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Evolution of a 4-Benzyloxy-benzylamino Chemotype to Provide Efficacious, Potent, and Isoform Selective PPAR α Agonists as Leads for Retinal Disorders

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Abstract

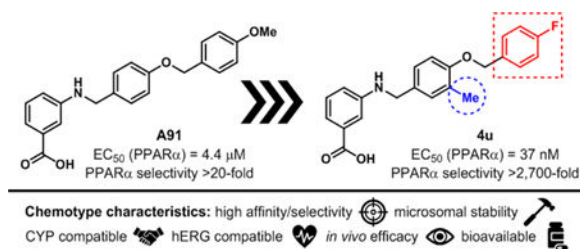
Peroxisome proliferator-activated receptor alpha (PPAR α) is expressed in retinal Müller cells, endothelial cells, and in retinal pigment epithelium, and agonism of PPAR α with genetic or pharmacological tools ameliorates inflammation, vascular leakage, neurodegeneration, and neovascularization associated with retinal diseases in animal models. As such, PPAR α is a promising drug target for diabetic retinopathy and age-related macular degeneration. Herein, we report proof-of-concept *in vivo* efficacy in an STZ-induced vascular leakage model (rat) and preliminary pharmacokinetic assessment of a first-generation lead **4a** (A91). Additionally, we present the design, synthesis, and evaluation of second-generation analogs, which led to the discovery of **4u** and related compounds that reach cellular potencies <50 nM and exhibit >2,700-fold selectivity for PPAR α over other PPAR isoforms. These studies identify a pipeline of candidates positioned for detailed PK/PD and pre-clinical evaluation.

Graphical Abstract

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ASSOCIATED CONTENT

The Supporting Information is available free of charge on the ACS Publications website at DOI: Molecular Formula Strings, two-concentration luciferase data, PPAR α expression and purification, cellular thermal shift western blots, LDH cytotoxicity data, *in silico* metabolic predictions, isothermal titration calorimetry thermograms and curve fits, ¹H and ¹³C NMR spectra for final compounds, HPLC traces for select leads, and pharmacokinetic data sets.



Keywords

PPAR α ; age-related macular degeneration; diabetic retinopathy; PPAR selectivity; structure-based design

INTRODUCTION

Diabetic retinopathy (DR) is a common complication of diabetes (~30% of diabetics) and is the leading cause of blindness in the working population.¹ Currently, >40% of the patient population fails to respond to the gold-standard treatment, direct intraocular injection of vascular endothelial growth factor (VEGF) antibodies.² New therapies that are superior or complementary to current approaches would be of great value to patients, especially for those that remain refractory to current options. Ideally, unlike current treatment options, new approaches should be non-invasive (to the eye), affordable, and not reliant on specialized facilities.

The promise of peroxisome proliferator-activated receptor alpha (PPAR α) agonism as a novel strategy for treating DR has been confirmed in two independent human clinical trials (FIELD and ACCORD), wherein orally administered Fenofibrate (Feno), a clinically used drug for hyperlipidemia, demonstrated robust protective effects against DR and retinal neovascularization (NV) in type 2 diabetic patients.³⁻⁴ This protection greatly reduced the need for patients to undergo disease modifying laser photocoagulation therapy. Since the report of these results, mechanistic and genetic studies have determined that the protective effects of Feno are unrelated to its lipid-lowering activity, but rather result from agonism of PPAR α by the Feno metabolite fenofibric acid (FenoFA).⁵⁻⁶ Feno/FenoFA however, suffer from low ocular distribution, low affinity and selectivity for PPAR α , and dose-limiting toxicities, all of which will likely limit its use as a DR therapy. Proof-of-concept, however, for targeting retinal PPAR α with systemically available small molecules has been established with Feno and revealed that small molecule agonists with improved PK/PD profiles have high promise to become first-in-class non-invasive treatment options for DR and related ocular conditions (e.g., age-related macular degeneration and retinopathy of prematurity).⁷⁻⁸

As important members of the nuclear receptor superfamily, PPARs play essential roles in regulating cellular differentiation, development, metabolism, and tumorigenesis of higher organisms. Thus, these receptors have been considered attractive targets for treating metabolic disorders and inflammatory conditions.⁹⁻¹⁰ PPARs consist of 3 subtypes: PPAR α , PPAR γ and PPAR δ/β . Although the ligand-binding domains share high sequence similarity

(60–70%), the isoforms exhibit distinct tissue distribution, biological functionality, and can be targeted selectively.¹¹ Within this family, PPAR α has been implicated in hyperlipidemia, diabetes and inflammatory disorders; and agonism of this isoform provides pharmacological benefits for these conditions.^{12–13} Only recently, however, have the roles of PPAR α in regulating VEGF, inflammation, mitochondrial function, apoptosis, and NV in diabetic retiniae been revealed, thus elucidating a new avenue for PPAR α agonists as therapeutics for oculo-vascular diseases.^{14–15} Specifically, previous studies demonstrate that PPAR α knockout alone results in retinal mitochondrial dysfunction, a declined electroretinogram (ERG) response, and retinal cell apoptosis.¹⁶ Furthermore, PPAR α levels in the retina are decreased in diabetic animal models and in diabetic humans and that agonism of PPAR α restores balance.^{14–15} Although PPAR α agonism was intensely investigated in the 1990s as a therapeutic approach for dyslipidemia, and numerous potent agonists exist in literature, none of the known chemotypes have been evaluated in animal models for efficacy against retinal conditions. Rather than attempting to repurpose known compounds, often lacking full disclosure of physicochemical and PK/PD profiles, we decided to embark on a discovery process to identify and develop novel PPAR α agonistic chemotypes as new compositions for DR and AMD leads.

Recently, we published first-generation structure activity relationship (SAR) studies on a new PPAR α agonistic chemotype, 4-carboxy-2-phenylquinolines, represented by Y-0452 (Figure 1).¹⁷ Guided by *in-silico* studies, we proposed that deconstruction of the rigid quinoline core would provide a more synthetically tractable and flexible scaffold, expected to exhibit improved complementarity to the U-shaped ligand binding pocket. Based on this design strategy, we developed the 4-benzyloxy-benzylamino chemotype to assess our hypothesis. In general, this small molecule family exhibits improved potency and selectivity over both Y-0452 (initial hit) and Fenofibrate, thus validating our rationale for further development of this chemotype and pursuing proof of concept studies. Compound A91 (Figure 1), a representative analogue from this series was advanced through rigorous biochemical evaluation to demonstrate typical downstream responses of PPAR α agonism including PPAR α upregulation, induction of target genes (e.g., *Acadm*, *Cpt1a*, *Fabp3*, and *Slc25a20*) and attenuation of cell migration.¹⁷ Having identified a new class of PPAR α agonists and demonstrating useful SAR, the purpose of the work presented herein was to assess the preliminary PK and *in vivo* protection against vascular leakage of A91 to establish proof-of-concept and develop second-generation derivatives aimed at improving potency and selectivity of this chemotype.

RESULTS AND DISCUSSION

PK/PD Profile of First-Generation Lead.

To gain insight about the metabolic stability and rate of clearance, both of which influence bioavailability, A91 was assessed for intrinsic clearance in human and rat (Sprague-Dawley) liver microsomes. As shown in Figure 2A, compound A91 proved to be rather stable, with the mean half-life ($t_{1/2}$) exceeding 60 minutes in both human and rat microsomes. These results indicate that this chemotype is likely to exhibit low clearance and that phase-I hepatic metabolism is unlikely to limit bioavailability.

In parallel, A91 was assessed for evidence of time-dependent inhibition against the major drug metabolizing cytochrome P450 isoforms (i.e., 1A, 2C9, 2C19, 2D6, and 3A). In this assay, peak areas corresponding to the metabolite of control substrates were recorded in the presence or absence of A91. This analysis was completed in the presence and absence of NADPH, to reveal time-dependence of inhibition. Test compounds which exhibit higher inhibition of metabolite formation in the absence of NADPH are deemed time-dependent inhibitors. As shown in Figure 2B, A91 exhibits no evidence of time-dependent inhibition for any of the CYP450s evaluated. This observation is especially important for CYP3A, as this family is responsible for the metabolism of statins, a drug class required for a majority of diabetics. These results indicate that CYP drug-drug interactions due to irreversible or tight binding are unlikely with this chemotype.

Lastly, A91 was evaluated for inhibition of the human ether-á-go-go (hERG) potassium channel. Inhibition of this channel is a flagship sign of cardiac toxicity. For this experiment, the automated whole cell patch-clamp (Qpatch 48) technique was used to record outward potassium currents from a single cell in the presence and absence of the compound of interest. As shown in Figure 2C, compound A91 exhibits low potency for the hERG channel (compounds that exhibit 50% inhibition at 10 μ M are flagged as toxic), indicating that this chemotype is unlikely to exhibit hERG-related cardiotoxicity, especially if potency for PPAR α is improved. It is also worth noting that A91 exhibits no measurable cytotoxicity in the retinal pigment epithelium photoreceptor cell-line 661W at concentrations as high as 200 μ M as measured in a colorimetric enzyme coupled LDH detection cytotoxicity assay (supplementary).

***In vivo* Efficacy of First-Generation Lead.**

With a promising initial PK profile, and cellular potency and selectivity surpassing Fenofibrate, we advanced A91 to proof-of-concept *in vivo* efficacy studies in a well-established streptozotocin (STZ)-induced rat model of DR, which assesses the ability of a compound of interest to attenuate retinal vascular leakage—a major culprit behind diabetic macular edema and consequential vision loss. As shown in Figure 3A, systemic administration (i.p.) of compound A91 reduces retinal vascular leakage in diabetic rats to the range of the non-diabetic cohort at relatively the same dose as Fenofibrate. Of interesting note, A91 seems to lack signs of hepatomegaly (Figure 3B), a common side-effect of Fenofibrate/Fenofibrate observed in rodent models but not in humans. These *in vivo* results paired with the initial PK assessment, demonstrate that A91 1) exhibits *in vivo* efficacy in a relevant DR model following systemic administration, 2) crosses the blood-retinal barrier and reaches the retina, 3) is bioavailable, 4) survives first-pass metabolism and clearance mechanisms well enough to maintain efficacy, and 5) demonstrates a relatively safe profile (no observable toxicity) after daily i.p. injection for one-month.

Inspired by promising PK and *in vivo* results for A91, we commenced second-generation SAR studies with an aim of enhancing potency and isoform selectivity of this chemotype for PPAR α . Since no co-crystal structure of A91 existed at the onset of this work, we developed an *in-silico* model using the Schrodinger Drug Discovery Suite to gain insight into the putative binding mode and guide our evolution of this chemotype. For these studies we

selected PDB 2P54, a co-crystal structure of a known PPAR α selective agonist GW590735 bound to human PPAR α (hPPAR α). To validate our docking approach, constraints, and parameters, GW590735 was extracted, exposed to MM2 energy-minimization, and redocked into the hPPAR α ligand binding domain to ensure that the results reproduced the bound conformation of the ligand. The overlay of co-crystallized and docked GW590735 showed excellent congruence (RMSD = 0.155 Å). Maintaining the same constraints and parameters, A91 was docked into hPPAR α and the results were visually inspected for strategies to improve or introduce key interactions. Since our gatekeeper assay is a cell-based assessment and thus not based on direct binding affinity, scores were not used to prioritize ligands at this stage. Instead, visual inspection and structural modification hypotheses dictated the prioritization of derivatives selected for synthesis and assessment.

Our approach focused on three key principles. First, maintain a suitably positioned carboxylate (or isostere) to capture key hydrogen bonding interactions. Previous studies demonstrate the significance of hydrogen bond interactions between hPPAR α Ser280, Tyr314, His440, and Tyr464 on the AF-2 helix and the carboxylate of known PPAR α ligands.^{18–25} The ability to interact with all four of these residues is believed to be responsible for triggering full agonism of hPPAR α .¹⁹ Poorer agonists tend to only interact with some of these hydrogen-bonding partners. It is worth noting that while historical PPAR α literature suggests a carboxylic acid is required for activity (a pharmacophore), examples of PPAR α agonists lacking this functionality have been reported,^{13, 26} demonstrating the -COOH motif can be altered. Secondly, we wanted to maintain the N-benzyl linkage to provide a complementary geometry for the U-shaped architecture of the ligand-binding domain. Third, our *in-silico* studies consistently predicted the aryl tail (C-ring) to bind a hydrophobic pocket comprised of helices 6 and 7 and defined by Ile241, Leu247, Ala250, Leu254, and Ala333. Literature precedence demonstrates that improvement of this interaction with other PPAR α chemotypes has proven beneficial to enhancing ligand potency and selectivity.^{23–25}

Stage I SAR: Linker Extension, C-Ring Modification, Carboxylate Location.

The first stage of our SAR (Figure 4) focused on assessing various hydrophobic substituents on the C-ring, probing the effects of linker extension, and determining optimal location for the carboxylate motif (or isostere).

To generate stage I analogues, 4-hydroxybenzaldehyde (**1**) was coupled with various benzyl bromides (**2a-l**) to yield the corresponding 4-benzyloxy-benzaldehydes **3a-l** (Scheme 1A). These precursors were then coupled with 3-aminobenzoic acid (or 4-aminobenzoic acid in the case of **4b***) via reductive amination to obtain **4a-4l** and **4b***. Following a similar synthetic approach, **5a**, **5b**, **5b***, and **5f** that include the typical fibrate head-group were synthesized (Scheme 1A). The only difference in the generation of **5a**, **5b**, **5b*** and **5f** was that the carboxylate remained protected as the ethyl ester until being revealed in the final step following saponification. Our previous studies indicated that the 4-benzyloxy-benzylamino chemotype deviates from the fibrates in that incorporation of the classical α -gem-dimethyl carboxylate “head-group” leads to *pan*-agonism and decreases potency. While

these effects are obviously opposite to our goals, these analogues were generated to determine if this trend continued to propagate through this series.

To generate derivatives with extended linkers the synthesis began with the transformation of primary alcohols **6a-b** into the corresponding alkyl bromides **7a-b** (Scheme 1B). These precursors were then utilized to *O*-alkylate 4-hydroxybenzaldehyde to generate **8a-b**, which were coupled with 3-aminobenzoic acid through reductive amination to provide **4m** and **4n** that included an extra $-\text{CH}_2-$ between the B and C-rings (Scheme 1B). Analogue **4o** with an additional $-\text{CH}_2-$ between the A and B rings and the B and C rings was also generated. Chemoselective alkylation of tyrosol with **7b** afforded **9**, which was converted to the corresponding alkyl bromide **10** that was used to *N*-alkylate 3-aminobenzoic acid (Scheme 1B).

Stage I derivatives were evaluated for *in vitro* human PPAR α (hPPAR α) agonism in a commercially available hPPAR α luciferase cell reporter assay (Indigo Biosciences). This assay utilizes non-human mammalian cells engineered to express hPPAR α with the luciferase reporter gene functionally linked to a PPAR α -response promoter. Thus, quantifying changes in luciferase expression in compound treated reporter cells provides a sensitive method to quantify changes in PPAR α activity. As an initial first-pass assessment, all compounds were evaluated at two concentrations, 5 μM and 50 μM . This two-concentration approach is a cost-effective means to provide insight into 1) rough estimates of potency, 2) therapeutic window, and 3) relative levels of agonism. A 5:50 ratio $\ll 1.0$ indicates the compound is much more active at 50 μM than 5 μM , meaning the compound has poor potency. A 5:50 ratio ~ 1.0 indicates equal activity at both concentrations, meaning the EC_{50} is $< 5 \mu\text{M}$ and is non-cytotoxic to the cells at a 10-fold higher concentration, as cell death results in a decrease in luminescence production. As one would presume, a 5:50 ratio of > 1.0 indicates a potency $< 5 \mu\text{M}$, but also demonstrates toxicity at 50 μM . Therefore, 5:50 ratios of ~ 1.0 are desirable. GW590735, a highly selective (~ 500 -fold over PPAR γ and PPAR δ) PPAR α partial agonist was employed as a positive control.²⁴

In line with our hypothesis, the incorporation of lipophilic or electron withdrawing substituents on the C-ring of the benzoic acid chemotype (**4b-l**) provides an improvement in cellular activity in comparison to A91 (**4a**). Most analogues in the **4b-l** series exhibit 5:50 ratios > 0.5 , indicating EC_{50} values $< 5 \mu\text{M}$ (Table 1). Comparison of **4b** and **4b*** demonstrates that the 3-carboxylate is favored over the 4-carboxylate. Within this class, we advanced 8 derivatives to multi-point dose-response analysis in the same luciferase assay. Compounds were chosen based on a combination of interest, 5:50 ratio, and level of agonism. As shown in Table 1, the 4-F (**4b**), 4-Br (**4d**), 2,4-difluoro (**4j**), and 3,4-difluoro (**4k**) analogues all exhibit submicromolar activity and are ~ 2 – 3 fold more active than the 4-I (**4e**) and 4-CF $_3$ (**4f**). This result was somewhat forecasted by our *in-silico* results, which predicted that substituents larger than a bromine at the 4-position would be problematic. Even though the 3-F (**4i**) is less active than the 4-F (**4b**), it was still more active than A91. Thus, it is not surprising that inclusion of fluorine at both the 3- and 4-positions (**4k**) of this scaffold results in additive properties and is the most potent compound of this series. Inclusion of the typical fibrate head-group (**5a**, **5b**, **5b***, and **5f**) once again leads to decreased potency (Table 2) but often higher levels of agonism than the benzoic acid

relatives (**5a** vs **A91**, **5b** vs **4b**, and **5f** vs **4f**). As shown in Table 3, analogs containing extended linkers **4m–4o** exhibited poor potency (**4m** and **4n**) or poor activation (**4o**) in comparison to related non-homologated derivatives.

From the stage I cohort, analogues **4b** and **4j** were selected to move forward to selectivity assessment, wherein equivalent luciferase assays were conducted on isogenic cell-lines engineered to overexpress either hPPAR γ or hPPAR δ . Neither compound produced significant levels of hPPAR γ or hPPAR δ agonism up to 100 μ M compound concentration, indicating a selectivity of >125-fold (**4b**) and >175-fold (**4j**) for hPPAR α . Rosiglitazone, a known selective PPAR γ agonist, and GW0742, a known selective PPAR δ agonist, were employed as positive controls in these isogenic cell-lines.

Stage II SAR: Strategic Methylation, Metabolism Blocking, Carboxylate Isosteric Replacement.

From the stage I results, compound **4b** was selected as the base structure from this family for advancement. Three strategies were leveraged to advance the SAR in stage II 1) methylation at strategic positions within the scaffold that were expected to enhance complementarity with the shape and hydrophobicity of the PPAR α ligand binding pocket; 2) blockage of predicted sites of metabolism; and 3) isosteric replacement of the carboxylate.

In silico prediction of major sites of metabolism for A91 with Schrodinger's P450 Site of Metabolism software identified the A-ring as a potential liability for CYP450 oxidation (supplementary). As such, we synthesized **4p** and **4q** (Scheme 2) to interrogate the effects of fluorine blockage on activity. These two analogues were generated by coupling **3b** to 5-amino-2-fluorobenzoic acid (**11a**) or 5-amino-3-fluorobenzoic acid (**11b**) to produce **4p** and **4q**, respectively. The 5-fluoro variant (**4q**) exhibits a ~3-fold improvement in activity over the 4-fluoro derivative (**4p**) and is nearly equipotent to **4b** (Table 4). Interestingly, incorporation of a chlorine at the 5-position of the A-ring (**4r**) greatly decreases potency (5:50 ratio = 0.11), while incorporation of a bromine at the 5-position (**4s**) maintains potency in comparison to **4b** but introduces apparent cytotoxicity at elevated concentrations (5:50 ratio = 1.3). These results suggest that fluorine incorporation should be a viable strategy to block sites on the A-ring if proven to be problematic as this series progresses through development.

Inspired by frequently reported methyl effects in drug discovery,²⁷ analogues containing one or two methyl groups at different positions were synthesized (Scheme 3) using variations of chemistry discussed in Schemes 1 and 2 and evaluated in the luciferase assay (**4t–z**, Table 5). Specifically, we interrogated the effects of methylation on the 2- and 3-position of the B-ring, the 4-position of the A-ring, the nitrogen linker, and all possible variations on the benzylic position of the linker between the B- and C-rings (another predicted site of metabolic liability). Interestingly, installation of a methyl group at the 3-position of B-ring (**4u**) resulted in a drastic improvement in cellular activity, dropping the EC₅₀ to ~40 nM, a ~21-fold improvement from the *des*-methyl relative, **4b**, and improving the selectivity for the PPAR α isoform to >2,700-fold (Figure 5). However, methylation at alternate sites (**4t** and **4v–z**) failed to induce similar effects on the activity, indicating that the improvement

observed with **4u** is not likely to arise from an increase in general hydrophobicity but rather from improving aspects of target engagement.

While the reason for the drastic improvement in activity arising from incorporation of the methyl group at the 3-position of the B-ring is not obvious, we have interrogated this phenomenon briefly by *in silico* methods. Results from these studies suggest that methylation at the 3-position (**4u**) results in a binding shift that allows for improved hydrogen bonding with the Ser280, Tyr314, His440, and Tyr464 tetrad (Figure 6A). Additionally, **4u** is predicted to bind with the aromatic ring of the benzoic acid perpendicular to His440 thus allowing for T-shaped edge-to-face pi-interactions (Figure 6A). Methylation at the 2-position (**4t**), however, produces no such shift and **4t** overlays very well with the *des*-methyl analogue **4b** (Figure 6B). As such, it is not surprising that the activity of **4t** and **4b** are relatively similar.

Coupling **3b** to a variety of anilines containing carboxylic acid surrogates (**18a-e**) allowed us to assess the effect of isosteric replacement on PPAR α agonistic activity (**19a-e**, Scheme 4). As seen in Table 6, replacement of the carboxylate with any of the isosteres resulted in dramatic decreases in activity, as noted by either a low 5:50 ratio or low overall signal. The exception being the tetrazole (**19d**), which exhibits decreased potency but comparable activation levels to the original carboxylate. This result confirms that the carboxylic acid is a key contributor to triggering PPAR α agonism but can indeed be modified. As part of this series, the thio-variant of the fibrate head-group was also synthesized and assessed (**19f**). Interestingly, this derivative exhibits an attenuated level of agonism compared to the parent analogue, **5b**. Differences in agonism levels now observed between the benzoic acid, fibrate head-group, and the thio-fibrate head-group strongly suggest that manipulation of this motif may provide a useful means to tune the level of agonism induced by a chemotype. This could be important as it is still unknown what level of stimulation (partial or full-agonism) is optimal for PPAR α to have a therapeutic effect in the retina.

Additional Assessment of **4u**.

