

## Design and Synthesis of 1,2-Deoxy-pyranose Derivatives of Spliceostatin A toward Prostate Cancer Treatment

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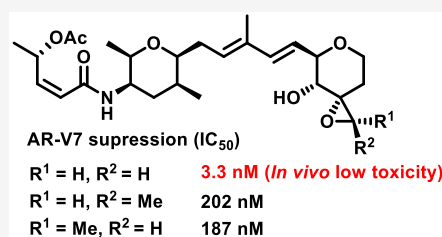
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**ABSTRACT:** We designed and synthesized a novel 1,2-deoxy-pyranose and terminal epoxide methyl substituted derivatives of spliceostatin A using Julia–Kocienski olefination as a key step. With respect to the biological activity, the 1,2-deoxy-pyranose analogue of spliceostatin A suppressed AR-V7 expression at the nano level ( $IC_{50} = 3.3$  nM). In addition, the *in vivo* toxicity test showed that the 1,2-deoxy-pyranose analogue was able to avoid severe toxicity compared to spliceostatin A.



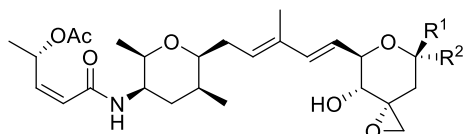
**KEYWORDS:** Prostate cancer, splicing, splicing variant, spliceostatin A, toxicity

In men, one of most common cancers is prostate cancer, which is treated with androgen deprivation therapy (ADT). However, after ADT, most of them develop castration-resistant prostate cancer (CRPC). These CRPC patients are then treated with hormonal therapies using androgen receptor (AR) signaling pathway inhibitors such as enzalutamide or abiraterone, but many CRPC patients become resistant to these medicines after a few years.<sup>1–3</sup> One of the reasons for the resistance is the expression of the splicing variant of AR, specifically AR-V7, which gets activated constitutively without hormone ligands and promotes prostate cancer growth.<sup>4,5</sup> Therefore, the development of inhibitors targeting AR-V7 expression is important for drug resistant CRPC therapy, and, to this end, some inhibitors have been developed (SF3b complex inhibitor and ROR- $\gamma$  antagonist).<sup>6–10</sup>

Previously, it was discovered that the target protein of the splicing modulators spliceostatin A (**1**) (Figure 1), pladienolide B, and GEX1A (herboxidiene) (Scheme S1), which showed strong antitumor activity toward solid tumors in mice and several cancer cells, was an SF3b complex in U2SnRNP (main factor of pre-mRNA splicing).<sup>11–19</sup> Until now, several synthetic studies (including analog synthesis) and mRNA splicing inhibition activity studies of **1**, pladienolide B, and GEX1A have been conducted by several research groups,<sup>20–37</sup> but there are only a few reports of splicing modulators which can inhibit the expression of AR-V7.

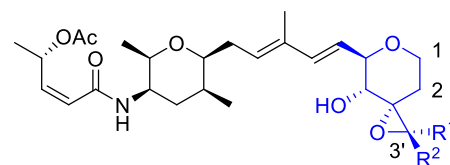
Recently, our group and the Hsieh group discovered that **1** and its relative pladienolide B and thailanstatins (Scheme S1) can inhibit the expression of AR-V7.<sup>6–8</sup> Specifically pladienolide B significantly reduced the tumor volume of CRPC model mice overexpressing AR-V7.<sup>7</sup> However, pladienolide B analogue E7107 entered phase I clinical trials, and as a result some patients suffered from vision loss.<sup>38,39</sup> Consequently, we

Previous work



- 1: Spliceostatin A ( $R^1 = Me, R^2 = OMe$ )
- 2: Phenyl C-glycoside derivative ( $R^1 = H, R^2 = Ph$ )

This work



- 3: 1,2-Deoxy-pyranose derivative ( $R^1 = H, R^2 = H$ )
- 4: 3'R Methyl epoxide derivative ( $R^1 = H, R^2 = Me$ )
- 5: 3'S Methyl epoxide derivative ( $R^1 = Me, R^2 = H$ )

**Figure 1.** Structure of spliceostatin A and its derivatives (1–5).

focused our research on **1** and its derivatives.<sup>40</sup> Very recently, we designed and synthesized **2** which was more easily synthesized and acid stable than **1** but had weak AR-V7 suppression inhibitory activity ( $IC_{50} = 132$  nM).<sup>8</sup> To discover

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easily synthesizable derivatives of **1** with potent AR-V7 suppression inhibitory activity, we designed and synthesized new derivatives **3–5** based on much more detailed theoretical considerations.<sup>41</sup>

Our considerations in designing derivatives **3–5** are shown in Figure 2.

