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From one species to another: A review on the interaction between chemistry and microbiology in relation to cleaning in the built environment

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Abstract

Since the advent of soap, personal hygiene practices have revolved around removal, sterilization, and disinfection—both of visible soil and microscopic organisms—for a myriad of cultural, aesthetic, or health-related reasons. Cleaning methods and products vary widely in their recommended use, effectiveness, risk to users or building occupants, environmental sustainability, and ecological impact. Advancements in science and technology have facilitated in-depth analyses of the indoor microbiome, and studies in this field suggest that the traditional "scorched-earth cleaning" mentality—that surfaces must be completely sterilized and prevent microbial establishment—may contribute to long-term human health consequences. Moreover, the materials, products, activities, and microbial communities indoors all contribute to, or remove, chemical species to the indoor environment. This review examines the effects of cleaning with respect to the interaction of chemistry, indoor microbiology, and human health.

KEYWORDS

antimicrobial resistance genes, chemical intervention, indoor microbiology, occupant health, surface microbiology, urface chemistry

1 | WHY WE CLEAN AND WHAT WE LEAVE BEHIND

The idea of cleanliness has shaped our social and political landscape for millennia and has been as nuanced as the ways in which we accomplish it. The earliest recorded use of soap dates to 2800 BC, but historically, different cultures either prioritized hygiene and the removal of visible soil as well as recommended against it.¹ Cleaning was often used for rituals or hospitality, or to remove odor, more

than it was used to remove human or animal waste as a potential source of infection.¹ Yet, urbanization over the last millennium fueled rapid human social and habitational development which altered our interactions and our environment, contributing to recordsetting, multinational epidemics, and intensifying the issues of waste management, sanitation practices, and public health.²⁻⁴

Over time, our reasons for cleaning evolved to incorporate management or prevention of infectious disease³; however, early infection control was frequently based on fear and anecdotal evidence.¹

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Town-scale quarantine was used to prevent the spread of seemingly uncontainable epidemics beginning in the second century, and by the fifth century, hospitals became more common, though patients often languished in decrepit and unsanitary conditions—these places were regarded as areas in which to die rather than to heal.³ Not until the initial discovery of microorganisms in the seventeenth century did scientists, medical professionals, and the general public begin to understand the mechanism of cleaning for infection control, but new ideas and implementations of public health initiatives were sometimes met with resistance.^{2,3}

By the twentieth century, increasingly rapid developments in microbiology and chemistry yielded multitudinous consumer products with which to design, manipulate, and clean the built environment, most of which attempt to imbue or associate their product with pleasing odors. Our intricate sensory perceptions drive our desire for multifaceted indoor spaces which gratify visual, auditory, tactile, and olfactory senses. Whether driven by chemistry or microbiology, our "obsession with clean" continues to shape our world view and is intertwined with cultural beliefs, practices, and social, political, and economic dynamics. 1,5,6 Yet, it also affects our health; we spend up to 90% of their time indoors, during which we are exposed to chemicals generated by building materials, as well as products for cleaning, personal care, and hygiene (Figure 1). Cleaning imbues the air we breathe and coats the surfaces we touch with a myriad of chemicals, the health or environmental effects of many of which have not been established.⁸⁻¹¹ Indoor chemistry is superbly complex.¹¹ due to the diversity of chemical compounds present, the ability of materials to

Practical Implications

- Simple interventions, such as hand washing, can dramatically improve health and reduce infectious disease.
- Chemical intervention, while effective, may encourage the development of microbial resistance over time if not implemented properly.
- Microbial communities adapt, reassemble, and persist, and recent theory in microbial ecology suggests that curating microbial communities may be more sustainable than perpetually attempting to remove them.

emit or adsorb, the propensity for gases to become trapped indoors, the proximity to occupants, and local environmental conditions—for example, ultraviolet (UV) light, temperature, humidity, ozone—which can alter chemical reactivity. The unintended consequence of cleaning is that occupants can be exposed to a myriad of chemicals and their byproducts.

Of particular interest are volatile organic chemicals (VOCs), which have a high reactivity, are frequently off-gassed by materials in the built environment, and are likely to become entrapped indoors (Figure 1). The majority of well-described VOCs, and those in focus here, create displeasing odors, irritate mucous membranes, cause corrosion damage to cells, exacerbate respiratory problems, and can produce secondary or tertiary chemical reactions. 12-14



FIGURE 1 Chemical species indoors are sourced from building materials, material goods, cleaning and hygiene products, human or other biological occupant activities, microbial activities, and a variety of chemical reactions. The interaction between microorganisms, chemicals, and human occupants is complex and often affected by architectural or environmental factors. Airflow through a building (indicated by arrows) can affect dispersal and mixing of chemical and microbial species indoors

Recent consumer demands have led to the development of "green products"—so named because they are marketed as less hazardous and more environmentally sustainable, as compared to synthetic cleaners. Yet "green" is an inexact term and regulation has not always kept pace with marketing; "green" products thus far do not inherently exhibit lower emission rates of all, or even some, classified hazardous VOCs, despite consumer perceptions. 10,13,15 Semi-volatile organic compounds (SVOCs), a subcategory of VOCs with a higher molecular weight, are commonly incorporated into materials in the built environment for a variety of purposes and can also be found in detergents (Figure 1); but these, too, may leach and accumulate in dust on surfaces. 16,17 Once leached into and incorporated into dust, chemicals can be resuspended into indoor air by occupant traffic, airflow from ventilation, and routine cleaning activities, allowing these chemical species to be inhaled, ingested, or otherwise come into contact with human skin. 18-21 Collectively, the chemicals present in cleaning solutions, as well as the reactions caused by solutions interacting with surfaces, can cause negative health effects in building occupants, especially those with close or frequent contact to the products. 14,21-23

Similarly, microorganisms can either process VOCs to create microbial volatile organic compounds (MVOCs) or independently produce MVOCs (Figure 1). Microorganisms can subsist off organic material found in dust, especially sloughed human cells, ^{24,25} and their production of ammonia and volatile fatty acids can be sufficient to spur odor complaints. ²⁵ However, low MVOC production, the plurality of microbial and chemical sources of chemicals indoors, and instrument detection limits can make it difficult to accurately classify a chemical as an MVOC. ^{26,27} As such, drawing conclusions from in situ MVOC studies has been difficult.

