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Chalcones, semicarbazones and pyrazolines as inhibitors of cathepsins B, H and L



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ABSTRACT

Cathepsin B [EC 3.4.22.1], cathepsin H [EC 3.4.22.16] and cathepsin L [EC 3.4.22.15] are the most versatile lysosomal cysteine proteases and are responsible for intracellular protein degradation. These are involved in a number of pathological conditions including tissue degenerative processes. In the present work, we report the synthesis and systematic evaluation of differently substituted chalcones, chalconesemicarbazones, and diarylpyrazolines on cathepsins B, H and L activity. It was found that after a preliminary screening as cysteine protease inhibitors, chalconesemicarbazones were better inhibitors to these cysteine proteases than diarylpyrazolines followed by chalcones. All the synthesized compounds were identified as the best inhibitors to cathepsin L followed by cathepsin B and then cathepsin H. The results are compared with docking studies and it was found that all the compounds resulted in decrease in energy while interacting with the active site of the enzyme.

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1. Introduction

Semicarbazones are important bioactive compounds possessing different biological activities such as anticonvulsant [1], antimicrobial [2], antiviral [3], antitumoral [4], antihypertensive [5] and antitubercular [6]. These compounds have also been found to be potent inhibitors to cruzain, cysteine protease of *Trypanosoma cruzi* and been shown to be trypanocidal [7]. The trypanocidal activity is related to cruzain inhibiting tendencies of the semicarbazones. Falcipain, another protozoal cysteine protease has also been reported to be inhibited by variety of semicarbazones. Therefore, semicarbazone inhibitors of cysteine proteases have potential use for prevention and treatment of protozoan infections such as trypanosomiasis, malaria and leishmaniasis. It has also been reported that semicarbazones of aryl and alkyl carbonyl compounds inhibit cysteine proteases of parasites more as compared to mammalian proteases and therefore indicate the possibility of their therapeutic use. The compounds also find use in inhibiting cysteine proteases associated with carcinogenesis, including cathepsins B, H and L [8].

Chalcones, another class of biologically active molecules, known to possess antimalarial [9], anticancer [10], antiprotozoal [11], anti-inflammatory [12], antibacterial [13], antioxidant [14], antifungal [15] activities are also reported to inhibit various enzymes like tyrosinase, alpha-amylase and beta-lectamase [16–18]. These,

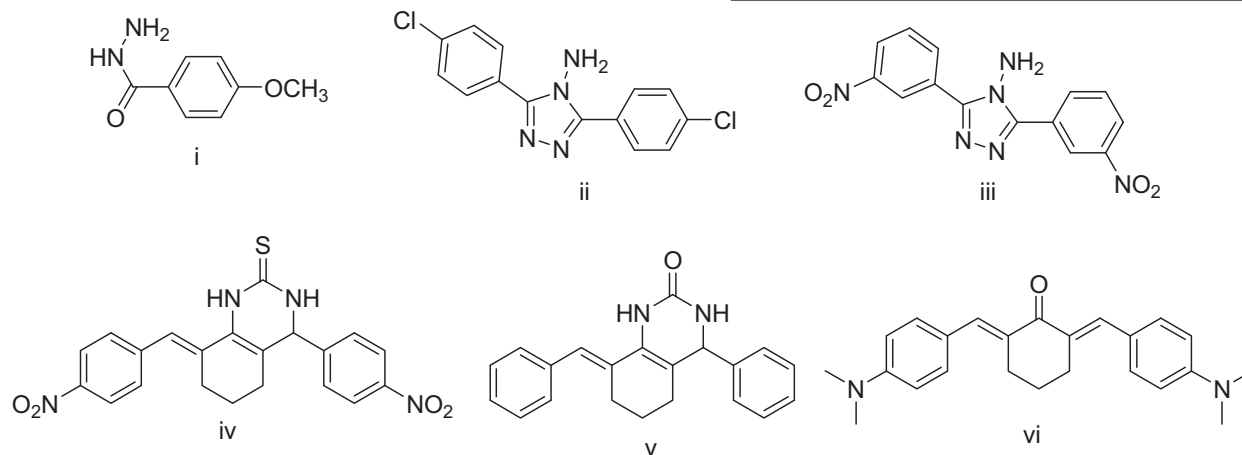
α,β -unsaturated compounds have been found to interact with serum albumin [19–24], the protein responsible for the transportation of various molecules including drugs. In the present work we have synthesized chalcones semicarbazones by combining α,β -unsaturated carbonyl- and semicarbazone-pharmacophores, which resulted in formation of two isomers not reported earlier and have evaluated their effect on three important lysosomal cysteine proteases, cathepsin B [EC 3.4.22.1], cathepsin H [EC 3.4.22.16] and cathepsin L [EC 3.4.22.15]. Cysteine proteases are key factors in the pathogenesis of cancer invasion, arthritis, osteoporosis, and microbial infections [25]. Cathepsin B possesses exo as well as endopeptidase activity [26] and is also capable of peptidyl-dipeptidase [27,28] and carboxypeptidase activities [29,30]. The enzyme has been found to be involved in various pathological conditions such as chronic inflammatory diseases of airways and joints, cancer and pancreatitis [31], atherosclerosis [32], rheumatoid arthritis [33] Alzheimer's disease [34], etc. Cathepsin H an endo and amino-peptidase is also involved in many diseased conditions including breast carcinoma [35], melanoma and tumour metastasis [36], head and neck carcinoma [37], malignant prostate cancer [38]. Similarly, cathepsin L activity has also been reported to be involved in diseases such as osteoarthritis [39,40], tumorigenesis [41–43], Ebola haemorrhagic fever, severe acute respiratory syndrome and Leishmaniasis [44–46]. Targeting these enzymes is therefore one of the strategies in the development of new chemotherapeutic agents for a number of diseases including tissue degenerative disorders.

We are working in the identification of small molecular weight compounds as inhibitors to endogenous proteolytic

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activities [47–52]. In quest for the identification of some novel inhibitors to cathepsin B and cathepsin H, we have recently reported various non-peptidyl inhibitors such as bischalcones based quinazoline-2(1H)-ones, quinazoline-2(1H)-thiones (iv-vi) [53] and acyl hydrazides, triazoles (i-iii) [54].



Effects of hydrazones [55], hydroxyl chalcones [56] and their cyclized derivatives, formyl and benzoyl pyrazolines [57] have also been established on these enzymes. To explore further in this direction the present work is focused on chalcones, chalconesemicarbazones and their cyclized derivatives as inhibitors to cathepsins B, H and L. The intention to synthesize semicarbazones of chalcones was undertaken keeping in view that the semicarbazones possessing azomethine group in conjunction with α,β -unsaturation will have extended pharmacological activities. The compounds were further cyclized to pyrazolines which are also reported to possess diverse biological activities such as antimicrobial [58], antiameobic [59], anti-inflammatory [60], anticancer [61], antidepressant [62] and antitubercular [63] activities. Effect of synthesized compounds was observed on cathepsins B, H and L to establish a structure activity relationship between the 1,3-diphenylprop-2-en-1-ones, substituted chalconesemicarbazones and substituted 3,5-diphenyl-2-pyrazoline-1-carboxamide derivatives. Structure activity relationship (SAR) studies of open chain and cyclized derivatives can result in identification of potential inhibitors among related class of compounds [55,56]. A comparative account of different moieties introduced in the synthesized compounds on cathepsin B, H and L, three lysosomal cysteine proteases inhibiting tendency has also been evaluated. The studies can lead to the development of new inhibitors of these enzymes.

2. Experimental

2.1. Materials

All the chemicals were of analytical grade. Fast Garnet GBC (*o*-aminoazotoluene diazonium salt), α -N-benzoyl-D,L-arginine-2-naphthylamide (BANA), Leu- β NA and ZPheArg-4m β NA were purchased from Bachern Feinchemikalien AG, Switzerland. The protein sample was concentrated using Amicon stirred cells with YM 10 membrane under nitrogen pressure of 4–5 psi. The source of enzyme was fresh goat liver obtained from local slaughter house.

2.2. Methods

Melting points were taken in open capillaries and are uncorrected. The progress of the reactions was monitored on silica gel G plates using iodine vapor as visualizing agent. Elisa plate reader was used for measuring absorbance in the visible range. The spectrophuge

was used for centrifugation purpose. IR spectra were recorded on Horizon 300 MHz spectrometer. NMR spectra were recorded on Bruker 300 MHz instrument. The chemical shifts are expressed in ppm units from an internal TMS standard. All commercially available reagents were used as-received.

2.3. Synthesis

The title compounds were synthesized according to Scheme 1.