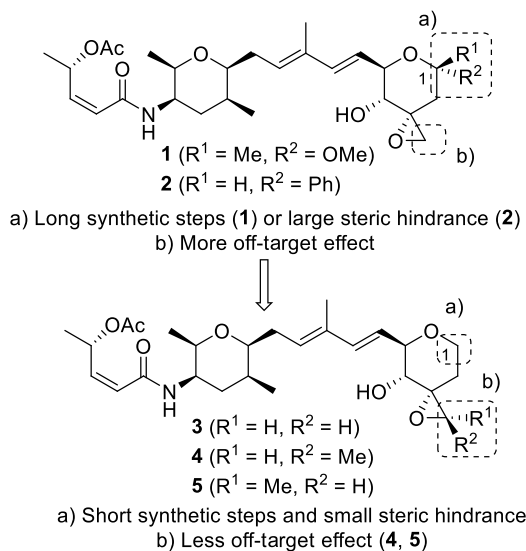


Figure 2. Design strategy for derivatives **3–5**.

For **3**, the AR-V7 expression inhibitory activity of phenyl-C-glycoside derivative **2** was weaker than that of **1**. We hypothesized that the weaker biological activity could be due to the presence of the bulky phenyl group, and therefore, we designed 1,2-deoxy-pyranose derivative **3** which had no functional group at the anomeric site and could have higher acid stability than **1**. In addition, it was considered that the C1 unsubstituted pyran fragment of **3** could be synthesized more easily than the C1 methyl ketal pyran fragment of **1**. (Figure 1a).

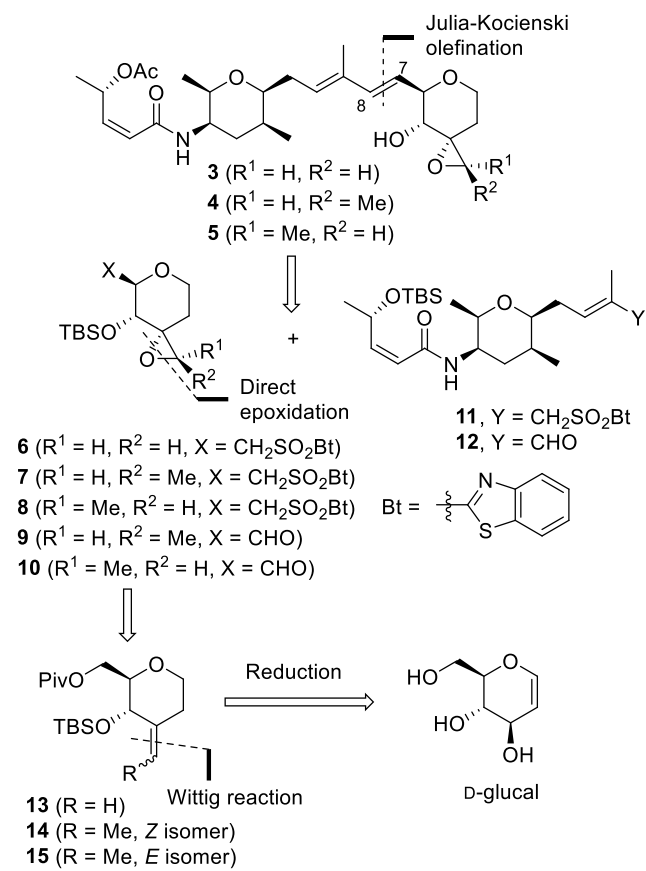
For **4** and **5**, we hypothesized that the presence of the terminal epoxide of **1** and/or **2** was required for biological activity expression. However, the epoxide is typically associated with instability or toxicity derived from off-target effects (forming covalent bonding with amino acid residues of other proteins).<sup>42</sup> Thus, we designed derivatives **4** (3'*R*) and **5** (3'*S*) to a methyl group on the epoxide of derivative **3** (Figure 2b), which could improve the stability of the epoxide.

