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Evaluation of the cytotoxic and cellular proteome impacts of food-grade TiO₂ (E171) using simulated gastrointestinal digestions and a tri-culture small intestinal epithelial model

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Abstract

Engineered nanomaterials (ENMs) are widely used in the food industry; however, regulations for ENMs in food are still in the early stages of development due to insufficient health data. This study investigated the cytotoxicity and changes to the proteomic profile in an *in vitro* small intestinal epithelium model after exposure to digested food models containing the ubiquitous engineered particulate food additive, TiO₂ (E171) with an average size around 110 nm. TiO₂ at 0.75% or 1.5% (w/w) concentrations in either a fasting food model (FFM) or a standardized food model (SFM) based on American diet were digested using an *in vitro* oral-gastric-small intestinal simulator, and the resulting digestas were applied to a small intestinal epithelium tri-culture cellular model. Effects on cell layer integrity, cytotoxicity, and oxidative stress were assessed. In order to explore the impact on cellular processes beyond basic cytotoxicity, mass spectrometry-based quantitative proteomic analyses of control and exposed tri-culture cells was performed. TiO₂ in FFM, but not in SFM, produced significant, dose-dependent cytotoxicity (24%, p<0.001), and at the higher dose caused significant oxidative stress (1.24-fold, p<0.01), indicative of a food matrix effect. No significant perturbations of the cellular proteome were observed with TiO₂ in either FFM or SFM food models. However, proteins involved in energy metabolism and protein synthesis were up-regulated by digestas from SFM compared to those from FFM, indicative of a food matrix effect on the cellular proteome. Interestingly, the differences in profiles between the two food models was more pronounced in the presence of TiO₂. Together, these results indicate that TiO₂ in a fasting diet may be slightly cytotoxic, and that ingested TiO₂ does not significantly alter the epithelial proteome, whereas the food matrix alone can have a dramatic effect on the proteome.

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Conflicts of interest

There are no conflicts to declare.

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Keywords

Engineered Nanomaterials; Food additive; Titanium Dioxide; Toxicity; Proteomics

Introduction

Nanotechnology has been rapidly developing over the past two decades. The unusual properties of engineered nanomaterials (ENMs) make them particularly useful in the food industry, where ENMs are widely used to improve food quality, safety, and nutritional benefits^{1–8}. In some cases, ENMs are added to food products intentionally, such as when ENMs are used to modify color and rheology, or as delivery systems. In other cases, ENMs are not added to the food directly, but may find their way into the final food products through contact with other materials containing ENMs, (*i.e.*, ENMs used for food packaging and sensors) or in agriculture applications⁹. With the increasing number of novel applications, and a growing number of ENMs with potential uses in the agriculture and food industries, human exposures are inevitable¹⁰.

The growing human oral exposure to expanding libraries of inorganic and organic ENMs raises concerns about the potential health risks^{11–13}. Emerging nano-safety research shows that the smaller size of the nano-scale materials gives them unique physicochemical properties, including much larger surface-to-mass ratios compared to bulk or micron-scale materials, which renders them more bioactive^{12,14,15}. More importantly, it was shown that nanoscale materials can bypass biological barriers and become systemic and interfere with cellular processes and functions^{16–20}.

In addition, interactions of ENMs with complex food matrices, and their transformations across the gastrointestinal tract (GIT) may alter an ENM's properties, such as electrical charge, solubility, agglomeration state, and surface composition^{1,21–23}. As a result, the biological fate and potential toxicity of ingested ENMs may be considerably different from those of the pristine ENMs. Therefore, understanding and accounting for these transformations is essential to the study of ingested ENMs^{13,21}. Nonetheless, the majority of *in vitro* cell culture studies of ingested ENMs have ignored the impact of food matrix and GIT bio-transformations²⁴.

The widely used food additive titanium dioxide (TiO₂), known as E171, is used as a whitening agent in over 900 food products, such as chewing gums, dairy products, and sweets^{25,26}. The human exposure to TiO₂ through oral ingestion has been estimated to be about 1 mg of Ti /kg body weight/day for US adults, and between 2 and 4 times that amount for US children²⁶. Food-grade TiO₂ (E171) has a wide range of particle sizes. It has been reported that about 40% of the particles in food-grade TiO₂ are in the nanoscale (are smaller than 100 nm in at least one dimension)^{26–28}.

Results from *in-vitro* cellular and animal studies on the oral toxicity, and tissue accumulation of TiO₂ continue to grow and are contradictory^{28,29}. This is likely because the different studies used different oral doses, particle sizes and crystal forms. For example, a single oral dose of TiO₂ nanoparticles (25, 80, or 155 nm at 5000 mg/kg bw) induced

hepatic injury, nephrotoxicity, and myocardial damage, and also resulted in TiO₂ accumulation in liver, lung, spleen, and kidney tissues of mice³⁰. In contrast, in a chronic study in which rats received repeated doses of up to 1041 mg/kg of 21 nm mixed anatase/rutile TiO₂ nanoparticles, no significant toxicity or accumulation in tissues or urine was observed, and high concentrations of TiO₂ were detected in feces, suggesting elimination primarily by the fecal route³¹. Due to growing concerns about the safety of E171, the French food safety authority has recently decided to ban foods containing E171 from their markets beginning January 1, 2020. However, US regulations pertaining to TiO₂ and other ENMs in food products are still in the early stages of development, largely due to a lack of relevant health data.^{23,32}

In this *in vitro* study, TiO₂ (E171) was incorporated into either a fasting food model (FFM) or a standardized food model (SFM) based on the typical American diet³³. The food models, with or without TiO₂, were digested using a three phase (oral, gastric, small intestinal) GIT simulator, and the resulting digesta was applied to an *in vitro* small intestinal epithelial tri-culture cellular model for toxicity assessments²¹. It is worth noting that this integrated methodology takes into account the effects of the food matrix and transformations of ENMs across the GIT²¹. Liquid chromatography coupled with tandem mass spectrometry (LC-MS/MS) was employed to determine the cellular proteome response to ingested TiO₂. A highly reproducible proteomic workflow with label-free quantification made it possible to quantitatively profile the potential alterations of the proteomes of the gut tri-culture cellular system. This study adds important information to the knowledge of the potential impacts of ingested nanomaterials under physiologically relevant conditions.

Materials and Methods

Materials

Food-grade TiO₂ (E171) was purchased from Pronto Foods Co. (Chicago, IL, USA). Detailed physico-chemical characterization of the TiO₂ (E171) was performed and described in detail by the authors in previous publication^{27,28}. Based on Title 21 of Code of the Federal Regulations (21 CFR §73.575), the allowable maximum concentrations of the color additive TiO₂ (E171) is 1% by weight, of the food. Therefore, two starting concentrations of food-grade TiO₂ (E171) were chosen for this study, 0.75% and 1.5% w/w in food models. Assuming average human food intake is 2.5 kg per day and average human body weight is 75 kg, the doses used in this study is equal to 150 and 300 mg Ti/kg body weight/day.

Food models

Phosphate buffer (5 mM) at pH 7.0 was used as a fasting food model (FFM). The standardized food model (SFM), based on the typical American diet, was previously described by the authors³³. In summary, SFM is prepared as an oil-in-water emulsion consisting of 5.2% digestible carbohydrate (modified corn starch), 4.6% sugar (sucrose), 3.4% protein (sodium caseinate), 3.4% fat (corn oil), 0.7% dietary fiber (pectin), and 0.5% sodium chloride, and spray-dried to produce a powder for storage at 4 °C. To reconstitute 100 g of the SFM, 17.8 g of the powder was combined with 82.2 g distilled water and stirred at 900 RPM for 30 minutes.

Colloidal characterization before digestion and in small intestinal phase digesta

A laser diffraction particle size analyzer (Mastersizer 3000, Malvern Instruments, Ltd) was used to obtain the surface-weighted mean diameters ($D[3,2]$), volume-weighted mean diameters ($D[4,3]$), and volume-weighted size distributions of particles before digestion and in small intestinal phase digesta from the simulated GIT digestion (described below). Zeta-potential (ζ) were determined using a Zetasizer Nano-ZS (Malvern Instruments, Ltd., Worcestershire, UK). All measurements were performed in triplicate, at 25 °C, using disposable optical polystyrene micro-cuvettes.

