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## Metabolism, Biochemical Actions, and Chemical Synthesis of Anticancer Nucleosides, Nucleotides, and Base Analogs

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### Abstract

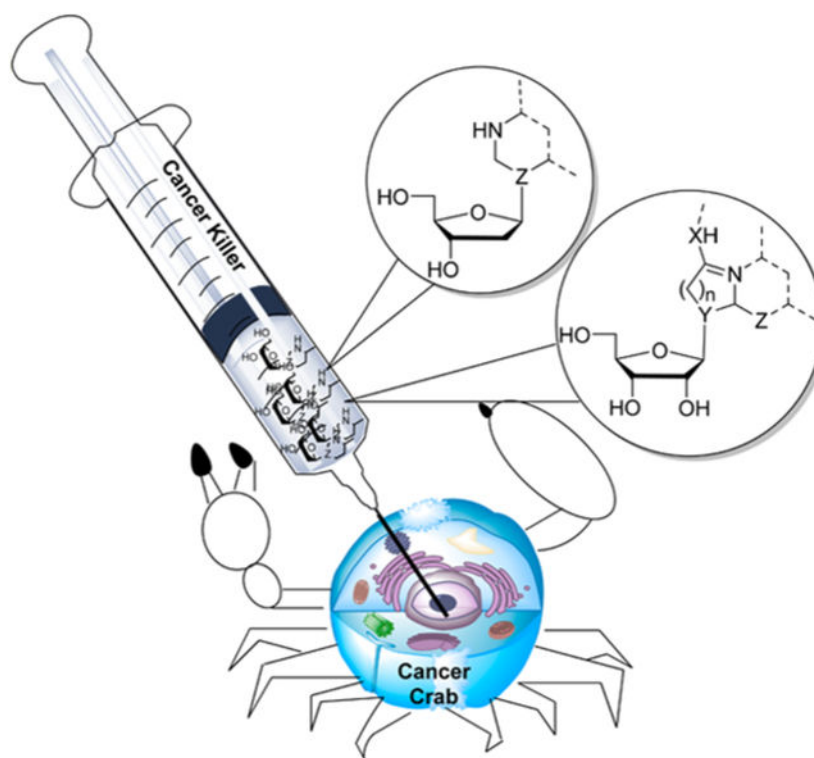
Nucleoside, nucleotide, and base analogs have been in the clinic for decades to treat both viral pathogens and neoplasms. More than 20% of patients on anticancer chemotherapy have been treated with one or more of these analogs. This review focuses on the chemical synthesis and biology of anticancer nucleoside, nucleotide, and base analogs that are FDA-approved and in clinical development since 2000. We highlight the cellular biology and clinical biology of analogs, drug resistance mechanisms, and compound specificity towards different cancer types. Furthermore, we explore analog syntheses as well as improved and scale-up syntheses. We conclude with a discussion on what might lie ahead for medicinal chemists, biologists, and physicians as they try to improve analog efficacy through prodrug strategies and drug combinations.

### Graphical Abstract

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## 1. INTRODUCTION AND MOTIVATION

Cancer is the second leading cause of death in the United States, accounting for 1 in 4 deaths annually. Importantly, cancer incidence continues to increase worldwide.<sup>1</sup> Cancer encompasses a broad range of diseases in which host cells escape their normal cell cycle regulation. This phenomenon can be linked to several factors, including gender, ethnicity, age of onset, and lifestyle,<sup>2</sup> but cancer can also be caused by cellular transformation linked to viral infection, chemical exposure, or radiation exposure, or the cause can be unknown (spontaneous) in nature.<sup>3–5</sup> Advances in oncology treatments have increased the number of cancer survivors from 3 million in 1971 to more than 13 million in 2012.<sup>6</sup> This has led to a 39% improvement in 5-year survival rates for different cancers and stages of disease as indicated by the 2010 data, and rates should continue to improve with recent advances in new therapies.<sup>1,7</sup>

The primary standard-of-care treatment for cancer includes surgery, radiation, chemotherapy, hormone therapy, immuno-therapy, and targeted therapy.<sup>1</sup> Cutting-edge therapies may include one or more of these procedures listed above depending upon the type and stage of the cancer being treated. Sometimes surgery and radiation may be a second tier treatment, with chemotherapy being used first to reduce the tumor burden. Chemotherapies are divided into several main drug classes: alkylating agents,<sup>8</sup> antimetabolites, anti-tumor antibiotics,<sup>9</sup> topoisomerase inhibitors,<sup>10</sup> mitotic inhibitors,<sup>11,12</sup> and corticosteroids.<sup>13</sup> Recently, antibodies directed against programmed cell death 1 (PD-1) protein, PD-1 ligands 1 and 2, and cytotoxic T-lymphocyte-associated protein 4 (CLTA-4) have been found to

activate the immune system to attack tumors, and they are therefore used with varying success to treat some types of cancer.<sup>14–16</sup>

Chemotherapeutic nucleoside, nucleotide, and base analogs, herein referred to as “nucleoside analogs”, are antimetabolites. They are chemically modified analogs of natural nucleosides, nucleotides, and bases, which are endogenous metabolites involved in many essential cellular processes, such as DNA and RNA synthesis, and purinergic signaling. Nucleoside analogs have been a cornerstone of anticancer and antiviral chemotherapy for decades. Currently, there are 15 FDA-approved nucleoside analogs used to treat various cancers, and they account for a large percentage of the current cancer chemotherapeutic arsenal.<sup>17</sup> Whereas chemotherapeutic nucleoside analogs alone seldom lead to a cure for cancer, they do provide a valuable treatment option for cancer patients. In addition, many other nucleoside analogs are currently being investigated in clinical trials as monotherapy or combination therapy for multiple cancers. These studies attempt to increase the potency or bioavailability of these agents while diminishing associated dose-limiting toxicity.

In this review, we focus on both the biology and chemical syntheses of the current FDA-approved anticancer nucleoside analogs (Figure 1), investigational anticancer nucleoside analogs in development since 2000 or analogs used outside of the United States (Figure 2), and nucleoside analogs in development since 2000 that have stalled in clinical trials (Figure 3). Additionally, we present an overview of the cellular metabolism, the biochemical mode of action, and the types of cancers treated with these agents, as well as focusing on current methodologies for their chemical synthesis.

## 2. GENERAL BIOLOGICAL ACTIVITY OF ANTICANCER NUCLEOSIDE ANALOGS

The two main routes for administering nucleoside analogs are by intravenously infusion and oral formulation. Once inside the body, the analogs can face a major bottleneck in cellular uptake. Both protein facilitated diffusion using either concentrative nucleoside transporters (CNT or SLC28; three members) and/or equilibrative nucleoside transporters (ENT or SLC29; four members) are involved in the cellular uptake for the majority of agents.<sup>18,19</sup> Passive diffusion is limited in regard to cellular uptake for parent nucleoside analogs, with troxacitabine **48** being the exception, requiring very high extracellular concentrations in order to occur.<sup>20</sup> Certain nucleoside analog prodrugs, such as NUC-1031 **32**, NUC-3373 **37**, or elacytarabine **51**, however, can enter the cell independent of transporters.

Nucleoside analogs are prodrugs and require cellular metabolism to be converted to their active metabolites. This is accomplished by the cellular salvage pathway—a group of enzymes that can phosphorylate nucleoside and nucleotide substrates. After transport across the plasma membrane, nucleoside analogs undergo an initial phosphorylation—or if a base analog, ribosylation followed by phosphorylation—to generate nucleoside-5'-monophosphate forms,<sup>21</sup> which is often the rate-limiting step within the cell in the process to generate active metabolites (Figure 4).<sup>22</sup> The cell primarily uses four kinases for nucleoside-5'-monophosphate phosphorylation: 2'-deoxycytidine kinase, thymidine kinase 1, thymidine kinase 2, and deoxyguanine kinase.<sup>23</sup> Each kinase has multiple substrates

and can have feedback inhibition to regulate their activities.<sup>24,25</sup> The initial phosphorylation is often performed by 2'-deoxycytidine kinase. Cancer cells usually express 2'-deoxycytidine kinase at a 3- to 5-fold higher protein level than most normal cells, affording some degree of selectivity.<sup>26</sup> Phosphoramidate prodrugs such as NUC-1031 **32** or NUC-3373 **37**, however, enter the cell as masked monophosphates and pass through a different enzymatic pathway in order to be converted to their nucleoside-5'-monophosphate forms.

Nucleoside-5'-monophosphate analogs are converted to nucleoside-5'-diphosphate and nucleoside-5'-triphosphate forms by various cellular kinases. Nucleoside-5'-triphosphate analogs are substrates for DNA polymerases and can be incorporated into DNA during replication or DNA excision repair synthesis, which gives rise to stalled replication forks and chain termination. These events activate various DNA damage sensors, which stimulate DNA repair, halt cell progression, and often lead to apoptosis.<sup>27,28</sup> Since most cancer cells replicate their genome more frequently than a majority of normal adult cells, which are quiescent and not actively synthesizing their DNA, this phenomenon allows for a degree of cancer cell selectivity.<sup>17</sup> Moreover, certain nucleoside-5'-triphosphate analogs can be incorporated into RNA, leading to transcriptional termination, and messenger RNA (mRNA) and ribosomal RNA (rRNA) instability.<sup>29,30</sup>

Nucleoside and nucleotide (mono-, di-, or triphosphates) derivatives can also inhibit key cellular enzymes, providing a secondary mode of action that inhibits cell growth. Such enzymes include ribonucleotide reductase,<sup>31</sup> which removes the 2'-OH group from the ribose sugar in order to generate *de novo* 2'-deoxyribonucleoside diphosphate,<sup>32</sup> purine nucleoside phosphorylase,<sup>33,34</sup> which is involved in purine metabolism catalyzing the reversible phosphorolysis of purine nucleosides,<sup>35</sup> and thymidylate synthase,<sup>36</sup> discussed below (see Section 5.1.1.1). A few newer nucleoside analogs, however, exert their anticancer activities as adenosine receptor antagonists, or by inhibiting cellular enzymes that are not involved in nucleic acid synthesis. These cellular enzymes include NEDD8-activating enzyme, which is involved in the ubiquitin-proteasome degradation system and has an important role in cellular processes associated with cancer cell growth,<sup>37</sup> and DOT1L (histone-lysine N-methyltransferase, H3 lysine-79 specific), which is involved in post-translational gene modification.<sup>38</sup> Furthermore, the enzymatic activity of DOT1L represents an oncogenic driver of MLL-r leukemia.<sup>39</sup>

Enzymes that dephosphorylate nucleotides are also present in the cell. 5'-Nucleotidases (5'-NTs) can dephosphorylate noncyclic nucleoside-5'-monophosphates to nucleosides and inorganic phosphates, leading to the regulation of cellular nucleotide and nucleoside levels.<sup>40,41</sup> There are seven intracellular 5'-NTs: cN-IA, cN-IB, cN-II, cN-III, cN-II-like, cdN, and mdN in the cell,<sup>42,43</sup> and they have been reported to modulated antiviral and anticancer nucleotide levels.<sup>41</sup> Their cellular functions involve cell-cell communication and signal transduction,<sup>44</sup> as well as nucleic acid repair.<sup>45</sup> Other enzymes involved in nucleotide degradation include eN (CD73), an ecto-5'-nucleotidase that converts AMP to adenosine plus phosphate,<sup>43</sup> and ectonucleoside triphosphate diphosphohydrolase-1 (CD39), which can hydrolyze nucleoside-5'-triphosphates into nucleoside-5'-monophosphate and nucleoside-5'-diphosphate products. Tumor expression of these two enzymes, in an animal



model, appears to enhance tumor growth by negatively regulating the immune response.<sup>46,47</sup> Furthermore, alkaline phosphatases are also present and can be targets for anticancer treatment, but they might have a lesser impact on antimetabolite metabolism.<sup>48</sup> Recently, a sterile alpha motif and histidine/aspartic acid domain containing protein 1 (SAMHD1) was discovered to be a dGTP/GTP-dependent deoxynucleotide triphosphohydrolase,<sup>49,50</sup> generating 2'-deoxynucleosides and inorganic triphosphates from the cellular canonical 2'-deoxyribonucleoside-5'-triphosphates. Clofarabine-5'-triphosphate has been shown to be a substrate for SAMHD1.<sup>51</sup> Furthermore, the SAMHD1 gene was shown to be mutated in CLL cancer,<sup>52</sup> and the protein level downregulated in lung cancer.<sup>53</sup>

Cancer chemotherapy is limited in effectiveness by off target toxicity and drug resistance, with the latter leading to a decreased ability to deliver an adequate concentration of drug to cancer cells. Drug resistance can arise from pre-existing intrinsic properties of cancer cells or by acquired traits, which cancer cells develop in response to limited drug exposure.<sup>54</sup> Cancer cells can acquire resistance to nucleoside analogs by several different mechanisms. First, many studies using cell lines exposed to nucleoside analogs have shown transporter mutations, which generally do not occur in cancer patients. Although a decrease in nucleoside transport in patients was detected, it was dependent on cell differentiation state: myeloblasts versus lymphoblasts.<sup>55,56</sup> For example, high expression of hENT1 was a potential co-determinant of poor clinical response to 5-FU **2** in cells *ex vivo* from colorectal cancer patients,<sup>57</sup> or using RNA1 to downregulate hENT1 in the pancreatic cell line.<sup>58</sup> In contrast, high hENT1 expression was reported as a biomarker for survival in patients with pancreatic cancer treated with gemcitabine.<sup>59</sup> Second, modulation of phosphorylating enzymes, such as 2'-deoxycytidine kinase, has been reported when using cancer cell lines.<sup>60</sup> In patients, however, resistance to nucleoside analogs by phosphorylating enzymes has been reported to be linked to decreases in mRNA levels and a decrease in enzyme activity.<sup>61,62</sup> Third, resistance to nucleoside analogs can also occur by augmentation of nucleotide excision repair machinery.<sup>63</sup>

An increase in cellular nucleoside analog efflux provides another mechanism of resistance.<sup>64</sup> Cellular nucleoside analog efflux is performed by several ATP-Binding Cassette (ABC) transporters, such as ABCB1 (MDR1), ABCC4 (MRP4), ABCC5 (MRP5), and ABCC11 (MRP8).<sup>65,66</sup> Furthermore, metabolic inactivation of nucleoside analogs can occur by the actions of cellular enzymes such as adenosine deaminase, cytidine deaminase, and purine nucleoside phosphorylase.<sup>67-69</sup>

### 3. GENERAL SYNTHETIC APPROACHES TO ANTICANCER NUCLEOSIDE ANALOGS

Syntheses highlighted in this review use three general approaches to access anticancer nucleoside analogs: divergent, convergent, and trans-glycosylation (Figure 5).<sup>70</sup> The divergent approach begins with an intact nucleoside that is subsequently modified at either the sugar or base. A major advantage of this method is that the stereochemistry, especially of the key glycosidic bond ( $\beta$ -anomer), is already fixed. A drawback of this approach is the

presence of two or three hydroxyl groups, which are relatively similar in chemical reactivity, and often require multiple protection/deprotection steps.

The convergent approach couples a nitrogenous base with a modified sugar in a glycosylation reaction. This method allows for more synthetic diversity, and it is more widely used, although the stereochemistry of the glycosylation can be an issue. The reaction is most often performed under Silyl–Hilbert–Johnson (Vorbrüggen) conditions or metal salt coupling conditions.

Benzoyl-protected furanose sugars are often used in glycosylations performed under Vorbrüggen conditions, because the stereochemistry of the glycosylation reaction is almost completely selective. Anchimeric assistance of the 2'-*O*-benzoyl group, protected furanose sugar **II**, leads to intermediate **III**, which effectively has its  $\alpha$ -face blocked. This allows for selective formation of the protected  $\beta$ -anomer **IV**. Difficulties arise when dealing with 2'-deoxy furanose sugars. The absence of a 2'-*O*-benzoyl group precludes formation of an intermediate with a sterically hindered  $\alpha$ -face, often leading to an anomeric mixture of products **VII**, and many times requires laborious chromatographic separation.

The metal salt glycosylation generally couples a furanose sugar possessing a leaving group (often halogens) with  $\alpha$ -stereochemistry and a salt of a nitrogenous base. Conditions are sought to make the reaction predominantly (or exclusively if possible)  $S_N2$  in nature, in an attempt to elude anomerization of the leaving group or avoid the formation of an oxonium ion intermediate similar to **VI**, favoring a  $S_N1$ -like reaction.

Enzyme-mediated base exchange is an addition method for nucleoside synthesis. Purine nucleoside phosphorylase or uridine phosphorylase can be used to replace one nucleoside base with another in a trans-glycosylation reaction. Overall, enzymes can selectively generate the desired  $\beta$ -anomer nucleoside.

## 4. PURINE AND ACYCLIC ANALOGS

### 4.1. Thiopurines

**4.1.1. Thiopurines Biology.**—In the early 1950s, Elion's group, while at the Wellcome Research Laboratories, discovered that hypoxanthine and guanine substituted with sulfur at the 6 position led to purine synthesis inhibition, which in turn decreased growth of *Lactobacillus casei*.<sup>71</sup> Subsequently, 6-mercaptopurine (6-MP, **1**) and 6-thioguanine (6-TG, **3**) were demonstrated to have potent anticancer activities against a wide variety of rodent tumors. The more promising 6-MP **1** was rapidly advanced into clinical trials; in 1953 it received FDA approval for treating pediatric acute lymphocytic leukemia (ALL), a rapid-developing cancer of the blood and bone marrow in which immature lymphocytes are overproduced.

Currently, 6-MP **1** and 6-TG **3** have various clinical applications. They can effectively treat hematologic cancers (childhood and adult leukemias).<sup>72</sup> 6-TG **3** is used to treat acute myelogenous leukemia, a cancer involving abnormal white blood cells of myeloid lineage that rapidly divide and interfere with normal white blood cell production, whereas 6-MP **1** is

used in combination with other chemotherapeutic agents and remains part of the standard of care for acute lymphocytic leukemia.<sup>73</sup> These compounds are also being used to treat autoimmune disorders, such as rheumatoid arthritis, psoriasis, and ulcerative colitis,<sup>74–76</sup> and they have also been employed to prevent solid organ transplant rejection.<sup>30,77,78</sup>

Thiobases require further metabolism by cellular enzymes to reach their active forms (Figure 6).<sup>79</sup> 6-MP **1** is converted by hypoxanthine/guanine phosphoribosyl transferase to 6-thioinosine-5'-monophosphate **III** and then converted to 6-thio-guanosine-5'-monophosphate **VI**, which is also the product of 6-TG **3** and hypoxanthine/guanine phosphoribosyl transferase. Further phosphorylation leads to 6-thio-guanosine-5'-triphosphate **VIII**, which can be incorporated into RNA.<sup>29</sup> However, Nelson et al. have reported that 6-MP **1** activity was not directed toward specifically inhibiting RNA synthesis, suggesting that 6-MP-5'-triphosphate **VIII** incorporation into RNA was not a major mechanism for its anticancer activity *in vitro*.<sup>80</sup>

A major active metabolite of both 6-MP **1** and 6-TG **3** is 6-thio-2'-deoxyguanosine-5'-triphosphate **XI**, which is produced from ribonucleotide reductase-mediated deoxygenation of 6-TG-5'-diphosphate **VII**, followed by phosphorylation.<sup>17</sup> 6-Thio-2'-deoxyguanosine-5'-triphosphate **XI** can be incorporated into DNA. DNA polymerases continue polymerization after 6-thio-2'-deoxyguanosine-5'-monophosphate incorporation, resulting in multiple thiobase derivatives embedded in the newly synthesized DNA strand. The reactive thiol group of the incorporated active metabolite is susceptible to nonenzymatic methylation, leading to incorporated 6-methylthioguanine-5'-monophosphate **IX**, which preferentially base-pairs with a thymine during DNA replication.<sup>81</sup> The 6-methylthioguanine:thymine base-pairings resemble DNA replication errors, which are recognized by cellular mismatch repair enzymes. It is hypothesized that 6-methylthioguanine:thymine base-pairing leads to a futile cycle of repair, which ultimately generates cytotoxic DNA damage.<sup>81</sup>

Another major active metabolite of 6-MP **1** is methylthioinosine-5'-monophosphate **IV**, which is formed by the methylation of thioinosine-5'-monophosphate **III** by thiopurine S-methyltransferase.<sup>82</sup> Metabolite **IV** inhibits phosphoribosylpyrophosphate (PRPP) amidotransferase, the first enzyme in the *de novo* purine biosynthesis pathway.<sup>83</sup> PRPP amidotransferase inhibition leads to a decrease in purine nucleotide pools, which can disrupt DNA synthesis and repair, ultimately leading to cell death *in vitro*.<sup>84</sup> 6-Thioguanine-5'-monophosphate **VI** can also be methylated by S-methyltransferase to form methylthioguanosine-5'-monophosphate **IX**. However, this metabolite is a weak inhibitor of PRPP amidotransferase.

Two enzymes involved in the detoxification of 6-MP **1** and 6-TG **3** are thiopurine methyltransferase and xanthine oxidase. The former directly methylates 6-MP **1** and 6-TG **3**, to metabolites **II** and **V**, respectively. Xanthine oxidase catabolizes 6-MP **1** to the 6-thiouric acid derivative **I**. However, because xanthine oxidase expression is low in hematopoietic cells, thiopurine S-methylation is considered to be the predominant pathway of detoxification of 6-MP **1** and 6-TG **3**.<sup>79</sup> In addition, six different variant alleles have been identified from clinical samples that lead to lower thiopurine methyltransferase activity.<sup>79</sup>

Thiopurine resistance can arise due to the action of ATP-Binding Cassette (ABC) transporters. 6-MP-5'-monophosphate **III** and 6-TG-5'-monophosphate **IV** are substrates for ABC transporters: MRP4 and MRP5, for active transport out of cancer cells.<sup>85,86</sup> Additional mechanisms of resistance identified using cell lines are the up regulation of P-glycoprotein<sup>87</sup> and decreased hypoxanthine-guanine phosphoribosyltransferase activity.<sup>88</sup> A genomic approach revealed that overexpression of the ATM/p53/p21 pathway, overexpression of TNFRSF10D, and overexpression of CCNG2 may also contribute to thiopurine resistance in cell lines.<sup>89</sup> Besides the natural mutations in thiopurine S-methyltransferase gene,<sup>79</sup> mutations in NT5C2 transporter gene led to an increase in nucleotidase activity and resistance to 6-mercaptopurine **1** and 6-thioguanine **3** in ALL patients.<sup>90,91</sup> Over 170 clinical trials evaluating thiopurines as a cancer treatment are listed at [ClinicalTrials.gov](https://clinicaltrials.gov).

**4.1.2. Thiopurines Synthesis.**—Elion and Hitchings reported the preparation of 6-MP **1** in 1952 (Scheme 1).<sup>92</sup> For their studies, they developed a multigram-scale synthesis of hypoxanthine **57** starting from 4-amino-6-hydroxy-5-nitrosopyrimidine-2-thiol **54**.<sup>93</sup> The heterocycle **54** was reduced, dethiolated, and cyclized, forming **57**, which was subsequently treated with P<sub>2</sub>S<sub>5</sub>, furnishing the desired 6-MP **1**. Conversion of hypoxanthine **57** to 6-MP **1** using P<sub>2</sub>S<sub>5</sub> has also been used in a recent process synthesis, which was done in toluene on a 200-g-scale with a yield of 93%.<sup>94</sup> Additionally, this transformation has been effected on a gram-scale using the crystalline and storable P<sub>4</sub>S<sub>10</sub>-pyridine complex reagent **58**.<sup>95</sup>

Elion and Hitchings synthesized 6-thioguanine (6-TG, **3**) in 1954 (Scheme 2).<sup>96</sup> Guanine **59** was converted to 6-TG **3** by treatment with P<sub>2</sub>S<sub>5</sub> in refluxing pyridine. Diacylated guanine **60** has also been used to produce **3**. Sulfuration with P<sub>2</sub>S<sub>5</sub>, followed by deprotection, furnished the product on 100-g-scale (Scheme 2).<sup>97</sup> In addition, the sulfuration of **60** has been effected by treatment with Lawesson's reagent.<sup>98</sup>

## 4.2. GS-9219

**4.2.1. GS-9219 Biology.**—Several acyclic nucleotide derivatives, such as 9-(2-phosphonylmethoxyethyl)guanine (PMEG), 9-(2-phosphonylmethoxyethyl)-2,6-diaminopurine (PMEDAP), and 9-(2-phosphonylmethoxyethyl)adenine (PMEA), display antiproliferative activities in cancer cells (Figure 7).<sup>99</sup> The derivatives bear a phosphonate group, which is a bioisostere of a phosphate, but metabolically and chemically more stable.<sup>100</sup> The acyclic nucleotide-5'-diphosphates are structural analogs of naturally occurring 2'-deoxyribonucleoside-5'-triphosphates and can be used as substrates for cellular DNA polymerases, acting as absolute chain terminators.<sup>101</sup>

PMEG has been shown to be the most cytotoxic among the acyclic derivatives studied.<sup>99</sup> Within the cell, PMEG is converted to its diphosphate form, which is the active metabolite. The PMEG-5'-diphosphate potently inhibits DNA polymerases  $\alpha$ ,  $\delta$ , and  $\epsilon$  (DNA chromosomal replication enzymes), leading to inhibition of DNA synthesis and repair.<sup>102</sup> The 3'  $\rightarrow$  5' exonuclease activity of DNA polymerase can remove PMEG incorporated during DNA synthesis.<sup>99</sup> However, if not excised from the growing DNA strand, the

PMEG-5'-monophosphate, which lacks a 3'-OH moiety, acts as an absolute DNA synthesis chain terminator.

Unfortunately, PMEG displays poor cellular permeability properties, in addition to kidney and gastrointestinal tract associated toxicities.<sup>102</sup> Therefore, researchers at Gilead developed GS-9219 (**53**, VDC-1101), a double prodrug of PMEG, in an attempt to preferentially target lymphoid tissues.<sup>103</sup> The *N*<sup>6</sup>-cyclopropyl moiety was installed to increase specific intracellular activation as well as decrease plasma exposure to the toxic PMEG. The phosphorodiamidate prodrug was installed to increase lymphoid cell and tissue loading efficacy. Such prodrugs are also often employed to increase compound solubility, cellular penetration, and selective tissue accumulation.<sup>104</sup> GS-9219 **53** showed potent antiproliferative activity against activated lymphocytes and hematopoietic cancer cells *in vitro* and in canines.<sup>103</sup> It was proposed that GS-9219 **53** is intracellularly converted to PMEG-5'-diphosphate in three steps: enzymatic hydrolysis of the amino acid of the prodrug (involving cathepsin A), deamination of the cyclopropyl amino moiety (involving *N*<sup>6</sup>-methyl-AMP-aminohydrolase), and phosphorylation of PMEG (Figure 7).<sup>105</sup>

Mutations in human adenosine deaminase-like proteins (H286R and S180N) have recently been found to promote resistance to GS-9219 **53** and to 9-(2-phosphonylmethoxyethyl)-*N*<sup>6</sup>-cyclopropyl-2,6-diaminopurine, an intermediate in the conversion pathway of GS-9219 **53** to PMEG.<sup>106</sup> Furthermore, point mutations in human guanylate kinase (S35N and V168F) were shown to promote GS-9219 resistance in cell lines.<sup>107</sup>

Unfortunately, GS-9219 **53** has stalled in human clinical trials for treatment of non-Hodgkin's lymphoma, chronic lymphocytic lymphoma, and multiple myeloma due to an unacceptable safety profile ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00499239) identifier: NCT00499239—terminated—no results posted). However, the agent was FDA-approved for canine lymphoma in June 2013,<sup>108,109</sup> and an additional Phase II study was done evaluating the effectiveness in canine cutaneous T-cell lymphoma.<sup>110</sup> GS-343074 and GS-424044 are two additional prodrugs synthesized after GS-9219 **53** and are being evaluated as anticancer prodrugs, particularly for canine cancers. No structures have been revealed for these two compounds.

**4.2.2. GS-9219 Synthesis.**—In 2005, Gilead disclosed their discovery synthesis of GS-9219 **53** (Scheme 3).<sup>111</sup> 2-Amino-6-chloropurine **61** was selectively *N*<sup>9</sup>-alkylated, and subsequent nucleophilic aromatic substitution with cyclopropylamine furnished alkyl diester **63**. TMSBr-mediated deprotection led to the phosphonic acid derivative **64**, which was sufficiently pure to require no chromatography after the reaction. Bis-amidate formation was effected by treatment of acid **64** with 2,2'-dithiodipyridine, triphenylphosphine, and *D*-alanine ethyl ester HCl.

Recently, Jansa *et al.* reported an improved route to bisamidate prodrugs.<sup>112</sup> These researchers surmised that the intermediate **65** would react directly with 2,2'-dithiodipyridine, triphenylphosphine, and the amino acid to produce the bisamidate (Scheme 4). This would obviate the need for the laborious purification step often needed to isolate and purify the intermediate diacid. The approach was successful and was used to synthesize GS-9219 **53**.<sup>113</sup> TMS bromide-mediated deprotection of the phosphonate diester **63**<sup>113</sup>

formed the bis(TMS)ester intermediate **65**, maintained by strictly avoiding any contact with air. The bis(TMS)ester intermediate was then directly converted to GS-9219 **53**.

## 5. PYRIMIDINE ANALOGS

### 5.1. Fluorinated Pyrimidines

#### 5.1.1. Fluorouracil (5-FU), Prodrugs, and Combinations Biologies.

**5.1.1.1. Fluorouracil (5-FU) Biology.:** In 1954, Rutman and colleagues observed that rat liver tumor cells utilized more uracil **18** as compared to normal liver cells.<sup>114</sup> Thymidylate synthase can catalyze the conversion of deoxyuridine-5'-monophosphate to thymidine-5'-monophosphate using 5,10-methylenetetrahydrofolate as a methyl source. These observations led Heidelberger and colleagues to synthesize fluorouracil (5-FU, **2**). Their hypothesis was that the 5-FU **2** would be converted to 5-fluorouracil-5'-monophosphate, and this product would selectively inhibit thymidylate synthase by forming a complex that could not breakdown due to prevention of elimination by the 5-fluorine (Figure 8).<sup>36,115</sup> Furthermore, they surmised that this would selectively target tumor cells, since 5-FU-5'-monophosphate would be found at higher concentrations in these cells. Moreover, the inhibition of thymidylate synthase and subsequent decrease of thymidine would be deleterious for cancer cells growth.

These researchers subsequently validated their hypothesis by showing that 5-FU **2** inhibited the production of thymine *in vitro*,<sup>36</sup> and soon thereafter reported that the agent inhibited tumor growth in animals.<sup>116</sup> Other reports have shown that the main mechanism of action for 5-FU **2** was the inhibition of synthesis,<sup>117</sup> although RNA metabolism was also influenced.<sup>118</sup>

5-FU **2** is used as a palliative treatment for colorectal, head and neck, and breast cancers.<sup>119</sup> Treatment can also include a biomodulator, such as leucovorin, which increases folate cofactors and the efficacy of 5-FU **2**.<sup>120,121</sup> Although introduced over 50 years ago, 5-FU **2** is still extensively studied, with over 2000 clinical trials listed at [ClinicalTrials.gov](https://clinicaltrials.gov).

Mechanisms of action for 5-FU **2** are complex, and an overview of its metabolism is shown in Figure 9. The agent is administered by intravenous infusion and is rapidly eliminated from the plasma, having a half-life of 8–20 min.<sup>122</sup> 5-FU **2** enters the cell by facilitated transport,<sup>123</sup> and subsequent ribosylation and phosphorylation occur by orotate phosphoribosyltransferase or in two steps via uridine phosphorylase and uridine kinase.<sup>119,124</sup> Nucleotide kinases convert 5-FU-5'-monophosphate **IV** to the active metabolite 5-FU-5'-triphosphate **VIII**, which may be incorporated into many species of RNA. This leads to several levels of RNA disruption, including rRNA maturation inhibition<sup>101,125</sup> and disruption of post-translational modifications of tRNAs,<sup>126</sup> which can lead to cell death.

5-FU-5'-diphosphate **VII** is also a substrate of ribonucleotide reductase, resulting in 2'-deoxyribose-5-FU-5'-diphosphate generation. It is further phosphorylated to 2'-deoxyribose-5-FU-5'-triphosphate **VI** (an additional active metabolite), which then can be incorporated into DNA by cellular polymerases. 2'-Deoxyribose-5-FU-5'-monophosphate embedded into DNA does not act as a chain terminator or delayed chain terminator as



compared to other antimetabolites. Instead, uracil glycosylase can excise the embedded 2'-deoxyribose-5-FU-5'-monophosphate from the DNA, or the activation of homologous recombination can occur.<sup>127</sup> This apyrimidinic site is recognized by apurinic/apyrimidinic endonuclease 1 and can promote a DNA single-strand break.<sup>17</sup> DNA repair enzymes recognize this break in the DNA, and a futile cycle of excision, repair, and nucleotide misincorporation ensues, which eventually causes cell death.

2'-Deoxyribose-5-FU-5'-triphosphate **VI** is converted by dUTPase to 2'-deoxyribose-5-FU-5'-monophosphate **II**, which then inhibits the enzyme thymidylate synthase. Inhibition of thymidylate synthase leads to a decrease in dTMP cellular concentration and an increase in dUMP concentration. As a result, less dTTP is available for DNA polymerases during DNA synthesis. This can promote the misincorporation of dUTP into newly synthesized DNA, which may result in the futile cycle mentioned above. Incorporation of dUTP into DNA has been shown to be mutagenic and promote genomic instability.<sup>128-130</sup>

There are several major obstacles for the effective use of 5-FU **2** as a chemotherapeutic. First, up to 80% of administered 5-FU **2** is catabolyzed by dihydropyrimidine dehydrogenase, which is abundantly expressed in the liver and acts as an innate resistance mechanism. Dihydropyrimidine dehydrogenase converts 5-FU **2** to dihydrofluorouracil **III** and is the rate-limiting enzyme in the catabolic pathway.<sup>122</sup> Secondly, 2'-deoxy-5-FU-5'-monophosphate and 5-FU-5'-monophosphate metabolites (**II** and **IV**) can be exported from cancer cells via the ABC transporters MRP5 and MRP8, limiting the therapeutic impact of this compound *in vitro*.<sup>131,132</sup>

Several mechanisms for 5-FU **2** resistance have been reported. The miRNA-320a has been shown to downregulate PDCD4 (programmed cell death protein 4) gene expression, thereby promoting 5-FU **2** resistance *in vitro*.<sup>133</sup> Also, miRNA-21 induces resistance by regulating PTEN and PDCD4 levels *in vitro*, which correlate to cell survival and cell death pathways, respectively.<sup>134</sup> Overexpression of nicotinamide *N*-methyltransferase is common in many cancer cell lines and has been linked to 5-FU **2** resistance by regulating nitric oxide production *in vitro*. Reducing reactive oxygen species leads to inactivation of the ASK1-p38 mitogen-activated protein kinase (MAPK) pathway, leading to a reduction in 5-FU **2**-induced apoptosis *in vitro*.<sup>135</sup> Furthermore, putrescine released by macrophage has been shown to promote 5-FU **2** resistance *in vitro*, which might be important for regulating 5-FU **2** concentration in the microenvironment of solid tumors *in vivo*.<sup>136</sup> Overall, 5-FU **2** resistance appears to be a multifactorial event which includes transport mechanisms, metabolism, molecular mechanisms, protection from apoptosis, and resistance via cell cycle kinetics.

In patients, 5-FU **2** resistance has been linked to thymidylate synthase methylenetetrahydrofolate reductase, dihydropyrimidine dehydrogenase, and thymidine phosphorylase, which influence 5-FU **2** metabolism.<sup>137,138</sup> There have been over 1500 clinical trials performed that evaluated 5-FU **2** therapy listed at [ClinicalTrials.gov](https://clinicaltrials.gov).

**5.1.1.2. Doxifluridine Biology:** 5-FU **2** is administered intravenously due to the high level of dihydropyrimidine dehydrogenase in the gut wall, which rapidly degrades the agent.



Researchers at Roche used prodrug strategies to develop an orally available prodrug that overcomes degradation in the gut leading to enhanced absorption and an improved pharmacokinetic profile.<sup>139</sup> This work led to the development of doxifluridine (5'-deoxy-5-fluorouridine) **23**, an orally available prodrug which is converted to 5-FU **2** in tumors by either thymidine phosphorylase or pyrimidine-nucleoside phosphorylase.<sup>140,141</sup>

High expression levels of pyrimidine-nucleoside phosphorylase have been reported for esophageal squamous cell carcinoma and colorectal cancer patients.<sup>142,143</sup> Furthermore, thymidine phosphorylase is expressed at relatively high levels in tumor tissues such as esophageal, breast, cervical, pancreatic, and hepatic,<sup>144,145</sup> which increase drug specificity. However, high thymidine phosphorylase expression is also found in the human intestinal tract. Therefore, doxifluridine treatment can result in dose-limiting toxicity (diarrhea) in some individuals.<sup>146</sup> In addition, the most frequent adverse effects for doxifluridine **23** were neurotoxicity and mucositis, whereas leukopenia and nausea were reported for 5-FU **2** in patients.<sup>147</sup> Several clinical trials have been done using doxifluridine **23** as a monotherapy or in combination therapy with mitomycin-C and cisplatin (administered intravenously).<sup>148</sup>

**5.1.1.3. Capecitabine Biology.:** As mentioned above, doxifluridine **23** treatment had gastrointestinal toxicity in patients. This led Roche to develop capecitabine **10** (5'-deoxy-5-fluoro-*N*-[(pentoxycarbonyl)cytidine], a third-generation prodrug of 5-FU **2** that was FDA-approved in 1998.<sup>149,150</sup> Capecitabine **10** is almost 100% orally bioavailable and is not a substrate for thymidine phosphorylase.<sup>151</sup> After passage through the intestinal mucosa, carboxylesterases in the liver cleave the *N*<sup>4</sup>-pentyl carbamate. Next, cytidine deaminase, in either the liver or tumor, catalyses the formation of doxifluridine **23**, which is subsequently converted by thymidine phosphorylase to 5-FU **2**.<sup>152</sup> Capecitabine **10** was highly effective in HCT116 human cancer xenograft models and showed better selective tumor delivery as compared to either doxifluridine **23** or 5-FU **2**.<sup>153</sup> In addition, it was found that radiation therapy stimulated thymidine phosphorylase activity, and combination therapy with capecitabine **10** enhanced tumor cell selectivity.<sup>154</sup>

Drug resistance to capecitabine **10** appears to involve the interplay between dihydropyrimidine dehydrogenase and thymidine phosphorylase, which are involved in the conversion of capecitabine **10** to 5-FU **2** and the degradation of 5-FU **2** within tumors.<sup>155</sup> In addition to these resistance mechanisms, drug efflux is associated with gene polymorphisms of the ABCB1 transporter in patients.<sup>66</sup>

Currently, capecitabine **10** is FDA-approved for treatment of metastatic colorectal and breast cancers. Moreover, several clinical trials are underway using capecitabine **10** in combination therapy. A Phase II study for capecitabine **10** and bendamustine (DNA alkylating agent) in women with pretreated locally advanced or metastatic Her2-negative breast cancer (MBC-6) is being conducted ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01891227) identifier: NCT01891227—active status—not recruiting patients). A Phase III study is being conducted that examines capecitabine **10** maintenance therapy following treatment with capecitabine **10** in combination with docetaxel (an antimetabolic taxane) in treatment of mBC (CAMELLIA study) ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01917279) identifier: NCT01917279—recruiting patients). Capecitabine **10** is also being evaluated as a follow up therapy for efficacy of capecitabine **10** metronomic

chemotherapy to triple-negative breast cancer (SYSUCC-001 study) ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01112826) identifier: [NCT01112826](https://clinicaltrials.gov/ct2/show/study/NCT01112826)—recruiting patients). It is also under examination for maintenance therapy: a Phase II study of maintenance capecitabine **10** to treat resectable colorectal cancer (CAMCO study) ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01880658) identifier: [NCT01880658](https://clinicaltrials.gov/ct2/show/study/NCT01880658)—recruiting patients).

**5.1.1.4. NUC-3373 Biology.:** In 2011 a novel 5-FU **2** prodrug NUC-3373 **37** (aka. NUC-3073 and FUDR) was reported having potent anticancer activity *in vitro*.<sup>156,157</sup> NUC-3373 **37**, a phosphoramidate prodrug of floxuridine **5**, was shown to be resistant to thymidine phosphorylase activity, and showed (in comparison to **5**) a significantly decreased loss of potency in CEM cells deficient in hENT1 transporters.<sup>157</sup> The authors also proposed a mechanism for the prodrug decomposition sequence including enzyme-mediated carboxyl ester hydrolysis, spontaneous cyclization with concomitant loss of the aryloxy leaving group, water-mediated opening of the mixed anhydride, and phosphoramidase-mediated P–N bond cleavage, resulting in 2'-deoxyribose-5-FU-5'-monophosphate **II** (Figure 10).

NUC-3373 **37** is a mixture of phosphoramidate diastereomers, and it has been shown to have many impressive prodrug attributes, overcoming the major 5-FU **2** resistance mechanisms. The agent retains activity in thymidine kinase deficient cells, is resistant to thymidine phosphorylase activity, and maintains activity in hENT1 deficient cells by entering independently of nucleoside transporters. It is resistant to dihydropyrimidine dehydrogenase catabolism, shows lower toxicity than 5-FU **2** in a variety of cell lines, and achieves high intracellular levels of 2'-deoxyribose-5-FU-5'-monophosphate **II**. Furthermore, the agent rapidly distributes to tissues following bolus infusion, shows low plasma levels of degradation metabolites, and reduces tumor weight in HT29 colorectal xenografts significantly more than 5-FU **2**.<sup>157</sup> This impressive profile has led to NUC-3373 **37** entering a Phase I clinical trial in late 2015 for treatment of advanced solid tumors ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02723240) identifier: [NCT02723240](https://clinicaltrials.gov/ct2/show/study/NCT02723240)—recruiting patients).

**5.1.1.5. Floxuridine and 5'-Fluoro-2'-deoxycytidine with Tetrahydrouridine Biologies.:** The National Cancer Institute was an early developer of floxuridine **5** (5-fluoro-2'-deoxyuridine, FdUrd), which has been reported to be 10–100 times more effective at inhibiting tumor cell proliferation *in vitro* than 5-FU **2**.<sup>158</sup> Floxuridine **5** gained FDA approval in 1970, and was marketed by Roche.<sup>159</sup> It is approved, but not widely used, for colon and colorectal cancers that have metastasized to the liver, kidney, and stomach.<sup>160</sup>

Floxuridine **5** is a hydrophilic compound, and is likely transported into cells by pyrimidine nucleoside transporters.<sup>161</sup> Thymidine kinase converts floxuridine **5** to floxuridine-5'-monophosphate **II**, which, as mentioned above, inhibits thymidylate synthase (Figure 9).

In the intestine, a significant amount of floxuridine **5** is converted to 5-FU **2** by thymidine phosphorylase. Therefore, hepatic arterial infusion has been employed to bypass the intestine and deliver more intact floxuridine **5** to the liver.<sup>162</sup> Furthermore, the agent exhibits two properties associated with drugs that are best fit for hepatic arterial infusion: a high hepatic extraction and a short plasma half-life.<sup>160</sup> These properties can lead to increased hepatic levels of floxuridine **5** and decreased systemic levels of the compound, which can

decrease overall toxicity. This procedure has shown improvements for floxuridine **5** treatment for colorectal cancers.<sup>163–165</sup>

Several mechanisms of floxuridine resistance have been reported. Studies have shown a decrease in orotate phosphoribosyltransferase activity and thymidine kinase activity *in vitro*, as well as an increase of thymidylate synthase mRNA expression *in vitro*.<sup>166</sup> In patients, an increase in thymidylate synthase levels has been correlated with floxuridine resistance.<sup>167</sup>

Several clinical trials that include floxuridine **5** are currently underway. These include hepatic arterial infusion with floxuridine **5** and the anti-inflammatory and immunosuppressive steroid derivative dexamethasone (ClinicalTrials.gov identifier: [NCT01862315](#)—recruiting patients). Another trial examines hepatic arterial infusion with floxuridine **5** and dexamethasone in combination with gemcitabine hydrochloride **9** or the antiangiogenic monoclonal antibody bevacizumab (ClinicalTrials.gov identifiers: [NCT01938729](#)—recruiting patients and [NCT00200200](#)—active status—not yet recruiting patients). Recent results from a Phase I trial using floxuridine **5** with modified oxaliplatin, 5-fluorouracil **2**, and leucovorin (m-FOLFOX6) were reported.<sup>164</sup>

A major problem for anticancer cytidine analogs (gemcitabine, cytarabine **4**, azacytidine **11**, decitabine **14**, capecitabine **10**, and 5-fluoro-2'-deoxycytidine **38**) is that they are substrates for cytidine deaminase and can be rapidly converted to uridine analogs, which are often inactive metabolites.<sup>168</sup> For example, the liver expresses high levels of cytidine deaminase that provide a microenvironment to protect cancer cells from decitabine **14** treatment.<sup>169</sup> In attempts to overcome the problem of deamination, cytidine deaminase inhibitors, tetrahydrouridine (**39**, NSC-112907) and (4*R*)-2'-deoxy-2',2'-difluoro-3,4,5,6-tetrahydrouridine (Figure 11), are being investigated with nucleoside analogs as potential combination therapies.

Tetrahydrouridine **39** has been investigated since the 1970s and been evaluated in cell lines and mice with cytarabine **4**, gemcitabine, azacytidine (**11**, and decitabine **14**).<sup>170–174</sup> However, clinical data has shown that patients with the cytidine deaminase gene G208A polymorphism had several adverse reactions to gemcitabine treatment.<sup>175,176</sup> In addition, the cytidine deaminase gene K27Q polymorphism promotes greater catalytic activity toward deoxycytine and cytarabine **4**.<sup>177</sup> Collectively, these data suggest prescreening patients for polymorphisms would be helpful in guiding treatment options.

Currently 5-fluoro-2'-deoxycytidine (FdCyd, **38**) is being studied in combination with tetrahydrouridine **39** in mice.<sup>168,174</sup> Coadministration of **38** with tetrahydrouridine **39** led to higher levels of FdCyd-5'-monophosphate, produced by 2'-deoxycytidine kinase in mice.<sup>178</sup> Once in the cell, FdCyd **38** can be converted to FdUrd by the action of cytidine deaminase, and subsequently, phosphorylation leads to FdUrd-5'-monophosphate, which inhibits thymidylate synthase. Furthermore, FdCyd-5'-monophosphate can be phosphorylated to FdCyd-5'-triphosphate and incorporated into DNA. Once incorporated into DNA, FdCyd **38** has been shown to inhibit DNA methyltransferase with comparable activity to azacytidine **11** (see Section 5.2.1).<sup>179</sup> Therefore, FdCyd **38** exposure can lead to the following direct and indirect cytotoxic activities in cells: (1) FdCyd-5'-monophosphate

inhibition of DNA methylation, (2) 2'-deoxyribose-5-FU-5'-monophosphate **II** inhibition of thymidylate synthase, and (3) 2'-deoxy-5-FU-5'-triphosphate **VI** and 5-FU-5'-triphosphate **VIII** incorporation into DNA and RNA, respectively.<sup>180</sup>

Both preclinical and clinical trials are currently underway using FdCyd **38** with tetrahydrouridine **39**. In 2016, a preclinical evaluation of FdCyd **38** and tetrahydrouridine **39** in pediatric brain tumors was reported,<sup>181</sup> and a methods development report was published evaluating a combination treatment of tetrahydrouridine **39** and FdCyd **38** from human serum using HPLC-MS analysis.<sup>182</sup> In 2015, Newman *et al.* reported a Phase I trial that examined the toxicity, plasma exposures, peak response, and dosing for the two compounds.<sup>183</sup> Additional clinical studies are underway for patients with advanced non-small cell lung cancer, breast cancer, bladder cancer, head and neck cancer, and solid tumors ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00978250) identifiers: [NCT00978250](https://clinicaltrials.gov/ct2/show/study/NCT00978250)—recruiting patients and [NCT01534598](https://clinicaltrials.gov/ct2/show/study/NCT01534598)—recruiting patients). Recently, the oral and intravenous pharmacokinetics of FdCyd **38** and tetrahydrouridine **39** were reported for cynomolgus monkeys and humans trial.<sup>184</sup>

(4*R*)-2'-Deoxy-2',2'-difluoro-3,4,5,6-tetrahydrouridine is a newer cytidine deaminase inhibitor being evaluated in preclinical studies.<sup>185</sup> The agent has an IC<sub>50</sub> of 0.4 μM for cytidine deaminase and exhibited enhanced acid stability at its *N*-glycosyl bond as compared to tetrahydrouridine **39**. Rhesus monkeys cotreated with (4*R*)-2'-deoxy-2',2'-difluoro-3,4,5,6-tetrahydrouridine and decitabine **14** had higher detectable levels of **14** in the serum.

