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Phyto-phospholipid complexes (phytosomes): A novel strategy to improve the bioavailability of active constituents



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ARTICLE INFO**Article history:**

Received 23 November 2017

Revised 14 April 2018

Accepted 29 May 2018

Available online 27 July 2018

Keywords:

Phyto-phospholipid complexes

Active constituents

Hydrogen bonds

Bioavailability

ABSTRACT

Although active constituents extracted from plants show robust *in vitro* pharmacological effects, low *in vivo* absorption greatly limits the widespread application of these compounds. A strategy of using phyto-phospholipid complexes represents a promising approach to increase the oral bioavailability of active constituents, which is consist of “label-friendly” phospholipids and active constituents. Hydrogen bond interactions between active constituents and phospholipids enable phospholipid complexes as an integral part. This review provides an update on four important issues related to phyto-phospholipid complexes: active constituents, phospholipids, solvents, and stoichiometric ratios. We also discuss recent progress in research on the preparation, characterization, structural verification, and increased bioavailability of phyto-phospholipid complexes.

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1. Introduction

Active constituents extracted from plants have been used to treat various diseases since ancient times [1,2]. Silybin extracted from milk thistle fruit has been used to provide liver support for 2000 years [3]. Curcumin extracted from turmeric have been reported to exhibit antioxidant and anticancer properties [4,5]. Moreover, some of the most

widely studied active constituents are polyphenols, such as flavonoids, terpenoids, and phenolics [6,7]. However, many active constituents extracted from plants are poorly absorbed when administered orally, which limits their widespread application [8,9]. The poor absorption of these compounds results from two properties. First, the multi-ring structures of polyphenols are too large to be absorbed by passive diffusion

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Peer review under responsibility of Shenyang Pharmaceutical University.

or non-active absorption. Second, the poor water or lipid solubility of these compounds prevents them from passing across the outer membrane of gastrointestinal cells [10,11].

Active constituents extracted from natural plants have been shown to exhibit robust *in vitro* pharmacological effects, but poor *in vivo* absorption. A variety of solutions have been proposed to counter the problem of poor absorption [12], such as the preparation of emulsions [13], liposomes [14], and nanoparticles [15], as well as the modification of chemical structures [16] and delivery as prodrugs [17]. Among the potential strategies, phyto-phospholipid complexes (known as phytosomes) have emerged as a promising strategy to enhance the bioavailability of active constituents [11]. Phyto-phospholipid complexes are prepared by complexing active constituents at defined molar ratios with phospholipids under certain conditions [18]. Amphiphatic phospholipids mainly act as “ushers” of active constituents to help them pass through the outer membrane of gastrointestinal cells, eventually reaching the blood [11]. After forming phospholipid complexes, the membrane permeability and oil-water partition coefficient of constituents are greatly improved. Thus, phyto-phospholipid complexes are more readily absorbed and generate higher bioavailability compared to free active constituents [19,20]. Encouragingly, the technique of phospholipid complexes has overcome the obstacle of poor bioavailability for many active constituents [21,22].

Therefore, the preparation of phyto-phospholipid complexes has recently received increased attention [23]. Based on the literature, we review various aspects of phyto-phospholipid complexes, including interactions, structures, characterization, structural verification, and increased bioavailability. We also highlight recent advances in the types of active constituents, phospholipids, solvents, and stoichiometric ratios.

2. Structure of phyto-phospholipid complexes

Phyto-phospholipid complexes are formed by interactions between active constituents and the polar head of phospholipids [23]. Interactions between active constituents and phospholipids enable phospholipid complexes to be an integral part in which the phospholipids head group is anchored, but the two long fatty acid chains do not participate in complex formation. The two long fatty acid chains can move and encapsulate the polar part of complexes to form a lipophilic surface. Phyto-phospholipid complexes form agglomerates when diluted in water, which resemble a small cell that shows some similarity to liposomes; the differences between liposomes and complexes are shown in Fig. 1 [24]. As can be seen from Fig. 1, the biggest difference between phytosomes and liposomes is that, in liposomes, the active ingredient is distributed in the medium contained the cavity or in the layers of the membrane, whereas in phytosomes, it is an integral part of the membrane, being the molecules stabled through hydrogen bonds to the polar head of the phospholipids.

Liposomes are closed vesicles formed by lipid bilayers that can encapsulate compounds within an aqueous compartment or multiple lipid bilayers, but do not mix with compounds [25].

