Identification of a ligand binding hot spot and structural motifs replicating aspects of tyrosyl-DNA phosphodiesterase I (TDP1) phosphoryl recognition by crystallographic fragment cocktail screening

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ABSTRACT

Tyrosyl DNA-phosphodiesterase I (TDP1) repairs type IB topoisomerase (TOP1) cleavage complexes generated by TOP1 inhibitors commonly used as anticancer agents. TDP1 also removes DNA 3' end blocking lesions generated by chain-terminating nucleosides and alkylating agents, and base oxidation both in the nuclear and mitochondrial genomes. Combination therapy with TDP1 inhibitors is proposed to synergize with topoisomerase targeting drugs to enhance selectivity against cancer cells exhibiting deficiencies in parallel DNA repair pathways. A crystallographic fragment screening campaign against the catalytic domain of TDP1 was conducted to identify new lead compounds. Crystal structures revealed two fragments that bind to the TDP1 active site and exhibit inhibitory activity against TDP1. These fragments occupy a similar position in the TDP1 active site as seen in prior crystal structures of TDP1 with bound vanadate, a transition state mimic. Using structural insights into fragment binding, several fragment derivatives have been prepared and evaluated in biochemical assays. These results demonstrate that fragment-based methods can be a highly feasible approach toward the discovery of small-molecule chemical scaffolds to target TDP1, and for the first time, we provide co-crystal structures of small molecule inhibitors bound to TDP1, which could serve for the rational development of medicinal TDP1 inhibitors.

INTRODUCTION

A member of the phospholipase D superfamily, tyrosyl-DNA phosphodiesterase I (TDP1) (E.C. 3.1.4.1) was discovered as a DNA-repair enzyme that cleaves the phosphodiester bond between a tyrosine residue of type IB topoisomerase (TOP1) and the 3' phosphate of DNA that occurs in stalled TOP1-DNA cleavage complexes (TOP1cc) (1,2). Human cells encode six DNA topoisomerases that regulate DNA topology by transiently cleaving the DNA backbone to remove DNA supercoiling, unlinking postreplication catenanes and resolving DNA knots (3). The TOP1 active site tyrosine reversibly 'breaks' the phosphodiester linkage of one strand of the double-stranded DNA by a nucleophilic attack, resulting in a TOP1-DNA covalent reaction intermediate that links this Tyr to the 3' end of the broken strand via a phosphodiester moiety. Inhibitors of TOP1 that block the religation of TOP1cc by binding at the enzyme–DNA interface (4) and generate stalled TOP1cc are used as anticancer drugs in the clinic (3,5). TDP1 has been hypothesized to be a pharmacological target for the treatment of cancer (6-10).

TDP1 repairs DNA lesions that are created by the trapping of TOP1 following treatment by camptothecin and its derivatives such as topotecan and irinotecan as well as

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those that accumulate under other physiological conditions where TOP1 acts on DNA alterations. (2.11-13). It carries out these functions by hydrolyzing the covalent bond between the TOP1 catalytic tyrosine residue and the 3' end of the DNA in a two-step reaction (Figure 1) (2,8,13,14). First, it cleaves the 3' tyrosyl bond by forming a transient covalent intermediate between a conserved histidine (H263 for human TDP1) and the 3' phosphate end. The covalent TDP1-DNA bond is then released by hydrolysis, which requires a second histidine of TDP1 (H493 for human TDP1) (1,15,16) (Figure 1). Subsequently, polynucleotide kinase phosphatase (PNKP) removes the residual 3' phosphate and installs a phosphate on the 5' end at the opposing side of the break. At last, DNA ligase III reseals the DNA (12). The repair function of TDP1 is not just limited to TOP1 cleavage complexes. It also removes 3' phosphoglycolate caused by oxidative DNA damage and 3' blocking lesions generated by oxygen radicals and alkylating agents (12, 17, 18). It can also serve as a backup repair pathway for DNA lesions generated by the trapping of DNA topoisomerase II (TOP2 α and TOP2 β) on DNA (17,19–21). These observations highlight a potentially broad and important role for TDP1 in the maintenance of genomic stability. TDP1-dependent repair pathways are normally redundant with other DNA damaging response pathways, such as Mre11 and XPF (22,23) that are often compromised in cancer cells.

Checkpoint and repair deficiencies are common in many cancer cells and, in these cases, TDP1 becomes the main mechanism for removal of TOP1-mediated DNA damage (3,24). It has been proposed that combination chemotherapy with TDP1 inhibitors and topoisomerase-targeting drugs should work synergistically to enhance selectivity toward cancer cell lines with pre-existing deficiencies in parallel repair pathways such as XPF-ERCC1, giving rise to synthetic lethality (25,26). Based on the broad spectrum of TDP1 substrates, TDP1 inhibitors should not only synergize with TOP1-targeting agents such as irinotecan, topotecan and other camptothecin derivatives, but also with the indenoisoquinoline derivative non-camptothecin TOP1 inhibitors in clinical development (5) and with genetiabine, a drug that is used for the treatment of various cancers (27). Despite the potential therapeutic value of TDP1 as an anti-cancer drug target, there are no drugs currently in the clinic that specifically target TDP1 (7,10). Although, there are a limited number of TDP1 inhibitors that have been reported in the literature (7,10,14,28,29) and the first crystal structures of TDP1 were solved nearly 15 years ago (15,16), crystal structures of TDP1 bound to a small molecule inhibitor have yet to be reported. The determination of a crystal structure of TDP1 bound to a small molecule, therefore, could potentially facilitate structure-based drug design efforts focused on the development of novel TDP1 inhibitors.

Within the past decade, fragment-based drug discovery has grown into an effective and viable method for deriving new pharmacophores which have often progressed into lead compounds for a variety of challenging targets (30,31). Fragments found in many commercial screening libraries are typically low molecular weight compounds (<250 Da) that adhere to the Lipinski rule of three and are readily amenable to further chemical modification (32,33). Such fragments provide chemical scaffolds that are smaller and less complex than those typically found in high-throughput screening assays (34). In turn, this often leads to an increased probability that a fragment will bind to a target protein, resulting in higher hit rates and more efficient screening of diverse chemical space. A major advantage of fragmentbased drug discovery is that it can identify lead compounds that possess high-ligand efficiency with the opportunity to fine-tune physiochemical properties during elaboration and optimization. Hits identified from fragment screens can also reveal unanticipated binding pockets or hot spots within a target protein (35,36). A wide array of biophysical techniques can be used to rapidly identify weak protein-binding fragments (37,38). X-ray crystallography and nuclear magnetic resonance (NMR) spectroscopy were among the first to be utilized in fragment-based drug discovery (38-40). Although still time-consuming and labor intensive, X-ray crystallography is a powerful primary method for screening fragment libraries that has noteworthy advantages over other biophysical screening methods (31,41). Many initial fragments tend to exhibit very weak binding affinities from the high micromolar to low millimolar range. X-ray crystallography provides the sensitivity needed to confirm the binding of such weak compounds and it concurrently reveals the structural information that is needed to immediately advance these fragments from hits to leads using structure-guided drug design (30).

In an attempt to identify novel chemicals that bind to and inhibit TDP1, we carried out a crystallographic fragment screening campaign using a library of \sim 640 fragment compounds. Two types of fragments were identified that bind to the active site pocket and members from both series exhibit inhibitory activity against TDP1. Herein, for the first time we provide the high-resolution crystal structures of TDP1 bound to small molecules. These may provide structural insights for structure-guided drug design efforts to identify and optimize inhibitors of TDP1.

MATERIALS AND METHODS

Cloning and expression

A derivative of the Gateway[™] vector pDONR221 containing the open reading frame (ORF) encoding the human TDP1 catalytic domain (residues S148-S608) preceded by a tobacco etch virus (TEV) protease recognition site was provided by Dr Tinoush Moulaei, Protein Structure Section, Macromolecular Crystallography Laboratory, National Cancer Institute. The ORF was moved by recombinational cloning into the destination vector pDEST-527 to produce pDN2454. This plasmid directs the expression of human TDP1 (S148-S608) with an Nterminal hexahistidine tag and an intervening TEV protease cleavage/recognition site (ENLYFQ/S148) (42). The fusion protein was expressed in the Escherichia coli strain BL21 Star[™] (DE3) (Invitrogen ThermoFisher Scientific). Cells containing pDN2454 were grown to mid-log phase $(OD_{600} \sim 0.5)$ at 37°C in Luria Bertani broth (Cellgro) containing 100 µg/ml ampicillin and 0.2% glucose. Overproduction of fusion protein was induced with isopropyl-β-Dthiogalactopyranoside at a final concentration of 1 mM for 18–20 h at 18°C. The cells were pelleted by centrifugation and stored at -80° C.