**5.1.1.6. Tegafur-Uracil, TS-1, Carmofur, and Flucytosine Biologies.:** Tegafur **19**, TS-1, carmofur **24**, and flucytosine **31** are other examples of prodrugs of 5-FU **2** (Figure 9). The tegafururacil combination, TS-1 (a triple drug combo including tegafur **19**, chloro dihydropyridine **20**, and potassium oxonate **21**), and carmofur (**24**, mifurof), are used globally in cancer treatments.<sup>186</sup> In 1983, tegafur–uracil (**18** + **19**) was approved for use in Japan. The orally bioavailable combination is given in a 4:1 mol ratio of tegafur **19** and uracil **18**. Tegafur **19** is metabolized to 5-FU **2**, which can be converted to the inactive dihydrofluorouracil mainly by dihydropyrimidine dehydrogenase in the liver.<sup>187</sup> The coadministration of uracil **18** inhibits this enzyme, thus helping maintain high levels of 5-FU **2** in the liver and in the circulation. Tegafur–uracil (**18** + **19**) is approved in Japan and Taiwan for various advanced gastrointestinal cancers. Patients are still being recruited for a Phase II clinical trial using metronomic therapy with tegafur–uracil (**18** + **19**) to treat head and neck cancer ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00855881) identifier: [NCT00855881](https://clinicaltrials.gov/ct2/show/study/NCT00855881)—recruiting patients). While a Phase II study in combination with sorafenib (kinase inhibitor) for advanced hepatocellular carcinoma has been terminated in Jan 2015, with no results posted ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01539018) identifier: [NCT01539018](https://clinicaltrials.gov/ct2/show/study/NCT01539018)—terminated in 2015—no results posted).

TS-1 includes tegafur **19**, chlorodihydropyridine (gimeracil) **20**, and potassium oxonate **21** used in a respective molar ratio of 1:0.4:1.<sup>188</sup> TS-1 (**19** + **20** + **21**) is an orally administered triple combination drug regiment, which was approved in Japan in 1999. Chlorodihydropyridine **20** is a more potent inhibitor of dihydropyrimidine dehydrogenase than uracil **18**.<sup>188,189</sup> Potassium oxonate **21** may decrease gastrointestinal toxicity by selectively inhibiting 5-FU **2** activity within the intestinal lumen. TS-1 (**19** + **20** + **21**) is

approved for use and is currently undergoing multiple clinical trials. Some of these studies include coadministration with cisplatin (DNA alkylating agent) for treatment of advanced non-small-cell lung cell cancer ([ClinicalTrials.gov](#) identifier: [NCT01874678](#)—completed status—with no results), a Phase II study for combination treatment of advanced hepatocellular carcinoma with the DNA alkylating agent oxaliplatin ([ClinicalTrials.gov](#) identifier: [NCT01429961](#)—unknown status—no updates for 2 years), and a Phase II study for monotherapy treatment of advanced metastatic breast cancer ([ClinicalTrials.gov](#) identifier: [NCT01492543](#)—unknown status—no updates for 2 years). In addition, Phase I clinical studies using combination therapy of TS-1 (**19** + **20** + **21**) and TAS-114, a first-in-class dual inhibitor of dUTPase (deoxyuridine-5'-triphosphate nucleotide hydro-lase) and dihydropyrimidine dehydrogenase,<sup>190</sup> is being conducted ([ClinicalTrials.gov](#) identifier: [NCT01610479](#)—active status—not recruiting).

Product marketing for carmofur **24** started in 1981. Carmofur **24** is a lipophilic-masked analog of 5-FU **2** that can be administered orally. The carbamoyl moiety of the drug is removed *in vivo* to release 5-FU **2**. Carmofur **24** has been used to treat colorectal cancer in China, Japan, and Finland for many years.<sup>186</sup> However, the agent has been shown to induce delayed leukoencephalopathy, characterized by progressive damage to white matter in the brain with stroke-like symptoms.<sup>191–193</sup> There are currently no open clinical trials for this agent. A trial treating patients with stage II hepatocellular carcinoma was suspended prematurely, because 56% of the treated patients had unacceptable side effects and offered no survival advantage for certain cancers in stage 1 and 2 of the disease.<sup>194</sup> This may be a reason why carmofur **24** was never pursued for FDA-approval in the US.

Flucytosine **31** is another prodrug of 5-FU **2** in clinical trials. It is converted to 5-FU **2** by cytosine deaminase, which is not encoded by the human genome. Therefore, flucytosine **31** is used in combination with gene therapy of cancer cells in an attempt to have drug-specific targeting.<sup>195</sup> Engineered mesenchymal stromal cells have been examined to convert flucytosine **31** into its active compound,<sup>196</sup> and modified neural stem cells are currently being tested for the same purpose ([ClinicalTrials.gov](#) identifiers: [NCT02015819](#)—recruiting status, and [NCT01172964](#)—completed status—no results posted). Flucytosine **31** is currently in a Phase I/II clinical trial in combination with maltose and APS001F, a recombinant anaerobic bacteria *Bifidobacterium* engineered to express the cytosine deaminase gene, for treatment of solid tumors ([ClinicalTrials.gov](#) identifier: [NCT01562626](#)—recruiting status).

The progression of fluoropyrimidine analogs shows the importance of a biological observation–chemical modification interplay. The intravenously-administered 5-FU **2** and floxuridine **5** was followed by the orally-bioavailable carmofur **24** and flucytosine **31**. The next step in the progression involved drug combinations TS-1 (**19** + **20** + **21**), tegafur–uracil (**18** + **19**), and the recently FDA-approved TAS-102 (**15** and **16**), in order to decrease 5-FU **2** degradation. Further chemical modifications led to double (doxifluridine **23**) and triple (capecitabine **10**) prodrugs, which showed improvement in toxicity and pharmacokinetics profiles. Currently, NUC-3373 **37** appears to overcome both innate and drug-derived resistance mechanisms. Newer analogs could be used as combination therapy with DNA

damaging agents such as mitomycin-C and cisplatin, and immunotherapies—antiCTLA-4 and anti-PD-1, to improve patient survival rates.

### 5.1.2. Fluorouracil (5-FU), Fluorouracil Prodrugs, and Combinations Syntheses.

**5.1.2.1. Fluorouracil (5-FU) Synthesis.:** In 1957, a synthesis of 5-FU **2** was reported in which the compound was prepared from pseudourea salts and  $\alpha$ -fluoro- $\beta$ -keto ester enolates.<sup>197–199</sup> Twenty years later, Robins *et al.* reported a two-step synthesis of 5-FU **2** from uracil **18** (Scheme 5).<sup>200</sup> Trifluoromethyl hypofluorite was used to introduce a fluorine atom into the uracil ring, and the resulting fluorinated adduct **66** underwent base-catalyzed elimination of MeOH, generating 5-fluoro-pyrimidine **2**. An alternative synthesis used the cheap and readily available orotic acid (vitamin B13, **67**) as the starting material.<sup>201</sup> Fluorination of **67** with fluorine gas and trifluoroacetic acid furnished acid **68**, which in turn was decarboxylated to produce 5-FU **2** on a gram-scale.

Baasner *et al.* reported a 100-g-scale preparation of 5-FU **2**.<sup>202</sup> This procedure used selective fluorine/chlorine exchange and chlorine hydrogenolysis reactions (Scheme 6). Tetrafluoropyr-imidine **72**, which can be prepared in four steps from urea **69** and diethylmalonate **70** via 2,4,5-trichloropyrimidine **71**, was treated with HCl gas forming the dichlorinated intermediate **73**. Dechlorination by selective palladium-catalyzed hydrogenolysis was followed by hydrolysis with aqueous sodium hydroxide to complete the synthesis.

**5.1.2.2. Doxifluridine and Capecitabine Syntheses.:** The first synthesis of capecitabine **10** was reported in the mid-1990s (Scheme 7).<sup>150,203</sup> Compound **79** is a key intermediate and is synthesized in six steps from D-ribose through a cyclization, mesylation, iodidation, reductive dehalogenation, hydrolysis, and cyclization/acetylation sequence. 5-Fluorocytosine, prepared by treating cytosine with trifluoromethyl hypofluorite in MeOH,<sup>204</sup> was glycosylated with protected sugar **79** in good yield.  $N^4$ -Amino acylation followed by selective deprotection of the 2',3'-acetyl groups by treatment with aqueous sodium methoxide afforded crude capecitabine **10**. Finally, pure capecitabine **10** was precipitated as a white solid from ethyl acetate/hexanes on a 100 gram-scale.

Two large-scale process syntheses of capecitabine **10** were reported by Hoffman-La Roche<sup>205</sup> and Gore *et al.*<sup>206</sup> Both syntheses use 5'-deoxy-5-fluorocytidine **85** as a key intermediate (Scheme 8). 5-Fluorocytidine **81**, synthesized by treating protected cytidine with CF<sub>3</sub>OF/CCl<sub>3</sub>F,<sup>200</sup> was protected, tosylated, and subsequently converted to 5'-iodo **83**.<sup>207</sup> Reductive dehalogenation followed by deprotection produced intermediate **85**.<sup>208</sup> This same sequence can also be employed to prepare doxifluridine **23**, although using 5-fluorouridine as the starting material.

The Hoffmann-La Roche synthesis converted **85** to the trisacylated intermediate **86**, using 3 equiv of *n*-pentyl chloroformate in cold dichloromethane and pyridine. Selective removal of 2',3'-ester groups was accomplished by treatment with aqueous sodium hydroxide in cold methanol, and precipitation from cold ethyl acetate afforded pure capecitabine **10**.



The Gore *et al.* approach converted **85** directly to the product **10** by selective  $N^4$  acylation. In refluxing THF, intermediate **85** was reacted with pentyloxycarbonyl-1-hydroxybenzotriazole **87**, which can be easily made from pentyl chloroformate and HOBt. This was followed by aqueous workup and precipitation to afford **10**. Other acylating agents, such as pentyloxycarbonyl-imidazole, pentyloxycarbonyl-4-nitrophenoxy, and pentyloxycarbonyl-pentafluorophenoxy were also used, and produced similar yields.

The Jamison group used continuous flow chemistry to prepare capecitabine **10**.<sup>209</sup> The reaction exhibited impressive yields, “green” conditions, and very short reaction times (Scheme 9). The glycosylation reaction between triacetate intermediate **88** and silylated pyrimidine **90**, while under Brønsted acid-catalyzed conditions, was completed in 20 min. This procedure does not require aqueous workup. Mixing conditions were carefully monitored to form the  $N^4$ -carbamate intermediate. Deprotection with a mixture of NaOH/MeOH/H<sub>2</sub>O occurred at room temperature in only 2 min. The overall process produced milligrams of capecitabine **10** in less than 1 h and a 72% yield from starting materials **88** and **90**.

In the same report,<sup>209</sup> doxifluridine **23** was also synthesized using protected sugar **88**, which was condensed with silylated 5-FU **2** in a glycosylation reaction. Deprotection of the resulting intermediate with sodium methoxide furnished the product. This one-flow, two-step process produced doxifluridine **23** in 89% yield in 10 min.

**5.1.2.3. NUC-3373 Synthesis.:** In 2011, McGuigan and co-workers reported the synthesis of NUC-3373 **37**.<sup>156</sup> The agent was prepared by reaction of phosphochloridate **93** (produced using L-alanine benzyl ester salt and 1-naphthyl-dichlorophosphate **92**) with floxuridine **5** in THF in the presence of *t*-BuMgCl (Scheme 10). The synthesis produces milligrams of the final product and requires column chromatography to produce the purified mixture of diastereomers. Furthermore, no large-scale syntheses of NUC-3373 **37** have been reported.

**5.1.2.4. Floxuridine, 5-Fluoro-2'-deoxycytidine, and Tetrahydrouridine Syntheses.:** In the late 1950s, floxuridine **5** was originally prepared enzymatically and later converted to 5-fluoro-2'-deoxycytidine **38** via chemical synthesis.<sup>210–212</sup> Robin's method (see Sections 5.1.2.1) was used to prepare both compounds in only two steps from 2'-deoxyuridine **94a** and 2'-deoxycytidine **94b**, with the key ring fluorination and sugar deprotection occurring in one pot (Scheme 11). This approach has the advantage of starting from a deoxynucleoside, which already has the needed stereochemistry fixed.

Surprisingly few syntheses of tetrahydrouridine **39** have been reported. In 1967, Hanze reported two approaches to the compound (Scheme 11).<sup>213</sup> Cytidine **96** was “overreduced” to tetrahydrocytidine **97** in one step using rhodium on aluminum in water, and subsequent hydrolysis afforded tetrahydrouridine **39**. Uridine **98** was reduced to 5,6-dihydrouridine **99** also using rhodium and aluminum, and subsequent treatment with sodium borohydride furnished the target compound. The syntheses suffered from complex reaction mixtures, and no yields were reported.



Aoyama reported a gram-scale synthesis of floxuridine **5**,<sup>214</sup> in which the key step was the condensation of 5-fluoro-2,4-bis(trimethylsilyloxy)pyrimidine **102** with 3,5-bis(*O*-*p*-chorobenzoyl)-2-deoxy- $\alpha$ -D-ribofuranosyl chloride **101**. Chlorosugar **101** was made as the pure  $\alpha$  anomer in three steps from 2'-deoxyribose (Scheme 12).<sup>215,216</sup> The condensation was performed in the presence of *p*-nitrophenol, which led, interestingly, to the stereoselective production of the desired  $\beta$ -anomer **103**, whereas the combination of *p*-nitrophenol and pyridine as catalysts led to formation of the  $\alpha$ -anomer. Recrystallization from acetic acid gave the pure  $\beta$ -anomer, which was deprotected with methanolic ammonia, concluding the synthesis.

**5.1.2.5. Tegafur, Carmofur, and Flucytosine Syntheses.:** In 1976, an original synthesis of tegafur **19** was reported that coupled 2-acetoxytetrahydrofuran with silylated 5-FU **2** in the presence of sodium iodide in 95% yield.<sup>70,217</sup> Five years later, Miyashita *et al.* reported an alternative preparation (Scheme 13).<sup>218</sup> The synthesis started with a one-pot condensation of urea **69**, triethyl orthformate **104**, and methyl malonate **105** under solvent-free conditions. The resulting methyl ureidomethylenemalonate **106** was then cyclized in methanolic sodium methoxide to give 5-methoxycarbonyluracil **107**. Condensation of **107** with 2,3-dihydrofuran followed by ring fluorination and subsequent hydrolysis concluded the synthesis of **19**.

Carmofur **24** has been synthesized by treating 5-FU **2** with phosgene and hexylamine.<sup>219,220</sup> An alternative approach also utilized 5-FU **2** as starting material but used aqueous formaldehyde to effect N-formylation. This was followed by oxidation with potassium bromate and finally condensation with hexylamine to conclude the synthesis of **24** (Scheme 13).<sup>221</sup>

Flucytosine **31** was synthesized by reacting cytidine **110** with CF<sub>3</sub>OF/CCl<sub>3</sub>F in methanol for 5 min followed by addition of triethylamine in water (Scheme 14).<sup>200</sup>

**5.1.2.6. Gimeracil and Oteracil Syntheses.:** In 1953, Kolder and Hertog reported a synthesis of the TS-1 additive gimeracil **20**, which was completed in seven steps using 4-nitropyridine *N*-oxide as starting material.<sup>222</sup> Later, Yano *et al.* reported an alternative gram-scale synthesis (Scheme 15).<sup>223</sup> The one-pot, three component condensation of malononitrile **111**, 1,1,1-trimethoxyethane, and 1,1-dimethoxytrimethylamine generated the dicyano intermediate **112**, which was into 2(*1H*)-pyridinone **113**.<sup>224</sup> Selective chlorination of **113** was followed by acid-mediated demethylation, hydrolysis, and decarboxylation, to afford gimeracil **20**. Interestingly, Xu *et al.* found that treatment of intermediate **113** with sulfuryl chloride resulted in dichloro **115** formation, which could still be converted to gimeracil **20** by treatment with sulfuric acid.<sup>225</sup>

Poje *et al.* reported a two-step, gram-scale preparation of the TS-1 additive oteracil **21** (Scheme 16).<sup>226</sup> Iodine-mediated-oxidation of uric acid **116** produced dehydroallantoin **117** as the major product, and subsequent treatment with potassium hydroxide resulted in the rearranged product oteracil **21**.<sup>227</sup>

**5.1.3. Trifluorothymidine and Tipiracil Hydrochloride (TAS-102) Biology.**—In 1964, Heidelberger and co-workers synthesized trifluorothymidine (TFT, **15**) and demonstrated that it had promising anticancer activity.<sup>228–231</sup> Initial animal studies showed

that TFT **15** was degraded to trifluorothymidine and catabolite 5-carboxyuracil, in the liver, spleen, and intestines.<sup>230</sup> The agent showed reductions in some tumor sizes during a clinical trial but, like other antimetabolites, exhibited a plasma half-life of 15 min in cancer patients.<sup>232</sup> Research on TFT **15** was discontinued due to inadequate information on the pharmacokinetics and toxicity profile.<sup>233,234</sup>

TFT **15** is a substrate for thymidine kinase, generating TFT-5'-monophosphate (an active intracellular metabolite). Similar to 5-FU-5'-monophosphate **II**, TFT-5'-monophosphate inhibits thymidylate synthase (Figure 8).<sup>235</sup> In contrast to 5-FU-5'-monophosphate **II**, TFT-5'-monophosphate does not form a ternary complex with thymidylate synthase but inhibits it by binding to the active site of the enzyme.<sup>236</sup> Furthermore, TFT-5'-monophosphate is a reversible inhibitor of thymidylate synthase, and removal of the agent results in rapid recovery of enzyme activity, whereas inhibition caused by formation of the 5-FU-5'-monophosphate **II** ternary complex has prolonged effects.<sup>237</sup> TFT-5'-monophosphate can be further phosphorylated to its triphosphate form (another active metabolite) and then incorporated into DNA to promote single-strand breaks.<sup>238</sup> DNA-incorporated TFT-5'-monophosphate is resistant to DNA glycosylase,<sup>239</sup> and incorporated TFT-5'-monophosphate can promote DNA instability and double-stranded breaks.<sup>240,241</sup>

The major drawback of TFT **15** monotherapy is its rapid degradation within the body, primarily by the action of thymidine phosphorylase, providing a natural TFT resistance mechanism.<sup>242</sup> However, coadministration of TFT **15** and a thymidine phosphorylase inhibitor improved the pharmacokinetic profile in animal models and antitumor activity in cell lines and animal models.<sup>243</sup> Currently, Taiho Pharmaceuticals is developing TAS-102 (**15** + **16**) as a combination therapy that uses TFT **15** with tipiracil hydrochloride **16**, a thymidine phosphorylase inhibitor with an IC of 35 nM,<sup>243</sup> in a respective molar ratio of 2:1. This combination was shown to greatly decreased the biodegradation of TFT **15** *in vitro*.<sup>240</sup> Interestingly, tipiracil hydrochloride **16** has also been shown to have antiangiogenic activity.<sup>228</sup>

TAS-102 (**15** + **16**) was FDA-approved in Sept 2015 for treatment in patients with colorectal cancer,<sup>244</sup> and it is also currently being evaluated in clinical trials for the treatment of advanced solid tumors and metastatic refractory colorectal cancer<sup>233</sup> ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01607957) identifier: [NCT01607957](https://clinicaltrials.gov/ct2/show/study/NCT01607957)—active status, not recruiting). A study was initiated evaluating TAS-102 (**15** + **16**) plus nivolumab in patients with microsatellite stable refractory metastatic colorectal cancer ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02860546) identifier: [NCT02860546](https://clinicaltrials.gov/ct2/show/study/NCT02860546)—recruiting patients). In addition, TFT **15** has been pursued as an antiviral agent (registered as Viroptic) for use against the herpes simplex virus.<sup>245,246</sup>

#### 5.1.4. Trifluorothymidine and Tipiracil Hydrochloride (TAS-102) Syntheses.—

Heidelberger *et al.* reported the first synthesis of TFT **15** (Scheme 17), featuring an enzyme-mediated transglycosylation of 5-trifluoromethyluracil **126** with thymidine.<sup>229</sup> The trifluoro base **126** was prepared in an eight-step sequence. The synthesis began with treatment of trifluoroacetone **118** with hydrogen cyanide to produce cyanohydrin **119**, and subsequent acetylation and ester pyrolysis afforded alkene **120**. Treatment of **121** with anhydrous HBr and urea provided ureidoamide **123**, which was then refluxed in acid to generate 5,6-

dihydro-5-trifluoromethyluracil **124**. This intermediate was subjected to a bromination–dehydrobromination sequence, furnishing 5-trifluoromethyluracil **126**.

Komatsu et al. have reported a more modern and enzyme-free preparation of TFT **15**.<sup>247</sup> The preparation featured a green glycosylation reaction performed on the 100-g-scale using the TMS-protected analog of thymine **127** and the corresponding chlorosugar **101**, made in two steps from 2'-deoxyribose. Conditions were sought to increase the rate of glycosylation (e.g., **101 $\alpha$**  to **128**) and decrease the rate of chlorosugar anomerization (e.g., **101 $\alpha$**  to **101 $\beta$** ), thereby increasing the yield of the  $\beta$ -glycosylation product **128** (Scheme 18).<sup>248</sup>

Nonpolar solvents were used to decrease chlorosugar anomerization, and temperature and reactant concentrations were explored empirically. Optimized large-scale conditions included condensation of silylated trifluorothymidine with an equimolar amount of chlorosugar **101** in minimal amounts of anisole (96% w/w of **101**) at 50 °C, producing an 85:15  $\beta/\alpha$  anomeric mixture of **128/129** in 71% yield. Deprotection with NaOMe at 4 °C, precipitation and filtration, and finally washing with AcOBu (to remove residual methyl 4-chlorobenzoate) produced pure trifluorothymidine **15**.

A concise 100-g-scale synthesis of tipiracil hydrochloride **16** has been reported (Scheme 19).<sup>249</sup> Amination of **134** with methanolic ammonia led to the cyclized intermediate 2-iminopyrrolidine **135**.<sup>250</sup> In parallel, 6-chloromethyl uridine, which can be prepared in four steps from urea **69** and ethyl 2-acetoacetate **130** via intermediate **131**,<sup>251</sup> was chlorinated at the 5-position to furnish the dichloromethyl uracil derivative **133**.<sup>252</sup> Finally, condensation of **133** with **135** produced tipiracil hydrochloride **16**.<sup>249</sup>

## 5.2. Azanucleosides

**5.2.1. Decitabine, Azacytidine, CP-4200, and SGI-110 Biologies.**—The azanucleosides, decitabine (Dacogen, **14**), and azacytidine (Videza, **11**), were first developed as cytostatic agents nearly half a century ago.<sup>253,254</sup> These cytostatic agents were later found to inhibit DNA methylation in human cell lines and were developed as epigenetic drugs.

Decitabine **14** is a 2'-deoxy-5-azanucleoside analog of cytidine that enters the cell by ENT-1.<sup>255</sup> It is subsequently converted to decitabine-5'-monophosphate by 2'-deoxycytidine kinase. Further phosphorylation leads to the active metabolite decitabine-5'-triphosphate, which is a substrate for DNA polymerase  $\alpha$  and is incorporated into the DNA.<sup>256</sup> The incorporated decitabine-5'-monophosphate cannot be methylated, which can influence epigenetic gene regulation.<sup>257</sup> DNA methylation usually involves the transfer of a methyl group from *S*-adenosyl-L-methionine, catalyzed by DNA methyl transferases (DNMTs), to the 5-position of a cytosine base within a cytosine-phosphate-guanosine dinucleotide.<sup>258,259</sup> Furthermore, DNA hypermethylation often silences tumor suppressor genes in hematopoietic malignancies,<sup>260</sup> and inhibition of DNA methylation at regions of decitabine-5'-monophosphate incorporation in the DNA most likely restores the expression of some of these genes.<sup>261–264</sup>

The ability of decitabine **14** to inhibit DNA methylation is often attributed to the formation of a complex between the azacytosine-guanine dinucleotide and DNMT1.<sup>265</sup> This

may possibly occur via the covalent trapping paradigm (Figure 12).<sup>266</sup> A covalent bond is formed between DNMT1 and the C6-position of the azacytosine base. Methylation of the nitrogen at position 5 of the base then occurs; however,  $\beta$ -elimination cannot occur due to the lack of a hydrogen atom at this position. As a consequence, DNMT1 remains covalently bound to the aza-base and is unable to continue its methylation activity. The covalent complex also triggers DNA damage signaling to result in DNMT1 degradation.<sup>267</sup> A recent study, however, indicated that decitabine **14** could induce degradation of methyltransferases without the formation of the covalent complex, suggesting that the agent may work by additional mechanisms as well.<sup>268</sup>

Decitabine **14** suffers from some chemical stability issues. The triazine ring of the drug is prone to hydrolytic opening and deformylation, whereas the sugar is susceptible to anomerization. Several groups have reported a half-life of decitabine **14** ranging from 3.5 to 21 h at physiological pH and temperature,<sup>269</sup> whereas the *in vivo* half-life of decitabine **14** has been reported to be 15–20 min.<sup>270</sup> Once the prodrug is metabolized to release decitabine **14** in the liver or cells, the high levels of cytidine deaminase protein will generate an inactive nucleoside byproduct,<sup>169</sup> which in turn limits the intracellular concentration and toxicity of decitabine **14**. Finally if patients have reduced levels of 2'-deoxycytidine kinase activity, this may also contribute to natural decitabine resistance, as observed with decitabine.<sup>271</sup>

At low concentrations, decitabine **14** has the ability to inhibit DNA methyltransferase 1, but it is cytotoxic at high concentrations *in vitro*.<sup>272,273</sup> Optimal treatment with decitabine **14** has been tested using continuous treatment at a low concentration in patients.<sup>274</sup> However, the agent has poor oral bioavailability (currently administered by intravenous infusion) and is also a substrate for cytidine deaminase, which provides a natural resistance mechanism. Additionally, *ex vivo* studies suggest that reduction in phosphorylation by 2'-deoxycytidine kinase may also contribute to decitabine resistance.<sup>271</sup> Attempts to overcome these problems include the preparation of a decitabine mesylate salt,<sup>275</sup> designed to increase the oral bioavailability, as well as coadministration of oral decitabine **14** with oral tetrahydrouridine **39** in patients.<sup>276</sup>

Azacytidine **11**, a riboside analog of decitabine **14**, is given by injection and enters cells by the uridine/cytidine transport system.<sup>277</sup> After phosphorylation by uridine-cytidine kinase,<sup>278,279</sup> the active metabolite azacytidine-5'-triphosphate is incorporated into RNA to disrupt RNA metabolism and protein synthesis.<sup>279</sup> Azacytidine-5'-diphosphate can also be reduced by ribonucleotide reductase to form 2'-deoxyazacytidine-5'-diphosphate and then follows the mechanism of action of decitabine **14**.

After years of clinical research and dosage refinement, conditions were finally produced to effectively treat myelodysplastic syndrome, with azacytidine **11** and decitabine **14** being FDA-approved in 2004 and 2006, respectively.<sup>280,281</sup> Myelodysplastic syndrome, also known as preleukemia, is a group of related diseases originating in the bone marrow in which hematopoietic stem cells produce ineffective myeloid cells.<sup>282</sup> A genetic polymorphism in the cytidine deaminase gene (79A>C) and promoter depletion (-31delC) can lead to a rapid-deaminator phenotype and an increase in mRNA expression, leading to increased toxicity in patients treated with azacytidine **11**.<sup>283,284</sup> Sorting patients based on

cytidine deaminase genotypes remains difficult due to genotype to phenotype relationships, whereas developing a functional cytidine deaminase assay would be beneficial.<sup>285</sup>

Azacytidine **11** and decitabine **14** treatments are being evaluated in numerous clinical studies. An oral formulation of azacytidine (CC-486, Celgene) is currently in Phase I clinical trials (alone and in combination) for treatment of refractory solid tumors and Japanese myelodysplastic syndrome (ClinicalTrials.gov identifiers: [NCT01478685](#)—completed—no results posted, and [NCT01908387](#)—terminated status—no results posted). A Phase I/II Study of azacytidine **11** and with capecitabine **10** and oxaliplatin (DNA alkylating agent) is underway (ClinicalTrials.gov identifier: [NCT01193517](#)—active status—not recruiting). A Phase II study is examining the kinase inhibitor sorafenib with azacytidine **11** for primary response and secondary toxicity profile (ClinicalTrials.gov identifier: [NCT02196857](#)—actively recruiting patients). For decitabine **14** and tetrahydrouridine **39** clinical trials, one is currently recruiting patients— adjuvant oral decitabine **14** and tetrahydrouridine **39** with or without celecoxib in people undergoing pulmonary metastasectomy (ClinicalTrials.gov identifiers: [NCT02839694](#)—recruiting patients). Four additional studies have yet to start recruiting patients with refractory/relapsed lymphoid malignancies, pancreatic cancer, and non-small cell lung cancer using different combination treatments (ClinicalTrials.gov identifiers: [NCT02846935](#)—not yet recruiting patients, [NCT02847000](#)—not yet recruiting patients, [NCT02795923](#)—not yet recruiting patients, and [NCT02664181](#)—not yet recruiting patients).

Clavis Pharma is developing CP-4200, an elaidic acid derivative of azacytidine, which has strong epigenetic modulatory potency in human cancer cell lines (Figure 13).<sup>286</sup> The aim of this drug is to circumvent therapy resistance in patients with MDS/AML due to decrease in drug uptake by hENT1, because the agent enters the cell by a hENT1-independent mechanism.<sup>287</sup> CP-4200 has not yet moved past pre-clinical development. CP-4200 might have an acceptable bioavailability profile for oral application.

Potential hurdles for CP-4200 include the following. First, upon prodrug metabolism to release azacytidine **11** in the liver or tumor cells, high levels of cytidine deaminase activity will generate an inactive nucleoside byproduct.<sup>288</sup> Second, patients with reduced levels of 2'-deoxycytidine kinase activity may have some level of natural azacytidine resistance, as observed with decitabine **14**.<sup>271</sup> Third, two other elaidic acid derivative prodrugs: elacytarabine (**51**, Phase I trial) and CP-4126 (**49**, two Phase III trials) have not been successful in clinical trials.

Astex Pharmaceuticals has developed the dinucleotide SGI-110 **22**, a second-generation prodrug of decitabine **14**, which couples the agent with deoxyguanosine via a phosphodiester bond.<sup>289</sup> It was synthesized in order to increase the *in vivo* exposure and efficacy of decitabine **14** by protecting it from cytidine deaminase activity.<sup>290,291</sup> SGI-110 **22** is being evaluated as a hypomethylation agent with an epigenetic mechanism of action in tumors.<sup>292,293</sup> In a Phase I/II dose-escalating study, SGI-110 **22** showed a favorable pharmacokinetic profile.<sup>294</sup> The dinucleotide was efficiently converted to decitabine **14** *in vivo*. Moreover, it exhibited a longer apparent half-life, lower  $C_{max}$ , and prolonged plasma exposures of decitabine **14** than equivalent amounts of IV-administered decitabine **14**.

Furthermore, SGI-110 **22** exhibited a greater stability, and lower doses of the agent achieved equal or better methylation inhibition than decitabine **14** monotherapy. SGI-110 **22** is currently in several Phase I/II studies (as monotherapy and combination therapy) for treatment of ovarian cancer, metastatic colorectal cancer, myelodysplastic syndrome, acute myelogenous leukemia, and advanced hepatocellular carcinoma ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT01696032](https://clinicaltrials.gov/ct2/show/study/NCT01696032)–active status–not recruiting patients, [NCT01896856](https://clinicaltrials.gov/ct2/show/study/NCT01896856)–recruiting patients, [NCT01261312](https://clinicaltrials.gov/ct2/show/study/NCT01261312)–active status–not recruiting patients, [NCT01752933](https://clinicaltrials.gov/ct2/show/study/NCT01752933)–completed in 2015–no results posted, and [NCT01966289](https://clinicaltrials.gov/ct2/show/study/NCT01966289)–recruiting patients).

**5.2.2. Azacytidine, Decitabine, and SGI-110 Syntheses.**—In 1964, Piskala *et al.* reported the first synthesis of azacytidine **11** (Scheme 20).<sup>295</sup> Chlorination of the 1'-position of tetraacetate riboside **136** followed by treatment with silver isocyanate produced 1-glycosyl isocyanate **137**. Conversion of **137** to isobiuret **138** using 2-methylisourea was followed by cyclization with triethyl orthoformate and deprotection with methanolic ammonia to give azacytidine **11**.

In 2006, Ionescu and Blumbergs reported a large-scale synthesis of azacytidine (**11**, Scheme 21).<sup>296</sup> Cyanoguanidine **139** was hydrolyzed and subsequently cyclized to produce the highly water-sensitive triazine **141**. The glycosylation reaction between silylated triazine **142** and 1,2,3,5-tetra-*O*-acetyl- $\beta$ -D-ribofuranose was performed in cold dichloromethane, with CF<sub>3</sub>SO<sub>3</sub>H added at 0 °C, followed by reaction warming to room temperature. Cooled solutions were used during the quenching/extraction step, which also minimized decomposition of the triazine base. Deprotection with a sodium methoxide/methanol solution, followed by recrystallization from DMSO/MeOH afforded azacytidine **11** on kilogram-scale. The last three steps, from glycosylation to recrystallization, were also performed in one pot starting with 5 g of the triazine, producing azacytidine **11** in 41% yield.

An additional process synthesis of azacytidine **11** using triflic acid in the glycosylation was reported.<sup>297</sup> Detailed empirical studies of the reaction found that a 1:1.2:0.95 ratio (silylated triazine **142**:triflic acid:1,2,3,5-tetra-*O*-acetyl- $\beta$ -D-ribofuranose) produced the highest yield. Exposure times to water or acid during the workup were minimized in order to protect the triazine base of intermediate **143**. Further empirical studies found optimized conditions for acetyl group deprotection of **143**, which involved *n*-butylamine in refluxing methanol. Recrystallization of the crude azacytidine **11** from DMSO/toluene gave the pure product in 99.9% purity on a kilogram-scale.

In 1964, the first synthesis of decitabine **14** was reported, which used a multistep sequence similar to the preparation of azacytidine **11** shown in Scheme 20. However, no yield was reported, as the  $\alpha/\beta$ -anomers were not separated.<sup>298</sup> Several years later, Winkley and Robins reported a synthesis of pure  $\beta$  decitabine (**14**, Scheme 22).<sup>299</sup> Triacetyl protected sugar **144** was converted to 3',5'-di-*O*-acetyl-2'-deoxy-D-ribofuranosyl chloride and then treated with the silylated 5-azacytosine **142**, forming **145** as a mixture of diastereomers. Deprotection with ethanolic ammonia followed by  $\alpha/\beta$  anomer separation, involving fractional crystallization followed by preparative layer chromatography on silica gel, gave decitabine



**14** in 7% yield. Unfortunately, the undesired  $\alpha$ -anomer was the predominant product, forming in 52% yield.

Two difficulties for large-scale synthesis of decitabine **14** are the lack of stability of the triazine base and the absence of anomeric selectivity during the glycosylation step. Indeed, when conjugated to a carbohydrate, 5-azacytosine is sensitive to hydrolytic decomposition under acidic, basic, and neutral conditions.<sup>300</sup> Furthermore, coupling of the silylated triazine with a 2'-deoxyribosugar generally leads to a 1:1 mixture of  $\alpha/\beta$  anomers.

Henschke *et al.* recently performed detailed studies under various conditions of the glycosylation step in attempts to improve anomeric selectivity (Scheme 23).<sup>301</sup> Solvent choice, reaction temperature, and ratio of the silylated triazine to the protected ribosugar were all found to affect anomeric selectivity, producing glycosylation products ranging from an anomeric ratio of 1:1 to 1:3 ( $\alpha:\beta$ ). Optimal large-scale conditions included a 1:1 molar ratio of the protected sugar to silylated triazine **142** with the addition of 1.05 equiv of TMSOTf in cold DCM. Kilograms of protected decitabine **147** were produced in a 1:2.7  $\alpha/\beta$  ratio in up to 55% yield. Furthermore, it was discovered that immediate quenching of the reaction with an organic amine (usually a primary amine) was needed in order to preserve the  $\alpha/\beta$ -ratio. Since they had observed isomerization of the  $\beta$ -anomer under different conditions, the authors hypothesized that prolonged reaction standing led to the resultant decrease of the  $\beta$ -anomer. Moreover, the decrease in the resultant was a result of its anomerization to the  $\alpha$ -product and not due to decomposition of the triazine base of protected decitabine. The synthesis of decitabine **14** was concluded by deprotection of a finely milled **147**; high quality of this intermediate was required for reaction success, in methanol with or without sodium methoxide. Subsequent filtration and recrystallization from DMSO and methanol afforded kilograms of API-grade decitabine **14**.

Kolla *et al.* have recently reported a process synthesis of decitabine (**14**, Scheme 24).<sup>302</sup> Bis-acetyl protected sugar **149**, formed by treating 2'-deoxy- $\alpha/\beta$ -D-ribose **100** with acetyl chloride in methanol followed by acetyl chloride in pyridine/DCM, was condensed with silylated 5-azacytosine in the presence of TMSOTf, giving a 1:1  $\alpha/\beta$  mixture of acylated **150**. Deprotection with methanolic ammonia treatment followed by crystallization afforded decitabine **14** in an HPLC purity of 90–99%. The resulting solid was dissolved in DMSO, filtered, washed with MeOH/EtOAc, and suction dried to produce crystalline decitabine **14** at 99.8% purity on a gram-scale.

Recently, Redkar reported an approach to SGI-110 **22** which increased the scale of compound production from 500 mg to 1 kg (Scheme 25).<sup>303</sup> Decitabine **14** was selectively protected at the  $N^4$ - and 5'-positions by treatment with DMF dimethyl acetal and then vinyl acetate in the presence of Lipozyme RM IM. Conversion to the 3'-phosphoramidite **151** was followed by coupling with 3', $N^2$ -protected guanosine **152**, and subsequent oxidation by *tert*-butyl hydrogen peroxide furnished dinucleotide **154**. Deprotection in methanolic ammonia followed by hydrolysis produced SGI-110 **22**, which precipitated as a sodium salt.



### 5.3. Ribosugar-Modified Cytidine Analogs

#### 5.3.1. Gemcitabine and Gemcitabine Prodrugs Biologies.

**5.3.1.1. Gemcitabine Biology:** Hertel and colleagues at Eli Lilly originally synthesized gemcitabine (2',2'-difluoro-2'-deoxycytidine) hydrochloride **9** as an antiviral nucleoside analog agent.<sup>304</sup> However, it was also discovered to be very effective at inhibiting cancer proliferation in cell culture.<sup>304</sup> Gemcitabine was originally screened against hematological cancers, but was also surprisingly shown to have activity in solid tumors. The agent is the only FDA-approved cytidine analog to be effective in solid tumors.<sup>305,306</sup>

Gemcitabine is imported into the cell by hENT-1,<sup>307</sup> and to a lesser extent by hCNT-1 and hCNT-3.<sup>308-310</sup> The agent is initially phosphorylated by 2'-deoxycytidine kinase, which is the rate-limiting step in its activation. Subsequent phosphorylations lead to the active metabolites: gemcitabine-5'-triphosphate and gemcitabine-5'-diphosphate. Gemcitabine-5'-triphosphate inhibits DNA synthesis by being incorporated into the newly synthesized DNA strand. Once incorporated into DNA, one additional nucleotide is inserted past gemcitabine-5'-monophosphate before chain elongation is inhibited.<sup>311,312</sup> This "masked chain termination" cannot be detected by exonucleases and eventually causes cell cycle arrest primarily in the S phase, resulting in apoptosis.<sup>313</sup> Alternatively, gemcitabine-5'-triphosphate can also be incorporated into DNA without chain elongation termination,<sup>17,314</sup> and is recognized by a DNA-dependent protein kinase/p53 protein complex, which coincides with cellular apoptosis.<sup>27</sup> Gemcitabine was also shown to have anticancer activity by being incorporated into RNA.<sup>315</sup> Furthermore, synergistic interaction of gemcitabine and sphingomyelin can induce ceramide-mediated apoptosis through Fas-induced cytotoxicity.<sup>316</sup>

Gemcitabine has self-potential mechanisms.<sup>317,318</sup> Gemcitabine-5'-diphosphate is a major active metabolite that inhibits ribonucleotide reductase, leading to one self-potential mechanism. Blocking ribonucleotide reductase activity decreases the natural deoxyribonucleotide pools, promoting more gemcitabine-5'-triphosphate to be incorporated into genomic DNA by DNA polymerase.<sup>31,319,320</sup> Furthermore, inhibition of ribonucleotide reductase leads to an indirect self-potential mechanism involving the regulation of 2'-deoxycytidine kinase. The enzyme is inhibited by a high dCTP concentration by a negative feedback loop regulation, and gemcitabine-induced ribonucleotide reductase inhibition decreases the dCTP and dATP concentrations.<sup>321</sup> This in turn leads to an increase in the activity of 2'-deoxycytidine kinase and more gemcitabine-5'-monophosphate accumulation in the cell. Gemcitabine-5'-triphosphate induces another direct potential mechanism involving the inhibition of deoxycytidylate, an enzyme that deaminates deoxycytidine-5'-monophosphate as well as gemcitabine-5'-monophosphate, preventing deamination of gemcitabine-5'-monophosphate into 2',2'-difluoro-2'-deoxyuridine-5'-monophosphate, an inactive metabolite.<sup>322</sup>

There are several mechanisms of resistance for gemcitabine reported from tissue culture studies using cell lines, and some (but not all) of the proteins and pathways being discovered include the following. The deaminated byproduct of gemcitabine, 2',2'-difluoro-5'-deoxyuridine, is a competitive inhibitor for its uptake by the hENT transporter,<sup>323</sup> and

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mutations in nucleotide transporters and 2'-deoxycytidine kinase prevent the uptake and initial phosphorylation of gemcitabine in the cell.<sup>60,324</sup> Intracellularly, gemcitabine-5'-monophosphate can be deaminated to form 2',2'-difluoro-2'-deoxycytidine-5'-monophosphate, and high gene expression of NT5C3 has been correlated with lower levels of phosphorylated gemcitabine metabolites.<sup>325</sup> Furthermore, upregulation of ABCC5 and hENT1 mRNA or protein was detected in gemcitabine resistance cell lines but has not been directly linked to promoting drug resistance in patients.<sup>326</sup> Additional studies have shown that histone methyltransferase G9a,<sup>327</sup> p38MAPK,<sup>328</sup> and Nutlin-3<sup>329</sup> might be correlated with gemcitabine resistance. Furthermore, gemcitabine resistance can arise from changes in high expression of drug efflux pumps and in nucleotide metabolism enzymes, inactivation of the apoptosis pathway, changes in miRNA expression, and modulation of Hedgehog, Wnt, and Notch pathways.<sup>330</sup>

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In patients, gemcitabine deamination appears to be a major mechanism limiting drug efficacy. Gemcitabine can be deaminated at the *N*<sup>4</sup> position of the cytidine base in the serum to the inactive 2',2'-difluoro-5'-deoxyuridine metabolite by cytidine deaminase,<sup>331</sup> or gemcitabine-5'-monophosphate can be deaminated by deoxycytidine deaminase in the cell. This metabolism is responsible for the very short half-life of the drug, thus limiting availability at the tumor.<sup>332</sup> Changes in ribonucleotide reductase subunit protein levels or polymorphisms, which regulate dNTP cellular concentrations, have been demonstrated.<sup>333–335</sup> Furthermore, collective cell line data suggest that modulation of mRNA expression, cell survival pathways, and tumor suppressor genes all might contribute to gemcitabine resistance. These mechanisms for gemcitabine resistance will need to be validated in the future using *ex vivo* models to determine their importance in patient populations. Since gemcitabine resistance is polygenic, combination treatments with alkylating agents, topoisomerase inhibitors, mitotic inhibitors, and biologicals (such as anti-PD-1 and anti-CTLA-4) appear to be a logical choice for new investigatory treatments.

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Gemcitabine hydrochloride **9** is currently being used to treat non-small cell lung, pancreatic, breast, bladder, and ovarian cancers.<sup>336–338</sup> The agent is also being used to treat pancreatic cancer; however, treatment with gemcitabine hydrochloride **9** monotherapy showed only a modest benefit.<sup>339,340</sup> Currently, various clinical trials studying using gemcitabine hydrochloride **9** in combination therapy are underway. Some of these trials include: gemcitabine-cisplatin (DNA alkylating agent) combination therapy for advanced non-small-cell lung cancer,<sup>341</sup> Hodgkin's lymphoma, and non-Hodgkin's lymphoma,<sup>342</sup> gemcitabine-erlotinib (tyrosine kinase inhibitor) regimen to treat locally advanced and metastatic pancreatic cancers,<sup>343</sup> and gemcitabine-oxaliplatin (DNA alkylating agent) treatment for biliary adenocarcinoma,<sup>344</sup> pancreatic cancer,<sup>345</sup> hepatocellular cancer,<sup>346</sup> testicular cancer,<sup>347</sup> and epithelial ovarian cancer.<sup>348</sup> Additional combination treatments of gemcitabine with necitumumab (a monoclonal antibody which binds the epidermal growth factor receptor) and radiation treatments are under investigation ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT01606748](https://clinicaltrials.gov/ct2/show/study/NCT01606748)–active status–not recruiting patients and [NCT02254681](https://clinicaltrials.gov/ct2/show/study/NCT02254681)–recruiting patients).

**5.3.1.2. LY2334737 Biology.:** As an *N*<sup>4</sup>-valproic acid prodrug, LY2334737 **52** was designed to improve the oral bioavailability of gemcitabine hydrochloride **9**, primarily by blocking the site of deamination and decreasing first-pass metabolism.<sup>349,350</sup> The prodrug

passes through the intestinal epithelium and enters the systemic circulation largely intact,<sup>350</sup> thereby reducing gut exposure to gemcitabine. LY2334737 **52** is hydrolyzed by slow systemic cleavage via the actions of carboxyesterase 2, liberating gemcitabine and valproic acid, and leads to prolonged gemcitabine exposure *in vitro*.<sup>351</sup> Preclinical studies showed that low-dosing of LY2334737 **52** was efficient for human colon and lung tumor xenograft models.<sup>352</sup> Moreover, elevated carboxyesterase 2 activity and ENT1 expression may enhance LY2334737 tumor response.

In a clinical study to determine the maximum tolerated dose and dose-limiting toxicities of LY2334737 **52**, stable disease was achieved in 22 of the 65 patients with advanced solid tumors.<sup>353</sup> In 2012, a Phase I clinical trial was conducted in persons with solid and metastatic tumors ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01648764) identifier: [NCT01648764](https://clinicaltrials.gov/ct2/show/study/NCT01648764)—completed in 2014—no study results were posted). However, in 2013 Eli Lilly discontinued the development of LY2334737 **52**,<sup>354</sup> due to results from a Japanese study that showed hepatic toxicities in patients.<sup>355</sup>

**5.3.1.3. CP-4126 Biology.:** CP-4126 (**49**, CO-1.01) is an elaidic acid ester derivative of gemcitabine developed by Clavis Pharma. The lipophilic side chain was installed to increase the ability of gemcitabine to diffuse through the plasma membrane. This modification makes CP-4126 **49** a transporter-independent analog as compared to gemcitabine, which enters the cell primarily by hENT1.<sup>356</sup> Once in the cell, cleavage to gemcitabine is believed to occur via the action of carboxyesterases, and subsequent phosphorylation to gemcibine-5'-monophosphate is performed by deoxycytidine kinase.<sup>357</sup> Furthermore, resistance studies showed that CP-4126 **49** and gemcitabine can arise by down regulation of 2'-deoxycytidine kinase *in vitro*.<sup>324</sup>

CP-4126 **49** has been assessed in several Phase I/II clinical trials for safety and efficacy as monotherapy or combination therapy,<sup>357</sup> involving non-small-cell lung cancer, metastatic pancreatic adenocarcinoma, and other advanced solid tumors. The results demonstrated no marked difference in overall survivals of individuals treated with CP-4126 **49** *versus* gemcitabine hydrochloride **9** having pancreatic ductal adenocarcinoma and a low hENT1 protein expression level.<sup>358</sup> Two Phase I clinical trials have been completed ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01392976) identifiers: [NCT01392976](https://clinicaltrials.gov/ct2/show/study/NCT01392976)—completed in 2013—results were published). For one clinical trial, CP-4126 **49** was poorly absorbed and was subject to metabolism before systemic exposure. Overall, this led to further testing of CP-4126 **49** in patients by intravenous administration.<sup>359</sup> Another Phase I clinical trial had CP-4126 **49** given intravenously and reported positive results ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00778128) identifiers: [NCT00778128](https://clinicaltrials.gov/ct2/show/study/NCT00778128)—completed in 2010—results were published).<sup>360</sup> Another Phase I trial testing CP-4126 **49** and cisplatin was stopped ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01641575) identifiers: [NCT01641575](https://clinicaltrials.gov/ct2/show/study/NCT01641575)—terminated in 2013—no results posted). As a result of these studies, Clavis suspended development of CP-4126 **49**.

We have not covered all the potential prodrug combinations or technologies under development for gemcitabine. Other unique modifications of the agent include mesoporous silica nanoparticles,<sup>361</sup> amino acid monoester prodrugs,<sup>362</sup> Hoechst conjugates,<sup>363</sup> and further base modifications.<sup>364</sup>

**5.3.1.4. NUC-1031 Biology.:** The ProTide concept has been widely used in the development of antiviral and anticancer nucleoside-5'-monophosphate prodrugs.<sup>104,365</sup> Protides were designed as protected nucleotide-5'-monophosphate isomers to increase cellular penetration as well as bypass the often rate-limiting monophosphorylation of the nucleoside analog.<sup>104</sup> Furthermore, they are also believed to improve the delivery of the parent compound to the liver.<sup>366</sup> NUC-1031 (Acelarin, **32**) is a 5'-phosphoramidate prodrug of gemcitabine.<sup>367</sup> This agent is able to overcome gemcitabine resistance mechanisms: decreased uptake by cellular transporters, decreased activity of 2'-deoxycytidine kinase (by mutation), and cytosine deamination.<sup>368</sup> However, it would still be sensitive to gemcitabine-5'-monophosphate deamination by deoxycytidine deaminase. Like gemcitabine, NUC-1031 **32**, is administered by intravenous injection.

