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Interfacial inhibitors: targeting macromolecular complexes

Yves Pommier, Christophe Marchand

Laboratory of Molecular Pharmacology, Center for Cancer Research, National Cancer Institute, US National Institutes of Health, Bethesda, Maryland 20892, USA.

Abstract

Interfacial inhibitors belong to a broad class of natural products and synthetic drugs that are commonly used to treat cancers as well as bacterial and HIV infections. They bind selectively to interfaces as macromolecular machines assemble and are set in motion. The bound drugs transiently arrest the targeted molecular machines, which can initiate allosteric effects, or desynchronize macromolecular machines that normally function in concert. Here, we review five archetypical examples of interfacial inhibitors: the camptothecins, etoposide, the quinolone antibiotics, the vinca alkaloids and the novel anti-HIV inhibitor raltegravir. We discuss the common and diverging elements between interfacial and allosteric inhibitors and give a perspective for the rationale and methods used to discover novel interfacial inhibitors.

Living cells rely on polymers for cellular architecture (for example, membranes and filaments) and information storage (for example, nucleic acids). Their molecular machines can be extremely complex, consisting of many components — many of them bound to each other non-covalently and reversibly. For instance, the eukaryotic ribosome consists of three RNA chains (5S, 28S and 18S RNAs) and at least 80 polypeptides, the spliceosome and the proteasome consist of approximately 40 different polypeptides each, and the replicative DNA polymerase and RNA polymerase II complexes consist of at least 10 protein subunits each. These molecular machines must also function at high speed, under conditions in which their parts move relative to each other with great precision and robustness. For instance, DNA and RNA polymerases track along their DNA substrates while incorporating nucleobases into new DNA and RNA polymers at average speeds of 30 bases per second¹. DNA topoisomerases I and II (TOP1 and TOP2, respectively) function at up to 100 cycles per second and 4 cycles per second, respectively (K. C. Neuman, personal communication). Moreover, to allow the movement of the polymerase complexes relative to their DNA substrate, DNA helicases and topoisomerases must open and recoil the DNA in concert with the polymerases, thereby creating ‘supermolecular machines’. Uncoupling helicases, topoisomerases and polymerases can be highly lethal, as discussed below.

Molecular machines have a large number of fast-moving parts and interdependent velocities; this makes them particularly vulnerable to small molecules that can infiltrate their hinges and interfere with the movements of individual parts relative to each other. It is therefore

Correspondence to Y.P. pommier@nih.gov.

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relatively easy to understand how small-molecule interfacial inhibitors can alter the function of large molecular machines (such as topoisomerases, integrase, polymerases, GTPases, and so on) and the proper organization of large cellular polymers (such as microtubules).

The interfacial inhibitor concept

Molecular pharmacology and structural studies of topoisomerase inhibitors have led to the conceptualization and demonstration of the interfacial inhibitor concept. The first evidence that topoisomerase inhibitors could trap topoisomerase as the enzyme cleaves DNA (that is, that the drugs trap topoisomerase cleavage complexes) was revealed approximately 30 years ago with the discovery that the antibiotics nalidixic acid and oxolinic acid produce gyrase-mediated DNA double-strand breaks^{2,3}. Furthermore, it was shown that anticancer drugs — including doxorubicin (also known as adria mycin), amsacrine, etoposide and camptothecins — target eukaryotic topoisomerases via a comparable mechanism; at the time, it was proposed that the drugs were trapping topoisomerase cleavage complexes by forming ternary complexes with a drug molecule bound at the interface of the enzymes and the cleaved DNA⁴. It took about 10 years to validate the ternary complex hypothesis, which was achieved when the crystal structure of the TOP1 ternary complex was solved with topotecan (a clinical derivative of camptothecin) intercalated in the cleaved DNA and specifically bound to TOP1 (REF. 5). Soon after, the ternary complex structure was confirmed for the natural alkaloid camptothecin and the indenoisoquinolines, the non-camptothecin inhibitors of TOP1, which are currently in clinical trials^{6,7}.

We proposed the interfacial inhibitor concept because of the common mechanistic principle underlying the mechanism of action of distantly related natural drugs^{8,9}. Camptothecin and brefeldin A served as the landmark examples, each targeting different macromolecular machines: camptothecin targets TOP1–DNA complexes, whereas brefeldin A targets the protein–protein complex formed by ADP-ribosylation factor and guanine nucleotide exchange factor⁸. The macromolecular structures of natural inhibitors have been extensively analysed in the literature; this has extended the relevance of the interfacial inhibitor concept to a range of drugs and molecular machines, including cyclosporine, tacrolimus (also known as FK506), forskolin, fusicoccin, rapamycin, colchicine, vinca alkaloids, paclitaxel (also known as Taxol) and α -amanitin⁸. Since then, anti biotics have been shown to act as interfacial inhibitors for bacterial type II topoisomerases^{10–12}. The interfacial inhibition mechanism has recently been extended to purely synthetic compounds such as the anti-AIDS drugs that target the HIV integrase–viral DNA complexes. Stunning co-crystal structures of integrase with its viral substrates have revealed the trapping of the integrase–DNA complex as the drugs bind at its interface^{13–15}. The interfacial hypothesis for the targeting of eukaryotic TOP2 by etoposide has also recently been confirmed¹⁶, as discussed below.

Examples of interfacial inhibitors

Several drug classes and targets of interfacial inhibitors are listed in TABLE 1. Below, we describe several examples of interfacial inhibitors, to clarify the concept of interfacial inhibition.

Camptothecins and non-camptothecin anticancer TOP1 inhibitors.

Topoisomerases belong to a large family of enzymes that regulate the topology of DNA without changing its primary sequence. They are ubiquitously expressed in eukaryotes, prokaryotes and archaea, and are also encoded by some viruses. Topoisomerases can be viewed as '3 in 1 machines' because a single enzyme performs three consecutive reactions. First, a topoisomerase cleaves the DNA backbone (endonuclease-equivalent function); second, it allows changes in DNA supercoils (DNA strand passage or untwisting activity); and third, it re-ligates the cleaved DNA (ligase or recombinase-like activity). Topoisomerases are classified as type I or type II, depending on whether they cleave one or both DNA strands in concert, respectively¹⁷.

The human genome contains six topoisomerase genes that encode two type IB enzymes (TOP1 and mitochondrial TOP1 (TOP1MT)), two type IA enzymes (TOP3 α and TOP3 β) and two type IIA enzymes (TOP2 α and TOP2 β). TOP1MT is only found in vertebrates. It is encoded in the nucleus but only acts on mitochondrial DNA⁴. The bacterium *Escherichia coli* encodes four topoisomerases: two type IA enzymes (topoisomerase I and topoisomerase III) and two type IIA enzymes (topoisomerase IV and DNA gyrase).

Topoisomerase inhibitors are important medicinal drugs. They are widely used as antibiotics and anticancer drugs. The quinolone derivatives (nalidixic acid, oxolinic acid, norfloxacin, ciprofloxacin, levofloxacin, moxifloxacin and gemifloxacin) target the bacterial type II enzymes DNA gyrase and topoisomerase IV without affecting the host type II enzymes (TOP2 α and TOP2 β), which explains their high antibacterial selectivity⁴. There are 14 anticancer drugs targeting topoisomerases that have been approved by the US Food and Drug Administration (FDA), and several more are in clinical development. Two of these 14 drugs — topo tecan and irinotecan — target TOP1, whereas the other 12 (doxorubicin, daunorubicin, doxil, myocet, idarubicin, epirubicin, aclarubicin, dexrazoxane, mitoxantrone, etoposide, etopophos and teniposide) target TOP2 (both TOP2 α and TOP2 β). Notably, anticancer drugs targeting topoisomerases are highly selective for either TOP1 or TOP2 but they do not target both enzymes.

