



The Heterotrophic Bacterium *Cupriavidus pinatubonensis* JMP134 Oxidizes Sulfide to Sulfate with Thiosulfate as a Key Intermediate

Yufeng Xin,^{a,b} Rui Gao,^a Feifei Cui,^a Chuanjuan Lü,^a Honglei Liu,^a Huaiwei Liu,^a Yongzhen Xia,^a 💿 Luying Xun^{a,c}

^aState Key Laboratory of Microbial Technology, Shandong University, Qingdao, People's Republic of China ^bCollege of Life Sciences, Qufu Normal University, Qufu, People's Republic of China ^cSchool of Molecular Biosciences, Washington State University, Pullman, Washington, USA

ABSTRACT Heterotrophic bacteria actively participate in the biogeochemical cycle of sulfur on Earth. The heterotrophic bacterium *Cupriavidus pinatubonensis* JMP134 contains several enzymes involved in sulfur oxidation, but how these enzymes work together to oxidize sulfide in the bacterium has not been studied. Using genedeletion and whole-cell assays, we determined that the bacterium uses sulfide:quinone oxidoreductase to oxidize sulfide to polysulfide, which is further oxidized to sulfite by persulfide dioxygenase. Sulfite spontaneously reacts with polysulfide to produce thiosulfate. The sulfur-oxidizing (Sox) system oxidizes thiosulfate to sulfate. Flavocytochrome *c* sulfide dehydrogenase enhances thiosulfate oxidation by the Sox system but couples with the Sox system for sulfide oxidation to sulfate in the absence of sulfide:quinone oxidoreductase. Thus, *C. pinatubonensis* JMP134 contains a main pathway and a contingent pathway for sulfide oxidation.

IMPORTANCE We establish a new pathway of sulfide oxidation with thiosulfate as a key intermediate in *Cupriavidus pinatubonensis* JMP134. The bacterium mainly oxidizes sulfide by using sulfide:quinone oxidoreductase, persulfide dioxygenase, and the Sox system with thiosulfate as a key intermediate. Although the purified and reconstituted Sox system oxidizes sulfide, its rate of sulfide oxidation in *C. pinatubonensis* JMP134 is too low to be physiologically relevant. The findings reveal how these sulfur-oxidizing enzymes participate in sulfide oxidation in a single bacterium.

KEYWORDS sulfide, sulfane sulfur, thiosulfate, sulfate, heterotrophic bacteria, sulfur oxidation pathway, sulfate reduction

Sulfur oxidation is a key step in the biogeochemical cycling of sulfur on Earth. Under anoxic conditions, sulfur-reducing bacteria use sulfate as the terminal electron acceptor for the oxidation of organic compounds, generating hydrogen sulfide (H₂S), which can be oxidized back to sulfate by chemolithoautotrophic bacteria under oxic conditions or used as the reducing power for photosynthesis by phototrophic bacteria under anoxic conditions (1, 2). Further, many heterotrophic bacteria are also capable of oxidizing H₂S (3, 4). H₂S is a common product of the microbial metabolism of L-cysteine, and heterotrophic bacteria carrying sulfide: quinone oxidoreductase (SQR) and persulfide dioxygenase (PDO) oxidize self-produced H₂S or exogenously introduced H₂S to sulfite and thiosulfate (3, 5, 6). *Pseudomonas putida* oxidizes thiosulfate to tetrathionate (7, 8), *Hyphomicrobium denitrificans* oxidizes thiosulfate to tetrathionate and sulfate (9), and other bacteria, including bacteria isolated from marine sediments and hydrothermal vents, oxidize thiosulfate to sulfate to sulfate to sulfate to tetrathionate (7, 8), *Hyphomicrobium denitrificans* oxidizes thiosulfate to tetrathionate and sulfate (9), and other bacteria, including bacteria isolated from marine sediments and hydrothermal vents, oxidize thiosulfate to sulfate to sulfate to sulfate to sulfate to tetrathionate (7, 8) and persulfate to tetrathionate and sulfate (9), and other bacteria, including bacteria isolated from marine sediments and hydrothermal vents, oxidize thiosulfate to oxidize sulfide to sulfate. **Citation** Xin Y, Gao R, Cui F, Lü C, Liu H, Liu H, Xia Y, Xun L. 2020. The heterotrophic bacterium *Cupriavidus pinatubonensis JM*P134 oxidizes sulfide to sulfate with thiosulfate as a key intermediate. Appl Environ Microbiol 86:e01835-20. https://doi.org/10.1128/AEM .01835-20.

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Address correspondence to Yongzhen Xia, xiayongzhen2002@sdu.edu.cn, or Luying Xun, luying_xun@vetmed.wsu.edu.

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Accepted 2 September 2020 Accepted manuscript posted online 11 September 2020 Published 28 October 2020 The soil bacterium *Cupriavidus pinatubonensis* (formerly *Ralstonia eutropha* and *Alcaligenes eutrophus*) JMP134 has been widely used to study the microbial degradation of aromatic compounds (11, 12). It also oxidizes H_2S during normal growth on organic compounds (3), offering an opportunity to investigate how the heterotrophic bacterium oxidizes H_2S . When its SQR and PDO genes are introduced into *Escherichia coli*, the recombinant cells oxidize H_2S to sulfite with polysulfide as an intermediate, and the sulfite spontaneously reacts with polysulfide to produce thiosulfate (5). From genome sequence analysis, *C. pinatubonensis* JMP134 also contains genes coding for the flavocytochrome *c* sulfide dehydrogenase (FCSD) system, the sulfur-oxidizing (Sox) system, and sulfite:cytochrome *c* oxidoreductase (SOR).

The FCSD system consists of a flavin adenine dinucleotide-containing protein (FccB) and a cytochrome *c* (FccA), which are soluble proteins in the periplasmic space. FccB and FccA are also called SoxF and SoxE, respectively, in chemolithoautotrophic bacteria, as their genes are often clustered with other *sox* genes (13). FCSD uses cytochrome *c* as the electronic acceptor, while SQR uses ubiquinone. FCSD was first identified in the purple photosynthetic bacterium *Allochromatium vinosum* (14), and it has subsequently been found to be widely present in autotrophic bacteria (15) and in heterotrophic bacteria (16). FCSD from *Cupriavidus necator* H16 oxidizes sulfide to polysulfide, which is further oxidized by PDO to sulfite and thiosulfate when both are cloned in *Pseudomonas aeruginosa* (16). Since *C. pinatubonensis* JMP134 contains both SQR and FCSD, it is unclear which one is preferentially used to oxidize H₂S.

The Sox system is a versatile enzyme system for the oxidation of reduced sulfur species, and the Sox proteins are soluble in the periplasmic space (17). The most extensively studied Sox system is from Paracoccus pantotrophus GB17 (ATCC 35512^T or DSM 2944), a facultative chemolithotroph able to growth on thiosulfate (17). The core enzyme of the Sox system consists of four enzymes, SoxYZ (a heterodimer of SoxY and SoxZ), SoxXA, SoxCD, and SoxB, encoded by seven sox genes. How the Sox system oxidizes thiosulfate has been characterized. SoxYZ is a carrier protein with the sulfur molecule to be oxidized being covalently linked to a conserved cysteine residue of SoxY (18, 19). SoxXA catalyzes the loading of thiosulfate to SoxYZ-SH, producing SoxYZ-Sthiosulfate (20). SoxB hydrolytically removes the terminal sulfonate group, producing SoxYZ-SSH and sulfate (21). SoxYZ-SSH is oxidized by SoxCD to SoxYZ-S-SO₃ $^{-}$ (22, 23), and the sulfonate group is again released by SoxB. The electrons from sulfur oxidation enter the electron transfer chain via a cytochrome c-type cytochrome (24). The purified and reconstituted Sox system also oxidizes sulfide, elemental sulfur, and sulfite to sulfate (24), but it has not been confirmed whether P. pantotrophus or any other bacteria use the Sox system to oxidize these sulfur species.

SOR consists of two proteins, SorA and SorB. SorA is a large protein containing molybdopterin, and SorB is a small protein containing cytochrome *c*. SOR, located in the periplasmic space, oxidizes sulfite to sulfate (25). It is unknown if SOR oxidizes the sulfite generated by PDO to sulfate in *C. pinatubonensis* JMP134.

In this study, we investigated how *C. pinatubonensis* JMP134 uses SQR/PDO, FCSD, the Sox system, and SOR to oxidize sulfide to sulfate. On the basis of genomic analysis and our experimental data, we identified a new pathway of H_2S oxidation in which SQR and PDO collectively oxidize sulfide to thiosulfate and then the Sox system oxidizes thiosulfate to sulfate. FCSD's main function was to enhance thiosulfate oxidation by the Sox system; however, FCSD oxidized H_2S to zero-valent sulfur, part of which was directly oxidized by the Sox system when *sqr* was deleted in *C. pinatubonensis* JMP134.