A Phase I study in people with advanced solid tumors has been launched in 2012.<sup>368,369</sup> The agent could be given at four times the maximum dose of gemcitabine, and over half the patients achieved stable disease, with a few patients having a reduction in tumor size. In addition, administration of NUC-1031 **32** produced a 13-fold increase in intracellular gemcitabine concentration as compared to gemcitabine intravenous injection. An additional study funded by NuCana BioMed Limited examining combination treatment of NUC-1031 **32** with carboplatin (DNA alkylating agent) for ovarian cancer is planned ([ClinicalTrials.gov](http://ClinicalTrials.gov) identifiers: [NCT01621854](https://clinicaltrials.gov/ct2/show/study/NCT01621854)—completed June 2015—no results posted, and [NCT02303912](https://clinicaltrials.gov/ct2/show/study/NCT02303912)—recruiting patients).

### 5.3.2. Gemcitabine and Gemcitabine Prodrugs Syntheses.

**5.3.2.1. Gemcitabine Syntheses.:** Hertel and co-workers reported the discovery synthesis of gemcitabine hydrochloride **9** (Scheme 26).<sup>370</sup> It commenced with a Reformatsky reaction involving ethyl bromodifluoroacetate and aldehyde **155**, which is easily prepared by isopropylidene protection and subsequent oxidative cleavage of D-mannitol, forming **156** as a mixture of diastereomers (3:1 *anti/syn*) which were separated by HPLC. Acid-mediated deprotection and cyclization were followed by 3',5'-O-silylation, affording the protected lactone **157**. Reduction, mesylation, and glycosylation under Vorbrüggen conditions produced protected sugar **158** as a 4:1 mixture of  $\alpha/\beta$  anomers. Deprotection allowed for reverse phase HPLC separation of the anomers. The unfortunate  $\alpha/\beta$  anomeric ratio produced under these conditions gave gemcitabine hydrochloride **9** in only 10% yield from intermediate **158**.

In 2010, Chang *et al.* reported a large-scale synthesis of gemcitabine **163** using 2,2-fluoro- $\alpha$ -ribofuranosyl bromide **162** as a key intermediate to increase the  $\beta$ -anomeric selectivity in the glycosylation reaction (Scheme 27).<sup>371</sup> The stereoisomers of **156** were protected with *p*-phenylbenzyl chloride and subjected to ester hydrolysis under basic conditions. Fortunately, the reaction volume reduction led to selective precipitation of the desired potassium *erythro*-pentonate **159**, and filtration furnished the pure diastereomer, representing a marked increase in efficiency when compared to the laborious diastereomeric separations of other syntheses (e.g., HPLC purification in the Hertel method). Deprotection, cyclization, and 5'-O-benzoylation of intermediate **159** were followed by lactone reduction and phosphorylation of the crude anomeric mixture, directly furnishing intermediate **161** (1:10.8  $\alpha/\beta$  ratio).

Subsequent bromination led to a 10.8:1  $\alpha/\beta$  mixture, and recrystallization from isopropanol afforded the pure  $\alpha$ -anomer **162**, with the phenylbenzoyl protecting group being necessary for recrystallization.

In order to maximize the  $S_N2$ -displacement of  $\alpha$ -bromide **162**, the glycosylation conditions were optimized. Therefore, non-polar solvents were used, and *in situ* generated TMSBr was removed via continuous distillation, in an attempt to decrease anomerization of **162** or formation of the oxocarbenium ion. Indeed, this led to an improved  $\beta/\alpha$  anomeric mixture of 5.5:1. Deprotection with methanolic ammonia, followed by isolation of the pure  $\beta$ -diastereomer by simple extraction, evaporation, and crystallization from water, produced the hemi- or dihydrate of gemcitabine, depending on workup conditions.

Jiang *et al.* also recently reported a process preparation for gemcitabine hydrochloride **9** on a 100-g-scale (Scheme 28).<sup>372</sup> A diastereomeric mixture of difluoro lactone **164** was acylated with cinnamoyl chloride, and when the solution was cooled, stereospecific crystallization afforded the desired lactone **165**. Reduction and tosylation gave a 1:1 mixture of anomeric tosylates **166**, which was found to crystallize in ethyl acetate/petroleum ether. Subsequent glycosylation of the anomeric mixture **166** with *N*-acetyl cytosine in the presence of TMSOTf, followed by direct treatment of the reaction mixture with sodium carbonate, allowed for the precipitation of the undesired  $\alpha$ -nucleoside analogue. Deprotection of  $\beta$ -isomer **167** with methanolic ammonia at room temperature gave crude gemcitabine **163**, which was acidified with HCl and crystallized in acetone/H<sub>2</sub>O, affording the target salt in 99.9% purity.

**5.3.2.2. LY2334737, CP-4126, and NUC-1301 Syntheses.:** The gemcitabine prodrug LY2334737 **52** was formed by directly coupling valproic acid with gemcitabine hydrochloride **9**, using peptide coupling conditions (EDC/HOBt/NMM).<sup>350</sup> Prodrug LY2334737 **52** is not crystalline; therefore, it was purified as a *p*-toluenesulfonic acid (pTSA) salt co-crystal (Scheme 28).

CP-4126 **49** was formed by direct esterification of the 5'-OH of gemcitabine **163** using elaidic acid chloride as reagent (Scheme 29).<sup>373</sup> NUC-1301 **32** was prepared by reaction of phosphochloridate **169**, which was produced by treating L-alanine benzyl ester hydrochloric salt **168** with phenyl-dichlorophosphate **92**, with gemcitabine **163** in THF/pyridine in the presence of *N*-methyl imidazole.<sup>374</sup>

An improved synthesis for NUC-1031 **32** has recently been reported, which affords the agent on a 20-gram-scale.<sup>375</sup> Boc protection of the 3'-OH and *N*<sup>4</sup>-amino group of gemcitabine **163** was followed by reaction with phenyl-dichlorophosphate **92** and L-alanine benzyl ester hydrochloride **168**, which installed the protide. Subsequent Boc deprotection completed the synthesis.

**5.3.3. Sapacitabine Biology.**—In the early 1990s, Matsuda *et al.* published the synthesis and anticancer activity of 2'-*C*-cyano-2'-deoxy-1- $\beta$ -D-*arabino*-pentofuranosylcytosine (CNDAC) along with the rationale for the structure of the agent.<sup>376</sup> Since a cyanoethyl group beta to a phosphate diester of a nucleoside was reported to undergo

$\beta$ -elimination under alkaline conditions,<sup>377</sup> it was hypothesized that a 2'- $\beta$ -C-cyano nucleoside when incorporated into DNA could lead to  $\beta$ -elimination of the 3'-phosphodiester moiety and produce DNA strand breaks (Figure 14). This mechanism of anticancer activity would be similar to that of radiation therapy, which was shown to produce similar DNA strand breaks that were believed to result in tumor cell death.<sup>378,379</sup>

Sapacitabine **25** (Cyclacel Pharmaceuticals) is an  $N^4$ -palmitoyl derivatized analog of CNDAC and is orally bioavailable. It is primarily metabolized in the plasma, gut, and/or liver to CNDAC, and eventually transported into cells.<sup>380</sup> CNDAC is phosphorylated by 2'-deoxycytidine kinase and further phosphorylated by other cellular kinases to generate the triphosphate form.<sup>381</sup> DNA polymerases incorporate the CNDAC-5'-triphosphate into the growing DNA strands, and further elongation of the chain is sluggish due to the steric effects of CNDAC analog.<sup>26</sup> Upon incorporation of CNDAC-5'-monophosphate into DNA,  $\beta$ -elimination of the 3'-phosphodiester moiety can result in a severed DNA strand that is terminated with 2'-C-cyano-2',3'-didehydro-2',3'-dideoxycytidine-5'-monophosphate (Figure 14).<sup>382</sup> This represents a unique anticancer mechanism of a nucleoside derivative, and results in cell cycle arrest at the G<sub>2</sub> phase, which is different from the majority of other anticancer nucleoside analogs that cause arrest in the S phase.<sup>383</sup>

Sapacitabine **25** is a weak substrate for cytidine deaminase, whereas CNDAC is a substrate and is converted into the inactive uracil derivative: 2'-C-cyano-2'-deoxy-1- $\beta$ -D-*arabino*-(pentofuranosyl)uridine. In addition, unlike other nucleoside analogs in their corresponding di- or triphosphates forms, CNDAC-5'-diphosphate does not effectively inhibit ribonucleotide reductase, and CNDAC-5'-triphosphate does not act as a feedback inhibitor of 2'-deoxycytidine kinase.<sup>26</sup>

Cells deficient in nucleotide excision repair (NER) of endonuclease XPF are 4- to 5-fold more sensitive to CNDAC exposure,<sup>384</sup> suggesting that the NER pathway is used to excise the agent from the DNA strand. If DNA single-strand breaks caused by CNDAC exposure are not repaired, they become more lethal DNA double-stranded breaks during the following round of DNA replication.<sup>385</sup> These CNDAC-induced DNA double-stranded breaks can occur when the replication fork encounters a DNA single-stranded break or after collapse of a stalled replication fork,<sup>386</sup> which are repaired primarily by homologous recombination. Cells with defects or mutations in the homologous recombination pathway (ATM, RAD51, Brac2, Xrcc3) are up to 100-fold more sensitive to CNDAC exposure.<sup>387</sup> In addition to DNA incorporation, CNDAC exposure was also shown to inhibit RNA synthesis to a limited extent *in vitro*.<sup>26</sup>

Sapacitabine **25** is in several clinical trials (Phase I–III) either as a monotherapy or in combination with other drugs for treatment of acute myelogenous leukemia, chronic lymphocytic leukemia, non-small cell lung cancer, and some solid tumors.<sup>349,387,388</sup> However, it failed to achieve end point efficacy in a Phase III study using oral sapacitabine **25** in elderly patients with newly diagnosed acute myelogenous leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01303796) identifier: NCT01303796–unknown, >2 years since reported). In 2013, Cyclacel Pharmaceuticals announced that sapacitabine **25** had activity against 75% of primary ovarian cancer samples isolated from individuals. DFP-10917 (**26**, intravenously administered



hydrochloride salt form of CNDAC) is in Phase I/II studies to treat relapsed or refractory acute leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01702155) identifier: [NCT01702155](https://clinicaltrials.gov/ct2/show/study/NCT01702155)—recruiting patients).

**5.3.4. Sapacitabine Synthesis.**—The discovery synthesis of CNDAC is shown in Scheme 30.<sup>376,389</sup> *N*<sup>4</sup>-Acylation of cytidine **96** followed by 3',5'-TIPS protection and 2'-OH oxidation provided ketosugar **170**. Treatment of **170** with sodium cyanide followed by reaction with phenyl chlorothionocarbonate produced an epimeric mixture of 2'-cyanohydrins **171**. Subsequent deoxygenation under standard Barton conditions produced **172** diastereoselectively, presumably due to steric hindrance on the  $\beta$ -face. Silyl deprotection followed by *N*<sup>4</sup>-acetyl removal with HCl in MeOH afforded DFP-10917 (**26**, the hydrochloride salt of CNDAC). Interestingly, attempted deacetylation of **172** with NH<sub>3</sub>/MeOH produced mainly cytosine, adding further evidence for the increased acidity of the 2'-*α* of **26**.

Soon after the discovery synthesis of CNDAC hydrochloride (DFP-10917, **26**), a process synthesis for sapacitabine **25** was reported.<sup>390</sup> The *N*<sup>4</sup>-palmitoyl group was installed in the first step, and the rest of the synthesis followed a route similar to that of Scheme 30, with a few variations in reagent choice (i.e., pyridinium dichromate for the 2'-oxidation) producing sapacitabine **25** in nine steps in 20% overall yield.

More recently, Westwood *et al.* reported a kilogram preparation of sapacitabine **25**.<sup>391,392</sup> The synthesis followed a similar route as the two above-mentioned preparations with some variation in reagent use and synthetic order. Deoxygenation was performed using the economical lauroyl peroxide with collidine on a substrate similar to intermediate **171**, except with a less-bulky ethyl xanthate ester. Additionally, 3',5'-OH silyl protection came before *N*<sup>4</sup>-acylation, generating a solid intermediate that was easy to purify and use in subsequent steps. The final step of the preparation involved *N*<sup>4</sup>-acylation with palmitic anhydride, producing sapacitabine **25** on a kilogram-scale in nine steps and 37% overall yield.

**5.3.5. TAS-106 Biology.**—Many antineoplastic agents exert their cytotoxic effects during the S phase (DNA synthesis) of the cell cycle. This poses a problem for more slow growing solid tumors, which have decreased DNA replication rates as compared to leukemias, making the incorporation of 2'-deoxynucleoside-5'-triphosphate analogs very inefficient. However, RNA synthesis occurs throughout the cell cycle except in the M phase. Therefore, a drug that targets both DNA and RNA was hypothesized to offer a broader interval in the cell cycle to exert anticancer effects.<sup>26</sup>

Drawing on the observation that CNDAC-5'-triphosphate (see Section 5.3.3) was shown to have some RNA synthesis inhibition, TAS-106 **46** (ECyd; [1-(3-C-ethynyl- $\beta$ -D-ribofuranosyl)cytosine]) was developed by Taiho Pharmaceuticals in hopes that the TAS-106-5'-triphosphate analog would inhibit RNA polymerase.<sup>382,393</sup> Indeed, the main mechanism of action of TAS-106 **46** was shown to be inhibition of RNA polymerases I, II, and III, which essentially blocks RNA synthesis to cause apoptosis.<sup>394</sup> In addition, TAS-106 **46** was shown to inhibit DNA repair proteins (BRAC2 and Rad51) in the homologous repair pathway, suggesting it might also act via some of the mechanisms of sapacitabine **25**.<sup>395</sup>



TAS-106 **46** is phosphorylated to the monophosphate form by uridine/cytidine kinase, which is preferentially expressed in tumor cells versus normal cells.<sup>396</sup> It is subsequently converted to the TAS-106-5'-diphosphate and TAS-106-5'-triphosphate forms by uridine/cytidine monophosphate kinase and nucleoside diphosphate kinase, respectively.<sup>397</sup> TAS-106-5'-triphosphate remains in cells for a prolonged period of time even after short-term exposure to TAS-106 **46**, producing significant exposure of cells to the active form of the drug *in vitro*.<sup>398</sup> TAS-106 **46** is a poor substrate for cytidine deaminase, although the deaminated uridine analog of the agent also exhibited potent cytotoxicity *in vitro*.<sup>399,400</sup> TAS-106-5'-diphosphate was anticipated to also block activity of the enzyme due to the precedent that other 2'-deoxyribonucleoside anticancer agents are ribonucleotide reductase inhibitors. However, contrary to the original hypothesis, TAS-106-5'-diphosphate was not found to inhibit ribonucleotide reductase. Additional investigational studies have examined duplex drug linking of 5FdU(5'-5')TAS-106 against gastric adenocarcinoma cell lines<sup>401</sup> and 5-FdU(3'-5')TAS-106 exposure to hepatoblastoma cells lines.<sup>402</sup> Both studies showed positive cytotoxic results *in vitro*, but further studies are needed to examine the full potential of using these duplex analogs.

TAS-106 **46** therapy has produced modest results in clinical trials. In a Phase II clinical trial for salvage metastatic or recurrent head and neck squamous cell cancers and nasopharyngeal cancer, TAS-106 **46** treatment was reasonably well tolerated for platinum-failure patients, but had myelosuppression as its main toxicity.<sup>403</sup> However, the agent showed no anticancer efficacy ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00737360) identifier: NCT00737360—terminated in 2012—limited number of patients). TAS-106 **46** treatment in combination with carboplatin, a DNA alkylating agent, recently underwent a Phase I dose-escalating study in subjects with solid tumors.<sup>404</sup> The combination was well tolerated, and a few individuals experienced stabilized disease, in which the cancer did not progress (>4 months). There are no current clinical trials underway for TAS-106 **46**.

**5.3.6. TAS-106 Synthesis.**—The discovery synthesis of TAS-106 **46** is shown in Scheme 31.<sup>399</sup> The chosen starting material was 1,2-*O*-isopropylidene- $\alpha$ -D-xylofuranose **173**, which was 5'-OH protected and then 3'-oxidized, forming intermediate **174**.<sup>405</sup> Stereoselective addition of lithiated (trimethyl)acetylene to **174** gave the tertiary alcohol **175** as the desired  $\beta$ -adduct. Subsequent desilylation, selective 5'-OH benzylation, isopropylidene deprotection, and 2'-OH benzylation generated **178**. Glycosylation of **178** with cytosine under Vorbrüggen conditions followed by deprotection furnished the desired TAS-106 **46** in good overall yield.

In preparation for clinical trials, a large-scale synthesis of TAS-106 **46** was developed (Scheme 32).<sup>406</sup> Most of the intermediates were crystalline and required no chromatography. Selective 5'-OH protection of 1,2-*O*-isopropylidene-D-xylofuranose **173** with *p*-chlorobenzoyl chloride was followed by TEMPO-mediated oxidation of the 3'-OH group. Stereoselective introduction of a TMS-ethynyl group followed by protecting group manipulation set the stage for the key glycosylation reaction. Multiple leaving groups at the anomeric position were screened, and the isobutyryloxy leaving group proved to be most effective, affording the protected intermediate **184** in good yield and stereoselectivity. Final

deprotection and crystallization from aqueous methanol furnished 128 g of crystallized TAS-106 **46**.

**5.3.7. Tezacitabine Biology.**—Tezacitabine **45** [MDL 101,731; (*E*)-2'-deoxy-2'-(fluoromethylene)cytidine] was designed by Hoechst Marion Roussel Inc. as a mechanism-based irreversible ribonucleotide reductase inhibitor.<sup>407</sup> Such inhibitors are modified at the 2'-position, and often form covalent complexes with the enzyme. Tezacitabine **45** employs a 2'-fluorovinyl moiety, which was proposed to aid in the ribonucleotide reductase-catalyzed abstraction of the 3'-hydrogen atom from tezacitabine-5'-diphosphate, by resonance stabilization of the resultant 3'-radical (Figure 15).<sup>408</sup> The design was successful, as tezacitabine-5'-diphosphate was indeed found to be an irreversible inhibitor of ribonucleotide reductase.<sup>409</sup> Furthermore, tezacitabine-5'-triphosphate was found to be a DNA polymerase chain terminator.<sup>410</sup>

Although a cytidine nucleoside analog, tezacitabine **45**, is relatively resistant to cytidine deaminase activity,<sup>411</sup> upon entering the cell, the agent is phosphorylated by 2'-deoxycytidine kinase and then by other endogenous kinases to the active metabolites: tezacitabine-5'-diphosphate and tezacitabine-5'-triphosphate. Tezacitabine-5'-triphosphate is a substrate for DNA polymerase  $\alpha$ , being incorporated into the growing DNA strand in the place of cytidine, and it prevents further chain elongation by DNA polymerases.<sup>412</sup>

In tissue culture tumor cell lines and mouse tumor models, tezacitabine **45** has been examined as a candidate for treatment of colorectal and hematological solid cancers, wherein hydroxyurea and cytarabine **4** have limited antitumor activities.<sup>413,414</sup> Furthermore, the radiosensitizing effects of tezacitabine **45** in the presence of zidovudine (a nucleoside analog inhibitor of reverse-transcriptase) were tested in the colon cancer cell line WiDr.<sup>415</sup> In addition, tezacitabine **45** treatment was also shown to have antiangiogenic activity in xenograft models.<sup>416</sup>

A phase I clinical study showed that tezacitabine **45** treatment caused myelotoxicity in 53% of the patient and 83% had febrile episodes.<sup>412</sup> An additional Phase I study showed that tezacitabine **45** plus 5-FU **2** treatment had activity in patients with advanced esophageal and other gastrointestinal carcinomas.<sup>417</sup> However, tezacitabine **45** treatment failed to have significant activity as a stand-alone therapy.<sup>417</sup> In 2004 Chiron reported that it had discontinued development of tezacitabine **45** after it failed to meet expectations in a Phase II trial ([ClinicalTrials.gov](https://clinicaltrials.gov) identifier: [NCT00051688](https://clinicaltrials.gov/ct2/show/study/NCT00051688)—terminated in 2004).

**5.3.8. Tezacitabine Synthesis.**—McCarthy *et al.* reported a synthesis of tezacitabine **45**,<sup>418</sup> which proceeded in seven steps from uridine in 10% overall yield, requiring five column chromatography purifications. Two years later, an improved process was published that used cytidine as starting material, allowing for the preparation of multigram quantities of the compound in five steps with a 29% overall yield (Scheme 33).<sup>419</sup> A one-pot protection of the 3'-OH and 5'-OH of cytidine **96** by treatment with TIPSCl and of the *N*<sup>4</sup>-amino group by the addition of *N,N*-dimethylformamide dimethyl acetal gave intermediate **185**. This derivative underwent swern oxidation to provide the keto intermediate **186**, which was worked up by a nonaqueous silica gel plug filtration in order to avoid ketone hydration.

A Horner–Wadsworth–Emmons reaction between **186** and the *in situ*-generated diethyl-1-fluoro-1-(phenylsulfonyl)-methylphosphonate produced a separable mixture of fluorovinyl sulfones (*Z/E* ratio 10:1, respectively). Deprotection of the *N*<sup>4</sup>-amino group with methanolic ammonia yielded the corresponding  $\alpha$ -fluorovinyl sulfone **187**, which readily crystallized from hexanes. Radical-mediated sulfone/stannane interchange provided **188** with retention of configuration, and treatment of the intermediate with cesium fluoride in refluxing methanol allowed for the simultaneous removal of the TIPS and tributyltin groups.

**5.3.9. Troxacitabine Biology.**—Troxacitabine **48**,  $\beta$ -L-dioxolane-cytidine (Troxaty, SGX Pharmaceuticals), was originally studied as an anti-HIV agent, and later it was shown to have potential as an anticancer agent.<sup>420</sup> Moreover, this agent is the first L-nucleoside studied for cancer.<sup>421</sup>

Troxacitabine **48** enters cancer cells by passive diffusion, possibly due to the lack of 3'-OH making it not a good substrate for facilitated transporters.<sup>422</sup> The agent is not a substrate for cytidine deaminase.<sup>423</sup> However, it is recognized by 2'-deoxycytidine kinase, suggesting more chiral specificity from the former enzyme and less from the latter.<sup>424</sup>

Troxacitabine-5'-monophosphate is further phosphorylated to the di- and triphosphate forms by cellular enzymes. In contrast to a majority of other anticancer nucleoside analogs, troxacitabine-5'-diphosphate is the predominant intracellular metabolite, and a large pool accumulates intracellularly.<sup>311,425</sup> Phosphoglycerate kinase converts troxacitabine-5'-diphosphate to troxacitabine-5'-triphosphate, the major intracellular active metabolite of troxacitabine **48**.<sup>426</sup>

Troxacitabine-5'-triphosphate is incorporated into newly synthesized DNA, being a substrate for DNA polymerases  $\alpha$ ,  $\beta$ ,  $\delta$ ,  $\gamma$ , and  $\epsilon$ , leading to polymerase inhibition by an absolute DNA chain termination mechanism.<sup>427,428</sup> The incorporated troxacitabine-5'-monophosphate is not readily excised from the DNA chain by proof reading 3'  $\rightarrow$  5' exonuclease activities associated with DNA polymerases, due to the chiral specificity of the enzymes. However, removal of the incorporated troxacitabine-5'-monophosphate from DNA can occur by the APE-1 excision repair mechanism.<sup>420</sup> Resistance to troxacitabine **48** exposure can arise by decreased expression of 2'-deoxycytidine kinase *in vitro*.<sup>429</sup>

Troxacitabine **48** initially showed promising anticancer activity against solid tumors and leukemias.<sup>424,427,430–432</sup> Tumor cells with catalytically inactive p53 have increased sensitivity to troxacitabine **48** exposure.<sup>429</sup> In addition, pancreatic cell lines were shown to have increased sensitivity to troxacitabine prodrugs bearing *N*<sup>4</sup> single carbon chain amides to increase drug lipophilicity.<sup>433</sup> However, no additional information is available addressing preclinical efficacy study models for this compound.

Unfortunately, despite the promising preclinical results, troxacitabine **48** therapy had limited success in clinical trials for leukemias and neoplasms (administered by intravenous infusion)<sup>434,435</sup> and ended in the termination of multiple trials ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT00129948](https://clinicaltrials.gov/ct2/show/study/NCT00129948)—terminated in 2006, [NCT00104468](https://clinicaltrials.gov/ct2/show/study/NCT00104468)—terminated in 2007, and [NCT00104286](https://clinicaltrials.gov/ct2/show/study/NCT00104286)—terminated in 2006). In 2007, results were reported for a Phase II study

evaluating troxacitabine **48** therapy in relapsed or refractory lymphoproliferative neoplasms or multiple myeloma.<sup>436</sup> All the patients had one adverse event, with 62% with one serious adverse event. Overall, troxacitabine **48** treatment had limited benefit in patients with these advanced disorders.<sup>436</sup> Furthermore, the agent also exhibited significant toxicity issues, indicating that further development of the agent is unlikely.

**5.3.10. Troxacitabine Synthesis.**—An initial diastereoselective synthesis of troxacitabine **48** used L-ascorbic acid as starting material (Scheme 34).<sup>437</sup> Acid **189** was condensed with benzyloxyacetaldehyde dimethyl acetal in the presence of tosylic acid, affording dioxolane **190** as a mixture of diastereomers. Oxidative degradation followed by benzyltriethylammonium chloride-catalyzed Wolfe oxidation produced a mixture of carboxylic acid products **192**, which could be separated with flash chromatography. Oxidative decarboxylation followed by glycosylation, separation of the  $\alpha$  and  $\beta$  isomers, and deprotection furnished troxacitabine **48**.

Kim *et al.* reported a gram-scale synthesis (Scheme 35)<sup>438</sup> starting from 2,3:5,6-di-*O*-isopropylidene-L-gulofuranose **194**, which is made by protecting and reducing L-gulono-6,3-lactone. Acid-mediated rearrangement of **194** to pyranose **195** followed by oxidative cleavage and reduction produced dioxolane **196**. Protecting group manipulation, oxidation, and Pb(OAc)<sub>4</sub>-mediated decarboxylation furnished intermediate **198**. Glycosylation of benzoyl-protected cytosine with **198** followed by deprotection with methanolic ammonia afforded troxacitabine **48**.

**5.3.11. Thiarabine Biology.**—Thiarabine **47** was first prepared in the 1970s and showed significant anticancer activity. However, difficulty in its synthetic preparation limited further research.<sup>439,440</sup> Two decades later, Southern Research Institute drew upon this finding as well as the report that 4-thionucleo-sides showed resistance to purine nucleoside phosphorylase cleavage<sup>441</sup> to continue the development of this next-generation anticancer nucleoside analog.<sup>442,443</sup>

Thiarabine **47** (T-ara C; 4'-thio-arabinofuranosylcytosine; OSI-7836) is phosphorylated to the monophosphate form via 2'-deoxycytidine kinase, although the phosphorylation rate of thiarabine **47** is 1% as compared to cytarabine **4**.<sup>444</sup> The thiarabine-5'-monophosphate exhibits reduced susceptibility to deaminase activity yet is still a good substrate for uridine/cytidine monophosphate kinase, leading to thiarabine-5'-diphosphate.<sup>445,446</sup>

Thiarabine-5'-diphosphate is further converted to the triphosphate form (the active metabolite of thiarabine **47**). Thiarabine-5'-triphosphate accumulates slowly in the cell but was shown to have long intracellular retention,<sup>447</sup> and yet was still efficiently degraded to the nucleoside when using cellular extracts.<sup>446</sup>

Thiarabine-5'-triphosphate is a substrate for the nuclear DNA polymerase and is incorporated into DNA, inhibiting DNA synthesis and generating DNA lesions,<sup>444,448</sup> with the primary mechanism of action being disruption of DNA polymerase activity.<sup>449</sup> In addition, HCT 116 colon carcinoma cells exposed to thiarabine **47** led to cleavage of caspase

3 and PARP that promoted cell death via apoptosis.<sup>447,450</sup> Thiarabine **47** was also reported to have antiangiogenic properties.<sup>442</sup>

Preclinical studies for thiarabine **47** were promising, having broad spectrum activities against human solid tumors and leukemia/lymphoma xenografts in mice.<sup>17,446</sup> In addition, cross-resistance in multidrug resistant cell lines did not occur for thiarabine **47**, suggesting that it potentially could be used in combination drug therapy.<sup>451,452</sup> Previously, a Phase I/II clinical trial showed positive results when using clofarabine **12** and cytarabine **4**, having a proposed mechanism of increased cytarabine-5'-triphosphate concentration in cells.<sup>453</sup> These results led to a combination therapy of clofarabine **12** and thiarabine **47** study, which showed synergistic activity of **12** and **47** leading to delayed tumor growth in mice.<sup>17</sup>

A few clinical trials have been reported for thiarabine **47**. In 2006, two Phase I trials were reported for thiarabine **47**.<sup>454,455</sup> Study results indicated excessive fatigue reversible grade-3 lymphopenia. Changes in the schedule and duration of thiarabine **47** did not improve its tolerability. Development of the agent has stalled, although some have called for continued clinical evaluation of the agent for hematological and/or solid tumors.<sup>449</sup>

**5.3.12. Thiarabine Synthesis.**—The first synthesis of thiarabine **47** was accomplished by Whistler *et al.* (Scheme 36)<sup>439,456–459</sup> through a long and arduous sequence. The protected glucanofuranose **199**, made from glucose in two steps (isopropylidene installation followed by benzylation), was transformed into the protected tosyl sugar **200** in six steps after protecting group manipulation and tosylation. Subsequent treatment of **200** with sodium methoxide, sodium benzyloxide, and tosyl chloride furnished **201**, which was further converted to intermediate **202** by exposure to potassium thioacetate. Isopropylidene removal followed by oxidative cleavage gave the ring-opened aldehyde **203**, and subsequent cyclization, deprotection/protection, and bromination furnished **204** as a mixture of diastereomers. The glycosylation step afforded an anomeric mixture, with the desired  $\beta$ -adduct crystallizing from solution, and deprotection concluded the synthesis of thiarabine **47**.

Secrist *et al.* developed an improved gram-scale synthesis of thiarabine **47** in order to access sufficient quantities for biological study (Scheme 37).<sup>441,443</sup> L-Xylose **205** was converted into the benzyl-protected dithioacetal **208** by selective methoxylation, O-benylation, and dithioacetalization. Treatment of **208** with a triphenylphosphine-iodine-imidazole reagent resulted in a single inversion cyclization at C-4, affording the desired 4-thiosugar **209** in the *D-arabino* configuration.<sup>441</sup> Coupling of **209** with silylated uracil led to a separable anomeric mixture of **210**; as compared to making the cytosine analog, this reaction has several advantages, including a higher  $\beta$ : $\alpha$  ratio, higher yield, and easier purification process. **210 $\beta$**  was converted to **212** via C-4 carbonyl activation and amination. Subsequent benzyl deprotection gave thiarabine **47**.<sup>443</sup>

**5.3.13. RX-3117 Biology.**—RX-3117 **29**, a fluorocyclopentenyl-cytosine nucleoside analog, was shown to have anticancer activity<sup>460,461</sup> with an IC<sub>50</sub> range of 0.4 to >30  $\mu$ M for 59 cell lines.<sup>462</sup> The agent was also shown to inhibit DNA and RNA synthesis and induce apoptotic cell death of tumor cells.<sup>349</sup> Interestingly, preclinical studies using nude mice

implanted with various tumors, including colon, lung, renal, and pancreatic, indicated that the ribose form and not the 2'-deoxyribose form was the active compound.<sup>463</sup>

Biological data were recently published on the metabolism and mechanism of action of RX-3117 **29**.<sup>462</sup> Cellular uptake of the agent occurs by hENT, and the agent is a poor substrate for cytidine deaminase, preventing its degradation within the cell.<sup>462</sup> Furthermore, uridine-cytidine kinase generates RX-3117-5'-monophosphate, which is subsequently phosphorylated to the di- and triphosphate forms by cellular kinases. RX-3117-5'-triphosphate is incorporated into RNA and inhibits RNA synthesis at high concentrations, but at IC<sub>50</sub> values it does not affect RNA structural integrity. RX-3117-5'-diphosphate is a substrate for ribonucleotide reductase, and the 2'-deoxy-RX-3117-5'-triphosphate can be incorporated into DNA, potentially providing another mechanism of cancer cell growth inhibition. The 2'-deoxy-RX-3117-5'-triphosphate also downregulates DNA methyltransferase-1 protein (at IC<sub>50</sub> values for specific cell line) and Cdc2 protein expression, which are both involved in the cell cycle division process.<sup>462</sup>

Preclinical and clinical studies of RX-3117 **29** showed potent efficacy in xenograph models.<sup>464</sup> In 2012, Rexahn Pharmaceuticals reported the completion of a Phase I clinical trial for RX-3117 **29** in Europe and indicated that the drug had 56% orally bioavailable, had a plasma half-life of 14 h, and was well tolerated in cancer patients. Currently, a Phase Ib trial is ongoing to test the dose-finding and safety of RX-3117 **29** as an oral monotherapy to treat advanced malignancies ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02030067) identifier: [NCT02030067](https://clinicaltrials.gov/ct2/show/study/NCT02030067)—recruiting patients).

**5.3.14. RX-3117 Synthesis.**—A synthesis of RX-3117 **29** is illustrated in Scheme 38.<sup>463</sup> Protection of D-ribose **213** followed by Wittig olefination, Swern oxidation, and vinyl Grignard addition afforded diene **217**. Silyl ether deprotection and selective primary alcohol benzylation, followed by ring-closing metathesis and oxidative rearrangement via a [3,3]-sigmatropic shift yielded benzylated cyclopentenone **221**. Subsequent iodination of the double bond, lactone reduction, and protection of the resulting lactol provided TBDPS ether **224**. Electrophilic vinyl fluorination followed by desilylation gave fluorocyclopentenol **225**, which was coupled with *N*<sup>3</sup>-benzoyluracil under Mitsunobu conditions, providing protected nucleoside analogue **226**. Removal of the *N*<sup>3</sup>-benzoyl group by methanolic ammonia and the 5'-benzyl group by BBr<sub>3</sub> generated **227**. Finally, triacetatyl intermediate **227** was converted to the cytidine analog RX-3117 **29** by a sequence involving formation of a 4-triazole intermediate and nucleophilic displacement with ammonia.

An alternative and slightly improved approach to RX-3117 **29** uses a trityl protecting group at the 5'-position of intermediate **214**, which avoids a deprotection/reprotection step, decreasing the total number of steps of the synthesis by two.<sup>465</sup> However, both processes include long and laborious syntheses, producing only small quantities of the final product in low yield. A large-scale synthesis is most likely needed for significant clinical treatment of RX-3117 **29**.

## 5.4. Cytarabine and Cytarabine Prodrugs

**5.4.1. Cytarabine Biology.**—In the 1940s, Werner Bergmann isolated two nucleosides from the marine sponge *Cryptotethya crypta*, which he termed spongothymidine and



spongouridine. These spongonucleosides were determined to have a 2'-OH in the *arabino* configuration. In the late 1950s, various unnatural nucleosides (including spongonucleosides) were prepared with the hypothesis that they would kill cancer cells by interfering with DNA replication, and indeed, many were found to inhibit DNA synthesis *in vitro*. In the 1960s, Seymour Cohen reported promising cytotoxic effects of a new spongonucleoside in cancer cells: cytarabine **4** (aka spongocytidine, cytosine arabinoside, or ara-C).<sup>466</sup> The anticancer agent rapidly progressed through rodent models and clinical trials to gain FDA approval as a cancer therapy.

Cytarabine **4** is transported into the cell primarily by hENT-1.<sup>467,468</sup> Cytarabine **4** is monophosphorylated by 2'-deoxycytidine kinase, which is the rate-limiting step in the formation of the active form: cytarabine-5'-triphosphate.<sup>469</sup> Cytarabine-5'-triphosphate competes with dCTP for incorporation into DNA and is a weak competitive inhibitor of DNA polymerases. Cytarabine-5'-triphosphate is also a substrate for DNA polymerase  $\alpha$  and is incorporated into both the leading and lagging strands of DNA. This incorporation leads to delayed chain termination, allowing several additional nucleotides to be incorporated before the DNA polymerase stops extending and falls off the replication fork. Cell cycle arrest occurs with the DNA repair mechanism deployed. If the embedded cytarabine-5'-monophosphate is not removed, the cell will enter apoptosis.<sup>470-472</sup>

Various factors decrease the effectiveness of cytarabine **4**. The agent is deaminated by cytidine deaminase, and cytarabine-5'-monophosphate can be deaminated by deoxycytidylate deaminase, leading to the inactive uridine analogs.<sup>473</sup> In addition, cytarabine-5'-triphosphate is a feedback inhibitor of 2'-deoxycytidine kinase, limiting the formation of cytarabine-5'-monophosphate.<sup>474</sup>

Resistance to cytarabine **4** exposure is afforded by several different mechanisms, including nucleotide transporter and 2'-deoxycytidine kinase gene mutations and decreased expression levels,<sup>475</sup> and changes in cytosolic nucleotidase II,<sup>476</sup> cytidine deaminase, and deoxycytidylate deaminase activities.<sup>477</sup> Recently, a correlation was reported between increased expression of the ABC transporter MRP8 and a poor treatment response of individuals with acute myelogenous leukemia treated with cytarabine **4**.<sup>478</sup>

Cytarabine **4** has been FDA-approved since 1969 for the treatment of acute myeloid leukemia,<sup>479</sup> with the exception of acute promyelocytic leukemia.<sup>480</sup> It is usually given in combination with a topoisomerase II inhibitor such as daunorubicin, idarubicin, or mitoxantrone.<sup>481-483</sup> Several clinical trials are underway using cytarabine **4** alone or in combination treatment studies. ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT01191541](#)—recruiting patients, [NCT01289457](#)—recruiting patients, and [NCT01756677](#)—recruiting patients).

**5.4.2. Elacytarabine Biology.**—Elacytarabine (**51**, CP-4055), an oleic acid ester of cytarabine **4**, was designed to enter cancer cells independent of nucleoside transporters, which would also overcome the low expression level of hENT-1 at the plasma membrane or loss of function mutation in hENT-1.<sup>484,485</sup> The lipophilic prodrug diffuses through the phospholipid bilayer, being retained in a membrane fraction within the cell.<sup>486</sup> After the action of an unidentified esterase, cytarabine **4** is released into the cytosol.<sup>349</sup> In comparison

to cytarabine **4**, elacytarabine **51** exposure led to an increased intracellular level of active metabolite, which was also retained longer in cells after drug removal.<sup>486</sup>

Elacytarabine **51** showed promising results in both preclinical and clinical studies.<sup>487–489</sup> However, in April 2013, despite the initial success, Clavis Pharma announced that elacytarabine **51** did not perform better than the regular standard of care therapy in a pivotal Phase III trial<sup>354</sup> ([ClinicalTrials.gov](https://clinicaltrials.gov) identifier: [NCT01147939](https://clinicaltrials.gov/ct2/show/study/NCT01147939)—completed in 2013). Moreover, patients with relapsed/refractory acute myeloid leukemia had no meaningful benefit from elacytarabine **51** when directly compared to seven other treatment regimens.<sup>490</sup> As a result, Clavis has suspended all further developmental work on elacytarabine **51**.

**5.4.3. MB 07133 Biology.**—Ligand Pharmaceuticals Inc. developed MB 07133 **34**, a phosphoramidate prodrug of cytarabine **4**. In 2003, MB 07133 **34** entered a phase I/II clinical trial as an intravenous infusion agent for the treatment of unresectable hepatocellular carcinoma ([ClinicalTrials.gov](https://clinicaltrials.gov) identifier: [NCT00073736](https://clinicaltrials.gov/ct2/show/study/NCT00073736)—completed—no results posted). MB 07133 **34** was well tolerated with evidence of disease stabilization. In 2011, Ligand Pharmaceuticals granted the license to Chiva Pharmaceuticals to develop MB 07133 **34** in China.

**5.4.4. Cytarabine, Elacytarabine, and MB 07133 Syntheses.**—In 1959, cytarabine **4** and 3- $\beta$ -D-arabinofuranosyluracil were originally synthesized by the Dekker group (Scheme 39).<sup>491</sup> Cytidine **93** was treated with polyphosphoric acid and purified by gradient elution chromatography to afford the diphosphate intermediate **228** as a crystalline cyclohexylammonium salt. Subsequent enzymatic dephosphorylation with prostatic phosphatase furnished cytarabine **4**. In order to verify the hypothesis that the transformation proceeded via a cyclonucleoside-5'-diphosphate intermediate, *O*<sup>2</sup>-2'-cyclocytidine **229** was isolated by ion-exchange fractionation. The intermediate could further be converted to cytarabine **4** by enzyme-mediated dephosphorylation and hydrolysis.

Two different process syntheses of cytarabine **4** used 1- $\beta$ -D-arabinofuranosyl uracil **231** as a key intermediate, which can be synthesized on a large-scale in two steps from uridine **98** via 2,2'-anhydro-uridine **230** (Scheme 40).<sup>492</sup> One process treated **231** with HMDS under high pressure and high temperature, leading to the intermediate **232**, which upon treatment with MeOH provided cytarabine **4** on a 50-g-scale.<sup>493</sup> The other process converted **231** to the *O*-acyl-substituted, 1,2,4-triazole derivative **233**, and subsequent deprotection produced cytarabine **4** on a kilogram-scale.<sup>494</sup>

Elacytarabine **51** can be prepared in one step by enzyme-mediated esterification of the 5'-OH of cytarabine **4** with oleic anhydride in DMF,<sup>495</sup> whereas the cyclic prodrug MB 07133 **34** has been synthesized from cytarabine **4** in four steps (Scheme 41).<sup>496,497</sup> Selective 2',3'-silyl protection followed by treatment with *N,N*-dimethylformamide dimethyl acetal generated intermediate **236**.<sup>496</sup> In parallel, *trans-p*-nitrophenylphosphate **239** was prepared in three steps from methyl 3-hydroxy-3-(pyridin-4-yl)-propanoate **237** via esterification (with *L-N,N*-dimethyl phenylalanine) and subsequent diastereomer separation, reduction with LiAlH<sub>4</sub>, and coupling with *p*-nitrophenyl phosphorodichloridate. Condensation of

protected nucleoside **236** with *p*-nitrophenyl phosphorochloridate **239** followed by TBS deprotection furnished MB 07133 **34**.<sup>497</sup>

## 6. PURINE ANALOGS

### 6.1. Deoxypurines

**6.1.1. Cordycepin Biology.**—In 1951, cordycepin **27** was isolated from the fungus *Cordyceps militaris*<sup>498</sup> and exhibited anticancer, anti-inflammatory, and antifungal activities.<sup>499,500</sup> It is currently thought to be the active ingredient in various traditional Chinese medicines.<sup>501</sup> The mechanism of cellular uptake of this agent has not been fully elucidated. However, the 3'-OH has been shown to be required for efficient uptake of nucleosides and nucleoside analogs by hENTs, suggesting that cordycepin **27** does not efficiently use this mechanism for uptake.<sup>502</sup> Once in the cell, the agent is converted to the cordycepin-5'-monophosphate by adenosine kinase,<sup>503</sup> and then to the corresponding di- and triphosphate forms by other cellular kinases. Cordycepin-5'-triphosphate is incorporated into the poly A tail of the growing mRNA chain, presumably causing chain termination due to its lack of a 3'-OH.<sup>504</sup> Furthermore, cordycepin **27** can inhibit ribose-phosphate pyrophosphokinase and 5-phosphoribosyl-1-pyrophosphate amidotransferase *in vitro*.<sup>505,506</sup>

Cordycepin **27** exposure was shown to cause DNA double-stranded break in breast cancer cells.<sup>507</sup> A possible explanation for this observation is the inhibitory effect this agent has on poly(ADP-ribose) polymerase, an enzyme involved in the repair of DNA single-strand breaks.<sup>28</sup> Unrepaired DNA single-strand breaks could lead to more extensive DNA damage, such as double-stranded breaks. In addition, cordycepin **27** is a substrate for adenosine deaminase, which can lead to significant degradation *in vivo*. Other mechanisms of action for cordycepin **27** are binding A3 adenosine and DR3 receptors, activating caspase-8/-3 pathways for cell death, suppressing cyclin D1 and c-myc expression cell signal pathways, inhibiting cAMP formation, and decreasing glycogen synthase kinase-3  $\beta/\beta$ -catenin activation.<sup>508-511</sup>

In 2008, Oncovista initiated a Phase I/II clinical trial of cordycepin **27** in combination with pentostatin **6** (adenosine deaminase inhibitor; see Section 6.1.5) for the treatment of refractory TdT-positive leukemia (ClinicalTrials.gov identifier: NCT00709215—unknown status >2 years without an update). Currently, no study updates have been reported. Earlier in 2000, a Phase I trial was conducted using cordycepin **27** plus pentostatin **6** in treating patients with refractory acute lymphocytic or chronic myelogenous leukemia (ClinicalTrials.gov identifier: NCT00003005—completed in 2001—no results reported). Development of cordycepin **27** seemed to have stalled, until Rainforest Pharmacal announced plans to reactivate the IND in 2016 and start a Phase I/II trial in TdT-positive refractory leukemia patients in 2017. However, issues with deamination will have to be addressed before cordycepin **27** can progress significantly in clinical trials.

**6.1.2. Cordycepin Synthesis.**—In 1961, Lee *et al.* reported a pioneering synthesis of cordycepin **27**, which involved desulfurization of 3'-ethylthioadenosine by treatment with Raney nickel in refluxing 2-methoxyethanol.<sup>512</sup> Later, Hansske and Robins used 2-acetoxyisobutyryl bromide (Mattock's bromide) to synthesize cordycepin (Scheme 42).<sup>513</sup>

Thus, reaction of adenosine **240** with 2-acetoxyisobutyryl bromide afforded the corresponding *trans*-bromoacetates (**241a** and **241b**). Cordycepin **27** was finally obtained after treatment with acidic resin to provide epoxide **242** and regioselective reduction with lithium triethylborohydride.

Aman *et al.* used a similar approach for a process synthesis of cordycepin **27** (Scheme 43).<sup>514</sup> Initially, the authors thought to use the epoxide **242** as a key intermediate, but they had difficulty controlling the regioselectivity of the subsequent reduction. They noticed, however, that *trans*-bromoacetate **243b** fortuitously precipitated selectively from solution. This intermediate was isolated and subsequently deprotected at the 2'- and 5'-positions in order to increase compound solubility and reactivity (decreased steric hindrance) in the final reduction reaction. The deprotection required a delicate balance of reaction temperature, acid concentration, and methanol concentration in order to fully hydrolyze **243b** to **244**, without leading to compound decomposition via depurination. Depurination also plagued the final dehalogenation step, and conditions had to be optimized. Finally, hydrogenation in the presence of 4–5 molar equivalent of sodium acetate in a 1:1 solvent mixture of ethanol and water effectively debrominated the hydrochloride salt of **244**. Ethyl acetate was used to extract crude cordycepin **27** from the reaction mixture, which had a high salt load. Subsequent addition of methanol to the organic solution followed by cooled agitation precipitated cordycepin **27**. The overall process required no chromatography and produced the product cordycepin **27** on a 100-g-scale at 99% purity.

**6.1.3. Cladribine Biology.**—Inherited adenosine deaminase deficiencies were reported to cause lymphopenia, an immunodeficiency in humans.<sup>515</sup> Furthermore, the accumulation of deoxyadenosine nucleotides had been implicated in the pathogenesis of this disorder. An increase in dATP levels was shown to kill lymphocytes.<sup>516</sup> Experimentally, a high dADP level in lymphoid cells led to negative feedback inhibition of ribonucleotide reductase, and the subsequent decrease in dNTP pools resulted in DNA synthesis inhibition and cell death.<sup>517</sup> It is now known that inhibition of adenosine deaminase causes an increase in both dATP and dADP levels.