FIGURE 1 shows the structure of the ternary complex of camptothecin bound at the interface of TOP1 and its nicked DNA substrate^{6,7,18}. It is crucial to note that camptothecin binds to both DNA and TOP1. The binding of camptothecin to DNA is established by π - π interactions between the heterocyclic ring system of the drug and the nucleobases flanking the cleavage site⁷, which hinders the rotation of the DNA¹⁹. When binding to the TOP1 polypeptide, the drug forms hydrogen bonds between its B, D and E rings and three amino acid residues (Asp533, Arg364 and Asn722)^{5,6,18}. Mutations in any of these single residues render TOP1 highly resistant to camptothecin and its clinical derivatives, despite the fact that camptothecin can still be seen in crystal structures^{18,20}. This apparent contradiction highlights that interfacial inhibitors are primarily kinetic inhibitors and that all the contacts are essential to retain the drugs in the ternary complex (that is, reduce the 'off-rate'). Off-rates cannot be inferred in crystal structures in which the ligand is used at saturating concentrations.

The stereospecific requirement for the optimum binding of camptothecin²¹, which is a general characteristic of interfacial inhibitors, is reflected by the fact that only the natural 20*S* isomer fits inside the TOP1 interfacial binding pocket (FIG. 1); the synthetic 20*R* isomer, which is inactive against TOP1, cannot fit inside the interfacial binding pocket²¹.

The exquisite selectivity of camptothecin for TOP1 is also reflected by two genetic observations. First, yeast cells in which the *TOP1* gene is inactivated are completely immune to camptothecin^{22,23}. Second, camptothecin-producing plants are immune to the drug because the TOP1 produced by these plants is homogeneously mutated at a single asparagine residue next to the catalytic tyrosine residue²⁴. Notably, the corresponding mutation in human TOP1 (Asn722 to Ser722) had previously been identified in camptothecin-resistant human leukaemia cells²⁵.

Understanding how the trapping of TOP1 leads to DNA damage and anticancer activity is crucial for rationalizing the use of TOP1-targeted drugs and their combination with other drugs^{26,27}. As indicated above, the transient stabilization of the TOP1 cleavage complex slows down the TOP1 catalytic cycles, which leads to DNA damage as the fast movements of the replication and transcription complexes produce collisions with the drug-stalled TOP1 cleavage complex.

Etoposide and anticancer TOP2 inhibitors.

Anticancer TOP2 inhibitors were the first drugs that were hypothesized to form ternary complexes by intercalating between the base pairs flanking the cleavage sites generated by a topoisomerase^{28,29}. This hypothesis was derived from the observation that each chemical class of inhibitors selectively traps TOP2 at different sites^{30,31}. Furthermore, sequencing a large collection of drug-induced TOP2 cleavage sites revealed that these differences are defined by the base pairs that flank those sites³². Namely, doxorubicin, daunorubicin, epirubicin and other anthracyclines trap TOP2 cleavage complexes with an adenine at the 3' end of the break (the A-1 site)^{28,33}, whereas amsacrine preferentially traps the A+1 sites; etoposide, teniposide and mitoxantrone preferentially trap TOP2 cleavage complexes at C-1 sites^{29,34}.

Chan and colleagues¹⁶ have successfully co-crystallized etoposide in a ternary complex with TOP2 β , based on the preference of etoposide and teniposide for the trapping of TOP2 at the C-1 base³². The crystal structure confirms the prediction²⁹ that the etoposide drug molecule stacks against the C-1 or G+5 base pair and forms a network of hydrogen bonds and van der Waals interactions with TOP2 (FIG. 2). Furthermore, the same group has successfully co-crystallized mitoxantrone in the TOP2 cleavage complex and demonstrated that mitoxantrone has a similar mechanism of action to etoposide (N. L. Chan, personal communication).

Quinolone antibiotics: inhibitors of bacterial type II topoisomerases.

Antibiotics have also been successfully co-crystallized with the bacterial type II topoisomerases topoisomerase IV^{10,11} and DNA gyrase¹², and have been shown to act as interfacial inhibitors. Because topoisomerase IV and DNA gyrase are essential for bacterial replication and are sufficiently different from the eukaryotic TOP2 α and TOP2 β enzymes,

which are insensitive to the antibiotics, bacterial type II topoisomerases are prime targets for antibiotics⁴. Moreover, the same antibacterial topoisomerase inhibitors usually target both DNA gyrase and topoisomerase IV owing to the high structural similarities between the two prokaryotic enzymes⁴.

Quinolone antibiotics represent the archetype of bacterial type II topoisomerase poisons. They were originally limited to Gram-negative bacteria but the introduction of a fluorine atom in their structure broadened their antibacterial spectrum. Several generations of fluoroquinolone derivatives have led to some of the most potent antibiotics available to date, such as ciprofloxacin and levofloxacin⁴ (FIG. 3). Similarly to other topoisomerase poisons, fluoroquinolones trap bacterial type II topoisomerases in a stabilized cleavage complex by stacking between the base pairs flanking the cleavage sites at the interface of the parC and parE subunits of topoisomerase IV^{10,35} (FIG. 3).

In the context of DNA gyrase and interfacial inhibition, it is relevant to mention the *ccd* operon of the F plasmid of *E. coli*, which represents an archetype of the toxin–antitoxin modules that are present in many bacteria and archaea, and function as biological interfacial inhibitors³⁶. These plasmid-encoded toxin–antitoxin systems may be used for the development of novel antibiotics³⁷. The *ccd* operon on the F plasmid encodes two proteins: the toxin CcdB and its antitoxin CcdA. CcdB poisons DNA gyrase by binding as a homodimer to both subunits A of DNA gyrase in the centre of the heterodimer, keeping the gyrase in an open conformation with the cleaved DNA³⁸. CcdA also functions as a homodimer³⁶; it neutralizes and rejuvenates DNA gyrase by binding to the CcdB dimer interface, thereby interfering with the binding of CcdB to DNA gyrase subunit A and forming CcdB–CcdA tetramers that act as repressors of the *ccd* operon (reviewed in REF. 36). Hence, the toxin–antitoxin *ccd* operon represents a highly accomplished interfacial inhibitor system that uses DNA gyrase as its endogenous target.

Another type of interfacial inhibition has been described for the catalytic inhibitors of eukaryotic TOP2, represented by the bisdioxopiperazine ICRF-193 and its chemotherapeutic derivative dexrazoxane (ICRF-187)^{39–41}. In this complex TOP2 does not cleave DNA, and the enzyme homodimer is trapped encircling both DNA double helices after re-ligation of the passing strand⁴. Dexrazoxane binds to a site at the interface of the ATP domains of two TOP2 molecules and consequently stabilizes TOP2 in this trapped intermediate state⁴¹.

Tubulin inhibitors.

Interfacial inhibitors can also target the protein–protein interface of structural rather than enzymatic macromolecular complexes such as microtubules. Microtubules have important roles in cellular development, proliferation, trafficking, signalling and migration. They represent key components of the cytoskeleton. Microtubules have been the targets of anticancer therapy for decades, and there are currently 42 microtubule-binding agents that have either been approved or are in clinical trials for the treatment of cancer⁴². Microtubules are composed of α - and β -tubulin heterodimers arranged in the form of long filamentous, tube-shaped polymers that can be several micrometers long. Microtubule dynamics are crucial for mitosis and protein trafficking. Mitotic spindle microtubules have a very high exchange rate (with half-times as low as 10 seconds) with the tubulin molecules in the

soluble pool. Any reduction in microtubule dynamics results in a mitosis block that may lead to apoptosis⁴³.

Microtubule-binding agents are often classified into two main categories — the microtubule-destabilizing agents and the microtubule-stabilizing agents — owing to their differential effects on microtubule mass at high doses. The destabilizing agents either bind to the vinca domain or to the colchicine domain of tubulin. Vinca domain-binding agents include the vinca alkaloids such as vinblastine and vincristine. Colchicine domain-binding agents include colchicine and its analogues, such as podophyllotoxin and combretastatin. The stabilizing agents bind to the taxane-binding site, and include taxanes such as paclitaxel and docetaxel. However, this classification of stabilizing or destabilizing agents has been shown to be oversimplistic, as both categories of microtubule-binding agents exhibit the same powerful suppression of microtubule dynamics at concentrations up to 100-fold lower than needed to observe either the stabilization or destabilization of microtubules⁴².