Typically, MVOC production is associated with damp buildings, and with fungal growth, especially molds. 27,28 Surface properties of different materials can contribute to localized areas of high moisture which, along with the chemicals absorbed there, can support microbial growth and MVOC production.²⁹ There is a clear association between health and the dampness of buildings, especially with visible mold, 30-32 yet a connection between MVOCs and health has yet to be firmly established beyond a mouse model. 33 Some MVOC studies report a correlation between MVOCs and self-reported adverse mucosal/respiratory symptoms (ie, stuffiness, sneezing, coughing) in occupants. 26,34 However, these correlations are always not statistically significant, 34 suggesting a more intricate relationship between individual MVOCs and health, or a limitation of using self-reported symptoms. This may also reflect the diversity of MVOCs and other microbial metabolites, very little of which is typically represented in sampling efforts.³⁵ MVOCs produced by fungi, in particular, may be poorly characterized, and it has been hypothesized that many of these MVOCs could also be classified as mycotoxins.²⁷ Collectively, MVOCs, microbial metabolites, and building-sourced VOCs apparently affect indoor chemistry, human health, and well-being, ^{25,34,36} and the generalization of poor indoor air quality and resulting negative effects on occupants are known as sick building syndrome. 8,34,37

The mechanisms of interaction between indoor microbes and chemicals are multifaceted. In addition to producing chemicals indoors, some microorganisms have the capacity to degrade chemical contaminants, like phthalates that are emitted from home vinyl flooring.³⁸ However, a high relative humidity is required to achieve this in dust.³⁸ Microbial activity indoors is strongly correlated with moisture availability, 24,39,40 which is generally limited in most circumstances. However, a number of cleaning activities, for example mopping, involve adding water to an indoor environment, which may have a large amount of microbial biomass suspended in an inactive state. Additionally, many chemical compounds found in cleaning products can be metabolized by microorganisms. 41 Thus, cleaning may aid in sustaining indoor microbial communities which are genetically dispositioned to react to intermittent growth conditions and resources to survive. 42 And, while cleaning surfaces often removes microbial biomass, the microbial community can reestablish within a matter of days. 43,44 This dispersal is aided by various occupant activities, as well as by certain chemical species, such as ozone, which may deteriorate or kill microorganisms, as well as chemically react with VOCs to generate particulate matter that can carry microorganisms. 45 Relatively little work has been done to understand how whole communities of microorganisms—bacteria, fungi, viruses, protozoa, and the oxygen-tolerant archaea in dust react to chemicals from cleaning or consumer products or from building materials. 46,47 Further research is needed to determine whether community structure, microbial activity, and biochemical capabilities are altered and how these might impact human health.

Moreover, cleaning chemistry may have intensified a microscopic arms race: Faster evolution occurs in harsh environments. 48,49 Microorganisms may express phenotypic resistance or resilience to our attempts to remove or destroy them, as well as acquire additional genetic resistance through mobile gene elements, thus enhancing their pathogenicity and virulence. 49-51 Despite their prodigious use in cleaning and hygiene products in the late twentieth and early twenty-first century, antimicrobial products do not always reduce infections compared to traditional cleaning products. 52-55 Antimicrobial compounds may persist in built or natural environments, potentially driving microbial evolution as well as environmental exposure to humans, causing alteration of our resident microbiome and the functional benefits it provides. 51,56-58

Chemical disinfection or sterilization of surfaces varies widely by the active ingredients present in products, the target materials and types of microbes, and mechanisms of action. ^{59,60} A non-exhaustive summary is provided in Table 1. However, the presence of certain organic materials can increase microbial tolerance or resistance to cleaning, for example, to sodium hypochlorite. ⁶¹ Therefore, effective cleaning in settings containing significant amounts of carbohydrates, fats, proteins, or other organic materials may benefit from antimicrobial activity coupled with removal or disambiguation of organic materials, that is, "soil." Similarly, cleaning solutions are less effective when microbial attachment and biofilm formation preclude the ability to access cells. In the same way that agriculture is embracing a "many little hammers" approach ⁶² to prevent herbicide resistance



