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Nutrient Zinc at the Host-Pathogen Interface

Zachery R. Lonergan^{1,2}, Eric P. Skaar^{1,3,*}

¹Department of Pathology, Microbiology, and Immunology, Vanderbilt University Medical Center, Nashville, TN;

²Microbe-Host Interactions Training Program, Vanderbilt University School of Medicine, Nashville, TN;

³Vanderbilt Institute for Infection, Immunology, and Inflammation, Vanderbilt University Medical Center, Nashville, TN

Abstract

Zinc is an essential cofactor required for life, and as such, mechanisms exist for its homeostatic maintenance in biological systems. Despite the evolutionary distance between vertebrates and microbial life, parallel mechanisms exist to balance the essentiality of zinc with its inherent toxicity. Vertebrates regulate zinc homeostasis through a complex network of metal transporters and buffering systems that respond to changes in nutritional zinc availability or inflammation. The fine-tuning of this network becomes critical during infections, where host nutritional immunity attempts to limit zinc availability from pathogens. However, accumulating evidence demonstrates that pathogens evolved mechanisms to subvert host-mediated zinc withholding, and these metal homeostasis systems are important for survival within the host. Here we discuss mechanisms of vertebrate and bacterial zinc homeostasis and mobilization, as well as recent developments in our understanding of microbial zinc acquisition.

Zinc is Required for Life

Transition metals (see Glossary) are essential micronutrients required to carry out biological processes in all domains of life [1]. Their requirement stems from unique biochemical properties attributed to late *d*-block elements that selected for their incorporation into catalytic and structural components of proteins during evolution. The essential metal zinc (Zn) is unique among the first row *d*-block elements in that it possesses a filled *d*-orbital and does not undergo redox cycling. Zn is ubiquitous in life and is required for the structure or function of thousands of metalloproteins [2]. Zn is an essential micronutrient for the survival and proliferation of bacteria, including pathogens that are major causes of morbidity and mortality worldwide [3]. Given this essentiality, both vertebrate hosts and pathogens evolved processes to maintain Zn homeostasis. While

*correspondence: Eric Skaar, Ph.D., M.P.H.; eric.skaar@vumc.org.

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mechanisms of maintenance and storage of Zn within vertebrates have been well described in recent decades [4], the diverse array of systems that pathogens possess to compete for this essential nutrient during infections has only recently been appreciated [3]. In this review, we discuss vertebrate Zn homeostasis and mobilization during infections, followed by a summary of our current understanding of bacterial Zn uptake and utilization systems.

Vertebrate Zinc Homeostasis is a Highly Regulated and Dynamic Process

As the second most abundant **trace metal** in humans, around 2 to 4 grams of Zn are distributed throughout the body in various tissue sites [5]. In vertebrates, the gastrointestinal tract serves as a major regulator of Zn homeostasis by tuning absorption and excretion via metal transporter proteins [4] (Figure 1). While the range of total Zn within the body is relatively stable for most humans, changes in dietary Zn availability are associated with poor health [4]. Human genetic deficiencies in Zrt/Irt-like Protein (ZIP) family members, which are the primary Zn importers in eukaryotes, cause severe Zn deficiency [6]. Additionally, dietary Zn deficiency is a major public health burden, with at least 25% of the global population being at risk of inadequate dietary Zn [7, 8]. Consequences of Zn deficiency include impaired immune function, delayed wound healing, diarrhea, and increased susceptibility to infection [9]. Conversely, Zn toxicity occurs in acute and chronic forms and is linked to several sources, including Zn supplements and parenteral nutrition, among others [10, 11]. Acute symptoms of Zn intoxication include symptoms like nausea and vomiting [12], while chronic intoxication may result in reduced immune function and altered copper (Cu) and iron (Fe) levels, demonstrating the interconnectedness between nutrient metals within the body [13].

Host tissues and circulating cells evolved mechanisms to control Zn levels to mitigate against the adverse effects of Zn deficiency and toxicity on human health. In circulation, serum Zn is mostly bound by albumin, transferrin, and α_2 -macroglobulin but remains accessible to transporters to balance Zn levels within cells [14]. The primary regulators of mammalian Zn transport are the ZIP Zn importers and the Zn-transport (ZNT) exporters (Figure 1). There are 14 isoforms of mammalian ZIP transporters that share many structural and functional properties, where they facilitate the import of Zn, and sometimes other cations, into the cytoplasm from the extracellular space or from intracellular vesicles and organelles [15]. However, aside from ZIP1 which is found on all cells, expression of the other ZIPs varies depending on the cell type [16]. Likewise, the ZNT family includes 10 isoforms, some of which are expressed ubiquitously, while others are cell-type specific [16]. ZNTs generally function in reverse of ZIPs, where they export Zn out of the cytoplasm into circulation or cellular vesicles and organelles. Expression of a subset of ZIPs and ZNTs is controlled by the Metal response element-binding Transcription Factor-1 (MTF-1) [17]. MTF-1 is a cytoplasmic transcription factor that undergoes nuclear translocation following Zn binding, where it regulates certain ZIPs and ZNTs to maintain Zn homeostasis [15]. However, not all Zn transporters are MTF-1-regulated, and the diversity of expression and localization of ZIPs and ZNTs highlights the importance of fine-tuning Zn levels depending on the tissue and cell type.

Aside from direct import and export of Zn, intracellular Zn sequestration also must occur to prevent toxicity. A major component of Zn chelation within eukaryotic cells occurs through the action of **metallothionein** and **glutathione**. Metallothionein and glutathione are cysteine-containing molecules that link reversible Zn binding with the cellular redox state [18, 19]. This reversible Zn binding allows these molecules to serve as a component of the intracellular Zn buffering system that includes the potential delivery of Zn to apofoms of **metalloenzymes** [20]. Metallothionein is synthesized in several organs, including the liver and kidneys, and can be found in circulation [21]. *In vitro* estimates demonstrate that sulfur-containing molecules like metallothionein control approximately 30% of a cell's Zn buffering capacity [22]. However, the exact dynamics of Zn buffering are likely dependent on cell type and environmental conditions.

Alterations in nutrient Zn availability affect many biological processes, and mechanistic studies have identified how immune system function changes in response to Zn fluctuations [14]. The impact of Zn deficiency on immune system function has been reviewed in-depth previously [14, 16]. Importantly, altered Zn concentrations affect both innate and adaptive branches of the immune system. Within innate immunity, excess Zn can induce chemotaxis of neutrophils, which are one of the primary cell types responsible for innate immune-mediated pathogen clearance [23]. Zn deficiency impedes the antimicrobial activity of neutrophils and macrophages by altering the oxidative burst and inhibiting phagocytosis [24, 25]. The maturation of dendritic cells (DCs), which connect the innate and adaptive immune system through antigen presentation, is also modulated by Zn availability. DCs experiencing Zn deficiency increase expression of major histocompatibility complexes and costimulatory molecules, while Zn excess inhibits this upregulation [26]. In the adaptive immune system, T cells and various T cells subsets are susceptible to alterations in Zn availability; insufficient Zn decreases T cell maturation while increasing apoptosis [27]. Changes in Zn homeostasis alter the balance of T_H1 , T_H2 , and T_H17 subsets, and Zn supplementation promotes regulatory T cell induction and T_H9 differentiation [28–31]. B cells appear to be impacted less by Zn deprivation, but they too experience a reduction in total cell numbers as well as alterations in development and antibody production [32, 33]. Collectively, these studies establish Zn homeostasis as a critical determinant of vertebrate survival and immune cell function.

