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Enzymatically crosslinked silk and silk-gelatin hydrogels with tunable gelation kinetics, mechanical properties and bioactivity for cell culture and encapsulation

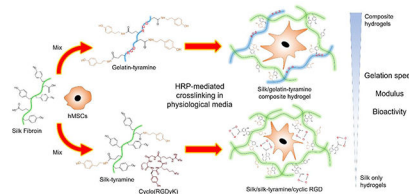
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Abstract

Silk fibroin (SF) was enzymatically crosslinked with tyramine-substituted silk fibroin (SF-TA) or gelatin (G-TA) to fabricate hybrid hydrogels with tunable gelation kinetics, mechanical properties and bioactivity. Horseradish peroxidase (HRP)/hydrogen peroxide (H_2O_2) mediated crosslinking of SF in physiological buffers results in slow gelation and limited mechanical properties. Moreover, SF lacks cell attachment sequences, leading to poor cell-material interactions. These shortcomings can limit the uses of enzymatically crosslinked silk hydrogels in injectable tissue fillings, 3D bioprinting or cell microencapsulation, where rapid gelation and high bioactivity are desired. Here SF/SF-TA and SF/G-TA composite hydrogels were characterized for hydrogel properties and the influence of conjugated cyclic arginine-glycine-aspartic acid (RGD) peptide or G-TA content on bioactivity was explored. Both SF-TA and G-TA significantly increased gelation kinetics, improved mechanical properties and delayed enzymatic degradation in a concentration-dependent manner. β -sheet formation and hydrogel stiffening were accelerated by SF-TA content but delayed by G-TA. Both cyclic RGD and G-TA significantly improved morphology and metabolic activity of human mesenchymal stem cells (hMSCs) cultured on or encapsulated in composite hydrogels. The hydrogel formulations introduced in this study provide improved control of gel formation and properties, along with biocompatible systems that can be utilized in tissue engineering and cell delivery applications.

Graphical Abstract



Keywords

silk hydrogels; HRP-mediated crosslinking; tyramine substitution; gelatin composite hydrogels; cyclic RGD peptide; cell encapsulation

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1. Introduction

Hydrogels are porous three-dimensional (3D) polymeric networks that can retain a large amount of water. Tunable physical, chemical and mechanical properties of hydrogel materials make them excellent candidates for the delivery of therapeutic agents [1, 2], tissue engineering and regenerative medicine applications [3–5]. One of the major uses of hydrogels is as injectable tissue fillers, where an empty or cell-laden pre-hydrogel solution is delivered to the defect area and cured *in situ* to guide cell and tissue growth [6, 7]. Besides tissue fillers, hydrogels also find use as bioinks for 3D bioprinting or as microscale particles for cell or drug delivery. Hydrogel scaffolds with a wide range of chemistry, shapes and structures have been reported for the fabrication of 3D tissue and organ systems for implantation [8, 9] or as models for *in vitro* drug testing [10]. Microgels fabricated by emulsification of an aqueous precursor solution in an oil phase, have been proposed for the *in vivo* delivery of encapsulated therapeutics and/or cells [11–15]. Silk fibroin (SF) of the domesticated silkworm *Bombyx mori* is a natural protein that displays excellent mechanical properties owing to hierarchical self-assembly, biocompatibility and biodegradation, and [16]. In addition to hydrogels physically crosslinked through the self-assembly of β -sheets induced by vortex [17], sonication [18], or polyalcohol treatment [19], aqueous SF solutions were also shown to form highly elastic, biocompatible hydrogels in the presence of horseradish peroxidase (HRP) and hydrogen peroxide (H_2O_2) through the oxidation of phenol groups in tyrosine residues into covalent crosslinks. [20] Mild crosslinking conditions in aqueous media, tunable mechanical properties, biodegradability and biocompatibility of enzymatically crosslinked silk hydrogels [21–23] make these materials excellent candidates for cell culture and encapsulation applications.

Enzymatic crosslinking of SF in physiological buffers such as phosphate buffered saline (PBS) or cell growth media was found to be much slower and result in weaker hydrogels compared to crosslinking in water [24], likely because of the salting out effect of metal ions inducing self-assembly of hydrophobic domains [25–27] which contain many tyrosine groups required for enzymatic crosslinking. Incubation of cells in water solution of SF, on the other hand, might impair cell viability due to osmotic stress [28]. This situation significantly limits the use of enzymatically crosslinked silk hydrogels in 3D bioprinting or microfluidic cell encapsulation, where rapid gelation is desired [29–31]. Another limitation of SF is that fibroin from the domesticated silkworm lacks cell attachment sequences [32], which resulted in a spherical morphology of the encapsulated cells [21, 24, 33]. The limited cell-matrix interactions may induce anoikis (homelessness), the programmed cell death of adherent cells due to missing survival signals provided by integrin-ligand interactions [34]. It is therefore useful to improve both gelation kinetics in physiological buffers and bioactivity of the enzymatically crosslinked silk hydrogels for cell encapsulation applications to preserve short or long-term cell viability.

Faster gelation of SF can be achieved by increasing the concentration of HRP [35, 36] and/or H_2O_2 [37]. However, high concentrations of HRP and H_2O_2 may cause potential immunological responses *in vivo* [36, 38] and apoptotic or necrotic cytotoxicity [39], respectively. Adding phenol red in silk solution was also found to increase gelation speed through bonding of phenolic hydroxyl groups to silk tyrosines [24], but the red color of the

hydrogels and the risk of dye leakage at high concentrations may raise aesthetic and cytotoxicity concerns, respectively. Here we proposed faster gelation of silk by increasing the number of phenol groups available for crosslinking. Introduction of phenol groups on polymers has been reported for enzymatic crosslinking of gelatin [40–42], poly(glutamic acid) (PGA) [43], hyaluronic acid (HA) [44–48], dextran [49–51], alginate [52, 53], heparin [54], cellulose [55], chondroitin sulfate [56], poly(vinyl alcohol) (PVA) [57], poly(ethylene glycol) (PEG) [58] and their blends [59–61]. It is also possible to modify SF on the aspartic and glutamic acid residues through carbodiimide coupling [62, 63], to introduce new phenol groups via the conjugation of tyramine residues. The modified polymers can be mixed with unmodified SF at various ratios to modulate gelation kinetics and mechanical properties of resulting hydrogels. The bioactivity can also be improved through conjugation of peptides with arginine-glycine-aspartic acid (RGD) sequence, which is found in many extracellular matrix (ECM) proteins such as vitronectin, fibrinogen, collagen and laminin [64]. RGD peptides have been shown to improve cell attachment, spreading and proliferation on hydrogels of bio-inert polymers such as PEG [65–71], HA [72–75] and alginate [76], but was not conjugated into silk hydrogels before. Based on the previous reports with *in situ* conjugation of peptides with a tyrosine group into the hydrogels of phenol-substituted polymers during HRP-mediated crosslinking [77–79], here we proposed that silk/silk-tyramine hydrogels can be supplemented with cyclic RGD peptides with a tyrosine residue to improve cell-matrix interactions.

Another strategy to improve bioactivity of silk hydrogels would be blending with a protein that contains intrinsic RGD sequences, such as gelatin. Gelatin is a soluble protein produced by partial acid (type A) or base (type B) hydrolysis of collagen, which is the main ECM component of connective tissues including bone, cartilage and skin [80]. Gelatin is rich in RGD sequence [81], and it has been incorporated into HA, alginate and PEG to improve their bioactivity for cartilage, adipose, muscle and nerve tissue engineering applications [82]. Silk-gelatin composite hydrogels have also been reported, and the most common strategy has been the immobilization of gelatin within physically crosslinked silk network through methanol treatment [83, 84] or sonication [85]. However, significant mass losses were reported at physiological temperatures due to the sol-gel transition of gelatin at around 30°C [82, 83]. To prevent this, UV crosslinking of methacrylated gelatin prior to methanol treatment [86] or covalent crosslinking of silk and gelatin via genipin after ultrasonication [87] have been attempted, but potential cytotoxicity of methanol [88] and sonication [89] present significant issues for cell encapsulation. Silk-gelatin hydrogels directly crosslinked via genipin [90] or tyrosinase [91, 92] treatment have also been reported, but gelation times were long. It is possible to achieve rapid HRP/H₂O₂ mediated crosslinking of gelatin through conjugation of phenol groups [40, 41, 93–97], and hydrogel bioactivity was improved when crosslinked with other tyramine-substituted bioinert polymers such as alginate [98], HA [61, 99] and PEG [100]. This strategy, however, has not been reported for enzymatically crosslinked silk-gelatin composite hydrogels. Our group has previously demonstrated crosslinking of silk with HA-tyramine through the formation of tyrosine-tyramine bonds, resulting in significantly faster gelation compared to SF only hydrogels [101]. Using the same strategy, unmodified SF can be crosslinked with G-TA to improve both gelation kinetics in physiological buffers and bioactivity of silk hydrogels.

In this study, our aim was to improve gelation kinetics, mechanical properties and bioactivity of SF hydrogels crosslinked in physiologically relevant media through combining SF with SF-TA and cyclo(RGDyK) cyclic peptide, or only with G-TA. The influence of SF-TA and G-TA ratios on gelation kinetics, mechanical properties, β -sheet content and enzymatic degradation were explored and the effects of cyclo(RGDyK) or G-TA content on the behavior of human mesenchymal stem cells (hMSCs) on and within composite hydrogels were characterized. Our results suggest that the hydrogel formulations introduced in the present study offer biocompatible systems with tunable gelation kinetics, mechanical properties, enzymatic degradation and bioactivity that can be utilized for culture and encapsulation of mammalian cells in tissue engineering and regenerative medicine applications.

2. Materials and Methods

2.1. Synthesis of SF-TA and G-TA

Aqueous silk solutions were prepared as described before [21]. Briefly, *B. mori* cocoons were degummed to remove sericin protein by boiling 5 g of cut cocoons in 2 L of 0.02 M sodium carbonate (Sigma-Aldrich, St. Louis, MO) solution for 60 min and rinsing three times in deionized (DI) water. Degummed fibers were dried overnight and solubilized in 9.3 M lithium bromide (Sigma-Aldrich, St. Louis, MO) solution at a concentration of 20% (w/v) for 4 h at 60°C. The solution was then dialyzed against distilled water using regenerated cellulose dialysis tubing (3.5 kD MWCO, Spectrum Labs Inc, Rancho Dominguez, CA). Dialysis water was changed 6 times over 3 days and the resulting solution was centrifuged 2 times at 9,000 rpm at 4°C for 20 min to remove insoluble particles. The concentration of silk solution was determined by weighing a known volume of sample before and after drying overnight at 60°C.

SF-TA and G-TA derivatives were synthesized via carbodiimide-mediated reaction as described elsewhere [40]. Briefly, 2% (w/v) SF and gelatin solutions were prepared in 0.05 M 2-(N-morpholino)ethanesulfonic acid (MES) buffer (pH 6.0) and reacted with tyramine hydrochloride (Sigma-Aldrich, St. Louis, MO) (500 mg per 1 g protein) in the presence of 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDC) (Thermo Fisher Scientific, Rockford, IL) (184 mg per 1 g protein) and N-hydroxysuccinimide (NHS) (57 mg per 1 g protein) under stirring at room temperature (24°C) for 18 h. Solutions were dialyzed against distilled water using 3.5 MWCO tubing with 6 changes over 3 days. G-TA solution was subsequently lyophilized and stored at -20°C while SF-TA was kept as a solution at 4°C for up to a month.

2.2. Gelation Kinetics

To prepare SF/SF-TA and SF/G-TA hydrogels, aqueous SF was mixed with SF-TA or G-TA solutions at a final protein concentration of 50 mg/mL or 30 mg/mL in 40 mM hydroxyethylpiperazine-1-ethanesulfonic acid (HEPES) (Sigma-Aldrich, St. Louis, MO) buffer (pH 7.4) or 0.5X Dulbecco's Modified Eagle Medium (DMEM) High glucose with final SF-TA or G-TA ratios ranging between 0-100% (Table 1). Pre-hydrogel, HRP and H₂O₂ solutions were sterile filtered using Millex-GV 0.22 μ m polyvinylidene fluoride

(PVDF) syringe-driven filter units (Millipore Sigma, Darmstadt, Germany) for the rest of the study. Crosslinking kinetics of SF/SF-TA and SF/G-TA hydrogels were monitored at 37°C by measuring fluorescence emission using a SpectraMax M2 multi-mode microplate reader (Molecular Devices, Sunnyvale, CA). Gelation of 150 µL solutions was initiated with 10 U/mL HRP (type VI, Sigma-Aldrich, St. Louis, MO) and 0.01 or 0.005 wt.% H₂O₂ (Sigma-Aldrich, St. Louis, MO) in a black 96-well plate and fluorescence emission at 415 nm after excitation at 315 nm was monitored for 4500 s. 5% G-TA only in HEPES was omitted from material characterization due to very rapid gelation, resulting in non-homogenous hydrogels. To determine the advantage of tyramine substitution of SF over simply increasing HRP concentration, gelation kinetics of SF only and SF-TA only solutions in 0.5X DMEM with 0.01 wt% H₂O₂ were monitored and compared upon induction with 10 to 100 U/mL and 0.1 to 10 U/mL HRP, respectively. Results are reported after normalizing to a blank measurement taken before H₂O₂ was added (n=5).

2.3. Rheology

Rheological properties of SF/SF-TA and SF/G-TA hydrogels were measured at 37°C using an ARES-LS2 rheometer (TA Instruments, New Castle, DE) with a 25 mm stainless steel upper cone and temperature controlled Peltier bottom plate. A 420 µL aliquot of pre-hydrogel solutions with 10 U/mL HRP was loaded onto the Peltier and the cone was lowered to 47 µm. To initiate gelation, 4.2 µL of 1% or 0.5% H₂O₂ was injected into the gap during a 10 s precycle at a steady shear rate of 100/s. The gap was sealed with mineral oil to prevent evaporation during analysis. A dynamic time sweep was performed at 1 Hz with a 1% applied strain for 4,000 s to determine gelation kinetics and storage moduli. Dynamic frequency sweeps (0.1–100 rad/s at 1% strain) and strain sweeps (0.1%– to failure, at 1 Hz) were conducted to analyze elastic behavior of resulting hydrogels (n = 3). Rheological properties were measured in the linear viscoelastic region, where the storage modulus was independent of the applied strain.

