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Urea Derivatives in Modern Drug Discovery and Medicinal Chemistry

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Abstract

The urea functionality is inherent to numerous bioactive compounds, including a variety of clinically approved therapies. Urea containing compounds are increasingly used in medicinal chemistry and drug design in order to establish key drug-target interactions and fine-tune crucial drug-like properties. In this perspective, we highlight physicochemical and conformational properties of urea derivatives. We provide outlines of traditional reagents and chemical procedures for the preparation of ureas. Also, we discuss newly developed methodologies mainly aimed at overcoming safety issues associated with traditional synthesis. Finally, we provide a broad overview of urea-based medicinally relevant compounds, ranging from approved drugs to recent medicinal chemistry developments.

Graphical Abstract



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1. INTRODUCTION

The synthesis of urea by the German chemist Fredrich Wöhler in 1828 marked the beginning of organic chemistry.^{1,2} Since then, rapid evolution of organic chemistry and further developments in synthesis enabled medicinal chemistry and drug discovery in the latter half of the 20th century.^{3,4} Urea and its derivatives (**1** and **2**, Figure 1) have a central role in drug development and medicinal chemistry due to the capability of the urea functionality to form multiple stable hydrogen bonds with protein and receptor targets. Such drug-target interactions are responsible for specific biological activity, drug actions, and drug properties. It is not surprising that a large number of urea derivatives are utilized in a broad range of medicinal applications. In particular, a urea functionality is incorporated to modulate drug potency and selectivity and improve drug properties in the development of anticancer, antibacterial, anticonvulsive, anti-HIV, antidiabetic agents, and other medicinal compounds. ^{5,6} With the advances of protein structures and identification of new disease targets, there is a growing interest in urea-based derivatives for drug design and development.^{7,8}

One of the early examples of urea derivatives as a medicinal agent is the development of compound **3** by Bayer's laboratories in Germany. Urea derivative **3** is a colorless derivative of trypan red and has shown potent antitrypanosomal activity. Further optimization of urea compound **3** led to the discovery of suramin (**4**) with potent antitrypanosomal properties. Suramin is used as an effective therapy during the early stage of sleeping sickness in humans that is caused by the protozoan parasites *T. Gambiense* and *T. Rhodesiense*.^{9,10} Urea-derived Glibenclamide (**5**), also known as glyburide, is a potent antidiabetic drug that prolongs the hypoglycemic effect. It was used to treat patients with type II diabetes.¹¹ Today, there are many urea containing compounds that are FDA approved drugs for a variety of human diseases.^{12–15} We will outline these approved therapies in the next section.

Beyond their presence in approved drugs, oligoureas play a pivotal role as starting motifs for the generation of artificial β -sheets^{16,17} and peptidomimetics.^{18,19} Urea derivatives are also widely employed as linkers for the development of antibody-drug conjugates as well as in combinatorial chemistry building blocks.^{20,21} Due to the importance of urea derivatives in the syntheses of medicinal agents and other applications in material science and organocatalytic reactions, a variety of methods have been developed for their syntheses.^{22,23} Traditional syntheses of urea-containing compounds involve the reaction of amines with phosgene, carbon monoxide, or isocyanates. However, these synthetic procedures have relevant safety and environmental issues due to the use of toxic agents. Alternative routes involving reactions of amines with less toxic agents such as ethylene carbonate or diethyl carbonate have been developed. Also, many other direct routes to urea derivatives from amines and CO₂ in the presence of numerous catalysts were investigated. A number of ionic liquids were also examined as suitable solvents for the syntheses of urea derivatives.²⁴

In recent years, urea containing compounds have received much attention due to their growing application in drug design and medicinal chemistry.^{25,26} In the present perspective, we plan to outline physicochemical properties of urea derivatives, highlighting the key role of the urea functionality in the drug-target interaction. We outline the chemical methodologies for the synthesis of urea derivatives. We cover the most recent methodologies

that highlight the progress in terms of both process safety and efficiency. Finally, we provide a broad overview of the relevance of the urea functionality and urea derivatives in modern drug discovery and medicinal chemistry.

2. CONFORMATION OF UREA DERIVATIVES

The urea functionality shows a certain degree of conformational restriction due to the presence and delocalization of nonbonded electrons on nitrogens into the adjacent carbonyl group. Accordingly, three resonance structures can be drawn for ureas (namely A, B, and C; Figure 2).²⁷ Data on the X-ray structure of N,N'-diphenyl-N,N'-diethylurea revealed that the two urea nitrogens adopt a geometry between trigonal and tetrahedral. The amide groups displayed a nonplanar distortion of approximately 30° with amide C-N bond lengths of 1.37 Å.²⁷

Substitution on the nitrogen atoms of the urea moiety plays a key role on the conformational preferences of urea derivatives.²⁸ As shown in Figure 3, three conformations (**6a-c**) are possible for N,N'-diphenylureas. In solution and in solid state, they are generally characterized by a *trans, trans* conformation.^{29,30} Interestingly, sequential introduction of *N*-methyl group(s) to the free NHs of N,N'-diphenylurea prompted the shift from *trans, trans* (**7**) to *cis, cis* (**8a**) conformation. Thus, N,N'-diaryl-N,N'-dimethylurea exhibits intrinsic preference for a *cis, cis* conformation (**8a**) over *trans, cis* conformation (**8b**), characterized by the two aromatic portions located in a face-to-face arrangement, thus allowing π - π stacking interactions.²⁸

Extensive NMR studies and ultraviolet spectra analysis in solution on several substituted diphenylureas were performed to obtain insight into the stacking of the aromatic rings. Interestingly, the conformation from the X-ray crystallographic data and the conformations adopted in solution are in good agreement. These studies supported similar hybridization of the nitrogens and relative positions of the aromatic rings.^{31,32}

Theoretical studies suggest that the conformation showing the two aromatic rings with a face to face disposition in a mirror image relationship is disfavored as shown in Figure 4. The X-ray structural studies revealed a staggered relationship where the overlap of the HOMO of one ring with the LUMO of the other is enhanced (Figure 4).³¹

More recent studies on the conformational preferences of urea derivatives employing a combination of IR, NMR, and computational studies (DFT-D, M06–2X, quantum calculations, molecular dynamics) highlighted a more dynamic behavior of the N,N'-aryl ureas and N,N'-diaryl-N,N'-dialkyl ureas, although still in favor of the predominant isomer from the solid state studies.^{33,34} In studies with synthetic retinoids and cytokinins, the stereochemical alteration following *N*-methylation of aromatic amides was utilized in modulating the biological activity.³⁵ Similarly, stereochemical switching through *N*-methylation of ureas from a transoid structure to the preferred bis-*cis-N,N'*-dimethyl-N,N'-diphenylureas has been exploited. This strategy is particularly attractive in molecular design due to ease of introduction of the *N*-methylurea substructure in organic molecules. As shown in Figure 5, the N,N'-dimethyl-N,N'-dinapthylurea (**9**) and *N*-methyl derivative of poly-

(phenyleneureido)benzene (10) formed nice π -stacked aromatic arrays as observed based upon NMR and X-ray crystallographic studies.³⁶ This concept has been successfully applied to the construction of conformationally defined oligomers (foldamers) and dynamic helical structures as biomimetics.³⁷

Employing N,N'-diaryl ureas to relay stereochemical information over long distances, thus achieving stereochemical control over conformation, is another important feature of this functional group. As shown in Figure 6, reduction of N,N'-diaryl urea **11** containing a chiral sulfinyl group proceeded with excellent diastereoselectivity, providing alcohol **13** as the major diastereomer (dr 95:5), even though the chiral sulfinyl group is localized many bond lengths away. Presumably, the observed remote diastereoselectivity is due to the defined conformation of the chiral N,N'-diaryl urea **12** which resulted in the preferential nucleophilic attack from the *Re*-face of the carbonyl center affording alcohol **13**.³⁸

Oligomeric aromatic ureas characterized by N, N'-dimethylated urea moieties display a multilayered structure due to the (*cis,cis*)-urea arrangement. Such derivatives, containing a chiral center on the R2 group, display a dynamic helical architecture (all-*S* or all-*R* axis configuration) in solution when the phenyl moieties are connected through urea linkages at the *meta* positions. These dual dynamic helical and aromatic multilayered properties can be exploited to construct aromatic functional molecules with unique physicochemical properties.^{39,40}

3. PHYSICOCHEMICAL PROPERTIES OF UREAS

The oral bioavailability of a drug depends upon many factors including solubility, dissociation, permeability, first-pass metabolism, and efflux properties. Drug solubility in water is vitally important for drug absorption, bioavailability, and drug administration. The development of appropriate formulations for poorly soluble drugs is a major challenge in drug design and development. The structure of a drug molecule typically contains multiple functional groups which determine the overall hydrophobic or hydrophilic nature of the molecule. This hydrophobic/hydrophilic balance affects the capability of a given drug to cross biological membranes.⁴¹

The presence of a urea functionality plays an important role in a drug's aqueous solubility and permeability due to its dual nature as a hydrogen bond donor and acceptor. For a central nervous system (CNS) acting drug, a moderate level of lipophilicity would allow the drug to cross the blood-brain barrier (BBB) by a passive diffusion process. The hydrogen bonding capability⁴² in addition to ionization, polar surface area, and flexibility also strongly affects drug transport across the BBB.^{43,44}

Hydrotropic solubilization involves the addition of one solute to promote the solubility of another and is a strategy explored in pharmaceutical formulations. Hydrotropes are mostly aromatic or nonaromatic anions but can also be represented by neutral compounds, such as ureas.⁴⁵ Hydrotropic solubilization strategy was employed for the solubilization of nifedipine, a poorly soluble antihypertensive drug, by a series of urea analogues in an aqueous environment. The solubilizing effect ranking followed the trend butylurea >

ethylurea > methylurea > urea.⁴⁶ Sometimes, urea containing drugs show poor pharmacokinetic properties due to solubility and permeability issues. Many strategies have been developed to circumvent these issues. They are shown below.

3.1. Modulation of the Hydrogen Bonding Ability of Ureas.

In order to modulate hydrogen bonding capability, electron donating and electron withdrawing functionalities have been introduced on the substituents on the urea nitrogen atoms. The nature of aliphatic moieties placed on the urea nitrogen has been shown to affect self-association properties thus controlling drug solubility in nonpolar solvents.⁴⁶ A representative example is the N,N'-di(2,6-diisopropylphenyl)urea (DIPPU, **14**) which is the only diaryl urea that is soluble in carbon tetrachloride among eight other symmetrical diaryl ureas examined. DIPPU does not form a supramolecular polymer. When the branched alkyl moieties are not too large, the corresponding dialkylureas self-assemble to form supramolecular polymers in solution (Figure 7A).⁴⁷ However, DIPPU stabilizes the out-*trans* conformation due to large steric bulk of the alkyl chain on the aromatic rings. Therefore, the formation of dimers occurs through hydrogen bonding of the N-H groups in the out conformation as shown (Figure 7B).

