

Supercritical CO₂ Fluid Extraction for the Identification of Compounds from *Citrus reticulata* Semen by Ultra-High-Performance Liquid Chromatography Combined with Q-Exactive Orbitrap Tandem Mass Spectrometry

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Cite This: *ACS Omega* 2020, 5, 2180–2186



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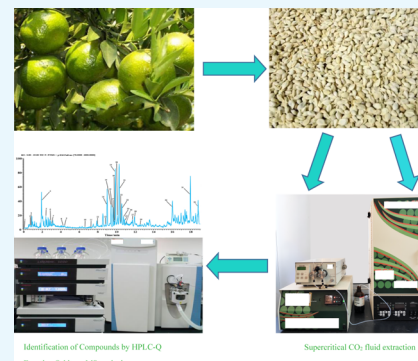


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ABSTRACT: A rapid and simple method based on the coupling of supercritical fluid extraction and ultra-high-performance liquid chromatography combined with Q-Exactive Orbitrap tandem mass spectrometry (SFE-UHPLC-Q-Exactive Orbitrap-MS) detection for the identification of compounds from *Citrus reticulata* semen (CRS) was developed for the first time in this study. Through the optimization of the SFE parameters including extractive pressure, extractive temperature, and time, most of the compounds were successfully extracted at 50 °C, 33 MPa, and 2 h without an entraining agent, among which 32 compounds were successfully identified. Moreover, the operating conditions of UHPLC-Q-Exactive Orbitrap-MS were also optimized for the analysis of the SFE extracts, and the extracts in the CRS showed good separation performance in 20 min. A total of 28 compounds from the SFE extract were identified by comparing the standard sample together with full scan and related literature data, among which esters and flavonoids were the major compounds identified in the CRS extracts. In addition, 2 phenols, 2 aldehydes, 2 triterpenes, and 5 other compounds were identified. The SFE-UHPLC-Q-Exactive Orbitrap-MS method was successfully validated and applied for the identification of compounds from the CRS.



1. INTRODUCTION

The *Citrus reticulata* semen (CRS) is the dry, mature seed of *C. reticulata* Blanco and its cultivated varieties and has been consumed as a traditional Chinese medicine for centuries with its properties including qi regulation, knot dispersion, pain relief, and so on.^{1,2} However, compared with *Citri Reticulatae* Pericarpium, the chemical composition of CRS has not been fully studied; most *Citrus* seeds are discarded rather than being exploited, resulting in resource waste and environmental pollution. Therefore, the study of its chemical components is of great significance for its further utilization in medicine and raw materials. Therefore, the purpose of this study is to establish a fast and efficient method for the comprehensive analysis and identification of phytochemical components in supercritical fluid extraction (SFE) extracts of CRS.

As a kind of low-temperature extraction technology, supercritical fluid CO₂ extraction (SFE) has the advantages of low toxicity, noncombustibility, high diffusion, gas-like viscosity, liquid-like density, sample processing automation, and high extraction efficiency.³ In addition, liquid chromatography (LC)–MS is often used for the analysis of phytochemical components, and in recent years, ultra-high-performance LC combined with Q-Exactive Orbitrap tandem

mass spectrometry (UHPLC-Q-Exactive Orbitrap-MS) has become a faster, more efficient, and more sensitive tool than HPLC–MS for the rapid analysis of plant extracts and bioactive compounds in some traditional Chinese medicines. As far as we know, only a few analytical methods are used for the determination of CRS at present. Previously reported analytical methods were only used to analyze the oil contained in it and one or several specific chemical components such as limonin and obacunone by GC–MS analysis or HPLC method, lacking systematic analysis. Thus, in order to evaluate the quality of CRS more comprehensively, an efficient method based on the coupling of SFE-UHPLC-Q-Exactive Orbitrap-MS detection was developed in this paper, and the chemical components in the SFE extracts of CRS were systematically analyzed. This study is of great significance and provides a