2.3.1. Synthesis of substituted 1,3-diphenylprop-2-en-1-ones (1a–1j)

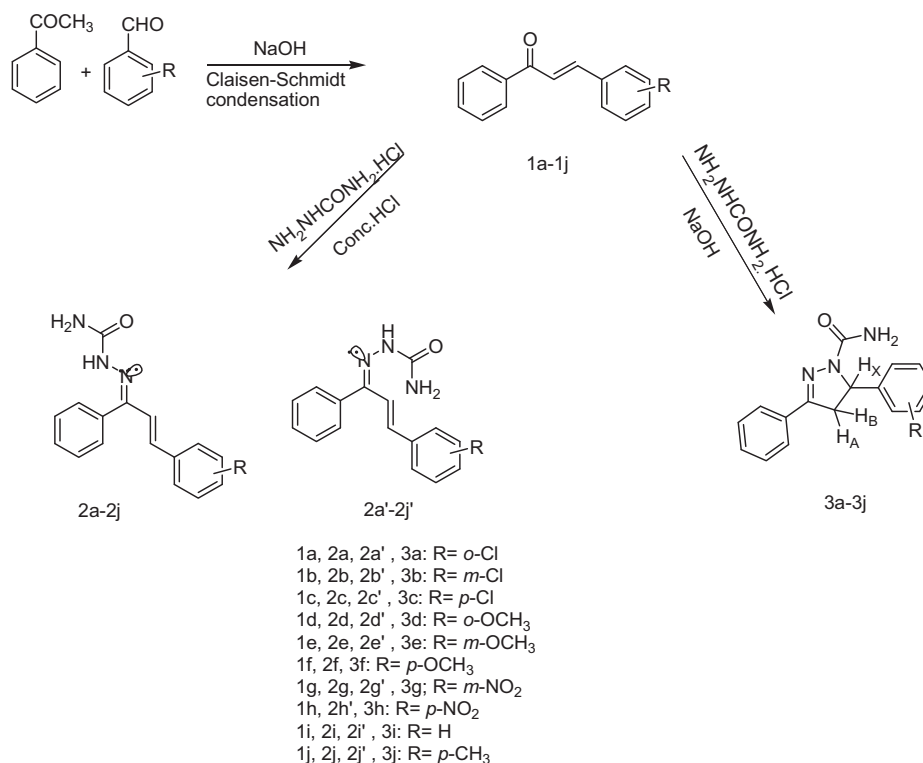
Acetophenone (0.01 mol) was added to equimolar quantity of substituted benzaldehyde (0.01 mol) dissolved in methanol (25 ml). To this solution equimolar NaOH pellets (0.01 mol) were added at once and the reaction mixture was stirred for 40 min at room temperature. Excess of methanol was again added and then again stirred for next 40 min at room temperature. It was then diluted with cold water. The product was separated, filtered and washed with water until neutral. The resulting chalcones were purified by recrystallization with methanol. The structures were characterized using melting point compared with literature melting point [19,64], IR and NMR.

2.3.2. Synthesis of substituted chalconesemicarbazones (2a–2j, 2a'–2j')

To an aqueous semicarbazide hydrochloride (0.02 mol), sodium acetate (0.03 mol) was dissolved. This solution was added to the ethanolic solution of substituted 1,3-diphenyl-2-en-1-ones (0.01 mol) (1a–1j). Then, few drops of concentrated hydrochloric acid were added. Reaction mixture was stirred at 40 °C and was monitored on TLC. After completion, the reaction mixture was poured into ice and precipitate, so obtained was filtered, washed with water, dried and recrystallized with ethanol. The product thus obtained was found to possess two stereoisomeric forms (Fig. 1) which were separated on column. The two isomers have been characterized using ¹H NMR.

2.3.3. Synthesis of substituted 3,5-diphenyl-2-pyrazoline-1-carboxamide derivatives (3a–3j)

To an aqueous sodium hydroxide (0.017 mol), semicarbazide hydrochloride (0.0085 mol) was dissolved. This solution was added to the solution of substituted 1,3-diphenyl-2-en-1-ones (0.005 mol) (1a–1j) in ethanol (25 ml) and then, refluxed for 6–8 h. TLC monitoring was extensively done. The solvent was evaporated and the solid so obtained was filtered, washed with water to neutral reaction, dried and recrystallized from ethanol. The structure was characterized using melting point compared



Scheme 1. Synthesis of substituted chalcones (1a–1j), chalconesemicarbazones (2a–2j, 2a'–2j') and pyrazolines (3a–3j).

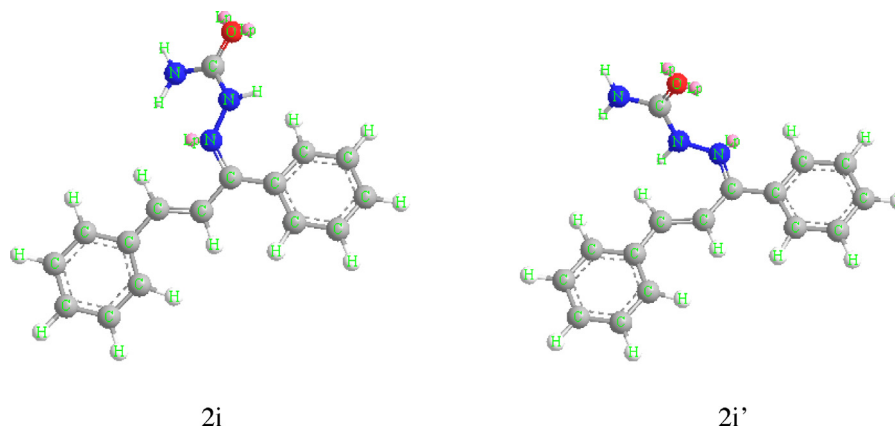


Fig. 1. 3-D structures (<http://media.cambridgesoft.com/cbou130/cbou1302.exe>) of two isomers 2i and 2i' from chalconesemicarbazones showing shielding and deshielding of C-H proton with lone pair of nitrogen, respectively.

with literature melting point [65], IR and NMR spectroscopic techniques.

2.4. Pharmacology

2.4.1. Purification of cathepsins B, H and L

All the purification steps were carried out at 4 °C. Cathepsins B, H and L were extracted and purified from goat liver by the established procedure reported previously [53] including the steps of acetone powder preparation, homogenization, acid autolysis, 30–70% $(\text{NH}_4)_2\text{SO}_4$ fractionation, molecular sieve chromatography on sephadex G-100 and ion exchange chromatographies on CM Sephadex C-50 and DEAE A-50 sephadex. The specific activities of the cathepsin B, H and L were equal to ~11.28 nmol/min/mg, ~24.01 nmol/min/mg and ~16.78 nmol/min/mg, respectively.

2.4.2. Autoproteolytic studies

The experiments were conducted in triplicates as described previously for endogenous proteolytic studies [47–52].

2.4.3. Enzyme inhibition studies

Cathepsin B and L activity was determined using BANA [66] and ZPheArg-4mBNA [67] substrate at pH 6.0, respectively whereas cathepsin H activity was determined using Leu-βNA [68] substrate at pH 7.0. Effect of synthesized compounds was observed on the activities of cathepsins B, H and L at 1×10^{-5} M, 1×10^{-4} M and 1×10^{-8} M final concentration of each compound, respectively. First of all, enzyme was equilibrated in buffer of appropriate pH at 37 °C. Then 20 μl of individual compound was added in the reaction mixture separately to effect the final concentration of each compound as 0.1 mM concentration. After an incubation time of 30 min residual enzyme activity was estimated by the usual enzyme

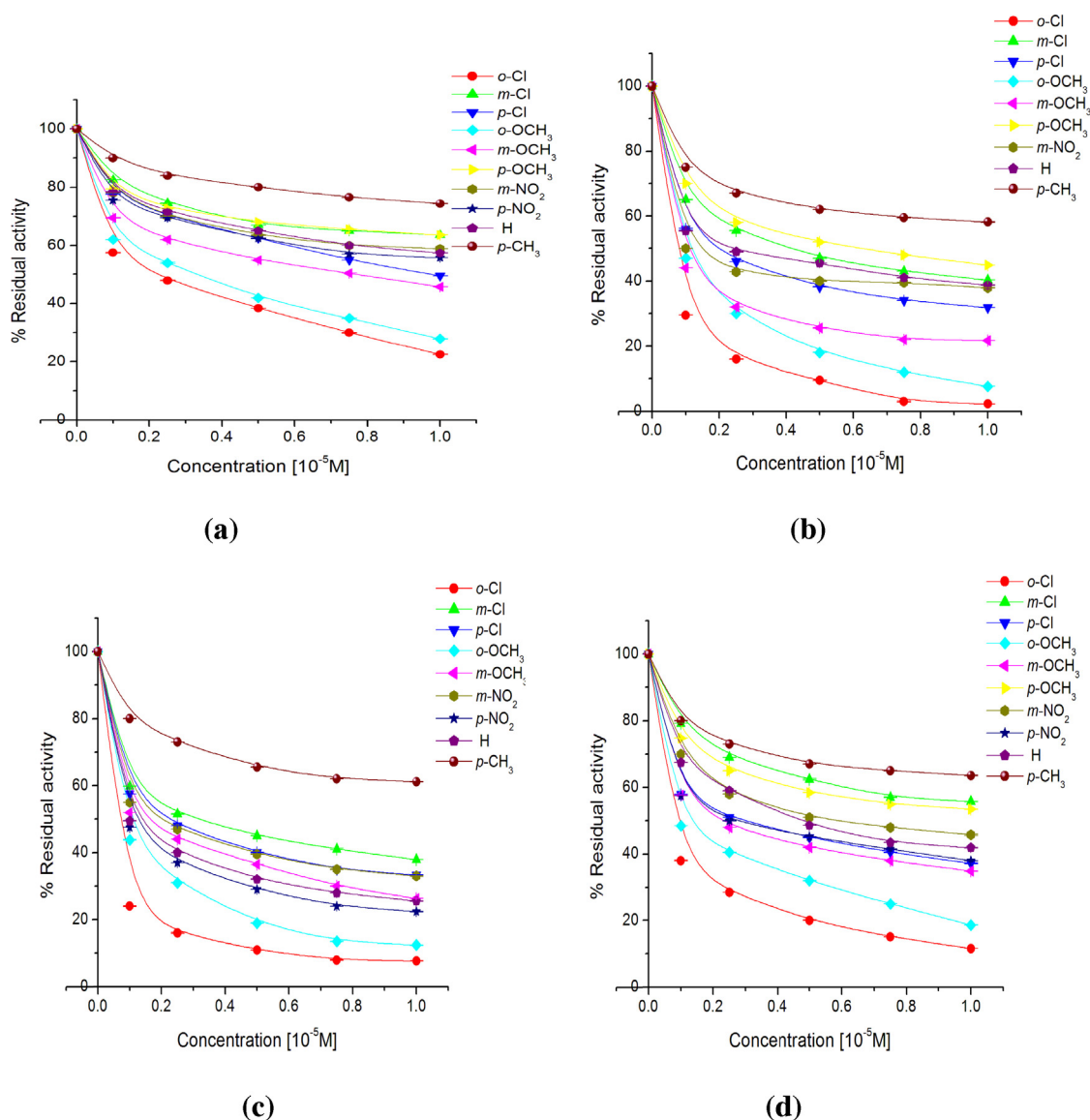


Fig. 2. The results are depicted as % residual activities of cathepsin B in presence of different concentrations (0.1, 0.25, 0.50, 0.75 and 1.0×10^{-5} M) of various chalcones (1a–1j; a), chalconesemicarbazone isomers (2a–2j; b), (2a'–2j'); c) and diarylpyrazolines (3a–3j; d) after an interaction time of 30 min. The experiments were conducted in triplicate and were calculated w.r.t. the control having no compound but an equivalent amount of solvent was added.

assay using the respective substrates. The experiments were performed in triplicate for each concentration and % activity has been calculated with respect to the control where no compound was added but an equivalent amount of solvent was present. Enzyme assays were similarly conducted at lower concentrations of each compound to observe the inhibitory effect of compounds at varying concentrations. The results are presented in Figs. 2–4 for cathepsins B, H and L, respectively.