3.2. Engagement of the Urea Moiety in the Formation of Intramolecular Hydrogen Bonding.

This strategy can be exploited to increase permeability and solubility with the formation of a transient or pseudoring structure. This can be achieved by engagement of a urea functionality in a novel monocyclic template that mimics a bicyclic structure.⁴⁸ As shown in Figure 8, based on the structure of a bicyclic kinase inhibitor (**15**), a series of pyrimidin-4-ylureas (**16**) were designed. This monocyclic structural template could function as suitable bioisostere due to its tendency to form resonance-assisted intramolecular hydrogen bonds.⁴⁹ The feasibility of this approach was demonstrated since this compound class exhibits multikinase inhibitory activity. A cocrystal structure revealed the hypothetical binding mode. Furthermore, NMR and theoretical studies substantiated the existence of an intramolecular hydrogen bond and its existence in water medium.⁵⁰

3.3. Disruption of the Planarity of Urea Derivatives.

In this approach, water solubility is enhanced by disrupting the molecular planarity of the solutes in order to reduce their crystal packing energy.⁵¹ The strategy involves the breaking of the urea symmetry by introducing a substituent on one of the urea nitrogens which disrupts planarity.^{52–54} A practical example is shown in Figure 9. *N*-Methyl-*N*-1-naphthyl urea (**18**) acts as a disruptor of cytokine-mediated STAT1 signaling in β -cells. Incorporation of a methyl group on the urea nitrogen of **17** provided *N*-methyl-*N*-1-naphthyl urea (**18**). The presence of a methyl group disrupts the existence of planar conformations, due to steric clash between hydrogens at C-2 or C-8 positions of the quinoline ring and the appended *N*-methyl group. Compound **18** showed a 110-fold solubility increase over compound **17**. This data also well correlated with a decreased melting point for compound **18** compared to compound **17** (~145 °C vs ~171 °C, respectively).⁵³

Another strategy to disrupt planarity is represented by the insertion of substituents at the *ortho* position of the *N*-aryl group of arylureas. Thus, the presence of two halogen atoms at the *ortho* position of the urea functionality could promote the formation of intermolecular hydrogen bonds due to a better conformational preorganization of the monomer. FT-IR and *ab initio* calculations revealed that an intramolecular hydrogen bond is formed when phenylureas are functionalized with chlorine or bromine atoms at the *ortho* positions. This halogen effect is at least in part due to the significant influence of the substituents on the dihedral angle (φ) between the urea functionality and the aromatic groups. As shown in Figure 10, in the absence of substituents in compound **19**, the most stable conformation for the compound is coplanar ($\varphi = 0^\circ$), while in the presence of chlorine or bromine substituents (compounds **20** and **21**), the energy surface is completely flat with φ values ranging from 60° to 120°.⁵⁵

3.4. Stacking Interactions of Ureas with Aromatic Rings.

Although the hydrogen bonding ability of the urea moiety has been long known and studied, this functionality can also be involved in other types of important interactions, especially those involving urea functionality and aromatic side chains in proteins.^{56,57} Several crystal structures of proteins bound to urea/urea derivatives have been identified to exhibit π -stacking-like favorable interactions. Moreover, perpendicular orientation of urea with respect to aromatic groups leads to NH- π type interactions. Such a complex hydrophilic/ hydrophobic combination of interactions involving the urea moiety could likely be exploited in the future design of ligand molecules for drug discovery purposes.⁵⁸

Several protein structures in the Protein Data Bank contain urea and urea derivatives as ligands or cosolvents. Quite interestingly, 38% of the structures containing urea and 25% of those containing urea derivatives were found to fall within the geometric requirements for a stacking arrangement with aromatic rings (e.g., PDB ID: 3IPU and 4EV9).^{59,60} The relevance of stacking interactions between urea and aromatic side chains in urea transporters (UTs) has been modeled by Priyakumar and co-workers.⁶¹ UTs are trimeric transmembrane proteins with a parallel arrangement of aromatic rings in the pore which allows stacking of urea and urea homologues (e.g., formamide, acetamide, and dimethylurea). This results in lowering of the energy barrier for urea based solute to transport.^{62,63}

The aromatic ring stacking interactions play an important role in urea-assisted protein denaturation. Using Trp-cage mini-protein, Priyakumar and co-workers⁶¹ investigated distortion of the protein hydrophobic core in the presence of urea and showed that aqueous urea optimally solvates aromatic groups in the protein. Furthermore, studies suggested the presence of stacking and NH- π interactions involving aromatic groups and urea. In particular, three kinds of interactions were shown between the Trp6 side-chain and urea: (i) stacking, where the urea moiety and indole ring were parallel to each other, (ii) NH- π interaction, where one of the NH bonds of the urea was perpendicular to the indole ring; (iii) typical hydrogen-bonding interaction between the urea oxygen atom and the nitrogen atom of the Trp side chain. Stacking and NH- π interactions with urea are responsible for stabilizing Trp6 in the unfolded state (Figure 11, PDB ID: 1L2Y).⁶¹

Urea-assisted RNA unfolding is another phenomenon which can be explained by the engagement of an extrahelical state via favorable π - π , NH- π , and hydrogen bonding interactions with the urea moiety. Both urea and water can establish NH- π and hydrogen bonding interactions with nucleic acid structures. However, stacking interactions are only possible with urea, and this explains its strong denaturing actions. These effects also rely upon the base composition and nature of nucleic acids.^{64,65}

4. IMPORTANCE OF UREA FUNCTIONALITY IN DRUG-RECEPTOR INTERACTION

The bioactivity of drugs is governed by molecular recognition through drug and target protein interactions. Among the multiple forces that are involved in protein-ligand interactions, hydrogen bonds play a key role due to their ability to stabilize the drug-receptor interaction.^{44,66} The donor-acceptor hydrogen bonding capability of urea derivatives is one of the most important elements of their molecular recognition and bioactivity. A few examples describing key interactions involving urea functionality in biologically active compounds are presented here.

A series of potent HIV protease inhibitors have been developed featuring a urea substructure. In 1993, Getman and co-workers reported a novel class of HIV-1 protease inhibitors incorporating the (hydroxyethyl)urea isostere. Compound **22** (SC-52151) exhibited good inhibitory activity on HIV-1 protease (IC₅₀ = 6.3 nM) and excellent selectivity over other aspartyl proteases, such as renin and cathepsin D. As shown in Figure 12, X-ray structural studies of a related inhibitor **23** containing a *n*-butyl chain complexed with HIV-1 protease revealed that the inhibitor binds in an extended conformation with the (*R*)-hydroxyl group positioned symmetrically between the two catalytic aspartates of the enzyme. The urea carbonyl group appears to be involved in water-mediated hydrogen bonding with the Ile50 and Ile50' residues. The P2' *n*-butyl substituent of the urea derivative is occupying the S1'-binding site, and the P1' isobutyl group has occupied the S2' subsite.⁶⁷

HIV protease inhibitors featuring a cyclic urea substructure have also been designed and developed into a number of preclinical candidates. One of the specific design objectives was to replace the structural water molecule found in cocrystal structures of HIV protease and linear inhibitors using the urea functionality. The resulting inhibitors maximized interactions in the protease active site and maintained good activity against mutant proteases.⁶⁸ Compound **24** (DMP450) and a number of other urea-based inhibitors exhibited broad spectrum antiviral activity against multidrug-resistant HIV-1 variants.⁶⁹ The cocrystal structure (PDB id: 1DMP) of DMP450-bound HIV-1 protease revealed that the urea functionality nicely mimicked the flap ligand-bridging water molecule (Figure 13). Also, the conformationally locked cyclic core allowed optimal interaction and accommodation of ligands within the S1, S1['], S2, and S2['] subpockets of the HIV protease active site.⁶⁸

The urea moiety also plays a prominent role in molecular recognition of sorafenib (**25**) and lenvatinib (**26**) by the protein kinase active site (Figures 14 and 15). The X-ray crystal structures of vascular endothelial growth factor receptor 2 (VEGFR2) in complex with either lenvatinib or sorafenib were determined. These protein-ligand complexes with lenvatinib and

sorafenib highlighted the importance of the urea functionality. As shown, the urea functionality is involved in critical hydrogen bonding with main-chain atoms of Asp1046 and the side-chain atoms of Glu885 (Figures 14 and 15).⁷⁰ The quinoline moiety of lenvatinib and the pyridine core of sorafenib

Urea derivative **27** is a potent Map kinase inhibitor that has undergone human clinical trials for the treatment of inflammatory diseases. Interestingly, the *N*-pyrazole-*N*-aryl urea occupies the binding domain on p38 that is exposed when the conserved binding loop containing Asp168, Phe169, and Gly170 adopts a DFG-out conformation (Figure 16). The X-ray structural studies revealed that both urea NHs form strong hydrogen bonds with the Glu71 side chain carboxylic acid while the urea carbonyl forms a strong hydrogen bond with the Asp168 backbone amide NH. The naphthalene moiety binds in the kinase specificity pocket. The ethoxymorpholine chain adopts a Gauche conformation and orients the morpholine oxygen to form a strong hydrogen bond with the backbone NH of Met109 in the ATP binding region of p38.⁶

Another example of the key role of urea functionality in drug-target interactions is shown in urea-derived soluble epoxide hydrolase (sEH) inhibitors. The sEH is involved in the metabolism of endogenous mediators (e.g., epoxides of linoleic acid, arachidonic acid, and other lipids).⁷¹ The catalytic mechanism of sEH as shown in Figure 17A, involves a SN2-type reaction in which the epoxide is first activated by forming hydrogen bonds with Tyr381 and/or Tyr465 residues. Subsequent nucleophilic attack by Asp333 forms the acylenzyme intermediate which is attacked by a water molecule activated by His523. The resulting tetrahedral intermediate finally collapses to provide the diol product (Figure 17A). 1,3-Disubstituted ureas have become interesting structural templates for sEH inhibition.^{72–75} The X-ray structural studies with the urea-derived inhibitor 4-(3-cyclohexylureido)-ethanoic acid (**28**) demonstrated that the urea moiety is able to mimic the transition state of the sEH-catalyzed epoxide ring opening, establishing strong hydrogen bonding interactions with Asp333 and Tyr381 residues (Figure 17B).^{76–78}

PF-04457845 (**29**) is a highly potent inhibitor of fatty acid amide hydrolase (FAAH).⁷⁹ FAAH is a serine hydrolase responsible for the metabolism of signaling lipids, such as the endocannabinoid anandamide.⁸⁰ FAAH is considered a potential therapeutic target for the treatment of pain and nervous system disorders. Thus, small-molecule FAAH inhibitors would exert their beneficial effects without the potential drawbacks associated with direct stimulation of cannabinoid receptors.^{81,82}

Compound **29** contains a benzylidenepiperidine pyridazine urea. The X-ray and kinetic studies demonstrated that these urea-containing derivatives are able to covalently and irreversibly inhibit FAAH through involvement of the Ser241-Ser217-Lys142 catalytic triad as shown in Figure 18. The urea functionality interacts with the catalytic Ser241 nucleophile, thus generating a tetrahedral intermediate (**30**) and sub-sequently inducing carbamoylation (inactivation) of the FAAH enzyme through formation of carbamate **31**.^{83,84} Recently, PF-04457845 has been shown to reduce cannabis withdrawal symptoms and cannabis use, in a phase IIa clinical trial.⁸⁵

5. UREA CONTAINING FDA APPROVED DRUGS

Urea substructures are characteristic of several FDA approved drugs as shown in Figure 1 and Figure 19. Many urea containing compounds are also undergoing clinical development which will be discussed later. In this section, approved urea derivatives for therapeutic applications are highlighted.

5.1. Sorafenib.

Sorafenib (25, Nexavar, Bayer and Onyx Pharmaceuticals, Figure 19) is a diaryl urea multikinase inhibitor acting on c-RAF, B-RAF, c-KIT, FLT3, platelet derived growth factor receptor (PDGFR) α and β , and vascular endothelial growth factor receptor (VEGFR) 1, 2, and 3, among others.¹² It was approved in 2007 for the therapy of hepatocellular carcinoma and advanced renal cell carcinoma and is under evaluation for acute myeloid leukemia (AML).^{13,86} Sorafenib is converted by UDP-glucuronosyltransferase 1A9 (UGT1A9) to its glucuronide and by CYP3A4 to its active metabolite sorafenib *N*-oxide.^{87,88}

5.2. Boceprevir.

Boceprevir (**32**, Victrelis, Merck, Figure 19) is a urea containing protease inhibitor useful against hepatitis C virus (HCV). Boceprevir reversibly binds the active site of nonstructural protein 3 (NS3) and displays high potency in the replicon system alone or combined with interferon a-2b and ribavirin.¹⁴ Boceprevir is a diastereomeric mixture of two compounds differing in the stereochemical configuration at the cyclobutylmethyl functionality appended to its ketoamide end. In January 2015, Merck announced market withdrawing for boceprevir due to the striking superiority of newer agents (e.g., ledipasvir/sofosbuvir). Oxidative metabolites of boceprevir are formed upon interaction with CYP3A4 and CYP3A5, while the keto-reduced metabolites mostly form upon transformations mediated by aldo-keto reductases AKR1C2 and AKR1C3. Since boceprevir metabolism encompasses two diverse enzymatic pathways, the drug undergoes drug-drug interaction to a lesser extent.⁸⁹

5.3. Ritonavir.

Ritonavir (**33**, Norvir, ABT-538, A-84538, AbbVie, Inc., Figure 19) features a thiazolyl methyl urea substructure. In 1996, it was approved as inhibitor of the HIV protease.⁹⁰ Ritonavir displayed an EC₅₀ of 0.025 μ M, plasma half-life of 1.2 h, and bioavailability of 78%. Ritonavir showed an excellent pharmacokinetic profile due to the higher stability of the thiazole moieties toward oxidative metabolism. Ritonavir inhibits CYP3A4 irreversibly binding to the heme iron via the thiazole nitrogen⁹¹ thus increasing plasma concentrations of other CYP3A4 substrate anti-HIV drugs, improving their efficacy.