Mesh sizes (ξ) of 3% hydrogels were estimated from the storage moduli determined by rheology using the equation below based on the rubber elasticity theory, which assumes purely elastic hydrogels without any physical interactions between the polymer chains

$$\xi = \left(\frac{G' N_A}{RT} \right)^{-\frac{1}{3}}$$

where G' is the storage modulus, N_A is the Avogadro constant, R is the molar gas constant, and T is the temperature [102].

2.4. Attenuated total reflectance -Fourier transform infrared spectroscopy (ATR-FTIR)

Secondary structure of the hydrogels was analyzed using a JASCO FTIR 6200 spectrometer (JASCO, Tokyo, Japan) with a MIRacle™ attenuated total reflection (ATR) with germanium crystal. Hydrogel discs (4 mm diameter, ~2–3 mm height) were prepared by gelling 200 µL aliquots of solutions in polydimethylsiloxane (PDMS) molds incubated in 1 mL DPBS for 1, 7, 14, 21 and 28 days (n=3). After washing in deuterium oxide (Sigma-Aldrich, St. Louis, MO) three times for 30 min each, measurements were conducted by averaging 32 scans with

a resolution of 4 cm^{-1} between 600 and 4000 cm^{-1} . After deconvolution (Supplementary figure Fig.S1), β -sheet content was calculated as the ratio of areas of peak absorbances at 1616 - 1621 , 1622 - 1627 , 1628 - 1637 and 1697 - 1703 cm^{-1} to total area between 1580 - 1720 cm^{-1} .

2.5. Unconfined compression

Unconfined compression was performed using an Instron 3366 Uniaxial Tensile Testing System (Instron, Norwood, MA) with a 10 N load cell. Hydrogel discs (4-5 mm height) prepared with $400\text{ }\mu\text{L}$ solutions in polydimethylsiloxane (PDMS) molds were incubated in 1 mL DPBS for 1, 7, 14, 21 and 28 days ($n=3$). After cutting with an 8 mm biopsy punch, hydrogel discs were placed between stainless steel parallel plates and stress response and elastic recovery were monitored during 30% strain at a rate of $0.667\% \text{ s}^{-1}$. Non-stiffened samples were objected to two pre-cycles at the same strain rate to eliminate artifacts. Tangent moduli were calculated in the linear region between 0 and 10% strain ($n=5$). Day 1 samples were also compressed to 80% strain to monitor the elastic behavior of the hydrogels.

2.6. *In vitro* enzymatic degradation

Hydrogel discs prepared from $200\text{ }\mu\text{L}$ solutions in PDMS molds were incubated in DPBS at room temperature for 4 h. Hydrogels were then transferred into $300\text{ }\mu\text{L}$ of 0.001 U/mL of protease (type XIV from *Streptomyces griseus*, Sigma-Aldrich, St. Louis, MO) dissolved in DPBS and incubated for 1, 2, 4, 6 or 8 days. The enzyme solution was changed every 2 days. After removal of enzyme solution, hydrogels were washed in Ultrapure™ distilled water (ThermoFisher Scientific, Waltham, MA) over night at room temperature, lyophilized and weighed. Results are reported as the mass fraction of the initial weight at day 0 ($n = 4$).

2.7. hMSCs culture on and in hydrogels

Human bone marrow mesenchymal stem cells (hMSCs) (ATCC, Manassas, VA) were cultured in DMEM High glucose supplemented with 10% fetal bovine serum (FBS), 1% Penicillin-Streptomycin (Life Technologies, Carlsbad, CA), 1% non-essential amino acids (Sigma-Aldrich, St. Louis, MO) and 1 ng/mL of fibroblast growth factor-2 (FGF-2) (Invitrogen, Carlsbad, CA). 5% SF/20% SF-TA solutions supplemented with 0, 0.1, 0.25, 0.5 or 1 mM cyclo(RGDyK) and 5% SF/G-TA solutions with G-TA weight ratios of 0, 5, 10, 15, 20, 25% were prepared in 40 mM HEPES and $300\text{ }\mu\text{L}$ aliquots with 10 U/mL HRP and 0.01% H_2O_2 were allowed to cure in 24-well plates for 1 h at 37°C in an incubator. 5% G-TA only hydrogels were prepared using 0.1 U/mL HRP due to very rapid gelation at 1 U/mL or 10 U/mL resulting in non-homogenous gels. Cells at passage 3 were seeded onto tissue culture plastic (TCP) controls and hydrogel surfaces at a density of $4,000\text{ cells/cm}^2$ in 1 mL of growth medium without FBS. At day 1, seeding media was replaced with regular growth media with FBS and it was changed every 3 days. For cell encapsulation, $300\text{ }\mu\text{L}$ pre-hydrogel solutions with 10 U/mL HRP were prepared in 40 mM HEPES or 0.5X DMEM and mixed with cells at passage 3 at a concentration of $2 \times 10^5\text{ cells/mL}$. Cell-laden precursor solutions with 0.01 or 0.005 wt. % H_2O_2 were then allowed to cure in 24 well-plates for 1 h at 37°C in an incubator. The wells were then flooded with 1 mL of culture media and media was changed every 3 days.

2.8. Fluorescence and confocal laser scanning microscopy (CLSM) imaging

To monitor the viability and morphology of hMSCs cultured on and encapsulated in hydrogels, cells were stained with Live/Dead viability kit (Invitrogen, Carlsbad, CA) and imaged with a BZ-X700 Fluorescence Microscope (Keyence Corp., Itasca, IL) or TCS SP8 microscope from Leica Microsystems (Wetzlar, Germany), respectively. After incubating the cells on hydrogel surface with calcein AM and ethidium homodimer-1 (EthD-1) for 20 min or the encapsulated cells for 1 h, samples were washed with DPBS and imaged. Day 1 fluorescence images of the cells cultured on the hydrogels were analyzed using ImageJ (1.48v, NIH, USA) to quantify cell spread areas ($n = 200$).

2.9. Metabolic activity

Metabolic activity of hMSCs cultured on and within hydrogels was determined at days 1, 3, 7, 14, 21 and 28 by alamarBlue viability assay (Invitrogen, Carlsbad, CA) according to the manufacturer's directions. After rinsing with DPBS, cells on hydrogel surfaces or encapsulated in hydrogels were incubated in 500 μL of 10% alamarBlue reagent in DMEM high glucose without phenol red (Sigma-Aldrich, St. Louis, MO) supplemented with 1% non-essential amino acids for 1 h or 3 h, respectively, at 37°C with 5% CO_2 . Following incubation, 130 μL aliquots were transferred into 96 well plates and absorbance at 570 and 595 nm were measured using a microplate reader. Results are reported as the fold change in % dye reduction at day 3, 7, 14, 21 and 28 after normalization to initial measurement at day 1 ($n=4$).

2.10. In vivo animal tests

Hydrogel discs of 8 mm diameter and 4 mm height were fabricated in autoclaved PDMS molds using 200 μL aliquots of 3% SF-TA only or SF/25% G-TA solutions using 10 U/mL HRP and 0.01 wt% H_2O_2 and implanted subcutaneously into 12-week-old female Friend Virus B NIH Jackson (FVB/NJ) mice (The Jackson Laboratory, Bar Harbor, ME) under the protocol approved by the Tufts Institutional Animal Care and Use Committee (M2019-121) ($n = 3$, 2 hydrogel discs per animal). The mice were anesthetized initially at 3% isoflurane and maintained at 2% for the duration of the surgery. Prior to making the incision, the area was shaved, sterilized using disinfectant and ethanol swabs (3x) and a subcutaneous injection of the sustained-release buprenorphine (0.5 mg/mL) analgesic was administered at a dose of 1 mg/kg. Two independent incisions (6-10 mm) were made on the back of each mouse to create a subcutaneous pocket using scissors and a blunt probe. The SF-TA gels were inserted into the upper back pocket (Supplementary Figure S2A) and the SF/25% G-TA discs were inserted in the lower back pocket (Supplementary Figure S2B). Incisions were closed with 1 or 2 stainless steel wound clips. The mice were monitored daily and euthanized by CO_2 asphyxiation after 3 days. After resection and fixation in 4% paraformaldehyde for 48 hours, the hydrogel discs submerged in increasing concentrations of sucrose (15-30% w/v) in PBS for 48 hours, flash frozen using dry ice in optimal cutting temperature (OCT) media and stored at -80°C freezer until cryo-sectioning. The samples were sliced into 15 μm cross-sections using a cryostat and collected on Fisherbrand tissue path superfrost plus gold slides (Fisher Scientific, Hampton, NH). The samples were stained with hematoxylin and eosin (H&E) and imaged using fluorescence microscope.

2.11. Statistical analyses

All data are expressed as means \pm standard deviations for $n = 3$. GraphPad Prism (GraphPad Software, La Jolla, CA) was used to perform One- or Two-way analysis of variance (ANOVA) with Tukey's post hoc multiple comparison test to determine statistical significance (* $p < 0.05$, ** $p < 0.01$, *** $p < 0.001$).

3. Results

3.1. Influence of buffer composition on hydrogel properties and cell encapsulation

The influence of the buffer used for hydrogel crosslinking on the shear mechanical properties of 5% SF hydrogels was assessed through rheology. It is known that salt solutions with higher ionic strength than distilled water induce self-assembly of hydrophobic domains of SF that contains many tyrosine residues into crystalline β -sheets (Fig.S3A). The 5% hydrogels crosslinked in 40 mM HEPES, 0.5X DMEM and particularly 1X PBS started gelling later (Fig.S3B) and had significantly lower storage moduli at $t=4000$ s (1.36, 0.74 and 0.27 kPa, respectively) compared to the hydrogels crosslinked in distilled water (3.1 kPa) (Fig.S3C). To assess the time required for hydrogels to reach mechanical stability, gel point (defined as the time point when the ratio of loss modulus to storage modulus (G''/G') permanently drops below 0.05) of the hydrogels crosslinked in various buffers were compared (Fig.S3D). The 5% SF solutions crosslinked in HEPES, DMEM and PBS reached the gel point at ~ 6 , 8 and 23 min, respectively, indicating a significant delay in gelation, particularly in PBS, compared to crosslinking in distilled water (~ 4 min).

Before encapsulation in the hydrogels, cells were incubated in 5% SF solutions prepared in distilled water, 1X PBS, 40 mM HEPES or 0.5X DMEM for 1 h at room temperature and then seeded on tissue culture plastic (TCP) to analyze the influence of buffer composition on cell viability. Almost no cells incubated in distilled water solution survived as expected due to osmotic stress. The metabolic activity of the cells incubated in HEPES buffer was reduced by half, while no significant reduction was recorded for 1X PBS or 0.5X DMEM compared to the untreated controls (Fig.S3E). Interestingly, when the cells were encapsulated in 5% hydrogels using the same buffer compositions, their viability in 1X PBS group was ~ 2 -fold lower compared to the untreated control cells just like the HEPES group, while no significant reduction was found for 0.5X DMEM group (Fig.S3F). Distilled water and 1X PBS were eliminated from the rest of the study due to inferior cell viability and hydrogel properties, respectively.

3.2. Characterization of SF-TA and G-TA

Tyramine-conjugated biopolymers were obtained by carbodiimide coupling of tyramine groups on SF and gelatin (Fig.1A). After conjugation, UV absorbance between 250 – 300 nm and absorbance peak at 275 nm significantly increased for gelatin (Fig.S4A), while no significant change was recorded for SF due to high background UV absorbance by tyrosine residues. Liquid chromatography tandem mass spectroscopy (LC-MS/MS) analysis (Fig.S4B) was employed to demonstrate the presence of tyramine groups on SF-TA and G-TA, and SF-TA was found to have a significantly larger peak area for tyrosine but a lower peak area for tyramine compared to G-TA. To further confirm tyramine conjugation on SF,

¹HNMR spectra of SF-TA and SF were compared (Fig.S4C) and alkyl proton peaks of tyramine were observed on SF-TA spectrum next to Tyr β proton peaks between 2.4-3.2 ppm. Molecular weight distributions of gelatin and silk before and after tyramine substitution were digitally quantified on polyacrylamide gel images after electrophoresis (Fig.S5A). Using the calibration curve of log molecular weight versus distance from the wells in pixels (Fig.S5B), frequency of each molecular weight in the protein smears were plotted (Fig.S5C). Interestingly, molecular weight with the highest frequency was found to decrease for SF upon tyramine substitution (122 to 57 kDa), while no significant change was observed in the mean molecular weight (SF: 85.5 kDa, SF-TA: 86.7 kDa) (Fig.S5D). For gelatin, on the other hand, molecular weight with the highest frequency shifted from 170 to 246 kDa upon tyramine substitution, while mean molecular weight slightly decreased from 131 to 124 kDa.