Carson *et al.* reasoned that deoxyadenosine analogs, which are resistant to adenosine deaminase activity, may be phosphorylated into their triphosphate forms in the cell. Moreover, a deoxyadenosine analog that inhibits adenosine deaminase activity might be successful for the treatment of lymphocytic malignancies.<sup>518</sup> In collaboration with John Montgomery, Carson studied 2-chloro-2'-deoxyadenosine (cladribine, **8**), which was shown to have significant cytotoxicity in malignant T lymphocytes.<sup>518</sup> The presence of the C2 chlorine in the purine base was believed to make the agent resistant to adenosine deaminase activity.<sup>519</sup> Furthermore, cladribine **8** was shown to be converted to the cladribine-5'-triphosphate *in vitro*, and to exert significant cytotoxicity both *in vitro* and *in vivo*.<sup>519–521</sup>

Cladribine **8** is transported by both hCNT and hENT into cells.<sup>522</sup> It is phosphorylated by cytosolic 2'-deoxycytidine kinase<sup>516</sup> and mitochondrial deoxyguanosine kinase to generate cladribine-5'-monophosphate. This phosphorylation is the rate-limiting step in the activation of the parent nucleoside analog.<sup>523</sup> Subsequent phosphorylation generates cladribine-5'-diphosphate and cladribine-5'-triphosphate forms; both are active metabolites of cladribine

**8**. Cladribine-5'-triphosphate is readily incorporated into DNA.<sup>524</sup> Chain extension by DNA polymerase continues past a single DNA incorporated cladribine-5'-monophosphate, but elongation is terminated when the polymerase encounters three successive incorporated cladribine molecules in the growing DNA strand.<sup>17</sup> In addition, cladribine-5'-triphosphate inhibits ribonucleotide reductase activity.<sup>525</sup> Collectively, these actions can promote cell death by dNTP imbalance and DNA double-stranded breaks.<sup>526,527</sup> Furthermore, cladribine **8** was shown to interfere with methyltransferases and may promote hypomethylation *in vitro*.<sup>528–530</sup> Other effects of cladribine **8** include the induction of caspase-3 activation, thereby promoting cell apoptosis *in vitro*.<sup>531</sup> Interestingly, the agent has also been shown to impede mononuclear cells from crossing the blood brain barrier and to inhibit T cells and B cells in the central nervous system.<sup>532</sup>

Resistance to cladribine **8** exposure can arise from compound metabolism and detoxification as well as increased egress of the agent from cancer cells *in vitro*.<sup>533</sup> Interestingly, a recent report examining metabolism data for cladribine **8** treatment in animals and humans indicated the formation of 2-chlorodeoxyinosine and 2-chlorohypoxanthine (inactive metabolites), suggesting a sensitivity to adenosine deaminase degradation.<sup>534</sup> Furthermore, cladribine **8** was shown to be a substrate for the ABCG2 transporter, leading to cell egress.<sup>535</sup>

Cladribine **8** gained FDA approval in 2004 for use as a first-line monotherapy for hairy cell leukemia<sup>536–538</sup> and has a favorable toxicity profile in comparison to other lymphocytic drugs.<sup>539</sup> It has been used in combination treatment for chronic lymphocytic leukemia<sup>35</sup> and abnormal mast cells debulking.<sup>540</sup> Cladribine **8** has also been reported to kill immature and mature monocyte-derived dendritic cells.<sup>537</sup> In addition, the agent was shown to be effective for treatment of Langerhans cell histiocytosis.<sup>541</sup> Cladribine **8** therapy has also been investigated for relapsingremittent multiple sclerosis, but has not gained FDA approval for this disease due to lack of safety information.<sup>542</sup>

**6.1.4. Cladribine Synthesis.**—Interestingly, the first reported preparation of cladribine **8** was as an intermediate in a synthesis of 2'-deoxyguanosine.<sup>543,544</sup> In 1972, Christensen *et al.* reported a nonstereoselective fusion glycosylation synthesis of cladribine **8** from diprotected sugar **245**, made by treatment of 2'-deoxyribose in methanol with a catalytic amount of acid followed by *p*-toluoyl protection (Scheme 44).<sup>545</sup> The anomeric mixture of glycosylated product **246** was resolved by silica gel chromatography, and the  $\beta$ -anomer was treated with methanolic ammonia, to give cladribine **8**.

In 2006, Robins reported a new synthesis of cladribine **8** that focused on improving the yield of the glycosylation step between  $N^6$ -imidazolyl purine **248** and  $\alpha$ -1'-chlorosugar **247**, which is formed by treating 2'-deoxyribose with HCl in methanol followed by tosylation and stereoselective crystallization of the  $\alpha$ -anomer (Scheme 45).<sup>546,547</sup> The modified purine base exhibited improved solubility in organic solvents as well as better glycosylation regioselectivity, due to steric hindrance of the  $N^7$  position of the base, leading to exclusive formation of the  $N^9$  adduct. A binary solvent system was employed in the glycosylation in attempts to increase the solubility of the modified purine base as well as improve the stereoselectivity of the reaction. The rationale for this system drew on a previously reported

observation that preferential solvation of reactants can lead to different free energies of activation, and therefore alter the ratios of two competitive pathways.<sup>548</sup> Indeed, a judicious solvent choice of cold dichloromethane and acetonitrile (1:1 mixture) allowed for the selective formation of the desired  $\beta$ -nucleoside in 95% yield. Furthermore, the use of cold solvent led to decreased anomerization of chlorosugar **247**. The synthesis was concluded by converting the imidazol group to an imidazolium group by treatment with *in situ*-generated benzyl iodide, and the resulting salt was heated in methanolic ammonia to give, after recrystallization, pure cladribine **8**, in >90% from chlorosugar **247**.

More recently, a kilogram-scale synthesis has been reported for cladribine **8** (Scheme 46).<sup>549</sup> The researchers chose to explore Vorbrüggen conditions for the glycosylation. Generally, when this reaction involves 2'-deoxy lactones, it usually gives a product ratio of roughly 50%  $\alpha/\beta$  nucleosides. This was indeed found to be the case when the glycosylation of the 4-chlorobenzyl protected sugar **249** and silylated 2-chloroadenine was performed in solvents such as tetrahydrofuran and dichloromethane. However, when solvents such as acetonitrile or lithium hexamethyldisilazane were employed, the desired  $\beta$ -anomer fortuitously precipitated from solution. It was hypothesized that a dynamic equilibrium existed in the reaction, involving glycosylation and retroglycosylation of the  $\alpha$ -anomer along with concomitant precipitation of the  $\beta$ -anomer, driving the equilibrium to the desired  $\beta$ -nucleoside product **253**. The reaction was complete within 1 h; however, an aging step at 60 °C was used to further increase the yield. Furthermore, only a true catalytic amount of the Lewis acid trimethylsilyl trifluoromethanesulfonate was needed, which greatly simplified the purification of the product by obviating the need for a catalyst quench and aqueous workup. Isolation was accomplished by direct filtration. Deprotection with 10–20 mol % of sodium methoxide in MeOH led to *in situ* crystallization of high purity cladribine **8**, in 99.8–99.9% HPLC purity with an overall yield of 43%.

**6.1.5. Pentostatin Biology.**—Pentostatin (**6**, [(*R*)-3-(2-deoxy- $\beta$ -D-erythro-pentofuranosyl)-3,6,7,8-tetrahydro-imidazo-[4,5-*d*][1,3-diazepin-8-ol]) is an antibiotic isolated from a culture broth of *Streptomyces antibioticus*.<sup>550</sup> It is an irreversible, stoichiometric inhibitor of mammalian adenosine deaminase, with a  $K_i$  in the nanomolar range.<sup>551,552</sup> Importantly, pentostatin **6** directly inhibits this enzyme in order to truly mimic adenosine deaminase deficiencies, which is in contrast to cladribine **8**, which was designed as an adenosine deaminase-resistant antimetabolite (see Section 6.1.3.). Pentostatin **6** is unique among anticancer nucleoside analogs in that it is active without metabolism (cellular phosphorylation). It has been proposed that the tetrahedral carbon at position 8 of the diazepine ring of pentostatin **6** mimics the purported transition state tetrahedral carbon in the deamination of adenosine to inosine.<sup>553</sup> In addition to this effect, pentostatin **6** also targets cellular methyltransferases, hindering the ability of the cell to methylate both DNA and mRNA.<sup>554–556</sup>

Pentostatin **6** is used to treat hairy cell leukemia and is effective against lymphoid malignancies with high adenosine deaminase activities.<sup>557,558</sup> The agent has also been used to prevent and to treat both acute and chronic graft-versus-host disease.<sup>559</sup> Pentostatin **6** is currently being tested in several clinical trials. An additional Phase I/II clinical trial uses



pentostatin **6** and cyclophosphamide (DNA-alkylating agent) cotreatment for low-intensity stem cell transplantation with multiple lymphocyte infusions and to treat advanced kidney cancer ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00923845) identifier: NCT00923845—active status—primary objective of trial was not achieved). A Phase II trial combines pentostatin **6**, cyclophosphamide (DNA alkylating agent), and rituximab (monoclonal antibody against B cell protein CD20) followed by lenalidomide (antiangiogenic/immunomodulatory agent) treatment for relapsed or refractory B cell chronic lymphocytic leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00074282) identifier: NCT00074282—unknown status—no updates in 2 years). Finally, a Phase I trial involves the combination of pentostatin **6**, bendamustine (DNA-alkylating agent), and ofatumumab (monoclonal antibody against B cell protein CD20) for the treatment of chronic lymphocytic leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01352312) identifier: NCT01352312—active status—not recruiting patients).

**6.1.6. Pentostatin Synthesis.**—Parke-Davis developed a large-scale industrial production of pentostatin **6** using fermentation cultures of *Streptomyces antibioticus* NRRL 3238. To date, this approach is the major form of production of the compound, although a few other syntheses have been reported.<sup>560</sup>

In 1982, the first synthesis of pentostatin **6** was reported by chemists at Warner-Lambert/Parke-Davis (Scheme 47).<sup>561,562</sup> Knoevenagel condensation of azole **254** with benzaldehyde produced styrylimidazole **255**, and subsequent benzylation formed regioisomers **256** and **257**. The mixture was subjected to ozonolysis, and **258** was separated from its isomeric product by differential precipitation. Carbon homologation of **258** was followed by nitro reduction, ring closure, and debenylation, producing heterocycle **261**. Due to the instability of heterocycle **261**, glycosylation conditions had to be optimized. The use of tin(IV) chloride as Lewis acid and BSTFA as silylating agent were found to produce the desired product in a 1:1 mixture of  $\alpha/\beta$  diastereomers. The mixture can be separated by preparative HPLC on a small-scale and by fractional crystallization on a larger-scale. Deprotection of the  $\beta$ -anomer **262** in the presence of sodium methoxide followed by ketone reduction produced a separable 1:1 mixture of pentostatin **6** and its stereoisomer. It is noteworthy that Zhang *et al.* improved the conclusion of the synthesis by using a chiral ruthenium catalyst for asymmetric transfer hydrogenation of ketone **262**.<sup>563</sup> This reaction was completely diastereoselective and gave a yield of 80%.

Phiasivongsa and Redkar reported a ring expansion of protected 2'-deoxyinosine **263** using diazomethane.<sup>564</sup> This was followed by deprotection and reduction to furnish a diastereomeric mixture of pentostatin **6**, which was separated by following the Warner-Lambert/Parke-Davis procedure in Scheme 48.<sup>564</sup> Unfortunately, no yields were reported.

**6.1.7. Clofarabine Biology.**—Clofarabine **12** is a second-generation purine nucleoside analog designed to overcome the limitations of fludarabine-5'-monophosphate **7** (see Section 6.2.3) and cladribine **8** (see Section 6.1.3),<sup>565</sup> which are both susceptible to glycosidic bond cleavage.<sup>521</sup> Indeed, the introduction of a C2'-fluorine in the *arabino* configuration in clofarabine **12** significantly increased the stability of the glycosidic bond in acidic solution and toward phosphorolytic cleavage.<sup>566,567</sup> Chlorine substitution instead of fluorine at the 2-position of the adenine base was chosen in order to avoid production of the

2-fluoroadenine, a precursor to the toxic 2-fluoro-adenosine-5'-triphosphate, should the glycosidic bond be cleaved.<sup>568</sup>

Clofarabine **12** is usually administered intravenously; however, the increased stability of the glycosidic bond also allows for oral administration.<sup>566,569,570</sup> Clofarabine **12** enters the cell via hENT1, hENT2, and hCNT2, and to a much lesser extent by passive diffusion.<sup>429</sup> 2'-Deoxycytidine kinase generates clofarabine-5'-monophosphate;<sup>571</sup> however, the enzyme has a lower  $K_m$  for clofarabine **12** than cladribine **8**.<sup>572</sup> Moreover, clofarabine resistance arises from decreased 2'-deoxycytidine kinase activity *in vitro*.<sup>572</sup> Formation of clofarabine-5'-diphosphosphate is generated by purine nucleotide monophosphate kinase, and is the rate-limiting step in the activation of the agent.<sup>573</sup> This is in contrast to many other nucleoside analogs, in which the rate-limiting step is the generation of clofarabine-5'-monophosphate.

Clofarabine-5'-diphosphate serves as an intracellular reservoir for the active-metabolite clofarabine-5'-triphosphate, which is retained in the cell longer than clofarabine-5'-triphosphate in cells from CLL and AML patients *ex vivo*.<sup>572</sup> Clofarabine-5'-triphosphate has several mechanisms of action in the cell. It inhibits ribonucleotide reductase (presumably by allosteric inhibition), which results in a decrease in dCTP and dATP concentrations.<sup>520</sup> The reduction in the amount of dCTP likely leads to DNA synthesis inhibition, and a decrease in the concentrations of dATP leads to greater DNA incorporation of clofarabine-5'-triphosphate (self-potential).<sup>568</sup> A low intracellular concentration of clofarabine-5'-triphosphate can be incorporated by DNA polymerases  $\alpha$  and  $\epsilon$ <sup>574</sup> into DNA and can promote polymerase arrest at the replication fork, inhibiting DNA repair and inducing strand breaks *in vitro*.<sup>573,574</sup> This DNA damage leads to the induction of cytochrome c-mediated apoptosis *in vitro*.<sup>575</sup> Studies have shown that clofarabine-5'-triphosphate can also be incorporated into RNA in cell lines.<sup>573</sup>

Clofarabine **12** is an FDA-approved drug for the treatment of relapsed or refractory pediatric acute lymphoblastic leukemias.<sup>35,576</sup> It is being evaluated for the treatment of acute myeloid leukemia in both children and the elderly.<sup>577</sup> The agent was examined for treatment of non-Hodgkin's lymphomas, myelodysplastic syndrome, and solid tumors.<sup>520</sup> Oral clofarabine **12** has been studied in relapse/refractory non-Hodgkin's lymphomas<sup>578</sup> and in high-risk myelodysplastic syndrome patients.<sup>579</sup> Currently clofarabine **12** in combination treatments is being investigated in over a dozen clinical trials for chronic lymphocytic leukemia, acute myelogenous leukemia, myelodysplastic syndrome, and mixed phenotype acute leukemia with nucleoside analogs such as cytarabine **4**, gemcitabine, and fludarabine-5'-monophosphate **7** and with other chemotherapies such as but not limited to busulfan, idarubicin, etoposide, and mitoxantrone.

**6.1.8. Clofarabine Synthesis.**—In the early 1990s, Montgomery *et al.* utilized differentially protected sugar **272** in order to access various nucleoside analogs, including clofarabine **12** (Scheme 49).<sup>566,580,581</sup> The synthesis began with the stereoselective fluorination of diprotected tosylate **267**, which can be formed by isopropylidene protection and tosylation of  $\alpha$ -D-glucopyranose, followed by selective deprotection, leading to fluorinated sugar **268**. Benzoylation of the 5'-OH followed by acid-mediated deprotection afforded lactol **269**. Oxidative cleavage and subsequent rearrangement produced **270**, which

was eventually converted to bromosugar **272**. Condensation of 2,6-dichloropurine with **272** occurred in refluxing dichloroethane, and chromatographic separation of the anomeric isomers produced nucleoside derivative **273**. Three days of treatment of **273** in a steel bomb with ethanolic ammonia afforded a mixture of clofarabine **12** and the partially deprotected 5'-*O*-benzoyl derivative. Treatment of the mixture with LiOH in MeCN/H<sub>2</sub>O removed the residual protecting group, and pure clofarabine **12** was produced by three crystallizations from water.

Twelve years later, Bauta *et al.* reported an improved and scalable process synthesis of clofarabine **12** (Scheme 50).<sup>582</sup> Studies were performed on bromosugar **279**,<sup>583</sup> with the desire to increase  $\beta$ -anomer formation in the glycosylation step. Compound **279** was easily prepared by first conversion of 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose **274** to protected sugar **276**<sup>584</sup> via formation of the benzoxonium ion intermediate and solvolysis. Subsequent installation of the imidazolylsulfonate leaving group, stereoselective deoxyfluorination using excess KHF<sub>2</sub>, and finally bromination under acidic conditions completed the synthesis of **279**.<sup>583</sup> Empirical studies of the glycosylation step of 2-chloroadenine found potassium *tert*-butoxide to be the optimum base, with the potassium counterion being particularly important for both yield and  $\beta$ -selectivity. Solvent choice was guided by the precedent that low dielectric constants of a solvent favor the formation of the  $\beta$ -anomer in glycosylations involving bromosugar **279**. It is believed that this effect arises from suppression of the formation of the oxonium ion intermediate **280** and subsequent S<sub>N</sub>1-type condensation, which would lead to both  $\alpha$ - and  $\beta$ -glycosylation products. Nonpolar solvents would therefore be needed in order to favor a S<sub>N</sub>2 reaction pathway. However, solubility in such solvents would also be an issue, particularly with the 2-chloroadenine base. In order to find a balance between the two conflicting requirements of anomeric selectivity and substrate solubility, a 1:2:1 mixture of MeCN/*tert*-amyl alcohol/DCE was eventually chosen after considerable experimentation. In addition, the additive CaH<sub>2</sub> was found to increase overall yield and anomeric ratio, due in part to removal of trace amounts of water from the solvent.

A number of challenges were overcome to separate protected clofarabine **281** from the heterogeneous reaction mixture. Excess 2-chloroadenine was filtered off. A subsequent solvent exchange to *n*-butyl acetate dissolved the mixture to allow for heptane-induced precipitation of crude **281**, which after two subsequent pH-maintained methanol slurries was isolated in a >50:1  $\beta/\alpha$  ratio. Finally, deprotection with a catalytic amount of sodium methoxide in methanol followed by recrystallization formed drug-pure clofarabine **12** on a 25-g-scale.

## 6.2. Arabinose Purine Analogs

**6.2.1. Nelarabine Biology.**—Arabinosylguanine (9- $\beta$ -D-arabinofuranosylguanine; ara-G) was discovered in the 1960s, but was not significantly developed at that time. In the 1970s, studies on the rare autosomal disorder purine nucleoside phosphorylase deficiency rekindled interest in ara-G.<sup>393,585</sup> A deficiency in purine nucleoside phosphorylase, an enzyme that catalyzes the phosphate-mediated deribosylation of certain purine nucleosides, results in a profound dGTP-mediated imbalance in the endogenous nucleotide pools that

leads to T cell lymphopenia without concomitant decreases in B cell populations.<sup>586</sup> This sensitivity of human T cells (especially immature T cells) to purine nucleoside phosphorylase deficiency is hypothesized to arise from their relatively high ratio of kinase activity versus nucleotidase activity (in contrast to other cell types) which results in a high accumulation of intracellular dGTP.<sup>587</sup> Deoxyguanosine, but not ara-G, is a substrate for purine nucleoside phosphorylase.<sup>588,589</sup>

Nelarabine (**13**, 2-amino-9- $\beta$ -D-arabinosyl-6-methoxy-9H-guanine; 506U78) is a prodrug of ara-G and has a 10-fold higher solubility as compared to ara-G.<sup>590,591</sup> Nelarabine **13** is converted to ara-G in the serum by adenosine deaminase.<sup>592</sup> Ara-G is phosphorylated by either cytosolic 2'-deoxycytidine kinase or mitochondrial deoxyguanine kinase into ara-G-5'-monophosphate<sup>593</sup> and, subsequently, into ara-G-5'-diphosphate and ara-G-5'-triphosphate by cellular kinases. The active metabolite ara-G-5'-triphosphate is a substrate for DNA polymerases and is incorporated into DNA. This incorporation leads to inhibition of the DNA polymerases, causing inhibition of DNA replication and resulting in cell death via apoptosis.<sup>594</sup> Moreover, it was believed that ara-G-5'-triphosphate would accumulate to high intracellular levels and would promote cytotoxic effects similar to those created by high dGTP concentrations. When T cell lymphoma cell lines were exposed to nelarabine, this resulted in FasL-mediated cytotoxicity, whereas myeloid and B cells did not accumulate ara-G-5'-triphosphate, but were stopped in the S phase of the cell cycle.<sup>595</sup> High concentrations of ara-G-5'-triphosphate accumulated in malignant T cells from human bone marrow when tested in a rodent model.<sup>596</sup>

In 2005, nelarabine **13** was approved by the FDA for the treatment of relapse T cell acute lymphocytic leukemia and relapse T cell lymphoblastic lymphoma.<sup>597</sup> Additional clinical trials are being conducted studying pharmacokinetic and pharmacodynamics properties in addition to efficacy in combination with etoposide (topoisomerase inhibitor) and cyclophosphamide (DNA-alkylating agent) treatment in lymphomas ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT01094860](https://clinicaltrials.gov/ct2/show/study/NCT01094860)—recruiting patients and [NCT00981799](https://clinicaltrials.gov/ct2/show/study/NCT00981799)—terminated status for Phase I/II in 2015).

**6.2.2. Nelarabine Synthesis.**—An early synthesis of nelarabine **13**, shown in Scheme 51,<sup>598</sup> goes through a key enzyme catalyzed trans-glycosylation between 1- $\beta$ -D-arabinosyl uracil **231** and 2-amino-6-methoxy-9H-purine **282**. The reaction uses uridine phosphorylase and purine nucleoside phosphorylase and is completely regio- and stereoselective, furnishing only the  $N^9$ ,  $\beta$ -anomer nelarabine **13** (Scheme 51).

A more recent process synthesis for nelarabine **13** was reported by Zong *et al.* (Scheme 52).<sup>599</sup> Glycosylation of 2-amino-6-chloropurine **61** with 2,3,5 tri-*O*-benzyl-arabifuranyl **283 $\beta$**  (see Section 6.2.4 for preparation) followed by methoxy displacement of the chlorine at position 6 of the purine base furnished benzylated intermediate **285**.<sup>600,601</sup> Debonylation by catalytic transfer hydrogenation, using ammonium formate as the *in situ* hydrogen source, afforded nelarabine **13** on a 30-g-scale at 99.8% purity after HPLC.

**6.2.3. Fludarabine and Fludarabine-5'-monophosphate Biologies.**—

Arabinofuranosyladenine (ara-A; vidarabine) has been around since the 1960s and was

originally studied for its antiviral effects.<sup>602</sup> The agent was also tested for anticancer effects, but showed limited success due to its rapid degradation by adenosine deaminase. In order to avoid adenosine deaminase-mediated deactivation, a fluorine was installed at the 2-position of the purine base of ara-A, producing fludarabine **291**.<sup>603</sup> This agent, however, had limited solubility and difficulties with its formulation. Therefore, the fludarabine-5'-monophosphate prodrug **7** (Fludara or fludarabine phosphate) was subsequently explored in clinical trials, which led to its FDA approval in 1991. Fludarabine-5'-monophosphate **7** is a commonly used drug against chronic lymphoid leukemia and hairy cell leukemia.<sup>538</sup>

The negatively charged fludarabine-5'-monophosphate **7** is dephosphorylated in the plasma by ecto-5'-nucleotidase, and fludarabine **291** is transported into cells by nucleoside transporters hENT1, hENT2, and hCNT3.<sup>429,592</sup> Although fludarabine-5'-monophosphate is adenosine deaminase resistant,<sup>604</sup> the drug is susceptible to glycosidic bond cleavage, which results in the formation of 2-fluoroadenine, a precursor to the toxic 2-fluoroadenosine-5'-triphosphate (see section 6.1.7).<sup>605</sup> After entering the cell, fludarabine is initially phosphorylated by cellular 2'-deoxycytidine kinase, although it is a relatively poor substrate for the enzyme.<sup>606,607</sup> Deoxyguanosine kinase has also been reported to phosphorylate fludarabine.<sup>608</sup> Fludarabine-5'-monophosphate **7** is then phosphorylated to the fludarabine-5'-diphosphate and fludarabine-5'-triphosphate by adenylate kinase and nucleoside diphosphate kinase, respectively, in the cell.

Fludarabine-5'-triphosphate is the active cytotoxic agent. It is a ribonucleotide reductase inhibitor, which leads to a decrease in cellular dNTP concentrations, and therefore negatively influences DNA synthesis. As the concentrations of dNTP diminish during cell division, this results in more fludarabine-5'-triphosphate being incorporated into the host DNA (self-potential).<sup>572,592</sup> Furthermore, the incorporated fludarabine-5'-monophosphate prevents further DNA chain elongation by restricting the addition of more dNTPs, leading to effective chain termination.<sup>603</sup> Moreover, fludarabine-5'-triphosphate can inhibit DNA polymerase, DNA primase, and DNA ligase, causing DNA strand breaks, which may lead to apoptosis.<sup>592,609</sup> Fludarabine-5'-triphosphate was also shown to be incorporated into RNA and can inhibit RNA synthesis.<sup>312,610</sup> In addition, fludarabine **291** exposure can also suppress the excision repair enzyme ERCC1 and led to cytotoxic synergy with SJG-136 (DNA minor groove binding agent) in chronic lymphocytic leukemia cells.<sup>611</sup> However, the side effects of fludarabine-5'-monophosphate **7** treatment are becoming an increasing concern, because secondary myelodysplastic syndrome and leukemia have been reported in at least 3% of persons treated with fludarabine-5'-monophosphate **7** (Fludara).<sup>609</sup>

Several different mechanisms have been reported that lead to fludarabine resistance *in vivo*. Changes in the expressions of miRNA-29a, miRNA-181a, and miRNA-221 have been reported.<sup>612</sup> The amplification of *MYC* gene and its transcript level miRNAs can promote fludarabine resistance.<sup>613</sup> Differential overexpression of sulfatase was correlated to fludarabine resistance.<sup>613</sup>

Fludarabine-5'-monophosphate **7**, cyclophosphamide (DNA-alkylating agent), and rituximab (monoclonal antibody against B cell protein CD20) combination therapy is

generally considered the standard treatment for younger fit patients with chronic lymphocytic leukemia.<sup>614</sup> A combination treatment of fludarabine-5'-monophosphate **7**, bendamustine (DNA-alkylating agent), and rituximab for chronic lymphocytic leukemia is being investigated, in addition to the fludarabine-5'-monophosphate **7**, clofarabine **12**, SAHA (vorinostat, histone deacetylase inhibitor), and busulfan (antineoplastic alkylating agent) combination for acute leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT01096992](https://clinicaltrials.gov/ct2/show/study/NCT01096992)–active status–not recruiting patients yet, and [NCT02083250](https://clinicaltrials.gov/ct2/show/study/NCT02083250)–recruiting patients).

**6.2.4. Fludarabine and Fludarabine-5'-Monophosphate Syntheses.**—In 1969, Montgomery and Hewson reported the discovery synthesis of fludarabine **291** (Scheme 53).<sup>615</sup> Key chlorosugar **283** was prepared from D-ribose by first cyclization in methanolic sulfuric acid to methyl- $\beta$ -D-ribofuranoside, followed by 3',5'-O-benylation, and subsequent hydrolysis, *p*-nitrobenzylation, and chlorination of the anomeric position.<sup>616,617</sup> The glycosylation step between chlorosugar **283** and 2,6-dichloropurine **286** gave a column-chromatography-separable mixture of  $\alpha/\beta$  diastereomers, with the desired  $\beta$ -anomer **287** formed in only 11%. Treatment with sodium azide followed by catalytic reduction furnished the diaminopurine intermediate **289**. Due to solubility issues of **289**, a mixed solvent system was employed for the modified Balz–Schiemann reaction, which introduced the  $N^2$ -fluorine. Debenzylation concluded the synthesis, producing fludarabine **291**.

Montgomery later reported a gram-scale preparation of fludarabine **291**, which utilized the same basic route, although the glycosylation reaction was performed using a silylated or acylated derivative of 2,6-diaminopurine.<sup>618</sup> Soon thereafter, he reported a gram-scale preparation of fludarabine-5'-monophosphate **7** by treatment of fludarabine **291** with phosphoryl chloride and triethyl phosphate (Scheme 53).<sup>619</sup>

Blumbers *et al.* developed a kilogram-scale preparation of fludarabine-5'-monophosphate **7**, which followed a similar route used by Montgomery (Scheme 54).<sup>620,621</sup> Yields were improved, allowing the synthesis of the drug in five steps from chlorosugar **283a** (12% yield). Improvements included the use of 2,6-di(methoxyacetamido)purine **293** during the glycosylation, which increased base solubility and greatly reduced the volume of solvent (ethylene dichloride) required for the reaction. Furthermore, chlorosugar **283a**, which was produced almost exclusively as the  $\alpha$ -anomer by treating the *p*-nitrophenol intermediate **294** with hydrogen chloride in dichloromethane,<sup>617</sup> was used in the glycosylation step, which allowed for higher yield of the desired  $\beta$ -anomer of **295**.

Another highlight of the synthesis involved the use of hydrochloric acid in the debenzylation step to avoid the partial-defluorination problem associated with Montgomery's catalytic hydrogenation, and to simplify the purification procedure. Furthermore, the phosphorylation step yield was significantly increased when using scrupulously dried nucleoside as the starting material. Recently, another publication effected the debenzylation by transfer hydrogenation, using ammonia formate as an *in situ* hydrogen donor, to produce kilograms of fludarabine **291** at 99.8% purity in 90–95% yield.<sup>622</sup>

Farina *et al.* prepared fludarabine **291** through enzymatic transglycosylation involving *Enterobacter aerogenes*, 2-fluoroadenine, and 9- $\beta$ -D-arabinofuranosyl-uracil (Scheme 55).



<sup>623</sup> The crude product was treated with acetic anhydride, forming 2',3',5'-tri-*O*-acetyl-9- $\beta$ -D-arabinofuranosyl-2-fluoroadenine, which was hydrolyzed and recrystallized to the pure fludarabine **291**. Subsequent phosphorylation afforded fludarabine-5'-monophosphate **7**.

### 6.3. Base Modified Purine Nucleosides

**6.3.1. 8-Chloro-adenosine and Tocladesine Biologies.**—In the 1980s, 8-chloro-adenosine **28** and tocladesine (**50**; 8-chloroadenosine 3',5'-monophosphate; 8-Cl-cAMP) were reported to have potent anticancer activities.<sup>624–627</sup> Tocladesine **50** is converted extracellularly to 8-chloro-adenosine **28**,<sup>628</sup> which is then transported by hCNT1 and hCNT2 into the cell.<sup>629</sup> Once in the cell, this agent is converted via adenosine kinase to 8-chloro-adenosine-5'-monophosphate, and ultimately to the active metabolite 8-chloro-adenosine-5'-triphosphate by cellular kinases. 8-Chloro-adenosine-5'-triphosphate accumulates to almost millimolar concentration in the cell.<sup>630</sup> 8-Chloroadenosine is a poor substrate for adenosine deaminase, and therefore, deamination of the agent is minimal.<sup>603</sup>

The 8-chloro-adenosine-5'-triphosphate affects cellular energetics by decreasing the levels of intracellular ATP, leading to RNA synthesis inhibition. Furthermore, RNA polymerase II incorporates 8-chloro-adenosine-5'-triphosphate into mRNA, and inhibits polyadenylation of full-length mRNA transcripts.<sup>631,632</sup> A decrease in these short-lived mRNA transcripts can also lead to a reduction in protein levels, which promote growth and survival of tumor cells. 8-Chloro-adenosine **28** exposure has been found to deplete the cyclin E level in breast cancer cell lines,<sup>624</sup> decrease the level of the receptor tyrosine kinase MET in multiple myeloma cell lines,<sup>633</sup> and depress the level of Mcl-1 in chronic lymphocytic leukemia cells.<sup>630</sup> Collectively, all of these mechanisms promote apoptosis.

Although many effects of 8-chloro-adenosine **28** are RNA-directed, 8-chloro-adenosine-5'-triphosphate has also been found to inhibit topoisomerase II and induce DNA double-stranded breaks in human myelocytic leukemia K562 cells.<sup>634</sup> 8-Chloro-adenosine **28** exposure also inhibits the rate of DNA synthesis and decreases dATP pools in mantle cell lymphoma cell lines.<sup>635</sup> Furthermore, decrease in cancer cell proliferation by 8-chloro-adenosine **28** exposure has been linked to AMP-activated protein kinase and p38 mitogen-activated protein kinase inhibitions *in vitro*.<sup>636,637</sup>

Glucose-6-phosphate dehydrogenase has been linked to 8-chloro-adenosine **28** resistance in multiple myeloma and chronic lymphocytic leukemia cell lines.<sup>638</sup> Also, a report indicated that the agent influences glucose consumption, leading to autophagy activation in multiple myeloma cells.<sup>639</sup>

In 2008, the M.D. Anderson Cancer Center sponsored a Phase I clinical trial of a 8-chloro-adenosine **28** therapy to study dosing tolerance in subjects with chronic lymphocytic leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00714103) identifier: [NCT00714103](https://clinicaltrials.gov/ct2/show/study/NCT00714103)—active status—not recruiting patients yet). In June 2015, the City of Hope Medical Center began a Phase I/II clinical trial evaluating 8-chloro-adenosine **28** in treating patients with relapsed or refractory acute myeloid leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00004902) identifier: [NCT00004902](https://clinicaltrials.gov/ct2/show/study/NCT00004902)—currently recruiting participants). Tocladesine **50** was evaluated in two clinical trials. In 2000, it was evaluated in a Phase II clinical trial for the treatment of persons with recurrent or refractory multiple

myeloma ([ClinicalTrials.gov](https://clinicaltrials.gov) identifier: [NCT00004902](https://clinicaltrials.gov/ct2/show/study/NCT00004902)—completed status). In 2001 the agent was studied in a Phase I trial for the treatment of progressive metastatic colorectal cancer ([ClinicalTrials.gov](https://clinicaltrials.gov) identifier [NCT00021268](https://clinicaltrials.gov/ct2/show/study/NCT00021268)—unknown status—greater than 2 years since updated). No trial results have been reported.

**6.3.2. 8-Chloroadenosine and Tocladesine Syntheses.**—Synthesis of 8-chloroadenosine **28** has been achieved in one step from adenosine **240** by using either tetrabutylammonium iodotetrachloride as the chlorinating reagent<sup>640</sup> or *m*-CPBA and acid (HCl or acetyl chloride) in aprotic solvents (Scheme 56).<sup>641,642</sup> Additionally, the synthesis of 8-chloroadenosine **28** has been achieved by direct coupling of 8-chloroadenosine **28** with 2,3,5-tri-*O*-benzoyl-D-ribofuranosyl bromide. However, the major product of the reaction is the *N*<sup>3</sup> glycosylated product and not the desired *N*<sup>9</sup> product.<sup>643</sup> Tocladesine **50** can be prepared by treatment of 8-chloroadenosine **28** with phosphoryl chloride in triethyl phosphate.<sup>642</sup>

**6.3.3. Forodesine Hydrochloride Biology.**—Forodesine hydrochloride (**36**, (1*S*)-1-(9-deazaguanin-9-yl)-1,4-dideoxy-1-D-4-imino-D-ribitol; BCX-1777; Immucillin H) was developed as a purine nucleoside phosphorylase inhibitor (see Section 6.2.1). Transition state analysis of the purine nucleoside phosphorylase-catalyzed phosphorolysis of inosine was used as a model system in the design of the agent (Figure 16).<sup>33,34</sup> The protonated nitrogen of the iminoribitol ring of forodesine exhibits a similar charge distribution to the ribosyl moiety oxocarbenium ion of the transition state. Furthermore, the 9-deazapurine of forodesine can be protonated at N7, which provides additional interaction with a residue of the purine nucleoside phosphorylase active site.<sup>644</sup> In addition, instead of a natural glycosidic bond, forodesine has a carbon–carbon bond which is not readily cleaved by purine nucleoside phosphorylase.<sup>630</sup>

Forodesine was shown to be a potent purine nucleoside phosphorylase inhibitor. Complete inhibition of the homotrimeric complex of the enzyme at a concentration as low as 72 pM has been reported, with the agent exhibiting a high equilibrium binding constant for and a very slow dissociation rate from the enzyme.<sup>337,630</sup> Inhibition occurs at a 1:1 molar ratio of inhibitor to purine nucleoside phosphorylase homotrimeric enzyme complex, with just one of the three substrate sites needing to be occupied by the agent to block enzyme activity.<sup>34</sup>

Forodesine hydrochloride **36** is administered by intravenous infusion. Like pentostatin **6** (see Section 6.1.5), the agent is active without metabolism, meaning it does not require phosphorylation. It blocks purine nucleoside phosphorylase, causing a high plasma concentration of deoxyguanosine, and it is not incorporated into DNA.<sup>645</sup> Deoxyguanosine is further converted to dGMP, dGDP, and dGTP by cellular kinases. High cellular levels of dGDP inhibit the ribonucleotide reductase complex and lead to imbalances of the nucleotide pool concentrations, resulting in p53-based apoptosis.<sup>591</sup> Because of this, forodesine works best in malignant cells containing high deoxyguanosine activity.<sup>646</sup>

Forodesine hydrochloride **36** has undergone clinical trials for the treatment of T-cell malignancies, including T-cell acute lymphocytic leukemia.<sup>35,647</sup> Additionally, the agent is being examined for treatment of chronic lymphocytic leukemia B cells, which may require

an additional nucleoside treatment for effectiveness.<sup>35,648</sup> Forodesine hydrochloride **36** did not inhibit the growth of colon cancer cell lines or normal human nonstimulated T cells, indicating some specificity of the drug toward normal versus malignant tissues.<sup>33</sup> In 2007 a phase IIb clinical trial of forodesine hydrochloride **36** in patients with leukemia/lymphoma was terminated due to manufacturing issues ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00419081) identifier: NCT00419081–terminated), indicating a need to improve the synthesis process in order to produce sufficient quantities of the agent for further clinical trials. In January 2013, Mundipharma began a Phase I/II clinical trial evaluating forodesine hydrochloride **36** in recurrent/refractory peripheral T cell lymphoma subjects in Japan ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01776411) identifier NCT01776411–active status–not recruiting patients).

**6.3.4. Forodesine Synthesis.**—The discovery synthesis of forodesine involved formation of the key iminoribitol intermediate **306** (see Scheme 57) followed by a 16-step linear construction of the base, producing only milligram quantities of the compound.<sup>649</sup> Therefore, a different route was designed, which could produce the required kilogram quantities of the agent needed for drug development. This new synthesis involved the coupling of imine **307** and brominated 9-deazahypoxanthine **313** (Scheme 57). One metric ton of **306** was prepared from D-ribose **213** by first bromine oxidation, protection, lactone epimerization,<sup>650</sup> and Fleet chemistry (five steps).<sup>651</sup> The same quantity of 9-deazahypoxanthine **310** was produced by condensing (ethoxymethylene)cianoacetate **308** with diethyl aminomalonate followed by cyclization and acid-mediated decarboxylation.<sup>652</sup> Three more steps were required to produce the bromo hypoxanthine derivative **312**: chlorination, protection and methoxy displacement, and finally bromination. At this stage, the condensation of lithiated heterocycle **313** (generated *in-situ* from **312**) with imine **307** afforded the  $\beta$ -anomer of **314** exclusively, due to steric hindrance of the  $\alpha$ -face by the isopropylidene protecting group. In order to avoid a hard-to-remove tarry byproduct, **314** was not converted directly to the product, but instead passed through a process of protection and sequential deprotection, eventually furnishing pure forodesine hydrochloride **36**.<sup>653</sup>

In order to avoid some of the cumbersome manipulations from the above-mentioned route, a different approach to forodesine hydrochloride **36** explored a novel route to prepare the aza-sugar core (Scheme 58).<sup>654</sup> Lactam **317**, which is prepared by esterification and reduction of L-pyrogutamic acid, was transformed into unsaturated bicycle **318** by treatment with benzaldehyde and catalytic acid followed by treatment with Meyer's reagent (methyl phenolsulfonate).<sup>655</sup> Osmium tetr-oxide-catalyzed dihydroxylation of compound **318** gave the desired diol stereochemistry, and subsequent acetal protection furnished the intermediate **319**. The coupling of lithiated 9-deazahypoxanthine to lactam **319** led to the ring-opened product **320**. Treatment of the intermediate **320** with BBr<sub>3</sub> gave the cyclized product **321**, which was then reduced with BH<sub>3</sub>·Me<sub>2</sub>S and deprotected under acidic condition to give forodesine hydrochloride **36** as a 4:1  $\beta$ : $\alpha$  mixture.

In 2016, a 50-g-scale synthesis of forodesine hydrochloride **36** was reported.<sup>656</sup> Differences from the original preparation (Scheme 57) include coupling of imine **307** with lithiated heterocycle **313** under milder condition, and the use of concentrated HCl instead of hydrogen for BOM removal in **315**.

## 7. ENZYME INHIBITOR AND CELL SURFACE RECEPTOR ANTAGONIST NUCLEOSIDE ANALOGS

### 7.1. Triciribine and Triciribine-5'-Monophosphate Biologies

In 1971, Schram and Townsend synthesized triciribine **30**, a tricyclic derivative of a purine nucleoside as a potential anticancer drug,<sup>657</sup> which has also been studied as an antiretroviral agent against HIV-1 and HIV-2.<sup>658</sup> Triciribine **30**, however, is poorly soluble, and this suboptimal property led to the development of the highly soluble prodrug triciribine-5'-monophosphate **33**.

Interestingly, triciribine **30** does not work by blocking host DNA or viral polymerases. It inhibits the phosphorylation of Akt1, Akt2, and Akt3,<sup>659</sup> and appears to influence the PI3K/Akt/mTOR cell survival pathway. The PI3K/Akt/mTOR pathway regulates a variety of cellular pathways, including cell proliferation and survival.<sup>660,661</sup> Akt is dysregulated in a variety of malignancies.<sup>662</sup> Initially, triciribine **30** was reported to inhibit *de novo* purine nucleotide and DNA synthesis in cell lines.<sup>425,663</sup> However, this would not explain the antiretroviral activity in HIV-1 and HIV-2 infected cell lines, because the reverse transcription step for proviral DNA synthesis of the viral life cycle was completed. Moreover, viral replication was no longer influenced by cellular dNTP concentrations or nucleoside reverse transcriptase inhibitors, which are chain terminators. Based on current data, the antiviral and anticancer activities of triciribine metabolite appear to be associated with Akt inhibition (Figure 17).<sup>664</sup> In addition to breast cancer, triciribine **30** may be effective when used as a combination therapy in bladder cancer patients prescreened for alterations in the Akt pathway.<sup>665,666</sup>

Triciribine-5'-monophosphate **33** is currently being evaluated in solid tumors, such as breast cancer.<sup>662,667,668</sup> A Phase I/II study is currently underway in patients with metastatic and locally advanced breast cancer that uses triciribine-5'-monophosphate monohydrate with paclitaxel (antimitotic taxane), which is followed by dose-dense doxorubicin (DNA intercalating agent) and the DNA alkylating-agent cyclophosphamide ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01697293) identifier: [NCT01697293](https://clinicaltrials.gov/ct2/show/study/NCT01697293)—recruiting patients). The H. Lee Moffitt Cancer Center and Research Institute recently began a Phase I/II clinical trial for the treatment of ovarian cancer with a combination of triciribine **30** and the DNA alkylating agent carboplatin ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01690468) identifier: [NCT01690468](https://clinicaltrials.gov/ct2/show/study/NCT01690468)—recruiting patients).

### 7.2. Triciribine and Triciribine-5'-Monophosphate Syntheses

Triciribine **30**<sup>657</sup> and triciribine-5'-monophosphate **33**<sup>669</sup> were originally prepared from the intermediate **322** by Townsend and Schram (Scheme 59). Aryl chloride displacement of **322** followed by tricycle formation led to triciribine **30**, and subsequent phosphorylation produced triciribine-5'-monophosphate **33**.

More recently Shen *et al.* developed a new approach to prepare triciribine **30** from commercially available intermediates in only 4 steps (Scheme 60).<sup>659</sup> Thus, reaction of 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose with 6-chloro-7-iodo-7-deazapurine **324** under classical glycosylation conditions gave intermediate **325**, which was then selectively

cyanated by treatment with tributyltin cyanide in the presence of catalytic Pd(0) to give compound **326**. Aryl chloride displacement with methyl hydrazine followed by deprotection and ring closure with sodium methoxide in refluxing methanol afforded triciribine **30** in a 37% overall yield.

### 7.3. MLN4924 Biology

Whereas the function of most FDA-approved nucleoside analogs involves their incorporation into DNA or RNA after phosphorylation, MLN4924 (**44**; Pevonedistat; TAK-924) suppresses tumor development mainly through blocking the proteasomal degradation pathway. MLN4924 **44** is a structural analog of adenosine-5'-monophosphate. MLN4924 **44** tightly binds to NEDD8-activating enzyme (NAE), which is an ubiquitin-like protein involved in the ubiquitin-proteasome degradation system (Figure 18).<sup>37</sup> An initial high throughput screening of adenosine-5'-monophosphate analogs identified *N*<sup>6</sup>-benzyl adenosine as an inhibitor of NAE, and further structural elaboration lead to MLN4924 **44**.<sup>37</sup>

Functionally, MLN4924 **44** was shown to selectively inhibit NAE, and did not inhibit transfer RNA synthetases or other ATP-consuming enzymes.<sup>37</sup> NAE is the enzyme that initiates the important pathway to maintain protein homeostasis, by forming a tight binding adduct with NEDD8.<sup>670,671</sup> This is a multistep enzymatic process in which Cullin-RING E3 ligases (CRLs) add the NEDD8 polypeptide onto target proteins.<sup>672</sup> MLN4924 **44** was shown to block polypeptide addition to target proteins. Furthermore, in cancer cells MLN4924 **44** blocks the important homeostatic balance between protein synthesis and degradation, which is essential for cell survival, cell division, and regulating both innate and adaptive immune responses.<sup>673-675</sup> As a result, CRLs are deactivated and consequently multiple known substrates of CRLs accumulate inside the target cell (e.g., DEPTOR and HIF-1 $\alpha$ ), leading to cell cycle arrest and apoptosis of the cancer cells (Figure 17).<sup>37,670,676</sup>

Cells in the S phase are more sensitive to exposure to MLN4924 **44**, a unique feature of the agent.<sup>37</sup> A recent study using the A375 human melanoma-derived cell line indicated that MLN4924 **44** induced cell cycle arrest might largely be triggered by a p21-mediated intra-S phase checkpoint.<sup>676</sup> Parallel to this study, an *in vivo* study with diffuse large B cell lymphoma models indicated that MLN4924 **44** might also suppress tumor growth via its inhibition of the NF- $\kappa$ B pathway.<sup>677</sup> Studies in mice demonstrated good toleration for MLN4924 **44**, as well as potent antiproliferative activity against human tumor cell xenografts derived from human colorectal and lung carcinomas.<sup>37</sup> However, HLC116 cells, a human colorectal carcinoma cell line, exposed to MLN4924 **44** developed a point mutation at A171T in the UBS3 subunit of NAE, which greatly decreased sensitivity to the drug.<sup>678</sup> Another study attempted to address a similar issue and found that both HIF1-REDD1-TSC1-mTORC1 and DEPTOR might play important roles in MLN4924-induced autophagy, which is a survival strategy in cancer cells in response to chemotherapy-mediated DNA damage.<sup>679</sup> This study suggested that cells exposed to an autophagy inhibitor may have a marked improvement in MLN4924 **44** response.<sup>679</sup>

MLN4924 hydrochloride affords better solubility than the parent compound, MLN4924 **44**. However, clinical studies indicate that MLN4924 **44** pevonedistat is being administered.

Currently the safety and efficacy of MLN4924 **44** therapy are being assessed in several Phase I clinical studies involving solid tumors, melanoma, acute myeloid leukemia, refractory or relapsed/refractory multiply myeloma, non-Hodgkin's lymphoma, Hodgkin's lymphoma, and cisplatin resistant ovarian cancers.<sup>680,681</sup> Current studies also include investigating fluconazole and itraconazole (antifungals often administered with cancer chemotherapeutics) effects on MLN4924 **44** safety and efficacy in solid tumors. Phase I trials using MLN4924 **44** in combination therapy with docetaxel (antimitotic taxane), gemcitabine hydrochloride **9**, or a combination of carboplatin (DNA alkylating agent) and paclitaxel (antimitotic taxane) in solid tumors are underway ([ClinicalTrials.gov](https://clinicaltrials.gov) identifiers: [NCT02122770](https://clinicaltrials.gov/ct2/show/study/NCT02122770)–recruiting patients, and [NCT01862328](https://clinicaltrials.gov/ct2/show/study/NCT01862328)–active status–not recruiting patients).

#### 7.4. MLN4924 Synthesis

A preparation of MLN4924 **44** is summarized in Scheme 61.<sup>682,683</sup> Starting from diene **217**, ring closing metathesis led to the tertiary  $\beta$ -alcohol product **328**, which had the required stereochemistry that allowed oxidative rearrangement to cyclopentenone **329**.<sup>682</sup> Stereoselective catalytic hydrogenation, due to hydrogen incorporation from the convex face of the intermediate, followed by Luche reduction with  $\text{CeCl}_3$ , provided **330** as a single diastereomer. Regioselective cleavage of the acetonide group of **330** was followed by treatment with  $\text{SOCl}_2$  and  $\text{RuCl}_3/\text{NaIO}_4$  to furnish the cyclic sulfate **331**. Subsequent condensation with *N*<sup>6</sup>-indanyl-7-deazaadenine in the presence of NaH afforded the nucleoside analogue **332**. Standard radical-mediated deoxygenation and O-TBDPS deprotection provided the intermediate **333**, which was then reacted with chlorosulfonamide and trifluoroacetic acid to furnish MLN4924 **44**.<sup>683</sup>

In 2007, Langston *et al.* published an alternative synthesis of MLN4924 **44** (Scheme 62).<sup>684</sup> The starting diol **334**,<sup>685</sup> formed in a four step sequence from cyclopentadiene and glycolic acid involving a hetero-Diels–Alder and oxidation/reduction reactions, was oxidized and treated with *p*-anisaldehyde to produce the cyclic epoxide **335**. Condensation with pyrrolopyrimidine **337** was followed by deoxygenation, deprotection, selective 5'-OH silylation, and 3'-OH acetylation. Finally, desilylation, sulfamate ester formation, and deacetylation led to MLN4924 **44**.