In vitro simulated digestions

FFM and SFM alone (control, 0% TiO₂) or FFM and SFM containing 0.75% or 1.5% w/w TiO₂ (E171), were used as initial food inputs for simulated digestions. Six samples were generated in total, *as follows*: (1) FFM control (5 mM phosphate buffer alone); (2) FFM with TiO₂ (0.75% w/w); (3) FFM with TiO₂ (1.5% w/w); (4) SFM control; (5) SFM with TiO₂ (0.75% w/w); (6) SFM with TiO₂ (1.5% w/w).

In vitro simulated digestion of the above samples was performed using a 3-phase (oral, gastric, small intestinal) simulator (Figure 1) as previously described in detail by the authors²¹. Briefly, in the oral phase, food inputs were brought to 37 °C, mixed with pre-warmed (37 °C) simulated saliva fluid, and inverted by hand for 10 seconds to mimic agitation in the mouth. The resulting oral phase digestas were then combined with pre-warmed simulated gastric fluid and incubated at 37 °C for two hours in a shaking incubator at 200 RPM to complete the stomach phase. In the small intestinal phase, stomach digestas were combined with pre-warmed solutions containing additional salts, bile extract and lipase to simulate intestinal fluid, and incubated at 37 °C for 2 hours, while maintaining a constant pH of 7.0, using a pH stat titration device (TitroLine® 7000, SI Analytics, GmbH, Germany).

It is worth noting that the initial concentrations of TiO₂ (E171) were diluted by a factor of 1/48 in total with digestive fluids in the oral (1/2), gastric (1/2) and small intestinal (1/3) phases, and with cell culture medium (1/4) before being applied to cells. Therefore, the final concentrations of TiO₂ (E171) used to expose the triculture cellular model were 150 µg/mL and 300 µg/mL for TiO₂ (E171) at starting concentrations of 0.75% and 1.5% w/w in food model, respectively.

Tri-culture small intestinal epithelial model and treatments

The tri-culture small intestinal epithelial model illustrated in Figure 1, was previously described and characterized in detail by the authors^{21,43}. In summary, Caco-2 cells and HT29-MTX cells were first co-cultured for 2 weeks on an 0.4 µm pore membrane transwell insert in a 6-well cell culture plate, during which time the Caco-2 cells differentiate to resemble intestinal enterocytes, and HT29-MTX cells, similar to intestinal goblet cells, secrete large amounts of mucus. Raji B cells were then introduced to the basolateral compartment of the transwell plate to stimulate some of the matured Caco-2 cells to differentiate and acquire an M-cell-like phenotype. This tri-culture model of intestinal epithelium mimics the absorptive enterocytes, goblet cells and M cells that populate the

human small-intestinal lining and is therefore a suitable *in vitro* cell model for prediction of small intestinal toxicity in humans.

Cells were obtained from Sigma, Inc. Briefly, Caco-2 and HT29-MTX cells were grown in high-glucose DMEM supplemented with 10% heat-inactivated fetal bovine serum (FBS). Raji B cells were cultured in RPMI 1640 media supplemented with 10% FBS. For transwell inserts, Caco-2 and HT29-MTX cells were combined in a ratio of 3:1 (Caco-2 :HT29-MTX). 1.5 mL portion of the cell mixture was seeded in the apical chamber, and 2.5 mL of complete DMEM media was added to the basolateral compartment of a 6-well transwell plate. After four days, media was changed every other day, until day 15. On day 15 and 16, the media in the basolateral compartment was replaced with 2.5 mL of a suspension of Raji B cells at a concentration of 1×10^6 cells/mL in 1:1 DMEM: RPMI complete media. Cell layer integrity, cytotoxicity (LDH release), and proteomic analysis using tri-cultures on transwells were initiated on day 17.

Co-cultures of Caco-2/HT29-MTX in 96-well plates were used for oxidative stress (ROS production) studies (see below) which require closed-bottom adherent cell cultures suitable for plate reader fluorescence measurements. Raji B cells were not used here, since they are suspension feeder cells (not part of the epithelium) which were added to the transwell basolateral compartments to promote M-cell differentiation of some apical Caco-2 cells. Raji cells could adhere to mucus, or become incorporated in the epithelial layer, if applied apically in closed 96-well plates. To prepare co-cultures, Caco-2 and HT29-MTX cells at a 3:1 ratio was seeded at a total 3×10^4 cells/well (100 μ L of cell mixture) in black-walled, clear optical bottom plates (BD Biosciences). After four days, media was changed every other day, until day 17.

The final small intestinal phase digestas from simulated digestions were combined with DMEM media at a ratio of 1:3, and the mixtures were applied to the cells on day 17 (1.5 mL to the apical compartment for transwell inserts, 200 μ L per well for 96-well plates).

TEER, oxidative stress and cytotoxicity analysis

Transepithelial electrical resistance (TEER) was measured after 24 h treatment in transwell plate using an EVOM2 Epithelial Volt/Ohm Meter with a Chopstick Electrode Set (World Precision Instruments).

Cytotoxicity test was performed after 24 h treatment in transwell tri-cultures using the Pierce LDH assay kit (Sigma Aldrich, St. Louis, MO, USA) according to manufacturer's instructions. Briefly, untreated wells were used to measure spontaneous LDH release control. Lysed wells were used to measure maximum LDH release control. Apical fluid in each well was transferred to a tube which was then centrifuged, and 50 μ L of the supernatant from each tube was dispensed in a fresh 96-well plate. 50 μ L of reaction mixture was added. Plates were incubated at room temperature for 30 minutes, and 50 μ L stop solution was added and mixed by tapping. Absorbance was measured at 490 nm (A_{490}) and 680 nm (A_{680}). To calculate LDH activity, A_{680} values were subtracted from measured A_{490} values to correct for instrument background. To correct for digesta background, LDH activities from no-cell controls were subtracted from test well LDH activities. Percent cytotoxicity

was calculated by subtracting spontaneous LDH release values from treatment values, dividing by total LDH activity (Maximum LDH activity – Spontaneous LDH activity), and multiplying by 100.

ROS analysis was performed after 6 h treatment in 96-well co-cultures. Production of ROS was assessed using the CellROX® green reagent (Thermo Fisher Scientific, Waltham, MA, USA) according to manufacturer's instructions. Cells treated with 100 μ M menadione for 1 h at 37°C were used as positive controls. Briefly, media was removed from test wells and replaced with 100 μ L of working solution, and plates were incubated for 30 minutes at 37 °C. Wells were then washed 3 times with PBS, and fluorescence was measured at 480 nm (excitation)/520 nm (emission).

Sample preparation for proteomics

Cells were removed from transwell membrane using a cell scraper, and then lysed in 100 μ l of lysis buffer (8 M urea in 50 mM NH_4HCO_3 , pH 7.8) with sonication for 30s (Branson 1500R, Branson) and vortexing. Total solubilized proteins in the lysates was quantified using the Bicinchoninic acid (BCA) assay (Thermo Fisher Scientific). An aliquot of 300 μ g of proteins was taken from each sample, and the sample volume was adjusted to 100 μ l with lysis buffer. Reduction of proteins was carried out by incubation with 5 mM dithiothreitol (DTT) at 37 °C for 1h, followed by alkylation with 10 mM iodoacetamide at room temperature for 1 h in the dark. The resulting samples were diluted five-fold with 50 mM NH_4HCO_3 , pH 7.8, and were then subjected to overnight digestion with sequencing grade modified trypsin (Promega) at a 1:50 enzyme to protein ratio in the presence of 1 mM CaCl_2 at 37 °C. Protein digests were further purified using solid phase extraction with Discovery C18 columns (Sigma-Aldrich). Prior to LC-MS/MS analysis, the peptides were resolubilized in 0.1% formic acid, quantified by BCA, and adjusted to a concentration of 0.1 μ g/ μ l.

