</sup> the trend was for differences to progressively increase as the complexity of dissolved substances increased (Figure 2). For example, 59% of the Clinitol™ samples showed a positive bias when examining 20–250  $\mu\text{L}$  volumes. For Protinol™, positive bias was 88% (240), 90% (280), and 96% (320).

## 4 | DISCUSSION

The primary purpose of this study was to examine the potential impact of sample volume on Posm measures. Furthermore, we sought to examine whether sample volume would influence calculated osmolality and the corresponding osmole gap assessment. Our primary conclusion is that 20  $\mu\text{L}$  Posm measures are both significant and importantly different from 250  $\mu\text{L}$  Posm measures. Approximately half of this effect may be due to plasma proteins. Furthermore, our results suggest that large sample volumes (ie, 250  $\mu\text{L}$ ) improve Posm measures as reflected by the proximity to calculated Posm values (ie, reduced osmole gap).

In our previous investigation, we conducted preliminary work to assess whether sample volume (20  $\mu\text{L}$  vs 250  $\mu\text{L}$ ) independently impacts the Posm values obtained.<sup>18</sup> In a limited sample size ( $n = 10$ ), those results demonstrated that the difference in Posm between sample volumes was  $4 \pm 2$  mmol/kg ( $P < 0.01$ ; 20  $\mu\text{L}$  samples being higher), but differences  $\leq 4$  mmol/kg were considered marginal.<sup>19</sup> Other investigations have similarly compared the impact of sample volume, but have also been limited in sample size, comparing a single Posm measure in 30 emergency department patients using 100  $\mu\text{L}$  vs 200  $\mu\text{L}$  (2-fold) sample volumes<sup>20</sup> or examining the difference between 10  $\mu\text{L}$  vs 50  $\mu\text{L}$  (5-fold) sample volumes in 24 euhydrated subjects.<sup>17</sup>

The current investigation sought to further expand our previous results by investigating a larger number of observations ( $>100$  measurements) through an increased range of Posm values tested (ie, euhydrated through dehydrated values) and using two common sample volumes that differ in range by more than 10-fold. The results of the present investigation unequivocally demonstrate that the sample volume size, 20  $\mu\text{L}$  volumes compared to 250  $\mu\text{L}$ , independently produces a significant and meaningful  $\sim 7$  mmol/kg higher value.

The sports medicine field uses Posm to assess hydration status with a commonly cited dehydration threshold value of 290 mmol/kg.<sup>11,25</sup> The results of our current investigation demonstrate that sample volume can dramatically alter hydration status classification. For example, a value of 290 mmol/kg (euhydrated) when obtained using a 250  $\mu\text{L}$  volume would concurrently read 297 mmol/kg (dehydrated) from a 20  $\mu\text{L}$ . Therefore, if investigators are using

20  $\mu\text{L}$  samples, our results suggest that Posm values would have to be  $<283$  mmol/kg to be considered euhydrated. Importantly, if dehydration is being monitored over time (ie, dynamic state measurement) while using reference change values (RCVs),<sup>26,32</sup> smaller sample volumes would still be appropriate since our results demonstrate that the error is constant and linear from the euhydrated through dehydrated Posm range.

In clinical settings, Posm measurements are used for the calculation of the osmole gap.<sup>33</sup> This assessment allows clinicians the ability to detect various toxic substance ingestion or abnormal metabolic states within a patient.<sup>28,34</sup> Typically, a gap of  $>10$  mmol warrants concern.<sup>9,10</sup> Previously, we determined that only 5 of 36 commonly used equations, when tested on a healthy population, are most accurate due to the small calculated osmole gap (range: 0.7–4.5 mmol).<sup>27</sup> Our current results demonstrate that the 20  $\mu\text{L}$  sample volume results in a osmole gap of  $\sim 7$  mmol/kg compared to the 250  $\mu\text{L}$  osmole gap of  $<1$  mmol/kg when using one of the most accurate (closest to measured) equations.<sup>28,34</sup> Thus, if a user does not choose an accurate equation and measures osmolality using a small sample volume, the potential for compounded measurement and calculation errors could result in a significantly inflated osmole gap ( $>10$  mmol), potentially altering a clinical diagnosis. Our results demonstrate that sample volume is an important consideration when assessing the osmole gap, with larger volumes (ie, 250  $\mu\text{L}$ ) being more advantageous. Importantly, when replicates of larger sample volumes are desired (eg, research), planned blood volume sampling requirements must be adjusted accordingly.

We observed no significant osmolality differences between 20 and 250  $\mu\text{L}$  samples of Clinitol. However, 20  $\mu\text{L}$  Protinol samples were 2–4 mmol/kg higher than 250  $\mu\text{L}$  samples and contributed proportionally more with increasing protein content (Figure 2). This suggests a role for plasma proteins in contributing to approximately half the  $\sim 7$  mmol/kg differences observed between 20 and 250  $\mu\text{L}$  samples of plasma (Figure 1A). Blood plasma is a complex biological fluid containing various osmotically active components.<sup>35</sup> When the components of plasma are not taken into account when measuring osmolality, results can be easily confounded.<sup>36</sup> Previously, we demonstrated that whole blood osmolality was significantly different from plasma osmolality when using small sample volumes.<sup>18</sup> Specifically, when large sample volumes were used (250  $\mu\text{L}$  vs 20  $\mu\text{L}$ ), the differences between whole blood and plasma were dramatically reduced leading us to hypothesize that a volume-dependent physical phenomenon was taking place. Our current work further validates this hypothesis. Furthermore, we used incrementally different calibration standards of dissimilar compositions to demonstrate that solutions with higher concentrations and increased complexity (ie, the presence of proteins) produce larger discrepancies between small and large sample volumes (see Figure 2), thus validating our presumption that when using freezing point osmometry, small sample volumes are influenced more when complex fluids—especially those with proteins—are analyzed. It is important to note that these conclusions are valid for the freezing point depression methodology only and may not directly apply to vapor pressure osmometry. For

example, vapor pressure osmometry may not be affected by sample viscosity or suspended particles such as freezing point depression osmometry. However, vapor pressure osmometry is impacted by the presence of volatile compounds in the sample (eg, alcohols), which are of interest when assessing the osmole gap; thus, freezing point depression osmometry was used in this study because it is a historically preferred clinical methodology.<sup>36</sup>

## 5 | CONCLUSION

In conclusion, this investigation provides strong evidence that when Posm is measured using freezing point depression osmometry, larger sample volumes will enhance the accuracy of the measurement; however, the ultimate use of the Posm measure will dictate the sample volume needed. Our results demonstrate that whether clinicians or researchers are using Posm to assess static hydration state, larger volumes would be recommended. If a dynamic dehydration assessment is desired, and RCVs are utilized, sample volume has little impact. However, in clinical situations where the osmole gap is of concern, larger sample volumes should be utilized since small sample volumes consistently and meaningfully create artificially higher osmole gaps. Clinicians and researchers alike should carefully consider sample volume when making Posm measures since sample volume can independently impact the outcome measure.

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