Received: September 23, 2019

Accepted: January 20, 2020

Published: January 30, 2020



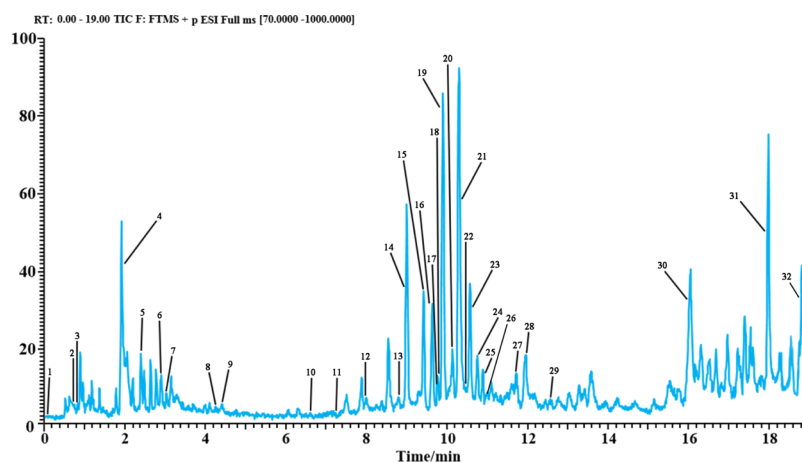


Figure 1. Total ion chromatogram in positive mode of CRS.

valuable basis for further scientific research and application of CRS in food, medicine, health care products, and other fields.

2. RESULTS AND DISCUSSION

The SFE extracts of CRS were analyzed by UHPLC-Q-Exactive Orbitrap-MS analysis. By optimizing a series of parameters such as flow rate and elution gradient of mobile phase, better analytical conditions are obtained. Under the optimal conditions, the components of SFE extracts in CRS could be separated in the positive mode within 19 min. The method is fast and efficient. The TIC diagram is shown in Figure 1. A total of 32 compounds were identified and separated and numbered 1–32 according to the sequence of peak time (Table 1). Totally, 29 SFE extracts constituents were divided into the following groups: 10 esters, 7 flavones, 2 phenols, 2 aldehydes, 2 triterpenes, and 6 other compounds. The identification of all these compounds was mainly based on MS/MS fragments and the analysis of literature. For a compound with a standard, the retention times and fragments of the compound should be compared with the standard under the same test conditions. Among them, 8 compounds were identified through comparison with standard products, and 20 compounds were identified based on MS/MS-positive ion fragments and relevant data from previous studies. Their chemical structures were illustrated in Figure 2.

2.1. Identification of Esters. Esters are major constituents extracted from the SFE extracts of CRS. Ten esters were identified in SFE extracts including *N*-butylidenephthalide, ligustilides, atractylenolide-I, atractylenolide-II, scoparone, ethyl ferulate, linolenic acid ethyl ester, and linderalactone. In positive mode, characteristic fragment ions such as CO, H₂O, and alkyl side chain were prone to lose in these kinds of compounds in high-energy collisions.⁴

In this paper, for example, identification and characterization of structures were discussed about ligustilides, atractylenolide-I, and atractylenolide-II. Peak 27 produced a relatively strong molecular ion peak at m/z 191.1064 for $[M + H]^+$ and relatively low intensity peaks at m/z 173.0959 for $[M + H - H_2O]^+$, 163.1115 for $[M + H - CO]^+$, 145.1010 for $[M + H - H_2O - CO]^+$, and at m/z 117.0700 for $[M + H - H_2O - CO - C_2H_4]^+$. Based on the characteristic fragmentation profiles, peak 27 were identified as ligustilides by comparing with the characteristic ions from previous study.⁵

Peak 22 was assigned to be atractylenolide-I by comparison with literature,⁶ which gave a molecular ion peak $[M + H]^+$ at

m/z 231.1378 and fragment ion peak at m/z 213.1274 for $[M + H - H_2O]^+$ and m/z 189.0910, 185.1324, 175.0754, 163.0752, 143.0855, 105.0702, and 91.0547. The detailed fragmentation process is shown in Figure 3a.

Peak 29 provided precursor ion peak of $[M + H]^+$ at m/z 233.1533 and fragment ion peaks at m/z 215.1428 for $[M + H - H_2O]^+$, m/z 187.1479 for $[M + H - H_2O - CO]^+$, m/z 159.0803 for $[M + H - H_2O - CO - C_2H_4]^+$, m/z 145.1010 for $[M + H - H_2O - CO - C_3H_6]^+$, m/z 131.0854 for $[M + H - H_2O - CO - C_4H_8]^+$, and m/z 105.0701 for $[M + H - H_2O - CO - C_6H_{10}]^+$. Furthermore, combined with literature⁷ and reference formula provided by MS, it was therefore confirmed as atractylenolide-II. The detailed fragmentation pattern of peak 29 is shown in Figure 3b.

