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Activity-Based Sensing for Site-Specific Proteomic Analysis of Cysteine Oxidation

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CONSPECTUS:

Oxidative post-translational modifications (OxiPTMs) of cysteine residues are the molecular foundation of thiol-based redox regulation that modulates physiological events such as cell proliferation, differentiation, and migration and, when dysregulated, can lead to biomolecule damage and cell death. Common OxiPTMs of cysteine thiols (—SH) include reversible modifications such as S-sulfenylation (—SOH), S-glutathionylation (—SSG), disulfide formation (—SSR), S-nitrosylation (—SNO), and S-sulfhydration (—SSH) as well as more biologically stable modifications like S-sulfinylation (—SO₂H) and S-sulfonylation (—SO₃H). In the past decade, our laboratory has developed first-in-class chemistry-based tools and proteomic methods to advance the field of thiol-based redox biology and oxidative stress. In this Account, we take the reader through the historical aspects of probe development and application in our laboratory, highlighting key advances in our understanding of sulfur chemistry, in the test tube and in living systems.

Offering superior resolution, throughput, accuracy, and reproducibility, mass spectrometry (MS)based proteomics coupled to chemoselective "activity-based" small-molecule probes is the most rigorous technique for global mapping of cysteine OxiPTMs. Herein, we describe the evolution of this field from indirect detection to state-of-the-art site-centric quantitative chemoproteomic approaches that enable mapping of physiological and pathological changes in cysteine oxidation. These methods enable protein and site-level identification, mechanistic studies, mapping foldchanges, and modification stoichiometry. In particular, this Account focuses on activity-based methods for profiling S-sulfenylation, S-sulfinylation, and S-sulfhydration with an eye toward new reactions and methodologies developed in our group as well as their applications that have shed new light on fundamental processes of redox biology. Among several classes of sulfenic acid probes, dimedone-based C-nucleophiles possess superior chemical selectivity and compatibility with tandem MS. Cell-permeable dimedone derivatives with a bioconjugation handle are capable of detecting of S-sulfenylation in living cells. In-depth screening of a C-nucleophile library has yielded several entities with significantly enhanced reactivity over dimedone while maintaining selectivity, and reversible linear C-nucleophiles that enable controlled target release. C-Nucleophiles have also been implemented in tag-switch methods to detect S-sulfhydration. Most recently, activity-based detection of protein S-sulfinylation with electrophilic nitrogen species

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(ENS), such as C-nitroso compounds and electron deficient diazines, offers significant advantages in simplicity-of-use and target specificity compared to label-free methods.

When feasible, the rich information provided by site-centric quantitative proteomics should not be tainted by oxidation artifacts from cell lysis. Therefore, chemoselective probes that function in a native environment with low cytotoxicity, good cell-permeability, and competitive kinetics are desired in modern redox chemoproteomics approaches. As our understanding of sulfur chemistry and redox signaling evolves, newly discovered cysteine OxiPTMs in microorganisms, plants, cells, tissues, and disease models should innovatively promote mechanistic and therapeutic research.

Graphical Abstract



OXIDATIVE POST-TRANSLATIONAL MODIFICATION OF CYSTEINE

Post-translational modification (PTM) exponentially expands the chemical repertoire available to proteins beyond the 20 standard amino acids. The human proteome contains more than 200 types of PTMs that modulate protein function.¹ Among these, oxidative posttranslational modifications (OxiPTMs) of cysteine residues have emerged as a fundamental mechanism in thiol-based redox regulation and signaling in physiological processes such as cell proliferation, differentiation, and migration.² The sulfur atom of cysteine can exist in various oxidation states from -2 to +6. Two electron oxidation of cysteine thiol (—SH) by the reactive oxygen species (ROS) hydrogen peroxide (H₂O₂) produces cysteine sulfenic acid (-SOH), which may be stabilized by the protein microenvironment or form a disulfide with an adjacent cysteine residue (-SSR) or mixed disulfide with glutathione (-SSG). Likewise, S-nitrosothiols (-SNO) and persulfides (-SSH) are generated from reactive nitrogen/sulfur species. The aforementioned cysteine OxiPTMs can be biologically reduced back to their thiol form with cellular enzymes like glutaredoxin (Grx) or thioredoxin (Trx), as a key mechanism in maintaining redox homeostasis.³ On the other hand, higher oxidation states of cysteine, like cysteine sulfinic acid (-SO₂H) and sulfonic acid (-SO₃H), are more stable and can accumulate over time during oxidative stress.⁴ The distinct reactivity of each cysteine OxiPTM provides the chemical basis for differential redox regulation of protein function. Cysteine oxidation also has a profound impact on PTMs that require a reduced thiol for modifications such as palmitoylation and drug pharmacology with covalent thiol-reactive inhibitors.^{5,6} Protein cysteines can exhibit a range of reactivity, dictated by

factors such as pK_a , electrostatic interactions, and solvent exposure.⁷ The reaction rate between cysteine thiolates and H₂O₂ can vary substantially: antioxidant enzyme peroxiredoxins (Prxs) are hyperactive ($10^5-10^8 \text{ M}^{-1}\cdot\text{s}^{-1}$), followed by redox-sensor proteins like GAPDH and OxyR ($10^3-10^5 \text{ M}^{-1}\cdot\text{s}^{-1}$), while the bulk of the cysteinome reacts slowly (~ $10 \text{ M}^{-1}\cdot\text{s}^{-1}$).^{8,9}

