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Absolute quantification of RNA or DNA using acid hydrolysis and mass spectrometry

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Abstract

Accurate, traceable quantification of ribonucleotide or deoxyribonucleotide oligomers is achievable using acid hydrolysis and isotope dilution mass spectrometry (ID-MS). In this work, formic acid hydrolysis is demonstrated to generate stoichiometric release of nucleobases from intact oligonucleotides, which can be then measured by ID-MS, facilitating true and precise absolute quantification of RNA, short linearized DNA, or genomic DNA. Surrogate nucleobases are quantified with a liquid chromatography-tandem mass spectrometry (LC-MS/MS) workflow, using multiple reaction monitoring (MRM). Nucleobases were chromatographically resolved using a novel cation exchange separation, incorporating a pH gradient. Trueness of this quantitative assay is estimated from agreement among the surrogate nucleobases, and by comparison to concentrations provided for commercial materials, or Standard Reference Materials (SRMs) from the National Institute of Standards and Technology (NIST). Comparable concentration estimates using NanoDrop Spectrophotometry or established from droplet-digital PCR (ddPCR) techniques agree well to the results. Acid hydrolysis-ID-LC-MS/MS provides excellent quantitative selectivity and accuracy while enabling traceability to mass unit. Additionally, this approach can be uniquely useful for quantifying modified nucleobases, or mixtures.

Graphical Abstract

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Supporting Information Available:

Figures and tables are provided in Supporting Information: MRM mass chromatograms of standard nucleobases; quantitative results from hydrolysis of DNA and RNA samples; nucleobase stability timecourses; dNMP stability timecourses; mass spectrometry analysis of 5mC; hydrolysis timecourse of DNA samples; nucleobase internal standards; MRM fragmentation transitions for MS analysis of nucleobases.



Introduction

Oligonucleotide quantification supports a diverse list of clinical and commercial fields, including microbial analysis, food safety (GMOs), medicine, biotherapeutics development, forensics, and more. Accurate and traceable quantitative assays to underpin these disciplines will be essential as those communities moves towards standardized measurement systems¹⁻⁴.

Conventionally, UV and fluorescence spectroscopy techniques have been broadly applied towards quantification of oligonucleotides^{2, 5, 6}. But although spectroscopic approaches may be fast, inexpensive, and simple, these relative quantification methods lack specificity, traceability, and calibration potentially resulting in significant quantitative bias. Still, fluorescence approaches can be highly sensitive, and UV can be fit-for-purpose when absolute accuracy is not essential. PCR-based techniques such as droplet digital polymerase chain reaction (ddPCR) can be robust, precise, and highly sensitive means toward oligonucleotide quantification, and have broad application in this area. Caution must be exercised however, as ddPCR assumes stoichiometric amplification of nucleic acid targets, which is not always true. Incomplete amplification is especially prevalent in reverse transcription assays (RT-dPCR) for RNA measurements^{7, 8}. Furthermore, ddPCR quantitative trueness is directly correlated to droplet size, which has been demonstrated to vary among different master stocks, between instrumentation, and over time^{6, 9, 10}; and ddPCR faces analytical limitations including for very large, truncated, high G/C, or non-linearized sequences.

Isotope dilution (ID) techniques are considered the gold-standard for biomolecule quantification. ID employs pre-spiked stable-isotope labeled (SIL) analogs (i.e., ¹³C or ¹⁵N) to account for quantitative biases assumed during sample preparation or subsequent detection by mass spectrometry (MS). In order to overcome the analytical limitations of

detecting large oligos by MS, or the costs/challenges associated with synthesizing SIL oligos, bottom-up approaches have been previously described where RNA or DNA could be enzymatically digested into constituent nucleosides^{11–19} or nucleotides^{3, 20, 21} prior to ID-MS analysis with nucleosides or nucleotides serving as quantitative surrogates of oligo concentration. Previously, this was achieved by other groups using endonucleases (RNAse A, nuclease P1, S1, DNAse I...) or exonuclease cocktails (phosphodiesterase I, alkaline phosphatase...). Unfortunately, enzymatic approaches have limitations – complete digestion may be inhibited by oligomeric size, by higher order structure, enzyme specificity, or covalent modifications²². In this manuscript, chemical hydrolysis rather than enzymatic cleavage is proposed as an alternative method to exploit bottom-up quantification of oligonucleotides.

