



Published in final edited form as:

Cell Cycle. 2009 December 15; 8(24): 4072–4078. doi:10.4161/cc.8.24.10242.

Peroxiredoxin 1 and its role in cell signaling

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Abstract

Peroxiredoxins (Prdxs) are a family of small (22–27 kDa) nonseleno peroxidases currently known to possess six mammalian isoforms. Although their individual roles in cellular redox regulation and antioxidant protection are quite distinct, they all catalyze peroxide reduction of H₂O₂, organic hydroperoxides and peroxynitrite.^{1,2} They are found to be expressed ubiquitously and in high levels,³ suggesting that they are both an ancient and important enzyme family. Prdxs can be divided into three major subclasses: typical 2-cysteine (2-Cys) Prdxs (Prdx1–4), atypical 2-Cys Prdx (Prdx 5) and 1-Cys Prdx (Prdx 6). Recent evidence suggests that 2-Cys peroxidases are more than “just simple peroxidases”. This hypothesis has been discussed elegantly in recent review articles, considering “over”-oxidation of the protonated thiolate peroxidatic cysteine and post-translational modification of Prdxs as processes initiating a mechanistic switch from peroxidase to chaperon function.^{4–6} The process of over-oxidation of the peroxidatic cysteine (C_P) occurs during catalysis in the presence of thioredoxin (Trx), thus rendering the sulfenic moiety to sulfenic acid,⁷ which can be reduced by sulfiredoxin (Srx).^{8,9} However, further oxidation to sulfonic acid is believed to promote Prdx degradation or, as recently shown, the formation of oligomeric peroxidase-inactive chaperones¹⁰ with questionable H₂O₂-scavenging capacity. In the light of this and given that Prdx1 has recently been shown by us¹¹ and by others^{12–17} to interact directly with signaling molecules, we will explore the possibility that H₂O₂ regulates signaling in the cell in a temporal and spatial fashion via oxidizing Prdx1. Therefore, this review will focus on H₂O₂ modulating cell signaling via Prdxs by discussing: (1) the activity of Prdxs towards H₂O₂; (2) sub cellular localization and availability of other peroxidases, such as catalase or glutathione peroxidases; (3) the availability of Prdxs reducing systems, such as thioredoxin and sulfiredoxin and lastly, (4) Prdx1 interacting signaling molecules.

Keywords

oxidative stress; peroxidases; cell signaling; phosphatases; peroxidases; transformation

Prdx1: “More than Just” a Peroxidase

Catalytic cycle.

All Prdxs share a conserved Cys residue, which corresponds to the N-terminal Cys⁵¹ (peroxidatic cysteine = C_P) in mammalian Prdx1. The majority of peroxidases (Prdxs 1–

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4) contain an additional conserved Cys residue in the C-terminal region that corresponds to Cys¹⁷² (resolving cysteine = C_R) in mammalian Prdx1. The N-terminal Cys⁵¹ is oxidized by H₂O₂ to cysteine-sulfenic acid (Cys⁵¹-SOH) and reacts with Cys¹⁷²-SH of the other subunit to produce an intermolecular disulfide (head to tail dimer), which can only be reduced by Trx, but not by GSH or glutaredoxin. Thus, the reducing equivalents stem from NADPH via thioredoxin reductase (TrxR) and Trx.^{18–20}

Prdxs can become easily over-oxidized on their catalytically active cysteine from sulphydryl to sulfinic or sulfonic acid in vitro and in vivo.^{5,10} It has been postulated that this is due to the fact that during catalysis Cys⁵¹-SH exists as a thiolate anion (Cys⁵¹-S⁻), whereas the other cysteines (Cys¹⁷, Cys⁸⁰ and Cys¹⁷²) remain protonated at neutral pH. The thiolate Cys⁵¹ is very unstable and highly reactive with any accessible thiol to form either a disulfide or to further oxidize to sulfinic or sulfonic acid.²¹ Crystal structures of Prdx1, Prdx2 and their yeast or bacterial counterparts revealed that in reduced Prdxs, the sulfur atoms of the N- and C-terminal conserved cysteine residues are too far apart to react with each other. Thus, disulfide formation requires significant conformational changes, such as unwinding of the active site N-terminal helix and the movement of four loops. Surprisingly, Cys⁵¹ in crystallized Prdx2, isolated from aged erythrocytes, exists as a sulfinic moiety, buried within the active-site pocket with Cys¹⁷¹ partially exposed. There, sulfinic Cys⁵¹ forms a salt bridge with Arg¹²⁷, similarly to as reduced Cys⁵¹ does.²² Based on this, it has been proposed that in a catalytic cycle, peroxidation of the catalytic cysteine in Prdx2 occurs in the fully folded active site, where the C_P (Cys⁵⁰ in Prdx2) is stabilized as a thiolate anion (Cys⁵⁰-S⁻) by Thr⁵² and the opposing Arg¹²⁷; Pro⁵¹ shield the active site from water. This active site environment lowers the pK_a value of Cys⁵⁰ to the range of 5–6, making the thiolate anion nucleophilic to attack the terminal oxygen of the peroxy bond (RO-OH), generating Cys⁵⁰-SOH and H₂O.²³ Cys⁵⁰-SOH in turn, builds a disulfide bridge with the sulphydryl group from the Cys¹⁷¹-SH concurrently releasing H₂O. Lastly, the Trx systems reduces the disulfide bond by thiol-disulfide exchange to regenerate free thiols in Cys⁵⁰ and Cys¹⁷¹.²⁴