2.4.4. Enzyme kinetic studies

After establishing the inhibitory action of 1,3-diphenylprop-2-en-1-ones (1a–1j), substituted chalconesemicarbazones (2a–2j, 2a'–2j') and substituted 3,5-diphenyl-2-pyrazoline-1-carboxamide derivatives (3a–3j) on cathepsins B, H and L experiments were designed to evaluate the type of inhibition and to determine the K_i value of these compounds on cathepsin B, H and L. For that, enzyme activity was evaluated at different substrate concentration in presence and absence of a fixed concentration of inhibitor. Lineweaver Burk plots were drawn between $1/[S]$ and $1/V$ (Figs. 5–7). The K_m value of cathepsin B, H and L for BANA, Leu β NA and

ZPheArg-4m β NA was found to be 4.3×10^{-4} M, 5.0×10^{-4} M and 0.5×10^{-4} M, respectively. The K_i values have been summarized in Table 1.

2.4.5. Drug modelling studies

Docking studies were performed using iGEMDOCK software. To conduct these, small molecular weight ligands were prepared using Marvin Sketch and were saved as MDL Mol File. Enzyme structure active site was retrieved from the Protein Data Bank (<http://www.rcsb.org/>) as cav2IPP B.PYS.pdb, cav8PCH H.NAG.pdb and cav3BC3L.CSW [69–71]. The prepared ligands and the binding site was loaded in the iGEMDOCK programme and docking was run by setting GA parameters for Standard Docking Accuracy Settings. Docking experiments show a decrease in energy when enzyme and ligands interact. The E_{total} resulting after H-bonding and van der Waals interactions are presented in Tables S1–S3 (Supplementary data). The docking poses of the most inhibitory compounds 1a, 2a, 2a', 3a for cathepsin B, 1g, 2g, 2c', 3b for cathepsin H and 1b, 2c, 2a', 3g for cathepsin L are shown in Figs. 8–10.

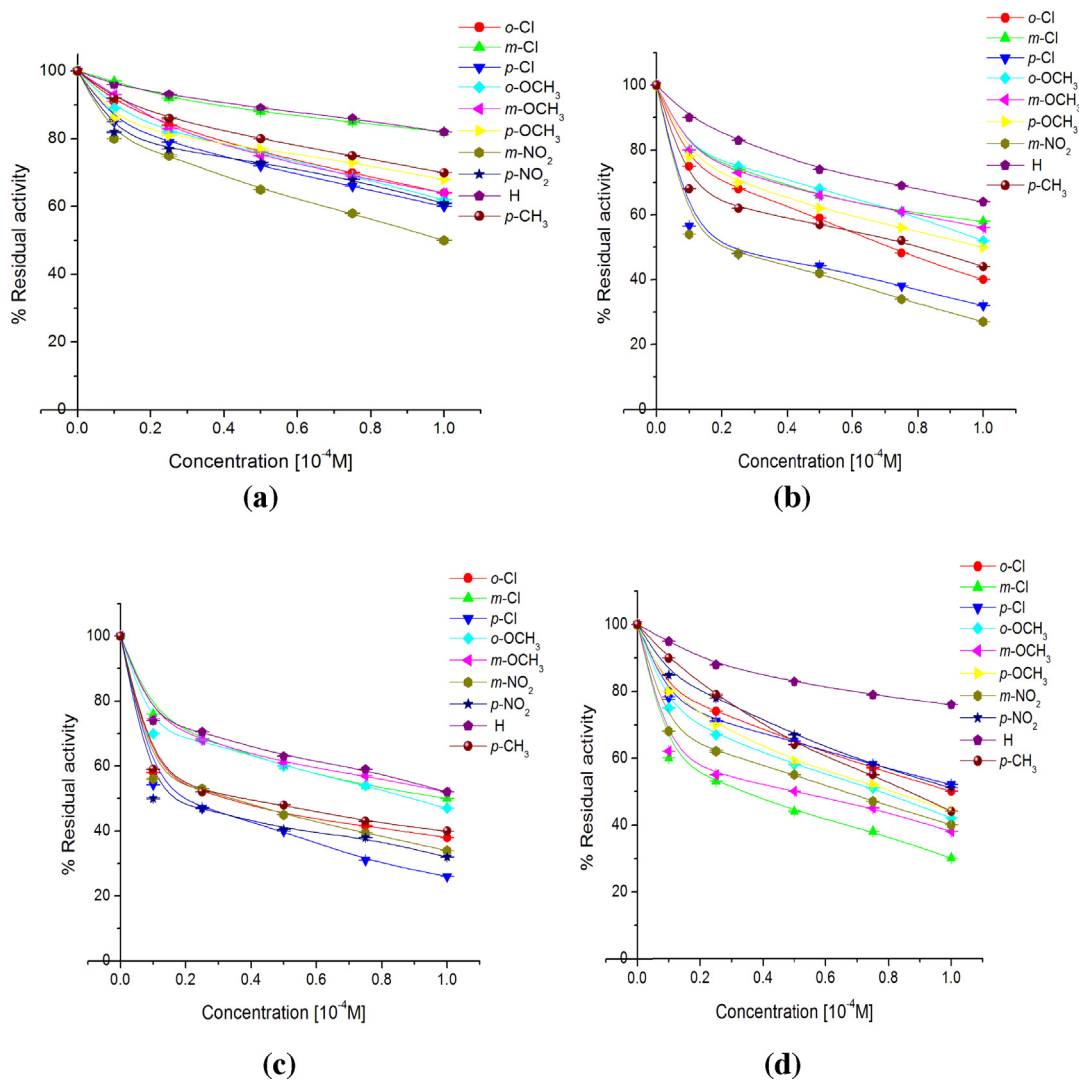


Fig. 3. The results are depicted as % residual activities of cathepsin H in presence of different concentrations (0.1, 0.25, 0.50, 0.75 and 1.0×10^{-4} M) of various chalcones (1a–1j; a), chalconesemicarbazone isomers (2a–2j; b), (2a'–2j'; c) and diarylpyrazolines (3a–3j; d) after an interaction time of 30 min. The experiments were conducted in triplicate and were calculated w.r.t. the control having no compound but an equivalent amount of solvent was added.

3. Result and discussion

The synthesis of chalcones (1a–1j) from substituted benzaldehyde and acetophenone was carried out using Claisen-Schmidt condensation. The substituted Chalconesemicarbazones (2a–2j, 2a'–2j') and substituted 3,5-diphenyl-2-pyrazoline-1-carboxamide (3a–3j) were synthesized [65] from these chalcones using semicarbazide hydrochloride (Scheme 1). After establishing the structure of chalcones their corresponding semicarbazones were synthesized and the structures were elucidated using melting point, IR and NMR spectra.

The synthesis of chalconesemicarbazones resulted in two isomeric form 2a–2j and 2a'–2j'. These two isomeric forms were separated by column chromatography on silica gel using petroleum ether and ethyl acetate as eluent. The separated isomers were characterized by ^1H NMR data. The isomers obtained were characterized on the basis of interaction of lone pair on nitrogen with C–H proton as explained by Curphy-Morrison chemical shift [72]. When lone pairs on nitrogen are syn to C–H bonds (2a–2j), the proton is shifted upfield showing the lower δ value. When lone pairs on nitrogen are anti to C–H bonds (2a'–2j'), the proton is shifted downfield. The conclusion was arrived on the basis of 3D

molecular modelling studies. In 2i where lone pair on nitrogen is syn to $\text{C}_2\text{-H}$ (Scheme 1) on minimization of energy in 3D molecular modelling (Fig. 1) [73] clearly indicate that lone pair on nitrogen is in optimal position to affect shielding of $\text{C}_2\text{-H}$ and also to $\text{C}_3\text{-H}$ to some extent because of extension of conjugation. In 2i' the lone pair on nitrogen is anti to $\text{C}_2\text{-H}$ and after 3D molecular modelling in minimized energy structure shown in Figure 1, it is clear that these can't have any effect on $\text{CH}=\text{CH}$ protons. On comparing the ^1H NMR, peaks for $\text{C}_2\text{-H}$ and $\text{C}_3\text{-H}$ for chalcones and chalconesemicarbazones, distinctive shielding effect is observed due to substitution of adjacent $\text{C}=\text{O}$ group with $\text{C}=\text{N}-\text{NHCONH}_2$. These isomeric forms are reported here for the first time, although a large number of reports are available in literature on the synthesis of these compounds [74–76]. In case of *p*- OCH_3 and *p*- NO_2 substituted isomers only one isomer was formed i.e. 2f and 2h, respectively.

On cyclization of chalconesemicarbazones, diarylpyrazolines are obtained and each of the two isomeric forms (2a–2j) and (2a'–2j') resulted in single product (3a–3j). In contrast, two isomeric forms of pyrazolines have been reported by Shekarchi et al. [77]. In the ^1H NMR spectra of (3a–3j), the characteristic ABX pattern has been obtained.