Chemical approaches for RNA or DNA hydrolysis have been around for decades^{23,24}, and have evolved continuously with improvements in accuracy and specificity. Predominantly though, acid hydrolysis has been applied for atypical or modified nucleobase detection (DNA damage or carcinogenic adduction^{25–32}), rather than with the intent to quantify the intact oligonucleotide. Recently chemical hydrolysis was demonstrated using double-stranded genomic DNA from bacteriophage lambda (\approx 48.5 kbp) showing that DNA can be accurately quantified using just such an approach³³. In this work, we demonstrate a modified chemical hydrolysis workflow for any sized DNA using formic acid hydrolysis and nucleobase calibrators applied to small dsDNA (7.9 kbp linearized plasmid containing 5.2 kbp BK virus), and for the first time, to human genomic DNA (3.2×10^9 base-pairs). But most importantly, we demonstrate for the first time that this approach can be applied to ribonucleotide oligomers (RNA).

Purine and pyrimidine bases are known to be disproportionately or incompletely hydrolyzed under acidic conditions²⁴. A previous report³⁴ estimated nucleic acid concentrations in cellular material using 6 mol/L hydrochloric acid (HCl) hydrolysis with diode array detection (DAD), noting significant deamination or other losses for all nucleobases in 6 mol/L HCl hydrolysis, while separately hydrolyzing purines and pyrimidines. Another study³⁵ using dilute HCl (0.2 mol/L) observed complete purine hydrolysis from DNA, but with diminutive release of pyrimidines. Quantitative analysis of purine base adducts in genomic DNA has been reported previously using mild acidic treatment for short hydrolysis times^{26, 36}. Most recently, the National Metrology Institute of Japan (NMIJ) demonstrated formic acid's utility for hydrolyzing purines and pyrimidines from intermediate-sized bacterial dsDNA³³. Here, we expand on that effort by demonstrating hydrolysis at extreme sizes (for human genomic dsDNA) and for quantification of ribo-oligos (RNA). Neat formic acid (≈26 mol/L) can be used to hydrolyze samples in either the gas phase or liquid phase. Purine and pyrimidine stability is demonstrated for the nucleobases A, G, T, and U under the optimized hydrolysis conditions, while some deamination of cytosine to uracil was observed \approx 3 % (m/m), as reported previously³⁷. Here, we demonstrate that acid hydrolysis is wellsuited to RNA hydrolysis and DNA hydrolysis, and in conjunction with ID-MS can achieve traceable, accurate absolute quantification.

A 'double isotope dilution' scheme is designed for this work, involving the use of SIL nucleobase analogs (¹³C and/or ¹⁵N-adenine, -cytosine, -guanine, -thymine, -uracil) pre-

spiked as internal standards into samples *and* nucleobase calibrators, enabling true and precise quantification, traceable through reference standards to the US Pharmacopeia (USP), European Pharmacopeia (EP), and British Pharmacopeia (BP) primary standards. (Certified Reference Materials (CRMs) necessary to establish SI-traceability do not yet exist for pure nucleobases or nucleotides.) This unique approach establishes a process for using a single set of SIL nucleobases along with a single set of well-characterized nucleobase standards to be broadly applied to calibrate quantitative measurement of multiple, unrelated RNA or DNA oligomers.

Experimental

Materials

Nucleobase calibrators were purchased from Millipore Sigma. Calibrators are categorized as "Pharmaceutical Secondary Standard Certified Reference Materials" with traceability to the USP, EP (PhEur) and BP primary standards. No further chemical purity or water analysis was performed in-house. [Adenine (PHR1383, traceable to USP 1012101, PhEur A0230000); Cytosine (PHR1350, traceable to USP 1162148); Guanine (PHR1243, traceable to BP 879 and USP 1302156); Thymine (PHR1345, traceable to USP 1754532); Uracil (PHR1581, traceable to USP 1705753 and PhEur Y0000764); 5-methylcytosine hydrochloride (Sigma M6751, 99% pure)].

Stable isotope-labeled nucleobase standards were purchased through Cambridge Isotope Laboratories, Inc. (Andover, MA) for use as internal standards, as described below (Table S1):

Deoxynucleoside monophosphate standards (dNMPs) were purchased from commercial sources.