What causes the additional or “over”-oxidation of C_P-SOH? It has been hypothesized that C_P-SOH is also protected in the active-site pocket and similarly shielded from further oxidation residing in the active site pocket as the sulfinic Cys⁵¹ does by forming a salt bridge with Arg¹²⁷.²² Yang et al. found that over-oxidation of Cys⁵¹ occurs during catalysis and correlates with increasing amounts of Trx.⁷ Thus, one could speculate that the binding of Trx to Prdx1-disulfides reduces Cys⁵¹-SOH, but then renders reduced Cys⁵¹-SH to destabilization and over-oxidation in the presence of H₂O₂, by inhibiting the sequestration of reduced Cys⁵¹ in the active-site pocket. Along those lines, Yang et al. further demonstrated that the rate of Prdx1 inactivation by H₂O₂ was increased in the range of 0.1–1 mM, despite an observed K_M for H₂O₂ <20 μM. These observations indicate that the initial oxidation of Cys⁵¹-SH to sulfenic acid is achieved by H₂O₂ attracted to the active-site pocket with an affinity constant of <20 μM, but all subsequent oxidation to Cys-SO₂H depends on increased collision with more H₂O₂ molecules.⁷ The above data illustrate one of the major differences between the peroxidase function of Prdxs and catalase: catalase decomposes H₂O₂ following an exponential decay, since its rate of H₂O₂ decomposition depends linearly on H₂O₂ concentration.²⁵ This suggests that under high amounts of H₂O₂, catalase decomposes H₂O₂

fast and efficiently, whereas Prdxs become readily over-oxidized in conditions of high H_2O_2 . Under conditions of low cellular H_2O_2 levels, Prdxs scavenge H_2O_2 more efficiently than catalase due to their high affinity towards H_2O_2 . Therefore, one could propose that depending on the cellular H_2O_2 load, Prdxs and catalase function rather sequentially than synergistically as peroxidases (Fig. 1). When focusing on the efficiency of H_2O_2 elimination, Prdxs may seem limited as peroxidases compared to catalase. However, this limitation offers an efficient switch panel for controlling signaling via H_2O_2 , since under increasing H_2O_2 -stress Prdx1 exhibits less complex formation with signaling partners, including the kinases c-Abl^{13,26} and JNK¹⁴ and the phosphatase PTEN¹¹ probably due to its own over-oxidation. Such loss of complex formation results in activation of c-Abl^{13,26} and JNK¹⁴ and inactivation of PTEN's phosphatase activity.¹¹

Regulation of 2-Cys Prdxs over-oxidation.

Recent studies have discovered that Prdxs have an additional function when over-oxidized, acting as molecular chaperones (reviewed most recently by Barranco-Medina et al.⁵). As discussed above, over-oxidation of 2-Cys Prdxs occurs only during catalysis in the presence of Trx, Trx-reductase and NADPH. It was previously believed that additional oxidation of C_P-SOH initiated degradation of the Prdxs, since it was not known if a system reducing the Cys-sulfinic or the sulfonic moiety exists. However, studies measuring the protein half life of sulfinylated Prdx1 and Prdx2 indicated reversibility of the sulfinic protein product,²⁷ suggesting a sulfinic acid reduction system may exist. Indeed, Biteau et al. identified in yeast a protein Srx, which reduced the sulfinic C_P of 2-Cys yeast Prdx (Tsa1) to C_P-SOH in a process which requires ATP hydrolysis, Mg²⁺ and thiol as an electron donor.²⁸

