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Structure and mechanism of a unique diiron center in mammalian stearyl-CoA desaturase

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Abstract

Stearyl-CoA desaturase 1 (SCD1) is a membrane-embedded metalloenzyme that catalyzes formation of a double-bond on a saturated acyl-CoA. SCD1 has a diiron center and its proper function requires an electron transport chain composed of NADH (or NADPH), cytochrome b₅ reductase (b₅R), and cytochrome b₅ (cyt b₅). Since SCD1 is a key regulator in fat metabolism and is required for survival of cancer cells, there is intense interest in targeting SCD1 for various metabolic diseases and cancers. Crystal structures of human and mouse SCD1 were reported recently, however, both proteins have two zinc ions instead of two iron ions in the catalytic center and as a result, the enzymes are inactive. Here we report a general approach for incorporating iron into heterologously expressed proteins in HEK293 cells. We produced mouse SCD1 that contains a diiron center, and visualized its diiron center by solving its crystal structure to 3.5 Å. We assembled the entire electron transport chain using the purified soluble domains of cyt b₅ and b₅R, and the purified mouse SCD1, and we showed that three proteins coordinate to produce proper products. These results established an *in vitro* system that allows precise perturbations of the electron transport chain for the understanding of the catalytic mechanism in SCD1.

Keywords

SCD1; membrane enzyme; diiron center; electron transfer; double bond formation

Introduction

Stearyl-CoA desaturase (SCD) is embedded in the endoplasmic reticulum (ER) membrane and converts saturated acyl-CoAs to monounsaturated acyl-CoAs which are precursors for the biosynthesis of phospholipids, cholesterol esters, and triglycerides [1, 2]. Humans have

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Competing interests

The authors declare no competing interests.

two SCD genes encoding SCD1 and SCD5 that are of 53% sequence identity, while mice have four (SCD1–SCD4) that are of 77–86% sequence identity. SCD1 is a key enzyme that helps maintain balance between fat consumption and accumulation [3, 4]. Mice with SCD1 knocked out are resistance to high-fat diet, leading to the perspective that drugs that inhibit SCD1 activity could be used to treat metabolic diseases such as obesity and diabetes [5, 6]. Expression of SCD1 is significantly higher in cancer cells due to a higher demand for membrane synthesis, and preclinical research have shown that inhibiting SCD1 activity stops propagation of cancer cells [7–10]. Early biochemical studies demonstrated that the desaturation reaction is catalyzed by a diiron center in SCD1 in a highly regio- and stereo-specific manner [1, 11, 12]. The double bond is formed at the 9th position (9) of the acyl chain (regiospecificity) and with the *cis* configuration (stereospecificity). SCD1 can utilize saturated acyl-CoAs with an acyl chain length of 14–19 carbons as substrates but favors stearoyl-CoA (18:0 CoA) over others [13].

Crystal structures of both mouse and human SCD1 were reported previously [14, 15]. The mouse SCD1 has an 84% sequence identity with human SCD1 and the two structures are almost identical with a root mean squared deviation (RMSD) of 0.35 Å. SCD1 has a structural fold that is different from soluble desaturases such as acyl-ACP desaturase, which catalyzes a similar reaction [16–18]. SCD1 is a monomer and has a mushroom-like shape with a dimension of 55 × 53 × 41 Å. Each SCD1 has four transmembrane helices (TM1–4) that form the stem of the mushroom and a cytosolic domain that forms the crown of the mushroom (Figure 1a). TM2 and TM4 are longer than TM1 and TM3 and protrude into the cytosolic domain to provide three of the nine histidine residues that coordinate the two metal ions (Figure 1a). The other histidine residues come from the soluble domain. However, further analysis showed that the two metal ions in the structure are zinc ions [14, 15], which is an artefact likely due to overexpression of SCD1 proteins in a heterologous system. In both the mouse and human SCD1 structures, a large non-protein electron density was present, and a stearoyl-CoA was built into the electron density. The coenzyme A moiety is recognized by the surface of the protein while the acyl chain accommodated by a V-shaped tunnel inside the protein (Figure 1b). The dimension of the tunnel is such that 9 and 10 carbons are positioned at the inflection point of the V-shape and is close to the dimetal center. The V-shaped tunnel also ensures that only the two hydrogens in *cis* are positioned close to the metal ions for extraction (Figure 1b). Thus, the conformation of the bound stearoyl-CoA provides a simple explanation for both the regioselectivity and stereospecificity imposed by SCD1s.

Although Zn²⁺ serves as a surrogate to maintain the structural integrity because of its similar ionic radius and charge property to a ferrous ion (Fe²⁺), misincorporation of Zn²⁺ renders the enzyme inactive and thus hampers further investigation of the enzymatic reaction mechanism. The misincorporation of Zn²⁺ is not unique to SCD1, and it was also reported in the production of other enzymes such as yeast fatty acid α -hydroxylase [19] and quinol-dependent nitric-oxide reductase [20].

Proper function of SCD1 requires two additional membrane-embedded proteins, cytochrome b₅ reductase (b₅R) and cytochrome b₅ (cyt b₅). b₅R contains a single transmembrane helix at its N-terminus, and a soluble cytosolic domain composed of an N-terminal and C-terminal

halves. A flavin adenine dinucleotide (FAD) binds to the N-terminal half while a NAD(P)H binds to its C-terminal half. The primary redox partner of b₅R is cyt b₅, which is composed of a soluble cytosolic domain and a single transmembrane domain at the C-terminus. The soluble domain contains a b-type heme, which is ligated to two axial histidine ligands. b₅R and cyt b₅ are both anchored to the ER membrane by their single transmembrane domain. An electron transport chain is formed sequentially from NADH/NADPH, b₅R, and cyt b₅, which then interacts with downstream enzymes such as oxygenase and desaturase (Figure 2) [21, 22]. Since it is not known how the three proteins interact, there are two potential ways the electrons can be delivered. Cyt b₅ may need to shuttle between its two partners so that binary complexes b₅R-cyt b₅ and cyt b₅-SCD1 form alternately. It's also possible that a ternary complex of b₅R-cyt b₅-SCD1 could form and allow for continuous electron transfer.