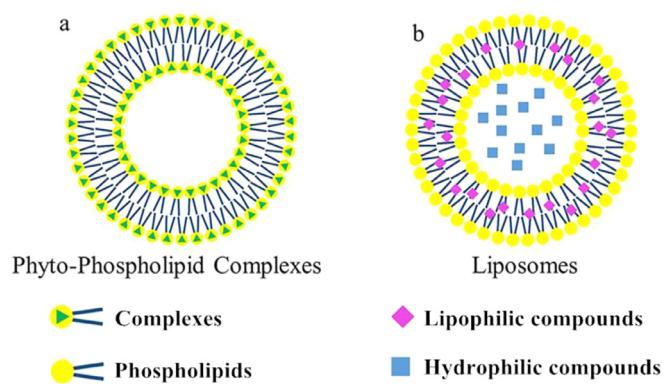


Fig. 1 – Structure of phyto-phospholipid complexes and liposomes.

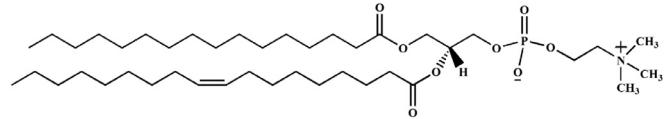


Fig. 2 – Structure of phosphatidylcholine (PC).

3. Phyto-phospholipid complex components

Bombardelli proposed that phyto-phospholipid complexes can be created from the reaction of phospholipids at a stoichiometric ratio with active constituents that are extracted from plants [26]. Based on subsequent studies, this initial description of phyto-phospholipid complexes has been challenged. According to the literature, we have proposed an updated list of the four essential components needed: phospholipids, phyto-active constituents, solvents, and the stoichiometric ratio involved in the formation of phyto-phospholipid complexes [24].

3.1. Phospholipids

Phospholipids are abundant in egg yolk and plant seeds. Currently, industrially produced phospholipids are available [24]. Phospholipids can be divided into glycerophospholipids and sphingomyelins depending on the backbone. Additionally, glycerophospholipids include phosphatidylcholine (PC), phosphatidylethanolamine (PE), phosphatidylserine (PS), phosphatidic acid (PA), phosphatidylinositol (PI), and phosphatidylglycerol (PG) [27].

PC, PE, and PS are the major phospholipids used to prepare complexes that are composed of a hydrophilic head group and two hydrophobic hydrocarbon chains [28]. Among these phospholipids, PC is the most frequently used to prepare phospholipid complexes. The structure of the PC is as follows (Fig. 2). The benefits of PC include its amphiphatic properties that give it moderate solubility in water and lipid media. Moreover, PC is an essential component of cell membranes, and accordingly it exhibits robust biocompatibility and low toxicity. PC molecules exhibit hepatoprotective activities, and have been reported to show clinical effects in the treatment of

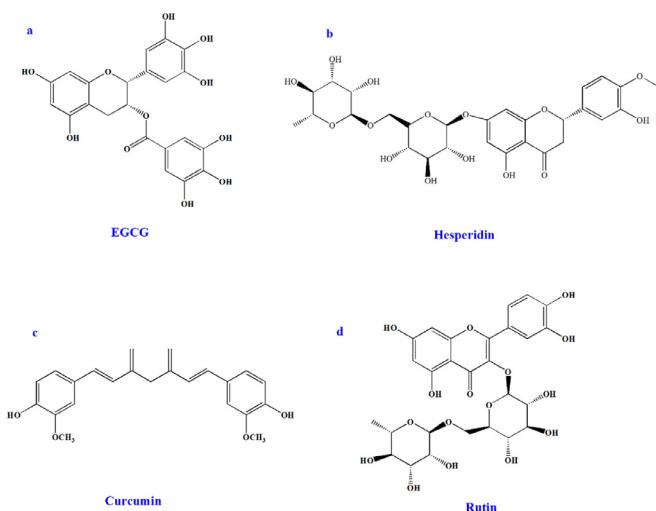


Fig. 3 – Structure of selected polyphenols: (a) EGCG, (b) Hesperidin, (c) Curcumin, and (d) Rutin.

liver diseases, such as hepatitis, fatty liver, and hepatocirrhosis [11,29]. Patel et al. prepared high-affinity small molecule-phospholipid complexes of siramesine and PA [33]. To date, the use of PG and PI to prepare phospholipid complexes has not yet been reported.