We now know that Srx regulates the chaperone function of 2-Cys Prdxs for the following reason: the molecular chaperone function for 2-Cys Prdxs is supposedly independent of its peroxidase activity and it correlates with the oligomerization status of the enzyme. Five homo-dimers of 2-Cys Prdxs form high molecular weight decamers, which can then further aggregate to form higher molecular complexes. It appears that a critical factor for determining the dimerdecamer equilibrium is the redox state of the catalytic cysteines, since the sulfinic enzymes favor decameric forms and the disulfide enzymes mainly prefer dimers. Under conditions of low H_2O_2 concentrations produced under normal cellular homeostasis, 2-Cys Prdxs form predominantly low-molecular weight oligomeric protein structures, which besides displaying peroxidase activity also protect proteins from degradation. However, following an unusual increase in H_2O_2 , the 2-Cys Prdxs experience structural changes in which the low molecular weight complexes convert into high molecular complexes and act as chaperones.^{10,29} This provides the mechanistic basis for Srx to modulate H_2O_2 -signaling in the cell by dissociating Prdxs high molecular weight complexes through reduction of Cys-sulfinic acid to Cys-sulfinic phosphoryl ester, which in turn becomes further reduced to a Cys-sulfenic acid by the Trx system.

H₂O₂ Modulates Cell Signaling through Oxidizing Prdx1

In order to better understand how H₂O₂ regulates signaling via Prdx1, we have to examine several aspects of H₂O₂ scavenging in the cell, by considering peroxidase activities of other peroxidases, localization of such peroxidases and availability of electron donors.

How Prdxs compare to other peroxidases in activity and localization.

Based on existing data, a direct biophysical and chemical comparison of H₂O₂ decomposition of each of the most common peroxidases (catalase, glutathione peroxidase (Gpx) and Prdx) in one study has not been done. So far each enzyme has been studied in individual studies making a direct comparison of Michaelis Menten kinetics difficult. However, reviewing the literature suggests that Prdx2 in erythrocytes possess a high affinity towards H₂O₂ since its oxidation occurred even in the presence of catalase, although its reaction rate with H₂O₂ was found comparable to catalase's: $1.3 \times 10^7 \text{ M}^{-1} \text{ s}^{-1}$. Catalase does not follow the Michaelis Menten constant since it decomposes H₂O₂ following an exponential decay, where its rate of H₂O₂ decomposition depends linearly on H₂O₂ concentration.²⁵ This suggests then that under high amounts of H₂O₂, catalase decomposes H₂O₂ fast and efficiently, whereas in conditions of low H₂O₂ Prdxs may scavenge H₂O₂ more efficiently due to their higher affinity for H₂O₂. Next to catalase and glutathione peroxidase, Prdx1 and 2 are the third most abundant antioxidant proteins in erythrocytes and play a crucial role in erythrocytic H₂O₂ removal, since we and others have shown, in contrast to catalase or glutathione peroxidase,^{30,31} loss of Prdx1 or 2 resulted in hemolytic anemia in mice.^{32,33} While one study suggested that Prdx2 is easily over-oxidized in those cells when exposed to excess H₂O₂,²⁵ another study found that erythrocytic Prdx2 appears mostly in a dimeric state. This was supported by the finding that TrxR activity is rather slow in red blood cells, compared to Jurkat cells where Prdx2 is easily found over-oxidized (sulfinic acid). The authors therefore suggested that Prdx2 is protected from over-oxidation since most of the protein is in a dimeric state and 2-Cys Prdxs function as non-catalytic scavengers of H₂O₂ in red blood cells.³⁴ These data exemplify how cell signaling is modulated by H₂O₂ through the availability of Prdxs scavenging systems. Thus when Prdxs sulfenic acids are not properly reduced Prdxs over-oxidation with a functional switch from peroxidases to chaperones is promoted.

Similar to glutathione peroxidases, Prdxs are present in all subcellular compartments, however reports on the localization of Prdxs in the subcellular compartments are inconsistent, which is most likely due to the difference of species, tissues, cells and methods used in the studies. Electron micrographs of rat hepatocytes showed that Prdx1 is not only localized to the cytoplasm and nucleus, but also to the matrixes of mitochondria and peroxisomes. This is an interesting finding, since Prdx1 seems amongst the Prdx-family to possess the widest cellular distribution and also, to display the highest abundance in various tissues.³⁵ Glutathione peroxidases are ubiquitously expressed with high levels in erythrocytes, liver and kidney, in which the enzyme is found in the mitochondria and the cytosol.³⁶ Catalase on the other hand is mainly found in the peroxisomes (together with Prdx1, GPX, MnSOD, Cu, Zn SOD, and other ROS decomposing enzymes)³⁷ whereas in transgenic catalase-overexpressing mice, it could also be found extraperoxisomal in the