A wealth of cysteine OxiPTMs in microorganisms, plants, animal tissues, and disease models await discovery. To this end, proteome-wide profiling of cysteine oxidation is an essential molecular technique to elucidate redox-regulated signaling pathways and offer new insights into pathological states involving oxidative stress, such as cancer, diabetes, and neurodegenerative and cardiovascular diseases.¹⁰ The rapidly growing area of chemical proteomics or chemoproteomics has been fueled by advances in instrumentation and bioinformatics as well as small-molecule probes that enable tagging, isolation, identification, and quantification.¹¹⁻¹³ In the past decade, researchers have developed MS-based techniques to identify low pK_a or "reactive cysteines" as well as cysteine OxiPTMs with detailed information including the site, type, dynamic fold-change, and extent of modification. Herein, we spotlight recent developments in profiling cysteine reactivity and oxidation with emphasis on site-centric quantitative approaches.

ORTHOGONAL REACTIVITY

Pioneered by Bertozzi and co-workers, bioorthogonal chemistry enables researchers to investigate native biomolecules under complex biological settings.¹⁴ Understanding the reactivity profile of each cysteine "chemotype" is key to designing an effective activitybased proteomic workflow. To achieve orthogonality, two types of approaches are commonly employed: (1) blocking undesired forms of cysteine (e.g., thiols), reducing cysteine OxiPTMs, followed by trapping nascent thiols with detectable tags ("tag-switch" method); and (2) probing distinct classes of cysteine OxiPTMs with reagents that are chemically selective. In the experimental design, all biologically relevant forms of cysteine must be taken into consideration. Summarized in Table 1, cysteine OxiPTMs can be categorized on the basis of sulfur charge density. Electronrich species or nucleophiles (when deprotonated) include cysteine thiols (—SH), persulfides (—SSH), and sulfinic acids (—SO₂H). Electrondeficient species or electrophiles include disulfides (—SSR), S-glutathione adducts (—SSG), and S-nitrosothiols (—SNO). Some species, like sulfenic acids (—SOH), are unique in that they exhibit both nucleophilic and electrophilic reactivity.

SITE-CENTRIC QUANTITATIVE PROTEOMICS

Reversible or labile cysteine OxiPTMs are extremely susceptible to perturbation during sample preparation and subsequent analyses. Therefore, chemical labeling is widely recognized in redox proteomics, because oxidation states are "trapped" by chemical tags, which may also enable affinity enrichment for sensitivity improvement. Stable isotope-coded tags are often used in quantitative proteomics, since corresponding peptides have identical ionization ability, and therefore can serve as internal MS standards.¹⁵ In addition, tagged peptides must be stable and have a clear fragmentation pattern on tandem MS to reveal modified sites. Such site-centric quantitative approaches not only show the location

Indirect Profiling of Cysteine Oxidation

With estimated concentration up to 50 mM, protein-bound cysteine thiols are mostly reduced and constitute the majority (up to 70%) of reduced thiols in cells.^{16,17} Iodoacetamide (IAM) and its derivatives are among the most frequently used thiol-reactive alkylating agents, but at the high concentrations typically employed they are plagued by off-target modification of basic amino acid residues (e.g., N-terminus, lysine and histidine residue)^{18,19} and other cysteine OxiPTMs including —SOH, —SO₂H, and —SSH.²⁰⁻²² Several novel reagents with enhanced selectivity, like methylsulfonyl benzothiazole (MSBT),²³ do not react with —SOH, —SO₂H, and —SNO (but thiol alkylation generates sulfinic acids, which may cross react with other species, including —SNO)²⁴ (Figure 1a). However, alkylation of persulfides (—SSH) is expected, due to its enhanced nucleophilicity originated from low p K_a (CysSSH p $K_a = 4.3$; CysSH p $K_a = 8.3$).²⁵ In fact, MSBT and IAM derivatives have been used in detection of —SSH (Figure 1b).^{26,27}

