Original Article

## Synthesis and cytotoxic evaluation of some quinazolinone- 5-(4-chlorophenyl) 1, 3, 4-oxadiazole conjugates

Farshid Hassanzadeh<sup>1</sup>, Hojjat Sadeghi-Aliabadi<sup>1</sup>, Elham Jafari<sup>1,\*</sup>, Azadeh Sharifzadeh<sup>1</sup>, and Nasim Dana<sup>2</sup>

<sup>1</sup>Department of Medicinal Chemistry, Isfahan Pharmaceutical Sciences Research Center, School of Pharmacy and Pharmaceutical Sciences, Isfahan University of Medical Sciences, Isfahan, I.R. Iran. <sup>2</sup>Applied Physiology Research Center, Isfahan University of Medical Sciences, Isfahan, I.R. Iran.

#### Abstract

1, 3, 4- Oxadiazoles and quinazolinones are privileged structures with extensive biological activities. On account of reported anticancer activity of them, in this study, a multi-step reaction procedure has been developed for the synthesis of some quinazolinone-1, 3, 4-oxadiazole derivatives. Reaction of the synthesized 3-amino-4(3H) quinazolinone derivatives with chloroacetyl chloride in the presence of dichloromethane/triethylamine yielded 2-chloro -N-(4-oxo-2-quinazolin3 (3H)-yl) acetamide derivatives as intermediate. Treatment of the resultants with 5- (4-chlorophenyl) 1, 3, 4-oxadiazole-2-thiol in dry acetone and potassium carbonate gave coupled derivatives of quinazolinone-1, 3, 4-oxadiazole. The cytotoxic effect of final compounds was tested against MCF-7 and HeLa cell lines using MTT assay. Compound 2-(5-(4-chlorophenyl)-1,3,4-oxadiazol-2-ylthio) N-(4-oxo-2-propylquinazolin)3(4H)acatamide 6a exhibited remarkable cytotoxic activity at 10 and 100  $\mu$ M against HeLa cell line. The alteration of substituents on C<sub>2</sub> of quinazolinone ring revealed that the introduction of propyl moeity improved cytotoxic activity against HeLa cell line.

**Keywords:** Cytotoxicity; Quinazolinone; Oxadiazole.

#### INTRODUCTION

Amongst heterocyclic compounds, oxadiazole one of the attractive for the development constructions new drugs. In drug discovery program, oxadiazole ring can be used as a main pharmacophore interfering with receptor, an aromatic linker, modulating molecular and bioisosteres of properties, and esters (1-5).

There are four known isomers of this five-membered ring including: 1,2,4-, 1,2,3-, 1,2,5-, and 1,3,4-oxadiazole. 2, 5 Disubstituted 1, 3, 4 oxadiazole isomer is associated with diverse pharmacological activities anticancer, antibacterial, such as antifungal, antitubercular, anticonvulsant, anti-inflammatory effects and (5-8).proposed anticancer mechanisms for oxadiazoles include inhibition of tubulin polymerization and epidermal growth factor receptor (EGFR). Quinazolinone is another nitrogenated scaffold that was extensively used in drug development programs and its derivatives showed diverse biological activities anticancer, antifungal, as anti-inflammatory, and antimicrobial (9-15). Quinazolinone-based structures their anticancer activity through different mechanisms including inhibition of the DNA repair enzyme system, inhibition of EGFR (9), thymidylate enzyme inhibition, and inhibitory effects for tubulin polymerize (14). The efficacy 1.3.4-oxadiazole and quinazolinone derivatives has been demonstrated in many literatures. Also quinazolinone-oxadiazole structures with antimicrobial, anti-inflammatory, antioxidant, anticancer, and analgesic effects have been reported (5,10,16-20).



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Because of the remarkable cytotoxic of 1, 3, 4-oxadiazole (6-8)quinazolinone (9-13)derivatives, and in the hope of achieving synergistic presence response due the to both quinazolinone and 1,3,4-oxadiazole moieties, in this study, some quinazoline derivatives containing the 1, 3, 4-oxadiazole were synthesized and evaluated for their cytotoxic activity.

#### MATERIALS AND METHODS

#### Instrumentation

All starting materials, reagents, and solvents were purchased from commercial like Merck (Germany) Aldrich (USA) companies. Merck silica gel 60 F<sub>254</sub> plates (Germany) were applied for analytical thin layer chromatography (TLC). Proton nuclear magnetic resonance (HNMR) spectra were recorded using a Bruker 400 MHz spectrometer (Germany), and chemical shifts are expressed as ppm with tetramethylsilane (TMS) as internal standard. Infrared (IR) (KBr discs) was recorded with a WOF-510 fourier-transform IR (FT-IR) spectrophotometer (China). Melting points were determined using electrothermal 9200 melting point apparatus (United Kingdom) and are uncorrected.