Scheme 1 illustrates our retrosynthetic analysis for **3–5**. An internal double bond on C7 and C8, was constructed by Julia–Kocienski olefination between an appropriate aldehyde **9**, **10**, or **12**<sup>8</sup>) and a sulfone (**11**<sup>21</sup> or **6–8**). The structures **6–10** were prepared from the corresponding olefins **13–15** by diastereoselective direct epoxidation, and the olefins **13–15** could be prepared by a Wittig reaction of the corresponding ketone, which was derived from D-glucal.

The synthesis of olefin key intermediates **13–15** is summarized in Scheme 2.

For **13** in Scheme 2a, the hydroxy group of the protected 1,2-deoxy pyranose **16**, which was prepared from commercially available D-glucal,<sup>43</sup> was oxidized to the corresponding ketone with Dess–Martin periodinane (DMP) and the subsequent Wittig reaction produced the olefin **17** in 55% yield over two steps. The *p*-methoxy benzyl acetal **17** was reduced to the

Scheme 1. Retrosynthetic Analysis for **3–5**

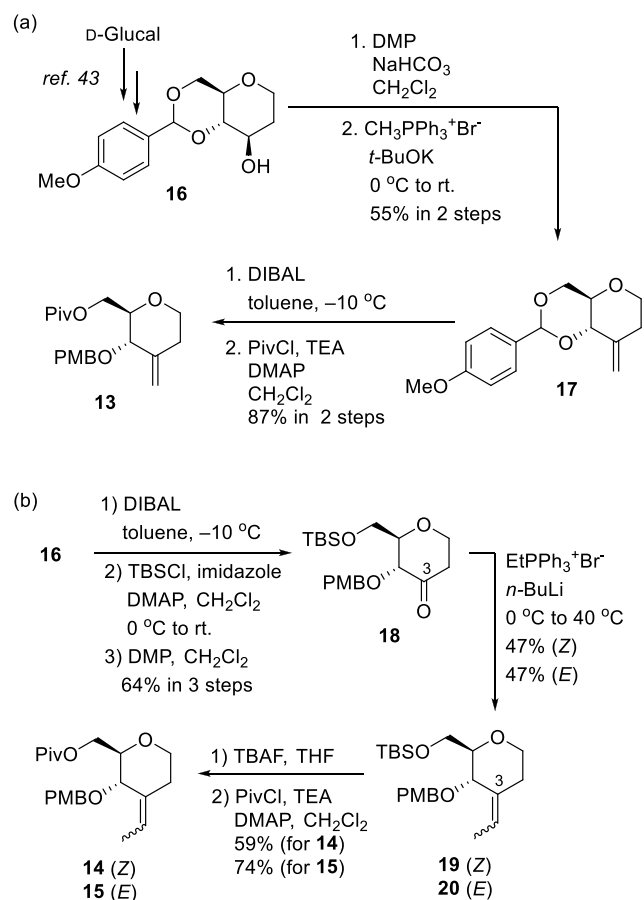


corresponding primary alcohol by *i*-Bu<sub>2</sub>AlH (DIBAL) at  $-10$  °C, and the generated primary alcohol was protected with *t*-BuCOCl (PivCl) to generate **13** in 87% yield (two steps).