RESULTS

Sulfur-oxidizing genes in *C. pinatubonensis* **JMP134.** The genome of *C. pinatubonensis* JMP134 consists of two chromosomes, chromosome A (3.8 Mb) and chromosome B (2.72 Mb), and two large plasmids (11). The *sqr-pdo2* gene cluster, located on chromosome B (Fig. 1; see also Table S1 in the supplemental material), codes for four possible proteins, SQR, PDO, a possible sulfite exporter protein (TauE), and a gene regulator (FisR) (3, 26). Another *pdo* gene (*pdo1*) is located on chromosome A (Table S1)

Chromosome 1 (A)



SQR/PDO system

FIG 1 Schematic overview of the sulfur-oxidizing genes in *C. pinatubonensis* JMP134. The genome of *C. pinatubonensis* JMP134 includes two chromosomes, A and B. Genes encoding the SOR, Sox, and FCSD systems are located on chromosome A, and the *sqr-pdo* operon is on chromosome B. The SOR system is encoded by *sorA* (GenBank accession number AAZ62443.1) and *sorB* (GenBank accession number AAZ62442.1). The Sox system genes include *soxB* (GenBank accession number AAZ62608.1), *soxX* (GenBank accession number AAZ62610.1), *soxA* (GenBank accession number AAZ62611.1), *soxZ* (GenBank accession number AAZ62613.1), *soxY* (GenBank accession number AAZ62614.1), *soxD* (GenBank accession number AAZ62611.1), *soxZ* (GenBank accession number AAZ62617.1). The FCSD system is encoded by *fccB* (GenBank accession number AAZ62620.1) and *fccA* (a possible cytochrome *c*; GenBank accession number AAZ62621.1). An operon coding for the SQR/PDO system genes consists of *sqr* (GenBank accession number AAZ62949.1). The locus tag of each gene is given below the gene (e.g., Reut_A3252 is the tag for *soxA*).

and encodes a type I PDO (27). The sox genes on chromosome A consist of seven sox genes (soxB, soxX, soxA, soxZ, soxY, soxD, soxC), encoding the core components of the Sox system (Fig. 1; Table S1). Downstream (locus tags, Reut_A3261 and Reut_A3262) of the sox genes are soxF and soxE, which are also known as the fccB and fccA genes, respectively, coding for FccB and FccA of the FCSD system. The sequence similarities of FccB and FccA in C. pinatubonensis JMP134 to FccB (GenBank accession number CAJ94633.1) and FccA (GenBank accession number CAJ94634.1) in C. necator H16 were 85% (query cover, 99%) and 79% (query cover, 84%), respectively. In C. necator H16, which does not have SQR, the FCSD system oxidizes H_2S (16). SOR has two subunits, SorA and SorB, whose genes are also located on chromosome A (Fig. 1; Table S1). Genes coding for thiosulfate dehydrogenases, oxidizing thiosulfate to tetrathionate (28, 29), and tetrathionate hydrolase, converting tetrathionate to sulfate, sulfur, and thiosulfate (30), were not found in C. pinatubonensis JMP134. Genes involved in dissimilatory sulfur reduction, such as *qmoABC*, *aprAB*, and *sat* (31), were not found in the bacterium. The gene coding for a ribulose bisphosphate carboxylase, necessary for carbon fixation and chemolithoautotrophic growth, was not found in the bacterium either.

Deletion of genes involved in sulfur oxidation. Selected genes were deleted via homologous recombination by using a suicide plasmid carrying the DNA fragments before and after the gene. All the deletions were in frame, with about 0 to 10 amino acid residues at the N terminus and about 0 to 36 residues at the C terminus remaining in the mutants to avoid affecting downstream and upstream genes. All mutants grew similarly to the wild type when growing in Luria-Bertani (LB) medium or in a mineral medium (MM) with sulfate, sulfite, or thiosulfate as the sole sulfur source (data not shown), suggesting that these genes associated with sulfur oxidation are not essential for the bacterium during heterotrophic growth.

SQR was the primary enzyme responsible for sulfide oxidation. To investigate whether SQR, FCSD, or the Sox system was mainly responsible for sulfide oxidation in *C. pinatubonensis* JMP134, we tested the rate of sulfide oxidation by sulfide-induced cells of the wild type and the *sqr*, *fccB*, or *soxY-soxZ* deletion mutants. The $\Delta soxYZ$ mutant oxidized sulfide at a rate similar to that for the wild type (Table 1; Fig. 2), the $\Delta fccB$, Δsqr , Δsqr $\Delta fccB$, and Δsqr $\Delta fccB$ $\Delta soxYZ$ strains oxidized sulfide at 98%, 27%, 11%, and 8% of the wild-type rate, respectively (Table 1; Fig. 2). The slow sulfide oxidation by

	Oxidation rate (nmol/mg/min) ^b				
Strain	HS [_]	HS _n -	\$ ₂ 0 ₃ ²⁻		
JMP134	6.2 ± 0.1	11.8 ± 0.4	1.7 ± 0.1		
Δsqr	$1.7 \pm 0.4^{**}$	_	_		
ΔfccB	6.1 ± 0.2	_	$0.4\pm0.1^{*}$		
$\Delta fccB/fccB$	6.1 ± 0.5	_	$0.9\pm0.1^{*}$		
Δsqr Δpdo1 Δpdo2	$1.2 \pm 0.2^{**}$	_	_		
Δsqr Δpdo1 Δpdo2/sqr pdo2	6.1 ± 1.0	_	_		
$\Delta sqr \Delta pdo1 \Delta pdo2 \Delta soxYZ$	$1.6 \pm 0.7^{**}$	_	_		
$\Delta sqr \Delta fccB$	$0.7 \pm 0.1^{**}$	_	_		
$\Delta sqr \Delta fccB \Delta soxYZ$	$0.5 \pm 0.4^{**}$	_	_		
$\Delta soxYZ$	6.3 ± 1.0	10.8 ± 1.3	$0.2\pm0.1^{*}$		
$\Delta sox YZ/sox YZ$	6.4 ± 0.2	_	1.2 ± 0.3		
Δpdo1	6.1 ± 0.5	11.0 ± 0.6	_		
Δpdo2	$3.5 \pm 0.2^{**}$	$2.6\pm0.5^{**}$	_		
Δpdo1 Δpdo2	$2.2\pm0.4^{**}$	$1.3 \pm 0.2^{**}$	_		
Δpdo1 Δpdo2/pdo2	5.7 ± 0.2	$8.1 \pm 0.4^*$	_		
Δpdo1 Δpdo2 ΔsoxYZ	$1.9 \pm 0.3^{**}$	$0.7 \pm 0.1^{**}$	_		
ΔsorA	6.0 ± 0.9	_	_		
∆tauE	6.1 ± 0.2	_	_		
Buffer	$0.6\pm0.3^{**}$	_			

^{*a*}Cells were induced, harvested, and resuspended in 100 mM HEPES buffer (pH 7.4) at an OD of 2. The cell suspensions were used to oxidize 500 μ M sulfide, 800 μ M polysulfide, or 600 μ M thiosulfate at 30°C. The induction was with the corresponding substrate before harvesting.

^bFor HS⁻ and HS_n⁻, the rates were calculated with data obtained at 0.5 h by using the cell dry weight. For $S_2O_3^{2-}$, the rates were calculated with data obtained at 2 h by using the cell dry weight. When calculating all rates, the rate for the heat-inactivated cell as a control was deducted. The rates of sulfide, polysulfide (HS_n⁻), and thiosulfate oxidation of heat-inactivated cells were 6.0 ± 0.1 , 0.7 ± 0.1 , and 0 nmol/mg/min, respectively. —, not measured; *, the result for the mutant was different (P < 0.05) from that for the wild type; **, the result for the mutant was significantly different (P < 0.01) from that for the wild type. The data are the averages for at least three samples with standard deviations.

the $\Delta sqr \Delta fccB \Delta soxYZ$ mutant is likely due to nonspecific oxidation by other enzymes, as superoxide dismutase and catalase are able to oxidize sulfide (32, 33). The reduced rates of sulfide oxidation in the mutants were recovered when the deleted genes were complemented in *trans* on a plasmid (Table 1; Fig. S1). These results indicate that for sulfide oxidation, SQR is the main enzyme, FCSD plays a supplemental role, and the Sox system is only marginally active.