In 2015, a 100-g-scale process synthesis of MLN4924 hydrochloride **350** was reported (Scheme 63).<sup>686</sup> The key intermediate amino-diol **346** can be prepared from commercially available carboxylic acid **342** in four steps: bromolactonization, reductive ring opening, Boc deprotection, and hydrogenation. Subsequent coupling of **346** with acetaldehyde and (*S*)-(+)-1-aminoindane afforded diol **348**. Selective sulfamoylation of diol **348** was carried out using a Burgess-type reagent **349**, which was produced on a 100-kg-scale by reaction of chlorosulfonyl isocyanate with *tert*-butanol and subsequent treatment with DABCO. MLN4924 **44** was then separated from the disulfamoylated byproduct by stepwise hydrolysis with increasing concentration of HCl, requiring no chromatography. Finally, salt formation and crystallization from ethanol furnished MLN4924 hydrochloride **350**.



## 7.5. EPZ-004777 and EPZ-5676 Biologies

Histone methyl transferases (HMT) use *S*-adenosyl-L-methionine as a cofactor to methylate histone amino acid side chains, leading to post-translational gene modification. An important cellular HMT is DOT1-like histone H3K79 methyltransferase (DOT1L histone methyltransferase). It targets residue Lys79 of histone 3 (H3K79) for methylation, as illustrated in Figure 19.<sup>687</sup> The DOT1L histone methyltransferase is unique among HMTs for several reasons, making it an attractive therapeutic target.<sup>688</sup> The enzyme has recently been implicated as a therapeutic target for acute leukemias bearing rearrangements in the mixed lineage leukemia (MLL) gene.<sup>689,690</sup>

A translocation in the MLL gene occurs in 5–10% of acute leukemias and is especially common in childhood acute leukemias.<sup>691,692</sup> These cancers are particularly aggressive and have a relatively poor prognosis with a five-year survival of <40%.<sup>693</sup> In MLL-rearranged leukemia cells, DOT1L histone methyltransferase has been shown to physically bind to a complex of the rearranged MLL protein fused with sequences from genes such as AF4, AF9, AF10, and ENL. This interaction recruits DOT1L, leads to hypermethylation of H3K79 and aberrant expression of leukemogenic genes such as *Hoxa9*, *Hoxa7*, and *Meis1*.<sup>559</sup> Furthermore, preclinical models have shown that MLL-rearranged leukemias depend on abnormal DOT1L histone methyltransferase methylation of H3K79.<sup>39</sup>

Epizyme Inc. recently developed a potent and selective inhibitor of DOT1L histone methyltransferase. The compound, EPZ-004777 **351** (Figure 19) is a 7-deaza-5'-amino-alkylurea substituted analogue of adenosine. The agent inhibits *S*-adenosyl methionine in the enzyme-binding pocket. EPZ-004777 binding leads to a conformational change in DOT1L histone methyltransferase, which opens a hydrophobic pocket past the amino acid site and produces additional interaction surfaces.<sup>694</sup> Consequently, **351** selectively binds to DOT1L histone methyltransferase with high affinity. Furthermore, the agent was shown to selectively inhibit H3K79 methylation *in vitro* and inhibit the proliferation of MLL-rearranged cells. It also blocked MLL fusion target gene expression and led to prolonged survival in a mouse xenograft model of MLL leukemia.<sup>695</sup> However, the poor pharmacokinetic properties of EPZ-004777 **351** led to weak *in vivo* activity and precluded clinical development.

To overcome the suboptimal pharmacokinetic profile of **351**, Epizyme synthesized additional analogs. In 2013, the company reported promising preclinical data for EPZ-5676 **43**, which exhibited improved drug-like properties and potency.<sup>696</sup> Structural modifications were made to the 5'-side chain in order to decrease the number of hydrogen bonds and conformational freedom. EPZ-5676 **43** also occupies the *S*-adenosyl methionine binding pocket of DOT1L histone methyltransferase, and it was found to be a potent and selective inhibitor of the enzyme with a  $K_i$  of <0.08 nM, having >37,000-fold binding selectivity against 15 other human methyltransferases.<sup>696</sup>

EPZ-5676 **43** demonstrated selective inhibition of MLL-rearranged versus non-MLL-rearranged leukemia cells. In particular, an  $IC_{50}$  of 3.5 nM for the agent was demonstrated in the MV4-11 cell line following 14 days of exposure, with antiproliferative activity being most evident after 7 days.<sup>696</sup> This delayed onset could be explained by a progressive series mechanism involving a decrease of H3K79 methylation, lowered MLL-fusion target gene

expression, and inhibition of the expression of leukemogenic genes.<sup>696</sup> EPZ-5676 **43** produces a concentration-dependent decrease of both cellular methylated H3K79 levels as well as mRNA transcript levels of the MLL-fusion target genes HOXA9 and MEIS1. Continuous intravenous infusion of EPZ-5676 **43** (70 mg/kg) for 21 days produced a complete regression of tumors in a rat model of MLL-rearranged leukemia. Little or no tumor regrowth was detected 30 days after treatment cessation. Evaluation of subcutaneous administration of EPZ-5676 **43** has also been reported.<sup>697</sup>

However, EPZ-5676 **43** still exhibits suboptimal pharmacokinetics with a short plasma half-life and with fecal excretion being the main route of excretion.<sup>698,699</sup> A carbocyclic analogue of EPZ-5676 **43** was synthesized **352** in an attempt to produce increased metabolic stability, and this compound (**352**; see Figure 19) did indeed show increased stability in human plasma and liver microsomes while maintaining a selective DOT1L histone methyltransferase  $K_i$  of 1.1 nM.<sup>559</sup>

In September 2012, Epizyme began a phase I clinical trial of EPZ-5676 **43** for subjects with MLL-rearranged leukemia. Currently there are two ongoing Phase I clinical studies examining the agent for dose escalation and the first-in-human study for malignancies with rearrangement of the MLL gene (ClinicalTrials.gov identifiers: NCT02141828—competed in June 2016—no results posted, and NCT01684150—competed in February 2016—no results posted).

## 7.6. EPZ-5676 Synthesis

The synthesis of EPZ-5676 **43** is shown in Scheme 64.<sup>700,701</sup> Amine **353** is prepared from adenosine in four steps: 2',3'-*O*-isopropylidene protection, installation of a 5'-*O* leaving group, 5'-azide displacement, and azide reduction. Reductive amination involving 5'-amino **353** and methyl 3-oxocyclobutanecarboxylate produced a *cis/trans* mixture (separated using chiral HPLC) furnishing intermediate **354**. Subsequent N-alkylation, combined reduction, and Wittig olefination, alkene reduction, and saponification furnished intermediate **357**. A two-step process was used to form the benzoimidazole ring of **359**, without isolation of the intermediate amide **358**. Finally, acid-mediated deprotection completed the synthesis of EPZ-5676 **43**.

In 2014, a 700-g-scale process of EPZ-5676 trihydrate was reported (Scheme 65).<sup>702</sup> Cyclobutanone **366** can be prepared from **360** on kilogram-scale in a six-step sequence: protection, [2+2] cycloaddition, dechlorination, deprotection, acyl substitution, and iron-catalyzed reduction followed by condensation. **353** was subsequently converted to isopropyl amine **367**, which was further converted to **368** as a mixture of diastereomers. Deprotection, followed by two crystallizations from MeCN/H<sub>2</sub>O, afforded pure EPZ-5676 **43**.

## 7.7. CF102 Biology

In 2007, preclinical studies developed by Can-Fite Biopharma showed that CF102 (2-chloro-N6-(3-iodobenzyl)-adenosine-5'-N-methyl-uronamide) **42** had potent anticancer activity in hepatocellular carcinoma,<sup>703</sup> activity against hepatitis B virus, and anti-inflammatory activity (demonstrated in pre-clinical animal models of liver inflammation).<sup>704</sup>

CF102 **42** is an antagonist for the A3 adenosine receptor. Hepatocellular carcinoma cells treated with CF102 had Wnt and NF- $\kappa$ B signaling pathways deregulation, which promoted apoptosis.<sup>703</sup> Preclinical studies in rats with N1S1 hepatocellular carcinoma tumors, which highly express the A3 adenosine receptor, showed that CF102 **42** encouraged tumor apoptosis.<sup>703</sup> Furthermore, A3 adenosine receptor expression appears to be inflammation induced.<sup>705,706</sup> The A3 adenosine receptor is also highly expressed on PBMCs in hepatocellular carcinoma patients and patients with rheumatoid arthritis, psoriasis, dry eye syndrome, and glaucoma, in addition to hepatocellular carcinoma and hepatitis.<sup>707,708</sup>

Currently, a Phase I/II study for dose escalation using CF102 **42** was reported in patients with unresectable hepatocellular carcinoma.<sup>709</sup> Overall study results showed that CF102 **42** was well tolerated and had favorable pharmacokinetic characteristics. A Phase II study examining the safety and efficacy of CF102 **42** is being planned ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02128958) identifier: [NCT02128958](https://clinicaltrials.gov/ct2/show/study/NCT02128958)—recruiting patients).

### 7.8. CF102 Synthesis

Kim *et al.* reported the discovery synthesis of CF102 **42** in 1994 (Scheme 66).<sup>710</sup> Protecting group manipulation of 1-*O*-methyl- $\beta$ -D-ribose **370** produced the intermediate **371**. Subsequent oxidation and esterification of the 5'-OH followed by methyl amide introduction, 2',3'-*O*-benzoylation, and 1'-acetylation produced intermediate **374**. Condensation of intermediate **374** with silylated *N*<sup>6</sup>-(3-iodobenzyl)-2-chloroadenine **376**, made by reacting 2,6-dichloropurine **375** with 3-iodobenzylamine, followed by debenzoylation produced CF102 **42** in milligram quantities in an overall yield of 1.7%.

Hou *et al.* recently reported an alternative preparation of CF102 **42** starting from the commercially available 1,2,3,5-tetra-*O*-acetyl- $\beta$ -D-ribofuranose **136 $\beta$** , which is easily made from D-ribose **213** (Scheme 67).<sup>711</sup> The synthesis performed the glycosylation before constructing the 5'-methyl amide and furnished the target compound on a gram-scale in 27% overall yield.

## 8. ANTIVIRAL NUCLEOSIDE ANALOGS BEING EXPLORED FOR CANCER TREATMENT

### 8.1. Valacyclovir Hydrochloride and Valganciclovir Hydrochloride Biologies

Valacyclovir hydrochloride **17** and valganciclovir hydrochloride **35** are L-valyl ester prodrugs of acyclovir and ganciclovir, respectively. They were FDA-approved in 2009 and 2010, for the treatments of cytomegalovirus and herpes infections, respectively. After oral administration, the parent drug is released *in vivo*.<sup>712</sup> Importantly the monophosphorylated agents occur by the viral-specific thymidine kinases, making them less toxic to uninfected cells. The monophosphorylated agents are further converted to their active triphosphate forms to block viral DNA synthesis.<sup>713</sup> Since the viral thymidine kinase does not exist in human cells, gene therapy involving an adenoviral vector (disabled virus) designed to express the herpes simplex virus thymidine kinase gene in combination with valacyclovir hydrochloride **17** or valganciclovir hydrochloride **35** may lead to targeted cancer treatment

(see Section 5.1.1.6).<sup>714</sup> Furthermore, bystander killing of cancer cells has been proposed to be one mechanism of action for ganciclovir and acyclovir.<sup>715,716</sup>

Several clinical studies are evaluating valacyclovir hydrochloride **17**. A Phase I trial is underway using combination therapy with adenovirus expressing viral-specific thymidine kinase (AD-VTK) and radiation for pediatric brain tumors ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00634231) identifier: [NCT00634231](https://clinicaltrials.gov/ct2/show/study/NCT00634231)—active status—not recruiting patients). Another Phase I trial is using intrapleural AD-VTK therapy with valacyclovir hydrochloride **17** in patients with malignant pleural effusion ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01997190) identifier: [NCT01997190](https://clinicaltrials.gov/ct2/show/study/NCT01997190)—active status—not recruiting patients). A Phase I/II trial of the **17** in combination with HSV-TK and brachytherapy for recurrent prostate cancer is ongoing ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01913106) identifier: [NCT01913106](https://clinicaltrials.gov/ct2/show/study/NCT01913106)—recruiting patients), as well as a Phase III trial with AD-VTK and valacyclovir hydrochloride **17** combination therapy for localized prostate cancer ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01436968) identifier: [NCT01436968](https://clinicaltrials.gov/ct2/show/study/NCT01436968)—recruiting patients). Currently, valganciclovir hydrochloride **35** is being used in a Phase IV trial for chronic lymphocytic leukemia, but the action of the agent is to inhibit cytomegalovirus (CMV) replication in an attempt to stop chronic lymphocytic leukemia propagation ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01552369) identifier: [NCT01552369](https://clinicaltrials.gov/ct2/show/study/NCT01552369)—recruiting patients). The agent has also been examined as an add-on treatment for controlling CMV in patients being treated for the malignant brain tumor glioblastoma. Overall, results showed that the addition of valacyclovir hydrochloride **17** to the standard treatment improved patient survival,<sup>717</sup> which has encouraged a discourse on the possibility that CMV infection influences the development of brain tumors.<sup>718,719</sup>

## 8.2. Valacyclovir and Valganciclovir Syntheses

Valganciclovir hydrochloride **35** has been prepared almost exclusively from its parent compound ganciclovir **387**, following the chemistry described in Scheme 68.<sup>720</sup> Epichlorohydrin **383** was converted to dibenzyl-protected **384**, which was subsequently subjected to chloromethylation by treatment with paraformaldehyde and HCl. Exposure of the resulting intermediate to potassium acetate in acetone afforded **385**. Acid catalyzed condensation of **385** with diacetylguanane furnished a mixture of  $N^7:N^9$  regioisomers, and the desired  $N^9$  isomer was crystallized from toluene. Debenzylation followed by deacetylation completed the synthesis of ganciclovir **387**, on a 250-g-scale.

In 2004, a 100-g-scale process synthesis of triacetyl-protected ganciclovir **389** was reported that significantly increased the yield of the base condensation step.<sup>721</sup> The reaction coupled diacetylguanane **60** and protected glycerol (**388** in the presence of TsOH (Scheme 69). This step occurs under thermodynamic conditions, with the desired  $N^9$  isomer being the thermodynamic product. Recycling of the undesired  $N^7$  isomer was possible, allowing for increased yields of the desired  $N^9$  isomer (overall 70%). Compound **389** can be deprotected to furnish ganciclovir **387**.

In 1998, Arzeno *et al.* reported a preparation of valganciclovir hydrochloride **35** from ganciclovir **387** (Scheme 70).<sup>722</sup> Treatment of **387** with trityl chloride and DMAP in DMF followed by coupling with Boc-L-valine furnished the intermediate **390**. Subsequent treatment with a TFA/TFE mixture, in order to remove the trityl groups, and hydrogenolysis

of the resulting Boc-protected intermediate with palladium on carbon in a methanol/HCl solution afforded kilograms of valganciclovir hydrochloride **35** with an HPLC purity of 98.4%.

Khanduri *et al.* reported a preparation of valganciclovir hydrochloride **35** directly from ganciclovir **387** (Scheme 71).<sup>723</sup> Treatment of ganciclovir **387** with Cbz-L-valine, DCC, and DMAP in DMSO led to a mixture of ganciclovir **387**, mono-L-valyl ester **391**, and bis-L-valyl ester **392**. Addition of water and ethyl acetate caused **387** and **391** to precipitate, while **392** was removed in the organic layer. Ganciclovir **387** was removed from the **387/391** mixture by treatment with TFA followed by extraction with dichloromethane. Mono-L-valyl ester **391** was then separated from the organic layer by recrystallization. Subsequent hydrogenolysis of **391** in acidic solution produced valganciclovir hydrochloride **35**, which was purified by recrystallization to give the target compound on a 60-g-scale in greater than 99% HPLC purity. Both ganciclovir **387** and bis-L-valyl ester **392** were recovered and recycled.

The discovery synthesis of valacyclovir hydrochloride **17** was reported in 1989 (Scheme 72). Acyclovir **393**, which is prepared in ways similar to those of ganciclovir **387**, was condensed directly with Cbz-L-valine.<sup>724</sup> Palladium-catalyzed deprotection of the intermediate followed by treatment with aqueous HCl produced **17**. An alternative preparation used Boc-L-valine instead of Cbz-L-valine, furnishing the compound on a 50-g-scale.<sup>725</sup>

An alternative route utilized azido-intermediate **395** to avoid protection–deprotection steps that increase risks of racemization (Scheme 72).<sup>726</sup> Sodium azide **394** was treated with sulfuryl chloride and imidazole followed by L-valine and copper sulfate, producing the intermediate **395**. Condensation of **395** with acyclovir followed by reduction with Rainey nickel furnished valacyclovir hydrochloride **17** in three steps.

### 8.3. Ribavirin Biology

Ribavirin (1- $\beta$ -D-ribofuranosyl-1,2,4-triazole-3-carboxamide) **40** was first discovered as an effective antiviral agent against influenza A and B viruses in the 1970s using cell lines and *in vivo* models.<sup>727,728</sup> It was FDA-approved 15 years ago for the combination treatment of hepatitis C virus.<sup>729</sup> The agent's antiviral activity is primarily attributed to ribavirin-5'-triphosphate, which can inhibit RNA polymerase.<sup>730</sup> Ribavirin-5'-monophosphate is also a potent inhibitor of inosine monophosphate dehydrogenase (IMPDH),<sup>731</sup> an enzyme involved in the rate-limiting step of purine *de novo* synthesis. It is an important factor in maintaining guanine nucleoside pools, and a virus is highly dependent on this process. IMPDH Type I is ubiquitously expressed in normal cells, whereas a high level of IMPDH Type II is also associated with many hematological cancers.<sup>732</sup> In addition, ribavirin **40** has a negative impact on eukaryotic translation initiation factor eIF4E activity, which is dysfunctional in 30% of cancers.<sup>733–735</sup> High levels of eIF4E have been correlated with clinical progression, increased angiogenesis, and poor prognosis in breast cancer,<sup>736,737</sup> and ribavirin **40** was shown to suppress growth in breast cell lines.<sup>738</sup>

Ribavirin **40** is currently being evaluated in a Phase I/II study for malignant solid tumors ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01309490) identifier: [NCT01309490](https://clinicaltrials.gov/ct2/show/study/NCT01309490)—unknown status—no updates for over 2 years)

and in a Phase I trial using combination therapy for head and neck cancer and squamous cell cancer ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01721525) identifier: NCT01721525–active status–not recruiting patients). Ribavirin **40** is also being examined in combination with an inhibitor of hedgehog proteins, which regulate cell growth, differentiation, and survival, with or without azacytidine **11** in acute myelogenous leukemia patients ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02073838) identifier: NCT02073838–recruiting patients). The pharmacodynamic effects of ribavirin **40** therapy are being investigated for patients with tonsil and/or base of tongue squamous cell carcinoma ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01268579) identifier: NCT01268579–active status–not recruiting patients). Ribavirin **40** is being evaluated in combination with decitabine **14** in high-risk acute myelogenous leukemia patients ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT02109744) identifier: NCT02109744–recruiting patients). Furthermore, a Phase I/II study of ribavirin **40** given as monotherapy in solid tumor cancer patients is ongoing ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01309490) identifier: NCT01309490–unknown status–no updates for over 2 years).

#### 8.4. Ribavirin Synthesis

Witkowski *et al.* reported the discovery synthesis of ribavirin **40** in 1972 (Scheme 73).<sup>727,739</sup> The compound was prepared by acid-catalyzed fusion of acylated sugar **136 $\beta$**  with 1,2,4-triazole-3-carboxylate **397**, which was prepared by reacting aminoguanidine bicarbonate with oxalic acid and subsequent esterification of the resulting 1,2,4-triazole-3-carboxylic acid. Treatment of methyl ester **398** with methanolic ammonia effected concomitant deprotection and amide formation, completing the synthesis.

In 2007, Banfi *et al.* reported a large-scale synthesis, which allowed for the preparation of ribavirin **40** on a 400-g-scale (Scheme 74).<sup>740</sup> Improvements over other preparations that allowed for scale up include the following: (1) a glycosylation involving direct coupling of triazole **397**, thereby avoiding the need for silylating reagents, (2) intermediates pure enough to be used directly in the following stages without difficult purifications, and (3) relatively mild reaction conditions.

Another recent preparation makes use of cheap nucleosides as starting materials.<sup>741</sup> Inosine, adenosine, and guanosine were cleaved to the tetra-acylated sugar **136 $\beta$**  and the corresponding acetylated purine base by treatment with acetic acid and TFA as catalyst. The glycosylation between **136 $\beta$**  and triazole **397** was catalyzed by TfOH, and after deprotection, ribavirin **40** was produced on a 30-g-scale.

Ribavirin **40** can also be synthesized enzymatically. It has been formed on gram-scale. A transglycosylation involving purine nucleoside phosphorylase, 7-methylguanosine and 1,2,4-triazole-3-carboxamide has been used to produce grams of the agent.<sup>742</sup> Furthermore, purine nucleoside phosphorylase has been used in conjunction with phosphopentomutase to couple ribose-5'-monophosphate **399** with the triazole **400**, forming ribavirin **40** (Scheme 75).<sup>743</sup>

#### 8.5. Acadesine Biology

Acadesine (AICA-Ribonucleotide (AICAR); 5-amino-1- $\beta$ -D-ribofuranosyl-imidazole-4-carboxamide, **41**) was previously under investigation for preventing ischemia-reperfusion injury in persons undergoing coronary artery bypass graft surgery.<sup>744</sup> The agent was found



to exert its cardio-protective effect via regulation of adenosine monophosphate activated protein kinase (AMPK) activity.<sup>745</sup> Acadesine-5'-monophosphate is a highly potent activator of AMPK, and it can bind to the allosteric site of the enzyme as an AMP mimic. In a recent study, this activation was found to exert anti-proliferative and apoptotic effects on acute lymphocytic leukemia cells by regulating different signaling pathways (Figure 17).<sup>746,747</sup> For this reason, cell-specific cytotoxicity of acadesine **41** might ascertain its future usage in anti-cancer therapy.

Preclinical trials with using acadesine **41** and rituximab showed synergistic anticancer activity in tissue culture and animal models.<sup>748</sup> Acadesine **41** therapy is currently being evaluated in a Phase I/II study for myelodysplastic syndromes and acute myelogenous leukemia ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT01813838) identifier: NCT01813838—trial status suspended in 2015) and a Phase I/II open label dose escalation study for safety and tolerability in B cell chronic lymphocytic leukemia patients ([ClinicalTrials.gov](https://clinicaltrials.gov/ct2/show/study/NCT00559624) identifier: NCT00559624—completed in 2010). Clinical trial results indicated that acadesine **41** had a manageable toxicity profile and might be a valuable agent for the treatment of relapsed/refractory CLL patients.<sup>749</sup>

### 8.6. Acadesine Synthesis

Acadesine **41** is now produced industrially from the fermentation broth of glucose,<sup>750–752</sup> but Baddiley *et al.* reported the first synthesis of the agent in 1959 (Scheme 76).<sup>753</sup> The key intermediate **403** was synthesized from D-ribose **213** in a five-step sequence involving cyclic acetal formation and benzylation, followed by anomeric bromination, acetylation, and chlorination.<sup>754</sup> In the presence of silver chloride, protected chlorosugar **403** was condensed with amide **405**, formed from 4-nitro-5-styryl glyoxaline **404** in three steps via a carboxylic acid and a methylester intermediate.<sup>755</sup> The resulting 1:1 mixture of regioisomers **406a** and **406b** was separated using descending paper chromatography. Hydrogenation of **406b** using Adams' catalyst afforded acadesine **41**.

Kohyama *et al.* reported a gram-scale synthesis of acadesine **41**, which features a hydrolysis of N-MEM protected inosine **408** by treatment with aqueous ammonia in methanol (Scheme 77).<sup>756</sup> The resulting deprotected triol was subjected to aqueous alkaline hydrolysis, furnishing the ring-opened product acadesine **41** in four steps from inosine.

## 9. CONCLUSIONS

Nucleoside analogs are used to treat a high percentage of clinical cancers, usually in combination with other agents. They interact with various biological targets, such as cellular kinases, ribonucleotide reductase, and cellular polymerases in order to produce cytotoxic effects. Overall, a majority of cancers treated with nucleoside analogs are haematological. Newer nucleoside analogs, however, such as RX-3117 **29**, tricyribine-5'-monophosphate **33**, NUC-1031 **32**, and NUC-3373 **37** are exhibiting promising activity against solid tumors in patients, but further testing is needed. With the exceptions of 6-MP **1** and cytarabine **4**, nucleoside analogs usually do not produce cures as standalone agents.

As is the case with most classes of compounds explored in the clinic, attrition rates for nucleoside analogs remains high. Continuing efforts are directed toward improving the

selectivity and targeting of anticancer nucleoside analog therapies. Nucleoside analogs interact with various biological targets, such as cell membrane transporters for cellular uptake, kinases to promote their phosphorylation, as well as DNA and RNA polymerases in order to produce their cytotoxic effects. Therefore, the tendency to increase chemical modifications to the parent nucleoside analog in hopes to design a better derivative should be considered with caution. Major efforts to improve analog biological efficacy currently focus on developing new prodrug strategies and using antimetabolites in combination treatments.

Prodrug strategies have emerged as promising approaches in the search for new anti-cancer nucleoside analogs. Next generation nucleoside analogs have been developed in an attempt to overcome cellular uptake deficiencies (e.g., elacytarabine **51**, SGI-110 **22**, and sapacitabine **25**) as well as overcome sensitivity to cytidine deamination and adenosine deamination (e.g., FdCyd **38**, MB 07133 **34**, fludarabine-5'-monophosphate **7**, clofarabine **12**). Compared to the parent compound, nucleoside analog prodrug candidates generally offer enhanced efficacy and less toxicity, with the most successful being capecitabine **10**.

Nucleoside analog prodrugs that have progressed to Phase III trials include elacytarabine **51**, SGI-110 **22**, and sapacitabine **25**. Elacytarabine **51** and CP-4126 **49** have stalled in development, with the former failing to achieve end point efficacy in a Phase III clinical trial involving elderly patients with acute myeloid leukemia,<sup>490</sup> indicating that 5'-elaidic acid ester conjugates are not successful clinical candidates. Elacytarabine **51** is still under clinical investigation, but questions remain about its clinical prospects as a monotherapy. The *N*<sup>4</sup>-alkylester moiety was not a successful approach for LY2333727 **52**, but it may be for sapacitabine **25**, which is still being investigated in a Phase III clinical trial as a monotherapy. Furthermore, in June 2016, Cyclacel Pharmaceuticals announced positive results from a Phase I clinical trial involving sapacitabine **25** with seliciclib (CDK inhibitor) treatment for advanced solid tumors. SGI-110 **22** is also currently being evaluated in a Phase III clinical trial in patients with AML. In addition, in May 2016 BioCryst Pharmaceuticals submitted regulatory paperwork to obtain forodesine hydrochloride **36** approval in Japan.

The phosphoramidate prodrug strategy has been successful in HCV treatment.<sup>757</sup> Currently, ProTide analogs NUC-3373 **37** and NUC-1031 **32** are still being evaluated in Phase I clinical trials. Furthermore, the phosphoramidate prodrugs of many of the nucleoside analogs in this review (e.g., cladribine **8**, fludarabine-5'-monophosphate **7**, clofarabine **12**, thiarabine **47**, troxacitabine **48**, etc.) have been claimed in patents, but no interesting biological effects have been reported.<sup>433,494</sup> Beyond prodrugs, parent nucleoside analogs such as L-nucleosides seem to have an uncertain future, given the poor performance of troxacitabine **48**, which produced multiple terminated clinical trials.

There are several other prodrug strategies being evaluated in preclinical studies. SGI-110 **22** employs a duplex linked drug of decitabine(3'-5')deoxyguanosine. Additional duplex drug linking of 5FdU(5'-5')TAS-106<sup>401</sup> and 5-FdU(3'-5')TAS-106<sup>402</sup> has been investigated *in vitro*. They may provide a means of delivering two nucleoside analogs at once to allow targeting of different cellular pathways. Preliminary data of the duplex drug linking approach seem promising, but further evidence is needed to show synergy and clinical efficacy. Other unique ways to modify nucleoside analogs being tested *in vitro* are

mesoporous silica nanoparticles,<sup>361</sup> 5'-mono-amino acid monoester nucleoside prodrugs,<sup>362</sup> and Hoechst nucleoside conjugates.<sup>363</sup>

A better understanding of cellular degradation pathways may greatly aid in maintaining high intracellular triphosphate forms of the anticancer nucleosides. SAMHD1 hydrolyzes dNTPs into 2'-deoxynucleoside and inorganic triphosphates.<sup>49,50</sup> The tight catalytic pocket of the enzyme excludes ribonucleoside-5'-triphosphates.<sup>758</sup> Clofarabine-5'-triphosphate was shown to be a substrate for SAMHD1 in a biochemical assay.<sup>51</sup> Therefore, we postulate that nucleoside analog triphosphates such as decitabine-5'-triphosphate, cladribine-5'-triphosphate, floxuridine-5'-triphosphate, ara-C-5'-triphosphate, ara-A-5'-triphosphate, ara-G-5'-triphosphate, fludarabine-5'-triphosphate, cladribine-5'-triphosphate, and clofarabine-5'-triphosphate may be sensitive to dNTP hydrolysis activity by SAMHD1 both *in vitro* and *in vivo*. Cotreatment of these nucleoside analogs with an effective inhibitor of SAMHD1, which still has not been developed, may increase the intracellular antimetabolite concentrations. Moreover, decreasing SAMHD1 protein levels have been shown to promote dNTP imbalance both *in vitro* and *in vivo*,<sup>759,760</sup> and inhibition of the dNTP hydrolytic activity of the enzyme might lead to higher mutation rates in cancer cells. An alternative approach has been suggested to increase SAMHD1 expression in cancer cells. Methylation of the promoter of SAMHD1 has been shown to decrease mRNA and protein levels in certain cancers,<sup>53,761,762</sup> and gene dosage of SAMHD1 has been shown to influence cellular dNTP levels in a mouse model.<sup>763</sup> The current concept is to increase the SAMHD1 protein level within cells in order to decrease cellular dNTPs, which, in turn, might inhibit cell proliferation.<sup>764,765</sup>

Continued investment in basic synthetic organic chemistry provides the ability to access novel nucleoside analogs on both discovery and production scales. For example, the Jamison group's syntheses of capecitabine **10** and doxifluridine **23** (see section 5.1.2.2) using continuous flow chemistry show great potential for the synthesis of nucleoside, nucleotide, and base analogs. Flow chemistry is widely used in academia and currently is attracting great interest in the pharmaceutical industry (e.g., establishment of the MIT-Novartis Center for Continuous Manufacture).<sup>766</sup> This burgeoning technology offers great potential for both microscale synthetic drug discovery efforts and macroscale production of active pharmaceutical agents. On the laboratory scale, the microreactor allows for rigorously controlled reaction conditions, leading to better heat and mass transfer, while also requiring much smaller reaction volumes.<sup>767</sup> This allows for much greater efficiency in comparison to batch systems—a reduction in reaction times (minutes versus hours), and a significant drop in waste (particularly solvent) production. On the macroscale, the significantly reduced reaction time has the potential to produce larger quantities of drug per unit time, in comparison to batch. Furthermore, the continuous flow process joins well with other enabling technologies (microwave irradiation, electrochemistry, 3D-printing, etc.), potentially leading to fully automated processes.<sup>768</sup> Although questions remain about anomeric selectivity in glycosylations involving 2'-deoxysugars as well as phosphoramidate prodrug synthesis, application of this technology to nucleoside analog drug discovery and drug production seems promising.

The recent FDA approval of TAS-102 (**15 + 16**) illustrates that possibilities still exist for anticancer nucleoside analogs. The wealth of information available for these analogs produced from decades of research offers a strong foundation for future pursuits, including prodrug and combination strategies. In addition, further development of novel nucleoside analog chemical entities as well as continued refinement of biological applications of current nucleoside analogs will continue to contribute positively to the fight against cancer.

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## Biographies

Jadd Shelton received his B.S. in Biology (Chemistry minor) from Brigham Young University—Idaho in 2008, where he worked in the lab of Jason Hunt, studying mink glycogen metabolism as a result of hormonal treatment. Also as an undergraduate, and NIH-funded Idaho INBRE fellow in the laboratory of Ron Strohmeyer at Northwest Nazarene University, he researched pharmacologic manipulation of microglial cell inflammatory gene expression related to Alzheimer's disease. In 2012, he received his Ph.D. in Chemistry from Brigham Young University, working in the lab of Matt Peterson, where his research focused on the synthesis of anticancer nucleoside analogs. From 2012 to 2015, Jadd worked as a postdoctoral researcher in the laboratory of Raymond F. Schinazi at Emory University, synthesizing and studying nucleoside analogs for antiviral applications. He currently works as an R&D chemist for Autoliv Inc. His research interests include the design and synthesis of energetic materials for pyrotechnic formulations.

Xiao Lu received his B.S. degree in Pharmaceutical Sciences in 2005 and his M.S. degree in Chemical Biology in 2008 from Peking University. He then earned his Ph.D. degree in Pharmaceutical and Biological Sciences from The University of Georgia in 2012. After finishing his doctoral studies, he joined Prof. Schinazi's Laboratory of Biochemical Pharmacology (LOBP) as a postdoctoral fellow at Emory University, working on medicinal chemistry for viral infectious diseases. He is currently a postdoctoral researcher at The National Center for Advancing Translational Sciences (NCATS), where he is contributing to developing drugs that block malaria transmission.

Joseph A. Hollenbaugh received his doctorate from the Department of Cellular and Molecular Biology at University of Vermont in 2006. He was trained as a cellular immunologist under the mentorship Richard Dutton (Trudeau Institute in Saranac Lake, NY) and examined CD8+ T cell responses to EG7OVA tumor in C57BL/6 mice. He did postdoctoral training in immunology and virology in the laboratories of David Topham and Baek Kim (University of Rochester). He is currently an Instructor in the Pediatrics Department, in the Laboratory of Biochemical Pharmacology at Emory University School of Medicine. His research interests include viral biology, cancer biology, and nucleoside biosynthesis regulation at the cellular level. His recent focus has been on the newly identified cellular protein SAMHD1, which degrades nucleoside triphosphates, in order to

understand their biological implications in diseases and to develop a nucleoside inhibitor for potential therapeutic applications.

Jong Hyun Cho was born in Jangyumyeon, Gimhae-si, South Korea, in 1967. In 2002, he received his Ph.D. in Organic Chemistry from Seoul National University, working on biologically active peptide mimetics under the direction of B. M. Kim. He then joined David C. K. Chu's group at the Department of Pharmaceutical and Biomedical Sciences, College of Pharmacy, University of Georgia, where he studied nucleoside chemistry and medicinal chemistry from 2003 to 2006. He is currently working as an instructor research fellow for Raymond F. Schinazi at Emory University. His research interests include synthetic approaches to nucleosides and peptidomimetics with antiviral activities against DNA and RNA viruses.

Franck Amblard was born in Châteauroux, France. He studied chemistry at the University of Orléans (France), where he received his Ph.D. in 2004 under the guidance of Luigi A. Agrofoglio, working on the synthesis of new nucleoside analogs using metathesis and palladium-catalyzed reactions. In 2005, he moved to the USA to join Raymond F. Schinazi's research group at Emory University (Atlanta, GA) and worked, as a postdoctoral fellow, on new nucleosides and nucleotides prodrugs. He is now Assistant Professor at the Department of Pediatrics, Emory University School of Medicine. His main research interests include the study of nucleosides analogs and small molecules as potential antiviral agents.

Raymond F. Schinazi is the Frances Winship Walters Professor of Pediatrics, Director of the Laboratory of Biochemical Pharmacology at Emory University, and Adjunct Professor at Georgia State University. He also serves as Director of the HIV Cure Scientific Working Group within the NIH-sponsored Emory University Center for AIDS Research. Dr. Schinazi is a world leader in the area of nucleoside chemistry and biology, having published more than 500 papers and having 100 US patents issued. He is the founder of several biotechnology companies focusing on antiviral drug discovery and development, including Pharmasset, Inc., Triangle Pharmaceuticals, Idenix Pharmaceuticals, and Cocrystal Pharmaceuticals, Inc. He is best known for his innovative and pioneering work on stavudine, lamivudine, emtricitabine, telbivudine, and sofosbuvir, all of which have been approved by the United States Food and Drug Administration and are widely used clinically. More than 94% of HIV-infected individuals take at least one of the antiretroviral agents he invented as a fixed dose combination.

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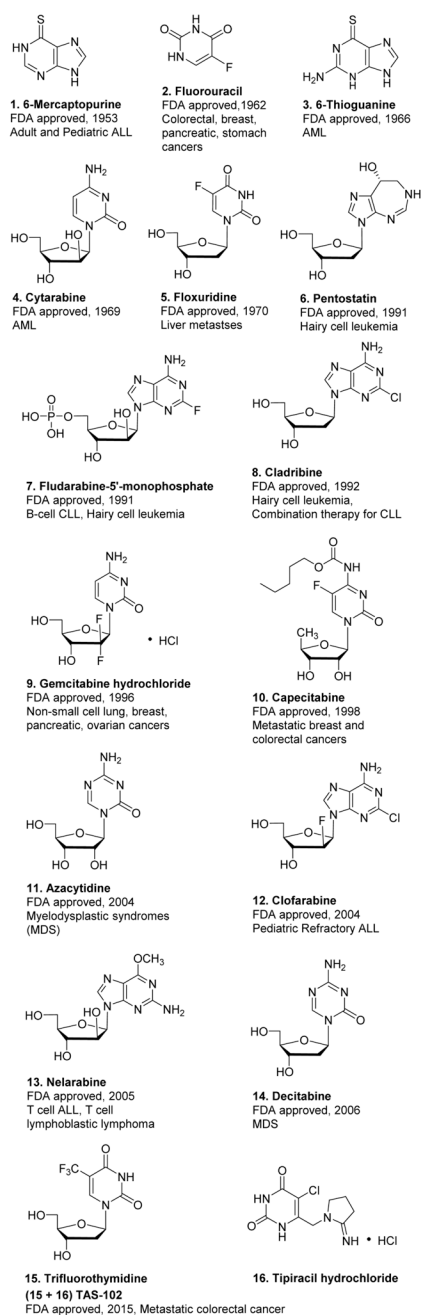


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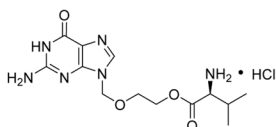
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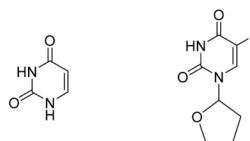
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**Figure 1.**  
FDA-approved anticancer nucleoside analogs.

**17. Valacyclovir hydrochloride**

Phase III, Combination therapy for localized prostate cancer  
Phase I/II, Combination therapy for recurrent prostate cancer  
Phase I, Combination therapy for pediatric brain tumors

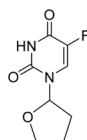
**18. Uracil**

(18 + 19) Tegafur-uracil

Used in Japan and Taiwan, Advanced GI cancers

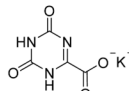
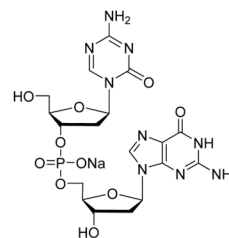
Phase I/III, Gastric cancer

Phase II, Colon cancer, combination therapy for hepatocellular carcinoma

**19. Tegafur**

(19 + 20 + 21) TS-1

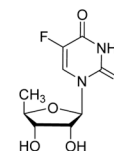
Phase III, Combination therapy with cisplatin for advanced gastric cancer  
Multiple Phase I/II, Combination therapy

**20. Gimeracil****21. Oteracil****22. SGI-110 (Dinucleotide of 14)**

Phase III, AML

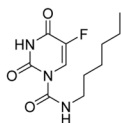
Phase I/II,

Metastatic colorectal and ovarian cancer, MDS, AML, HCC

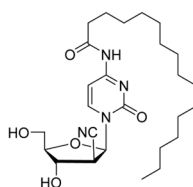
**23. Doxifluridine**

Phase III,

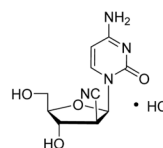
Combination therapy for GI cancer

**24. Carmofur**

Used in China, Japan and Finland  
Colorectal cancer

**25. Sapacitabine**

Phase III, newly diagnosed AML in elderly  
Multiple Phase I/II, AML, CLL, non-small cell lung cancers, solid tumors

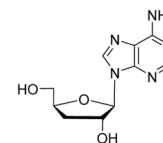


(HCl salt of CNDAC)

**26. DFP-10917**

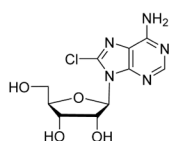
Phase I/II,

Relapsed or refractory AML

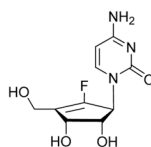
**27. Cordycepin**

Phase I/II,

Combination therapy for TdT-positive leukemia

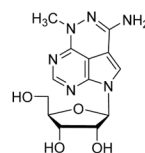
**28. 8-Chloro-adenosine**

Phase I,  
Chronic lymphocytic leukemia

**29. RX-3117**

Phase I/II, Solid tumor

Phase I,  
Advanced Malignancies

**30. Triciribine**

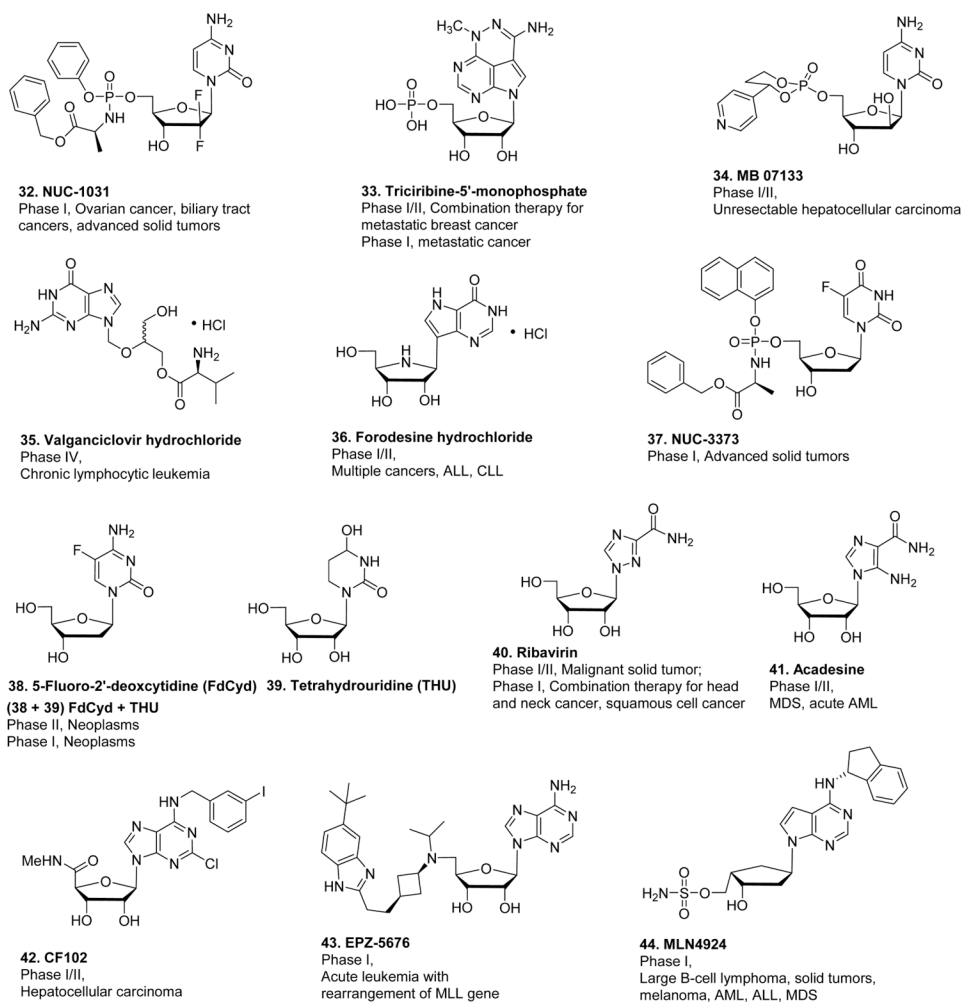
Phase I/II, ovarian cancer

Phase I,  
Combination therapy for ovarian cancer

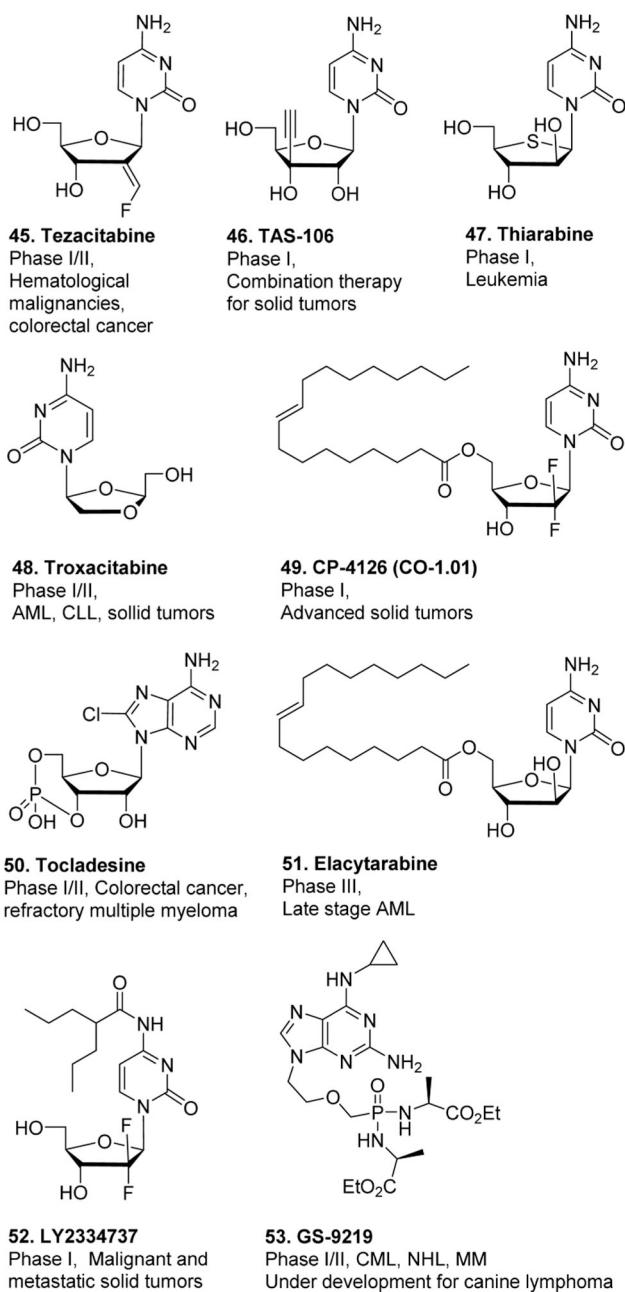
**31. Flucytosine**

Used in gene therapy,  
Phase I/II, Combination therapy for solid tumors  
Phase I, brain tumors

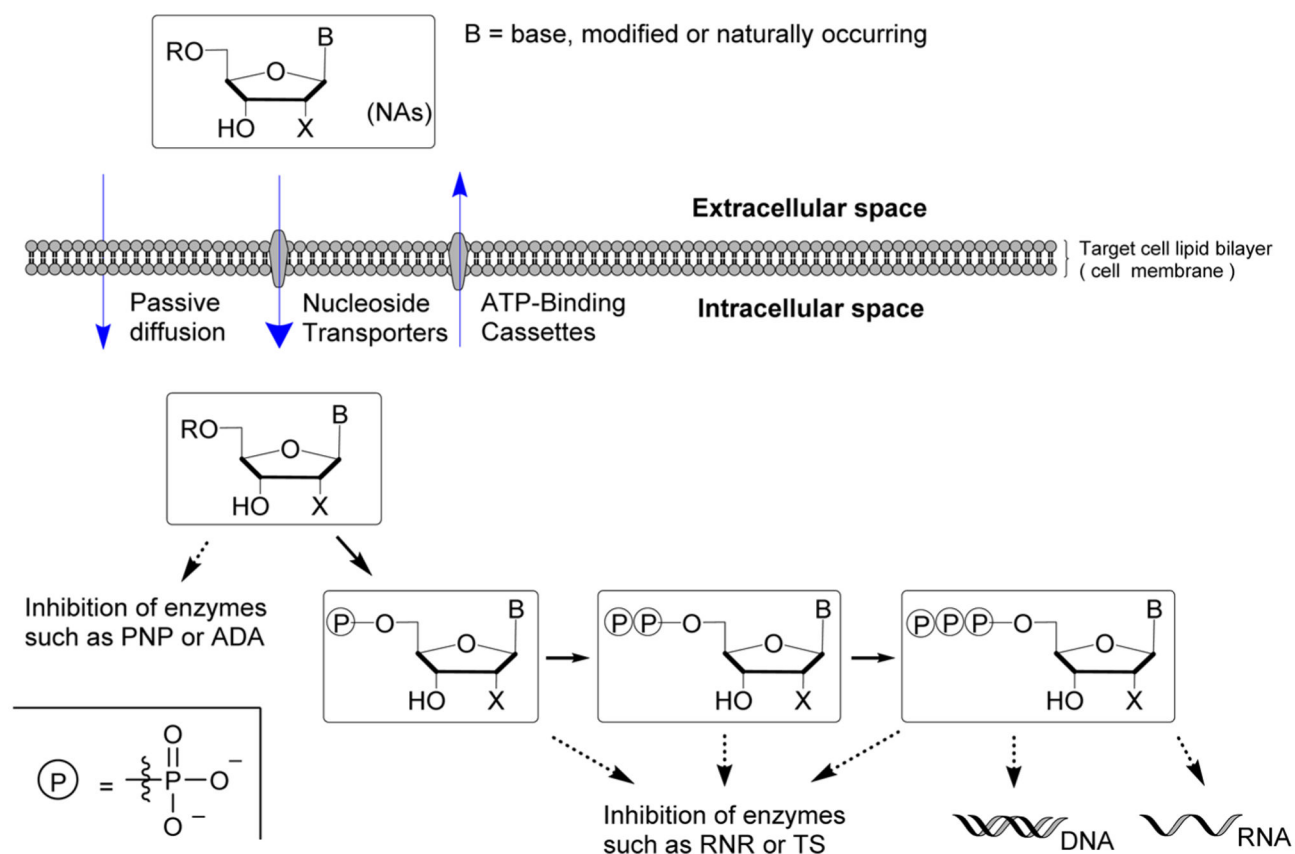




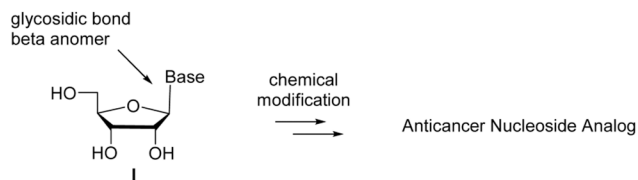
**Figure 2.**  
Various nucleoside analogs in clinical trials or used outside of the U.S.