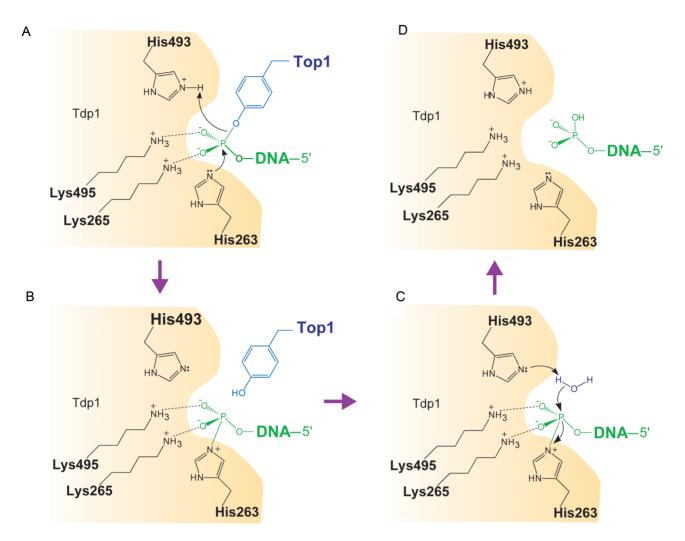


Figure 1. Schematic representation of TDP1 catalytic mechanism. Residue numbers refer to human TDP1. Curved arrows denote the transfer of electron pairs. (A) The imidazole N2 atom of H263 nucleophilically attacks the phosphate of the phosphotyrosyl group (B) Phosphohistidine covalent intermediate. (C) Hydrolysis of the phosphohistidine intermediate by H493-activated water molecule. (D) Generation of a final 3' phosphate product and free TDP1.

Purification

All procedures were performed at 4-8°C. A total of 20-25 g of E. coli cell paste expressing human TDP1 (S148-S608) with an N-terminal hexahistidine tag were suspended in 250 ml ice-cold buffer A (50 mM Tris-HCl, 500 mM NaCl, 25 mM imidazole, pH 7.5) containing 5 mM benzamidine HCl (Sigma Aldrich Corporation) and Complete[™] (ethylenediaminetetraacetic acid) EDTA-free protease inhibitor cocktail (Roche Diagnostics). The cells were lysed with an APV-1000 homogenizer (Invensys APV Products) at 10 000 psi and centrifuged at 30 000 \times g for 30 min. The supernatant was filtered through a 0.45 µm polyethersulfone membrane and applied to a HisTrap FF column (GE Healthcare Life Sciences) equilibrated in buffer A. The column was washed to baseline with buffer A and eluted with a linear gradient of imidazole to 1 M. Fractions containing the His₆-TDP1 fusion protein were pooled, concentrated using an Ultracel[®] 30 kDa ultrafiltration disc (EMD Millipore Corporation), diluted with 50 mM Tris-HCl, 500 mM NaCl, pH 7.5 buffer to reduce the imidazole concentration to about 25 mM, and digested overnight at 4°C with

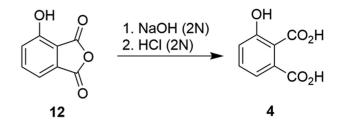
His₇-tagged TEV protease (43). The digest was applied to the HisTrap FF column equilibrated in buffer A and the TDP1 emerged in the column effluent. The effluent was incubated overnight at 4°C with 10 mM dithiothreitol, concentrated using an Ultracel[®] 10 kDa ultrafiltration disc, and applied to a HiPrep 26/60 Sephacryl S-300 HR column (GE Healthcare Life Sciences) equilibrated in 25 mM Tris-HCl, 500 mM NaCl, 2 mM tris(2-carboxyethyl) phosphine, pH 7.5 buffer. Fractions containing TDP1 (S148-S608) were pooled, concentrated as above, diluted with 50 mM Tris-HCl pH 7.5 buffer to reduce the NaCl concentration to 25 mM and applied to a HiTrap Q HP column (GE Healthcare Life Sciences) equilibrated in 50 mM Tris-HCl, 25 mM NaCl, pH 7.5 buffer. Recombinant protein that emerged in the column effluent was pooled and applied to a HiTrap SP HP column (GE Healthcare Life Sciences) equilibrated in 50 mM HEPES, 25 mM NaCl, pH 7.2 buffer. The column was washed to baseline with equilibration buffer and eluted with a linear gradient of NaCl to 500 mM. Fractions containing pure TDP1 were pooled, concentrated as above and applied to the HiPrep 26/60 Sephacryl S-300 HR

column equilibrated in 25 mM Tris–HCl, 150 mM NaCl, 2 mM tris(2-carboxyethyl) phosphine, pH 7.2 buffer. The peak fractions containing recombinant TDP1 (S148–S608) were pooled and concentrated to 30–50 mg/ml (estimated at 280 nm using a molar extinction coefficient of 112 300 M^{-1} cm⁻¹ (derived using the Expasy ProtPram tool) (44). Aliquots were flash-frozen in liquid nitrogen and stored at -80° C. The final product was judged to be >95% pure by sodium dodecyl sulfate-polyacrylamide gel electrophoresis. The molecular weight was confirmed by electrospray ionization mass spectroscopy (LC/ESMS).

Crystallization

The catalytic domain of TDP1 (residues S148–S608) was screened for crystallization conditions using several sparsematrix crystallization screens from Hampton Research, Qiagen, Molecular Dimensions and Anatrace. The most promising conditions were obtained using the Morpheus crystallization screen (45). Crystallization plates were prepared using a Gryphon crystallization robot (Art Robbins Instruments). Diffraction-quality crystals used for crystallographic fragment screening were obtained from drops consisting of 2 µl TDP1 (22 mg/ml) with 2 µl well solution consisting of 0.1 M MOPS/HEPES-Na pH 7.5, 10% (w/v) PEG 8000, 20% (v/v) ethylene glycol, 0.03 M sodium fluoride, 0.03 M sodium bromide, 0.03 M sodium iodide and sealed over 500 µl well solution in a Nextal 15-well crystallization plate (Qiagen). Crystallographic fragment screening was carried out using the Zenobia Fragment Library 1 and Zenobia Fragment Library 2.2 (Zenobia Fragments). In preparation for crystallographic screening, cocktail mixtures were prepared by mixing four fragments together (single fragment stock, 200 mM, final concentration 50 mM) in dimethyl sulfoxide (DMSO). A 4-µl soaking drop was then prepared by adding stock concentration of the cocktail mixture to well solution to achieve a final concentration of 5 mM for each fragment (10% (v/v) DMSO). TDP1 crystals were then transferred to the soaking drop and soaked for a period of 24 h. The crystals were retrieved from the drop using a litholoop (Microlytic) and immediately flashcooled by plunging into liquid nitrogen without the need of further cryoprotectant. For structural studies of TDP1 in complex with fragment derivatives (Table 1), crystals of TDP1 were transferred to well solution consisting of compound in DMSO (concentration listed in Table 2) and 0.1 M MOPS/HEPES-Na pH 7.5, 10% (w/v) PEG 8000, 20% (v/v) ethylene glycol, 0.03 M sodium fluoride, 0.03 M sodium bromide and 0.03 M sodium iodide. The crystals were soaked for 24 h and retrieved from the drop using a litholoop and immediately flash-cooled by plunging into liquid nitrogen.

X-ray diffraction datasets were collected at beamlines 22-ID and 22-BM of the SER-CAT facility, Advanced Photon Source, Argonne National Laboratory. Diffraction datasets were processed using HKL3000 (46). For structure solution, molecular replacement was performed using the coordinates of a previously solved crystal structure of TDP1 (PDB code: 1JY1) (15) stripped of solvent molecules as a search model and searching for two molecules in the asymmetric unit using the program Phaser in the PHENIX suite (47).



Scheme 1. Synthesis of 3-hydroxyphthalic acid (4).

The resulting electron density maps were examined for difference electron density features (contoured at 3.0 σ level) to identify bound fragments. Fragment coordinates were prepared using the molinspiration server (www.molinspiration. com) and .cif files for the fragment coordinates were prepared using the eLBOW feature in PHENIX (48). Model rebuilding and water picking were carried out in COOT (49,50) and refined with phenix.refine (51). Model quality and validation were analyzed using MolProbity (52) (Table 2).

Recombinant TDP1 assay

A 5' ³²P-labeled single-stranded DNA oligonucleotide containing a 3' phosphotyrosine (N14Y) (10) was incubated at 1 nM with 10 pM recombinant full-length TDP1 in the absence or presence of inhibitor for 15 min at room temperature in the assay buffer containing 50 mM Tris-HCl, pH 7.5, 80 mM KCl, 2 mM EDTA, 1 mM dithiothreitol (DTT), $40 \,\mu\text{g/ml}$ bovine serum albumin and 0.01% Tween-20 (53). Reactions were terminated by the addition of 1 volume of gel loading buffer [99.5% (v/v) formamide, 5 mM EDTA. 0.01% (w/v) xylene cyanol and 0.01% (w/v) bromophenol blue]. Reaction products were separated on a 16% denaturing polyacrylamide gel electrophoresis (PAGE). Gels were dried and exposed to a PhosphorImager screen (GE Healthcare Life Sciences). Gel images were scanned using a Typhoon FLA 9500 (GE Healthcare Life Sciences), and densitometry analyses were performed using the ImageQuant software (GE Healthcare Life Sciences).