TABLE 1 Summary of cleaning strategies presented

Classification, examples	Benefits	Limitations/risks
Osmolarity disruption		
Acids; acetic acid, chlorine, citric acid	Disrupts cell membranes Easily neutralized	 Microbial tolerance to or metabolization renders substance ineffective
Bases; ammonia, sodium bicarbonate	Low or no toxicity at dilute concentrations	
Alcohols; isopropanol, ethanol		
Quaternary ammonia; Alkyldimethylbenzylammonium chloride, benzalkonium chloride, benzethonium chloride, dialky- ldimethylammonium chloride	 Electrostatic damage to anionic microbial cell membranes Effective against Gram-positive bacteria Stable over time Low odor 	 Less effective against gram-negative bacteria Little activity against bacterial endospores or viruses
Oxidation		
Halogens; iodine, chlorine, fluorine	Creates reactive-oxygen species which disrupt chemical bonds on DNA, RNA, proteins, fatty acids	 Chlorine products may only be effective at certain pH ranges
Hydrogen peroxide Sodium hypochlorite	Immediate effect Broad targeting	
Ozone	Reactivity reduces persistence	Reactivity may damage surfaces
Coagulation		
Alcohols and phenols Aldehydes	 Alters chemical bonds that denature and coagulate proteins Easily neutralizes or evaporates 	Microbial tolerance to or metabolization renders substance ineffective
Ammonia compounds		
Halophenol; chloroxylenol	 Disrupts cell membranes, blocks adenosine triphosphate production 	Resistance common in molds or Pseudomonas aeruginosa
Detergents and surfactants		
Detergents; sodium laureth sul- fate, ammonium lauryl sulfate, amphoteric sodium deoxycho- late, bile salts, Tween, and Triton	 Disrupts lipids, and may lyse cells, via hydrophobic interactions Can decrease cell metabolism Disrupts attachment to surface Cationic detergents disrupt lipopolysaccharide and peptidoglycan Anionic detergents disrupt lipopolysaccharide 	 Disinfection capacity diminished at temperatures <27°C Insufficient detergent activation can stimulate bacterial metabolism Associated with irritation and endocrine disruption of animal cells
Enzyme targeting		
Triclosan	Prevents formation of membranes in bacteria	 Persistence in animal tissues, water, soil, and dust increases human exposure risk and potential for antimicrobial resistance Endocrine disruption of animal cells Suppresses immune cell response in some immunological disorders Not been shown to reduce disease transmission
Triclocarban	Disrupts peptidoglycan in cell wall	
Microbial-based products		
Containing microbial byproducts (enzymes)	 Highly effective against organic material May be effective against microbial biofilms 	 Shorter shelf-life than most commercial cleaning products Limited by enzyme specificity for target, temperature range
Containing microorganisms	 May degrade organic material Successfully incorporated into waste water treatment systems Promising in vitro work but in situ experimental design challenges preclude assessment of efficacy of surface cleaners 	 Promising in vitro work but in situ experi- mental design challenges preclude assess- ment of efficacy of surface cleaners

TABLE 1 (Continued)

Classification, examples	Benefits	Limitations/risks
Metals		
Copper	Immediate effectDisrupts osmolarity and basic cell function; promotes cell death	 Wet copper is less effective than a dry copper surface Oxygen may limit effectiveness
Titanium dioxide	 Photocatalytic properties make it more effective with UV light Effective against prokaryotic and eukaryotic cells 	May be toxic to animal cells

in weeds by using a combination of control methods, so, too, antimicrobial resistance can be delayed or thwarted by combining multiple strategies for cleaning, disinfection, and sterilization.

and may be able to metabolize these cleaning products, particularly if the microorganism was growing in environments containing similar chemicals at tolerable concentrations. 15,65-68

2 | ANTIMICROBIAL CLEANING

Many cleaning products can be considered antibiological more than specifically antimicrobial, as they are intended to remove, inactivate, or decay biological material. However, in the scope of cleaning, we aim to deactivate, destroy, or remove microbial communities on our surfaces; typically, bacteria, fungi, and viruses. As such, most cleaning aims to kill cells or inhibit their activity by targeting cell mechanisms that promote reproduction, proliferation, and survival (Table 1). These include targeting cell wall structure or ability to control transmembrane transport, stifling of target DNA replication, RNA translation, or inhibition of other sites for protein production, with variable efficacy or specificity of celltype targeting (Figure 2). While several antimicrobials have broad action against multiple taxa or domains of microorganisms, a number have specific action (Figure 3), include those listed above. Antibiotics refer to the subtype with action against bacteria; as such, antivirals target viruses, and antifungals target fungi (Figure 3). This distinction seems superfluous, but is a critical component of cleaning product selection by the lay user and the resulting effectiveness at controlling infectious disease.

2.1 | Osmolarity disruption

Simple solutions can be used to lyse and destroy cells by disrupting osmolarity around the cell or chemical disruption of the cell membrane. Simple acid cleaners, such as dilute (10%) solutions of acetic acid (vinegar), can be very effective against cellular debris and non-hazardous to building occupants. ⁶³ The same principle applies to chemicals such as ammonia or alcohols, like isopropanol or ethanol, which are widely reported as effective at minimum (ie, ≤10%) concentrations. ⁶⁴ The benefits to these cleaners include a lack of toxicity toward humans at working concentrations, provided they are not mixed with other chemicals, as well as the relative ease at which they are neutralized, thus reducing environmental contamination. Due to the reactive nature of acids, bases, chlorines, and alcohols, these cleaning products may not remain stable over time, and evaporation or dissociation may change the formulation rendering working batches ineffective. Microorganisms can be tolerant to acidic, basic, or alcoholic conditions

2.2 | Oxidation

The choice of oxidizing or non-oxidizing coagulating cleaners is often down to the target microorganisms, ⁶⁰ as cell wall composition, growth phase, and ability to form spores (sexual structures of fungi), endospores (dormant and resilient structures of some bacteria), or cysts (for some free-living protozoans such as amoeba). Oxidizing agents,

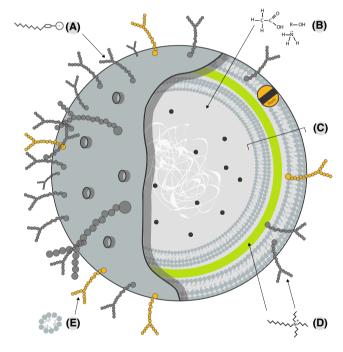
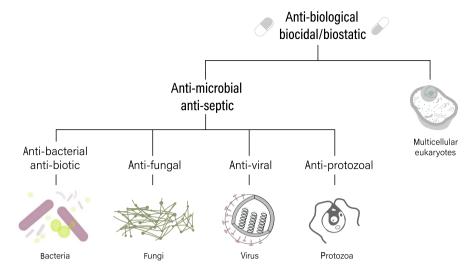


