

# Synthesis and Efficacy of the *N*-carbamoyl-methionine Copper on the Growth Performance, Tissue Mineralization, Immunity, and Enzymatic Antioxidant Capacity of Nile tilapia (*Oreochromis niloticus*)

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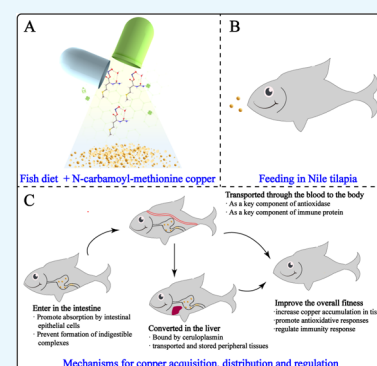


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**ABSTRACT:** Immunogenic, methionine copper-induced response had proven to be precedent in providing resistance against certain diseases in fish. This study allocates the fitness strategy for *Oreochromis niloticus* by introducing and incorporating the well-designed, stabilized, and biocompatible *N*-carbamoyl-methionine copper (NCM-Cu) as a Cu potent source in diet that enhances the bioavailability and fitness. The synchronized NCM-Cu complex was characterized by directing ultraviolet and visible spectrophotometry (UV-vis), Fourier-transform infrared (FTIR), X-ray diffractometry (XRD), thermogravimetric analysis (TGA), and single-crystal X-ray diffraction. Results revealed blue columnar crystalline, NCM-Cu complex with an empirical formula as  $C_{12}H_{30}CuN_4O_{10}S_2$ . Anonymously, the overall growth performance of the fish remained unaltered with NCM-Cu adjunct feed. NCM-Cu significantly raised the Cu accumulation in the fish muscles, liver, gill, and intestine in contrast to the basic Cu-rich feed. The serum antioxidant enzyme activity elevated up to (ceruloplasmin: 19.38 U/L) and the lowest liver malondialdehyde (MDA) content (8.81 nmol/mg prot.) and triglyceride content (0.39 nmol/g prot.) were observed in the NCM-Cu group as compared to the basic Cu and  $CuSO_4$  groups, suggesting that NCM-Cu promoted antioxidative responses and alleviated lipid peroxidation of *O. niloticus*. Overweening, the synthesized complex, NCM-Cu significantly regulated the expression levels of lysozyme, immunoglobulin M, complement 4, and complement 3 up to 10.93 U/mL, 0.72, 0.77, and 1.18 mg/mL in serum, respectively. Thus, such endorsed results reveal the preeminence of NCM-Cu-supplemented diet for the fitness in *O. niloticus*.



## INTRODUCTION

Methionine (Met) is an essential sulfur amino acid and is generally considered as the primary limiting amino acid in aquafeeds. Dietary Met deficiency would affect the uptake of essential nutrients by alteration of gene expression of amino acid transporters, which results in the health problem of animals.<sup>1,2</sup> In biological systems, Met is involved in five metabolic pathways, including transmethylation, transculturation, remethylation, aminopropylation, and salvage.<sup>3</sup> However, Met was found to be sensitive to oxidative damage caused by a variety of oxidants, which can result in loss of structural integrity and disruption of protein functions.<sup>4</sup> To minimize this shortcoming, a potential strategy for stabilizing Met against oxidative damage is to functionalize it by adding a non-oxidizable group such as a *N*-carbamoyl group.<sup>5</sup> Indeed, *N*-carbamoyl amino acids are considered as very stable compounds, the hydrolysis of which requires drastic basic conditions or the use of enzymes.<sup>6</sup> Since most fish digest feed through the gastrointestinal tract, we only discuss the behavior of an *N*-carbamoyl-methionine copper (NCM-Cu) sample within two differential pH environments (i.e., acid and

neutral).<sup>7</sup> It has been reported that a chelated metal complex could be dissolved by stomach acid (acid environment), but its dissolution rate is much lower than that of inorganic salts.<sup>8,9</sup> Additionally, some previous in vitro bioavailability experiments displayed that the chelated metal complex was slowly dissolved in the succus entericus (neutral environment) and was kept relatively constant.<sup>10,11</sup> Therefore, the prepared NCM-Cu sample theoretically has good gastrointestinal stability as compared to inorganic salt. Interestingly, *N*-carbamoyl-methionine (NCM) can be reconverted to Met in the body. In prokaryotes, NCM is also the starting amino acid for biological proteins.<sup>12</sup>

Cu is an essential trace mineral element for aquatic animals, which is the primary component of key enzymes in biological