General synthetic procedures

Proton (¹H) and carbon (¹³C) NMR spectra were recorded on a 400 MHz spectrometer (Varian) or 500 MHz spectrometer (Varian) and are reported in ppm referenced to the solvent in which the spectra were collected. Solvent was removed by rotary evaporation under reduced pressure and anhydrous solvents were obtained commercially and used without further drying. Purification by silica gel chromatography was performed using Combiflash units (Teledyne) with EtOAc-hexanes solvent systems. Preparative high-pressure liquid chromatography (HPLC) was conducted using a Waters 2535 Quaternary Gradient Module having a Waters 2998 photodiode array detector (Waters) and a C₁₈ column (Phenomenex, Cat. No. 00G-4436-P0-AX, 250 mm × 21.2 mm 10 µm particle size, 110 Å pore). Binary solvent systems consisting of A = 0.1% aqueous trifluoroacetic acid (TFA) and B =0.1% TFA in acetonitrile were employed with gradients

Table 1. Compounds discussed in the text

Comp.	Name	STRUCTURE	TDP1 IC ₅₀
1	benzene-1,2,4-tricarboxylic acid	HO O OH HO O OH	0.74 mM ^a
2	4-oxo-1,4-dihydroquinoline-3- carboxylic acid	ООН	7.2 mM
3	4-hydroxyphthalic acid		1.9 mM
4	3-hydroxyphthalic acid	но он	3.1 mM
5	4-aminophthalic acid		80 µM
6	8-isopropyl-4-oxo-1,4- dihydroquinoline-3-carboxylic acid	O N H H	N/A ^b
7	4-oxo-1,4-dihydroquinoline- 3,8-dicarboxylic acid	но он	N/A
8	8-bromo-4-oxo-1,4- dihydroquinoline-3-carboxylic acid		N/A
9	8-nitro-4-oxo-1,4- dihydroquinoline-3-carboxylic acid		15% at 1 mM°
10	4-(3- carboxypropanamido)phthalic acid	но н	1.01 mM
11	4-(4- carboxybutanamido)phthalic acid	HO J J HO J OH OH	0.96 mM

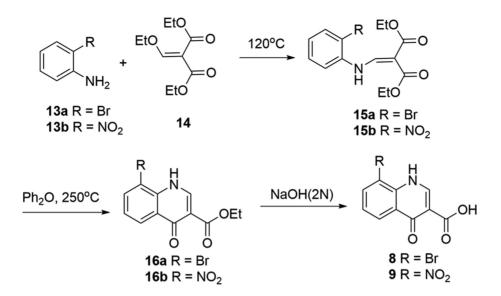
^aCompound 1 inhibited TDP1 in non dose-dependent manner, a midpoint between highest inactive and lowest active concentrations was chjosen as IC_{50} . ^bN/A, not active up to the highest tested concentration of 1 mM. ^cCompound 9 inhibited 15% of TDP1 activity at highest tested concentration of 1 mM.

	TDP1– Compound 1 (cocktail soak)	TDP1– Compound 1 (single soak)	TDP1– Compound 2 (cocktail soak)	TDP1– Compoiund 2 (single soak)	TDP1- Compound 3	TDP1- Compound 4	TDP1- Compound 5	TDP1- Compound 6	TDP1- Compound 7	TDP1- Compound 8	TDP1- Compound 9	TDP1- Compound 10	TDP1- Compound 11
Fragment concentration (mM)	5	20	5	20	20	5.5	17	20	22.5	4.5	13.6	6.6	16.4
Diffraction Source	SER-CAT 22-ID	SER-CAT 22-BM	SER-CAT 22-BM	SER-CAT 22-BM	SER-CAT 22-ID	SER-CAT 22-BM	SER-CAT 22-BM	SER-CAT 22-ID	SER-CAT 22-ID	SER-CAT 22-ID	SER-CAT 22-ID	SER-CAT	SER-CAT
Wavelength (Å)	1.0000	1.0000	1.0000	1.0000	1.0000	1.00000	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000	1.0000
Space group a- $b = c = (Å)$	$P2_{I}2_{I}2_{I}$ 50.05 105 26	$P2_I2_I2_I$ 50.09 105.63	$P2_{I}2_{I}2_{I}$ 50.11.105.64	$P2_{I}2_{I}2_{I}$ 50 12 105 41	$P2_{I}2_{I}2_{I}$ 49 91 105 59	$P2_{I}2_{I}2_{I}$ 49 77 104 99	$P2_I2_I2_I$ 49 94 105 14	$P2_{I}2_{I}2_{I}2_{I}$ 50 06 105 37	$P2_{I}2_{I}2_{I}2_{I}$ 50 16 105 31	$P2_{I}2_{I}2_{I}2_{I}$ 50.07,105.14	$P2_I2_I2_I$ 50.16.105.59	$P2_{I}2_{I}2_{I}$ 50.05 105 41	$P2_{I}2_{I}2_{I}$ 49 96 105 06
u, o — , c — (m)	193.91	194.03	194.95	194.56	194.19	194.11	193.37	194.58	193.98	194.45	193.98	194.1	193.32
$\alpha =, \beta =, \gamma = (^{\circ})$ Data Collection Statistics	90,90,90	90, 90, 90	90, 90, 90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90	90,90,90
Resolution Range (Å)	50-1.63	50-1.78	50-1.81	50-1.78	50-1.78	50-1.75	5-1.74	50-1.70	50-1.67	50-1.91	50-1.85	50-1.62	50-1.50
	$(1.66-1.63)^{a}$	(1.81-1.78)	(1.84 - 1.81)	(1.81 - 1.78)	(1.81 - 1.78)	(1.78 - 1.75)	(1.77 - 1.74)	(1.73 - 1.70)	(1.70 - 1.67)	$(1.94 - 1.91)^{a}$	(1.88 - 1.85)	(1.65 - 1.62)	(1.53 - 1.50)
Total No. of reflection	876 873	632 682	395 972	492 373	1 443 515	630 560	536 200	781 610	841 105	531 475	544 340	908 312	1 022 403
Total No. of unique reflections	128 455	99 895	93 819	98 896	100 750	105 159	105 222	112 529	120 178	78 029	83 875	131 858	157 054
Completeness (%)	99 6 (98 4)	100 (100)	97 8 (98 5)	(2 86) 2)	0017666	99 6 (94 3)	00 0 (07 8)	(1 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2 2	00 6 (99 3)	98 3 (96 5)	95.0.(93.8)	100 (99 8)	96 2 (92 9)
CC1/2 ^b	0.754	0.769	0.622	0.733	0.891	0.925	0.956	0.790	0.870	0.754	0.817	0.760	0.802
Multiplicity	6.8 (6.1)	6.3 (6.1)	4.2 (3.8)	5.0(4.9)	14.3 (14.0)	6.0(5.1)	5.1 (4.4)	6.9 (5.7)	7.0 (6.5)	6.8 (6.6)	6.5(6.0)	6.9 (6.1)	6.5 (5.7)
$I/(\sigma) I$	38.2 (2.3)	32.7 (3.1)	29.2 (2.3)	29.3 (2.3)	48.3 (4.5)	41.7 (4.9)	39.8 (6.7)	35.4 (3.0)	37.8 (3.5)	28.4 (2.6)	25.7 (2.5)	28.1 (2.2)	37.2 (2.0)
$R_{\rm sym}$	0.062 (0.711)	0.078 (0.918)	0.071 (0.882)	0.071 (0.561)	0.117 (0.780)	0.051 (0.327)	0.050 (0.215)	0.081 (0.811)	0.072 (0.771)	0.088 (0.957)	0.104(0.801)	0.071 (0.046)	0.046 (0.806)
R pim	0.025 (0.311)	0.034(0.405)	0.036(0.501)	0.035(0.289)	0.032(0.219)	0.023 (0.157)	0.024 (0.115)	0.033 (0.352)	0.030 (0.316)	0.036 (0.343)	0.043 (0.342	0.027 (0.349)	0.023(0.353)
Kennement Statistics	25 (1 1 12	02 1 00 20	10 00 1 01	02 1 01 20	EE 1 10 01	30 1 01 70	101000	25 (0 1 20	E7 1 0E 30	101 00 22	20 1 01 05	071 0 20	021 020
Kesolution Kange (A)	0.1-00.00	5/.U&-1./8	40.99-1.61	5/.12-1./8 0.196	40.91-1.70	C/ .1-71.05	50.90-1.74 0.174	0/.1-00.00	0.1-0/.05	14.1-00.76	02.1-24.05	20.1-1.02 0.100	00.1-0.00
Final K _{work} Einel D	10110	6/T/0	0.100	0.180	0.176	0.181	0.1/4	161.0	0.100	8/1.0 8/C 0	0.1.0 0.118	0.100	0.100
No. of non H-atoms/Average B factor (Å ²)	v.211 rage B factor (Å ²)	017-0	177.0	0.441	0.210	117.0	0.02.00	177.0	0.412	077.0	0.410	707.0	107.0
Protein chain A	3640/38.5	3599/26.1	3605/28.6	3612/25.7	3627/24.8	3633/22.7	3625/19.0	3609/24.7	3616/23.9	3616/26.5	3626/23.7	3623/23.7	3654/24.8
Protein chain B	3619/30.3	3601/31.5	3610/34.9	3582/32.1	3596/29.1	3619/25.9	3621/23.0	3587/32.0	3616/30.2	3601/33.9	3591/30.6	3618/28.3	3606/28.5
Water	695/38.9	676/38.7	582/39.5	727/38.8	711/38.2	779/34.7	886/33.4	648/38.6	619/37.6	586/38.9	580/36.9	842/38.9	796/40.0
Ligand chain A	15/37.2	15/29.8	14/36.3	14/32.7	13/28.6	13/26.5	13/20.0	17/31.5	17/38.2	15/37.2	17/36.4	20/29.7	21/33.5
Ligand chain B	15/39.5	15/35.1	14/45.3	14/44.0	13/30.8	13/26.4	13/21.7	17/36.2	17/39.1	15/37.2	17/39.3	20/34.5	17/31.1
Ethylene glycol	40/37.9	16/31.7	24/37.7	16/29.5	12/26.0	24/28.6	28/27.6	16/30.6	16/30.8	12/29.9	12/32.5	20/32.6	12/30.4
R.m.s.d from ideal	200.0	200.0	1000	10000	0 007	200.0	200.0	200.0	200.0	100 0	200.0	200.0	1000
Bond lengins (A)	0.000	0.006	0.00/	0.00/	0.000	0.000	0.000	0.000	0.000	0.007	0.000	0.000	c00.0
Bond angles (~) Ramachandran nlot	0.8	0.8	0.8	0.8	0.8	0.8	0.8	0.8	0.8	6.0	0.8	0.8	0.8
Favored (%)	98.3	98.0	27.7	97.1	98.2	97.9	97.9	97.6	97.6	97.6	97.9	97.8	98.0
Allowed (%)	1.6	1.9	2.0	2.6	1.7	2.0	2.0	2.3	2.1	2.3	1.9	1.9	1.9
Disallowed (%)	0.1	0.1	0.3	0.3	0.1	0.1	0.1	0.1	0.3	0.1	0.2	0.3	0.1
Molprobity score	1.08 (99th	0.98 (100th	1.06 (100th	1.17 (99th	1.05 (100th	1.12 (99th	1.06 (100th	1.24 (98th	1.19 (99th	1.23 (99th	2.09 (100th	3.7 (97th	2.5 (99th
	percentile)	percentile)	percentile)	percentile)	percentile)	perecentile)	percentile)	percentile)	percentile)	percentile	percentile)	percentile)	percentile)
PDB code:	6DHU	6DIE	6DIM	(DJD	HIQ9	IfQ9	(DJJ	6DJE	6DJF	HID3	6MJ5	6N17	6119