2'-deoxyadenosine 5'-monophosphate disodium salt hydrate (MP Biomedicals 02150795; MW 331.225)

2'-deoxycytidine 5'-monophosphate sodium salt (Sigma D7625; MW 351.16)

2'-deoxyguanosine 5'-monophosphate disodium salt hydrate (Sigma 852228; MW 391.18)

thymidine 5'-monophosphate disodium salt hydrate (Sigma T7004; MW 366.17)

2'-deoxyuridine 5'-monophosphate disodium salt (Sigma D3876; MW 352.15)

An RNA standard was acquired from Ambion (Invitrogen) through ThermoFisher (AM7155, RNA Control 250). DNA standards were acquired in-house. BK virus linearized plasmid DNA was acquired from a stock solution used in the preparation of NIST SRM 2365. Human genomic DNA was from NIST SRM 2372a (Component A)³⁸. 5-methylcytosine powder was purchased through Sigma-Aldrich (M6751, 5mC•HCl). Formic acid was LC-MS grade from Honeywell (56302).

Acid hydrolysis

RNA or DNA samples were gravimetrically added to 400 μL glass flat-bottomed autosampler inserts (Agilent, 5181–3377), and dried to dryness in a speed-vac without heat. A solution of a mixture of SIL-nucleobases was pre-spiked into samples and calibrants and used as an internal standard to normalize MS signal. Glass inserts were placed into acid-resistant, temperature-safe Teflon vessels within a steel compression pressure-bomb. Approximately two (2) mLs of neat formic acid was pipetted into the vessels, external to the glass inserts. Vessels were tightened to ensure vapor pressure retention, and then placed into an oven at 140 °C for gas-phase hydrolysis. Hydrolysis was performed for 24–48 hours depending on the sample type. After hydrolysis, glass inserts were dried to dryness in a speed-vac and samples were reconstituted in water for subsequent LC-MS analysis.

LC-MS/MS analysis

For the chromatographic separation of nucleobases, an ion-exchange mechanism with a decreasing pH gradient on a mixed-mode column enabled complete resolution of nucleobases (Figure 1). Gradient elution began at 99 % mobile phase A (0.5 mL/L TFA) and increased linearly with additional mobile phase B (4.5 mL/L TFA in 0.2 L/L aqueous acetonitrile (ACN)) to 10 % over 10 min., and to 40 % over the next 20 min. followed by a column wash at 95 % B and re-equilibration. The column was regularly washed with 95 % ACN in-between batches. A constant mobile phase flow rate of 200 µL/min was used. A mixed-mode ion exchange / reversed-phase Primesep 100 column (SIELC Technologies, Wheeling, IL), 2.1×250 mm, 3 µm particles, was maintained at 50 °C in a thermostatted column compartment. Samples were maintained at 5 °C in the autosampler. Fresh mobile phases were prepared weekly.

For quantification, a multiple-reaction monitoring (MRM) assay was developed targeting two specific fragmentation transitions of each nucleobase and two fragmentation transitions of a stable-isotope-labeled analog of each nucleobase (Figure S1). Precursor-to-product ion fragmentation transitions and ionization conditions were optimized using purified standards of nucleobases. An Agilent 1290 Infinity liquid chromatography system (Agilent Technologies, Santa Clara, CA) was used in-line with an Agilent 6460A triple quadrupole (QQQ) mass spectrometer. The LC separation was optimized for complete chromatographic resolution of all nucleobases in order to ensure specificity of the quantitative assay. Electrospray ionization (ESI) was achieved in positive ion polarity using one continuous MRM scan segment. All analyses were performed with the following Agilent 6460A MS source parameters: source gas temperature = $300 \,^{\circ}$ C, source gas flow = $13 \, \text{L/min}$, nebulizer = 345 kPa (50 psi), sheath gas temperature = 250 °C, sheath gas flow = 12 L/min, capillary voltage = 3500 V, nozzle voltage = 1500 V. Each MRM fragmentation scan was acquired in unit resolution for MS¹ and MS², over a 100 ms dwell time, and with a cell accelerator voltage of 7 V. All other instrument parameters for MRM scans used in the measurement of each set of labeled and unlabeled nucleobases are listed in Table S2.

Agilent MassHunter Workstation software (version B.09.01) was used for peak selection and integration. Peak retention times and integrated peak areas were automatically determined by MassHunter. All peak integrations were visually inspected, and in some cases, manual

integration was necessary. Peak area ratios were exported into Microsoft Excel for quantitative analysis. Unlabeled/labeled integrated peak area ratios were calculated from calibrant data and plotted against gravimetric mass ratios into calibration curves. For quantitative analysis of sample, molar mass ratios were extrapolated from the calibration curves according to the measured peak area ratios. From this data, nucleobase concentrations were calculated and converted to intact oligonucleotide concentration.