3.3. Gelation of SF/SF-TA and SF/G-TA composite hydrogels

Covalent crosslinking of SF with SF-TA and G-TA was expected in the presence of HRP and H₂O₂ through the formation of tyrosine-tyramine linkages as well as dityrosine and dityramine crosslinks (Fig.1B). The sol-gel transition of SF/SF-TA and SF/G-TA composite hydrogels occurred upon mixing with HRP and H₂O₂. Unlike transparent SF only hydrogels, hybrid hydrogels were opaque, the degree of which increased with increasing SF-TA or G-TA content (Fig. 2A). Crosslinking kinetics of the hybrid hydrogels were monitored by quantifying the fluorescence emission by biphenyl-2,2'-diol groups of dityrosine, tyrosine-tyramine and dityramine at 415 nm after excitation at 315 nm. Curves of 5% SF/SF-TA (Fig.2B-i) and SF/G-TA (Fig.2B-ii) composite hydrogels crosslinked in 40 mM HEPES buffer were hyperbolic-like and reached to a plateau faster with increasing SF-TA or G-TA content, while SF only hydrogels displayed a sigmoidal shaped curve and did not reach a plateau at t = 4500 s, indicating that crosslinking was not completed. Relative intensity of fluorescence emission was found to increase significantly with increasing SF-TA content from 0% to 50% and a slight increase was observed in SF with 25% and 50% G-TA hydrogels compared to SF only group (Fig.2B-iii). For 3% hydrogels crosslinked in 0.5X DMEM buffer, all gelation curves including the SF only group had a hyperbolic shape and increasing SF-TA (Fig.2C-i) or G-TA (Fig.2C-ii) content accelerated crosslinking compared to SF only hydrogel. This increase in crosslinking speed, however, did not result in a significant increase in fluorescence intensity (Fig.2C-iii). Despite almost instant gelation, fluorescence emission by G-TA only hydrogels was significantly lower compared to the other groups. LC-MS/MS analysis revealed that the content of dityrosine bonds and unreacted tyrosines decreased while the abundance of tyrosine-tyramine and dityramine crosslinks increased with increasing SF-TA or G-TA ratio in 3% hydrogels (Fig.S6).

The advantage of tyramine substitution on SF over increasing HRP concentration in terms of gelation kinetics was assessed by monitoring gelation of SF only (Fig.S7A-i) and SF-TA only (Fig.S7B-i) solutions in 0.5X DMEM upon induction with HRP concentrations of 10 to 100 U/mL and 0.1 to 10 U/mL, respectively. By increasing the HRP concentration used for gelation of 3% unmodified SF solutions in 0.5X DMEM from 10 to 40 U/mL, the time required for reaching 90% of the maximum fluorescence intensity decreased from ~2400 s to ~1600 s, but it did not change significantly from 40 to 100 U/mL HRP (Fig.S7A-ii). SF-

TA solution, on the other hand, was found to reach 90% of maximum fluorescence in only 690 s at 10 U/mL, and it increased up to 2400 s when HRP concentration was dropped to 0.1 U/mL (Fig.S7B–ii), which was comparable with unmodified SF solution with 10 U/mL HRP. Maximum fluorescence intensities of SF gels at 20 or 40 U/mL HRP were significantly higher than those at 10 or 100 U/mL (Fig.S7A–iii), while a statistically significant sharp and mild increases were recorded for 0.1 to 1 U/mL and 1 to 10 U/mL for SF-TA gels (Fig.S7B–iii).

As a proof of concept, 3% SF-TA solutions prepared in 0.5X DMEM were used for printing of hydrogel patterns and fabrication of hydrogel microspheres using co-flow microfluidics approach. Upon mixing of the streams with HRP and H₂O₂ right before extrusion (Fig.S8A, B) through a 3D bioprinter-compatible (Fig.S8C) 2-inlet needle, the aqueous solution was observed to gel rapidly, resulting in opaque hydrogel patterns on the printing surface (Fig.S8D). Similarly, when extruded through a co-axial needle into a continuous oil phase (Fig.S9A), droplets of SF-TA pre-hydrogel solution with HRP rapidly gelled into individual hydrogel microspheres (Fig.S9B) upon diffusion of H₂O₂ dispersed in the oil phase. Droplets of unmodified SF solution, however, were found to fuse to each other due to slow gelation (data not shown).

Residual H₂O₂ in SF hydrogels supplemented with 30% w/w SF-TA or 25% w/w G-TA was quantified using the tetramethylbenzidine (TMB) colorimetric assay. A calibration curve of H₂O₂ concentration versus optical density at 450 nm (OD₄₅₀) with a linear trend was obtained using a concentration range of 10⁻⁵ (~3.2 μM) to 5x10⁻⁴ wt.% (~163 μM) (Fig.S10A). The concentration of residual H₂O₂ collected by diffusing out by shaking the hydrogel discs or homogenizing them in ultrapure water was found to be below the detection limit of the assay as no significant difference was recorded between the OD₄₅₀ of the blank ultrapure water or the experimental groups (Fig.S10B).

3.4. Shear mechanical properties

The influence of SF-TA and G-TA on shear mechanical properties of composite hydrogels was studied through rheology. For 5% hydrogels prepared in 40 mM HEPES buffer, time point that gelation started and the gel point ($G''/G' < 0.05$) decreased with increasing SF-TA (Fig.3A–i, iii) and G-TA (Fig.3B–i, iii) ratio. Shear storage moduli of the hydrogels at $t=4000$ s were found to increase up to a certain point, reach a plateau and then start decreasing with increasing SF-TA (Fig.3A–ii) or G-TA (Fig.3B–ii). The maximum storage moduli recorded were ~2.2 kPa for SF with 30% SF-TA and ~ 2 kPa for 25% G-TA in comparison to 1.4 kPa for SF only hydrogels. SF with 25% unmodified gelatin started gelling much later and resulted in a significantly lower storage modulus compared to SF only or SF/25% G-TA hydrogels. The same trend of faster gelation and increase in storage modulus up to a certain point with increasing SF-TA or G-TA ratios was also observed in 3% hydrogels crosslinked in 0.5X DMEM (Fig.3C, Fig.S11), but they had lower storage moduli (0.1-0.24 kPa) compared to 5% hydrogels crosslinked in HEPES buffer (1.3-2.1 kPa). The 3% G-TA only hydrogels gelled instantly and displayed the highest storage modulus, but this measurement might not be very accurate as the solution was rapidly crosslinked and the gel was ruptured during 10 s of precycle shearing at high speed. All gels

tested regardless of SF-TA or G-TA content, protein concentration or crosslinking media were frequency independent and withstood at least 100% shear strain, except for the 3% G-TA only hydrogels, which started failing at around 60% strain (Fig.S12A, B).

Rheological properties of the 3% hydrogels crosslinked in 0.5X DMEM (Fig.S13A) also depended on H₂O₂ concentration used for crosslinking. At 0.005 wt.% H₂O₂, G''/G' ratio of SF only hydrogels (Fig.S13B) did not permanently drop below 0.05 and storage modulus fluctuated with increasing frequency. It took much longer for SF/30% SF-TA hydrogels to reach gel point (~400 s) compared to 0.01 wt.% H₂O₂ (~120 s), while it did not change significantly for SF/25% G-TA hydrogels (Fig.S13B). Storage moduli of all 3% hydrogels crosslinked using 0.005% H₂O₂ were lower more than 2-fold compared to 0.01 wt.% H₂O₂ condition (Fig.S13C). Although no significant difference was found between SF/30% SF-TA (0.23 kPa) or 25% G-TA (0.24 kPa) at 0.01% H₂O₂, SF/25% G-TA (0.1 kPa) was significantly stiffer than SF/30% SF-TA (0.03 kPa) and SF only (0.02 kPa) hydrogels at 0.005% H₂O₂. LC-MS/MS analysis showed that the dityrosine, tyrosine-tyramine, and dityramine content gradually decreased when H₂O₂ concentration was lowered from 0.02 wt. % to 0.005 wt. % (Fig.S14). Cyclo(RGDyK) peptides included in 3% SF/30% SF-TA hydrogels were expected to be conjugated to silk chains during crosslinking (Fig.S15A). The hydrogels with 0.25 mM or 0.5 mM cyclo(RGDyK) started to gel (Fig.S15B) and reached the gel point (Fig.S15C) much later, and resulted in significantly lower storage modulus (Fig.S15D) compared to no peptide control.

Mesh sizes of 3% hydrogels were calculated based on rubber elasticity theory. Mesh sizes of the hydrogels crosslinked using 0.01% H₂O₂ were determined to range from 20 to 35 nm, which decreased from 35.1 ± 1.3 nm of SF only to 26.7 ± 2.1 of SF/30% SF-TA but then increased to 32.7 ± 1.7 nm in SF/50% SF-TA; and decreased to 26.3 ± 1.6 nm in SF/25% G-TA (Fig.S16A). G-TA only hydrogels had the lowest mesh size, which was 22.2 ± 0.5 nm. Both decreasing the H₂O₂ concentration to 0.005 wt% and increasing cyclic RGD peptide concentration increased mesh size (Fig.S16B), confirming that crosslinking density was decreased.

3.5. Influence of SF-TA and G-TA content on β -sheet content

IR absorption spectra of hydrogels between 1580-1720 nm (Fig.4A & D, Fig.S17) were analyzed using ATR-FTIR to quantify β -sheet content over 4 weeks of incubation in PBS at 37°C. β -sheet content of 5% hydrogels at day 1 was in the range of 20-35% and it was found to decrease with increasing SF-TA (Fig.4B) or G-TA (Fig.4C) concentration except SF-TA only hydrogels, which had the highest initial content of 32%. At day 7, SF only and SF/SF-TA hydrogels reached a β -sheet content of approximately 40% and remained approximately the same over 4 weeks, while that of SF-TA only hydrogels increased up to 48%. β -sheet content of SF with 10 to 25% G-TA, on the other hand, remained at ~25% for at least 2 weeks and then increased to 35-40% at day 21 (Fig.4C). For the 3% hydrogels prepared in 0.5X DMEM, the increase in β -sheet content over time was slower compared to 5% hydrogels. At day 1, β -sheet content of SF-TA only hydrogels was significantly lower than those of SF only or SF/30%SF-TA hydrogels. The crystalline content remained at around 20% for 3% SF only hydrogels during the first week and gradually increased to ~40% over 4

weeks, while those of SF/30% SF-TA and SF-TA only hydrogels reached above 35% and 40%, respectively (Fig.4E). After 4 weeks, SF-TA only hydrogels reached the highest crystalline ratio of 46% among the other 3% hydrogels. For SF/25% G-TA, however, no significant change was recorded over 28 days and β -sheet content remained significantly lower (~20%) compared to SF only hydrogels.

3.6. Hydrogel morphology and compressive mechanical properties

The 5% and 3% composite hydrogels prepared in 40 mM HEPES and 0.5X DMEM, respectively, were incubated in PBS at 37°C and their morphologies and compressive properties were analyzed over 4 weeks. The SF only, SF/SF-TA and SF-TA only hydrogel discs shrank by day 14 compared to day 1 (Fig.S18). The 5% SF/10% or 25% G-TA hydrogels, however, roughly preserved their initial diameters (Fig.S18, marked with red arrows) by day 14, but they also shrank significantly till day 28. Similar to G-TA only gels, no significant change was observed in the dimensions of 3% SF/25% G-TA hydrogels (Fig.S18, marked with yellow arrows), from day 1 to day 28. Unconfined compression results were parallel with the hydrogel morphologies. Stress-strain curves of all 5% (Fig.5A) and 3% (Fig.5C) hydrogels except 5% SF-TA only indicated elastic behavior with full recovery at day 1, and their compressive moduli ranged between 1.5-3 kPa and 0.3-0.6 kPa, respectively (Fig.5B, D). Initial modulus decreased with increasing SF-TA concentration but was not changed significantly by the G-TA content. 5% SF-TA only hydrogels, however, were already stiff with an average compressive modulus of 21.8 ± 2.7 kPa and displayed a large hysteresis even at day 1. All 5% hydrogels gradually stiffened over 4 weeks with an increase in hysteresis (Fig.5A, Fig.S19), and this stiffening was quicker in SF with 30% or 50% SF-TA (Fig. 5B). At day 7, elastic moduli of SF only and with 10% SF-TA hydrogels were around 4 kPa, while those with 30%, 50% or 100% SF-TA had moduli around 45, 60 and 80 kPa, respectively. At day 14, average moduli of SF only and all SF/SF-TA hydrogels were above 45 kPa and reached 60-100 kPa at day 28, while that of SF-TA only hydrogels was above 300 kPa, which could not be compressed more than 20% strain due to exceeding the detection limit of the load cell. Despite having the highest stiffness, stress-strain curves of 5% SF-TA hydrogels did not display mechanical failure over 4 weeks while 5% SF only or all SF/SF-TA hydrogels did from day 7 to day 28. Moduli of SF with 10% or 25% G-TA were maintained at around 3 kPa at day 14 and increased above 65 kPa at day 21. Among the 3% hydrogels, G-TA only (1.39 kPa) and SF-TA only (0.76 kPa) hydrogels had the highest average moduli at day 1, which were followed by SF with 30% SF-TA (0.59 kPa) or 25% G-TA (0.53 kPa) and SF only (0.34 kPa) hydrogels (Fig.5D). Compressive moduli of SF/30% SF-TA and SF-TA only increased to ~11 and ~26 kPa, respectively, at day 7 and to ~90 kPa at day 28, while it was 3.8 kPa for SF only hydrogels at day 7 and gradually increased to 39 kPa at day 28. After a slight increase from day 1 (0.53 kPa) to day 7 (1.47 kPa), compressive modulus of SF/25% G-TA hydrogels did not change significantly over 4 weeks and was measured as 1.94 kPa at day 28, where hydrogels were still able to recover. No significant change was observed in the compressive properties of G-TA only hydrogels over 28 days of incubation in PBS. At day 1, all 5% and 3% hydrogels except 5% SF-TA only gels were able to withstand 80% compressive strain and displayed elastic behavior to varying degrees (Fig.S20). 5% SF-TA only hydrogels could not be compressed above 70% strain due to the detection limit of the load cell.

3.7. Influence of SF-TA and G-TA content on proteolytic degradation

The influence of SF-TA and G-TA content on enzymatic degradation of composite hydrogels was determined by monitoring the percent initial mass over 8 days of incubation in protease XIV solution (Fig.6). Increasing SF-TA or G-TA content was found to delay the degradation of 5% hydrogels crosslinked in HEPES buffer. Less than 5% of the initial mass of SF only hydrogels remained at day 4 while those of SF/50% SF-TA and SF/25% G-TA were around 60% and 50% their original weights at day 4, and 30% and 15% at day 8, respectively (Fig.6A). SF-TA only hydrogels degraded significantly slower and maintained ~65% of its initial weight at day 8. A similar trend was also recorded for the 3% hydrogels crosslinked in 0.5X DMEM buffer, but their degradation was slower compared to 5% hydrogels (Fig.6B). Less than 40% and 5% of the SF only hydrogels were left at day 4 and 8, respectively, while SF/30% SF-TA and SF/25% G-TA were approximately 60% and 30% of their initial weights at days 4 and 8, respectively. SF-TA only hydrogels, on the other hand, were ~48% of their original weight at day 8, while G-TA only hydrogels displayed the fastest degradation and completely dissolved away by day 6.