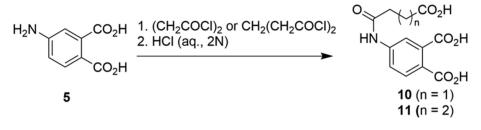
Table 2. X-ray diffraction data collection and refinement statistics

Nucleic Acids Research, 2019, Vol. 47, No. 19 10139

 $^{\rm a}$ Values in parenthesis are for the highest-resolution shell of data. $^{\rm b}$ Value is for the highest-resolution shell of data.



Scheme 2. Synthesis of 8-bromo-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8) and 8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (9) (55,59,60).



Scheme 3. Synthesis of 4-(3-carboxypropanamido)phthalic acid (10) and 4-(4-carboxybutanamido)phthalic acid (11).

as indicated. Products were obtained as amorphous solids following lyophilization. Electrospray ionization-mass spectrometry (ESI-MS) was performed with an Agilent LC/MSD system equipped with a multimode ion source (Agilent). The compounds, benzene-1,2,4-tricarboxylic acid (1), 4-oxo-1,4-dihydroquinoline-3-carboxylic acid (2), 4-hydroxyphthalic acid (3), 4-aminophthalic acid (5), 8-isopropyl-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (6), and 4-oxo-1,4-dihydroquinoline-3,8-dicarboxylic acid (7) were obtained from commercial sources. The compounds, 3-hydroxyphthalic acid (4) (54), 8-bromo-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (8) (55), and 4-oxo-8-sulfo-1,4-dihydroquinoline-3-carboxylic acid (9) (56), 4-(3-carboxypropanamido)pthalic acid (10) and 4-(4-carboxybutanamido)pthalic acid (11) were prepared according the published procedures with modifications as described below.

Preparation of 3-hydroxyphthalic acid (4)

In a variation of a previously reported procedure (54), commercially available 4-hydroxyisobenzofuran-1,3-dione (12, 169 mg, 1.0 mmol) was dissolved in aqueous NaOH (2N, 4.0 ml). The reaction mixture was stirred at room temperature (15 h) and then aqueous HCl (2N) was added to adjust the pH to 2. The resulting solution was purified by preparative HPLC using a linear gradient of 20% B to 50% B over 30 min at a flow rate of 10 ml/min; retention time = 17.4 min. Lyophilization yielded product **4** as a white amorphous solid (80 mg, 43% yield) (Scheme 1). ¹H NMR (400 MHz, DMSO- d_6) δ 10.21 (bs, 1H), 7.33–7.27 (m, 2H), 7.08 (dd, J = 7.4, 1.8 Hz, 1H). ¹³C NMR (101 MHz, DMSO- d_6) δ 168.97, 167.64, 155.12, 130.60, 130.32, 123.67, 120.26, 120.03. ESI-MS m/z: 183.0 (MH⁺).

Preparation of 8-bromo-4-oxo-1,4-dihydroquinoline-3carboxylic acid (8) and 8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylic acid (9).

General procedure A to prepare compounds 15 (a, b). The mixture of commercially available aniline (13a or 13b, 3.5 mmol) and diethyl 2-(ethoxymethylene)malonate (14, 3.5 mmol) was heated (120°C, 4 h). The reaction mixture was cooled to room temperature and purified by silica gel column chromatography. Compound diethyl 2-((phenylamino)methylene)malonate (15a or 15b) was afforded (Scheme 2).

General procedure B to prepare compounds 14 (a, b). Diethyl 2-((phenylamino)methylene)malonate (**15a** or **15b**, 0.6 mmol) was suspended in Dowtherm A (0.5 ml) (eutectic mixture of 26.5% diphenyl + 73.5% diphenyl oxide) and the reaction mixture was heated (250° C, 1 h). The resultant mixture was cooled to room temperature and filtered to afford ethyl 4-oxo-1,4-dihydroquinoline-3-carboxylate (**16a** or **16b**) (Scheme 2).

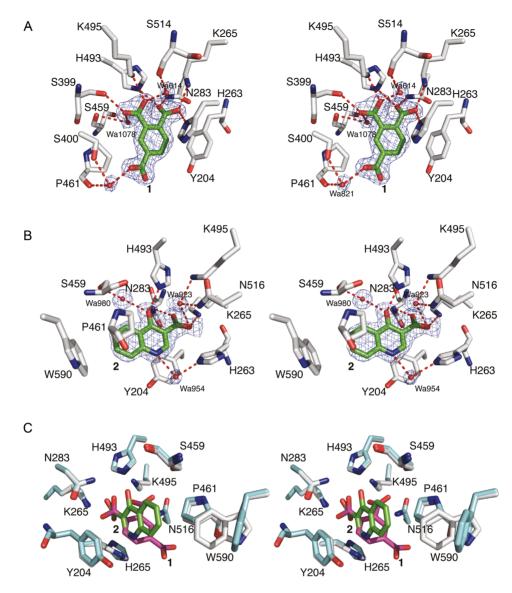


Figure 2. (A) Stereoview of the active site of TDP1 from the crystal structure of TDP1 complexed with compound 1 obtained from soaking of TDP1 crystals with ZT0911 alone (single soak, PDB code: 6DIE). TDP1 carbon atoms are colored in gray, nitrogen atoms in blue and oxygen atoms in red. For clarity, some main chain atoms have been removed from selected residues. The carbon atoms of compound 1 are shaded in green. Hydrogen bonds are highlighted as red dashes. The fit of compound 1 and bound waters is shown to the $2F_o$ - F_c electron density maps (blue) at 1.78 Å resolution and contoured at 1.0 σ level. (B) Stereoview of the active site of TDP1-compound 2 complex crystal structure at 1.78 Å obtained after soaking a TDP1 crystal with compound 2 (PDB code: 6DJD, carbon atoms in green). The fit of 2 and bound waters to the $2F_o$ - F_c map is shown at 1.0 σ level contour. (C) The superimposed coordinates of the active site residues from the crystal structure of the TDP1-compound 1 complex (TDP1 carbon atoms in cyan, fragment carbon atoms in magenta) and TDP1-compound 2 complex (TDP1 carbon atoms in green) illustrating the unique binding modes of the fragments.

