

HHS Public Access

Chem Commun (Camb). Author manuscript; available in PMC 2021 August 27.

Published in final edited form as:

Author manuscript

Chem Commun (Camb). 2020 August 27; 56(69): 9998-10001. doi:10.1039/d0cc04011h.

Natural-abundance ¹⁷O NMR spectroscopy of magnetically aligned lipid nanodiscs

Thirupathi Ravula^a, Bikash Sahoo^a, Xiaofeng Dai^a, Ayyalusamy Ramamoorthy^a

^a·Biophysics and Department of Chemistry, University of Michigan, Ann Arbor, MI 48109-1055, USA.

Abstract

Natural-abundance ¹⁷O NMR experiments are used to investigate the hydrated water in magnetically alligned synthetic polymer based lipid-nanodiscs. Residual quadrupole couplings (RQCs) measured from the observed five ¹⁷O (central and satellite) transitions, and molecular dynamics simulations, are used to probe the ordering of water moecules across the lipid bilayer.

Graphical Abstract



Water is central to all life, and crucial for the structure and function of all biological systems. Study of hydration dynamics can provide valuable insights into the role of water in the function of such biological systems including protein, RNA and biomaterials.^{1, 2} For example, water significantly influence the structure, dynamics and the variety of functional properties of the cell membrane.^{2, 3} However, there are numerous challenges for high-resolution probing of the structure and dynamics of water molecules in a membrane environment. Herein, we demonstrate the use of natural-abundance ¹⁷O NMR spectroscopy to investigate the interaction of water molecules with lipid-nanodiscs.

Ayyalusamy Ramamoorthy, ramamoor@umich.edu.

Conflicts of interest

There are no conflicts to declare.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here].

While the very low-abundance ($\sim 0.037\%$) and a small gyromagnetic ratio ($\sim 1/7$ th of ¹H) render ¹⁷O NMR to be insensitive, magic angle spinning (MAS) studies on solids well utilized the large chemical shift span and quadrupolar interaction of ¹⁷O.^{4, 5} ¹⁷O quadrupole central transition has also been used to study chemical and biological macromolecules in solution as reported in the literature.^{4, 6–9} In this study, polymer nanodiscs are used as membrane mimetics to investigate the interaction of water with lipid bilayer using naturalabundance ¹⁷O NMR spectroscopy. Polymer nanodiscs consists of a planar lipid bilayer encased by the polymer belt (Fig.1A).^{10, 11} Lipid nanodiscs technology is increasingly used for the structure and functional studies of membrane proteins.^{12, 13} While isotropic nanodiscs are used in solution NMR studies, the magnetic-alignment properties of polymer nanodiscs (diameter >20 nm) have been shown to be unique in enabling the application of solid-state NMR experiments to study lipids and membrane proteins, and to measure RDCs (residual dipolar couplings) using solution NMR studies.^{14–17} Natural-abundance ¹⁷O NMR spectra of magnetically-aligned and flipped nanodiscs are remarkable in showing all the five transitions for the spin 5/2 system, and the exciting feasibility of using the experimentally measured residual quadrupolar couplings (RQCs) to determine the orientation of the water molecules in the lipid bilayer. A combination of NMR and molecular dynamic simulation presented here reveals the molecular level ordering of water molecules across the lipid bilayer.

Nanodiscs were prepared by the addition of a synthetic polymer (styrene maleimide quaternary ammonium (SMA-QA); synthesis, purification and characterization can be found elsewhere¹⁵) and DMPC (1,2-dimyristoyl-sn-glycero-3-phosphocholine) lipids as described in the Supporting Information, and characterized using biophysical experiments as reported previously.¹⁸ The polymer nanodiscs prepared in various concentrations and their ability to magnetically align in the presence of an external magnetic field were confirmed using ³¹P NMR as reported previously.^{16, 18} In addition, TEM images were used to confirm the approximate size and shape of the nanodiscs (see Figure 1B). ¹⁷O NMR experiments on varying concentrations of SMA-QA-DMPC nanodiscs were performed at 310 K, and the acquired spectra are shown in Figure 1(C). At a low lipid concertation (10 % w/v), a single peak corresponding to the isotropic chemical shift frequency of water was observed. The scalar coupling between ¹⁷O and ¹H resulting in a triplet for isotropic water is not observed for water in the nanodiscs sample as the observed line is broader (line width is ~167 Hz) than the scalar coupling (\sim 78 Hz). When the lipid concertation was increased to 20% w/y, the line-width of the observed ¹⁷O peak increased to ~606 Hz and also showed a triplet pattern. Further increase in the concentration of nanodiscs to 30% w/v showed a wellresolved pentet pattern as shown in Fig.1C. Motion of water molecules in the nanodiscs sample averaged the ¹⁷O quadrupole coupling to a smaller value (called RQCs) such that all the five (one central and four satellite) transitions observed in the form of a pentet pattern. The RQC values are obtained by measuring the frequency difference between any of the two adjacent peaks from the pentet signal; a value of \sim 492 Hz is measured for 30% w/v nanodiscs. These results suggest that water in magnetically-aligned nanodiscs is partially aligned.