3.8. Effects of cyclo(RGDyK) and G-TA on response of hMSCs cultured on hydrogel surfaces

To determine the effects of cyclo(RGDyK) (0.1-1 mM) and G-TA (5-25%) on cell behavior, the area of hMSC spreading at day 1 and metabolic activity over 4 weeks of culture were analyzed. Cells were seeded onto hydrogels in FBS free medium immediately after crosslinking to evaluate the cell-surface interactions without adsorption of any serum proteins onto the surfaces as they guide cell adhesion and fate of adherent cells [103]. Many cells on no RGD hydrogels (controls) displayed a round morphology at day 1 while those on the hydrogels supplemented with cyclic RGD or G-TA flattened and spread well (Fig.7A). The highest and lowest cell spreading were observed on G-TA only and SF only hydrogels, respectively, and the area of cell spreading increased and reached a plateau with increasing cyclo(RGDyK) or G-TA concentrations in composite hydrogels (Fig.7B). The maximum spread area on SF/G-TA hydrogels was much higher than that on SF/SF-TA with cyclic RGD. At day 14, cells on all hydrogels spread well and reached confluency at day 28 except the ones on G-TA only gels (Fig.S21). Metabolic activity at day 1 decreased (Fig.S22A) but the fold change over 4 weeks increased significantly with increasing cyclo(RGDyK) concentration. A slight decrease in metabolic activity was also recorded with increasing G-TA ratio at day 1 (Fig.S22B), but the fold change over 4 weeks increased significantly even though a plateau was reached after 15% G-TA (Fig.7C). The fold change in metabolic activity on G-TA only gels was significantly higher than the other groups during the first two weeks but leveled with others at day 14. At day 28, the highest average fold change was observed on G-TA only hydrogels (4.88 ± 0.96 -fold) followed by SF/20% G-TA (4.63 ± 0.13 -fold), which were higher than that recorded on SF/SF-TA with 1 mM cyclo(RGyK) (3.92 ± 0.09 -fold).

3.9. *In vitro* response of encapsulated hMSCs

hMSCs were encapsulated in SF/SF-TA with or without cyclo(RGDyK) and SF/G-TA composite hydrogels and their viability and morphologies were monitored over 4 weeks.

First, cells were encapsulated in 5% hydrogels crosslinked in HEPES buffer to determine whether accelerated gelation with increasing SF-TA ratio improved initial viability. Indeed, fewer dead cells were observed in the hydrogels (Fig.8A–i) and % dye reduction at day 1 increased significantly with increasing SF-TA concentration (Fig.8A–ii). At day 7, however, metabolic activity decreased in all groups compared to day 1 and the fold change in metabolic activity for SF/30%SF-TA was 0.18 at day 28 compared to day 1 (Fig.8A–iii). The same decrease over time was also observed with the SF/25% G-TA hydrogels, but the fold change over 28 days (0.41-fold) was higher than those in SF only (0.21-fold) SF/30% SF-TA hydrogels. When hydrogels were crosslinked in 0.5X DMEM instead of 40 mM HEPES (Fig.8A–iv), the fold change was still below 1 in SF only (0.64-fold) and even lower in SF/30% SF-TA hydrogels (0.41), while a 1.4-fold increase was recorded in SF /25% G-TA hydrogels at day 28. 0.5X DMEM was used for preparation of hydrogel precursor solutions for the rest of the study.

To determine whether lowering protein and/or H₂O₂ concentration improves long term cell response, morphologies and metabolic activity of hMSCs encapsulated in 3% composite hydrogels crosslinked with 0.01 wt.% or 0.005 wt.% H₂O₂ were monitored over 28 days. The SF only group was omitted since the cells settled to the bottom of the plate and proliferated there due to the slow gelation. The majority of the cells encapsulated in the SF/SF-TA hydrogels crosslinked using 0.01% H₂O₂ remained spherical over 4 weeks, while some cells in SF/G-TA hydrogels and almost all cells in G-TA only gels had cytoplasmic extensions branching from their bodies at day 28 (Fig.8B–i, Fig.S23). No significant change was found in metabolic activities in any SF/SF-TA or SF/15% G-TA hydrogels over 4 weeks, while there was a statistically insignificant 1.5 fold increase in SF/20% G-TA hydrogels and statistically significant 1.9-fold and 2.9-fold increase in G-TA only and SF/25% G-TA hydrogels, respectively (Fig.8B–ii). When the H₂O₂ concentration was lowered to 0.005 wt.%, however, more cells exhibited elongated or branched morphologies, particularly in SF/G-TA hydrogels. An increase in % dye reduction up to 4-fold compared to day 1 was recorded in SF hydrogels with 15-25% G-TA, while the increase in G-TA only gels was still around 2-fold. Again, no significant change was observed in SF/SF-TA hydrogels (Fig.8B–iii).

Cell response to SF/SF-TA hydrogels with cyclic RGD peptides was also determined by monitoring cell morphology and metabolic activity over 28 days. At both 0.01 and 0.005 wt.%, more and more cells displayed cytoplasmic extensions with an increase in cyclo(RGDyK) concentration (Fig.8C–i, Fig.S24A). No significant change was found in metabolic activity over 28 days in any group at 0.01 wt.% H₂O₂ (Fig.S24B), while 1.6 and 2.7-fold increases were recorded in the hydrogels supplemented with 0.25 and 0.5 mM cyclo(RGDyK), respectively at 0.005 wt. % H₂O₂ (Fig.8C–ii).

3.10. *In vivo* response to SF-TA only and SF/G-TA composite hydrogels

Preformed hydrogel discs of 3% w/v SF-TA only and SF/25% G-TA were implanted subcutaneously in mice and no adverse effects from the implants were observed over 3 days. The samples were extracted at 3 days post-implantation (Fig.9A, B) and examined histologically. H&E staining showed formation of a thin inflammatory capsule surrounding

the gels (pointed with black arrows) indicating a typical foreign body response without excessive inflammation or cell infiltration into the bulk of the gels (Fig.9C, Fig.S25). SF/25% G-TA discs mostly remained intact (Fig.9C–i) while SF-TA only hydrogel discs broke up into several pieces (Fig.9C–ii).

4. Discussion

Enzymatically crosslinked SF/SF-TA and SF/G-TA hydrogels were characterized for gelation kinetics, mechanical properties, secondary structure and enzymatic degradation. In addition, the influence of cyclic RGD or G-TA on the bioactivity of composite hydrogels were explored for the culture and encapsulation of hMSCs. In the presence of HRP and H₂O₂, aqueous solutions of SF formed highly elastic and biodegradable hydrogels through oxidation of tyrosine residues into dityrosine crosslinks; complete gelation took 10 to more than 30 min in distilled water depending on the average molecular weight of the regenerated SF chains [21, 23]. Even though cell encapsulation in distilled water was cytocompatible when the cells were added shortly before gelation was complete [21], this approach is impractical for 3D bioprinting or microencapsulation, where cells may need to be incubated in the pre-hydrogel solution for extended periods of time, thus, the ionic strength of the solution must be controlled carefully [104, 105]. Crosslinking of SF in physiological buffers, however, takes much longer for gelation start and completion, and results in mechanically weaker hydrogels likely because of the high ionic strength. Solutions of the salts of Na⁺ and K⁺ have been shown to induce the self-assembly of the hydrophobic domains in SF sequence into β -sheets through salting out effect [25, 27], which has been utilized before for the fabrication of physically crosslinked silk microparticles [106] or aqueous-derived silk scaffolds [26] using high concentration salt solutions. In the case of enzymatic crosslinking, however, physiological buffers with relatively high ionic strength could be expected to reduce the accessibility of tyrosine residues available in these hydrophobic domains, impairing the gelation speed as well as the overall crosslinking density. Slow gelation is an issue for tissue fillers, bioinks for 3D bioprinting, and cell encapsulating microgels, where rapidly gelling hydrogel materials with high structural integrity are desired [14, 104, 107]. One of the most commonly used rapidly gelling hydrogel system in these applications has been the ionically crosslinked polysaccharide alginate [108–110], which have very limited stability at physiological conditions due to the ongoing ion exchange mechanisms and their structural and mechanical integrities have been shown to deteriorate very quickly [111, 112]. Slow degradation and long-term structural stability [113] of silk hold promise as an alternative hydrogel system and therefore the gelation kinetics of enzymatically crosslinked silk hydrogels in physiological buffers should be improved for widespread biomedical applications.

Our first approach to improve gelation kinetics was to conjugate tyramine residues onto SF to increase the total phenol content. Moreover, we hypothesized that tyramine conjugation on aspartic and glutamic acid residues that are located in the amorphous regions of SF [114] would provide more accessible phenol groups for crosslinking at high ionic strength. Despite the low theoretical increase in the number of phenol groups upon 100% conversion of all carboxylic acid residues (1.1 mol% [115]), even 20 wt. % SF-TA significantly decreased the time required for the initiation of gelation, from 5-10 min to tens of seconds. Gelation

kinetics of aqueous silk solutions could also be modulated by changing the concentrations of H₂O₂ and/or HRP. Besides potential cytotoxicity [39], high concentrations of H₂O₂ could also impair gelation as it was shown that increasing it above 2.4 mM (~0.007 wt%) gradually decreased crosslinking density of 2.5% w/v SF hydrogels, likely because of the inhibition of the enzymatic activity of HRP [37]. Using high concentrations of HRP, on the other hand, may raise concerns of immune response since HRP is a plant-derived enzyme and both its protein core and N-glycans were shown to induce immune reaction in rodents [36]. Moreover, we showed that tyramine-substitution enables gelation speeds at a moderate concentration (10 U/mL) of HRP that cannot be achieved with unmodified silk using a high (100 U/mL) concentration of HRP. An important advantage of SF-TA over other tyramine-substituted polymers is the control over gelation kinetics through modulation of SF and SF-TA ratios. This control is not possible with other polymers such as gelatin, HA or PEG, which have few or no phenol residues, therefore unmodified polymers cannot be crosslinked with tyramine-modified chains. SF, on the other hand, is already rich in phenol residues because of its high tyrosine content, which enables crosslinking of unmodified SF with SF-TA at any weight ratio to modulate gelation kinetics.

As an alternative strategy, we crosslinked SF with tyramine-substituted gelatin (G-TA) to improve both bioactivity and gelation kinetics of SF hydrogels. Dityrosine crosslinking of SF with gelatin was demonstrated before through tyrosinase treatment, but complete gelation took more than 40 min [91]. Here we also found that unmodified gelatin increases the gelation time and deteriorates shear mechanical properties of enzymatically crosslinked silk hydrogels due to relatively low tyrosine content (0.26 mol%) of porcine skin gelatin [116] compared to SF. After chemical modification, G-TA had a significantly higher tyramine content than SF-TA because of higher aspartic and glutamic acid content of porcine skin gelatin (12.4 mol% [116]) available for carbodiimide coupling compared to SF (1.1 mol% [115]). Since both SF and gelatin have free amine and carboxylic acid residues in their primary structure, crosslinking of polymer chains during carbodiimide coupling of tyramine groups could be an issue. To prevent this, tyramine was used at molar excess of aspartic and glutamic acid residues on SF (~15X) and gelatin (~3X), and it was added to protein solutions before EDC and NHS to ensure carboxylic acids would already be saturated by amine groups of the tyramine residues upon activation by EDC and NHS. Still, gel electrophoresis showed that although mean molecular weight did not significantly changed for neither groups, molecular weight with highest frequency increased significantly for gelatin, suggesting an increase in chain length likely because of crosslinking. Interestingly, molecular weight with highest frequency in the distribution of SF-TA chains was remarkably lower compared to unmodified silk. This could be explained with precipitation and removal of the crosslinked large chains due to low solubility of SF. G-TA increased the gelation speed of silk hydrogels to a higher degree than SF-TA at the same weight ratio, as expected. A similar increase in gelation rate was also reported before for SF crosslinked with HA-TA, but high viscosity and hydrophilicity of HA were stated as issues for effective mixing of the two polymers at high HA-TA ratios [101]. Unlike HA-TA, G-TA had a low viscosity at 37°C and was completely miscible with SF-TA at any ratio. At room temperature, on the other hand, solutions with final G-TA solutions above 1% w/v undergoes thermal gelation similar to unmodified gelatin. Our results indicate that oxidative crosslinking of silk with G-TA

through tyrosine-tyramine bridges is more advantageous over other strategies such as genipin-mediated crosslinking of primary amine residues, completion of which was found to increase up to 48 h with increasing SF ratio due to low basic amino acid content of SF [90]. Oxidative crosslinking of SF with SF-TA or G-TA introduced in this study is also compatible with enzyme-free fabrication strategies such as using syringes packed with HRP-immobilized beads [117], riboflavin-mediated photocrosslinking [118] or Fe(III) ion-mediated Fenton reaction [119].