## General procedures for synthesis of compounds

2-Amido-benzoic acid derivatives (2a-2c) were prepared by reaction of anthranilic acid (1) with acylchloride derivatives. The reaction was followed by dehydrative cyclization of 2- amido-benzoic acid derivatives to form benzoxazinone intermediate (3a-3c). Reaction between benzoxazinone and hydrazine hydrate in ethanol under reflux condition produced 3-amino quinazolinone derivatives in high yield (4a-4c). Treatment of 3-amino quinazolinone with chloro acetylchloride in the presence of dichloromethane/triethylamine afforded 2-chloro -N-(4-oxo-2-quinazolin3 (3H)-yl) acetamide derivatives Final compounds (6a-6c) obtained through the nucleophilic displacement of the chloride with thiol of 5-(4-chlorophenyl) 1, 3, 4oxadiazole-2-thiol in dry acetone and potassium carbonate (Scheme 1) (21,22).

Scheme1. Synthesis of the target compounds (6a-6c).

### Synthesis of 2- amido-benzoic acid derivatives (2a-2c)

To a magnetically stirred solution of anthranilic acid (1) (0.04 mol) in dimethyl formamide (35 mL) was added dropwise a solution of acylchloride (butyryl chloride, benzoyl chloride and 4-nitrobenzoyl chloride) (0.045 mol) over 15 min. The mixture was stirred at room temperature for 3 h until a solid product was formed. Then the mixture was poured into water and the precipitate was collected by filtration, washed with water, and dried under reduced pressure to achieve compounds **2a-2c** (Scheme 1) (23).

#### Synthesis of benzoxazone derivatives (3a-3c)

A solution of compounds **2a-2c** (0.01 mol) in acetic anhydride (30 mL) was heated for 1 h with vigorous stirring. After completion of the reaction which confirmed by TLC, the solvent was removed by distillation under reduced pressure to obtain derivatives **3a-3c** (Scheme 1)(23).

### Synthesis of 3-aminoquinazolinone derivatives (4a-4c)

A mixture of related benzoxazone **3a-3c** (0.01 mol) and hydrazine hydrate (0.02 mol) in ethanol was refluxed for 3 h. After the reaction was completed, the mixture was cooled and the separated solid was collected by filtration and recrystallized from ethanol or isopropanol (21,22).

## Synthesis of 2-chloro -N-(4-oxo-2-quinazolin3 (3H)-yl) acetamide derivatives (5a-5c)

Chloroacetylchloride (0.01 mol) was added solution of 3-amino-quinazoline derivatives **4a-4c** (0.01 mol) in dichloromethane (20 mL) and triethylamine (0.01 mol), mixture was stirred at room temperature for 30 min. Then the reaction mixture was poured into ice water and with dichloromethane ethyl acetate. The extracted ethyl acetate was washed with sodium bicarbonate solution (3%) and dried over anhydrous magnesium sulfate, which upon evaporation afforded the products **5a-5c** (21,22).

## Synthesis of quinazolinone-oxadiazole hybrid derivatives (6a-6c)

Title compounds **6a-6c** were synthesized by refluxing 2-chloro -N-(4-oxo-2-quinazolin3 (3H)-yl) acetamide derivatives **5a-5c** (0.01 mol) with 5-(4-chlorophenyl) 1, 3, 4-oxadazole-2-thiol in dry acetone (20 mL) and anhydrous potassium carbonate (0.01 mmol) for 6 h. The reaction mixture was filtered while hot. The organic solution was concentrated and purified by preparative TLC (21,22).

#### Cytotoxicity assay

Sample and culture media preparation

MCF-7 (breast cancer), and HeLa (cervical cancer) cells were purchased from pasture institute of Iran (Tehran, I.R. Iran) and maintained at 37 °C in a humidified atmosphere (90%) containing 5% CO2. Both cell lines were grown in RPMI 1640 completed with 5% v/v fetal bovine serum, 100 U/mL penicillin, and 100 mg/mL streptomycin. After 2-3 subcultures, 180  $\mu$ L of the cell suspensions (5 × 10<sup>4</sup> cells/mL) were seeded in 96well plates and incubated for 24 h.

The stock solutions of compounds (10 mM, 1 mL) were prepared in minimum volume of dimethyl sulfoxide (DMSO) and diluted with the medium to obtain 10, 100, 1000 μM concentrations. After 24 h incubation, 20 µL of different concentrations of the derivatives were added\and the microplates were further incubated for 48 h. Paclitaxel was used as positive control. To evaluate cell survival, 20 μL of MTT solution (5 mg/mL in phosphate buffer solution) was added to each well and incubated for 4 h. Afterwards, the media in each well was gently replaced with 150 µL DMSO to dissolve formazan crystals. The absorbance of each well was measured at 540 nm using an ELISA plate reader. Each experiment was repeated three times. Analysis of variance (ANOVA) followed by Tukey test was used to determine the differences between various groups.

