Note

Toward the Target: Tilorone, Quinacrine, and Pyronaridine Bind to Ebola Virus Glycoprotein

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he most recent Ebola virus (EBOV) outbreak in the L Democratic Republic of the Congo resulted in the deaths of over 2200 people according to the World Health Organization.¹ While there is an approved vaccine for prevention of EBOV disease,² we have no approved antivirals to treat patients, although there are several treatments that have reached the clinic³⁻⁶ with the most promising results for biologics only. One target of particular interest for antibody therapies' is the glycoprotein which is composed of GP1 and GP2 subunits and is involved in attachment to the cell and entry.⁸ Earlier high throughput screens had identified benzodiazepine analogues as GP1 inhibitors which act early in viral entry.^{8,9} Several structurally diverse U.S. Food and Drug Administration approved drugs have also been identified including toremifene, benztropine, bepridil, paroxetine, sertraline, and ibuprofen¹⁰ which all bind in the same site at the tunnel entrance to the glycoprotein, destabilizing it and resulting in release of GP2, thus preventing fusion between virus and endosome.9 Several other research groups have identified small molecules that inhibit glycoprotein, suggesting that a wide array of molecules may bind.^{11–15} Others have used computational approaches to perform virtual screens to identify inhibitors of GP2.¹⁶ Recent efforts to identify small molecule drugs for EBOV have included using computational methods in the form of a Bayesian machine learning (ML) approach trained with EBOV inhibitors.^{17,18} This ML model enabled a virtual screen and selection of three compounds, tilorone, quinacrine, and pyronaridine tetraphosphate¹⁹ (Figure 1). All these drugs inhibited EBOV in vitro in HeLa cells (IC₅₀'s of 1.14-1.48,

1.05–1.48, and 0.82–1.30 μ M for tilorone, quinacrine, and pyronaridine tetraphosphate, respectively, across three strains of EBOV²⁰) and *in vivo* in the mouse-adapted EBOV (ma-EBOV) efficacy model.^{21–24} Pyronaridine tetraphosphate also demonstrated significant activity in the guinea pig adapted model of EBOV infection²⁰ and is currently used as an antimalarial in combination with artesunate (Pyramax). Most recently, pyronaridine was shown to be a potent lysosomotropic agent while artesunate which is also active against EBOV *in vitro* was not.²⁵ Combining these two drugs had an additive effect on inhibiting EBOV replication *in vitro* and reduced cytotoxicity.²⁵ Pursuit of the potential target of these computationally identified drugs pointed us to docking compounds in the glycoprotein structure and experimental validation.

Tilorone had the best libdock score of 132.91 followed by pyronaridine (119.02) and quinacrine (111.56) (Figure 2). Binding energies for these docking poses were also calculated post ligand minimization but did not correlate with the libdock scores (data not shown). This may be due to the limitations of the CHARMM force fields used or other parameters, which may not have accurately estimated binding energy due to the

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Figure 1. Molecular structures of compounds evaluated in this study.



Figure 2. Docking of ligands in EBOV glycoprotein crystal structure showing lowest energy poses from the rigid docking (libdock). Cartoon representations of EBOV glycoprotein (GP1(Cyan)-GP2(Red)) with a glycan cap (purple). (A) Crystal structure of EBOV GP in complex with toremifene (PDB ID: 5JQ7; 2.69 Å). Top scoring docked poses of pyronaridine (B, libdock score: 119.016), tilorone (C, libdock score: 132.906), and quinacrine (D, libdock score: 111.558) in EBOV GP using libdock.

complex π -system interactions and may require more robust calculations such as density functional theory. As the binding site used for docking is speculated based on the crystallographic position of toremifene, we also do not rule out the distinct possibility that the site of interaction may be different for one or

all of these molecules. The main proposed electrostatic interaction between tilorone and quinacrine is a cation- π interaction with K64, with tilorone having a more idealized distance (Figure 3A and B). This differs with pyronaridine which has a π -stacking with Y517 (Figure 3C). Many studies have shown that cation $-\pi$ interactions can enhance binding energies by 2-5 kcal/mol,²⁶ and model calculations on benzene and toluene dimers (parallel and off center) suggest that these energies of interaction are in the range of 1-4 kcal/mol.^{27,2} This would suggest that these interactions may be comparable. The docking suggests that there are additional hydrophobic interactions that occur between tilorone and the EBOV GP protein. Tilorone and pyronaridine share interactions with residues L515 and M548. However, tilorone has additional substantial hydrophobic interactions with residues V66 and Y517. In addition, tilorone has nonpolar interactions with R164. Pyronaridine does have an additional interaction with I544 that is not shared with tilorone. Quinacrine shares some of the same interactions, specifically with I544 and V66 with a weaker cation $-\pi$ interaction with R64 and nonpolar interactions with E100 and R164. All of these interactions are visualized in Figure 3.