Reversibly oxidized cysteine modifications can be indirectly detected by thiol-blocking, reduction of oxidized species, followed by labeling of nascent thiols (Figure 2a). DTT and TCEP (standard reduction potential = -0.33 and -0.32 V, respectively)²⁸ are powerful reducing agents that are capable of converting reversible cysteine OxiPTMs to thiols, but they do not effectively reduce —SO₂H or —SO₃H. Alternatively, a given OxiPTM may be selectively reduced and tagged for detection. Sodium ascorbate and sodium arsenite were considered as selective reducing agents for —SNO and —SOH, respectively.^{29,30} Likewise, enzymatic reduction of —SSG by Grxs has been employed to detect S-glutathionylation.³¹ Nevertheless, extreme caution is advised when interpreting data obtained from these strategies, as incomplete blocking, reduction, or sample degradation can occur during lengthy sample processing workflows, especially for labile modifications. In addition, clear examples of cross-reactivity with other cysteine OxiPTMs indicate that ascorbate and arsenite are far less selective than previously believed.^{9,29}

In early studies, cysteine OxiPTMs were typically profiled with isotope-coded affinity tag (ICAT)-based platforms;^{32,33} however, many recent improvements to this workflow have been made (Figure 2b). Cysteine-reactive isobaric tags have been introduced for multiplexed redox quantitation, reducing run-to-run variance. For example, 6-plex tandem mass tag (TMT) reagents have been used to quantify —SNO and other reversible cysteine OxiPTMs in three competitive samples.^{34,35}

The application of smaller electrophiles has expanded the coverage of targetable cysteines due to easier access to sterically encumbered sites. Alkynyl iodoacetamide probes IA-alkyne and IPM can be conjugated to isotope-coded reporter groups via copper-catalyzed azide– alkyne cycloaddition (CuAAC) or "click chemistry". With the IPM probe, over 6500 cysteines in multiple human cell lines were profiled, and most of them responded to H_2O_2 in a cell-specific manner.³⁶ The Weerapana group has developed cost-effective isotope-coded electrophiles and applied these reagents to identify almost 1000 reversibly oxidized cysteine sites in HeLa lysate.³⁷ Moreover, the inherent cytotoxicity of IA-alkyne (LC₅₀ = 16 μ M) can

be mitigated, at least in part, by using photocaged electrophiles. Epidermal growth factor (EGF)-induced cysteine oxidation in A431 cells has been analyzed using the photocaged probe, CBK1, which displayed little detectable cytotoxicity below 250 μ M and allowed a degree of temporal control over cysteine labeling.³⁸

In combination with other methodologies, researchers aim to integrate quantitative information on relative cysteine oxidation, termed fold-change, with total protein levels. For example, the SILAC-iodoTMT (metabolic labeling), GELSILOX (¹⁸O proteolytic labeling), cysTMTRAQ (coupled isobaric labeling), and oxSWATH (integrative chemical labeling)³⁹⁻⁴² platforms have been recently reported to comprehensively evaluate the effect of oxidation on cysteine proteomes.

The Chemical Odyssey of Sulfenic Acid (—SOH) Detection

A chemical probe is a small-molecule reagent that allows mechanistic and phenotypic studies on its protein targets.⁴³ In the context of redox proteomics, the chemical probe is expected to target a specific class of cysteine OxiPTM among the whole proteome, with minimal cross-reactivity. Because of its dynamic nature and central role in OxiPTM-based redox regulation, the odyssey of trapping cysteine sulfenic acids (—SOH) with nucleophilic compounds began in 1974, when Allison and co-workers introduced the cyclic 1,3-diketone, dimedone, and other nucleophiles that reacted with cysteine sulfenic acid in GAPDH.⁴⁴ Thereafter, nucleophiles functionalized with biotin, azide, or alkyne reporter groups and other handles for bioorthogonal ligation based upon the dimedone structure have been reported.⁹ Our laboratory has performed a comprehensive study, wherein a library of ~100 carbon-based nucleophiles (C-nucleophiles) were designed, prepared, and screened, ultimately identifying several entities with exceptional reactivity (up to 3170 M⁻¹·s⁻¹, a more than 300-fold increase over dimedone) while maintaining selectivity.₄₅₋₄₇Furthermore, our survey of linear C-nucleophiles showed reversible covalent labeling of sulfenic acids for the first time, enabling target release under defined conditions (Figure 1c).⁴⁶

Complementary to nucleophilic probes, several strained alkynes and alkene electrophilic reagents have been engineered to capture —SOH.⁴⁸⁻⁵¹ Because these reactions are strain-promoted and the alkenes/alkynes are not as electron-deficient as Michael acceptors, thiol–ene/yne side reactions can be suppressed, yet incompletely.^{49,52} The issue of thiol–ene/yne side reaction is magnified given the much higher abundance of —SH than —SOH in cells. Moreover, —SSH was found to participate in the reaction with strained alkynes under a mechanism analogous to that of —SOH.⁵³