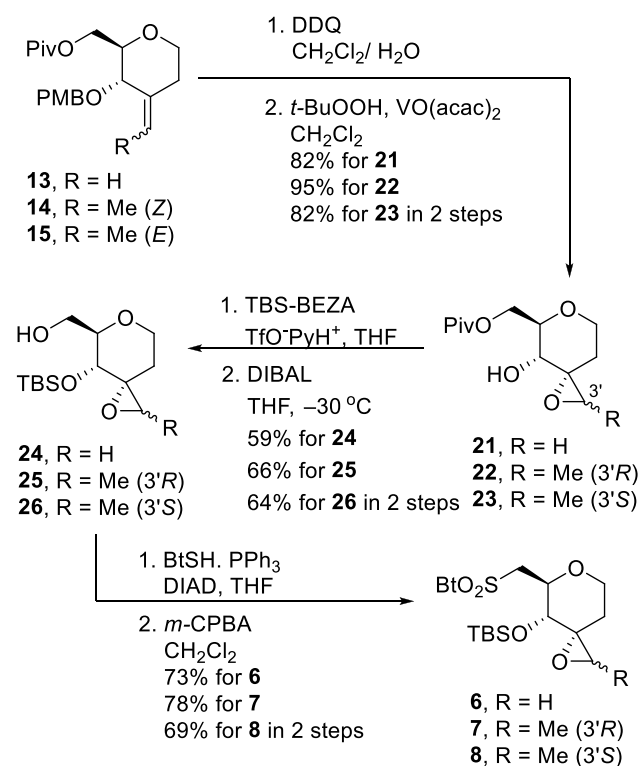
For **14** and **15** in Scheme 2b, the acetal **16** was reduced by DIBAL in the same manner as described above, the primary hydroxyl group was protected with a Si-*t*-BuMe<sub>2</sub> (TBS) group and the generated secondary alcohol was oxidized with DMP to produce ketone **18**. Subsequent Wittig reaction proceeded to furnish internal olefins **19** and **20**. The reaction proceeded smoothly (EtPPh<sub>3</sub><sup>+</sup>Br<sup>−</sup>, *n*-BuLi, THF, 0–40 °C) to create the geometric isomer **19** (*Z*) and **20** (*E*) in 47% yield, respectively. To construct trisubstituted olefins on the C3 position, other reaction conditions are actually applied (changing reaction temperature, base, substrate). The details of the reaction conditions are shown in Scheme S2. Finally, the TBS groups of **19** (*Z*) and **20** (*E*) were removed with Bu<sub>4</sub>NF (TBAF) to synthesize the corresponding primary alcohol. The alcohol was further protected with PivCl to produce ester **14** (*Z*) and **15** (*E*) in 59% and 74% yield over two steps, respectively.

The synthesis of benzothiazoles **6–8** is summarized in Scheme 3. The *p*-methoxy benzyl groups of **13–15** were removed by 2,3-dichloro-5,6-dicyano-*p*-benzoquinone (DDQ) to produce the corresponding alcohol. Subsequent diastereoselective epoxidation of the double bond with catalytic vanadyl acetylacetonate (VO(acac)<sub>2</sub>) (10 mol %) and *t*-BuOOH afforded the corresponding epoxide **21–23** as a single diastereomer in 82%, 95%, and 82% yield over two steps, respectively. The relative configuration of **22** was determined by X-ray crystal structure analysis (See SI CIF file). Next, the secondary hydroxyl groups of **21–23** were protected as TBS-

Scheme 2. (a) Synthesis of Intermediate 13 and (b) Synthesis of Intermediates 14 and 15



Scheme 3. Intermediates 6–8



ether with *O*-(*t*-butyldimethylsilyl)benzamide (TBS-BEZA) and TfO<sup>-</sup>PyH<sup>+</sup>. The pivalic ester was removed by DIBAL in THF at -30 °C to give the alcohols 24–26 in 59%, 66%, and 64% yield over two steps, respectively. Finally, the alcohols 24–26 were treated with 2-mercaptobenzothiazole (BtSH), PPh<sub>3</sub>, and diisopropyl azodicarboxylate (DIAD) to afford the benzothiazolyl sulfide, which was oxidized with *m*-chloroperoxybenzoic acid (*m*-CPBA) to generate the sulfones 6–8 in 73%, 78%, and 69% yield over two steps, respectively.