LC-MS/MS proteomic analysis

LC-MS/MS data were acquired on a Q-Exactive mass spectrometer (Thermo Scientific) coupled with a nanoAcquity UPLC system (Waters). Seven μ L of peptide samples were injected and separated by a custom packed analytical C18 column (70 cm \times 75 μ m i.d., 3 μ m particle size of Jupiter C18, Phenomenex) in a 3 h gradient. The flow rate of the mobile phase (buffer A: 0.1% formic acid in water; buffer B: 0.1% formic acid in acetonitrile) was 300 nL/min. For peptide elution buffer B was increased to 12% at 36 min, 30% at 135 min, 45% at 175 min and finally 95% at 180 min. A data-dependent acquisition method was applied, in which a full MS scan was followed by up to 12 data-dependent MS/MS scans of the most abundant peptide precursors. A dynamic exclusion of 30 s was implemented to prevent repeated sequencing of previously selected peptides. For MS scans (400–2000 m/z; resolution of 35000 at 400 m/z), the automatic gain control (AGC) was set to 3×10^{-6} with the maximum injection time (IT) of 20 ms. An isolation window of 2 m/z in the quadrupole was used for selecting precursor ions, which were fragmented by higher-energy collisional dissociation (HCD) at a normalized collision energy of 35. For MS/MS scans (200–2000 m/z with first fixed mass of 100 m/z; resolution of 17500 at 400 m/z), the AGC was set to 1×10^{-5} and maximum IT was 100 ms.

Proteomic data analysis

Raw MS/MS data files were processed with MaxQuant (version 1.6.1.0). The non-redundant human UniProt proteome database (with 20,198 protein entries as of the April 2017 release) was used for searching. The searching parameters included carbamidomethylation on cysteine residues as fixed modifications, and acetylation on the peptide N-terminal, as well as oxidation on methionine, as variable modifications. Trypsin was selected as the enzyme for digestion, and a maximum of two missed cleavages was allowed. Mass tolerances for the first peptide search and main peptide search were 20 and 4.5 ppm, respectively. Match tolerance for MS/MS was 0.5 Da. A decoy database approach was used to limit the false discovery rate (FDR) to less than 0.01 for peptide spectrum matches. Proteins were also identified at FDR less than 0.01 with at least one unique peptide. For relative quantification, the label-free quantification (LFQ) algorithm was used with the minimum LFQ ratio count of 2. The “match between runs” feature was enabled with a 0.7 min match time window.

The LFQ protein intensities obtained from MaxQuant were first log₂ transformed. Proteins quantified in all samples were used for further analysis. For data visualization in a clustered heatmap, a median-centered normalization was performed by subtracting the protein abundance from the median for each protein. The R package “pheatmap” was used to perform clustering analysis (“complete” as the clustering method) and to generate the heatmap. For statistical analysis, the protein abundance data were subjected to two-sample t test in Perseus (version 1.6.1.1). Permutation-based FDR calculation was used to control false discoveries in multiple-hypothesis testing. Significant proteins were defined with FDR < 5% as a cutoff. Gene ontology analyses of the significant proteins were performed using DAVID (the database for annotation, visualization and integrated discovery, <https://david.ncifcrf.gov/>).

Statistical analysis

All experiments, with the exception of proteomics, were performed in triplicate. Four biological replicates were used in the proteomic study. Results of TEER, cytotoxicity and ROS generation studies were analyzed by one-way ANOVA with Dunnett’s multiple comparisons test.

Results

Detailed physicochemical characterization of TiO₂ (E171) was reported previously by the authors²⁷. Briefly, XRD patterns revealed an anatase crystal form, with a mean particle size by TEM of 113.4 ± 37.2 nm. Approximately 40% of particles had at least one dimension <100 nm, in agreement with what has been reported by others²⁶.

The surface charges (ζ -potentials), average particle diameters, and particle size distributions of the standardized food model (SFM) alone, of both SFM and fasting food model (FFM) with TiO₂, and of digestas of both food models with and without TiO₂, are summarized in Table 1 and Figure 2. Because undigested FFM alone is a simple salt solution with no particles its colloidal properties were not assessed. Moreover, it should be understood that the results of these colloidal analyses represent a heterogeneous mixture of particles,

including a variety of agglomerated food and digestive proteins, and lipid droplets in the case of SFM, any of which may or may not include TiO₂ particles, as well as separate TiO₂ particles and agglomerates, with biocoronas composed of various food and/or digestive macromolecules. The methods used for these analyses (laser Doppler electrophoresis for ζ -potentials and Multi-angle laser diffraction for size characterization) do not distinguish among these different types of particles.

Particles in undigested SFM, SFM + TiO₂, and FFM + TiO₂ were strongly negatively charged (ζ -potentials = -44.3 ± 1.7 , -44.0 ± 0.6 , and -40.5 ± 0.5 , respectively). Particles in the corresponding small intestinal digestas, as well as in digestas of FFM alone, were also negatively charged, but less strongly than those in undigested suspensions. The ζ -potentials of digestas of FFM, with or without TiO₂, were about half that of undigested FFM with TiO₂ (-20.9 ± 0.2 and -20.9 ± 0.7 , respectively), and ζ -potentials of SFM digestas, with or without TiO₂, were roughly three fourths those of the corresponding undigested suspensions (-27.6 ± 0.8 and -32.2 ± 0.9 , respectively).

The volume-weighted size distributions of SFM alone, and of SFM and FFM with TiO₂, prior to digestion (Figure 2A) all contained a dominant major peak at about 1 μm , and a smaller peak in the 0.2–0.5 μm range, with a left tail extending into the nano (<100 nm) range. Distributions of SFM, with or without TiO₂, unlike that of FFM with TiO₂, also featured broad peaks of larger agglomerates between 10 and 100 μm in diameter. The mean and cumulative percentile diameters from MALD analysis (Table 1) were consistent with the size distributions. The surface area-weighted mean, D[3,2], also known as the Sauter mean, was smallest in FFM + TiO₂ ($0.14 \pm 0.01 \mu\text{m}$), consistent with the lack of large agglomerates in the size distribution of that suspension, and somewhat larger in SFM and SFM + TiO₂ (0.46 ± 0.01 and $0.52 \pm 0.01 \mu\text{m}$). The volume-weighted size averages, denoted D[4,3], also known as the DeBroukere means, of these undigested suspensions were considerably larger than the corresponding Sauter means (0.60 ± 0.01 , 9.89 ± 0.05 , and $9.39 \pm 1.81 \mu\text{m}$ in FFM + TiO₂, SFM + TiO₂, and SFM alone, respectively), as is commonly the case with MALD analysis, but were also much larger in SFM suspensions (with or without TiO₂) than in FFM with TiO₂ (9.39 ± 1.81 in SFM and $9.89 \pm 0.05 \mu\text{m}$ in SFM + TiO₂ vs. $0.60 \pm 0.01 \mu\text{m}$ in FFM + TiO₂). The presence of large agglomerates in the SFM suspensions, and the absence of such agglomerates in the FFM + TiO₂ suspension, were most strongly reflected in the cumulative volume percentile size values (Table 1), denoted Dv10, Dv50, and Dv90, representing the 10th, 50th, and 90th size percentiles of the cumulative particle volume of a sample. Specifically, the smallest 90% of particles in FFM + TiO₂ had sizes equal to or smaller than $1.07 \pm 0.01 \mu\text{m}$ (Dv90), whereas particles in the corresponding 90% of particles in SFM and SFM + TiO₂ suspensions were as large as 29.9 ± 2.3 and $37.5 \pm 0.4 \mu\text{m}$, respectively.

The size distributions of digestas from SFM and FFM suspensions (Figure 2B) featured single dominant, broad peaks with prominent shoulders or adjoining minor peaks, and in the cases of digestas with TiO₂, small peaks in the millimeter range. The major peaks of TiO₂-containing digestas were substantially left-shifted relative to those of the corresponding food model only suspensions (from ~5 to ~1 μm for FFM, and from ~30 to ~2 μm for SFM suspensions), suggesting that the presence of TiO₂ somehow reduced or prevented

agglomeration of digestive proteins during digestion. However, digestas containing TiO₂ also contained small amounts of very large agglomerates represented by peaks at ~0.6 mm in FFM + TiO₂ and ~1.5 mm in SFM + TiO₂.

As with undigested suspensions, the mean and cumulative percentile diameters of digestas (Table 1) were, for the most part, consistent with their size distributions. The surface area-weighted Sauter means, D[3,2], were several fold larger than those of the undigested suspensions (e.g., 1.46 ± 0.05 vs. 0.14 ± 0.01 μm for FFM + TiO₂, and 2.29 ± 0.07 vs. 0.52 ± 0.01 μm for SFM + TiO₂), and about 3 to 4 fold larger in digestas without TiO₂ than in those with TiO₂ (e.g., 6.73 ± 0.02 μm for SFM alone vs. 2.29 ± 0.07 μm for SFM + TiO₂), consistent with the left shift of major peaks in the distributions of TiO₂-containing digestas noted above. The volume-weighted DeBroukere means, D[4,3], in contrast, were roughly 2 fold smaller in digestas without TiO₂ (e.g., 16.7 ± 1.7 μm for FFM alone vs. 35.8 ± 9.6 μm for FFM + TiO₂). This was somewhat puzzling, given the apparent strong left shift of most of the AUC of the distribution when TiO₂ was present. It is possible that the calculation of D[4,3] gave greater weight to larger particles, such as those in the small peaks at ~0.6 mm in FFM + TiO₂ and at ~1.5 mm in SFM + TiO₂, than their apparent weights in the volume distribution histograms.