cytosol and nucleus but not in the mitochondria.³⁸ Taken together, these findings demonstrate a clear difference in cellular localization of the main peroxidases in the cell. Important to point out is that cytosolic H₂O₂ can only be scavenged by catalase after diffusing into the peroxisomes. This may also explain that H₂O₂ removal is a sequential event as mentioned above. Growth factor induced signaling increases H₂O₂ levels in the cell via mitochondrial and NADPH oxidase activity. Superoxide induced by growth factor signaling is either spontaneously or enzymatically dismutated to oxygen and H₂O₂. Such spontaneous rise in H₂O₂ is presumably low and localized if the stimulation is short and is controlled by peroxiredoxins and glutathione peroxidases. If however, H₂O₂ production continues due to prolonged receptor signaling, as found for example in cancer cells, Prdxs may shift to chaperone function and catalase activity scavenges more of the cellular H₂O₂ since H₂O₂ diffuses then into the peroxisomes. In general, catalase peroxidase activity probably works best under circumstances of higher cellular H₂O₂, as found in cancer cells or after acute high cellular exposure to H₂O₂.

Which factors modulate H₂O₂-induced signaling?

All three main peroxidases depend functionally on NADPH while glutathione peroxidase and Prdxs additionally require reductases and a suitable oxidant. Reduction of Trx by TrxR is needed in order to reduce disulfide-bridged Prdxs and glutathione reductase requires NADPH to reduce oxidized glutathione. Catalase utilizes NADPH differently by binding to it and thereby stabilizing catalase's compound II. This in turn protects the enzyme from oxidation-induced inactivation,³⁹ confirming that cellular catalase activity towards H₂O₂ correlates with intracellular concentrations of NADPH.⁴⁰ Oxidation of glucose is the main source to generate either mitochondrial NADPH, as a product of the citric acid cycle in the mitochondrial matrix, or cytosolic and nuclear NADPH, as a product of the pentose phosphate pathway. Therefore, glucose availability is an important resource used to protect cells against H₂O₂-insult. This has been confirmed by experiments showing that glucose deprivation reduces the GSH/GSSG ratio through a decrease of NADPH levels.⁴¹

Several studies suggest that subcellular localization of reductant proteins influences cellular redox. For example, decreased protein expression of Trx, primarily in an oxidized, nuclear form, is found in exponentially growing cell cultures, compared to confluent quiescent cells in which Trx is primarily reduced and cytoplasmic.⁴² In addition, exponentially growing cells have higher overall H₂O₂-content and increased levels of GSSG. Along those lines, in proliferating cells (S + G₂/M phase) glutathione was primarily found nuclear, whereas in resting cells (G₁/G₀ phase) it was found mostly cytoplasmic.⁴³ Noh et al. have recently demonstrated that the Prdx sulfanyl reductase Srx translocates from the cytosol to the mitochondria in response to oxidative stress to reduce sulfinic acid of Prdx3,⁴⁴ thereby most likely leaving part of the cytosolic sulfinic Prdxs in a sulfinic state and prone to further oxidation or formation of oligomeric structures.

Prdx1 was originally identified as a serum-induced gene that was three-fold overexpressed in Ras-transformed breast epithelial cells compared to non-transformed cells.⁴⁵ It was subsequently found to have higher expression in S-phase of cells challenged with agents inducing ROS.⁴⁶ Interestingly, phosphorylation of Prdx1 on Thr 90 by Cdc2 substantially

decreases Prdx1 peroxidase activity, thereby elevating H₂O₂ towards the end of G₂, allowing mitosis to occur.⁴⁷ However at the same time, inactivation of Prdx1 presumably promotes H₂O₂-induced inactivation of the protein tyrosine phosphatases Cdc25.⁴⁸ Oxidized Cdc25 phosphatases form intramolecular disulfides, which in turn are readily reduced by Trx.⁴⁹ This emphasizes the interplay of H₂O₂ and antioxidants in normal cycling cells and appoints Prdx1 as an inhibitory regulator of the cell cycle. Since Ras transformed cells proliferate faster and have higher intracellular H₂O₂ levels due to induction of NADPH oxidases,⁵⁰ the question arises to whether Prdx1 is functioning as a cell cycle promoter or inhibitor in cancer cells. Given all the above and considering that nuclear Prdx1 levels are increased in Ras-transformed fibroblasts compared to untransformed cells (Neumann CA, et al. unpublished data), it is plausible to conclude that in cancer cells, Prdx1 may be “immune” to Cdc2 induced phosphorylation due to conformational unavailability of Thr90, or that higher levels of Prdx1 expression may outcompete a possible inactivation by Cdc2. Moreover, the higher levels of ROS in cancer cells (recently reviewed by Trachootham et al.⁵¹) could result in Prdx oligomerization and chaperone activity. Since Cdc25 activity is preserved in cancer cells, it could be speculated that Cdc25 phosphatases are chaperoned by Prdxs in cancer cells. The exact biology of Prdxs chaperone function needs yet to be determined, not only in the context of cancer, where presumably cellular H₂O₂ is chronically elevated, but also in the context of an acute H₂O₂ insult. In cancer cells, elevated H₂O₂ is probably due to continuous signaling of either ligand binding-independent growth factor receptor signaling or mutations rendering kinases hyperactive or phosphatases inactive. The question is, if under such circumstances, Prdxs are mostly oligomerized and if such high molecular structures can then still be reduced by Srx, or if Srx itself is then mostly inactivated due to high H₂O₂ levels?