Cell viability calculated using the following equation:

Cell survival (%) =  $\frac{Absorbance\ of\ treated\ well-\ absorbance\ of\ blank}{Absorbance\ of\ control\ well-\ absorbance\ of\ blank} \times 100$ 

#### **RESULTS**

## 2-(5-(4-Chlorophenyl)-1,3,4-oxadiazol-2-ylthio)N-(4-oxo-2-propylquinazolin)3(4H) acatamide (6a)

Yield: 31%, m.p.162.5-163 °C, IR  $v_{max}$ , 3243(NH), 2931 (C-H), 1677 (C=O), 1611(C=N), cm<sup>-1</sup>; <sup>1</sup>HNMR: (400 MHz; CDCl<sub>3</sub>): δ 9.79 (b, NH), 8.07 (1H, dd, J = 8 Hz, J = 4 Hz, H<sup>a</sup>), 7.87 (2H, d, J = 8 Hz, H<sup>b</sup>), 7.66 (1H, m, H<sup>c</sup>), 7.59 (1H, d, J = 8 Hz, H<sup>d</sup>), 7.44(2H, d, J = 8 Hz, H<sup>e</sup>), 7.34 (1H, t, J = 8 Hz, H<sup>f</sup>), 4.17-4.21 (1H, d, J = 16 Hz, CH<sub>2</sub><sup>g</sup>), 3.94-3.98 (1H, d, J = 16 Hz, CH<sub>2</sub><sup>g</sup>), 2.64 (2H, t, J = 8 Hz, CH<sub>2</sub><sup>h</sup>), 1.72 (2H, m, CH<sub>2</sub><sup>i</sup>), 0.92 (3H, t, J = 8 Hz, CH<sub>3</sub><sup>J</sup>).

## 2-(5-(4-Chlorophenyl)-1,3,4-oxadiazol-2-ylthio)N-(4-oxo-2-phenylquinazolin)3(4H) acatamide (6b)

Yield: 30%, m.p.150-151 °C, IR  $\nu_{max}$ , 3216 (NH), 1696 (C=O) cm<sup>-1</sup>; <sup>1</sup>HNMR δ: (400 MHz; CDCl<sub>3</sub>), 8.73 (1H, s, NH), 8.23 (1H, d, J=8 Hz, H<sup>a</sup>), 7.72-7. 78 (3H, m, H<sup>b,c</sup>), 7.57(2H, d, J=8 Hz, H<sup>d</sup>), 7.38-7.50(7H, m, H<sup>e</sup>, f), 4.08-4.12(1H, d, J=16 Hz, CH<sub>2</sub><sup>g</sup>), 3.87-3.91(1H, d, J=16Hz, CH<sub>2</sub><sup>g</sup>).

# 2-(5-(4-Chlorophenyl)-1,3 ,4 - oxadiazol - 2 - ylthio)N(2-4nitrophenyl)4-oxoquinazolin) 3(4H)acatamide(6C)

Yield: 33%, m.p.177 °C (decomposed), IR v<sub>max</sub>, 3313 (NH), 1696 (C=O), 1596, 1349

NO<sub>2</sub> cm<sup>-1</sup>; <sup>1</sup>HNMR  $\delta$ : (400 MHz; CDCl<sub>3</sub>), 8.74 (1H, s, NH), 8.24-8.28 (3H, m, H<sup>a,b</sup>), 7.73-7.83 (5H, m, H<sup>c,d,e</sup>), 7.53 (1H, t, J = 8 Hz, H<sup>f</sup>), 7.1-7.24 (3H, m, H<sup>g,h</sup>), 4.12-4.16 (1H, d, J = 16 Hz, CH<sub>2</sub><sup>i</sup>), 3.91 -3.95 (1H, d, J = 16 Hz, CH<sub>2</sub><sup>i</sup>).

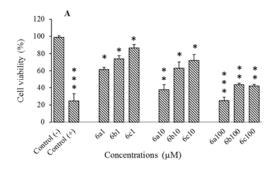
#### Cytotoxic effects of the derivatives (6a-6c)

The cytotoxicity of compounds were evaluated against HeLa and MCF-7 cell lines at different concentrations (1, 10, and 100  $\mu$ M) using MTT assay. Results are shown in Fig. 1A, 1B, and Table 1.

Compound **6a** exhibited remarkable cytotoxic effect at all tested concentrations (1, 10, and 100  $\mu$ M) on HeLa cell line and cell viability reduced to about 61%, 37% and 24% respectively. Compounds **6b** and **6c** indicated similar effects in the same concentrations on HeLa cell line. These two compounds reduced cell viability to about 42% at 100  $\mu$ M.