***In vitro* toxicity of food-grade TiO₂ (E171)**

In vitro cellular toxicity was evaluated for small intestinal digestas of the food model controls, and of food models containing TiO₂ (E171) at concentrations of 0.75% and 1.5% w/w. Toxicity results are summarized in Figure 3 A, B and C for ENMs in fasting food model (FFM) and in Figure 3 D, E and F for ENMs in standardized food model (SFM).

No differences in TEER values were observed between untreated cells, food model controls, and food models containing TiO₂ (E171) at either concentration tested, indicating that cell layer and tight junction integrity were not affected by any of the treatments. Changes in ROS production after 6 hours of exposure to digestas (relative to cells cultured in fresh medium) are shown in Figure 3 B and E. The positive control (100 μM menadione) induced a 2.4-fold change in ROS production. The difference in ROS production between FFM alone and untreated cells was negligible. In contrast, SFM alone induced a slight but statistically significant increase in ROS production (1.24-fold, $p < 0.05$). Similar results were observed in cytotoxicity measured by LDH assay after 24 hours of exposure to digestas. FFM alone caused no increase in cytotoxicity, whereas SFM alone caused a slight although statistically insignificant increase in cytotoxicity (7.5%, $p > 0.05$).

At the lower concentration of 0.75% w/w in FFM, TiO₂ had no effect on ROS production, but produced a small, statistically significant increase (7%, $p < 0.01$) in cytotoxicity. At 1.5% w/w in FFM, TiO₂ caused a statistically significant increases over FFM control in both ROS production (1.24-fold, $p < 0.01$), and cytotoxicity (24%, $p < 0.001$). In contrast, TiO₂ in SFM at either dose caused no significant cytotoxicity, and produced only a slight increase in ROS production compared to SFM alone, suggesting that the effect of a more complex food matrix in this case was to eliminate the slight toxicity observed with TiO₂ in FFM..

Proteomic analysis of triculture exposed to TiO₂

The proteomic analysis resulted in the identification of 4944 proteins in total, in which 2271 proteins were quantified across all samples (Supplemental Data 1). Reproducibility analysis among replicates showed a less than 10% coefficient of variation (CV) for most proteins (Figure S1), demonstrating the robustness of the workflow. In order to visualize the overall quantitative patterns across all samples, an unsupervised clustering was used to analyze the proteomic data (Figure. S2). The result revealed two distinct sample groups corresponding to the two food models used in the study. Within each food matrix cluster, similar overall patterns of protein abundances were observed between the food model control (FFM alone and SFM alone) and TiO₂-treated samples, suggesting that TiO₂ within each food matrix induced a minimum impact on the cellular proteomes.

Impact of TiO₂ and food matrix on the cellular proteome

The two-factor experimental design made it possible to perform multiple comparisons to determine the effect of ENM treatment (TiO₂ vs control) and food matrix (control or TiO₂ treatment with two different food matrices). Statistical analyses of the proteomes were represented as volcano (scatter) plots, in which the change in protein abundance (log₂ fold change) and the statistical significance (-log adjusted p value) of each protein is depicted as a point (Figure 4 and Figure 5). No significantly altered proteins were identified in cells treated with digestas from FFM containing TiO₂ at either concentration (Figure 4A and B). A similar result was observed in digestas from TiO₂ in SFM, with only 28 significant proteins found at the low dose of TiO₂ treatment (Figure 4C, Supplemental Data 2), and no significant proteins detected at the high dose of TiO₂ (Figure 4D). Visualization of the proteome analyses shows that the log₂ fold change was narrowly distributed (majority of proteins falling between -0.5 and 0.5), with only a few proteins showing more than a 2-fold change (log₂ fold change > 1, Figure 4).

The impact of food matrix on cellular proteomes was evaluated by comparing samples with the same TiO₂ concentration (0% (control), 0.75% or 1.5%) but with different food matrices. In these comparisons, hundreds of proteins were identified as statistically significant with abundance differences (Figure 5, Supplemental Data 2). Significant differences were observed between the two food model controls (245 proteins). The number of differentially expressed proteins between FFM containing high dose of TiO₂ and SFM containing high dose of TiO₂ was somewhat greater (530 proteins) (Figure 5 A vs. 5 C). At the low TiO₂ dose, the number of differentially expressed proteins between FFM and SFM was considerably lower (85 proteins). Together, these data show that the food matrix significantly affects the proteome of the triculture system, whereas TiO₂ itself has relatively little impact, but may amplify or modulate the differences caused by the food model.

Food matrix-associated differential proteins and cellular processes

To identify cellular pathways and biological processes affected by the food matrix, differentially expressed proteins between the two food matrixes (alone and with TiO₂ at each dose) were subjected to gene ontology (GO) based enrichment analysis. Cellular processes identified, and the significance of their enrichment (represented as -log₁₀ p value), in each of the three comparisons were shown in Figure 6A, in which the significance of the

enrichment was indicated by the negative log p value. Several biological processes, including cell-cell adhesion, translation, and tricarboxylic acid cycle (TCA) were significantly enriched in all comparisons of the two food matrixes, suggesting a consistent food matrix effect on the cellular proteome, regardless of the presence of TiO₂. Most biological processes were less significantly enriched by food model in the presence of TiO₂ at lower dose, consistent with the relatively small number of significant proteins observed in this comparison.

In order to visualize food matrix-associated protein abundance changes in these biological processes, we compared the relative protein abundance between FFM and SFM for selected proteins (Figure 6 B). Higher abundances of all selected proteins were observed in the triculture system after treatment with digesta from SFM, relative to treatment with FFM (regardless of presence or concentration of TiO₂). A prominent example of a biological processes up-regulated by SFM was the TCA cycle, where core enzymes (SUCA, MDHC, SDHA, ACON, and IDHC) had significantly higher abundances in cells treated with SFM digestas than those treated with FFM digestas. Likewise, SFM caused upregulation in proteins involved in gluconeogenesis and protein synthesis (translation), as well as in mitochondrial proteins involved in oxidation-reduction processes, including MMSA (methylmalonate-semialdehyde dehydrogenase, mitochondrial) and DHE3 (glutamate dehydrogenase 1, mitochondrial). The upregulation of metabolic activities and oxidation-reduction processes by SFM relative to FFM suggests a higher level of energy metabolism and redox control after ingestion of a standardized complete diet relative to a fasting diet.

Discussion

The increasing use of engineered nanomaterials, such as TiO₂, in food raises concerns about potential health risks from various stakeholders²⁶. To accelerate the science based implementation of related regulations, a comprehensive understanding of the toxicity of ingested ENMs is required.

In this study, we assessed the cytotoxicity and proteomic effects of food-grade TiO₂ (E171), which contains a considerable portion of nano-scale particles (approximately 40% by number). In order to bring physiological relevance to this *in vitro* cellular study, two different food models (FFM and SFM) were evaluated, and a gastrointestinal tract (GIT) simulator was used to mimic oral, gastric, and small intestinal digestion. This integrated approach take into consideration interactions of TiO₂ with food matrix and potential property transformations across the GIT which are known modulators of bioactivity^{13,14,21,33}. The final small intestinal phase digesta was applied to a tri-culture small intestinal epithelium model to assess the cytotoxicity and impact on the cellular proteome of food-grade TiO₂ (E171) rather than using the ENM mixed with culture medium alone which is usually the practice in nanotoxicology research of ingested ENMs.