Regardless of the chemistry or probe used for trapping, sulfenic acid models are essential touchstones in reactivity and selectivity studies (Figure 3). A small-molecule anthraquinone-1-sulfenic acid (Fries' acid) is stabilized via intramolecular hydrogen bonding, but not "bio-representative".⁹ In comparison, protein-stabilized sulfenic acids are typically preferred choices, including C165S AhpC and C64,82S Gpx3.^{45,48} On the other hand, certain electrostatic or steric effects can clearly modulate the rate of probe reactivity with protein —SOH. Therefore, an alternative model developed by our group consists of a dipeptide cyclic sulfenamide, which is stable in dry, powdered form and organic solvent, but readily hydrolyzes to form sulfenic acid in aqueous solution; this model has been employed

with great success to benchmark our extensive library of —SOH probes and reactive fragments.⁴⁵⁻⁴⁷ Thermal-driven sulfoxide elimination reactions also produce sulfenic acids. ^{9,54} Recently, we have reported an approach for the generation of small-molecule and protein sulfenic acids from photocaged sulfoxide precursors with applications in unnatural amino acid (UAA) incorporation.⁵⁵

The Selectivity of Dimedone-Based Probes for Sulfenic Acid (-SOH)

Dimedone-based C-nucleophile probes for —SOH do not possess intrinsic reactivity with biological nucleophiles, such as the thiol in cysteine, hydroxyl in serine, amino in lysine, or cysteine sulfinate (—SO₂⁻).⁴⁵ They are also inert toward weak electrophiles, such as the disulfide of a protected cysteine (Z-Cys-OH)₂, oxidized glutathione (GSSG), or S-nitrosoglutathione (GSNO) (Figure 4a).^{45,56} No reaction between dimedone-based C-nucleophiles and protein disulfides (such as those in Trx, Gpx3, and Prx I) has been observed either. It has been noted that the activated methylene in dimedone can undergo Knoevenagel-type condensation with nonprotein bound aldehydes, including pyridoxal, acrolein, and glyceraldehyde.⁵⁷ From a chemoproteomic standpoint, however, this reaction is a red herring. Indeed, the rate constant for the reaction of dimedone with glyceraldehyde ($0.06 \text{ M}^{-1} \cdot \text{s}^{-1}$) is approximately 200-fold less than that of —SOH (~10 M⁻¹ \cdot \text{s}^{-1}). Also, no reaction was observed between butyraldehyde (a representative model of an aldehyde-modified protein) and dimedone at physiological pH and temperatures up to 100 °C (Gupta, unpublished results).

Another electrophilic species that could conceivably react with dimedone-based probes is the cyclic sulfenamide, which exists in equilibrium with the corresponding sulfenic acid (Figure 4c).⁴⁵ Protein tyrosine phosphatase 1B (PTP1B) is the first and best-characterized example of this cysteine OxiPTM, whose crystal structure showed a cyclic sulfenamide linkage in its active site.⁵⁸ The nitrogen atom on an amide ($pK_a \sim 17$) is less nucleophilic than the nitrogen of an amine, due to the resonance stabilization afforded by the amide carbonyl group. However, in the case of PTP1B, the pK_a of the Ser216 amide nitrogen is decreased via extensive hydrogen bonding within the phosphate-binding loop (P-loop), which enables nucleophilic attack on the neighboring sulfur atom of the Cys215-SOH precursor. Although quite interesting, this phenomenon is rare and there are few wellcharacterized examples.^{58,59} Additionally, dimedone does not react with PTP1B in its cyclic sulfenamide state.⁶⁰ Due to the high energy barrier for cyclization, cyclic sulfenamide formation is quite slow.^{54,61} On this basis, after —SOH is formed, sulfenamidation would be outcompeted by nucleophilic attack of a thiol or, if present, dimedone-based probes (Figure 4b), in contrast to some recent assertions.⁶² Similarly, it is energetically unfavorable for weakly electrophilic species such as -SSR, -SSG, -SSH, and -SNO to form a cyclic sulfenamide and subsequently be tagged with dimedone-based probes.⁶³

Other studies have suggested that dimedone labeled polysulfides (— S_nH) or polythiosulfenic acid (— S_nOH), but they were performed in vitro under nonphysiological conditions, with 0.2–0.3 mM Na₂S₂.^{64,65} Such adducts have never been reported in any proteome-wide experiments, most likely due to their exceedingly rare existence in cells.

Given consideration to all aspects above, —SOH is the only meaningful protein target of C-nucleophile probes in biological systems.

Profiling Protein Sulfenic Acids (—SOH)

Compared to indirect thiol reactivity-based approaches that involve blocking, reduction, and reprobing steps, chemoselective reactions allow direct tagging of a specific cysteine chemotype, providing precise molecular information about the OxiPTM identity and minimizing artifacts stemming from incomplete blocking and reduction, which can be significant in cell lysates and lengthy workflows. In order to capture dynamic cysteine OxiPTMs in living systems, successful activity-based probes should be both selective and kinetically competitive, while maintaining high stability and membrane permeability as well as low cytotoxicity.