C. pinatubonensis JMP134 oxidizes sulfide to sulfate with sulfane sulfur and thiosulfate as detectable intermediates. To understand the pathway, we monitored the intermediates and products during sulfide oxidation by *C. pinatubonensis* JMP134. When 500 μ M NaHS was added to a suspension of sulfide-induced *C. pinatubonensis* JMP134 cells at an optical density at 600 nm (OD₆₀₀) of 2, about 100 μ M sulfide remained at 1 h (Fig. 2A) and the cells produced about 32 μ M sulfane sulfur (S⁰) (including polysulfide and elemental sulfur), 120 μ M thiosulfate, and 45 μ M sulfate (Fig. 2B to D). Sulfane and thiosulfate were further consumed and gradually decreased to almost zero (Fig. 2B and C), and the cell suspension produced 362 μ M sulfate at 7 h (Table 2; Fig. 2D). Sulfite was not detected during sulfide oxidation by the bacterium. The rate of sulfide oxidation was lower than that of polysulfide oxidation but higher than that of thiosulfate oxidation by *C. pinatubonensis* JMP134 (Table 1), reflecting the increased accumulation of thiosulfate during sulfide oxidation (Fig. 2).

SQR, PDO, and the Sox system sequentially oxidized sulfide to sulfate. Whether *C. pinatubonensis* JMP134 used PDO or the Sox system to oxidize the sulfane sulfur produced by SQR was further investigated by the construction of mutants with deletions of the corresponding genes. The deletion of *pdo1* did not affect sulfide oxidation, but the deletion of *pdo2* significantly slowed sulfide oxidation (Table 1). The $\Delta pdo1 \Delta pdo2$ mutant showed a further reduction in the sulfide oxidation rate (Table 1; Fig. 2A), with the elevated production of sulfane sulfur, the decreased production of thiosulfate, as well as the decreased production of sulfate (Table 2; Fig. 2B to D). This observation implies product inhibition: SQR oxidizes sulfide to sulfane sulfur, which



FIG 2 The function of SQR, PDO, the Sox system, and FccAB in *C. pinatubonensis* JMP134. Cells were harvested, washed, and resuspended in 100 mM HEPES buffer, pH 7.4, at an OD₆₀₀ of 2.0. Sulfide was added to 500 μ M to initiate the reaction. Sulfide (A), sulfane sulfur (B), thiosulfate (C), and sulfate (D) concentrations were determined. There was no apparent difference in the decrease in sulfide levels in HEPES buffer with or without heat-inactivated cells, but the heat-inactivated cells produced more sulfate (53 ± 13 μ M) than the buffer (5 ± 3 μ M) at 7 h (see Fig. S1 in the supplemental material). The apparent decrease in the buffer with heat-inactivated cells was likely due to evaporation via shaking and autoxidation. All data are averages for at least three samples with standard deviations (error bars).

accumulates and inhibits SQR activity in the absence of PDO, which oxidizes sulfane sulfur (5). The resting state of cells of the $\Delta soxYZ$ mutant did not affect sulfide oxidation or the transitory accumulation of sulfane sulfur but largely abolished thiosulfate oxidation and sulfate production (Table 2; Fig. 2A to D); complementation of soxYZ partially restored thiosulfate oxidation and sulfate production (Table 2; Fig. 2A to D); complementation of soxYZ partial recovery seen after complementation could have been due to the level of gene expression or an imbalanced ratio of the proteins that make up the Sox system. A downstream effect is also possible, as it is not clear whether there are internal promoters within the deleted gene, albeit the deletion was in frame.

Cells in the polysulfide-induced resting state were also used to determine the rates of polysulfide oxidation by *C. pinatubonensis* JMP134 and its mutants (Table 1). The $\Delta soxYZ$, $\Delta pdo1 \Delta pdo2$, and $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutants oxidized polysulfide at rates of 91%, 11%, and 6% of the wild-type rate, respectively, suggesting that the two PDOs are the primary enzymes for the oxidation of added polysulfide, which is a specific form of sulfane sulfur and which is likely the direct product of SQR (5, 34). The $\Delta pdo1$ mutant and the $\Delta pdo2$ mutant oxidized polysulfide at rates of 93% and 22% of the rate of the wild type, respectively (Table 1), suggesting that PDO2 is the main PDO in the bacterium. The complementation of the $\Delta pdo1 \Delta pdo2$ mutant with pdo2 led to recovery of the rate of polysulfide oxidation to 69% of the wild-type rate (Table 1).

Cells in the thiosulfate-induced resting state were used to determine the rates of thiosulfate oxidation by *C. pinatubonensis* JMP134 and its mutants (Table 1), as thiosulfate induces the expression of *sox* genes in *P. pantotrophus* GB17 (17). The $\Delta soxYZ$ cells oxidized thiosulfate at a rate of 12% of the wild-type rate. Complementation with

	Concn (μ M)	Recovery of			
Strain	HS ⁻	Sulfane sulfur	S ₂ O ₃ ²⁻	\$0 ₄ ²⁻	sulfur (%)
JMP134	3 ± 1	3 ± 1	7 ± 5	362 ± 16	76
∆sqr	51 ± 6	6 ± 5	2 ± 0	189 ± 21	44
ΔfccB	2 ± 2	5 ± 2	62 ± 6	187 ± 11	63
∆fccB/fccB	4 ± 1	2 ± 1	1 ± 0	252 ± 17	52
$\Delta sqr \Delta pdo1 \Delta pdo2$	92 ± 8	6 ± 3	1 ± 0	200 ± 11	51
$\Delta sqr \Delta pdo1 \Delta pdo2/sqr pdo2$	5 ± 2	3 ± 1	0 ± 0	436 ± 38	89
$\Delta sqr \Delta pdo1 \Delta pdo2 \Delta soxYZ$	100 ± 5	24 ± 1	40 ± 1	33 ± 2	34
$\Delta sqr \Delta fccB$	141 ± 5	7 ± 1	5 ± 2	123 ± 4	39
$\Delta sqr \Delta fccB \Delta soxYZ$	143 ± 15	10 ± 1	69 ± 3	28 ± 1	49
$\Delta sox YZ$	4 ± 2	8 ± 2	139 ± 2	83 ± 6	74
$\Delta sox YZ/sox YZ$	20 ± 4	3 ± 3	1 ± 0	448 ± 17	94
Δpdo1 Δpdo2	29 ± 7	8 ± 2	1 ± 0	158 ± 9	36
Δpdo1 Δpdo2/pdo2	7 ± 2	3 ± 1	8 ± 4	327 ± 18	70
$\Delta pdo1 \ \Delta pdo2 \ \Delta soxYZ$	65 ± 9	4 ± 1	38 ± 4	30 ± 1	25
ΔsorA	3 ± 1	3 ± 0	0 ± 0	357 ± 32	72
∆tauE	7 ± 1	5 ± 3	7 ± 4	363 ± 5	77
Buffer	176 ± 10	3 ± 2	16 ± 5	5 ± 3	12
Heat-inactivated cell	172 ± 7	8 ± 2	22 ± 6	53 ± 13	32

TABLE 2 Substrates and products	at /	hc	of sulfide	oxidation I	oy who	ole cells ^o
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^aThe oxidation of 500 μM sulfide by cell suspensions at an OD of 2 in 100 mM HEPES buffer, pH 7.4, at 30°C was used. Sulfur compounds were analyzed at 7 h. All data are the averages for at least three samples with standard deviations.

soxYZ led to the partial recovery of the rate of thiosulfate oxidation to 71% of the wild-type rate (Table 1).

Collectively, the data suggest that SQR oxidizes sulfide to sulfane sulfur; PDO oxidizes sulfane sulfur to sulfite, which reacts with sulfane sulfur to generate thiosulfate in *C. pinatubonensis* JMP134 (5); and the Sox system further oxidizes thiosulfate to sulfate.

SQR did not couple with the Sox system for sulfide oxidation. The direct coupling of SQR and the Sox system was not apparent, as the $\Delta pdo1 \Delta pdo2$ and $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutants oxidized sulfide at similar rates (Table 1; Fig. 3A), suggesting that the product (sulfane sulfur) inhibition in SQR was not alleviated by the Sox system in the $\Delta pdo1 \Delta pdo2$ mutant. The level of sulfane sulfur was higher for both the $\Delta pdo1 \Delta pdo2$ and the $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutants than for the wild type (Fig. 2B and 3B). Thiosulfate was also produced by the $\Delta pdo1 \Delta pdo2$ and $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutants than for the wild type (Fig. 2B and 3B). Thiosulfate was also produced by the $\Delta pdo1 \Delta pdo2$ and $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutants (Fig. 2C and 3C), suggesting that sulfane sulfur is slowly oxidized to thiosulfate by other enzymes, like catalase (33), or by autoxidation (35). The $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutant accumulated thiosulfate (Fig. 3C and D). However, the $\Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutant also produced 30 μ M sulfate (Table 2; Fig. 3D), suggesting nonspecific oxidation of either sulfane sulfur or thiosulfate at a very low rate by the mutant. Thus, the sulfane sulfur produced by SQR is mainly oxidized by PDO to thiosulfate, which is oxidized by the Sox system to sulfate.