As the most potent analogue identified in stages I and II, **4u** was selected for additional assessment. To confirm target engagement, evaluate binding potency, and provide some insight into selectivity in a relevant human retinal cell line (MIO-M1, Müller cells) we subjected **4u** to a CETSA (cellular thermal shift assay).²⁸⁻²⁹ Unlike the luciferase assay, MIO-M1 cells are not engineered to specifically report engagement with PPAR α and thus provide a more realistic picture of compound behavior in a disease related cell setting. In an initial heat gradient, non-ligand associated PPAR α displayed a temperature-dependent decay, with T_{agg}(50) and T_{agg}(75) at 40.8 °C and 45.5 °C, respectively (Figure 7A). In a subsequent test, dose-dependent potency of the PPAR α agonists, A91 and **4u** were tested at 45.5 °C, the temperature point at which 75% of free PPAR α protein melted. As shown in Figure 7B, A91 improved PPAR α stability incrementally with an EC₅₀ of 0.3 μ M. On the other hand, **4u** rescued PPAR α stability by 3-fold with an EC₅₀ of 93 nM. The results for **4u** correlate well with the EC₅₀ value obtained from the luciferase assay (EC₅₀ = 37 nM), demonstrating consistency of this compound within different contexts and eluding to excellent intracellular selectivity.

Compound **4u** was assessed in the same PK studies as **4a** (A91). As seen in Figure 8, this analogue exhibits significantly improved stability in human and rat liver microsomes (Figure 8A), shows no signs of irreversible inhibition of the major drug metabolizing CYP450s (Figure 8B) and is not considered a hERG liability (Figure 8C). It is important to note that while this compound exhibits 10% inhibition of hERG at 10 μ M (the same as A91), this analogue is ~20-fold more potent than A91.

Stage III SAR: Combining Lessons Learned.

Due to information gained from the stage I and II cohorts we designed and synthesized a series that combined beneficial features from each stage. Utilizing established synthetic routes, compounds **20-33** were synthesized and evaluated in similar fashion to stage I and II derivatives (Scheme 5). As shown in Table 7, all stage III analogues exhibit submicromolar activity in the cell luciferase assay, with many derivatives revealing EC₅₀ values under 50 nM. Comparison of **24-29** to the corresponding *des*-methyl relatives demonstrates that the methyl effect is ~10–20-fold more potent for each derivative and thus exhibits exquisite consistency. Most notably, **22**, which contains an iodine at the 4-position on the C-ring, exhibits a 50-fold improvement in activity in comparison to the *des*-methyl B-ring relative. Based on the predicted shift in the binding pocket (Figure 6) we suspect that room in the C-ring pocket is subsequently freed, allowing space for the larger iodine substituent that is not as easily accommodated with the *des*-methyl compound, **4e**. Consistent with the effect observed between **4w** and **4b**, N-methylation of the linker results in ~2-fold decrease in potency (**4u** vs. **30**). Furthermore, comparison of **4u** and **31** reveals that incorporation of the 3-F on the A-ring results in a maintenance in potency, a phenomenon first observed between **4b** and **4q**. Overall, the results from stage III analogues confirm trends seen in stages I and II, corroborate *in silico* docking predictions, and provide compelling evidence that the SAR for this chemotype is consistent, a desirable yet often elusive property in lead optimization.

Thermodynamic Binding Profiles.

To determine thermodynamic parameters relating to the formation of PPAR α /ligand complexes, we subjected structurally related **4a**, **4b**, and **4u** along with GW590735 (as a known agonist) to isothermal titration calorimetry (ITC). These analogues were selected, as they only differ in single functionalities that produce significant changes in cellular activity and thus allowed for the correlation of binding thermodynamics to specific molecular features. Replacement of the C-ring 4-OMe (**4a**) with a 4-F (**4b**) produces a ~3-fold improvement in binding affinity, an observation consistent with cellular activity. Based on the thermodynamic profile (Figure 9, Table 5), the observed improvement in binding affinity is likely due to a 4.73 kcal/mol lower entropic penalty of binding for **4b**. Introduction of a methyl group on the B-ring (**4u**) dramatically improves (~38-fold) the binding affinity, also consistent with cellular results. As depicted in Figure 9, the incorporation of the methyl group at the 3-position of the B-ring produces a favorable $-T \Delta S$ term while also maintaining a favorable enthalpic contribution. The full thermodynamic binding profile of **4u** reveals enthalpically driven binding to PPAR α , helping to explain the significant jump in affinity.

Combined, these results suggest that the observed improvements in the cellular luciferase assay are likely due to improved binding profiles and not alterations in cellular permeability,

solubility, or other physicochemical properties that may affect activity. Likewise, the proximities of K_d and EC_{50} values provides compelling evidence that this chemotype exhibits high specificity for PPAR α . Attempts to obtain a K_d and thermodynamic binding profile for the fibrate headgroup containing member of this series (**5b**) failed to produce reproducible data due to solubility issues under assay conditions.

2.9 Conclusion—In conclusion, we have advanced our initial studies on the 4-benzyloxy-benzylamino chemotype as a novel PPAR α agonistic scaffold by providing proof-of-concept *in vivo* efficacy, preliminary pharmacokinetic assessment and the design, synthesis, and evaluation of second-generation analogues that provide highly potent and isoform selective leads. The data demonstrate that systemic A91 (**4a**) treatment (i.p. injection) reduces retinal vascular leakage in diabetic rats to the level of non-diabetic controls at relatively the same dose as Fenofibrate, without signs of hepatomegaly. A full PK profile of this chemotype has yet to be determined (due to resource limitations), limiting PK/PD correlations for the *in vivo* data in comparison to Fenofibrate. However, embedded in these results is confirmation that A91 1) exhibits *in vivo* efficacy in a relevant DR model following systemic administration, 2) crosses blood-retinal barrier and reaches the retina, 3) is bioavailable, 4) survives first-pass metabolism and clearance mechanisms well enough to reach the site of disease and maintain efficacy, and 5) demonstrates a relatively safe profile (no observable toxicity) after daily intraperitoneal injection for one-month.

Utilizing A91 as a template we leveraged structure-guided design to develop second generation derivatives focused on improving the potency, affinity, and selectivity for PPAR α (Figure 10). These efforts led to the discovery of **4u** and related compounds that reach cellular potencies <50 nM and exhibit >2,700-fold selectivity for PPAR α over other PPAR isoforms. Importantly, we discovered that incorporation of a methyl at the 3-position of the B-ring leads to dramatic improvements in target engagement, selectivity, and cellular activity. Cellular thermal shift analysis confirms tight binding of the top candidates to hPPAR α and isothermal titration calorimetry further confirms a much-improved binding affinity and thermodynamic profile for **4u** in comparison to A91 (**4a**). Initial pharmacokinetic assessment of **4u** reveals microsomal stability in both human and rat microsomes, no evidence of irreversible inhibition of major drug metabolizing CYP450 enzymes, and a low risk for hERG interactions at relevant concentrations.

Detailed determination of a full PK/PD profile of this chemotype and advanced *in vivo* studies are ongoing and will be communicated in due time. Priority studies moving forward will focus on determining drug exposure in the retina after systemic administration and determining a more comprehensive metabolic profile, including phase-II conjugations. The data reported within, however, emphasize the promise of the 4-benzyloxy-benzylamino chemotype as a PPAR α agonistic family with efficacy in the retina after systemic administration and identify a pipeline of candidates positioned for detailed PK/PD and pre-clinical evaluation.

EXPERIMENTAL SECTION

General Synthetic Information.

Starting materials, ACS grade DCM, methanol, ethyl acetate, toluene, anhydrous DMF, THF were purchased from TCI Chemicals, Oakwood, Alfa Aesar, Boron Molecular, Fisher Scientific or Sigma-Aldrich. All reactions requiring anhydrous conditions were run under a nitrogen atmosphere. NMR were collected on a 400, 500, and 600 MHz (specified below) Varian VNMRs Direct Drive spectrometer equipped with an indirect detection probe. NMR data were collected at 25 °C unless otherwise indicated. Pulse sequences were used as supplied by Varian VNMRJ 4.2 software. All NMR data was processed in MestReNova v11.0. High resolution mass spectrometry was obtained from and analyzed by the Mass Spectrometry Facility at the University of Oklahoma. All analogues evaluated in biological assays were >95% purity based on HPLC and NMR analyses.

Purity Analysis of Final Analogues.

Ultra-high-pressure liquid chromatography (UPLC) separation of samples (25 μ M) was performed using an Agilent 1290 UPLC system equipped with an Agilent Zorbax Eclipse Plus reverse-phase column (ODS-18; 50 mm x 2.1 mm, 1.8- μ m particle size). Mobile phase A consisted of 100% Optima® LC/MS submicron filtered water (Fisher Scientific). Mobile phase B consisted of 100% Optima® LC/MS acetonitrile. A binary gradient at 0.65 mL/min flow rate was applied as follows: 100% solvent A from 0 to 1 min linear gradient to 20% solvent B, linear gradient to 80% solvent B from 1 to 8 min, linear gradient to 100% solvent B from 8 to 9.1 min, 100% solvent B from 9.1 to 10.3 min, and linear gradient to return the mobile phase to 100% solvent A from 10.3 to 10.4 min, which was maintained for an additional 3 min before the next sample was injected. The UPLC column eluent was introduced into an Agilent 1290 Infinity Diode Array Detector. Absorbance was scanned from 190 to 400 nm. Absorbance intensity at 154 nm was quantitated by area integration. For accurate mass confirmation the eluent traveled in series from the DAD to an Agilent 6545 UHD Accurate Mass QTOF device equipped with an electrospray ionization source operated in negative ion mode. Nitrogen was used as a nebulizing gas (40 pounds/inch²) and a drying gas (325 °C and 10 L/min flow rate). Fragmentor, skimmer, and capillary voltages were 180, 45, and 2500 V, respectively. Data was collected with Mass Hunter Acquisition (B.08.00) and analyzed with Mass Hunter Qualitative (B.07.00).

General Procedure A (Reductive amination).

Requisite aniline (1.0 eq.) was dissolved in toluene (30 mL), and the requisite aldehyde (1.02 eq.) was added. The reaction mixture was heated for 2 h under reflux using a Dean–Stark trap. Subsequently, the solvent was evaporated under reduced pressure. The crude product was dissolved in THF (0.1 M) and cooled to 0 °C prior to the addition of neat sodium triacetoxyborohydride (2.0 eq.). The resulting reaction mixture was stirred for 15 min, after which acetic acid (0.01 eq.) was added to the reaction. The mixture was then allowed to stir for 12 h at 25 °C while monitoring by TLC. Upon complete consumption of the aniline, the reaction was quenched through slow addition of water. The mixture was extracted three times with EtOAc, the organic layers were combined and washed with brine. The organic layer was dried over anhydrous sodium sulfate, filtered, and evaporated under

vacuum. The product was purified by column chromatography (SiO₂) using dichloromethane/methanol (99.5:0.5) as the eluent.

General Procedure B (Ester saponification).

To a stirred solution of the requisite ethyl ester (1.0 eq.) in THF/MeOH/H₂O mixture (0.1 M, 3:1:1) was added lithium hydroxide monohydrate (5.0 eq.) and the reaction mixture was stirred for 12 h at 25 °C. The reaction mixture was concentrated to remove organic solvents. The aqueous layer was acidified to pH 2.0 with 1 N aq. HCl and then extracted with EtOAc. The organic layer was washed with water and brine. The resulting organic layer was dried over anhydrous Na₂SO₄, filtered, and evaporated under vacuum. The product was purified by column chromatography (SiO₂) using dichloromethane/methanol (99:1) as the eluent.

3-((4-((4-methoxybenzyl)oxy)benzyl)amino)benzoic acid (4a): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4a** was prepared following General Procedure A with aldehyde **3a** (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (210 mg, 80% yield). ¹H NMR (400 MHz, Chloroform-*d*) δ 7.46 (d, *J* = 7.7 Hz, 1H), 7.36 (d, *J* = 9.2 Hz, 3H), 7.27 (q, *J* = 6.3, 4.1 Hz, 4H), 6.94 (dd, *J* = 14.6, 8.1 Hz, 4H), 6.84 (d, *J* = 8.3 Hz, 1H), 4.99 (s, 2H), 4.30 (s, 2H), 3.82 (s, 3H) ppm; ¹³C NMR (101 MHz, Chloroform-*d*) δ 171.8, 159.6, 158.4, 148.3, 131.1, 130.2, 129.4, 129.3 (2C), 129.1, 128.9 (2C), 119.4, 118.2, 115.2 (2C), 114.2 (2C), 114.0, 70.0, 55.4, 47.8 ppm. HRESI *m/z*: 364.1449 (C₂₂H₂₁NO₄ + H⁺ requires 364.1543).

3-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4b): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4b** was prepared following General Procedure A with aldehyde **3b** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (40 mg, 32% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.48 (dd, *J* = 8.3, 5.5 Hz, 2H), 7.27 (d, *J* = 8.2 Hz, 2H), 7.21 (dd, *J* = 9.3, 8.9 Hz, 2H), 7.16 (dd, *J* = 1.9 Hz, 1H), 7.13 (dd, *J* = 7.7 Hz, 1H), 7.09 (d, *J* = 7.6 Hz, 1H), 6.96 (d, *J* = 8.7 Hz, 2H), 6.82 – 6.60 (m, 1H), 6.44 (t, *J* = 6.0 Hz, 1H), 5.04 (s, 2H), 4.21 (d, *J* = 5.8 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 161.7 (d, *J* = 243.5 Hz), 157.1, 148.7, 133.4 (d, *J* = 3.1 Hz), 131.9, 131.4, 129.8 (d, *J* = 8.4 Hz, 2C), 128.8, 128.3 (2C), 116.6, 116.3, 115.2 (d, *J* = 21.3 Hz, 2C), 114.6 (2C), 112.9, 68.4, 45.7. TOFMSESI *m/z*: 352.1357 (C₂₁H₁₈FNO₃ + H⁺ requires 352.1304).

4-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4b*): Reaction scale: 50 mg (0.36 mmol) of 4-amino benzoic acid. Compound **4b*** was prepared following General Procedure A with aldehyde **3b** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (60 mg, 52% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.75 (d, *J* = 8.8 Hz, 2H), 7.45 (dd, *J* = 8.5, 5.5 Hz, 2H), 7.27 (d, *J* = 8.5 Hz, 2H), 7.13 – 7.04 (m, 2H), 6.95 (d, *J* = 8.6 Hz, 2H), 6.59 (d, *J* = 8.8 Hz, 2H), 5.04 (s, 2H), 4.30 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 170.8, 168.3 (d, *J* = 244.5 Hz), 168.1, 166.5, 162.6, 159.3, 154.4, 133.0 (d, *J* = 3.2 Hz), 132.6 (2C), 130.7, 130.6 (2C), 129.6 (d, *J* = 8.6 Hz, 2C), 118.5, 116.3 (d, *J* = 21.6 Hz, 2C), 116.0 (2C), 112.5 (2C), 70.3, 47.3. TOFMSESI *m/z*: 352.1357 (C₂₁H₁₈FNO₃ + H⁺ requires 352.1304).

3-((4-((4-chlorobenzyl)oxy)benzyl)amino)benzoic acid (4c): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4c** was prepared following General Procedure A with aldehyde **3c** (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (203.3 mg, 76% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.41 (d, *J* = 8.5 Hz, 2H), 7.36 (d, *J* = 8.5 Hz, 2H), 7.29 (d, *J* = 6.3 Hz, 2H), 7.28 (s, 1H), 7.25 (dt, *J* = 7.6, 1.3 Hz, 1H), 7.15 (t, *J* = 7.8 Hz, 2H), 6.93 (d, *J* = 8.6 Hz, 1H), 6.82 (ddd, *J* = 8.0, 2.5, 1.1 Hz, 1H), 5.03 (s, 2H), 4.26 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 169.3, 157.6, 148.8, 136.3, 133.1, 132.0, 131.0, 128.7 (2C), 128.5, 128.2 (2C), 128.1 (2C), 117.6, 117.0, 114.5 (2C), 113.2, 68.7, 46.5. TOFMSESI *m/z*: 366.0897 (C₂₁H₁₈ClNO₃ - H⁺ requires 366.0902).

3-((4-((4-bromobenzyl)oxy)benzyl)amino)benzoic acid (4d): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4d** was prepared following General Procedure A with aldehyde **3d** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (96 mg, 65% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.68 (s, 1H), 7.57 (d, *J* = 8.5 Hz, 2H), 7.39 (d, *J* = 8.0 Hz, 2H), 7.27 (d, *J* = 8.2 Hz, 2H), 7.18 (dd, *J* = 2.0 Hz, 1H), 7.13 (dd, *J* = 7.6 Hz, 1H), 7.10 (ddd, *J* = 7.5, 1.5 Hz, 1H), 6.95 (d, *J* = 8.8 Hz, 2H), 6.78 (ddd, *J* = 8.0, 1.8 Hz, 1H), 6.44 (s, 1H), 5.05 (s, 2H), 4.21 (s, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 168.3, 157.4, 149.1, 137.1, 132.5, 131.9, 131.8 (2C), 130.2 (2C), 129.3, 128.8 (2C), 121.3, 117.1, 116.7, 115.1 (2C), 113.4, 68.8, 46.2. TOFMSESI *m/z*: 412.0512 (C₂₁H₁₈BrNO₃ + H⁺ requires 412.0543).

3-((4-((4-iodobenzyl)oxy)benzyl)amino)benzoic acid (4e): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4e** was prepared following General Procedure A with aldehyde **3e** (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (210 mg, 80% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.74 (d, *J* = 8.3 Hz, 2H), 7.27 (d, *J* = 8.4 Hz, 2H), 7.24 (d, *J* = 8.0 Hz, 2H), 7.17 (dd, *J* = 1.9 Hz, 1H), 7.14 (dd, *J* = 7.7 Hz, 1H), 7.13 – 7.07 (m, 1H), 6.94 (d, *J* = 8.6 Hz, 2H), 6.78 (ddd, *J* = 8.3, 1.7 Hz, 1H), 5.03 (s, 2H), 4.20 (s, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.7, 157.0, 148.6, 137.2 (2C), 137.0, 132.0, 131.2, 129.8 (2C), 128.8, 128.3 (2C), 116.7, 116.4, 114.6 (2C), 112.9, 93.7, 68.4, 45.7. TOFMSESI *m/z*: 460.0424 (C₂₁H₁₈INO₃ + H⁺ requires 460.0404).

3-((4-((4-(trifluoromethyl)benzyl)oxy)benzyl)amino)benzoic acid (4f): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4f** was prepared following General Procedure A with aldehyde **3f** (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (170 mg, 58% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 7.75 (d, *J* = 8.1 Hz, 2H), 7.65 (d, *J* = 8.1 Hz, 2H), 7.28 (d, *J* = 8.7 Hz, 2H), 7.16 (dd, *J* = 2.6, 1.9 Hz, 1H), 7.13 (dd, *J* = 8.5, 7.7 Hz, 1H), 7.09 (ddd, *J* = 7.6, 1.6 Hz, 1H), 6.98 (d, *J* = 8.7 Hz, 2H), 6.78 (ddd, *J* = 8.1, 2.6, 1.2 Hz, 1H), 6.45 (t, *J* = 6.0 Hz, 1H), 5.20 (s, 2H), 4.21 (d, *J* = 5.8 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 156.9, 148.7, 142.1, 132.1, 131.3, 128.8, 128.4 (2C), 128.2 (q, *J* = 29.2, 30.8, 33.0 Hz), 127.9 (2C), 125.3 (q, *J* = 3.6 Hz, 2C), 124.2 (q, *J* = 271.2,

271.8, 272.3 Hz), 116.6, 116.3, 114.6 (2C), 112.8, 68.2, 45.7. TOFMSESI m/z: 400.1139 (C₂₂H₁₈F₃NO₃ - H⁺ requires 400.1166).

3-((4-((4-cyanobenzyl)oxy)benzyl)amino)benzoic acid (4g): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4g** was prepared following General Procedure A with aldehyde **3g** (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (110 mg, 42% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.63 (s, 1H), 7.84 (d, *J* = 8.2 Hz, 2H), 7.62 (d, *J* = 7.9 Hz, 2H), 7.28 (d, *J* = 8.3 Hz, 2H), 7.19 (dd, *J* = 2.0 Hz, 1H), 7.14 (dd, *J* = 7.7 Hz, 1H), 7.12 (d, *J* = 8.7 Hz, 1H), 7.00 – 6.90 (m, 2H), 6.79 (dt, *J* = 7.8, 2.0 Hz, 1H), 6.45 (s, 1H), 5.18 (s, 2H), 4.22 (s, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.9, 156.8, 148.7, 143.0, 132.4 (2C), 132.2, 131.3, 128.9, 128.4 (2C), 128.0 (2C), 118.8, 116.7, 116.4, 114.7 (2C), 112.9, 110.4, 68.2, 45.8. TOFMSESI m/z: 359.1406 (C₂₂H₁₈N₂O₃ + H⁺ requires 359.1390).

3-((4-((2-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4h): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4h** was prepared following General Procedure A with aldehyde **3h** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (122 mg, 97% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 7.54 (t, *J* = 7.4 Hz, 1H), 7.41 (q, *J* = 6.4 Hz, 1H), 7.29 (d, *J* = 8.2 Hz, 2H), 7.24 (d, *J* = 10.1 Hz, 1H), 7.22 (d, *J* = 7.4 Hz, 1H), 7.18 (s, 1H), 7.14 (t, *J* = 7.7 Hz, 1H), 7.10 (d, *J* = 7.6 Hz, 1H), 6.98 (d, *J* = 8.5 Hz, 2H), 6.79 (d, *J* = 7.8 Hz, 1H), 6.44 (s, 1H), 5.10 (s, 2H), 4.22 (d, *J* = 4.4 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 160.4 (d, *J* = 245.9 Hz), 157.1, 148.7, 132.1, 131.3, 130.6 (d, *J* = 4.1 Hz), 130.3 (d, *J* = 8.3 Hz), 128.9, 128.4 (2C), 124.5 (d, *J* = 3.5 Hz), 123.9 (d, *J* = 14.7 Hz), 116.7, 116.3, 115.4 (d, *J* = 21.1 Hz), 114.6 (2C), 112.9, 63.5 (d, *J* = 3.4 Hz), 45.8. TOFMSESI m/z: 374.1169 (C₂₁H₁₈FNO₃ + Na⁺ requires 374.1163).

3-((4-((3-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4i): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4i** was prepared following General Procedure A with aldehyde **3i** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (96 mg, 65% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.61 (s, 1H), 7.46 – 7.38 (m, 1H), 7.29 – 7.28 (m, 2H), 7.28 – 7.27 (m, 1H), 7.26 – 7.24 (m, 1H), 7.20 – 7.17 (m, 1H), 7.16 – 7.14 (m, 1H), 7.14 – 7.12 (m, 1H), 7.13 – 7.08 (m, 1H), 6.99 – 6.95 (m, 2H), 6.81 – 6.75 (m, 1H), 6.44 (t, *J* = 5.8 Hz, 1H), 5.10 (s, 2H), 4.21 (d, *J* = 4.4 Hz, 2H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 167.8, 162.2 (d, *J* = 243.5 Hz), 157.0, 148.7, 140.2 (d, *J* = 7.3 Hz), 132.0, 131.3, 130.4 (d, *J* = 8.3 Hz), 128.9, 128.4 (2C), 123.4 (d, *J* = 2.8 Hz), 116.7, 116.4, 114.7 (2C), 114.5 (d, *J* = 20.9 Hz), 114.1 (d, *J* = 21.9 Hz), 112.9, 68.3, 45.8. TOFMSESI m/z: 352.1360 (C₂₁H₁₈FNO₃ + H⁺ requires 352.1343).