Zinc is Mobilized During Microbial Infections

In addition to altered immune function in response to changes in dietary Zn, Zn availability can also modulate the response of the vertebrate host to infection and inflammation. During the acute phase of inflammation, serum Zn drops substantially due to proinflammatory cytokines altering Zn transporter expression, which results in the accumulation of Zn-bound metallothioneins by hepatocytes that capture incoming Zn [34] (Figure 2). This decrease in bioavailable Zn is part of an immune response known as **nutritional immunity**, which was first described for sequestration of Fe from invading microbes [35]. However, our understanding of nutritional immunity has expanded in recent years to include other nutrients, such as manganese (Mn) and Zn [36]. Nutritional immunity is implicated as a critical host defense mechanism for many types of pathogens, and the role of nutrient metal withholding during fungal infections has been reviewed previously [37].

Host strategies to limit Zn availability from bacterial pathogens include both cell-mediated Zn restriction and extracellular Zn sequestration. Cell-mediated Zn restriction occurs mainly through the activity of Zn transporters. For example, the ZIP8 Zn transporter is expressed by immune cells, where it associates with the lysosomal transmembrane glycoprotein Lamp1 and decreases lysosomal Zn levels [38] (Figure 2). These findings suggest that Zn is actively removed from the lysosome as a strategy to limit nutrient Zn from pathogens trapped within this cellular compartment.

Extracellular Zn sequestration mechanisms implicate members of the S100 protein family as being critical for Zn limitation during infections (Figure 2). S100 proteins are EF-hand calcium-binding proteins found in vertebrates that serve important functions in basic physiology and in the host inflammatory response [39]. A unifying feature among S100 proteins is that they form dimers and may form transition metal binding sites at the dimer interface [39]. S100A8 and S100A9 are unique among S100 proteins in that they preferentially form heterodimers [40]. The heterodimeric S100A8/S100A9 protein complex is called calprotectin (also known as calgranulin A/B or myeloid-related protein 8/14). Calprotectin (CP) is involved in many biological processes, including serving as a **damage-associated molecular pattern (DAMP)** and as ligands for Toll-like receptor 4 (TLR4), the receptor for advanced glycation end products (RAGE), and CD33 [41–43]. CP is readily detected at infectious foci during infections [44] and accounts for more than 40% of the cytoplasmic protein content of neutrophils [45], underscoring CP's importance during an immune response.

Part of CP's immunological importance is due to the protein's ability to chelate nutrient metals. At the dimer interface between S100A8 and S100A9, two metal binding sites are formed that are termed Site I and Site II. Site I possess broad metal-binding capabilities, including ability to bind Zn, Mn, Fe, Cu, and nickel (Ni) [44, 46–50]. Site II only coordinates Zn with high affinity [47, 51, 52]. The two metal binding sites are important for CP's antimicrobial activity, as demonstrated by numerous microbial pathogens displaying growth inhibition *in vitro* in the presence of CP [44, 50, 53–57]. Importantly, addition of exogenous Zn and Mn is generally sufficient to reverse CP-mediated growth inhibition, which demonstrates that metal-binding by CP is adequate to limit microbial growth [39]. However, other nutrient metals have been implicated in binding by CP, including Ni and Fe, among others. While the relative contribution of CP to Ni withholding is not yet known, Fe is increasingly recognized as a metal restricted by CP [57, 58]. Further, mice deficient in producing the CP heterodimer (*S100A9*^{-/-}) have altered infection susceptibility, demonstrating that CP is critical to infection outcome [44, 46, 53, 54, 59–62].

Other S100 proteins are also implicated in Zn withholding at the host-pathogen interface, including S100A7 and S100A12 (Figure 2). S100A7, also called psoriasin, functions as a homodimer and is constitutively expressed in the skin and at mucosal surfaces. S100A7 binds two Zn ions across the dimer interface [63]. Similar to CP, Zn withholding by S100A7 may contribute to limiting metal availability from bacteria [64]. S100A12, also known as calgranulin C, also functions as a homodimer and binds Zn and Cu at its dimer interface [65]. Recombinant S100A12 can inhibit microbial growth through Zn chelation [66, 67], but its broader contribution to immunity has been difficult to define due to its absence in mice

[39]. Additionally, the contribution of S100A12 to Cu withholding during bacterial infections is largely unexplored.

Zn mobilization occurs during infections not only to sequester the metal from invading pathogens, but perhaps to be trafficked within immune cells to impart toxicity (Figure 2). While Zn is redox-inactive, the **Irving-Williams series** predicts that the high affinity of Zn for metal binding sites promotes aberrant loading of Zn to non-Zn proteins, which leads to toxic effects through **mismetallation** or other indirect mechanisms [68]. Macrophages infected with *Mycobacterium tuberculosis* accumulate Zn within the cell that is sufficient to induce bacterial Zn intoxication [69]. Additionally, internalization of *Streptococcus pyogenes* by human neutrophils results in Zn mobilization that may induce bacterial Zn poisoning [70]. Conversely, macrophages infected with the fungal pathogen *Histoplasma capsulatum* also accumulate Zn, but this Zn is shuttled from the phagosome to the Golgi apparatus in a granulocyte macrophage-colony stimulating factor (GM-CSF)-dependent manner [71]. Downstream consequences of this Zn shuttling include generation of reactive oxygen species (ROS) to inhibit *H. capsulatum* growth, presumably by preventing appropriate metalation of Zn/Cu superoxide dismutase enzymes [72]. These results imply that Zn accumulation within immune cells, mediated by shuttling of Zn out of the phagosome, indirectly diminishes pathogen viability [72]. These findings suggest that both Zn starvation and toxicity are employed by the host to limit microbial survival, but precise situations in which starvation or toxicity may be utilized is not well-defined.

Metalloregulators Control Bacterial Zinc Homeostasis

In response to metal restriction by the host, microbes have evolved mechanisms to subvert nutritional immunity during infections. The bacterial response to changes in metal availability is primarily mediated by **metalloregulatory proteins**, although metabolite-sensing and riboswitch-mediated sensing systems are also described [73]. Generally, these metalloregulators sense changes in cellular metal concentrations and alter gene expression of metal homeostatic systems. These metalloregulators are widely distributed in bacteria, and they respond to metal limitation, metal intoxication, or both conditions. The diversity in bacterial metalloregulators has been reviewed previously [74], therefore we will focus here specifically on mechanisms of bacterial Zn homeostasis.