2.2. Identification of Flavonoids. Flavonoids are important components of natural flavonoids in *Citrus*. According to the classification of flavonoids, they were divided into normal flavones, flavanones, and polymethoxylated flavones.^{8,9}

With the high-resolution UHPLC-Q-Exactive Orbitrap-MS system, a total of seven flavonoids were identified in the SFE extracts of CRS including peaks 11, 16, 17, 18, 20, 23, and 26. Among them, peaks 26, 20, 16, and 23 were unambiguously identified as 5-hydroxy-6,7,8,3',4'-pentamethoxyflavone, 3,5,6,7,8,3',4'-heptamethoxyflavone, nobiletin, and tangeretin by comparing with the literature and the reference standard with the same fragmentation pattern. Nobiletin was selected as a typical example of flavonoids to illustrate the fragmentation pathways. Nobiletin is a polymethoxylated flavone with six methoxy substituents in its structure, which produces strong molecular ion peaks at m/z 403.1385 and formed relatively lower intensities of MS² fragments at m/z 388.1148 for $[M + H - CH_3]^+$, m/z 373.0914 for $[M + H - 2CH_3]^+$, m/z 358.0676 for $[M + H - 3CH_3]^+$, and m/z 327.0854 for $[M + H - 3CH_3 - CH_3O]^+$, among others, with a series of characteristic losses of CH₃ and CH₃O. Figure 3c depicts the proposed fragmentation pattern.

2.3. Identification of Limonins. Limonoids are secondary metabolites of tetracyclic triterpenes in Rutaceae and Meliaceae widely distributed in the seeds and membranes of *Citrus* fruits, which are a major group of chemical constituents from CRS. Two limonoids were identified in SFE extracts, including limonoid and obacunone, after comparison with fragments of the standard reference and literature.^{10,11}

Table 1. Summary of Chemical Constituents About Supercritical CO₂ Fluid Extraction from CRS by Ultra-HPLC Combined with Q-Exactive Orbitrap Tandem Mass Spectrometry