In normal cells mitogenesis did not cause hyperoxidation of either Prdx1 or Prdx2, whereas H₂O₂-dependent cell cycle arrest led to hyperoxidation of Prdx2 resulting in 66 and 140 Kda complexes, independent of the H₂O₂ dosage used. Alternatively, over-oxidation of Prdx1 was H₂O₂-dependent and was not detectable at higher concentrations of H₂O₂, but its formation of high oligomeric structures (around 600 KDa) was, suggesting a distinct difference of those two proteins, although they share a nearly 90% homology.⁵² Along those lines, Lee et al. demonstrated that the degree of H₂O₂-induced inactivation was higher for Prdx1 compared to Prdx2. However, replacing Cys83, which is only present in Prdx1 and not in Prdx2, with serine desensitized Prdx1 to H₂O₂-induced inactivation, abolished its chaperone function and supported the further finding that Cys83 is required to form disulfide structures in oligo-decameric structures at the dimerdimer interphase. This explains then why Prdx1 is mostly present as decamers while Prdx2 and Prdx1^{Cys83Ser} are primarily found as dimers.⁵³ If Prdx1 is more susceptible to over-oxidation, followed by a molecular switch to function as chaperone, the question arises if its reduction by Srx is slower or less efficient compared to Prdx2. Data from Chevallet and al. may have answered that question by showing that after over-oxidizing Prdx1 by using *t*-butylhydroperoxide treatment, Prdx1 recovered via retro-reduction far slower than Prdx2 did.⁵⁴

Very recent evidence now adds a new layer of complexity to the regulation the peroxidase/chaperone function of Prdx1. Park and et al. showed that Srx catalyzes deglutathionylation of Prdx1. In this study Prdx1 was found glutathionylated on Cys 51, 83 and 172, only after

small dosage of H₂O₂. The H₂O₂ dosages chosen were below the one inducing oligomerization. Interestingly, H₂O₂ dosages needed to induce glutathionylation varied between cell lines, correlating with the Trx levels expressed. Moreover, while Srx catalyzed the de-glutathionylation of Cys 83 and 172, glutaredoxin 1 (Grx1) catalyzed the de-glutathionylation of Cys 51, adding another level of control in Prdx1 function.⁴⁴ These data suggest that regulation of Prdx1 activity is finely tuned and influenced by cellular H₂O₂ concentrations and the availability of its regulating enzymes such as Trx, Srx and Grx1 (summarized in Fig. 1).

How H₂O₂ and Prdx1 regulate cell signaling via interacting with signaling proteins.

Although over the past years all three major peroxidases have been found to interact with cellular signaling proteins, more binding partners have been identified for Prdx1 overall.^{11–17} More than 10 years ago, Prdx1 was identified by Wen et al. to regulate c-Abl tyrosine kinase activity by interacting with its SH-3 and kinase domains.¹³ Later, Prdx1 was also found to bind to JNK together with GSTpi, thereby suppressing radiation-induced JNK signaling.¹⁴ In both instances, it was evident that increased oxidative stress (H₂O₂ in the case of c-Abl and ionizing radiation in the case of JNK), inhibited these interactions and resulted in kinase activation. Although Kim and et al. speculated about a role of Prdx1 oxidation in regulating a loss of interaction with JNK, further evidence is still needed to understand exact details of the interaction, including stoichiometry and oxidation status of both binding partners. Prdx1, Gpx1,⁵⁵ and catalase⁵⁶ have also been found to associate directly with the c-Abl SH-3 domain. Since SH3 binding domains bind only one protein at a time, it would be interesting to know the biology determining the binding preference of c-Abl to Prdx1, Gpx1 or catalase. Of interest also is that Gpx1, like Prdx1, binds to c-Abl to inhibit its kinase activity and that this interaction is lost under H₂O₂-induced stress, resulting in activation of the c-Abl kinase.⁵⁵ Catalase on the other hand, binds to c-Abl only under stress and is a substrate of c-Abl, whereas Prdx1 or Gpx are not. Phosphorylation of catalase by c-Abl has been shown on four of its tyrosine residues, which then either contribute to catalase activation⁵⁶ or its degradation.⁵⁷ Taken together, c-Abl activity seems well regulated by peroxidases, which reflects its important role in stress signaling in normal as well as in cancer cells.