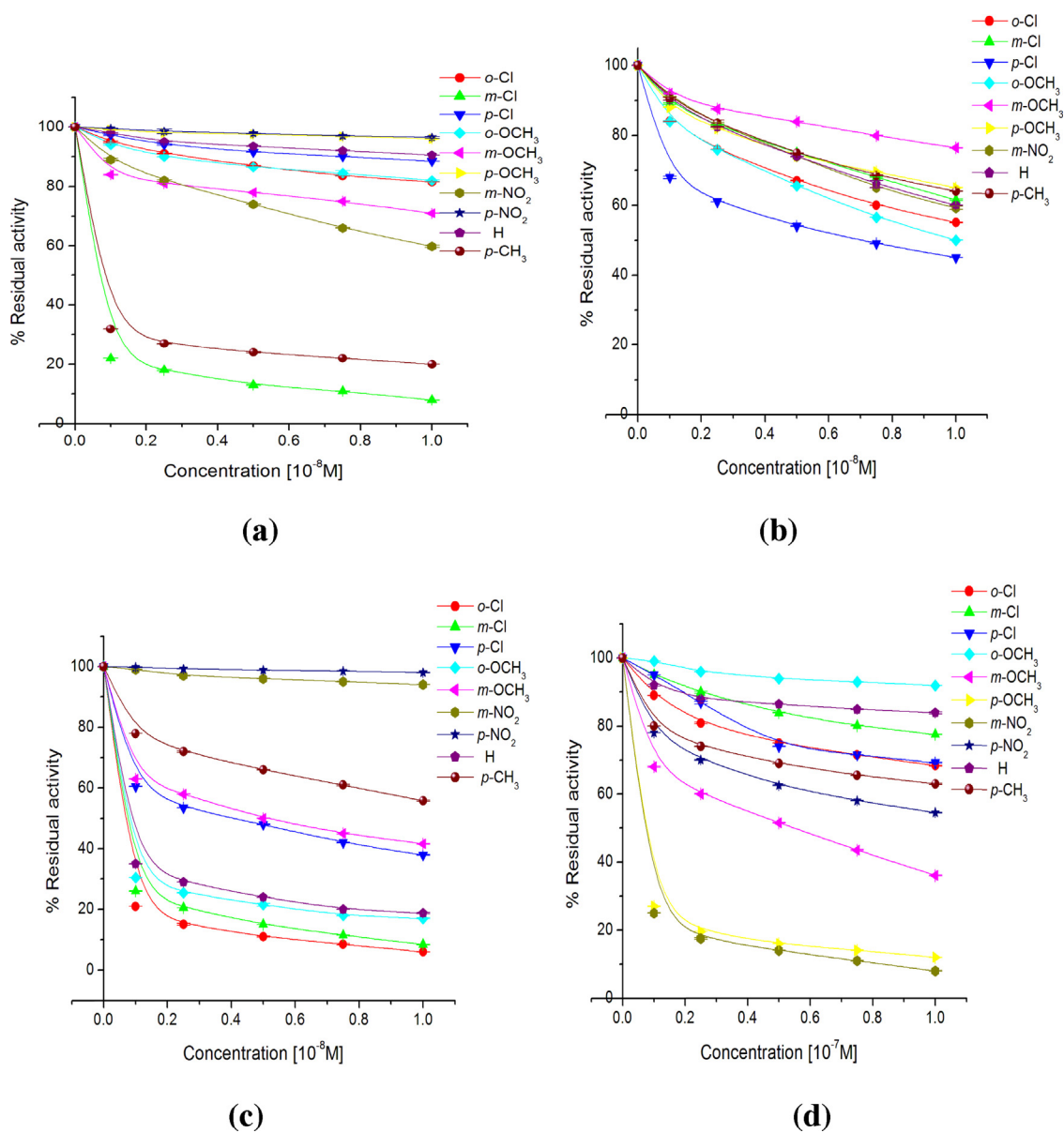


Fig. 4. The results are depicted as % residual activities of cathepsin L in presence of different concentrations (0.1, 0.25, 0.50, 0.75 and 1.0×10^{-8} M) of various chalcones (1a–1j; a), chalconesemicarbazone isomers (2a–2j; b), (2a'–2j'; c) and (0.1, 0.25, 0.50, 0.75 and 1.0×10^{-7} M) of various diarylpyrazolines (3a–3j; d) after an interaction time of 30 min. The experiments were conducted in triplicate and were calculated w.r.t. the control having no compound but an equivalent amount of solvent was added.

3.1. Effect of synthesized compounds on *in vitro* endogenous proteolysis in liver homogenate

Proteolytic activity is inhibited appreciably in presence of these compounds. In some cases $\sim 100\%$ inhibition is achieved at 1×10^{-4} M concentration. Moreover the inhibition was more at 3.0 h and less at 24.0 h, emphasizing that the compounds inhibit the proteolytic activity reversibly. In case of chalcones, chloro and unsubstituted compounds showed 100% inhibition. For chalconesemicarbazones and diaryl pyrazolines, chloro and nitro substituted compounds showed 100% inhibition. In addition, *o*- and *p*-methoxy substituted chalconesemicarbazone and *o*-methoxy pyrazoline also exhibited 100% inhibition while unsubstituted and methyl substituted chalconesemicarbazones and pyrazolines exerted inhibition to lesser extent.

After establishing the inhibition of endogenous proteolytic activity in presence of chalcones and their derivatives at pH 5.0, where most of the proteolytic activity is attributed to cysteine

proteases it was thought proper to study the effect of synthesized compounds on purified cathepsins B, H and L.

3.2. Effect of synthesized compounds on the activity of cathepsins B, H and L

The effect of differently substituted chalcones (1a–1j), chalconesemicarbazones (2a–2j, 2a'–2j') and diarylpyrazolines (3a–3j) on the activity of cathepsins B, H and L at varying concentrations is shown in Figs. 2–4. From these plots of % residual activities versus the concentrations of different compounds, it can be observed that at a particular concentration all the synthesized compounds inhibited cathepsin L activity more than cathepsin B and H.

3.3. The inhibition type and K_i values

The type of inhibition caused by various compounds was determined through Lineweaver–Burk double reciprocal plot. In order

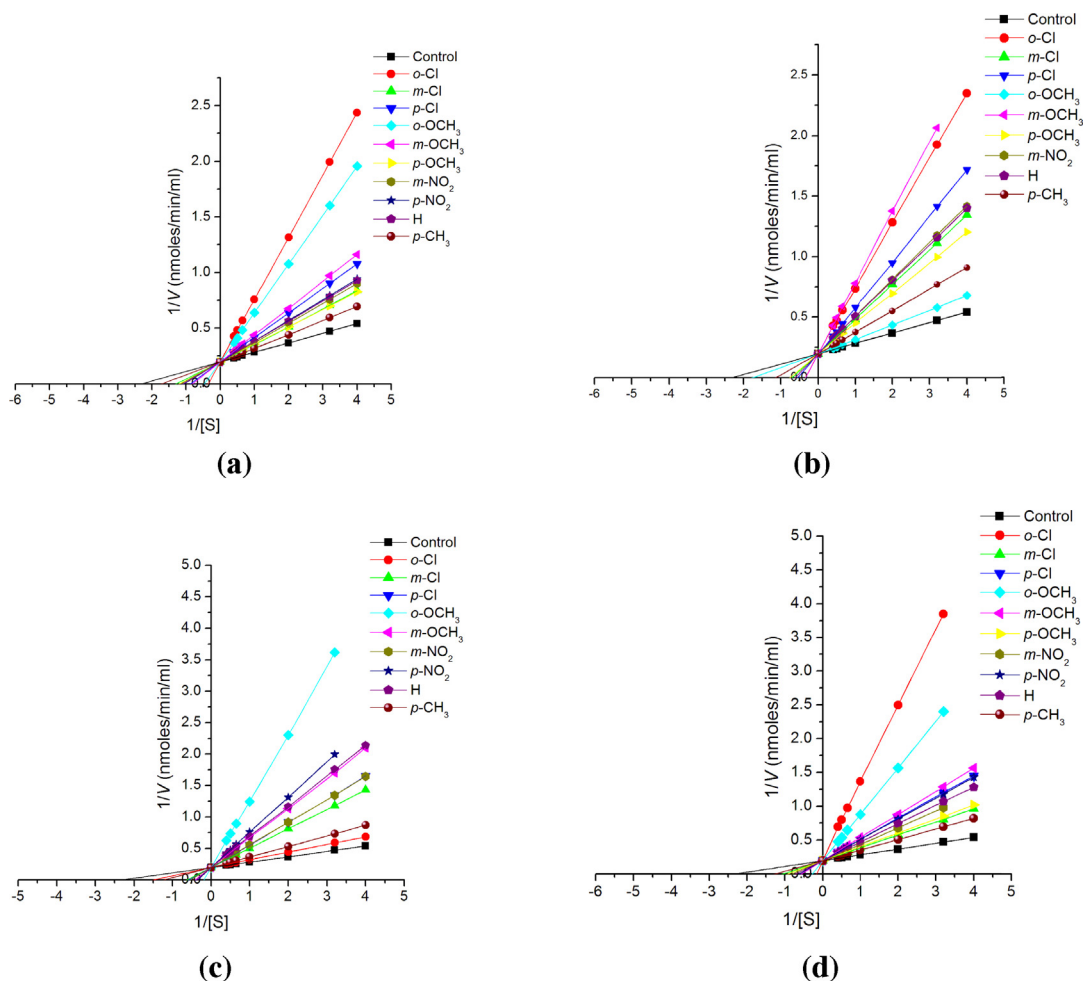
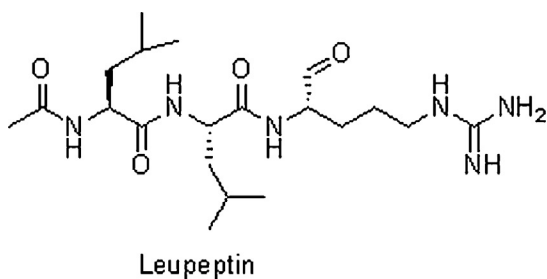


Fig. 5. Lineweaver–Burk plots for inhibition of substituted chalcones (1a–1j); a), chalconesemicarbazone isomers (2a–2j); b), (2a'–2j'); c) and diarylpyrazolines (3a–3j); d) on cathepsin B at fixed concentration (10^{-5} M) of inhibitor and varying substrate i.e. BANA concentration (2.5, 2.0, 1.5, 1.0, 0.5, 0.3, 0.25 mM).