As a bacterial cell experiences Zn starvation, transcriptional changes must occur to counterbalance these conditions. Many diverse Zn-responsive transcriptional regulators have been identified and reviewed previously [75]. In many bacterial pathogens, the primary regulator for Zn homeostasis is the Zn uptake regulator (Zur) [76, 77]. Zur is a member of the ferric uptake regulator (Fur) family of metallosensing DNA-binding proteins. Metal-sensing by Fur family members is directly mediated by metal binding to the metalloregulator, which induces conformational changes and alters the regulator's affinity for DNA [78]. Zur is exquisitely sensitive to Zn fluctuations and senses changes to Zn concentrations in the femtomolar (10^{-15}) range to regulate transcription [79]. In Zn-replete conditions, the metal binding sites of Zur are predicted to be occupied. The increased affinity of Zn-bound Zur for DNA permits the metalloregulator to recognize and bind to conserved palindromic inverted repeat regions, termed Zur boxes, in the DNA promoter

region of its **regulon** [78]. The Zur box location generally overlaps with motifs required for effective RNA polymerase recruitment, thereby inhibiting gene transcription. Upon Zn starvation, an increasing proportion of Zur is no longer bound by Zn, which diminishes the affinity of Zur binding to DNA and results in derepression of the Zur regulon [80].

Derepression of the Zur regulon induces several physiological changes, and general themes will be explored for Gram-positive and Gram-negative bacteria.

Zinc Sensing in Gram-Positive Bacteria

Mechanistic studies into the response of Gram-positive bacteria to Zn starvation have been conducted in several organisms [73]. In *Bacillus subtilis*, Zn starvation is sensed by the metalloregulator Zur [77]. However, non-Fur family Zn metalloregulators have been identified as well; for example, *Streptomyces pneumoniae* controls Zn uptake through the MarR family member AdcR [81]. Precise investigations into Zn sensing by *B. subtilis* Zur revealed that the regulator possesses differential activity, corresponding to varying DNA affinity, depending on the number of Zn-binding sites occupied [82] (Figure 3). This differential activity permits a fine-tuned response to Zn starvation that occurs in a step-wise fashion with three distinct stages [83]. First, non-Zn utilizing ribosomal proteins L31* and L33* are expressed, which replace Zn-requiring isoforms to effectively decrease the total cellular requirement of Zn and promote Zn mobilization. Second, the high affinity Zn uptake ABC transport system genes, *znuABC*, are derepressed to promote Zn acquisition. The ZnuABC system is widely conserved across many species and therefore represents a major metal acquisition system, although other transporters also promote Zn uptake (Table 1). Along with *znuABC*, the predicted Zn metallochaperone gene *zagA* (formerly *yciC*) is also derepressed [84, 85]. Finally, further Zn starvation leads to induction of an additional alternative ribosomal protein S14* to sustain protein synthesis and the Zn-independent GTP cyclohydrolase I FolE2, which permits the continuation of *de novo* folate biosynthesis, which is a critical metabolite for life [83]. While the extent to which this graded response occurs in other bacteria is not well-defined, the Gram-negative pathogen *Salmonella enterica* serovar Typhimurium has some features of a graded transcriptional response [86]. Additionally, these same general transcriptional changes occur in other Gram-positive organisms during Zn limitation. These responses include expression of the Zur-regulated Zn uptake systems in *Listeria monocytogenes*, *Bacillus anthracis*, *Staphylococcus aureus*, and *Streptococcus pyogenes* [87–90] and Zn mobilization via induction of non-Zn requiring ribosomal proteins in *Streptomyces coelicolor* [91].

Bacteria also respond to Zn toxicity through metal-sensing transcriptional regulators. While some metalloregulators can function as both repressors and activators [92], others are functionally divided. In *B. subtilis*, excess Zn is sensed by the ArsR family of metalloregulators, CzrA [93], which effectively functions in reverse of Zur (Figure 3). When CzrA is not metallated, the protein represses its regulon. Upon metalation, CzrA undergoes a conformational change that lowers its DNA binding affinity and leads to derepression of metal efflux genes that encode a P-type ATPase named CadA and a cation diffusion facilitator type transporter named CzcD [93]. Analogous proteins are involved in detoxification of other divalent cations [94], demonstrating that metal efflux is a broadly conserved bacterial strategy to overcome metal intoxication.

Zinc Sensing in Gram-Negative Bacteria

Consistent with findings in Gram-positive bacteria, Gram-negative organisms primarily sense Zn starvation through the metalloregulator Zur (Figure 4). Structural insights into *Escherichia coli* Zur-DNA interactions have provided important details about protein conformational changes that occur during Zn binding by Zur. High sequence similarity among Zur homologs from different species suggests these mechanisms are conserved [80]. As is the case with Gram-positive Zur homologs, Zur-regulated derepression in Gram-negative organisms includes induction of genes encoding the high affinity ZnuABC Zn transporters, and these transporters are important for virulence of important human pathogens [53, 95] (Table 1). Additionally, the Zur regulon has been defined for several Gram-negative bacteria, including *Yersinia pestis* [96], *Neisseria meningitidis* [97], and *Acinetobacter baumannii* [98]; in addition to the highly conserved ZnuABC transporters, non Zn-binding ribosomal proteins are also typically increased in expression in response to Zn limitation [74]. Further, Zn starvation induces expression of an outer membrane TonB-dependent Zn transporter named ZnuD in *N. meningitidis* [99] (Figure 4). Structural studies into *N. meningitidis* ZnuD demonstrate that large extracellular loops directly interact with Zn ions and suggest active uptake of free Zn from the extracellular space. Interestingly, ZnuD possesses structural homology to bacterial heme transporters but does not bind heme [100]; these findings demonstrate that ZnuD is capable of binding free Zn but does not exclude the possibility that ZnuD may bind Zn in some chelated form. *A. baumannii* also encodes ZnuD homologues that are directly regulated by Zur [98]. In addition to the ZnuD with high homology to the *N. meningitidis* ZnuD, certain *A. baumannii* strains encode an additional candidate *znuD*, denoted *znuD2* [98]. However, the relative contribution of each of these ZnuD transporters to *A. baumannii* Zn homeostasis is unknown.

Similar to *B. subtilis* CzrA, *E. coli* possesses a separate metalloregulator named ZntR to respond to Zn excess [101] (Figure 4). ZntR is a member of the MerR family of transcriptional regulators. ZntR recognizes conserved inverted repeat sequences in the promoter region of the gene encoding a P-type ATPase named ZntA [101]. Apo-ZntR binds this inverted repeat and causes DNA distortions that prevent gene transcription [102]. Following Zn binding, Zn-ZntR induces DNA untwisting and unkinking that promotes efficient RNA polymerase recruitment and *zntA* expression [102]. Induction of P-type ATPases and cation diffusion family (CDF) transporters during Zn intoxication have been shown in other Gram-negative organisms as well. In *A. baumannii*, Zn intoxication leads to significant induction of a wide array of P-type ATPases and CDF transporters that also deplete cellular copper levels [103]. While the transcriptional regulator responsible for these changes is undefined, an *A. baumannii* strain lacking Zur has increased expression of predicted cation efflux systems and other transporters, which suggesting Zur plays a regulatory role in Zn efflux [98].