An additional binding partner of Prdx1 is c-Myc.⁵⁸ Prdx1 binds to the highly conserved Myc boxII, which is critically important for transformation and transcriptional activity. Prdx1-induced myc inhibition causes a broad but selective loss of c-Myc target gene regulation¹⁶ and Prdx1^{-/-}MEFs show evidence of c-Myc activation.¹² ASK-1, a kinase upstream of the JNK activating signaling cascade, interacts with Prdx1 after H₂O₂-induced stress.⁵⁹ However, it is not clear from this study if such interaction is inhibiting ASK-1 activity. Considering all this, it becomes clear that Prdx1 is a promiscuous binding partner of stress signaling proteins, thereby regulating stress signaling cascades on multiple levels. An exciting prospect for future studies is the importance of the catalytic cysteines in Prdx1 in the regulatory process of modulating kinases activity given the differences in H₂O₂-induced binding or non-binding of Prdx1 with the kinases discussed above. We and others have taken first steps to investigate this problem in more detail. Park et al. showed that Prdx1 interacts with the androgen receptor (AR), enhancing its trans-activation. Interestingly, this process

was independent of Prdx1 peroxidase activity suggesting that the AR is subject to Prdx1 chaperone function.¹⁵ The yeast protein Tsa1, corresponding to the human Prdx2, associates with ribosomes in an oxidation-dependent manner, since H₂O₂ treatment resulted in a loss of binding between Tsa1 WT, but not with a catalytically inactive Tsa1 (Tsa^{C47S}), suggesting that oxidation of Tsa1 results in (a) complex disruption of Tsa and the ribosomes and (b) a switch from the peroxidase to the chaperone function of Tsa1.⁶⁰

Peroxides, are known to modify protein tyrosine phosphatases (PTPs) by oxidizing the catalytic cysteine in their active site. PTP catalytic property depends on a thiolate anion of a low pK_a cysteine residue (pK_a 4.7–5.4) located in the conserved motif of its active site.^{61,62} This highly nucleophilic group makes the initial attack on the phosphate group of the substrate, but renders the PTP extremely susceptible to oxidation, which results in its inactivation. The PTP and tumor suppressor PTEN exhibits phosphatase activity towards inositol and tyrosine phosphates and is known to be similarly inactivated through H₂O₂-mediated oxidation.⁶³ A recent study showed that PTEN oxidation in stimulated macrophages resulted in the temporary inhibition of its phosphatase activity and in the downstream activation of Akt through its phosphorylation on Serine473.⁶⁴ Analysis of human recombinant PTEN revealed that two of the five cysteines in its N-terminal phosphatase domain (Cys71 and Cys124) form a disulfide after oxidation, which resulted in the transient inhibition of its phosphatase activity.⁶⁵ We have recently shown that Prdx1 associates with PTEN, protecting its lipid phosphatase from H₂O₂-induced inactivation.¹¹ Our data demonstrated that Prdx1 most likely interacts with PTEN as a monomer, since adding Prdx1 to a PTEN lipid phosphatase reaction challenged by H₂O₂ preserved PTEN's phosphatase activity fully when Prdx1 was added in a 1:1 ration and could not be further enhanced when additional Prdx was added. Treatment of cells with small amounts of H₂O₂ resulted in disruption of the complex and in oxidation of Prdx1. We think that Cys⁵¹ in Prdx1 regulates complex disruption for the following two reasons: (a) Testing the role of the catalytic Prdx1 cysteines showed that Prdx1^{C51S} bound stronger to PTEN than Prdx1^{C172S} or Prdx1^{C51,172S}; (b) PTEN inhibited Prdx1 peroxidase activity in the presence of Trx, which may suggest that PTEN hinders Cys⁵¹-SH from folding back into the shielding pocket, thereby promoting its exposure to further oxidation and more importantly, the disruption of the Prdx1:PTEN complex. Yang et al. also demonstrated that the rate of Prdx1 inactivation by H₂O₂ was increased in the range of 0.1–1 mM, despite an observed K_M for H₂O₂ <20 μM. These observations indicate that the initial oxidation of Cys⁵¹-SH to sulfenic acid is achieved by H₂O₂ attracted to the active-site pocket with an affinity constant <20 μM. All subsequent finding, since the amount of H₂O₂-inactivated PTEN in the presence of Prdx1 was independent of the H₂O₂ amount given (25 μM–500 μM H₂O₂).