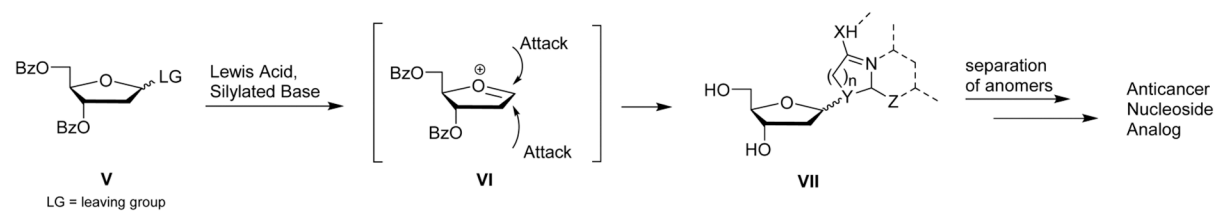
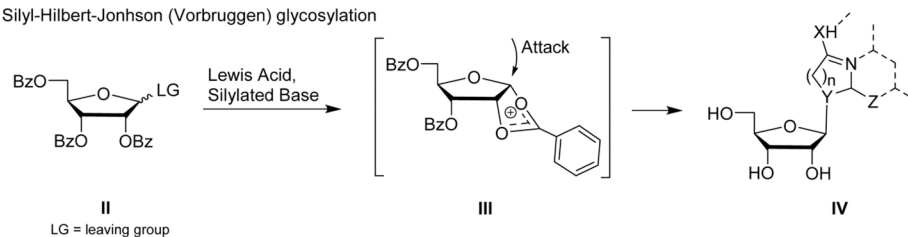
**Figure 3.**  
Various nucleoside analogs that have stalled in clinical trials.



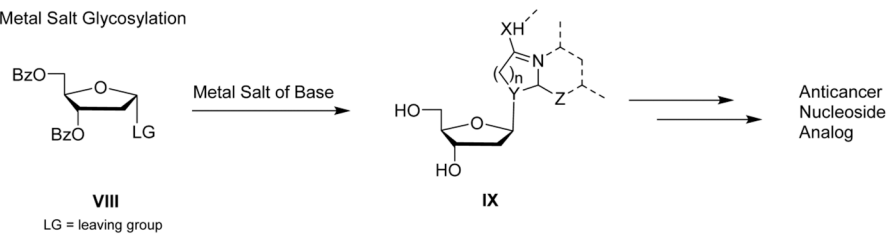
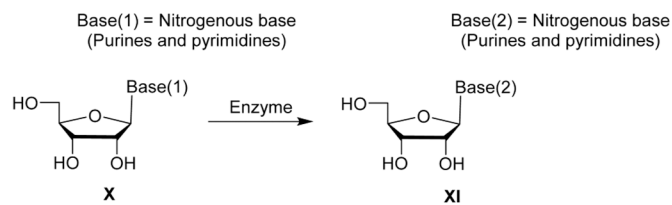
**Figure 4.**  
General biological mechanism of action of anticancer nucleoside analogs.

**Divergent Approach:****Convergent Approach:**

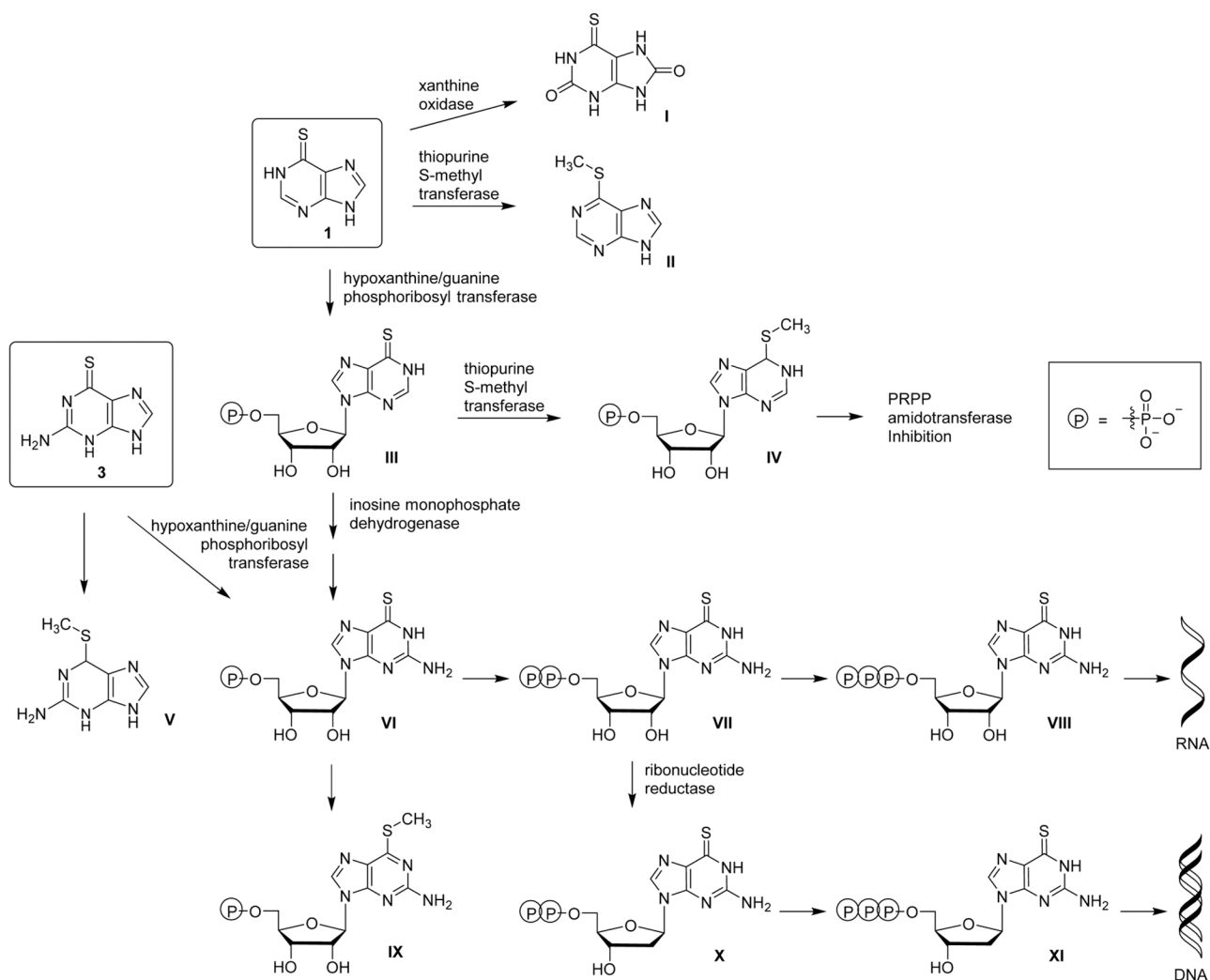
## Silyl-Hilbert-Johnson (Vorbruggen) glycosylation



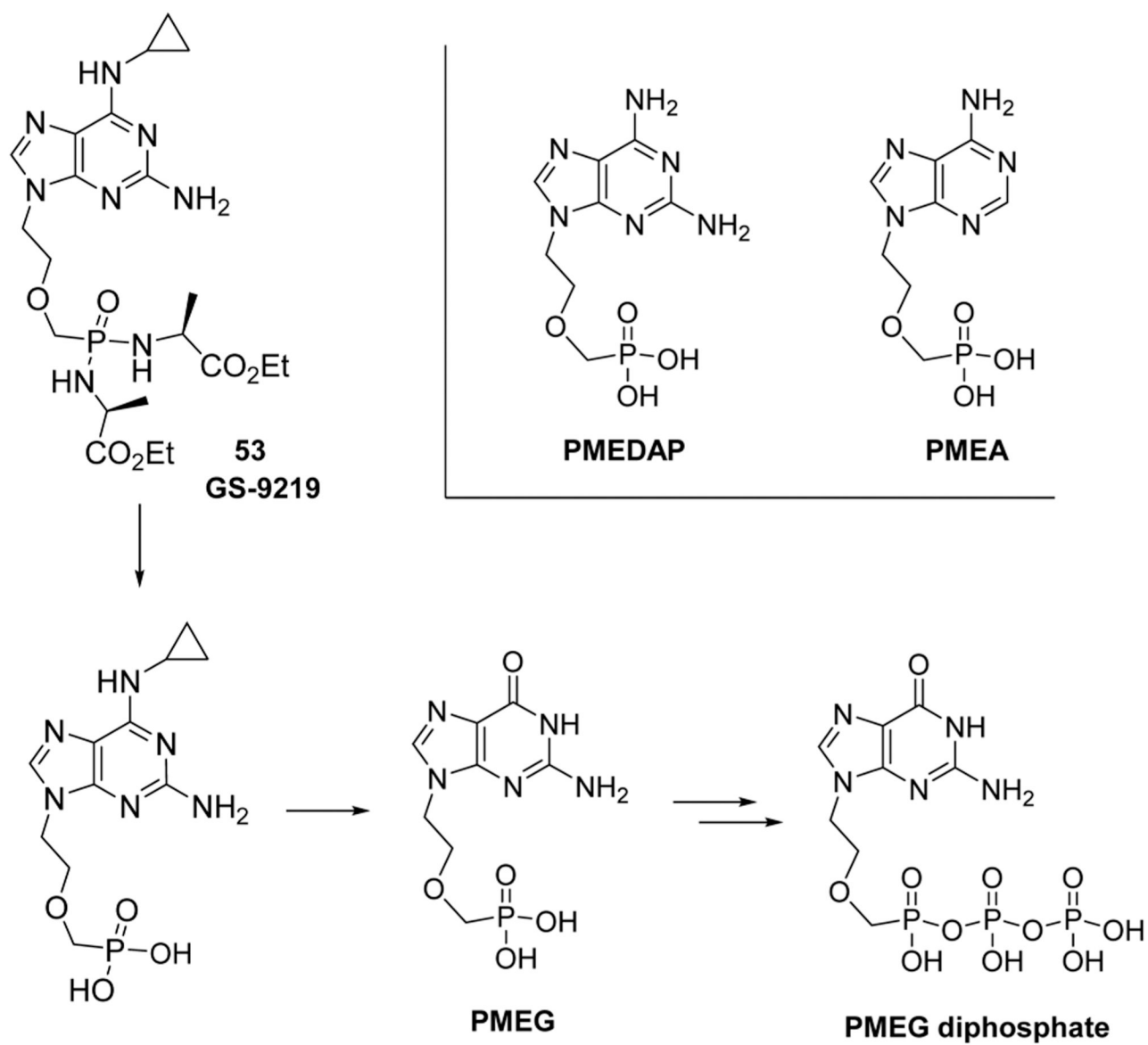
## Metal Salt Glycosylation

**Transglycosylation:**

**Figure 5.**  
General synthetic approaches to anticancer nucleoside analogs.

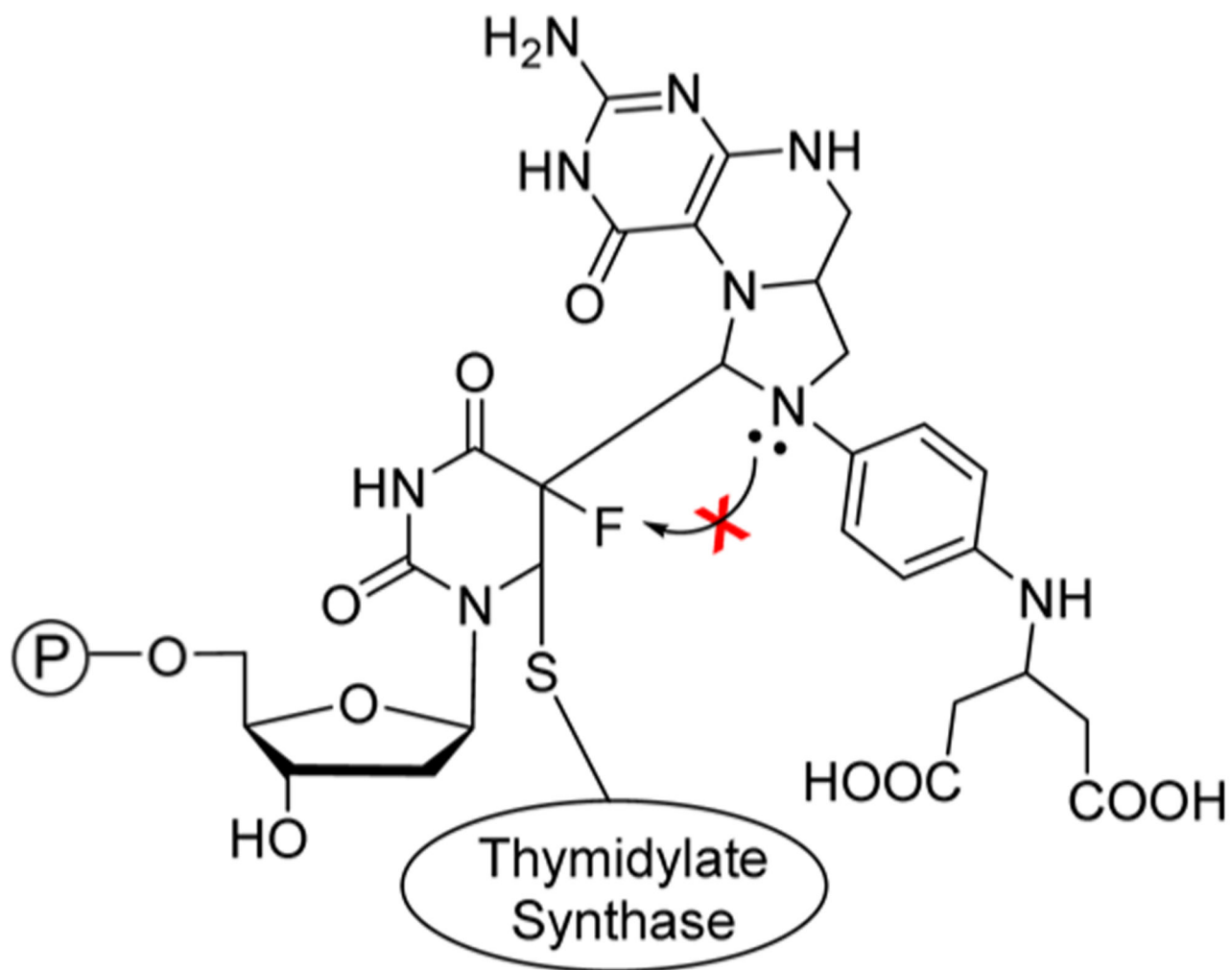


**Figure 6.**  
Metabolism of 6-MP **1** and 6-TG **3**.

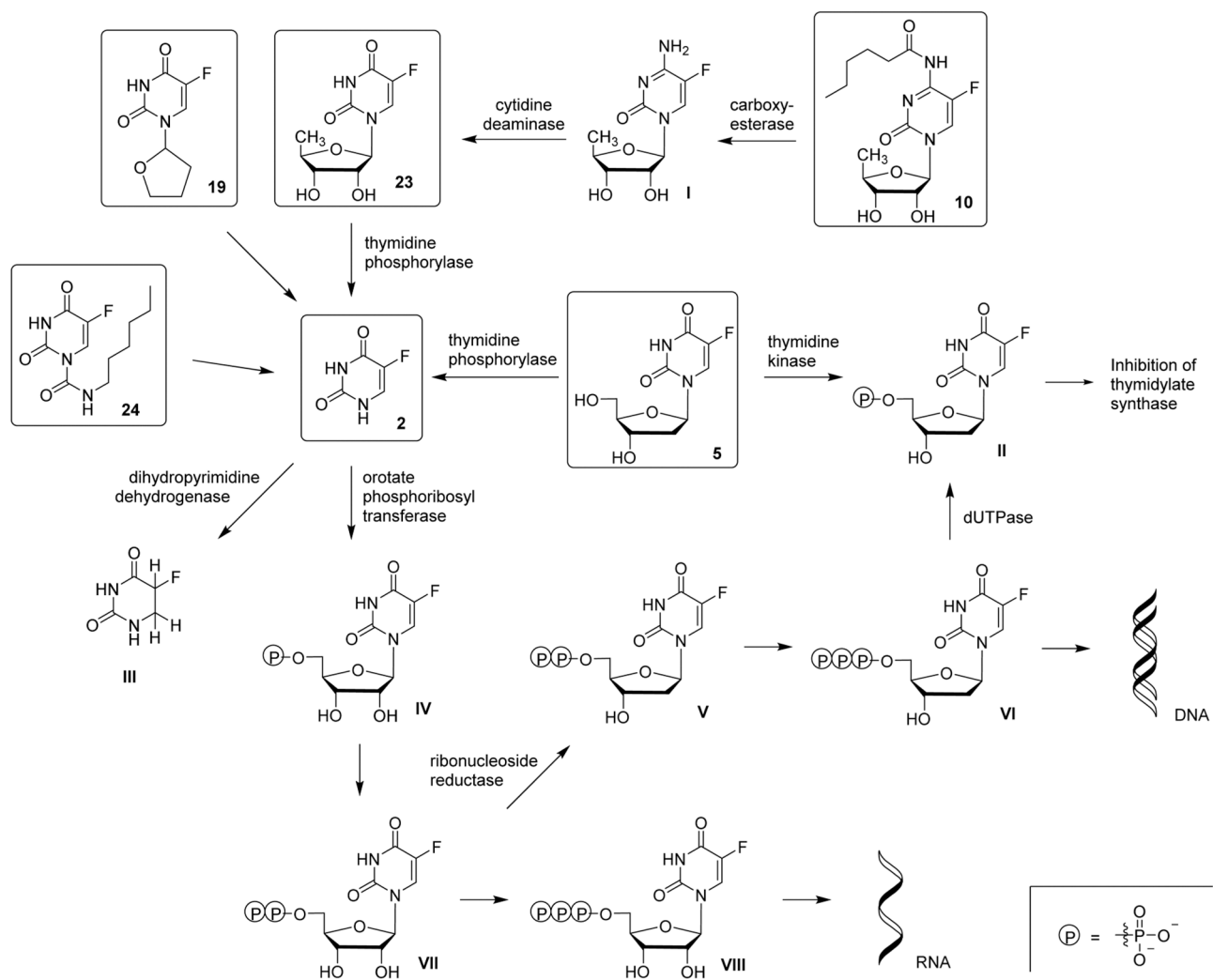


**Figure 7.** Structures of PMEG, PMEDAP, and PMEAs, and proposed cellular metabolism of GS-9219 53.

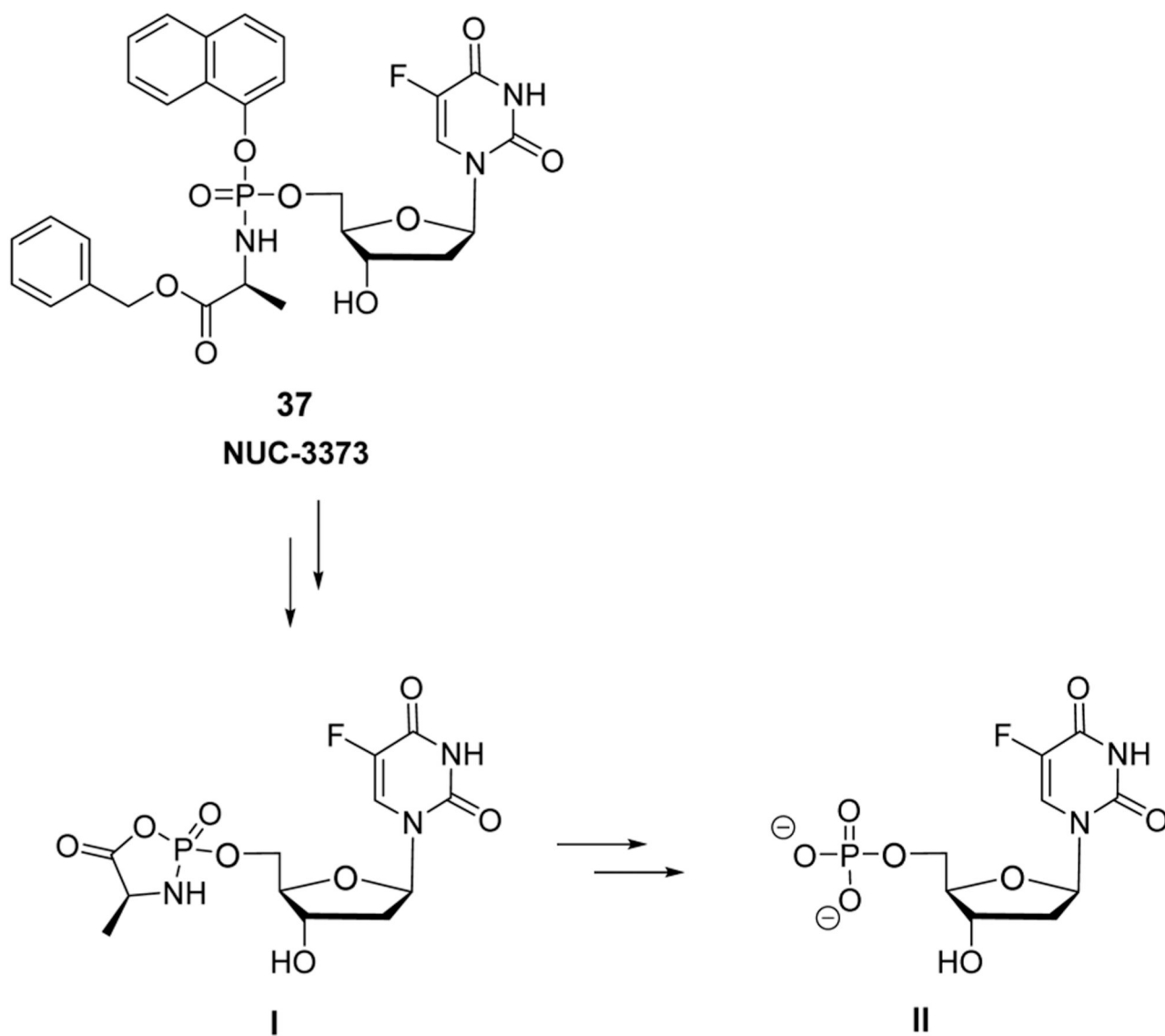




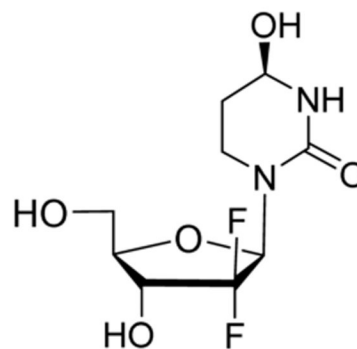
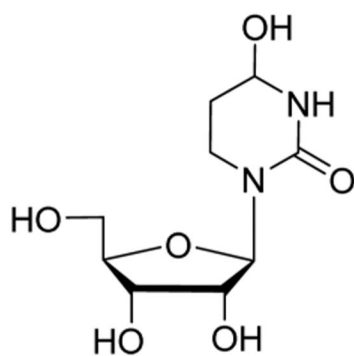
**Figure 8.**  
Complex of 5,10-methylenetetrahydrofolate and thymidylate synthase with 5-FU-5'-monophosphate.



**Figure 9.**  
Metabolism of 5-FU 2 and derivatives.

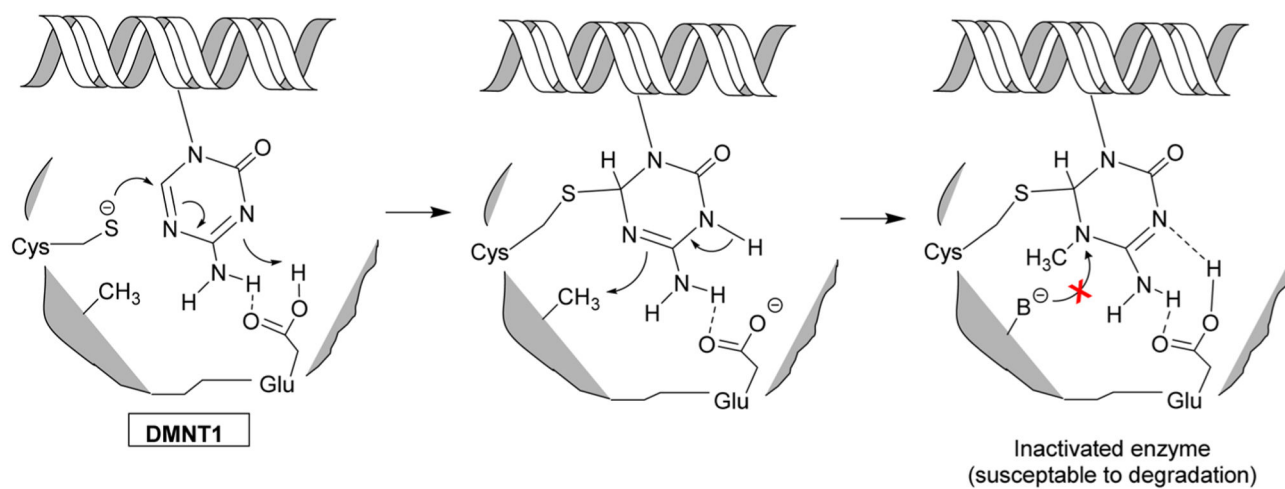


**Figure 10.**  
Proposed mechanism for phosphoramidate decomposition of NUC-3373 **37**.

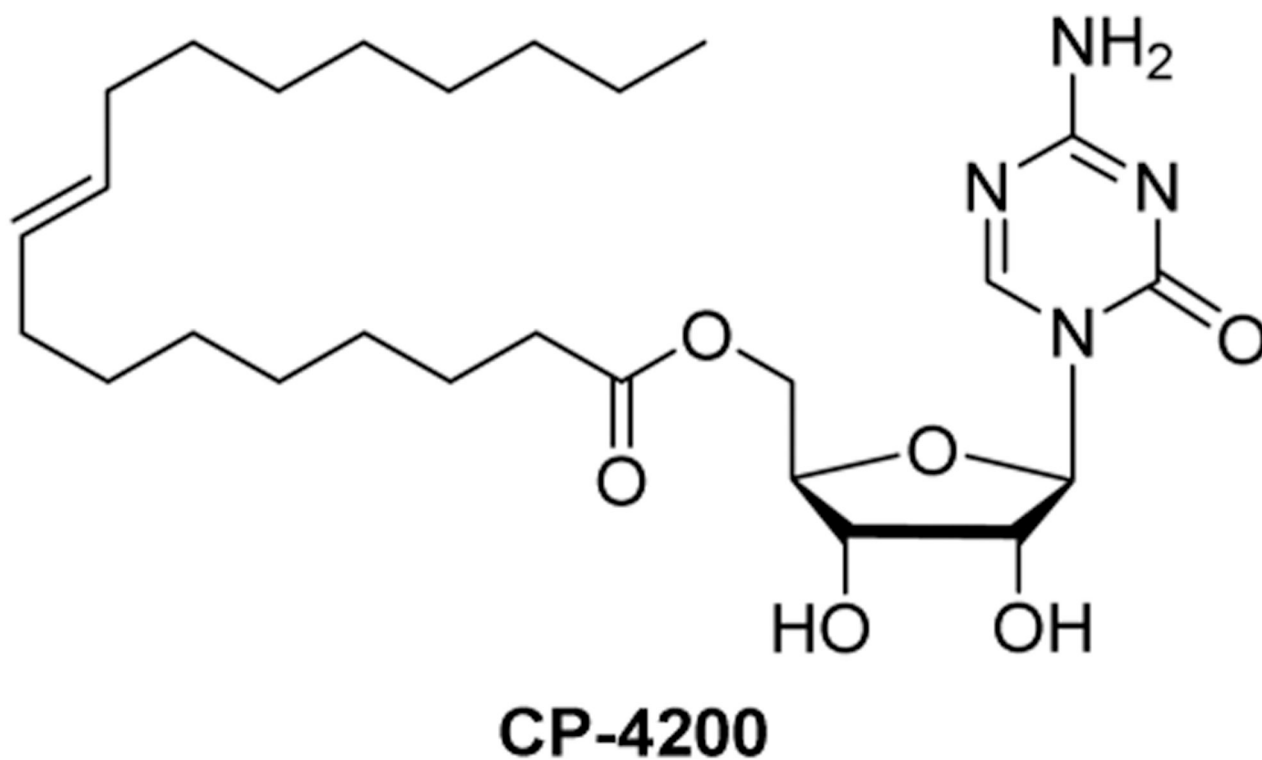


**Tetrahydrouridine**      **(4*R*)-2'-Deoxy-2',2'-difluoro-3,4,5,6-tetrahydrouridine**

**Figure 11.**  
Structures of tetrahydrouridine **39** and (4*R*)-2'-deoxy-2',2'-difluoro-3,4,5,6-tetrahydrouridine.

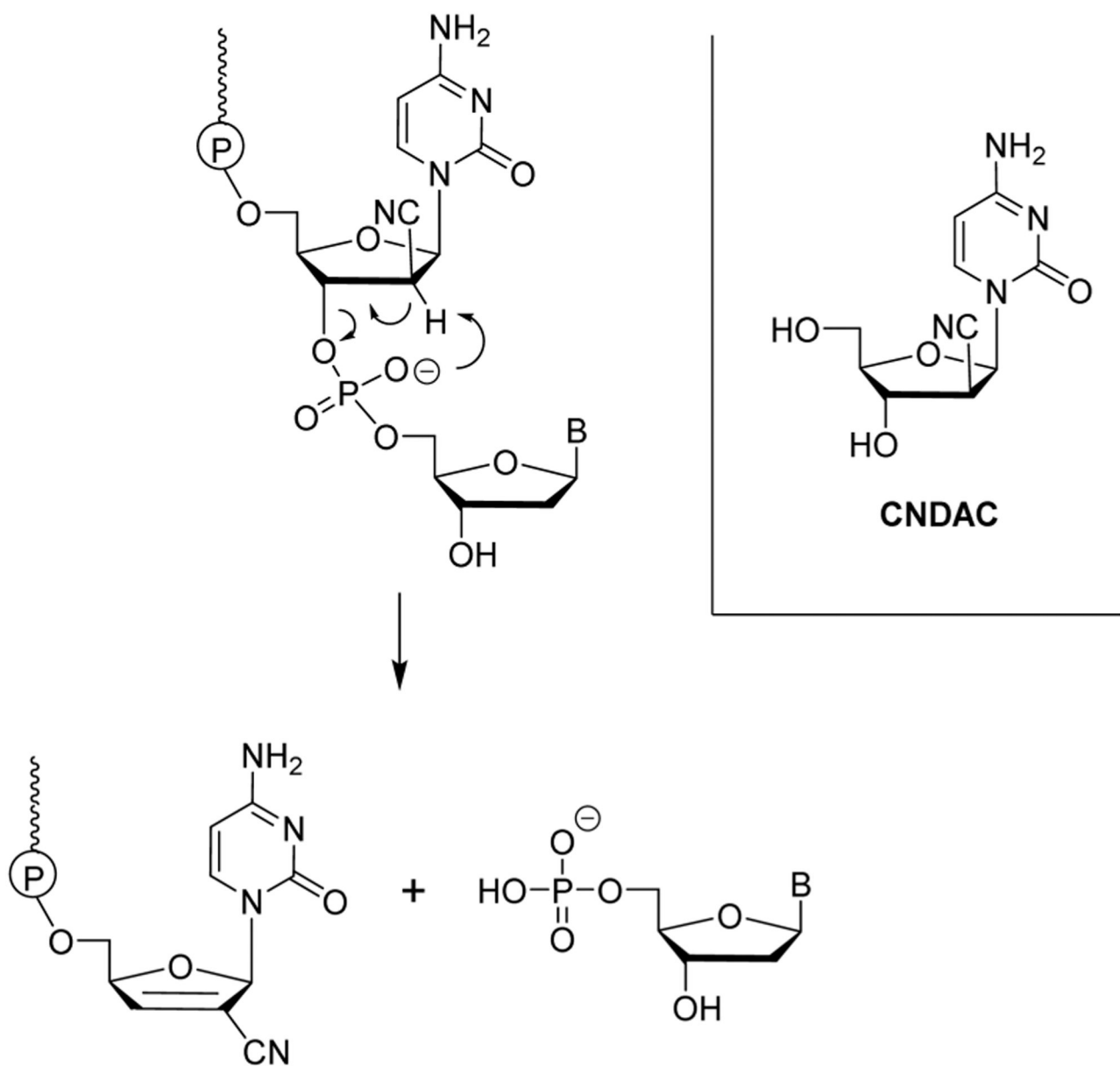


**Figure 12.**  
Covalent trapping paradigm: mechanism-based DNMT1 degradation by decitabine **14**.



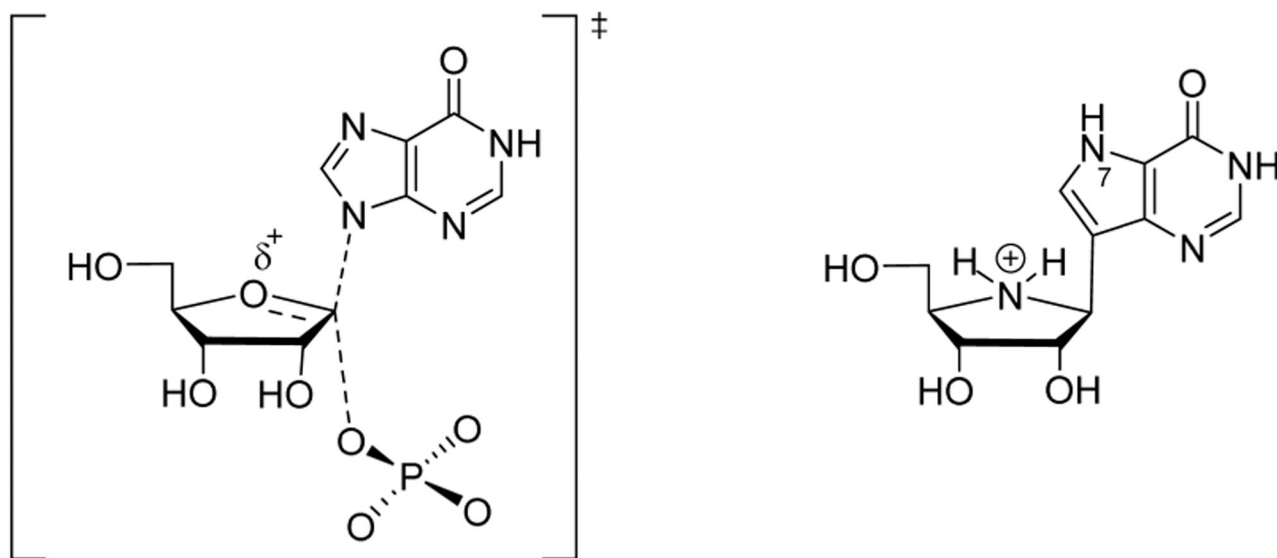
**Figure 13.**  
Chemical structure of CP-4200.



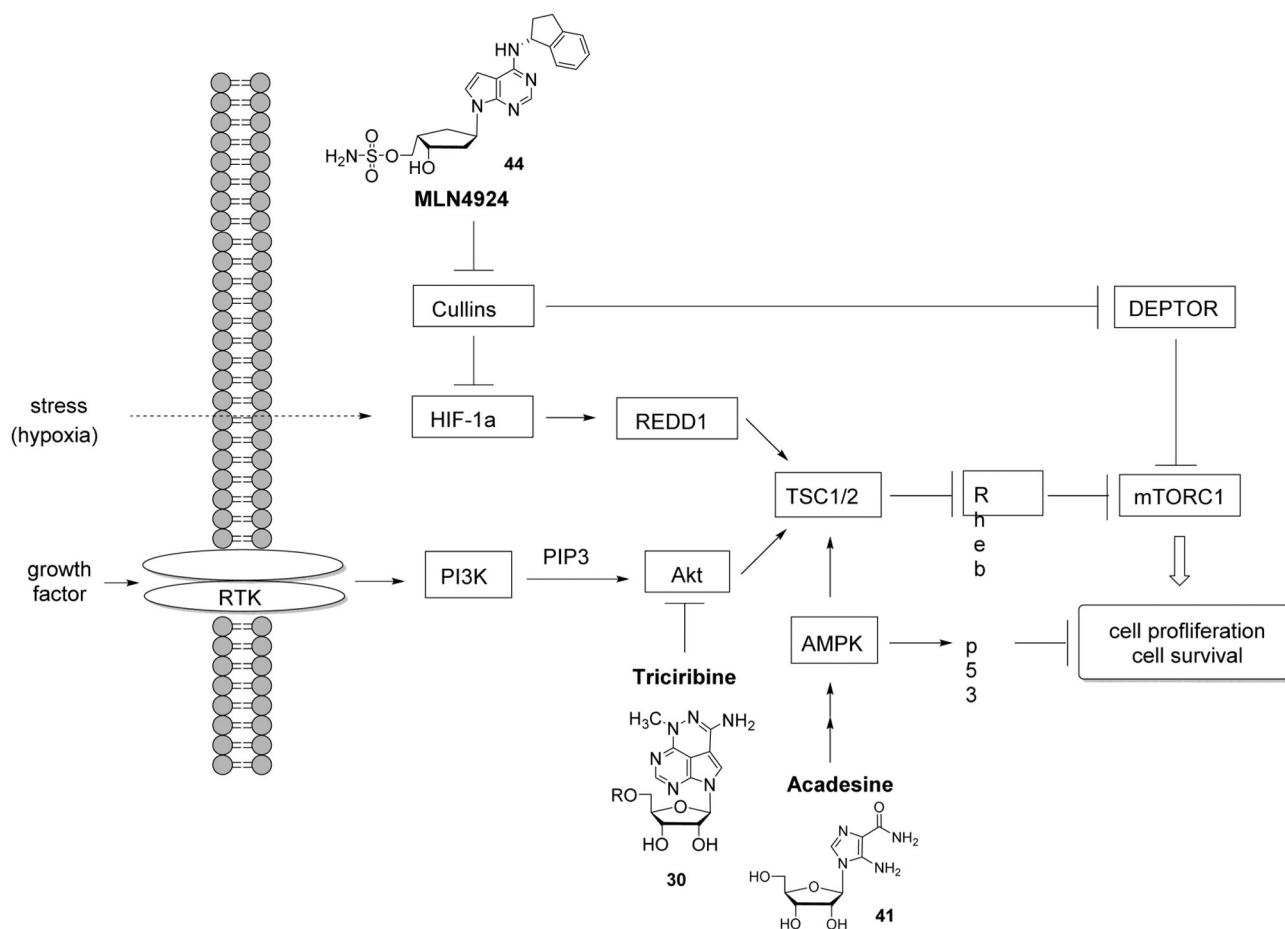


**Figure 14.** CNDAC structure and hypothesized mechanism of DNA strand break with incorporated CNDAC.

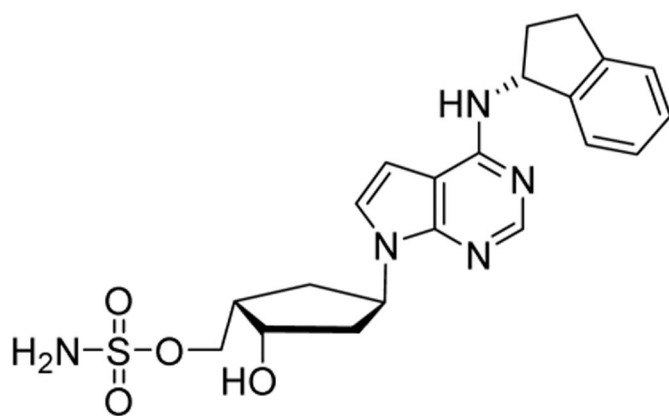




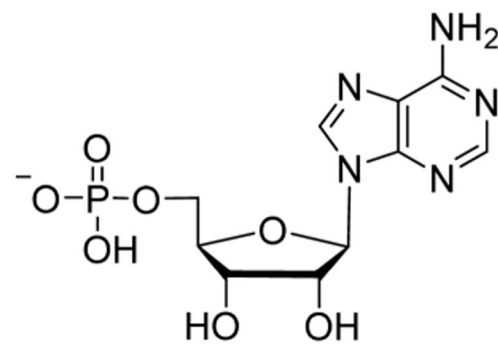
**Figure 16.** Transition state model for the phosphorolysis of inosine by purine nucleoside phosphorylase and the protonated form of forodesine.



**Figure 17.** Proposed biological mechanisms of actions of triciribine **30** as an Akt inhibitor (see Section 7.1, Triciribine Biology); MLN4924 **44** as a Cullins inhibitor (see Section 7.3, MLN4924 Biology); and acadesine **41** as an AMPK activator (see Section 8.5, Acadesine Biology).

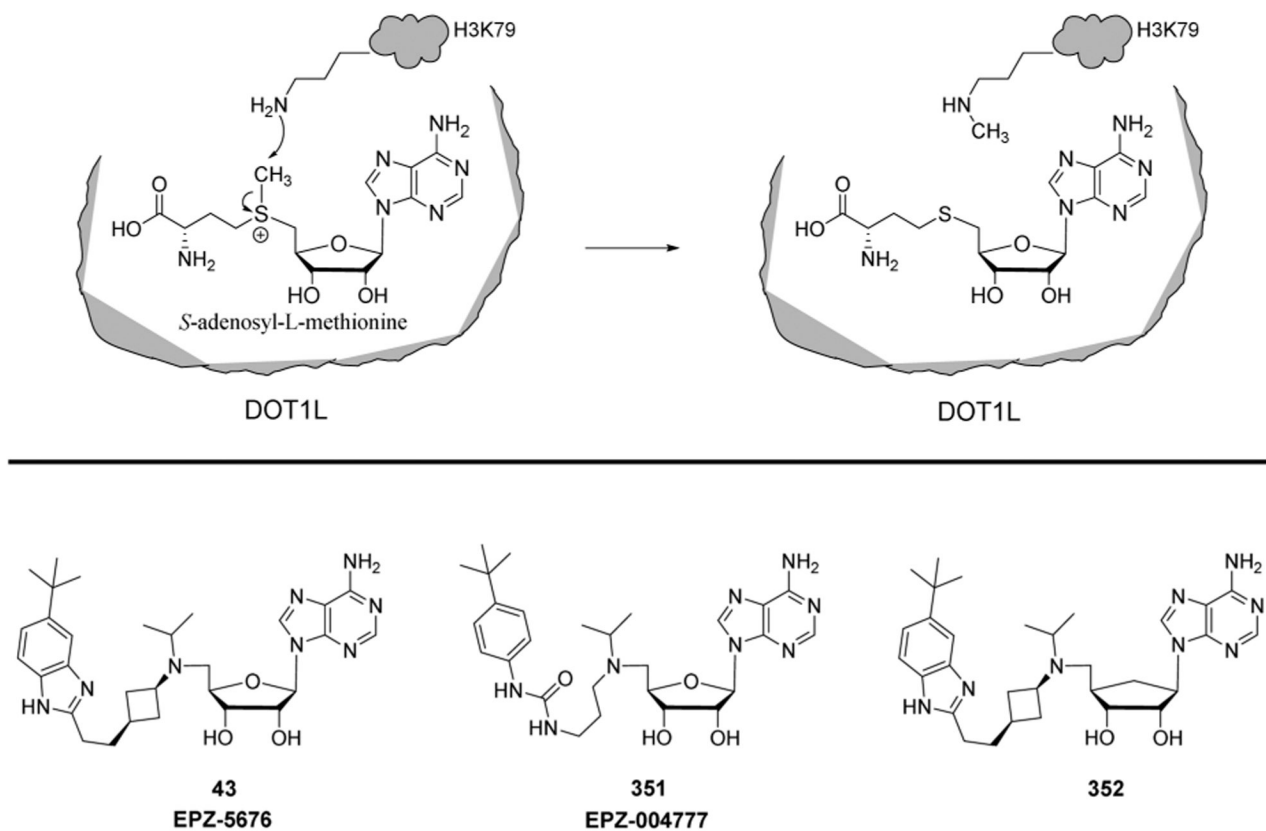


**44**  
**MLN4924**



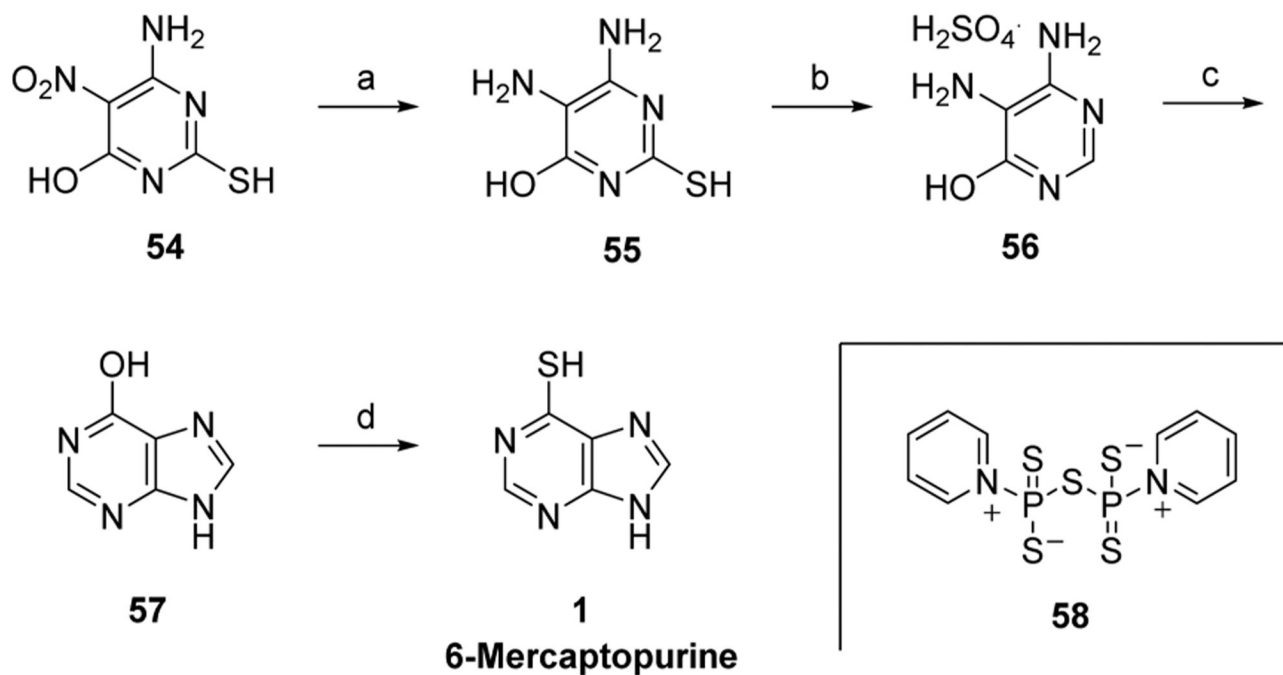
**Adenosine-5'-monophosphate**

**Figure 18.**  
Structures of MLN4924 44 and adenosine-5'-monophosphate.



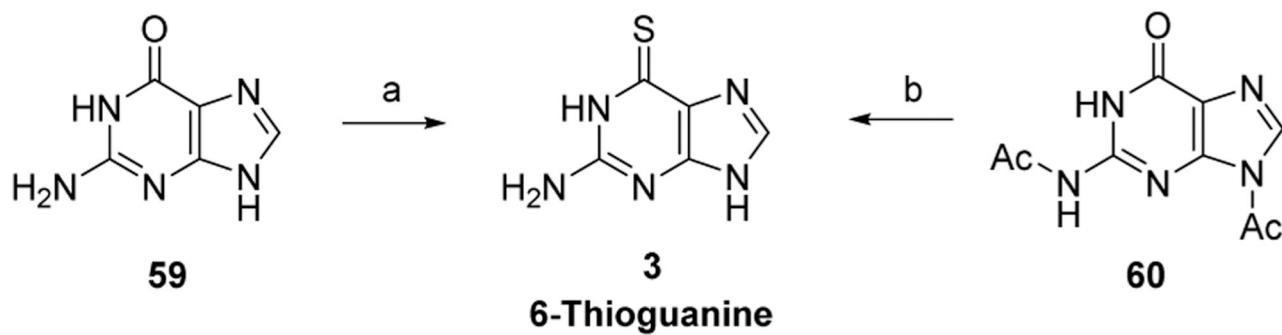
**Figure 19.**  
DOT1L-catalyzed methylation of H3K79.