The vinca alkaloid vinblastine (FIG. 4a,b) binds at the interface of an α - and β -tubulin heterodimer⁴⁴ in a site involving residues from both monomers (FIG. 4c,d). The effect of vinblastine binding results in the suppression of spindle microtubule dynamics and stabilizes curved microtubule polymers, which interferes with their proper extension and leads to the slowing or blocking of mitosis and the induction of apoptotic cell death.

HIV integrase strand transfer inhibitors: synthetic interfacial inhibitors.

As shown in TABLE 1, most of the interfacial inhibitors that were historically identified were natural products. The elucidation of the crystal structure of integrase inhibitors bound to the enzyme–DNA complex^{13,15,45} demonstrates that synthetic molecules can be successfully developed into archetypical interfacial inhibitors.

HIV integrase is the most recently validated target for the treatment of AIDS⁴⁶. The FDA approved raltegravir (Isentress; Merck), an integrase inhibitor (FIG. 5), in 2007 based on the remarkable activity and good tolerance of the drug. Several second-generation integrase inhibitors are in development, including elvitegravir (GS-9137; Gilead Sciences) and dolutegravir (GSK-1349572; GlaxoSmithKline & Shionogi-ViiV), which are in Phase III clinical trials with the aim of overcoming the resistance to raltegravir and improving their pharmacokinetics profile^{45,47,48}. HIV integrase is one of the three enzymes encoded by the HIV *pol* gene, together with pro-tease and reverse transcriptase (reviewed in REF. 49). HIV integrase inserts the viral genome into a host chromosome following reverse transcription. Integration is required for viral replication because only the integrated genome is efficiently transcribed and able to produce both viral RNA genome copies and the viral mRNA that generates the proteins for new virus particles.

Integrase acts in two consecutive steps. First, it activates the viral DNA by generating 3′-deoxyribose hydroxyl ends as the enzyme catalyses the removal of a few (generally two) nucleotides from the 3′ ends of the viral DNA immediately after the conserved CA dinucleotide sequences that are common to retroviruses. This first reaction, which is referred to as 3′ processing, takes place in the cytoplasm immediately after reverse transcription. The activated viral DNA is then packaged into pre-integration complexes with integrase bound to

the viral ends. After 2–4 hours, once they have passed through the nuclear membrane, the pre-integration complexes reach a host chromosome. The second step of the integration reaction is referred to as strand transfer. It is an integrase-mediated electrophilic attack from the 3'-hydroxyl ends of the viral DNA towards the phosphodiester DNA backbones of a host chromosome. This results in the insertion of the viral DNA into the host chromosome^{49,50}.

The integrase-mediated strand transfer reaction relies on the precise positioning of the phosphodiester backbone of the host DNA and the viral 3' ends of the DNA (the attacking nucleophiles). The host DNA is held in place by two divalent metal cations (Mg^{2+} or Mn^{2+} ; FIG. 5) that also coordinate the three crucial catalytic residues of HIV integrase (Asp64, Asp116 and Glu152 for human HIV integrase) (FIG. 5). Crystal structures of raltegravir bound to the integrase from prototype foamy virus (which has an active site that is highly conserved with HIV integrase) showed that raltegravir binds at the interface of the integrase–DNA– Mg^{2+} complex^{13,15}. FIGURE 5 summarizes the essential features of the interfacial binding of raltegravir. Raltegravir occupies the active site of integrase and inactivates integrase by blocking the binding of the host DNA as it displaces the terminal adenylyl nucleotide. Raltegravir is held in the active site of integrase by three key contacts: chelation of both Mg^{2+} ions by the drug carbonyl groups; π -stacking of the halophenyl ring of the drug with the penultimate cytosine base of the viral DNA; and π -stacking of the oxadiazole ring of the drug with the Tyr212 residue of prototype foamy virus integrase (Tyr143 in HIV integrase) (FIG. 5).

In spite of — and possibly owing to — its exclusive targeting of integrase–DNA complexes, raltegravir selects for drug-resistant viruses in some patients, who then become refractory to the drug⁴⁷. The identification and characterization of the corresponding integrase mutations are being effectively used to generate second-generation integrase inhibitors to overcome such mutations^{45,47,48,51}. Co-crystal structures of elvitegravir and dolutegravir with the prototype foamy virus intasome have recently revealed that both drugs also act as interfacial inhibitors^{15,45}.

Definition of interfacial inhibition

Here, we summarize and further clarify the common properties of interfacial inhibitors based on the five examples described above and the widespread nature of the interfacial inhibition paradigm. It should be noted that targeting a protein interface does not always qualify for our definition of interfacial inhibitors. We also discuss interfacial inhibition in the context of other inhibition mechanisms, and define the common and differential elements between interfacial and allosteric inhibition in more detail.

Common and essential characteristics of interfacial inhibitors that differentiate them from drugs that target unbound (free) macromolecular interfaces.

BOX 1 summarizes the salient features of interfacial inhibitors, and TABLE 1 lists a number of interfacial inhibitors. One of the essential characteristics of interfacial inhibitors is their binding to a site generated at the interface of two (or more) biomolecules that are bound to each other and engaged in a functional complex. As illustrated by the examples in FIGS 1–5 and TABLE 1, these biomolecules can be polypeptides, nucleic acids (DNA or RNA) or

even divalent metal cations. If the interface is generated by proteins (such as α -tubulin or β -tubulin; FIG. 4), the polypeptides have to be encoded by separate genes. However, if the interface is generated by different protein domains from a single protein encoded by one gene, we do not refer to this binding as interfacial. Although some biological interfaces could be generated by a small endogenous metabolite or mediator, such as a nucleotide or a hormone bound to a polypeptide, such examples have not yet come to our attention. Hence, it seems most appropriate to use the term 'biomolecules' to define the individual components that generate the binding interface.

It is important to clarify that our definition of an interfacial inhibitor excludes a drug that targets a macromolecular interface by blocking the binding of another endogenous macromolecule (such as a protein or nucleic acid) to that same surface. For instance, the B cell lymphoma-extra large (BCL-X_L) inhibitor ABT-737, which blocks the heterodimerization of the proapoptotic BCL-2-homologous antagonist/killer (BAK) or BCL-2 antagonist of cell death (BAD) to BCL-X_L, does not qualify as an interfacial inhibitor; rather, it qualifies as a competitive inhibitor of BAK or BAD to BCL-X_L. Similarly, nutlins block the binding of cellular tumour antigen p53 to the oncoprotein HDM2 (also known as MDM2) by binding to the p53-binding pocket of HDM2. However, they do not bind to the p53–HDM2 interface and therefore do not qualify as interfacial inhibitors but instead as competitive inhibitors (BOX 2) (reviewed in REF. 52).

Historically, interfacial inhibitors were discovered before their target (or targets) had been identified. These first interfacial inhibitors were generally natural products (TABLE 1), which led us to propose in our first description of the interfacial inhibitor concept that interfacial inhibitors represented “one of Nature’s paradigms” for generating inhibitors or toxins^{8,9}. It is important to keep this principle in mind when dealing with natural products and looking for their molecular targets. This is even more relevant if the natural product contains racemic centres and if the natural isomer is markedly more active than the synthetic stereoisomers. As indicated above, this is the case with camptothecin; only the natural 20*S* isomer, and not the synthetic 20*R* isomer, targets TOP1.

In recent years, several examples of synthetic interfacial inhibitors have emerged (TABLE 1). For instance, the indenoisoquinoline drugs targeting TOP1 were developed based on the principles established by studying the camptothecins⁷. Raltegravir was discovered as a specific HIV integrase inhibitor by searching chemical libraries of compounds with antiviral activity. These two examples of synthetic inhibitors demonstrate the feasibility of synthetic drug development programmes for the discovery of interfacial inhibitors.

Comparison of interfacial and allosteric inhibitors.