FccAB coupled with the Sox system for sulfide oxidation. The $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant produced some thiosulfate during sulfide oxidation, suggesting that the sulfane sulfur produced by FccAB is oxidized to thiosulfate by other enzymes, as catalase slowly oxidizes polysulfide (33). Further, three lines of evidence supported the suggestion that the sulfane sulfur produced by FCSD was directly oxidized to sulfate by the Sox system in the absence of SQR. First, the $\Delta pdo1 \Delta pdo2$ mutant produced less sulfate than the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant did (Table 2; Fig. 3D), even though it oxidized sulfide faster (Table 1; Fig. 3A) and transitorily accumulated more sulfane sulfur and thiosulfate (Fig. 3C). Here (Fig. 3C), the reduced production of thiosulfate by the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant in comparison with that by the $\Delta pdo1 \Delta pdo2$ mutant suggests the direct coupling of FCSD with the Sox system, in which thiosulfate is not produced. Second, the rates of sulfide oxidation and the final levels of production of sulfate by the Δsqr and the $\Delta sqr \Delta pdo1 pdo2$ mutants were similar (Tables 1 and 2), indicating that PDO is not



FIG 3 Sulfide oxidation by the $\Delta pdo1 \Delta pdo2$ mutant and its derivatives. Cells were harvested, washed, and resuspended at an OD₆₀₀ of 2.0 in 100 mM HEPES buffer, pH 7.4. Sulfide was added to 500 μ M to initiate the reaction. Sulfide (A), polysulfides (B), thiosulfate (C), and sulfate (D) levels were determined at different times. All data are averages for at least three samples with standard deviations (error bars).

necessary during sulfide oxidation in that Δsqr mutant and that FCSD can couple with the Sox system for sulfide oxidation. Third, the $\Delta sqr \Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutant accumulated more sulfane sulfur than the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant did during sulfide oxidation (Table 2; Fig. 3B), suggesting that the Sox system oxidizes sulfane sulfur in the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant. Thus, in the *sqr* deletion mutants, the sulfane sulfur produced by FCSD is oxidized either by nonspecific enzymes, like catalase, or by the Sox system. When FCSD couples with the Sox system, thiosulfate is not an intermediate during sulfide oxidation to sulfate.

The sulfite produced during sulfide oxidation is not released into the medium. Recombinant *E. coli* with cloned *sqr* and *pdo2* from *C. pinatubonensis* JMP134 oxidizes sulfide to sulfite and thiosulfate (5), but we did not detect a transitory accumulation of sulfite in *C. pinatubonensis* JMP134 or its mutants during sulfide oxidation. *C. pinatubonensis* JMP134 contains a SOR system, encoded by *sorA* and *sorB*; however, the Δ *sorA* mutant metabolized sulfide in a similar way as the wild type did (Fig. 2; Fig. S2), and sulfite was still undetectable. The results indicate that *C. pinatubonensis* JMP134 produces sulfite during sulfide oxidation and that sulfite reacts with sulfane sulfur to generate thiosulfate inside the cell without being released.

Sulfite was oxidized to sulfate by the SOR system and the Sox system. The sulfite-induced resting cells of *C. pinatubonensis* JMP134 quickly consumed 500 μ M sulfite within 2 h, producing about 500 μ M sulfate; the $\Delta sorA$ mutant completely lost the ability to oxidize sulfite; sulfite oxidation was restored in the $\Delta sorA/sorA$ strain. The $\Delta soxYZ$ mutant oxidized sulfite as fast as the wild type did (Fig. S3), suggesting that the Sox system is not involved in sulfite oxidation in the wild type. When the cells were induced with thiosulfate and then used to oxidize sulfite, the $\Delta sorA$ mutant oxidized sulfite at a significantly lower rate (0.6 \pm 0.3 nmol/mg [dry weight]/min) than the wild type did (5.4 \pm 0.2 nmol/mg [dry weight]/min), and the $\Delta sorA$ $\Delta soxYZ$ double mutant



FIG 4 Proposed pathways of sulfide oxidation in *C. pinatubonensis* JMP134 and its $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant. (A) Sulfide oxidation in the wild type. Sulfide is oxidized to sulfane sulfur (S⁰) by SQR; PDO oxidizes S⁰ to sulfite, which spontaneously reacts with S⁰ to generate thiosulfate in the cytoplasm. Thiosulfate is transported to the periplasmic space and is oxidized by the Sox system to sulfate. (B) Sulfide oxidation by the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant without SQR and PDO. FCSD oxidizes sulfide to S⁰, which is then oxidized by the Sox system to sulfate. This pathway is marginal in the wild type.

completely lost the ability to oxidize sulfite (Fig. S4). Thus, the Sox system also oxidized sulfite but did so at a rate of 11% of the rate that SOR did in *C. pinatubonensis* JMP134.

DISCUSSION

Our results indicate that C. pinatubonensis JMP134 oxidizes sulfide to sulfate via a new pathway (Fig. 4A). The pathway has not been previously observed in any bacteria. First, H₂S enters the cytoplasm via diffusion, as membrane transporters are not required for the process (36). Second, SQR oxidizes H₂S to sulfane sulfur (S⁰) in the cytoplasm (37). Third, PDO oxidizes sulfane sulfur to sulfite, which spontaneously reacts with sulfane sulfur to generate thiosulfate (5). The reaction between sulfane sulfur and sulfite likely occurs inside the cytoplasm, as both reactants are produced in the cytoplasm (37). Fourth, the produced thiosulfate is transported to the periplasmic space by an unknown transporter. Although tauE, encoding a hypothetical sulfite exporter (see Table S1 in the supplemental material), is next to the sqr-pdo2 operon, its role is not essential, as its deletion did not have detectable effects on sulfide oxidation and sulfate production (Tables 1 and 2). Fifth, the Sox system oxidizes thiosulfate to sulfate in the periplasmic space (Fig. 4A). In C. pinatubonensis JMP134, FCSD does not play a significant role in sulfide oxidation but enhances thiosulfate oxidation by the Sox system, as the $\Delta fccB$ mutant oxidized sulfide at a rate similar to that for the wild type but oxidized thiosulfate at a significantly reduced rate (Table 1). The role of FCSD in enhancing thiosulfate oxidation by the Sox system also occurs in P. pantotrophus GB17 (38). In sqr deletion mutants, such as the Δsqr and the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutants, FCSD is likely able to direct couple with the Sox system for sulfide oxidation to sulfate (Fig. 4B), but at a significantly reduced rate (Table 1). Thus, C. pinatubonensis JMP134 contains a main pathway (Fig. 4A) and a contingent pathway (Fig. 4B) for sulfide oxidation.

The main pathway is different from other documented pathways in bacteria. The anoxygenic purple sulfur bacterium *Allochromatium vinosum* uses H_2S as the electron donor for photosynthesis, producing sulfate. It oxidizes sulfide to polysulfide by SQR and then to sulfite by reverse dissimilatory sulfite reductase (34). Sulfite is mainly oxidized to sulfate by a polysulfide reductase-like iron–sulfur molybdoprotein (SoeABC) (39).

The dimethylsulfide-degrading bacterium *Hyphomicrobium denitrificans* degrades dimethylsulfide to sulfide and then further oxidizes sulfide to sulfate under oxic conditions. Interestingly, the bacterium also produces thiosulfate as a metabolic intermediate. Although it contains PDO, its role in sulfane sulfur oxidation during dimethylsulfide oxidation is not observed. It employs heterodisulfide reductase to oxidize sulfane sulfur to sulfite, which reacts with sulfane sulfur to generate thiosulfate (9). Since the bacterium contains the incomplete Sox system without SoxCD, the system

cannot completely oxidize thiosulfate to sulfate. The incomplete Sox system works with heterodisulfide reductase to oxidize thiosulfate to sulfate and sulfite (9). Sulfite reacts with sulfane sulfur to generate thiosulfate. The net process leads to the oxidation of reduced sulfur compounds to sulfate (40).

The chemolithotroph *Acidithiobacillus caldus*, a bioleaching agent, and members of the genus *Thioalkalivibrio*, a genus of haloalkaliphilic sulfur-oxidizing bacteria, contain genes similar to those seen in *H. denitrificans* (41, 42), and they may produce thiosulfate as a key intermediate during sulfur oxidation. However, *Acidithiobacillus* spp. and *Thioalkalivibrio* spp. also possess sulfite:quinone oxidoreductase, which may directly oxidize sulfate to sulfate (41, 42).

Both SQR and FCSD oxidize sulfide to sulfane sulfur (Fig. 2 and 3). Sequence analysis of 4,929 bacterial genomes showed that 1,014 bacteria contain SQR (3) and 190 bacteria contain FccB (16). The popularity of SQR may suggest its significance in H₂S oxidation. Among the 190 bacteria with FccB, 121 bacteria (63.7%) also carried SQR (16). The frequent co-occurrence of FccB and SQR within a single bacterium indicates that they may have different physiological roles. In *C. pinatubonensis* JMP134, SQR is the primary sulfide oxidase, and the deletion of *fccB* did not reduce the sulfide oxidation rate but significantly reduced the rate of thiosulfate oxidation (Table 1). This finding is similar to that for *A. vinosum*, which contains both *sqr* and *fccAB*, and the inactivation of *fccAB* does not affect its rate of sulfide oxidation (43). The enhancement of the Sox activity by FCSD has previously been reported (38). Further, FCSD oxidizes sulfide at a lower rate in the Δsqr mutant (Table 1). Thus, FCSD oxidizes sulfide in bacteria that do not contain SQR, as previously reported (16). The choice of SQR over FCSD is likely due to energy conservation, as SQR uses ubiquinone as its electron acceptor (44), producing more membrane potential than FCSD, which uses cytochrome *c* as its electron acceptor (16).