At starting concentrations of 0.75% and 1.5% w/w in FFM or SFM (final concentrations in the digesta-cell culture medium mixture of 150 µg/mL and 300 µg/mL), TiO₂ (E171) caused little or no cytotoxicity in the tri-culture small intestinal epithelium model. This contrasts somewhat with numerous reports of cell culture studies, which found that industrial grade

but not a food-grade, uniform 20 nm TiO₂ (P25), used for paintings, printing inks, and plastics, as well as other very small (5–10 nm) TiO₂ nanoparticles, cause considerable cytotoxicity in concentrations ranging from 50 µg/mL to 2 mg/mL in cell culture medium^{44–46}. The cytotoxicity was presumably from the extremely small size of TiO₂ used in these studies. However, in the case of food-grade TiO₂ (E171) with an average size around 110 nm, the low cytotoxicity reported here is consistent with previous reports. Chen et al. extracted TiO₂ particles from chewing gum to evaluate their cytotoxicity in gastrointestinal cells, and found that gum nano-TiO₂ particles caused little cytotoxicity after a 24 h exposure at 200 µg/mL in cell culture medium²⁵. Similarly, in Dorier et al.'s study with co-culture intestinal epithelial cells, in which cells were either continuously or repeatedly exposed to TiO₂ (E171) at 100 µg/mL in cell culture medium⁴⁷, no differences were observed in viability, tight junction or microvilli morphology, or mucus secretion, between treatment and control cells. However, a slight increase in intracellular ROS was detected (similar to the slight increase in ROS reported here for digested E171 in FFM, but not in SFM), and continuous exposure was found to induced significant oxidative damage to DNA. The aforementioned studies provide valuable information about the interaction between an *in vitro* model of small intestinal epithelium and pristine food-grade TiO₂ in cell culture medium, but do not account for the effects of interactions of the TiO₂ with the food matrix and GIT fluids and conditions during digestion, which may modify the oral toxicity of ingested ENMs.

Interestingly, SFM eliminated the slight toxicity of TiO₂ (E171) observed in FFM. It is well known that adsorption of proteins, lipids, and other molecules from the surroundings alter the surface characteristics of ENMs^{11,23,27,48}. The formation of a biological corona impacts the cellular and tissue responses to ENMs^{49–51}. For example, the binding of blood proteins to carbon nanotubes (CNTs) reduced their cytotoxic effect on human umbilical vein endothelial cells and acute monocytic leukemia cells by protecting them from direct contact with the bare surfaces of CNTs⁵². In the present study, TiO₂ nanoparticles may have adsorbed some of the constituents in SFM, thereby masking harmful surface atoms and groups, and otherwise altering surface characteristics and possibly cellular uptake resulting in reduced cytotoxicity. Moreover, the observed differences between the use of the two food models can also be a result of differential agglomeration and settling of formed agglomerates which contribute to differences in doses delivered to the cells. However, the delivered dose in digesta cannot be measured using our previous dosimetry model which was developed for nanoparticles administrated to cells in cell culture medium^{53–55}. The complexity of digestas make measurement of effective density of ingested ENMs impossible⁵⁶. More mechanistic studies are needed to understand the role of biological coronas and delivered dose in reducing the bioactivity in the presence of SFM model.

MS-based quantitative proteomics has become a robust tool that provides identification and quantification of thousands of proteins in biological samples. It has been increasingly used to study the biological impact of exposure to ENMs in biological systems, including TiO₂^{34–41}. These studies have shown that ENMs can alter many cellular processes, including metabolism, protein translation, response to ROS, cytoskeleton remodeling, and immune responses. Despite the wealth of information about potential biological effect of ENMs provided by these proteomic studies, the oral route of exposure and impact of the food

matrix on the ability of ENMs to trigger cellular responses has not been evaluated to the best of our knowledge. In addition, many of these investigations employed laborious and time-consuming gel-based proteomic approaches^{41,42}, in which low proteome coverage and poor quantitative reproducibility are limiting.

The proteomic approach employed in this study provides an unbiased quantitative analysis of the impact of TiO₂, as well as that of the food matrix, at the proteome-wide level. The similarity of the proteomes of TiO₂-treated and untreated samples, regardless of food model, validated our observation that food-grade TiO₂ in either food models causes little or no cytotoxicity. However, results from previous *in vitro* and *in vivo* studies of the impact of TiO₂ on the proteome are inconsistent^{34–36,38}. For example, Chen *et al.* found that the abundance of nearly 1000 proteins in mouse bone marrow derived macrophages were significantly changed after 24 h treatment with 100 µg/mL TiO₂ nanoparticles (10 nm, anatase) in cell culture media⁵⁷. In contrast, Triboulet *et al.* employed the same dose (100 µg/mL) of TiO₂ nanoparticles (25 nm, anatase) in mouse macrophages and observed relatively few changes in the proteome, as exemplified by the identification of only seven proteins whose abundance was significantly changed (>20%) by exposure to TiO₂ nanoparticles (in cell culture media)⁴¹. This discrepancy is likely due to differences in particle sizes, cell lines, and experimental schemes (i.e. different sizes and composition TiO₂ nanoparticles were used and their dissolution kinetics and transformations across the GIT were ignored²⁷).

It is notable that in the present study, food-grade TiO₂ (E171) with an average size of 110 nm was first dispersed in a food model, and the resulting mixture was then subjected to physiological digestion, before applied to cells. This approach recapitulates the interactions of TiO₂ with food and digestive tract secretions that would normally occur during digestion, and prior to contact with the small intestinal epithelium. These interactions could alter surface charges or oxidation states, or result in adsorption of food and digestive molecules, which could neutralize or mask reactive surface groups, and otherwise render the TiO₂ particles less toxic. Moreover, the tri-culture model used in this study represents a realistic hybrid model of the complete small intestinal epithelium.

Although the direct impact of TiO₂ on the proteome was negligible, more significant and substantial differences in protein abundance profiles were found between the FFM- and SFM-treated cells. The consistent expression patterns within each category of samples (FFM and SFM) and the differential regulation between the two categories clearly separated the FFM-proteomes and the SFM-proteomes. This confirms that the effect of TiO₂ itself on the proteome is minimal in both food models and reveals a more dominant effect of the food matrix on the small intestinal epithelial cellular proteome. Compared to FFM, SFM induced proteome changes suggesting increases in cellular activities involved in energy metabolism (TCA cycle, gluconeogenesis) and protein biosynthesis (translation). The TCA cycle is the metabolism hub, producing ATP, NADH and other metabolite intermediates through the oxidation of acetyl-CoA. Food availability would likely trigger the catabolism of carbohydrates, fats, and proteins, generating acetyl-CoA as the common substrate for the TCA cycle. The increased availability of acetyl-CoA might, through as yet unknown mechanisms, increase synthesis of TCA cycle enzymes. Indeed, multiple enzymes in the

TCA cycle were up-regulated in SFM samples. Such increased metabolic activity in mitochondria is also consistent with the increased expression of many metabolic enzymes involving in the oxidation-reduction process since many of these enzyme activities are redox-dependent⁵⁸. Increased mitochondrial energy metabolism in SFM digesta treated cells could also lead to higher levels of ROS generation, as mitochondria are one of the major sources of ROS production in mammalian cells^{59,60}. This is consistent with the finding of slightly higher levels of ROS with SFM than with FFM digesta treatments (Figure 3 E). Moderate levels of ROS can activate the antioxidant defense system, which is reflected in the up-regulation of the peroxidase GPX4 observed in SFM digesta-treated cells. It is notable that other proteins involved in the oxidation-reduction process, such as HIG1A, showed higher abundance in FFM than in SFM digesta-treated cells. HIG1A is a mitochondrial inner membrane protein that has been shown to play a protective role against oxidative stress by promoting mitochondrial homeostasis⁶¹. Thus, it is possible that higher ROS generation in the SFM digesta treated cells was quenched by the antioxidant system, not necessarily leading to oxidative stress. Instead, the nutrient-poor FFM could trigger slight cellular stress, which further activate an increase of HIG1A. Interestingly, SFM treatments also caused an up-regulation of cell-cell adhesion molecules, including two septins (SEPT7 and SEPT9), plasma membrane-associated cytoskeletal components required for cadherin-mediated cellular junctions⁶²⁻⁶⁴. Given that cell-cell adhesion molecules, such as cadherins and integrins, participate in intercellular adhesion of Caco2/HT-29^{65,66}, we speculate that this up-regulation of cell-cell adhesion proteins may promote the maintenance and integrity of these critical intercellular junctions.

