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Photoaffinity probes for nematode pheromone receptor identification†

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Abstract

Identification of pheromone receptors plays a central role for uncovering signaling pathways that underlie chemical communication in animals. Here, we describe the synthesis and bioactivity of photoaffinity probes for the ascaroside *ascr#8*, a sex-pheromone of the model nematode, *Caenorhabditis elegans*. Structure–activity studies guided incorporation of alkyne- and diazirine-moieties and revealed that addition of functionality in the sidechain of *ascr#8* was well tolerated, whereas modifications to the ascarylose moiety resulted in loss of biological activity. Our study will guide future probe design and provides a basis for pheromone receptor identification *via* photoaffinity labeling in *C. elegans*.

Introduction

The nematode *Caenorhabditis elegans* has become an important model for inter-organismal signaling *via* pheromones. *C. elegans* and other nematode species communicate with conspecifics *via* ascarosides, a family of small-molecule pheromones based on the 3,6-dideoxysugar, L-ascarylose, linked to fatty acid-like side chains (Fig. 1). Ascarosides can be further decorated with building blocks from diverse primary metabolic pathways, *e.g.* the likely folate-derived *p*-aminobenzoic acid in *ascr#8*¹ (**1**), or the neurotransmitter octopamine in *osas#9*² (**5**). Ascarosides are involved in almost every aspect of the life history of *C. elegans*, including developmental changes,^{3–5} aging,^{6,7} and diverse behaviors *e.g.* dispersal,⁸ aggregation,^{9,10} and mating.^{11,12} Ascarosides are typically sensed as cocktails of compounds,¹³ often functioning in synergy.^{1,8,12} Notably, even small changes in ascaroside structures can result in dramatic differences in biological responses.⁷

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Conflicts of interest

There are no conflicts to declare.

Three ascarosides (ascr#2 (6), ascr#3 (8), and ascr#4 (7)) have been shown to synergistically attract males,¹² whereas the structurally distinct ascr#8 (1) acts as a potent male attractant even in the absence of other ascarosides.^{1,11} ascr#8 (1) and the related ascr#81 (2) are unique among identified ascaroside pheromones in that they incorporate a *p*-aminobenzoate moiety.^{14,15}

Identification of ascaroside receptors has not kept pace with the discovery of new ascarosides and activities, and it appears that multiple G protein-coupled receptors (GPCRs) are involved in sensing individual compounds.^{16,17} Receptor identification has been hampered in part by the large number of GPCRs in *C. elegans*, which likely exceeds 1000.¹⁸ In order to identify ascaroside receptors, previous studies have employed reverse genetics screens,^{16,19} as well as quantitative trace locus analyses.^{20–22} The use of ascaroside-based photoaffinity probes could enable targeted receptor identification and, importantly, has the advantage to demonstrate direct ascaroside-receptor binding. In a previous study, an ascr#2-based photoaffinity probe was used to confirm binding to the GPCR DAF-37, which was originally identified *via* immunoprecipitation;¹⁷ however, the ascr#2 probe used in this study featured extensive structural modifications and correspondingly was much less biologically active than unmodified ascr#2 and thus unlikely to facilitate *de novo* receptor identification.

Toward identification of receptor(s) of the sex pheromone ascr#8, we aimed to design a photoaffinity probe that would enable covalent linking and receptor pull-down, yet retain most bioactivity. We synthesized four ascr#8 probe designs, two of which retained potent activity. The inactive probes provide insight into structure-activity relationships and may help guide design for probes of other ascaroside pheromones.

Results

Previous studies showed that ascr#8 elicits a robust attractive response in young adult male *C. elegans*. We confirmed attraction to ascr#8 (1) using a spot retention assay, in which the dwell time of worms in areas treated either with vehicle control or the chemical of interest is measured (Fig. 2A).^{1,11,12}

For photoaffinity labeling, the installation of two types of groups into ascr#8 is required, a photoreactive group and a bioorthogonal reactive group for affinity purification. Since large photoreactive groups may interfere with binding between the ligand and its receptor,²³ we focused on diazirine derivatives, which have additional advantages including short lifetime following UV irradiation and subsequent high reactivity.^{24,25} As a click-chemistry handle for affinity purification, we chose an alkyne, since azides may become subject to metabolic reduction.²⁶ To assess whether additional functional groups affect the biological activity of ascr#8, chemical modifications were introduced one at a time, followed by testing the bioactivity of the resulting ascr#8 derivatives.