These docking predictions were then validated using microscale thermophoresis in order to calculate dissociation constants (K_d). The K_d values for pyronaridine (7.34 μ M), tilorone (0.73 μ M), and quinacrine (7.55 μ M) were lower than that for the positive control toremifene (24.83 μ M) (Figure 4) which is very similar to the literature value of 16 μ M run under similar conditions.¹⁰ Additionally, we identified clomiphene (30.74 μ M) (Figure 1), whereas favipiravir and artesunate did not bind to the EBOV glycoprotein (Figure 4). Favipiravir is a



Figure 3. Cartoon representation of the crystal structure of EBOV GP (PDB ID: 5JQ7; 2.69 Å) with side chains of the residues that form predicted interactions with the compounds tilorone (A), quinacrine (B), and pyronaridine (C) shown in a stick representation.



Figure 4. Ebola glycoprotein K_d values generated using microscale thermophoresis for test compounds.

well-known EBOV RNA polymerase inhibitor and would not be expected to bind to the glycoprotein. $^{29-32}$

We have previously demonstrated that artesunate does not inhibit lysotracker²⁵ and is likely inhibiting EBOV via a different mechanism from that of pyronaridine with which it is used in combination as the drug Pyramax for malaria.³³ The lack of inhibition of EBOV glycoprotein by artesunate provides further confirmation of this. Prior work on pyronaridine has demonstrated that it has promising *in vitro* activity against EBOV as well as excellent *in vitro* absorption, distribution, metabolism, and excretion (ADME) properties and a very long half-life across mice and humans.²² It also has statistically

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significant in vivo efficacy in mice and guinea pigs infected with EBOV.²⁰ While the mechanism to date was unclear, it has been shown to inhibit lysostracker, suggesting a lysosomotropic effect.²⁵ When this is coupled with the potential to bind to glycoprotein, this dual mechanism will likely prevent EBOV entry. Similarly, quinacrine is a small-molecule, orally bioavailable drug which has also been used clinically as an antimalarial. This also has very favorable ADME properties (apart from potent CYP2D6 inhibition), and it demonstrated in vivo activity in mice infected with EBOV.²¹ Tilorone is structurally different and is used in eastern Europe as an antiviral. It has excellent in vitro ADME properties, and it was recently demonstrated to have potent in vitro inhibition of EBOV as well as efficacy in mice infected with EBOV.²³ Interestingly, this molecule did not demonstrate efficacy in the guinea pig EBOV model likely due to the significant species differences in metabolism when compared to the mouse model.²⁰ Tilorone has also demonstrated in vitro inhibition of MERS,³⁴ is under evaluation as a potential treatment for other coronaviruses such as SARS-CoV-2,²⁴ and has shown a low μ M IC₅₀.³⁵ This current work now provides further detail as to how these three molecules target EBOV glycoprotein as well as being lysosomotropic, enabling their previously reported blocking of viral entry. This information may aid in structure guided modification of these compounds and future X-ray crystallography. Such mechanistic insights may also aid drug discovery for other viruses (e.g., SARS-CoV-2). To date tilorone would appear to be the highest affinity compound for EBOV glycoprotein compared to the previously reported toremifene.^{10,15}

In conclusion, the accumulated *in vitro* and *in vivo* data gathered for tilorone, quinacrine, and pyronaridine points to them sharing a common target or mechanism for the inhibition of EBOV.^{20-23,25,36} It would also appear all of these molecules block viral entry, are lysosomotropic, and are now identified to bind to the glycoprotein. Shedding further light on how these molecules work *in vitro* may provide further justification for clinically repurposing these compounds in future outbreaks.