In addition, a greater number of significantly different proteins were observed between the SFM- and FFM-treated cells in the presence of high dose of TiO₂ than without TiO₂ (Figure 5), suggesting that the interaction of TiO₂ and food models may affect the proteome. The actual cause of this effect is currently unknown and requires further studies. The effects of TiO₂ on digestion and absorption of some food components have been reported. For example, TiO₂ can decrease the bioaccessibility and bioavailability of several nutrients, including iron, zinc, fatty acids⁶⁷, and dietary flavonoids⁶⁸. TiO₂ can also disrupt the structure of the dietary protein, casein, resulting in a decrease in its hydrolysis in the stomach⁶⁹. On the other hand, TiO₂ can increase the toxicity and bioavailability of pesticide residues in foods²⁸. All these findings highlight the importance of considering the food matrix when studying the implications of ingested ENMs, not only because the food matrix may alter the cytotoxicity of the ENMs, but also because the ENMs may affect the cellular responses to the food matrix. Nevertheless, since no *in vitro* model can perfectly mimic the mechanical, biochemical, and biological conditions or cellular responses in the human gastrointestinal tract, future animal studies are required for a comprehensive assessment of the oral toxicity of the food additive TiO₂ (E171).

Conclusion

In this study, the cytotoxicity, and effects on the cellular proteome, of the food additive TiO₂ (E171) in a tri-culture small intestinal epithelium, were investigated. Food matrix effects on bioactivity were also assessed using a fasting food model (FFM), and a standardized food model (SFM) which is based on the average American diet. The results of these studies

demonstrated that the food additive TiO₂ (E171) is relatively non cytotoxic for the small intestinal epithelium under the experimental conditions tested. However, significant and substantial differences in protein abundance profiles were observed between the FFM digesta- and SFM digesta-treated cells, and a greater number of significantly different proteins were observed between the SFM digesta- and FFM digesta-treated cells in the presence of high dose of TiO₂ than without TiO₂. Additional studies, and corresponding *in vivo* studies, focused on the interactions of ingested TiO₂ ENMs with the food matrix and digestive processes and cells are necessary. Understanding the full implications of the increasing amounts of ENMs entering the food stream will provide regulators with useful data and facilitate development of safer-by-design food-grade ingested nanomaterials.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Highlights

- Food additive TiO₂ (E171) in fasting food model (FFM), but not in standardized food model (SFM), produced significant, dose-dependent cytotoxicity, and at the higher dose caused significant oxidative stress, indicative of a food matrix effect.
- No significant perturbations of the cellular proteome were observed with TiO₂ in either FFM or SFM food models.
- Proteins involved in the tricarboxylic acid (TCA) cycle, gluconeogenesis, and translation were up-regulated by digestas from SFM compared to those from FFM, also indicative of food matrix effects in the case of food models alone.
- The differences in profiles between the two food models was more pronounced in the presence of TiO₂ indicative of food matrix effects

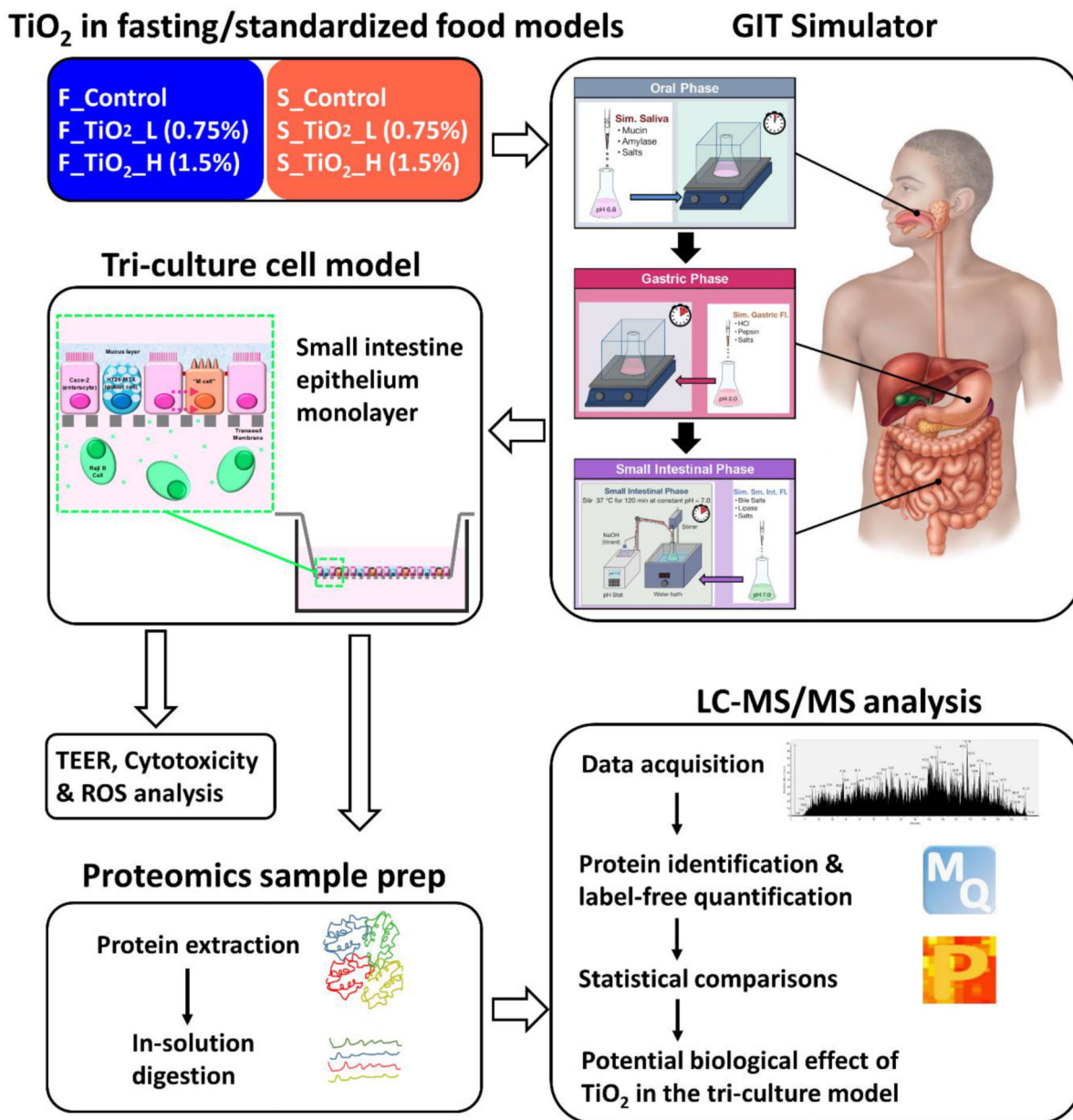


Figure 1. Schematic diagram of the study design.