Functional studies on SCD1 have been reported previously [21]. SCD1 activity in isolated rat liver microsomes was monitored by oxidation of NADPH [23, 24]. SCD1 has also been produced in a cell-free translation system and its activity was demonstrated using a radiolabeled substrates [25]. Both systems have provided information on required components for the desaturation reaction. However, these systems are inadequate for structural and spectroscopic studies that require large amount of protein. To further understand the mechanism of SCD1 catalysis, it is imperative that we produce fully functional SCD1 so that the redox state of the bound irons can be tracked by precise spectroscopic studies such as UV-vis spectroscopy and electron paramagnetic resonance and by structural studies via X-ray crystallography. As a first step towards this goal, we developed a protocol to consistently produce iron-containing SCD1 and solved its crystal structure to 3.5 Å. We also purified the soluble domains of cyt b₅ and b₅R and reconstituted the stearyl-CoA desaturation reaction *in vitro*.

Results

Production of iron-containing SCD1 and other proteins

As an initial effort to replace zinc with iron in SCD1s, we added iron in cell culture media but the majority of the purified SCD1 still has zinc ions. This indicates that iron was not efficiently taken up by the insect cells and led us to try the HEK cell expression system. We reasoned that the endogenous transferrin receptor could facilitate iron uptake. We expressed mouse SCD1 in human embryonic kidney (HEK) 293 cells and supplemented the cell culture media with transferrin and ferric ions (Figure 3a). SCD1 produced by this protocol has an iron occupancy of more than 90% as determined by inductively coupled plasma mass spectrometry (ICP-MS) (Figure 3c). After optimizing the expression and purification protocol, we routinely produce ~1 mg pure iron containing SCD1 per liter of cell culture. The iron-loaded SCD1 shows a small and yet significant absorbance peak around 340 nm ($A_{340}/A_{280} \approx 0.082$), which is typical for a non-heme diiron center [26] (Figure 3b).

Crystal structure of SCD1 with diiron center

We proceeded to crystallize the iron containing mouse SCD1 and obtained crystals using the lipidic cubic phase (LCP) method [27]. These crystals diffracted to ~3.5 Å and the structure was solved by molecular replacement using the previous SCD1 structure as a search model

and with both the metal ions and the histidine residues that coordinate the metal ions omitted. Metal ions and histidine residues were then built into electron densities of the “omit” map. A composite omit map covering the entire contents of the unit cell was also calculated to further reduce model bias (Supplementary Figure 1). The final structural model was refined to R/R_{free} of 21.9%/ 27.6%; the final structure includes residues 41 to 361, two iron ions, and one oleoyl CoA. The structure of the current iron containing SCD1 is almost identical to the previous zinc-containing SCD1 with a RMSD of only ~ 0.3 Å for all backbone atoms (Figure 4a). The structures also show that the conserved histidine residues that coordinate the di-metal center have the same coordination geometry (Figure 4a). These results suggest that although Zn^{2+} cannot catalyze the redox reaction, it is a suitable surrogate for Fe^{2+} to maintain the three-dimensional structure likely due to its similar ionic radius to Fe^{2+} (0.88 Å for Zn^{2+} and 0.92 Å for Fe^{2+}) and charge property.

In addition to the ICP-MS results, the presence of two iron ions was confirmed by anomalous x-ray scattering signal, which unequivocally locates the iron ions in the structure (Figure 4b). The two iron ions are separated by ~ 6.4 Å, similar to that between the two zinc ions in the previous SCD1 structures (Figure 4c) [14, 15]; This distance exceeds the range allowed for the formation of known reaction intermediates, such as cis- μ -1,2 peroxo or diferryl, proposed in other enzymes with a diiron center. One of the iron ions (Fe1) is coordinated by five histidine residues while another (Fe2) by four and a putative water molecule. These histidine residues are highly conserved and segregate into four histidine-rich motifs that are distant in the primary sequence [14]. Both the coordination by all histidine residues and the distance between the ions are different from all other diiron centers found in soluble redox enzymes, such as acyl-ACP desaturase [28], ribonucleotide reductase [29], and methane monooxygenase [30, 31] (Supplementary Figure 2). There is no obvious electron density between the two iron ions, suggesting that an oxo-bridge, which is a common feature in other diiron centers, is either not present or short-lived. These properties of the diiron center suggest that the activation of molecular oxygen in SCD1 has a novel mechanism.

Oleoyl-CoA is present in both structures

As is observed in the zinc-containing SCD1, one acyl-CoA molecule is present in our structure (Figure 5a). In the current structure, clear and strong electron density is present for the CoA moiety, while the electron density for the acyl chain is less continuous, likely due to the modest resolution and a lower occupancy of the acyl-CoA. In the previous structure which has a higher resolution, the electron density is continuous for the entire acyl-CoA (Figure 5b) [14].

We developed a protocol to extract and identify the bound acyl-CoA from the SCD1 and examined the SCD1 produced from the previous insect cell expression system and the current HEK cell expression system. We found that the bound ligand is oleoyl-CoA and its molar ratio to protein is close to 1:1 (Methods). Therefore, we built an oleoyl-CoA into the previous mouse SCD1 structure (PDB ID: 4YMK) and refined the model. It turns out that an 18:1 acyl chain fits better at the 9 and 10 position than the previously built 18:0 acyl chain (Figure 5b). This geometry restraint imposed by the double bond results in slightly

different conformation of the zigzag string of acyl chain between the 10 carbon and the thioester sulfur, whose positioning is stabilized by the extensive interactions between CoA group and SCD1. The 9 carbon on the oleoyl-CoA is $\sim 3 \text{ \AA}$ to the water bound to Fe²⁺ (Figure 5c).