For high-resolution MS analysis of modified nucleobases (5mC) on an Orbitrap Elite MS system (Thermo), a data-dependent analysis with dynamic exclusion was used with FTMS resolution of 30,000 for MS¹ acquisition in positive polarity within the scan range of m/z 50.0–300.0. Standard source and fragmentation parameters were used with a CID normalized collision energy of 35, and MS² data acquisition in the ion-trap.

Isotope dilution and sample preparation

A double exact-matching isotope dilution (ID) workflow was designed for the quantitative assay. This design requires the use of well-characterized SIL internal standard nucleobases pre-spiked into samples and external calibrants to normalize for instrument and sample preparation variability³⁹. Calibration is achieved with a five-point bracketing calibration curve using traceable, pure standards of nucleobases with known concentrations. For hydrolysis assays of intact RNA and DNA, six replicates were prepared gravimetrically and were pre-spiked with a mixture of SIL internal standard nucleobases prior to hydrolysis. Samples were incubated at 140 °C for 24 h –48 h using gas phase formic acid hydrolysis, dried, and reconstituted in mobile phase A prior to LC-MS/MS analysis. Calibrants consisted of nucleobase standards in a mixture with SIL nucleobases, prepared at unlabeled/ labeled mass ratios of approximately 0.5, 0.75, 1.0, 1.25, and 1.5 (m/m %), where the midpoint of the calibration range is estimated as the expected concentration of nucleobase content of the RNA or DNA sample. SIL internal standard concentrations varied among RNA or DNA samples and were targeted at a ≈ 1.1 molar equivalence to the expected sample concentration, while remaining constant among calibrators for a given sample. All calibration plots were reported with coefficients of determination (R^2) for regression lines > 0.99 for both MRM transitions of all nucleobases.

Results and Discussion

Chromatography

The most commonly used liquid chromatographic approach for nucleic acids is reversedphase separation^{40, 41}. Although C_{18} reversed-phase is both robust and MS-friendly, nucleobase retention on this phase is relatively weak. Here, we demonstrate better retention and complete resolution of nucleobases using a mixed-mode column consisting of a C_{18} stationary phase embedded with acidic ion-pairing groups. This phase (SiELC, Primesep 100) improves retention of basic compounds (nucleobases) by cation-exchange mechanisms while using gradient elution of decreasing pH and increasing organic mobile phase composition. Figure 2 provides representative MRM extracted ion chromatograms of canonical nucleobase standards on a) C_{18} analytical column using common mobile phases (A:B = acidified H₂O: acidified ACN), b) on a porous-graphitized carbon column (HyperCarb, Thermo) using the same mobile phases, and c) using the mixed-mode Primesep column as described in Experimental. Retention under several gradient conditions on the C_{18} column (shown in Fig. 2) proved inadequate for complete peak resolution resulting in less precise quantification. However, it should be noted that not all C_{18} columns are alike, and many chromatographic options are available. PGC is a unique phase with distinct separating mechanisms for polar or non-polar analytes. Under MS-friendly conditions PGC retains nucleobases well with near baseline resolution, however, purines tend to tail under these LC conditions leading to integration bias. Ion exchange/ reversed-phase separations using the Primesep column were observed with moderate retention using MS-amenable mobile phases and with baseline resolution of A, C, T, G, and U (Figure 1). The separation of canonical nucleobases, this column was demonstrated capable of retaining other modified nucleobases including 5-methylcytosine (5mC).

Acid hydrolysis optimization

Time-course assays of calibration standards were designed to test nucleobase stability under hydrolysis conditions (Figure S2). Hydrolysis was ultimately achieved in this work using formic acid, however, initially three acids were tested: 1) trifluoroacetic acid (TFA; CF₃CO₂H) rapidly degraded nucleobases in less than one hour at each temperature tested (60 °C, 120 °C, and 140 °C); 2) hydrochloric acid (HCl) was tested at 8 mol/L and \approx 2 mol/L (120 °C and 140 °C) with only slight differences observed – nucleobase pyrimidines degraded rapidly (less than one hour), while purines were more tolerant but degraded partially or entirely within 2–4 hours. Hydrolysis using ≈ 0.2 mol/L HCl yielded incomplete pyrimidine release and significant deamination of cytosine to uracil at longer timepoints; 3) hydrolysis using neat formic acid (HCOOH) was demonstrated to preserve purines over the full time-course at 140 °C. For pyrimidines, both uracil and thymine were shown stable when subjected to these hydrolysis conditions, while cytosine was observed to be slightly deaminated into uracil at longer time-points. Semi-quantitative assays estimated cytosine deamination rates of \approx 3 % (relative peak area) at 24 hours of hydrolysis (Figure 3). A 5methylcytosine (5mC) standard was tested using only formic acid and although some deamination into thymine was expected, little to no thymine degradation product was detectable.