To measure the effect of temperature on the alignment of water molecules in nanodiscs, ¹⁷O spectra were recorded as a function of temperature (Fig.1D). As demonstrated using ³¹P and

¹⁴N NMR experiments, the nanodiscs are isotropic at temperatures below the main gel to liquid crystalline phase transition temperature of DMPC ($T_m \sim 24$ °C) and magnetically align above T_m.¹⁶ It is remarkable that ¹⁷O RQC values observed for water in nanodiscs reveal the magnetic-alignment of nanodiscs as shown in Fig.1D. ¹⁷O spectra show an isotropic peak below T_m at 295 and 300 K with line-widths <200 Hz, while increasing temperature resulted in line-broadening near T_m (see the spectrum at 305 with line-width ~508 Hz) exhibiting a poorly resolved splitting from the motionally-averaged RQC from nanodsics. On the other hand, a multiplet pattern is observed above T_m as shown for the spectrum at 310 K with an RQC of ~492 Hz. Further increase in the temperature to 315 K, increased the observed quadrupolar coupling to ~556 Hz. These observations clearly demonstrate the use of ¹⁷O NMR spectra to probe the water molecules ordering using magnetic-alignment of nanodiscs. To gain further insights into the dynamics of water, we measured the longitudinal relaxation (T_1) times for all five ¹⁷O peaks observed (Figure 3). The T_1 values for water molecules in the nanodiscs are in the range of 4 ms, and much smaller than that for bulk water (~9 ms, Fig.S2), suggesting the restricted motion of water molecules in nanodiscs under these conditions.

Since the observed RQC depends on the order parameter as well as on the orientation of bilayer-associated water molecules, a single ¹⁷O ROC value obtained from aligned nanodiscs is not sufficient to completely describe the alignment of water molecules. A complete description of the alignment of water molecules requires RDC values measured under different alignment conditions. To measure the RQCs of ¹⁷O under different alignment conditions, we used a paramagnetic salt, YbCl₃, to flip the orientation of magneticallyaligned nanodiscs as described in our previous study.^{16 31}P and ¹⁴N NMR experiments were used to confirm the flipping of the aligned nanodiscs upon the addition of YbCl₃. The effect of different paramagnetic salts on the NMR spectra and relaxation parameters of nanodiscs is reported in our previous study.¹⁹ The observed chemical shift for ³¹P (from -13.6 ± 0.5 ppm to ~22.7 \pm 3.7 ppm upon flipping) and doubling of ¹⁴N ROC (from ~8.9 \pm 0.6 kHz to ~16.8 \pm 0.7 kHz upon flipping) confirmed the 90° flipping of nanodiscs to result in a parallel orientation of the lipid bilayer normal to the external magnetic field direction (Figure 2). ¹⁷O NMR spectra of water in the flipped nanodiscs showed a significant increase in the RQC value: from 492 Hz to 690 Hz upon flipping as shown in Figure 2(A and B). ¹⁷O NMR spectra of varying concentrations of flipped and unflipped nanodiscs are shown in Fig.S1 along with deconvoluted spectra illustrating the multiplet pattern and ROC values. While it is remarkable that the motionally averaged water RQCs are sensitive to the alignment of nanodiscs, the 90° flipping of aligned nanodiscs did not follow $(3\cos^2\phi - 1)$ for the observed ¹⁷O RQCs unlike ³¹P and ¹⁴N spectra; where ϕ is the angle between the bilayer normal and the magnetic field axis. The observed changes for ³¹P and ¹⁴N are because the axis of motional averaging for lipids (collinear with the bilayer normal) is oriented perpendicular (Figure 2E) and parallel (2J) to the magnetic field axis. On the other hand, the bilayerassociated water molecules are in exchange with the free water molecules and are randomly oriented with respect to the bilayer normal. Even though the bilayer-associated water molecules prefer to orient with their symmetry (or the bisector) axes collinear with the bilayer normal as reported for zwitterionic lipids,²⁰ the observed RQC values indicate that the motional averaging of the ¹⁷O quadrupolar coupling is different in flipped and unflipped