no.	t_R (min)	$[M + H]^+$ (m/z)	fragment ions (m/z)	compound formula	identification
1	0.03	195.0875	177.0912, 167.0847, 153.0544, 146.9612, 138.0662, 135.0442, 123.0432, 114.6795, 110.0716, 107.0856, 89.0602, 79.0550, 69.0455, 58.0659	C ₈ H ₁₀ N ₄ O ₂	caffeine
2	0.70	124.0394	119.0366, 106.0655, 101.0255, 96.0448, 90.0198, 80.0501, 78.0343, 68.0503, 62.9905, 53.0393	C ₆ H ₅ NO ₂	nicotinic acid
3	0.82	123.0553	118.0349, 108.9585, 105.0374, 100.0247, 96.0448, 90.5265, 88.0236, 86.0262, 80.0500, 76.5157, 67.5105, 62.9906, 56.9656, 53.0392	C ₆ H ₆ N ₂ O	nicotinamide
4	1.92	156.1019	138.0913, 122.8464, 113.0840, 110.0967, 108.0810, 96.0812, 88.0762, 86.0606, 84.0451, 82.0657, 71.0498, 69.0342, 67.0548, 53.0394	unknown	unknown
5	2.40	137.1073	122.0837, 96.0811, 88.2350, 80.0502, 69.0706, 60.5088, 55.0550	C ₈ H ₁₂ N ₂	tetramethylpyrazine
6	2.91	153.0546	144.6615, 134.0600, 125.0597, 122.4801, 113.0483, 111.0442, 107.0494, 97.0652, 93.0339, 88.9528, 79.0548, 70.9424, 65.0393, 63.0239	C ₈ H ₈ O ₃	vanillin
7	3.01	183.0652	173.9848, 165.0906, 159.9690, 155.0702, 145.9897, 140.0467, 131.9742, 123.0442, 113.9639, 105.0450, 98.9845, 95.0495, 90.9481, 86.9537, 81.0341, 72.9378, 67.0550, 65.0393, 56.9430	C ₉ H ₁₀ O ₄	3,5-dimethoxy-4-hydroxybenzaldehyde
8	4.21	207.0652	201.9320, 192.0416, 179.0704, 175.0391, 163.0390, 151.0754, 148.0517, 121.0650, 107.0495, 91.0547, 79.0548, 67.0548	C ₁₁ H ₁₀ O ₄	scoparone
9	4.41	147.0440	137.9799, 129.0700, 123.9644, 119.0857, 105.9538, 103.0545, 95.0496, 91.0547, 84.0813, 73.0845, 64.9280, 53.0393	C ₉ H ₆ O ₂	coumarin
10	6.61	133.0648	119.9576, 117.9598, 115.0544, 109.5154, 105.0701, 103.0546, 98.0076, 96.0099, 91.0547, 89.0023, 83.8491, 79.0548, 77.0390, 72.0814, 65.0394, 58.0336	C ₉ H ₈ O	cinnamaldehyde
11	7.24	303.0859	285.0751, 262.1411, 243.0657, 219.0645, 201.0540, 185.0447, 177.0546, 163.0389, 153.0182, 135.0441, 117.0337, 111.0080, 89.0390, 67.0185	C ₁₆ H ₁₄ O ₆	hesperitin
12	8.13	223.0964	214.0120, 205.0857, 199.9964, 186.0551, 177.0546, 163.0389, 149.0597, 145.0284, 135.0441, 121.0650, 117.0337, 107.0859, 93.0703, 89.0391, 79.0548, 67.0550	C ₁₂ H ₁₄ O ₄	ethyl ferulate
13	8.99	471.2012	453.1918, 425.1955, 409.2001, 383.2280, 367.1904, 339.1947, 321.1846, 279.1379, 225.0913, 213.0910, 187.0753, 161.0597, 133.0648, 105.0702, 95.0132, 79.0548	C ₂₆ H ₃₀ O ₈	limonin ^a
14	9.07	151.1117	146.0297, 139.0752, 128.0193, 123.0806, 121.0650, 113.9638, 109.0650, 105.0700, 95.0859, 87.0045, 81.0704, 79.0547, 67.0550, 65.0393, 55.0549	C ₁₀ H ₁₄ O	2-adamantanone
15	9.41	245.1170	227.1066, 217.1214, 203.1063, 199.1116, 189.0544, 171.1166, 156.0934, 145.0647, 131.0492, 119.0857, 105.0702, 95.0495, 81.0341, 76.9062, 67.0550, 55.0186	C ₁₅ H ₁₆ O ₃	linderlactone
16	9.63	403.1385	388.1148, 373.0914, 358.0676, 327.0854, 301.0701, 284.0666, 258.0519, 229.0339, 211.0235, 193.0125, 183.0287, 165.0546, 122.70389, 99.0443, 69.0341	C ₂₁ H ₂₂ O ₈	nobiletin ^a
17	9.76	287.0910	269.0804, 245.0814, 203.0699, 179.0332, 171.0286, 161.0596, 153.0181, 133.0647, 118.0414, 103.0547, 67.0185	C ₁₆ H ₁₄ O ₅	isosakuranetin
18	9.78	343.1173	327.0856, 313.0701, 299.0909, 282.0881, 267.0650, 239.0699, 199.0233, 171.0285, 153.0181, 133.0647, 122.0363, 86.0967, 69.0339	C ₁₉ H ₁₈ O ₆	6-demethoxytangeretin
19	9.89	515.2271	469.2221, 455.2037, 419.1846, 411.2158, 393.2051, 349.1804, 321.1849, 279.1374, 243.0992, 215.1061, 187.0752, 161.0596, 133.0647, 95.0131, 81.0340	unknown	unknown
20	10.16	433.1488	418.1254, 403.1019, 385.0911, 345.0599, 317.0650, 289.0700, 255.0640, 211.0230, 183.0290, 165.0544, 149.0235, 127.0387, 94.0416, 70.9067	C ₂₂ H ₂₄ O ₉	3,5,6,7,8,3',4'-heptamethoxyflavone ^a
21	10.31	137.0596	122.0734, 119.0493, 116.5949, 109.0650, 107.0493, 105.0451, 95.0495, 93.0703, 91.0547, 81.0705, 79.0549, 67.0549, 54.4142, 53.0394	C ₈ H ₈ O ₂	p-anisaldehyde
22	10.53	231.1378	213.1274, 189.0910, 185.1324, 175.0754, 163.0752, 143.0855, 135.0442, 119.0857, 105.0702, 91.0547, 69.0706, 55.0550	C ₁₅ H ₁₈ O ₂	atractylenolide-I
23	10.57	373.1280	358.1041, 343.0809, 328.0570, 297.0753, 283.0602, 271.0599, 254.0569, 229.0319, 211.0235, 193.0132, 183.0287, 168.0051, 135.0440, 99.0444, 68.9978	C ₂₀ H ₂₀ O ₇	tangeretin ^a
24	10.76	455.2060	437.1934, 419.1848, 409.2013, 391.1895, 359.1282, 349.1430, 315.1386, 303.1373, 263.068, 235.1124, 197.0964, 175.0754, 161.0596, 133.0648, 105.0702, 95.0132, 79.0548, 67.0550	C ₂₆ H ₃₀ O ₇	obacunone
25	10.82	183.0804	173.9850, 159.9689, 141.9586, 131.9742, 127.9794, 118.9675, 113.9638, 105.0338, 100.9570, 95.0495, 81.0705, 72.9378, 67.0549, 59.9310, 56.9430	C ₁₃ H ₁₀ O	atractyloidin
26	11.10	389.1227	374.0992, 359.0757, 341.0652, 331.0807, 316.0574, 285.0761, 260.0672, 227.0544, 215.0184, 197.0080, 187.0235, 169.0130, 163.0752, 148.0518, 113.0237, 85.0291, 55.0185	C ₂₀ H ₂₀ O ₈	5-hydroxy-6,7,8,3',4'-pentamethoxyflavone ^a