Electron transfer from cyt b₅ to SCD1

We expressed and produced the soluble domains of cyt b₅ and b₅R (Supplementary Figure 3a and 3c). Spectroscopic features of the two proteins confirm the presence of appropriate cofactors [32, 33]. In the presence of b₅R, addition of NADH can reduce the heme group in cyt b₅ as demonstrated by a shift of the Soret peak from 413 nm to 423 nm, which indicates that both proteins function properly (Supplementary Figure 3b and 3d).

To examine the redox communication between cyt b₅ and SCD1, we measured the kinetics of cyt b₅ oxidation in presence of SCD1. We first prepared ferrous cyt b₅ with b₅R and NADH. Reduced cyt b₅ can be gradually oxidized by molecular oxygen in solution, which interferes with the electron transfer between cyt b₅ and SCD1. Therefore, the reactions were carried out anaerobically. The SCD1 accelerates the oxidation of cyt b₅, which exhibits a biphasic time course with a $k_1 = 1.86 \text{ min}^{-1}$ and $k_2 = 0.246 \text{ min}^{-1}$ as observed by following the decrease of Soret peak at 423 nm (Figure 6). These results indicate that the purified cyt b₅ and SCD1 are both functional and form an electron transport pair.

In vitro assembly of the continuous turnover system

We next tested if we can assemble an *in vitro* enzymatic reaction. Detergent-solubilized SCD1, the cytosolic domains of cyt b₅ and b₅R, and stearoyl-CoA were included in the reaction mixture, and the reaction was initiated by the addition of NADH. The reactions were stopped at different time points and the amount of oleoyl-CoA was determined (Methods) (Figure 7a). Detection of *de novo* synthesis of oleoyl-CoA provides unequivocal evidence that we have assembled an *in vitro* system for steady-state turnover. The initial rate of oleoyl-CoA production at different concentrations of stearoyl-CoA was measured and a Michaelis-Menten curve was fitted to the data with a K_M and a k_{cat} of $6.3 \pm 1.2 \text{ \mu M}$ and $2.78 \pm 0.59 \text{ min}^{-1}$, respectively (Figure 7b).

The production of oleoyl-CoA stalled after ~ 10 minutes well before the exhaustion of substrate stearoyl-CoA, and no more oleoyl-CoA was produced even with addition of more NADH (Figure 7c). The maximum oleoyl-CoA production was proportional to the amount of SCD1 in the system, and the reaction seemed to stall with a turnover number of 7.52 ± 0.047 (Figure 7d). We speculate that product inhibition occurs because of limited capacity of a detergent micelle around SCD1.

Discussion

When the SCD1 was misincorporated with Zn²⁺, the bound Zn²⁺ resists extraction by extensive dialysis or high-affinity zinc chelators like TPEN (Methods). The current protocol provides an efficient procedure for large-scale production of iron-containing SCD1. We anticipate that this protocol can be adapted for production of other iron-containing proteins.

The structures of SCD1 show a tightly bound acyl-CoA, which we have identified to be an oleoyl-CoA even in the non-functional Zn-containing SCD1. After refinement, the distance between $\text{Fe}1$ and $\text{Fe}2$ is 4.7 Å, which is slightly less than that in the previous structure (Figure 5c). We speculate that the release of oleoyl-CoA may require larger conformational changes of SCD1 because the acyl chain becomes more rigid when a double bond is present. We suspect that product inhibition could be due to a combination of higher apparent affinity towards oleoyl-CoA and limited diffusion of oleoyl-CoA in detergent micelles surrounding SCD1. Such product inhibition may be lifted when SCD1 are on the ER membrane, where free diffusion and drainage of products by downstream enzymes, such as DGAT and ACAT [21], will reduce product concentration, thus sustaining continuous turnover.

SCD1 is embedded in ER membrane while $\text{cyt } b_5$ and b_5R are anchored to the ER membrane by their single transmembrane helix (Figure 2). This configuration limits their diffusion to 2-dimension and enhances their interactions. The rate-limiting step in our *in vitro* assembled chain is the interactions between $\text{cyt } b_5$ and SCD1, as the rate of electron transfer from $\text{cyt } b_5$ to SCD1 ($k_T = 1.86 \text{ min}^{-1}$) is comparable to the steady-state rate of oleoyl-CoA formation ($k_{cat} = 2.78 \text{ min}^{-1}$). Since our experiments are performed with the three proteins all in the solution, their communications are likely not as efficient as in the native environment. Further optimization could come from reconstitution of all three proteins into a nanodisc or generating fusion proteins of either two or all three partners. Fusion of a $\text{cyt } b_5$ -like domain occurs in certain desaturases, such as OLE-1 [34] and FAT5–6 [35], which may reflect a strategy of achieving efficient electron transfer in the evolution of these enzymes.

Methods

Iron incorporation into Sf9-expressed SCD1

Mouse SCD1 was expressed and purified from Sf9 (*Spodoptera frugiperda*) cells as previously described [14]. We tried several approaches, and found that the most efficient way of replacing zinc ions was to dialyze SCD1 first against two strong zinc chelators, N,N,N',N'-tetrakis(2-pyridinylmethyl)-1,2-ethanediamine (TPEN) (100 μM) and dipicolinic acid (DPA) (10 mM), and then against high concentrations of ferrous ions (10 mM). However, the maximum iron occupancy was less than 70% as determined by ICP-MS and there was a significant loss of protein during the required prolonged dialysis. Since both zinc ions have to be replaced in order for the protein to be functional, a 70% occupancy translates to less than 50% (0.7²) of functional protein.

Construct of mouse SCD1

Mouse SCD1 gene with deletion of residue 2–23 at the N-terminus [14] (m SCD1) was codon-optimized for human cell expression and subcloned into a BacMam expression vector [36] with green fluorescent protein (GFP) tag attached to the C-terminus. A tobacco etch virus (TEV) protease recognition site was inserted between the C-terminus of the protein and the GFP tag.