5.4. Lisuride.

Lisuride (**34**, Dopergin, Proclacam, or Revanil, Figure 19) is a urea-based ergot-related dopamine agonist used for the therapy of Parkinson's disease (PD). Lisuride also binds the serotonin 5-HT1A, 5-HT2*A*/2C, and histamine H1 receptors.⁹² Studies aimed at examining the metabolism of 14C-lisuride hydrogen maleate in humans and rhesus monkeys demonstrated that 2-keto-3-hydroxy-lisuride was the main metabolic derivative. Parallel metabolic transformations of lisuride were identified, including hydroxylation of the phenyl

ring, oxidative *N*-deethylation, monooxygenation at C2 and C9, and oxidation of C2/C3 and C9/C10 double bonds.⁹³

5.5. Cariprazine.

Cariprazine (**35**, Vraylar in the United States and Reagila in Europe, Gedeon Richter and Actavis, Figure 19) is a urea-containing dopamine D3/D2 receptor partial agonist, acting as atypical antipsychotic for treating schizophrenia and bipolar disorders.^{94,95} Cariprazine received FDA approval on September 2015. It is highly metabolized by cytochrome CYP3A4 with the formation of active metabolites and, to a limited extent, by CYP2D6.⁹⁶ Cariprazine displays a mean half-life of 2–5 days (1.5–12.5 mg dose). It generates two clinically relevant metabolites by modification of its urea moiety: desmethyl-cariprazine and didesmethyl-cariprazine, this latter displaying an extended half-life with respect to cariprazine.⁹⁷

5.6. Telcagepant.

Telcagepant (**36**, code name MK-0974, Figure 19) was a urea containing oral calcitonin gene-related peptide (CGRP) receptor antagonist launched by Merck as an investigational drug for the treatment and prevention of migraine; it was the first orally available drug in this class.⁹⁸ A Phase IIa clinical trial evaluating telcagepant for the prophylaxis of episodic migraine was halted on March 26, 2009 after the observation of a significant increase of serum transaminase levels upon treatment.⁹⁹ MK-0974 undergoes significant oxidative metabolism (CYP3A) in monkey intestinal microsomes. The drug is mainly metabolized as pyridine *N*-oxide, and its metabolism is slower in rat than in monkey intestinal microsomes. ¹⁰⁰

5.7. Zileuton.

Zileuton (**37**, Zyflo, Cornerstone Therapeutics Inc., Figure 19) is a benzothiophene *N*-hydroxyurea. It is an inhibitor of 5-lipoxygenase (5-LOX), and it alleviates allergic and inflammatory states by suppressing leukotriene biosyn-thesis.^{101,102} Zileuton inhibits 5-LOX by coordinating the iron ion in the active site, and it also displays weak reducing properties. Zileuton and its *N*-dehydroxylated metabolite are oxidatively metabolized by CYP450 isoenzymes 1A2, 2C9, and 3A4. Interactions of zileuton with other drugs are related to inhibition of CYP1A2.¹⁰³

5.8. Celiprolol.

Celiprolol (**38**, brand names Cardem, Selectol, Celipres, Celipro, Celol, Cordiax, Dilanorm, Acer Therapeutics, Figure 19) is a β -blocker characterized by a unique pharmacologic profile: it behaves as a β 1-andrenoceptor antagonist with partial β 2 agonist activity. Based on these properties, it can be defined as a selective adrenoreceptor modulator with antihypertensive and antianginal properties.¹⁰⁴ Recently, the unique properties of celiprolol prompted investigation into its use for the treatment of a rare connective tissue disorder, namely Ehlers-Danlos syndrome.¹⁰⁵ Celiprolol is minimally metabolized, with only a very low percentage of a dose being excreted.¹⁰⁶

5.9. Carmofur.

Carmofur (**39**, HCFU, 1-hexylcarbamoyl-5-fluorouracil, Figure 19) is a pyrimidine analogue used as an antineoplastic agent, and it is an orally available lipophilic-masked derivative of 5-fluorouracil. Its carbamoyl moiety is cleaved *in vivo* to release 5-FU. Carmofur has been employed for the therapy of colorectal cancer,¹⁰⁷ although causing delayed leukoencephalopathy.¹⁰⁸ Significant side effects and no survival advantage in patients with stage II hepatocellular carcinoma led to clinical trial discontinuation.¹⁰⁹

5.10. Lenvatinib.

Lenvatinib (**26**, Figure 19) is a multi-kinase urea-based inhibitor targeting VEGF receptors 1–3, fibroblast growth factor receptors 1–4, PDGFR *a*, and RET and KIT proto-oncogenes. ^{110–112} Lenvatinib is approved for treating radioiodine-refractory differentiated thyroid cancer, and combined with Everolimus, it is used to treat advanced renal cell carcinoma. ¹¹³ Moreover, lenvatinib is under study for the treatment of hepatocellular carcinoma. The main metabolic products isolated from human liver microsomes derive from demethylation, decyclopropylation, *O*-dearylation, and *N*-oxidation. ¹¹⁴ CYP-related metabolism was reported in human, monkeys, dogs, and rats, while non-CYP-mediated metabolism, mainly ascribable to the action of aldehyde oxidase, has been found in monkeys and humans. ¹¹⁵

5.11. RG7112.

RG7112 (**40**, Figure 19) is a urea-containing small-molecule acting as an MDM2 antagonist. MDM2 downregulates the tumor suppressor p53.¹¹⁶ RG7112 underwent a phase I clinical trial in patients with hematologic malignancies. The main goal of the study was the estimation of the posology and safety profile. The secondary goals involved the assessment of pharmacokinetics, pharmacodynamics, and preliminary clinical efficacy.¹¹⁷ RG7112 is mainly metabolized by CYP3A4/5 and also behaves as a moderate inhibitor of CYP3A4/5 and P-glycoprotein (P-gp).¹¹⁸

6. PROCEDURES FOR THE SYNTHESIS OF UREA DERIVATIVES

The urea functionality is present in drug molecules, agro-chemicals, resins, and dyes. For this reason, a variety of synthetic methodologies for the preparation of urea derivatives have been developed (Scheme 1). The classical approach for the preparation of urea derivatives involves reagents such as phosgene or its derivative, triphosgene. Urea synthesis typically proceeds through an isocyanate intermediate, which upon reaction with a second amine forms the urea derivatives (Scheme 1). Over the years, a series of environmentally friendly phosgene substitutes were developed in order to overcome the hazardous nature of phosgene. In particular, carbonates, N,N'-carbonyldiimidazole, 1,1'-carbonylbisbenzotriazole, *S*-methylthiocarbamate, formamides, and chloroformates have been utilized for the synthesis of urea derivatives. Also, Curtius, Lossen, and Hofmann rearrangements have been employed to generate urea derivatives through the isocyanate intermediates.

More recently, carbon monoxide has been utilized as a reliable alternative to phosgene. Accordingly, catalytic oxidative carbonylation employs amines, an oxidant, and carbon

monoxide as starting materials to provide ureas. These reactions are appealing in terms of atom economy standards, with the only byproducts being the reduced form of the oxidant. Various transition metal catalysts including Pd, Co, Ni, Ru, Mn, and Au have been employed to afford urea derivatives. However, outcomes varied and often reaction conditions were harsh. As a result, complex mixtures of ureas, oxamides, and formamides were obtained.

6.1. Synthesis of Urea Derivatives Using Phosgene or Phosgene Equivalents.

The reaction of amines with phosgene represents the most traditional methodology for the generation of urea derivatives. This methodology is commonly preferred for the generation of symmetrical ureas, but also unsymmetrical urea derivatives can be prepared efficiently. In general, amines react with phosgene in the presence of a base to provide the desired isocyanate intermediates.^{119,120} Subsequent reactions of the isocyanates with diverse amine nucleophiles provide *N*,*N*-disubstituted or *N*,*N*,*N'*-trisubstituted unsymmetrical urea derivatives. This methodology is widely employed in drug discovery and pharmaceutical research since it represents a convenient method to generate urea derivatives.⁵ A synthesis of a triazole urea derivative containing a β -lactam ring is shown in Scheme 2. Beta-lactam derivative **41** was reacted with phosgene and 1,2,4-triazole (**42**) to provide triazole urea **43**, a potent and selective monoacylglycerol lipase (MAGL) inhibitor.¹²¹

Bis(trichloromethyl)carbonate (**46**, BTC, or triphosgene) is a crystalline and stable solid, allowing safe handling. For this reason, it has been considered a safer replacement to phosgene.⁵ However, the vapor pressure is sufficiently high to easily result in toxic concentrations, therefore safety concerns are also associated with its prolonged use.^{122,123} The use of triphosgene in the synthesis of urea derivative **47**, a potent antitrypanosomal agent, is shown in Scheme 3.¹²⁴ Aniline derivative **44** was reacted with BTC in the presence of Et₃N. The resulting intermediate was reacted with amine **45** to provide **47**.

6.2. Synthesis of Urea Derivatives via Safer Phosgene Substitutes.

N,N'-Carbonyldiimidazole (**50**, CDI) is a widely exploited alternative to phosgene reagent for the preparation of ureas. CDI is a crystalline solid and is commercially available. CDI does not produce chlorine or chlorinated byproducts and is thus a safer and less toxic alternative to the use of BTC and phosgene for the generation of biologically active unsymmetrical ureas.^{125,126} As shown in Scheme 4, CDI has been used for the synthesis of urea derivative **51**, a potent inhibitor of the DCN1-UBE2M interaction, starting from amine **48** and aniline **49**.¹²⁷

1,1'-Carbonylbisbenzotriazole (CBT, **53**) was proposed as a safer and milder alternative to phosgene for the preparation of unsymmetrical urea derivatives. As an example, compound **53** reacts with secondary amine **52** to provide the corresponding carbamoyl benzotriazole derivative **54**. Intermediate **54** is subsequently treated with a different secondary amine (**55**), to smoothly provide tetrasubstituted unsymmetrical urea derivative **56** (Scheme 5). Steric hindrance of the substituents placed on the secondary amine is the main factor influencing reaction outcome. For cyclic and aliphatic amine derivatives, the reaction proceeds at room temperature and provides good yields, while it may require higher temperatures or reflux condition for aromatic amines.¹²⁸

The aminolysis of a urethane derivative is a well-known procedure for the synthesis of ureas. Urethanes are usually generated from carbonates or chloroformates.^{129,130} Urethanes can also be prepared by reacting amine derivatives with carbon dioxide or by utilizing other safer and more recent methodologies.¹³¹ Urethanes react with amines with varying efficacy, depending upon the nature of the leaving alkoxide group.^{5,129,132}

As an example, addition of one equivalent of the indoline **58** to phenylcarbamate **57** in the presence of triethylamine followed by heating to reflux resulted in almost quantitative conversion to the urea **59** (Scheme 6). The product was isolated in 78% yield and >99% purity through the addition of water and collection by filtration.¹²⁹

For the effective synthesis of unsymmetrical tetrasubstituted ureas, Batey and co-workers formed imidazolium salts and developed effective carbamoyl transfer reagents.¹³³ As shown in Scheme 7, carbamoylimidazolium salts are readily synthesized by reaction of secondary amines with CDI and successive alkylation with iodomethane to form salt **61**.¹³¹ Addition of primary or secondary amines to these salts in the presence of triethylamine efficiently provided disubstituted or tetrasubstituted ureas, respectively. The resulting byproducts (namely *N*-methylimidazole and triethylamine hydrochloride salt) could be washed away by treatment of the organic layer with a dilute acid. The protocol was not successful for poorly nucleophilic aromatic amines. However, their conversion to anilide anions, upon treatment with KHMDS or *n*-BuLi led to more reactive nucleophiles. Reaction of these anions with imidazolium salts smoothly provided the corresponding ureas in good yields.¹³⁴

Artuso and co-workers validated *S*,*S*-dimethyl dithiocarbonate (DMDTC, **63**) as a valuable replacement of phosgene for the in-water carbonylation of amines for the convenient generation of mono-, di-, and trisubstituted urea derivatives (Scheme 8). Symmetrical disubstituted ureas were prepared using a 1:2 molar ratio of DMDTC and amine, respectively, at 60 °C. Unsymmetrical urea derivative **65** was synthesized in two steps involving: (i) formation of the *S*-methyl *N*-alkylthiocarbamate derivative **64** (room temperature); and (ii) reaction of this intermediate with butylamine (50–70 °C). This protocol provided urea derivatives in excellent yields and purity. The protocol allowed for easy and straightforward isolation of reaction products.¹³⁵

As part of safer phosgene substitutes to be exploited under mild reaction conditions, Yoon and co-workers examined the use of phenyl 4,5-dichloro-6-oxopyridazine-1(6*H*)-carboxylate (**68**). Compound **68** was synthesized from 4,5-dichloropyridazin-3(2*H*)-one (**66**) and phenyl chloroformate (**67**) in the presence of triethylamine (Scheme 9). The compound showed stability even at high temperature. Unsymmetrical ureas were synthesized through this protocol in a one pot fashion starting from compound **68**. Upon reaction of anilines with **68** to form the corresponding *N*-aryl carbamates, aliphatic amines were added and refluxed until disappearance of the carbamate to efficiently provide *N*-aryl-*N*-alkyl urea derivatives. 136

Padiya and co-workers described a useful protocol for synthesis of urea derivatives (Scheme 10) in water. It turns out, CDI reacts with amines in aqueous medium to afford the corresponding *N*-substituted carbonylimidazolides in good yields. Carbonylimidazolides

(such as compound **69**) derived from primary amines react in situ with an amine nucleophile (e.g., compound **70**) providing the corresponding urea **71**. Compounds **72** and **73** are other examples of urea derivatives which can be efficiently prepared through this procedure. Due to product precipitation from the reaction mixture, a high degree of purity was observed.¹³⁷

Recently, Moroz and co-workers investigated *bis*(2,2,2-trifluoroethyl) carbonate (**76**) as a convenient alternative to traditional carbonates. Carbonate **76** was prepared by reaction of triphosgene (**75**) and trifluoroethanol (**74**). The reaction provided **76** and its trifluoroethylchloroformate counterpart in a nearly 1:3 ratio. Moreover, trifluoroethylchloroformate could provide **76** by further treatment with trifluoroethanol and triethylamine (Scheme 11).