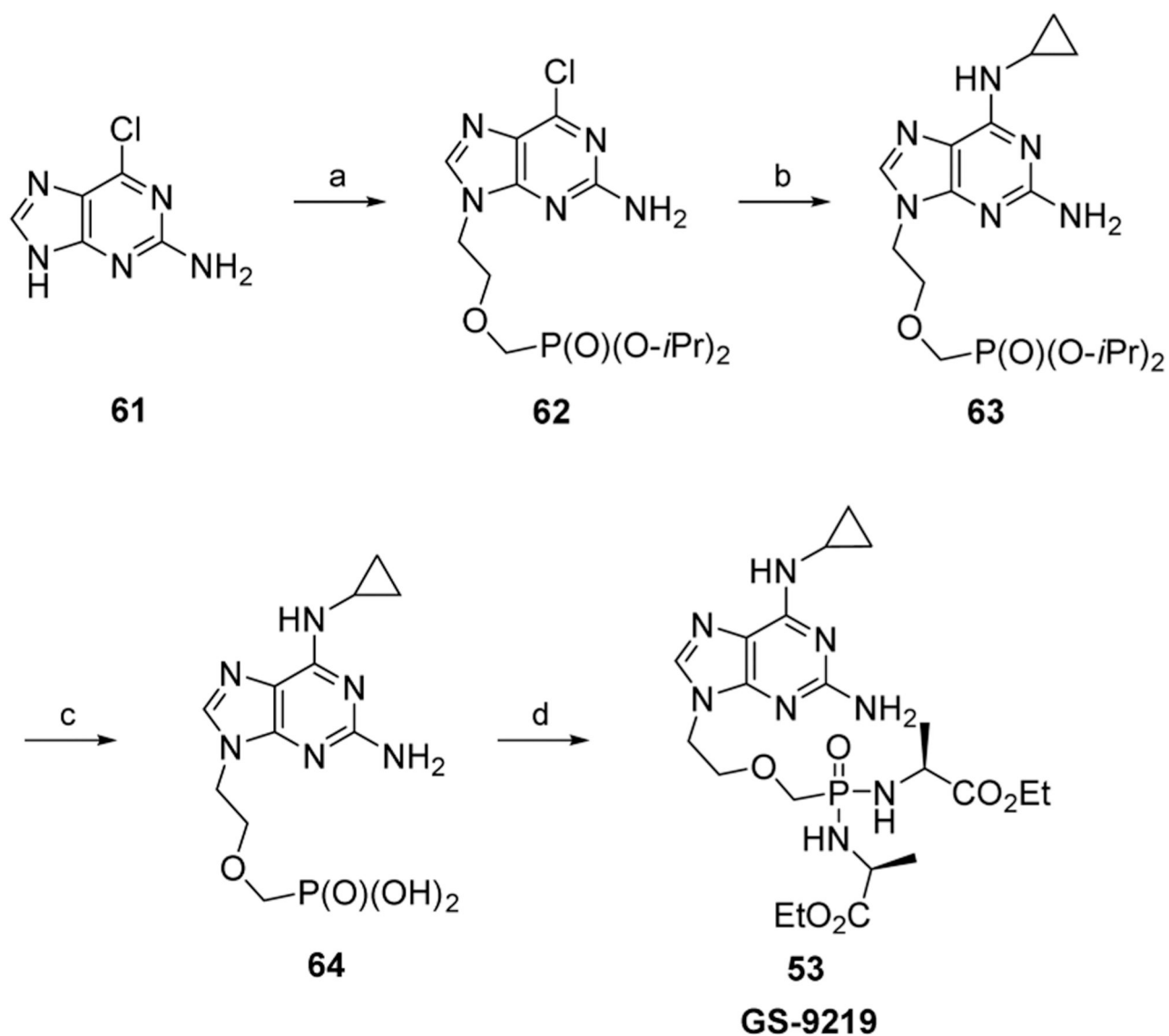
**Scheme 1. Synthesis of 6-Mercaptopurine 1<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $\text{Na}_2\text{S}_2\text{O}_4$ ,  $\text{H}_2\text{O}$ , 60–70 °C, 80–85%; (b) Raney Ni,  $\text{Na}_2\text{CO}_3$ ,  $\text{H}_2\text{O}$ , reflux, 2 h then  $\text{H}_2\text{SO}_4$ , 88%; (c)  $\text{HCO}_2\text{H}$ , reflux, 2 h then aq NaOH, AcOH, 93%; (d)  $\text{P}_2\text{S}_5$ , tetralin, 190–200 °C, 12 h, 40%; or **58**, DMSO<sub>2</sub>, 165–170 °C, 85%.

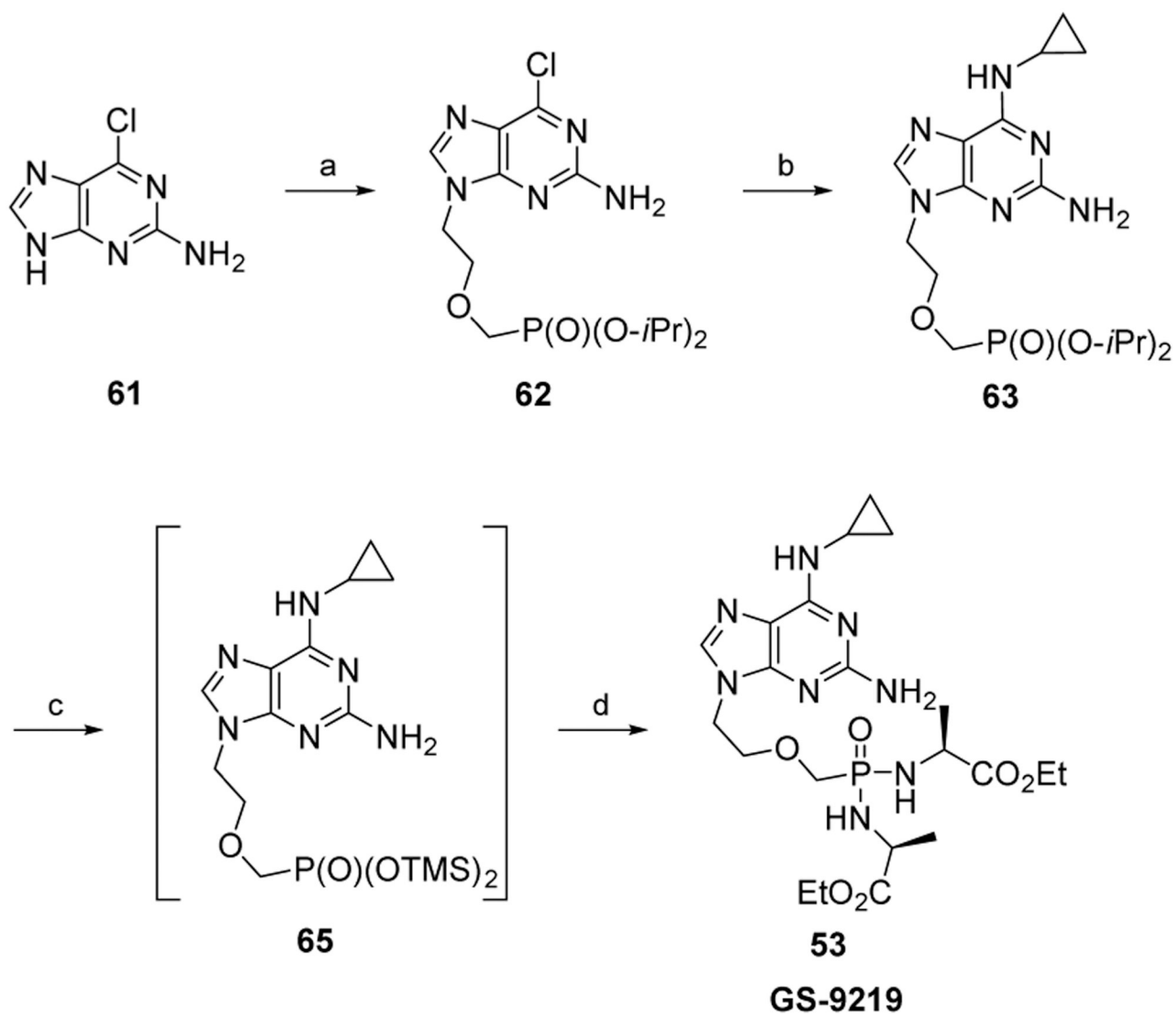


**Scheme 2. Synthesis of 6-Thioguanine 3<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) P<sub>2</sub>S<sub>5</sub>, pyr, reflux, 2.5 h, 32%; (b) P<sub>2</sub>S<sub>5</sub>, pyridine hydrochloride, pyr, 110 °C, 4 h, then HCl, H<sub>2</sub>O, pH 4, recrystallization, 75%.

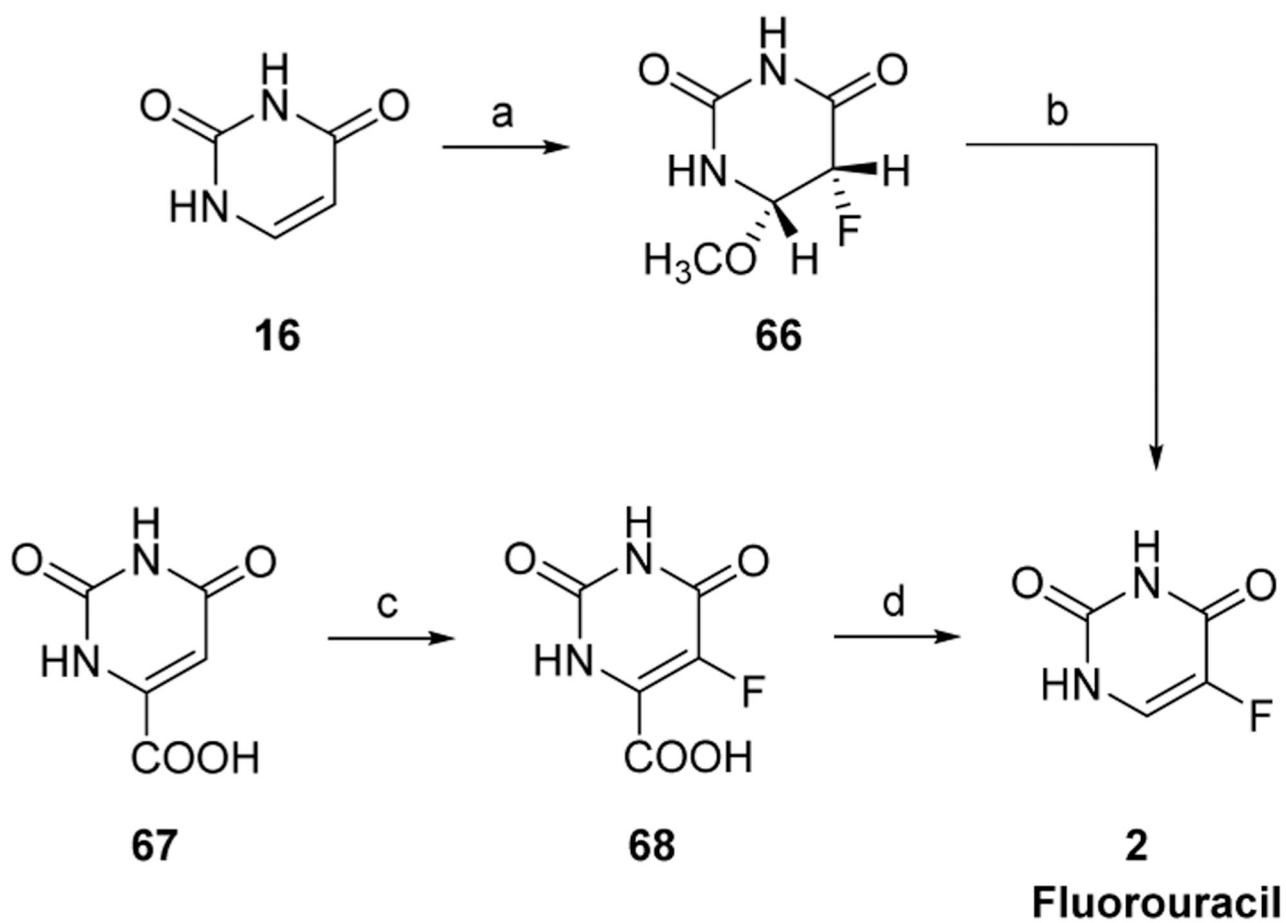
**Scheme 3. Synthesis of GS-9219 53<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) HO(CH<sub>2</sub>)<sub>2</sub>OCH<sub>2</sub>PO(*O*-*i*Pr)<sub>2</sub>, DIAD, PPh<sub>3</sub>, DMF, -15 °C to rt, 4 h, 63%; (b) cyclopropylamine, CH<sub>3</sub>CN, 100 °C, 4 h, 90%; (c) TMSBr, CH<sub>3</sub>CN, rt, overnight, 90%; (d) D-alanine ethyl ester HCl, 2,2'-dithiodipyridine, PPh<sub>3</sub>, Et<sub>3</sub>N, pyr, 60 °C, overnight, 50%.

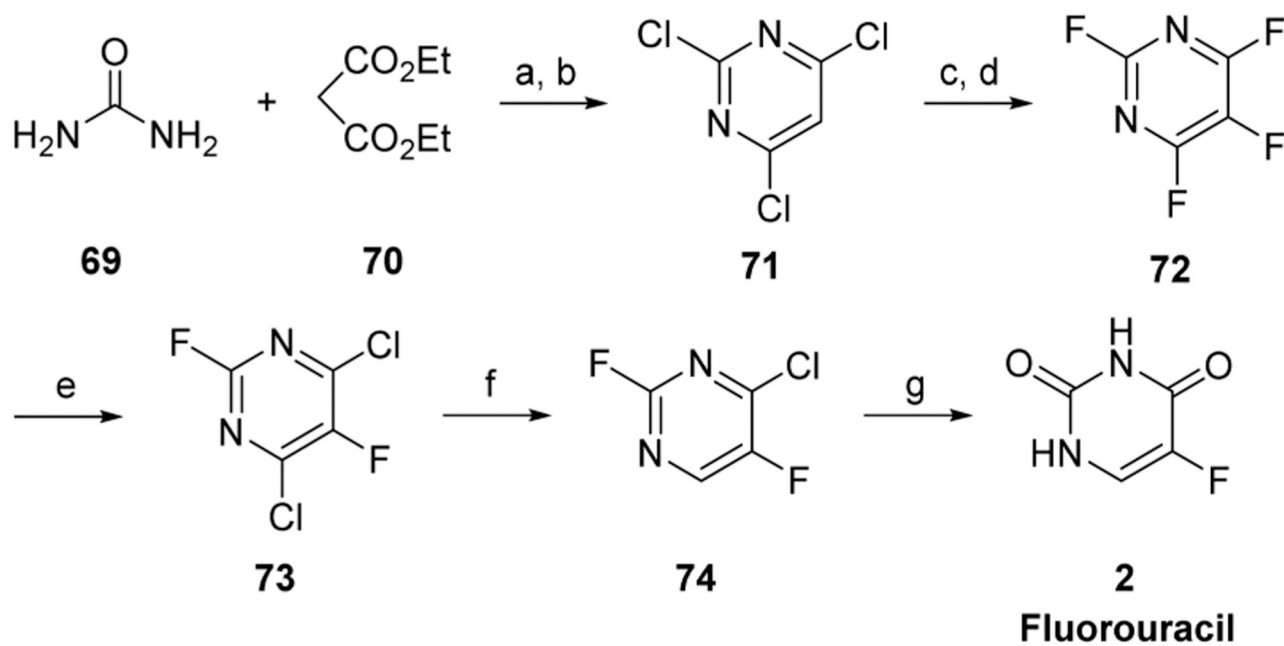


**Scheme 4. Synthesis of GS-9219 53<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $\text{Cl}(\text{CH}_2)_2\text{OCH}_2\text{P}(\text{O})(\text{O}-i\text{Pr})_2$ ,  $\text{Cs}_2\text{CO}_3$ , DMF, 80 °C, 8 h, 56%; (b) cyclopropylamine, EtOH, reflux, 3 h, 65%; (c) TMSBr,  $\text{CH}_3\text{CN}$ , rt, overnight; (d) D-alanine ethyl ester, aldrithiol-2,  $\text{PPh}_3$ ,  $\text{Et}_3\text{N}$ , pyr, 50 °C, 3 h, 92–98% (over two steps).

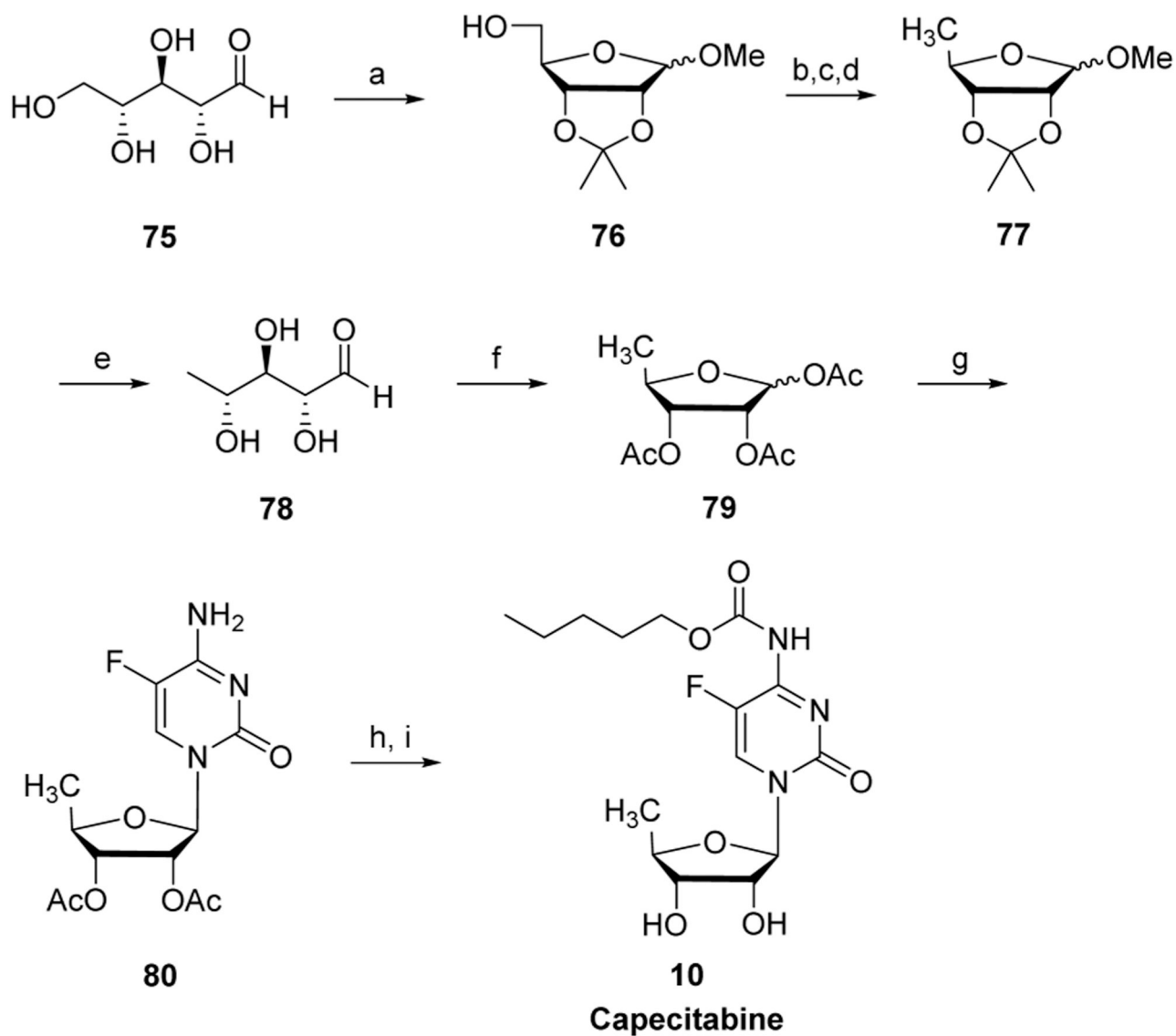
**Scheme 5. Synthesis of Fluorouracil 2<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $\text{CF}_3\text{OF}/\text{CCl}_3\text{F}$ , MeOH,  $-78\text{ }^\circ\text{C}$ , 5 min; (b)  $\text{Et}_3\text{N}$ , MeOH,  $\text{H}_2\text{O}$ , rt, 11 min, 90% (over two steps); (c)  $\text{F}_2$  (g), TFA,  $0\text{ }^\circ\text{C}$  to  $10\text{ }^\circ\text{C}$ , 3 h, 80%; (d) triethyleneglycol dimethyl ether,  $130\text{ }^\circ\text{C}$  to  $200\text{ }^\circ\text{C}$ , 20 min, 90%.

**Scheme 6. Synthesis of Fluorouracil 2<sup>a</sup>**

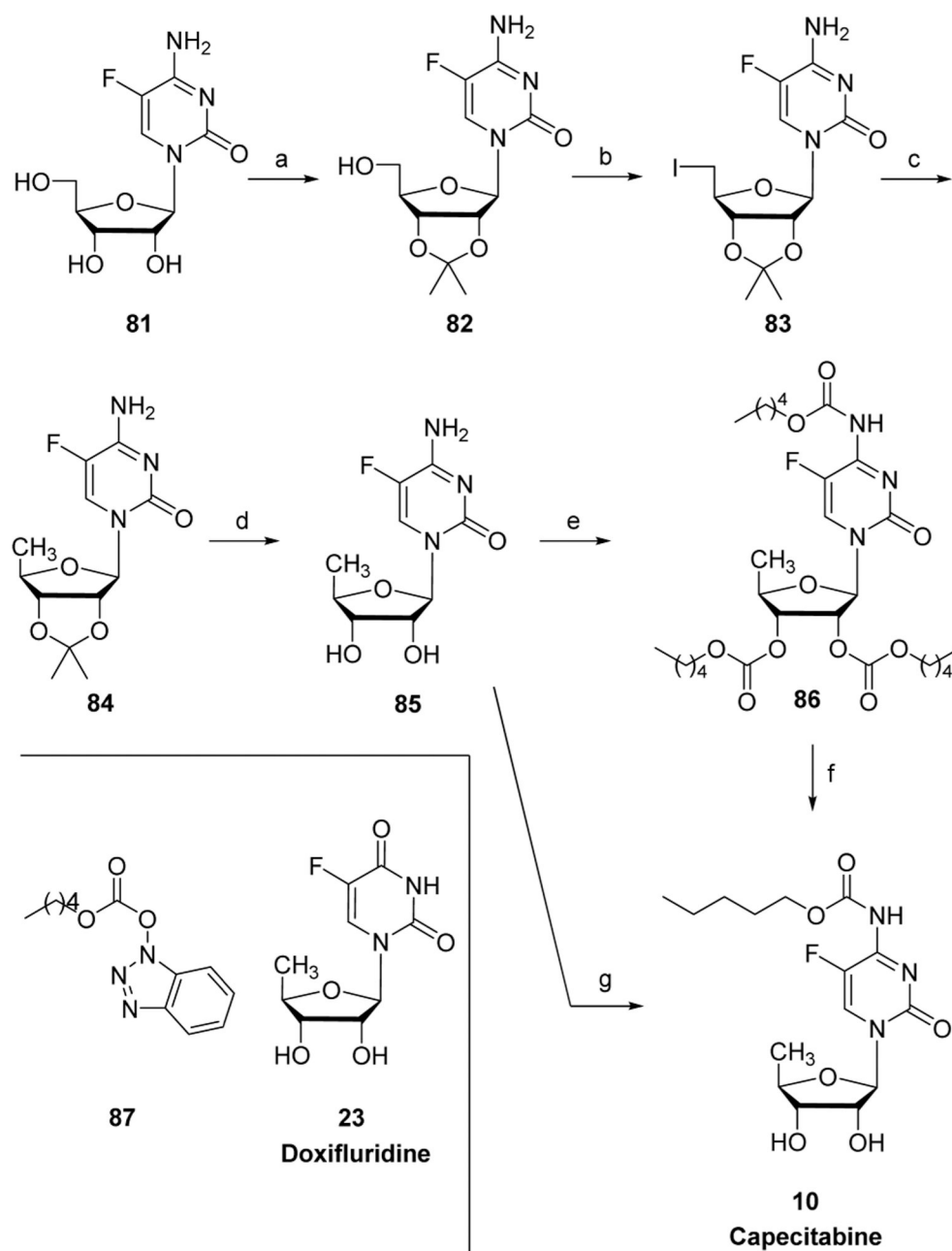
<sup>a</sup>Reagents and conditions: (a) Na, MeOH; (b) POCl<sub>3</sub>, PhNMe<sub>2</sub>; (c) proton sponge hydrofluoride; (d) NaF, F<sub>2</sub>, CFC<sub>3</sub>; (e) HCl (g), 160 °C, 4 h, 38%; (f) Pd/C, H<sub>2</sub>, Et<sub>3</sub>N, EtOAc, 3.5 h, 82%; (g) aq NaOH, 80 °C, 4 h, 93%.





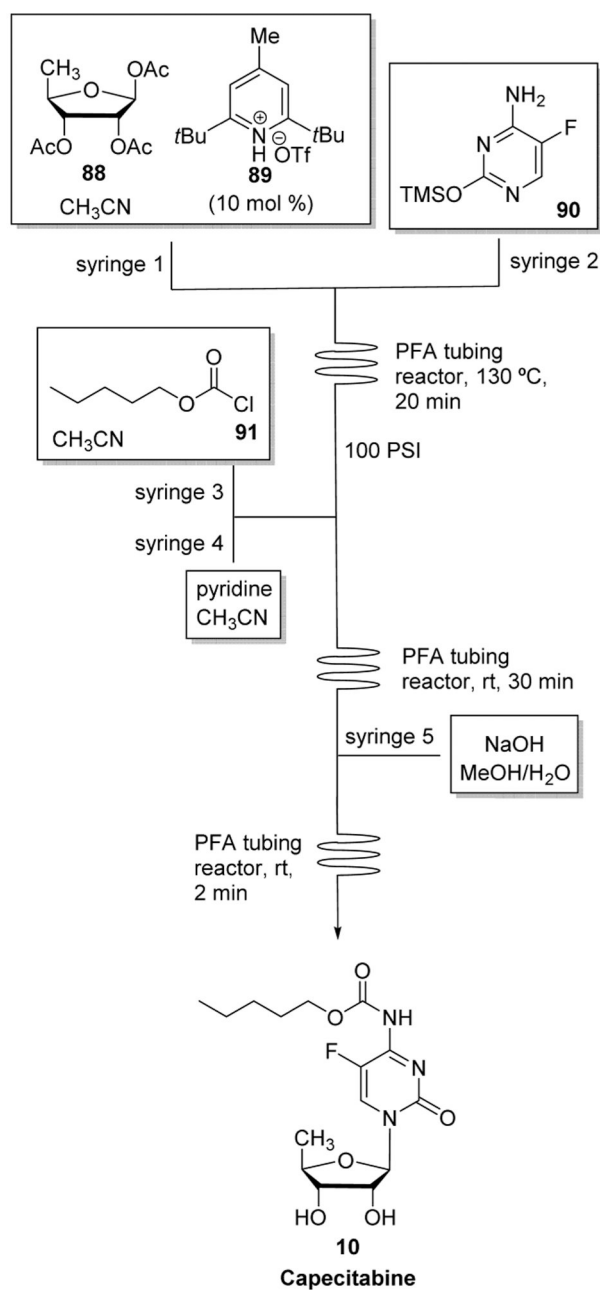
**Scheme 7. Synthesis of Capecitabine 10<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) HCl, MeOH/Me<sub>2</sub>CO, 78%; (b) MsCl, pyr; (c) NaI, DMF; (d) Pd/C, H<sub>2</sub>, 27%; (e) aq HCl, 100 °C, 97%; (f) Ac<sub>2</sub>O, pyr, 64%; (g) 5-fluorocytosine, HMDS, SnCl<sub>4</sub>, DCM, 76%; (h) *n*-pentyl chloroformate, pyr; (i) aq NaOH, MeOH.

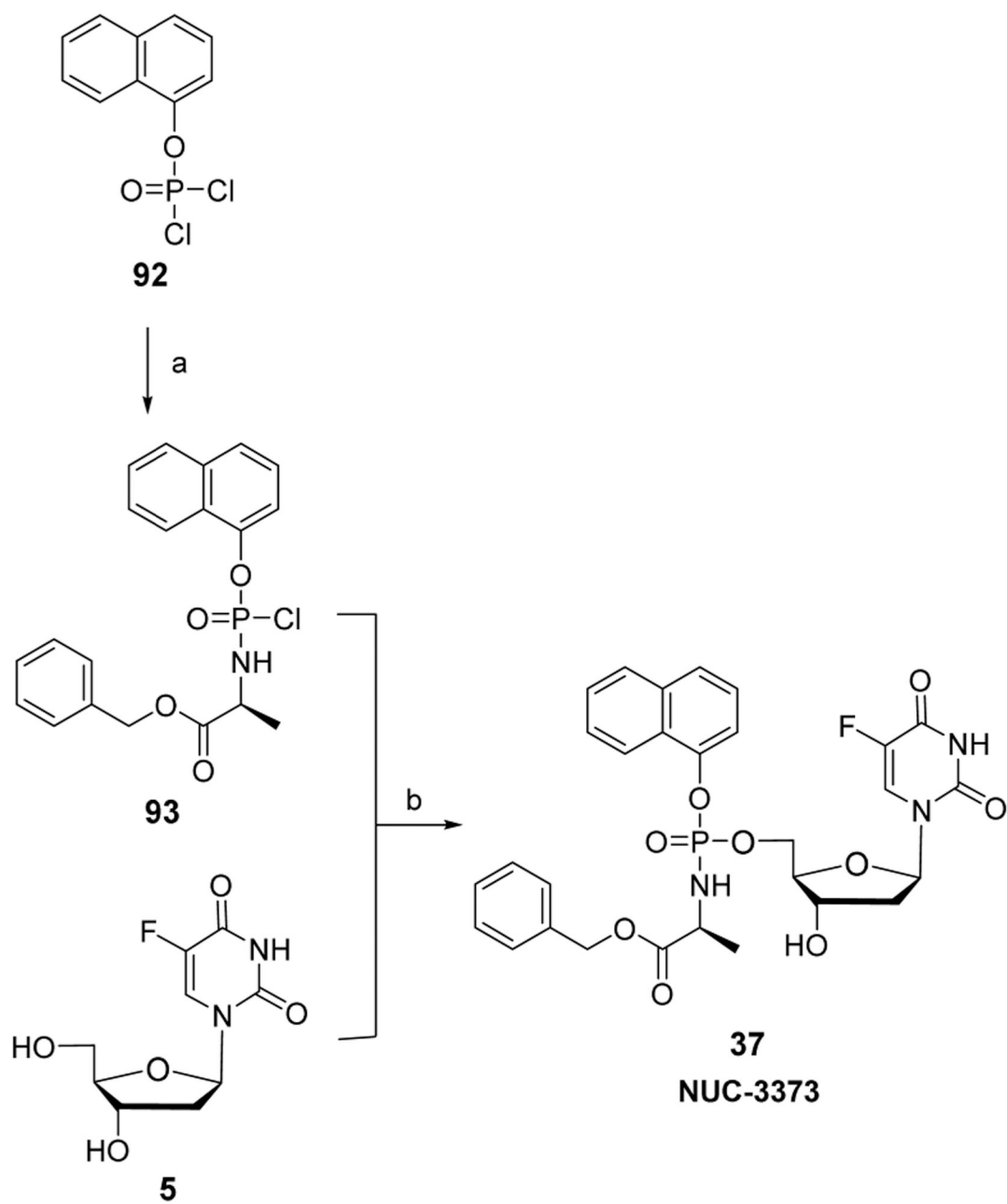


**Scheme 8. Synthesis of Doxifluridine 23 and Capecitabine 10<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 2,2-dimethoxypropane, TsOH, Me<sub>2</sub>CO, rt, 2 h, 95%; (b) methyltriphenoxyphosphonium iodide, DMF, rt, 2.5 h, then MeOH, rt, 1 h, 74%; (c) Et<sub>3</sub>N, H<sub>2</sub>, Pd/C, MeOH, rt, 1.5 h, 93%; (d) TFA, 40 min, 78%; (e) CH<sub>3</sub>(CH<sub>2</sub>)<sub>4</sub>OCOCl, pyr, DCM, -5 °C to rt, overnight, 92%; (f) NaOH, MeOH, -10 °C, 15 min, then concd HCl, 87%; (g) 87, THF, reflux, 66%.

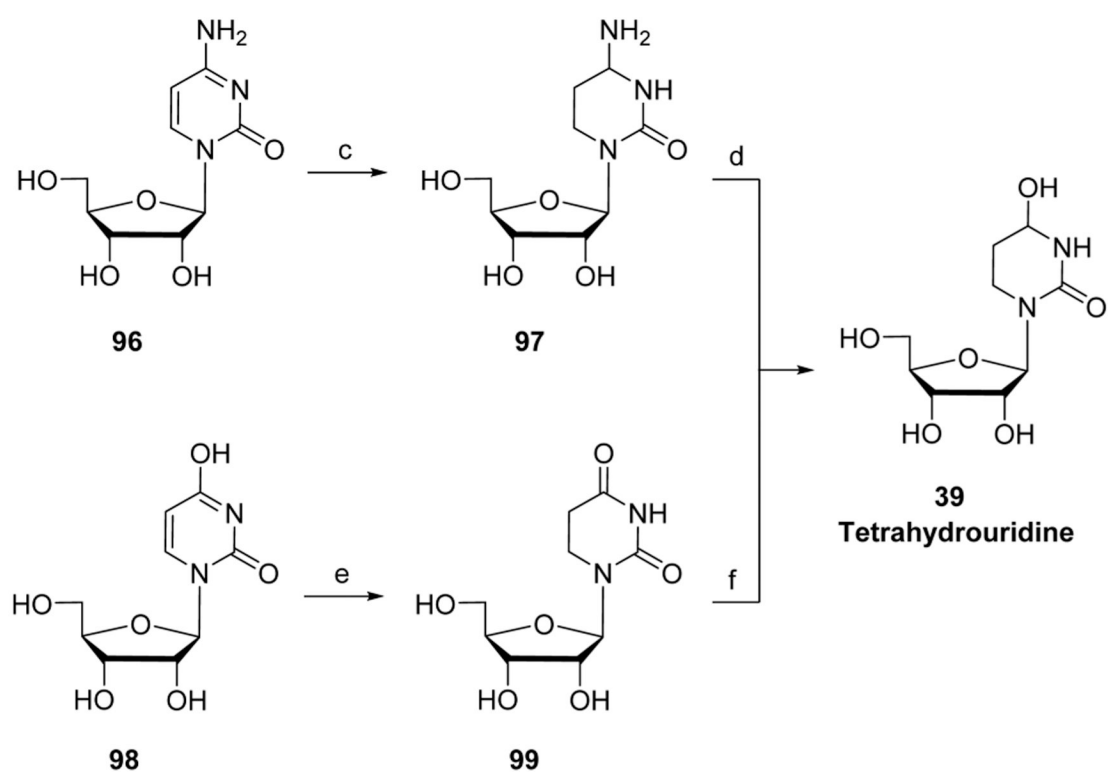
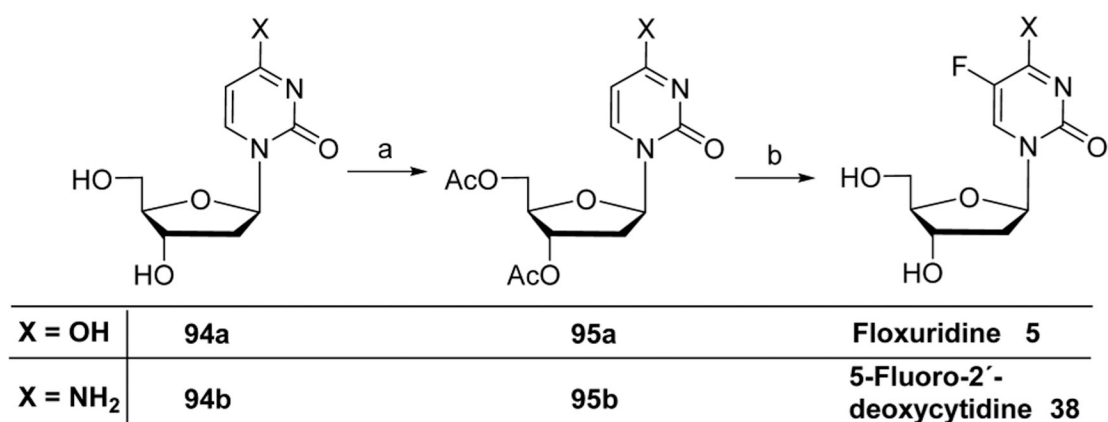


**Scheme 9.**  
Flow Synthesis of Capecitabine 10

**Scheme 10. Synthesis of NUC-3373 37<sup>a</sup>**

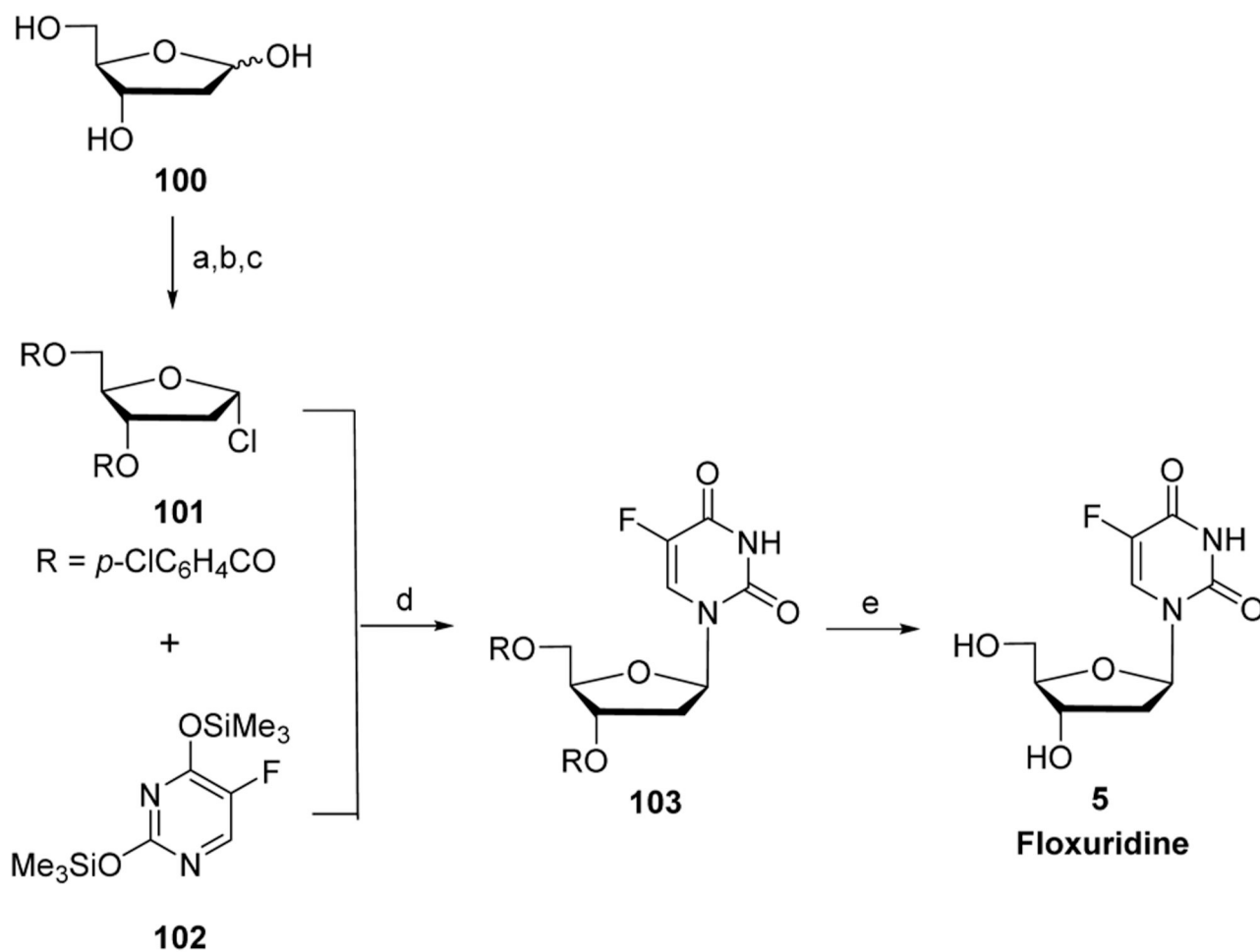
<sup>a</sup>Reagents and conditions: (a) L-alanine benzyl ester salt, Et<sub>3</sub>N, DCM, -78 °C, 1–3 h, 74%;

(b) *t*-BuMgCl, THF, rt, 18 h, 8%.



**Scheme 11. Synthesis of Floxuridine 5, 5-Fluoro-2'-deoxycytidine 38, and Tetrahydrouridine 39<sup>a</sup>**

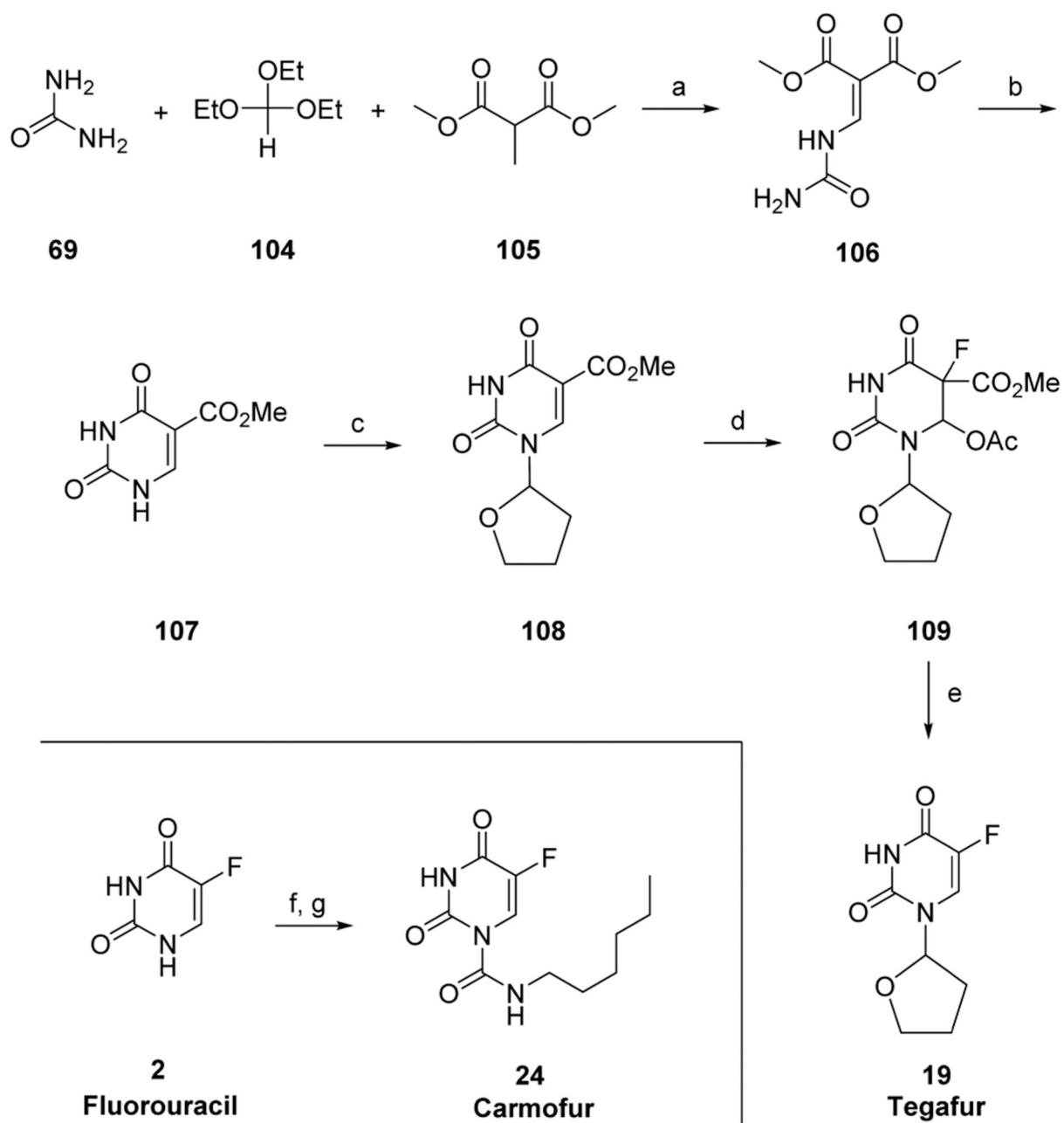
<sup>a</sup>Reagents and conditions: (a) Ac<sub>2</sub>O, DMAP, rt, 24 h, 88% (for **95a**); Ac<sub>2</sub>O, pyr, rt, 4.5 h, 99% (for **95b**); (b) CF<sub>3</sub>OF, CCl<sub>3</sub>F, CHCl<sub>3</sub>, -30 °C, evaporation, then Et<sub>3</sub>N, MeOH, H<sub>2</sub>O, rt, (55% for **5**), (69% for **38**); (c) Rh/Al, H<sub>2</sub> (g), H<sub>2</sub>O, overnight; (d) H<sub>2</sub>O, pH 6, 92% (over two steps); (e) Rh/Al, H<sub>2</sub> (g), NaOH, H<sub>2</sub>O, 24 h, 45% of **99**, 35% of **39**; (f) NaBH<sub>4</sub>, H<sub>2</sub>O, freezer, overnight.



**Scheme 12. Synthesis of Floxuridine 5<sup>a</sup>**

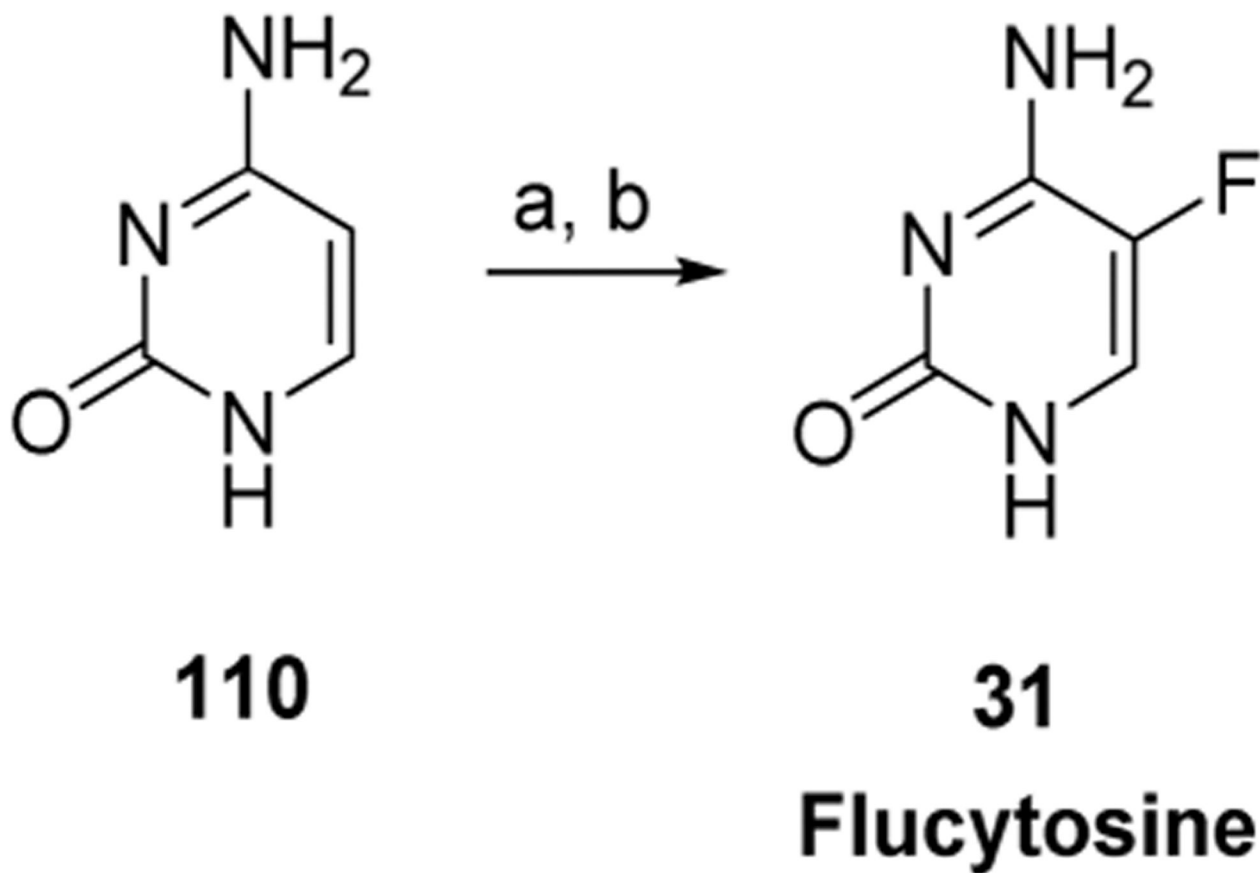
<sup>a</sup>Reagents and conditions: (a) AcCl, MeOH, 25 °C, 45 min, then pyr; (b) pyr, DMAP, 4-CIBzCl, 0 °C, 1 h then 25 °C, 12 h; (c) HOAc, HCl (g), 0 °C, 7–10 min, 82% (*α*-anomer crystallizes from the solvent); (d) *p*-nitrophenol, CHCl<sub>3</sub>, 30 °C, 12 h, 92%; (e) NH<sub>3</sub>/MeOH, 30 °C, 16 h, 81%.