Time-course assays were performed on dNMP standards to determine a target hydrolysis time which ensures reaction completion. dNMPs were hydrolyzed with formic acid, as above for nucleobase standards, with liberated nucleobases being detected by LC-MS/MS. Saturation of signal was monitored for reaction completion (Figure S3). Several temperatures were tested over the 24 h time-course. It was observed that at the highest temperature tested (160 °C), deamination of cytosine was accelerated. At lower temperatures (60 °C and 120 °C), complete hydrolysis of pyrimidine-based dNMPs was not guaranteed (dUMP and dTMP specifically). The optimal temperature was determined to be 140 °C, with complete hydrolysis being observed for all dNMPs by \approx 24 h. It was observed that for the RNA material full release of the pyrimidines (T and U) was not complete until >24 h, as detailed previously³⁴. A hydrolysis time of 24 h (or 48 h) was possible for analysis of the intact DNA samples described in this work, and 48 h of hydrolysis was used for RNA

samples, to ensure complete hydrolysis. For any given target oligonucleotide, whether DNA

or RNA, it is important to test for hydrolysis completion under the given conditions and ensure that the optimized conditions are appropriate.

Absolute Quantification of RNA

A single-strand RNA control solution used routinely for verifying the accuracy of NanoDrop Spectrophotometer measurements was acquired from commercial sources (Invitrogen, part AM7155). The RNA Control 250 material is provided as an ultra-pure RNA transcript of a precisely defined sequence, molecular weight, and concentration $(255 \pm 13 \text{ ng/}\mu\text{L})$, as determined by the manufacturer using absorbance @ 260nm. Individual nucleobase concentrations ([A], [C], [G], [U]) were inferred from the provided RNA sequence and its stated concentration, assuming stoichiometric release following hydrolysis, and expressed as nmol/L (nM). The manufacturer-provided concentrations of A, C, G, and U, respectively, are 176.0 \pm 8.8, 224.3 \pm 11, 194.3 \pm 9.7, and 185.0 \pm 9.3 nmol/L. RNA samples were thawed only once from –20 °C, and replicates were gravimetrically prepared with SIL-nucleobase internal standard. Approximately three (3) μ Ls of the RNA 250 control solution was hydrolyzed per replicate, providing abundant signal for robust quantification (typical MRM peak height 10⁵–10⁶). Hydrolysis was completed after 48 hours at 140 °C using gas-phase hydrolysis with neat formic acid. ID-MS analysis was performed on three (3) separate days using six sample replicates per day and freshly prepared calibration curves.

The average (μ) results from ID-LC-MS/MS analyses of the RNA control 250 solution are provided in Table 1, reporting standard deviation (σ) and relative standard deviation (RSD) of each nucleobase (throughout this text, statistical analysis is represented as $\mu \pm \sigma$, with RSD expressed as appropriate). Expected values and measured values of nucleobase concentrations agree within $\pm 1\sigma$ (Figure S4). The measurement precision for nucleobases ranged from 3.4 % – 4.5 % RSD. Both purines trended slightly high but fell within uncertainty estimates. Purine nucleobases were observed to be completely released from the intact oligomer quickly under the hydrolysis conditions described here. Both adenine and guanine reached steady-state within one hour of hydrolysis (Figure 4). Due to practical limitations on rapid heating and cooling of the steel hydrolysis bomb, it is difficult to make estimates on shorter time-points. Regardless, it is possible to quantify the purines at quite early timepoints using this ID-MS approach as compared with pyrimidine nucleobases. Pyrimidine bases were released slowly under formic acid hydrolysis, requiring greater than 24 h to reach steady-state for this intact RNA. Fortuitously, due to the robust stability of nucleobases under these hydrolysis conditions, it is possible to measure the purines and pyrimidines simultaneously using a single assay. Pre-spiked SIL-nucleobase internal standards further guarantee any potential quantitative biases associated with losses or degradation during the sample preparation are accounted for.