Table 1. continued

no.	t_R (min)	$[M + H]^+$ (m/z)	fragment ions (m/z)	compound formula	identification
27	11.72	191.1064	173.0959, 163.1115, 149.0596, 145.1010, 135.0438, 130.0778, 121.0651, 117.0700, 105.0701, 93.0702, 91.0546, 81.0706, 79.0548, 69.0704, 67.0549, 55.0550	$C_{12}H_{14}O_2$	ligustilides
28	11.91	189.0907	171.0802, 166.0290, 156.0927, 143.0854, 133.1015, 128.0620, 125.0960, 117.0698, 105.0701, 97.1014, 91.0547, 83.0860, 67.0550, 55.0550	$C_{12}H_{12}O_2$	<i>N</i> -butyridenepthalide
29	12.52	233.1533	215.1428, 205.1582, 197.1323, 187.1479, 177.0908, 173.0961, 159.0803, 151.0752, 145.1010, 131.0854, 119.0857, 105.0701, 95.0859, 81.0704, 67.0549	$C_{15}H_{20}O_2$	atractylenolide-II
30	16.07	295.2263	277.2158, 259.2050, 241.1943, 223.1689, 205.1582, 187.1482, 173.1321, 161.1323, 147.1165, 135.1167, 121.1012, 107.0858, 95.0859, 81.0704, 67.0549	Unknown	unknown
31	17.98	324.2893	307.2624, 263.2367, 245.2264, 161.1329, 133.1013, 109.1013, 95.0858, 67.0549, 62.0608	Unknown	unknown
32	18.79	307.2627	261.2208, 243.2102, 233.2252, 219.2103, 187.1481, 173.1322, 145.1009, 137.1322, 123.1168, 109.1013, 95.0858, 81.0704, 67.0549	$C_{20}H_{34}O_2$	linolenic acid ethyl ester

^aFurther confirmation in comparison with standard compound.

Taking limonin, for example, according to the literature, the fragmentation model of the standard substance and the fragment ions provided by MS can be known, and the positive ion of limonin was detected at m/z 471.2012. The generation of m/z 425.1955 fragment ions was due to neutral loss of the ester group in lactonic ring D. The fragment ion peak at m/z 409.2001 indicates that the epoxy group can transform into dihydroxyl after deleting the carboxyl group. The loss of the epoxy group and the lactone ring A produced fragments of ions at m/z 367.1904 and m/z 339.1947, respectively. In addition, the fragment ions at m/z 161.0597 were generated because of the cleavage of ring C. The detailed fragmentation schemes are shown in Figure 3d.