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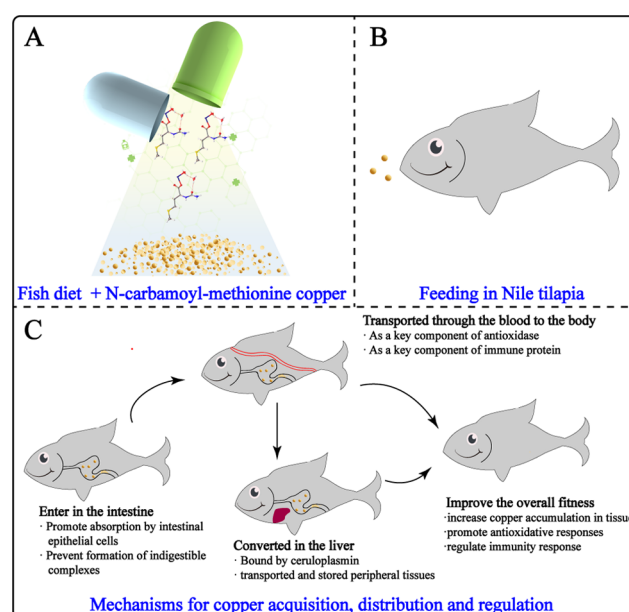
processes such as lysyl oxidase, cytochrome *c* oxidase, ferroxidase, and tyrosinase.<sup>13,14</sup> Adequate dietary Cu levels had a positive effect on growth performance, antioxidant status, and immune response.<sup>15,16</sup> Therefore, Cu should be supplemented to diets for maintaining the normal growth in fish. Whereas, the use of Cu supplements requires careful consideration of its levels in feed diet and may lead to negative impacts on the environment. On the one hand, signs of impaired growth, increased mortality, oxidative injury, and immune barrier malfunctions may occur in fish fed on Cu deficiency diet.<sup>17,18</sup> On the other hand, unnecessarily high additions of Cu in fish can cause toxicity in the tissue structure, leading to changes in osmotic response and acid–base regulatory system.<sup>19,20</sup> In an aquaculture system, dietary intake is the major way of Cu acquisition for fish.<sup>21,22</sup> Accordingly, it is necessary for aquaculture producers to give priority to Cu sources with higher bioavailability in the intensification of the breeding system.

In biological systems, the presence of protons ( $H^+$ ) could facilitate the dissolution of dietary Cu due to the interactions of endogenous inhibitors (e.g., phytate and tricalcium phosphate) in the feed; dissolved  $Cu^{2+}$  reprecipitates to form insoluble or indigestible complexes and reduce the bioavailability of dietary Cu.<sup>15</sup> Over presence of  $HCO_3^-$  (>50 mmol/L) limits the bioavailability of  $Fe^{2+}$  (precipitation of  $Fe(HCO_3)_2$ ), which may also be applicable for  $Cu^{2+}$ . Interestingly, studies found that the chelated trace elements may protect metal ions against antinutritional factors present in practical diets.<sup>23,24</sup> Moreover, the bioavailability of organic chelated Cu is higher than Cu inorganic salts in fish such as *Oncorhynchus mykiss* and *Epinephelus malabaricus*.<sup>25</sup> It has been reported that chelated Cu as an organic Cu supplement had great benefits for animal growth performance, antioxidant status, and immunity.<sup>26</sup> Therefore, using *N*-carbamoyl-methionine copper (NCM-Cu) as the Cu source may be more beneficial to meet the nutritional needs of aquatic animals.

*Nile tilapia* (*Oreochromis niloticus* L.) is the most important aquaculture products economically, being farmed in more than 135 countries, with a global creation around 5.9 million tones.<sup>27</sup> In China, Nile tilapia aquaculture production has increased significantly to provide a valuable source of protein for the increasing market demand. In this study, the NCM-Cu complex was fabricated and a new organic Cu source was used in the feed to improve the overall fitness of *O. niloticus*. The molecular structure and interactions of the NCM-Cu complex were determined by a variety of physicochemical methods including ultraviolet and visible spectrophotometry (UV–vis), Fourier-transform infrared (FTIR), the X-ray diffractometry (XRD), thermogravimetric analysis (TGA), and single-crystal X-ray diffraction. Furthermore, the efficacy of NCM-Cu on the growth performance, tissue mineralization, antioxidant status, and immune response of a freshwater fish, *O. niloticus* was evaluated.

## RESULTS AND DISCUSSION

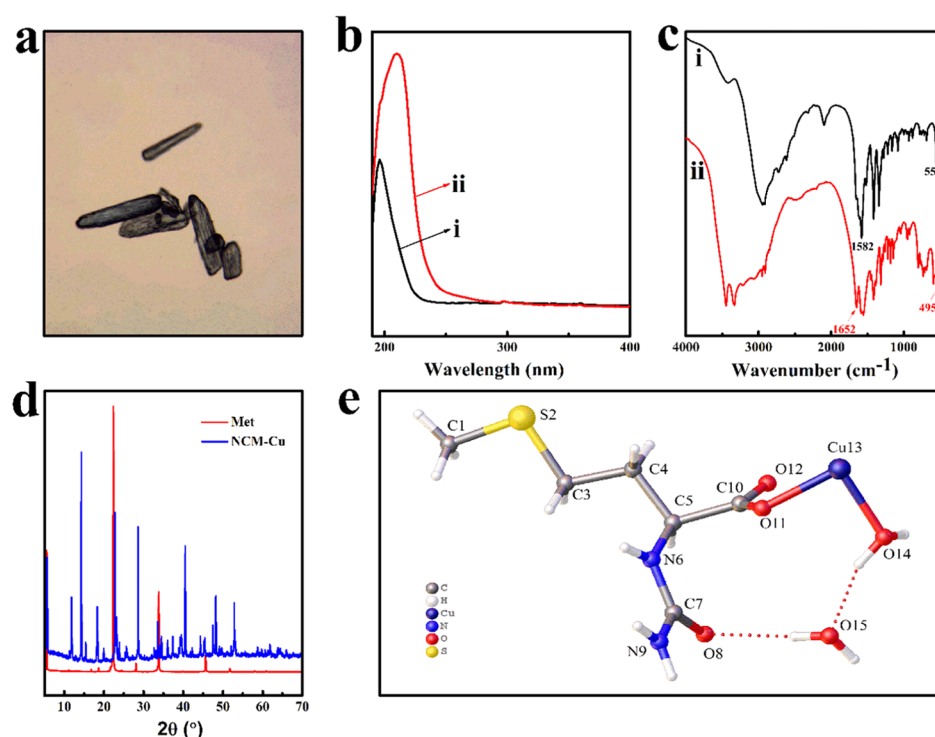
Present work was focused on the synthesis and efficacy of *N*-carbamoyl-methionine copper on the growth performance, tissue mineralization, immunity, and enzymatic antioxidant capacity of Nile tilapia (*O. niloticus*). The overall illustration of intake of *N*-carbamoyl-methionine copper as diet and their biological effects on Nile tilapia (*O. niloticus*) are shown in Figure 1.