The Sox system was extensively studied in *P. pantotrophus*, which is able to grow on organic compounds as well as on thiosulfate (45). Because the purified and reconstituted Sox system from *P. pantotrophus* oxidizes sulfide, thiosulfate, sulfur, and sulfite, the Sox system has been proposed to be a general sulfur-oxidizing system (24, 46). Early genetic analysis further supported the involvement of the Sox system in the oxidation of sulfide and thiosulfate in the anaerobic phototrophic bacterium *Rhodovulum sulfidophilum* (47). However, the role of Sox in sulfide oxidation should be revisited, as it should be confirmed whether *R. sulfidophilum* contains SQR and FCSD. Our data support the suggestion that the physiological substrate of the Sox system is thiosulfate in *C. pinatubonensis* JMP134 (Table 1). In the $\Delta sqr \Delta pdo1 \Delta pdo2$ mutant (Fig. 3) and the $\Delta sorA$ mutant (Fig. S2), the Sox system can oxidize S^o and sulfite but that the efficiency is much lower than that for PDO and SOR in *C. pinatubonensis* JMP134.

The Sox system oxidized the sulfane sulfur generated by FCSD, but not by SQR or added polysulfide (Table 1). Since both FCSD and the Sox system are in the periplasmic space, it is possible that FccB directly transfers sulfane sulfur to SoxY, producing SoxYZ-SSH, which can be further oxidized by the Sox system to sulfate (Fig. 4B). When polysulfide is added to neutral solutions, it is rapidly converted to elemental sulfur in the form of S₈ (48); perhaps the reaction of S₈ with SoxYZ-SH is kinetically slow or needs additional reducing power. Since the active site of SQR in *C. pinatubonensis* JMP134 is in the cytoplasm (37), the membrane separation of SQR-produced sulfane sulfur and the Sox system may contribute to the uncoupling of SQR and the Sox system.

Our evidence does not support the suggestion that the bacterium uses the Sox system for the efficient oxidation of sulfide. First, the $\Delta soxYZ$ mutant oxidized sulfide at the same rate as the wild type (Table 1; Fig. 2). *A. vinosum* also does not use the Sox system for sulfide oxidation, as the deletion of the *sox* genes in *A. vinosum* does not affect its sulfide oxidation (29). Second, the $\Delta sqr \Delta fccB$ mutant with the Sox system did not show any meaningful rate of sulfide oxidation (Table 1; Fig. 2), suggesting that *C. pinatubonensis* JMP134 does not use the Sox system for H₂S oxidation. Since *P. pantotrophus* contains SQR (GenBank accession number RKS43125.1) and FCSD (49), it should be investigated whether it uses the Sox system to oxidize sulfide.

TABLE 3 Strains and plasmids used in this study

Strain or plasmid	or plasmid Characteristic or target protein	
Strains		
E. coli S17-1	recA pro thi hsdS, RP4 tra functions, supE44	Invitrogen
C. pinatubonensis JMP134	Wild type	Our labs
Plasmids	Characteristics	Source
pBBR1MCS2	Kanamycin resistance, mob ⁺ , pBBR1 replicon, cloning vector	Qi Qingsheng
pK18mobsacB	Widely used gene-knockout vector, kanamycin resistance	Our labs
pBBR-soxYZ	pBBR1MCS2 containing soxY and soxZ	This study
pBBR-pdo2-sqr	pBBR1MCS2 containing pdo2 and sqr	This study
pBBR-fccB	pBBR1MCS2 containing fccB	This study
pBBR-pdo2	pBBR1MCS2 containing pdo2	This study
pBBR-sorA	pBBR1MCS2 containing sorA	This study
Mutants of JMP134		
Δsqr	sqr deleted	This study
ΔfccB	fccB deleted	This study
$\Delta sqr \Delta pdo1 \Delta pdo2$	sqr, pdo2, and pdo1 deleted	This study
$\Delta sqr \Delta pdo1 \Delta pdo2 \Delta soxYZ$	sqr, pdo2, pdo1, soxY, and soxZ deleted	This study
$\Delta sqr \Delta fccB$	sqr and fccB deleted	This study
$\Delta sgr \Delta fccB \Delta soxYZ$	sqr, fccB, soxY, and soxZ deleted	This study
ΔsoxYZ	soxY and soxZ deleted	This study
∆pdo1	pdo1 deleted	This study
Δpdo2	pdo2 deleted	This study
Δpdo1 Δpdo2	pdo1 and pdo2 deleted	This study
$\Delta pdo1 \Delta pdo2 \Delta soxYZ$	pdo1, pdo2, soxY, and soxZ deleted	This study
ΔsorA	sorA deleted	This study
ΔtauE	tauE deleted	This study
Complemented strains		
ΔfccB/fccB	$\Delta fccB$ mutant with plasmid pBBR-fccB	This study
Δsqr Δpdo1 Δpdo2/sqr pdo2	$\Delta sqr \Delta pdo1 \Delta pdo2$ mutant with plasmid pBBR-pdo2-sqr	This study
$\Delta sox YZ/sox YZ$	$\Delta sox YZ$ mutant with plasmid pBBR-sox YZ	This study
Δpdo1 Δpdo2/pdo2	$\Delta pdo1 \ \Delta pdo2$ mutant with plasmid pBBR-pdo2	This study
∆sorA/sorA	ΔsorA mutant with plasmid pBBR-sorA	This study
Δsqr Δpdo1 Δpdo2 ΔsoxYZ/soxYZ	$\Delta sqr \Delta pdo1 \Delta pdo2 \Delta soxYZ$ mutant with plasmid pBBR-soxYZ	This study

In *C. pinatubonensis* JMP134, SQR prefers to work with PDO, and the Sox system may oxidize the sulfane sulfur produced by FCSD. The preference may be related to the subcellular localization of the enzymes (Fig. 4). SQR is a membrane protein with its active site in the cytoplasm, and PDO is also in the cytoplasm (37), whereas FCSD and the Sox system are soluble proteins in the periplasmic space (16, 46, 50). The preference is also reflected in the linkages of these genes on the chromosome, with *sqr* and *pdo2* being organized in an operon on chromosome B and the *sox* genes and *fccAB* being adjacently located on chromosome A (Fig. 1).

The H₂S oxidation network of SQR-PDO, FCSD, and the Sox system of *C. pinatubonensis* JMP134 may also be present in other bacteria, including *P. pantotrophus* GB17, *Roseobacter denitrificans* OCh 114, and *Ruegeria pomeroyi* DSS-3, as they all contain these genes (Table S2). These bacteria may use the same pathway to oxidize H₂S. *C. pinatubonensis* JMP134 is a soil bacterium (12); *P. pantotrophus* GB17 is from a denitrifying, sulfide-oxidizing effluent treatment plant in the Netherlands (51); *R. denitrificans* OCh 114 and *R. pomeroyi* DSS-3 are the model bacteria of the *Roseobacter* clade, dominant in coastal seawater and surface sediments (52). These bacteria all belong to the *Proteobacteria* and are capable of heterotrophic growth. Their wide distribution in soil, wastewater treatment plants, and marine waters implies the potential role of these bacteria in H₂S oxidation in the natural environment.

MATERIALS AND METHODS

Strains, plasmids, and primers. The strains and plasmids used in this study are listed in Table 3. All the primers are listed in Table 4.