2.4. Identification of Others. In addition, except for four unknown compounds, nine other compounds including peak 25, peak 1, peak 3, peak 2, peak 5, peak 14, peak 6, peak 7, peak 21, and peak 10 were identified as atractyloidin, caffeine, nicotinamide, nicotinic acid, tetramethylpyrazine, 2-adamantanone, two phenols (vanillin, 3,5-dimethoxy-4-hydroxybenzaldehyde), and two aldehydes (*p*-anisaldehyde, cinnamaldehyde), respectively, combined with literature^{12–21} and reference formula provided by MS. Using peak 25 as example, peak 25 showed an ion signal at m/z 183.0804 $[M + H]^+$ in the MS spectrum. In the MS² spectrum, the major fragment ion peaks were at m/z 159.9689, m/z 131.9742, m/z 113.9638, and m/z 105.0338 $[M + H - C_6H_6]^+$. Peak 25 was identified as atractyloidin by comparing the reference formula and previous research.

3. CONCLUSIONS

In the present study, all samples from CRS were extracted by supercritical carbon dioxide under optimized conditions, and a total of 32 compounds released from all of these varieties of SFE extracts were separated and identified by UHPLC-Q-Exactive Orbitrap-MS including 10 esters, seven flavonoids, two phenols, two aldehydes, two triterpenes, and nine other compounds, among which nine esters were first identified in CRS. The analysis results showed that the chemical composition of SFE extracts and methanol extracts was significantly different. According to our previous studies, methanol extracts of CRS mainly consist of hesperidin, neohesperidin, and other compounds with greater polarity, as well as medium-polarity compounds such as nobiletin and tangeretin, hardly containing ester and other low-polarity compounds. The SFE extracts used in this study were mainly low-polarity compounds such as esters and medium-polarity compounds such as flavonoids.

According to the literature studies, the triterpenoids and ester compounds in CRS have good pharmacological activities. For example, the triterpenoid limonin has antitumor,^{22,23} analgesic, and anti-inflammatory effects,²⁴ while obacunone has anti-inflammatory,²⁵ antioxidation,²⁶ and antitumor effects.^{27,28} Esters, moreover, as ligustilides, were effective for limiting the oxidative stress²⁹ and anti-inflammatory effect.³⁰

Since ancient times, China has a history of using CRS as a traditional Chinese medicine. However, because of the lack of systematic research on the chemical components contained in CRS and the lack of research on the biological activity of its chemical components, the application of CRS is limited and the quality is varied. Therefore, the composition of compounds in the middle- and low-polar compounds in CRS was investigated through the SFE-UHPLC-Q-Exactive Orbitrap-MS method for the first time in this study, which laid a