FIGURE 2 Chemical-based cleaning acts on various cellular components which may be specific to a particular domain or cell type or generalizable across many types. (A) Anionic detergents disrupt lipopolysaccharides in the cell membrane of Gramnegative bacteria. (B) Vinegars, ammonia, and alcohols disrupt the osmolarity of a cell. (c) The cell wall of a Gramnegative bacterium is complex, including outer cell membrane, periplasmic space, peptidoglycan, and inner cell membrane. (D) Cationic detergents disrupt the normal activities of peptidoglycan in cell walls and lipopolysaccharides in cell membranes. (E) Detergents (eg, sodium laureth sulfate) disrupt the attachment of cells to surfaces and disrupt lipid membranes through hydrophobic interactions with glycopolysaccharides

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FIGURE 3 The action of a chemical against a particular domain of life or cell type informs its effectiveness at killing or preventing the growth and division of target microorganisms



such as hydrogen peroxide, sodium hypochlorite (bleach), and halogens (iodine, chlorine, and fluorine), create reactive-oxygen species which degrade DNA, RNA, proteins, and fatty acids by disrupting chemical bonds. ^{60,63,69-71} Chlorine products are often only useful at specific pH ranges and can react with organic materials leaving behind toxic byproducts. ⁷² Similarly, oxidizing agents can form harmful secondary compounds when reacting with fragrances and other VOCs. ⁷³

Oxidative sterilization is also achieved using gaseous or aqueous ozone, which is effective against a variety of microorganisms, including known pathogens, relative to the concentration of microbial cells present. Ozone is highly reactive and is not only immediately effective but reacts or dissipates swiftly, thus reducing application and "turn-around time" during cleaning. Due to its reactivity, however, it cannot be used with building occupants present, and it readily reacts with chemicals in air or on surfaces indoors, causing physical damage or creating chemical byproducts which are more toxic to building occupants.

2.3 | Coagulation

Similar to oxidizing agents, chemicals causing coagulation damage DNA, RNA, proteins, and other cellular components; however, this is mainly achieved using alteration of chemical bonds that create cross-linkage and a disfigurement and coagulation of cell components. Non-oxidizing agents or coagulating agents include alcohols and phenols, aldehydes, ammonia compounds, and other chemical mixtures. Alcohol products work better in aqueous solutions, as water acts as a reaction catalyst and it allows proper penetration of cells that would not be otherwise possible in 100% alcohol solutions which have protein coagulase properties. Alcohols are commonly metabolized by microorganisms, allowing them to adapt to the use of alcohol-based cleaners.

2.4 Detergents and surfactants

Detergents or surfactants are not considered disinfectants, as their primary mode of action is to envelope or disrupt attachment of dirt and grease, ultimately removing them from their attached surface or skin. However, their action against lipids can cause them to lyse cells. This is done by using their polar and nonpolar regions to interact with other nonpolar molecules. 78,79 Active ingredients of detergents and surfactants are soluble in water, but contain a hydrophobic portion which makes them ideally suited to disrupting lipids. Common detergents include sodium laureth sulfate (sodium dodecyl sulfate), ammonium lauryl sulfate, amphoteric sodium deoxycholate, bile salts, Tween, and Triton. The effectiveness of detergents is based on their active ingredients and ionic charge-cationic, anionic, or ampholytic-as well as the type of microorganism being treated. Cationic detergents are more effective and work on Gram-positive or Gram-negative bacteria. 80 Anionic detergents, however, may only act against Grampositive bacteria. 80 Not only can detergents lyse cells, but they can decrease their metabolism as well, effectively inactivating them. 80,81 However, the disinfecting capacity of detergents can be diminished at temperatures below 27°C82 and insufficient detergent activation can actually stimulate bacterial metabolism. 80,81 Furthermore, detergents have been associated with irritation and health risks in humans, particularly in endocrine (hormone) disruption. 14,78,83 Recently, detergents have been implicated in altering the human gut microbial community, both directly and indirectly through the alteration of host physiology. 84,85

2.5 | Enzyme-targeting

Lately, a number of new antimicrobial chemicals, which were originally thought to only target specific microbial enzymes, have become popular in cleaning products used in homes, hospitals, and other built environments. As previously stated, "antimicrobial" is a broad term which encompasses the action of many different cleaning strategies. However, it has become the archetype term for a group of relatively new chemical compounds with specific action against microorganisms and which were originally thought to have no effect on other domains of life. Primarily, these have included triclosan (TCS) and triclocarban (TCC). The use of TCS and TCC in

consumer products has been remarkably widespread and includes personal products such as toothpaste, soaps, lotions, shaving cream, and deodorant. Additionally, TCS and TCC are used as a material preservative and thus found in consumer products such as adhesives, fabrics, plastics, textiles, and exposed interior surface coatings. They may also be used as disinfectants which can be sprayed on any surface and wiped down.

Triclosan prevents the formation of membranes in bacteria and causes mitochondrial dysfunction in eukaryotes, but is toxic to a number of mammalian cell types and has adverse effects in many biological organisms, including humans. Not only is TCS a known endocrine disrupter and carcinogen, but it also suppresses immune cell response in immunological disorders such as asthma, allergies, and atopy. TCC is particularly effective against Gram-positive bacteria, but its specific mechanism of action is not well understood. TCC is a suspected endocrine disruptor, a nanomolar inhibitor of soluble epoxide hydrolase (sEH) enzyme, and exhibits high environmental persistence. 89

In response to the information regarding the effect of triclosan and triclocarban on human health, these chemicals have recently been banned in hand soaps and a few other consumer products in the United States, yet are still present in many other products globally. In reaction, benzalkonium chloride has been substituted in as an antimicrobial additive, yet similar concerns over its safety are being voiced. Antimicrobials, like these compounds, are effective at deactivating microorganisms on non-porous surfaces under laboratory conditions. However, the use of these products has no statistical difference in effectiveness for reducing disease transmission when compared to traditional plain soap. 52,54,55,91-93 Promoting basic hand washing hygiene and community education on hygiene, even without the use of antimicrobial additives, reliably reduces incidences of certain transmissible diseases. Thus, their risks may outweigh their benefits to the extent that they are not worth including in products.