Culture conditions. *C. pinatubonensis* JMP134 and its mutants were grown at 30°C in Luria-Bertani (LB) medium containing 10 g of NaCl per liter or in a mineral medium (MM) (53). MM consisted of 0.58

TABLE 4 Primers used in this study

Target gene	Primer ^a	Sequence (5′–3′) ^b
Deletion		
soxY and soxZ	Up-f	CAGGAAACAGCTATGACATGATTACGAATTCCGTCATTTTCGGATATCGC
	Up-r	TGGTCGCTTCCCGCAGTACTTCTCTTCGTTTC
	Down-f	AGTACTGCGGGAAGCGACCATCGCCTGATT
	Down-r	TTCAGGATCCCCGGGTACCGAGCTCGAATTCGCTGTTCGAGATCCATGAC
	V-f	GATGGTAGGGTGGATTCTTGAGGC
	V-r	CTACGGGGCCTTCAAAGGTGTT
fccB	UP-f	CAGGAAACAGCTATGACATGATTACGAATTGTCAACGAACTGGAAATCACGTC
	UP-r	TAGCTCAGCACTTCAGAAAGTTGCGTCGTTGC
	Down-f	CTTTCTGAAGTGCTGAGCTAGCCGTTCCGCAT
	Down-r	TTCAGGATCCCCGGGTACCGAGCTCGAATTAACCGGCCCCACATGGTGTAG
	V-f	GATCAAGCCTCCACCTCGCAGAAC
	V-r	CGCAAAGCGTCAACAGAAACCCG
sqr	Up-f	CAGGAAACAGCTATGACATGATTACGAATTGACGGGGGCCTTGAACTTTTATC
	Up-r	GACCATGCAATTGCAGGGATAACACCCGAAG
	Down-f	ATCCCTGCAATTGCATGGTCGTCTGTTCCTTG
	Down-r	TTCAGGATCCCCGGGTACCGAGCTCGAATTAATCGCAACGCTCTGCTAACC
	V-f	GCACCGGTGCCTTTGTATTG
	V-r	TCCTGTACATGTGCCACGAC
pdo1	Up-f	<u>CAGGAAACAGCTATGACATGATTACGAATT</u> AGACGATTACCTGGTCTACACCTTC
	Up-r	CAGCTGTTCGTACAGGCGCGTCAAATCCTTCTAT
	Down-f	CGCGCCTGTACGAACAGCTGATAGAAGGTTTGCAT
	Down-r	TTCAGGATCCCCGGGTACCGAGCTCGAATTGGCTGATGATGGAGAACGAAC
	V-f	TATTGGCTGCCATCTGCT
	V-r	GCTCTACAAGCTCAATGCG
pdo2	Up-f	<u>CAGGAAACAGCTATGACATGATTACGAATT</u> CGAGGTCGTAGCGGTAGTTG
	Up-r	ACACACATGAGCTATCTGAAGATTCCCCTCAAC
	Down-f	TTCAGATAGCTCATGTGTGTCTATCCGTGGTTAGC
	Down-r	TTCAGGATCCCCGGGTACCGAGCTCGAATTCCATTTCATCGAGGAATAGCGT
	V-f	ATGGCGTCCCAATCCAGCTT
	V-r	TTGCCTGGAGAGTGGCTTTG
sorA	Up-f	<u>CAGGAAACAGCTATGACATGATTACGAATT</u> ATCCGTCGGCAGGTAACAGC
	Up-r	CATATAGCCGATTGTCAGGCGCATTCATGG
	Down-f	GCCTGACAATCGGCTATATGCGCAACGTGG
	Down-r	TICAGGATCCCCGGGTACCGAGCTCGAATTGCGCGTGGAGGAAGTGATCC
	V-f	TCTTTCGCTTGCGTCTGATG
	V-r	
tauE	Up-f	
	Up-r	
	Down-f	
	Down-r	
	V-t	
	V-r	CIGIIGAAGGGACAGGGAIAGG
Complementation		
soxY and sox7	Forward	
	Reverse	
fccB	Forward	
	Reverse	TAACAAAATATTAACGCTTAGCTCAGCATGTCGGC
sorA	Forward	
56071	Reverse	TTCCATTCGCCATTCAACTGCGGTAACGCGCGTGGA
sar and pdo2	Forward	
· · · · · · · · · · · · · · · · · · ·	Reverse	TTCCATTCGCCATTCATTATCCCTGCAACTCGGGTGTCG
pdo2	Forward	CACACAGGAAACAGCTATGACACCGACCATGCCAAG
r	Reverse	TTCCATTCGCCATTCATCAGAGGGGGGGGGG
Linearization of pBBR1MCS2	Forward	TGAATGGCGAATGGAAATTGTAAG
	Reverse	AGCTGTTTCCTGTGTGAAATTGTTATC

^aUp, primers used to clone the upstream sequence of the target gene; Down, primers used to clone the downstream sequence of the target gene; V, primers used to verify the mutants.

^bFor the deletion mutants, underlining represents the sequences overlapping the plasmid pK18mobsacB sequence. For the complemented strains, underlining represents the sequences overlapping the plasmid pBBR1MCS2 sequence.

g of K_2 HPO₄, 0.19 g of KH_2 PO₄, 0.25 g of NaNO₃, 0.5 g of MgCl₂, and 1 ml of a trace element solution per liter of deionized water, and its pH was adjusted to 7.0. The carbon source was 0.5% (wt/vol) monoso-dium glutamate. NaSO₄, Na₂SO₃, or Na₂S₂O₃ (1 mM) was used as the sole sulfur source. The trace element solution contained 10 ml of concentrated HCl, 4.74 mg of ZnCl₂, 2.53 mg of MnCl₂·4H₂O, 30 mg of H₃BO₃, 20 mg of CoCl₂·6H₂O, 1 mg of CuCl₂·2H₂O, 2 mg of NiCl₂·6H₂O, 3.24 mg of Na₂MoO₄·2H₂O per liter.

Sulfur compound preparation. NaHS, Na_2SO_3 , or $Na_2S_2O_3$ was freshly prepared in 100 mM HEPES buffer (pH 7.4). The polysulfide was prepared according to a published method by mixing sulfur power and sodium sulfide in anoxic distilled water under argon gas (5).

Gene deletion and complementation. The method used to delete the sulfur-related genes in *C. pinatubonensis* JMP134 is essentially similar to that previously reported (3). The primers used in the deletion process are shown in Table 4. After the upstream and downstream fragments of the target gene were obtained by PCR, these two fragments were ligated with the linearized plasmid pK18mobsacB by a modified In-Fusion method (54) to construct a deletion plasmid, and the deletion plasmid was transformed into *E. coli* S17-1 and then transferred to *C. pinatubonensis* JMP134 by conjugation. After two rounds of screening by using colony PCR, the correct deletion strain was obtained.

The complementation strain was generated by transforming a recombinant plasmid into the corresponding mutant. The recombinant plasmid was constructed by assembling the PCR-amplified gene in the broad-host-range plasmid pBBR1MCS2 (linearized via PCR) by using a modified In-Fusion method (54). The primers used to amplify the target gene and linearized plasmid pBBR1MCS2 are listed in Table 4.

Whole-cell assay. A single colony of the wild type, deletion strains, or complementation strains was inoculated in LB medium (with antibiotics, as necessary) and incubated at 30°C overnight (200 rpm). The culture was transferred to 100 ml fresh medium (1% inoculation) and incubated at 30°C with shaking (200 rpm) to an OD₆₀₀ of 1.0 to 1.5, and then 20 μ M NaHS, 20 μ M polysulfides, 100 μ M Na₂SO₃, or 100 μ M Na₂S₂O₃ was added for induction with continuous incubation to an OD₆₀₀ of 2.5 (about 1 h). Cells were collected by centrifugation (4,000 × *g*, 5 min) and resuspended in HEPES buffer (pH 7.4, 100 mM) at an OD₆₀₀ of 2.0. The heat-inactivated cells were prepared by incubating 10 ml of cell suspension in a boiling water bath for 15 min.

Ten milliliters of the suspension was transferred to a 50-ml large plastic centrifuge tube (Labcon, USA); NaHS, polysulfide, Na_2SO_3 , or $Na_2S_2O_3$ was added; and the tube was covered with a silicone stopper. The sample was mixed and incubated with shaking (30°C, 100 rpm). One milliliter of sample was taken at 15 min, 30 min, 1 h, 2 h, 4 h, and 7 h, and the concentrations of the sulfur compounds were measured. Suspensions were used directly for the detection of sulfide and sulfane sulfur. When sulfite, thiosulfate, and sulfate were tested, the suspension was centrifuged (13,000 × *g*, 3 min) and the supernatant was taken for detection.

Sulfur compound detection. Sulfide was detected by using the methylene blue method, and sulfane sulfur (including polysulfide, persulfide, and elemental sulfur) was detected by using the cyanide method (5). The detection of sulfite, thiosulfate, and sulfate in the supernatant was carried out by using an ion chromatograph system (ICS; model ICS-1100; Dionex, USA) in the anion detection mode with an Ion Pac AS19 column, an eluent automatic generation device (RFC-30), a column temperature of 30°C, an ASRS_4 mm suppressor, and an eluent of 22 mM KOH with isocratic elution (1 ml/min KOH). Under these conditions, the elution times of sulfite, sulfate, and thiosulfate were 7.9 min, 8.5 min, and 27.5 min, respectively. This method had a sulfite detection limit of 10 μ M.