3-((4-((2,4-difluorobenzyl)oxy)benzyl)amino)benzoic acid (4j): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4j** was prepared following General Procedure A with aldehyde **3j** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (100

mg, 70% yield). ^1H NMR (600 MHz, DMSO- d_6) δ 12.62 (s, 1H), 7.60 (ddd, J = 8.6, 6.7 Hz, 1H), 7.32 – 7.27 (m, 1H), 7.29 – 7.26 (m, 2H), 7.17 (dd, J = 1.9 Hz, 1H), 7.14 (d, J = 7.7 Hz, 1H), 7.13 – 7.11 (m, 1H), 7.10 – 7.09 (m, 1H), 6.98 (d, J = 8.6 Hz, 2H), 6.83 – 6.73 (m, 1H), 6.46 (t, J = 6.3 Hz, 1H), 5.07 (s, 2H), 4.22 (d, J = 4.7 Hz, 2H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 167.8, 161.4 (dd, J = 4.0, 12.6 Hz), 161.4 (dd, J = 12.1, 499.9 Hz), 157.0, 148.7, 132.2, 132.1 (dd, J = 5.7, 10.1 Hz), 131.3, 128.8, 128.4 (2C), 120.4 (dd, J = 3.4, 14.4 Hz), 116.7, 116.3, 114.6 (2C), 112.9, 111.5 (dd, J = 3.8, 21.3 Hz), 104.0 (t, J = 25.8 Hz), 63.1 (d, J = 2.9 Hz), 45.7. TOFMSESI m/z : 392.1085 ($\text{C}_{21}\text{H}_{17}\text{F}_2\text{NO}_3 + \text{Na}^+$ requires 392.1069).

3-((4-((3,4-difluorobenzyl)oxy)benzyl)amino)benzoic acid (4k): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4k** was prepared following General Procedure A with aldehyde **3k** (0.37 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (106 mg, 79% yield). ^1H NMR (600 MHz, DMSO- d_6) δ 12.61 (s, 1H), 7.51 (ddd, J = 11.6, 7.8, 2.0 Hz, 1H), 7.44 (ddd, J = 10.8, 8.4 Hz, 1H), 7.30 (d, J = 4.0 Hz, 1H), 7.27 (d, J = 8.4 Hz, 2H), 7.16 (dd, J = 1.9 Hz, 1H), 7.13 (dd, J = 7.7 Hz, 1H), 7.09 (ddd, J = 7.6, 1.3 Hz, 1H), 6.96 (d, J = 8.7 Hz, 2H), 6.77 (ddd, J = 8.0, 2.6, 1.2 Hz, 1H), 6.45 (t, J = 6.1 Hz, 1H), 5.06 (s, 2H), 4.21 (d, J = 5.7 Hz, 2H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 167.8, 156.9, 149.3 (dd, J = 12.7, 245.6 Hz), 148.9 (dd, J = 12.4, 245.3 Hz), 148.7, 135.0 (dd, J = 3.6, 5.8 Hz), 132.1, 131.3, 128.8, 128.3 (2C), 124.4 (dd, J = 3.5, 6.5 Hz), 117.5 (d, J = 17.1 Hz), 116.7 (d, J = 17.5 Hz), 116.6, 116.3, 114.6 (2C), 112.9, 67.8, 45.7. TOFMSESI m/z : 370.1238 ($\text{C}_{21}\text{H}_{17}\text{F}_2\text{NO}_3 + \text{H}^+$ requires 370.1249).

3-((4-((3,5-difluorobenzyl)oxy)benzyl)amino)benzoic acid (4l): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **4l** was prepared following General Procedure A with aldehyde **3l** (0.74 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (196 mg, 73% yield). ^1H NMR (600 MHz, DMSO- d_6) δ 12.61 (s, 1H), 7.28 (d, J = 8.3 Hz, 2H), 7.19 (dd, J = 2.4, 9.1 Hz, 1H), 7.17 (dd, J = 1.8 Hz, 2H), 7.15 (d, J = 3.7 Hz, 1H), 7.13 (d, J = 7.7 Hz, 1H), 7.10 (ddd, J = 1.4, 7.6 Hz, 1H), 6.97 (d, J = 8.6 Hz, 2H), 6.78 (ddd, J = 1.2, 2.5, 8.0 Hz, 1H), 6.45 (t, J = 5.8 Hz, 1H), 5.11 (s, 2H), 4.22 (d, J = 5.1 Hz, 2H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 167.8, 162.4 (dd, J = 13.0, 246.5 Hz, 2C), 156.7, 148.7, 141.9 (t, J = 9.2 Hz), 132.2, 131.3, 128.8, 128.4 (2C), 116.7, 116.3, 114.7 (2C), 112.9, 110.3 (dd, J = 5.2, 20.3 Hz, 2C), 103.0 (t, J = 25.8 Hz), 67.8, 45.7. TOFMSESI m/z : 368.1120 ($\text{C}_{21}\text{H}_{17}\text{F}_2\text{NO}_3 - \text{H}^+$ requires 368.1104).

3-((4-phenethoxybenzyl)amino)benzoic acid (4m): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4m** was prepared following General Procedure A with aldehyde **8a** (0.37 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (121 mg, 96% yield). ^1H NMR (600 MHz, Methanol- d_4) δ 7.39 (s, 1H), 7.28 (t, J = 2.0 Hz, 1H), 7.27 (s, 2H), 7.26 (d, J = 3.5 Hz, 2H), 7.24 (s, 2H), 7.19 (h, J = 4.1 Hz, 1H), 7.14 (t, J = 7.8 Hz, 1H), 6.84 (d, J = 8.6 Hz, 2H), 6.81 (dd, J = 1.9, 8.4 Hz, 1H), 4.23 (s, 2H), 4.13 (t, J = 6.9 Hz, 2H), 3.02 (t, J = 6.9 Hz, 2H). ^{13}C NMR (151 MHz, Methanol- d_4) δ 169.3, 157.9, 148.8, 138.5, 131.6, 131.1,

128.6 (2C), 128.5, 128.2 (2C), 128.0 (2C), 125.9 (2C), 117.6, 117.0, 114.2, 113.3, 68.5, 46.7, 35.3. TOFMSESI m/z: 346.1443 (C₂₂H₂₁NO₃ - H⁺ requires 346.1449).

3-((4-(4-methoxyphenoxy)benzyl)amino)benzoic acid (4n): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4n** was prepared following General Procedure A with aldehyde **8b** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (130 mg, 96% yield). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.28 (d, *J* = 2.2 Hz, 1H), 7.25 (d, *J* = 8.1 Hz, 2H), 7.25 (s, 1H), 7.18 (d, *J* = 8.2 Hz, 2H), 7.14 (t, *J* = 7.8 Hz, 1H), 6.84 (d, *J* = 8.8 Hz, 2H), 6.83 (d, *J* = 8.7 Hz, 2H), 6.83 – 6.78 (m, 1H), 4.24 (s, 2H), 4.09 (t, *J* = 6.9 Hz, 2H), 3.75 (s, 3H), 2.96 (t, *J* = 6.9 Hz, 2H). ¹³C NMR (151 MHz, Methanol-*d*₄) δ 170.7, 159.8, 159.4, 150.3, 133.0, 132.5, 131.8, 130.9 (2C), 129.9, 129.6 (2C), 119.0, 118.4, 115.7 (2C), 114.9 (2C), 114.8, 70.1, 55.7, 48.1, 35.9. TOFMSESI m/z: 378.1709 (C₂₃H₂₃NO₄ + H⁺ requires 378.1700).

3-((4-(4-methoxyphenoxy)phenethyl)amino)benzoic acid (4o): 3-amino benzoic acid (50 mg, 0.36 mmol) and compound **10** (113 mg, 0.37 mmol) were dissolved in DMF. Triethylamine (102 uL, 0.72 mmol) was added at 25 °C and the reaction was stirred for 12 h upon which the reaction was deemed complete by TLC. The reaction was quenched with aq. 1N HCl (3 mL). The solution was diluted with EtOAc (30 mL). The organic layer was washed with water (3 × 20 mL) and brine (3 × 20 mL). The organic layer was dried over Na₂SO₄, filtered, and concentrated under vacuum. The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (36 mg, 25% yield). ¹H NMR (500 MHz, Methanol-*d*₄) δ 7.36 – 7.29 (m, 1H), 7.27 (ddd, *J* = 7.7, 1.6, 1.1 Hz, 1H), 7.19 (d, *J* = 6.6 Hz, 2H), 7.17 (d, *J* = 6.7 Hz, 2H), 7.14 (d, *J* = 7.7 Hz, 1H), 6.90 (ddd, *J* = 8.0, 2.4, 1.1 Hz, 1H), 6.84 (t, *J* = 2.0 Hz, 2H), 6.83 (t, *J* = 2.1 Hz, 2H), 4.41 (t, *J* = 6.9 Hz, 2H), 4.09 (t, *J* = 6.9 Hz, 2H), 3.75 (s, 3H), 2.97 (t, *J* = 7.0 Hz, 2H), 2.96 (t, *J* = 6.9 Hz, 2H). ¹³C NMR (126 MHz, Methanol-*d*₄) δ 168.5, 159.8, 159.1, 149.3, 132.2, 131.8, 131.4, 131.0 (2C), 130.9 (2C), 130.1, 120.8, 119.8, 116.7, 115.7 (2C), 114.8 (2C), 70.1, 66.8, 55.6, 35.9, 35.3. TOFMSESI m/z: 392.1870 (C₂₄H₂₅NO₄ + H⁺ requires 392.1856).

2-fluoro-5-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4p): Reaction scale: 50 mg (0.32 mmol) of 5-amino-2-fluorobenzoic acid (**11a**). Compound **4p** was prepared following the General Procedure A with aldehyde **3b** (0.33 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (98 mg, 82% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.48 (dd, *J* = 8.5, 5.7 Hz, 2H), 7.26 (d, *J* = 8.4 Hz, 2H), 7.21 (t, *J* = 8.9 Hz, 1H), 7.01 – 6.97 (m, 1H), 7.00 – 6.93 (m, 3H), 6.75 (dt, *J* = 8.8, 3.5 Hz, 1H), 6.31 (s, 1H), 5.05 (s, 2H), 4.17 (s, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 165.6, 161.7 (d, *J* = 243.5 Hz), 157.1, 152.9 (d, *J* = 244.7 Hz), 144.9, 133.4, 131.8, 129.9 (d, *J* = 8.2 Hz, 2C), 128.4 (2C), 117.3 (d, *J* = 7.4 Hz), 116.9 (d, *J* = 23.3 Hz, 2C), 115.3, 115.1, 114.7 (2C), 113.7, 68.4, 46.2. TOFMSESI m/z: 368.1093 (C₂₁H₁₇F₂NO₃ - H⁺ requires 368.1104).

3-fluoro-5-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4q): Reaction scale: 50 mg (0.32 mmol) of 3-amino-5-fluorobenzoic acid (**11b**). Compound **4q** was

prepared following the General Procedure A with aldehyde **3b** (0.33 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (102 mg, 86% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.45 – 7.37 (m, 2H), 7.26 (d, *J* = 8.4 Hz, 2H), 7.11 (t, *J* = 1.8 Hz, 1H), 7.10 – 7.00 (m, 2H), 6.96 – 6.89 (m, 2H), 6.87 (dt, *J* = 9.1, 1.9 Hz, 1H), 6.47 (dt, *J* = 11.4, 2.3 Hz, 1H), 4.98 (s, 2H), 4.22 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 169.4 (d, *J* = 3.3 Hz), 165.1 (d, *J* = 241.1 Hz), 163.8 (d, *J* = 244.7 Hz), 159.2, 152.0 (d, *J* = 11.1 Hz), 134.8 (d, *J* = 3.2 Hz), 134.1 (d, *J* = 9.9 Hz), 132.8, 130.6 (d, *J* = 8.1 Hz, 2C), 129.6 (2C), 116.2, 116.0 (2C), 111.1 (d, *J* = 2.0 Hz), 104.5 (d, *J* = 24.0 Hz, 2C), 103.9 (d, *J* = 26.0 Hz), 70.3, 47.8. TOFMSESI *m/z*: 368.1096 (C₂₁H₁₇F₂NO₃ - H⁺ requires 368.1104).

3-chloro-5-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4r): Reaction scale: 50 mg (0.29 mmol) of 3-amino-5-chlorobenzoic acid (**11c**). Compound **4r** was prepared following the General Procedure A with aldehyde **3b** (0.30 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (88 mg, 78% yield). ¹H NMR (500 MHz, Methanol-*d*₄) δ 7.42 (dd, *J* = 8.4, 5.4 Hz, 2H), 7.26 (d, *J* = 8.3 Hz, 2H), 7.17 (dt, *J* = 16.7, 1.7 Hz, 2H), 7.07 (t, *J* = 8.8 Hz, 2H), 6.93 (d, *J* = 8.5 Hz, 2H), 6.75 (t, *J* = 2.1 Hz, 1H), 5.00 (s, 2H), 4.23 (s, 2H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 166.6, 161.7 (d, *J* = 243.5 Hz), 157.2, 150.1, 133.7, 133.3 (d, *J* = 3.0 Hz), 133.0, 131.3, 129.9, 129.8, 128.4 (2C), 115.5, 115.3, 115.1, 114.8, 114.7 (2C), 112.0, 68.5, 45.6. TOFMSESI *m/z*: 384.0812 (C₂₁H₁₇ClFNO₃ - H⁺ requires 384.0808).

3-bromo-5-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (4s): Reaction scale: 50 mg (0.23 mmol) of 3-amino-5-bromobenzoic acid (**11d**). Compound **4s** was prepared following the General Procedure A with aldehyde **3b** (0.23 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a pale amorphous solid (85 mg, 86% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.42 (dd, *J* = 8.5, 5.4 Hz, 2H), 7.29 (d, *J* = 1.7 Hz, 1H), 7.26 (d, *J* = 8.3 Hz, 2H), 7.21 (t, *J* = 1.8 Hz, 1H), 7.07 (t, *J* = 8.8 Hz, 2H), 6.93 (d, *J* = 8.4 Hz, 2H), 6.91 (t, *J* = 2.1 Hz, 1H), 5.00 (s, 2H), 4.23 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 169.1, 165.0, 161.9 (d, *J* = 146.3 Hz), 159.3, 151.5, 134.8, 134.2, 132.7, 130.6 (d, *J* = 8.2 Hz), 129.6 (2C), 123.7, 120.9, 119.9, 116.2, 116.04, 116.01 (2C), 113.6, 70.29, 47.67. TOFMSESI *m/z*: 428.0295 (C₂₁H₁₇BrFNO₃ - H⁺ requires 428.0303).

3-((4-((4-fluorobenzyl)oxy)-2-methylbenzyl)amino)benzoic acid (4t): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **4t** was prepared following General Procedure A with aldehyde **12a** (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (100 mg, 70% yield). ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.53 (s, 1H), 7.48 (dd, *J* = 8.5, 5.7 Hz, 2H), 7.25 – 7.16 (m, 2H), 7.18 (s, 1H), 7.17 (d, *J* = 4.0 Hz, 1H), 7.14 (d, *J* = 7.6 Hz, 1H), 7.11 (dt, *J* = 7.6, 1.5 Hz, 1H), 6.86 (d, *J* = 2.6 Hz, 1H), 6.79 (td, *J* = 8.6, 2.3 Hz, 2H), 6.31 – 5.95 (m, 1H), 5.05 (s, 2H), 4.15 (d, *J* = 4.5 Hz, 2H), 2.30 (s, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 167.7, 157.1, 148.8, 148.1 (d, *J* = 330.2 Hz), 137.3, 133.4 (d, *J* = 2.9 Hz), 131.3, 129.7, 129.6, 129.4, 128.7 (d, *J* = 7.3 Hz, 2C), 116.7, 116.5, 116.1, 115.1 (d, *J* = 21.4

Hz, 2C), 112.5, 111.6, 68.3, 44.2, 18.7. TOFMSESI m/z: 364.1360 (C₂₂H₂₀FNO₃ - H⁺ requires 364.1354).

3-((4-((4-fluorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (4u): Reaction scale: 100 mg (0.72 mmol) of 3-amino benzoic acid. Compound **4u** was prepared following General Procedure A with aldehyde **12b** (0.73 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (200 mg, 75% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 7.49 (dd, *J* = 8.5, 5.7 Hz, 2H), 7.21 (dd, *J* = 8.9 Hz, 2H), 7.17 (dd, *J* = 2.0 Hz, 1H), 7.15 (d, *J* = 2.4 Hz, 1H), 7.14 (s, 1H), 7.11 (s, 1H), 7.10 (ddd, *J* = 7.5, 1.4 Hz, 1H), 6.95 (d, *J* = 8.3 Hz, 1H), 6.82 – 6.67 (m, 1H), 5.06 (s, 2H), 4.17 (s, 2H), 2.16 (s, 3H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.7, 161.6 (d, *J* = 243.3 Hz), 155.1, 148.7, 133.7 (d, *J* = 2.9 Hz), 131.5, 131.2, 129.5, 129.4 (d, *J* = 8.2 Hz, 2C), 128.8, 125.8, 125.6, 116.6, 116.3, 115.2 (d, *J* = 21.3 Hz, 2C), 112.8, 111.6, 68.5, 45.8, 16.2. TOFMSESI m/z: 388.1328 (C₂₂H₂₀FNO₃ + Na⁺ requires 388.1319).

3-((4-((4-fluorobenzyl)oxy)benzyl)amino)-4-methylbenzoic acid (4v): Reaction scale: 50 mg (0.33 mmol) of 3-amino-4-methylbenzoic acid. Compound **4v** was prepared following the General Procedure A with aldehyde **3b** (0.34 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (111 mg, 92% yield). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.44 (t, *J* = 6.7 Hz, 2H), 7.31 (d, *J* = 8.1 Hz, 2H), 7.26 (s, 1H), 7.21 (d, *J* = 7.8 Hz, 1H), 7.08 (t, *J* = 8.6 Hz, 2H), 7.01 (d, *J* = 7.7 Hz, 1H), 6.95 – 6.91 (m, 2H), 5.02 (s, 2H), 4.36 (s, 2H), 2.18 (s, 3H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.4, 161.7 (d, *J* = 243.6 Hz), 157.0, 145.9, 133.4 (d, *J* = 3.0 Hz), 132.2, 129.9, 129.8, 129.4, 128.0 (2C), 127.9, 126.2, 117.1, 115.3, 115.0, 114.6 (2C), 114.4, 110.0, 68.4, 45.7, 17.8. TOFMSESI m/z: 364.1342 (C₂₂H₂₀FNO₃ - H⁺ requires 364.1354).

3-((4-((4-fluorobenzyl)oxy)benzyl)(methyl)amino)benzoic acid

(4w): Paraformaldehyde (5 mg, 0.16 mmol), compound **4b** (30 mg, 0.08 mmol), sodium triacetoxyborohydride (6.2 mg, 0.096 mmol) and 2 mL of ethanol were combined in a 10 mL of round bottom flask with a reflux condenser. The mixture was heated for 2 h at 100 °C and subsequently cooled to 25 ° for 12 h. The solvent was removed under vacuum and the residue was dissolved in EtOAc (60 mL). The organic layer was washed with aq. 1N HCl (20 mL), water (20 mL), and brine (20 mL). The organic layer was dried over Na₂SO₄, filtered, and concentrated under vacuum. The compound was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (13 mg, 43% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.4 (d, *J* = 5.5 Hz, 1H), 7.4 – 7.4 (m, 2H), 7.3 (dt, *J* = 1.3, 7.6 Hz, 1H), 7.2 (t, *J* = 7.9 Hz, 1H), 7.1 (d, *J* = 8.6 Hz, 2H), 7.1 (t, *J* = 8.8 Hz, 2H), 7.0 (ddd, *J* = 1.1, 2.8, 8.3 Hz, 1H), 6.9 (d, *J* = 8.7 Hz, 2H), 5.0 (s, 2H), 4.5 (s, 2H), 3.0 (s, 3H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 170.7, 163.8 (d, *J* = 244.4 Hz), 159.2, 151.1, 134.8 (d, *J* = 3.3 Hz), 132.4, 132.3, 130.6 (d, *J* = 8.3 Hz, 2C), 130.0, 129.2 (2C), 118.8, 118.2, 116.1 (d, *J* = 21.7 Hz, 2C), 116.0 (2C), 114.5, 70.3, 56.8, 38.9. TOFMSESI m/z: 388.1328 (C₂₅H₂₆FNO₄ + H requires 388.1319). TOFMSESI m/z: 366.1509 (C₂₂H₂₀FNO₃ + H⁺ requires 366.1500).

(R)-3-((4-(1-(4-fluorophenyl)ethoxy)benzyl)amino)benzoic acid (4x): Compound **13** (53 mg, 0.15 mmol), (*S*)-1-(4-fluorophenyl)ethan-1-ol (21 mg, 0.15 mmol), and PPh₃ (39 mg, 0.15 mmol) were dissolved in toluene (2.5 mL) and cooled to 0 °C. A solution of DIBAD (35 mg, 0.15 mmol) in toluene (5 mL) was added dropwise to the mixture. The reaction was warmed to 25 °C and stirred at for 12 h, at which time the reaction was deemed complete by TLC. The solvent was removed under reduced pressure. The residue was purified by flash column chromatography (SiO₂, hexanes/ethyl acetate = 2: 1) to provide the desired product **14** as colorless oil. The resulting product **14** (50 mg, 0.114 mmol) was dissolved in a THF/MeOH/H₂O (3:1:1) mixture and LiOH-added was (mmol 0.57, mg 24) H₂O. The reaction stirred at 25 °C for 12 h upon which the reaction was deemed complete by TLC. The reaction was quenched with aq. 1N HCl (3 mL) and extracted with EtOAc (3 × 10 mL). The organic layers were combined dried over Na₂SO₄, filtered and concentrated by vacuum. The product **4x** was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (25 mg, 65% yield). $[\alpha]_D^{20} = +22.0$ (c = 9.1 mg/mL, l = 0.5 dm, MeOH). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.35 (dd, *J* = 8.6, 5.4 Hz, 2H), 7.29 – 7.24 (m, 1H), 7.26 – 7.23 (m, 1H), 7.19 – 7.15 (m, 2H), 7.15 – 7.09 (m, 1H), 7.01 (t, *J* = 8.8 Hz, 2H), 6.80 (d, *J* = 6.6 Hz, 2H), 6.79 – 6.75 (m, 1H), 5.33 (q, *J* = 6.4 Hz, 1H), 4.17 (s, 2H), 1.54 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 170.7, 163.5 (d, *J* = 243.8 Hz), 158.2, 150.2, 140.7 (d, *J* = 3.1 Hz), 133.1, 132.4, 129.9, 129.5 (2C), 128.7 (d, *J* = 8.3 Hz, 2C), 119.0, 118.4, 117.1 (2C), 116.1 (d, *J* = 21.7 Hz, 2C), 114.7, 76.4, 48.0, 24.7. TOFMSESI *m/z*: 364.1363 (C₂₂H₂₀FNO₃ - H⁺ requires 364.1354).