Bacterial Zinc Homeostasis Beyond Transporters

More recently, our understanding of bacterial Zn uptake has been expanded to include additional systems other than the ZnuABCD transporters. One strategy that has become increasingly appreciated for Zn acquisition is the production and secretion of Zn-binding

small molecules (Figures 3 & 4). This strategy is well-defined for Fe, where Fe-binding molecules termed siderophores are produced that facilitate Fe acquisition in diverse environments [104]. However, some siderophores are capable of binding other nutrient metals, including Zn. The Gram-negative bacterium *Pseudomonas putida* produces the siderophore pyridine-2,6-bis(thiocarboxylic acid) (PDTC) that is capable of binding both ferric Fe and Zn [105, 106]. Other siderophores including pyochelin, micacocidin, and yersiniabactin have been shown to bind Fe, Zn, and potentially other metals [107–111]. Mechanistic studies into the role of yersiniabactin in Zn uptake revealed that the Gram-negative pathogen *Yersinia pestis* uses a dedicated Zn-yersiniabactin importer named YbtX to acquire Zn from the molecule; further, genetic inactivation of the *znu* system and yersiniabactin biosynthetic genes reduces *Y. pestis* virulence in a septicemic plague model [112]. Zn-binding **metallophores** have also been implicated in Gram-positive Zn acquisition. For example, *Streptomyces coelicolor* produces a small molecule termed coelibactin which may bind Zn [113], and *S. aureus* produces the metallophore staphylopin with broad metal-chelating abilities that affect Zn homeostasis [88, 114].

Type VI secretion systems (T6SSs) are multiprotein machines used by many Gram-negative bacterial species to translocate effectors into neighboring cells and have been implicated in Zn acquisition (Figure 4). *Burkholderia thailandensis* produces a Zn-scavenging molecule named TseZ that is secreted through a specific T6SS, termed T6SS4. Zn-bound TseZ is then imported into the bacterial cell using the heme transporter HmuR specifically during conditions of oxidative stress, where Zn may be used to populate Cu/Zn superoxide dismutase enzymes and ameliorate potential damage from reactive oxygen species [115]. A similar model also occurs in *Y. pseudotuberculosis*, where the oxidative stress regulator OxyR induces expression of T6SS4 [116]. Additionally, ZntR was recently identified as a transcriptional activator of the *Y. pseudotuberculosis* T6SS4 [117], which is consistent with the observation that Zn deficiency promotes increased oxidative damage [118]. This T6SS4 can translocate a Zn-binding molecule named YezP that aids in Zn uptake. While a dedicated importer for YezP is not known, there likely exists an energy-dependent transporter that facilitates YezP uptake, as is the case for other metal-binding small molecules [104, 116].

Cell wall modifications are necessary to construct complex secretion systems and other macromolecular structures. Given the induction of T6SSs and Zn uptake machinery during Zn limitation, there may be changes to the bacterial cell envelope that occur specifically during nutrient starvation. Indeed, members of the genus *Acinetobacter* are morphologically constricted to shortened, rounded cells during nutrient limitation and significantly alter the abundance of major peptidoglycan muropeptides [119, 120]. Further, the M15 family Zn-binding peptidase ZrlA contributes to these muropeptide changes and promotes efficient Zn uptake and cell envelope barrier function [120]. In *Vibrio cholerae* the M23 family Zn-binding endopeptidase ShyB is implicated in cell wall maintenance during Zn limitation [121]. Importantly, both ZrlA and ShyB are directly regulated by Zur and collectively demonstrate that bacterial pathogens encode peptidoglycan-modifying enzymes that are important for surviving Zn limitation (Figure 4).

In addition to the production of metal-chelating molecules by bacterial pathogens, some bacteria can utilize host-derived molecules as a metal source. For example, *S. aureus* can use

human hemoglobin as its sole Fe source through deployment of the iron-regulated surface determinant (Isd) system [122]. As the second most abundant trace metal in humans, Zn scavenging within the vertebrate host may be an effective strategy to subvert nutritional immunity. Consistent with this prediction, the *N. meningitidis* TonB-dependent outer membrane receptor protein CpbA is expressed during Zn starvation [99] (Figure 4). CpbA is capable of binding human CP, and the presence of CpbA permits *N. meningitidis* to use CP as its sole Zn source. Additionally, the CpbA homolog in *Neisseria gonorrhoeae* named TdfH also binds CP and permits Zn acquisition from the protein [123]. These “Zn piracy” mechanisms [124] represent an exciting new area of future investigation towards understanding bacterial Zn acquisition during vertebrate colonization.

Bacterial Zinc Buffering and Allocation

Metal availability varies widely across environments and niches. Therefore, bacterial survival is largely dependent on systems to maintain cellular metabolism despite fluctuations in available nutrients. Considering there is essentially no free Zn within a bacterial cell despite the relatively high total Zn level, Zn must exist in chelated forms that is accessible during Zn starvation [79]. In *B. subtilis*, Zn limitation induces expression of non-Zn binding ribosomal proteins and the Zn-independent folate biosynthesis enzyme FolE2 [83, 125] (Figure 5). In a system analogous to eukaryotic metallothionein, *B. subtilis* uses the low-molecular weight molecule bacillithiol to maintain an intracellular labile Zn pool [126]. Similarly, *E. coli* uses glutathione to buffer Zn and other divalent cations [127], *A. baumannii* utilizes the amino acid L-histidine as a component of its labile pool [128]; during conditions of Zn starvation, *A. baumannii* upregulates the histidine utilization (Hut) system, thereby catabolizing cellular Zn-histidine complexes and increasing levels of bioavailable Zn [128] (Figure 5). However, the complete inventory of molecules capable of aiding in Zn buffering is not well-defined and warrants further investigation.

The requirement of Zn for many cellular processes suggests that a mechanism exists to ensure appropriate metalation of cognate metalloproteins, particularly during times of Zn starvation. For metalloregulators, differences in standard free energies for metal complex formation compared to relative metal-binding affinities dictates regulator-metal specificity [86]. However, the process of appropriate metallation is likely more complex for diverse metalloenzymes. To aid in proper metal allocation, members of the G3E GTPase superfamily have been identified as metallochaperones and/or metal insertases [129]. Four subfamilies exist within the G3E superfamily. Two of the subfamilies, represented by the metallochaperones UreG and HypB, are involved in Ni incorporation into the Ni metalloenzymes urease and hydrogenase, respectively [130, 131]. A third family is represented by MeaB, which is involved in methylmalonyl-CoA mutase activation [132]. The fourth subfamily, denoted the COG0523 subfamily, is less defined but is conserved in all domains of life [129]. Genomic analyses suggest that a subset of COG0523 members are Zur-regulated and may therefore serve as Zn metallochaperones [129]. Representative members include *E. coli* YjiA and YeiR, *B. subtilis* ZagA, and *A. baumannii* ZigA. Each of these proteins bind Zn and possess GTPase activity [128, 133–135] (Figure 5). Since Zn is required for many essential cellular processes, COG0523 members may aid in the prioritization of Zn to core metabolic processes when the metal is limited [58]. Consistent

with this prediction, analyses of the response of *B. subtilis* and *A. baumannii* to Zn starvation revealed that folate and riboflavin biosynthesis are hindered, respectively [58, 85]. In *B. subtilis*, ZagA interacts with the Zn-dependent FolE enzyme and aids in folate biosynthesis during Zn starvation, and this interaction is dependent on the Z nucleotide ZTP [85]. In *A. baumannii*, severe Zn restriction hinders *de novo* flavin biosynthesis [58]. This flavin deficiency is exacerbated in a strain lacking *zigA* [58], which suggests that ZigA impacts flavin biosynthesis. These studies position Zn metallochaperones at important metabolic hubs, and uncovering other processes altered by COG0523 members represents an exciting area of future research.