**Figure 1.** Illustration of the synthesis of *N*-carbamoyl-methionine copper (a) intake as diet (b) and their biological effects on Nile tilapia (*O. niloticus*) (c).

Both methionine (Met) and Cu element are essential substances in maintaining the fish's normal life activities.<sup>1,13</sup> Besides, it has been reported that *N*-carbamoyl-methionine (NCM) can be reconverted to Met in the body.<sup>12</sup> In prokaryotes, NCM is also the starting amino acid for biological proteins. Thus, the NCM-Cu supplements are nontoxic and safe for fish. However, studies have shown that the excessive accumulation of Cu in fish can cause changes in the tissue structure, induce antioxidation, damage and stress response in tissues and organs, lead to changes in osmotic pressure and the acid–base regulatory system, induce immunotoxicity, hinder the cell circulation of immune organs, promote apoptosis, and affect the growth, development, and reproduction of fish.<sup>19</sup> Therefore, the use of NCM-Cu supplements requires careful consideration of Cu levels in feed diet.

The NCM-Cu complex was successfully synthesized, which presents a stable crystal form for the intake of fish. An optical microscope image (Figure 2a) indicated that the solid product of the NCM-Cu complex had a blue columnar morphology. For the sake of confirming the successful complexation of the NCM-Cu complex, UV–vis spectroscopy and FT-IR spectroscopy were applied. For the UV–vis spectra (Figure 2b), a bathochromic shift of the NCM-Cu complex was obviously observed, which may be due to the extension of the conjugated system with complexation.<sup>28</sup> For the FT-IR spectra (Figure 2c), two characteristic peaks of Met were detected at 1582 and 553  $cm^{-1}$  corresponding to  $-NH_2$  bending vibration and to C–S–C stretching vibration, respectively.<sup>29,30</sup> In contrast, the appearance of new absorption peak at 1652 and 495  $cm^{-1}$  indicated the existence of  $-OH$  and Cu–O in the NCM-Cu complex, respectively.<sup>31,32</sup> The existence of these characteristic peak confirmed that the NCM-Cu complex was formed. In addition, X-ray diffraction (XRD) was further employed to characterize the phase composition of the NCM-Cu complex (Figure 2d). Compared to Met, the appearance of a new and special diffraction pattern confirmed that a new crystal structure of the NCM-Cu complex was formed.<sup>33</sup> Besides, a series of intense and sharp peaks of XRD patterns displayed



**Figure 2.** Structural characterization. (a) Optical microscope image of the NCM-Cu complex. (b) UV-vis spectra: (i) Met and (ii) NCM-Cu. (c) FTIR spectra: (i) Met and (ii) NCM-Cu. (d) XRD patterns. (e) Molecular structure of NCM-Cu.

that the NCM-Cu complex was a crystalline compound. To further corroborate this notion, the chemical states of the NCM-Cu complex were investigated by the single-crystal X-ray diffraction. The results (Table 1) revealed that the NCM-Cu

**Table 1. Crystallographic Data and Structure Refinement of *N*-carbamoyl-methionine copper**

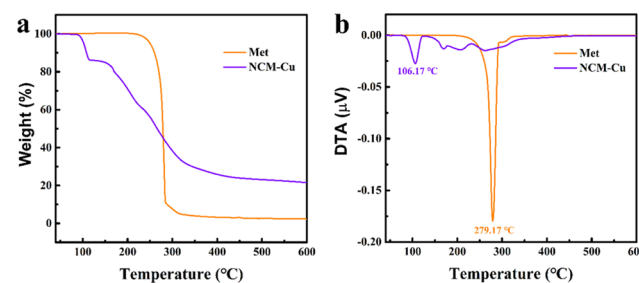
parameter	values
empirical formula	$C_{12}H_{30}CuN_4O_{10}S_2$
formula weight (g/mol)	518.07
$T$ (°C)	149.99
crystal system	monoclinic
space group	$P21/c$
$a$ (Å)	15.8914(2)
$b$ (Å)	4.65130(10)
$c$ (Å)	15.2181(3)
volume (Å <sup>3</sup> )	1094.89(4)
$Z$	2
final $R$ indices	0.0476/0.1329
$R$ indices (all data)	0.0491/0.1365

complex was crystallized in the monoclinic system and the  $P21/c$  space group. Moreover, the empirical formula of NCM-Cu was identified as  $C_{12}H_{30}CuN_4O_{10}S_2$ .

As depicted in Figure 2e, the aminocarbonyl ( $O=C-NH_2$ ) group was coordinated with secondary amine ( $-NH$ ) in the NCM-Cu complex. Simultaneously, the Cu atom in the NCM-Cu complex was coordinated with the carboxyl groups of *N*-carbamoyl methionine ligands via a Cu–O bond. Besides, two crystallization water was strongly trapped in the crystal lattice. These results were consistent with other analysis data. Noteworthy, the crystallization behavior may be affected by the central atom, which directly affected the packing of the structure.<sup>9,34</sup> The content of Cu of the NCM-Cu complex was

14.28% (w/w) as measured by inductively coupled plasma spectrometer (ICP).