It is useful to provide an estimated value or a 'predicted' value for the concentration of the intact RNA sequence based on the concentration of detected nucleobases. Although this work cannot establish the origin of the detected nucleobases after they are liberated from the oligomer, it is possible to provide a predicted concentration of RNA with the reasonable assumption that all nucleobases originated from an intact, homogenous, pure oligomer of

known structure. (Techniques like ddPCR or UV/Vis make similar assumptions by amplifying and detecting oligomer fragments, or detecting excited light from non-specific electronic transitions.) Predicted RNA concentration was calculated using the mean of the percent deviation of each nucleobase from its expected nucleobase concentration and multiplying this value against the RNA concentration provided by the manufacturer. This assumes that the mean measurement deviation of the nucleobases from theoretical is proportional to the measurement deviation of the intact RNA oligomer from the theoretical value. The predicted value from ID-MS analysis of the RNA Control 250 material of 263.6 ng/ μ L agrees statistically to the manufacturer-provided value of 255 ± 13 ng/ μ L.

Absolute Quantification of DNA

NIST SRM 2365 (BK virus DNA Quantitative Standard⁴²) is a 7934 bp, linearized plasmid DNA material containing BK virus DNA that is provided within a solution of yeast tRNA, for material stability. BK virus DNA copy number was certified at NIST for SRM 2365 using a robust ddPCR assay. In this work, a concentrated $(5.58 \times 10^9 \text{ copies}/\mu\text{L} \pm 1.2 \times 10^8)$ tRNA-free stock solution of SRM 2365 was used, as acid hydrolysis does not discriminate between nucleobases originating from DNA or from the tRNA. The stock solution used to prepare NIST SRM 2365 was subjected to acid hydrolysis using formic acid at 140 °C. Time-course analysis of the BK plasmid demonstrates full hydrolysis of its nucleobases by 24 h (Figure S5). As described above for the RNA material, the BK concentration can be expressed in terms of nucleobase concentration and for calculated in nmol/L: A, C, G, and T concentrations are 41.3 ± 0.91 , 32.0 ± 0.70 , 32.0 ± 0.70 , and 41.8 ± 0.92 nmol/L, respectively. The BK DNA is stable at 4 °C, and replicate samples were gravimetrically prepared with mixed SIL-nucleobase internal standard. Approximately fifteen (15) µLs of the BK solution was lyophilized and hydrolyzed per replicate. Hydrolysis was completed after 24 h at 140 °C using gas-phase hydrolysis with neat formic acid. ID-MS analysis was performed on three (3) separate days using six sample replicates per day and freshly prepared calibration curves.

The average (μ) results from ID-LC-MS/MS analyses of the BK solution are provided in Table 2, with standard deviation (σ) and relative standard deviation (RSD) of each nucleobase specified. Expected values and measured values of nucleobase concentrations agree within $\pm 1\sigma$ (Figure S6). The total precision of the assay was excellent, with RSDs for nucleobases A, C, G, and T equal to 2.8, 3.1, 3.9, and 3.0 %, respectively. As mentioned earlier, complete hydrolysis of purines was observed within one (1) hour, significantly faster than hydrolysis of pyrimidines. In addition to the canonical nucleobases expected from DNA, a non-negligible concentration of uracil was measured in this material. Although the precision of the uracil quantification was poor ($\approx 17\%$ RSD), the mean of the measurements suggested at least some contribution from sources other than deamination of cytosine to uracil. Based on the hydrolysis stability assays for a cytosine standard material (Figure 3), deamination of cytosine should contribute to roughly 1.1 nmol/L of uracil following hydrolysis of the BK material. RNA contamination of the stock solution is one potential source of the unexpected uracil signal. Another possible contribution is from a measurement calibration system bias at exceedingly low analyte concentrations.

A predicted value for the intact DNA oligomer can be estimated in a similar fashion to the calculation above of RNA concentration from its constituent nucleobases. For this ID-MS assay, the predicted value of the BK virus DNA oligomer of $5.42 \times 10^9 \pm 1.7 \times 10^8$ copies/µL is statistically equivalent to the value predicted from ddPCR of $5.58 \times 10^9 \pm 1.2 \times 10^8$ copies/µL.