Overexpression and purification of SCD1 from HEK293 cells

The resulting plasmid containing cDNA of mouse SCD1 was used to generate baculoviruses following the manufacturer's protocol (Invitrogen). HEK293S GnTI- suspension cells at 3×10^6 cells/mL in Freestyle 293 medium (GIBCO) supplemented with 2% FBS were infected with 10% P3 virus. Iron-saturated human transferrin (Sigma) in 50 mM sodium bicarbonate (pH 8.5) and 10 mM FeCl_3 solution were filtered and added to the cell culture to reach a final concentration of 5 mg/L transferrin and 5 μM FeCl_3 . Infected cells were incubated at 37°C overnight (~16 h). The protein expression was then induced by adding 10 mM sodium butyrate and temperature was decreased to 30 °C for two days (~48 h) before harvesting [36]. Cell pellets were homogenized in buffer A containing 20 mM HEPES (pH 7.5), 150 mM NaCl, 1 mM PMSF, 5 mM MgCl_2 , and DNaseI. 40 mM n-Decyl- β -D-Maltopyranoside (DM, Anatrace) was added to solubilize the membranes at 4 °C for 2 h under gentle agitation. After centrifugation (55,000g, 45 min, 4 °C) to remove the insoluble fraction, the supernatant containing the detergent-solubilized proteins was collected and loaded onto a column of CNBR-activated Sepharose beads (GE Healthcare) coupled to GFP nanobody (GFPnb). The GFPnb was derived from a llama single chain antibody and was expressed and purified as previously reported [37]. The columns were pre-equilibrated with buffer B [20 mM HEPES (pH 7.5), 150 mM NaCl, 4 mM DM]. After 1 h incubation at 4 °C, GFPnb resins were washed with 20 column volume of buffer B. The proteins were cleaved off the column with TEV protease at room temperature for 1 h. The TEV protease with a polyhistidine tag was removed by loading the supernatant to pre-equilibrated cobalt-based affinity resins (Talon, Clontech). The released proteins were collected, concentrated (Amicon 50 kDa cutoff, Millipore), and loaded onto a size-exclusion column (Superdex 200 10/300 GL, GE Health Sciences) equilibrated with buffer B. Peak fractions were collected and analyzed by SDS-PAGE. The predicted molecular mass and extinction coefficient at 280 nm of m SCD1 are 39,512 g/mol and 73,800 $\text{M}^{-1}\text{cm}^{-1}$ [38], respectively, and were used to calculate protein concentration.

ICP-MS

HEK-expressed and Sf9-expressed SCD1 were diluted to 15 μM in buffer B. 300 μL of each sample together with blank buffer were subjected to ICP-MS (Agilent 8800 Triple-Q-ICP-MS) at the ICP Analytical Research Lab at University of Houston. Iron and zinc standards were analyzed at the beginning and the end of each run to set up the calibration line. All samples were prepared in 2% HNO_3 prior to analysis.

Crystallization and structure determination

Crystallization trials were carried out in LCP methods [27]. Purified SCD1 were concentrated to 30–50 mg/mL. Protein solution was mixed with molten monoolein (Sigma) doped with 10% (w/w) cholesterol (Sigma) [39] at 1:1.5 (v/w) ratio with a coupled syringe device. Gryphon crystallization robot was used to dispense 50 nL protein-lipid mixture into a 96 well glass sandwich plate (Molecular Dimensions) and overlaid with 800 nL precipitant solution in each well. Crystallization conditions were optimized based on an initial crystal hit (100 mM Tris, pH 8.5, PEG400 35%, 50 mM $\text{Mg}(\text{Acetate})_2$) from a home-made grid screen. The typical condition for good crystal growth is: 100 mM Tris, pH 7.8–8.2, PEG400 37–

42%, 20 mM Mg(Ace)₂, and 50 mM MgCl₂. Crystals were harvested and flash frozen in liquid nitrogen directly without cryoprotection.

Crystals were screened at beamlines 24ID (NE-CAT) at the Advanced Photon Source at the Argonne National Laboratory. A dataset was collected with a resolution of 3.51 Å at a wavelength of 0.9791 Å. The diffraction images were indexed, integrated, and scaled using the XDS [40]. The phases were obtained by the molecular replacement (MR) method in Phenix [41] with mouse SCD1 structure (PDB: 4YMK) [14] as a search model. The obtained solution was subjected to successive rounds of refinement in phenix.refine [41] and manual correction in COOT [42]. Protein geometry was validated by MolProbity [43]. Datasets collected at 1.7389 Å near iron edge were of poorer quality, which diffracted to ~4 Å. X-ray fluorescence emission spectrum confirms the presence of iron in the crystals. The diffraction data were processed similarly by XDS, which indicated that anomalous signals exist in these datasets. Refined model from the datasets of normal wavelength were used as search model. After several rounds of refinement, the resulting anomalous difference map was inspected and four predominant peaks were found in one asymmetric unit, which corresponds to two diiron centers. All structure figures were produced with PyMOL (Schrödinger LLC.).

Overexpression and purification of soluble cyt b₅ and b₅R

The cDNAs of mouse cyt b₅ and b₅R were subcloned into a pET vector which appends a polyhistidine tag and a TEV protease site to the C-terminus (for cyt b₅) or the N-terminus (for b₅R) of the overexpressed protein. High-level expression was achieved by following the previous protocols for soluble cyt b₅ [44] and b₅R [32]. Cells expressing cyt b₅ were supplemented with 0.5 mM δ-aminolevulinic acid, 5 μM FeCl₃, and 1 μM hemin chloride, and those expressing b₅R with 100 μM FAD in the media. Cell pellets were resuspended in buffer A and then sonicated until cells appeared fully lysed. The following steps are similar to the SCD1 purification protocol. Target protein capture was achieved by cobalt-based affinity resins (Talon, Clontech) and proteins were cleaved off the resins with TEV protease. Buffers without DM and concentrators with 10 kDa cutoff were used.