Allosteric inhibitors generally act as interfacial inhibitors, as in many cases they bind at the interface of biomolecules. A classical example of an allosteric interfacial inhibitor is the nicotine molecule, which binds to the neuronal acetylcholine receptor⁵³. However, some allosteric inhibitors bind to a single polypeptide with flexible domains, such as allosteric inhibitors of the cyclic AMP-specific 3',5'-cyclic phosphodiesterase 4D, which bind and stabilize an inactive form of the enzyme with its regulatory domain closed across the active site⁵⁴. In fact, in the case of allosteric inhibitors that bind and distort a single polypeptide or

a single nucleic acid effector, the distinction between allosteric inhibitors and interfacial inhibitors is that allosteric inhibitors alter the movement or the arrangement of flexible protein domains that are encoded by a single gene, whereas in the case of allosteric interfacial inhibitors the domains are represented by separately encoded and non-covalently linked polypeptides.

Allostery has become a prominent biological and pharmacological topic⁵⁵; however, in some contexts its definition has evolved. For instance, allosteric effects are commonly used to refer to propagated effects that produce conformational changes within a biomolecule. This definition is different from the original one proposed by Changeux, Jacob and Monod^{55,56} (BOX 2) and cannot be readily related to interfacial interactions.

Concluding remarks and prospects

Targeting complex macromolecular systems is becoming mainstream in drug discovery. As protein surfaces tend to be relatively shallow, it is thermodynamically difficult to identify or synthesize small molecules that compete with large endogenous ligands. Targeting multicomponent molecular machines using interfacial inhibitors that bind at their interfacial hinges overcomes this difficulty because these inhibitors are thermodynamically more efficient and likely to be more selective than competitive inhibitors, which need to cover a large protein–protein or protein–nucleic acid interface. Interfacial inhibitors take advantage of two intrinsic characteristics of macromolecular machines. First, the binding of different molecules tends to generate deep molecular clefts (hinges) at their junction, which can be targeted by small interfacial inhibitors. Second, the macromolecular machines need to be flexible; they must undergo a range of movements around intermolecular hinges to create transient clefts that can be stabilized by the binding of interfacial inhibitors. The ‘jamming’ of the molecular machines, even if reversible, tends to be highly toxic because of the necessary orderly timing and concerted actions of biological systems.

Interestingly, interfacial inhibitors represent a large fraction of the drug market (TABLE 1). Quinolones represent one of the largest fraction (17%) of the antibiotics market, with annual sales above US\$7 billion. In the anti-HIV market, the recently approved integrase inhibitor raltegravir is now routinely used both as a first-line therapy and in patients who fail to benefit from the highly active antiretroviral therapy regimen. The market shares for raltegravir amounted to over \$1 billion in 2010 and are expected to increase in 2011. The outstanding activity and good tolerance of the two other integrase inhibitors in late-stage clinical trials, elvitegravir and dolutegravir, makes it likely that they will be approved in 2012. In addition, some non-nucleoside reverse transcriptase inhibitors may act not only as allosteric inhibitors but also as interfacial inhibitors. In the anticancer drug market, both the camptothecin and non-camptothecin drugs targeting TOP1 act as interfacial inhibitors, similarly to the anticancer drugs targeting TOP2, tubulin and target of rapamycin (TOR). Drugs that are clinically approved for treating human diseases are listed in TABLE 1.

The objective of this article was to define and elucidate the concept of interfacial inhibition by providing examples and defining its identity with respect to other modes of inhibition. As

described above, many allosteric inhibitors are interfacial inhibitors, but the converse is not always true.

There are several obvious implications of the interfacial inhibitor paradigm for drug discovery. First, the assays for discovering interfacial inhibitors need to take into account the stabilization of macromolecular complexes rather than only testing for inhibition of macromolecule binding, which — to our knowledge — is not common practice in drug discovery and high-throughput screening. Second, most interfacial inhibitors act as non-competitive or uncompetitive inhibitors, which emphasizes the importance of developing screening tests that go beyond competitive inhibition. Third, molecular modelling of interfacial inhibitors can be demanding. It requires generating structures of large complexes including at least three components, two or more biomolecules and the drug. Moreover, the dynamic nature of the biological interfaces generated by the movement of the targeted molecular machine can be difficult to model. In any case, we are hopeful that in the near future new drugs will be generated based on interfacial and allosteric principles. Such approaches are logical in the context of complex biological systems that need to utilize large-sized molecular machines with multiple components that move at a high speed relative to each other. Cell surface receptors, G protein-coupled receptors, tyrosine kinases, hormone receptors and transcription factors are all obvious candidates for allosteric and interfacial inhibition.

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Box 1 |**Characteristics of interfacial inhibitors**

- The drugs target a macromolecular system or machine consisting of two or more components. These components can be proteins or nucleic acid–protein complexes.
- The drugs bind at the interface generated by the binding of the macromolecules to each other (for example, protein–DNA interfaces or protein–protein interfaces).
- The drug-binding ‘hotspot’ is generated by the movements of the macromolecular machine as it opens sites at the interface of the individual macromolecules (for example, cleavage of DNA or bending of the microtubule filament).
- Only one drug stereoisomer tends to be active because of the tight fit between the drug and the interfacial site (most natural products follow this rule).
- Interfacial inhibitors generally bind reversibly to the macromolecular interface by hydrogen bonds, π -stacking or metal chelation (see examples described in the article).
- In addition to competitive and allosteric inhibition, interfacial inhibitors can act by selectively slowing down the targeted molecular machine (eventually bringing it to a grinding halt), which desynchronizes it from other components of the biological process (for example, slowed reversal of topoisomerase cleavage complexes leads to replication and transcriptional collisions).

Box 2 |**Drug inhibition mechanisms**

It is important to note that inhibitors can have more than one characteristic; for instance, they can be reversible, allosteric, non-competitive or allosteric interfacial.

- We refer to ligands as the substrates or cofactors that are chemically modified by enzymes. Ligands also include molecules that induce signals as they bind to a biomolecule (for instance, hormones or growth factors in the case of receptors). Ligands bind to active sites or receptor sites.
- Reversible inhibition can be overcome by increasing the ligand concentration or by dilution (for example, washout or diffusion after drug removal). Reversible inhibitors bind non-covalently via hydrogen bonds, van der Waals interactions, π -stacking, ionic bonds or chelation. Notably, some covalent bonds are reversible, such as Schiff bases or the binding of ecteinascidin 743 (also known as trabectedin) to DNA⁵⁷.
- Competitive inhibitors bind in a reversible way to the macromolecule active site by competing via steric hindrance for ligand binding; this is known as the mass action law. ATP-mimetic protein kinase inhibitors, for instance, are competitive inhibitors. This type of inhibition can be overcome at high ligand concentrations. For enzymatic reactions, the V_{\max} (velocity of the enzyme-catalysed reaction at infinite concentration of substrate) is not altered by competitive inhibitors but the K_m (the Michaelis constant) is increased. Competitive inhibitors are generally isosteric (see below). They can also be allosteric if they hinder a proximal ligand binding site.
- Non-competitive inhibitors are also reversible inhibitors. They do not bind to or alter the active sites and their binding is unaffected by ligand binding. They reduce biomolecule activity. V_{\max} decreases as the drug concentration increases. K_m remains unchanged, as non-competitive inhibitors do not affect substrate binding (for example, nifedipine inhibits cytochrome P450 2C9 via a non-competitive inhibitory mechanism).
- Uncompetitive inhibitors bind in a reversible way but alter the active site (for instance, by binding to the ligand–active site complex). This type of inhibition causes the V_{\max} to decrease and the K_m to increase.
- Mixed inhibitors bind reversibly to both the ligand-bound and ligand-free biomolecules (enzymes or receptors) but their affinities (the inhibition constant K_i) for the ligand-bound and ligand-free forms are different. They interfere with ligand binding (by increasing K_m) and hamper catalysis (by decreasing V_{\max}).
- Irreversible inhibitors usually covalently modify the biomolecule. However, in some cases they bind non-covalently with extremely slow off-rates (for example, raltegravir). Their effect cannot be overcome by increasing ligand

concentration or by dilution (washout or drug removal); it can only be overcome by the generation of intact biomolecules (enzymes or receptors).