TGA–DTA analysis is useful for evaluating the composition, thermal stability, and degradation behavior of metal complexes. In the preparation and processing of feed, heating at high temperatures may create a large change in the chemical properties of feed additives.<sup>35</sup> The change of chemical properties of feed additives can influence the stability and physiological efficacy of NCM-Cu samples.<sup>36</sup> The poor thermal stability of feed additives has an adverse effect on its development and zoological research.<sup>37</sup> Thus, it is important to evaluate the NCM-Cu samples prepared in this work. In another experiment, the thermal stability of Met and NCM-Cu over a range of 40–600 °C was tested (Figure 3). The TG



**Figure 3.** Thermal stability analysis. (a) Thermogravimetry (TG) curves; (b) derivative thermogravimetry (DTA) curves.

curve of Met showed mass loss within the temperature range of 214.01–340.47 °C with DTG peaks at 279.17 °C. In contrast, the first mass loss of NCM-Cu (8.0%) was related to the loss of crystallization water.<sup>38</sup> Besides, the second stage mass loss of NCM-Cu at 134.56–427.10 °C was 61.2%. The difference in the thermal stability of the NCM-Cu complex might be caused

**Table 2.** Growth Performance and Feed Utilization of *O. niloticus* by Feeding Experimental Diet with Various Forms of Cu for 60 Days<sup>a</sup>

experimental diets	WG (%)	FE (%)	CF (g/cm <sup>3</sup> )	VSI (%)	HSI (%)	Survival (%)
Cu-basic diet	234.1 ± 50.5	85.8 ± 18.5	2.05 ± 0.12	8.46 ± 0.78	1.94 ± 0.41	100
CuSO <sub>4</sub> diet	237.9 ± 37.1	85.5 ± 13.3	1.96 ± 0.19	8.87 ± 0.90	2.11 ± 0.41	100
NCM-Cu diet	225.0 ± 34.8	82.2 ± 12.6	2.01 ± 0.16	8.65 ± 0.95	2.16 ± 0.53	100

<sup>a</sup>WG: weight gain; FE: feeding efficiency; CF: condition factor; VSI: viscerosomatic index; HSI: hepatosomatic index.

by the insertion of metal, leading to reduction in the number of hydrogen bonding.<sup>32</sup> Interestingly, the residual amount of Met was smaller than that of the NCM-Cu complex, further proving the occurrence of complexation between *N*-carbamoyl methionine and Cu ions.

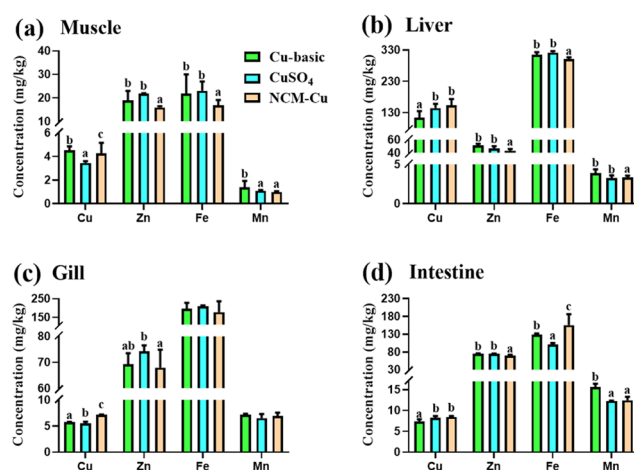
In summary, the NCM-Cu complex was successfully synthesized, with the formula NCM-Cu as C<sub>12</sub>H<sub>30</sub>CuN<sub>4</sub>O<sub>10</sub>S<sub>2</sub>. Multiple structural analysis indicated that the Cu atom in the NCM-Cu complex was coordinated with carboxyl groups of *N*-carbamoyl methionine ligands via a Cu–O bond. Besides, the original amino group in Met was substituted with the aminocarbonyl group such that the NCM-Cu compound did not undergo Strecker degradation.<sup>39</sup> This stable structure provided the possible idea that the NCM-Cu complex may have better bioavailability and biocompatibility in the body.

CCDC 1945764 contains the supplementary crystallographic data for this paper. These data can be obtained basic of charge from The Cambridge Crystallographic Data Centre via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif).

Results of the growth, feed utilization, and morphological parameters of *O. niloticus* by feeding the experimental diet with various forms of Cu are displayed in Table 2. According to one-way ANOVA analysis, there were no significant differences in weight gain (WG), feeding efficiency (FE), condition factor (CF), viscerosomatic index (VSI), hepatosomatic index (HSI), and survival of *O. niloticus* among all treatments ( $p > 0.05$ ). Based on the results, the growth performance and feed utilization of *O. niloticus* fed on three experimental diet were comparable. However, the basic Cu diet did not affect the fish feeding and growth performance of *O. niloticus*. Similar experimental results were found in *O. niloticus* and *Cyprinus carpio*.<sup>40,41</sup>

Besides, it had been reported that varying concentrations of Cu did not significantly affect the weight gain and feed efficiency of *Salmo salar*.<sup>42</sup> Some authors have suggested that the difference in fish growth performance was not only affected by chemical forms of Cu but also related to species, age of fish, and dietary factors. Therefore, the results indicated that *O. niloticus* was less susceptible to dietary Cu than other fish species.