Trifluoroethyl carbonate **76** smoothly reacts with alkyl amines with high regioselectivity providing trifluoroethyl carbamates (**77**), rather than symmetrical ureas. Further reaction with amines in the presence of a base provided urea derivatives. This protocol displays a wide scope and utility, since it allowed the straightforward synthesis of ureas in the presence of potentially competing functionalities (e.g., amino, hydroxyl and phenolic groups, pyrazole moieties) (Scheme 11).¹³⁸

6.3. Urea Synthesis Using Rearrangements.

The Hofmann rearrangement is a common method to access ureas through the formation of their amine precursors. This reaction involves the generation of a primary amine from a carboxamide derivative in the presence of sodium hydroxide and bromine, encompassing the formation of an intermediate isocyanate. The intermediate isocyanate could be converted into urea derivatives by in situ trapping with a suitable amine nucleophile.¹³⁹ To improve the overall process, a variety of alternative oxidants and different bases arose over years, such as diacetoxyiodobenzene PhI(OAc)₂,¹⁴⁰ MeOBr,¹⁴¹ *N*-bromosuccinimide (NBS)-CH₃ONa,¹⁴² (NBS)-KOH,¹⁴³ lead(IV) acetate,¹⁴⁴ and benzyltrimethylammonium tribromide.¹⁴⁵ Hofmann rearrangement has been utilized in the synthesis of 2-pyridylureas as glucokinase activators. As shown in Scheme 12, carboxamide **78** was subjected to Hofmann rearrangement with NBS and potassium hydroxide to provide aniline **79**. It was subsequently converted to urea derivative **80**, precursor of the target compound **81**.¹⁴⁶

Curtius rearrangement of an acyl azide provides the corresponding isocyanate. This can undergo in situ reaction with a variety of nucleophiles, thus affording urea and carbamate derivatives or other *N*-acyl compounds. A one-pot Curtius rearrangement starting from carboxylic acid derivatives using diphenylphosphoryl azide (DPPA) provides the corresponding isocyanate derivatives for urea synthesis.^{147–150} The use of Curtius rearrangement for the synthesis of urea derivative **83** is shown in Scheme 13. Compound **83** is a potent inhibitor of human influenza virus replication. Carboxylic acid precursor **82** was reacted with DPPA and triethylamine, and subsequent reaction with morpholine provided urea derivative **83**.¹⁵¹

The Lossen rearrangement is the conversion of hydroxamic acids into isocyanates. The transformation is initiated by exposure of the hydroxamic acid to base or heat. The resulting anion is rapidly converted into the corresponding isocyanate. The isocyanate could then

easily provide the desired urea derivatives. As shown in Scheme 14, valine-derived hydroxamic acid **84** was treated with alanine methyl ester **85** in the presence of EDCI, providing the urea derivative **86** in 81% yield.¹⁵²

Dubé and collaborators explored the potential of an inventive CDI-mediated Lossen rearrangement. The proposed protocol is characterized by operational simplicity and mildness, leading to only imidazole and CO_2 as the stoichiometric byproducts (Scheme 15). The authors proved the electronic nature of the transformation and provided a wide exploration of the reaction scope. Morpholine trapping of isocyanates quantitatively affords the corresponding urea derivatives. Therefore, this nucleophile was employed to examine the outcome of the protocol employing hydroxamates deriving from aryl, heteroaryl, and aliphatic hydroxamic acids. Diverse ureas were generated in good yields and acceptable purity grades according to this protocol, which also represents a green alternative to traditional Curtius and Hofmann rearrangements.¹⁵³

Sureshbabu and co-workers reported the ultrasonication-mediated formation of hydroxamates from aromatic acids and *N*-protected amino acids using 1-propanephosphonic acid cyclic anhydride (**87**, T3P) as the carboxy activator as shown in Scheme 16. Compound **87** activates hydroxamates, to form isocyanates via a Lossen rearrangement, thus ultimately affording, upon reaction with suitable nucleophiles, the corresponding urea derivatives. The reported methodology was applied to the synthesis of several N_a -protected amino acid hydroxamates and for the preparation of peptidomimetics.¹⁵⁴

Mandal and co-workers proposed the use of 2-cyano-2-(4-

nitrophenylsulfonyloxyimino)acetate (**88**, 4-NBsOXY) for both the generation of hydroxamic acids and the subsequent Lossen rearrangement. As shown in Scheme 17, activation of the carboxylic acids with reagent **88** (1 equiv) in the presence of DIPEA and catalytic DMAP followed by the addition of hydroxylamine led to the corresponding hydroxamic acid derivatives. For amino acids, there was no detectable racemization. Urea derivatives were obtained via Lossen rearrangement from the generated hydroxamic acids by reaction of **88** (1 equiv) with DIPEA (0 °C for 90 min) and subsequent addition of amine at room temperature. Interestingly, the side products 4-nitrobenzenesulfonic acid and Oxyma [ethyl 2-cyano-2-(hydroxyimino)acetate] can be used to regenerate **88**.¹⁵⁵

Lebel and co-worker synthesized aromatic carbamates and urea derivatives from aromatic acids exploiting a Curtius rearrangement in the presence of di-*tert*-butyl dicarbonate or chloroformates and sodium azide as shown in Scheme 18. The reaction possibly involved the generation of an intermediate azidoformate. Final amine trapping of the acyl azides affords urea derivatives.¹⁵⁶

Sureshbabu and collaborators developed a simple one-pot protocol for the preparation of neoglycopeptides and urea-containing peptidomimetics employing a Curtius rearrangement in the presence of Deoxo-Fluor (**90**, *bis*(2-methoxyethyl)-aminosulfur trifluoride) and TMSN₃ under ultrasonication conditions as shown in Scheme 19. The method could involve four sequential steps: (i) generation of acyl fluoride, (ii) azide formation, (iii) rearrangement

into isocyanate, and (iv) reaction with the selected amine to provide urea derivatives. The conditions represent a racemization-free procedure.¹⁵⁷

Thakur and co-workers reported a simple protocol for the preparation of unsymmetrical ureas starting from (hetero)-aromatic carboxylic acids and amines by using DPPA as outlined in Scheme 20. This one-pot, microwave-accelerated method was reported to provide urea derivatives in an extremely rapid fashion (1–5 min) and in excellent yields. The method was applied to the synthesis of cannabinoid 1 and α 7 nicotinic receptor ligands (e.g., compound **94**, PSNCBAM-1).¹⁵⁸

Hu and co-workers reported a Hofmann rearrangement using iodosylbenzene (PhIO) as the oxidizing agent to form the corresponding one-carbon shorter isocyanates from amide derivatives as shown in Scheme 21. Nucleophilic attack of amines finally generated the corresponding urea or carbamate derivatives. Several amides and amines were employed as substrates. Symmetric and asymmetric ureas were also prepared in good to excellent yield. 159

Chien and co-workers proposed the Tienmann rearrangement with benzenesulfonyl chlorides (TsCl or *o*-NsCl) to generate the corresponding *N*-substituted cyanamide derivatives as shown in Scheme 22. Acidic hydrolysis of the cyanamide derivatives would then afford *N*-monosubstituted ureas. *N*-monosubstituted ureas could also be prepared from nitriles in a one-pot fashion, encompassing the transformation of nitriles into amidoximes and subsequent Tienmann rearrangement followed by hydrolysis.¹⁶⁰

Recently, Sanz and co-workers reported the synthesis of novel benzylidene urea derivatives upon reaction of *o*-lithiated carbamates with nitriles according to a $1,5-O \rightarrow N$ carbamoyl migration, namely a Snieckus-Fries-type rearrangement as outlined in Scheme 23. Several *O*-aryl *N*,*N*-diethylcarbamates were treated with (hetero)aromatic nitriles behaving as the electrophiles, affording the desired benzylidene ureas in high yields.¹⁶¹

6.4. Synthesis of Ureas via Metal-Catalyzed Methods.

Recently, many routes utilizing phosgene alternatives have been developed employing carbon dioxide or carbon monoxide as the carbonyl group source.^{162,163} Carbonylation of amines employing Pd-based catalysts has been broadly investigated since 1966, when it was first described.¹⁶⁴ In this procedure, Pd(II) is reduced to Pd(0). Its reoxidation to the active form of the catalyst is a crucial step which is facilitated by the use of oxidizing agents such as a molecular oxygen, aromatic nitrocompounds, iodine, and quinones. The use of molecular oxygen as the oxidizing agent became attractive due to both sustainability and atom economy perspectives as water is the only coproduct formed. Accordingly, several methods encompassing the use of molecular oxygen are increasingly appealing. Unfortunately, most of the protocols take place in potentially harmful conditions, since they use carbon monoxide and molecular oxygen in volume proportions very close to the flammability range.¹⁶⁵

Fukuoka and co-workers¹⁶⁶ and Chaudhari and co-workers¹⁶⁷ described the preparation of urea derivatives form alkylamines by exploiting oxidative carbonylation in the presence of

Pd/C and molecular oxygen, promoted by iodide salts. Similar results were obtained by Gabriele and co-workers, who synthesized linear and cyclic ureas in moderate to good yields utilizing PdI₂ and molecular oxygen.¹⁶⁸ Mechanistically, the different reactivities of amine derivatives were ascribed to their capability to form the isocyanate intermediates upon conversion of the carbamoylpalladium complex generated in pre-equilibrium with starting materials. Pd(0) is then reoxidized to Pd(II) by addition of I₂, which is in turn regenerated through oxygen-mediated oxidation of HI.¹⁶³

In addition to Pd-based catalysts, Au, Co, Ni, Rh, Ru, and other transition metals have also proven useful in affording urea derivatives.¹⁶³

As shown in Scheme 24, dicobalt octacarbonyl was used as a CO-free methodology to synthesize symmetrical and unsymmetrical urea derivatives under high-density microwave irradiation conditions. Reaction of cyclohexylamine **95** with dicobalt octacarbonyl led to the corresponding isocyanate derivative **96**, which, upon reaction with an excess of secondary amine **97**, provided trisubstituted unsymmetrical urea derivative **98** in moderate yields.^{5,169}

Nitroarenes show a great potential for the preparation of ureas via reductive carbonylation. Homogeneous catalysts such as Pd and Ru are generally required for these reactions. In 1975, Heck described the synthesis of *N*,*N*-diarylureas through reaction of nitroarenes and CO with anilines in the presence of Pd(II) salts, organic phosphines, a tertiary amine, and tetrabutylammonium chloride at 90 °C. A combination of symmetrical and unsymmetrical ureas as well as amines derived from the reduction of the initial nitroarenes were isolated. Subsequently, it was found that portionwise addition of amines during the reaction could be a suitable strategy to increase the yield of unsymmetrical aryl-alkyl ureas. As shown in Scheme 25, nitroarene **99** was converted into urea derivative **100** using catalytic amounts of Pd(OAc)₂ and dppp as ligand in the presence of acetic acid as the cocatalyst. The reaction yielded the desired urea in 98% conversion and 57% selectivity.¹⁷⁰

Pd-catalyzed carbonylation is a renowned strategy to synthesize compounds containing carbonyl functionalities. Accordingly, transition metal mediated reaction of amines with CO and an oxidant provides a useful approach to generate urea derivatives.¹⁷¹ Buchwald and coworkers investigated a one-pot protocol for the preparation of aryl isocyanates employing a Pd-catalyzed cross coupling between aryl chlorides and triflates in the presence of sodium cyanate as depicted in Scheme 26.¹⁷² The resulting aryl isocyanates were reacted with an amine nucleophile to provide unsymmetrical N,N'-di- and N,N,N'-trisubstituted ureas. The methodology offers a broad substrate scope. The reaction protocol was applied in the synthesis of Omecamtiv Mecarbil (**102**), a phase-II clinical trial activator of cardiac myosin.