Remarkable differences were not observed between the cytotoxicity of these compounds on MCF-7 cell line. These compounds displayed the highest cytotoxic activities against MCF-7 cells at  $100~\mu M$  that cell viability reduced to about 50%.

These compounds exhibited significant differences in viability compared to the negative control on both cell lines which presented in Fig. 1A and 1B.



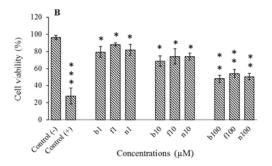


Fig. 1. Cytotoxic effect of compounds 6a-6c on (A) HeLa and (B) MCF-7 cells following exposure to different concentrations (1, 10, and 100  $\mu$ M). Data are presented as mean  $\pm$  SD, n = 3. \*P < 0.05, \*\*P < 0.01, and \*\*\* P < 0.001 Shows significant differences in comparison with negative control group, Paclitaxel was used as positive control.

**Table 1.** The IC<sub>50</sub> of tested compounds against MCF-7 and HeLa cell lines.

Target compounds	R	IC <sub>50</sub> (μM) MCF-7	IC <sub>50</sub> (μM) HeLa	<del></del>
6a	Propyl	$82.18 \pm 3$	$7.52 \pm 0.6$	
6b	Phenyl	$97.17 \pm 4$	$79.74 \pm 3$	
6c	Nitro phenyl	$101.47 \pm 4$	$79.32 \pm 3$	

#### **DISCUSSION**

1, 3, 4-Oxadiazole heterocycle as good bioisosteres of amides and esters, can improve pharmacological activity via hydrogen bonding interactions with the receptors (6,24). Literature survey revealed that little changes in the structure of substituted 1, 3, 4-oxadiazole can lead to quantitative and qualitative alterations in their biological activities (3).

A series of 2, 5-disubstituted-1, 3, 4oxadiazoles has been reported as tubulin polymerization inhibitors (8,25). Moreover, 1, 3, 4-oxadiazole derivatives possessing 1,4-benzodioxan moiety have been introduced anticancer agents as potential Besides, quinazolinone is a heterocyclic scaffold with extensive biological effects, in particular, anticancer activity. Derivatives of substituted quinazolinone at 2, 3 or 2, 4 positions have been reported as anticancer agents (22,27-29). Literature surveys have shown many reports on cytotoxic activities of quinqzolinone (9,22,27) and oxadiazoles (1,2,6,8).

Hybrid structures of quinazolinoneoxadiazole have presented anticancer (5,14,19) and antimicrobial (14,16,20) activities. Some of the 4-alkoxyquinazoline derivatives containing the 1, 3, 4-oxadiazole scaffold showed potent inhibitory activity against HeLa and MCF-7 cell lines (19).

We reported the synthesis of a novel series of quinazolin-4(3H)-one derivatives bearing oxadiazole, in the 3-position of the quinazolinone nucleus in a multiple-step reaction procedure. Amongst tested compounds, **6a** showed the highest cytotoxic activity against HeLa cell line at all tested concentrations while compounds **6b** and **6c** indicated mild cytotoxic effects against HeLa cell line at highest tested concentration. Three compounds displayed the highest cytotoxic activities against MCF-7 cells at 100 µM concentration.

According to the results shown in Table 1, compounds **6b** and **6c** bearing aromatic substituents on C<sub>2</sub> of the quinazolinone ring showed lowest cytotoxic activities on both cell lines, while compound **6a** containing aliphatic substituent on C<sub>2</sub> was more active on HeLa cell line with IC<sub>50</sub> value 7.52 μM.

Khodarahmi et al. reported the synthesis and cytotoxic evaluation of quinazolinonebenzimidazole hybrid derivatives against MCF-7 and HeLa cell lines. Cytotoxicity results revealed that compounds with phenyl and nitrophenyl substitutes on C<sub>2</sub> of the quinazolinone ring had the lowest cytotoxic activity against both cell lines and compounds with aliphatic substituents in this position had the highest potency (30). Other hybrids of quinazolinone-triazole were recently reported by Jafari et al. Cytotoxicity results exhibited that the presence of electron donating substituents on C2 of the quinazolinone ring could be in favor of the activity for these compounds (22). Collectively it could be assumed that the presence of electron-donating groups such as propyl substitution on C2 of quinazolinon ring could improve activity for these compounds while electron withdrawing groups such as phenyl and nitrophenyl substitutes have opposite effects.

#### **CONCLUSION**

In the present study, some of the conjugated oxadiazole-quinazolinone derivatives with amide linker were synthesized and evaluated for their cytotoxicity against HeLa and MCF-7 cell lines. Compound  $\bf 6a$  showed the highest cytotoxic activities with the IC50 value of 7.52  $\mu$ M against HeLa cell line. Substitution of propyl group at 2 position of quinazolinone improved the cytotoxic activity against HeLa possibly due to electronic effects.

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