The sulfone 6 and aldehyde 12 were treated with LiN(SiMe<sub>3</sub>)<sub>2</sub> (LHMDS) at -78 °C in THF (Julia–Kocienski olefination) to give the diene compound, which was treated with HCl/MeOH (0.05%) to give alcohol 27 in 63% yield over two steps (Scheme 4a).<sup>44</sup> Next, to synthesize the coupling compounds 28 and 29, we initially tried the Julia–Kocienski olefination with sulfones 7, 8 and aldehyde 12, but the reaction was unsuccessful. The details of the reaction conditions are shown in Scheme S3. As shown in Scheme 4b, we tried another coupling combination. The oxidation of the primary alcohols 24 and 25 with DMP afforded the corresponding aldehydes 9 and 10, and treatment of the sulfone 11<sup>21</sup> and aldehydes 9 and 10 with LHMDS produced the corresponding coupling product in acceptable yield, which was treated with HCl/MeOH (0.05–0.1%) to give alcohols 28 and 29 in 39% and 59% yields over three steps, respectively.<sup>45</sup> Acetylation of the alcohols 27–29 with Ac<sub>2</sub>O, triethyl amine (TEA), and *N,N*-dimethyl-4-amino pyridine (DMAP) gave the corresponding acetate, and after removal of the TBS group with TBAF it generated 3–5 in 72%, 82%, and 86% yields over two steps, respectively (Scheme 4c).

The biological activity study is presented in Table 1. Derivatives 3–5 suppressed AR-V7 splicing (IC<sub>50</sub> = 3.3, 202, and 187 nM, respectively). The biological activity of 4 and 5 was almost identical to that of 2 (IC<sub>50</sub> = 132 nM), but the activity of 3 was significantly improved over that of 2. The result of 3 suggested that the phenyl group at the C1 position could produce a steric hindrance within the SF3b complex and removing the phenyl group considerably improved the biological activity. The results of 4 and 5 suggested that the introduced methyl group on the epoxide might block the hydrogen bonding network around the epoxide and weaken the interaction with the SF3b complex. Since there were various amino acid residues that could interact with the epoxide of 4 or 5, the difference of stereochemistry (*R* and *S*) of the epoxy methyl did not affect the biological activity. With respect to the *in vivo* toxicity test, wild type mice administered 1 (280 nM/body, *n* = 6) showed severe toxicity (mice died within 24 h).

However, none of the mice treated with 3 (280 nM/body, *n* = 7) died, and they did not show any significant weight loss compared to mice that were treated with DMSO alone (*n* = 8) (Figure 3).<sup>46–49</sup>

In conclusion, we designed, synthesized, and biologically evaluated the derivatives of spliceostatin A, 1,2-deoxy-pyranose derivative 3, and terminal epoxide methyl substituted derivatives (4, 5). The 1,2-deoxy-pyranose fragment and its terminal epoxide methyl substituted fragment were synthesized from commercially available D-glucal, and the synthetic steps for 24 were rather short.<sup>21</sup> We also investigated Julia–Kocienski olefination for appropriate combinations. Furthermore, the IC<sub>50</sub> values on suppression of AR-V7 for compounds 3–5 were weaker than that of 1; however, compound 3 exhibited potency (IC<sub>50</sub>) in the low nanomolar range. The *in*

Scheme 4. (a) Synthesis for 27, (b) Synthesis for 28 and 29, and (c) 1,2-Deoxy-pyranose Derivative 3 and Its Epoxide Substituent Derivative 4 and 5.