Electron transfer kinetic measurement

3.1 μM soluble cyt b₅-b₅R and 3.6 μM SCD1 in buffer B were placed in two tonometers; the solutions were then made anaerobic by 5 cycles of alternate 30 s of degassing followed by 4.5 min of gas displacement with argon. The tonometers and a NADH stock prepared in anaerobic buffer were transferred into an anaerobic chamber. Anaerobic kinetic measurements were conducted with a SX-18MV stopped-flow apparatus (Applied Photophysics, Leatherhead, UK) placed inside the anaerobic chamber. Reduction of cyt b₅ domain was achieved by addition of 3.1 μM NADH. The time course of A₄₂₃ in the reaction between reduced cyt b₅ domain and SCD1 was monitored. The decay of A₄₂₃ was fit with a double exponential function: $A = A_0 + A_1 * e^{-k_1 * t} + A_2 * e^{-k_2 * t}$, where A and A_0 are absorbance and final absorbance, respectively; A_1 and A_2 are the amplitudes of the two phases; k_1 and k_2 are the rate constants of the two phases; t is the reaction time.

Continuous turnover reactions and product quantification

Typically, the steady-state reaction system includes detergent-solubilized SCD1 (2 μM), cyt b₅ (10 μM), b₅R (20 μM), and stearoyl-CoA (300 μM) in buffer B, and the reaction was triggered by the addition of NADH (1 mM). The reaction mixture was quenched at different time points with an acyl-CoA extraction buffer containing 25% iso-propanol and 50% acetonitrile (ACN) in phosphate buffer pH 5.3. We used a previously published protocol [45, 46] with modifications to separate and quantify acyl-CoAs. Briefly, the quenched reaction mixtures were clarified by a brief centrifugation to remove protein precipitates, and the supernatants were loaded in a C18 reversed phase column on a HPLC system. A linear gradient from 50% (v/v) ACN in phosphate buffer pH 5.3 to 70% was applied to elute acyl-CoAs, and the elution of acyl-CoAs was monitored at 254 nm. The identity of oleoyl-CoA and stearoyl-CoA was confirmed by comparing the retention times (RT) to the standard samples and by the mass spectrometry coupled to the HPLC (LC-MS). The amount of acyl-CoAs was calculated based on the integrated area under peak. To generate the Michaelis-Menten curve against [stearoyl-CoA], 1–75 μM of stearoyl-CoA was used in the reaction system. Time courses at different [stearoyl-CoA] were recorded. Straight lines were used to fit time courses of oleoyl-CoA production within 1 min after reaction started. The initial velocities were calculated from the slopes of these straight lines.

Accession Numbers

The coordinate and the structure factor have been deposited in the Worldwide Protein Data Bank (wwPDB) with accession number 6WF2.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Highlights

- A general approach for large-scale production of iron-containing proteins
- Crystal structure of mouse stearyl-CoA desaturase 1 with a diiron center
- *In vitro* assembly of an electron transfer chain composed of NADH, cytochrome b₅ reductase, cytochrome b₅, and stearyl-CoA desaturase 1

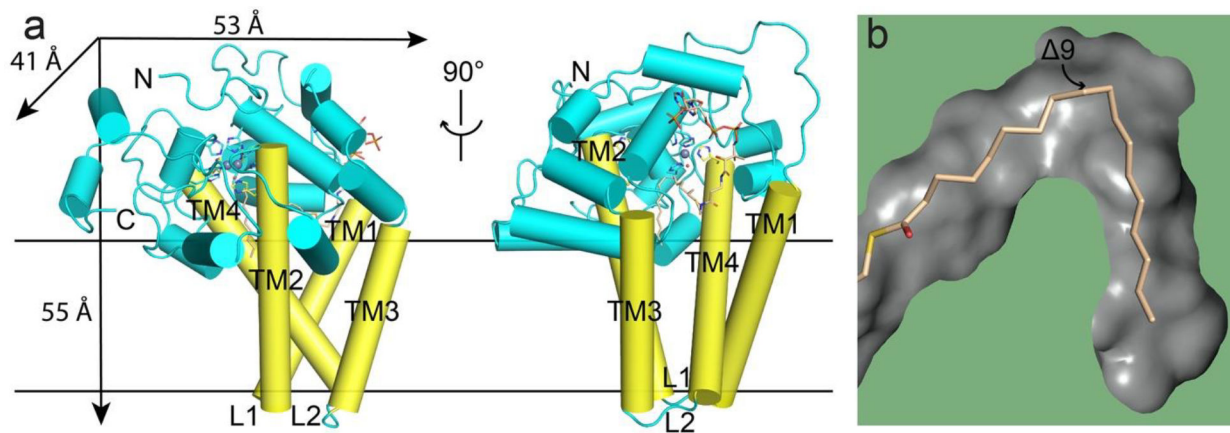


Figure 1.

Crystal structure of zinc-containing mouse SCD1 (PDB ID: 4YMK). **a.** Overall structure of SCD1 shows a mushroom-like architecture with two zinc ions (gray spheres) coordinated by nine histidine residues (side-chain sticks) and one water molecule (red sphere), and one bound stearyl-CoA (wheat stick). TM, transmembrane helix, in yellow; L, loop region; N and C, N-terminus and C-terminus. **b.** A long V-shape tunnel (gray surface) buried inside the protein accommodates the acyl chain of the bound acyl-CoA. The $\Delta 9$ carbon is in close proximity to the dimetal center located at the turn of the tunnel.