Bioinformatics. The basic sulfur metabolic pathway was derived from the analysis of *C. pinatubonen*sis JMP134 at the KEGG website (55). The query sequences were from *A. vinosum* DSM 180^T (locus tag, Alvin_0091; GenBank accession number ADC61061) and *Cupriavidus metallidurans* CH34 (locus tag, Rmet_5347; GenBank accession number ABF12206) for thiosulfate dehydrogenases, from *Acidithiobacillus ferrooxidans* (GenBank accession number AAB93983) for tetrathionate hydrolase, and from *Thiobacillus denitrificans* ATCC 25259 (GenBank accession number Q60028.3) for ribulose bisphosphate carboxylase. Similar proteins of each sulfur-related enzyme of *C. pinatubonensis* JMP134 were searched for in selected microbial genomes by using the TBLASTN program (56).

Data availability. Data and NCBI accession numbers related to sulfide oxidation by additional mutants and complementation strains and sulfur-metabolizing enzymes in *C. pinatubonensis* JMP134 and other bacteria are given in the supplemental material.

SUPPLEMENTAL MATERIAL

Supplemental material is available online only. **SUPPLEMENTAL FILE 1**, PDF file, 1 MB.

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REFERENCES

- Dahl C. 2015. Cytoplasmic sulfur trafficking in sulfur-oxidizing prokaryotes. IUBMB Life 67:268–274. https://doi.org/10.1002/iub.1371.
- Wang R, Lin JQ, Liu XM, Pang X, Zhang CJ, Yang CL, Gao XY, Lin CM, Li YQ, Li Y, Lin JQ, Chen LX. 2018. Sulfur oxidation in the acidophilic autotrophic *Acidithiobacillus* spp. Front Microbiol 9:3290. https://doi.org/ 10.3389/fmicb.2018.03290.
- Xia Y, Lu C, Hou N, Xin Y, Liu J, Liu H, Xun L. 2017. Sulfide production and oxidation by heterotrophic bacteria under aerobic conditions. ISME J 11:2754–2766. https://doi.org/10.1038/ismej.2017.125.
- Luebke JL, Shen J, Bruce KE, Kehl-Fie TE, Peng H, Skaar EP, Giedroc DP. 2014. The CsoR-like sulfurtransferase repressor (CstR) is a persulfide sensor in *Staphylococcus aureus*. Mol Microbiol 94:1343–1360. https:// doi.org/10.1111/mmi.12835.
- Xin Y, Liu H, Cui F, Liu H, Xun L. 2016. Recombinant *Escherichia coli* with sulfide:quinone oxidoreductase and persulfide dioxygenase rapidly oxidises sulfide to sulfite and thiosulfate via a new pathway. Environ Microbiol 18:5123–5136. https://doi.org/10.1111/1462-2920.13511.
- 6. Hou N, Xia Y, Wang X, Liu H, Liu H, Xun L. 2018. H₂S biotreatment with

sulfide-oxidizing heterotrophic bacteria. Biodegradation 29:511–524. https://doi.org/10.1007/s10532-018-9849-6.

- Mason J, Kelly DP. 1988. Thiosulfate oxidation by obligately heterotrophic bacteria. Microb Ecol 15:123–134. https://doi.org/10.1007/BF02011707.
- Sorokin DY, Tourova TP, Muyzer G. 2005. Oxidation of thiosulfate to tetrathionate by an haloarchaeon isolated from hypersaline habitat. Extremophiles 9:501–504. https://doi.org/10.1007/s00792-005-0465-0.
- Koch T, Dahl C. 2018. A novel bacterial sulfur oxidation pathway provides a new link between the cycles of organic and inorganic sulfur compounds. ISME J 12:2479–2491. https://doi.org/10.1038/s41396-018 -0209-7.
- Teske A, Brinkhoff T, Muyzer G, Moser DP, Rethmeier J, Jannasch HW. 2000. Diversity of thiosulfate-oxidizing bacteria from marine sediments and hydrothermal vents. Appl Environ Microbiol 66:3125–3133. https:// doi.org/10.1128/aem.66.8.3125-3133.2000.
- Lykidis A, Perez-Pantoja D, Ledger T, Mavromatis K, Anderson IJ, Ivanova NN, Hooper SD, Lapidus A, Lucas S, Gonzalez B, Kyrpides NC. 2010. The complete multipartite genome sequence of *Cupriavidus necator* JMP134, a versatile pollutant degrader. PLoS One 5:e9729. https://doi.org/10 .1371/journal.pone.0009729.
- Don RH, Pemberton JM. 1981. Properties of six pesticide degradation plasmids isolated from *Alcaligenes paradoxus* and *Alcaligenes eutrophus*. J Bacteriol 145:681–686. https://doi.org/10.1128/JB.145.2.681-686.1981.
- Gregersen LH, Bryant DA, Frigaard NU. 2011. Mechanisms and evolution of oxidative sulfur metabolism in green sulfur bacteria. Front Microbiol 2:116. https://doi.org/10.3389/fmicb.2011.00116.
- 14. Bartsch RG, Kamen MD. 1960. Isolation and properties of two soluble heme proteins in extracts of the photoanaerobe *Chromatium*. J Biol Chem 235:825–831.
- Kostanjevecki V, Brige A, Meyer TE, Cusanovich MA, Guisez Y, van Beeumen J. 2000. A membrane-bound flavocytochrome *c*-sulfide dehydrogenase from the purple phototrophic sulfur bacterium *Ectothiorhodospira vacuolata*. J Bacteriol 182:3097–3103. https://doi.org/10.1128/jb .182.11.3097-3103.2000.
- Lü C, Xia Y, Liu D, Zhao R, Gao R, Liu H, Xun L. 2017. Cupriavidus necator H16 uses flavocytochrome c sulfide dehydrogenase to oxidize selfproduced and added sulfide. Appl Environ Microbiol 83:e01610-17. https://doi.org/10.1128/AEM.01610-17.
- Friedrich CG, Rother D, Bardischewsky F, Quentmeier A, Fischer J. 2001. Oxidation of reduced inorganic sulfur compounds by bacteria: emergence of a common mechanism? Appl Environ Microbiol 67:2873–2882. https://doi.org/10.1128/AEM.67.7.2873-2882.2001.
- Grabarczyk DB, Berks BC. 2017. Intermediates in the Sox sulfur oxidation pathway are bound to a sulfane conjugate of the carrier protein SoxYZ. PLoS One 12:e0173395. https://doi.org/10.1371/journal.pone.0173395.
- Quentmeier A, Friedrich CG. 2001. The cysteine residue of the SoxY protein as the active site of protein-bound sulfur oxidation of *Paracoccus pantotrophus* GB17. FEBS Lett 503:168–172. https://doi.org/10.1016/ s0014-5793(01)02727-2.
- Grabarczyk DB, Chappell PE, Johnson S, Stelzl LS, Lea SM, Berks BC. 2015. Structural basis for specificity and promiscuity in a carrier protein/ enzyme system from the sulfur cycle. Proc Natl Acad Sci U S A 112: E7166–E7175. https://doi.org/10.1073/pnas.1506386112.
- Sauve V, Roversi P, Leath KJ, Garman EF, Antrobus R, Lea SM, Berks BC. 2009. Mechanism for the hydrolysis of a sulfur-sulfur bond based on the crystal structure of the thiosulfohydrolase SoxB. J Biol Chem 284: 21707–21718. https://doi.org/10.1074/jbc.M109.002709.
- Lu WP, Kelly DP. 1984. Properties and role of sulphite: cytochrome c oxidoreductase purified from *Thiobacillus versutus* (A2). Microbiology 130:1683–1692. https://doi.org/10.1099/00221287-130-7-1683.
- Zander U, Faust A, Klink BU, de Sanctis D, Panjikar S, Quentmeier A, Bardischewsky F, Friedrich CG, Scheidig AJ. 2011. Structural basis for the oxidation of protein-bound sulfur by the sulfur cycle molybdohemoenzyme sulfane dehydrogenase SoxCD. J Biol Chem 286:8349–8360. https://doi.org/10.1074/jbc.M110.193631.
- Rother D, Henrich HJ, Quentmeier A, Bardischewsky F, Friedrich CG. 2001. Novel genes of the sox gene cluster, mutagenesis of the flavoprotein SoxF, and evidence for a general sulfur-oxidizing system in *Paracoccus pantotrophus* GB17. J Bacteriol 183:4499–4508. https://doi.org/ 10.1128/JB.183.15.4499-4508.2001.
- Robin S, Arese M, Forte E, Sarti P, Giuffre A, Soulimane T. 2011. A sulfite respiration pathway from *Thermus thermophilus* and the key role of newly identified cytochrome c₅₅₀. J Bacteriol 193:3988–3997. https://doi .org/10.1128/JB.05186-11.