Quantification of human genomic DNA

The acid hydrolysis procedure was next applied to a human genomic DNA material – NIST SRM 2372a (Component A) – for a proof-of-principle experiment aiming to demonstrate that even the largest biological genomic materials are amenable to complete hydrolysis using formic acid and ID-MS quantification. This DNA material was prepared from human buffy coat genomic DNA from a single male donor (the buffy coat contains leukocytes and platelets). The certified value for DNA copy number was based on ddPCR quantification assays of ten targets on eight (8) chromosomes. The certified DNA mass concentration of 49.8 ± 5.0 ng/µL is calculated based on a reference human haploid genome equivalent and the mean MW of sodium salts of nucleotide monomers. In order to calculate nucleobase concentrations from the certified values provided in the NIST CoA it is necessary to make the assumption that the AT:GC for human genomic DNA \approx 60:40 and therefore the average MW of a nucleoside in genomic DNA is 330 g/mol⁴³. A second assumption was made that the density of the buffy coat sample ≈ 1 g/mL.

SRM 2373a(A) was subjected to 24 hours of formic acid hydrolysis as described previously. The analysis was performed in triplicate subjecting $\approx 15 \,\mu$ L of the SRM material per replicate to hydrolysis. Calibration was achieved using a similar scheme as described above. The absolute concentrations for nucleobases in SRM 2372a(A) were determined for A, C, G, and T as 43.3, 30.8, 32.9, and 43.3 nmol/L, respectively. RSDs for nucleobase measurements ranged from 2.8 – 8.4 % with standard deviations for A, C, T, and G equal to 1.80, 0.848, 1.98, and 3.62 nmol/L, respectively. It follows that the product of the average MW of nucleosides in genomic DNA (330 g/mol) with the summation of the molar concentrations (from ID-MS) of A, C, G, and T (~150.2 nmol/L) is approximately equal to 49.6 ng/µL of DNA in SRM 2372a(A). This value agrees well to the certified DNA mass concentration on the CoA (49.8 \pm 5.0 ng/µL) determined by ddPCR, and suggests that acid hydrolysis with ID-MS is well-suited to quantitatively measure complete, human genomic DNA. Other considerations beyond the scope of this manuscript that might affect the ID-MS work include the potential contribution of nucleobase signal from mitochondrial DNA in the buffy coat, which is expected to be negligible in this case when assuming the mass of mtDNA < 2% (m/m) that of nDNA. Also, the detection of uracil from deamination of cytosine, from RNA, or mtRNA was not addressed.

(It should be clearly noted that the work presented here did not in any way contribute to the certification of NIST SRM 2372a.)

Detection of covalently-modified nucleobases

Formic acid hydrolysis with LC-MS/MS analysis is capable of detecting or quantifying modified nucleobases – among others, 5-methyl cytosine (5mC). During LC-MS/MS

analysis of the BK sample described above, relative quantification was performed for the modified nucleobase 5mC using LC-MRM-MS, and later we confirmed the identity of 5mC by high-resolution LC-MS/MS analysis on an Orbitrap MS system. The 5mC standard was used to optimize chromatographic retention time, and the ionization and fragmentation conditions for LC-MS/MS assays. However, as there are no CRMs or primary quantitative standards available for 5mC analysis, this work demonstrates only a semi-quantification of 5mC in the BK solution. Figure S7 shows an extracted ion chromatogram (XIC) from the MRM analysis of a calibration solution and for a BK plasmid sample, as performed on a QQQ instrument. 5mC is chromatographically resolved from other nucleobases, enabling use of low-resolution techniques such as MRM analysis, even considering a mass difference () from thymine of only -1 Da. 5mC was monitored during method development using two MRM transitions (126.1 \rightarrow 82.4, 110.3), however, due to its low biological abundance, only the 82.4 m/z fragment ion was observed with good S/N in samples. A calibration curve was generated without SIL-internal standard for 5mC. Semi-quantification indicates ≈92 pmol/L of 5mC in the BK virus sample. This is equivalent to $\approx 0.3\%$ relative abundance to cytosine – a value that tends towards the low end of agreement with other semi-quantitative work⁴⁴. To verify the presence of 5mC in the BK sample, high-resolution LC-MS analysis was performed using an Orbitrap MS and fragmentation data acquired in the ion trap. Figure S8 provides XICs and MS² spectra of 5mC comparing standards to the BK sample.