nanodiscs, which is more pronounced when the concentration decreases (Fig.S1). Therefore, ¹⁷O RQCs can be useful to probe the molecular events at the membrane-water interface.

To explore the origins of an increase in the quadrupolar coupling upon 90° flipping of the nanodiscs, we used MD simulation using DMPC and water as a system (Fig.4A); additional details are given in the supporting information. The main axis of the ¹⁷O quadrupolar coupling tensor is perpendicular to the molecular plane of water as reported elsewhere.²¹ To obtain the relative orientations and average RQCs for the quadrupole coupling tensor across the lipid bilayer, $\cos\theta$ values were calculated. MD simulations show a gradual decrease in water density across the bilayer (i.e. along the z-axis) and becomes zero in the hydrophobic core of the bilayer (Fig.4B). With the DMPC bilayer placed along the xy-plane, space and time-averaged $<3\cos^2\theta-1>$ values were calculated (Fig.4C). The resulting $<3\cos^2\theta-1>$ values are positive along the z-axis and negative along x and y axes. The larger $<3\cos^2\theta-1>$ magnitude along the z-axis suggest that the motional averaging along x and y axes more effectively reduces the RQCs than that along the z-axis. These results are in agreement with the observed larger RQC values for the flipped-nanodiscs, which have the bilayer-normal (i.e. the z-axis in Fig.4) along the magnetic field axis (Fig.2J). Thus, the measured ¹⁷O RQCs can be used to probe the ordering and dynamics of water molecules associated with a lipid bilayer.

In conclusion, we have demonstrated that natural-abundance ¹⁷O NMR is an useful technique to study the magnetic-alignment of lipid-nanodiscs and also to probe membrane water interaction. At optimum hydration levels under the magnetic-alignment of nanodiscs, all five transitions of ¹⁷O nuclei are observed. The measured RQCs are explained based on the orientation preference of bilayer-associated water molecules. Overall, the reported results clearly indicate the molecular level ordering of water across the bilayer, and a combination of ¹⁷O NMR and MD simulation can be used to further probe water-membrane interactions. The reported ¹⁷O NMR experiments can also be used to investigate other aligned samples such as other types of nanodsics^{22, 23}, bicelles^{24–28}, and other alignment media used in NMR studies²⁹. We also expect the reported approach and results to be useful in the investigation of the role of ordered water in many biological processes including protein folding, misfolding, amyloid aggregation, and biomineralization process.^{30–32} The use of dynamic nuclear polarization experiments,^{33, 34} to enhance ¹⁷O NMR sensitivity for studies on aligned samples^{35, 36} would be exciting.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

Acknowledgment

This study was supported by NIH (GM084018 to AR). We thank Dr. Ganapathy, a visiting researcher in the Ramamoorthy lab, for help with initial $^{17}\mathrm{O}$ NMR experiments.