Concluding Remarks

Zn is required for life, which necessitates that both bacterial pathogens and vertebrate hosts have evolved strategies to acquire and maintain appropriate Zn levels. Members of the S100 protein family such as CP are capable of withholding Zn from invading pathogens; however, the extent to which Zn starvation occurs in diverse sites within the host is unexplored, but it is likely niche- and pathogen-specific. For example, *S. aureus* microcolonies experience heterogeneous Fe starvation even within a single tissue [136], which suggests a complex interplay in metallostasis between host and pathogen that has yet to be defined.

Within a bacterial cell, Zn starvation upregulates Zn uptake machinery, but it also has major consequences for Zn-dependent metabolic processes. Zn starvation has been shown to change ribosome composition, alter bacterial cell wall dynamics, and impact labile Zn pools within the cell. Additionally, representative COG0523 members ZigA and ZagA have been implicated in Zn allocation to metalloenzymes but further exploration is required to determine their precise mechanisms of action as well as the identity of their client proteins. Interestingly, COG0523 members are also present in humans [129], which suggests that understanding their functionality within bacteria may inform metal homeostasis more broadly. Interrogating systems used by both vertebrates and microbes to balance nutrient metals has the potential to improve human health while simultaneously broadening our understanding of metal biology (see Outstanding Questions).

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Glossary

Damage-associated molecular patterns (DAMPs)

molecules produced by damaged, stressed, or dying cells that can escalate an inflammatory response and can occur in the absence of a microbial infection

Glutathione

tripeptide containing glycine, glutamate, and cysteine that possesses metal-binding properties

Irving-Williams series

the relative stability by which transition metals form stable complexes. The series is as follows (from least stable to most stable): $Mn^{2+} < Fe^{2+} < Co^{2+} < Ni^{2+} < Cu^{2+} > Zn^{2+}$

Metalloenzyme

Enzymes that contain metal ions that are either covalently-bound or bound to prosthetic groups. The metal serves as a cofactor for enzymatic activity, as opposed to a metalloprotein, which may only contain metals for structural stability

Metallophore

Small molecules with the ability to bind diverse metals; derived from 'siderophore,' which refers to iron-binding small molecules

Metalloregulatory proteins

Transcriptional regulators that respond to changes in metal availability, usually by direct physical interactions with the metal the regulator is sensing

Metallothionein

small, cysteine-rich metal binding protein

Mismetallation

The process whereby the incorrect metal is bound to a metal-binding protein

Nutritional immunity

the process whereby a host limits the availability of nutrients to defend against infection

Regulon

The collective genes that are regulated by a specific transcriptional regulator

Trace metal

metals that are required for biological functions but exist in low abundance within a given system. The precise definition of a trace metal is field-specific, but within vertebrates trace metals are also known as micronutrients and include elements such as iron, copper, magnesium, and selenium

Transition metals

elements within the central block of the periodic table that have variable outer shell electrons. Zn is not a true transition metal, owing to its filled outer shell. Transition metals are also known as d-block elements

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Outstanding Questions

1. How do local changes in local metal availability within vertebrates impact infection progression?
2. What is the precise chemical speciation of zinc and other metal ions within biological systems?
3. Do zinc-specific siderophores (“zincophores”) exist broadly in bacteria and aid in zinc acquisition?
4. How do zinc-binding metalloenzymes ensure correct cofactor incorporation, and what is the relative contribution of COG0523 family members to this process?
5. How is host-mediated zinc intoxication and starvation balanced to limit virulence of diverse pathogens?

Highlights

Zinc is a redox-inactive nutrient metal required for catalytic activity and/or structural stability for thousands of proteins throughout life.

Vertebrate hosts and bacterial pathogens have evolved parallel mechanisms for balancing the essentiality of zinc with its inherent toxicity.

Zinc homeostasis relies on a complex network of metal transporters linked to zinc buffering systems.

Members of a GTPase subfamily are implicated as zinc-specific metallochaperones, which aid in metal delivery to cognate metalloenzymes.

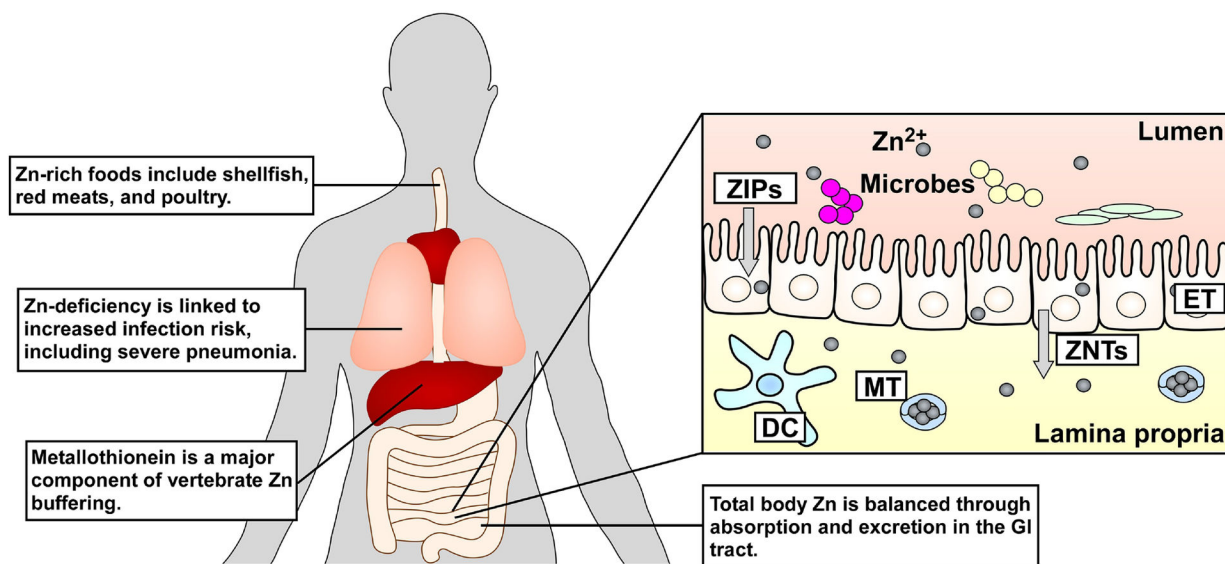


Figure 1.

Vertebrate Zn homeostasis systems. Changes in dietary Zn (grey) can be associated with adverse clinical outcomes and diarrheal disease. Zn buffering is regulated by changes in epithelial cell (ET) Zn uptake/efflux associated with the gastrointestinal (GI) tract primarily by the action of ZIP Zn importers and ZNT Zn exporters. ZIP activity results in Zn uptake from the intestinal lumen into ETs, and ZNT activity results in efflux of Zn from the ETs via the lamina propria into circulation and the extracellular space. Small molecule Zn chelators such as metallothioneins (MTs) that are produced in high abundance by the liver and kidneys also contribute to Zn buffering [4], Immune cells such as dendritic cells (DCs) may respond differentially depending on nutrient Zn availability.