EXPERIMENTAL PROCEDURES

Materials. Pyronaridine tetraphosphate [4-[(7-chloro-2-methoxybenzo[b][1,5]naphthyridin-10-yl)amino]-2,6-bis(1-pyrrolidinylmethyl)phenol phosphate <math>(1:4)]¹⁹ was purchased from BOC Sciences (Shirley, NY). Tilorone and quinacrine were purchased from Cayman Chemicals (Ann Arbor, MI). Toremifene and clomiphene were purchased from MedChemExpress (Monmouth Junction, NJ), and favipiravir was from TRC Canada (North York, ON, Canada). Zaire ebolavirus disulfide-linked glycoprotein heterodimer (GP1-GP2) was purchased from Novus Biologicals (Centennial, CO). According to the manufacturer, EBOV GP protein is purified from CHO-derived viral expression with previous internal verification of significant glycosylation.

Docking. Three EBOV inhibitors, tilorone, pyronaridine, and quinacrine, were docked into the EBOV glycoprotein structure (PDB: 5JQ7) at the same site as previously described for toremifene¹⁰ using Discovery Studio LibDock. Binding energies were calculated following a ligand minimization (rigid protein).

Microscale Thermophoresis. An amount of 200 μ g of lyophilized protein was resuspended in RED-NHS second Generation labeling buffer (NanoTemper; Cambridge, MA). This was followed by the labeling of the primary amines using the RED-NHS dye according to the manufacturer's protocol. Labeled protein was buffer exchanged into 10 mM MES, pH 5.0, 150 mM NaCl, 170 mM sodium malonate at pH 5.2 (MST buffer), which is similar to a buffer previously shown to be appropriate for EBOV GP,¹⁰ and then diluted to a final concentration of 1 μ M. For each experimental compound, 16 independent stocks were

made in DMSO using 2-fold serial dilution (10 mM initial concentration). The MST buffer used for a final dilution prior to MST was supplemented with 0.05% Tween 20 and 10 mM BME. The protein was diluted to 2.5 nM in the supplemented MST buffer, and 19.5 μ L of this was combined with 0.5 μ L of the compound stock and then mixed thoroughly. This resulted in 2-fold serial dilution testing series with the highest and lowest concentrations of 250 μ M and 7.629 nM, respectively, with a consistent final DMSO concentration of 2.5%. These reactions were incubated for 20–30 min prior to transferring to standard Monolith NT.115 capillaries. Experiments were run at 20% excitation and high MST power at 23.0 °C on a Monolith NT.115Pico instrument (NanoTemper). Favipiravir and toremifene were also run as the negative and positive controls, respectively. Each experimental compound was run in quadruplicate.

The data were acquired with MO.Control 1.6.1 (NanoTemper Technologies). Recorded data were analyzed with MO.Affinity Analysis 2.3 (NanoTemper Technologies). The dissociation constant K_d quantifies the equilibrium of the reaction of the labeled molecule A (concentration c_A) with its target T (concentration c_T) to form the complex AT (concentration c_{AT}) and is defined by the law of mass action as $K_{\rm d} = \frac{c_{\rm A}c_{\rm T}}{C_{\rm AT}}$ where all concentrations are "free" concentrations. During the titration experiments, the concentration of the labeled molecule A is kept constant and the concentration of added target T is increased. These concentrations are known and can be used to calculate the dissociation constant. The free concentration of the labeled molecule A is the added concentration minus the concentration of formed complex AT. The K_d is calculated as $K_d = \frac{(c_A^0 - c_{AT})(c_b^0 - c_{AT})}{c_{AT}}$. The c_{AT} fraction of bound molecules x can be derived from F_{norm} , where $F_{norm}(A)$ is the normalized fluorescence of only unbound labeled molecules A and $F_{\text{norm}}(AT)$ is the normalized fluorescence of complexes AT of labeled as shown by the equation: $x = \frac{F_{norm}(c_1^0) - F_{norm}(A)}{F_{norm}(AT) - F_{norm}(A)}.$ The MST traces that showed aggregation or outliers were removed from the data sets prior to K_d determination.

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T.R.L. performed all experimental work and data analysis and helped draft the manuscript. S.E. designed the study and wrote the manuscript.

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Notes

The authors declare the following competing financial interest(s): S.E. is CEO of Collaborations Pharmaceuticals, Inc. T.R.L. is an employee at Collaborations Pharmaceuticals, Inc. Collaborations Pharmaceuticals, Inc. has obtained FDA orphan drug designations for pyronaridine, tilorone, and quinacrine for use against Ebola.

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