Concentrations of four metals in tissues of the fish fed on three experiment diets for 60 days are shown in Figure 4 (the values are shown in Tables S1–S4). Obviously, diets supplemented with NCM-Cu increased the Cu absorption and deposition efficiency in *tilapia* tissues. Specifically, Cu concentrations in the muscle (4.99 mg/kg), liver (154.3 mg/kg), gill (7.18 mg/kg), and intestine (8.48 mg/kg) in the NCM-Cu group were significantly higher than those of the Cu-basic group. Simultaneously, a significant increase in the Cu concentration in the muscle and gill was also observed in fish fed on the NCM-Cu diet when compared to that of fish fed on a CuSO<sub>4</sub> diet, as shown in Figure 4a,c. The highest Cu concentration was observed in the liver, suggesting that the liver was the target for Cu accumulation. It has been reported



**Figure 4.** Effects of different Cu sources as the experimental diets on metal elemental concentration changes in tissues of *O. niloticus*: (a) muscle, (b) liver, (c) gill, and (d) intestine. Bars sharing the different alphabets are significantly different ( $p < 0.05$ ).

that organic forms of Cu were more effective than CuSO<sub>4</sub> in some aquatic animals such as *O. mykiss* and *Litopenaeus vannamei*.<sup>43,44</sup> In the gastrointestinal tract, Cu ions were easily precipitated by the interactions of the endogenous inhibitors. However, due to its stable structure, the possibility for chelated Cu to form insoluble or indigestible complexes was reduced. Some authors suggested that chelating minerals can be absorbed in an intact form by amino acid transporters. Therefore, tissue mineralization data also showed that the NCM-Cu complex had a higher bioavailability in *O. niloticus* than the inorganic form.

On the other hand, NCM-Cu diet might not be good for Zn accumulation in *tilapia* tissues. For example, Zn concentration in the muscle, liver, and intestine in the NCM-Cu group was significantly lower than the Cu-basic and CuSO<sub>4</sub> groups, as shown in Figure 4a,b,d. Due to similarities in the physicochemical attributes of Cu and other cations, there was an interaction between these minerals, which affected the absorption of the gastrointestinal tract.<sup>45</sup> Therefore, the type of the mineral source and the concentration also affected the minimal dietary inclusion levels and the absorption of other minerals. Evidence of competition between Cu and Zn was also found in *Larimichthys croceus*.<sup>46</sup> In addition, the Fe content in the fish muscle and liver in the NCM-Cu group was significantly lower than those in the Cu-basic and CuSO<sub>4</sub> groups, as shown in Figure 4a,b, but the opposite result was detected in the intestine, as shown in Figure 4d. Besides, Mn accumulation in *tilapia* tissues was significantly affected among three treatments. Mn content in the muscle, liver, and intestine in the NCM-Cu group was significantly lower than in the Cu-basic group. Interestingly, the gill appeared to be least affected by different sources of Cu supply. Fe and Mn contents in *tilapia* gill were comparable when fed on three different diets.

**Table 3. Content of Total Superoxide Dismutase (T-SOD), Cu-Zn Superoxide Dismutase (Cu-Zn SOD), Malondialdehyde (MDA), Triglyceride (TG), Total Cholesterol (T-CHO), and Ceruloplasmin (CP) in the Serum of *O. niloticus* Fed on Experimental Diets with Different Sources of Cu for 60 Days<sup>a</sup>**

experimental diets	T-SOD (U/mg prot.)	Cu-Zn SOD (U/mg prot.)	CP (U/L)	MDA (nmol/mg prot.)	TG (mmol/g prot.)	T-CHO (mmol/g prot.)
Cu-basic diet	35.48 ± 1.06	33.77 ± 2.97 <sup>a</sup>	16.16 ± 1.49 <sup>a</sup>	11.47 ± 1.14 <sup>b</sup>	0.54 ± 0.06 <sup>b</sup>	0.39 ± 0.07 <sup>b</sup>
CuSO <sub>4</sub> diet	35.12 ± 0.79	36.06 ± 0.20 <sup>b</sup>	16.26 ± 1.58 <sup>a</sup>	10.54 ± 1.17 <sup>b</sup>	0.47 ± 0.07 <sup>b</sup>	0.27 ± 0.03 <sup>a</sup>
NCM-Cu diet	36.07 ± 0.43	35.61 ± 0.91 <sup>ab</sup>	19.38 ± 0.50 <sup>b</sup>	8.81 ± 1.38 <sup>a</sup>	0.39 ± 0.06 <sup>a</sup>	0.26 ± 0.08 <sup>a</sup>

<sup>a</sup>Data in the same column with different superscript letters are significantly different ( $p < 0.05$ ).

**Table 4. Activity of Lysozyme (LZM), Immunoglobulin M (IgM), Complement 4 (C4), and Complement 3 (C3) Values in the Serum of *O. niloticus*, Fed on Experimental Diet Supplemented with Various Forms of Cu<sup>a</sup>**

experimental diets	LZM (U/mL)	IgM (mg/mL)	C4 (mg/mL)	C3 (mg/mL)
Cu-basic diet	10.48 ± 0.70 <sup>ab</sup>	0.25 ± 0.10 <sup>a</sup>	0.50 ± 0.12 <sup>a</sup>	0.84 ± 0.16 <sup>a</sup>
CuSO <sub>4</sub> diet	9.91 ± 0.46 <sup>a</sup>	0.32 ± 0.05 <sup>a</sup>	0.75 ± 0.17 <sup>b</sup>	0.94 ± 0.27 <sup>ab</sup>
NCM-Cu diet	10.93 ± 0.38 <sup>b</sup>	0.72 ± 0.11 <sup>b</sup>	0.77 ± 0.10 <sup>b</sup>	1.18 ± 0.19 <sup>b</sup>

<sup>a</sup>Data in the same column with different superscript letters are significantly different ( $p < 0.05$ ).