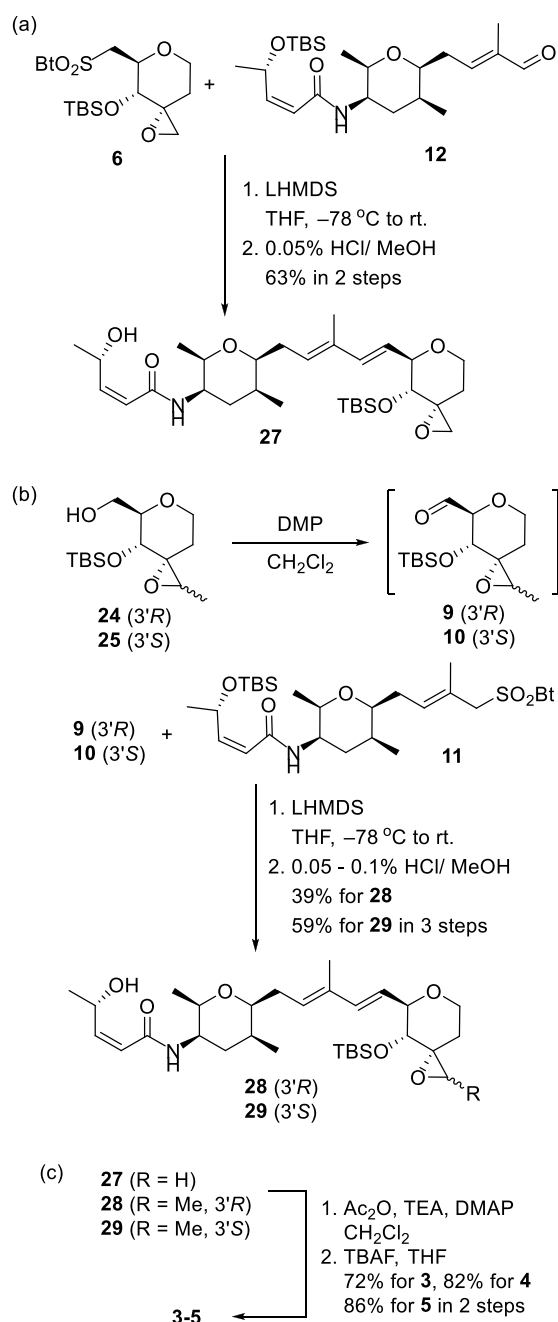


Table 1. AR-V7 Expression Inhibitory Activity Evaluation for 1 and 3–5

	IC <sub>50</sub> (nM)			
	3	4	5	spliceostatin A (1)
AR-V7-GFP negative population	3.3	202	187	0.6

*in vivo* toxicity test showed that wild type mice treated with **1** died within 24 h, but those mice treated with **3** did not die for 14 days and did not show weight loss. Therefore, we successfully created **3** with high AR-V7 expression inhibitory activity and low *in vivo* toxicity.

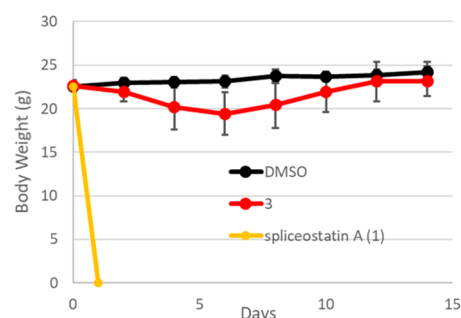


Figure 3. *In vivo* toxicity evaluation of DMSO, **3** and spliceostatin A (**1**) for wild type mice (injected on day 0, 2, 4, 280 nmol/body,  $n = 6-8$ ). Spliceostatin A killed all the treated mice ( $n = 6$ ).

## ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsmchemlett.0c00153>.

Experimental procedures, characterization data, and NMR spectra (PDF)

Crystallographic data of compound **22** (CIF)

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### Author Contributions

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### Notes

The authors declare no competing financial interest.

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## ABBREVIATIONS

ADT, androgen deprivation therapy; CRPC, castration-resistant prostate cancer; AR, androgen receptor; AR-V7, AR splicing variant 7; DMP, Dess–Martin periodinane; DIBAL, *i*-Bu<sub>2</sub>AlH; PivCl, *t*-BuCOCl; TBS, Si-*t*-BuMe<sub>2</sub>; TBAF, Bu<sub>4</sub>NF; VO(acac)<sub>2</sub>, vanadyl acetylacetonate; TBS-BEZA, *O*-(*t*-butyldimethylsilyl)benzanilide; BtSH, 2-mercaptobenzothiazole; DIAD, diisopropyl azodicarboxylate; *m*-CPBA, *m*-chloroperoxybenzoic acid; LHMDS, LiN(SiMe<sub>3</sub>)<sub>2</sub>; TEA, triethyl amine; DMAP, *N,N*-dimethyl-4-amino pyridine

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