(S)-3-((4-(1-(4-fluorophenyl)ethoxy)benzyl)amino)benzoic acid (4y): Compound **13** (60 mg, 0.17 mmol), (*R*)-1-(4-fluorophenyl)ethan-1-ol (24 mg, 0.17 mmol), and PPh₃ (45 mg, 0.17 mmol) were dissolved in toluene (2.5 mL) and cooled to 0 °C. A solution of DIBAD (39 mg, 0.17 mmol) in toluene (5 mL) was added dropwise to the mixture. The reaction was warmed to 25 °C and stirred at for 12 h, at which time the reaction was deemed complete by TLC. The solvent was removed under reduced pressure. The residue was purified by flash column chromatography (SiO₂, hexanes/ethyl acetate = 2: 1) to provide the desired product **15** as colorless oil. The resulting product **15** (24 mg, 0.05mmol) (23 mg, 0.049 mmol) was dissolved in a THF/MeOH/H₂O (3:1:1) mixture and LiOH-added was (mmol 0.25, mg 10) H₂O. The reaction stirred at 25 °C for 12 h upon which the reaction was deemed complete by TLC. The reaction was quenched with aq. 1N HCl (3 mL) and extracted with EtOAc (3 × 10 mL). The organic layers were combined, dried over Na₂SO₄, filtered, and concentrated by vacuum. The product **4y** was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (14 mg, 75% yield). $[\alpha]_D^{20} = -22.0$ (c = 9.1 mg/mL, l = 0.5 dm, MeOH). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.40 – 7.34 (m, 2H), 7.28 – 7.24 (m, 1H), 7.27 – 7.22 (m, 1H), 7.21 – 7.16 (m, 2H), 7.13 (t, *J* = 7.8 Hz, 1H), 7.02 (t, *J* = 8.8 Hz, 2H), 6.82 – 6.80 (m, 2H), 6.79 (ddd, *J* = 8.1, 2.5, 1.0 Hz, 1H), 5.36 (q, *J* = 6.4 Hz, 1H), 4.19 (s, 2H), 1.55 (d, *J* = 6.4 Hz, 3H). ¹³C NMR (151 MHz, Methanol-*d*₄) δ 170.8, 163.7 (d, *J* = 243.9 Hz), 158.3, 150.4, 140.9 (d, *J* = 2.9 Hz), 133.3, 132.5, 130.0, 129.7 (2C), 128.8 (d, *J* = 8.4 Hz, 2C), 119.2, 118.6, 117.3 (2C), 116.3 (d, *J* = 21.6 Hz, 2C), 114.8, 76.5, 48.2, 24.8. TOFMSESI *m/z*: 364.1365 (C₂₂H₂₀FNO₃ - H⁺ requires 364.1354).

4-((2-(4-fluorophenyl)propan-2-yl)oxy)benzaldehyde (17): To a solution of compound **16** (1.5 g, 8.2 mmol) in MeCN (10 mL) at 0 °C was added DBU (3.7 g, 24.6 mmol). TFAA (1.14 mL, 8.2 mmol) was added dropwise and the solution was stirred for 30 min. Then the mixture was added dropwise to a solution of compound **1** (1 g, 8.2 mmol) and CuCl₂·H₂O (1.4 mg, 0.0082 mmol) in MeCN (10 mL) over a period of 5 min at 0 °C. The mixture was allowed to warm to 25 °C and stir for 12 h. The solvent was removed under reduced pressure and the residue was diluted in DCM (40 mL). The organic layer was washed sequentially with aq. 1N HCl (2 × 20 mL), aq. 1N NaOH (2 × 20 mL), sat. aq. NaHCO₃ (2 × 20 mL), and brine (20 mL). The organic layer was dried over Na₂SO₄, filtered, and concentrated in vacuo. Compound **17** was purified by silica gel chromatography (SiO₂, hexanes/ethyl acetate = 20:1) to provide the desired product as colorless oil. Yield: 8%. ¹H NMR (500 MHz, Chloroform-*d*) δ 9.81 (s, 1H), 7.68 – 7.62 (m, 2H), 7.44 – 7.36 (m, 2H), 7.10 – 7.01 (m, 2H), 6.78 – 6.69 (m, 2H), 1.78 (s, 6H). ¹³C NMR (126 MHz, Chloroform-*d*) δ 190.9, 162.1 (d, *J* = 246.5 Hz), 161.6, 141.3 (d, *J* = 3.4 Hz), 131.4, 129.8, 126.9 (d, *J* = 8.0 Hz), 119.1, 115.7 (d, *J* = 21.4 Hz), 81.1, 29.8.

3-((4-((2-(4-fluorophenyl)propan-2-yl)oxy)benzyl)amino)benzoic acid

(4z): Compound **4z** was prepared following the General Procedure A with aldehyde **17** (0.62 mmol). The desired product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (60 mg, 25% yield). ¹H NMR (500 MHz, DMSO-*d*₆) δ 8.4 (s, 1H) 7.53 – 7.45 (m, 2H), 7.22 – 7.13 (m, 2H), 7.14 (d, *J* = 2.3 Hz, 1H), 7.16 – 7.08 (m, 2H), 7.12 – 7.05 (m, 1H), 7.07 (s, 1H), 6.70 (dt, *J* = 6.2, 2.9 Hz, 1H), 6.62 – 6.55 (m, 2H), 6.26 (t, *J* = 5.9 Hz, 1H), 4.13 (d, *J* = 5.4 Hz, 2H), 1.63 (s, 6H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.4, 161.1 (d, *J* = 243.0 Hz), 154.1, 148.6, 142.3 (d, *J* = 3.1 Hz), 132.8, 128.5, 127.9 (2C), 127.3 (d, *J* = 8.0 Hz, 2C), 119.9 (2C), 116.8, 115.5, 115.0 (d, *J* = 21.0 Hz, 2C), 112.9, 102.2, 79.4, 45.8, 29.1 (2C). TOFMSESI *m/z*: 402.1482 (C₂₃H₂₂FNO₃ + Na⁺ requires 402.1476).

2-(3-((4-((4-Methoxybenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoic acid (5a):

Reaction scale: 99 mg (0.22 mmol) of ethyl 2-(3-((4-((4-methoxybenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoate. Compound **5a** was prepared following General Procedure B. The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 100:1) to give a white amorphous solid (13 mg, 14% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.34 (d, *J* = 8.6 Hz, 2H), 7.25 (d, *J* = 8.5 Hz, 2H), 6.93 (d, *J* = 2.8 Hz, 1H), 6.92 (d, *J* = 3.1 Hz, 2H), 6.90 (d, *J* = 3.1 Hz, 2H), 6.34 – 6.26 (m, 1H), 6.19 (s, 1H), 6.16 (d, *J* = 8.6 Hz, 1H), 4.97 (s, 2H), 4.18 (s, 2H), 3.79 (s, 3H), 1.47 (s, 6H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 160.9, 159.3, 157.8, 151.1, 133.5, 130.8, 130.3 (2C), 130.2, 130.0, 129.6 (2C), 116.0 (2C), 114.8 (2C), 111.4, 109.2, 109.0, 105.6, 70.9, 55.7, 48.2, 25.8 (2C). HRESI *m/z*: 444.1771 (C₂₅H₂₇NO₅ + Na⁺ requires 444.1781).

2-(3-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoic acid

(5b): Reaction scale: 96 mg (0.22 mmol) of ethyl 2-(3-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoate. Compound **5b** was prepared following General Procedure B. The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 100:1) to give a white amorphous solid (19 mg, 21% yield). ¹H

NMR (600 MHz, Methanol- d_4) δ 7.58 – 7.39 (m, 2H), 7.26 (d, J = 8.4 Hz, 2H), 7.16 – 7.02 (m, 2H), 6.94 (d, J = 4.2 Hz, 1H), 6.92 (dd, J = 8.4, 1.5 Hz, 2H), 6.39 – 6.26 (m, 1H), 6.18 (t, J = 2.0 Hz, 1H), 6.16 (dd, J = 8.0, 2.0 Hz, 1H), 5.02 (s, 2H), 4.18 (s, 2H), 1.47 (s, 6H). ^{13}C NMR (151 MHz, Methanol- d_4) δ 168.9, 154.3 (d, J = 244.4 Hz), 149.6, 148.3, 141.6, 125.4 (d, J = 3.4 Hz), 124.2, 121.1 (d, J = 8.4 Hz, 2C), 120.7, 120.1 (2C), 106.6 (d, J = 21.8 Hz, 2C), 106.5 (2C), 99.8, 99.6, 96.2, 70.5, 60.8, 38.7, 16.3 (2C). TOFMSESI m/z : 410.1761 ($\text{C}_{24}\text{H}_{24}\text{FNO}_4 + \text{H}^+$ requires 410.1762).

2-(4-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoic acid (5b*):

Reaction scale: 96 mg (0.22 mmol) of ethyl 2-(4-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenoxy)-2-methylpropanoate. Compound **5b*** was prepared following General Procedure B. The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 100:1) to give a white amorphous solid (62 mg, 60% yield). ^1H NMR (400 MHz, DMSO- d_6) δ 7.48 (dd, J = 8.3, 5.5 Hz, 2H), 7.27 (d, J = 8.1 Hz, 2H), 7.20 (t, J = 8.7 Hz, 2H), 6.95 (d, J = 8.1 Hz, 2H), 6.65 (d, J = 8.3 Hz, 2H), 6.46 (d, J = 8.3 Hz, 2H), 5.04 (s, 2H), 4.10 (s, 2H), 1.36 (s, 6H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 175.3, 161.7 (d, J = 243.5 Hz), 157.1, 145.4, 144.5, 133.5 (d, J = 3.1 Hz), 132.5, 129.9 (d, J = 8.3 Hz, 2C), 128.6, 121.5, 115.3 (d, J = 21.3 Hz, 2C), 114.6, 112.6, 78.8, 68.4, 46.5, 25.0 (2C). TOFMSESI m/z : 460.1732 ($\text{C}_{24}\text{H}_{24}\text{FNO}_4 + \text{H}^+$ requires 410.1762).

2-methyl-2-(3-((4-((4-(trifluoromethyl)benzyl)oxy)benzyl)amino)phenoxy)propanoic acid (5f):

Reaction scale: 107 mg (0.22 mmol) of ethyl 2-methyl-2-(3-((4-((4-(trifluoromethyl)benzyl)oxy)benzyl)amino)phenoxy)propanoate. Compound **5f** was prepared following General Procedure B. The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 100:1) to give a white amorphous solid (30 mg, 30% yield). ^1H NMR (400 MHz, Methanol- d_4) δ 7.67 (d, J = 8.3 Hz, 2H), 7.62 (d, J = 8.2 Hz, 2H), 7.27 (d, J = 8.6 Hz, 2H), 6.98 – 6.90 (m, 2H), 6.92 (d, J = 8.0 Hz, 1H), 6.30 (ddd, J = 8.0, 2.2, 0.9 Hz, 1H), 6.17 (t, J = 2.2 Hz, 1H), 6.16 – 6.13 (m, 1H), 5.15 (s, 2H), 4.19 (s, 2H), 1.46 (s, 6H). ^{13}C NMR (151 MHz, Methanol- d_4) δ 168.8, 149.4, 148.2, 141.6, 134.0, 124.4 (d, J = 2.3 Hz), 121.3 (q, J = 32.1 Hz), 120.7, 120.2 (2C), 119.2 (2C), 116.8 (q, J = 3.9 Hz, 2C), 116.3 (q, J = 271.2 Hz), 106.4 (d, J = 2.2 Hz, 2C), 99.7 (d, J = 3.8 Hz), 99.6 (d, J = 1.8 Hz), 96.2 (d, J = 3.6 Hz), 70.5, 60.5 (d, J = 2.8 Hz), 38.7, 16.3 (2C). TOFMSESI m/z : 460.1732 ($\text{C}_{25}\text{H}_{24}\text{F}_3\text{NO}_4 + \text{H}^+$ requires 460.1730).

N-(3-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenyl)methanesulfonamide (19a):

Reaction scale: 50 mg (0.27 mmol) of N-(3-aminophenyl)methanesulfonamide (**18a**). Compound **19a** was prepared following General Procedure A with aldehyde **3b** (0.28 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (27 mg, 25% yield). ^1H NMR (500 MHz, DMSO- d_6) δ 9.39 (s, 1H), 7.48 (dd, J = 8.4, 5.7 Hz, 2H), 7.26 (d, J = 8.5 Hz, 2H), 7.20 (t, J = 8.8 Hz, 2H), 6.95 (d, J = 4.7 Hz, 2H), 6.94 (d, J = 4.1 Hz, 1H), 6.47 (d, J = 2.1 Hz, 1H), 6.38 – 6.32 (m, 1H), 6.31 (s, 1H), 6.33 – 6.26 (m, 1H), 5.05 (s, 2H), 4.13 (d, J = 5.9 Hz, 2H), 2.87 (s, 3H). ^{13}C NMR (101 MHz, DMSO- d_6) δ 161.7 (d, J = 243.6 Hz), 157.0, 149.4, 139.0, 133.4 (d, J = 3.1 Hz), 132.1, 129.8 (d, J = 8.2 Hz, 2C), 129.4, 128.5 (2C), 115.2 (d, J = 21.4

Hz, 2C), 114.6 (2C), 108.0, 107.2, 103.5, 68.4, 45.8, 38.8. TOFMSESI m/z: 423.1160 (C₂₁H₂₁FN₂O₃S + H⁺ requires 423.1149).

3-((4-((4-fluorobenzyl)oxy)benzyl)amino)-N-hydroxybenzamide (19b): Reaction scale: 50 mg (0.33 mmol) of 3-amino-N-hydroxybenzamide (**18b**). Compound **19b** was prepared following General Procedure A with aldehyde **3b** (0.34 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (96 mg, 80% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.44 (dd, *J* = 8.5, 5.5 Hz, 2H), 7.29 (d, *J* = 8.6 Hz, 2H), 7.15 (d, *J* = 7.8 Hz, 1H), 7.12 – 7.08 (m, 2H), 7.08 – 7.07 (m, 1H), 7.05 (d, *J* = 2.6 Hz, 1H), 6.93 (d, *J* = 8.6 Hz, 2H), 6.77 (dd, *J* = 7.8, 2.0 Hz, 1H), 5.02 (s, 2H), 4.27 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 201.5, 192.0 (d, *J* = 244.5 Hz), 187.3, 178.5, 163.8, 163.0 (d, *J* = 3.2 Hz), 161.6, 158.8 (d, *J* = 8.3 Hz, 2C), 158.2, 157.8 (2C), 145.5, 144.8, 144.3 (d, *J* = 21.8 Hz, 2C), 144.1 (2C), 141.0, 98.5, 76.1. TOFMSESI m/z: 351.1291 (C₂₁H₁₉FN₂O₃ OH⁻ + 2H⁺ requires 351.1503).

3-((4-((4-fluorobenzyl)oxy)benzyl)amino)benzenesulfonamide (19c): Reaction scale: 100 mg (0.58 mmol) of 3-aminobenzenesulfonamide (**18c**). Compound **19c** was prepared following General Procedure A with aldehyde **3b** (0.59 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (17 mg, 8% yield). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.44 (dd, *J* = 8.6, 5.5 Hz, 2H), 7.31 – 7.26 (m, 2H), 7.20 (t, *J* = 7.9 Hz, 1H), 7.14 (t, *J* = 2.0 Hz, 1H), 7.12 – 7.09 (m, 2H), 7.07 (s, 1H), 6.97 – 6.91 (m, 2H), 6.78 (ddd, *J* = 8.2, 2.4, 1.0 Hz, 1H), 5.02 (s, 2H), 4.27 (s, 2H). ¹³C NMR (151 MHz, Methanol-*d*₄) δ 163.8 (d, *J* = 244.4 Hz), 159.2, 150.5, 145.4, 134.8 (d, *J* = 2.9 Hz), 133.0, 130.6 (d, *J* = 8.1 Hz, 2C), 130.5, 129.7 (2C), 117.2, 116.1 (d, *J* = 21.8 Hz, 2C), 116.0 (2C), 114.6, 110.5, 70.3, 47.8. TOFMSESI m/z: 421.0645 (C₂₀H₁₉FN₂O₃S + Cl⁻ requires 421.0794).

N-(4-((4-fluorobenzyl)oxy)benzyl)-3-(1H-tetrazol-5-yl)aniline (19d): Reaction scale: 100 mg (0.62 mmol) of 3-(1H-tetrazol-5-yl)aniline (**18d**). Compound **19d** was prepared following General Procedure A with compound **3b** (0.63 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (100 mg, 43% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.45 (dd, *J* = 8.5, 5.6 Hz, 2H), 7.37 – 7.30 (m, 2H), 7.30 – 7.24 (m, 2H), 7.20 (dt, *J* = 7.5, 1.3 Hz, 1H), 7.14 – 7.04 (m, 2H), 7.00 – 6.92 (m, 2H), 6.83 (ddd, *J* = 8.1, 2.5, 1.1 Hz, 1H), 5.03 (s, 2H), 4.32 (s, 2H). ¹³C NMR (101 MHz, Methanol-*d*₄) δ 163.8 (d, *J* = 244.6 Hz), 159.2, 157.7, 151.0, 134.8 (d, *J* = 3.4 Hz), 133.2, 131.0, 130.6 (d, *J* = 8.1 Hz, 2C), 129.7 (2C), 125.6, 117.1, 116.1 (d, *J* = 18.6 Hz, 2C), 116.01, 115.99 (2C), 111.7, 70.3, 47.8. TOFMSESI m/z: 376.1575 (C₂₁H₁₈FN₅O + H⁺ requires 376.1568).

(3-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenyl)boronic acid (19e): Reaction scale: 100 mg (0.58 mmol) of (3-aminophenyl)boronic acid (**18e**). Compound **19e** was prepared following General Procedure A with aldehyde **3b** (0.59 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (109 mg, 42% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.43 (dd, *J* = 8.6, 5.5 Hz, 2H), 7.28 (d, *J* = 8.6 Hz, 2H), 7.11 – 7.08 (m, 1H), 7.07 (d, *J* = 8.8 Hz, 2H), 6.98

– 6.89 (m, 2H), 6.84 (s, 1H), 6.84 – 6.79 (m, 1H), 6.67 (ddd, $J = 8.1, 2.5, 1.2$ Hz, 1H), 5.01 (s, 2H), 4.23 (s, 2H). ^{13}C NMR (101 MHz, Methanol- d_4) δ 165.0, 162.6, 159.1, 149.3, 134.9 (d, $J = 3.3$ Hz), 133.8 (2C), 130.6 (dd, $J = 8.5, 23.6$ Hz), 129.6 (d, $J = 21.7$ Hz), 129.3 (d, $J = 6.9$ Hz), 123.0 (2C), 119.1 (2C), 116.7 – 116.0 (m), 115.9 (2C), 115.6 (d, $J = 11.9$ Hz), 70.3, 48.3. TOFMSESI m/z : 352.1537 ($\text{C}_{20}\text{H}_{19}\text{BFNO}_3 + \text{H}^+$ requires 352.1515).

2-((3-((4-((4-fluorobenzyl)oxy)benzyl)amino)phenyl)thio)-2-methylpropanoic acid (19f): Reaction scale: 190 mg (0.56 mmol) of ethyl 2-((3-aminophenyl)thio)-2-methylpropanoate (**18f**). Compound **19f** was prepared following General Procedure B with compound **3b** (0.57 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 100:1) to give a white amorphous solid (30 mg, 13% yield). ^1H NMR (400 MHz, Methanol- d_4) δ 7.47 – 7.38 (m, 2H), 7.30 – 7.23 (m, 2H), 7.13 – 6.97 (m, 2H), 7.01 (t, $J = 7.8$ Hz, 1H), 6.96 – 6.88 (m, 2H), 6.76 (dd, $J = 2.4, 1.6$ Hz, 1H), 6.72 (ddd, $J = 7.5, 1.7, 1.0$ Hz, 1H), 6.63 (ddd, $J = 8.2, 2.4, 1.0$ Hz, 1H), 5.01 (s, 2H), 4.21 (s, 2H), 1.37 (s, 6H). ^{13}C NMR (101 MHz, Methanol- d_4) δ 178.3, 163.8 (d, $J = 244.5$ Hz), 159.1, 150.3, 134.8 (d, $J = 3.1$ Hz), 133.5, 133.3, 130.6 (d, $J = 8.1$ Hz, 2C), 130.0, 129.7 (2C), 125.7, 121.6, 116.1 (d, $J = 21.7$ Hz, 2C), 115.9 (2C), 115.1, 70.2, 52.0, 48.0, 26.6 (2C). TOFMSESI m/z : 448.1358 ($\text{C}_{24}\text{H}_{24}\text{FNO}_3\text{S} + \text{Na}^+$ requires 448.1359).

3-((4-((4-chlorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (20): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **20** was prepared following General Procedure A with 4-((4-chlorobenzyl)oxy)-3-methylbenzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (50 mg, 36% yield). ^1H NMR (600 MHz, DMSO- d_6) δ 12.61 (s, 1H), 7.45 (d, $J = 8.6$ Hz, 2H), 7.43 (d, $J = 8.6$ Hz, 2H), 7.14 (d, $J = 1.9$ Hz, 1H), 7.13 (d, $J = 2.5$ Hz, 1H), 7.13 – 7.10 (m, 1H), 7.09 (d, $J = 6.7$ Hz, 1H), 7.08 – 7.04 (m, 1H), 6.91 (d, $J = 8.4$ Hz, 1H), 6.75 (ddd, $J = 8.0, 2.5, 1.2$ Hz, 1H), 6.41 (t, $J = 6.0$ Hz, 1H), 5.06 (s, 2H), 4.15 (d, $J = 5.6$ Hz, 2H), 2.15 (s, 3H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 177.3, 164.5, 158.2, 146.0, 141.7, 141.0, 140.8, 139.0, 138.6 (2C), 138.3, 137.9 (2C), 135.3, 135.1, 126.1, 125.8, 122.3, 121.1, 77.8, 55.3, 25.7. TOFMSESI m/z : 404.1006 ($\text{C}_{22}\text{H}_{20}\text{ClNO}_3 + \text{Na}^+$ requires 404.1024).

3-((4-((4-bromobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (21): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **21** was prepared following General Procedure A with 4-((4-bromobenzyl)oxy)-3-methylbenzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO_2 , $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH}$, 99.5:0.5) to give a white amorphous solid (70 mg, 45% yield). ^1H NMR (600 MHz, DMSO- d_6) δ 12.62 (s, 1H), 7.58 (d, $J = 8.4$ Hz, 2H), 7.41 (d, $J = 8.3$ Hz, 2H), 7.16 (d, $J = 2.1$ Hz, 1H), 7.17 – 7.14 (m, 1H), 7.16 – 7.12 (m, 1H), 7.13 – 7.09 (m, 1H), 7.12 – 7.07 (m, 1H), 6.93 (d, $J = 8.3$ Hz, 1H), 6.77 (dd, $J = 8.0, 1.1$ Hz, 1H), 6.42 (t, $J = 6.0$ Hz, 1H), 5.07 (s, 2H), 4.17 (d, $J = 5.4$ Hz, 2H), 2.17 (s, 3H). ^{13}C NMR (151 MHz, DMSO- d_6) δ 167.8, 155.0, 148.8, 137.0, 131.5, 131.3 (2C), 131.2, 129.5, 129.4 (2C), 128.8, 125.8, 125.6, 120.7, 116.6, 116.3, 112.9, 111.6, 68.3, 45.9, 16.2. TOFMSESI m/z : 426.0648 ($\text{C}_{22}\text{H}_{20}\text{BrNO}_3 + \text{H}^+$ requires 426.0699).