Analytical approach

The results of this work establish a robust and accurate, mass-traceable method for multiplexed nucleic acid measurement, suitable for absolute quantification of homogeneous RNA or DNA samples. The approach is broadly applicable to many classes of nucleic acid polymers, whether single or double stranded or consisting of ribo- or deoxyribo-nucleic acids, and this approach is demonstrated to be responsive to covalently-modified (stable) nucleobases. Measurement precision for this acid hydrolysis with ID-MS assay typically ranges between 2–5 % RSD for quantification of nucleobase concentration – equal to or better than many other measurement techniques. The principal benefit of using an MS-based approach for oligonucleotide quantification as compared to prevailing techniques is the capability to apply an isotope dilution (ID) framework. ID techniques make it feasible to use external and internal calibration of the measurement assay, simultaneously linking the quantitative traceability chain to a higher-order reference standard - a Standard or Certified Reference Material – while also accounting for experimental biases associated with sample preparation or chromatographic or mass spectrometric variability. Conventional techniques such as UV absorbance or fluorescence detection lack specificity and traceability and can be susceptible to interferences; copy number measurements by ddPCR require an accurate determination of droplet size to achieve absolute quantification and can measure only accessible, intact targets – while all three approaches require nucleic acid oligomer standards, or unique primers, that exclusively match only the sample of interest. In this work, calibration solutions and SIL-internal standards can be developed from nucleobase monomers. The unvarying result of hydrolyzing any RNA or DNA sequence is the same subset of five canonical nucleobases (ignoring covalently-modified nucleobases detectable at considerably lower levels). This congruence means that any pure RNA or DNA sample can be quantified by ID-MS using the same set of calibrators where all nucleobases are useful to

determine a 'consensus' quantity of the intact oligomer. Alternatively, is it possible to differentiate DNA from RNA in a mixture through quantification of unique surrogate bases – cytosine or uracil, respectively. In this approach, the deamination rate will necessarily need to be considered for mixtures where [DNA] >> [RNA].

There are clear limitations to this technology, as well. All nucleic acid-containing compounds contribute to the signal, so it is not possible to distinguish between, say, adenine from different sources. However, it is conceivable to differentiate a mixture of DNA and RNA based on differences in thymine or uracil signal. The other major limitation is analytical sensitivity when compared to amplification-based approaches like PCR, or fluorescence detection which might be >100 fold more sensitive than mass spectrometry. Nevertheless, acid hydrolysis with mass spectrometry detection should find its niche, being exactly well-suited to provide traceable quantification necessary for reference material development, and for applications such as "total" nucleobase quantification of mixtures, for quality control of droplet size for ddPCR, or for relative quantification or optimization of cell-based sample preparation procedures.

Supplementary Material

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Figure 1:

Liquid chromatographic separation of nucleobase standards using a mixed-mode ion exchange / reversed-phase stationary phase with a decreasing pH gradient. Fragmentation transitions were detected using multiple-reaction monitoring on a triple quadrupole mass spectrometer.



Figure 2:

MRM EICs for nucleobase standards using a) C_{18} column with reversed-phase chromatography, b) PGC column, or c) mixed-mode cation-exchange with C_{18} column using a pH and organic gradient elution.

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Figure 3:

Cytosine stability timecourse under hydrolysis conditions (gas-phase formic acid, 140°C). Uracil is quantified by LC-MS/MS (MRM) analysis following the deamination of cytosine into uracil.



Figure 4:

Purines and pyrimidines release from RNA – hydrolysis timecourse using gas-phase formic acid at 140°C. Purines (adenine and guanine) are rapidly released from RNA (or DNA) and can be quantified within 2 hours of hydrolysis, while pyrimidines (cytosine and uracil) do not reach complete hydrolysis until more than 24 hours after hydrolysis begins.

Table 1:

Quantitative results for RNA 250 control solution from Ambion (Invitrogen)

	adenine	cytosine	guanine	uracil
[average], nM	189.4	225.5	206.6	183.3
stdev, nM	7.9	10.2	7	7.3
% CV	4.2	4.5	3.4	4.0
[expected], nM	176.0	224.3	194.3	185.0
stdev, nM	8.8	11	9.7	9.3
% of theoretical	107.6	100.5	106.3	99.1

Table 2:

Quantitative results from ID-LC-MS/MS analysis of BK virus solution

	Adenine	Cytosine	Guanine	Thymine	Uracil
[average], nM	40.5	31.2	30.9	40.1	2.8
stdev, nM	1.1	1.0	1.2	1.2	0.5
%CV	2.8	3.1	3.9	3.0	16.7
[expected], nM	41.3	32.0	32.0	41.8	-
stdev, nM	0.91	0.70	0.70	0.92	-
% of expected	98.2	97.5	96.7	96.0	-