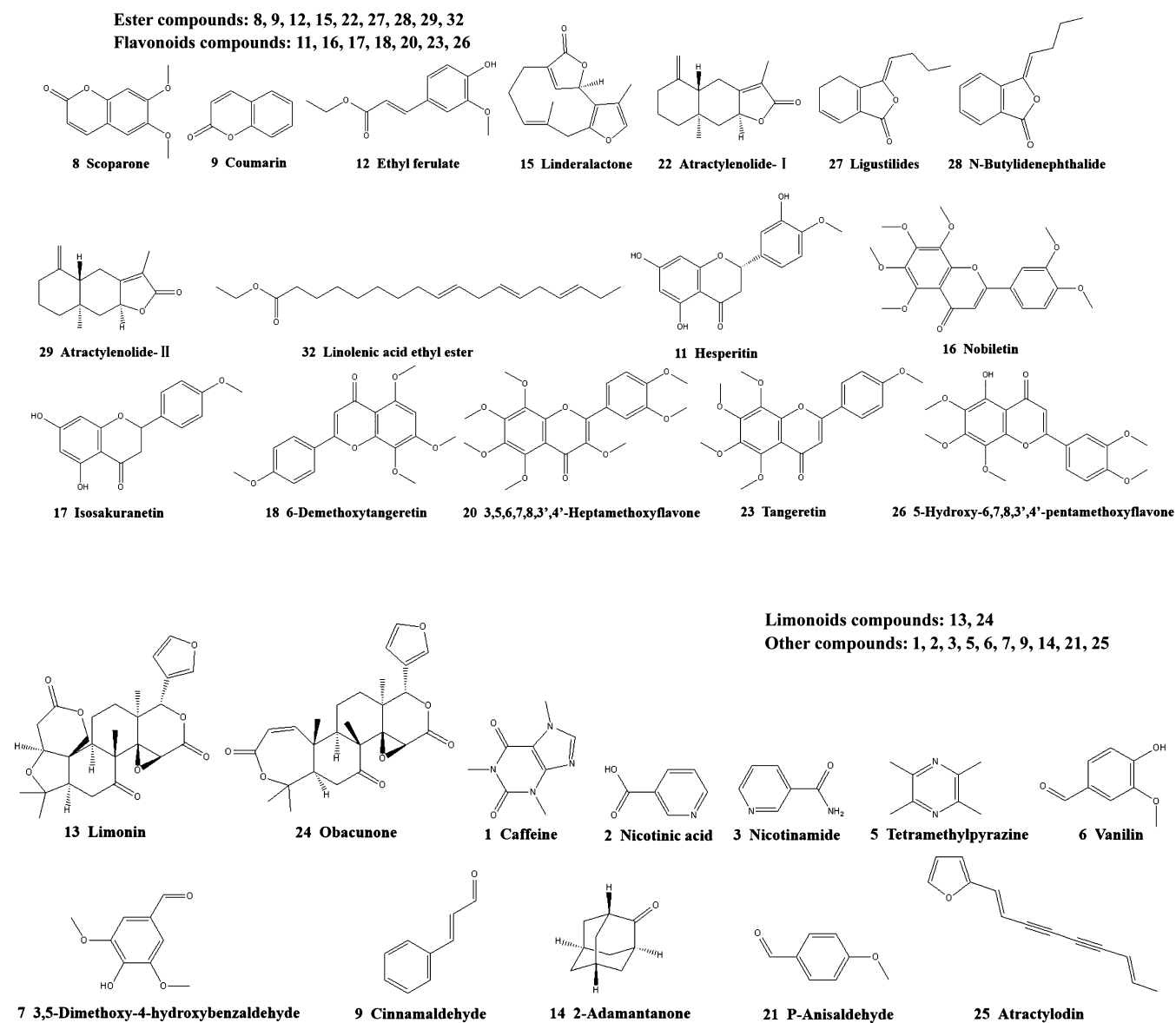


Figure 2. Chemical structures of compounds identified in CRS.

foundation for further scientific research and application of the CRS.

4. MATERIALS AND METHODS

4.1. Materials and Reagents. About 10 kg of fresh Xinhui *Citrus* was collected from Nantan Island, Xinhui District, Jiangmen City, Guangdong Province, in October 2017 and identified as the fruit of *C. reticulata* “Chachi” by Professor Zheng Guodong from the school of Pharmacy, Guangzhou Medical University. After removing the peel and pulp of the fresh *Citrus*, the seeds were collected, washed, dried, and stored in the Pharmacognosy Laboratory, School of Pharmacy, Guangzhou Medical University. Eight compounds were used as reference standards with purity greater than 98%. Among them, 3,5,6,7,8,3',4'-heptamethoxyflavone and 5-hydroxy-6,7,8,3',4'-pentamethoxyflavone were purchased from Sichuan Weikeqi Biotechnology Co., Ltd and Nanjing Chunqiu Biological Engineering Co. Ltd, respectively. All the others were purchased from Chengdu Mansite Biotechnology Co.

Ltd. In addition, acetonitrile and MS-grade formic acid were bought from Thermo Fisher Scientific (China) Co. and Merck KGaA Company of Germany, Ltd., respectively. Other reagents were all of analytical grade.

4.2. Sample Preparation. The 50 g powder samples of CRS were directly loaded into the SFE kettle having a specification of 50 g without the use of any kind of support material and entrainer for the SFE extraction. The SFE conditions were as follows: extraction temperature: 50 °C, extractive pressure: 33 MPa, extractive time: 2 h, and CO₂ flow rate: 0.05 L/min. The SFE sample was collected, placed into a vial, and stored at 4 °C. Prior to analysis, the reserved solution was further diluted with dichloromethane to an appropriate concentration and filtered through a needle-type microporous membrane with an aperture of 0.22 μm.