In the initial probe designs, we planned to integrate the click-chemistry moiety as part of the fatty acid side-chain (9). Replacing the ω -methyl group in the side chain with a terminal alkyne seemed most straightforward and least likely to perturb receptor binding. For introduction of the diazirine moiety, we envisioned replacing one of the two hydroxyl groups

of the ascarylose, as we felt such modification may cause only minimal changes to the overall size and shape of ascr#8 (**1**) (Fig. 2). For the synthesis of the alkyne-integrating side chain (Scheme 1), metathesis of 4-pentenoic acid (**13**) with ethyl acrylate produced intermediate **14**, which was converted into the TMS-protected alkyne **15**. Following reduction of the ketone, TBDMS protection of the resulting alcohol and hydrolysis of the ester, the acid **16** was coupled to ethyl *p*-aminobenzoate, followed by TBDMS deprotection. The alkyne-containing side chain **17** was then coupled to protected ascarylose using established procedures, furnishing probe A (**9**). Given that the alkyne moiety in Probe A is sterically somewhat encumbered, we confirmed the ability of such alkynes to undergo click chemistry using the analogous alkyne derivative of ascr#18 (**4**) (see ESI[†]). Probe A was found to elicit levels of attraction comparable to unmodified ascr#8 (Fig. 2B and S1[†]), indicating the integration of an alkyne moiety in place of the ω -methyl in the side chain does not significantly perturb activity of ascr#8 and may represent a useful entry for the design of other ascaroside receptor probes.

Next, we developed syntheses for introduction of the diazirine at the 2' and 4' carbons of ascarylose. Starting from key intermediate (**20**),²⁷ a diazirine moiety was installed in position 4 in 10 steps, furnishing probe B (**10**, Scheme 2A). Our approach to probe B necessitated late-stage introduction of benzoyl protection in position 2, since the 2-*O*-benzoyl-protected derivative of ketone **21** had a strong tendency to eliminate and would not survive under the conditions required to install the diazirine. The 2-*O*-benzoyl moiety is required for stereochemical control of the subsequent glycosylation step *via* neighboring group participation (NGP). Conversion of the benzyl ether (Bn) into the benzoyl ester (Bz) after installment of the diazirine group in **22** was achieved through ruthenium(III) chloride-catalyzed oxidation (Scheme 2A). Following a different synthetic strategy, we inserted the diazirine at the 2' carbon of the ascarylose, furnishing probe C (**11**, Scheme 2B). In the probe C synthesis, we opted to first establish the glycosidic linkage, followed by benzoyl deprotection, oxidation, and introduction of the diazirine moiety. Desilylation of **28** furnished probe C.

Testing probes B and C in the spot retention assay, we found that both probes had no activity (Fig. 2B and S1[†]). In fact, males spent less time in the probe C-treated spot compared to control, although the difference did not reach significance. Loss of biological activity with probe B (**10**) and probe C (**11**) suggests that both 2' and 4' hydroxyl groups of ascarylose participate in essential interactions during receptor binding. These data support previous studies that demonstrate the importance of these hydroxyl groups in determining the identity of individual ascarosides.^{8,12}

These results led us to reevaluate our design strategy, based on a biological evaluation of naturally occurring ascr#8 derivatives. Recent expansion of the known family of ascarosides has uncovered two derivatives of ascr#8, named ascr#81 and ascr#82.¹⁴ In these ascr#8 derivatives, L-glutamic acid or L-glutamyl-L-glutamic acid is attached to the *p*-aminobenzoate *via* amide linkages. We asked whether these alterations of the carboxy

[†]Electronic supplementary information (ESI) available: Experimental procedures and spectral data for all prepared compounds with copies of ¹H NMR and ¹³C NMR. See DOI: [10.1039/c9ob02099c](https://doi.org/10.1039/c9ob02099c)

terminus affect biological activity and tested a synthetic sample of ascr#81 (**2**) in the spot retention assay. We found that ascr#81 is at least as active as ascr#8 (**1**) in this assay (Fig. 2B and S1†). In fact, dwell times elicited by ascr#81 were higher than those measured for ascr#8, though the difference did not reach statistical significance.