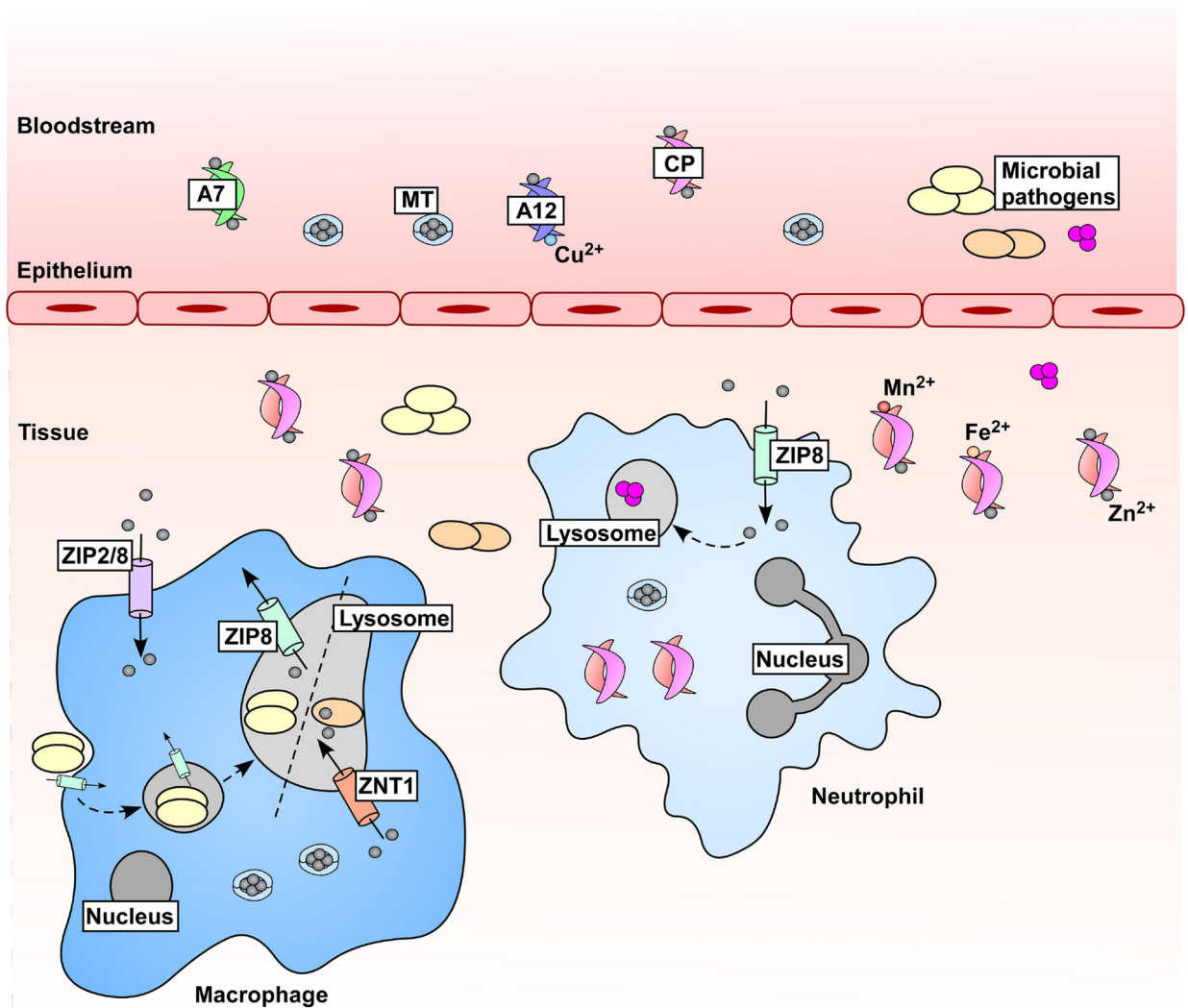


Figure 2.

Zn is mobilized in hosts during microbial infections and inflammation. In response to microbial challenge, serum Zn drops through increased expression of zinc transporters (not depicted) and mobilization and uptake of Zn-binding metallothioneins (MTs). Immune cells contribute to changes in Zn availability, where expression of ZIP Zn importers drives accumulation of Zn within these cells. Zn is also mobilized within immune cells; *Mycobacterium tuberculosis* (orange ovals) experiences Zn intoxication within macrophages in a ZNT1-dependent manner, and *Streptococcus pyogenes* (pink spheres) is poisoned by Zn within the neutrophil lysosome [69, 70]. Conversely, the fungal pathogen *Histoplasma capsulatum* (yellow ovals) has been shown to be Zn-starved within the macrophage lysosome [71]. These two mechanisms (Zn intoxication and Zn starvation) within the macrophage lysosome are denoted by a dashed line. Additionally, metal-chelating proteins are produced as part of the inflammatory response to further reduce Zn availability for pathogens, including those within the S100 protein family. S100A7 (A7) binds Zn and is produced in high abundance by keratinocytes, and S100A12 (A12) binds both Zn and Cu. The heterodimer of S100A8/S100A9, known as calprotectin (CP) [159], binds Zn with high

affinity, as well as other divalent cations. CP is a major component of the neutrophil cytoplasmic protein content and is abundant at sites of infection and inflammation within vertebrates.

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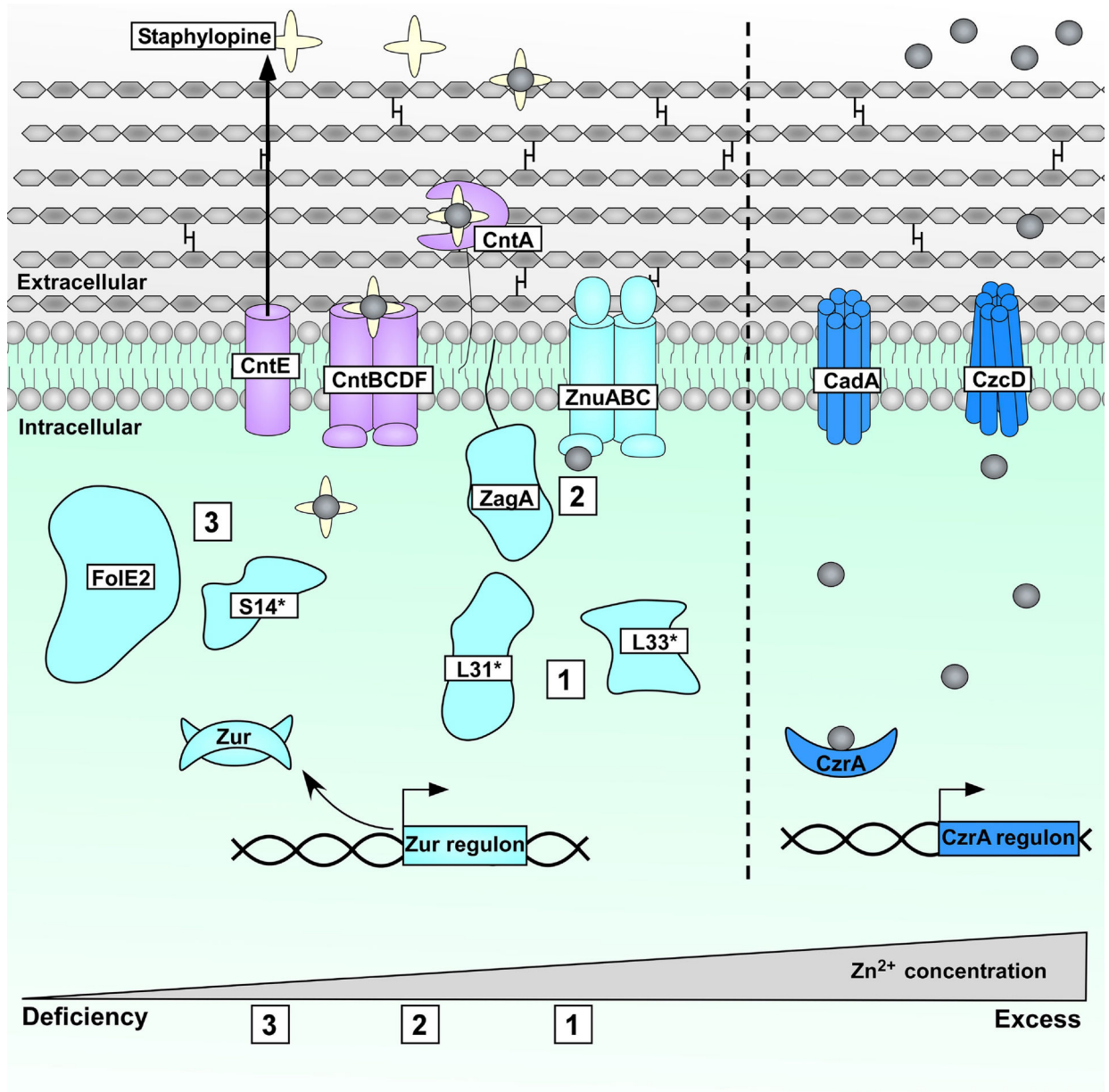