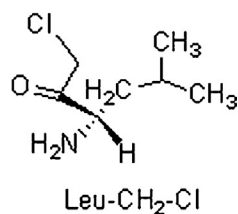
to establish inhibition ability of the under consideration compounds results were compared with potent inhibitors of cathepsin B, e.g. Leupeptin and cathepsin H e.g. Leu-CH₂-Cl, respectively. As reported in literature, Leupeptin being a potent peptide inhibitor of cathepsin B, inhibited the goat brain cathepsin B competitively with K_i value of 12.5×10^{-9} M [66] whereas K_i value for human liver cathepsin B [78] was reported to be 7.0×10^{-9} M. In contrast, K_i value for human liver cathepsin H was reported to be 9.2×10^{-6} M [79] and K_i value for goat brain cathepsin L was reported to be 1.45×10^{-9} M [67].



For evaluating the type of inhibition caused by different chalcones, chalconesemicarbazones and diarylpyrazolines, cathepsins B, H and L activity was measured at varying substrate i.e. BANA, Leu β NA and ZPheArg-4m β NA concentration in presence and absence of a fixed concentration of compound. The plots of $1/V$

and $1/[S]$ were straight lines intersecting at the Y-axis and shows that value of V_{max} remains constant in all the compounds whereas the value of $K_{m'}$ change with each compound. These studies suggested that chalcones, chalconesemicarbazones and diarylpyrazolines are competitive inhibitors to cathepsins B, H and L. Using the Lineweaver–Burk equation of competitive inhibition the K_i values were calculated, which has been presented in Table 1.

$$K_{m'} = K_m \left(1 + \frac{[I]}{K_i} \right)$$



Lineweaver–Burk plots of different chalcones (1a–1j), chalconesemicarbazones (2a–2j, 2a'–2j') and diarylpyrazolines (3a–3j) for cathepsins B, H and L are shown in Figs. 5–7.

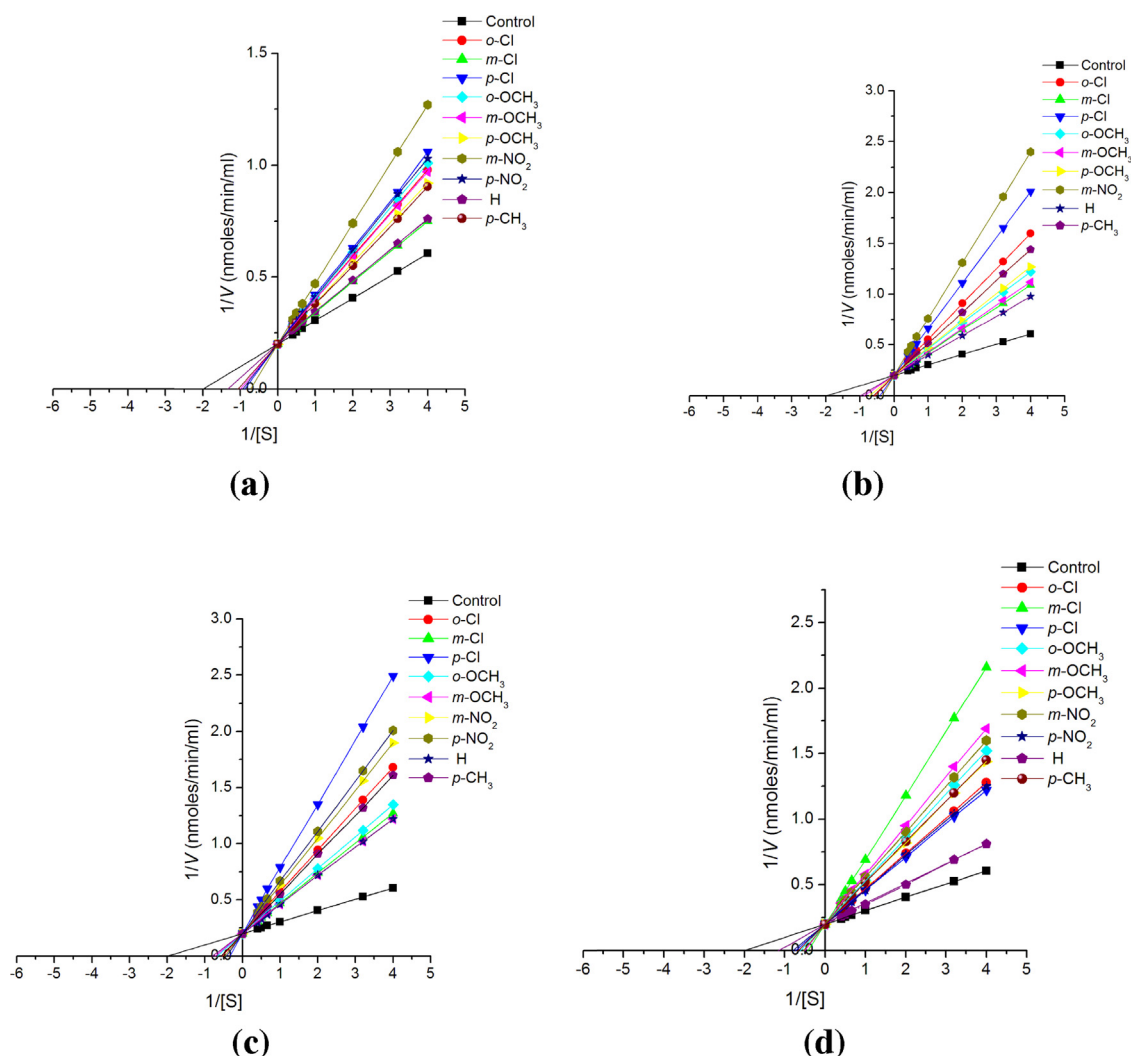


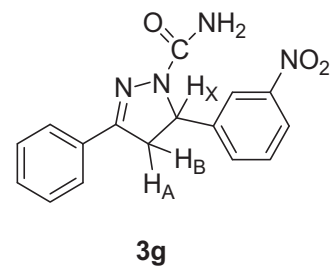
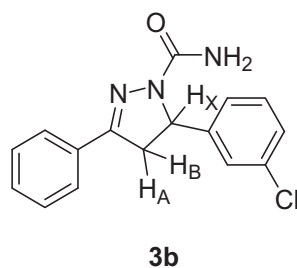
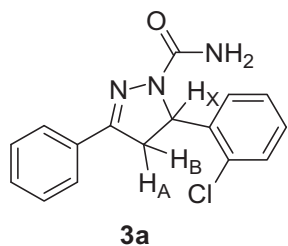
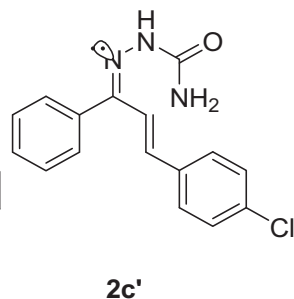
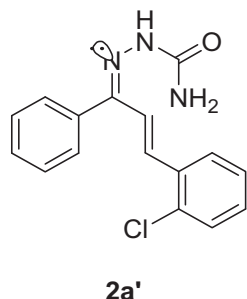
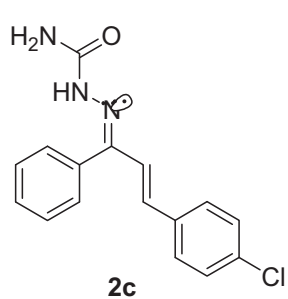
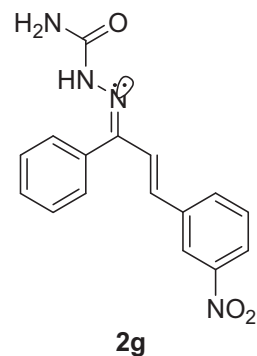
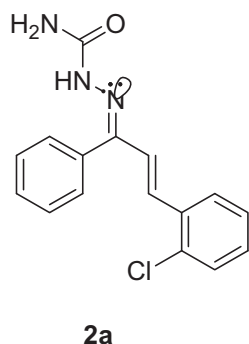
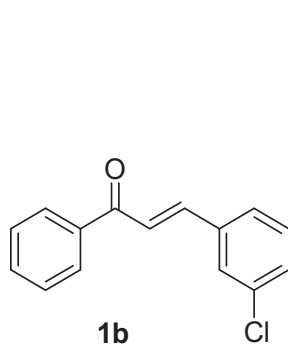
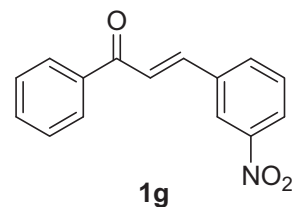
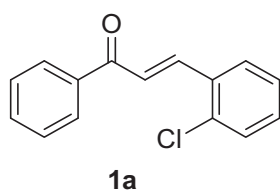
Fig. 6. Lineweaver–Burk plots for inhibition of substituted chalcones (1a–1j; a), chalconesemicarbazone isomers (2a–2j; b), (2a'–2j'; c) and diarylpyrazolines (3a–3j; d) on cathepsin H at fixed concentration (10^{-4} M) of inhibitor and varying substrate i.e. Leu β NA concentration (2.5, 2.0, 1.5, 1.0, 0.5, 0.3, 0.25 mM).

3.4. Structure activity relationship

Proteases contribute to tumour cell invasion and angiogenesis and are commonly associated with metastasis. It is now recognized that cysteine proteases play pivotal roles in cancer progression. Of all the cathepsins, studies have shown that cathepsins B, H and L are of significant importance as these are involved in various pathologies and oncogenic processes. Though these enzymes have been intensively studied as valuable targets for drug discovery and development, a number of peptide [80] and non-peptide [53,54] inhibitors have been described in literature. Towards this endeavour we have now synthesized chalcones and their semicarbazone and pyrazoline derivatives with different functionalities in order to explore their inhibitory potential on these important enzymes keeping in view the inhibitory potential of semicarbazones and pyrazolines as well as chalcones on cysteine proteases. The work has been envisaged by combining the two active pharmacophores.