As discussed above, dimedone-based nucleophilic probes show remarkable chemoselectivity with -SOH. We have also demonstrated that dimedone and related probes are well suited for tandem MS.⁶⁶ The first dimedone-based probes, reported by the Poole and King groups, were directly conjugated to biotin or fluorescent tags for visualization.⁹ However, these tools were incompatible with MS and suffered from poor cell permeability. To address this issue, our lab developed azide-functionalized dimedone derivatives, termed DAz-1 and DAz-2, that are compatible with bioconjugation techniques, including Staudinger ligation (Figure 5a). Although both probes are small in size, DAz-2 exhibited better reactivity and membrane permeability due to the replacement of the amide linkage at C-6 with a hydrocarbon linker at C-4. Using DAz-2, we reported the first first survey of S-sulfenylation in living cells, identifying both established redox-sensitive targets and more than 175 new candidates.⁶⁷ As a continuation of developing -SOH probes for in situ detection, we replaced the azide functional group on our probes with an alkynyl group, since alkyne reporters offer superior sensitivity and stability in vivo.⁶⁸ At the same time, we switched from Staudinger ligation chemistry to the more efficient CuAAC reaction. Subsequently, two platforms, termed SulfenM and SulfenQ, were developed in close collaboration with the Liebler group. These methods featured an alkyne-containing probe, DYn-2, to probe -SOH modifications in human cancer RKO and A431 cells (Figure 5d, e).⁶⁹ Both strategies enabled precise mapping of sulfenic acid modifications to individual cysteines in complex proteomes. SulfenM was employed to map more than 1000 S-sulfenylation sites from 700 proteins in RKO cells. SulfenQ featured a heavy isotopomer, DYn-2-d6, to quantify the fold-change in -SOH between samples (Figure 5b). Using the SulfenQ workflow, cells were treated with stimulus (H₂O₂ or EGF) or vehicle control in the presence of heavy or light DYn-2. Treatment of RKOs with exogenous H₂O₂ led to detectable changes in S-sulfenylation (heavy/light > 2) in >89% of 360 identified —SOH sites; similar observations were made with H₂O₂ treatment of A431 cells. By contrast, fewer than half of -SOH sites were modulated by stimulation of A431 cells with EGF, demonstrating that growth factor-driven cysteine S-sulfenylation was more target-selective.

Although DYn-2 is an effective probe, its moderate reactivity $(10 \text{ M}^{-1} \cdot \text{s}^{-1})$ poses some inherent limitations for quantitative detection, particularly —SOH with short cellular lifetimes. To address the demand for kinetically superior probes, we made structural

modifications to the dimedone 1,3-diketone scaffold, swapping one or both ketones to other electron withdrawing groups, such as amides, esters, sulfones, sulfonamides, and nitro or cyano groups on a linear or cyclic skeleton.⁴⁵⁻⁴⁷ Four representative cyclic nucleophiles equipped with alkyne handles (PD, PYD, PRD, BTD) showed diverse reactivity with 2–170-fold increased rates relative to DYn-2 (Figure 5c). These five probes were applied in the SulfenM platform, resulting in the discovery of 1283 sulfenylated sites on 761 proteins.⁷⁰ The most efficient probe, BTD, was also applied in a site-centric quantitative proteomics platform for global detection of —SOH in cell lysates and living cells. The platform is a modified version of SulfenQ, where paired samples are distinguished by isotope-coded UV-cleavable azido-biotin (Az-UV-biotin) instead of an isotope-coded —SOH probe (Figure 5f). Due to the enhanced reactivity of BTD relative to DYn-2, the input of lysate material was reduced more than 10-fold (e.g., from 30 mg of protein for DYn-2 to 2 mg of protein for BTD) while increasing the coverage of —SOH sites (1867 sites for BTD versus 1105 sites for DYn-2) in RKO cells.⁷¹

In addition to quantifying relative fold-changes in cysteine oxidation, another important goal is to measure the stoichiometry of —SOH modification. By installing an iodo group on dimedone at C-2, we created a new thiol-reactive alkylating agent, yielding the same reaction product as sulfenic acid labeling by dimedone. Isotope-coded dimedone and iododimedone (ICDID) reagents were subsequently prepared such that the two reagents were separable by 6 Da, which enabled absolute quantification of the fraction —SOH at a given cysteine (Figure 6a).⁷² In one proof-of-concept study, oxidation of Cys149 and Cys244 of GAPDH were profiled simultaneously. Importantly, only redox-sensitive Cys149 exhibited an increase in the fraction of sulfenic acid modification in response to H₂O₂. In an elegant modification of this chemical strategy, Guengerich and co-workers prepared BTD and iodo-BTD derivatives ${}^{13}C_6d_7$ -pBTD, d_7 -ipBTD, and d_0 -ipBTD to quantify sulfenic acids, free thiols, and reversible OxiPTMs in complex proteomes (Figure 6b).⁷³ A relatively low number (~600) of sulfenylated peptides can be attributed to the lack of an enrichment handle.