Figure 3.

Zn sensing in Gram-positive bacteria. In response to Zn starvation, the *Bacillus subtilis* Zur regulon experiences derepression in three distinct waves. **1)** The non-Zn binding ribosomal proteins L31* and L33* are synthesized to displace the Zn-binding L31/L33 ribosomal proteins, followed by **2)** upregulation of the ZnuABC high affinity Zn transport system and the putative metallochaperone ZagA [85]. Lastly, **3)** the Zn-independent GTP cyclohydrolase I enzyme F0IE2 and the non-zinc requiring S14* ribosomal protein are expressed. Some Gram-positive organisms also produce metal-binding small molecules to capture Zn from the extracellular space. For example, *Staphylococcus aureus* produces the metal-binding small molecule staphylopin that aids in Zn acquisition [88, 114]. Staphylopin is secreted via CntE, captured by CntA, and imported by the CntBCDF

system. Gram-positive bacteria also experience transcriptional changes in response to Zn excess; in *B. subtilis*, CzcA-regulated genes are expressed during Zn intoxication and includes the P-type ATPase CadA and the cation diffusion family (CDF) transporter CzcD [93]. Expression of these proteins results in Zn efflux from the bacterial cell.

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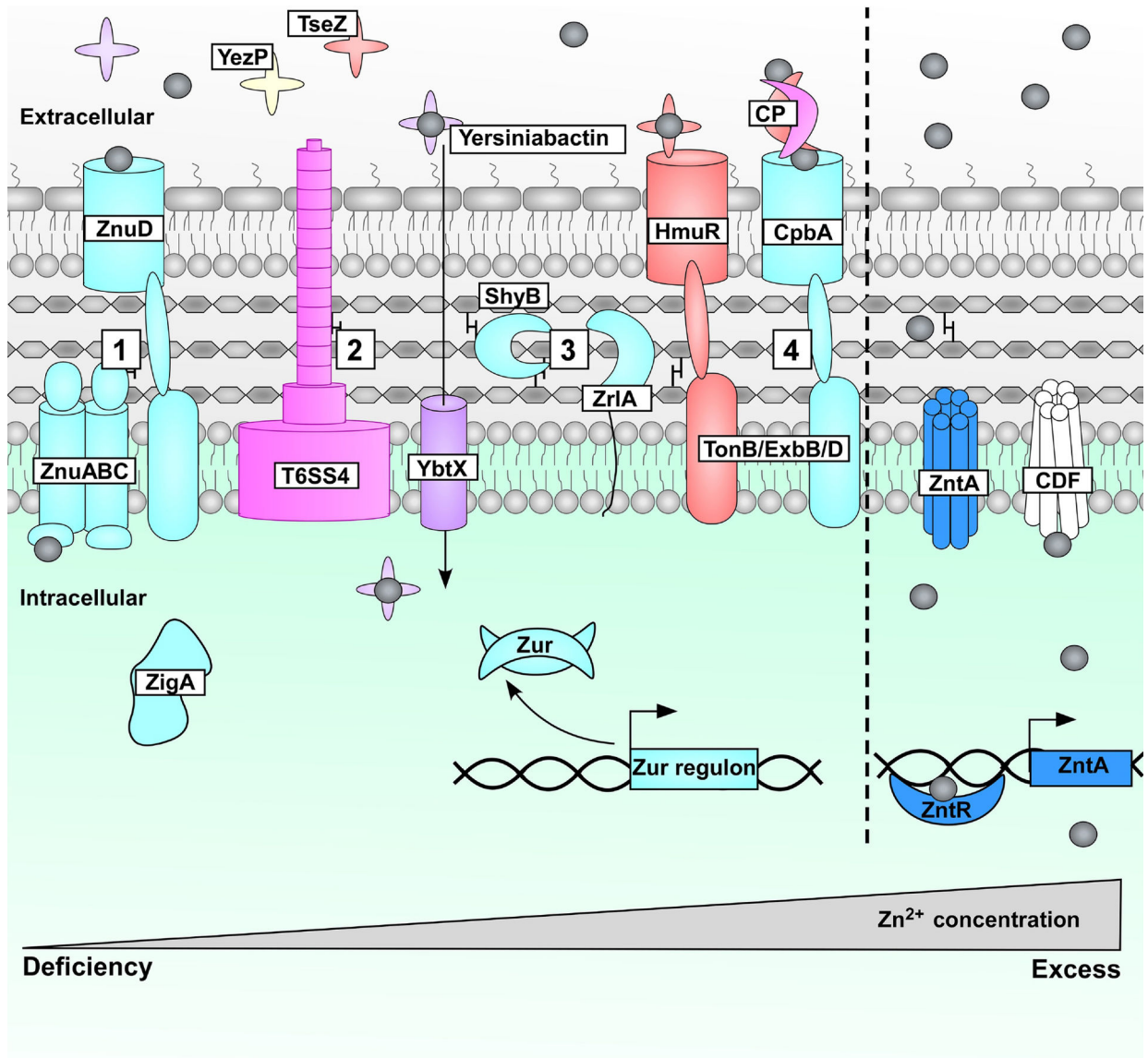


Figure 4.