2.6 | Resistance and sustainability

Microorganisms have an indefatigable ability to develop survival strategies to adverse conditions, including cleaning solutions and sterilization methods. The advent of modern-day antimicrobial chemicals was seen as the end of unwanted microbial growth, but this optimism was replaced with dismay that these too, were selecting for genetic resistance and fostering the evolution of "super-bugs." 48-51 However, the potential for and development of microbial resistance is nuanced. A year-long study reported that using products containing TCS did not significantly increase antimicrobial-resistant organisms on the skin, 92 but it has been argued that evolution is a product of time and a significantly longer study would be needed to make definitive claims about the ecological impact of antimicrobials. However, the bacterial community found on skin may not be a dependable system to test this, as it is largely comprised of Gram-positive species, which are often able to scavenge fatty acids from host epithelial cells rendering TCS ineffective.⁹⁴ Thus, resistance might not increase because the community is likely to be inherently resistant.

Further, it has also been suggested that there are other factors that contribute to antimicrobial drug resistance in organisms. ⁹² For example, antimicrobial resistance may simply be facilitated in ecosystems with higher microbial biomass and activity than the skin, as well as access to nutrients and moisture, such as the digestive tract or waste water systems. ^{95,96} Resistance may also be facilitated in areas with more consistent presence of antimicrobials such as building products, ^{46,47} which may be imbued with chemicals during manufacturing or may accrue chemicals from cleaning or hygiene produce residues. Mammalian host tissues readily absorb TCS, but sequester it in areas of the body apart from the host's microbial communities.

Antimicrobial compounds may persist in built or natural environments, driving microbial evolution as well as environmental exposure to humans, causing alteration of our resident microbiome and the functional benefits it provides. ^{51,56-58} Recent studies have shown that TCS can accumulate in indoor dust and potentially affect the microbial communities within, for example, by favoring organisms carrying antibiotic resistance genes ⁴⁷ or changes in cellular morphology and functions which can confer for antimicrobial resistance. ^{46,96}

The concern over development of microbial resistance to chemical cleaners extends beyond the location and scope of their use; leftover residues and environmental contamination from chemical cleaners can also contribute. Some surface disinfectants are able to dissociate into non-hazardous components, or may be used at low concentrations for extended periods without long-term health effects, 97 and are thus considered environmentally friendly and safer to use, including hydrogen peroxide, bleach, alcohol, and isopropanol. Others, such as halogens, are considered environmentally friendly at low concentrations. But large-scale use, particularly the use of chlorine in water and food decontamination, has led to concerns about volatilization, as well as the possibility of selecting for microbial resistance. 70,72,98,99 More complex compounds require disposal at specific facilities to prevent contamination of water systems, making them more suitable for use in industrial settings where disposal regulations are enforced. 100

As discussed, there are several different mechanisms by which antimicrobial cleaning works to destroy cells or disrupt their cellular functions. While some compounds may be effective at their proposed activity, they have not been proven to be significantly more effective at doing so than plain soap. Use of these antimicrobial products is likely propagating antimicrobial resistance among microbial communities. Although there are many products and methods that remove microbes from the system without using harsh chemicals, the byproducts produced can be just as or more harmful than the initial intervention.

3 | PRO-MICROBIAL CLEANING

Cleaning products which mimic or utilize microbial products, namely enzymes, have long been implemented as a method for degrading

residues of synthetic or organic chemicals, including antimicrobial compounds, and inactivating or killing pathogens. Biofilms on medical equipment can be difficult to remove and contribute to the spread of infectious disease between healthcare patients. ¹⁰¹ Certain solutions of enzymes mixed with biofilm removal cleaning solutions showed promise by removing biofilms more effectively than control solutions. ¹⁰¹ Enzyme-based cleaning formulations were even viable and effective after 24 weeks of storage at room temperatures. ¹⁰¹ However, enzymatic cleaning solutions are limited by enzyme specificity, only functioning at specific temperatures with limited activity duration, and many enzymes only inhibit growth for certain strains of bacteria.

Similarly, incorporating microorganisms into systems to degrade materials has been effectively used in water systems, including aquariums, wastewater treatment plants, bioswales, green infrastructure, and more. 102 More recently, alternatives to traditional household cleaners and disinfectants have become commercially available that include viable bacterial species chosen for their competitive abilities. Currently, most commercially available formulations are comprised of one or several bacterial species of Bacillus, with a few incorporating fungal species. Probiotic cleaning solutions aim to kill pathogenic bacteria while leaving a residual coating of non-threatening probiotic bacteria to form a protective biofilm. The probiotic microorganisms on the surface produce enzymes which are capable of removing dirt and grime from the surface, or antimicrobial compounds to prevent further microbial accumulation. Probiotic cleaners embrace the theory that microbial competition can be harnessed to effectively control microbial populations in the built environment, by removing or selectively altering biomass and antimicrobial resistance gene abundance. 103