Out of various synthesized compounds, (E)-3-(2-chlorophenyl)-1-phenylprop-2-en-1-one (1a), (1Z)-1-((E)-3-(2-chlorophenyl)-1-phenylallylidene)semicarbazide (2a), (1E)-1-((E)-3-(2-chlorophenyl)-1-phenylallylidene)semicarbazide (2a') and 5-(2'-chlorophenyl)-3-phenyl-2-pyrazoline-1-carboxamide (3a)

with K_i values $(1.95 \pm 0.02) \times 10^{-5}$ M, $(0.207 \pm 0.001) \times 10^{-6}$ M, $(2.70 \pm 0.04) \times 10^{-6}$ M and $(9.40 \pm 0.03) \times 10^{-6}$ M, respectively showed maximum inhibition on the activity of cathepsin B. However, activity of cathepsin H was maximally inhibited by (E)-3-(3-nitrophenyl)-1-phenylprop-2-en-1-one (1g), (1Z)-1-((E)-3-(3-nitrophenyl)-1-phenylallylidene)semicarbazide (2g), (1E)-1-((E)-3-(4-chlorophenyl)-1-phenylallylidene)semicarbazide (2c') and 5-(3'-chlorophenyl)-3-phenyl-2-pyrazoline-1-carboxamide (3b) with K_i values $(6.00 \pm 0.02) \times 10^{-5}$ M, $(2.30 \pm 0.03) \times 10^{-5}$ M, $(2.10 \pm 0.02) \times 10^{-5}$ M and $(2.50 \pm 0.05) \times 10^{-5}$ M, respectively and were found to be the best inhibitors for cathepsin H. Similarly, (E)-3-(3-chlorophenyl)-1-phenylprop-2-en-1-one (1b), (1Z)-1-((E)-3-(4-chlorophenyl)-1-phenylallylidene)semicarbazide (2c), (1E)-1-((E)-3-(2-chlorophenyl)-1-phenylallylidene)semicarbazide (2a') and 5-(3'-nitrophenyl)-3-phenyl-2-pyrazoline-1-carboxamide (3g) with K_i values $(0.04 \pm 0.0001) \times 10^{-9}$ M, $(1.17 \pm 0.02) \times 10^{-9}$ M, $(0.4 \pm 0.003) \times 10^{-10}$ M and $(0.04 \pm 0.0005) \times 10^{-8}$ M, respectively showed maximum inhibition on the activity of cathepsin L. Followed by these results it was concluded that the synthesized compounds showed more inhibition on activity of cathepsin L than on cathepsins B and H. This may lead to the development of selective inhibitors to cathepsins B, H and L.



3.5. Small molecular weight representative molecules

Further it can be observed that chalconesemicarbazones showed maximum inhibition than their cyclized precursor pyrazolines followed by chalcones on the activity of cathepsin B and cathepsin H. Whereas, for cathepsin L chalcones and chalconesemicarbazones were found to show more inhibition than pyrazolines. The results are discussed later in the next section along with the results obtained after molecular docking.

Among the substituted chalcones, chalconesemicarbazones and pyrazolines, the compounds bearing electron withdrawing moiety were found to be more inhibitory probably inducing more nucleophilic character at the site of attack shown in Scheme 2.

The results obtained were compared with potential inhibitors of cathepsin B and L e.g. Leupeptin and cathepsin H, e.g. Leu-CH₂-Cl, respectively [54]. It can be observed that Leupeptin showed ~98.8% inhibition at 10⁻⁶ M concentration for cathepsin B whereas it showed ~52.0% inhibition at 10⁻⁵ M concentration for cathepsin H and 89.2% inhibition at 10⁻⁹ M concentration for cathepsin L which is in accordance with the previously reported results. Similarly, Leu-CH₂-Cl showed ~9.5% inhibition at 10⁻⁵ M concentration

for cathepsin B whereas it showed ~93.5% inhibition at 10⁻⁵ M concentration for cathepsin H and 1.5% inhibition at 10⁻⁵ M concentration for cathepsin L. The results obtained are comparable with earlier results reported for brain cathepsin H [68], cathepsin B and cathepsin L [81].

3.6. Molecular docking experiment

The docking approach was used to study the interaction of compounds with the active site of cathepsin B, H and L to observe binding poses of individual compounds. Individual binding poses of each compound was assessed and their interactions in the active site of the enzyme were analyzed. The empirical scoring function of iGemDOCK is the estimated sum total of van der Waals, H-bonding and electrostatic energy.

Table S1 (Supplementary data) presents the data of docking studies of different chalcones, their acyclic and cyclic semicarbazone derivatives in cathepsin B active site (cav2IPP_B_PYS.pdb). The results clearly indicate that binding energy of chalcones is less than pyrazolone derivatives which is less than the open chain semicarbazones. The binding energy of chalcones (1a–1j),

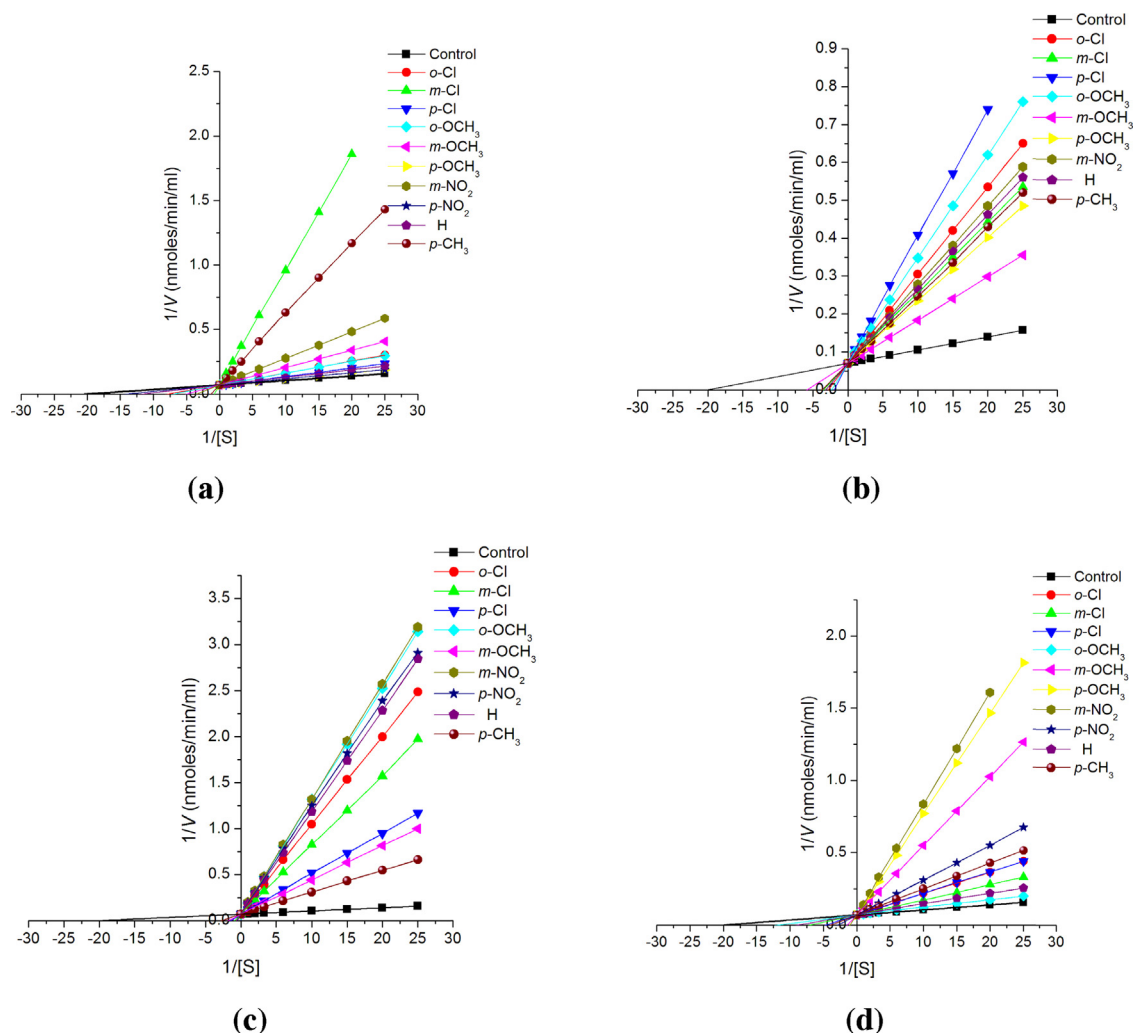
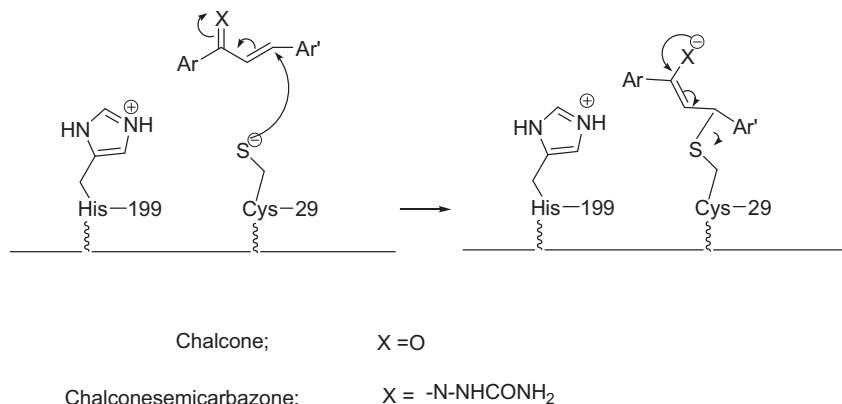


Fig. 7. Lineweaver–Burk plots for inhibition of substituted chalcones (1a–1j); a), chalconesemicarbazone isomers (2a–2j); b), (2a'–2j'); c) and diarylpyrazolines (3a–3j); d) on cathepsin L at fixed concentration (10^{-8} M of chalcones (1a–1j), chalconesemicarbazone (2a–2j, 2a'–2j') and 10^{-7} M of diarylpyrazolines (3a–3j)) of inhibitor and varying substrate i.e. ZPheArg-4m β NA concentration (1.0, 0.5, 0.3, 0.16, 0.1, 0.06, 0.05, 0.04 mM).

chalconesemicarbazones (2a–2j, 2a'–2j') and diarylpyrazolines (3a–3j) vary in the range -71.00 to -80.46 , -83.11 to -99.13 , -88.75 to -95.97 and -82.51 to -91.78 , respectively. The most inhibitory compound in each series was 1a, 2a, 2a' and 3a having E_{total} as -71.76 , -95.42 , -90.54 and -84.27 , respectively. These binding energies when compared to the binding energies of

leupeptin with cathepsin B are lesser to approximately 20 units. The docking results explain that the title compounds may not be potent inhibitors to the protease activity to an extent of specific peptidyl inhibitors but certainly suggest a differential inhibitory pattern of these compounds on cathepsin B activity. However, maximum interaction is observed for BANA with a score of -129.83 .