Prdx1 is the first antioxidant protein reported protecting protein function from inactivation through interaction. Analysis of human recombinant PTEN revealed that two of the five cysteines in its N-terminal phosphatase domain (Cys71 and Cys124) form a disulfide bond after oxidation, which resulted in the transient inhibition of its phosphatase activity,^{63,65} which is essential for its tumor suppressive function.^{66,67} Heterozygous loss of germline PTEN, or temporal loss of somatic PTEN, in mice promotes tumor formation^{68,69} and is accompanied by enhanced cell proliferation, decreased cell sensitivity to apoptosis, and increased Akt kinase activity.^{69–71} Despite the resemblance of mouse PTEN deficiency with

human Cowden's disease,⁷² spontaneous forms of human breast cancer rarely exhibit loss of both PTEN alleles or other identifiable PTEN mutations.⁷³ Since ROS are believed to be abnormally high in cancer cells,⁷⁴ and present hyperactive Akt signaling,⁷⁵ posttranslational modification such as oxidation may contribute to a loss of PTEN tumor suppressive function.^{72,73,76} It is known that PTEN negatively regulates Akt activity via dephosphorylating phosphatidylinositol (3,4,5) triphosphates (PIP₃s), which are essential for the membrane recruitment and full activation of Akt.^{69,71} PTEN lipid phosphatase activity is at optimum after membrane binding via phosphatidylinositol (4,5) biphosphates (PIP₂s)⁷⁷ and plays an important role in tumor suppression.^{66,67} Along those lines, Akt1 ablation protects (MMTV)-ErbB2/neu, and MMTV-v-H-Ras mice from breast cancer initiation.^{78,79} Lastly, loss of Prdx1 in H-Ras and ErbB-2 transformed PTEN positive MEFs increased transformation, whereas in transformed PTEN-negative MEFs it did not.¹¹ This clearly demonstrated that Prdx1 tumor suppressive function is mainly achieved via PTEN regulation. Therefore it can be proposed that Prdx1 prevents Akt-driven transformation by protecting PTEN from oxidation-induced inactivation, since oxidation of Prdx1 dissociates the Prdx1:PTEN complex resulting in hyperactive Akt signaling and oncogenesis (Fig. 2).

However, many questions still remain unanswered: for example, is Prdx1 reducing or preventing formation of the intramolecular disulfide build in PTEN following H₂O₂-induced stress? Or, is the interaction of Prdx1 and PTEN restricted to a certain cellular compartment?

Concluding Remarks

Over the recent years H₂O₂ has been recognized as a second messenger modifying cell signaling via protein oxidation.² Yet, all aspects involved in regulating the production and elimination of H₂O₂ are still not fully understood. We acknowledge now that many different factors, such as sub cellular localization and expression levels of antioxidant systems can vary in different phase of the cell cycle and cell types, or that differences in individual peroxidase enzyme kinetics, as discussed here for catalase and Prdx1, can result rather in a sequential than synergistic H₂O₂-scavenging. Since Prdxs are distributed throughout the cell and not like catalase only reduced to one cell organelle, they are capable to scavenge H₂O₂ where it is produced, for example, by NADPH oxidases and the mitochondria. Since NADPH oxidases are in close proximity to growth factor receptors and given that Prdxs have a high susceptibility to inactivation by oxidation, Prdxs function as "fine tuner" of cellular H₂O₂-signaling. By doing so, they dissociate from their binding partners, which in turn activates or inactivates the activity of the binding partner.

Acknowledgements

The authors wish to thank the Neumann lab and Drs. Yefim Manevich and Scott Eblen for fruit ful discussions. This work was supported by grants from the NIEHS-K22 ES012985, ACSIRG-97-219-05, Claudia Adams Barr Award-DFCI. All (C.A.N.) and Abney Research Foundation-MUSC (J.C.).

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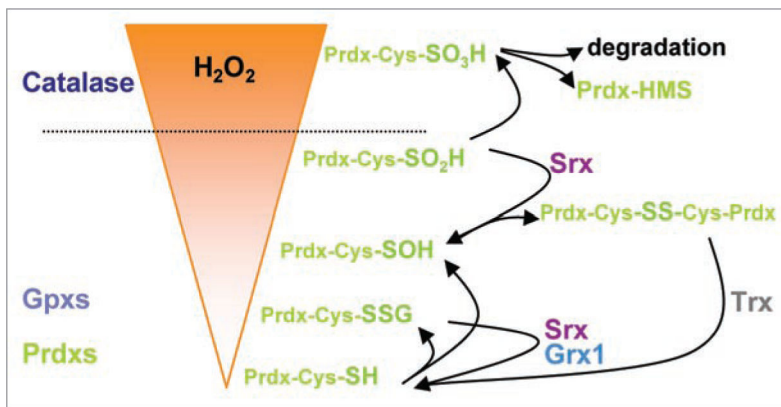