Lemaire and co-workers developed a Pd-catalyzed procedure for the reductive alkylation of monosubstituted ureas using aldehydes and molecular hydrogen as a clean reducing agent as outlined in Scheme 27. The protocol used dry methanol, which behaves as a dehydrating agent and allows the generation of an imine from a hemiaminal precursor. Imine hydrogenation ultimately provides *N*,*N*-disubstituted urea derivatives. The reaction scope was investigated in optimized conditions (H₂ at 5 bar, 5% Pd/C (2.5 mol %), MeOH as the solvent, 100 °C) employing decanal and diverse urea derivatives. The desired *N*,*N*-

disubstituted ureas were smoothly formed in good to excellent yields. The methodology was also implemented in solvent-free conditions.⁹²

The metal-catalyzed carbonylation of nitrenes with amines is gaining interest for the synthesis of unsymmetrical ureas. Recently, Zhang and co-workers described a Pd/C-catalyzed carbonylation of benzyl, alkyl, and aryl azides with amines to synthesize unsymmetrical urea derivatives as described in Scheme 28. The use of Pd/C offers advantages compared to general homogeneous catalysts in terms of safe handling, simple removal of catalyst by filtration, and recycling potential. Notably, the use of a phosphorus ligand such as XPhos increased the reaction yields, and toluene was selected as the best performing solvent. Due to the higher reactivity of aryl azides, reaction could be implemented at room temperature. This cross-coupling protocol proceeded even in an aqueous media containing 5% PhMe and 5% nBu_4NCL^{171}

Mancuso and co-workers reported a catalytic system involving the use of PdI_2 in the presence of an excess of KI for the oxidative carbonylation of primary amines providing 1,2-disubstituted symmetric urea derivatives. The reaction with a mixture of a primary and a secondary amine leads to the formation of trisubstituted urea derivatives as outlined in Scheme 29. The protocol has been applied to synthesis of β -amino alcohols bearing a primary amine group for their conversion into the corresponding oxazolidinones. Conceivably, either the carba-moyl-palladium complex I or the isocyanate derivative II could provide the desired symmetric urea derivatives.¹⁷²

The use of an appropriate ionic liquid such as 1-butyl-3-methylimidazolium tetrafluoroborate, ([bmim][BF₄]) in the presence of the PdI₂/KI catalytic system has been shown to provide tetrasubstituted oxamides from secondary amines through an oxidative double carbonylation.¹⁷³ Also, carbonylation protocols employing a Pd(II)/Ag(I) catalytic system have been developed.¹⁷⁴ Wu and collaborators developed several carbonylation protocols utilizing formic acid^{175–180} or Mo(CO)₆ as a CO source.

The synthesis of unsymmetrical urea derivatives through a Pd/C-catalyzed carbonylation of nitroarenes using stoichio-metric $Mo(CO)_6$ is shown in Scheme $30.^{181-183}$ An NaI additive was crucial for the conversion, but its exact function is not clear. Also, the use of Pd(OAc)₂ in place of Pd/C provided slightly lower yields. Monodentate ligands performed better than bidentate ligands. From inexpensive and stable nitroarenes several unsymmetrical ureas, including the herbicide neburon (**103**), were synthesized in good yields. Beyond palladium, ruthenium is another metal of growing use for the synthesis of urea derivatives.

Hong and co-workers very recently reported the synthesis of urea derivatives using ruthenium pincer complex as shown in Scheme 31.¹⁸⁴ The method shows high atom economy and produces hydrogen as the only byproduct, and only low Ru catalyst loadings are required. The method was extended to the synthesis of unsymmetrical ureas via a sequential one-pot two-step protocol with initial generation of the formamide from an amine and methanol, and subsequent reaction of the formamide with a second amine derivative.¹⁸⁴

Recently, Gunanathan and co-workers also reported "amine-amide" metal-ligand cooperation using a ruthenium pincer complex with DMF as the carbon monoxide surrogate

as shown in Scheme 32. The use of 2 mol % of the catalyst **105** and 10 equiv of DMF provided the best reaction conditions for the generation of symmetrical arylmethyl-, arylalkyl-, and *N*,*N*-dialkylurea derivatives. Moreover, the protocol was extended to the preparation of unsymmetrical urea derivatives. After the generation of formamide derivatives, diverse amines were added and temperature was increased to efficiently provide the desired unsymmetrical urea derivatives.¹⁸⁵

Although Pd, Ru, and Rh have been the metals of choice for oxidative carbonylation, there are issues with the excessive cost of the catalysts and with the toxicity of the solvents. The use of other metals such as tungsten,¹⁸⁶ cobalt,¹⁸⁷ and copper-complex¹⁸⁸ have been explored for the synthesis of cyclic urethanes. As shown in Scheme 33, Nacci and coworkers developed a copper-based system for the synthesis of 2-oxazolidinones, urea derivatives, and urethanes in an aqueous medium without the use of additives. Copper(II) chloride was the optimum catalyst, and the reactions proceeded efficiently in water under homogeneous conditions at 100 °C and in approximately 4 h.¹⁸⁹

6.5. Synthesis of Urea Derivatives Using Urethanes and Isonitriles.

Syntheses of ureas starting from Boc-protected and Cbz-protected amines were carried out. As shown in Scheme 34, trifluoromethanesulfonyl anhydride was used to generate the isocyanate intermediate which provided the desired urea derivative upon reaction with a suitable amine.^{190,191}

The use of 2-substituted pyridines such as 2-chloropyridine (**106**) demonstrated high efficacy. In order to obtain satisfactory yields, the optimum amount of nucleophile is 3 equiv. The protocol was successfully applied to anilines, sterically hindered *tert*-butylamines, and secondary amines, affording trisubstituted ureas. Moreover, cyclic amines and noncyclic secondary amines led to high yields. The protocol was applied to chiral substrates with no racemization or epimerization.

Bana and co-workers designed a continuous-flow system involving two contiguous microreactors to synthesize non-symmetric ureas from Boc-protected amines under mild conditions as outlined in Scheme 35. In-line FT-IR was employed for monitoring the process. The setup was applied to the synthesis of a series of urea derivatives, and it was successfully used for the preparation of the active pharmaceutical ingredient cariprazine.¹⁹² Riesco-Domínguez and coworkers recently described the continuous flow synthesis of two libraries of urea derivatives containing a piperidin-4-one scaffold.¹⁹³

Ganem and co-workers developed a simple and effective protocol for the oxidation of isonitrile derivatives to isocyanates employing DMSO as the oxidizing agent and trifluoroacetic anhydride as the catalyst as outlined in Scheme 36. Isocyanate formation is completed within a few minutes, with dimethyl sulfide being the sole byproduct. The resulting isocyanates were used to synthesize urea derivatives.¹⁹⁴

Recently, Wang and co-workers reported the synthesis of urea derivatives employing oximes as recyclable starting materials by an intermolecular oxidative C-O bond formation between oxime derivatives and isocyanides as shown in Scheme 37.¹⁹⁵

1-Phenylethan-1-one oxime (**107**) and cyclohexyl isocyanide were used to set up optimum reaction conditions to form the corresponding carbamoyl oxime as shown. The use of 20% NIS in the presence of aqueous TBHP, with 1,4-dioxane as the solvent at 90 °C, represented the best reaction conditions.

Peterson and co-workers reported the synthesis of urea libraries starting from carbamic acids prepared from the DBU-catalyzed reaction of amines with gaseous carbon dioxide as outlined in Scheme 38. The desired urea derivatives were synthesized by reacting carbamic acids from primary amines with Mitsunobu reagents to generate isocyanates in situ, which then reacted with primary and secondary amines. The resulting isocyanates efficiently formed trisubstituted urea derivatives upon treatment with secondary amines.¹⁹⁶ The methodology was used successfully with benzylamine, and then extended to anilines.

7. RECENT DESIGN OF UREA DERIVATIVES IN DRUG DISCOVERY

As described earlier, urea functionality is a structural feature in a variety of FDA approved drugs. In this section, we cover design of urea-based compounds in medicinal chemistry and drug discovery in the past 10 years in a range of medicinal chemistry fields.

7.1. Urea-Containing Anticancer Agents.

Urea derivatives have been designed and synthesized to show broad biological properties, particularly potent antitumor activity. In particular, diarylurea derivatives have been designed as anticancer agents, targeting protein kinase or microtubules.

7.1.1. Kinase Inhibitors.—In recent years, a series of kinase inhibitors containing aryl urea substructures have been introduced in the market for the treatment of cancers or tested clinically. Among them, linifanib was approved for the therapy of colorectal cancer,¹⁹⁷ Sorafenib is used for advanced renal cell carcinoma,¹⁹⁸ and tandutinib is employed in the treatment of kidney cancer¹⁹⁹ and acute leukemia.²⁰⁰ Meanwhile, Gedatolisib (PKI-587) is in phase II trials for treatment of acute myeloid leukemia.²⁰¹ In recent years, a considerable effort has been directed toward the design of antitumor agents containing urea derivatives, especially in the fields of kinase inhibitors.

Epidermal growth factor receptor (EGFR) overexpression has been observed in many cancer types.²⁰² Homo- or heterodimerization are able to activate EGFR; subsequent autophosphorylation of tyrosine at the intracellular kinase domain stimulates Ras/Raf/ MAPK and the PI3K/AKt/mTOR pathways.^{203,204} Several small molecule EGFR inhibitors have been developed for antitumor therapy. EGFR inhibitors belonging to the second generation feature a Michael acceptor group and show higher selectivity compared to first generation inhibitors. They are more prone to overcoming drug resistance.

Mowafy and collaborators reported a series of inhibitors incorporating a urea moiety at the C-6 position of the quinazoline core that behaves as a hydrogen bond acceptor. A number of derivatives were potent EGFR inhibitors. 6-Ureidoanilinoquinazoline derivative **109** (Figure 20) displayed an IC₅₀ value of 0.061 μ M, showing remarkable growth inhibitory activity against tumor cells characterized by EGFR over-expression, namely nonsmall cell lung

cancer (EKVX NCIH322 M cell line), renal cancer (A498, TK-10 cell lines), and breast cancer (MDA-MB-468 cell line). Docking studies unveiled a binding mode similar to gefitinib and supported an extra interaction with Cys-773 residue at the kinase gate-keeper. 205

SAR studies show that replacement of the amino-2-cyanoacrylate with amino-methylene malonate resulted in increased potency for compound **108**. Further substitution with urea resulted in the very potent compound **109**. The urea moiety of compound **109** established a key hydrogen bonding interaction with Cys793 which is absent in gefitinib. This hydrogen bonding interaction may be responsible for the enhanced binding affinity and also for particular ligand positioning.

Further modification of the anilinoquinazoline scaffold containing a C-6 urea and a C-7 tetrahydrofuranyl ether led to compound **110**, which is a reversible inhibitor of EGFR. Compound **110** effectively inhibited EGFR and its autophosphorylation in A431 human epithelial carcinoma cells. Further studies later established that compound **109** behaved as an irreversible EGFR inhibitor.²⁰⁶

A series of potent multikinase inhibitors was designed utilizing established pharmacophores for kinase inhibition such as 4-anilinoquinazoline and unsymmetrical diarylurea derivatives. Among them, compound **111** (Figure 21) exhibited significant potency toward BRAF, BRAF V600E, EGFR, and VEGFR-2. Docking studies showed that the compound might nicely fit the DFG-out conformation of BRAF.²⁰⁷

Further modification on the aryl group resulted in diaryl ureas (**112**) as dual EGFR and VEGFR-2 inhibitors. The SAR studies established that a terminal diaryl urea moiety with a chlorine atom at the *ortho*-position of the urea group was optimal for high inhibitory potency. Compound **113** exhibited potent inhibition of EGFR ($IC_{50} = 1 \text{ nM}$) and VEGFR-2 ($IC_{50} = 79 \text{ nM}$), and good antiproliferative activities on HT-29 (colorectal adeno-carcinoma), H460 (lung cancer), and MCF-7 (breast cancer) cell lines. Docking studies suggested binding of **113** with the DFG-out conformation of VEGFR-2.²⁰⁸

Angiogenesis is a complex process triggered by the production of a range of pro-angiogenic and antiangiogenic factors.²⁰⁹ The altered equilibrium between these factors is implicated in the etiology of different diseases, such as cancer.²¹⁰ The discovery of compounds that are able to block the autophosphorylation of the pro-angiogenic VEGFR-2 has represented a major area of therapeutic intervention for the treatment of several cancers.²¹¹

Eldehna and co-workers developed 1-(4-((2-oxoindolin-3-ylidene)amino)phenyl)-3arylureas as novel agents that are potentially effective against HepG2 hepatocellular carcinoma through VEGFR-2 inhibition. Urea derivative **114** (Figure 22) emerged as the most active compound (IC₅₀ = 0.31 μ M), exhibiting good activity against HepG2 cells as well. Docking studies suggested the involvement of the urea linker in two key hydrogen bonding interactions.²¹²

Gao and co-workers reported a set of VEGFR-2 inhibitors containing a biphenyl urea and an oxime moiety. Urea derivative **115** showed potent enzyme inhibitory activity ($IC_{50} = 5.3$

nM) and excellent antiproliferative potency against several cancer cell lines (among them A549, MCF-7, and HT29 cells).²¹² Structurally, the salicylaldoxime portion allowed the generation of a pseudo six-membered ring formed through intramolecular hydrogen bond interaction between the oxime functionality and the adjacent hydroxyl group in order to mimic the planar quinazoline of known ATP-competitive inhibitors. The intra-molecular hydrogen bond stabilizes the planar conformation and confers structural rigidity. The conformational similarity of salicylaldoxime to quinazoline led to the hypothesis that both of them could act with the hinge region of VEGFR-2. Docking studies highlighted three crucial hydrogen bonding interactions of urea and salicylaldoxime moieties with VEGFR-2.