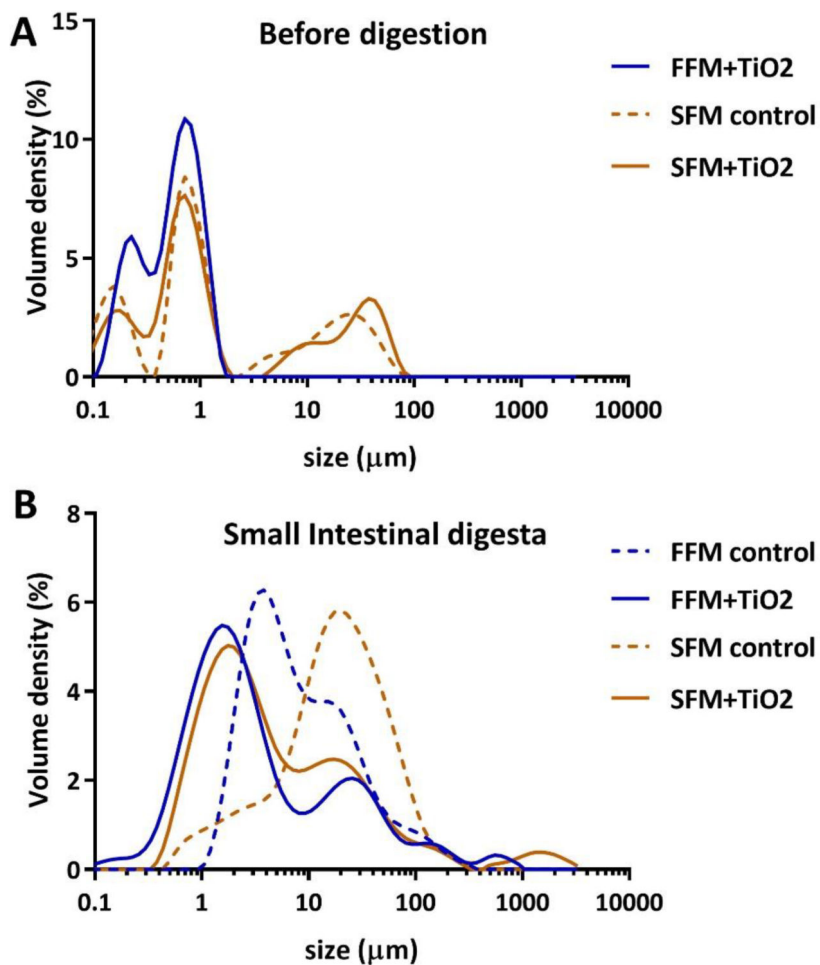


Figure 2. Size distributions of particles before digestion and in small intestinal phase digesta. (A) Volume density (%) of particle diameter distributions for TiO₂ in FFM (fasting food model, phosphate buffer), SFM (standardized food model) control, and TiO₂ in SFM. (B) Volume density (%) of particle diameter distributions for small intestinal phase digesta of FFM control, TiO₂ in FFM, SFM control, and TiO₂ in SFM.

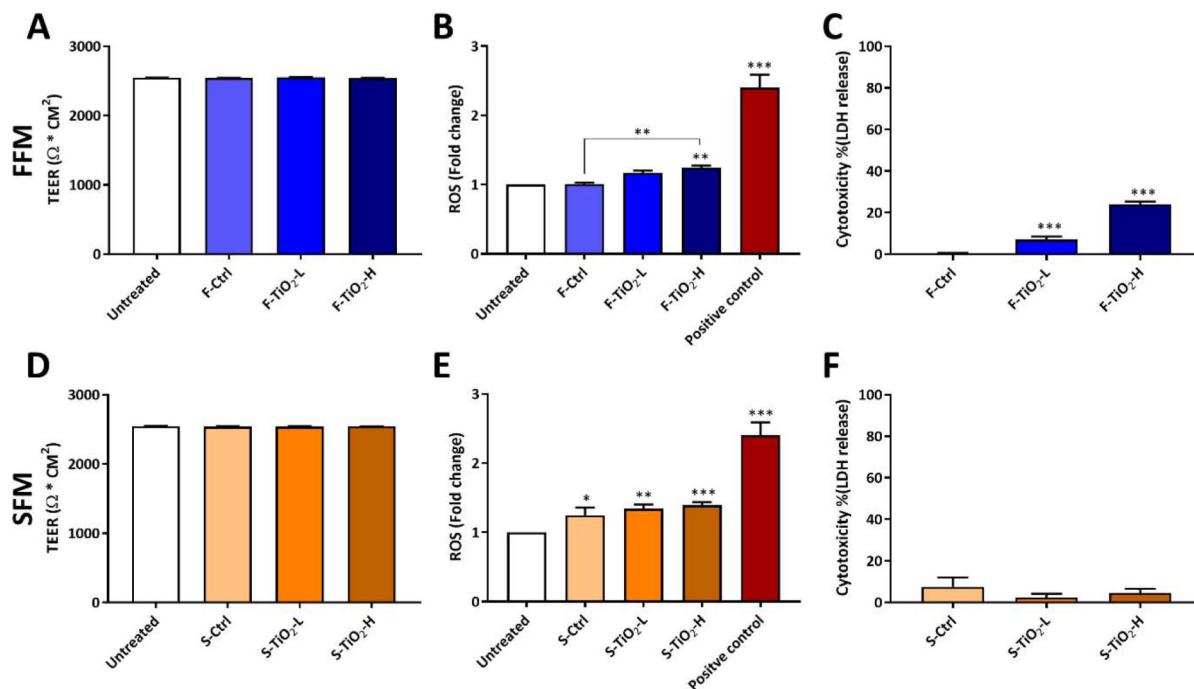


Figure 3.

In vitro toxicity of food-grade TiO₂ in fasting food model (A, B and C) and in standardized food model (D, E and F). TEER measurements after 24 h exposures to digesta from simulated digestion (A and D). ROS generation (fold change from untreated cells) following 6 h exposure to digesta from simulated digestion (B and E). Cytotoxicity (LDH assay) following 24 h exposure to digesta from simulated digestion (C and F). Error bars represent \pm SD, *= $p < 0.05$, **= $p < 0.01$, ***= $p < 0.001$.

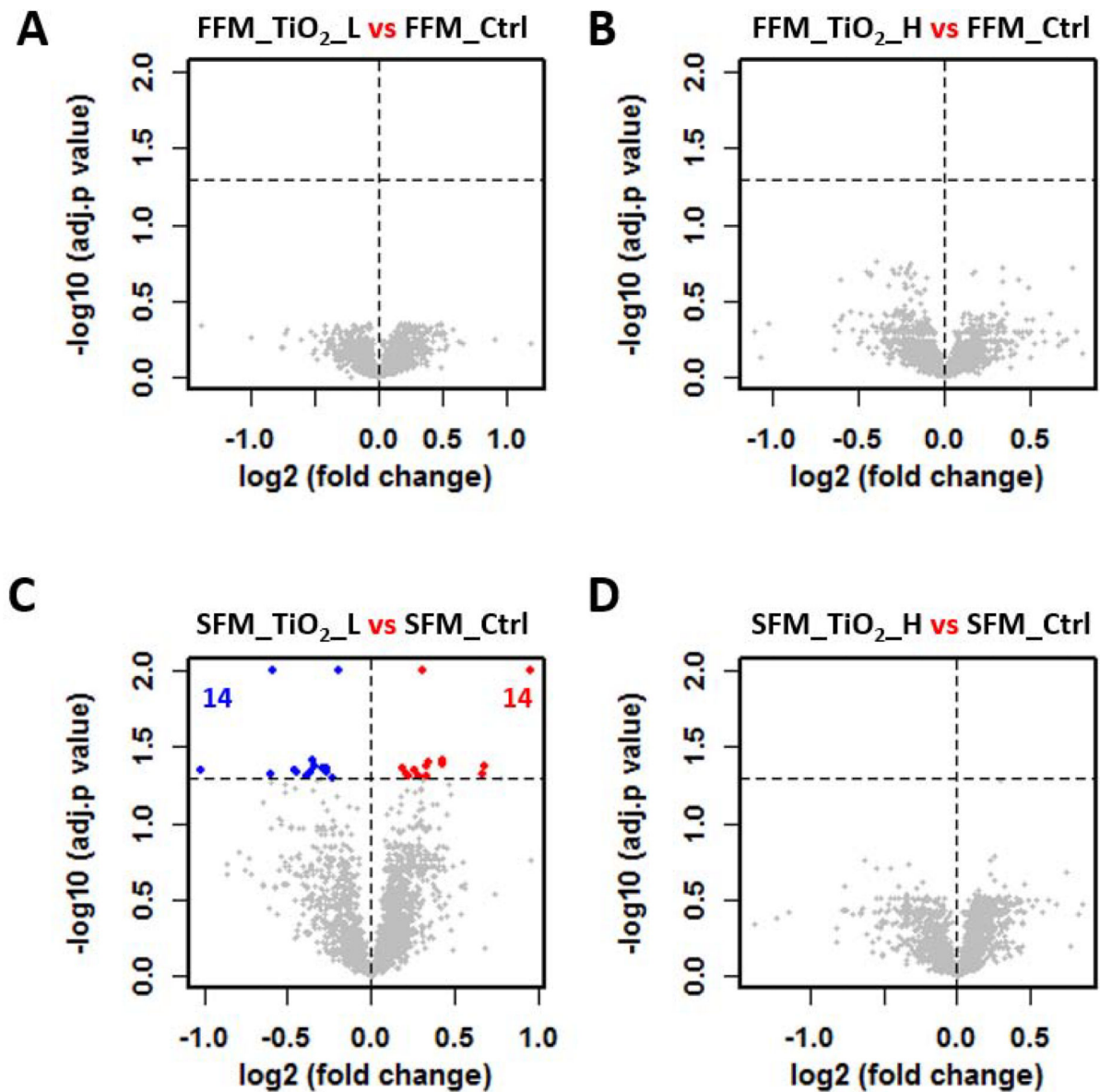


Figure 4.