3.2. Phyto-active constituents

The active constituents of herbal extracts identified by researchers are generally defined based on robust *in vitro* pharmacological effects, rather than on *in vivo* activities. Most of these compounds are polyphenols. Some of the polyphenols structural drugs are shown below (Fig. 3). Some of the biologically active polyphenolic constituents of plants show affinity for the aqueous phase and cannot pass through biological membranes, such as hesperidin. By contrast, others have high lipophilic properties and cannot dissolve in aqueous gastrointestinal fluids, such as curcumin and rutin. Phyto-phospholipid complexes cannot only improve the solubility of lipophilic polyphenols in aqueous phase but also the membrane penetrability of hydrophilic ones from aqueous phase. Furthermore, the production of complexes can protect polyphenols from destruction by external forces, such as hydrolysis, photolysis, and oxidation [2].

Apart from polyphenols, other molecules from herbal extracts can also be prepared in complexes with phospholipids, such as 20(R)-25-methoxyl-dammarane-3 β ,12 β ,20-triol (25-OCH₃-PPD) [30], evodiamine (EVO) [31], and siramesine (SRM) [32] (Fig. 4). Therefore, phyto-phospholipid complexes are no longer limited to polyphenols and, in theory, the strategy of generated complexes is suitable for any active molecule [2].

More phyto-phospholipid complexes on the market are shown in Table 1.

3.3. Solvents

Different solvents have been utilized by different researchers as the reaction medium for formulating phyto-phospholipid

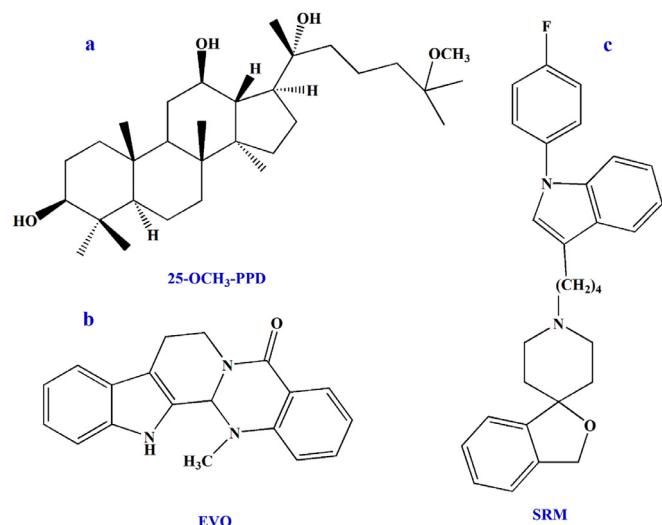


Fig. 4 – Structure of (a) 25-OCH₃-PPD, (b) EVO, and (c) SRM.

complexes. Traditionally, aprotic solvents, such as aromatic hydrocarbons, halogen derivatives, methylene chloride, ethyl acetate, or cyclic ethers etc. have been used to prepare phyto-phospholipid complexes but they have been largely replaced by protic solvents like ethanol [23,38]. Indeed, protonic solvents, such as ethanol and methanol, have been more recently been successfully utilized to prepare phospholipid complexes. For example, Xiao prepared silybin-phospholipid complexes using ethanol as a protonic solvent; subsequently, the protonic solvent was removed under vacuum at 40°C [22].

Various types of solvents have been successfully studied. When the yield of phospholipid complexes is sufficiently high, ethanol can be a useful and popular solvent that leaves less residue residual and causes minimal damage. Some liposomal drug complexes operate in the presence of water or buffer solution, where the phytosomes interact with a solvent with a reduced dielectric constant [33].

Recently, many studies have used the supercritical fluid (SCF) process to control the size, shape, and morphology of the material of interest. Supercritical anti solvent process (SAS) is one of the SCF technologies that are becoming a promising technique to produce micronic and submicronic particles with controlled size and size distribution [39]. In this technique, a supercritical fluid (usually CO₂) will be chosen as an anti-solvent to reduce the solute's solubility in the solvent.

3.4. Stoichiometric ratio of active constituents and phospholipids

Normally, phyto-phospholipid complexes are employed by reacting a synthetic or natural phospholipid with the active constituents in a molar ratio ranging from 0.5 to 2.0 [40]. Whereas, a stoichiometric ratio of 1:1 is considered to be the most efficient ratio for preparing phospholipid complexes [41]. For example, quercetin-phospholipid complexes were prepared by mixing Lipoid S 100 and quercetin at a molar ratio of 1:1 [42]. However, different stoichiometric ratios of active constituents and phospholipids have been used. Maryana et al. prepared silymarin-phospholipid complexes with dif-

Table 1 – Therapeutic applications of different phyto-phospholipid complexes on the market [33–37].