- Allosteric inhibitors bind to a site distinct from the active site on an intrinsically flexible macromolecule target. Their long-range effect on the active site influences the binding of the natural ligand (by reducing the V_{\max} and increasing the K_m). The two models of allosteric inhibition are: the concerted two-state model put forth by Monod, Wyman and Changeux (also referred to as the MWC model)⁵⁶; and the sequential model described by Khoshland, Nemethy and Filmer⁵⁸. The latter model postulates that the binding of an allosteric inhibitor induces a conformational allosteric transition in the macromolecular complex, whereas the MWC model assumes that the inhibitor binds to a relaxed conformation of the complex that is present before drug binding.
- Orthosteric inhibitors bind to the biomolecule target (an enzyme or a receptor) on the active site where endogenous ligands would normally bind.
- Interfacial inhibitors (BOX 1) can act as allosteric inhibitors (for example, curare alkaloids or nicotine for the nicotinic acetylcholine receptor) or as orthosteric inhibitors (for example, raltegravir can be viewed as an orthosteric inhibitor as it binds to the active site of HIV integrase, formed by the HIV integrase polypeptide, the viral end of the DNA and the divalent metal cofactor).

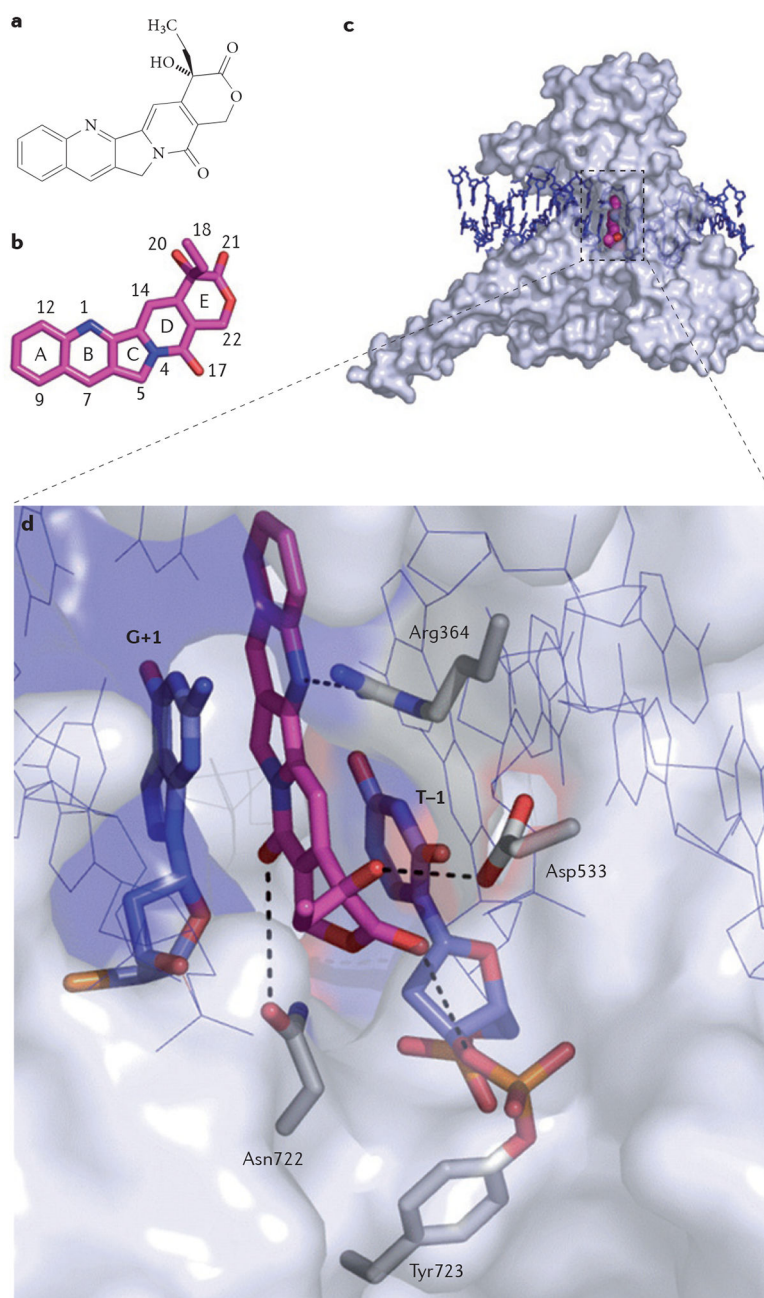


Figure 1 | Structure of a topoisomerase I cleavage complex trapped by camptothecin.

a | Chemical structure of camptothecin (CPT). **b** | Three-dimensional structure of CPT (stick representation with carbon, nitrogen and oxygen atoms coloured in magenta, blue and red, respectively; Protein Data Bank ID code 1T81⁶; numbering according to REF. 26). **c** | Topoisomerase I (TOP1) is shown in surface representation (grey) to highlight the depth of the CPT binding pocket. CPT (shown in magenta) is bound inside the catalytic core of TOP1, intercalated between the DNA base pairs (shown in blue) flanking the TOP1 cleavage site. This mechanism of action is common to other TOP1 poisons⁷. **d** | Interaction network between CPT, DNA and TOP1 in the drug–TOP1–DNA ternary complex. DNA contacts by π -stacking with the flanking DNA base pairs (G+1 and T-1) are indicated in blue. Protein

contacts are illustrated, with the following hydrogen bonds demonstrated using dashed lines: hydrogen bonding between the N1 nitrogen atom of CPT and the guanidinium group of Arg364; hydrogen bonding between the 20 hydroxyl group (see panel **a**) of CPT and the carboxylic functional group of Asp533; and hydrogen bonding between the C17 pyridone ring oxygen and the side chain nitrogen of Asn722.