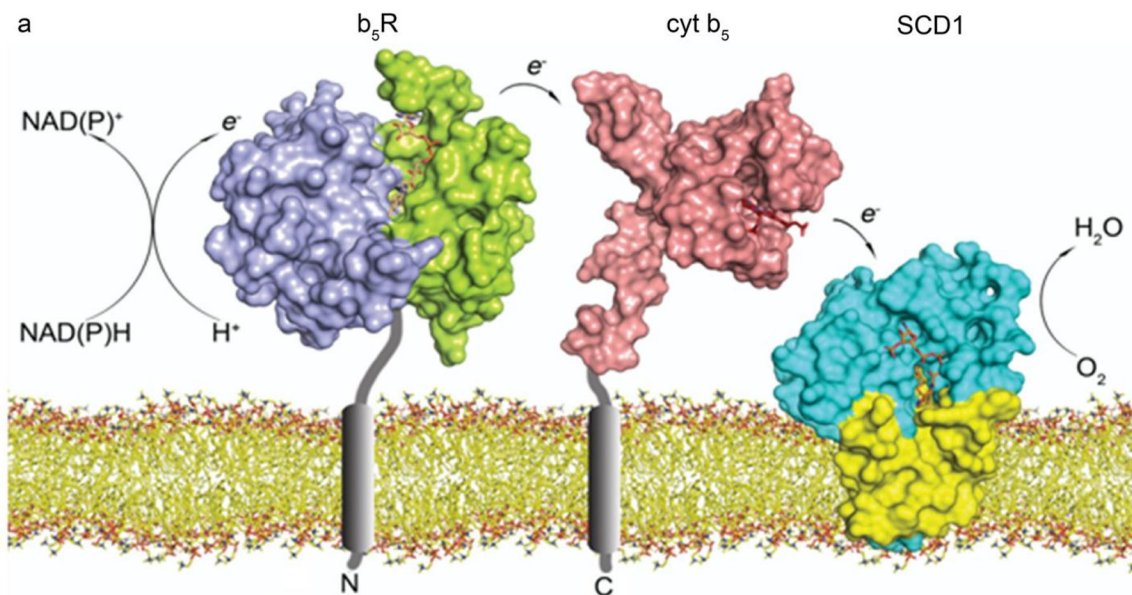


Figure 2. Proteins involved in the desaturation reaction. Electrons transfer sequentially from NADH, to b₅R (PDB ID: 1UMK), cyt b₅ (PDB ID: 2I96), SCD1 (PDB ID: 4YMK), and molecular oxygen. The single transmembrane domains in b₅R and cyt b₅ are not in their structures and shown as gray cylinders. Color coding: NADH-binding domain of b₅R (purple), FAD-binding domain of b₅R (green), FAD (wheat), cyt b₅ (pink), heme (red), soluble (cyan) and transmembrane (yellow) domains of SCD1. Lipid bilayer is shown as sticks.

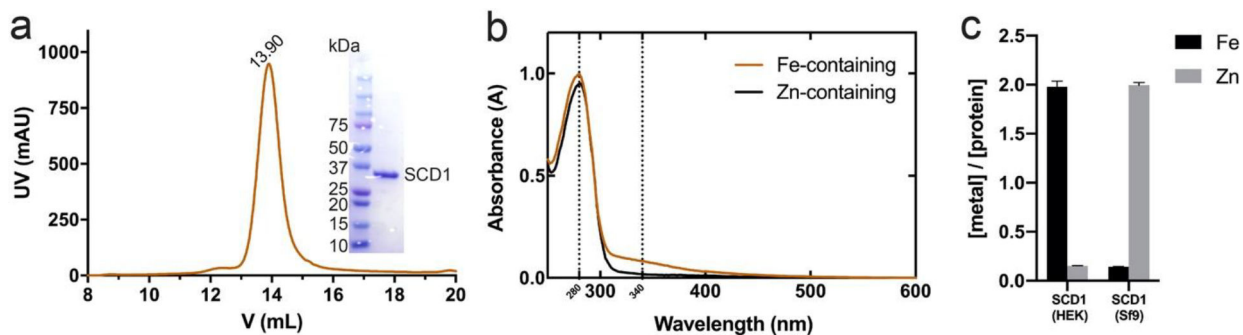


Figure 3.

Purification and characterization of HEK-expressed mouse SCD1. **a.** Size exclusion chromatography (SEC) profile and SDS-PAGE gel of the purified mouse SCD1. **b.** UV/Vis spectra of iron-containing and zinc-containing SCD1. SCD1 with the diiron center showed a characteristic broad peak around 340 nm. Vertical dash lines mark the absorbance at 280 nm and 340 nm. **c.** ICP-MS results of SCD1 expressed in HEK or Sf9 cells. The molar ratio of iron to protein in HEK-expressed SCD1 is 1.98 ± 0.06 .