S. no.	Trade name	Phytoconstituents complex	Indication
1.	Greenselct® phytosome	Epigallocatechin 3-O-gallate from <i>camelia sinensis</i> (green tea)	Systemic antioxidant. Protect against cancer and damage to cholesterol.
2.	Ginkgoselect® phytosome	Ginkgo flavono glycosides from <i>Ginkgo biloba</i>	Protects brain and vascular lining.
3.	Silybin phytosome	Silybin from silymarin	Provides antioxidant protection for the liver and skin.
4.	Glycyrrhiza phytosome	18-beta glycyrrhetic acid	Anti-inflammatory activity
5.	Grape seed (Leucoselect) phytosome	Procyanidins from <i>Vitis Vinifera</i>	Anti-oxidant, anticancer
6.	Curcumin (Merinoselect) phytosomes	Polyphenol from <i>Curcuma Longa</i>	Cancer chemo preventive agent improving the oral bioavailability of curcuminoids, and the plasma.
7.	Oleaselect™ phytosome	Polyphenols from olive oil	Inhibit harmful oxidation of LDL cholesterol, and provides an anti-inflammatory effect.
8.	Sabalselect® phytosome	An extract of saw palmetto berries through supercritical CO ₂ (carbon dioxide) extraction	It is beneficial to the normal functioning of the prostate
9.	PA2 phytosome horse Chestnut bark	Proanthocyanidin A2 from	Anti-wrinkles, UV protectant
10.	Zanthalene phytosome	Zanthalene from <i>zanthoxylum bungeanum</i>	Soothing, anti-irritant, anti-itching
11.	Centella phytosome	Terpenes	Vein and skin disorders
12.	Hawthorn phytosome™	Flavonoids from <i>Crataegus sp.</i>	Nutraceutical, cardio-protective and antihypertensive

ferent stoichiometric ratios of 1:5, 1:10, and 1:15; they found that the complexes with a stoichiometric ratio of 1:5 showed the best physical properties and the highest loading capacity of 12.18% ± 0.30% [43]. Yue et al. conducted a comparative study using the stoichiometric ratios of 1:1, 1.4:1, 2:1, 2.6:1, and 3:1 to generate oxymatrine-phospholipid complexes; they determined that optimal quantity was obtained at a ratio of 3:1 [20]. Therefore, a stoichiometric ratio of 1:1 is not always optimal for the formulation of phospholipid complexes. For different types of drugs, we should experimentally adjust the stoichiometric ratio of active constituents and phospholipids according to distinct purposes, such as the highest drug loading.

4. Interactions between active constituents and phospholipids

In 1989, Bombardelli reported a chemical bond between a flavonoid molecule and a phospholipid molecule [44]. In the past, there was controversy about the formation of phyto-phospholipid complexes [45,46]. A study on the interaction of the 20(S)-protopanaxadiol phospholipid complexes at the molecular level using molecular docking showed that a hydrogen bond formed between one of the -OH group in 20(S)-protopanaxadiol and the —P=O group in the phospholipids (Fig. 5) [47]. Phyto-phospholipid complexes, which employed by reaction of stoichiometric amount of phospholipids and the phytoconstituents complex, are revealed by the spectroscopic data that the phospholipid – active ingredient interaction is due to the formation of hydrogen bond between the polar head and the polar functionalities of the active ingredient. The ¹H NMR and ¹³C NMR data show that, the signal of the fatty chain has not changed both in free phospholipids and in the complex, which suggested that long aliphatic chains are wrapped around the active principle, producing lipophilic envelope [40]. The same conclusion can also be drawn from the

thermal analysis in other studies that, the interaction between the two molecules had been attributed to formation of hydrogen bond or hydrophobic interaction [48].

In summary, most researchers agree that interactions between active constituents and phospholipids occur via hydrogen bonds to generate intermolecular force rather than chemical or hybrid bonds.

5. Advantage

5.1. Enhance the bioavailability

Multiple studies have revealed that phyto-phospholipid complexes can boost the absorption of through oral topical route, hence it can increase the bioavailability and reduce the required dose. So, it can significantly improve therapeutic benefit (Table 2).