Besides improved gelation kinetics, one of the most significant effects of the SF-TA and G-TA content on SF hydrogels was the improved shear mechanical properties. Both tyramine-substituted polymers decreased the time required for gels to reach mechanical stability and increased their storage modulus up to a point, suggesting an increase in crosslinking density [101, 120]. This was confirmed by LC-MS/MS analysis revealing the presence tyrosine-tyramine and dityramine crosslinks in addition to dityrosine bonds. Interestingly, the increase in storage modulus was not linearly correlated with polymer-TA concentration, as it reached a plateau at ~20 wt. % SF-TA or G-TA and then sharply dropped at higher ratios. This is likely because of very rapid local gelation that prevents homogenous diffusion of H₂O₂ and limits chain mobility required for collision of phenolic radicals to form new crosslinks. Decreasing H₂O₂ concentration by half decreased crosslinking density and resulted in higher gelation times and lower storage moduli. The same trend was reported before for tyramine-substituted polymers such as PGA-TA [43] or HA-TA [121] as well as SF [101], which was explained with a decrease in H₂O₂/phenol mol ratio and the number of phenolic radicals available for crosslinking. SF-TA and particularly G-TA significantly improved shear mechanical properties at low H₂O₂ concentration as well, indicating that improved gelation was independent of H₂O₂ concentration. Another parameter that influenced the shear mechanical properties of SF/SF-TA hydrogels was the cyclic RGD content, which gradually increased gelation time and decreased storage moduli at increasing concentrations. This was an expected outcome since some of the phenolic radicals generated on the polymer chains are occupied by the tyrosine residues of the peptides, resulting in a decrease in crosslinking density. The same observation was also made by others for *in situ* conjugation of peptides into hydrogels of acrylate-crosslinked HA [74] and dithiol crosslinked PGA [71] as well as dityramine crosslinked tetronic-tyramine [77], gelatin-PEG-tyramine [122] and HA-tyramine hydrogels [78, 79]. In the SF/SF-TA system, the decrease in crosslinking density with increasing peptide concentration can be compensated by increasing the weight ratio of SF-TA to obtain desired gelation kinetics and mechanical properties.

Another notable effect of the polymer-tyramine content was the β -sheet content and gradual stiffening of SF hydrogels incubated in PBS. Stiffness of hydrogel matrices is an important parameter as it is sensed by the cells through mechanotransduction, which alters gene expression and influences morphology, proliferation and lineage commitment [123–125]. Modulation of the mechanical properties of enzymatically crosslinked silk hydrogels has been demonstrated before through changing the average molecular weight (boiling time), protein concentration [21], or the amount of H₂O₂ used for crosslinking [37]. Here we report modulation of secondary structure and stiffness of SF hydrogels by altering SF-TA or G-TA content without changing any other parameter. Initially, all 3% and 5% hydrogels except 5%

SF-TA only were very elastic at day 1 with maximum compressive moduli of 0.6 and 3 kPa, respectively, and could withstand 80% of compressive strain without failure. SF-TA ratio in SF/SF-TA composites was negatively correlated with the β -sheet content and compressive moduli of 5% hydrogels at day 1, likely because of increased crosslinking density that was proposed to restrict movement of SF chains in the network required for the formation of β -sheet domains [126]. Both SF only and SF/SF-TA composite hydrogels shrank and stiffened over 4 weeks with an increase in compressive modulus up to 100 kPa as a result of a gradual increase in β -sheet ratio. Stiffening of enzymatically crosslinked SF hydrogels has been reported before as a result of self-assembly of hydrophobic silk domains into crystalline β -sheet structures [101, 127, 128], which can be explained with SF chains in the gel network being still partly movable and capable of rearranging into crystalline domains [126]. Despite initially providing lower β -sheet content and stiffness, increasing SF-TA content accelerated β -sheet formation and stiffening compared to SF only control likely because of an increase in hydrophobicity of SF chains upon conversion of the charged carboxylic acid residues into hydrophobic phenol groups. 5% SF-TA only hydrogels, for instance, were found to stiffen rapidly even in a single day. Higher hydrophobicity might have favored water removal and formation of β -sheets through a similar mechanism with silk self-assembly induced by protonation of carboxylic acid residues at acidic pH [129, 130]. Moreover, the increase in the number of phenol residues on SF-TA chains might have enhanced π - π and π -OH group interactions, which were proposed to provide templating effects and foster silk self-assembly [131]. Unlike SF-TA, G-TA significantly delayed crystallization and stiffening of silk hydrogels. Hindered self-assembly of silk in physically crosslinked composite hydrogels has been reported before as a result of the topological constraints caused by entanglements with gelatin molecules [83, 86, 87]. Moreover, higher hydrophilicity of gelatin because of its significantly larger content of charged amino acids (26.2 mol%) [116] compared to SF (1.7 mol%) [115] might have provided higher water retention, which was shown to delay β -sheet formation in SF crosslinked with HA-TA that has a large negative charge [101]. Our findings indicate that a wide range of hydrogel stiffness with compressive moduli ranging between 1-100 kPa could be achieved simply by altering SF-TA or G-TA content. These hydrogels can be utilized as scaffold materials that favor stem cell differentiation towards a wide range of lineages including neurogenic (0.1-1 kPa), adipogenic (1-5 kPa), myogenic (10-20 kPa), and osteogenic (>25 kPa) cells [123, 132].

SF-TA and G-TA content modulated the enzymatic degradation of the composite hydrogels. Spatial and temporal control of biodegradation is an important parameter in terms of tissue regeneration and potential release of biochemical agents or encapsulated cells. Tissue engineering scaffolds should be able to guide cell growth and allow new ECM deposition while gradually degrading in the body until the original tissue architecture is restored and delivery vehicles should sustain desired release kinetics [133]. High crystalline content of silk-based hydrogels has been shown to provide slower degradation compared to other common biopolymers such as collagen/gelatin, elastin or HA due to its high crystalline content [102]. To determine the degradation behavior of composite hydrogels we incubated them in a solution of protease XIV, an enzyme that was shown to favor the degradation of non- β -sheet regions in silk hydrogels [134], and found that both SF-TA and G-TA reduced degradation rates, likely because of higher crosslinking densities of the composite hydrogels

than SF only control. Reduced enzymatic degradation rate with increased crosslinking density was reported before for enzymatically crosslinked silk hydrogels [135] as well as many other covalently crosslinked polymer networks including gelatin [40], HA [35], and PEG [136]. This can be explained with a decrease in average mesh size with increased crosslinking density [137], leading to reduced enzyme diffusivity into the gel. Interestingly, 3% hydrogels degraded more slowly compared to their 5% counterparts, which might seem to contradict the faster enzymatic degradation of physically crosslinked silk hydrogels at lower protein concentrations [33]. However, crosslinking density of enzymatically crosslinked silk hydrogels decreases with increasing protein concentration above 1.25% (w/v) as a result of lower chain mobility and limited collisions of phenolic radicals. Diffusivity of enzymatically crosslinked silk hydrogels was therefore found to increase above 1.25% silk [37]. Despite having the lowest mesh size, 3% G-TA only hydrogels completely degraded in 6 days, while SF-TA only hydrogels had a larger calculated mesh size but degraded the slowest. This observation can be explained with both the lack of protease sensitive sequences in SF primary structure and its high crystalline content that limits water retention and diffusivity. It should also be noted that the mesh sizes were calculated for day 0 samples based on rubber elasticity theory, which assumes no physical interactions between the polymer chains [102]. Therefore, it would not be accurate to apply it to silk hydrogels at later time points due to physical crosslinking through β -sheet formation.

To test the cell-matrix interactions of hydrogels supplemented with cyclic RGD peptides or with G-TA, hMSCs were first cultured on the hydrogel surfaces and their viability, spreading and metabolic activity were analyzed. RGD peptides regulate integrin mediated focal adhesion formation followed by cell adhesion, spreading and organization of actin cytoskeleton, which further modulate survival, proliferation, and differentiation via mechanotransduction [64]. We preferred a cyclic RGD peptide rather than a linear peptide because cyclic RGD peptides were found to improve hydrogel-matrix interactions to a higher degree compared to linear counterparts [67, 138] as they bind selectively to $\alpha_v\beta_3$ and $\alpha_v\beta_5$ integrins with higher affinity and display lower susceptibility to enzymatic degradation [139]. Even though tethering of linear RGD peptides improved cell-matrix interactions on silk films [63, 140, 141], fibers [142], and nanofibrous mats [143], conjugation of linear or cyclic RGD peptides into physically or enzymatically crosslinked silk hydrogels has not been reported before. The only example to our knowledge is the ionically crosslinked silk-alginate hydrogels supplemented with cyclic RGD peptides, which was shown to provide better adhesion of mouse embryonic stem cells compared to a no RGD control [144]. Here we showed that cyclic RGD peptides significantly improved initial cell spreading and the metabolic activity over 4 weeks on SF/SF-TA hydrogels. This is parallel with earlier studies reporting improved adhesion of murine osteoblasts on elastin-mimetic polypeptide hydrogels [138], better adhesion, spreading, and proliferation of endothelial cells on PEG diacrylate (PEGDA) hydrogels [67], and higher adhesion and proliferation of murine fibroblasts on PEGDA [70] and PGA [71] hydrogels after covalent conjugation of cyclic RGD peptides. Similarly, HRP-mediated conjugation of cyclo(RGDyK) on surface deposited resilin-like peptides was shown to improve fibroblast attachment significantly through the binding of integrin α_v chains [145]. Although we observed no significant cytotoxicity with live/dead staining, initial metabolic activity decreased gradually with increasing cyclic RGD

concentration, which might be due to the unconjugated peptides that diffused out and reduced cell attachment by blocking the surface integrins [69]. Crosslinking SF with G-TA improved cell spreading and metabolic activity of hMSCs on hydrogels to a higher extent than cyclic RGD peptides. Improved cell-hydrogel interactions upon blending silk with gelatin was also reported by others before. For example, Das et al. demonstrated that goat chondrocytes displayed prominent adhesion with many filopodial extensions and reach confluency earlier on physically crosslinked SF-gelatin hydrogels compared to SF only controls [75]. Similarly, attachment, spreading, and proliferation of human MG63 osteosarcoma cells increased with increasing gelatin content in silk-gelatin blend hydrogels [87]. Increasing the concentration of tyramine-substituted gelatin also improved adhesion, spreading, and proliferation of human aortic endothelial cells and murine embryo fibroblast on gelatin-alginate composite hydrogels [98]. The increase in cell spreading and metabolic activity recorded on SF/G-TA hydrogels, however, was not linearly correlated with G-TA concentration; it reached a plateau and then started declining after 15% G-TA, probably due to the increase in modulus, which was reported to reduce spread area and proliferation of hMSCs on hydrogel surfaces [124].

Encapsulating cells in hydrogel scaffolds has been an attractive approach for tissue engineering applications by providing a biodegradable and highly hydrated tissue-like environment for the growth of cells and tissues. Since the cells to be encapsulated are present in the pre-gel solution during the crosslinking process, pre-gel formulations and crosslinking chemistries should be cytocompatible [6]. Here we initially used 40 mM HEPES solution, a non-toxic tissue culture buffer [146], to maintain a constant physiological pH of SF solution without significantly increasing the ionic strength. SF solution in HEPES buffer, however, turned out to be hypotonic and could not maintain high viability of suspended or encapsulated cells. Interestingly, increasing SF-TA content significantly improved initial viability of hMSCs encapsulated in 5% hydrogels, likely because of faster gelation that lowered the duration of exposure to osmotic stress. However, cell viability gradually decreased in SF only or composite hydrogels from day 1 to 7, suggesting a long-term damage caused by initial osmotic stress. When the buffer was replaced with 0.5X DMEM to provide better physiological conditions [21, 24], there was still a decrease in metabolic activity in SF only and particularly in SF/SF-TA composite hydrogels as opposed to a slight increase observed in SF/G-TA hydrogels. This trend suggests that β -sheet formation and stiffening, which were enhanced by SF-TA but delayed by G-TA, impaired growth of encapsulated hMSCs. This was not unexpected as it was reported before that β -sheet transition in enzymatically crosslinked SF hydrogels led to an increase in hydrogel stiffness and induced apoptosis of encapsulated cancer cells [124] or limited the growth of encapsulated cardiac fibroblasts [117]. Decreasing protein concentration to 3% improved metabolic activity of encapsulated cells, likely because of a decrease in hydrogel stiffness [21] as well as an increase in hydrogel porosity [127], which was shown to enhance cell proliferation in hydrogel matrices [147]. The most remarkable improvement in the metabolic activity of encapsulated cells, however, was obtained when H_2O_2 concentration was reduced by half. It is known that H_2O_2 may induce apoptotic cell death even at low (0.1-10 mM) concentrations [39], but its cytotoxicity was shown to be minimized if rapidly consumed by the HRP mediated oxidative crosslinking reaction [36]. The concentration of residual H_2O_2

in hydrogels crosslinked using 0.01 wt% H₂O₂ could not be detected using TMB assay, suggesting that it was rapidly consumed by HRP and was lower than 3.2 μM, which was the detection limit we recorded. Indeed, cytotoxicity of the initial H₂O₂ concentration was not an issue as the metabolic activity at day 1 was not improved when H₂O₂ concentration was reduced to 0.005 wt%. The increase in fold change in metabolic activity over time could be attributed to lower crosslinking density at lower H₂O₂ concentrations [37] that results in a decrease in hydrogel modulus and an increase in mesh size [121], which were reported to improve proliferation and spreading of encapsulated cells [148, 149]. The fold change in metabolic activity in SF hydrogels was not affected significantly by SF-TA content, but it increased significantly when pre-gel solution was supplemented with 0.5 mM cyclic RGD peptide, indicating signs of cell growth. With increasing cyclic (RGDyK) peptide concentration, encapsulated cells also started to spread and displayed long cytoplasmic extensions. These observations are parallel with previous studies, where encapsulated MSCs were shown to display higher viability [66, 68] and spreading [74] in RGD-tethered PEG and HA hydrogels compared to no RGD controls, which could be explained with higher expression of integrins [57, 65]. It is likely that the decrease in crosslinking density with increasing cyclic peptide that is expected to result in larger mesh size also enhanced cell growth and spreading further. Despite increasing the crosslinking density and storage modulus, G-TA improved metabolic activity even to a higher degree than cyclic RGD peptides and provided cell spreading to some degree, particularly at low H₂O₂ concentration. An increase in spreading and proliferation upon crosslinking with gelatin-tyramine was also reported for myoblasts [61] and dermal fibroblasts [100] encapsulated in HA-tyramine and PEG-tyramine hydrogels, respectively, indicating that the intrinsic RGD peptides in gelatin sequence improve cell-matrix interactions and cell growth. In another study, MG-63 osteoblast-like cells were shown to spread and proliferate better in gelatin-alginate composite hydrogels compared to alginate only or RGD-conjugated alginate hydrogels [150], which is parallel with our observations. Besides providing cell-adhesive sequences, G-TA also inhibited crystallization and stiffening of 3% composite hydrogels and maintained a well-hydrated structure over 4 weeks of culture, which probably contributed to better growth of hMSCs that were shown to proliferate faster in softer hydrogel matrices [124, 151]. Interestingly, the fold change in metabolic activity of encapsulated hMSCs over 4 weeks of culture was significantly higher in SF/25% G-TA composite hydrogels than in G-TA control, likely because of the highest crosslinking density and therefore smallest mesh size of G-TA only gels among all groups, which might have limited cellular activity within the polymer network. Still, the cells in G-TA only gels were observed to spread earlier and more than SF/25% G-TA at both 0.01 and 0.005 wt% H₂O₂, suggesting that the primary sequence of protein network played a more important role in the morphology of encapsulated cells than the mesh size.