Figure 1. Cellular H_2O_2 levels are controlled sequentially by peroxidases. Prdxs and Gpxs scavenge smaller amounts of H_2O_2 whereas catalase catalyses higher amounts. Small amounts of H_2O_2 result in glutathionylation of Cys-thiols 51, 82 and 172 in Prdx1, which can be deglutathionylated by Srx and Grx1. The disulfide structure comprising sulfenic Prdx1 Cys⁵¹ and Cys¹⁷² from another Prdx1 protein can be reduced by Trx, Trx-reductase and NADPH. Further elevation of H_2O_2 promotes oxidation of Prdx1 Cys⁵¹ sulfenic acid to sulfinic acid. This process is reversible through an ATP and Mg^{2+} dependent reduction reaction induced by Srx. Oxidation of sulfinic Prdx1 Cys⁵¹ to sulfonic acid however is not reversible. Such over-oxidized Prdx1 proteins tend to form decamers with questionable peroxidase activity, but protein chaperone function. Such functional switch from peroxidase to chaperon elevates in turn cellular H_2O_2 , which is then scavenged by catalase. Compared to Prdxs, catalase is not readily over-oxidized and decomposes H_2O_2 following an exponential decay, since its rate of H_2O_2 decomposition depends linearly on H_2O_2 concentration.

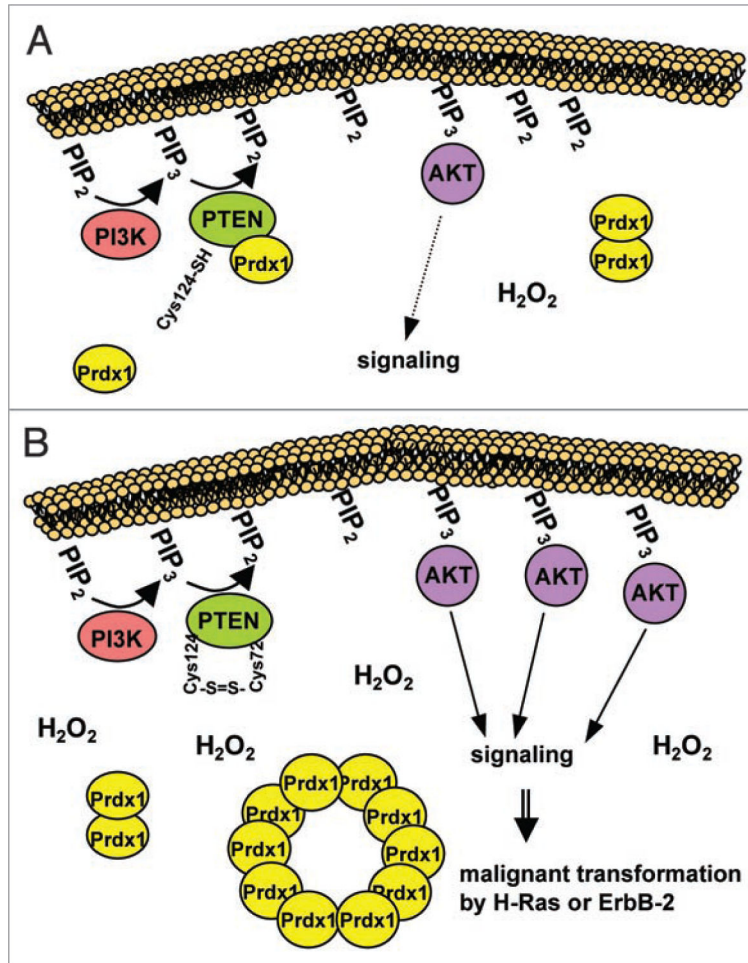


Figure 2. Prdx1 prevents Akt-driven tumorigenesis through protecting PTEN lipid phosphatase activity from oxidation-induced inactivation. (A) Prdx1 regulates PTEN phosphatase activity during oxidative stress, since binding of Prdx1 and PTEN occurs in conditions of mild or nil cellular stress. This constitutes a setting in which H₂O₂ is scavenged by Prdx1, which itself becomes in turn reversibly oxidized, in a controlled fashion. (B) However, under conditions of elevated oxidative stress, Prdxs are known to become irreversibly over-oxidized and dissociate from PTEN. Thereby, PTEN is inactivated by H₂O₂ resulting in hyperactivation of Akt. Hyperactive Akt then in turn can promote oncogenic signaling via ErbB-2- and Ras.