Further modification and introduction of aromatic-hetero-cyclic templates as hinge-binding fragments via a core-refining approach led to potent inhibitor **116**, which showed excellent VEGFR-2 inhibitory activity (IC₅₀ = 0.50 nM) and significant antiproliferative activity (A549 and SMMC-7721 cell lines).²⁰⁵

Ravez and co-workers reported multityrosine kinase inhibitors based on a 7-aminoalkoxy-4aryloxy-quinazoline urea derivative. Compound **117** (Figure 23) displayed one-digit nanomolar inhibition toward VEGFR-1 (IC₅₀ = 5 nM), VEGFR-2 (IC₅₀ = 4 nM), VEGFR-3 (IC₅₀ = 8 nM), PDGFR- β (IC₅₀ = 9 nM), and c-Kit (IC₅₀ = 9 nM). This compound also showed excellent activity on PC3, MCF7, and HT29 tumor cell lines and HUVEC (human umbilical vein endothelial cells). The potential of compound **117** to reduce the induction of a weblike network of capillary tubes and invasion was also evaluated in HUVEC cells.²¹³

Raf kinases, namely A-Raf, B-Raf, and C-Raf (Raf-1), play a crucial role in the MAPK cascade (Ras-Raf-MEK-ERK), thus regulating cell survival and proliferation.²¹⁴ In this context, BRaf represents an attractive target for cancer.²¹⁵ Substitution of a glutamic acid with a valine at residue 600 (V600E) represents the most common aberration in B-Raf (>90%).²¹⁶ This point mutation is mainly related to melanoma insurgence with a 60% frequency.²¹⁷

Yang and co-workers developed a series of bis-aryl ureas based upon a 2-amino-3purinylpyridine moiety as novel DFG-out BRaf^{V600E} inhibitors. Compound **118** displayed good inhibitory activity against B-Raf^{V600E} (IC₅₀ = 12 nM) and potent antiproliferative activities toward the melanoma A375 cell line (B-Raf^{V600E}, IC₅₀ = 27.27 μ M). The compound was further optimized and led to a subsequent series of bis-aryl amides active in an A375 xenograft mouse model.²¹⁸

El-Damasy and collaborators developed benzothiazole amides and ureas linked to a pyridylamide moiety by an aryl-ether tether at the 6-position of the benzothiazole ring. The best compound of the series was the 3,5-*bis*-trifluoromethylphenyl urea **119**, which effectively inhibited B-Raf^{V600E} and C-Raf with an IC₅₀ of 1.23 μ M and 0.566 μ M, respectively. The compound also showed antiproliferative activity on 57 human tumor cell lines. Docking studies indicated comparable binding mode with the two homologous kinases and the role of the urea moiety in forming crucial hydrogen bonding interactions, as well. Profiling on CYP450 and *h*ERG channel ruled out the possibility of drug-drug interactions and cardiac toxicity.²¹⁹

The same authors further designed 2-amido and ureido quinoline derivatives bearing a 2-*N*-methylamido-pyridin-4-yloxy group at the 5-position of quinoline. The 4-chloro-3-trifluoromethylphenyl urea derivative **120** was one of the best compounds of the series against BRAF^{V600E} (IC₅₀ values of 316 nM) and C-Raf kinases (IC₅₀ values of 61 nM). Compound **120** showed antiproliferative activity on several cancer cell lines such as RCC and NSCL.

Recently, researchers from Eli Lilly developed the urea-based clinical candidate LY3009120 (**121**, Figure 24).²²⁰ Compound **121** behaves as a type IIa kinase inhibitor toward A-Raf, B-Raf, and C-Raf kinases with an IC₅₀ of 44, 31–47, and 42 nM, respectively. Unlike more selective BRAF inhibitors, compound **121** displayed minimal paradoxical pathway activation in Raf wild-type cells. The compound was found to be active *in vivo* against B-Raf and K-Ras mutated colorectal cancer xenograft models.²²¹ The compound is currently under evaluation in phase I clinical trials.

FMS-like tyrosine kinase 3 (FLT3) belongs to the type III split-kinase domain family of receptor tyrosine kinases (RTKs).²²² Upon binding of FLT3 ligand to the receptor, dimerization and autophosphorylation occur, thus prompting FLT3 kinase activation and promoting proliferation of leukemic cells. Therefore, small molecule inhibitors of FLT3 have been explored as a viable therapeutic option for AML.²²³

Urea derivative **122** (Quizartinib, AC220) has been developed as an exceedingly potent and selective FLT3 inhibitor (IC₅₀ = 1.6 nM) with favorable drug properties and good tolerability in tumor xenografts. Compound **122** has also demonstrated a comparable efficacy and tolerability profile in humans and is presently in phase II clinical trials.^{224,225}

Increased levels of FLT3⁺ dendritic cells (DCs) are seen in human psoriatic lesions.²²⁶ Therefore, inhibition of FLT3 could be potentially exploited as a therapeutic option for psoriasis through interference with DCs. Recently, Li and colleagues developed compound **123** as a potent FLT3 inhibitor with an IC₅₀ value of 5 nM. *In vitro* studies indicated the ability of **123** to lower the production of DCs and reduce cytokine secretion of DCs. The compound demonstrated promising antipsoriatic effects *in vivo* in the murine K14-VEGF model, displaying no recurrence 15 days after the last administration.²²⁷

7.1.2. Microtubule-Targeting Agents.—Microtubules (MTs) are highly dynamic structures containing heterodimers of *a*- and β -tubulin. The MT polymerization and depolymerization balance plays a key role in mitosis and vesicular transport.²²⁸ Disruption and interference with MT dynamics would lead to mitotic catastrophe causing cell death.²²⁹ Several clinical anticancer agents target MTs, including some natural derivatives. Urea functionality has been incorporated in a variety of microtubule targeting agents (MTAs).

Song and co-workers synthesized a series of 3-haloacylamino benzoylurea derivatives as MTAs. 6-Fluoro derivative **124** (Figure 25) showed excellent potency toward nine human cancer cell lines, including Bel-7402 (hepatoma), MCF-7, DU-145, and PC-3 (prostate cancer) and LOVO (colon cancer), DND-1A (melanoma), and MIA Paca (pancreatic cancer) with IC₅₀ values ranging from 0.01 to 0.30 μ M. This compound showed a strong anticancer

effect against human hepatocarcinoma in a mouse model with over 86% tumor growth inhibition. The direct effects of **124** were assessed on purified tubulin, demonstrating that the compound selectively inhibited the process of MT assembly without affecting the disassembly process.²³⁰

Diaryl urea derivative **125** was reported as a dual acting compound displaying both receptor tyrosine kinase (Flt3, FGFR1, cKit, and PDGFR β) inhibitory activity and tubulin polymerization enhancer properties. The compound demonstrated antiproliferative effects in several cancer cell lines and inhibited tumor growth in LoVo and HT29 mice xenografts. Compound **125** showed significant antiangiogenic effect, and this may be due to combination of microtubule disruption and VEGFR inhibition.²³¹

Boger and co-workers reported several formerly inaccessible C20' urea derivatives of vinblastine with the aim of improving their affinity for tubulin and cellular activity. A very interesting finding was the steric tolerance for the size of a C20' substituent. The importance of the presence of a hydrogen bond donor on the C20' position was also demonstrated. Moreover, the insertion of a sterically demanding urea functionality is responsible for enhancing potency as much as 10-fold, and can be of great utility to modulate the physicochemical properties of these compounds. Representative compound 126, bearing the large biphenyl urea moiety, binds to tubulin with higher affinity than vinblastine. The compound also displayed good cell growth inhibitory activity against the HCT116 (human colon cancer) and HCT116/VM46 (resistant human colon cancer) cancer cell lines.²³² Gagné-Boulet and co-workers incorporated cyclic and acyclic urea substructures while developing combretastatin analogues. Representative compound 127, containing an imidazoline-2-one moiety and a trimethoxyphenyl group, displayed antiproliferative activity in the low nanomolar range on a variety of tumor cell lines (e.g., HT29, M21, MCF7, and HT1080). The compound efficiently arrested the cell cycle in the G2/M phase by binding to the colchicine-binding site and leading to the depolymerization of MTs.²³³

7.1.3. Urea-Derived Anticancer Agents with Unknown Mechanisms of

Actions.—Post-translational modifications of histones can influence chromatin architecture, thus holding a key importance in the regulation of gene expression. Histone acetyltransferases (HATs) and histone deacetylases (HDAC) act on histone lysine tails by attaching or removing acetyl groups, respectively.²³⁴ These processes play a key role in controlling cell division, cell-cell communication, survival, and invasion. Therefore, their inhibition is an attractive anticancer strategy.²³⁵

Kozikowski and co-workers developed a series of selective urea-based HDAC6 inhibitors. Compound **128** (Figure 26) shows potent inhibitory potency against HDAC6 with an IC₅₀ value of 5.02 nM and a selectivity of ~600-fold with respect to the HDAC1 isoform. This compound inhibited the growth of B16 melanoma cells.²³⁶

Schnekenburger and co-workers discovered urea derivative **129** as an HDAC class III sirtuins 1 and 2 (SIRT1/2) inhibitor with IC₅₀ values of 6.2 and 4.2 μ M on SIRT1 and 2, respectively. Compound **129** showed comparable potency and was evaluated on the NCI-60 cell line and exhibited promising antiproliferative activity on several cancer cell lines.

Moreover, neither mutation of p53 nor overexpression of MDR efflux pumps significantly reduced the anticancer activity of **129**. The therapeutic efficacy of **129** was confirmed in 3D glioblastoma spheroids and in zebrafish xenografts.²³⁷

Prostate-specific membrane antigen (PSMA) is a well-known biomarker for prostate cancer. ²³⁸ PSMA expression in prostate cancer is generally 10–100-fold higher than in healthy prostate tissue.²³⁹ Harada and co-workers developed unsymmetrical urea-containing compounds targeting PSMA through a nucleophilic conjugate addition between cysteineand maleimide-containing reagents. Compound **130** ([¹²³I]-IGLCE) displayed high affinity for PSMA, thus supporting the importance of the phenyl ring and succinimidyl moiety.²⁴⁰

Hepsin is a cell surface serine protease overexpressed in the vast majority of prostate tumors. ²⁴¹ PSMA and hepsin may constitute effective targets for imaging in prostate cancer since both of them feature an extracellular enzymatic active site. Azam and co-workers developed the heterobivalent urea containing compound **131**, targeting both hepsin and PSMA with IC_{50} values of 2.8 μ M and 28 nM, respectively. *In vitro* imaging studies demonstrated selective binding and retention to both PSMA and hepsin in cells highly expressing PC3/ML-PSMAHPN.²⁴²

7.2. Urea Derivatives for Neurological and Neuro-degenerative Diseases.

Urea derivatives are structural features in many compounds for the treatment of neurological and neurodegenerative diseases such as Parkinson disease (PD), Alzheimer's disease (AD), and many other CNS disorders. Azam and co-workers designed 3-phenyl/ethyl-2-thioxo-2,3-dihydrothiazolo[4,5-*d*]pyrimidin-7-yl urea derivatives that reduced haloperidol-induced catalepsy and oxidative stress in a PD animal model. Compound **132** (Figure 27), administered 30 min prior to haloperidol, was able to reduce oxidative stress, restore enzyme activities, and reinstate the levels of glutathione and malondialdehyde.²⁴³ Ghiron and co-workers developed urea containing selective agonists of *a*7 nicotinic acetylcholine receptor for possible treatment of schizophrenia and AD. Compound **133** displayed good *in vitro* profile. It was investigated in diverse *in vivo* behavioral models, showing efficacy after oral administration on perceptual processing and cognition and ability to reduce pharmacologically induced deficits.^{244,245} Further optimization of drug-like properties led to compound **134**, which displayed the desired activity/selectivity and showed a good preliminary ADME profile.²⁴⁶

Elkhamhawy and co-workers developed a series of quinazo-line urea derivatives that block β -amyloid peptide (A β)-mediated mitochondrial permeability transition pore (mPTP) opening as evaluated by JC-1 assay. Compound **135** (Figure 28) is the most active nonpeptidyl mPTP blocker possessing improved inhibitory profile with respect to the reference cyclosporin A (CsA), showing potential as a novel AD treatment. Compound **135** showed very low toxicity and excellent cell viability in an MTT assay.²⁴⁷

Recently, Hoang and co-workers developed human glutaminyl cyclase (QC) inhibitors as potent anti-AD agents. Based on the putative active conformation of a prototype inhibitor, novel *N*-substituted urea derivatives were developed as conformationally restricted analogues, potentially favoring a bent bioactive conformation (*Z*-*E* conformer). Among

them, compound **136** showed the best inhibitory profile on human QC ($IC_{50} = 6.1 \text{ nM}$) and favorable drug disposition properties.