Overall, our results indicate that SF/SF-TA with cyclic(RGDyK) and SF/G-TA composite hydrogels have several advantages over G-TA only hydrogels. Firstly, gelation kinetics, mechanical properties and bioactivity of SF/SF-TA/cyclic(RGDyK) and SF/G-TA composites can be tuned by changing the weight ratios of individual components, while G-TA only solutions gel very rapidly even at very low HRP concentrations, making it hard to obtain homogenous hydrogels due to improper mixing. Secondly, unlike aqueous G-TA

solutions, SF/SF-TA formulations or SF/G-TA blends with a G-TA weight ratio of 10 to 25% do not display thermal gelation. This enables storage and handling of aqueous solutions at room temperature without a significant increase in solution viscosity, which could otherwise cause issues during filter sterilization as well as 3D printing and microfluidic applications, such as a decrease in cell viability as a result of higher shear stress^[152] or a drop in the efficiency of microfluidic fabrication of monodisperse droplets^[153], respectively. Finally, SF/SF-TA and SF/G-TA composite hydrogels were found to be more resistant to enzymatic degradation, and SF/G-TA composite hydrogels allowed for a significantly higher increase in the metabolic activity of the encapsulated hMSCs compared to G-TA only hydrogels, suggesting that they could be a better alternative as tissue engineering scaffolds with longer-term stability to guide cell growth and tissue organization until native tissue is regenerated. SF/SF-TA/cyclo(RGDyK) composite gels also led to a higher fold change in metabolic activity compared to G-TA only controls and they could be utilized in hard-tissue engineering applications^[154] or disease models^[155] considering the self-assembly and gradual stiffening of SF/SF-TA or SF-TA only gels in physiological buffers, a pattern that is not observed with G-TA-based hydrogels.

5. Conclusions

Gelation kinetics, mechanical properties, enzymatic degradation and bioactivity of silk hydrogels were successfully modulated by combining SF with SF-TA and cyclic RGD, or only with G-TA. Both SF-TA and G-TA increased gelation speed and mechanical stability and slowed the enzymatic degradation of silk hydrogels crosslinked in physiological buffers. Temporal crystallization and stiffening of silk hydrogels were enhanced by SF-TA but delayed by G-TA, enabling modulation of mechanical properties within a wide range suitable for both soft and hard tissue engineering applications. We were able to modulate the bioactivity of the silk hydrogels by changing the cyclo(RGDyK) or G-TA content, which enables modification of cell microenvironment for specific scaffold or cell delivery applications. Our future studies will focus on utilizing these composite hydrogel formulations as bioinks for 3D bioprinting of cell-laden scaffolds and as mammalian cell-coating sheets using microfluidic encapsulation.

Supplementary Material

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Acknowledgements

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Statement of Significance

HRP-mediated covalent crosslinking of aqueous silk solutions allows for fabrication of highly elastic, mechanically tunable, degradable and biocompatible hydrogels. However, slow gelation and impaired mechanical properties of silk hydrogels enzymatically crosslinked in physiological media as well as the lack of cell-recognition sequences in silk fibroin limit potential cell encapsulation applications. In the present study, crosslinking of regenerated silk with tyramine-substituted silk or gelatin was performed for to improve gelation kinetics and mechanical properties of silk hydrogels by increasing the number of phenol groups available for enzymatic crosslinking. Moreover, *in situ* conjugation of cyclic RGD peptides into silk hydrogels was employed to enhance cell-matrix interactions. Our findings indicate significant improvement of gelation kinetics and modulus, as well as modulation of silk secondary structure and enzymatic degradation of composite hydrogels in addition to enhanced bioactivity provided by cyclic RGD peptides or tyramine-substituted gelatin. Silk hydrogel formulations introduced here hold great promise as injectable cell-laden tissue fillers and as bioinks for cell encapsulation through 3D bioprinting or droplet-based microgel fabrication.

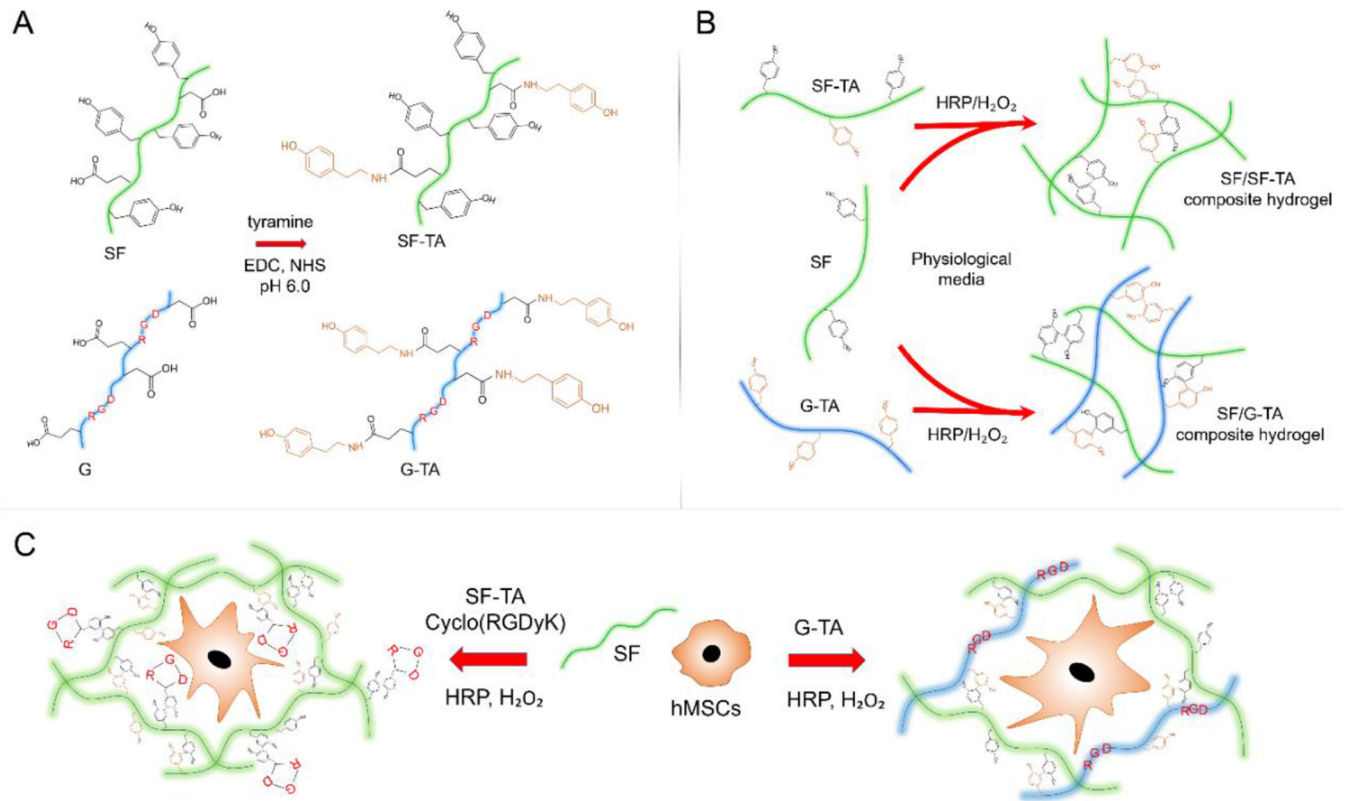


Figure 1.

Schematic representation of the SF/SF-TA and SF/G-TA composite hydrogel systems for cell encapsulation. **(A)** Carbodiimide coupling of tyramine (TA) groups to carboxylic acid residues on silk fibroin (SF) and gelatin (G) chains. **(B)** HRP-mediated covalent crosslinking of SF with SF-TA or G-TA. **(C)** Encapsulation of hMSCs within enzymatically crosslinked SF/SF-TA hydrogels with *in situ* conjugated cyclo(RGDyK) or in SF/G-TA composite hydrogels. SF: silk fibroin, G: gelatin, SF-TA: tyramine-substituted silk fibroin, G-TA: tyramine-substituted gelatin.

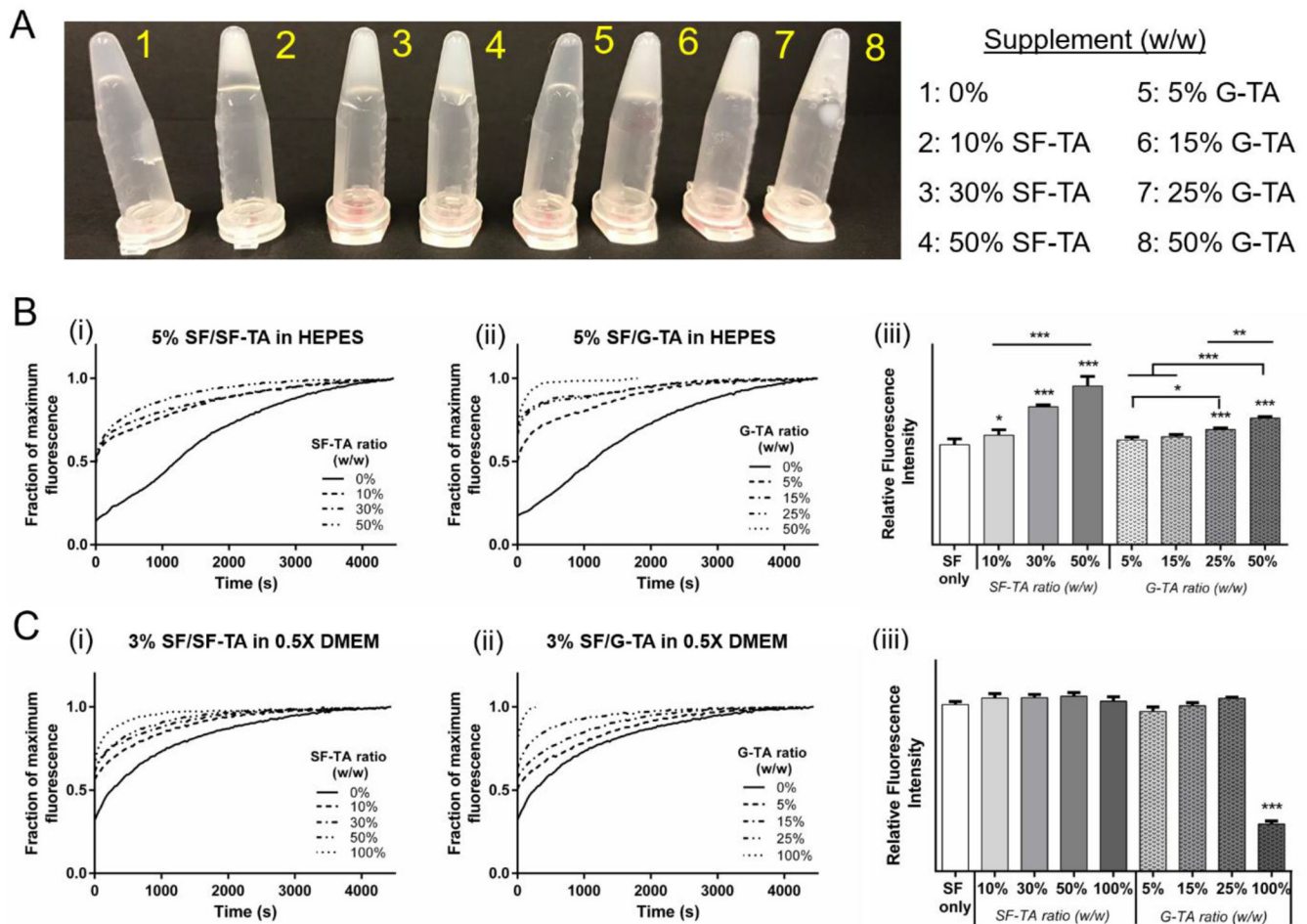


Figure 2. Gelation of silk hydrogels supplemented with tyramine-modified silk or gelatin. (A) Photograph showing 5% SF/SF-TA and SF/G-TA solutions gelled in the bottom of Eppendorf tubes. Influence of (i) SF-TA or (ii) G-TA weight ratios on crosslinking kinetics and (iii) maximum fluorescence intensities of (B) 5% hydrogels prepared in 40 mM HEPES buffer and (C) 3% hydrogels prepared in 0.5X DMEM buffer ($n=5$, $*p < 0.05$, $**p < 0.01$ and $***p < 0.001$, asterisks above the bars represent significance compared to SF only gels).