Although other chemical probes for sulfenic acids have been reported, few of them have actually been used to globally quantify sulfenomes using MS platforms. Fox and co-workers recently reported the SAM-TCO probe, a strained *trans*-cyclooctene with an axial hydroxyl group proposed to facilitate the reaction via an intramolecular attack. Additionally, a bioorthogonal tetrazine-alkene reaction was employed to quench excess SAM-TCO, which may help to minimize artifacts that can result from cell lysis.⁵¹

To date, site-specific mapping of protein S-sulfenylation in the aforementioned chemoproteomic studies has illuminated many novel thiol-based regulatory mechanisms. For example, application of DYn-2 in A431 cells revealed S-sulfenylation of epidermal growth factor receptor (EGFR) at a critical cysteine adjacent to its active site, Cys797. Further biochemical and cellular studies in our laboratory demonstrated that S-sulfenylation of this site augments EGFR tyrosine kinase activity, thus creating a positive feedback loop with the enzymatic source of EGF-induced generation of H₂O₂ and NADPH oxidase (Nox).^{6,74} In another example, we demonstrated that S-sulfenylation of SIRT6 Cys18 leads to intermolecular disulfide formation with the transcription factor, HIF1A, disclosing a new

mechanism for regulating the expression of multiple glycolytic genes.⁶⁹ A recent study of Ssulfenylation with BTD in *Arabidopsis thaliana* mapped more than 1000 protein targets including AtMAPK4, an orthologue to human MAPK1, which is a known redox switch.⁷⁵ Finally, application of BTD in mouse livers revealed the redox control of circadian pacemaker CLOCK via a reversibly oxidized cysteine.⁷⁶

In addition to the discovery of new mechanisms of thiol-based redox regulation, quantitative site-specific mapping of S-sulfenylation has applications in drug design. Profiling with structurally and chemically distinct cyclic C-nucleophiles (DYn-2, PD, PYD, PRD, and BTD) identified > 1280 S-sulfenylated cysteines present in "druggable" proteins and orphan targets, revealing both disparate reactivity profiles and target preferences.⁷⁰ Among the unique ligand–protein interactions, we identified a planar pyrrolidinedione nucleophile, PYD, that reacted preferentially with protein tyrosine phosphatases (PTPs). This study shows that fragment-based covalent ligand discovery with C-nucleophiles can be applied to generate an expansive view of the ligandable "redoxome" with significant implications for covalent inhibitor pharmacology.

Profiling Protein Sulfinic Acids (—SO₂H)

Protein S-sulfinylation was traditionally detected by antibodies or label-free MS. These methods suffer from limited affinity and specificity, or lengthy preparation steps. On the other hand, electrophilic nitrogen species (ENS) only form stable adducts with —SO₂H and can therefore serve as chemoselective probes for activity-based detection of S-sulfinylation (Figure 1d). Our group has designed two ENS-based —SO₂H probes: one compound is a C-nitroso benzoic ester named NO-Bio and the other probe, DiaAlk, is based upon an electron deficient diazine scaffold.⁷⁷⁻⁷⁹ The latter has superior sensitivity and compatibility with MS, facilitating site-centric quantitative analyses of protein sulfinic acids (Figure 7a). Biotinylated S-nitrosothiols have been used to detect —SO₂H,²⁴ but side reactions generate an aggressive electrophilic oxidant nitroxyl (HNO) that converts thiols to sulfinic acids.⁷⁹

Most recently, in close collaboration with the Yang laboratory, we have developed a sitecentric proteomic approach to profile cysteine sulfinic acids. Application of DiaAlk uncovered 387 —SO₂H sites from native lysates of A549 and HeLa cells, hundreds of which were previously unknown. Interestingly, S-sulfinylation levels under acute H₂O₂ exposure remained relatively static at the majority of sites, in contrast to the dynamic changes observed in the S-sulfenylome, suggesting that sulfinic acid modification likely takes place on a different time scale, with fewer biological pathways for repair. To investigate these observations in greater detail, we probed S-sulfinylome repair in a cysteine sulfinic acid reductase sulfiredoxin (SRX) knockout model. Comparative study of $Srx^{+/+}$ and $Srx^{-/-}$ mouse embryonic fibroblasts (MEFs) revealed more than 50 potential substrates of SRX, with subsequent biochemical validation of several new targets (Figure 7b). Our findings expand the substrate scope of SRX significantly beyond Prx isoforms 1–4, with new implications for the role of SRX in oxidative stress-associated diseases and drug development programs.⁷⁹

Profiling Protein Persulfides (—SSH)