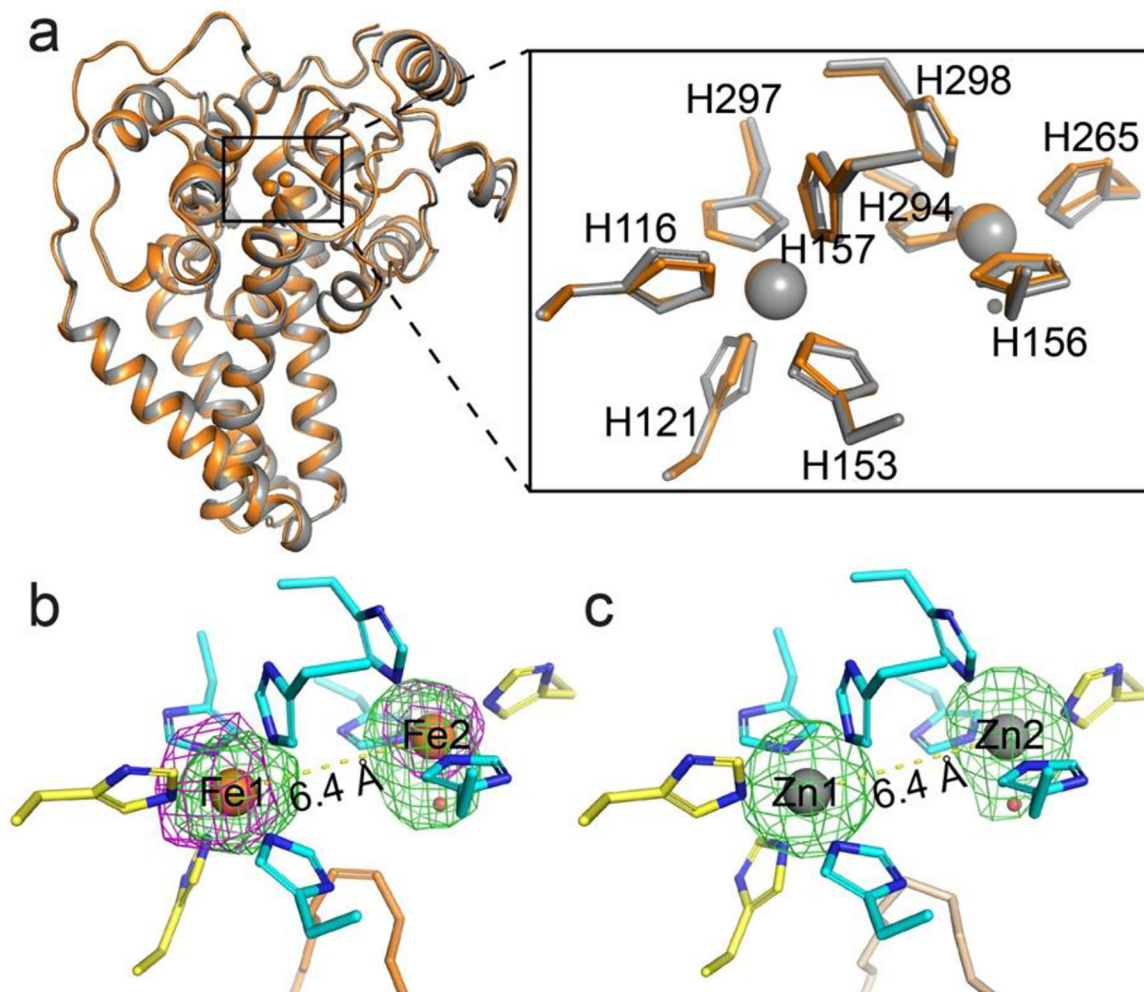


Figure 4. Crystal structure of iron-containing mouse SCD1. **a.** Structural alignment of iron-containing (orange) and zinc-containing (gray) SCD1. Inset, alignment of the dimetal center and the coordinating histidine residues. **b.** The reaction center is occupied by a diiron center with a unique coordination scheme that is preserved in the structure. The iron ions (orange) and a water molecule (red) are shown as spheres. Nine histidine ligands (residues from TMs in yellow; from the soluble domain in cyan) and the oleoyl-CoA (orange) are shown as sticks. Green mesh, $F_o - F_c$ map with iron ions omitted contoured at 3σ ; magenta mesh, anomalous difference map at 3.5σ . **c.** Metal coordination in the zinc-containing structure (PDB ID: 4YMK). An elongated density is also observed at Zn2. The stearoyl-CoA are shown as wheat stick. Green mesh, $F_o - F_c$ map with zinc ions omitted contoured at 4.0σ .

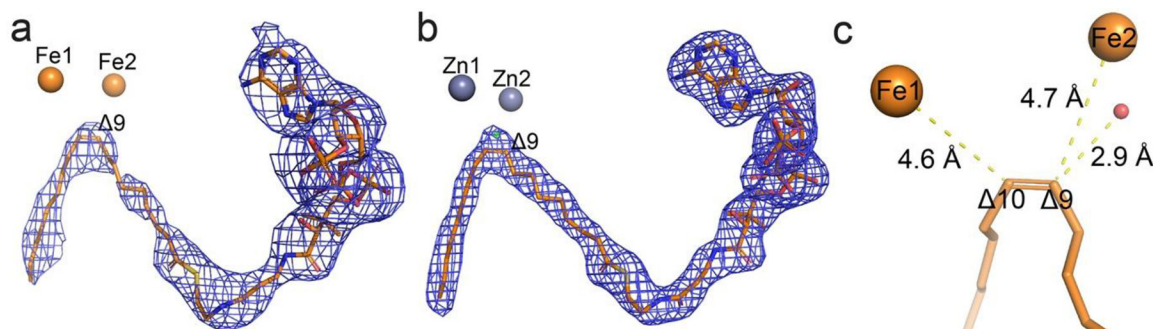


Figure 5.

An oleoyl-CoA is bound to the SCD1. **a.** The oleoyl-CoA molecule (orange stick) fits well in the iron-containing structure. Blue mesh, $2F_o - F_c$ map at 1.0σ . **b.** An oleoyl-CoA (orange stick) was modeled to into the non-protein electron density in the previous zinc-containing structure (PDB ID: 4YMK). Blue mesh, $2F_o - F_c$ map at 1.5σ ; green mesh, $F_o - F_c$ map at 4.0σ . The extra density is present whether an oleoyl-CoA or a stearoyl-CoA was modeled. **c.** Distances between the two irons and the $\Delta 9$ and $\Delta 10$ carbons on the oleoyl-CoA.