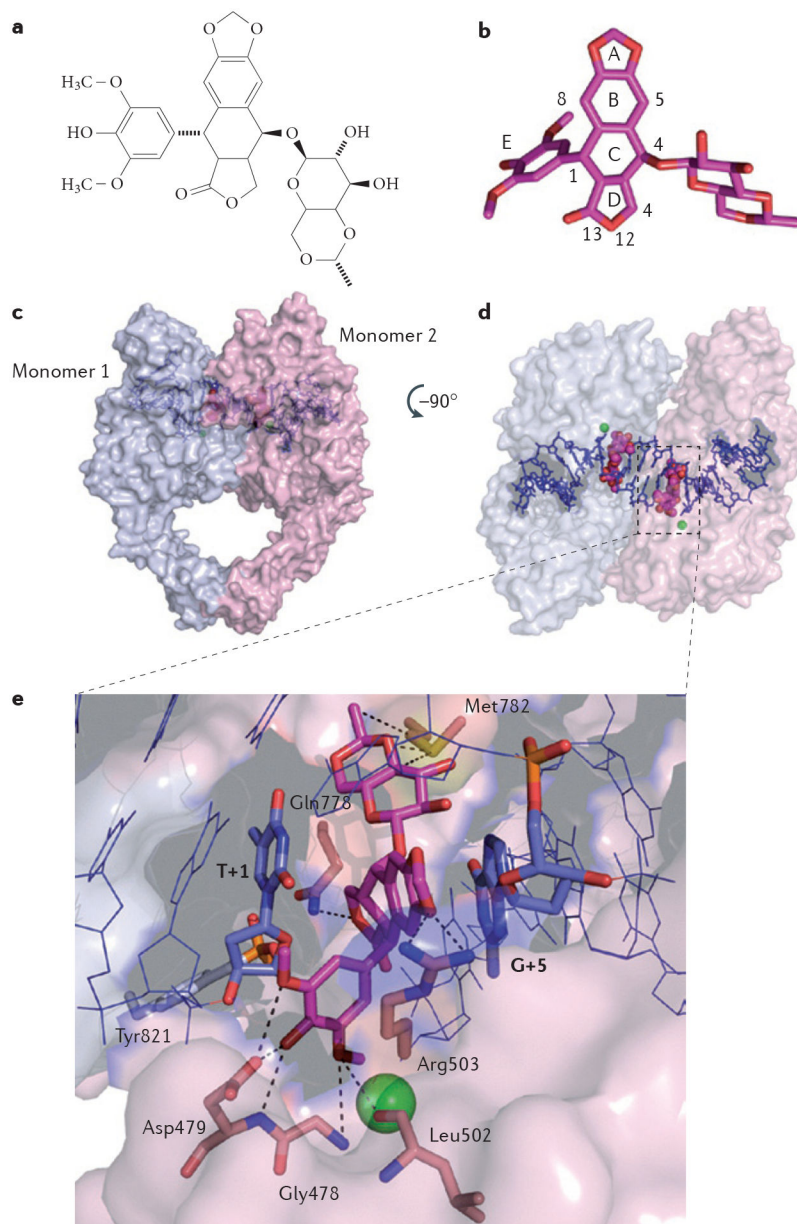


Figure 2 | Structure of a topoisomerase II β cleavage complex trapped by etoposide.
a | Chemical structure of etoposide. **b** | Three-dimensional structure of etoposide (stick representation with carbon and oxygen coloured in magenta and red, respectively; Protein Data Bank ID code 3QX3 (REF. 16); numbering according to REF. 16). **c** | Topoisomerase II β (TOP2 β) is shown in surface representation and functions as a homodimer (represented by the light blue and light pink subunits). Etoposide (shown in magenta) is bound inside the catalytic core of each TOP2 β subunit and stabilizes the cleavage complex by intercalating between the DNA base pairs (shown in blue) flanking the TOP2 β cleavage sites. **d** | The image shown is the surface representation of TOP2 β (in the left panel) after 90° rotation. **e** | Interaction network between etoposide in the drug–TOP2 β –DNA ternary complex. DNA contacts by π -stacking with the flanking DNA base pairs (T+1 on the cleaved strand and G

+5 on the uncleaved strand) are indicated in blue. The active site of TOP2 β is assembled in *trans* with the catalytic Tyr821 residue (shown in grey) from monomer 1 and the magnesium (green sphere)-chelating triad of acidic residues (not shown) from monomer 2. Protein contacts from monomer 2 with etoposide are shown, with the following hydrogen bonds and Van der Waals interactions depicted using dashed lines: hydrogen bonding and Van der Waals interactions between the E ring oxygen atoms of etoposide and Gly478, Asp479 and Leu502 residues; between the A ring oxygen of etoposide and the Arg503 residue; between oxygen 12 on the D ring of etoposide and the Gln778 residue; and between the glycosidic group of etoposide and the Met782 residue.

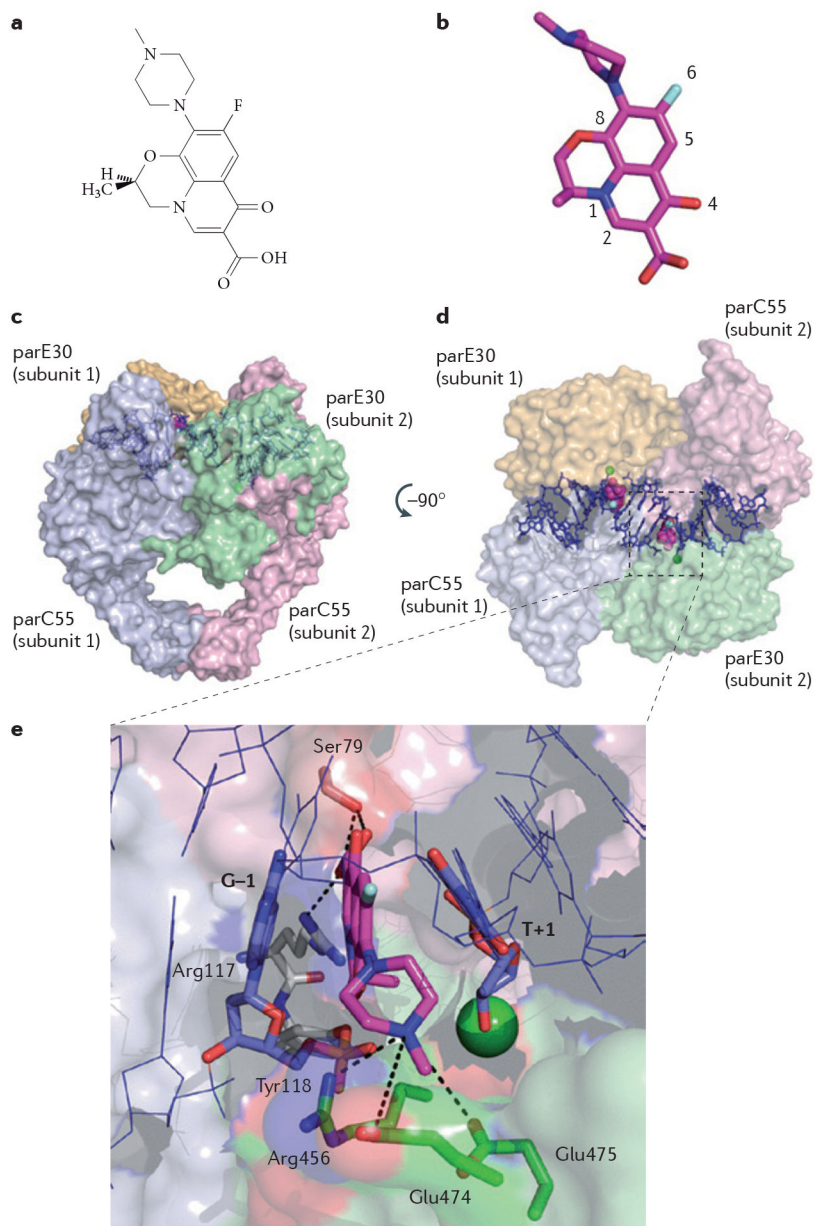


Figure 3 | Structure of a topoisomerase IV cleavage complex trapped by the fluoroquinolone antibiotic levofloxacin.

a | Chemical structure of levofloxacin. **b** | Three-dimensional structure of levofloxacin (stick representation with carbon, nitrogen, oxygen and fluorine atoms coloured in magenta, blue, red and grey, respectively; Protein Data Bank ID code 3K9F^{10,11}; numbering according to REF. 11). **c** | Bacterial topoisomerase IV functions as a tetramer of two parC55 (light blue and light pink) and two parE30 (light green and yellow) subunits⁴, shown in surface representation. Levofloxacin (shown in magenta) stabilizes the cleavage complex by intercalating between the DNA base pairs (shown in blue) flanking the topoisomerase IV cleavage sites. **d** | The image shown is the surface representation of topoisomerase IV (in the left panel) after 90° rotation. **e** | Interaction network between levofloxacin in the ternary

complex formed by the drug, topoisomerase IV and DNA. DNA contacts by π -stacking with the flanking DNA base pairs (G-1 and T+1) are indicated in blue. The active site of topoisomerase IV is assembled in *trans* with the catalytic Tyr118 residue (shown in grey) from the parC55 subunit 1 and the magnesium (green sphere)-chelating triad of acidic residues (not shown) from the parE30 subunit 2. Protein contacts are shown, with the following hydrogen bonds depicted using dashed lines: hydrogen bonding between the C3 carboxyl group of levofloxacin and the residues Ser79 and Arg117 of the parC55 subunit; and the hydrogen bonds involving the Arg456, Glu474 and Glu475 residues of the parE30 subunit, which hold together the N4 nitrogen atom of the piperazine ring at the other side of the levofloxacin molecule.