3-((4-((4-iodobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (22): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **22** was prepared following General Procedure A with 4-((4-bromobenzyl)oxy)-3-methylbenzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (90 mg, 52% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.62 (s, 1H), 7.75 (d, *J* = 8.2 Hz, 2H), 7.25 (d, *J* = 7.9 Hz, 2H), 7.16 (s, 1H), 7.15 (s, 1H), 7.14 (d, *J* = 7.9 Hz, 1H), 7.11 (d, *J* = 8.1 Hz, 1H), 7.09 (d, *J* = 7.8 Hz, 1H), 6.92 (d, *J* = 8.3 Hz, 1H), 6.77 (d, *J* = 8.0 Hz, 1H), 6.40 (t, *J* = 6.1 Hz, 1H), 5.05 (s, 2H), 4.17 (d, *J* = 5.5 Hz, 2H), 2.17 (s, 3H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 155.0, 148.7, 137.3, 137.2 (2C), 131.5, 131.3, 129.5 (2C), 129.4, 128.8, 125.8, 125.6, 116.6, 116.2, 112.9, 111.6, 93.6, 68.5, 45.9, 16.2. TOFMSESI *m/z*: 472.0399 (C₂₂H₂₀INO₃ - H⁺ requires 472.0415).

3-((4-((2,4-difluorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (23): Reaction scale: 40 mg (0.29 mmol) of 3-amino benzoic acid. Compound **23** was prepared following General Procedure A with 4-((2,4-difluorobenzyl)oxy)-3-methylbenzaldehyde (0.30 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (87 mg, 78% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.61 (s, 1H), 7.61 (ddd, *J* = 8.5, 6.6 Hz, 1H), 7.30 (ddd, *J* = 10.5, 2.6 Hz, 1H), 7.16 (dd, *J* = 1.9 Hz, 1H), 7.15 (s, 1H), 7.14 (s, 1H), 7.14 – 7.13 (m, 1H), 7.11 (d, *J* = 8.1 Hz, 1H), 7.09 (d, *J* = 7.5 Hz, 1H), 7.01 (d, *J* = 8.2 Hz, 1H), 6.80 – 6.72 (m, 1H), 6.43 (t, *J* = 6.0 Hz, 1H), 5.08 (s, 2H), 4.18 (d, *J* = 5.5 Hz, 2H), 2.12 (s, 3H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 161.5 – 160.8 (m), 159.6, 155.0, 148.7, 131.7 (d, *J* = 13.3 Hz), 131.7 – 131.5 (m), 131.3, 129.5, 128.8, 125.9, 125.7, 120.7, 116.6, 116.3, 112.9, 111.8, 111.5, 104.0 (t, *J* = 25.8 Hz), 63.3, 45.8, 16.1. TOFMSESI *m/z*: 406.1243 (C₂₂H₁₉F₂NO₃ + Na⁺ requires 406.1225).

3-((4-((3,4-difluorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (24): Reaction scale: 100 mg (0.73 mmol) of 3-amino benzoic acid. Compound **24** was prepared following General Procedure A with 4-((3,4-difluorobenzyl)oxy)-3-methylbenzaldehyde (0.74 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (170 mg, 61% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 7.51 (tt, *J* = 8.0, 4.0 Hz, 1H), 7.45 (ddd, *J* = 10.7, 9.9, 8.4 Hz, 1H), 7.31 (dd, *J* = 6.4 Hz, 1H), 7.16 (d, *J* = 1.8 Hz, 1H), 7.16 (s, 1H), 7.15 – 7.13 (m, 1H), 7.13 – 7.10 (m, 1H), 7.09 (d, *J* = 7.5 Hz, 1H), 6.93 (d, *J* = 8.4 Hz, 1H), 6.80 – 6.75 (m, 1H), 6.42 (t, *J* = 6.0 Hz, 1H), 5.07 (s, 2H), 4.17 (d, *J* = 5.4 Hz, 2H), 2.18 (s, 3H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 154.9, 149.9 (dd, *J* = 12.5, 73.4 Hz), 148.7, 148.3 (dd, *J* = 12.6, 73.0 Hz), 135.3 (dd, *J* = 3.7, 5.8 Hz), 131.7, 131.3, 129.5, 128.8, 125.8, 125.6, 124.1 (dd, *J* = 3.4, 6.7 Hz), 117.5 (d, *J* = 17.2 Hz), 116.6, 116.34 (d, *J* = 10.7 Hz), 116.26, 112.9, 111.6, 67.9, 45.8, 16.2. TOFMSESI *m/z*: 406.1243 (C₂₂H₁₉F₂NO₃ + Na⁺ requires 406.1225).

3-((4-((3,5-difluorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (25): Reaction scale: 40 mg (0.29 mmol) of 3-amino benzoic acid. Compound **25** was prepared following General Procedure A with 4-((3,5-difluorobenzyl)oxy)-3-methylbenzaldehyde (0.30 mmol). The product was purified by flash column

chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (83 mg, 74% yield). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.28 (d, *J* = 2.3 Hz, 1H), 7.25 (d, *J* = 7.7 Hz, 1H), 7.19 (s, 1H), 7.15 (d, *J* = 7.9 Hz, 1H), 7.13 (s, 1H), 7.08 – 7.02 (m, 2H), 6.89 – 6.87 (m, 1H), 6.86 (d, *J* = 8.5 Hz, 1H), 6.81 (dd, *J* = 7.9, 2.3 Hz, 1H), 5.10 (s, 2H), 4.24 (s, 2H), 2.26 (s, 3H). ¹³C NMR (151 MHz, Methanol-*d*₄) δ 164.6 (dd, *J* = 13.0, 247.4 Hz, 2C), 156.7, 150.3, 143.9 (t, *J* = 9.2 Hz), 133.4, 131.0 (2C), 129.8, 127.9, 127.0 (2C), 119.0, 118.2, 114.8, 112.6 (dd, *J* = 2.4, 4.2 Hz), 110.6 (dd, *J* = 5.3, 20.7 Hz, 2C), 103.6 (t, *J* = 25.8 Hz), 69.6, 48.1, 16.5. TOFMSESI *m/z*: 382.1268 (C₂₂H₁₉F₂NO₃ - H⁺ requires 382.1260).

3-((3-fluoro-4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (26): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **26** was prepared following General Procedure A with 3-fluoro-4-((4-fluorobenzyl)oxy)benzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (100 mg, 75% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.64 (s, 1H), 7.49 (dd, *J* = 8.5, 5.7 Hz, 2H), 7.23 – 7.20 (m, 2H), 7.20 – 7.19 (m, 1H), 7.18 (d, *J* = 1.9 Hz, 1H), 7.16 – 7.15 (m, 1H), 7.14 (d, *J* = 7.8 Hz, 1H), 7.11 (d, *J* = 1.4 Hz, 1H), 7.10 (q, *J* = 1.6 Hz, 1H), 6.77 (ddd, *J* = 7.9, 2.5, 1.2 Hz, 1H), 6.50 (t, *J* = 6.1 Hz, 1H), 5.12 (s, 2H), 4.23 (d, *J* = 5.9 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.7, 161.9 (d, *J* = 243.9 Hz), 151.7 (d, *J* = 243.9 Hz), 148.5, 144.6 (d, *J* = 10.7 Hz), 133.4 (d, *J* = 5.0 Hz), 132.8 (d, *J* = 2.9 Hz), 131.3, 130.0 (d, *J* = 8.5 Hz, 2C), 128.9, 123.0 (d, *J* = 3.0 Hz), 116.9, 116.4, 115.4, 115.3 (d, *J* = 21.3 Hz, 2C), 114.7 (d, *J* = 18.3 Hz), 112.9, 69.6, 45.3. TOFMSESI *m/z*: 368.1110 (C₂₁H₁₇F₂NO₃ - H⁺ requires 368.1104).

3-((3-chloro-4-((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (27): Reaction scale: 100 mg (0.72 mmol) of 3-amino benzoic acid. Compound **27** was prepared following General Procedure A with 3-chloro-4-((4-fluorobenzyl)oxy)benzaldehyde (0.73 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (120 mg, 43% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.65 (s, 1H), 7.50 (dd, *J* = 8.3, 5.6 Hz, 2H), 7.42 (d, *J* = 2.0 Hz, 1H), 7.27 (d, *J* = 8.5 Hz, 1H), 7.23 (t, *J* = 8.7 Hz, 2H), 7.19 (d, *J* = 8.4 Hz, 1H), 7.15 (d, *J* = 2.5 Hz, 1H), 7.14 (d, *J* = 7.7 Hz, 1H), 7.11 (d, *J* = 7.5 Hz, 1H), 6.79 – 6.69 (m, 1H), 6.51 (s, 1H), 5.15 (s, 2H), 4.23 (d, *J* = 5.9 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 161.8 (d, *J* = 243.5 Hz), 152.2, 148.4, 133.6, 132.8 (d, *J* = 3.1 Hz), 131.4, 129.7 (d, *J* = 8.4 Hz, 2C), 128.9, 128.5, 126.9, 121.4, 116.9, 116.3, 115.3 (d, *J* = 21.3 Hz, 2C), 114.3, 112.9, 69.3, 45.2. TOFMSESI *m/z*: 384.0820 (C₂₁H₁₇ClFNO₃ - H requires 384.0808).

3-((3,4-bis((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (28): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **28** was prepared following General Procedure A with 3,4-bis((4-fluorobenzyl)oxy)benzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (95 mg, 58% yield). ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.62 (s, 1H), 7.45 (dd, *J* = 2.1, 6.2 Hz, 2H), 7.44 (dd, *J* = 2.4, 5.8 Hz, 2H), 7.24 – 7.16 (m, 2H), 7.17 (d, *J* = 2.5 Hz, 2H), 7.19 – 7.12 (m, 1H), 7.13 (d, *J* = 7.6 Hz, 1H), 7.11 (s, 1H), 7.10 (d, *J* = 5.9 Hz, 1H), 6.99 (d, *J* = 8.2 Hz, 1H), 6.88 (dd, *J* = 1.9, 8.2 Hz, 1H), 6.77 (dt, *J* = 2.0, 7.7 Hz, 1H), 6.42 (t, *J* = 6.0 Hz, 1H), 5.06 (s, 2H), 5.05 (s, 2H), 4.19 (d, *J* = 5.3 Hz, 2H). ¹³C NMR (101 MHz,

DMSO-*d*₆) δ 167.8, 161.7 (d, *J* = 243.3 Hz), 160.5 (d, *J* = 243.5 Hz), 148.7, 148.1, 147.0, 133.6 (d, *J* = 3.2 Hz), 133.4 (d, *J* = 3.1 Hz), 132.9, 131.3, 129.8 (d, *J* = 8.3 Hz, 2C), 129.7 (d, *J* = 8.2 Hz, 2C), 128.8, 119.9, 116.7, 116.3, 115.3 (d, *J* = 21.3 Hz, 2C), 115.1 (d, *J* = 21.3 Hz, 2C), 114.6, 113.9, 113.0, 69.50, 69.47, 46.0. TOFMSESI *m/z*: 474.1525 (C₂₈H₂₃F₂NO₄ - H⁺ requires 474.1522).

3-((2,4-bis((4-fluorobenzyl)oxy)benzyl)amino)benzoic acid (29): Reaction scale: 50 mg (0.36 mmol) of 3-amino benzoic acid. Compound **29** was prepared following General Procedure A with 2,4-bis((4-fluorobenzyl)oxy)benzaldehyde (0.37 mmol). The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (102 mg, 59% yield). ¹H NMR (600 MHz, DMSO-*d*₆) δ 12.61 (s, 1H), 7.52 (dd, *J* = 5.7, 8.4 Hz, 2H), 7.47 (dd, *J* = 5.7, 8.5 Hz, 2H), 7.24 – 7.19 (m, 2H), 7.21 – 7.16 (m, 2H), 7.15 (d, *J* = 2.0 Hz, 1H), 7.13 (d, *J* = 8.1 Hz, 1H), 7.12 (d, *J* = 7.7 Hz, 1H), 7.09 (d, *J* = 7.6 Hz, 1H), 6.75 (t, *J* = 1.7 Hz, 1H), 6.74 (d, *J* = 2.3 Hz, 1H), 6.56 (dd, *J* = 2.3, 8.4 Hz, 1H), 6.29 (t, *J* = 6.1 Hz, 1H), 5.16 (s, 2H), 5.03 (s, 2H), 4.20 (d, *J* = 5.3 Hz, 2H). ¹³C NMR (151 MHz, DMSO-*d*₆) δ 167.8, 161.7 (d, *J* = 243.9 Hz), 161.6 (d, *J* = 243.5 Hz), 158.3, 156.6, 148.8, 133.4 (d, *J* = 2.9 Hz), 133.3 (d, *J* = 2.8 Hz), 131.3, 129.9 (d, *J* = 8.2 Hz, 2C), 129.5 (d, *J* = 8.2 Hz, 2C), 128.8, 128.6, 119.9, 116.5, 116.3, 115.19 (d, *J* = 21.3 Hz, 2C), 115.17 (d, *J* = 21.4 Hz, 2C), 112.6, 105.8, 100.5, 68.6 (2C), 40.8. TOFMSESI *m/z*: 476.1663 (C₂₈H₂₃F₂NO₄ + H⁺ requires 476.1668).

3-((4-((4-fluorobenzyl)oxy)-3-methylbenzyl)(methyl)amino)benzoic acid (30): Paraformaldehyde (5 mg, 0.16 mmol), compound **4u** (30 mg, 0.08 mmol), sodium triacetoxyborohydride (6.2 mg, 0.096 mmol) and 2 mL of ethanol were combined in a 10 mL round bottom flask and refluxed for 2 hr at 100 °C. The reaction was then cooled to 25 °C for 12 h. The solvent was removed by vacuum and the residue was dissolved in EtOAc (60 mL). The organic layer was washed with aq. 1N HCl (20 mL), water (20 mL), and brine (20 mL). The organic layer was dried over Na₂SO₄, filtered, and concentrated by vacuum. The compound was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (13 mg, 44% yield). ¹H NMR (600 MHz, Methanol-*d*₄) δ 7.47 – 7.42 (m, 2H), 7.42 (d, *J* = 1.7 Hz, 1H), 7.37 – 7.29 (m, 1H), 7.23 (t, *J* = 7.9 Hz, 1H), 7.08 (t, *J* = 8.8 Hz, 2H), 7.01 (d, *J* = 2.2 Hz, 1H), 6.98 – 6.96 (m, 1H), 6.96 – 6.94 (m, 1H), 6.85 (d, *J* = 8.3 Hz, 1H), 4.99 (s, 2H), 4.46 (s, 2H), 3.00 (s, 3H), 2.18 (s, 3H). ¹³C NMR (151 MHz, Methanol-*d*₄) δ 169.3, 162.3 (d, *J* = 244.4 Hz), 155.7, 149.7, 133.6 (d, *J* = 3.3 Hz), 131.0, 130.4, 128.9 (d, *J* = 8.3 Hz, 2C), 128.6, 126.7, 125.0, 117.3, 116.7, 114.7 (d, *J* = 21.8 Hz, 2C), 113.0, 111.39, 111.36, 68.9, 55.4, 37.4, 15.2. TOFMSESI *m/z*: 378.1509 (C₂₃H₂₂FNO₃- H⁺ requires 378.1511).

3-fluoro-5-((4-((4-fluorobenzyl)oxy)-3-methylbenzyl)amino)benzoic acid (31): Reaction scale: 50 mg (0.32 mmol) of 3-amino-5-fluorobenzoic acid. Compound **31** was prepared following General Procedure A with aldehyde **12b**. The product was purified by flash column chromatography (SiO₂, CH₂Cl₂/CH₃OH, 99.5:0.5) to give a white amorphous solid (104 mg, 84% yield). ¹H NMR (400 MHz, Methanol-*d*₄) δ 7.47 (dd, *J* = 8.4, 5.2 Hz, 2H), 7.38 (d, *J* = 2.2 Hz, 1H), 7.22 (dd, *J* = 8.5, 2.2 Hz, 1H), 7.13 – 7.02 (m, 4H), 6.88 (d, *J* = 9.2 Hz, 1H), 6.48 (d, *J* = 11.3 Hz, 1H), 5.09 (s, 2H), 4.25 (s, 2H). ¹³C NMR

(101 MHz, Methanol- d_4) δ 169.5, 165.1 (d, J = 241.1 Hz), 163.8 (d, J = 244.3 Hz), 157.2, 152.1, 135.1 (d, J = 2.3 Hz), 134.3 (d, J = 10.0 Hz), 132.5, 130.9, 130.4 (d, J = 8.2 Hz), 128.1, 126.9 (2C), 116.1 (d, J = 21.6 Hz), 112.8 (2C), 111.1, 104.4 (d, J = 24.0 Hz), 103.9 (d, J = 25.8 Hz), 70.4, 47.9, 16.6. TOFMSESI m/z : 382.1254 ($C_{22}H_{19}F_2NO_3 - H^+$ requires 382.1260).

3-((3-chloro-4-((4-fluorobenzyl)oxy)benzyl)amino)-5-fluorobenzoic acid

(32): Reaction scale: 50 mg (0.32 mmol) of 3-amino-5-fluorobenzoic acid. Compound **32** was prepared following General Procedure A with 3-chloro-4-((4-fluorobenzyl)oxy)benzaldehyde. The product was purified by flash column chromatography (SiO_2 , CH_2Cl_2/CH_3OH , 99.5:0.5) to give a white amorphous solid (109 mg, 84% yield). 1H NMR (400 MHz, Methanol- d_4) δ 7.51 – 7.43 (m, 2H), 7.38 (d, J = 2.2 Hz, 1H), 7.22 (dd, J = 8.4, 2.2 Hz, 1H), 7.09 (dt, J = 5.9, 2.1 Hz, 2H), 6.92 – 6.84 (m, 1H), 6.48 (dt, J = 11.4, 2.3 Hz, 1H), 5.09 (s, 2H), 4.25 (s, 2H). ^{13}C NMR (101 MHz, Methanol- d_4) δ 169.4 (d, J = 3.4 Hz), 165.1 (d, J = 241.2 Hz), 163.9 (d, J = 244.8 Hz), 154.5, 151.8 (d, J = 11.0 Hz), 134.5, 134.3 (d, J = 3.1 Hz), 134.2 (d, J = 3.7 Hz), 130.5, 130.4, 130.1, 127.8, 124.3, 116.3, 116.1, 115.6, 111.1, 104.7 (d, J = 24.1 Hz), 103.9, 71.3, 47.3. TOFMSESI m/z : 402.1317 ($C_{21}H_{16}ClF_2NO_3 - H^+$ requires 402.0714).

2-(3-((4-((4-fluorobenzyl)oxy)-3-methylbenzyl)amino)phenoxy)-2-methylpropanoic acid (33)

(33): Reaction scale: 99 mg (0.22 mmol) of ethyl 2-(3-((4-((4-fluorobenzyl)oxy)-3-methylbenzyl)amino)phenoxy)-2-methylpropanoate. Compound **33** was prepared following General Procedure B. The product was purified by flash column chromatography (SiO_2 , CH_2Cl_2/CH_3OH , 100:1) to give a white amorphous solid (30 mg, 32% yield). 1H NMR (400 MHz, Methanol- d_4) δ 7.41 (dd, J = 8.5, 5.5 Hz, 2H), 7.11 (d, J = 2.1 Hz, 1H), 7.09 (d, J = 6.0 Hz, 1H), 7.06 (t, J = 7.7 Hz, 2H), 6.92 (t, J = 8.0 Hz, 1H), 6.83 (d, J = 8.3 Hz, 1H), 6.30 (dd, J = 8.1, 2.1 Hz, 1H), 6.20 (t, J = 2.3 Hz, 1H), 6.17 (dd, J = 8.0, 2.3 Hz, 1H), 4.98 (s, 2H), 4.12 (s, 2H), 2.19 (s, 3H), 1.46 (s, 6H). ^{13}C NMR (101 MHz, Methanol- d_4) δ 178.2, 163.7 (d, J = 244.2 Hz), 157.7, 157.0, 151.0, 135.1 (d, J = 3.3 Hz), 133.1, 130.9, 130.3 (d, J = 8.1 Hz, 2C), 130.2, 127.9, 126.9, 116.1 (d, J = 21.7 Hz, 2C), 112.7, 109.2, 109.1, 105.7, 79.9, 70.3, 48.3, 25.8 (2C), 16.6. TOFMSESI m/z : 422.1784 ($C_{25}H_{26}FNO_4 - H^+$ requires 422.1773).

BIOLOGICAL ASSAY PROTOCOLS

***In vitro* PPAR functional reporter gene assay.**

Agonist activity of test compounds against human PPAR α , was analyzed using commercial kits from Indigo Biosciences (hPPAR α : IB00111, hPPAR γ : IB00101; hPPAR δ : IB00121). Protocols provided with the kit were followed for experimental execution. Briefly, a suspension of reporter cells was prepared in Cell Recovery Medium (CRM: containing 10% charcoal stripped fetal bovine serum (FBS)). 200 μ L of the reporter cell suspension was dispensed into wells of white 96-well assay plates provided within the kit for a 4–6 h pre-incubation period at 37 $^{\circ}C$ and 5% CO_2 . At the end of the pre-incubation period, the culture media was discarded and 200 μ L of Compound Screening Medium (CSM: containing 10% charcoal stripped FBS) containing the requisite concentration of derivative to be evaluated

was added to representative wells. Following a 22–24 h incubation at 37 °C and 5% CO₂, the treatment media were discarded and 100 µL of the provided Luciferase Detection Reagent was added to each well. Luminescence was quantified using a GloMax Explorer (Promega). Dose-response analyses of compounds were performed via non-linear curve-fitting with four parameters. Plots of relative light units vs. Log[compound, (µM)] were generated using GraphPad Prism software, which was also used to calculate EC₅₀ and SEM values. Unless otherwise noted, each compound evaluated was tested in three separate experiments in triplicate.

Cellular thermal shift assay.