Bacilli are well-known for their ability to survive adverse environmental conditions, form endospores, and produce antimicrobial compounds 104,105 and have also been demonstrated to antagonize a handful of bacterial pathogens, ¹⁰⁶ even on non-porous surfaces. ¹⁰⁷ Moreover, Bacillus subtilis poses little threat to individuals who are not immunocompromised and is commonly used as a dietary probiotic. 108 Bacilli are capable of degrading organic materials, ammonia, sulfides, and more, 109 and some are capable of producing surfactants which can disrupt a variety of biological and synthetic oils. 110 Yet, bacilli are commonly found in all environments and may compete with each other, possibly precluding the ability of the probiotic cleaner species to form a biofilm. 106,111,112 They may also compete with other common environmental bacteria and fail to persist on a surface, 113 potentially requiring continuous or frequent application. Further, environmental context can alter microbial communities: Exposure to outdoor dust enhanced the biofilm formation of a number of bacterial pathogens as well as their ability to proliferate in co-culture with human cells. 114 While plenty of information about commonly used probiotic species is available on their capacity in the laboratory, information justifying the selection of these as probiotic species is less abundant. For example, data stemming from studies involving in vitro experimentation using dust communities or in situ experimentation in complex, occupied built environments is lacking.

Studies have claimed that daily cleaning with the Probiotic Cleaning Hygiene System (PCHS), comprised of equal parts Bacillus subtilis, Bacillus pumilus and Bacillus megaterium, over sustained periods of time can lead to a reduction of viable bacterial pathogen colony forming units (CFU) present on hospital surfaces by >89%. 115,116 However, current literature lacks experimental design detail or rigor: one study did not include any experimental or technical replication and was not able to provide statistical support for reduction claims. 116 This same combination of bacilli showed a significant reduction in CFU/m² as well as fewer reported hospital acquired infections during the treatment phase. 117,118 However, compliance to hospital cleaning procedures is a persistent issue, and it is unclear from the reported study designs whether the probiotic treatment reduced pathogen biomass and hospital acquired infections, or simply the retraining of cleaning staff, though minimal, for use of the probiotic cleaning which caused a re-adherence to cleaning protocols was sufficient to decrease bacterial load. 54,119-121 The previous study, which did use replication, also utilized randomized cleaning product treatments to which hospital cleaning staff were blind, and a return to conventional cleaning methods after a probiotic treatment, 115 providing demonstrable promise to the use of PCHS in field trials. Yet it is also important to note that many hospital-acquired infections follow a seasonal trend in infection rates affected by humidity or other factors 122,123; thus, a short-term cleaning trial implemented during a downward trend in HAIs may yield false positives.

Another study showed that mixing probiotic cleaning solutions with phages in vitro yielded higher rates of destruction of multidrugresistant pathogens which can be extremely helpful in hospitals and schools where surfaces can hold larger amounts of pathogens due to the constant movement and transfer of people and microorganisms. 124 The study showed that this mixture can be effective at low dosages against microorganisms on fomites typical of a hospital or workplace, over a duration of time (up to 13 days), and the mixture is sustainable for long-term use meaning the probiotics or phages remain viable in solution longer than they would alone. While phages are often host-specific, a number of bacteriophages may also affect humans, and if ingested, inhaled, or contacted via skin, the phage present in cleaning solutions could potentially remodel the host microbiome. 125,126 However, phages die quickly after being dispersed into the environment; thus, the potential for exposure to most building inhabitants is low. Nonetheless, newly cleaned surfaces could be dangerous to the public immediately after phage application, and cleaning personnel are subject to deleterious effects of phage exposure due to their continual spatial proximity to the cleaning solutions.

Although many products already exist on the market, microbial-based cleaners are still a nascent technology that lack rigorous laboratory and field testing on long-term microbial community and human health effects. As food additives, microorganisms or products may be categorized by the United States' Food and Drug Administration (FDA) as "generally recognized to be safe" (GRAS), often based on their ubiquity in products and lack of evidence of harm. As a component of consumer cleaning products, there are no regulations in place for evaluating safety or efficacy.¹⁰⁹ Fortunately,

the vast majority of bacilli have not been identified as causative agents of disease in humans.

Another important consideration is eradication versus inactivation of transposable microbial elements, such as antimicrobial resistance genes. Microorganisms are capable of horizontal gene transfer (HGT) or moving genetic material between cells. HGT is accomplished with specialized cell components that trade DNA between cells, acquire relic DNA from the dead cells in the environment, and transfer viruses between cells. Bacilli-based cleaners did not completely eradicate all antibiotic-resistant strains of bacteria from surfaces. Rather, they appeared to kill or outcompete the other bacterial cells to ingest the antimicrobial resistance (AMR) genes. 118,127 Although several studies found correlations between the use of probiotic cleaning solutions and reductions in hospital-acquired infections, a vital part of their reasoning is missing: longevity. The persistence of a non-pathogenic biofilm that ultimately inhibits the growth of multidrug-resistant strains of bacteria has not been shown to survive longer than a few days.

One of the major limitations that comes with the growing interest of these biologically based cleaning solutions comes with quality control and quality assurance. As of recently, there has been little regulation by most government organizations, and this has been reflected in inconsistent cell counts in products, effectiveness of use, and formulation. Moreover, the wider implications toward health have not been considered, as these formulations may come into contact with skin, whereby they might be accidentally ingested, as well inhaled, all of which can modulate the infectious potential of strains. To date, the microbial species known to be included in these products fall in the category of GRAS, but the potential for infection, allergy, immune sensitization, or the production of undesirable MVOCs nevertheless remains. 128

4 | PHYSICAL REMOVAL OR ADHESION PREVENTION

The physical removal of microorganisms in air or water systems can be achieved using mechanical filtration. For stationary surfaces, physical removal is achieved via cell destruction, or detachment from surfaces using abrasion or suction. Depending on the microorganism(s) requiring removal, the surface or environment (eg, pipes, air ducts), and the accessibility of the microbial community, many methods of physical removal have limited applicability. However, the complete removal of all microorganisms is not always the goal of surface cleaning, nor is it an easily achievable one. Physical removal may not equate to destruction of the cells, and these methods do not rely on chemical intervention, although incorporating both physical and chemical methods may render cleaning more effective and discourage the development of antimicrobial resistance.