Scheme 2. Proposed mechanism for inhibition of cathepsin B with chalcone and chalconesemicarbazone analogues.

Table 1
 K_i values of various chalcones (1a–1j), chalconesemicarbazones (2a–2ja, 2a'–2j') and diarylpyrazolines (3a–3j) for cathepsins B, H and L.

Compounds	Cathepsin B $K_i \pm$ S.M.D [1×10^{-5} M]	Cathepsin H $K_i \pm$ S.M.D [1×10^{-5} M]	Cathepsin L $K_i \pm$ S.M.D [1×10^{-9} M]
1a	1.95 ± 0.02	11.0 ± 0.1	6.00 ± 0.02
1b	13.0 ± 0.2	31.2 ± 0.2	0.04 ± 0.0001
1c	6.61 ± 0.03	8.30 ± 0.03	12.50 ± 0.08
1d	2.40 ± 0.06	10.00 ± 0.02	6.20 ± 0.02
1e	6.32 ± 0.03	11.00 ± 0.007	3.50 ± 0.08
1f	13.0 ± 0.1	13.80 ± 0.01	26.30 ± 0.07
1g	10.0 ± 0.1	6.00 ± 0.02	2.05 ± 0.03
1h	8.77 ± 0.04	10.00 ± 0.005	27.7 ± 0.3
1i	9.14 ± 0.03	26.3 ± 0.2	15.15 ± 0.1
1j	22.67 ± 0.3	15.10 ± 0.03	0.06 ± 0.0002

Compounds	Cathepsin B $K_i \pm$ S.M.D [1×10^{-6} M]	Cathepsin H $K_i \pm$ S.M.D [1×10^{-5} M]	Cathepsin L $K_i \pm$ S.M.D [1×10^{-9} M]
2a	0.207 ± 0.001	4.30 ± 0.02	1.78 ± 0.02
2b	43.0 ± 0.4	9.60 ± 0.06	2.42 ± 0.01
2c	27.3 ± 0.4	2.90 ± 0.05	1.17 ± 0.02
2d	2.70 ± 0.02	6.60 ± 0.07	1.56 ± 0.02
2e	17.0 ± 0.2	7.40 ± 0.06	4.71 ± 0.02
2f	52.4 ± 0.8	5.40 ± 0.03	2.66 ± 0.03
2g	41.3 ± 0.5	2.30 ± 0.03	2.05 ± 0.02
2i	42.5 ± 0.2	10.00 ± 0.04	2.27 ± 0.04
2j	91.4 ± 0.9	4.50 ± 0.05	2.50 ± 0.02

Compounds	Cathepsin B $K_i \pm$ S.M.D [1×10^{-6} M]	Cathepsin H $K_i \pm$ S.M.D [1×10^{-5} M]	Cathepsin L $K_i \pm$ S.M.D [1×10^{-10} M]
2a'	2.70 ± 0.04	3.30 ± 0.04	0.4 ± 0.003
2b'	39.0 ± 0.5	6.00 ± 0.01	0.5 ± 0.005
2c'	36.4 ± 0.5	2.10 ± 0.02	9.3 ± 0.07
2d'	9.40 ± 0.05	5.40 ± 0.03	3.1 ± 0.05
2e'	24.0 ± 0.2	6.60 ± 0.03	11.0 ± 0.13
2g'	34.9 ± 0.2	3.30 ± 0.01	149.3 ± 2.0
2h'	19.50 ± 0.02	3.20 ± 0.02	168.0 ± 0.6
2i'	22.00 ± 0.04	6.60 ± 0.02	3.6 ± 0.04
2j'	104.0 ± 1.0	4.00 ± 0.04	16.9 ± 0.2

Compounds	Cathepsin B $K_i \pm$ S.M.D [1×10^{-6} M]	Cathepsin H $K_i \pm$ S.M.D [1×10^{-5} M]	Cathepsin L $K_i \pm$ S.M.D [1×10^{-8} M]
3a	9.40 ± 0.03	6.00 ± 0.03	3.16 ± 0.04
3b	82.7 ± 1.0	2.50 ± 0.05	5.05 ± 0.03
3c	39.0 ± 0.2	6.60 ± 0.04	3.30 ± 0.01
3d	14.80 ± 0.07	4.00 ± 0.01	21.74 ± 0.40
3e	34.9 ± 0.1	4.30 ± 0.01	0.82 ± 0.006
3f	69.30 ± 0.07	4.80 ± 0.02	0.05 ± 0.0002
3g	58.10 ± 0.05	3.80 ± 0.02	0.04 ± 0.0005
3h	39.00 ± 0.03	6.00 ± 0.03	1.69 ± 0.03
3i	52.40 ± 0.01	19.20 ± 0.04	8.62 ± 0.02
3j	116.2 ± 2.0	4.80 ± 0.05	2.34 ± 0.02

The experiments were conducted in triplicate using different concentrations of bana, leu β na and zphearg-4m β na as substrate for cathepsins b, h and l, respectively. The results are calculated using Lineweaver–Burk equation for competitive inhibitors.

Decrease in total energy for leupeptin-cathepsin B has come out be -116.42 of which the contribution of van der Waals interactions are more with a score of -86.37 as compared to H-bonds with a score of -29.44 . Leupeptin-cathepsin B binding energy is due to peptide–protein interaction. Leupeptin is peptidyl in nature and therefore being a flexible molecule binds effectively with the enzyme active site resulting in higher binding energy. iGem-DOCK provide algorithms for flexible docking approach for both ligands and proteins [82] therefore flexible ligands like leupeptin will show a larger decrease in total energy as compared to the molecules under study as these are smaller in structure and possess lesser flexibility compared to leupeptin. Therefore, the binding energy of title compounds is less than leupeptin. In the quest of

synthesizing potent inhibitors of enzymes a structure–activity relationship is required where cyclic and acyclic analogues are to be studied. There are various pharmaco–dynamic reactions which molecule acting as drug encounters in vivo such as cyclization, oxidation, reduction etc. so we have here, provided the comparative account of in vitro studies and *in-silico* study on the structurally related compounds or their isomers. These compounds are synthesized from the same starting material. The open chain analogues are the intermediates therefore there are a need to screen their inhibitory potential along with the products i.e. the cyclized derivatives, pyrazolines in order to establish structure activity relationship. In this direction the present study has been undertaken to evaluate the effect of 1, 3-diphenylprop-2-en-1-ones (1a–1j), substituted chalconesemicarbazones (2a–2j, 2a'–2j') which are open chain compounds and their cyclic analogues i.e. substituted 3,5-diphenyl-2-pyrazoline-1-carboxamide derivatives (3a–3j) on the activity of cathepsin B, a protease of immense importance. During the synthetic studies of semicarbazones we came across the presence of two stereoisomers in the preparation. Docking experiments were conducted on these two isomers also and a differential effect was observed (Table S1-Supplementary data). The results of the docking studies support the in vitro experimental studies conducted on liver cathepsin B. It has been observed that chalcones were less inhibitory to the enzyme activity whereas semicarbazones and pyrazoline derivatives. This was quite surprising as chalcones having an α - β unsaturated carbonyl group seems to be more responsive towards thiol attack of cysteine present at the active site of enzyme but both the results after, *in-silico* docking experiments as well as in vitro solution studies clearly indicate that presence of azomethine carboxamide might have provided larger interaction to the binding of compound with the enzyme active site and there by exerting more inhibition than its precursor chalcones, which is further confirmed by the enzyme–ligand interaction study.

Fig. 8 shows the binding of most inhibitory compound 1a, 2a, 2a' and 3a in the active site of cathepsin B. The ligands are shown in pink, residues involved in H-bonding and van–der Waals interactions are shown as green and grey, respectively. It is clearly observed that cys-29 and His-199 residues present at the catalytic site of the enzyme are involved in the binding of compounds. Based on these interactions, a mechanism of inhibition (Scheme 2) by chalcones and chalconesemicarbazones is proposed on the basis of a previously reported inhibition of cathepsin L, another important cysteine protease by thiosemicarbazones [83]. The thiolate of cys-29 attacks on the nucleophilic site in chalcones and chalconesemicarbazones, where the later is more stabilized to the binding site. In addition, Gly-198, Ala-200 and Trp-30 amino acids residues are also involved in the stabilization of compounds in binding site. Pyrazolines are also better inhibitors to cathepsin B activity than chalcones again emphasizes on the importance of binding of compound with the active site. Though pyrazolines lack any nucleophilic site in their structure but still they show better inhibitory potential probably due to the presence of side chain carboxamide and nitrogen in the ring altogether providing a larger binding area to the enzyme site. Among themselves the cyclized pyrazolines were less inhibitory than open chain semicarbazones. It can be interpreted that acyclic compounds interact more with the active site being straight in nature not to the extent of peptidyl inhibitors but certainly greater than the cyclized pyrazoline derivatives. This gives an understanding of the inhibition caused by the target compounds on structural basis.