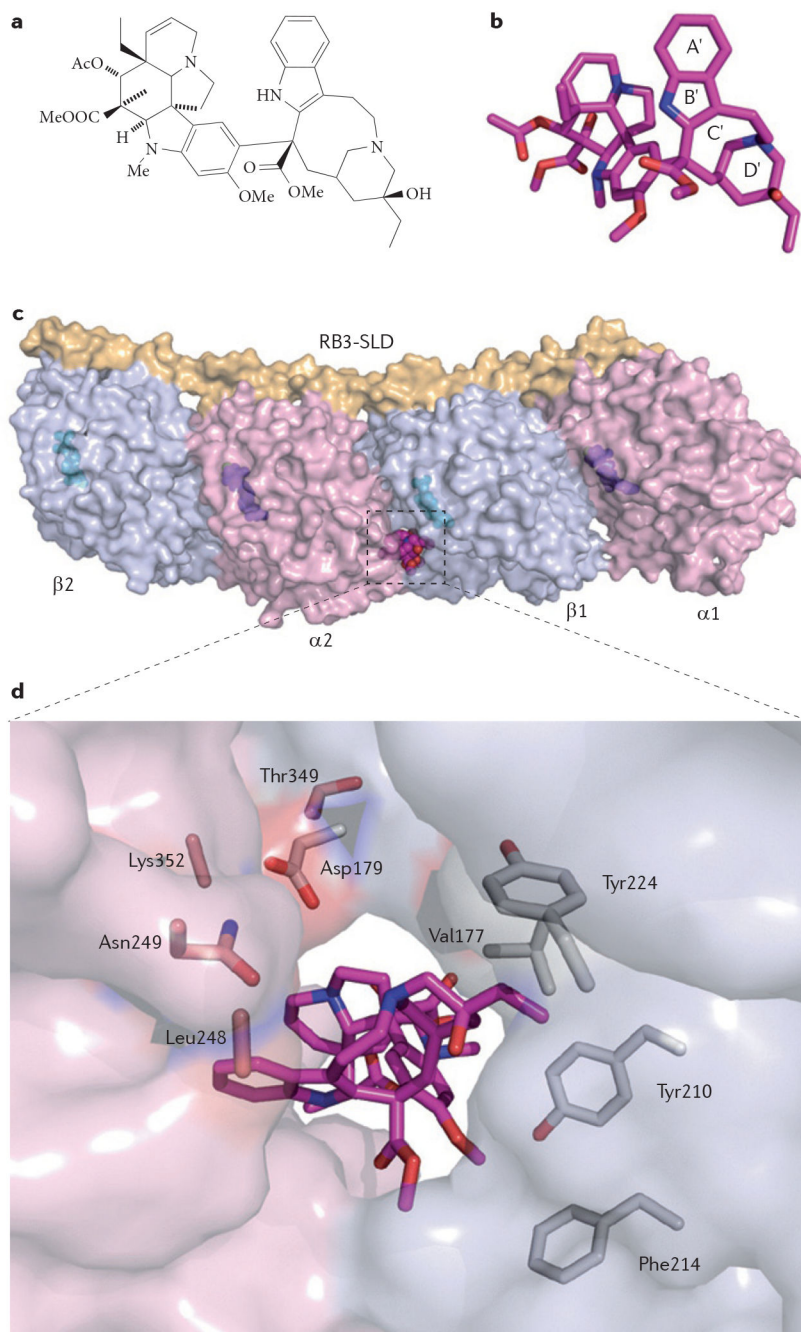


Figure 4 | Structure of vinblastine bound in a ternary complex with tubulin heterodimers.
a | Chemical structure of vinblastine. **b** | Three-dimensional structure of vinblastine (in sphere representation with carbon, nitrogen and oxygen atoms coloured in magenta, blue and red, respectively; Protein Data Bank (PDB) ID code 1Z2B⁴⁴). Heterocycles labelled A' to D' correspond to the catharanthine domain, and the other portion of the molecule corresponds to the vindoline portion of vinblastine⁴⁴. **c** | Vinblastine binds between an α -subunit ($\alpha 2$; shown in light pink) and a β -subunit ($\beta 1$; shown in light blue) of two tubulin α - β heterodimers, in complex with the *Rhodospirellula baltica* protein 3 (RB3) stathmin-like

domain (SLD), as shown in surface representation. GDP and GTP molecules are shown in sphere representation in cyan and magenta, respectively. **d** | Interaction network between vinblastine and tubulin heterodimers. Vinblastine is buried inside the complex and its orientation allows its catharanthine and vindoline moieties to interact with both tubulin heterodimers. Following vinblastine binding, the β -subunit residue Asp179 and the α -subunit residue Thr349 from each heterodimer move towards the drug and contribute to its binding. Vinblastine is stabilized on one side by the α -subunit residues Leu248, Asn249 and Lys352. On the other side, it is stabilized by the β -subunit residues Val177, Tyr210, Phe214 and Tyr224.

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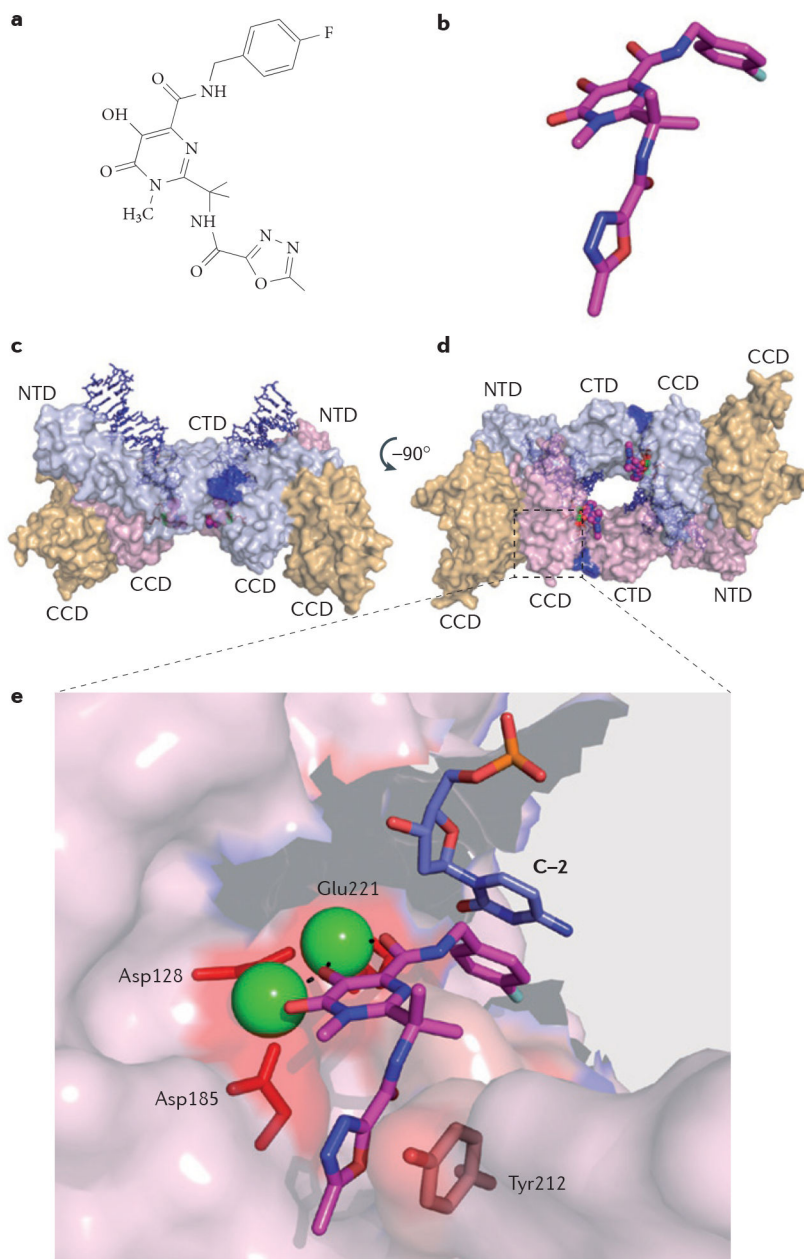


Figure 5 | Structure of the retroviral prototype foamy virus intasome in ternary complex with the strand transfer integrase inhibitor raltegravir.

a | Chemical structure of raltegravir. **b** | Stick representation of raltegravir with carbon, nitrogen and oxygen atoms coloured in magenta, blue and red, respectively; Protein Data Bank (PDB) ID code 3OYA^{15,59}. **c** | Surface representation of the retroviral intasome formed by a dimer of prototype foamy virus integrase, two viral DNA ends (shown in blue) and two molecules of raltegravir (shown in magenta). Each full-length monomer (shown in light blue and light pink) consists of three domains: the core catalytic domain (CCD), the carboxy-terminal domain (CTD) and the amino-terminal domain (NTD), and is associated in the crystal structure with a structural CCD (shown in yellow). **d** | Surface representation of the intasome rotated 90 around the horizontal axis to reveal the site of integration into the host

DNA, occupied by two molecules of raltegravir. **e** | Interaction network between raltegravir, viral DNA (stick representation; shown in blue) and integrase residues (shown in red). Raltegravir forms three types of contacts: chelation of the catalytic Mg^{2+} ions (represented by green spheres) via its carbonyl groups (the three catalytic acidic residues, Asp128, Asp185 and Glu221, that chelate the Mg^{2+} ion on the other side are represented by red sticks); DNA binding via π - π interactions with the penultimate base of the cleaved viral DNA (C-2; shown in blue); and integrase binding by π - π interactions with the Tyr212 residue.