Zhang et al. prepared phospholipid complexes of 25-OCH₃-PPD by solvent evaporation and found that the AUC_(0–24 h) of the complex increased from 26.65 to 97.24, which was 3.65-fold higher than free 25-OCH₃-PPD and resulted in a relative bioavailability of 365% [30]. Zhang prepared quercetin-phospholipid complexes and showed that the AUC_(0–t) of the quercetin-phospholipid complexes increased from 2.04 to 8.12, which was a 3.98-fold increase [42]. Chen et al. examined the pharmacokinetics and oral bioavailability of these complexes by comparing quercetin, kaempferol and isorhamnetin extracts from *Ginkgo biloba* (GBE) to their respective phospholipid complexes (GBP) in rats. The AUC_{0–T} of quercetin, kaempferol, and isorhamnetin for GBP increased by 2.42-, 1.95-, and 2.35-fold, respectively. This indicated that bioavailability was significantly enhanced by GBP compared with GBE [19].

After the formation of phyto-phospholipid complexes, the membrane permeability and oil–water partition coefficients of active constituents significantly improve. Thus, phyto-

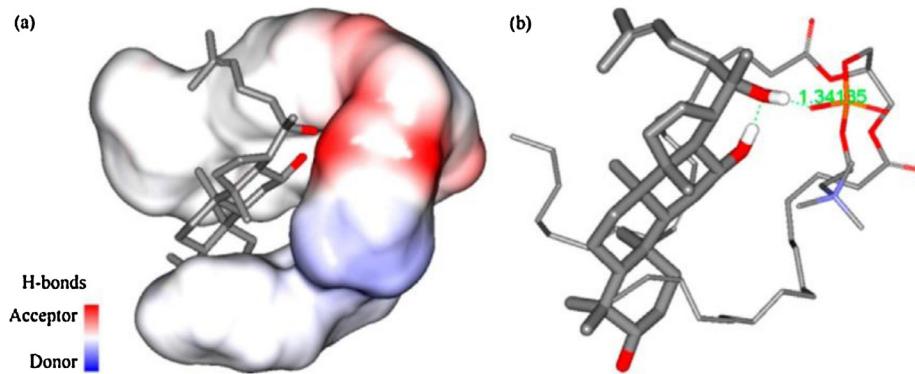


Fig. 5 – A 3D conformational model of 20(S)-protopanaxadiol and phospholipids after docking and optimization. (a) 20(S)-protopanaxadiol and phospholipids with an interpolated charge surface; (b) hydrogen bond interactions between the molecules (reproduced with permission from [47]). Copyright 2016 MDPI.

Table 2 – Major pharmacokinetic parameters [19,30,42,49–52].

Parameter	T _{max} (h)	C _{max} (ng/ml)	AUC _(0–24 h) (ng·h/ml)	Ref.
25-OCH ₃ -PPD	2.25 ± 2.47	7.18 ± 3.39	26.65 ± 7.11	[30]
25-OCH ₃ -PPD-phospholipid complexes	2.10 ± 0.55	28.07 ± 19.52	97.24 ± 40.17	[30]
Quercetin (10 ³)	0.50 ± 0.07	0.67 ± 0.08	2.04 ± 0.38	[42]
Quercetin-phospholipid complexes (10 ³)	0.75 ± 0.10	1.58 ± 0.11	8.12 ± 0.95	[42]
Quercetin	1.21	179.21	1368.26	[19]
Quercetin-phospholipid complexes	1.02	724.89	3321.05	[19]
Kaempferol	6.32	180.23	1139.59	[19]
Kaempferol-phospholipid complexes	5.83	323.56	2228.21	[19]
Isorhamnetin	7.21	195.96	1153.66	[19]
Isorhamnetin-phospholipid complexes	4.32	672.29	2722.37	[19]
Luteolin	2.5 ± 0.75	3410 ± 910	10,420 ± 2350	[49]
luteolin-phospholipid complexes	2 ± 0.63	9770 ± 1250	55,790 ± 5760	[49]
Berberine	0.5	66.01 ± 15.03	384.45 ± 108.62	[50]
berberine-phospholipid complexes	2	219.67 ± 6.02	1169.19 ± 93.75	[50]
Oleanolic acid (OA)	0.313 ± 0.12	59.5 ± 10.64	259.6 ± 22.98	[51]
Solidified OA-phospholipid complexes	0.46 ± 0.001	78.7 ± 10.18	360.6 ± 19.13	[51]
Rosuvastatin	1.94 ± 0.89	183.95*10 ³ ± 4.56*10 ³	671.56*10 ³ ± 5.64*10 ³	[52]
Rosuvastatin-phospholipid complexes	1.41 ± 0.95	674.17*10 ³ ± 1.24*10 ³	2018.59*10 ³ ± 3.79*10 ³	[52]