Physical removal with light abrasion and water, even without the addition of a soap emulsifier or disinfectant, can remove some microbial biomass, ^{53,71} as well as inactivate some viruses. ¹²⁹ For example, wet mopping and moist mopping are the most effective ways

to reduce the presence of organic material from floor surfaces. 130 Even though moisture can foster microbial activity, it also improves the ability of microorganisms to be removed from surfaces to which they are adhered. 131 However, the physical abrasion employed by common cleaning methods does not completely remove all microorganisms. 43,71,132 While the act of cleaning and repetitive use of cleaning materials (eg, using the same rag to clean multiple surface) can spread unwanted organisms. 132 the goal of physical abrasion during cleaning is to reduce microbial biomass and organic material on a surface. When microbial cells accrue, the density of the community may trigger phenotypic changes to cause the formation of a biofilm, the strength of which is often determined by oxygen or nutrient content, community structure, and surface characteristics. In the built environment, microbial biofilms commonly form in water systems or places of excess moisture. During their early stages of formation, biofilms can be removed using abrasive methods such as pressurized air or water, but once established, simple abrasion often fails to remove all or any of the microorganisms. 132 Once cells begin to adhere to the surface, chemical sterilization remains the most effective at removing biofilms. 133,134

Physical destruction of cells may involve radiation, desiccation, heat, and/or pressurization (eg, autoclaving). Visible spectra of light, as well as ultraviolet light, triggers damage to cellular components by changing their biochemistry, reduces microbial biomass in dust, 135 and inactivates some pathogens. 136 It is well-understood to be biocidal for many microorganisms, via disruption of chemical bonds in proteins and nucleic acids which cause lethal physical complications. Methods like non-ionizing UV-C and pulsed xenon UV systems have been shown to be effective at killing multidrug-resistant strains, and review of their efficacy showed they were both equally effective at destroying these strains on non-porous surfaces. 137-139 In the case of UV-C, it is assumed that microorganisms are unable to form resistance because UV-C does not normally reach Earth's surface due to atmospheric spectral filtering. However, laboratory trials of UV-A and UV-B exposure highlight the ability of microbial communities to enhance their radiation resistance over time due to insufficient exposure¹⁴⁰ and would suggest that UV-C resistance is possible if effective protocols are not established and adhered to. Further, UV light can cause increased emission and chemical changes to VOCs and physical degradation to building finishes and furnishing. 141

Photocatalytic chemicals, typically used in conjunction with UV light treatment, can imbue biocidal properties to surfaces. Titanium dioxide (${\rm TiO_2}$) has biocidal properties against bacterial and eukaryotic (including human) cells, but when used in conjunction with UV light treatment will act against microorganisms. 142,143 When coated or imprinted onto surfaces, ${\rm TiO_2}$ reduced adhesion of the pathogen *Streptococcus mutans*, and a synergistic effect between chemical concentration, type of application onto surface, and use of UV light treatment was observed. 144

A more proactive solution to physical removal is to prevent the accrual of microorganisms and their ability to adhere to surfaces, ¹⁴⁵ by utilizing the surface properties of different materials. Chemical composition and micro-topography affect the cleanability of indoor

surfaces; as a result, cleaning product efficacy varies between surface materials, and engineered self-cleaning surfaces are designed to lessen the necessity for regular cleaning intervention. Surface composition may be biostatic or biocidal to microorganisms, or their composition or physical structure may inhibit microbial attachment—"self-cleaning surfaces."

A number of metals are inherently antimicrobial through chemical toxicity, and most notable among them is copper for its low risk of toxicity to humans and its ability to kill microorganisms in as little as one minute. 146-148 While the mechanism is not known, it is hypothesized that once a microorganism contacts the copper surface. it is subject to cell membrane damage, allowing excess copper influx that results in lethal oxidative damage to DNA. 146 Copper can reduce pathogenic species on surfaces more effectively than traditional stainless steel appliances, which are common in hospitals. 148,149 However, dry copper surfaces are significantly more effective at contact-killing pathogens than wet copper surfaces, indicating a potential limitation to its application in the presence of oxygen and suggesting a mechanism for resistance to copper oxides which is not effective against copper ions. 147 Like other methods of disinfection, microorganisms have also displayed mutations and modulation of cellular mechanisms which promote tolerance or resistance to toxic metals, including copper. 150 Few studies have been conducted that show the effects of combined cleaning with copper surfaces, but those few indicate that it may actually be more beneficial to leave the surfaces alone rather than adding supplementary cleaning, as it can deteriorate the copper and render it less effective. 149,151

Hydrophobic surfaces are the most abundant and fundamental products in the self-cleaning market. 152 These surfaces were originally designed as a biomimicry of the lotus leaf, which effectively removes debris by maintaining water droplet surface tension down a funnel gradient. 153 This type of design is typically seen with stain-resistant fabric in which the substance applied to it beads and rolls off the fabric. 154 Generally considered as hydrophobic surfaces, Teflon, PEHD, and PVC have been found to increase cell adhesion in certain strains of bacteria such as Bacillus subtilis, Bacillus cereus, and Escherichia coli. 145 Interactions between the surface chemistry of the cell wall and the chemistry of the surface itself affect the cell's ability to adhere to the surface as previously stated. A second type of selfcleaning surface involves a hydrophilic response of the surface and the substance applied to it: Hydrophilic surfaces tend to use metal oxides to "sheet" the water which removes dirt from the surface. 154 These types of coatings are typically produced using nanostructured material which is water repellent, resistant to corrosion, and stable under ultraviolet radiation, and have several antimicrobial properties like preventing cell adhesion to surfaces. 154,155