TiO_2 induce little to no impact on the cellular proteome of the triculture system. Volcano plots were used to show the proteome difference involving four TiO_2 effect comparisons (A-D). Abbreviation scheme of comparing groups: food matrix (F for fasting and S for standard), followed by treatment (TiO₂_L and TiO₂_H for low and high dose of TiO_2 treatment, respectively, and Ctrl for the controls). x axis: the extent of fold change in protein intensity (in \log_2 scale); y axis: the level of statistical significance of the comparisons ($-\log$ adjusted p value). Red and blue dots represent significantly down- and up-regulated proteins, respectively. Grey dots represent non-significant regulated proteins. Black dotted horizontal line at $y = 1.3$, the cutoff to define significant proteins. The number of significantly down- and up-regulated proteins were shown in blue and red, respectively.

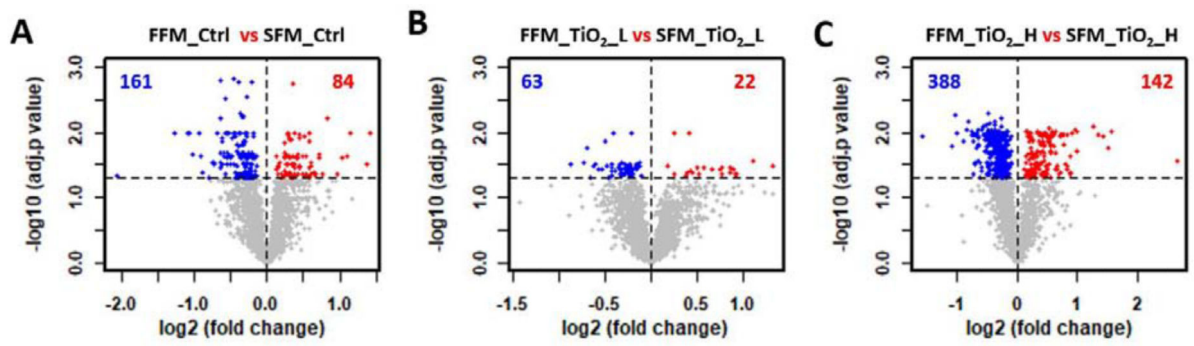


Figure 5. Differential proteomes associated with food matrices. Volcano plots were used to show the proteome difference involving three food matrix effect comparisons (A-C).

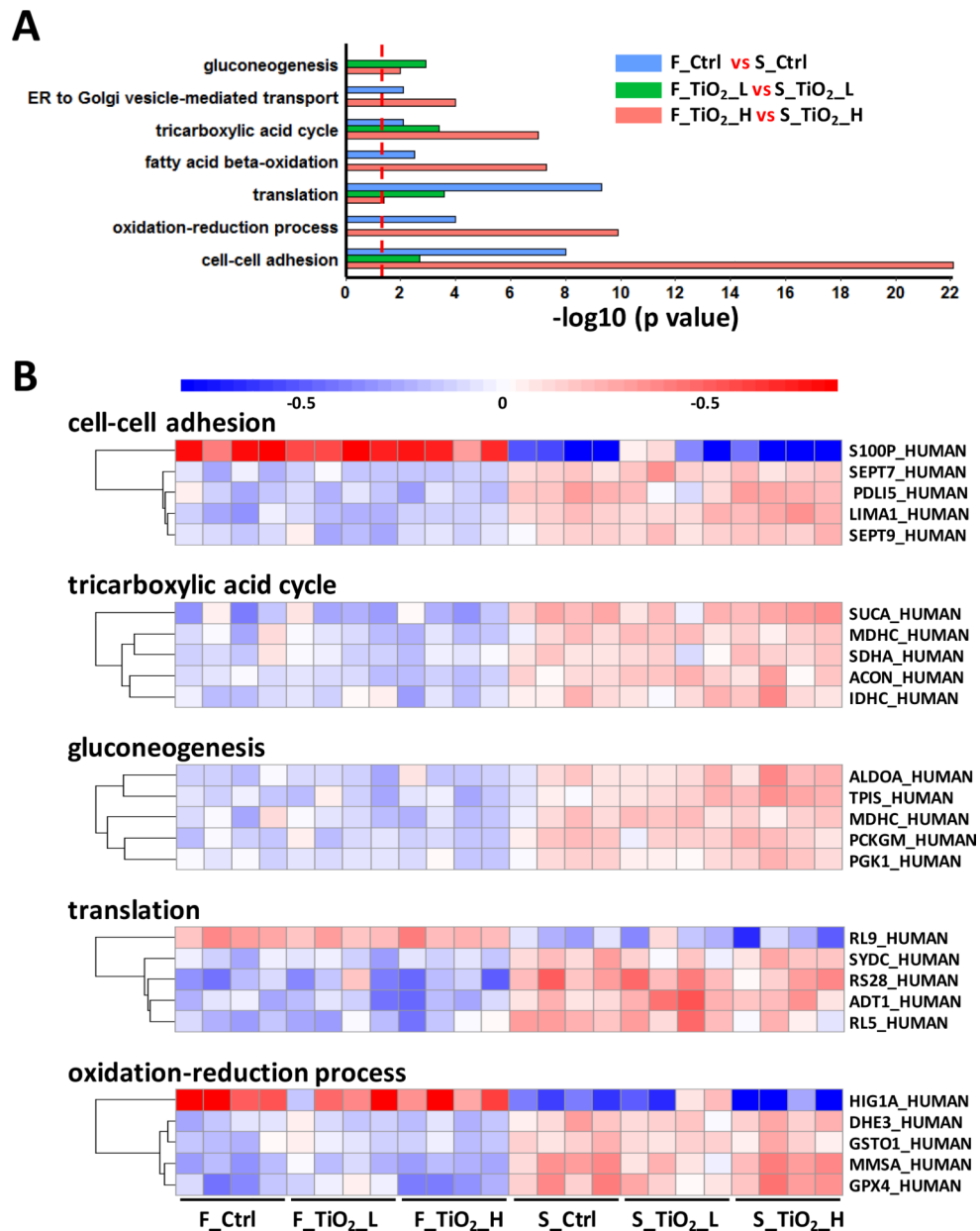


Figure 6. Pathways and proteins in response to different food matrices. (A) Biological processes enriched from three food matrix effect comparisons. Significant proteins from each comparison were subjected to gene ontology enrichment, and the figure illustrated biological processes ($p < 0.01$) found from at least two out of the three comparisons. (B) Protein abundance changes in selected biological processes. Significant proteins in each biological process were subjected to t test between the two food matrix groups, and proteins with top 5 smallest p values were selected for plotting.

Table 1.Colloidal characterization of TiO₂ (E171) before digestion and in small intestinal phase digesta.

	Sample	ζ (mV)	D[3,2] (μm)	D[4,3] (μm)	Dv10 (μm)	Dv50 (μm)	Dv90 (μm)
Before digestion	FFM control	-	-	-	-	-	-
	FFM+TiO ₂	-40.5 \pm 0.5	0.14 \pm 0.01	0.60 \pm 0.01	0.19 \pm 0.01	0.57 \pm 0.01	1.07 \pm 0.01
	SFM control	-44.3 \pm 1.7	0.46 \pm 0.01	9.39 \pm 1.81	0.145 \pm 0.01	0.87 \pm 0.01	29.9 \pm 2.3
	SFM+TiO ₂	-44.0 \pm 0.6	0.52 \pm 0.01	9.89 \pm 0.05	0.18 \pm 0.01	0.83 \pm 0.01	37.5 \pm 0.4
Small intestinal digesta	FFM control	-20.9 \pm 0.7	5.01 \pm 0.21	16.7 \pm 1.7	2.27 \pm 0.05	6.44 \pm 0.42	37.3 \pm 4.1
	FFM+TiO ₂	-20.9 \pm 0.2	1.46 \pm 0.05	35.8 \pm 9.6	0.66 \pm 0.01	2.32 \pm 0.01	42.5 \pm 2.4
	SFM control	-33.2 \pm 0.9	6.73 \pm 0.02	15.9 \pm 12.7	2.60 \pm 0.01	18.5 \pm 0.4	68.4 \pm 3.5
	SFM+TiO ₂	-27.6 \pm 0.8	2.29 \pm 0.07	42.2 \pm 29.8	0.91 \pm 0.01	3.44 \pm 0.20	43.9 \pm 10.6