phospholipid complexes are more readily absorbed and generate increased bioavailability compared with free active constituents. Therefore, the preparation of phyto-phospholipid complexes has recently gained increased attention. Phylosomal formulations have also gained important applications in the pharmaceutical industry. Furthermore, products based on phyto-phospholipid complexes have been available on the market to treat several diseases or improve human health, such as Silymarin Phytosome® from milk thistle seed for antihepatotoxic activities [53], Escin β-sitosterol Phytosome® from horse chestnut fruit for antioedema effects, Leucoselect® Phytosome® from grape seed for antioxidant activities, and Greenselect® Phytosome® from green tea leaf for weight management [25,54].

5.2. Enhance percutaneous absorption

Phyto-phospholipid complexes can easily transition from a hydrophilic environment into the lipophilic environment of the cell membrane and then enter the cell

[37]. Therefore, a large number of studies have displayed that the percutaneous absorption of phytoconstituents is improved because of the application of phytoconstituents in form of phytosome [51,55].

Due to the above characteristics of skin penetration, phyto-phospholipid complexes are widely employed in transdermal field [56,57]. According to a recent report, Stefano TOGNI et al. prepared a quercetin-phospholipids 2% cream to increase the safety and tolerability of the quercetin-based formulation [56]. For the sake of overcoming the problem of lower skin penetration of Citrus aurantium and Glycyrrhiza glabra, Damle and Mallya developed a novel phyto-phospholipid complex cream [57].

5.3. Hepatoprotective effect

Compare with carriers employed in other drug delivery systems, phosphatidylcholine are crude ingredients that also reveal a great therapeutic benefits [39]. Phosphatidylcholine not only acts as an ingredient added to the formulation of phyto-

phospholipid complexes, but also acts as a hepatoprotective. So, when phosphatidylcholine is taken by the patient, it will show the synergistic effect to protect the liver. In some situations, phospholipids also have the nutritional benefits.

Since 1994, Choline has been reported that it is necessary for normal liver function [58]. *In vitro* studies have shown that these PLs increase hepatic collagenase activity and may thus help prevent fibrosis and cirrhosis by encouraging collagen breakdown [59]. Besides, lecithin has been verified that it can also produce a protective effect in non-alcoholic fatty liver disorders and protection against various other toxic substances like hepatitis A and B [60]. Thus, it has a great protective effect of liver function. Khan et al. prepared a phospholipid complex of luteolin (LPC) by solvent evaporation and its hepatoprotective potential was assessed against D-galactosamine and lipopolysaccharide (GalN/LPS) induced hepatic damage [49].

5.4. Other advantage

Phyto-phospholipid complexes possess better drug complexation rate and the preparation of phyto-phospholipid complexes is not complicated [61]. In addition, phyto-phospholipid complexes show better stability trait because the formation of chemical bonds between plant extracts and phosphatidylcholine molecule. By increasing the solubility of bile to active constituents, phyto-phospholipid complexes improve the liver targeting [37]. In some case, the duration of drug is prolonged. Due to a shorter half-time of Naringenin and rapid clearance from the body, frequent administration of the molecule is required. Wherefore, phospholipid complexes of Naringenin were prepared to prolong its duration in blood circulation [62]. According to another report, the phospholipid enhanced the plasma concentration of andrographolide in a significant manner and the effect persisted for a longer period of time. And the half-life of andrographolide–phospholipid complexes is 3.34 times that of pure andrographolide [63].

6. Methods for the preparation of phyto-phospholipid complexes

6.1. Methods

There are three primary methods for preparation of phyto-phospholipid complexes, including solvent evaporation, freeze-drying, and anti-solvent precipitation.

Solvent evaporation is a traditional and frequently used method for preparing phospholipid complexes. Shan and colleagues applied the solvent evaporation method to prepare oleanolic acid-phospholipid complexes [64]. Yu et al. prepared a berberine-phospholipid complexes (P-BER) by a rapid solvent evaporation method followed by a self-assembly technique, for the sake of developing a more efficient berberine drug delivery system [50]. Briefly, a designated stoichiometric ratio of active constituents and phosphatidylcholines was placed in the shared round-bottom and dissolved in a favorable solvent by heating at an optimal unchanging temperature with a certain time (Fig. 6). Prepared complexes