This result suggested that modification of ascr#8 at the carboxy terminus may be well tolerated. Therefore, we revised our probe design to incorporate both the diazirine and alkyne handle into a single moiety linked *via* an amide to the *p*-amino-benzoic acid in ascr#8 (probe D (**12**), Scheme 3). Commercially available 2-(3-(but-3-yn-1-yl)-3*H*-diazirin-3-yl)ethan-1-amine was incorporated at the terminal carboxylic acid, analogous to the previously described synthesis of ascr#81 (**2**).¹⁴ Gratuitously, bioassays demonstrated that probe D retained biological activity matching that of unmodified ascr#8 (Fig. 2B and S1†).

Discussion

Together, these results suggest multiple avenues for testing and development of ascaroside receptor probes. The presence of the hydroxyl groups on the ascarylose sugar appears to be essential for maintaining ascaroside activity (Fig. 2, **10**, **11**).^{5,14} Introduction of an alkyne moiety at the ω -position of the side chain is well tolerated in the case of ascr#8. Except for ascr#5 (**3**), all ascarosides for which biological activity has been demonstrated in *C. elegans* feature an ω -methyl group in the side chain, suggesting that alkyne introduction at this position could be used for probing receptor interactions of biosynthesis of diverse ascarosides in *C. elegans*. However, in other nematode species, modification of the ω -methyl may be less well tolerated; for example, in *Caenorhabditis nigoni*, hydroxylation of the ω -side chain carbon was found to abolish the attraction of *C. nigoni* males to ascr#3 (**8**).²⁸

In the case of ascr#8, use of commercially available 2-(3-(but-3-yn-1-yl)-3*H*-diazirin-3-yl)ethan-1-amine, as in probe D (**12**), allowed for inclusion of both the photo-reactive group and bioorthogonal reactive group with relative ease. However, activity of many other ascarosides requires the presence of an unmodified carboxy terminus, and therefore installation of the alkyne at the ω -position of the side chain may be preferable, though a suitable location for the diazirine moiety remains to be found. For the identification of the receptor of ascr#8, our efforts are directed at crosslinking probe D with putative receptor candidates in heterologous expression systems.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgements

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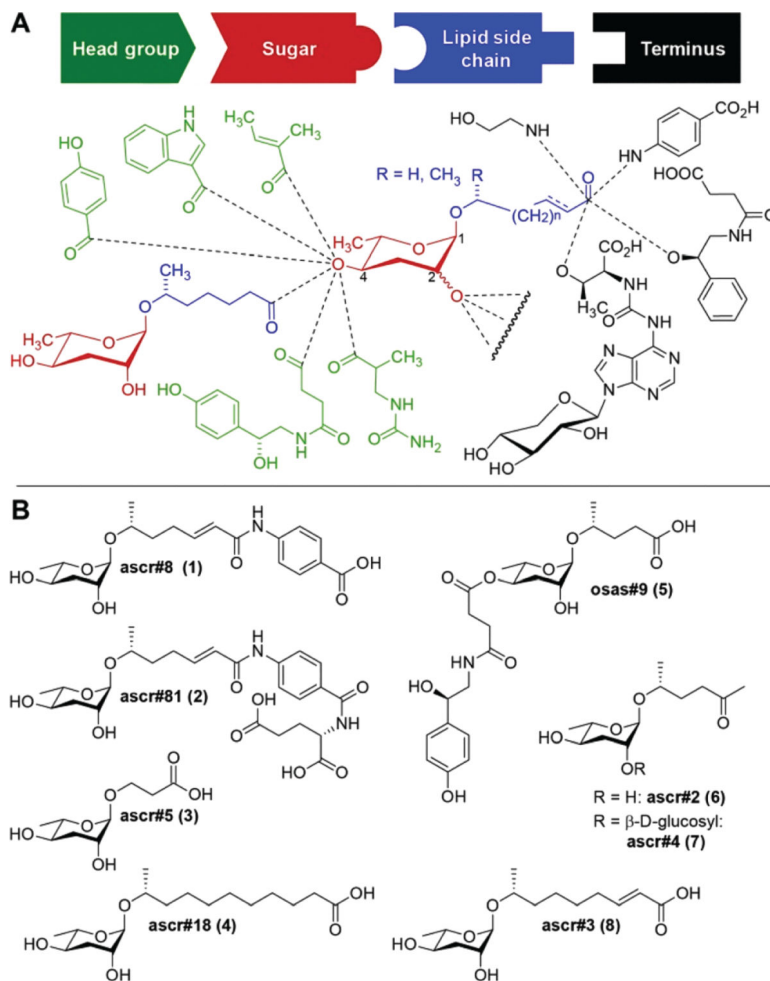