When assayed in the 5XFAD mouse model, compound **136** effectively reduced the concentrations of pyroform $A\beta$ in the brain after 4 weeks of i.p. administration.²⁴⁸

Disfunction of beta 2-adrenergic receptor (β_2AR) has been recently related to AD and other neurodegenerative disorders, although the mechanisms involved are still poorly understood. Gaiser and co-workers recently developed bitopic urea-based ligands in order to target the orthosteric binding site (OBS) and a metastable binding site (MBS) on β_2AR . The compounds were designed on the basis of docking studies involving the antagonist (*S*)alprenolol into the OBS and a MBS. In general, the conceived ligands showed potency and affinity comparable to a (*S*)-alprenolol. Among them, bitopic ligand **137** demonstrated pIC₅₀ of 8.8 on β_2AR , as determined by a cell-based cAMP assay.²⁴⁹

Kurt and co-workers developed new chemotypes as multi-functional agents in AD. In particular, urea substituted benzofuranylthiazole derivatives were shown to inhibit AChE and BuChE and behave as antioxidant agents. Among them, compound **138** (Figure 29) showed the best inhibitory potency against BuChE with an IC₅₀ value of 2.03 μ M, a value 8.5-fold more potent than that of galanthamine. Also, a number of compounds showed good ABTS cation radical scavenging ability.²⁵⁰

Prakash and co-workers developed a series of urea derivatives as potential anticonvulsant agents. Compound **139** showed significant activity in the mouse model of epilepsy (MES) revealing protection in the electrically induced seizures at 30 mg/kg and 100 mg/kg 0.5 and 4 h after IP administration, respectively. This compound also displayed protective potential in the subcutaneous pentylenetetrazole model (scPTZ) (at 300 mg/kg at both time intervals), thus emerging as a promising wide-spectrum antiepileptic lead devoid of neurotoxic effects. ²⁵¹ Siddiqui and co-workers reported a series of urea derivatives with anticonvulsant activity. Compound **140** showed relevant anticonvulsant potency at a dose of 100 and 300 mg/kg in both screenings without displaying neurotoxic effects. ²⁵²

Recently, Amato and co-workers reported the diarylurea-containing compound **141** (VU0463841, Figure 30) as a potent and selective mGluR5 (IC₅₀ = 13 nM, other mGlus >30 μ M) negative allosteric modulator with favorable CNS penetration in rats. The efficacy of **141** was demonstrated in suitable rat models of cocaine addiction. In particular, IP dosing of **141** in rats led to therapeutically relevant brain exposures, with **141** being able to dose-dependently reduce cocaine self-administration and cocaine place preference.²⁵³ Xu and co-workers identified urea-based compound **142** as a mean CNS-penetrating CXCR2 antagonist. The compound demonstrated promising activity upon oral administration in a cuprizone model of demyelination, thus validating the role of CXCR2 as promising target for the treatment of multiple sclerosis.²⁵⁴

7.3. Urea-Containing Antiviral, Antibacterial, and Antiparasitic Compounds.

The urea functionality is a structural feature in a variety of antiviral, antibacterial, and antiparasitic agents. We provide recent examples of urea containing compounds in the design and development of these agents.

7.3.1. Antiviral Agents.—Duan and co-workers reported the design of CCR5 antagonists based on a cyclic urea structure. These derivatives were designed relying upon the rationale that five- and six-membered cyclic ureas would function as stable pharmacophores. The transition from acyclic to cyclic ureas not only substantially improved the chemical stability but also led to the development of compound **143** (Figure 31) that showed very good *in vivo* pharmacokinetic properties.²⁵⁵ A number of compounds showed high antiviral potency against HIV-1 in both human osteosarcoma (HOS) cell and peripheral blood lymphocyte (PBL) cell assays.

Kazmierski and co-workers developed urea-based potent HCV protease inhibitors. A number of derivatives, including **144**, established improved potency in HCV replicon assays with respect to second generation inhibitor, such as danoprevir. Oral administration of compound **144** showed favorable drug concentrations in rat liver, suggesting promising anti-HCV and pharmacokinetic profiles.²⁵⁶

7.3.2. Antibacterial Agents.—DNA gyrase and topoisomerase IV (topoIV) attracted great attention as targets for effective antibiotics.²⁵⁷ They are heterotetrameric type II topoisomerase enzymes that can modify chromosome structure through breaking and joining DNA double strands. These enzymes are also important for bacterial DNA replication.^{258,259} Charifson and co-workers developed benzimidazole ureas as dual targeting agents exerting simultaneous inhibition of the ATPase function of topoisomerase IV and gyrase. Ureaderivative **145** (Figure 32) showed an *E. coli* Gyrase K_i value of 6 nM and *S. aureus* TopoIV K_i value of 6 nM. Several related compounds displayed strong antibacterial activity toward clinically relevant, multidrug resistant strains. Inhibitor **145** possesses low *in vitro* spontaneous resistance frequencies and significant bactericidal activity. The compound was also shown to be effective upon oral and intravenous administration in mouse models of pneumonia and skin infections.²⁶⁰

Compound **145** behaved as a potent CYP3A4 inhibitor and displayed poor drug disposition properties. Subsequently, a number of second-generation aminobenzimidazole ureas were designed to address these issues. Compound **146** provided a dose- and time-dependent antibacterial effect. Also, investigation of metabolic properties of **146** showed that unlike compound **145**, metabolic degradation occurs away from the urea functionality.²⁶¹

Stokes and co-workers replaced the aminobenzimidazole substructure and developed benzothiazole ethyl urea derivatives such as compound **147** which inhibited both DNA gyrase (GyrB) and topoisomerase IV (ParE) with IC₅₀ values of <0.1 μ g/mL. The compound did not display affinity for human topoisomerase II. Moreover, no cytotoxicity toward HepG2 cells was observed. Compound **147** showed antibacterial activity against several Gram-positive bacterial strains with minimal inhibitory concentrations (MICs) ranging from

0.008 to 0.12 $\mu\rm{g}/\rm{mL}.$ However, it showed contrasting outcomes against Gram-negative organisms.^{262}

Yule and co-workers reported the synthesis and SAR of pyridine-3-carboxamide-6-yl ureas targeting DNA gyrase. Compound **148** (Figure 33) demonstrated broad-spectrum activity on Gram positive microorganisms and antibacterial activity against a Gram-negative efflux pump mutant and single-step GyrB/ParE mutants.²⁶³ Linezolid was the first FDA-approved oxazolidinone antibiotic approved in 2000, and tedizolid followed after (2014). Linezolid is used to treat serious Gram-positive infections such as those caused by vancomycin-resistant *Enterococcus faecalis* (VRE) and methicillin-resistant *Staphylococcus aureus* (MRSA).²⁶⁴ However, linezolid-resistant strains have been described.^{265–267} In an effort to develop more effective oxazolidinone antibiotics, Suzuky and co-workers reported urea-substituted [1,2,5]-triazepane or [1,2,5]oxadiazepine derivatives of linezolid. A number of derivatives showed potent antibacterial activity *in vitro*, and a subset underwent *in vivo* testing on a mouse model infected with *S. aureus* SR3637 strain. Compound **149** showed the best *in vivo* therapeutic effect. Furthermore, compound **149** displayed negligible monoamine oxidase (MAO)-inhibitory activity. The seven-membered heterocycles linked to a urea moiety appear to be a valuable structural template.²⁶⁸

De Rosa and co-workers also reported linezolid analogues.²⁶⁹ They performed molecular modeling studies in order to design derivatives with improved binding, particularly to ribosomes carrying mutations conferring partial or complete resistance to linezolid. In general, thiourea derivatives displayed better activities than their urea counterparts. Representative compound **150** showed a MIC higher than 100 μ g/mL.

7.3.3. Antiparasitic Agents.—Malaria is a parasitic disease caused by parasites of the genus *Plasmodium*, and it represents a major global health and economic problem. Although several drugs are used therapeutically to treat malaria, all have now succumbed to parasite drug resistance, therefore displaying significantly decreased efficacy.^{270–272}

Several alkylated (bis)urea polyamine derivatives were reported by Verlinden and coworkers.²⁷³ Compounds were tested for antimalarial potency against chloroquine (CQ)sensitive and CQ-resistant strains of *Plasmodium falciparum*. The majority of compounds are structural analogs of compound **151** (Figure 34). These compounds displayed inhibitory activity against *P. falciparum* in the 100–650 nM range. Compound **151** was active against CQ-resistant *P. falciparum*. The compounds in this class also showed excellent selectivity toward the parasite (>7000-fold lower IC₅₀ against *P. falciparum*) compared to the HepG2 mammalian cell line.

Recently, Singh and co-workers reported β -lactam-4-amino-quinoline conjugates containing urea/oxalamide linkers as antimalarial agents.²⁷⁴ The presence of a substituent on the *N*-1 nitrogen of the β -lactam ring, along with the linker structure, and the nature and length of the alkyl chain influenced the activity profiles as well as the physicochemical properties. Urea derivative **152** is the most potent compound and displayed an IC₅₀ of 42.4 nM against the CQ-resistant W2 strain of *P. falciparum*.

A series of urea derivatives based upon the thiaplakortone A tricyclic scaffold was reported by Schwartz and co-workers.²⁷⁵ Compounds were assayed for their *in vitro* antimalarial potential and mammalian cell toxicity. ADME profiling was also evaluated for several analogues. Compound **153** showed an IC₅₀ value of 644 nM against 3D7 strain. This compound was well tolerated in a mouse model when administered subcutaneously at 32 mg/kg twice daily for 4 days. At this dose, compound **153** also suppressed blood stage *P. berghei* by 26%. However, oral administration at the same dose did not suppress parasitemia, most likely due to poor oral bioavailability.

Human African trypanosomiasis (HAT) is a parasitic disease caused by *Trypanosoma brucei* gambiense or *Trypanosoma brucei* rhodesiense. The World Health Organization (WHO) currently counts 20000 cases with 65 million people at infection risk.²⁷⁶ There is an urgent need for innovative anti-HAT drugs.

Patrick and co-workers reported 2-(2-benzamido)ethyl-4-phenylthiazole derivative **154** (Figure 35), as a promising hit against *Trypanosoma brucei*.²⁷⁷ However, **154** displayed poor metabolic stability. Further optimization led to urea-containing compound **47**, which exhibited good *in vitro* potency with an EC₅₀ value of 34.8 nM. This compound did not show any significant toxicity toward CRL-8155 and HepG2 human cell lines. It showed excellent half-lives in human and mouse liver microsomes (at least 60 min). Compound **47** also demonstrated promising oral bioavailability and optimum brain penetration in mice. Following oral administration, compound **47** led to 5/5 cures in both acute and chronic HAT murine models.¹²⁴

Shibata and co-workers reported methionyl-tRNA synthetase (MetRS) inhibitors based on a urea structure and assessed their potential against HAT.²⁷⁸ Urea derivative **155** showed high preference for the parasitic enzyme with respect to the closest human homologue (IC₅₀ *T. brucei* MetRS = 57 nM, IC₅₀ human mitochondrial MetRS > 10 μ M). Compound **155** also inhibited parasite growth (EC₅₀ = 450 nM) while displaying low toxicity toward mammalian cells (CRL-8155 and HepG2). Furthermore, compound **155** showed CNS penetration in mice and also showed optimum membrane permeation ability when tested in the MDR1-MDCKII model.