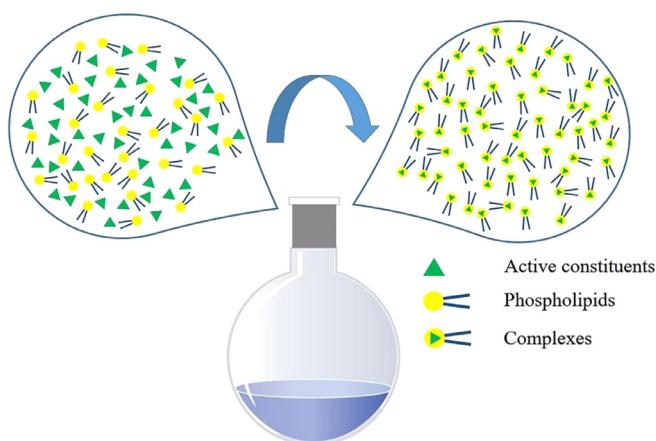


Fig. 6 – Schematic of the preparation of phyto-phospholipid complexes.

can be obtained by evaporation to remove solvent under vacuum.

Other methods for preparing phospholipid complexes include anhydrous co-solvent lyophilization [65,66] and anti-solvent precipitation [67,68]. For example, Diosmin was firstly dissolved in DMSO, and then the diosmin was added to the solution of SPC followed by stirring for 3 h until complex formation [69]. Kaempferol-phospholipid complexes were also prepared by lyophilization method [70]. In addition, allium cepa-phospholipid complexes (ACP), which showed a significant antidiabetic activity, were prepared by anti-solvent precipitation [71].

Out of these, some methods described in various studies are discussed here. Li et al. developed supercritical anti-solvent precipitation to produce puerarin phospholipid complexes and proposed that supercritical fluids were superior to conventional methods for drug phospholipid complexes preparation [72]. Damle and Mallya used Salting-Out Method and Film Formation Method to prepared a phyto-phospholipid complexes [57].

6.2. The influencing factors of phyto-phospholipid complexes

The factors that influence the formation of phyto-phospholipid complexes are mainly included solvent, stoichiometric ratio of active constituents, reaction temperature and reaction time [55,69,73–75]. Depending on the desired target, different process variables can be selected. For maximum yield, Saoji et al. studied the influence of process variables such as the phospholipid-to-drug ratio, the reaction temperature and the reaction time, and used a central composite design to acquire the optimal formulation [55]. For best solubility and skin permeation, Das and Kalita prepared a rutin phytosome in different stoichiometric ratios [74]. According to a recent report [75], by changing stoichiometric ratios and reaction temperature, the highest yield apigenin-phospholipid complexes are prepared by Telange and his colleagues.

7. Yield (complexation rate) of phyto-phospholipid complexes

The yield of active constituents in complex with phospholipids represents an extremely important index for prescription screening. The weight difference between the initial active constituent and free compounds is the amount of active constituent in complex with phospholipids. The formula is as follows:

$$\text{Yield}(\%) = [(a - b)/a] \times 100\%$$

Where “a” is the weight or content of initial active constituent, “b” is the weight or content of free active constituent, and “(a – b)” is the weight or content of the phospholipid complexes [76].

Based on the properties of the active constituents, high-performance liquid chromatography (HPLC) or ultraviolet spectrophotometry can also be used to calculate the yield. The choice of solvent, temperature, duration, drug concentration, and stoichiometric ratio of active constituents to phospholipids are the main factors that affect the yield of phospholipid complexes.

8. Characterization of phyto-phospholipid complexes

8.1. Solubility and partition coefficient

Determining solubility in either water or organic solvents and the n-octanol/water partition coefficient (P) is necessary to characterize active constituents, active constituent phyto-phospholipid complexes and physical mixtures. Generally, phyto-phospholipid complexes have better lipophilicity and hydrophilicity than active constituents, and typically exhibit improved lipophilicity [24]. Rahila confirmed that embelin in complex has greater solubility in n-octanol and water than embelin and its respective physical mixtures [77].

8.2. Particle size and zeta potential

Particle size and zeta potential are important properties of complexes that are related to stability and reproducibility. In general, the average phospholipid complexes particle size ranged from 50 nm to 100 μm . Mazumder prepared sinigrin phytosome complexes, and the average particle size and zeta potential of the complex were 153 ± 39 nm and 10.09 ± 0.98 mV, respectively [67].