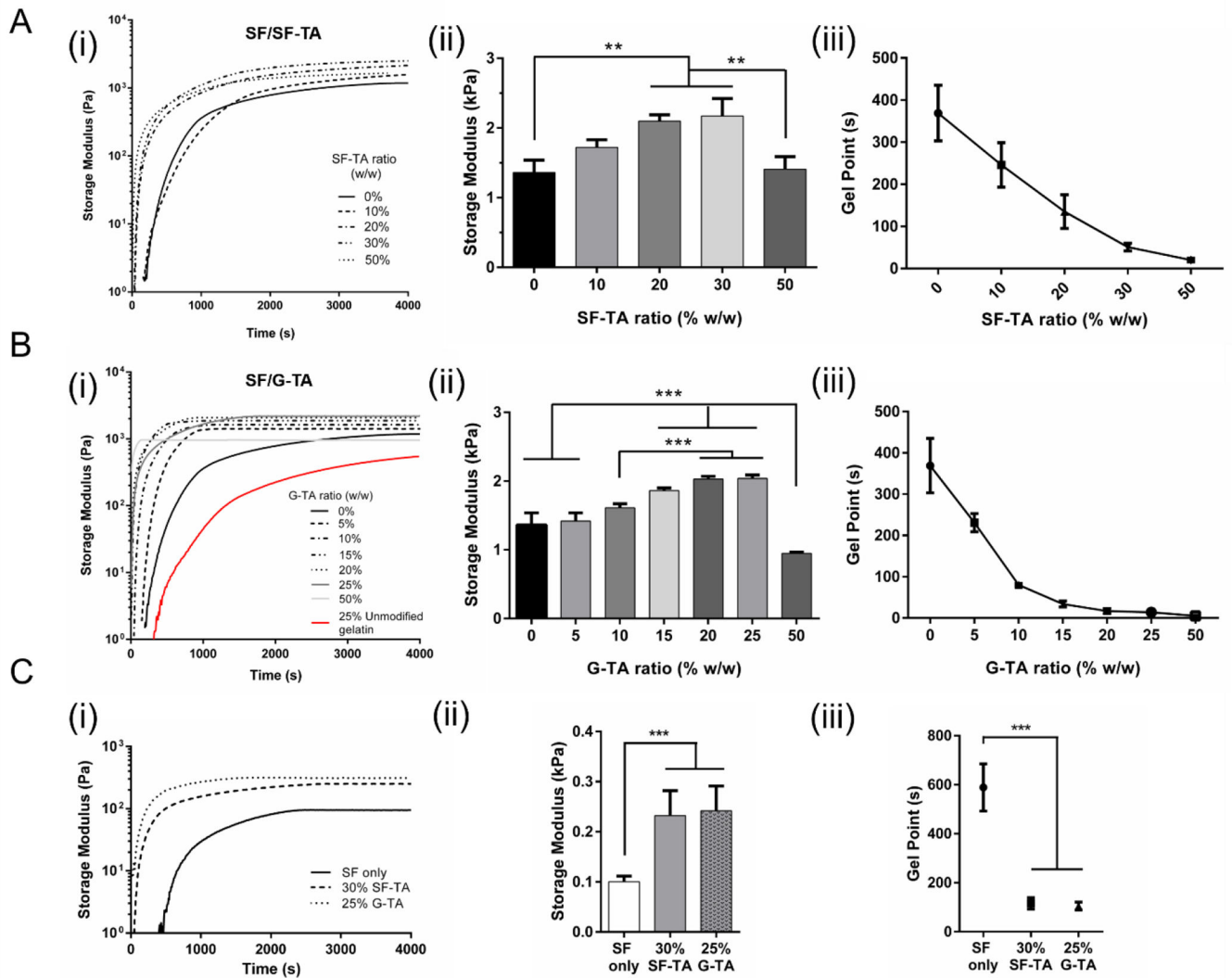


Figure 3. Rheological properties of composite hydrogels. (i) Representative time sweeps, (ii) shear storage moduli at $t = 4000$ s and (iii) gel ($G''/G' < 0.05$) points of 5% (A) SF/SF-TA and (B) SF/G-TA hydrogels prepared in 40 mM HEPES buffer and (C) 3% hydrogels prepared in 0.5X DMEM. ($n = 3$, * $p < 0.05$, ** $p < 0.01$ and *** $p < 0.001$).

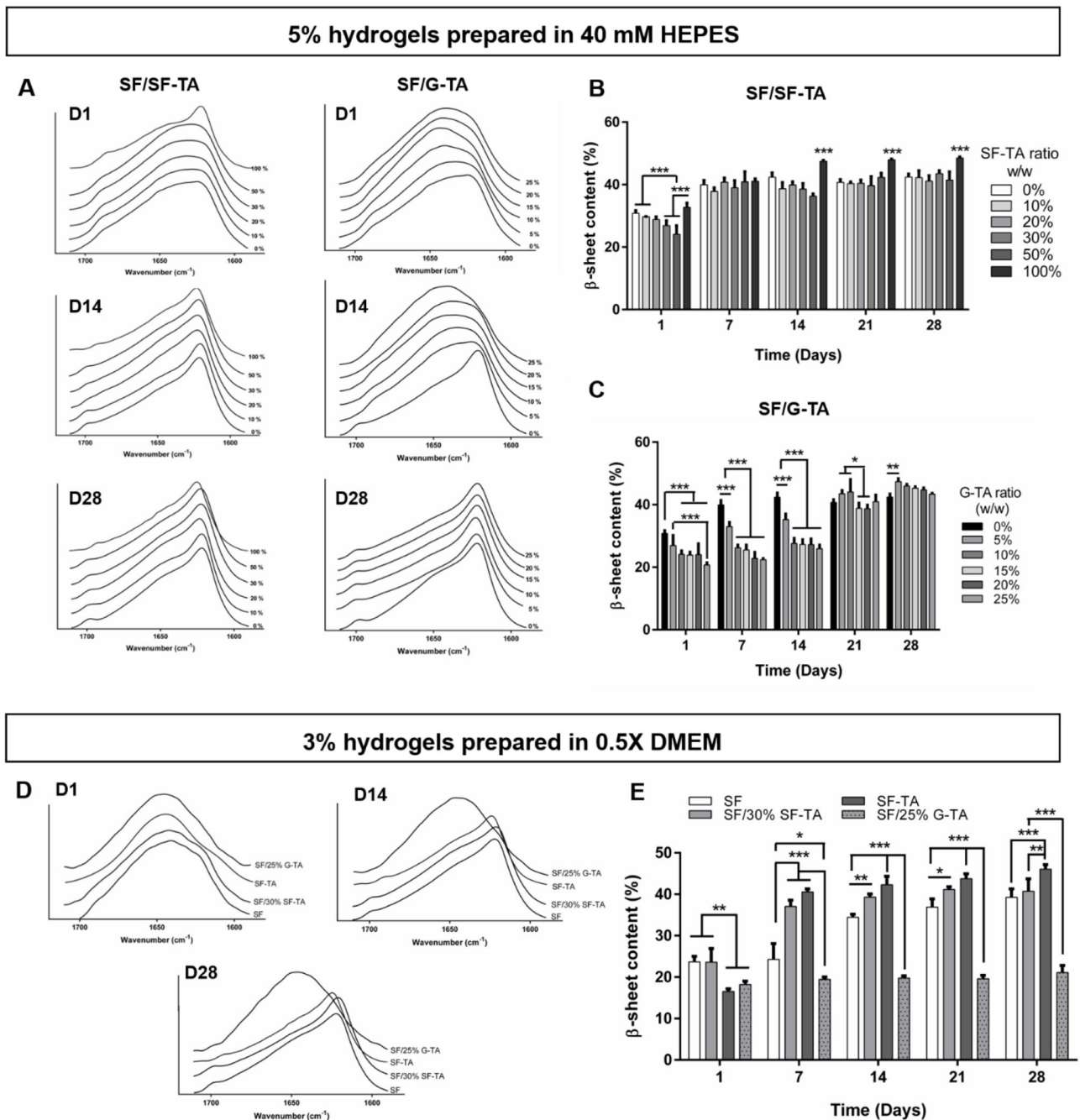


Figure 4. Secondary structure of composite hydrogels determined by FTIR absorbance. ATR-FTIR absorbance spectra of (A) 5% hydrogels prepared in 40 mM HEPES and (D) 3% hydrogels prepared in 0.5X DMEM. β -sheet contents of 5% (B) SF/SF-TA, (C) SF/G-TA and (E) 3% composite hydrogels over 4 weeks of culture in PBS at 37°C. (n = 3, *p < 0.05, **p < 0.01 and ***p < 0.001).

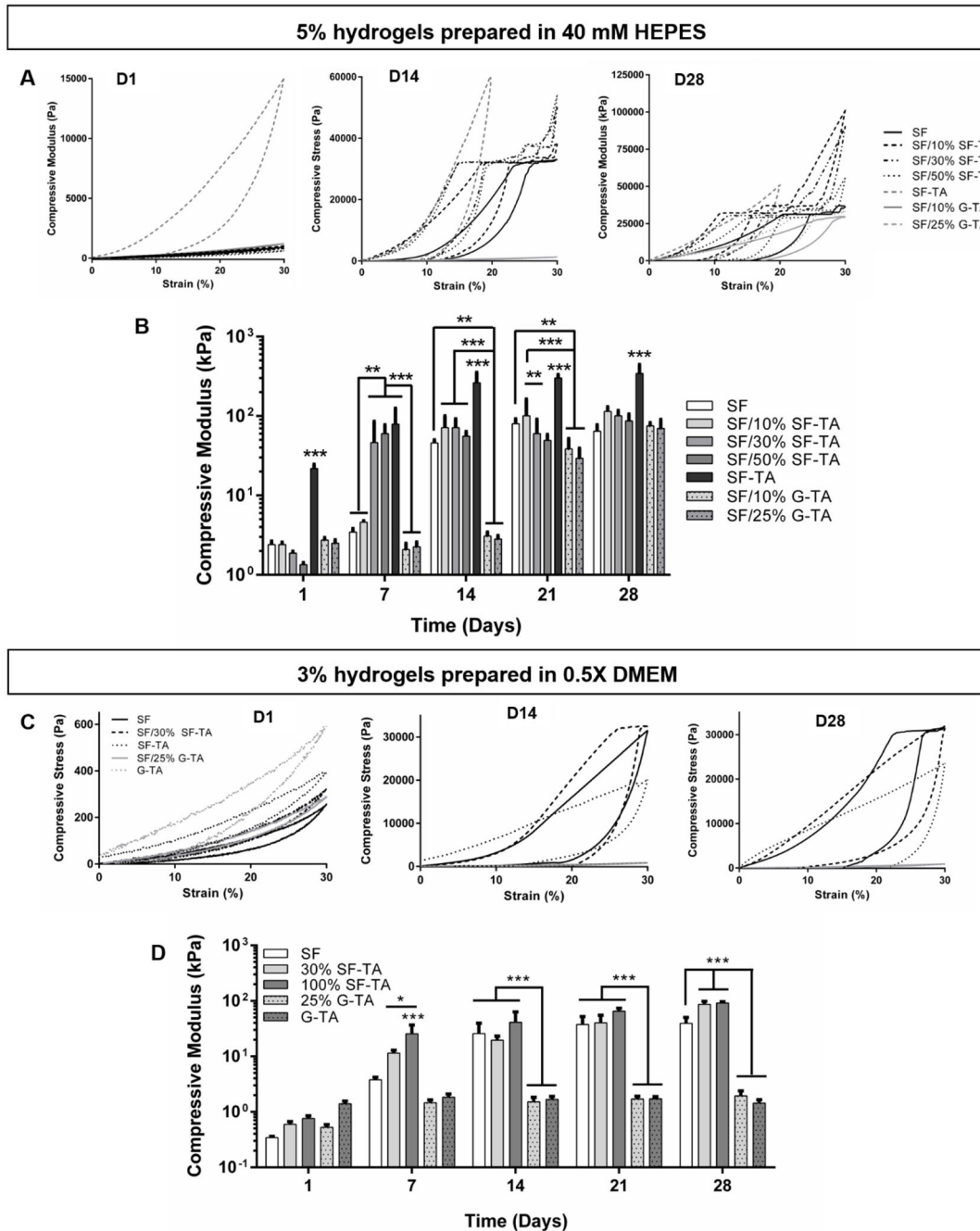
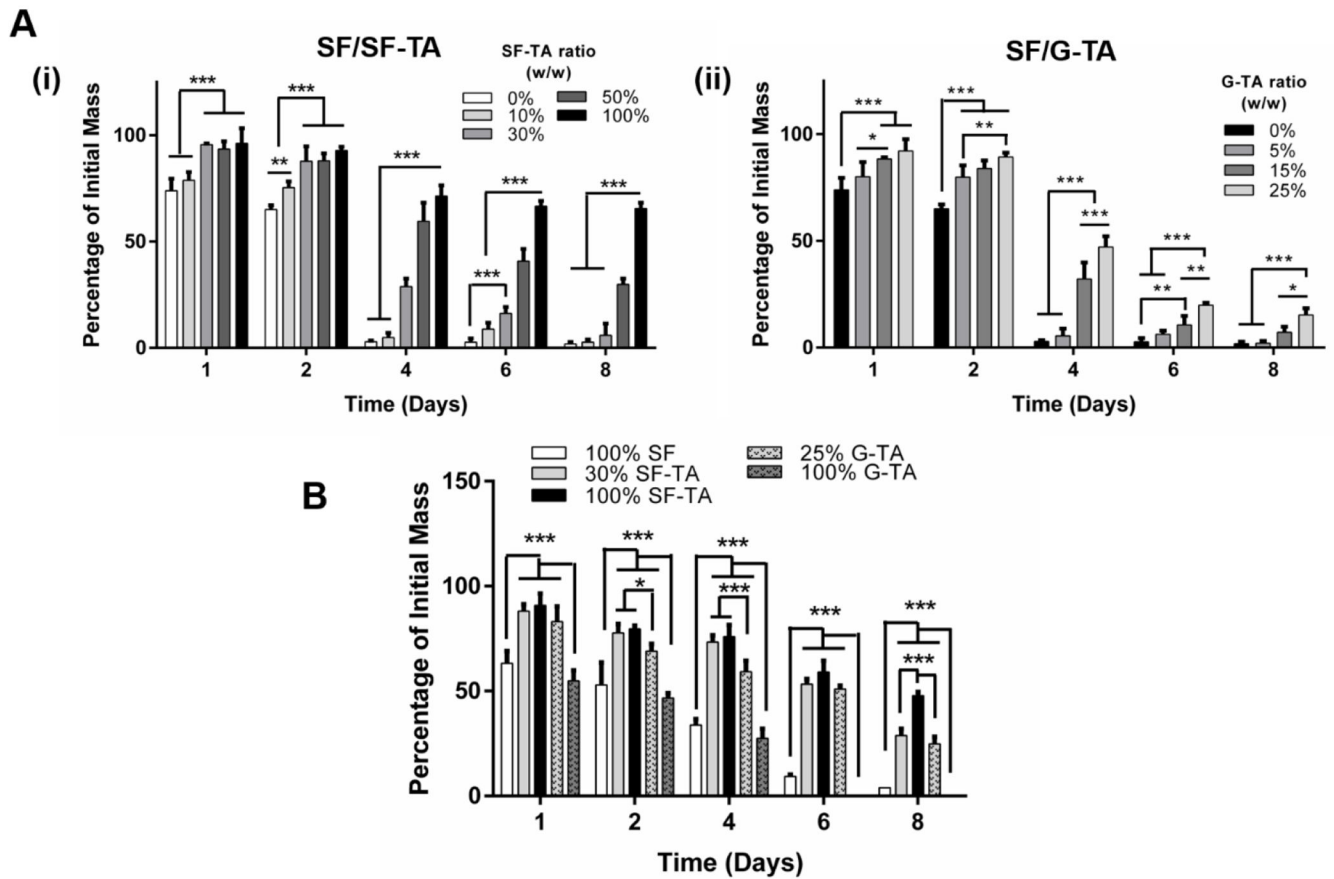


Figure 5. Unconfined compression tests of SF/SF-TA and SF/G-TA composite hydrogels. Representative stress-strain curves of (A) 5% hydrogels prepared in 40 mM HEPES and (C) 3% hydrogels prepared in 0.5X DMEM. Influence of SF-TA and G-TA on compressive moduli of (B) 5% and (D) 3% silk hydrogels over 4 weeks of incubation in PBS at 37 °C. (n = 5, compressive moduli are on log scale, *p < 0.05, **p < 0.01 and ***p < 0.001).