The liver antioxidant parameters and serum ceruloplasmin activity are shown in Table 3. The total superoxide dismutase (T-SOD) and Cu-Zn SOD activities in the NCM-Cu group were comparable when compared to the Cu-basic or the CuSO<sub>4</sub> group. However, biological antioxidant potential (CP) recorded the highest value (19.38 U/L) in the NCM-Cu group as compared to the Cu-basic (16.16 U/L) and the CuSO<sub>4</sub> groups (16.26 U/L) ( $p < 0.05$ ). Simultaneously, the content of malondialdehyde (MDA) (8.81 nmol/mg prot.) and TG (0.39 mmol/g prot.) in *tilapia* liver in the NCM-Cu group was significantly lower than those in the Cu-basic and CuSO<sub>4</sub> groups. Moreover, fish fed on diets supplemented with NCM-Cu had significantly lower T-CHO content (0.26 mmol/g prot.) when compared with the Cu-basic group. Similarly, previous results displayed that methionine Cu had higher antioxidant capacity than fish fed on the same Cu content in the CuSO<sub>4</sub> diet. It has been reported that adequate dietary Cu levels improved the nonspecific immune responses or antioxidation abilities of *Acipenser gueldenstaedtii*.<sup>47</sup> Noteworthy, a Cu ion is an active center of Cu-Zn SOD and CP and to its catalytic action. It has been reported that an increase in the hepatic Cu pool resulted in a sustained increase in the Cu-Zn SOD and CP concentrations in the liver. Thus, the activity of Cu-Zn SOD and CP enzymes is not determined by enzyme protein but also determined by the Cu<sup>2+</sup> content. On the contrary, Cu deficiency would affect the antioxidant defense and lipid metabolism. Therefore, our results suggested that NCM-Cu as a Cu additive alleviated the oxidative stress and lipid peroxidation.<sup>48</sup>

The activities of serum LZM, IgM, C3, and C4 of *O. niloticus* fed on three different experimental diets are presented in Table 4. The best immune response in *tilapia* was observed in the NCM-Cu treatment. Fish fed on diets supplemented with NCM-Cu had a significantly higher content of IgM (0.72 mg/mL), C3 (0.77 mg/mL), and C4 (1.18 mg/mL) when compared to the fish fed on diets supplemented with Cu-basic. Simultaneously, compared to CuSO<sub>4</sub>, a significant increase in the activity of LZM (10.93 U/mL) and IgM (0.72 mg/mL) of the fish fed on the NCM-Cu diet was observed, as shown in Table 4. This result denoted the positive effect of the NCM-Cu complex on the immunity response of *O. niloticus*, which may be related to Cu accumulation in tissues. A recent study also demonstrated that Russian

sturgeon fed with methionine Cu displayed significantly higher immune responses and resistance to bacterial infection than CuSO<sub>4</sub> treatment. Because Cu was the key component of functional protein (e.g., albumin) in an immunity system, the improved bioavailability of dietary Cu might stimulate the innate immune response. On the contrary, Cu deficiency had an adverse impact on immune responses and decreased the resistance in the fish.<sup>49</sup> Generally, the first consideration for aquaculture producers is to obtain the best fish production. There are many factors that influence fish populations including aquatic ecosystems conditions, species interactions, disease, and feed supplements.<sup>50,51</sup> From the science-based knowledge we have accumulated to-date, these factors can be enacted to mitigate, minimize, or reverse by some suitable management actions. Specifically, the use of fortification of foods or dietary supplements can increase the overall fish fitness and reduce the environmental pollution.<sup>52</sup> It is remarkable that aquaculture sustainability must balance the socioeconomic needs and ecological protection. Therefore, choosing fish feed additives with higher bioavailability and bioactivity will be beneficial for the development of the entire aquaculture industry.

## CONCLUSIONS

In the present study, a promising *N*-carbamoyl-functionalized NCM-Cu complex was synthesized, and its stable structure protected itself against oxidative damage. Besides, multiples structure analysis indicated that the aminocarbonyl group was coordinated with secondary amine and the Cu atom was chelated to the carboxyl groups of the NCM-Cu complex. Single-crystal X-ray diffraction result clearly confirmed the NCM-Cu complex as C<sub>12</sub>H<sub>30</sub>CuN<sub>4</sub>O<sub>10</sub>S<sub>2</sub>. Anonymously, the overall growth performance of the fish remained unaltered with NCM-Cu adjunct feed. NCM-Cu significantly raised the Cu accumulation in the fish muscles (4.99 mg/kg), liver (154.3 mg/kg), gill (7.18 mg/kg), and intestine (8.48 mg/kg) in contrast to the basic Cu-rich feed. The serum antioxidant enzyme activity elevated up to (ceruloplasmin: 19.38 U/L) and the lowest liver MDA content (8.81 nmol/mg prot.) and triglyceride content (0.39 mmol/g prot.) were observed in the NCM-Cu group as compared to the basic Cu and CuSO<sub>4</sub> groups, suggesting that NCM-Cu promoted antioxidative responses and alleviated lipid peroxidation of *O. niloticus*.