Protein persulfidation (S-sulfhydration) generates persulfides (-SSH) that were detectable by alkylating agents. Although there are limited reports from small-molecule models, possibly as a result of low yield from product decomposition,⁸⁰ our lab has convincingly demonstrated that a variety of persulfide models, including protein persulfide Gpx3-SSH. readily react with most thiol-reactive agents.²² However, they are often accompanied by issues in selectivity. An alkylation/tag-switch approach was introduced as a reliable method to detect persulfides, where the disulfide products of -SSH alkylation are substituted by a nucleophilic probe and alkylated -SH are irreversibly modified. For example, thiols and persulfides were alkylated with MSBt-A, and only the latter subsequently reacted with a cyanoacetate nucleophile with a biotin reporter (Figure 8a).²⁶ Recently, the Filipovic group used NBD-Cl and a dimedone derivative to detect protein persulfides. An evolutionaryconserved correlation between aging and a decay in persulfidation levels was observed, suggesting a protective role of —SSH against irreversible cysteine oxidation (Figure 8b).⁸¹ A direct method to simultaneously profile thiol and persulfide proteomes has also been reported.²⁷ In this method, termed low-pH quantitative thiol reactivity profiling (QTRP), the alkylation by IPM is performed at pH 5.0, because persulfides (pK_a of Cys—SSH = 4.3) are expected to maintain high reactivity, whereas more abundant free thiols (pK_a of Cys—SH = 8.3) are mostly protonated and less reactive (Figure 8c).

CONCLUSIONS AND OUTLOOK

LC-MS platforms featuring chemoselective probes (Table 2) offer compelling advantages on resolution, throughput, accuracy, and reproducibility. At the same time, our evolving understanding of cysteine OxiPTMs presents new challenges in detection. For example, reagents believed to be thiol-selective were found to react with —SOH, —SO₂H, and —SSH species, which are of increasing biological relevance. Certain cysteine OxiPTMs exhibit dual reactivity (e.g., —SOH is both nucleophilic and electrophilic), diverse reactivity (e.g., —SSR), or similar reactivity to other OxiPTMs (e.g., —SH and —SSH). Therefore, it is critical to conduct a comprehensive survey of cross-reactivity while keeping the biological relevance and abundance of cysteine OxiPTMs in mind. More importantly, caution is advised against extrapolating the reactivity of an entire cysteine chemotype on the basis of studies with compounds that are not biorepresentative, as such models which may not represent the true reaction profiles as in complex proteomes.

In order to profile cysteine OxiPTMs with high fidelity, many efforts have been made in recent years including but not limited to the following: (1) Direct "activity-based" sensing of OxiPTMs with chemoselective probes to minimize various sources of artifacts generated by tag-switch methods. (2) In situ detection with smaller, cell permeable, nontoxic probes, which makes discovery in live cells or tissues possible. (3) Temporal control of labeling with probes that can be switched on/off. This decreases the cytotoxicity of the probes and can reduce labeling after cell lysis. (4) Site-centric profiling, which is significantly more stringent than protein level discovery and provides mechanistic information. (5) Multiplex labeling that allows multiple samples or modifications to be analyzed simultaneously, reducing the cost and run-to-run variations.

Novel discovery of proteomic cysteine oxidation fuels the future development in the fields of redox signaling and disease mechanism. Moreover, complementary to covalent inhibitors that target free cysteine thiols, redox-based covalent inhibitors targeting cysteine OxiPTMs may be developed. For example, we have shown that sulfenylation of EGFR not only enhances its activity but also masks its reactivity toward thiol-targeting covalent inhibitors like afatinib.^{6,74} Proteome-wide ligand discovery on cysteine OxiPTMs is ongoing and is expected to reveal additional targets. Another interesting direction is organelle-specific profiling of cysteine oxidation. As one of the major source of ROS, mitochondria are an intriguing host of cysteine OxiPTMs. Along these lines, the Furdui and King laboratories have reported -SOH probes conjugated with mitochondria targeting vectors, such as fluorescent coumarin and rhodamine moieties, or triphenylphosphonium groups, which serve as proof-of-concept for localized profiling of OxiPTMs.^{82,83} Last but not least, direct probing of protein disulfides remains an extremely challenging but vital mission. To date, protein disulfides are detected by indirect tag-switch methods, hampering our ability to profile disulfide reactivity and assign disulfide pairs. Although a sophisticated MS-based technique to characterize disulfide linkages in IgG2 λ antibodies has been reported.⁸⁴ going forward, a reaction-based approach to map disulfides at the proteome level is still in great demand. This is an active area of research in our group, as we currently explore several promising chemical strategies; these findings are forthcoming and will be reported in due course.

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Biography

Yunlong Shi received his B.S. in Chemistry from Zhejiang University, where he conducted research on organometallic chemistry and bioengineering. He earned his Ph.D. degree in Organic Chemistry at North Carolina State University in the laboratory of Prof. Joshua G. Pierce, where he focused on the synthesis and medicinal chemistry of natural products. He is currently a postdoctoral associate in the laboratory of Prof. Kate S. Carroll, working on chemical biology and designing novel probes for proteomics and therapy.