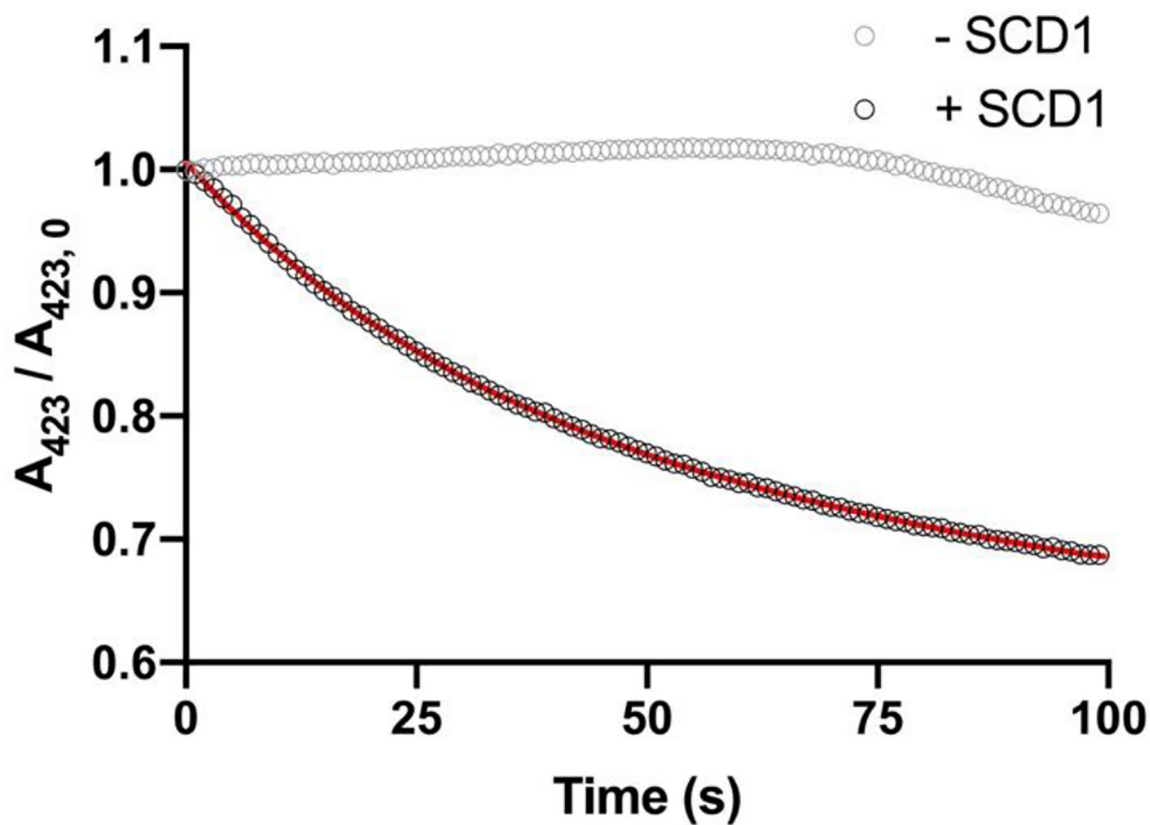


Figure 6. Electron transfer between cyt b_5 and SCD1. Ferrous cyt b_5 can be oxidized by diferric SCD1 as followed by Soret peak at 423 nm (black circle). A double exponential decay model was fitted to the data as shown in red line. Cyt b_5 remained in its reduced state without SCD1 in anaerobic environment (gray circle).

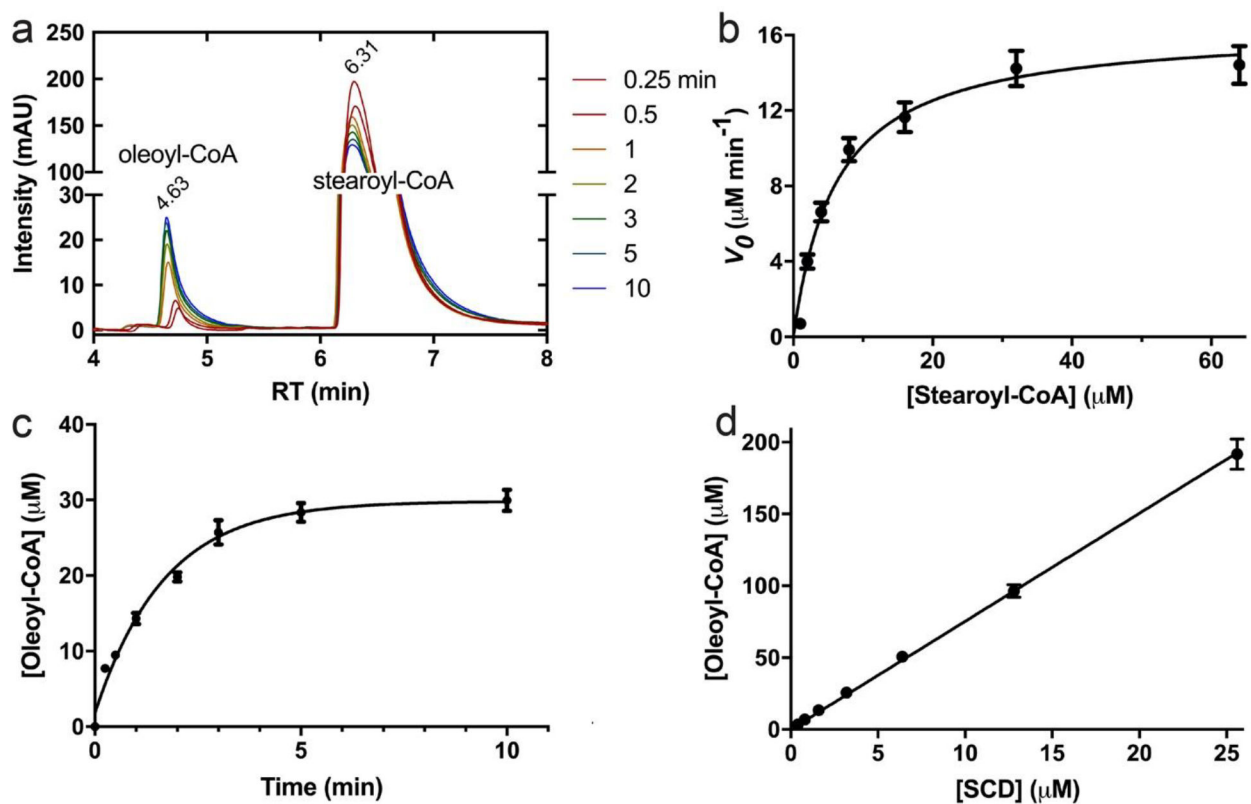


Figure 7.

Continuous turnover reaction of SCD1. **a.** HPLC profiles of reaction products in the assembled electron transfer chain including SCD1, cyt b_5 and b_5R , quenched at different time points (left). Increasing peak of oleoyl-CoA and simultaneous decreasing peak of stearoyl-CoA confirmed stearoyl-CoA to oleoyl-CoA conversion. **b.** Michaelis-Menten kinetics of the steady-state enzymatic reaction quantified by initial rates of oleoyl-CoA production before reaching product inhibition. All the error bars stand for experiments of three repeats. **c.** The time course of oleoyl-CoA production showed a profound product inhibition as most of the excess stearoyl-CoA was not reacted (>90%). **d.** Linear correlation of final [product] with the [SCD1] in the presence of excess substrate.