This exhibits the nurturing of the antioxidative reflexes and the alleviated lipid peroxidation in *O. niloticus* in response to an NCM-Cu adjutant. Overweening, the synthesized complex, NCM-Cu significantly regulated the expression levels of lysozyme, immunoglobulin M, complement 4, and complement 3 up to 10.93 U/mL, 0.72, 0.77, and 1.18 mg/mL in serum, respectively, proclaiming the commendatory immunogenic effects of the NCM-Cu adjutant. Thus, such endorsed results reveal the preeminence of the NCM-Cu-supplemented diet for the fitness in *O. niloticus*. These findings pointed out the potential use of NCM-Cu as a new Cu supplementation source in the aquaculture industry since its clear molecular structure would facilitate further studies on the mechanism of action in the body. Furthermore, the biological experiment displayed that the NCM-Cu complex had a good efficacy on Cu accumulation, antioxidant stress, alleviated lipid peroxidation, and immunity response of *O. niloticus*.

## MATERIALS AND METHODS

Methionine (Met), sodium cyanate (NaCNO), and  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  were supplied by Aladdin Co., Ltd. Juvenile tilapia of *O. niloticus* was bought from a commercial farm in Guangzhou, China. All of the diagnostic kits for measuring the enzymatic capacity and immune indices were purchased from Nanjing Jiancheng Bioengineering Institute, China.

**Preparation of NCM-Cu.** Briefly, 121.19 g of Met and 67.11 g of NaCNO were accurately weighed and added to 350 mL of deionized water at 90 °C with 90 min of stirring. Subsequently, the mixture solution was naturally cooled to a solution temperature of 40 °C. The pH of the above mixture was adjusted to 3 using dilute sulfuric acid and further stirred for 60 min. The reaction solution was separated after standing for 60 min. To obtain pure *N*-carbamoyl methionine (NCM), the upper solution was collected and dried at 50 °C for 5 h in a hot air oven.

Five grams of NCM and 13.02 g of  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  were accurately weighed and dissolved in 20 mL of deionized water, and the pH was adjusted to 4.5 using a 0.1 mol/L NaOH solution. The mixture was then magnetically stirred for 60 min at 60 °C, and the target solution was obtained after filtering. To obtain pure NCM-Cu, the target solution was collected and dried at 50 °C for 4 h in a hot air oven.

**Structural Characterization of NCM-Cu.** For structural characterization, UV-vis spectra of Met and NCM-Cu were recorded on a TU-1810 spectrophotometer (Beijing Purkinje General Instrument Co., Ltd., China) in the range of 190–400 nm. Besides, Met and NCM-Cu were measured by Fourier-transform infrared spectrometry (Spectrum 100, PerkinElmer Inc.) using a KBr tablet method. Moreover, the crystal phases of Met and NCM-Cu were investigated by an X-ray powder diffractometer (X'Pert PRO MPD, Panalytical, Holland) with Cu  $K\alpha$  radiation in the  $2\theta$  range of 5–70° at a scan rate of 1°/min. Furthermore, the structure data of NCM-Cu were collected via single-crystal X-ray diffraction (SMART APEX II, Bruker, Germany) and solved by direct methods using SHELXS, and refinement was done against  $F^2$  using SHELXL. Simultaneous thermogravimetric analysis (TGA) and differential thermal analysis (DTA) were performed for Met and NCM-Cu. The curves were obtained using TG-DTA modulus (TGA2, Mettler Toledo, Switzerland) under a nitrogen atmosphere from 40 to 600 °C with a heating rate of 10 °C/min.

**Experimental Diet Preparation.** To meet all known nutritional requirements for *O. niloticus* juveniles, the basal diet was prepared. The formulation and proximate composition of the diets are shown in Table 5. Specifically, fish meal, soybean

**Table 5. Formulation and Proximate Composition of the Basic Diets**

ingredient (g/kg)	basic Cu diet	$\text{CuSO}_4$ diet	NCM-Cu diet
fish meal	180	180	180
soybean meal	230	230	230
sesame meal	240	240	240
DDGS	50	50	50
wheat flour	230	230	230
soybean oil	25	25	25
monocalcium phosphate	15	15	15
choline chloride	10	10	10
vitamin premix <sup>a</sup>	10	10	10
mineral premix, Cu-basic <sup>b</sup>	10	10	10
proximate composition			
crude protein (%)	39.38	39.31	37.50
crude lipid (%)	7.82	7.83	7.66
ash (%)	10.22	10.08	9.97
moisture (%)	5.49	6.60	7.80
final actual concentration of Cu (mg/kg)	10.43	40.14	41.88

<sup>a</sup>Vitamin premix (mg/kg diet): vitamin B<sub>1</sub>, 12; vitamin B<sub>12</sub>, 0.05; vitamin B<sub>6</sub>, 8; nicotinic acid, 30; vitamin D<sub>3</sub>, 5; vitamin C, 100; vitamin E, 50; pantothenic acid, 40; biotin, 0.8; folic acid, 5; vitamin A, 25; vitamin K<sub>3</sub>, 8; riboflavin, 12; inositol, 100. <sup>b</sup>Mineral premix, zinc-basic (mg/kg diet):  $\text{FeSO}_4 \cdot \text{H}_2\text{O}$ , 400; KCl, 200;  $\text{MnSO}_4 \cdot \text{H}_2\text{O}$ , 150; KI 60;  $\text{Na}_2\text{SeO}_3 \cdot 5\text{H}_2\text{O}$ , 65;  $\text{ZnSO}_4 \cdot 7\text{H}_2\text{O}$ , 40.

meal, and sesame meal were used as protein sources; wheat flour was used as the carbohydrate source; and soybean oil was used as soybean oil. To evaluate the bioactivities in *O. niloticus* of supplementing diet with NCM-Cu (synthesized by this studies), the control (as basic Cu) and  $\text{CuSO}_4$  groups were formulated. The additional Cu sources in the basic diets of both  $\text{CuSO}_4$  and NCM-Cu groups were 30 mg/kg. The dietary Cu concentration was analyzed by inductively coupled plasma spectrometer (ICP-OES 2100, PerkinElmer), as shown in Table 5. The ingredients of the experiment diet were weighed and thoroughly mixed in a food mixer, then put in a pelletizer and cut into uniform sizes. The molded feeds were dried in a mechanical convection oven and then stored at 4 °C for use.