Notes and references

1. Ball P, Chem Rev, 2008, 108, 74–108. [PubMed: 18095715]

2. Chaplin M, Nat Rev Mol Cell Biol, 2006, 7, 861-866. [PubMed: 16955076]

- 3. Disalvo EA, Lairion F, Martini F, Tymczyszyn E, Frias M, Almaleck H and Gordillo GJ, Biochim Biophys Acta, 2008, 1778, 2655–2670. [PubMed: 18834854]
- 4. Castiglione F, Mele A and Raos G, in Annual Reports on NMR Spectroscopy, ed. Webb GA, Academic Press, 2015, vol. 85, pp. 143–193.
- McMahon MT, deDios AC, Godbout N, Salzmann R, Laws DD, Le H, Havlin RH and Oldfield E, Journal of the American Chemical Society, 1998, 120, 4784–4797.
- Keeler EG, Michaelis VK, Wilson CB, Hung I, Wang X, Gan Z and Griffin RG, J Phys Chem B, 2019, 123, 3061–3067. [PubMed: 30882222]
- 7. Wu G, Prog Nucl Magn Reson Spectrosc, 2019, 114–115, 135–191.
- 8. Kouril K, Meier B, Alom S, Whitby RJ and Levitt MH, Faraday Discuss, 2018, 212, 517–532. [PubMed: 30238100]
- Rezaei-Ghaleh N, Munari F, Becker S, Assfalg M and Griesinger C, Chem Commun (Camb), 2019, 55, 12404–12407. [PubMed: 31566648]
- Ravula T, Hardin NZ and Ramamoorthy A, Chem Phys Lipids, 2019, 219, 45–49. [PubMed: 30707909]
- Orwick MC, Judge PJ, Procek J, Lindholm L, Graziadei A, Engel A, Grobner G and Watts A, Angew Chem Int Ed Engl, 2012, 51, 4653–4657. [PubMed: 22473824]
- Dorr JM, Koorengevel MC, Schafer M, Prokofyev AV, Scheidelaar S, van der Cruijsen EA, Dafforn TR, Baldus M and Killian JA, Proc Natl Acad Sci U S A, 2014, 111, 18607–18612. [PubMed: 25512535]
- Lee SC, Knowles TJ, Postis VL, Jamshad M, Parslow RA, Lin YP, Goldman A, Sridhar P, Overduin M, Muench SP and Dafforn TR, Nat Protoc, 2016, 11, 1149–1162. [PubMed: 27254461]
- 14. Ravula T and Ramamoorthy A, Angew Chem Int Ed Engl, 2019, 58, 14925–14928. [PubMed: 31310700]
- Ravula T, Hardin NZ, Ramadugu SK, Cox SJ and Ramamoorthy A, Angew Chem Int Ed Engl, 2018, 57, 1342–1345. [PubMed: 29232017]
- 16. Ravula T, Kim J, Lee DK and Ramamoorthy A, Langmuir, 2020, 36, 1258–1265.
- 17. Radoicic J, Park SH and Opella SJ, Biophys J, 2018, 115, 22–25. [PubMed: 29914645]
- Ravula T, Ramadugu SK, Di Mauro G and Ramamoorthy A, Angew Chem Int Ed Engl, 2017, 56, 11466–11470. [PubMed: 28714233]
- Di Mauro GM, Hardin NZ and Ramamoorthy A, Biochim Biophys Acta Biomembr, 2020, 1862, 183332. [PubMed: 32360741]
- 20. Cheng JX, Pautot S, Weitz DA and Xie XS, Proc Natl Acad Sci U S A, 2003, 100, 9826–9830. [PubMed: 12904580]
- 21. Yazyev OV and Helm L, J Chem Phys, 2006, 125, 054503. [PubMed: 16942222]
- Nasr ML, Baptista D, Strauss M, Sun ZJ, Grigoriu S, Huser S, Pluckthun A, Hagn F, Walz T, Hogle JM and Wagner G, Nat Methods, 2017, 14, 49–52. [PubMed: 27869813]
- 23. Denisov IG and Sligar SG, Chem Rev, 2017, 117, 4669–4713. [PubMed: 28177242]
- 24. Lange OF, Lakomek NA, Fares C, Schroder GF, Walter KF, Becker S, Meiler J, Grubmuller H, Griesinger C and de Groot BL, Science, 2008, 320, 1471–1475. [PubMed: 18556554]
- Sanders CR, Hare BJ, Howard KP and Prestegard JH, Progress in Nuclear Magnetic Resonance Spectroscopy, 1994, 26, 421–444.
- 26. Tjandra N and Bax A, Science, 1997, 278, 1111–1114. [PubMed: 9353189]
- 27. Prestegard JH, Bougault CM and Kishore AI, Chem Rev, 2004, 104, 3519–3540. [PubMed: 15303825]
- Stelzer AC, Frank AT, Bailor MH, Andricioaei I and Al-Hashimi HM, Methods, 2009, 49, 167– 173. [PubMed: 19699798]
- 29. Chen K and Tjandra N, in NMR of Proteins and Small Biomolecules, ed. Zhu G, Springer Berlin Heidelberg, Berlin, Heidelberg, 2012, DOI: 10.1007/128_2011_215, pp. 47–67.
- 30. Wang T, Jo H, DeGrado WF and Hong M, J Am Chem Soc, 2017, 139, 6242–6252. [PubMed: 28406028]
- 31. Thirumalai D, Reddy G and Straub JE, Acc.Chem Res, 2012, 45, 83–92. [PubMed: 21761818]