While much of the research using surface micropattern and material for microbial control is promising and could potentially reduce the need for vigorous cleaning or chemicals, there are several limitations to the field itself. Mimicking nature using mathematical models to produce these self-cleaning surfaces has proven technically challenging. There are a number of additional considerations for developing these self-cleaning products, such as surface roughness, contact

time, synthesis or replacement of the material, and intended use. One of the main benefits to using a copper surface is that the age of the copper itself is not a detrimental factor to the effectiveness of it; one study showed that having 6-month-old copper features in hospital settings did not affect the efficacy of microbial reduction in the environment. Further, the appealing aspect of self-sterilizing or self-cleaning surfaces is not only their reduced need for cleaning, but also their presumed lack of toxicity toward humans, but this is often related to dose. The toxicity of metals toward microorganisms is increased when incorporated with nanoparticles, ¹⁵⁷ and the effects on humans are largely unknown. However, the chemistry of these surfaces may change over time, surfaces wear, ¹⁵² and a lack of moisture ¹⁵³ or accrual of debris can all render the surface ineffective, further complicating our knowledge of chemical leaching or reactions.

5 | COMPLEX SOLUTIONS FOR COMPLEX CHEMISTRY

In recent years, the primary motive for cleaning has been infection prevention, yet despite the advances in chemistry and microbiology in the last several centuries, evidence-based infection control remains elusive: Proving the effectiveness of particular cleaning interventions in reducing infectious organisms or disease symptoms varies widely by type of intervention or is often absent altogether. This is true for studies that pertain to infection control on nonhuman surfaces. For one, the definition of "clean" is vague and relies on personal or cultural definitions, as well as temporal context. How often do surfaces need to be cleaned to maintain them free of organic matter and organisms? How long do surfaces need to be free from organic matter? From all living microorganisms or just certain ones? On that point, the methodology for measuring the presence of microorganisms, cellular debris, or general soil varies 119,161 and is often measured indirectly through observed infection rates. 119,161

This disparity may have a number of root causes, including experimental design challenges, unexpected in situ conditions, resilience of the microbial community, and fluctuating adherence to cleaning product instructions and institutional protocols. Many cleaning studies in occupied buildings lack randomization or comparable control groups, implement multiple interventions simultaneously, or conflate treatment effects with seasonal trends in infections. Studies may also generate false positives when introducing a new product; making staff aware that samples are being collected (ie, The Hawthorne Effect ¹⁶²) can cause cleaning personnel to re-adhere to stringent cleaning practices, which have a tendency to lapse over time.⁵⁴ It is possible that by maintaining compliance to enacted protocols, regular staff trainings, and effective staff communication to prevent certain hospital equipment from being overlooked in cleaning protocols, 120,121,163-165 cleaning can be effective regardless of the products used. For example, basic hand hygiene is nearly uniformly effective at reducing disease transmission regardless of the cleaning products used, 4,6,52,166 while environmental cleaning can have mixed outcomes ranging from positive to unsatisfactory. 121,167-169 The efficacy of most chemical cleaning products is limited to the application, duration, and thoroughness of cleaning, the latter of which is subjective to each person. 120 While cleaning practices can be mandated in work environments, it is impossible to ensure total compliance to stringent cleaning policies. Even if practices are rigorously followed, built environments are difficult to maintain due to consistent traffic of individuals. This high rate of individual occupant turnover increases the rate of microorganismal and chemical exchange between the built environment and outside environment, making it difficult for cleaning protocols to serve a consistent environment.

While it is undeniable that the act of cleaning can be effective at reducing the microbial load on surfaces, it is an understatement to say that the influence of the act of cleaning is variable among methods. Many of the products and practices reviewed here and, elsewhere, may be effective for specific purposes but still require significant investigation to be understood in the context of human health or microbial ecology. Notably, the effectiveness of many cleaning products or behaviors is highly situational, a nuance which may be lost on consumers. Moreover, the action against monocultured organisms may not accurately reflect the action of a chemical against whole communities of microorganisms, highlighting the need for more study utilizing complex communities and broader genomic targets (ie, shotgun metagenomics). The disruption of the microbiome, not only human but that of the built environment, needs to be considered within the developing context of cleanliness and beneficial microbiota.

Many of our current practices, products, and ideology center around the context of "scorched-earth cleaning"-sterilizing and removing everything-leading to growing concern among the scientific community about the potential for evolutionary effects on microorganisms. Not only is the complete removal or sterilization of the indoor environment potentially unachievable, but recent work into host-associated microbial communities reveals this to be contrary to proper immune system development and health, though it is unclear, and outside the scope of this review, what role the microbial community of the built environment may play in the development of the human immune system. The antithesis to sterilization, fostering a surface microbial community or using probiotic cleaners, presents some promising results but are too nascent and in situ testing has proved difficult for evaluating these in controlled settings. We are only just beginning to understand the complexities that result from our symbiotic relationships with our resident microorganisms, and even more recently, to unravel the intricacies of our microbiome and health with respect to microbial and chemical exposures mediated by the built environment.

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CONFLICT OF INTEREST

The authors declare no conflict of interest.

AUTHOR CONTRIBUTIONS

SI conceived of scope and contributed to writing and editing. SV performed literature review, writing, and editing. KVDW, WG, LD, MF, PH, and SN contributed to writing and editing, and JH, JS, CB, YX, and EM contributed to editing and provided valuable insight.

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