Kate S. Carroll is an Associate Professor with tenure in the Department of Chemistry at The Scripps Research Institute in Jupiter, Florida. She received her B.A. degree in Biochemistry from Mills College in 1996 and Ph.D. in Biochemistry from Stanford University in 2003. Her postdoctoral work was completed at the University of California, Berkeley, where she was a Damon Runyon-Walter Winchell Chancer Fund Fellow with Prof. Carolyn Bertozzi. She was an Assistant Professor at the University of Michigan until 2010, when she joined the Chemistry faculty at Scripps. Her research interests span the disciplines of chemistry and biology with an emphasis on studies of sulfur biochemistry pertinent to human disease. Her lab focuses on the development of chemical probes to study redox modifications of cysteine thiols, profiling changes in protein oxidation associated with disease, and exploiting this information for development of diagnostic and therapeutic approaches.

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Figure 1.

Common chemical reagents that selectively target cysteine oxoforms. (a) Reagents that label free cysteine thiols. (b) Thiol-reactive reagents also label persulfides. (c) Sulfenic acid probes. (d) Sulfinic acid probes.

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Figure 2.

(a) Indirect detection of cysteine OxiPTMs with thiol-reactive reagents. (b) Recent advances in thiol-reactivity profiling.

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Figure 3.

Small-molecule and protein models used in developing sulfenic acid probes. (a) Stabilized small-molecule and protein sulfenic acids. (b) Hydrolytic equilibrium of cyclic sulfenamide.(c) Generation of sulfenic acids via sulfoxide elimination. (d) Photocaged sulfenic acid.



Figure 4.

Energy profiles of cysteine OxiPTMs. Graphs (a,b) are drawn qualitatively based on reactivity of a Cys-Val dipeptide representing an unperturbed cysteine residue. (a) Dimedone-based nucleophiles readily label —SOH but do not react with disulfides, indicating $E_1 < E_2$. (b) Cyclic sulfenamide (—SNR') rapidly hydrolyzes to form —SOH, indicating $E_1 < E_3$. (c) Representative rate constants of chemotype transformations of the dipeptide model compound. Data was extracted from ref 45.



Figure 5.

(a) Structures of DAz sulfenic acid probes. (b) Structures of DYn probes. (c) Structures of dimedone-based probes with enhanced reactivity. (d) General workflow of the SulfenM strategy. (e) General workflow of the SulfenQ strategy. (f) Modified workflow of the SulfenQ strategy with the BTD probe.



Figure 6.

(a) Basic principle of the ICDID strategy of sulfenic acid quantification. (b) BTD-derived probes used in the quantification of cysteine oxidation.

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Figure 7.

(a) DiaAlk reacts with sulfinic acids followed by hydrolysis to form a stable MS-compatible product. (b) Discovery of SRX substrates by a chemical proteomics approach. With a recovery period, putative SRX substrates showed a lower recovery/control (blue/red) ratio in wild type cells, and a higher ratio in SRX knockout cells.



Figure 8.

(a) Persulfide detection by MSBt-A alkylation and cyanoacetate tag-switch. (b) Persulfide detection by NBD-Cl alkylation and dimedone derivative tag-switch. (c) Direct thiol and persulfide activity profiling with the low-pH QTRP workflow.

Table 1.

Orthogonal Reactivity Chart of Common Cysteine Chemotypes

ОП	no	partial
no	no	partial
arsenite, c dimedone	yes	yes
ascorbate, c etc. d	no	no
no	ou	ou
glutaredoxin	no	no
ENS ^e	no	no
	no arsenite, ^c dimedone ascorbate, ^c etc. ^d no glutaredoxin ENS ^e	no no arsenite, c dimedone yes ascorbate, c etc. d no no no glutaredoxin no ENS no

 $^{\ensuremath{\mathcal{C}}}$ Arsenite and as corbate are not fully selective.

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 $d_{\rm Several}$ chemical probes for —SNO are not discussed here.

 $^{
m \ell}$ Electrophilic nitrogen species also reacts with thiols, but only forms stable products with sulfinic acids.

Table 2.

List of Chemical Probes Discussed in This Account

name of probe	targets	notes	refs
DYn-2, BTD	—SOH	BTD is ~170× faster than DYn-2	69-71,74
ICDID, iTROC reagents	—SH, —SOH	for stoichiometric analysis	72,73
BCN, Norb-bio, SAM-TCO	—SOH	may react with Cys-SH, -SSH	48-51
NO-bio, DiaAlk	—SO ₂ H	DiaAlk is MS-compatible	77-79
IPM	—SH, —SSH	operates at low pH	27,36
MSBT, NBD-Cl	—SH, —SSH	tag-switch with C-nucleophiles	26,81