**Bioactivity Experiments.** Juveniles of *O. niloticus* in the study were purchased from a commercial farm in Guangzhou, China. Upon arrival, the fish were acclimated in 400 L laboratory tanks for 2 weeks and fed with the basal diets. After acclimation, a total of 180 individuals of *O. niloticus* with an average initial weight of  $77.27 \pm 0.15$  g were randomly distributed into three groups, each group containing three replicates (20 fish/replicate). The first group served as the control group fed on a Cu basal diet. The second and third groups were fed on a  $\text{CuSO}_4$  diet and NCM-Cu diet, respectively. The experiment lasted for 60 days and was under ambient temperature and natural light and dark cycle. Besides, each tank was equipped with water inlet and outlet, and continuously aerated to maintain the dissolved oxygen concentration higher than 6 mg/L. *O. niloticus* was fed two times a day (daily ration was about 4% of biomass) at 9:00 am and 5:00 pm. During the feeding trial, *O. niloticus* were fed

closely to satiation and the uneaten feed after feeding 30 min was collected, dried, and weighed to calculate the feed intake. Moreover, the fish were weighed every 2 weeks and the daily ration diet was adjusted accordingly. Water temperature, dissolved oxygen, total ammonia–nitrogen, and pH were monitored daily by YSI Proplus (YSI, Yellow Springs, Ohio). The mean water quality parameters were recorded as follows: dissolved oxygen was  $\geq 6$  mg/L, total ammonia–nitrogen was  $0.055 \pm 0.012$  mg/L, pH was  $7.2 \pm 0.20$ , and the water temperature ranged from 27 °C to 32 °C.

**Sample Collection.** At the end of the feeding period, all fish in each tank were fasted for 24 h prior to final sampling. Subsequently, all of the fish from each replicate were randomly collected and weighed for the analysis of survival and growth performance including the weight gain (WG, %), feeding efficiency (FE, %), and condition factor (CF, g/cm<sup>3</sup>). Parameters were calculated according to the following formulae:

$$\text{survival (\%)} = \left[ \frac{\text{final number of fish}}{\text{initial number of fish}} \right] \times 100\%$$

$$\text{weight gain (WG, \%)} = \left[ \frac{\text{final body weight} - \text{initial body weight}}{\text{initial body weight}} \right] \times 100\%$$

$$\text{feed efficiency (FE, \%)} = \left[ \frac{\text{final body weight} - \text{initial body weight}}{\text{feed intake}} \right] \times 100\%$$

$$\text{condition factor (CF, \%)} = \left[ \frac{\text{final individual weight}}{\text{final individual length}^3} \right] \times 100\%$$

Then, blood samples from three fish of each tank were collected as quickly as possible from the caudal vein using sterilized syringes and pooled (three pooled samples per tank) [9, 25]. The serum was separated after being stored at 4 °C for 4 h and after that, centrifugation occurred at 3500 rpm for 15 min. The supernatant was collected to evaluate the antioxidant and immunological parameters in *O. niloticus* including lysozyme (LZM), immunoglobulin M (IgM), complement 4 (C4), complement 3 (C3), and ceruloplasmin (CP). In addition, viscera and liver were dissected out from 10 fish from each aquarium, weighed individually to get the viscerosomatic index (VSI) and hepatosomatic index (HSI). Moreover, muscles, liver, gill, and intestine were collected from three fish each tank to estimate the tissue mineralization (Cu, Zn, Fe, and Mn contents) of *O. niloticus*. Meanwhile, a small part of the liver from three fish of each tank was collected and pooled (three pooled samples per tank) for the measurement of the contents of total superoxide dismutase (T-SOD), Cu-Zn superoxide dismutase (Cu-Zn SOD), malondialdehyde (MDA), triglyceride (TG), and total cholesterol (T-CHO). Enzymatic activity and immune parameters were analyzed using an automatic biochemical analyzer (Sysmex-800, Sysmex Corporation, Kobe, Japan) and followed the methods and instructions of the specific kits. VSI and HSI were calculated according to the following formulae:

$$\text{hepatosomatic index (HSI, \%)} = \left( \frac{\text{hepatic weight}}{\text{body weight}} \right) \times 100\%$$

$$\text{viscerosomatic index (VSI, \%)} = \left( \frac{\text{viscera weight}}{\text{body weight}} \right) \times 100\%$$

**Statistical Analysis.** The data (means  $\pm$  SEM, standard error) in this study were statistically calculated using SPSS 25.0 statistical software and then analyzed by one-way ANOVA method. A *p* value of  $<0.05$  was judged to be statistically significant. Tukey's multiple comparison test was used when there was homogeneity of variances.

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.0c03220>.

Tissue mineralization analysis; raw data for Zn, Cu, Fe, and Mn concentrations in the fish tissue (PDF)

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### Notes

The authors declare no competing financial interest.

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