General procedure C to prepare compounds 8 and 9. Ethyl 4-oxo-1,4-dihydroquinoline-3-carboxylate (**16a** or **16b**, 0.3 mmol) was suspended in sodium hydroxide (2.7 mmol) (1.0 N, 4.0 ml) and EtOH (0.6 ml). The mixture was stirred (70°C, 16 h). The reaction was cooled and acidified by HCl (aq. 2.0 N) and the resulting white solid was filtered and washed with H₂O. The solid was further purified by preparative HPLC to afford 4-oxo-1,4-dihydroquinoline-3-carboxylic acid (**8** or **9**) (Scheme 2).

Preparation of diethyl 2-(((2bromophenyl)amino)methylene)malonate (15a). The title compound was afforded as a white solid in 88% yield from 2-bromoaniline (**13a**) and diethyl 2-(ethoxymethylene)malonate (**14**) using General Procedure A. ¹H NMR (400 MHz, CDCl₃) δ 11.28 (d, J = 13.3 Hz, 1H), 8.52 (d, J = 13.3 Hz, 1H), 7.61 (dd, J = 8.0, 1.4 Hz, 1H), 7.40 – 7.35 (m, 1H), 7.31 (dd, J = 8.3, 1.5 Hz, 1H), 7.05 – 6.99 (m, 1H), 4.37 (q, J = 7.1 Hz, 2H), 4.28 (q, J = 7.1 Hz, 2H), 1.41 (t, J = 7.1 Hz, 3H), 1.35 (t, J = 7.1Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 168.36, 165.65, 150.53, 137.73, 133.47, 128.75, 125.40, 115.96, 113.71, 95.54, 60.62, 60.30, 14.41, 14.35. ESI-MS *m/z*: 342.0, 344.0 (MH⁺).

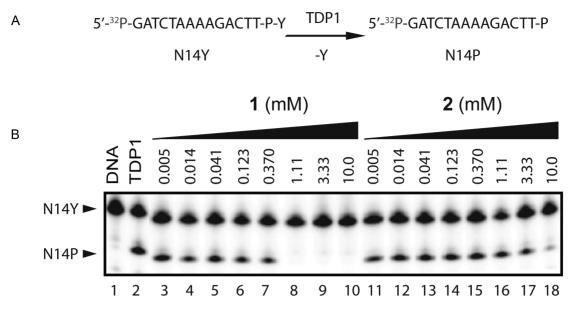


Figure 3. Inhibition of TDP1 by **1** and **2**. (A) Schematic representation 3' phosphotyrosyl DNA substrate (N14Y) conversion by TDP1 into cleaved products (N14P). (B) Representative gel for inhibition of TDP1-catalized hydrolysis by **1** and **2**: lane 1, N14Y only; lane 2, N14Y and TDP1; lanes 3–10 3-fold serial dilution of **1** from 10 mM to 5 μ M; lanes 11–18 3-fold serial dilution of **2** from 10 mM to 5 μ M.

Preparation 2-(((2of diethyl nitrophenyl)amino)methylene)malonate (15b). The title compound was afforded as a yellow solid in 81% yield from 2-nitroaniline and diethyl 2-(ethoxymethylene)malonate (14) using General Procedure A. ¹H NMR (400 MHz, CDCl₃) δ 12.60 (d, J = 13.0 Hz, 1H), 8.55 (dd, J = 13.0, 0.6 Hz, 1H), 8.28 (dd, J = 8.4, 1.5 Hz, 1H), 7.73 – 7.68 (m, 1H), 7.53 - 7.51 (m, 1H), 7.23 (ddd, J = 8.5, 7.2, 1.2Hz, 1H), 4.42 (q, J = 7.1 Hz, 2H), 4.30 (q, J = 7.1 Hz, 2H), 1.41 (t, J = 7.1 Hz, 3H), 1.37 (t, J = 7.1 Hz, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 166.74, 165.58, 148.28, 137.09, 136.08, 135.85, 126.81, 123.32, 116.73, 99.45, 60.96, 60.72, 14.36, 14.34. ESI-MS m/z: 309.1 (MH⁺), 639.2 (M₂Na⁺).

Preparation of ethyl 8-bromo-4-oxo-1,4-dihydroquinoline-3-carboxylate (16a). The title compound was afforded as a brown solid in 21% yield from diethyl 2-(((2bromophenyl)amino)methylene)malonate (**15a**) using General Procedure B. ¹H NMR (500 MHz, DMSO-*d*₆) δ 11.63 (s, 1H), 8.45 (s, 1H), 8.17 (dd, J = 8.1, 1.4 Hz, 1H), 8.04 (dd, J = 7.7, 1.4 Hz, 1H), 7.36 (t, J = 7.9 Hz, 1H), 4.23 (q, J =7.1 Hz, 2H), 1.28 (t, J = 7.1 Hz, 3H). ¹³C NMR (126 MHz, DMSO-*d*₆) δ 173.26, 164.76, 146.05, 137.10, 136.41, 129.29, 126.11, 126.06, 112.24, 110.75, 60.31, 14.74. ESI-MS *m/z*: 296.0, 298.0 (MH⁺).

Preparation of ethyl 8-nitro-4-oxo-1,4-dihydroquinoline-3carboxylate (16b). The title compound was afforded as a brown solid in 13% yield from diethyl 2-(((2nitrophenyl)amino)methylene)malonate (**15b**) using General Procedure B. ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.18 (s, 1H), 8.65 (dd, J = 8.1, 1.6 Hz, 1H), 8.59–8.57 (m, 2H), 7.60 (t, J = 8.0 Hz, 1H), 4.25 (q, J = 7.1 Hz, 2H), 1.29 (t, J = 7.1 Hz, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 172.48, 164.34, 146.68, 137.22, 134.36, 133.59, 130.69, 129.61, 124.19, 111.92, 60.53, 14.71. DUIS-MS *m*/*z*: 263.0 (MH⁺).

Preparation of 8-bromo-4-oxo-1,4-dihydroquinoline-3carboxylic acid (8). The title compound was afforded as a white fluffy solid in 75% yield from ethyl 8-bromo-4-oxo-1,4-dihydroquinoline-3-carboxylate (**16a**) using General Procedure C after HPLC purification using linear gradient of 20% B to 90% B over 30 min at a flow rate of 10 ml/min, retention time = 15.8 min. ¹H NMR (400 MHz, DMSO-*d*₆) δ 14.86 (brs, 1H), 12.61 (s, 1H), 8.66 (s, 1H), 8.31 (dd, *J* = 8.1, 1.4 Hz, 1H), 8.24 (dd, *J* = 7.7, 1.3 Hz, 1H), 7.53 (t, *J* = 7.9 Hz, 1H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 178.61, 166.15, 146.36, 137.93, 137.53, 127.46, 126.69, 125.58, 113.16, 108.71. ESI-MS *m/z*: 268.0, 269.9 (MH⁺).

Preparation of 8-nitro-4-oxo-1,4-dihydroquinoline-3carboxylic acid (9). The title compound was afforded as a yellow fluffy solid in 45% yield from ethyl 8-nitro-4-oxo-1,4-dihydroquinoline-3-carboxylate (16b) using General Procedure C after HPLC purification using linear gradient of 20% B to 90% B over 30 min at a flow rate of 10 ml/min, retention time = 15.1 min. ¹H NMR (400 MHz, DMSO-d6) § 12.97 (brs, 1H), 8.83–8.81 (m, 2H), 8.74 (d, J = 8.0 Hz, 1H), 7.78 (t, J = 8.1 Hz, 1H). ¹³C NMR (126) MHz, DMSO-d6) δ 177.91, 165.68, 147.21, 137.87, 133.71, 133.54, 132.11, 126.99, 125.70, 109.65. ESI-MS m/z: 235.0 $(\mathrm{MH}^{+}).$

Preparation of 4-(3-carboxypropanamido)phthalic acid (10) and 4-(4-carboxybutanamido)phthalic acid (11)

General Procedure D to prepare compounds 10 and 11. The appropriate carboxyl chlorides (0.67 mmol) were added to the mixture of 4-aminophthalic acid (5, 0.67 mmol) and Na₂CO₃ (1.68 mmol) in water (3.0 ml). The mixture was

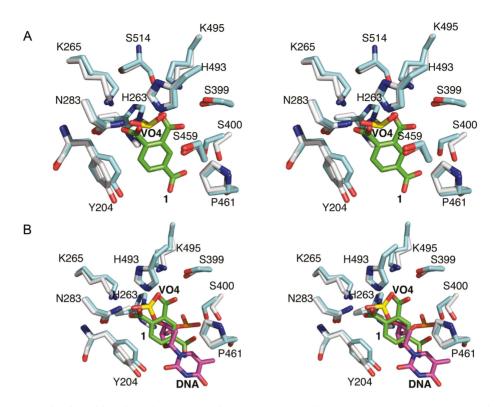


Figure 4. (A) Superimposed active site residues from the structures of TDP1 complexed with compound 1 (TDP1 carbon atoms in gray, 1 carbon atoms in green, PDB code: 6DIE) and TDP1 with bound vanadate (yellow, PDB code: 1NOP). (B) The same superimposed coordinates now highlighting the position of cytosine 806 from the bound DNA substrate.