The cellular thermal shift assay (CETSA) was performed with MIO-M1 cells, a cell line derived from human retinal Müller glial cells, cultured in DMEM medium supplemented with 10% FBS. For an initial determination of the thermal melting profile of PPAR α , cells were dispensed into a 96-well PCR plate in the above medium (6000 cells/well/50 µl) and subjected to a temperature gradient (40–60 °C) for 10 min. Cold non-denaturing lysis buffer (PBS supplemented with 0.1% TritonX-100 and 1x protease inhibitors) was added to the wells, and the plate was rocked, and then incubated for 15 min on ice. Subsequently, centrifugation was performed at 14,000 rpm to sediment the unstable protein content. Supernatant was collected and loaded for SDS-PAGE, and immuno-detection was performed using an anti-PPAR α antibody. The PPAR α band was quantified on a LI-COR C-Digit Blot Scanner, and subsequently T_{agg}(50) and T_{agg}(75) values were calculated for PPAR α . In a subsequent run, cells were treated at various doses (1, 0.3, 0.1, 0.037, 0.012, 0.004 and 0.0014 µM) of FA, **4a** (A91) and **4u** together with DMSO controls, for 3 h. Cells were then subjected to heat shock at T_{agg}(75) for 10 min, and unstable protein was removed by centrifugation. Following an immuno-blotting step, bands of stable PPAR α protein was quantified, normalized to loading control and plotted using GraphPad Prism software. EC₅₀ values of FA, **4a** (A91) and **4u** were calculated. Data obtained from a single experiment run in duplicate.

PHARMACOKINETIC ANALYSES

Studies were carried out by Eurofins Discovery Services. Experimental protocols presented below are pulled directly from company provided SOPs. All experiments were run in duplicate.

Microsomal stability.

The following procedure is designed to determine the stability of a test compound in pooled liver microsomes from human or rat in a 96-well format. The test compound is quantified at five time points by HPLC-MS/MS analysis. The test compound was pre-incubated with pooled liver microsomes in phosphate buffer (pH 7.4) for 5 min in a 37 °C shaking water bath. The reaction was initiated by adding NADPH-generating system and incubated for 0, 15, 30, 45, and 60 min. The reaction is stopped by transferring the incubation mixture to acetonitrile/methanol. Samples are then mixed and centrifuged. Supernatants are used for HPLC-MS/MS using selected reaction monitoring. The HPLC system consists of a binary LC pump with an autosampler, a C-18 column, and a gradient. Four reference compounds

are tested in each assay. Propranolol and imipramine are relatively stable, whereas verapamil and terfenadine are readily metabolized. Peak areas corresponding to the test compound are recorded. The compound remaining is calculated by comparing the peak area at each time point to time zero. The half-life is calculated from the slope of the initial linear range of the logarithmic curve of compound remaining (%) vs. time, assuming first order kinetics. In addition, the intrinsic clearance (Cl_{int}) is calculated from the half-life using the following equation: $Cl_{int} = 0.693/(t_{1/2} \times [\text{protein}])$ and units are $\mu\text{L}/\text{min}/\text{mg}$ protein. Human liver microsomes: mixed gender and pool of 50 donors; rat liver microsomes: male, Sprague-Dawley, pool of 100 or more.

Time-dependent CYP inhibition.

The following procedure is designed to assess if a test compound displays time-dependent inhibition on cytochrome P450 activity in pooled human liver microsomes in a 96-well plate format. Test compounds were assessed at 10 μM with 0.1% DMSO. The test compound was pre-incubated with human liver microsomes (mixed gender, pool of 50 donors, 1 mg/mL) in phosphate buffer (pH 7.4) for 30 min in a 37 °C shaking water bath. The pre-incubation is performed in two sets: in the presence and absence of NADPH-generating system, respectively. The reaction was then initiated by adding one volume of the pre-incubation mixture to nine volumes of a mixture of the substrate and the NADPH-generating system. The incubation was allowed for 10 min and stopped by transferring the reaction mixture to acetonitrile/methanol. Samples were mixed and centrifuged and supernatants used for HPLC-MS/MS analysis of the respective metabolite. Reference time dependent inhibitors were tested in each assay at multiple concentrations to obtain IC_{50} (concentration causing a half-maximal inhibition of the control value) values for both reaction conditions (\pm NADPH). Peak areas corresponding to the metabolite were recorded. The percent of control activity was calculated by comparing the peak area in the presence of the test compound to the control samples containing the same solvent. Subsequently, the percent inhibition was calculated by subtracting the percent control activity from 100. The IC_{50} value was determined by non-linear regression analysis of the concentration-response curve using the Hill equation. Reference compounds: phenacetin (CYP1A), diclofenac (CYP2C9), omeprazole (CYP2C19), dextromethorphan (CYP2D6), and midazolam (CYP3A).

hERG Automated Patch-Clamp.

The automated whole cell patch-clamp (Qpatch 48) technique was used to record outward potassium currents from a single cell. CHO-K1 (Chinese Hamster Ovary) cells stably transfected with human hERG cDNA were used. The cells were harvested by trypsinization and maintained in Serum Free Medium at room temperature before recording. The cells were washed and re-suspended in Extracellular Solution before being applied to the automated patch-clamp sites. The test solutions are prepared in the Extracellular Solution on the day of patch-clamp assay. The assay can tolerate up to 1% DMSO. Test compounds were assessed at three concentrations (0.1, 1 and 10 μM). Intracellular Solution: 130 mM KCl, 10 mM NaCl, 1 mM MgCl_2 , 10 mM EGTA, 5 mM MgATP, 10 mM HEPES (pH adjusted to 7.2 with KOH) Extracellular Solution: 137 mM NaCl, 4 mM KCl, 1.8 mM CaCl_2 , 1 mM MgCl_2 , 10 mM D(+)-Glucose, 10 mM HEPES (pH adjusted to 7.4 with NaOH) After whole cell configuration was achieved, the cell was held at -80 mV. A 50 ms pulse to -40 mV was

delivered to measure the leaking current, which was subtracted from the tail current on-line. Then the cell was depolarized to +20 mV for 2 s, followed by a 1 s pulse to -40 mV to reveal the hERG tail current. This paradigm was delivered once every 5 s to monitor the current amplitude. The assay was conducted at room temperature. The Extracellular Solution (control) was applied first and the cell was stabilized in the solution for 5 min. Then the test compound was applied from low to high concentrations sequentially on the same cell. The cells were incubated with each test concentration for 5 min. The reference compound E-4031 was tested concurrently at multiple concentrations to obtain an IC₅₀ value. The percent inhibition of hERG channel was calculated by comparing the tail current amplitude before and after application of the compound (the current difference was normalized to control).

***In vivo* retinal vascular permeability assay.**

All *in vivo* experiments were performed in compliance with the NIH guidelines for the Care and Use of Animals and approved by the Institutional Animal Care and Use Committees at the University of Oklahoma Health Sciences Center. Brown Norway rats were purchased from Charles Rivers Laboratories (Wilmington, MA). Diabetes was induced in male Brown Norway rats (at the ages of 8 weeks) using an intraperitoneal injection of streptozotocin (STZ, 55 mg/kg of body weight). Diabetes was confirmed 3 days following STZ injection and rats with blood glucose levels higher than 350 mg/dl were used as diabetic rats. Two weeks following the STZ injection, the animals received daily injections of the compound for another 4 weeks, with the same volume vehicle as control. The rats were then used for retinal vascular permeability assay using Evans blue as tracer as described previously³⁰ and described as follows. At the study end-point, the animals were anesthetized by intraperitoneal injection of Ketamine (75 mg/kg, Zetamine™, MWI, Boise, ID) and Xylazine (10 mg/kg, AnaSed™, LA, MWI, Boise, ID). Evans blue (30 mg/ml) (Sigma-Aldrich, St. Louis, MO) was administered over 10 s through the femoral vein using 30G needle under microscopic inspection at the dosage of 1 µl/g body weight. The animals were placed on a warm pad for 2 hrs to ensure the complete circulation of Evans blue. The animals were perfused via the left ventricle with 1% paraformaldehyde in citrate buffer (pH 4.2), which was pre-warmed to 37°C to prevent vasoconstriction. Immediately after perfusion, the eyeball was enucleated, and the retina was carefully dissected under an operating microscope. The retinal homogenate was incubated with formamide for 18 hrs at 70°C to extract the Evans blue. Then the extract was ultra-centrifuged (Optima TLX Ultracentrifuge, Beckman Coulter Inc., Brea, CA) at 80,000 rpm and 4°C for 30 minutes. The supernatant was used for measurement of absorbance at 620 nm with a spectrophotometer (DU-800 UV/Vis Spectrophotometer, the Lab World Group, Woburn, MA). Concentrations of Evans blue in the extracts were calculated based on a standard curve of Evans blue in formamide. The concentration of Evans blue was normalized by total protein concentration in the supernatant. The result of retinal vascular permeability was expressed in µg of Evans blue/mg of proteins. All values are expressed as mean ± SD. Student's t-test was used to compare 2 groups, and P values less than 0.05 were considered statistically significant.

DOCKING PROTOCOLS

Molecular modelling and docking procedures were executed using the standard protocol implemented in Maestro Molecule Builder of Schrödinger,^{19, 31} version 11.9. File preparation and docking protocol was completed as follows:

Protein Preparation.

The desired crystal structure (PDB: 2P54) was obtained from the Protein Data Bank and prepared with Schrödinger's *Protein Preparation Wizard* software. Proteins were preprocessed by assigning bond orders using the ccd database, adding hydrogens, creating zero-bond order to metals, creating disulfide bonds, filling missing side chains and loops using *Prime*, deleting water molecules beyond 5 Å, and generating Het state using *Epik* pH 7 ± 2. Het -COOH was modified to COO⁻. Structural refinement was done employing H-bond assignment by sampling water orientations, minimizing hydrogens of altered species at pH 7.0, and restraining minimization based on the OPLS3 force field. All remaining parameters were kept as default settings.

Ligand Preparation.

SMILES files of each ligand were loaded into Schrödinger's *LigPrep* software. Ligand preparation employed the OPLS3 force field. Ionization states for pH 7.0 ± 2 were generated with *Epik*. Ligands were desalted and tautomers were generated. Chirality specified in the ligand input file was retained.

Receptor Grid Generation.

i) Receptor tab: select the native ligand and show markers, set the Van der Waals scaling factor to 1.0 with a partial charge cut off of 0.25. In advanced settings, select read from the input structure file and Force field OPLS3. ii) Site tab: Select centroid of workspace ligand, and dock ligand with length < 10 Å. In advanced settings set the box length = 10 Å. iv) Constraints tab: H-Bond Tyr 464, His-440, Tyr-314 and Ser-280; v) Rotatable Groups tab: select all in the grid box.

Ligand Docking.

Docking studies of compounds were performed using Schrödinger's *Glide* docking module from suite 2017–4. Docking was executed with the following parameters: Using partial charge input, scaling of Van der Waal radii with a scaling factor of 0.8 and partial charge cut off of 0.15, with XP precision, sample nitrogen inversion, sample ring conformation, bias sampling of torsion for amide group only, add *Epik* state penalties to docking score, OPLS3 force field with no constraints selected.

Metabolism Prediction.

Metabolism prediction studies were performed using Schrödinger's P450 Site of Metabolism module. After conducting *Ligand Preparation* (see above), compound **4b** was subjected to metabolism prediction studies with 2C9, 2D6, and 3A4. For these calculations, default parameters were held constant. The algorithm calculates reactivity of each ligand

based on Hammett and Taft linear free energy schemes for these promiscuous P450 enzymes, which are believed to be the most independent on structural restrictions. The software then subjects the ligand of interest to induced-fit docking (2C9 and 2D6) to identify possible sites of metabolism based on Fe-accessibility. Intrinsic reactivity and site of metabolism data is then output as 1) Fe-accessibility, 2) Intrinsic reactivity, 3) and overall SOM score (the linear combination of the accessibility and the intrinsic reactivity), which identifies the major predicted sites of P450 metabolism.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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ABBREVIATIONS

CETSA	cellular thermal shift assay
DR	diabetic retinopathy
FenoFA	fenofibric acid
Feno	fenofibrate
ITC	isothermal titration calorimetry
NV	neovascularization
POC	proof of concept
SI	selectivity index
STZ	streptozotocin
T_{agg}	protein aggregation temperature
VEGF	vascular endothelial growth factor

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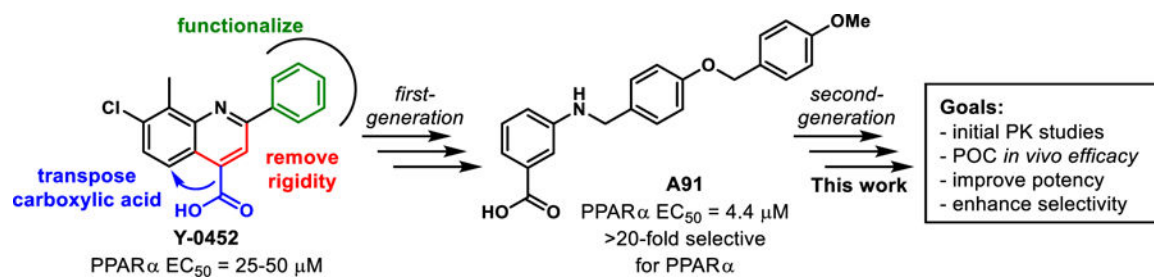


Figure 1.
Overall strategy for hit optimization and the objective of the studies presented herein.

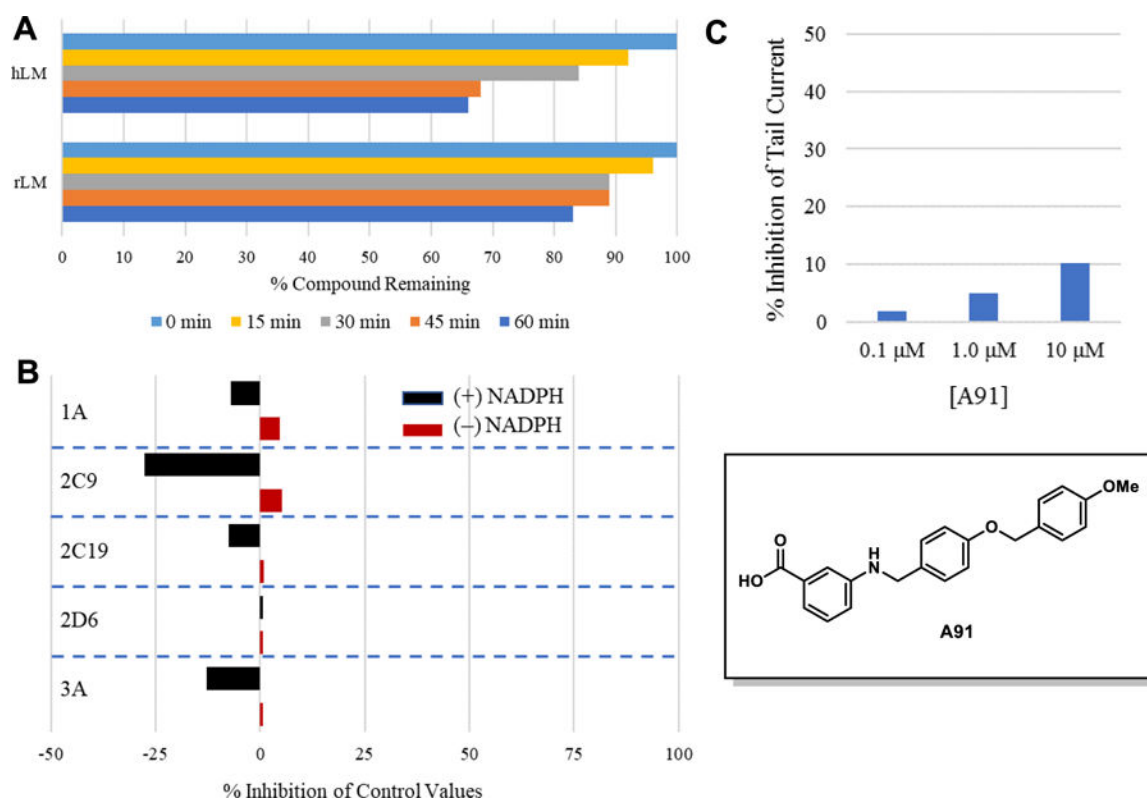


Figure 2. Pharmacokinetic profile of A91.

(A) Stability in liver microsomes. A91 was pre-incubated with pooled liver microsomes in phosphate buffer (pH 7.4) for 5 min at 37 °C. The system was activated by the addition of an NADPH-regenerating system and incubated for 0, 15, 30, 45, and 60 min before being quenched with an acetonitrile/methanol mixture. Samples were processed and analyzed by HPLC-MS/MS to determine the peak area remaining. Experiments completed in duplicate and mean values are depicted. (B) Assessment of A91 (10 μM) for irreversible inhibition of the major CYP450 drug metabolizing isoforms. Peak areas corresponding to the metabolite of known substrates for each isoform are recorded. The percent of control activity was calculated by comparing the peak area obtained in the presence of A91 to that obtained in the absence of A91. Percent inhibition was calculated by subtracting the percent control activity from 100. Time-dependent CYP inhibition is demonstrated if the percent inhibition from samples pre-incubated with the NADPH-regenerating system is larger than those without NADPH-regeneration. Negative values are a reflection of the variability in the background noise of the experiment. Experiments completed in duplicate and mean values are depicted. (C) Assessment of A91 for inhibition of the hERG channel. The degree of inhibition (%) was obtained by measuring the tail current amplitude, which is induced by a one-second test pulse to -40 mV after a two-second pulse to +20 mV, before and after drug incubation. Experiments completed in duplicate and mean values are depicted.

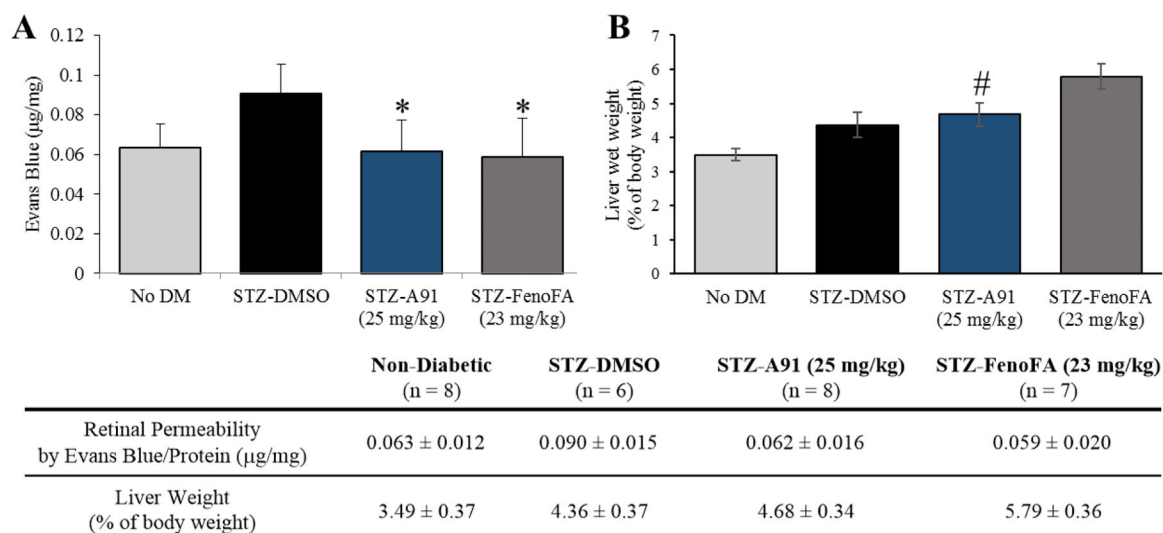


Figure 3.

(A) Effects on retinal permeability and (B) liver phenotype for compound A91 in comparison to FenoFA. 7–8 week-old male Brown Norway rats were injected with streptozotocin (STZ, 55 mg/kg) to induce diabetes. Two weeks after STZ injections, daily treatment (i.p. injection) with compound A91 or FenoFA commenced and lasted for 28 days. Student's t-test was used to compare groups for statistical significance and values are mean \pm SD, # $P < 0.05$ (vs. FenoFA), Retinal permeability was measured using Evans blue as tracer and normalized by total retinal protein. * $P < 0.05$ (vs. STZ-DMSO).

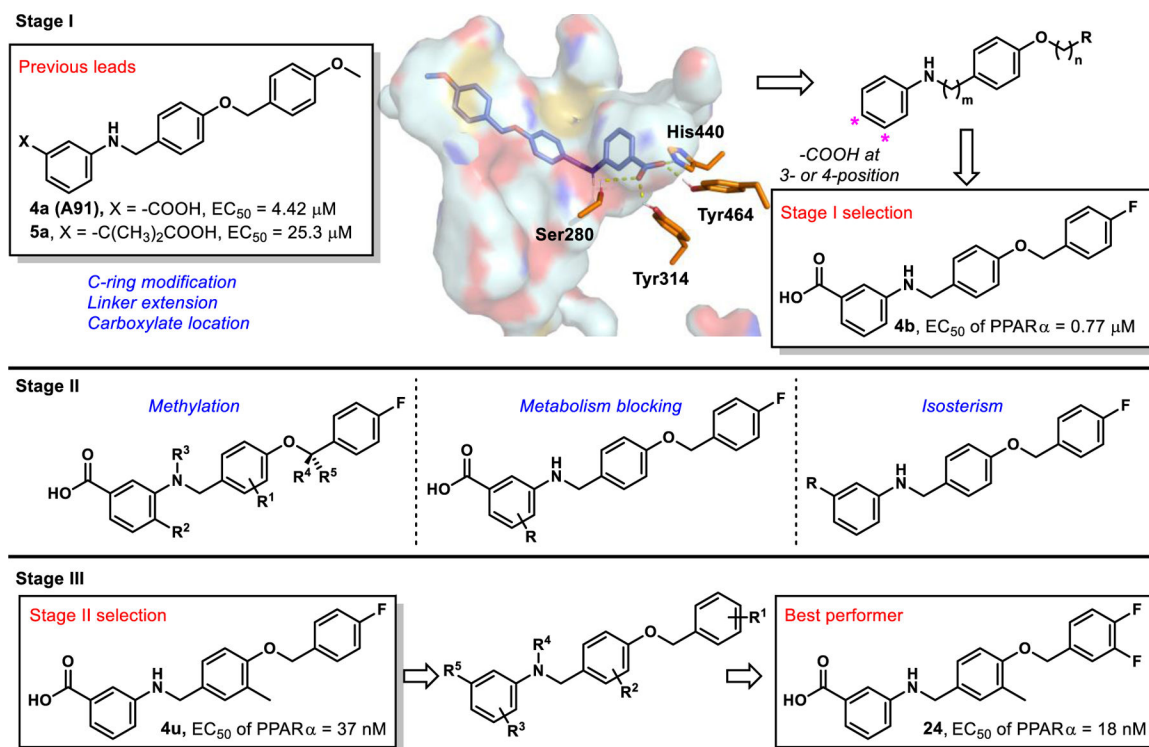


Figure 4. Multi-stage approach to structural improvement of A91. Model depicts **4a (A91)** docked to PDBID 2P54.