Similar trends have been observed for cathepsin H. However, in cathepsin H, the most inhibitory compound in each series have been found to be 1g, 2g, 2c' and 3b. Fig. 9 shows the binding of the most inhibitory compounds in the active site of cathepsin H. The binding energies of title compounds in the amino acyl

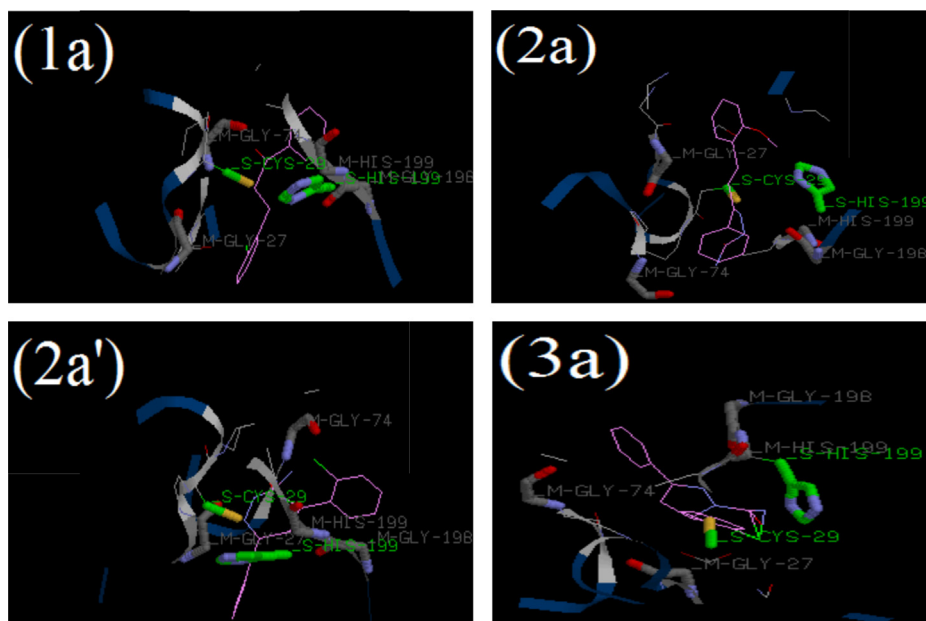


Fig. 8. Binding of most inhibitory chalcones 1a, chalconesemicarbazone isomer 2a, chalconesemicarbazone isomer 2a' and pyrazoline 3a into the binding site of cathepsin B (cav2IPP B.PYS.pdb).

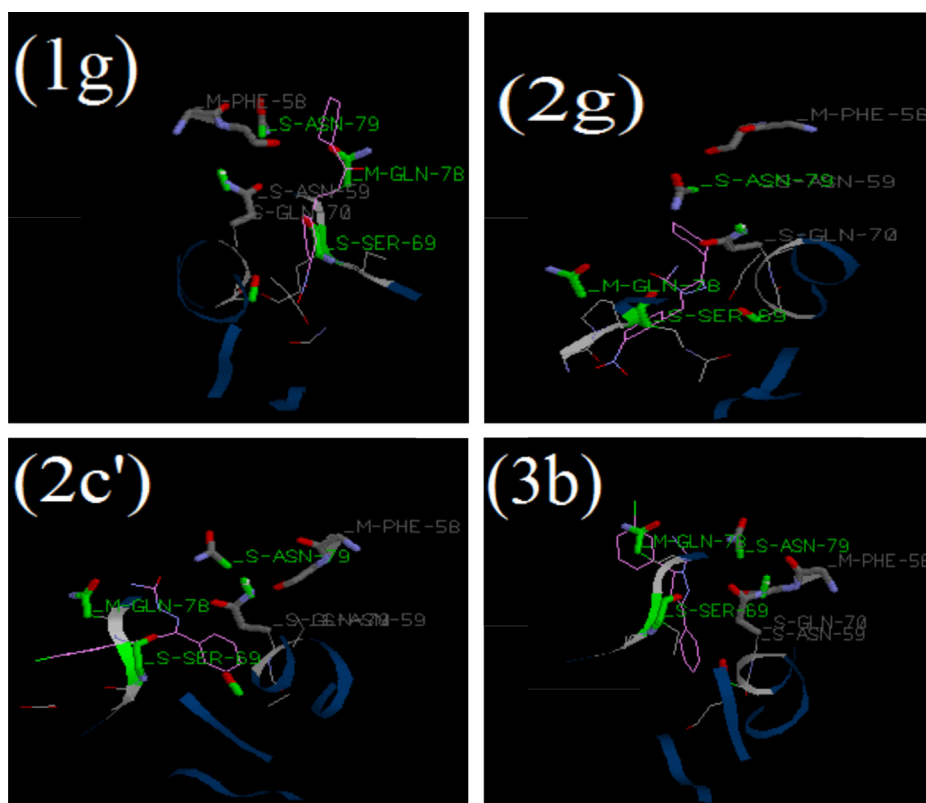


Fig. 9. Binding of most inhibitory chalcones 1g, chalconesemicarbazone isomer 2g, chalconesemicarbazone isomer 2c' and pyrazoline 3b into the binding site of cathepsin H (cav 8PCH H.NAG).

binding site of cathepsin H (cav8PCHH.NAG) is presented in Table S2 (Supplementary data). Experimental results obtained can be correlated with the ligand-binding interactions. It is observed that for 1g, 2g, 2c' and 3b, the binding energies computed come out to be -82.21 , -96.59 , -93.63 and -84.78 . In each series, these most inhibitory compounds show a decrease in binding energy towards higher side. The binding energies show effective

interaction between the enzyme binding site and inhibitory compounds may be responsible for these inhibition patterns.

Fig. 10 shows the binding of most inhibitory compounds 1b, 2c, 2a' and 3g in the active site of cathepsin L. The results of the docking studies support the in vitro experimental studies conducted on goat liver cathepsin L. The binding energies of title compounds in the amino acyl binding site of cathepsin L (cav3BC3L.CSW) is presented

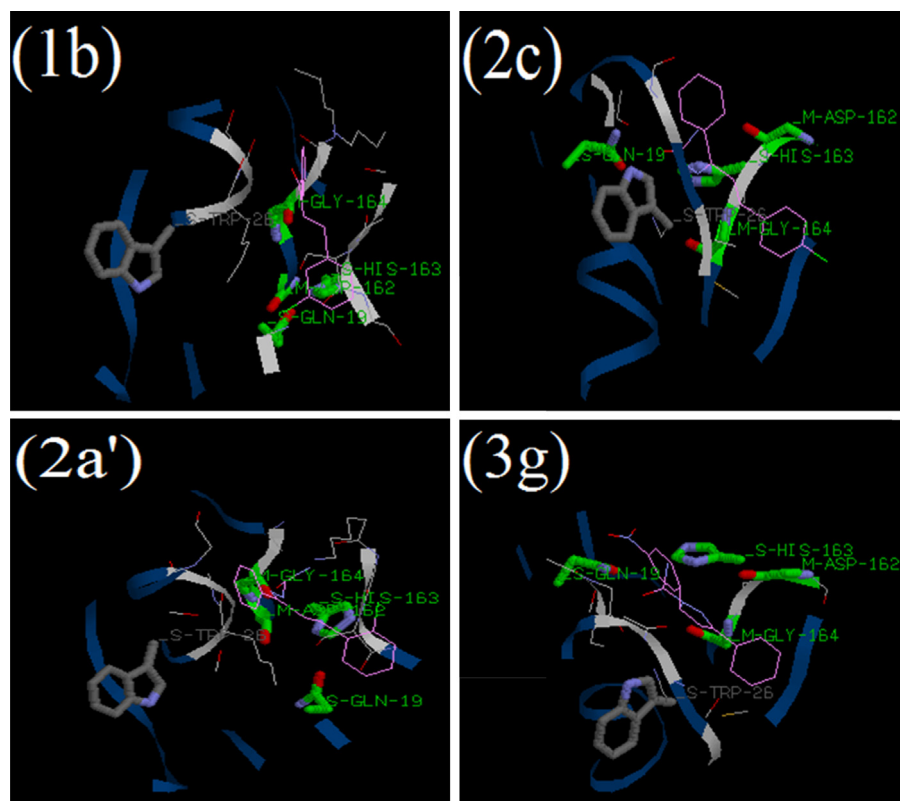


Fig. 10. Binding of most inhibitory chalcones 1b, chalconesemicarbazone isomer 2c, chalconesemicarbazone isomer 2a' and pyrazoline 3g into the binding site of cathepsin L (cav3BC3L.CSW).

in Table S3 (Supplementary data). The binding energies of 1b, 2c, 2a' and 3g were found to be -77.07 , -92.93 , -104.04 and -114.94 , respectively.

4. Conclusion

One of the important aspect of the present work comprise of synthesis and isolation of two stereoisomers of chalconesemicarbazones not reported earlier. Synthesis of these types of molecules has been largely reported in literature keeping in view of their vast biological activities and as precursors of pyrazolines and pyrazoles. The synthesized title compounds have been evaluated as better inhibitors for cathepsin L than cathepsin B followed by cathepsin H. Best inhibitor for cathepsin B has been evaluated as (1Z)-1-((E)-3-(2-chlorophenyl)-1-phenylallylidene) semicarbazide, (2a) with the K_i value of $(0.207 \pm 0.001) \times 10^{-6}$ M, for cathepsin H (1E)-1-((E)-3-(4-chlorophenyl)-1-phenylallylidene)semicarbazide (2c') showed maximum inhibition with a K_i value of $(2.10 \pm 0.02) \times 10^{-5}$ M and for cathepsin L, (1E)-1-((E)-3-(2-chlorophenyl)-1-phenylallylidene)semicarbazide (2a') showed maximum inhibition with the K_i value of $(0.4 \pm 0.003) \times 10^{-10}$ M Chalconesemicarbazones inhibited all the three enzymes effectively followed by pyrazolines and chalcones.

Conflict of interest

The authors have declared no conflict of interest.

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.ijbiomac.2015.07.029>

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