4.3. Analytical System. The ultra-high-performance LC was carried out on an Dionex UltiMate 3000 UHPLC system (Thermo Scientific, USA) equipped with an online degasser, a quaternary pump, an autosampler, and a column temperature

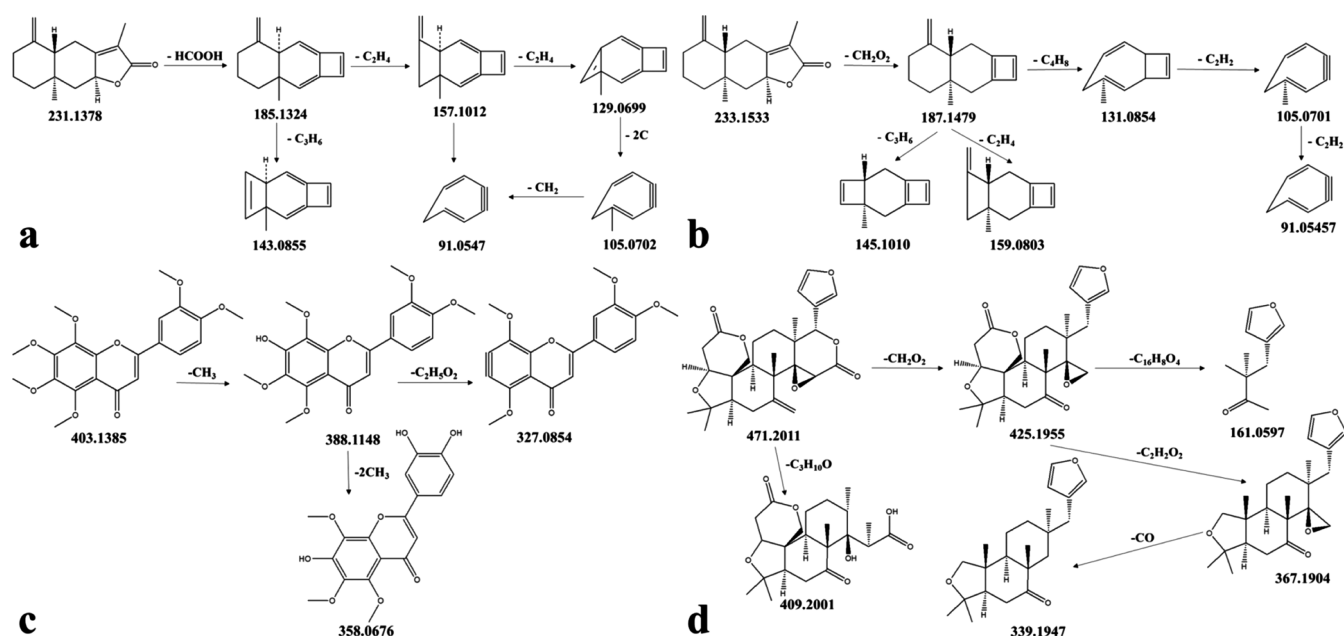


Figure 3. Proposed fragmentation pathways of main fragment ions for atractylenolide-I (a), atractylenolide-II (b), nobiletin (c), and limonin (d) in positive ion mode.

compartment, which was connected to a Q-Exactive Orbitrap tandem mass spectrometer (Thermo Scientific, USA) via an electrospray ionization interface and high-resolution MS detection was performed in the positive ion mode, and the data were acquired in full scan mode. A ZORBAX Eclipse Plus C₁₈ (2.1 mm × 50 mm, 1.8 μm) was utilized at 40 °C and eluted with a gradient solvent from A/B (90:10) to A/B (10:90) at a flow rate of 0.50 mL/min, where A is 0.1% (v/v) formic acid solution and B is 0.1% (v/v) formic acid solution of acetonitrile. The linear gradient elution was as follows: 0–2 min, 10–25% B; 2–4 min, 25–25% B; 4–10 min, 25–50% B; 10–14 min, 50–50% B; 14–18 min, 50–85% B; and 18–19 min, 85–90% B.

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Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

The authors greatly appreciate the financial support from the National Key R&D Program of China (nos. 2017YFC1701103 and 2017YFC1701105), National Modern Agricultural Industrial Park of China (no. njf [2017] 110), National Natural Science Foundation of China (no. 31401613), and Cultivation Plan for High-level University Academic Backbone of Guangzhou Medical University in 2017 (no. gydf [2017] 210).

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