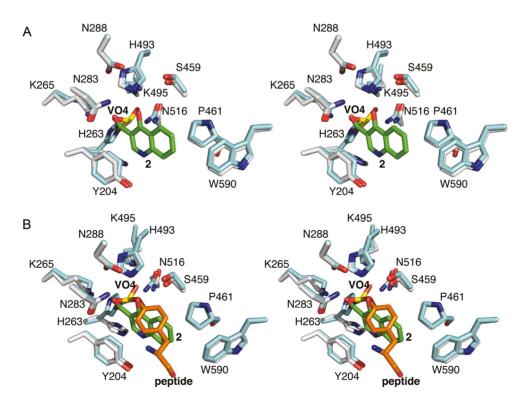


Figure 5. (A) Superimposed active site residues from the structures of TDP1 complexed with compound 2 (TDP1 carbon atoms in gray, 2 carbon atoms in green, PDB code: 6DJD) and TDP1 with bound vanadate (yellow, PDB code: 1NOP). (B) The same superimposed coordinates now highlighting the position of Y723 and L723 from the bound topoisomerase-derived peptide substrate.

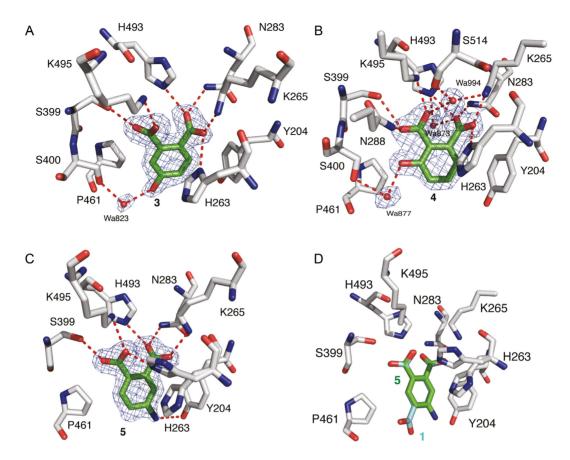


Figure 6. (A) Structure of TDP1 complexed with compound 3, (TDP1 carbon atoms in gray, fragment carbons in green, oxygen atoms in red and nitrogen atoms in blue, PDB code: 6DIH). Hydrogen bond atoms are shown as red dashes. The fit of 3 to the $2F_o$ - F_c electron density map (blue, 1.78 resolution, 1.0 σ level contour) is shown. (B) Structure of TDP1 bound to compound 4 (PDB code: 6DJI). The fit of 4 to the $2F_o$ - F_c electron density map (blue, 1.75 Å resolution, 1.0 σ level contour) is shown. (C) Structure of TDP1 bound to compound 5 at 1.74 Å (PDB code: 6DJJ). The fit of 5 to the $2F_o$ - F_c electron density map (blue, 1.75 Å resolution, 1.0 σ level contour) is shown. (C) Structure of TDP1 bound to compound 5 at 1.74 Å (PDB code: 6DJJ). The fit of 5 to the $2F_o$ - F_c electron density map (blue, 1.0 σ level contour) is highlighted. (D) Comparison of the binding modes of compound 1 (carbon atom in cyan) and compound 5 (carbon atoms in green).

stirred (rt, 16 h). The resultant mixture was acidified by HCl (2.0 N) and purified by HPLC to afford 4-amidophthalic acids (**7b-e**) (Scheme 3).

Preparation of 4-(3-carboxypropanamido)phthalic acid (10). The title compound was afforded as a white solid in 32% yield from 4-aminophthalic acid (5) and succinyl dichloride using general procedure D after HPLC purification using linear gradient of 10% B to 15% B over 20 min at a flow rate of 20 ml/min, retention time = 13.8 min. ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.68 (brs, 3H), 10.34 (s, 1H), 7.86 (s, 1H), 7.70 (s, 2H), 2.62 – 2.54 (m, 4H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 174.18, 171.30, 169.59, 168.02, 142.21, 135.87, 130.55, 125.61, 119.87, 118.01, 31.61, 29.07. ESI-MS m/z: 282.1 (MH⁺), 304.0 (MNa⁺).

Preparation of 4-(4-carboxybutanamido)phthalic acid (11). The title compound was afforded as a white solid in 26% yield from 4-aminophthalic acid (5) and glutaroyl dichloride using general procedure D after HPLC purification using linear gradient of 10% B to 40% B over 20 min at a flow rate of 20 ml/min, retention time = 7.6 min. ¹H NMR (400 MHz, DMSO- d_6) δ 12.59 (brs, 3H), 10.19 (s, 1H), 7.79 (d, J = 1.9 Hz, 1H), 7.64 (d, J = 2.0 Hz, 1H), 7.63 (s, 1H), 2.32 (t, J = 7.4 Hz, 2H), 2.21 (t, J = 7.3 Hz, 2H),

1.74 (p, J = 7.4 Hz, 2H). ¹³C NMR (101 MHz, DMSO d_6) δ 174.56, 171.88, 169.58, 168.03, 142.20, 135.84, 130.55, 125.70, 120.00, 118.17, 35.91, 33.36, 20.68. ESI-MS m/z: 296.1 (MH⁺), 318.1 (MNa⁺).

RESULTS

Fragment screening

A crystallographic fragment screening campaign targeting TDP1 was conducted using the Zenobia Fragments libraries 1 and 2.2. A total of 640 compounds were divided into cocktails that each contained four fragments. Two fragments, the hydrolysis product of ZT0911, 1,2,4benzenetricarboxilic acid (1) and fragment ZT1982, 4hydroxyquinoline-3-carboxylic acid (2), could be unambiguously identified from the F_0 - F_c difference electron density maps at 1.63 Å (PDB code: 6DHU) and 1.81 Å resolution (PDB code: 6DIM), respectively (Table 2). To confirm these results, TDP1 crystals were soaked with compounds 1 and 2, individually, and X-ray diffraction datasets were collected at 1.78 Å resolution for both complexes (Figure 2A and B; PDB codes: 6DIE and 6DJD, respectively). In the single component soaks, both fragments bound to several catalytic residues within the active site (57) in the same man-

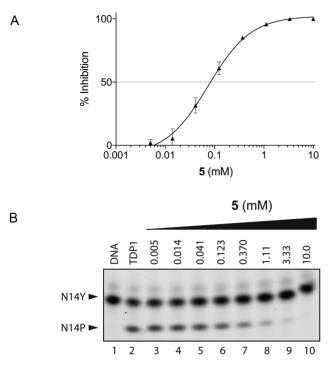


Figure 7. Inhibition of TDP1 by 4-aminonapthoic acid (5). (A) TDP1 inhibition dose response curve (n = 3, error bar not shown for data points where the bar is shorter than the height of the symbol). (B) Representative gel for inhibition of TDP1-catalyzed hydrolysis by 5: lane 1, N14Y only; lane 2, N14Y and TDP1; lanes 3–10 3-fold serial dilution of 5 from 10 mM to 5 μ M.

ner that was observed in the multicomponent soaks. Comparison with previously reported crystal structures of TDP1 reveal no major conformational changes in the protein upon fragment binding.

Compound 1 binds to the active site of TDP1 by direct hydrogen bonds involving its carboxylate oxygen atoms with the catalytic residues H263 (3.1 Å hydrogen bond distance), K265 (2.6 Å), N283 (2.8 Å), S399 (2.6 Å), H493 (2.8 Å) and K495 (2.7 Å) (Figure 2A, PDB code: 6DIE). Compound 1 also interacts with the active site via watermediated bridges between the side chains of S400 and S459 and backbone carbonyl oxygen atoms of P461 and S514. The side chains of Y204 and P461 engage in hydrophobic interactions with compound 1. The observed interactions involve several residues that play important roles in the catalytic activity of TDP1. Specifically, H263 serves as the nucleophile in the first step of the enzymatic reaction, while H493 functions as a general acid-base (see Figure 1). K265, N283 and K495 provide stabilizing interactions for the bound substrate and the transition state species during the reaction, while residues Y204 and S400 have been observed to interact with the DNA portion of the substrate (21,57,58). The fragment was evaluated in a gel-based TDP1 inhibition assay (Figure 3A). For this the 3' phosphotyrosyl DNA substrate N14Y was incubated with full-length TDP1 in presence of the tested compounds as concentrations up to 10 mM. Formation of the hydrolysis product N14P (Figure 3A) was monitored and analyzed to assess enzyme performance and drug inhibitory potency. Biochemical characterization demonstrates that **1** completely blocks TDP1catalyzed substrate conversion at concentrations above 1 mM in non-dose-dependent manner (Figure 3B).