- 26. Li H, Li J, Lu C, Xia Y, Xin Y, Liu H, Xun L, Liu H. 2017. FisR activates σ^{54} -dependent transcription of sulfide-oxidizing genes in *Cupriavidus pinatubonensis* JMP134. Mol Microbiol 105:373–384. https://doi.org/10.1111/mmi.13725.
- Liu H, Xin Y, Xun L. 2014. Distribution, diversity, and activities of sulfur dioxygenases in heterotrophic bacteria. Appl Environ Microbiol 80: 1799–1806. https://doi.org/10.1128/AEM.03281-13.
- Denkmann K, Grein F, Zigann R, Siemen A, Bergmann J, van Helmont S, Nicolai A, Pereira IA, Dahl C. 2012. Thiosulfate dehydrogenase: a widespread unusual acidophilic *c*-type cytochrome. Environ Microbiol 14: 2673–2688. https://doi.org/10.1111/j.1462-2920.2012.02820.x.
- Hensen D, Sperling D, Truper HG, Brune DC, Dahl C. 2006. Thiosulphate oxidation in the phototrophic sulphur bacterium *Allochromatium vinosum*. Mol Microbiol 62:794–810. https://doi.org/10.1111/j.1365-2958 .2006.05408.x.
- De Jong GA, Hazeu W, Bos P, Kuenen JG. 1997. Isolation of the tetrathionate hydrolase from *Thiobacillus acidophilus*. Eur J Biochem 243: 678–683. https://doi.org/10.1111/j.1432-1033.1997.00678.x.
- Ramos AR, Keller KL, Wall JD, Pereira IA. 2012. The membrane QmoABC complex interacts directly with the dissimilatory adenosine 5'phosphosulfate reductase in sulfate reducing bacteria. Front Microbiol 3:137. https://doi.org/10.3389/fmicb.2012.00137.
- Olson KR, Gao Y, Arif F, Arora K, Patel S, DeLeon ER, Sutton TR, Feelisch M, Cortese-Krott MM, Straub KD. 2018. Metabolism of hydrogen sulfide (H(2)S) and production of reactive sulfur species (RSS) by superoxide dismutase. Redox Biol 15:74–85. https://doi.org/10.1016/j.redox.2017.11 .009.
- Olson KR, Gao Y, DeLeon ER, Arif M, Arif F, Arora N, Straub KD. 2017. Catalase as a sulfide-sulfur oxido-reductase: an ancient (and modern?) regulator of reactive sulfur species (RSS). Redox Biol 12:325–339. https:// doi.org/10.1016/j.redox.2017.02.021.
- Prange A, Engelhardt H, Truper HG, Dahl C. 2004. The role of the sulfur globule proteins of *Allochromatium vinosum*: mutagenesis of the sulfur globule protein genes and expression studies by real-time RT-PCR. Arch Microbiol 182:165–174. https://doi.org/10.1007/s00203-004-0683-3.
- Kleinjan WE, de Keizer A, Janssen AJ. 2005. Kinetics of the chemical oxidation of polysulfide anions in aqueous solution. Water Res 39: 4093–4100. https://doi.org/10.1016/j.watres.2005.08.006.
- Mathai JC, Missner A, Kugler P, Saparov SM, Zeidel ML, Lee JK, Pohl P. 2009. No facilitator required for membrane transport of hydrogen sulfide. Proc Natl Acad Sci U S A 106:16633–16638. https://doi.org/10.1073/ pnas.0902952106.
- Gao R, Liu H, Xun L. 2017. Cytoplasmic localization of sulfide:quinone oxidoreductase and persulfide dioxygenase of *Cupriavidus pinatubonen*sis JMP134. Appl Environ Microbiol 83:e01820-17. https://doi.org/10 .1128/AEM.01820-17.
- Bardischewsky F, Quentmeier A, Friedrich CG. 2006. The flavoprotein SoxF functions in chemotrophic thiosulfate oxidation of *Paracoccus* pantotrophus in vivo and in vitro. FEMS Microbiol Lett 258:121–126. https://doi.org/10.1111/j.1574-6968.2006.00210.x.
- Dahl C, Franz B, Hensen D, Kesselheim A, Zigann R. 2013. Sulfite oxidation in the purple sulfur bacterium *Allochromatium vinosum*: identification of SoeABC as a major player and relevance of SoxYZ in the process. Microbiology (Reading) 159:2626–2638. https://doi.org/10.1099/mic.0 .071019-0.
- 40. Tanabe TS, Leimkuhler S, Dahl C. 2019. The functional diversity of the prokaryotic sulfur carrier protein TusA. Adv Microb Physiol 75:233–277. https://doi.org/10.1016/bs.ampbs.2019.07.004.
- 41. Chen L, Ren Y, Lin J, Liu X, Pang X, Lin J. 2012. Acidithiobacillus caldus sulfur oxidation model based on transcriptome analysis between the wild type and sulfur oxygenase reductase defective mutant. PLoS One 7:e39470. https://doi.org/10.1371/journal.pone.0039470.
- Berben T, Overmars L, Sorokin DY, Muyzer G. 2019. Diversity and distribution of sulfur oxidation-related genes in *Thioalkalivibrio*, a genus of chemolithoautotrophic and haloalkaliphilic sulfur-oxidizing bacteria. Front Microbiol 10:160. https://doi.org/10.3389/fmicb.2019.00160.
- Reinartz M, Tschape J, Bruser T, Truper HG, Dahl C. 1998. Sulfide oxidation in the phototrophic sulfur bacterium *Chromatium vinosum*. Arch Microbiol 170:59–68. https://doi.org/10.1007/s002030050615.
- Cherney MM, Zhang Y, Solomonson M, Weiner JH, James MN. 2010. Crystal structure of sulfide:quinone oxidoreductase from *Acidithioba-cillus ferrooxidans*: insights into sulfidotrophic respiration and detox-ification. J Mol Biol 398:292–305. https://doi.org/10.1016/j.jmb.2010.03.018.

- 45. Rainey FA, Kelly DP, Stackebrandt E, Burghardt J, Hiraishi A, Katayama Y, Wood AP. 1999. A re-evaluation of the taxonomy of *Paracoccus denitrificans* and a proposal for the combination *Paracoccus pantotrophus* comb. nov. Int J Syst Bacteriol 49:645–651. https://doi.org/10.1099/
- 00207713-49-2-645. 46. Friedrich CG, Quentmeier A, Bardischewsky F, Rother D, Kraft R, Kostka S, Prinz H. 2000. Novel genes coding for lithotrophic sulfur oxidation of *Paracoccus pantotrophus* GB17. J Bacteriol 182: 4677-4687. https://doi.org/10.1128/jb.182.17.4677-4687.2000.
- Appia-Ayme C, Little PJ, Matsumoto Y, Leech AP, Berks BC. 2001. Cytochrome complex essential for photosynthetic oxidation of both thiosulfate and sulfide in *Rhodovulum sulfidophilum*. J Bacteriol 183:6107–6118. https://doi.org/10.1128/JB.183.20.6107-6118.2001.
- Toohey Jl. 2011. Sulfur signaling: is the agent sulfide or sulfane? Anal Biochem 413:1–7. https://doi.org/10.1016/j.ab.2011.01.044.
- Quentmeier A, Hellwig P, Bardischewsky F, Wichmann R, Friedrich CG. 2004. Sulfide dehydrogenase activity of the monomeric flavoprotein SoxF of *Paracoccus pantotrophus*. Biochemistry 43:14696–14703. https:// doi.org/10.1021/bi048568y.
- Frigaard NU, Dahl C. 2009. Sulfur metabolism in phototrophic sulfur bacteria. Adv Microb Physiol 54:103–200. https://doi.org/10.1016/S0065 -2911(08)00002-7.

- Robertson LA, Kuenen JG. 1983. *Thiosphaera pantotropha* gen. nov. sp. nov., a facultatively anaerobic, facultatively autotrophic sulphur bacterium. J Gen Microbiol 129:2847–2855.
- Lenk S, Moraru C, Hahnke S, Arnds J, Richter M, Kube M, Reinhardt R, Brinkhoff T, Harder J, Amann R, Mußmann M. 2012. Roseobacter clade bacteria are abundant in coastal sediments and encode a novel combination of sulfur oxidation genes. ISME J 6:2178–2187. https://doi.org/10 .1038/ismej.2012.66.
- Louie TM, Webster CM, Xun L. 2002. Genetic and biochemical characterization of a 2,4,6-trichlorophenol degradation pathway in *Ralstonia eutropha* JMP134. J Bacteriol 184:3492–3500. https://doi.org/10.1128/jb .184.13.3492-3500.2002.
- Xia Y, Li K, Li J, Wang T, Gu L, Xun L. 2019. T5 exonuclease-dependent assembly offers a low-cost method for efficient cloning and site-directed mutagenesis. Nucleic Acids Res 47:e15. https://doi.org/10.1093/nar/ gky1169.
- Kanehisa M, Goto S. 2000. KEGG: Kyoto Encyclopedia of Genes and Genomes. Nucleic Acids Res 28:27–30. https://doi.org/10.1093/nar/28 .1.27.
- Altschul SF, Gish W, Miller W, Myers EW, Lipman DJ. 1990. Basic local alignment search tool. J Mol Biol 215:403–410. https://doi.org/10.1016/ S0022-2836(05)80360-2.