Fig. 1. Structures of ascaroside pheromones from *C. elegans* and other nematodes. (A) Overview of structural diversity of ascaroside pheromones. Building blocks from diverse metabolic pathways are combined to furnish a modular library of signaling molecules, with specific examples shown in (B).

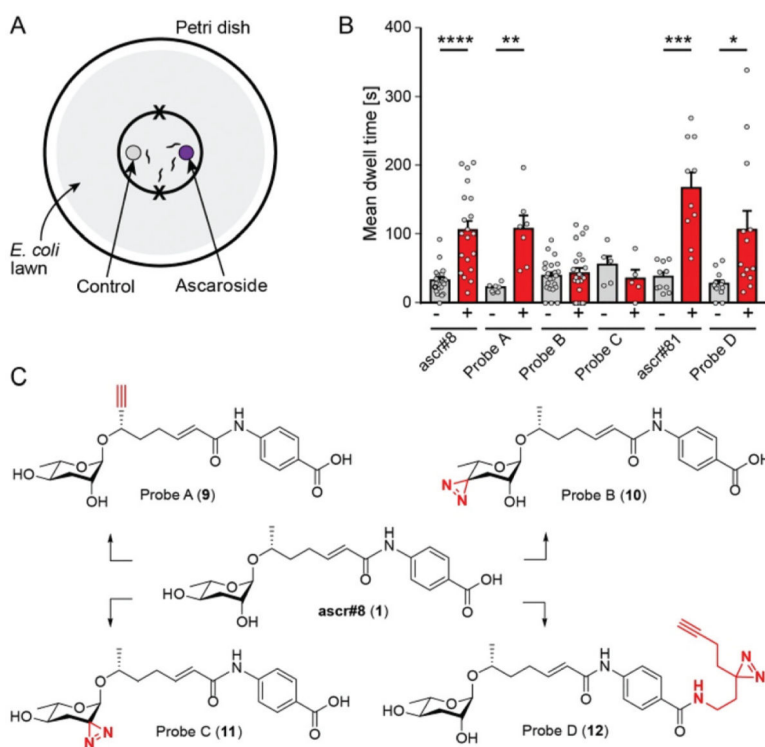
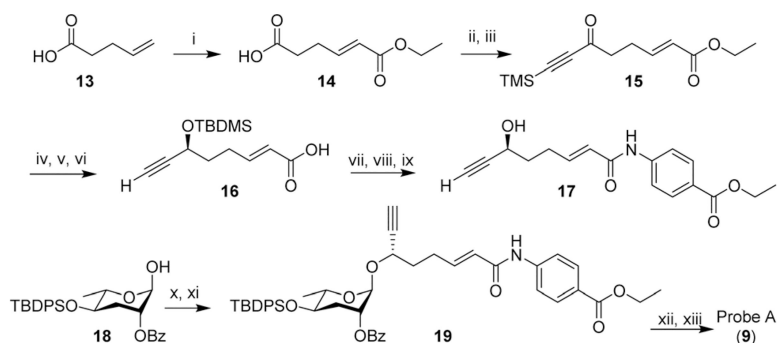
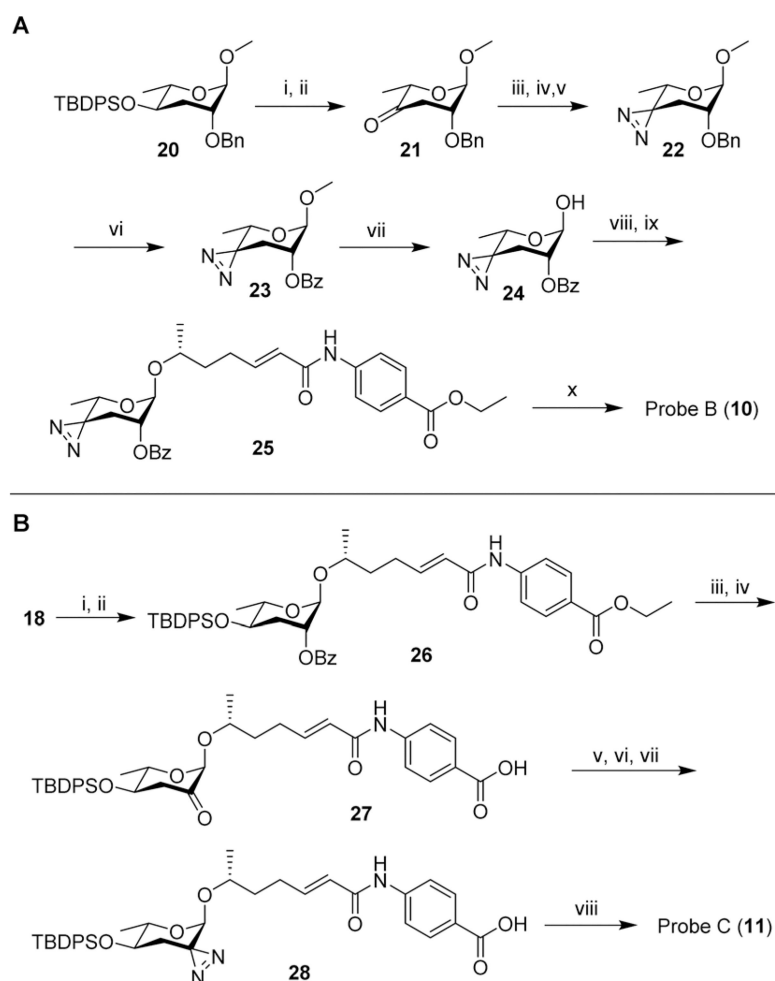


Fig. 2. Response of *C. elegans* males to ascr#8 and derivatives. (A) Schematic of the spot retention assay. Males are placed on each “X” and scored for time spent in each of the smallest circles. (B) Dwell times of *C. elegans* males in vehicle controls (–) and ascaroside-treated (+) circles. Data presented as mean \pm SEM. * $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$, **** $p < 0.0001$. (C) Structures of ascr#8 (1) and four probe designs (9–12).