Compound 2 also binds to the TDP1 active site through several direct hydrogen bonding interactions (Figure 2B, PDB code: 6DJD). The carboxylate oxygen atoms are engaged in hydrogen bonds with the side chains of the catalytic residues K265 (3.0 Å), N283 (2.9 Å) and the hydroxyl group of 2 to H493 (2.6 Å). Three water-mediated bridges between 2 and the active site are also observed: the nitrogen atom of 2 with the side chain of H263 via water 954 (Wa954), the hydroxyl oxygen of 2 with the side chain of S459 via Wa980 and the carboxylate oxygen of 2 with the side chains of K495 and N516 via Wa923. Additionally, the side chains of Y204, P461 and W590 participate in hydrophobic interactions with compound 2. Biochemical measurements indicate that 2 exhibits weak inhibitory activity against TDP1 by nearly completely inhibiting DNA substrate conversion at 10 mM (Figure 3B).

Comparison of binding modes

Compounds 1 and 2 both bind to the active site of TDP1 through several direct hydrogen bonds with some of the key catalytic residues. However, each exhibits a unique binding mode (Figure 2C). A common feature of the two fragments is that the carboxylate moieties occupy the similar position as observed in the crystal structure of TDP1 in complex with vanadate which represents a transitionstate mimic (16). In Figure 4A, the coordinates of TDP1 in complex with 1 are superimposed onto the coordinates of the crystal structure of TDP1 complexed with vanadate, DNA substrate and a TOP1 derived peptide (PDB code: 1NOP) (57). Oxygen atoms from both carboxylate moieties of 1 occupy the same space as the oxygen atoms of the vanadate molecule. The vanadate molecule is observed in a trigonal bipyramidal state that mimics the transition state that arises during the S_N2 nucleophilic attack on the phosphate atoms in the first step of the enzymatic reaction (57). Compound 1 is observed to interact with the TDP1 K495, N283, H493 and K495 side chains. Similarly, the superimposed coordinates of the TDP1-compound 2 complex with the 1NOP structure reveal one oxygen of the carboxylate moiety of 2 overlaps with the O3 oxygen of the vanadate molecule (Figure 5A). The O3 atom of vanadate makes direct hydrogen bonds with the side chains of N283 and K265, as does the hydroxyl oxygen in 2.

While 1 and 2 both bind in the same position as does the vanadate molecule in the 1NOP coordinates, each fragment occupies different spaces with respect to the DNA and topoisomerase-derived peptide components of the substrate. Fragment 1 superimposes with the cytosine 806 base of the DNA substrate (Figure 4B) while fragment 2 overlaps with the Y723 residue of the peptide substrate that is covalently attached to vanadate and mimics the phosphotyrosine bond (Figure 5B) (57). The two different binding modes of the fragments, combined with the common feature of overlapping with the vanadate binding spot, may present two possible pathways for inhibitor design against TDP1. The carboxylate moieties that interact with the catalytic TDP1 residues can potentially serve as an anchor

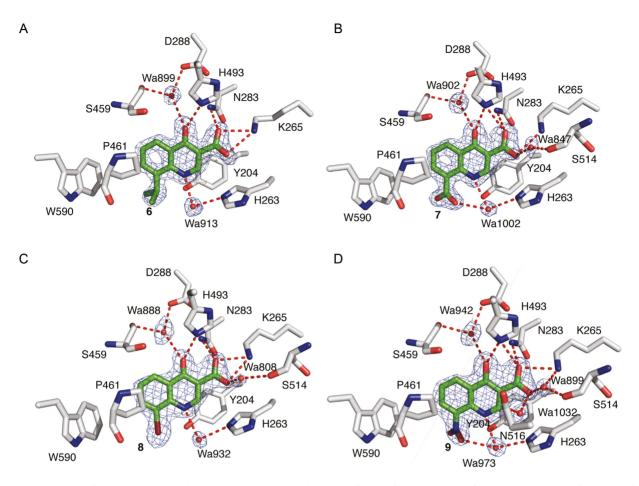


Figure 8. (A) Structure of TDP1 complexed with compound 6 (TDP1 carbon atoms in gray, fragment carbons in green, oxygen atoms in red and nitrogen atoms in blue, PDB code: 6DJE). Hydrogen bond atoms are shown as red dashes. The fit of 6 to the $2F_o$ - F_c electron density map (blue, 1.70 Å resolution, 1.0 σ level contour) is shown. (B) Structure of TDP1 bound to compound 7 (PDB code: 6DJF). The fit of 7 to the $2F_o$ - F_c electron density map (blue, 1.67 Å resolution, 1.0 σ level contour) is shown. (C) Structure of TDP1 bound to compound 8 at 1.91 Å (PDB code: 6DJH). The fit of 8 to the $2F_o$ - F_c electron density map (blue, 1.67 Å resolution, 1.0 σ level contour) is highlighted. (D) Structure of TDP1 bound to compound 9 (PDB code: 6MJ5). The fit of 9 to the $2F_o$ - F_c electron density map (blue, 1.85 Å resolution, 1.0 σ level contour) is shown.

to the active site while the different binding modes can be taken advantage of in targeting both the DNA and peptide substrate binding sites.

Fragment derivatives

Derivatives of both 1 and 2 were designed to demonstrate the potential for elaboration and optimization of the initial fragment hits. Three derivatives based on the chemical structure of 1 were examined for binding and inhibitory activity against TDP1. Compound 3 (Table 1), which is commercially available from Sigma-Aldrich (PH004941), features a hydroxyl substitution in place of the carboxylate moiety in position 4. The crystal structure of TDP1 complexed to compound 3 (PDB code: 6DIH) at a resolution of 1.78 Å reveals that this derivative retains the direct hydrogen bonding interactions between the carboxylate moieties and the side chains of K495 (2.7 Å), S399 (2.7 Å), H493 (2.7 Å), N283 (2.7 Å) K265 (2.5 Å) and H263 (3.0 Å) (Figure 6A). The hydroxyl at position 4 participates in a water-mediated bridge to the side chain of S400 via Wa823.

Compound 4 (Table 1) differs from 1 in that the carboxylate moiety at position 4 has been removed and a hydroxyl

group has been added at position 3. The 1.75 Å resolution crystal structure of TDP1 bound to 4 (PDB code: 6DJI) revealed that the compound binds to the active site via several direct hydrogen bonds and water-mediated bridges (Figure 6B). The carboxylate at position 1 forms direct hydrogen bonds with the side chains of H493 (2.8 Å) and N283 (2.7 Å), while the carboxylate group at position 2 forms direct hydrogen bonds with the side chain of S399 (3.0 Å), K495 (2.7 Å) and a water-mediated bridge to the backbone carbonyl oxygen of S514 via Wa994. The hydroxyl oxygen at position 3 forms a water-mediated bridge with the side chain hydroxyl of S400 through Wa877. The side chains of Y204 and P461 interact with 4 via hydrophobic interactions. As commonly observed for initial low molecular weight fragments, compounds 3 and 4 exhibited weak potencies with IC₅₀ values of 1.9 and 3.1 mM, respectively (30).

Higher inhibitory activity was observed for compound 5 (Table 1), in which the carboxylate moiety at position 4 in compound 1 was replaced by an amino group. The IC₅₀ in this case was estimated to be \sim 80 μ M (Figure 7). The 1.74 Å resolution structure of TDP1 complexed to 5 (PDB code:

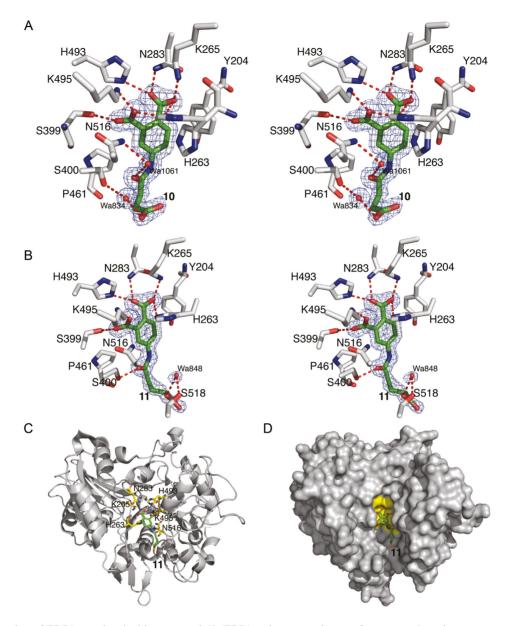


Figure 9. (A) Stereoview of TDP1 complexed with compound 10 (TDP1 carbon atoms in gray, fragment carbons in green, oxygen atoms in red and nitrogen atoms in blue, PDB code: 6N17). The fit of 10 to the the $2F_o$ - F_c electron density map (blue, 1.62 Å resolution, 1.0 σ level contour) is shown. (B) Stereoview of TDP1 complexed with compound 11 (TDP1 carbon atoms in gray, fragment carbons in green, oxygen atoms in red and nitrogen atoms in blue, PDB code: 6N19). The fit of 11 to the $2F_o$ - F_c electron density map (blue, 1.50 Å resolution, 1.0 σ level contour) is shown. (C) Ribbon diagram of TDP1 (gray) with the residues of the catalytic HKN motif highlighted in yellow sticks. Compound 11 is shown with carbon atoms in green. (D) Surface representation TDP1 (gray) with the residues of the catalytic HKN motif shaded in yellow. Compound 11 (carbon atoms in green) is shown extending into the TDP1 active site DNA-binding pocket.