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Table 1 |

Interfacial inhibitors identified/developed to date

Drugs	Origin	Substrates	Mechanism of action	Medical use*	Refs
α -amanitin	Black cap mushroom	RNA polymerase II	Binds to Rpb1–Rpb2 interface and prevents Pol II translocation	Not reported	60,61
Adriamycin (doxorubicin) and anthracyclines	Soil bacteria (<i>Streptomyces peuceitii</i>)	TOP2-DNA complex	Trap cleavage complexes by binding at the enzyme–DNA interface	FDA-approved	28
Anabasine	Sea worm (<i>Paranemertes peregrina</i>)	α 7 nicotinic acetylcholine receptor (homopentamer)	Agonizes α 7 nicotinic acetylcholine receptor by binding at the interface between each of the α 7 subunit monomers	Not reported	62
Anaesthetics (desflurane and propofol)	Synthetic	Pentameric ligand-gated ion channels	Act as ion channel blockers	FDA-approved	63
Benzodiazepines	Synthetic	GABA _A receptor	Increase conduction of chloride ion channels	FDA-approved	64
Brefeldin A	Fungus (<i>Eupenicillium brefeldianum</i>)	ARF•GDP–GEF complex	Stabilizes a kinetic protein–protein intermediate undergoing conformational changes	Not reported	65,66
Camptothecin	Chinese tree (<i>Camptotheca acuminata</i>)	TOP1–DNA complex	Stabilizes a kinetic intermediate of the enzymatic DNA cleavage reaction	FDA-approved	5,18
Colchicine	Crocus (<i>Colchicum autumnale</i>)	Tubulin heterodimer	Prevents tubulin polymerization by stabilizing a curved tubulin heterodimer	FDA-approved	67
Curares	Plant (<i>Strychnos toxifera</i>)	Nicotinic acetylcholine receptor	Act as nicotinic acetylcholine receptor blockers	Not reported	53
Cyclosporine	Soil fungus (<i>Tolypocladium inflatum</i>)	Cyclophilin–Calcineurin complex	Hinders access to the active site of the protein phosphatase calcineurin by artificially creating a protein–protein interface	FDA-approved	68,69
Dexrazoxane (ICRF-187)	Synthetic	TOP2–ATP complex	Stabilizes the closed TOP2 dimer	FDA-approved	41,70
Ecteinascidin 743	Sea squirt (<i>Ecteinascidia turbinata</i>)	Transcription-coupled nucleotide excision repair	Poisons the repair complex after incision of the DNA by preventing normal repair ligation	Clinically approved	71
Epothilone A	Soil-dwelling myxobacterium (<i>Sorangium cellulosum</i>)	β -tubulin	Stabilizes microtubules	FDA-approved	72
Etoposide and teniposide	Synthetic	TOP2–DNA complex	Stabilize a kinetic intermediate of the enzymatic DNA cleavage reaction	FDA-approved	16
Tacrolimus (FK506)	Soil bacteria <i>Streptomyces tsukubaensis</i>	FKBP–calcineurin complex	Hinders access to the active site of the protein phosphatase calcineurin by artificially creating a protein–protein interface	FDA-approved	73,74
Forskolin	Indian Coleus plant (<i>Coleus forskohlii</i>)	Heterodimer of cytoplasmic domains of adenylyl cyclase	Activates the enzyme by stabilizing the heterodimer catalytic site in an active conformation	Not reported	75
Fusicoccin	Plant parasite fungus (<i>Fusicoccum amygdali</i>)	14-3-3–ATPase complex	Overstabilizes a regulatory complex	Not reported	76
Fusidic acid	Fungus (<i>Fusidium coccineum</i>)	Bacterial ribosome	Stabilizes the translation elongation factor EF-G	Clinically approved	77,78
Indenisoquinolines	Synthetic	TOP1-DNA complex	Stabilizes a kinetic intermediate of the enzymatic DNA cleavage reaction	In Phase I trials	7,79

Drugs	Origin	Substrates	Mechanism of action	Medical use*	Refs
Kirromycin	Fungus (<i>Streptomyces collinus</i>)	Antibiotics, bacterial ribosomes	Stabilizes translation elongation factor Tu	Not reported	77,78
Mitoxantrone	Synthetic	TOP2-DNA complex	Stabilizes a kinetic intermediate of the enzymatic DNA cleavage reaction	FDA-approved	16 [‡]
Nalidixic acid and quinolone antibiotics	Synthetic	Gyrase-DNA or topoisomerase IV-DNA complexes	Trap cleavage complexes by binding at the interface between DNA gyrase or topoisomerase IV subunits and DNA	FDA-approved	10-12
Nicotine	Plant (<i>Nicotiana tabacum</i>)	Nicotinic acetylcholine receptor	Increases activity of nicotinic acetylcholine receptors	FDA-approved	53
Raltegravir, elvitegravir and dolutegravir	Synthetic	HIV-1 integrase-DNA complexes	Chelate metal ions in the catalytic site of the integrase-DNA complex	FDA-approved and in Phase III trials	13,15,45
Rapamycin (sirolimus)	Soil bacteria (<i>Streptomyces hygroscopicus</i>)	FKBP-mTOR complex	Promotes dimerization of FKBP12 with mTOR and inhibits mTOR	FDA-approved	80
SAHMI	Stapled peptide	Notch transcription factor complex	Antagonizes Notch signalling pathway and cell proliferation by binding to the intracellular domain of Notch 1 and the DNA-bound CSL transcription factor	Not reported	81
Simocyclinone	Bacteria (<i>Streptomyces antibioticus</i>)	DNA gyrase subunit A homotetramer	Poisons DNA gyrase by binding at the DNA gyrase A59 subunit homotetramer interface	Not reported	82
Paclitaxel and docetaxel	Trees (<i>Taxus baccata</i> , <i>Taxus brevifolia</i>)	β -tubulin	Stabilize microtubules	FDA-approved	83
Thiostrepton	Bacteria (<i>Streptomyces azureus</i> , <i>Streptomyces laurentii</i>)	Bacterial ribosome	Prevents EF-G binding by altering the interface between ribosomal protein L11 and ribosomal RNA	Veterinary use	84
Vinblastine	Madagascar periwinkle plant (<i>Catharanthus roseus</i>)	Tubulin heterodimer	Prevents tubulin polymerization by stabilizing a curved tubulin heterodimer	FDA-approved	44

ARE, ADP-ribosylation factor; CSL, DNA-binding protein complex consisting of the mammalian protein CBF1 (also known as RBP1- κ), the *Drosophila melanogaster* protein Su(H) and the *Caenorhabditis elegans* protein LAG1; FDA, US Food and Drug Administration; FKBP, FK506-binding protein; GABA-A, γ -aminobutyric acid type A; GEF, guanine nucleotide exchange factor; mTOR, mammalian target of rapamycin; Pol II, DNA polymerase II; Rpb1, RNA polymerase II subunit 1; Rpb2, RNA polymerase II subunit 2; SAHMI, stapled α -helical peptides derived from mastermind like-1 (MAML1); TOP1, topoisomerase I; TOP2, topoisomerase II.

* A large fraction (20 out of 30 drug classes listed) of these interfacial inhibitors are in clinical use (approved by the FDA, approved outside the United States, or in clinical trials).

[‡]The mechanism of action of mitoxantrone as a TOP2-DNA interfacial inhibitor was reported at the 2011 Topoisomerase conference in Taipei, Taiwan, October 2011, by the group of Nei-Li Chan, National Taiwan University.¹⁶