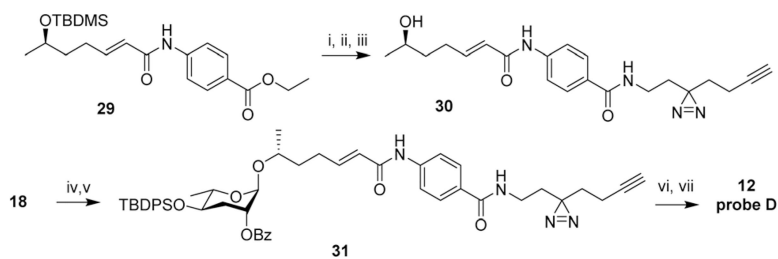
**Scheme 1.**

Synthesis of probe A (9). (i) ethyl acrylate, Grubb's second generation catalyst, DCM, 25 °C, 10 h, 75%; (ii) oxalyl chloride, cat. DMF, DCM, 0 °C, 20 min; (iii) bis(trimethylsilyl)acetylene, AlCl₃, DCM, 0 °C, 3 h, 26% ~2 steps; (iv) freshly prepared Terashima reagent (see Experimental section), -78 °C, 98%, ee 80%; (v) TBDMSCl, imidazole, DMF, 25 °C, 2.5 h, 94%; (vi) LiOH·H₂O, dioxane, H₂O, 60 °C, 12 h, 50%; (vii) oxalyl chloride, cat DMF, DCM, 0 °C, 20 min; (viii) benzocaine, DIEA, DCM, 0 °C, 1 h, 87% ~2 steps; (ix) HF in H₂O, MeCN, 25 °C, 1 h, 85%; (x) CCl₃CN, DBU, DCM, 25 °C, 2 h; (xi) **17**, TMSOTf, DCM, 0 °C, 1 h, 50% ~2 steps; (xii) TBAF, THF, 25 °C, 12 h, 93%; (xiii) LiOH·H₂O, dioxane, H₂O, 60 °C, 12 h, 62%.



Scheme 2.

(A) Synthesis of probe B (**10**). (i) TBAF, THF, 25 °C, 8 h, 95%; (ii) PCC, 4 Å molecular sieves, DCM, 25 °C, 4 h, 74%; (iii) 7N NH₃ in MeOH, pTsOH, MeOH, 0 °C, 3 h; (iv) NH₂OSO₃H, 0 °C → rt, 16 h; (v) NEt₃, I₂ in MeOH titration, 25 °C, 39% ~3 steps; (vi) RuCl₃·H₂O, NaIO₄, DCM : MeCN : H₂O = 1: 1: 1, 25 °C, 5 h, 72%; (vii) BBr₃, DCM, -78 °C, 30 min, 71% BRSM; (viii) CCl₃CN, DBU, DCM, 25 °C, 2 h; (ix) N-(6'*R*-hydroxy-2'*E*-heptenoyl)-4-aminobenzoic acid ethyl ester (prepared following previous reported method¹), TMSOTf, DCM, 0 °C, 2 h, 57%~2 steps; (x) LiOH·H₂O, dioxane, H₂O, 60 °C, 3 h, 79%. (B) Synthesis of probe C (**11**): (i) CCl₃CN, DBU, DCM, 25 °C, 2 h; (ii) N-(6'*R*-hydroxy-2'*E*-heptenoyl)-4-aminobenzoic acid ethyl ester (prepared following previous reported method¹), TMSOTf, DCM, 0 °C, 2 h, 15% ~2 steps; (iii) LiOH·H₂O, dioxane, H₂O, 60 °C, 12 h, 42%; (iv) Dess-Martin periodinane, DCM, 25 °C, 12 h, 51%; (v) 7 N NH₃ in MeOH, pTsOH, MeOH, 0 °C, 3 h; (vi) NH₂OSO₃H, 0 °C → rt, 16 h; (vii) NEt₃, I₂ in MeOH titration, 25 °C; (viii) TBAF, THF, 25 °C, 8 h, 13% ~4 steps.

**Scheme 3.**

Synthesis of probe D (**12**). Reagents and conditions: (i) LiOH·H₂O, dioxane, H₂O, 60 °C, 12 h, 72%; (ii) 2-(3-(but-3-yn-1-yl)-3*H*-diazirin-3-yl)ethan-1-amine, EDC HCl, DMAP, DCM/DMF, 25 °C, 8 h, 53%; (iii) HF in H₂O, MeCN, 25 °C, 1 h, 92%; (iv) CCl₃CN, DBU, DCM, 25 °C, 2 h; (v) **30**, TMSOTf, DCM, 0 °C, 1 h, 8% ~2 steps; (vi) TBAF, THF, 25 °C, 12 h, 77%; (vii) LiOH·H₂O, dioxane, H₂O, 60 °C, 12 h, 60%.