**Figure 6.**

In vitro degradation of hydrogels over 8 days of incubation in PBS with 0.001 U/mL protease XIV. (A) Degradation of 5% silk hydrogels prepared in 40 mM HEPES buffer and supplemented with varying weight ratios of (i) SF-TA or (ii) G-TA. (B) Degradation of 3% silk only, with 30% SF-TA or 25% G-TA hydrogels prepared in 0.5X DMEM ($n = 4$, $*p < 0.05$, $**p < 0.01$ and $***p < 0.001$).

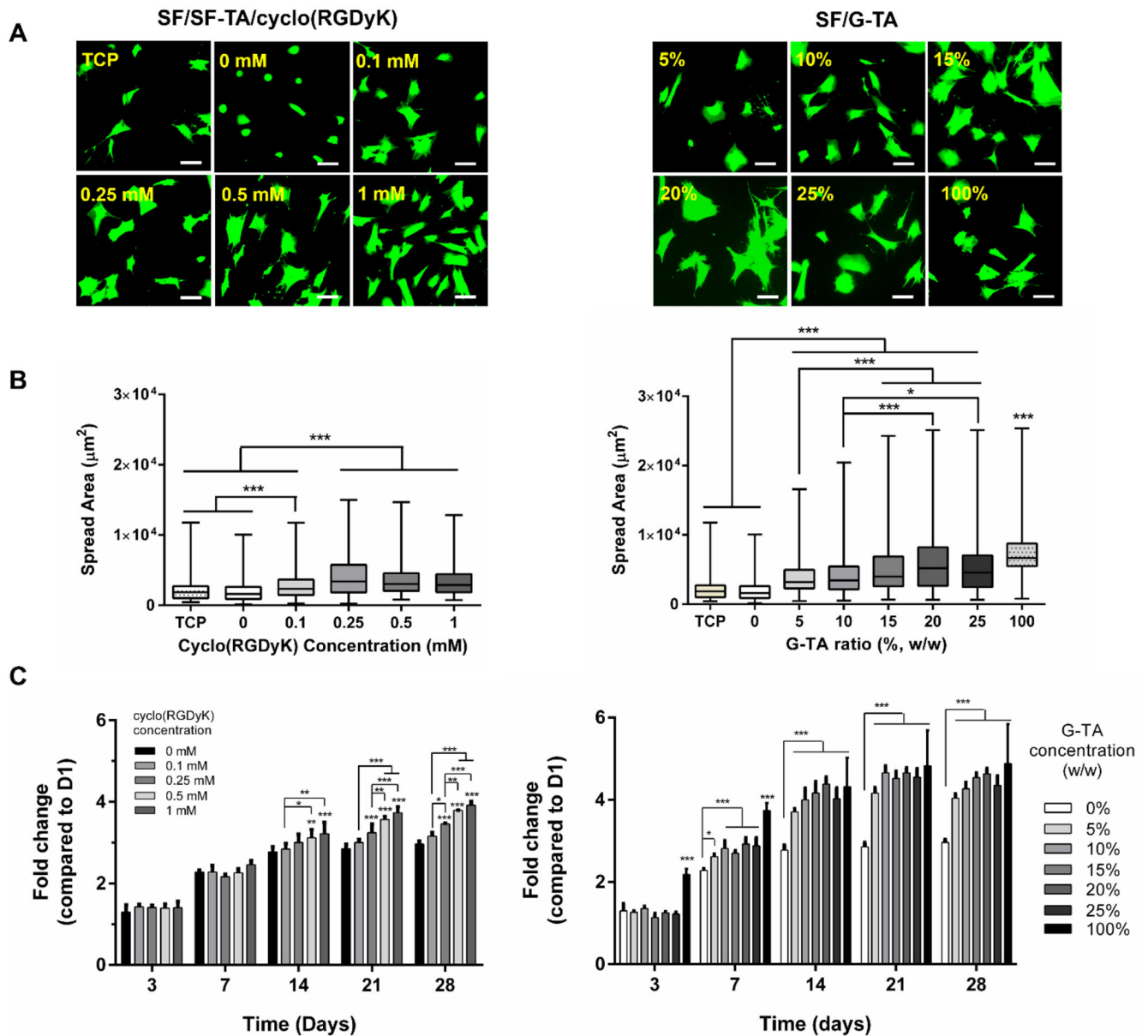


Figure 7. hMSCs behavior on 5% silk hydrogels supplemented with cyclo(RGDyK) or G-TA. (A) Fluorescence images of hMSCs 24 h after seeding on hydrogel surfaces (green: calcein, scale bars: 100 μm). (B) Box-whisker distribution graphs showing the effects of cyclo(RGDyK) or G-TA content on spread area of hMSCs on silk hydrogels at day 1 (spread areas of individual cells were quantified on fluorescent micrographs using ImageJ software, $n = 200$). (C) Fold changes in metabolic activity of hMSCs on 5% silk hydrogels supplemented with cyclo(RGDyK) or G-TA over 4 weeks of culture compared to day 1. ($n = 4$, * $p < 0.05$, ** $p < 0.01$ and *** $p < 0.001$, asterisks above the bars represent significance compared to no cycloRGD control).

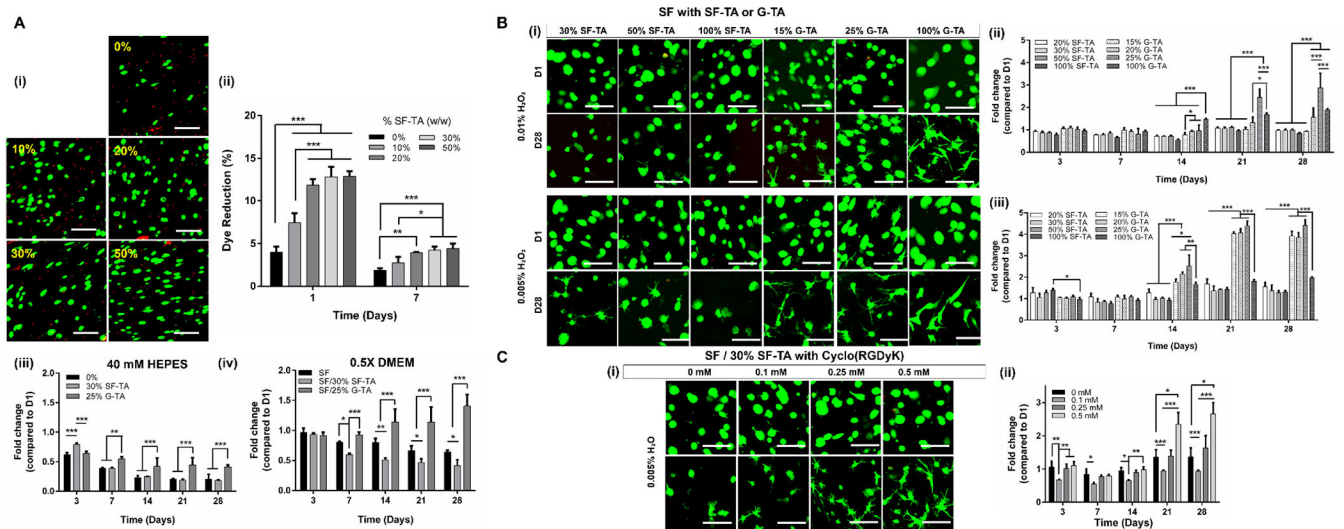


Figure 8.

Viability of hMSCs encapsulated in silk hydrogels. (A) (i) Day 1 fluorescent micrographs of hMSCs encapsulated in 5% silk hydrogels (prepared in 40 mM HEPES) with different SF-TA weight ratios (green: live (calcein), red: dead (EthD-1), scale bars: 100 μ m). (ii) % dye reduction by the cells encapsulated in 5% SF/SF-TA composite hydrogels. Fold changes in metabolic activity of hMSCs encapsulated in 5% SF, SF/SF-TA or SF/G-TA hydrogels prepared in (iii) HEPES or (iv) 0.5X DMEM buffers. (B) (i) Day 1 and 28 fluorescent micrographs of hMSCs encapsulated in 3% SF/SF-TA or SF/G-TA composite hydrogels prepared in 0.5X DMEM buffer using 0.01% or 0.005% H_2O_2 (scale bars: 100 μ m). Fold changes in metabolic activities of hMSCs compared to day 1 in composite hydrogels prepared using (ii) 0.01% or (iii) 0.005% H_2O_2 . (C) (i) Day 1 and 28 fluorescent micrographs and (ii) fold changes in metabolic activities of hMSCs encapsulated in 3% SF/30%SF-TA hydrogels supplemented with various concentrations of cyclo(RGDyK) and crosslinked using 0.005% H_2O_2 . (n = 4 for metabolic activity assays, *p < 0.05, **p < 0.01 and ***p < 0.001).

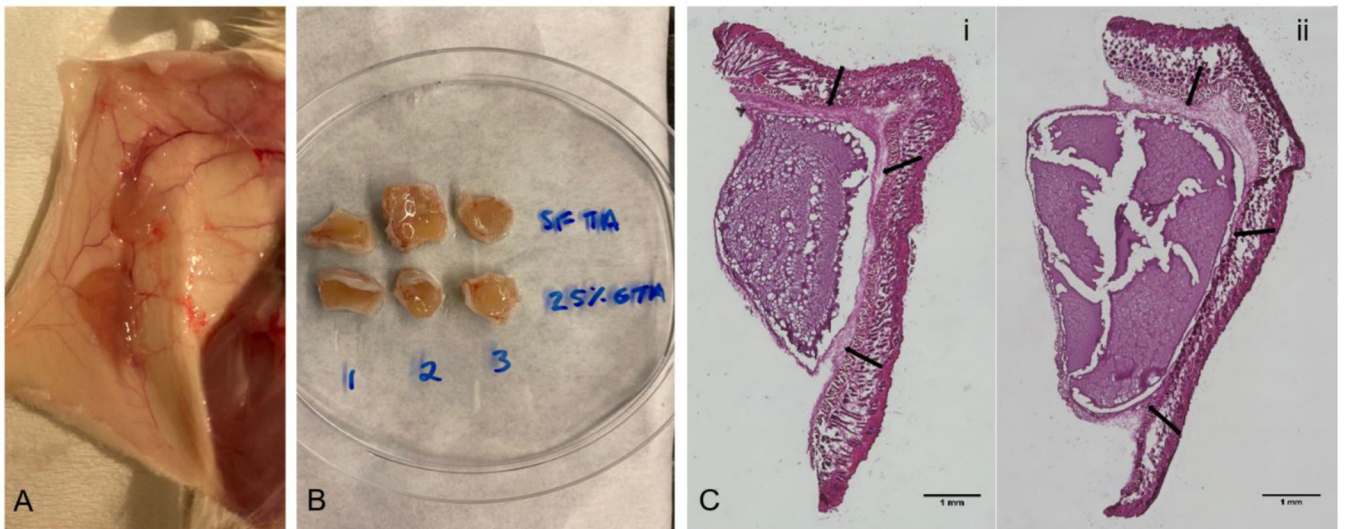


Figure 9.

In vivo animal tests. Day 3 photographs of subcutaneously implanted 3% SF-TA only and SF/25% G-TA composite hydrogels crosslinked in 0.5X DMEM using 0.01 wt% H₂O₂ and 10 U/mL HRP (A) before and (B) after removal from mice. (C) Hematoxylin and eosin stained slice of cutaneous tissues containing (i) SF/25% G-TA and (ii) SF-TA only hydrogel discs. Scale bars: 1 mm.

Table 1.

Concentrations of each component in 5% and 3% SF/SF-TA and SF/G-TA hydrogels. Samples are denoted by the weight ratio of SF-TA or G-TA to total polymer concentration.

SF/SF-TA			SF/G-TA		
Sample	SF concentration (mg/mL)	SF-TA concentration (mg/mL)	Sample	SF concentration (mg/mL)	G-TA concentration (mg/mL)
5% w/v					
SF only	50	0	5%	47.5	2.5
10%	45	5	10%	45	5
20%	40	10	15%	42.5	7.5
30%	35	15	20%	40	10
50%	25	25	25%	37.5	12.5
SF-TA only	0	50	50%	25	25
3% w/v					
SF only	30	0	5%	28.5	1.5
10%	27	3	15%	25.5	4.5
30%	21	9	25%	22.5	7.5
50%	15	15	G-TA only	0	30
SF-TA only	0	30			