6DJJ) demonstrates that the amino group substitution enhances the binding such that the amino group is positioned within 3.3 Å of the hydroxyl oxygen of the Y204 side chain residue. In addition, the carboxylate groups at positions 1 and 2 retain direct hydrogen bonding interactions with the side chains of S399 (2.7 Å), K495 (3.0 Å), H263 (2.6 Å), H493 (2.7 Å), N283 (2.7 Å) and K265 (2.6 Å), as in compound 1 (Figure 6C and D). The P461 and Y204 side chains also interact with the aryl portion of **5** via hydrophobic interactions.

Elaborated fragments based on the chemical scaffold of **2** were also examined. A crystal structure of TDP1 in complex

with compound **6** (Sigma-Aldrich compound CDS010292) was solved at 1.70 Å resolution (PDB code: 6DJE). Compound **6** features an isopropyl extension at position 8 (Table 1). The binding mode reveals that the core structure retains the same binding interactions as the parent molecule **2** while the isopropyl substituent extends into the DNA binding region of TDP1 and makes hydrophobic contacts with the side chains of P461 and W590 (Figure 8A). Compound 7 also yielded a high resolution crystal structure (PDB code: 6DJF). It has a carboxylate moiety instead of an isopropyl group at position 8 (Table 1), which also extends into the DNA binding pocket and forms a water-mediated bridge

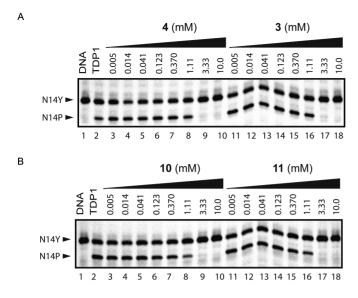


Figure 10. Inhibition of TDP1 by 3, 4, 10 and 11. (A) Representative gel for inhibition of TDP1-catalyzed hydrolysis by 4 and 3: lane 1, N14Y only; lane 2, N14Y and TDP1; lanes 3–10, 3-fold serial dilutions of 4 from 10 mM to 5 μ M; lanes 11–18, 3-fold serial dilutions of 3 from 10 mM to 5 μ M. (B) Representative gel for inhibition of TDP1-catalyzed hydrolysis by 10 and 11: lane 1, N14Y only; lane 2, N14Y and TDP1; lanes 3–10, 3-fold serial dilutions of 10 from 10 mM to 5 μ M; lanes 11–18, 3-fold serial dilution of 11 from 10 mM to 5 μ M.

via Wa1002 to the side chain of H263 (Figure 8B). New binding interactions are also observed between the carboxylate oxygen and the backbone carbonyl oxygen of S514 through a water-mediated bridge via Wa847. Compound 8 has a bromine at position 8. A water-mediated bridge via Wa808 between the 8 carboxylate oxygen and the backbone carbonyl oxygen of S514 is observed as well (Figure 8C, PDB code: 6DJH). The structure of TDP1 in complex with compound 9 (PDB code: 6MJ5), a derivative of compound 2 with a nitro group at position 8 (Table 1), reveals that one of the nitro oxygen atoms engages in a water-mediated bridge via Wa973 with the side chain nitrogen of H263. The water-mediated bridge between the carboxylate oxygen and S514 is retained as observed with compounds 7 and 8 (Figure 8D). Therefore, the crystal structures of the fragment 2 derivatives in complex with TDP1 reveal the potential for picking up new interactions with residues in the DNAbinding region via substitutions at position 8 of the parent fragment. However, the derivatives of compound 2 were only weak inhibitors compared to compound 5, a derivative of compound 1. Compounds 6, 7 and 8 did not inhibit TDP1 at 1 mM concentration and incubation of TDP1 with **9** resulted in only 15% TDP1 inhibition at 1 mM.

Potential for fragment elongation

A key step in the fragment approach to drug development is the process of elongating a fragment into a larger compound that can pick up new binding interactions with the target protein. Crystal structures of TDP1 bound to 6, 7, 8 and 9 demonstrate that fragment elongation at the 8position of the 4-hydroxyquinilone -3-carboxylic acid series can be used to pick up new binding interactions that

may enable the fragment to grow into the peptide binding pocket. Additionally, using structure-guided design, we prepared two pthalic acid derivatives, compounds 10 and 11 (Table 1) with linker groups at the fourth position (3carboxypropanoamido and 4-carboxybutanamido, respectively) that extend into the DNA-binding pocket of TDP1. Compound 10 (Figure 9A, PDB code: 6N17) is anchored in the active site by hydrogen bonds between the 1- and 2carboxylates to the side chains of H263 (2.6 and 2.8 Å, sidechain existing in two alternate conformations), K265 (2.5 Å), N283 (2.8 Å), S399 (2.7 Å), H493 (2.7 Å) and K495 (2.7 Å). The linker group of 10 picks up two water-mediated hydrogen bond bridges with TDP1. The carbonyl O07 group forms a hydrogen bond to Wa1061 (3.1 Å) which is hydrogen bonded to the ND2 atom of the side chain of N516 (2.9 Å). The O03 atom of 10 is hydrogen bonded to Wa834 (2.6 Å), which forms a water-bridge to the side chain of S400 (3.1 Å). Compound 11 (Figure 9B, PDB code: 6N19) forms hydrogen bonds between the 1- and 2-carboxylates and the side chains of the active site residues H263 (3.0 A), K265 (2.7 Å), N283 (2.8 Å), S399 (2.8 Å), H493 (2.7 Å) and K495 (2.7 Å). The linker group at the fourth position picks up a direct hydrogen bonding interaction between the carboxylate O08 atom and the side chain of S400 (2.5 Å). Wa848 mediates a water-bridge between the O03 atom (2.6 Å) and the side chain of S518 (2.7Å). These compounds show millimolar inhibition of TDP1 (IC₅₀ values of 1.01 and 0.96 mM, respectively, Figure 10). The addition of a growing linker that picks up new binding interactions demonstrates the potential to use structure-guided design to elongate the fragments into the DNA-binding pocket (Figure 9C and D).

DISCUSSION

To our knowledge, this study is the first to report crystal structures of organic chemical probes bound to the catalytic site of human TDP1. Two chemically distinct fragments were discovered that bind at the catalytic site as revealed by high resolution crystal structures. Both phthalic acids and quinolone-based fragments engage the catalytic core of TDP1 in a fashion similar to the previously described phosphate mimic vanadate. The difference in the chemical nature of the two fragment classes results in a somewhat different arrangement of their planar aromatic cores and overlapping polar contact networks. This finding points to the fact that polar groups within and in close proximity to the TDP1 catalytic site are capable of forming a broad interaction network with either the substrate or an inhibitor.

Each of the eleven compounds engage one of the two conserved catalytic triad residues of human TDP1 (H263, K265, N283) (Figure 1) (16,58). All phthalic and some 4-hydroxiquinoline-3-carboxylic acid analogues exhibited detectable enzyme inhibition. Their low potency is consistent with the fact that, at this stage, the compounds correspond to low molecular weight fragments (30,39,40). Further improvements in potency, which may be obtained by elaboration of the fragments, will be required before their efficacy can be evaluated *in vivo*. The greatest inhibitory potency was seen for compound **5**, which engages both HKN catalytic triads centered on H263 and H493, respectively (see Figure 1). The higher potency of compound **5** in compari-

son to other phthalic acid derivatives (compounds 1, 3 and 4) could be attributed to its additional interaction with the edge of the catalytic groove by forming a hydrogen bond with Y204. The crystal structures of TDP1 bound to 10 and 11 also demonstrate that further elaboration of molecules based on the current chemical fragments may lead to larger compounds that pick up additional binding interactions with the TDP1 active site. The crystal structures of TDP1 in complex with the fragments and their derivatives described here finally open the door for structure-based drug design on this important molecular target.

DATA AVAILABILITY

Protein Data Bank (https://www.rcsb.org/): 6DHU, 6DIE, 6DIH, 6DIM, 6DJD, 6DJE, 6DJF, 6DJH, 6DJI, 6DJJ, 6MJ5, 6N17, 6N19.

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