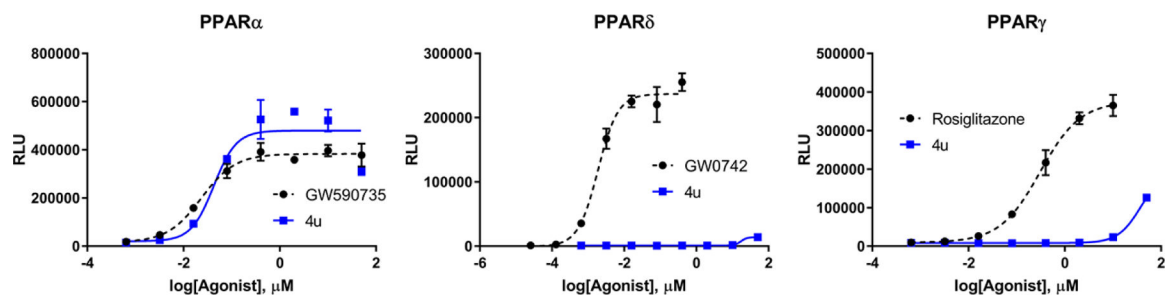


Figure 5.
 Dose-response curves for **4u** (blue squares, solid line) against PPAR α , PPAR δ , and PPAR γ .
 Respective controls are provided as black circles/dashed line.

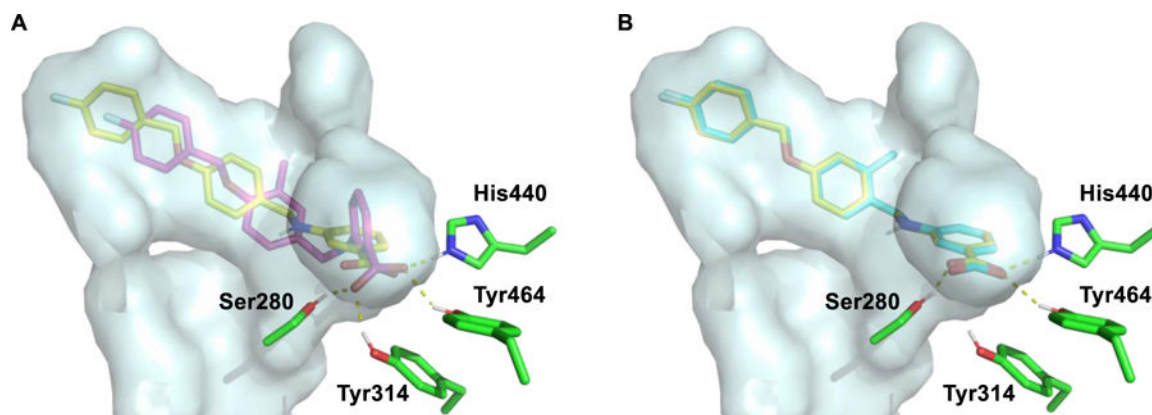


Figure 6. Predicted binding modes of **4b**, **4t**, and **4u**, demonstrating the possible effects of methyl installation on the 2- or 3-position of the B ring. (A) Overlay of the predicted binding poses for **4b** (yellow) and **4u** (magenta). (B) Overlay of the predicted binding poses for **4b** (yellow) and **4t** (cyan). PDBID: 2P54. The pdb file of the docking model is included in the Supporting Information.

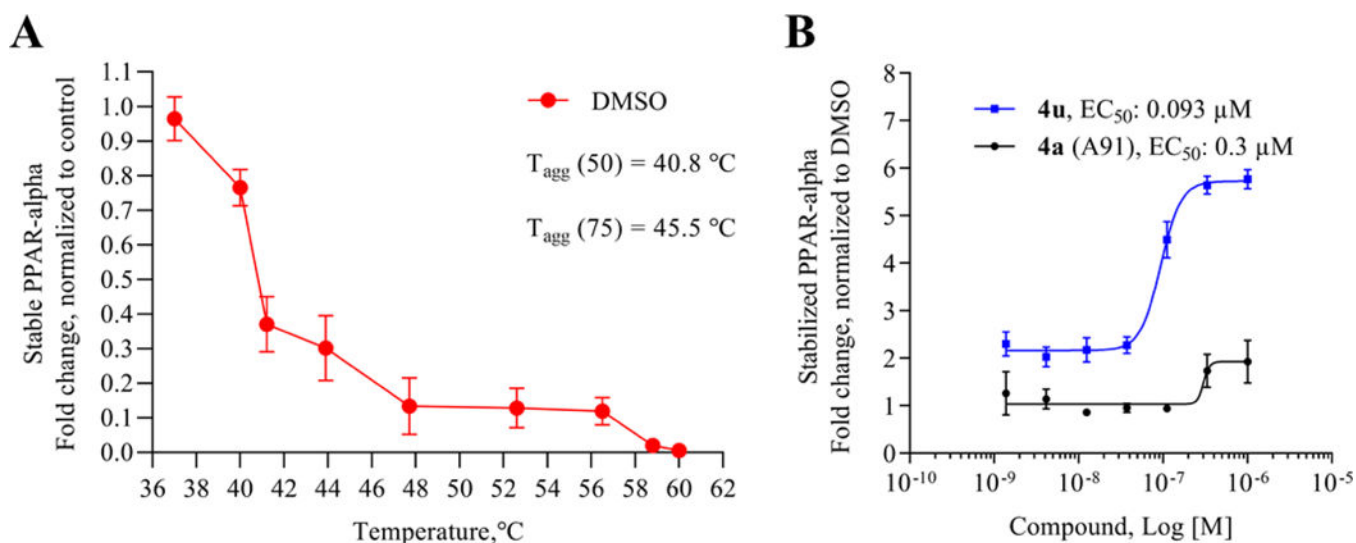


Figure 7. Cellular thermal shift assay results for A91 and **4u**. (A) Thermal melting profile of PPAR α . (B) Dose-dependent stabilization of PPAR α by A91 and **4u**. Experiment was run at 45.5 °C, the $T_{agg}(75)$.

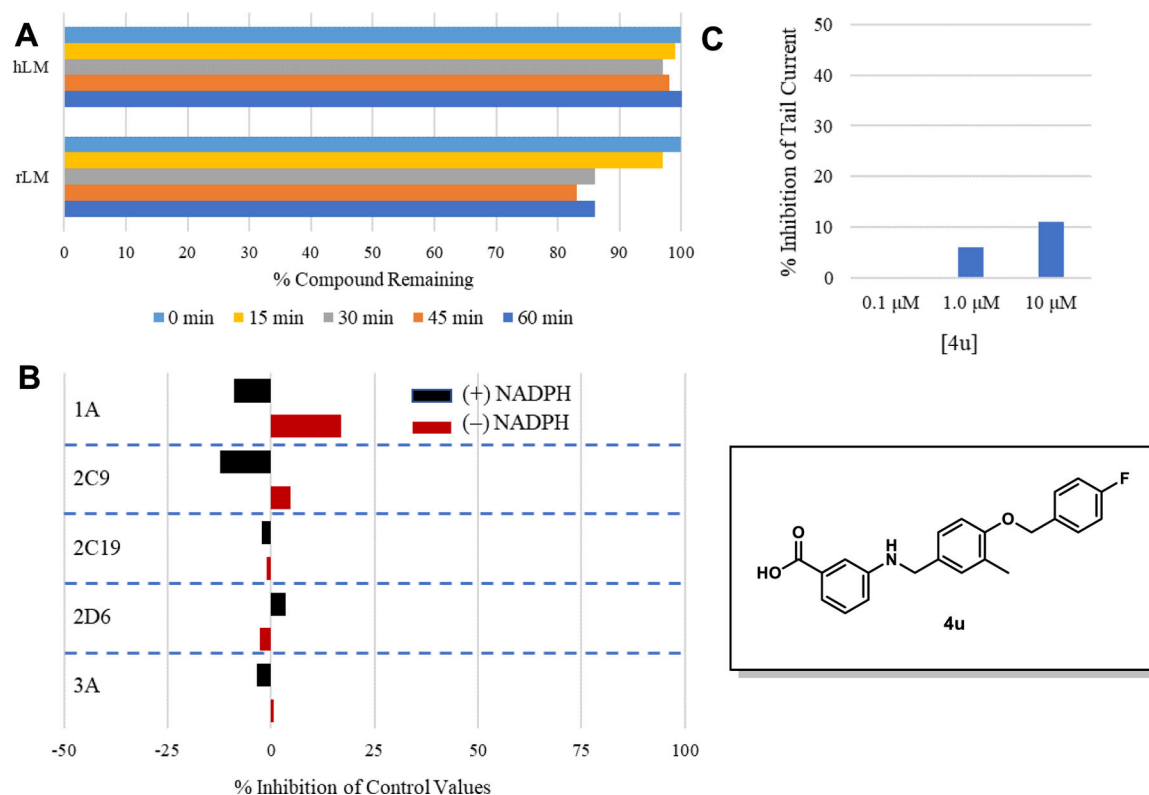


Figure 8. Pharmacokinetic profile of 4u.

(A) **4u** was pre-incubated with pooled liver microsomes in phosphate buffer (pH 7.4) for 5 min at 37 °C. The system was activated by the addition of an NADPH-regenerating system and incubated for 0, 15, 30, 45, and 60 min before being quenched with an acetonitrile/methanol mixture. Samples were processed and analyzed by HPLC-MS/MS to determine the peak area remaining. Experiments completed in duplicate and mean values are depicted. (B) Assessment of **4u** (10 μM) for irreversible inhibition of the major CYP450 drug metabolizing isoforms. Peak areas corresponding to the metabolite of known substrates for each isoform are recorded. The percent of control activity was calculated by comparing the peak area obtained in the presence of **4u** to that obtained in the absence of **4u**. Percent inhibition was calculated by subtracting the percent control activity from 100. Time-dependent CYP inhibition is demonstrated if the percent inhibition from samples pre-incubated with the NADPH-regenerating system is larger than those without NADPH-regeneration. Negative values are a reflection of the variability in the background noise of the experiment. Experiments completed in duplicate and mean values are depicted. (C) Assessment of **4u** for inhibition of the hERG channel. The degree of inhibition (%) was obtained by measuring the tail current amplitude, which is induced by a one-second test pulse to -40 mV after a two-second pulse to +20 mV, before and after drug incubation. Experiments completed in duplicate and mean values are depicted.

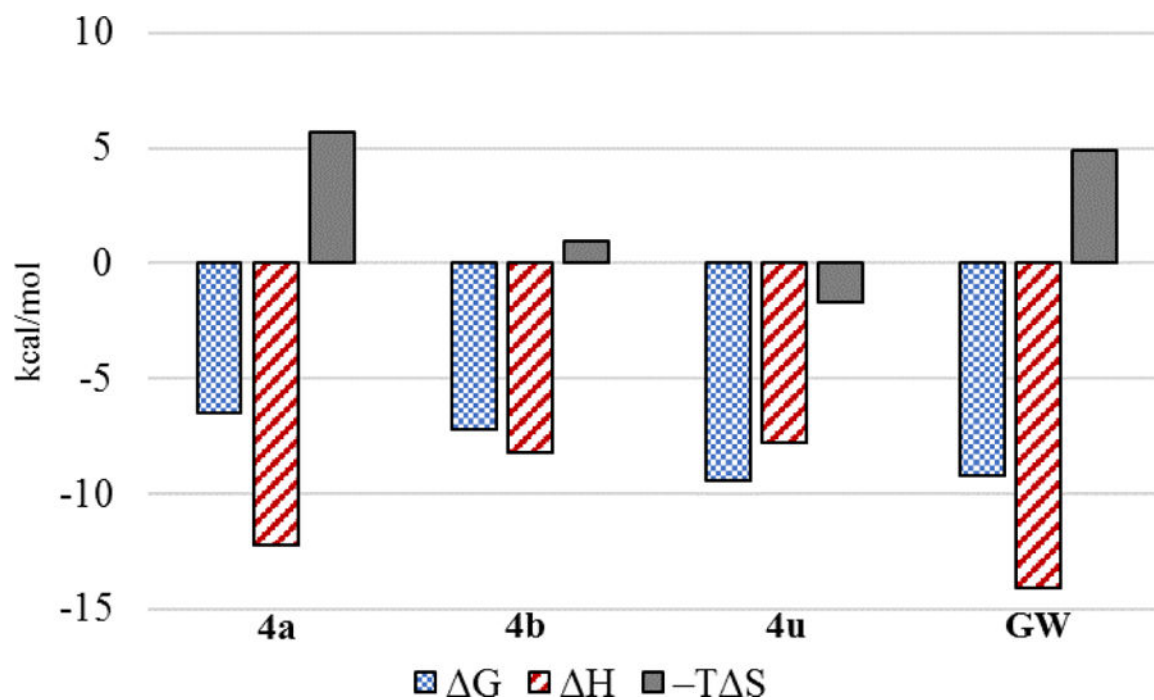


Figure 9.
Thermodynamic binding profiles of select ligands.

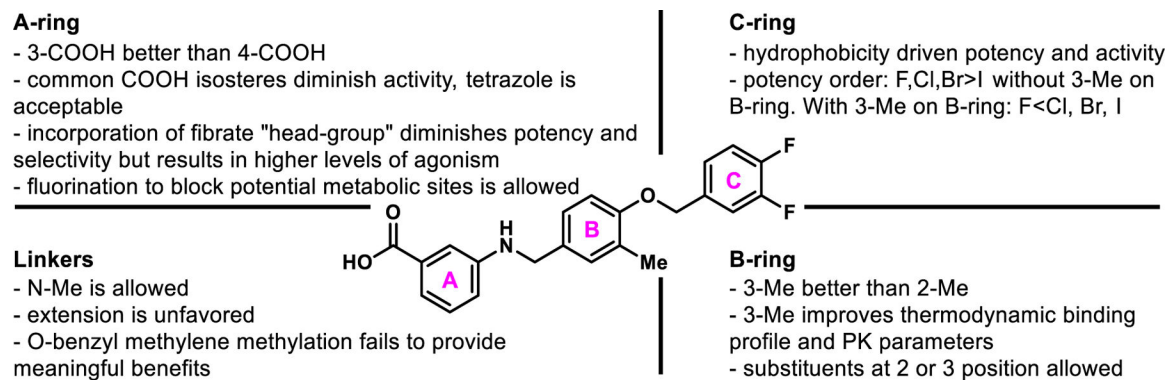
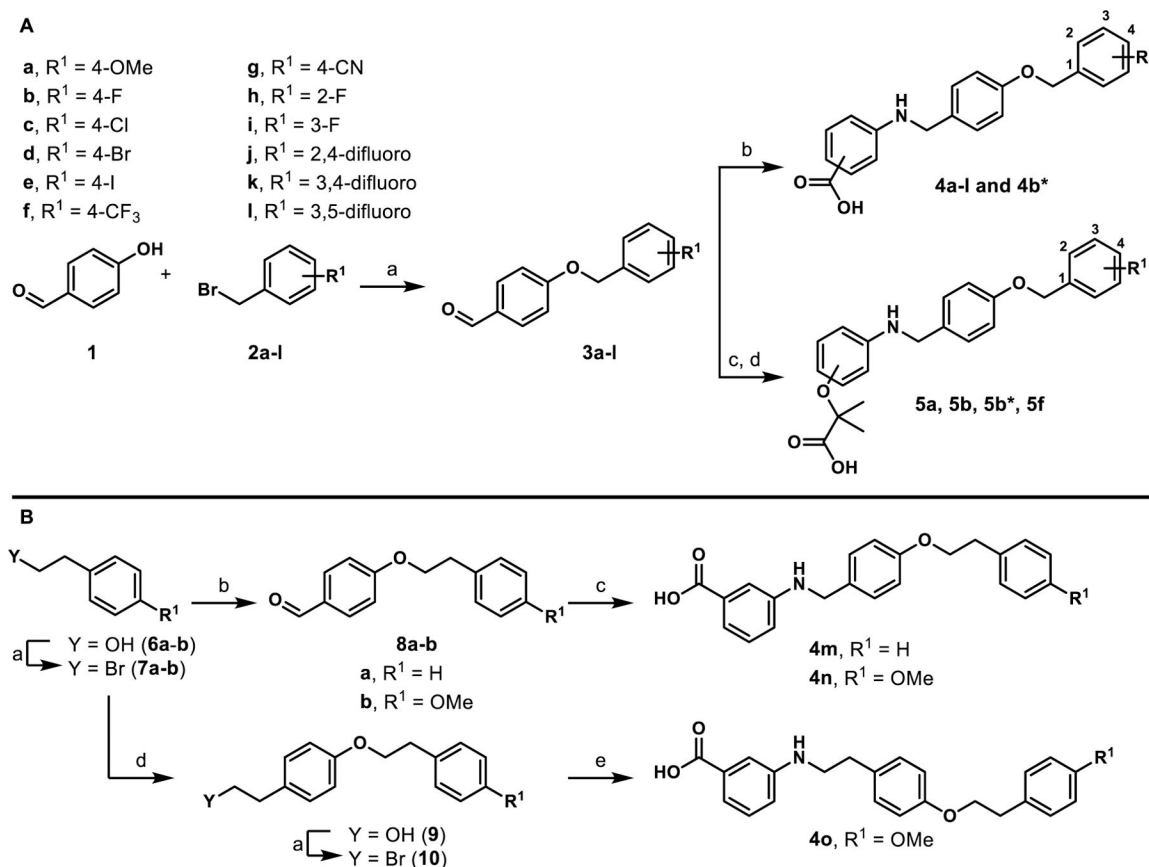
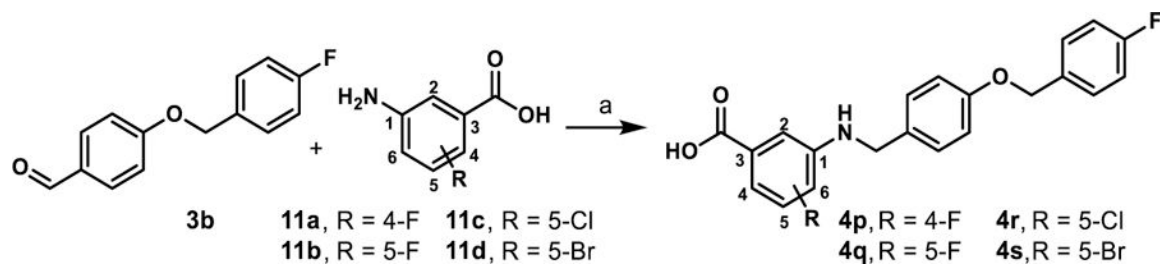


Figure 10.
Summary of current SAR on the 4-benzyloxy-benzylamino chemotype for PPAR α agonism.

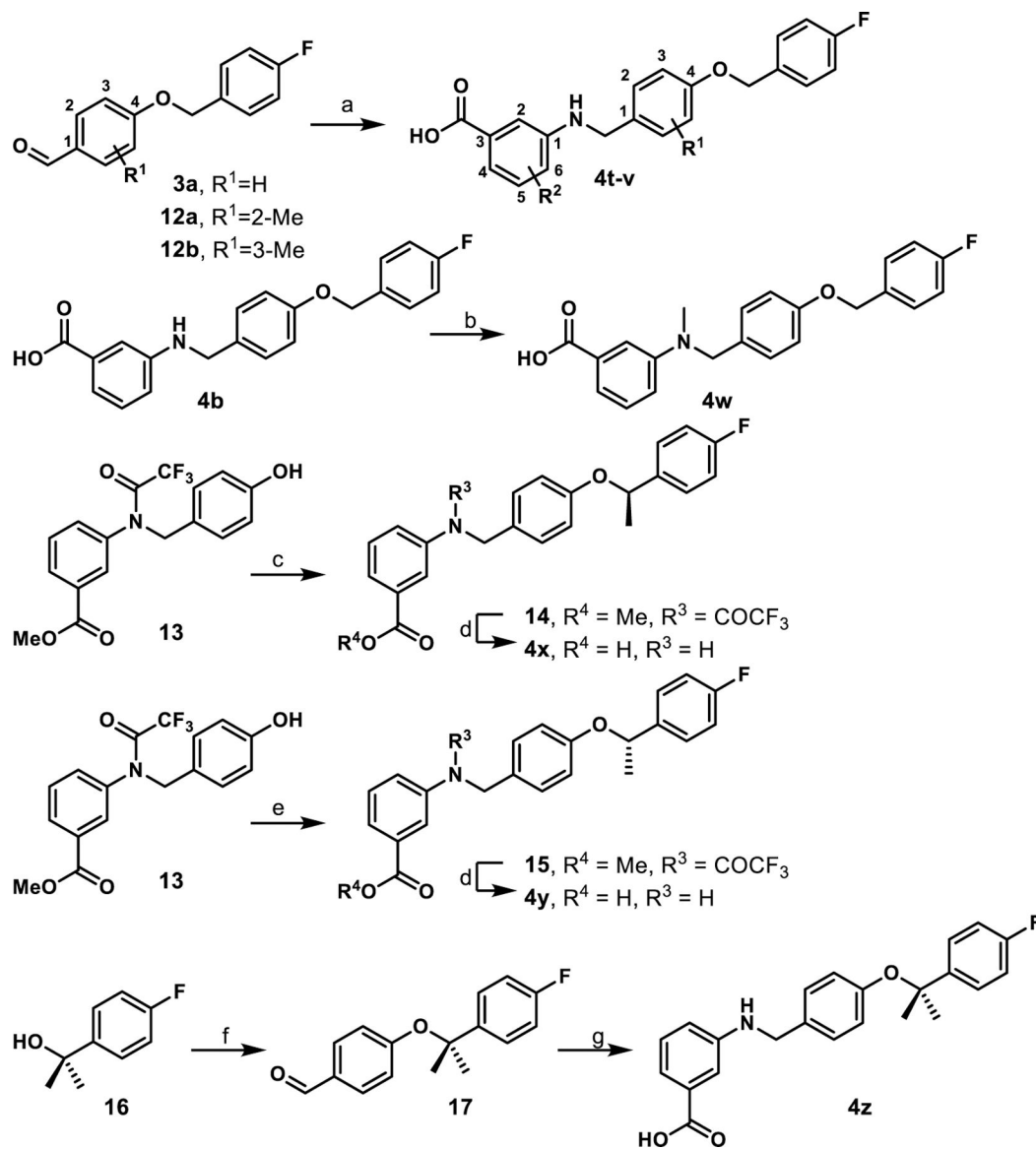
**Scheme 1.**

Synthesis of stage I analogues^a

^aReagents and conditions: A) (a) K₂CO₃, DMF; (b) i) 3-aminobenzoic acid or 4-amino benzoic acid, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 12 h, 32–97%; (c) i) ethyl 2-(3-aminophenoxy)-2-methylpropanoate or ethyl 2-(4-aminophenoxy)-2-methylpropanoate, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 12 h; (d) LiOH·H₂O, THF/CH₃OH/H₂O, 12 h, 14–60%. B) (a) PBr₃, diethyl ether; (b) 4-hydroxybenzaldehyde, K₂CO₃, DMF, 40–65%; (c) 3-aminobenzoic acid, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 12 h, 96%; (d) tyrosol, K₂CO₃, DMF, 26%; (e) 3-aminobenzoic acid, Et₃N, DMF, 25%. All yields reported are unoptimized. Asterisks indicate the -COOH motif is attached *para* instead of *meta*.