- 32. Rodriguez Camargo DC, Korshavn KJ, Jussupow A, Raltchev K, Goricanec D, Fleisch M, Sarkar R, Xue K, Aichler M, Mettenleiter G, Walch AK, Camilloni C, Hagn F, Reif B and Ramamoorthy A, Elife, 2017, 6, e31226. [PubMed: 29148426]
- 33. Su Y, Andreas L and Griffin RG, Annu Rev Biochem, 2015, 84, 465–497. [PubMed: 25839340]
- 34. Carnahan SL, Lampkin BJ, Naik P, Hanrahan MP, Slowing II, VanVeller B, Wu G and Rossini AJ, J Am Chem Soc, 2019, 141, 441–450. [PubMed: 30525547]
- 35. Salnikov E, Rosay M, Pawsey S, Ouari O, Tordo P and Bechinger B, J Am Chem Soc, 2010, 132, 5940–5941. [PubMed: 20392100]
- 36. Jakdetchai O, Denysenkov V, Becker-Baldus J, Dutagaci B, Prisner TF and Glaubitz C, J Am Chem Soc, 2014, 136, 15533–15536. [PubMed: 25333422]





(A) Schematic representation of SMA-QA polymer based lipid-nanodiscs. B) TEM image of SMA-QA:DMPC (0.5:1 w/w) nanodiscs. ¹⁷O NMR spectra of water from SMA-QA:DMPC (0.5:1 w/w) nanodiscs (~25 nm diameter) for varying lipid concentration at 310 K (C) and at the indicated temperatures for 30% w/v lipid concentration (D). All other details on the sample preparation, characterization, and experimental conditions are given in the supporting information.





¹⁷O (A, B, F and G), ³¹P (C and H) and ¹⁴N (D and I) NMR spectra of magnetically-aligned SMA-QA:DMPC nanodiscs obtained at 310 K with the lipid-bilayer-normal perpendicular (top row, as illustrated in E) and parallel (bottom row, as illustrated in J) to the external magnetic field direction. A 2 mM YbCl₃ was used to flip the nanodiscs (bottom row). All spectra were acquired from 20% w/v lipid concentration, except that 30% w/v was used for A and F. Deconvolution of ¹⁷O pentet spectra of aligned and flipped nanodiscs (20% and 30% w/v) are shown in Figure S1.



Figure 3. T₁ relaxation of ¹⁷O NMR transitions.

¹⁷O spin-lattice (T₁) relaxation for the central (CT) and satellite transitions (ST1, ST2, ST3, ST4) measured from SMAQA-DMPC (30% w/v) nanodiscs containing a 2 mM YbCl₃. ¹⁷O NMR spectrum labelled with the transitions and the measured T₁ values (inset). T₁ values were obtained from the best-fitting of the inversion-recovery intensities (data points). See Figure S2 for ¹⁷O T₁ experimental data obtained from isotropic water, and Figure S3 for ¹⁷O inversion recovery NMR spectra of water present in nanodiscs.



Figure 4.

(A) Selected region of the DMPC:water system (consisting of 256 DMPC and 10153 water molecules) used in molecular dynamic simulations; additional details can be found in the supporting information (Figs.S4 and S5). (*x*, *y*, *z*) coordinates are shown with the *z*-axis parallel to the lipid bilayer normal. (B) Total number of water molecules present in the *xy* plane plotted along the *z*-coordinate (i.e. lipid-bilayer normal) throughout the simulation time. (C) Space and time-averaged $<3Cos^2\theta-1>$ as a function of the *z*-coordinate; where θ is the angle between the normal to the water plane (i.e. the main principal axis of ¹⁷O quadrupole coupling tensor) and *x* (blue), *y* (red), and *z* (black) axes. Space averaging was done for all water molecules present in the *xy* plane for each point along the *z*-axis. The error bar represents the slandered deviation from the average value. The large error bars are due to less water molecules present in the core of the bilayer. See the supporting information for additional details on the simulations used in this study.