8. CONCLUSION

The historic first synthesis of urea by Wöhler occurred nearly 200 years ago. Today, urea and its derivatives are substructures in numerous approved drugs. The many applications of urea derivatives in modern medicinal chemistry and drug development demonstrate the broad scope and utility of urea containing compounds. The urea functionality is quite unique, and it is involved in critical drug-target interactions and modulation of drug properties. In this perspective, we highlighted the role of the urea functionality in medicinal chemistry, provided an overview of physiochemical properties, and emphasized its involvement in the formation of donor-acceptor hydrogen bonding interactions with the target enzymes or receptors. We outlined the predominant role of the urea substructure in approved drugs and compounds with clinical potential. Furthermore, we showcased recent designs and applications of urea derivatives in the development of numerous anticancer and

anti-infective agents as well as agents for the potential treatment of neurodegenerative diseases. We hope that the present perspective will further stimulate the use of urea in modern medicinal chemistry and drug design.

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Biographies

Arun K. Ghosh obtained his BS and MS with honors in Chemistry from the University of Calcutta and Indian Institute of Technology, Kanpur, India, respectively. He received his Ph.D. in chemistry from the University of Pittsburgh (1985). He then carried out postdoctoral research in Professor E. J. Corey's laboratory at Harvard University (1985–1988). He was a research fellow at Merck prior to joining the chemistry faculty at the University of Illinois, Chicago as an Assistant Professor in 1994. In 2005, he joined the Department of Chemistry and Medicinal Chemistry at Purdue University, West Lafayette. In 2009, he became the Ian P. Rothwell Distinguished Professor of Chemistry and Medicinal Chemistry. His research interests are in broad areas of synthetic organic, bioorganic, and medicinal chemistry.

Margherita Brindisi graduated cum laude from the University of Siena, where she also received her Ph.D. in Pharmaceutical Sciences in 2008. In 2010–2011, she worked as a postdoctoral fellow in the group of Professor Arun Ghosh at Purdue University on the development of aspartyl protease inhibitors. From 2012 to 2015 she worked as a fixed-term Researcher at the Department of Biotechnology, Chemistry and Pharmacy at the University of Siena on the development of novel treatments for cancer, parasitic diseases, and brain disorders. In 2016–2017, she again moved to Purdue University as a Visiting Researcher. Dr. Brindisi currently holds a position as Assistant Professor at the Department of Excellence of Pharmacy at the University of Naples Federico II.

ABBREVIATIONS USED

| Aβ | amyloid beta |
|--------------------------|---|
| AD | Alzheimer disease |
| AKR | aldo-keto reductase |
| AML | acute myeloid leukemia |
| BBB | blood-brain barrier |
| [bmim][BF ₄] | 1-butyl-3-methylimidazolium tetrafluoroborate |
| втс | bis(trichloromethyl)carbonate |

| CCR5 | chemokine receptor 5 |
|--------|---|
| CDI | <i>N,N</i> '-carbonyldiimidazole |
| CGRP | calcitonin gene-related peptide |
| CNS | central nervous system |
| CQ | chloroquine |
| CsA | cyclosporin A |
| CYP450 | cytochrome P450 |
| DBAD | dibenzyl azodicarboxylate |
| DBU | 1,8-diazabicyclo(5.4.0)undec-7-ene |
| DC | dendritic cell |
| DCM | dichloromethane |
| DIPEA | N,N-diisopropylethylamine |
| DIPPU | <i>N,N</i> -di(2,6-diisopropylphenyl)urea |
| DMAP | 4-(dimethylamino)pyridine |
| DMF | dimethylformamide |
| DMDTC | S,S-dimethyldithiocarbonate |
| DMSO | dimethyl sulf-oxide |
| DNA | DNA |
| DPPA | diphenylphosphoryl azide |
| EDCI | 1-ethyl-3-(3-(dimethylamino)propyl)carbodiimide |
| sHE | soluble epoxide hydrolase |
| EGFR | epidermal growth factor receptor |
| FAAH | fatty acid amide hydrolase |
| FDA | Food & Drug Administration (US) |
| FLT3 | FMS-like tyrosine kinase 3 |
| FTIR | Fourier transform infrared |
| HATs | histone acetyltransferase |
| НАТ | Human African trypanosomiasis |
| HCV | hepatitis C virus |

| HDAC | histone deacetylases |
|----------|---|
| HIV | human immunodeficiency virus |
| НОМО | highest occupied molecular orbital |
| HOS | human osteosarcoma |
| HSDD | hypoactive sexual desire disorder |
| IR | infrared |
| KHMDS | potassium bis(trimethylsilyl)amide |
| LDA | lithium diisopropylamide |
| 5-LOX | 5-lipoxygenase |
| LUMO | lowest unoccupied molecular orbital |
| MAGL | moanoacylglycerol lipase |
| ΜΑΟ | monoamine oxidase |
| MDR | multidrug resistant |
| MES | maximal electroshock seizure |
| MetRS | methionyl-tRNA synthetase |
| MIC | minimal inhibitory concentration |
| MRSA | methicillin-resistant Staphylococcus aureus |
| МТ | micro-tubule |
| МТА | microtubule targeting agent |
| NBS | N-bromosuccinimide |
| 4-NBsOXY | 2-cyano-2-(4 -nitrophenylsulfonyloxyimino)acetate |
| NMR | nuclear magnetic resonance |
| NS3 | nonstructural protein 3 |
| PBL | peripheral blood lymphocytes |
| PD | Parkinson's disease |
| PDGFR | platelet derived growth factor receptor |
| P-gp | P-glycoprotein |
| PSMA | prostate-specific membrane antigen |
| РТР | permeability transition pore |

| RNA | ribonucleic acid |
|--------|---|
| RTK | receptor tyrosine kinase |
| SAR | structure-activity relationships |
| scPTZ | subcutaneous pentylenetetrazole |
| SET | single-electron transfer |
| SIRT | sirtuin |
| SPECT | single photon emission computed tomography |
| TBHP | tert-butyl hydroperoxide |
| TEA | triethylamine |
| TFA | trifluoroacetic acid |
| TFAA | trifluoroacetic anhydride |
| THF | tetrahydrofuran |
| T3P | 1-propanephosphonic acid cyclic anhydride |
| UGT1A9 | UDP-glucuronosyltransferase 1A9 |
| VEGFR2 | vascular endothelial growth factor receptor-2 |
| VRE | vancomycin-resistant Enterococcus faecalis |
| WHO | World Health Organization |

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Figure 1. Structures of urea and selected urea derivatives.



Figure 2. Possible resonance structures for the urea moiety.



Figure 3.

Conformations of N, N'-diphenyl urea, N-methyl-N, N'-diphenyl urea, and N, N'-dimethyl-N, N'-diphenyl urea.

















(A) Association pattern of N, N'-disubstituted ureas; (B) proposed association pattern for DIPPU (14).













Figure 10. Disruption of urea planarity by introduction of *ortho*substituents at the aryl urea system.



Figure 11.

(A) Representation of the stacking interaction between Trp6 and urea; (B) NH- π interaction between the tryptophan 6 side-chain and urea. Part of the Trp-cage miniprotein is shown in magenta; carbon atoms, nitrogen, and hydrogen atoms are shown in green, blue, and yellow, respectively. The figure is modified based upon published structures.⁶¹

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Figure 12.

Structure of HIV-1 protease inhibitor **23** and highlight of the interaction with protease. Hydrogen bonding interactions are shown by blue dotted lines.

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24 (DMP450)

Figure 13.

X-ray structure of urea derivative **24** and HIV-1 protease complex (pdb code: 1DMP). Hydrogen bonding interactions are shown by black dotted lines.





Figure 14.

X-ray structure of sorafenib (25) and VEGFR2 complex (pdb code: 4ASD). Hydrogen bonding interactions are shown by black dotted lines.





Figure 15.

X-ray structure of lenvatinib (26) and VEGFR2 complex (pdb code: 3WZD). Hydrogen bonding interactions are shown by black dotted lines.



Figure 16.

X-ray structure of MAP kinase inhibitor BIRB 796 (**27**) with human p38 MAP kinase (PDB code: 1KV2). Hydrogen bonding interactions are shown by black dotted lines.

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(A) Transition state for sEH-catalyzed oxirane ring opening; (B) Interaction of inhibitor **28** in the enzyme active site.



Figure 18. Mechanism of FAAH inhibition by urea derivative PF-04457845 (**29**).





Figure 19. Structures of urea containing FDA approved drugs.







Figure 21. Urea-containing multikinase inhibitors **111–113**.





Figure 22. Urea-containing VEGFR-2 inhibitors **114–116**.




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Figure 25. Urea-containing microtubule targeting agents 124–127.



Figure 26. Urea containing anticancer agents **128–131**.

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EC₅₀ (α7) = 0.22 μM

Figure 27. Urea derivatives 132–134 for neurodegenerative diseases.





Figure 28. Urea derivatives 135–137 for neurodegenerative diseases.

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141

IC₅₀ (mGluR5) = 13 nM



pIC₅₀ (CXCR2) = 9.3 nM

Figure 30. Urea derivatives 141 and 142 for neurodegenerative diseases.







 K_i (S. aureus TopolV) = 6 nM



146

K_i (*E. coli* Gyrase) = 9 nM K_i (*S. aureus* TopolV) < 12 nM



Figure 32. Urea-derivatives 145–147 as antibacterial agents.

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MIC (*S. aureus*) > 100 μg/mL

150

0

Figure 33. Urea-containing antibacterial agents **148–150**.



Figure 34. Urea-containing antimalarial agents **151–153**.



Scheme 1. Traditional Methodologies for the Synthesis of Ureas

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Scheme 4. Formation of Urea Derivative 51 Using CDI (50)









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Scheme 7. Synthesis of Urea Derivatives Using Carbamoylimidazolium Salts









Scheme 9.

Synthesis of Urea Derivatives Using Phenyl 4,5-Dichloro-6-oxopyridazine-1(6*H*)-carboxylate (68)







Scheme 11. Synthesis of Urea Derivatives Using *Bis*(2,2,2-trifluoroethyl) Carbonate 76











Scheme 14. Formation of Urea Derivative 86 Using the Lossen Rearrangement



Scheme 15. Synthesis of Urea Derivatives Using CDI-Mediated Lossen Rearrangement

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Scheme 16.

Synthesis of Urea Derivatives Using a Lossen Rearrangement Mediated by 1-Propanephosphonic Acid Cyclic Anhydride (87, T3P)





Synthesis of Urea Derivatives Using a Lossen Rearrangement Mediated by 2-Cyano-2-(4nitrophenylsulfonyloxyimino)acetate (88, 4-NBsOXY)

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Scheme 18. Synthesis of Urea Derivatives Using Curtius Rearrangement via Formation of the Azidoformate

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PG = Fmoc, Cbz, Boc











Scheme 20. Synthesis of Urea Derivatives Using Curtius Rearrangement under Microwave Conditions



Scheme 21. PhIO-Induced Hofmann Rearrangement



Scheme 22. Synthesis of Urea Derivatives Using Tienmann Rearrangement

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Scheme 23. Synthesis of Urea Derivatives Using Snieckus-Fries-Type Rearrangement



Scheme 24. Dicobalt Octacarbonyl Promoted Generation of Urea Derivatives



Scheme 25. Generation of Urea Derivatives from Nitroarenes


Scheme 26.

Synthesis of Urea Derivatives Using Cross Coupling of Aryl Chlorides and Triflates with Sodium Cyanate in the Presence of Ligand 101





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Scheme 28.

Pd/C-Catalyzed Carbonylation of Benzyl, Alkyl, and Aryl Azides with XPhos ^{*a*}PhMe, 60 °C, 12 h. ^{*b*}5% *n*Bu₄NCl, H₂O/PhMe 20:1, rt, 24 h.



Scheme 29. Synthesis of Urea Derivatives by Oxidative Carbonylation Using PdI_2 and KI

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Scheme 30. Synthesis of Urea Derivatives by Carbonylation with Mo(CO)₆



104, Ru-MACHO-BH[Ru]















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Scheme 36. Synthesis of Ureas from Isonitriles via the Intermediate Isocyanates





Scheme 37. Synthesis of Urea Derivatives Using Oximes





Scheme 38. Synthesis of Urea Derivatives via Carbamic Acids





Figure 35. Urea-containing antitripanosomal agents **47**, **154**, and **155**.