**Scheme 2.**Synthesis of 4p-4s^a

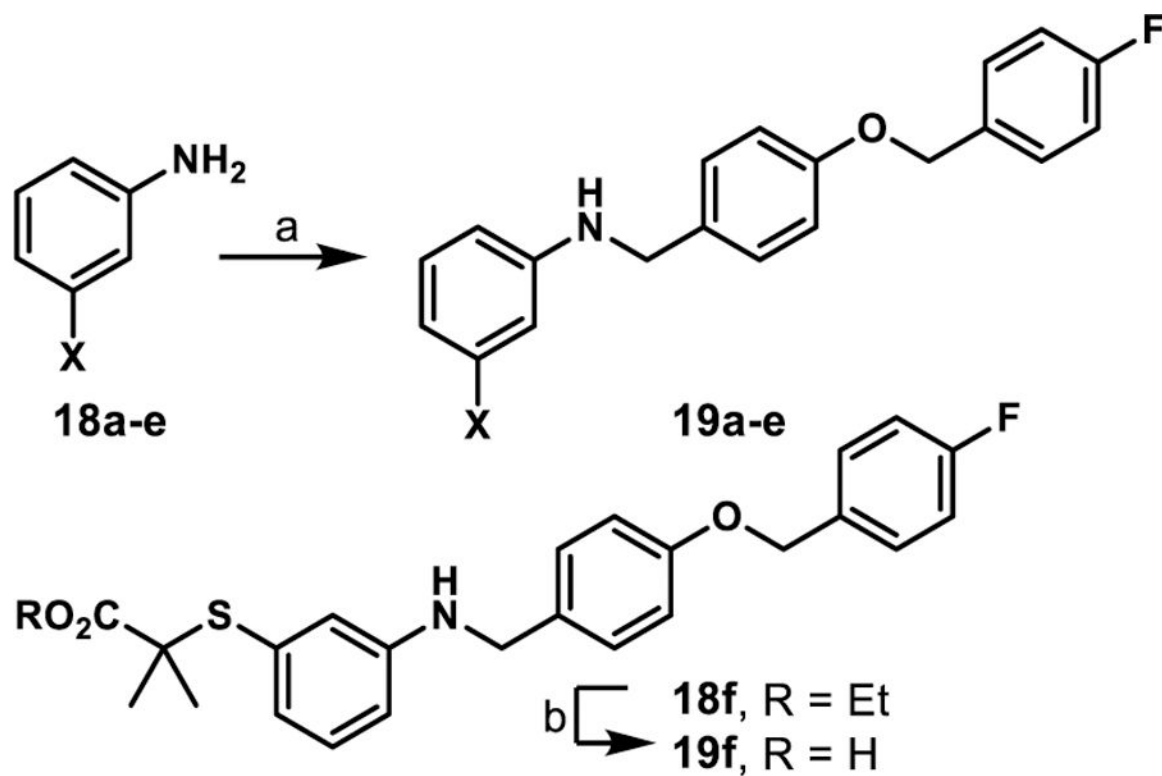
^aReagents and conditions: (a) i) **11**, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 25 °C, 12 h, 78–86%.



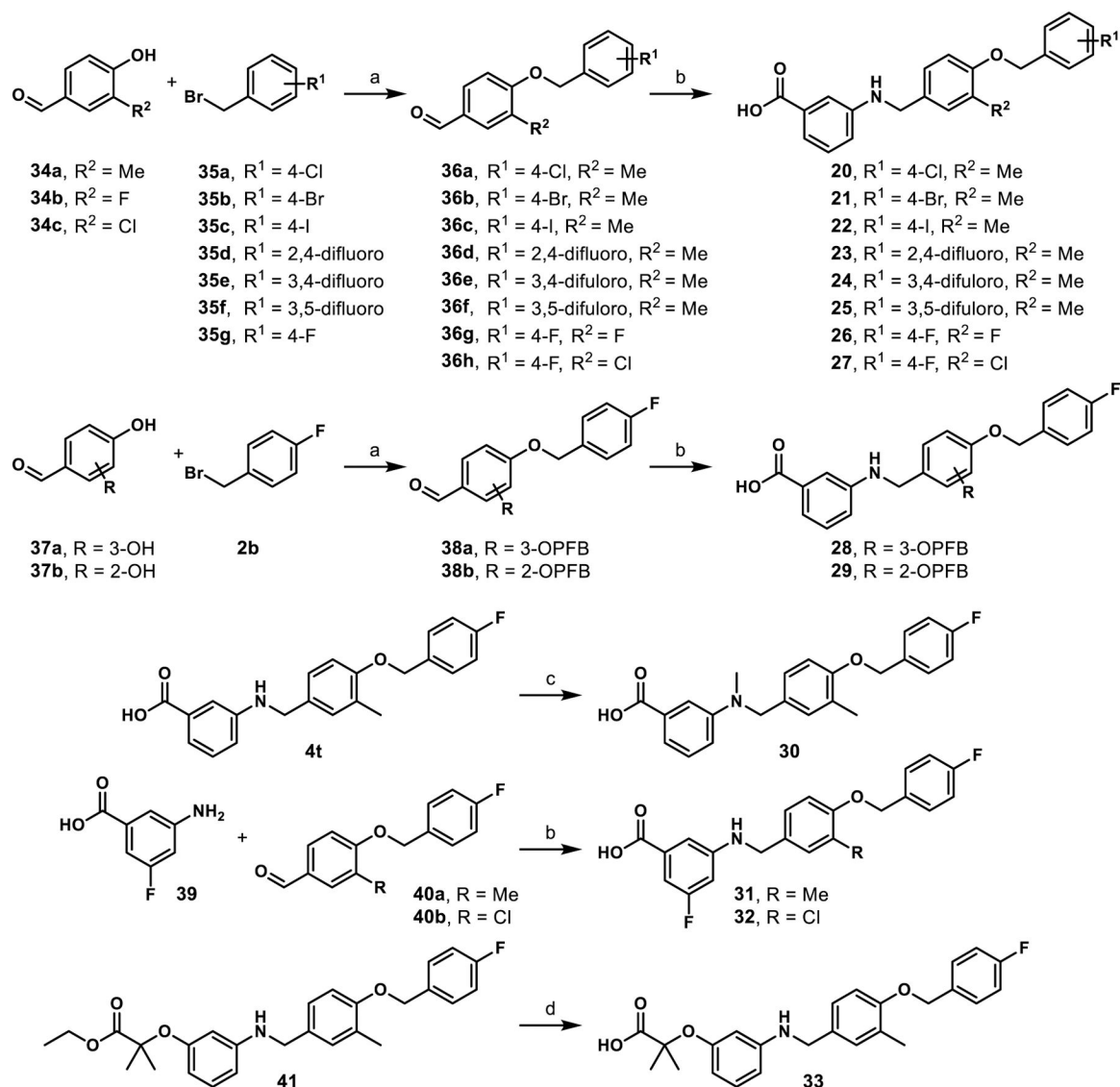
Scheme 3.

Synthesis of methylated derivatives 4t-z^a

^aReagents and conditions: (a) i) 3-aminobenzoic acid or 3-amino-4-methylbenzoic acid, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 25 °C, 12 h, 70–92%; (b) paraformaldehyde, NaBH(OAc)₃, EtOH, reflux 2 h, 25 °C, 12 h, 43%; (c) (*R*)-1-(4-fluorophenyl)ethan-1-ol, di-*tert*-butyl azodicarboxylate, PPh₃, toluene; (d) LiOH·H₂O, THF/CH₃OH/H₂O, 25 °C, 12 h, 65–75%.; (e) (*S*)-1-(4-fluorophenyl)ethan-1-ol, di-*tert*-butyl azodicarboxylate, PPh₃, toluene; (f) 4-hydroxybenzaldehyde, TFAA, DBU, CuCl₂, MeCN, 8%; (g) i) 3-aminobenzoic acid, toluene, 130 °C, ii) NaBH(OAc)₃, AcOH, THF, 12 h, 25%. All yields reported are unoptimized.

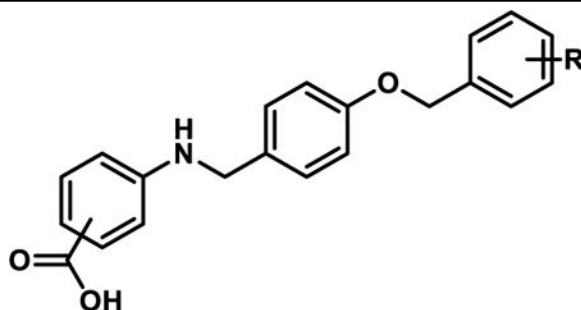
**Scheme 4.**Synthesis of isostere containing derivatives 19a-f^a

^aReagents and conditions: (a) **3b**, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 25 °C, 12 h, 8–80%; (b) LiOH·H₂O, THF/CH₃OH/H₂O, 12 h, 13%.

**Scheme 5.**Synthesis of Stage III Analogues 20–33^a

^aReagents and conditions: (a) K₂CO₃, DMF; (b) i) 3-aminobenzoic acid or 4-amino benzoic acid, toluene, 130 °C, 2–3 h; ii) NaBH(OAc)₃, acetic acid, THF, 12 h, 36–84%; (c) paraformaldehyde, NaBH(OAc)₃, EtOH, reflux 2 h, 25 °C, 12 h, 44%; (d) LiOH·H₂O, THF/CH₃OH/H₂O, 12 h, 32%.

Table 1.

In vitro Activity of 4a-l^a

Compound	R	Ratio (5/50) ^b	Fold-Signal ^c	EC ₅₀ hPPAR α (μ M) ^d	EC ₅₀ hPPAR γ (μ M)	EC ₅₀ hPPAR δ (μ M)	SI ^e
4a (A91)	4-OMe	0.5	1.4	4.43 \pm 0.01	>100	>100	>20
4b	4-F	0.8	1.3	0.77 \pm 0.03	>100	>100	>125
4b*	4-F	0.3	0.4				
4c	4-Cl	0.6	1.5	0.83 \pm 0.04			
4d	4-Br	0.8	1.3	0.97 \pm 0.14			
4e	4-I	0.7	1.1	1.45 \pm 0.02			
4f	4-CF ₃	0.9	0.9	0.96 \pm 0.11			
4g	4-CN	0.6	1.0				
4h	2-F	0.8	1.0				
4i	3-F	0.9	1.2	1.18 \pm 0.05			
4j	2,4-difluoro	1.2	1.2	0.54 \pm 0.09	>100	>100	>175
4k	3,4-difluoro	1.1	1.2	0.36 \pm 0.01			
4l	3,5-difluoro	1.1	1.1				
GW590735		1.0	1.0	0.015 \pm 0.002			
Rosiglitazone					0.28 \pm 0.3		
GW0742						0.0019 \pm 0.002	

^aAll analogues contain the carboxylate head-group at the *meta*-position of the A-ring unless otherwise indicated.

* indicates a *para*-carboxylate.

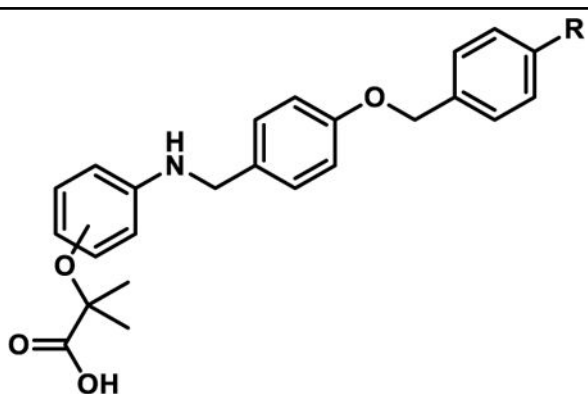
^bRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.

^cRatio of maximal signal strength observed for the compound of interest to that obtained with GW590735.

^dEC₅₀ values represent the mean \pm SEM of at least two separate experiments performed in triplicate.

^eSI = EC₅₀ (PPAR γ or PPAR δ) / EC₅₀ (PPAR α). Blank cells indicate compound was not selected for testing in the corresponding assay.

Table 2.

In vitro Activity of 5a, 5b, 5b*, 5f^a

Compound	R	Ratio (5/50) ^b	Fold-Signal ^c
5a	OMe	0.1	1.8
5b	F	0.2	2.0
5b*	F	0.6	1.8
5f	CF ₃	0.2	1.2

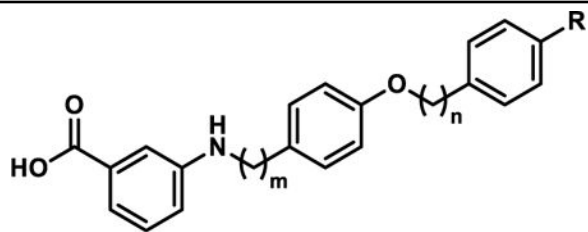
^aAll analogues contain the carboxylate head-group at the *meta*-position of the A-ring unless otherwise indicated.

* indicates a *para*-carboxylate.

^bRatio of relative light unit (RLU) signal at 5 μM and 50 μM compound concentrations.

^cRatio of maximal signal strength observed for the compound of interest to that obtained with GW590735.

Table 3.

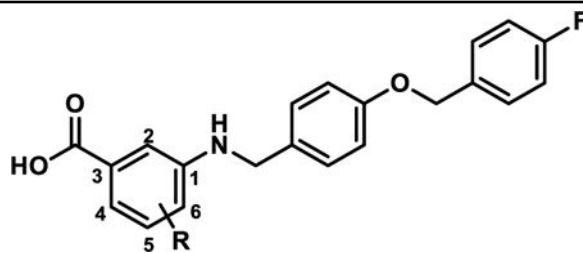
In vitro Activity of 4m-o

Compound	m	n	R	Ratio (5/50) ^a	Fold-Signal ^b
4m	1	2	H	0.1	0.9
4n	1	2	OMe	0.1	0.9
4o	2	2	OMe	1.1	0.1

^aRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.

^bRatio of maximal signal (RLU) strength observed for the compound of interest to that obtained with GW590735.

Table 4.

In vitro Activity of 4p-s

Compound	R	Ratio (5/50) ^a	Fold-Signal ^b	EC ₅₀ hPPAR α (μ M) ^c	EC ₅₀ hPPAR γ (μ M)	EC ₅₀ hPPAR δ (μ M)	SI ^d
4p	4-F	0.8	1.2	1.61 \pm 0.06			
4q	5-F	1.1	1.1	0.58 \pm 0.04	>100	>100	>170
4r	5-Cl	0.1	1.1				
4s	5-Br	1.3	1.4	1.12 \pm 0.08	>100	>100	>100

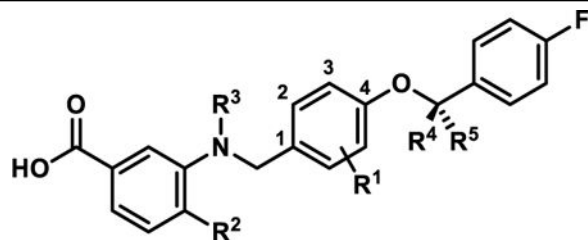
^aRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.

^bRatio of signal maximal signal (RLU) strength observed for the compound of interest to that obtained with GW590735.

^cEC₅₀ values represent the mean \pm SEM of at least two separate experiments performed in triplicate.

^dEC₅₀ (PPAR γ or PPAR δ) / EC₅₀ (PPAR α). Blank cells indicate compound was not selected for testing in the corresponding assay.

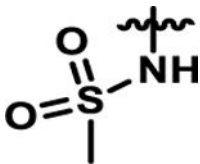
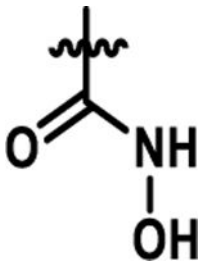
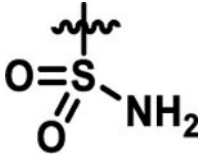
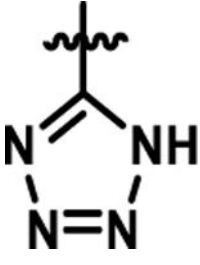
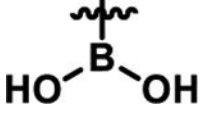
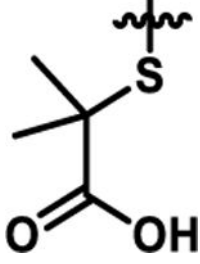
Table 5.

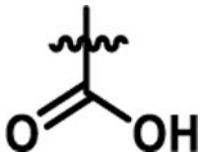
In vitro Activity of 4t-z

Compound	R ¹	R ²	R ³	R ⁴	R ⁵	Ratio (5/50) ^a	Fold-Signal ^b	EC ₅₀ hPPAR α (μ M) ^c	SI ^d
4t	2-Me	H	H	H	H	0.8	1.2	1.45 \pm 0.06	
4u	3-Me	H	H	H	H	1.4	1.2	0.037 \pm 0.01	>2700
4v	H	Me	H	H	H	0.2	0.3		
4w	H	H	Me	H	H	0.8	1.7	0.90 \pm 0.08	
4x	H	H	H	Me	H	0.1	0.6	25.6 \pm 3.5	
4y	H	H	H	H	Me	0.2	1.6		
4z	H	H	H	Me	Me	0.2	0.8	8.52 \pm 1.83	
4a (A91)						0.5	1.4	4.43 \pm 0.01	>20
4b						0.8	1.3	0.77 \pm 0.03	>125
GW590735						1.0	1.0	0.015 \pm 0.002	

^aRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.^bRatio of signal maximal signal (RLU) strength observed for the compound of interest to that obtained with GW590735.^cEC₅₀ values represent the mean \pm SEM of atleast two separate experiments performed in triplicate.^dEC₅₀ (PPAR γ or PPAR δ) / EC₅₀ (PPAR α). Blank cells indicate compound was not selected for testing in the corresponding assay.

Table 6.*In vitro* Activity of Carboxylate Isosteres

Compound	X	Ratio (5/50) ^a	Fold-Signal ^b
19a		1.0	0.1
19b		0.5	0.6
19c		1.2	0.1
19d		0.4	1.6
19e		0.9	0.1
19f		0.2	1.1

Compound	X	Ratio (5/50) ^a	Fold-Signal ^b
4b		0.8	1.3
GW590735		1.0	1.0

^aRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.

^bRatio of signal maximal signal (RLU) strength observed for the compound of interest to that obtained with GW590735.

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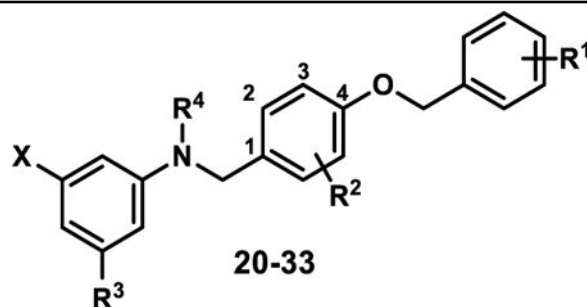
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Table 7.

In Vitro Activity of Stage III Analogues

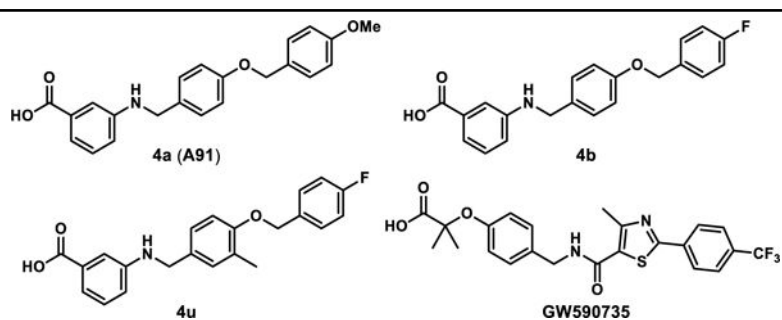


Compound	X	R ¹	R ²	R ³	R ⁴	Ratio (5/50) ^a	Fold-Signal ^b	EC ₅₀ hPPAR α (μ M) ^c
20	COOH	4-Cl	3-Me	H	H	1.2	1.2	0.029 \pm 0.001
21	COOH	4-Br	3-Me	H	H	1.2	1.3	0.031 \pm 0.008
22	COOH	4-I	3-Me	H	H	1.8	1.5	0.027 \pm 0.003
23	COOH	2,4-difluoro	3-Me	H	H	1.2	1.4	0.056 \pm 0.007
24	COOH	3,4-difluoro	3-Me	H	H	1.4	1.6	0.018 \pm 0.001
25	COOH	3,5-difluoro	3-Me	H	H	1.1	1.6	
26	COOH	4-F	3-F	H	H	0.7	1.7	
27	COOH	4-F	3-Cl	H	H	1.1	1.5	
28	COOH	4-OPFB	3-OPFB	H	H	1.0	1.4	0.81 \pm 0.17
29	COOH	4-OPFB	2-OPFB	H	H	0.9	1.2	0.93 \pm 0.02
30	COOH	4-F	3-Me	H	Me	0.8	2.1	0.067 \pm 0.01
31	COOH	4-F	3-Me	F	H	0.9	1.5	0.040 \pm 0.001
32	COOH	4-F	3-Cl	F	H	1.1	1.8	0.052 \pm 0.011
33	OC(CH ₃) ₂ COOH	4-F	3-Me	H	H	1.0	1.6	0.74 \pm 0.04
4b						0.8	1.3	0.77 \pm 0.03
4a (A91)						0.5	1.4	4.43 \pm 0.01
GW590735						1.0	1.0	0.015 \pm 0.002

^aRatio of relative light unit (RLU) signal at 5 μ M and 50 μ M compound concentrations.

^bRatio of signal maximal signal (RLU) strength observed for the compound of interest to that obtained with GW590735.

^cEC₅₀ values represent the mean \pm SEM of at least two separate experiments performed in triplicate. OPFB = *para*-fluorobenzyl. Numbering shown is based on the name of the resulting products. Blank cells indicate compound was not selected for testing in the corresponding assay.

Table 8.Thermodynamic Binding Profiles of Selected Ligands with the Ligand Binding Domain of PPAR α .^a

Compound	EC ₅₀ (μM)	K _d (μM)	G (kcal/mol)	H (kcal/mol)	-T S (kcal/mol)
4a (A91)	4.43 ± 0.01	16.90 ± 1.50	-6.50 ± 0.05	-12.20 ± 1.40	5.71 ± 1.31
4b	0.77 ± 0.03	5.32 ± 0.78	-7.20 ± 0.09	-8.18 ± 0.20	0.98 ± 0.27
4u	0.037 ± 0.005	0.14 ± 0.05	-9.39 ± 0.17	-7.75 ± 0.26	-1.65 ± 0.37
GW590735	0.015 ± 0.002	0.17 ± 0.03	-9.24 ± 0.09	-14.10 ± 0.50	4.90 ± 0.50

^aBuffer: 20 mM HEPES, 150 mM NaCl, pH 7.4. Ligand solution: 200–1000 μM ligand in buffer with 0.8–10% DMSO. Protein solution: 30–85 μM PPAR α LBD in buffer with same concentration of DMSO as corresponding ligand solution. Thermodynamic parameters reported as mean ± S.D. of at least three separate experiments. All experiments were performed at 25 °C. N.D. = not determined.