Zn sensing in Gram-negative bacteria. Zn starvation in Gram-negative bacteria leads to a number of physiological changes. **1)** Zur-derepression leads to expression of genes encoding the Zn uptake ZnuABCD system, as well as the predicted Zn metallochaperone ZigA in *Acinetobacter baumannii* [128]. **2)** Zn-binding molecules are produced such as YezP and TseZ, which are secreted by T6SS4, and yersiniabactin, which is captured by YbtX [112, 115, 116]. **3)** Enzymes implicated in cell wall homeostasis are induced, including *Vibrio cholerae* ShyB [121] and *A. baumannii* ZrIA [120]. **4)** TonB-dependent transporters aid in the uptake of Zn-bound molecules; *Burkholderia thailandensis* HmuR captures the TSS64-secreted TseZ [115], and *Neisseria meningitidis* and *Neisseria gonorrhoeae* use CpbA/TdfH to bind calprotectin (CP) for Zn acquisition as a form of Zn piracy [123, 124]. During Zn intoxication, *Escherichia coli* Zn-ZntR causes DNA conformational changes leading to expression of the P-type ATPase ZntA to alleviate the metal toxicity [101]. Many Gram-

negative organisms also encode cation diffusion family (CDF) transporters that contribute to overcoming Zn intoxication.

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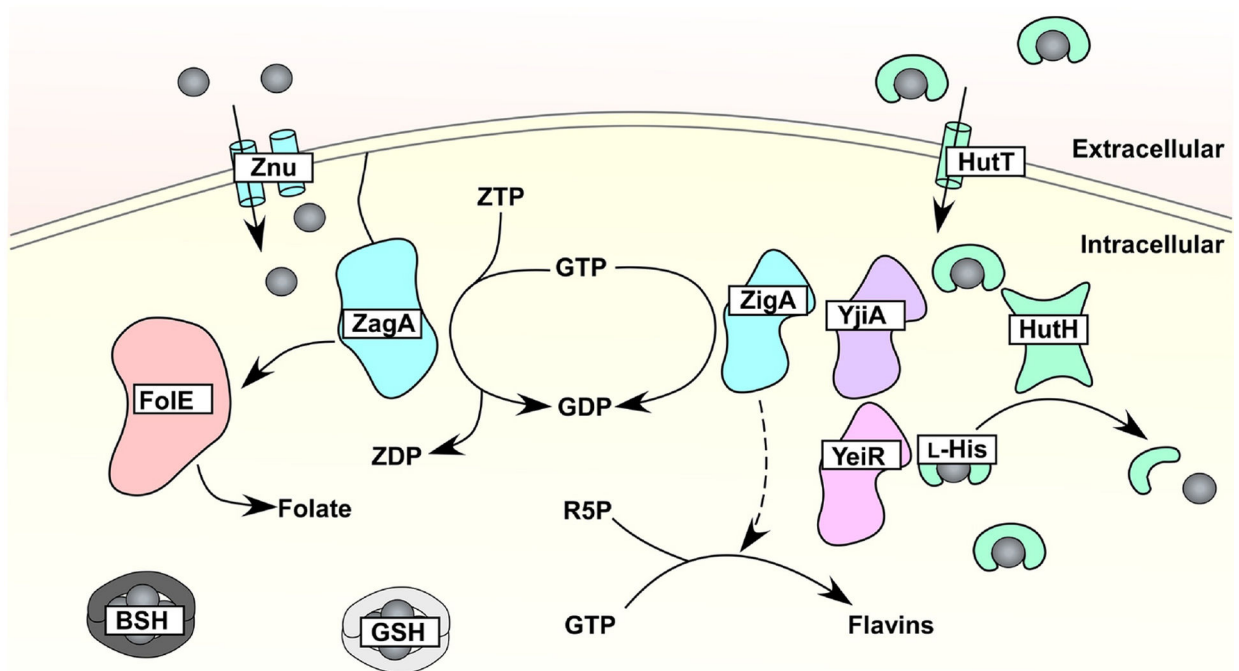


Figure 5.

Bacterial systems for Zn buffering. In order to buffer cellular changes to Zn availability, bacteria employ small molecules to bind excess Zn that can be accessed during zinc limitation. Bacillithiol (BSH) in *Bacillus subtilis*, glutathione (GSH) in *Escherichia coli*, and L-histidine in *Acinetobacter baumannii* serve as components of the labile Zn pool [126–128]. In *A. baumannii*, the histidine transporter HutT captures Zn-L-His, and its subsequent HutH-mediated degradation is hypothesized to liberate Zn [128]. The delivery of Zn to metalloenzymes remains an area of active investigation, but the mobilization of enzymes involved in essential cofactor biosynthesis implicates these pathways targets for Zn metallochaperone activity. Consistent with this prediction, *B. subtilis* folate biosynthesis and *A. baumannii* riboflavin biosynthesis are stressed during Zn limitation. In *B. subtilis*, the metallochaperone ZagA responds to Zn limitation and the purine alarmone ZTP to interact with, and possibly metallate, the Zn-dependent folate biosynthetic enzyme FolE [85]. In *A. baumannii*, the predicted metallochaperone ZigA contributes to maintenance of cellular flavin levels during Zn limitation through an undefined mechanism [58]. Possible metallochaperone interactions are denoted with arrows in the figure.

Table 1.

Bacterial pathogen zinc uptake systems.

Transporter	Pathogen	Role in Pathogenesis	Reference
ZnuABC	<i>Acinetobacter baumannii</i>	Lung colonization	[53]
	<i>Brucella abortus</i>	Macrophage survival and systemic infection	[137]
	<i>Campylobacter jejuni</i>	Cecal colonization	[138]
	<i>Escherichia coli</i>	Epithelial cell interactions and urinary tract infections	[76, 139, 140]
	<i>Francisella tularensis</i>	Macrophage survival	[141]
	<i>Moraxella catarrhalis</i>	Intracellular invasion and lung colonization	[142]
	<i>Neisseria gonorrhoeae</i>	Not defined	[143]
	<i>Pasturella multocida</i>	Systemic infection	[144]
	<i>Proteus mirabilis</i>	Urinary tract infections	[145]
	<i>Pseudomonas aeruginosa</i>	Not defined	[146]
	<i>Salmonella enterica</i> serovar Typhimurium	Systemic infection and cecal inflammation	[95, 147, 148]
	<i>Treponema pallidum</i>	Not defined	[149]
	<i>Vibrio cholerae</i>	Gut colonization	[150]
	<i>Vibrio parahaemolyticus</i>	Systemic infection	[151]
	<i>Yersinia pestis</i>	Systemic infection when yersiniabactin production is inactivated	[112]
ZnuD	<i>Neisseria meningitis</i>	Complement resistance	[124, 152]
AdcABC	<i>Streptococcus agalactiae</i>	Survival in human biological fluids	[153]
	<i>Streptococcus pneumoniae</i>	Survival in biological fluids	[154]
	<i>Streptococcus pyogenes</i>	Systemic infection	[90]
ZupT	<i>Escherichia coli</i>	Urinary tract infection	[140]
	<i>Francisella tularensis</i>	Macrophage survival	[141]
	<i>Salmonella enterica</i> serovar Typhimurium	Systemic infection	[155]
ZinT	<i>Escherichia coli</i>	Epithelial cell adherence	[139]
	<i>Salmonella enterica</i> serovar Typhimurium	Systemic infection when <i>znuA</i> is inactivated	[156]
ZinABC/ZurA	<i>Listeria monocytogenes</i>	Lethality following oral infection	[157]
ZevAB	<i>Haemophilus influenzae</i>	Lung colonization	[158]
TroABCD	<i>Treponema pallidum</i>	Not defined	[149]