8.3. Scanning electron microscopy (SEM) and transmission electron microscopy (TEM)

SEM has yielded important insights into the solid state properties and surface morphology of complexes. TEM is often used to study the crystallization and dispersion of nano-materials and to measure the particle size of nanoparticles. SEM has shown that active compounds can be visualized in a highly crystalline state, but the shaped crystals disappeared after

complexation. When diluted in distilled water under slight shaking, TEM showed that phyto-phospholipid complexes exhibit vesicle-like structures [67].

9. Structural verification of phyto-phospholipid complexes

9.1. Ultraviolet spectra (UV-spectra)

Samples that reflect different absorption in the UV wavelength range can be used to characterize own structural properties. Most studies have revealed no differences in the UV absorption characteristics of constituents before and after complexation. Xu et al. prepared luteolin-phospholipid complexes and found that the characteristic peaks of luteolin remained present [78]. Therefore, we conclude that the chromophores of compounds are not affected by being complexed with phospholipids.

9.2. Differential scanning calorimetry (DSC)

In DSC, interactions can be observed by comparing the transition temperature, appearance of new peaks, disappearance of original peaks, melting points, and changes in the relative peaks area [76]. Phyto-phospholipid complexes usually display radically different characteristic peaks compared to those of a physical mixture. It is assumed that, in addition to the two fatty chains of phospholipids, strong interactions occur in the active ingredients and the polar part of phospholipids also inhibits free rotation. Das and Kalita prepared phyto-phospholipid complexes that contained rutin and the resulting DSC thermogram showed two characteristic peaks that were lower than that of the physical mixture and the peaks of rutin and PC disappeared [79].

9.3. Fourier transform infrared spectroscopy (FTIR)

FTIR is a powerful method for structural analysis, and yields different functional groups that show distinct characteristics in band number, position, shape, and intensity. The formation of phyto-phospholipid complexes can be verified by comparing the spectroscopy of phospholipid complexes to that of physical mixtures. Separate studies may show different results. Indeed, Das and Kalita prepared phyto-phospholipid complexes composed of rutin. The FTIR of a physical mixture of rutin and phyto-phospholipid complexes was superimposable with that of pure rutin [79]. When Mazumder et al. prepared sinigrin-phytosome complexes, the FTIR of phytosome complex showed different peaks from that of sinigrin, phospholipids and their mechanical mixtures [67].

9.4. X-ray diffraction

Currently, X-ray diffraction is an effective method to examine the microstructure of both crystal materials and some amorphous materials. X-ray diffraction is usually performed on either active constituents or active constituent phyto-phospholipid complexes, PCs and their physical mixtures.

X-ray diffraction of an active constituent and physical mixture shows intense crystalline peaks that indicate a high crystal form. By contrast, active constituent phyto-phospholipid complexes do not exhibit crystalline peak, which suggests that the constituents in complex with phospholipids exhibit a molecular or amorphous form. That may account for the observation that phyto-phospholipid complexes have better lipophilicity and hydrophilicity than active constituents [24].

9.5. Nuclear magnetic resonance (NMR)

The ^1H NMR and ^{13}C NMR techniques play an important role in the identification of the structures of the complexes. As noted above, interactions between polyphenols and phospholipids are created by hydrogen bonds rather than chemical bonds. Angelico et al. established based on NMR data that hydrogen bonds can form between some polar phenolic functional groups of silybin A and phospholipids [80]. The spectra of different phyto-phospholipid complexes suggest that the hydrophobic side of lipids can act to cover the envelope on the central choline-bioactive parts of these complexes.

10. Conclusions

Phospholipids show affinity for active constituents through hydrogen bond interactions. In theory, a phospholipid complex strategy should be suitable for any active molecule not limited to polyphenols. With advances in research, we update the newest advances in phospholipids, phyto-active constituents, solvents, and stoichiometric ratios that are essential to prepare phospholipid complexes. The characterization and structural verification of phospholipid complexes has been well established. Bioavailability can be significantly improved with the help of phospholipids compared with chemically equivalent non-complexed forms. The potential of phyto-phospholipid complexes, with the effort of clinicians and other researchers, has a bright future for applications in the pharmaceutical field.

Conflict of interest

The authors report no conflicts of interest. The authors alone are responsible for the content and writing of this article.

Acknowledgments

This review did not receive any specific grant from funding agencies in the public, commercial, or not-for-profit sectors.

Supplementary materials

Supplementary material associated with this article can be found, in the online version, at doi:[10.1016/j.ajps.2018.05.011](https://doi.org/10.1016/j.ajps.2018.05.011).

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