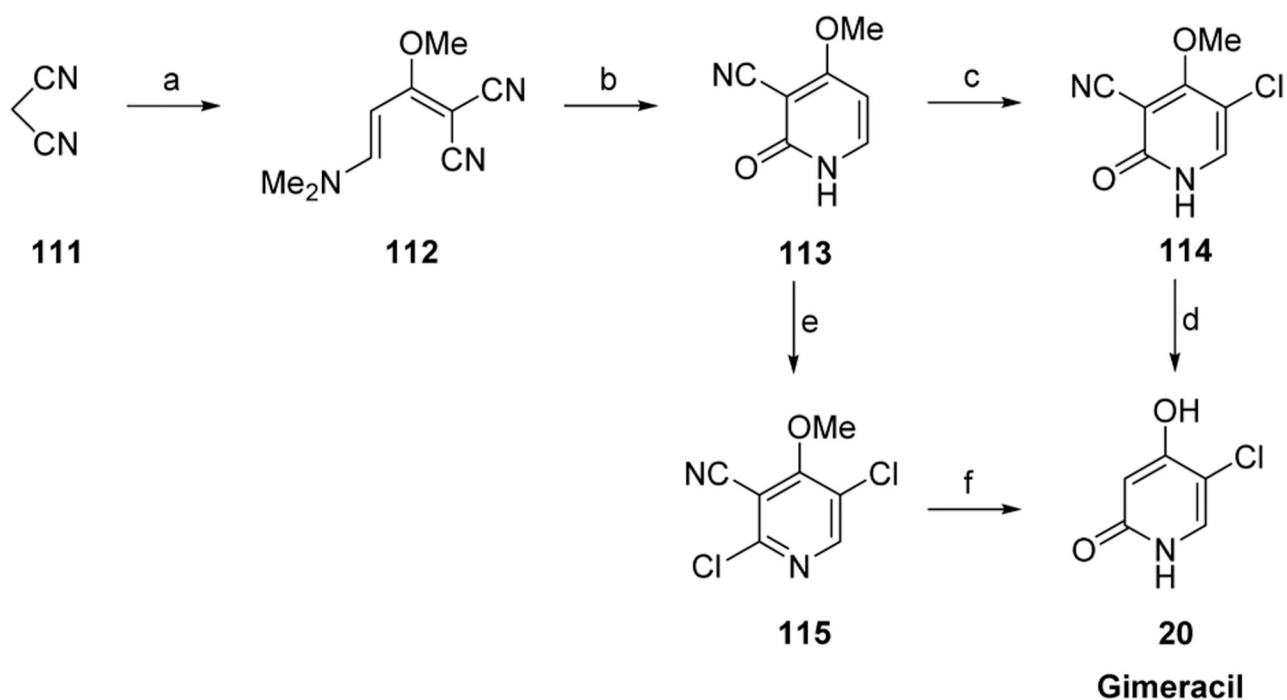


**Scheme 13. Synthesis of Tegafur 19 and Carmofur 24<sup>a</sup>**

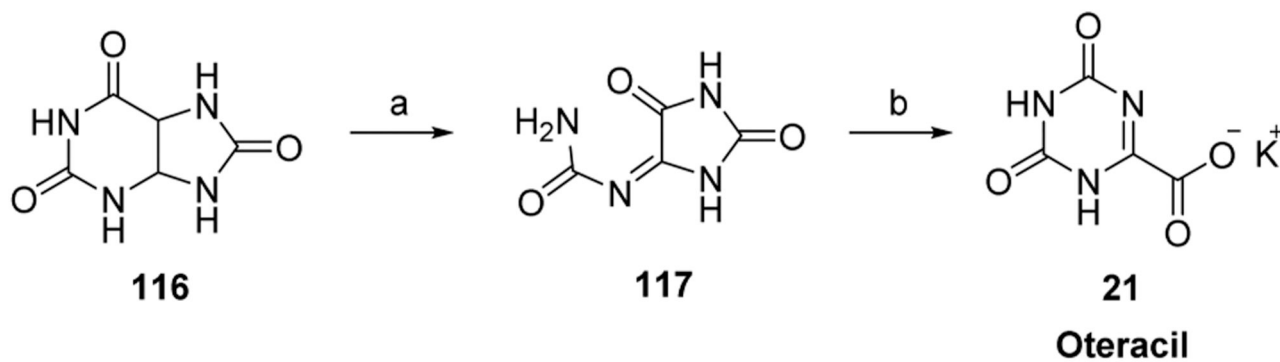
<sup>a</sup>Reagents and conditions: (a) 130 °C, 4 h, 69%; (b) NaOMe, MeOH, reflux, 10 min, 84%; (c) 2,3-dihydrofuran, pyr, 180 °C; (d) F<sub>2</sub>, AcOH, rt; (e) 1 N NaOH, rt, 1 h, 62% (over two steps); (f) aq HCHO, 55 °C, 6 h, 82%; (g) KBrO<sub>3</sub>, 80 °C, then H<sub>2</sub>N(CH<sub>2</sub>)<sub>5</sub>CH<sub>3</sub>, DCC, MeCN, 0 °C to rt, 6 h, 40%.

**Scheme 14. Synthesis of Flucytosine 31<sup>a</sup>**

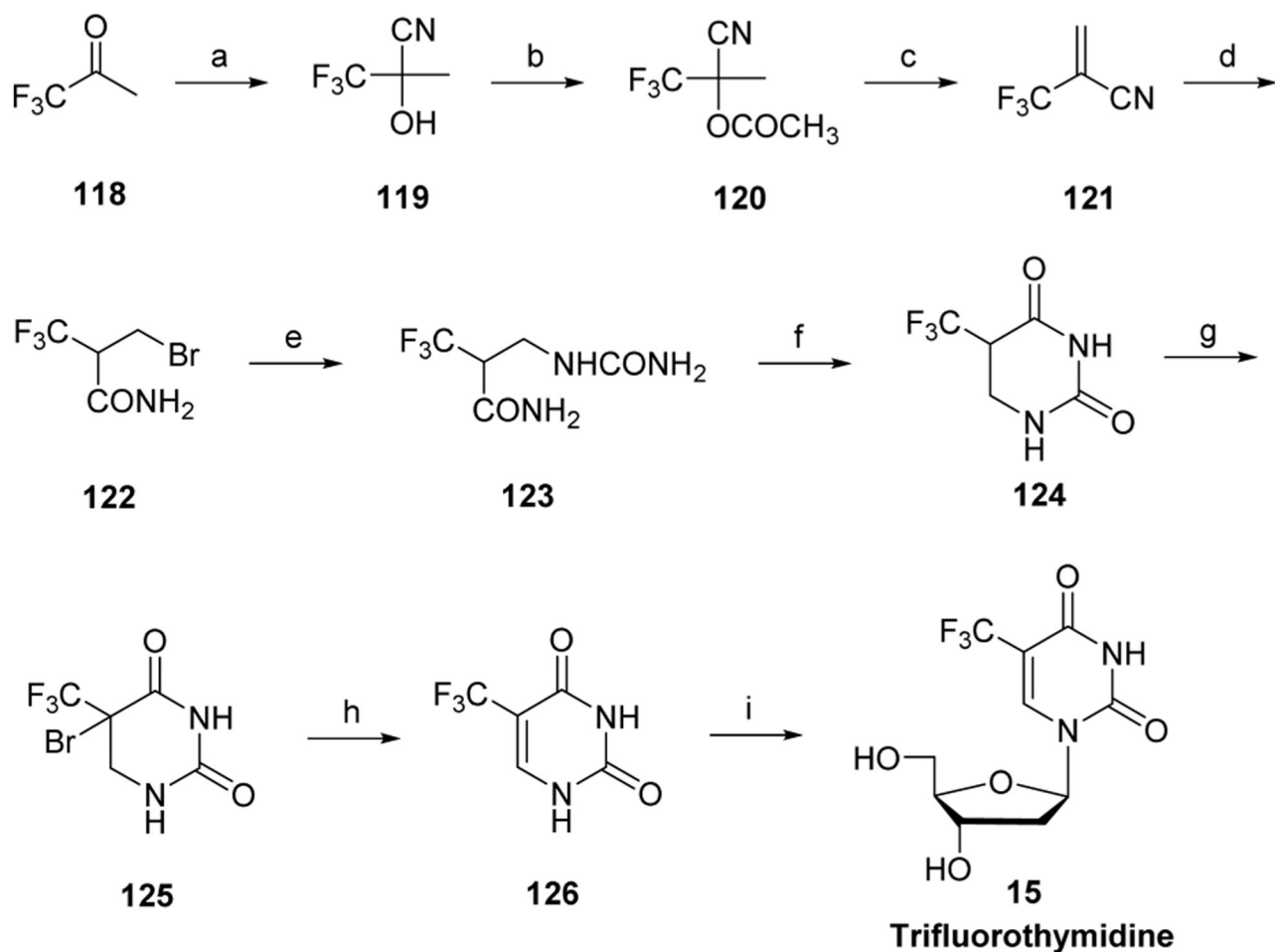
<sup>a</sup>Reagents and conditions: (a)  $\text{CF}_3\text{OF}$ ,  $\text{CCl}_3\text{F}$ , MeOH,  $-78\text{ }^\circ\text{C}$  to rt; (b)  $\text{Et}_3\text{N}$ , MeOH,  $\text{H}_2\text{O}$ , rt, 8 h, 85% (over two steps).

**Scheme 15. Synthesis of Gimeracil 20<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $\text{CH}_3\text{C}(\text{OCH}_3)_3$ , MeOH, then  $(\text{CH}_3)_2\text{NHCH}(\text{OCH}_3)_2$ , reflux, 92%; (b) aq AcOH, 130 °C, 2 h, 95%; (c)  $\text{SO}_2\text{Cl}_2$ , HOAc, 50 °C, 0.5 h, 91%; (d) 40%  $\text{H}_2\text{SO}_4$ , 130 °C, 4 h, 91%; (e)  $\text{SO}_2\text{Cl}_2$ , HOAc, 50 °C, 45 min, 86%; (f) 75%  $\text{H}_2\text{SO}_4$ , 140 °C, 3 h, then NaOH, then pH 4–4.5, 89%.

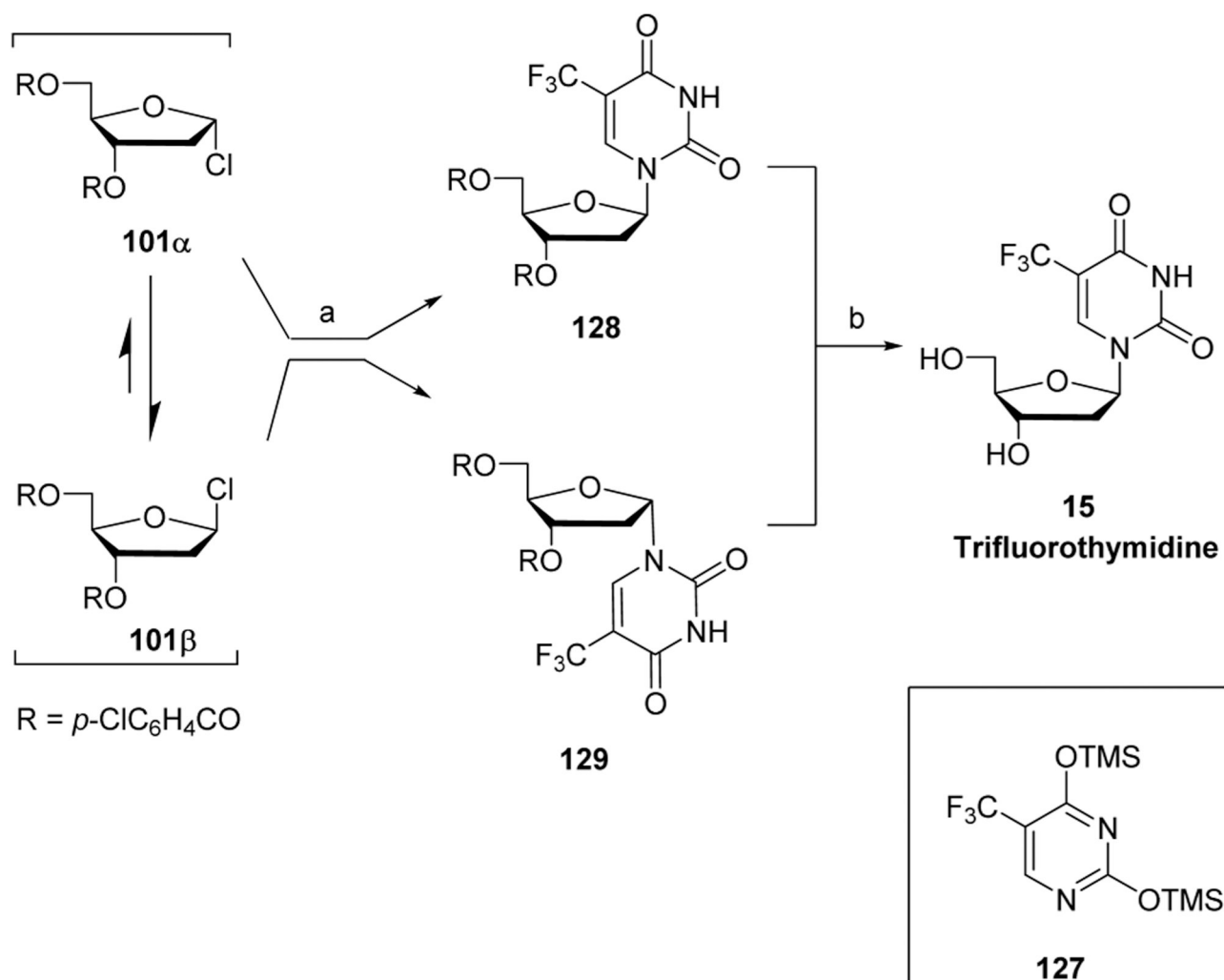
**Scheme 16. Synthesis of Oteracil 21<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) LiOH, I<sub>2</sub>, H<sub>2</sub>O, 5 °C, 5 min, then AcOH, 75%; (b) aq KOH, 20 min, rt, 82%.



**Scheme 17. Synthesis of Trifluorothymidine 15<sup>a</sup>**

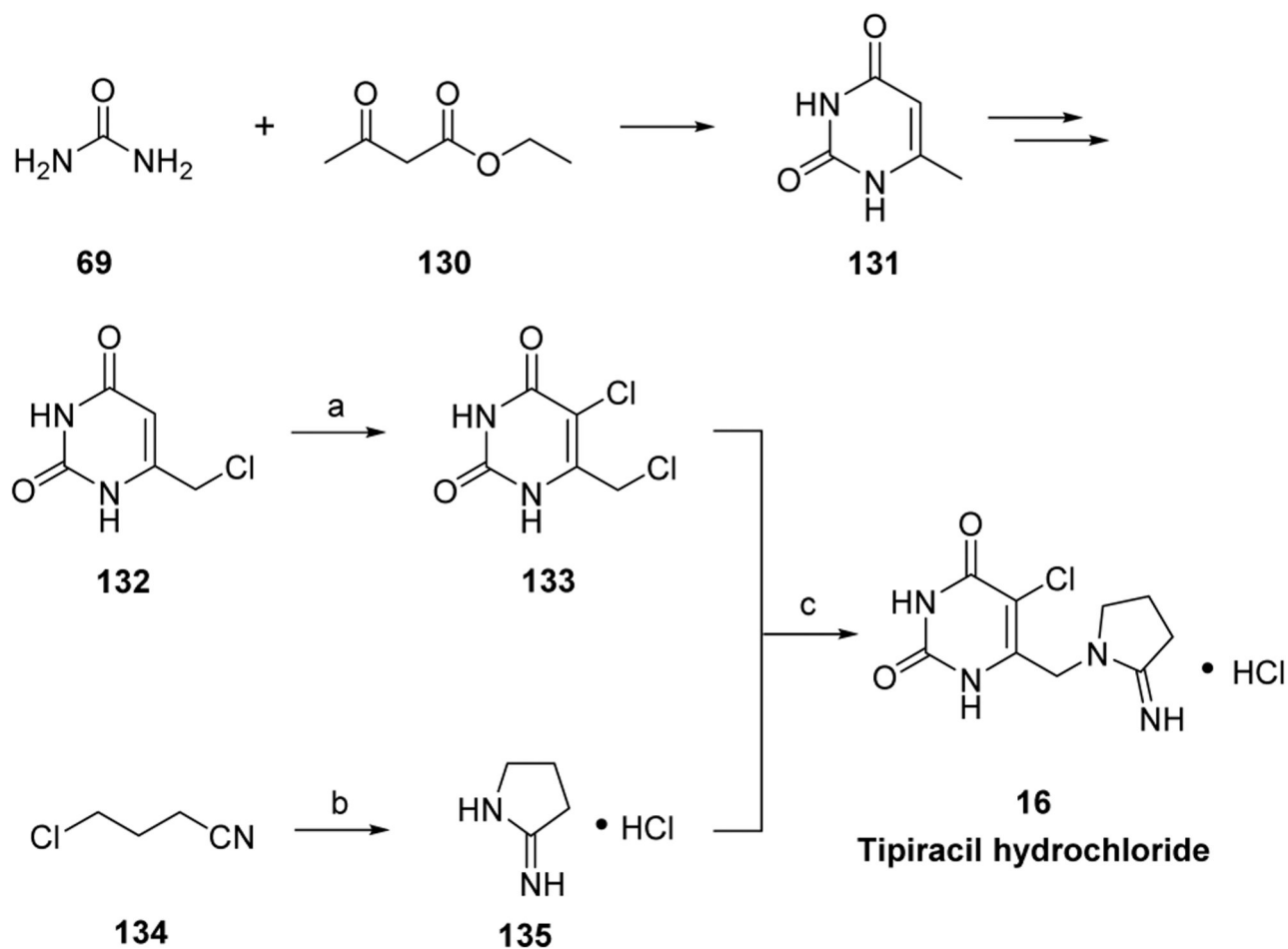
<sup>a</sup>Reagents and conditions: (a) NaCN, H<sub>2</sub>SO<sub>4</sub>, H<sub>2</sub>O, 10 °C, 3 h, 99%; (b) Ac<sub>2</sub>O, concd H<sub>2</sub>SO<sub>4</sub>, reflux, 1 h, 78%; (c) 500 °C, 4 h, 64%; (d) HBr (g), MeOH, 0 °C, 36 h, 82%; (e) (NH<sub>2</sub>)<sub>2</sub>CO, H<sub>2</sub>O, 100 °C, 30 min, 28%; (f) aq HCl, reflux, 1 h, 58%; (g) Br<sub>2</sub>/AcOH, reflux, 3 h, 85%; (h) DMF, 140 °C, 75 min, 80%; (i) thymidine, bactotryptone, NaCl, *Escherichia coli* B, 0.067 M phosphate buffer (pH 6.7), 37 °C, 3.5 h, 14%.



**Scheme 18. Synthesis of Trifluorothymidine 15<sup>a</sup>**

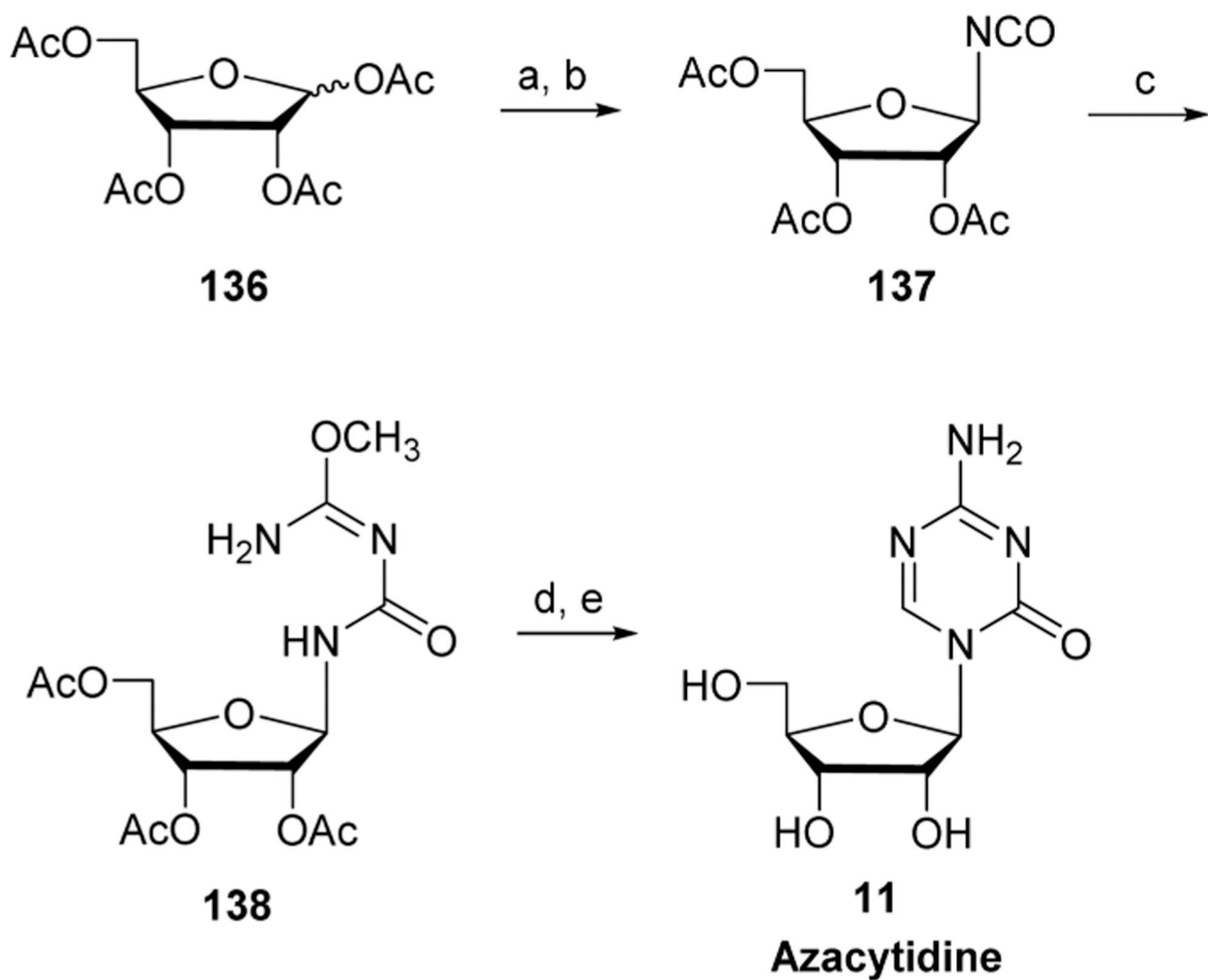
<sup>a</sup>Reagents and conditions: (a) AcCl, MeOH, 25 °C then pyr; (b) pyr, DMAP, 4-ClBzCl, 0 °C, 1 h, then 25 °C, 12 h; (c) HOAc, HCl (g), 0 °C, 7–10 min, 82% (over three steps); (d) 127, anisole, 50 °C, 3.5 h, 85% ( $\beta/\alpha = 85.7:14.3$ ); (e) NaOMe, MeOH, 4 °C, 3.5 h, then AcOBu, 97% ( $\beta/\alpha = 99.92:0.08$ ).



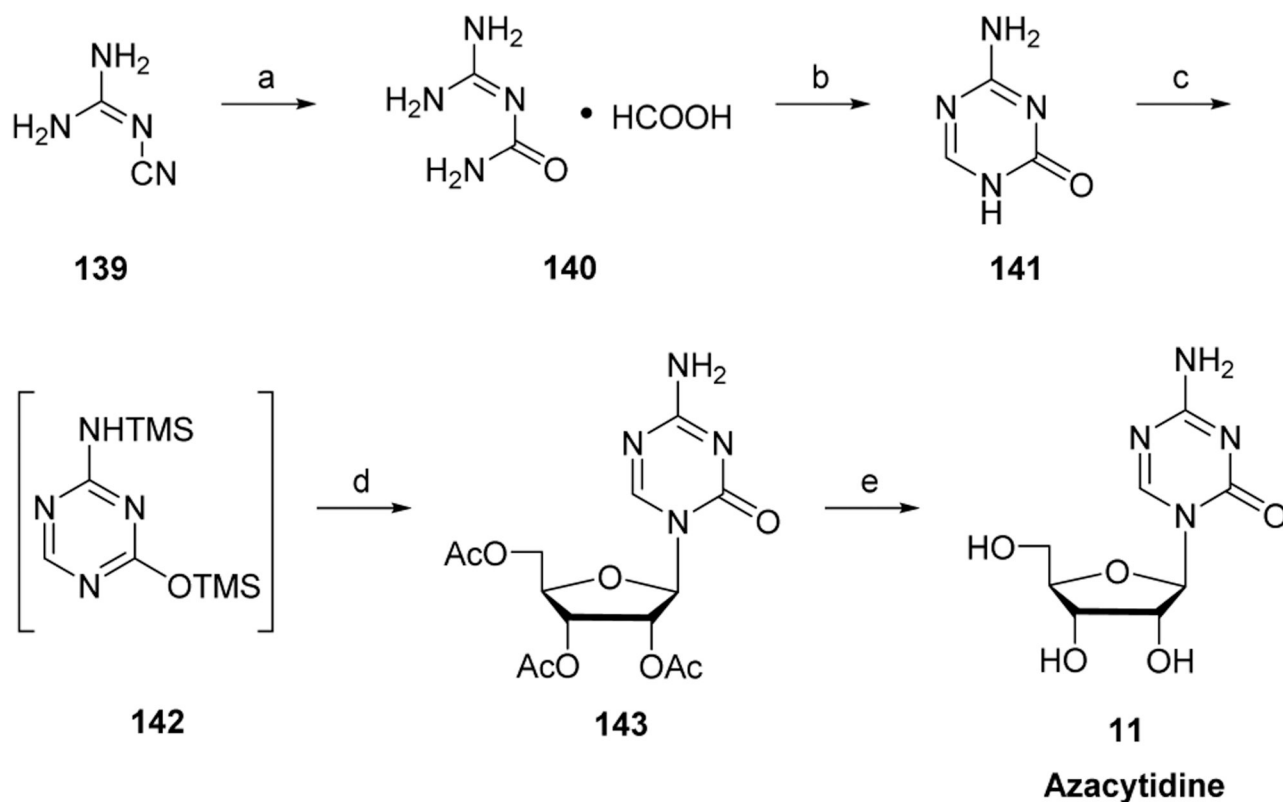


**Scheme 19. Synthesis of Tipiracil Hydrochloride 16<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) SO<sub>2</sub>Cl<sub>2</sub>, AcOH, 50 °C, 2.5 h, 83%; (b) NH<sub>3</sub>/MeOH, 120 °C, 10 h, 83%; (c) NaOEt, DMF, rt, 16 h, then aq HCl, 38%.

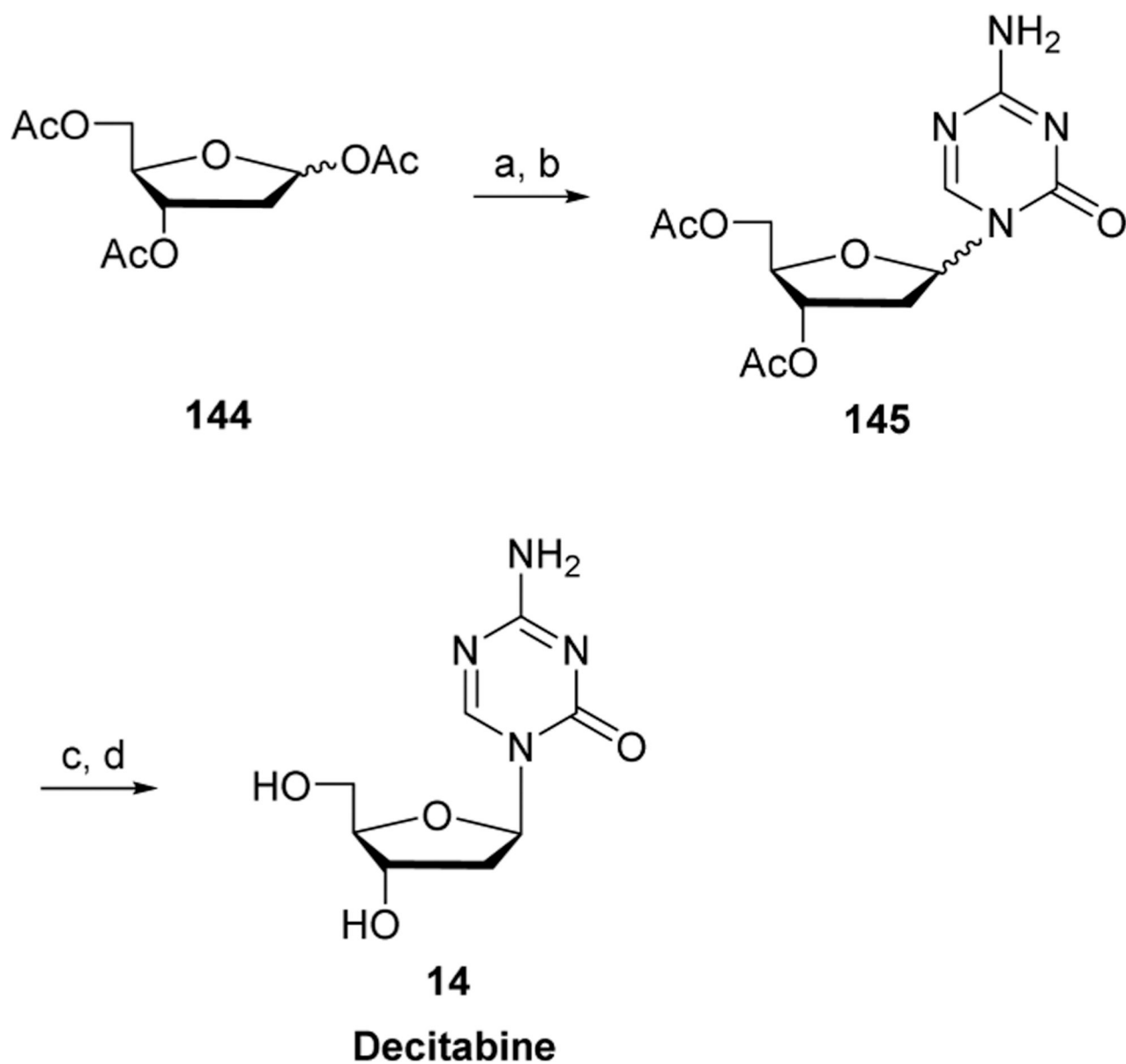
**Scheme 20. Synthesis of Azacytidine 11<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) HCl (g), Ac<sub>2</sub>O, Et<sub>2</sub>O; (b) AgNCO, 60%; (c) 2-methylisourea, 68%; (d) HC(OC<sub>2</sub>H<sub>5</sub>)<sub>3</sub>; (e) NH<sub>3</sub>/MeOH, 40%.



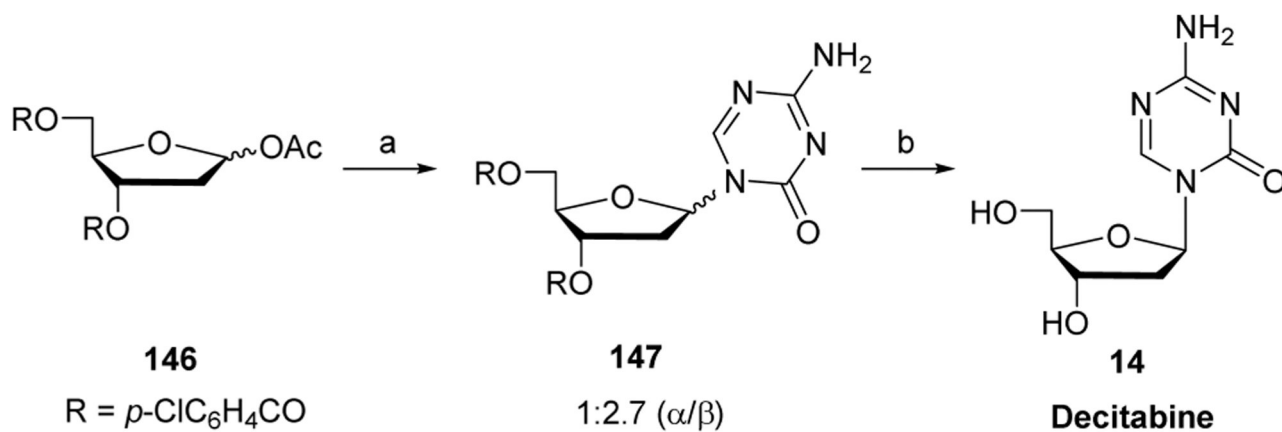
**Scheme 21. Synthesis of Azacytidine 11<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) HCO<sub>2</sub>H, 80 °C, 15 min, 85%; (b) (CH<sub>3</sub>)<sub>2</sub>NCH(OCH<sub>3</sub>)<sub>2</sub>, NaOMe, MeOH, 40–50 °C, 4 h, 90%; (c) HMDS, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, CH<sub>3</sub>CN, reflux, 4–8 h; (d) 1,2,3,5-tetra-*O*-acetyl-β-D-ribofuranose, CF<sub>3</sub>SO<sub>3</sub>H, EtOAc, 45 °C, 30 min, then HCl, DCM, rt, 45 min; (e) *n*-BuNH<sub>2</sub>, MeOH, 65 °C, 1 h, 49% (over three steps).

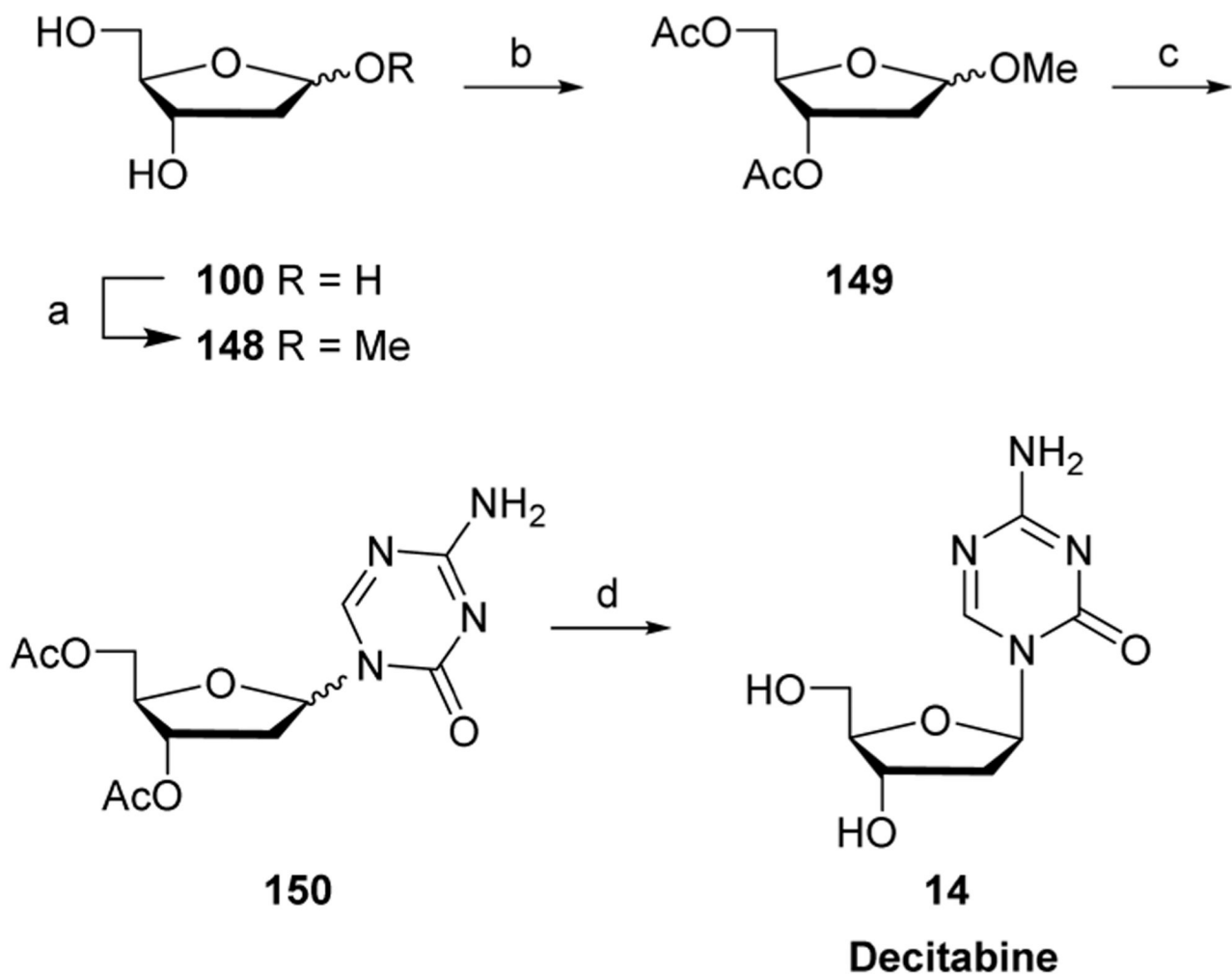


**Scheme 22. Early Synthesis of Decitabine 14<sup>a</sup>**

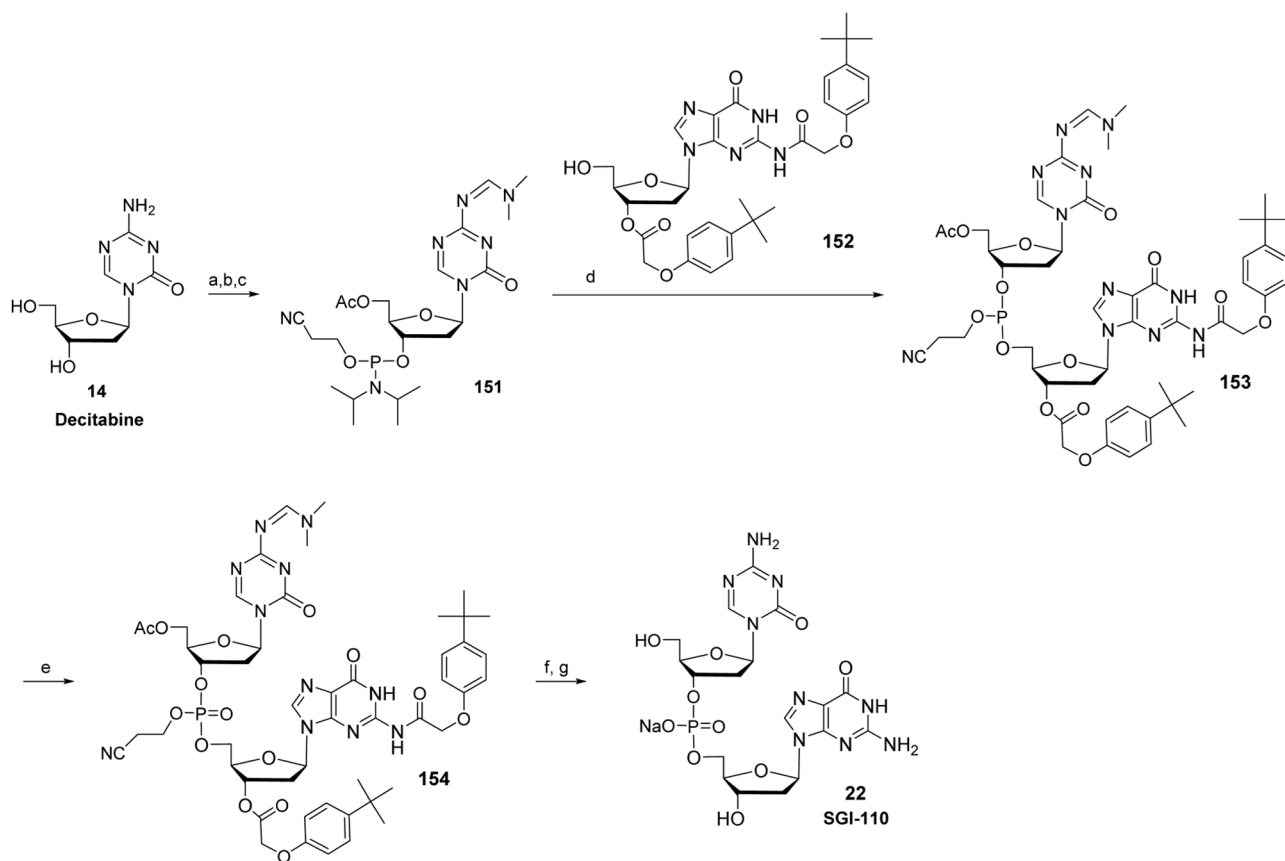
<sup>a</sup>Reagents and conditions: (a) AcCl, HCl, Et<sub>2</sub>O, 0 °C; (b) bis(trimethylsilyl)-5-azacytosine 142, CH<sub>3</sub>CN, 10%; (c) NH<sub>3</sub>/EtOH; (d) anomer separation, 7%.

**Scheme 23. Kilogram-Scale Synthesis of Decitabine 14<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) bis(trimethylsilyl)-5-azacytosine **142**, TMSOTf, DCM, cold, 55%; (b) NaOMe/MeOH, 65%.

**Scheme 24. Synthesis of Decitabine 14<sup>a</sup>**

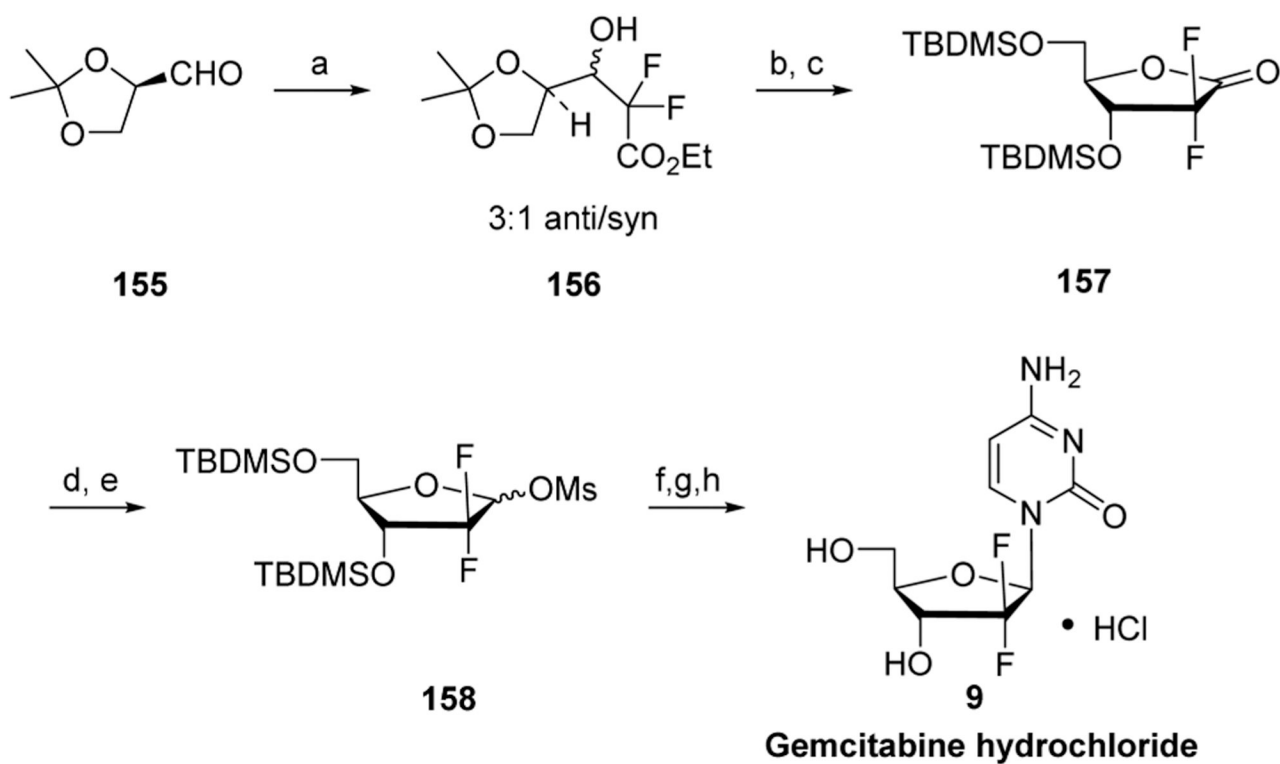
<sup>a</sup>Reagents and conditions: (a) AcCl/MeOH, 99%; (b) AcCl/pyr/DCM, 78%; (c) 5-azacytosine, HMDS, NH<sub>4</sub>SO<sub>4</sub>, TMSOTf, 60% ( $\beta/\alpha = 1:1$ ); (d) NH<sub>3</sub>/MeOH, 34%.



**Scheme 25. Synthesis of SGI-110 22<sup>a</sup>**

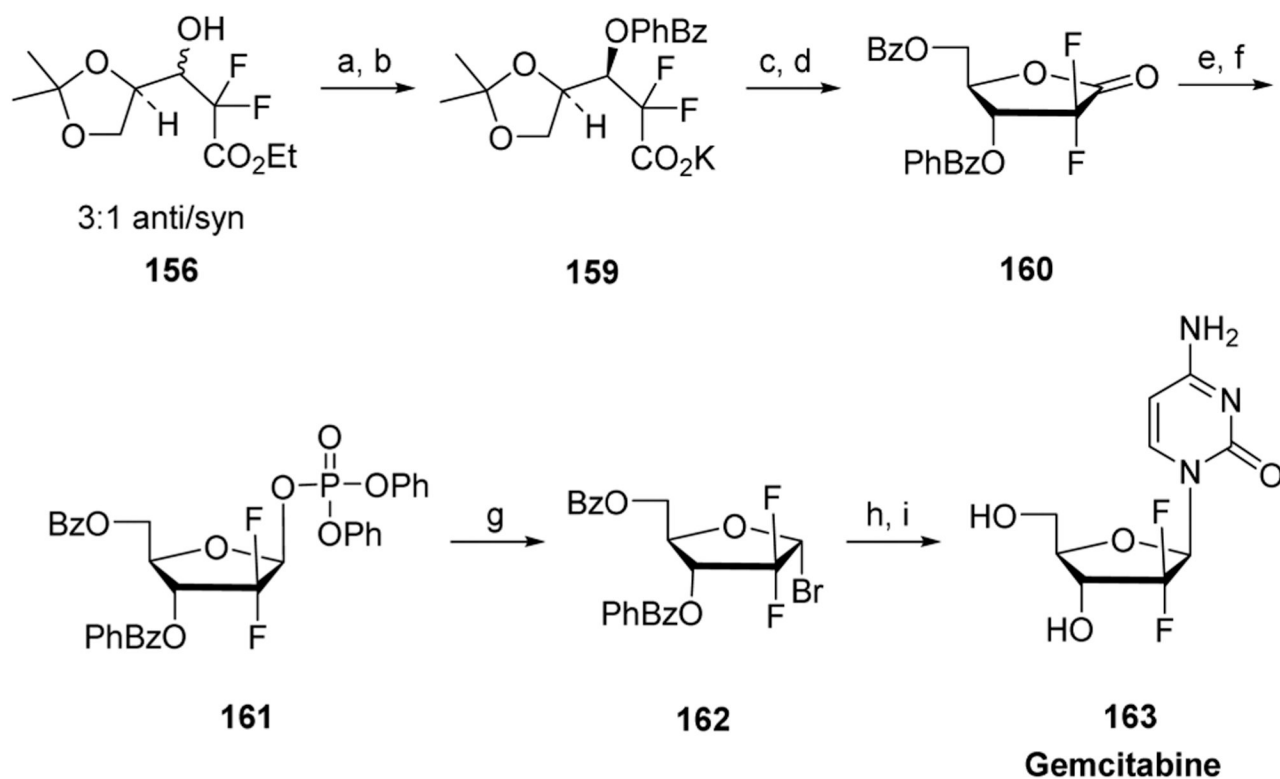
<sup>a</sup>Reagents and conditions: (a) DMF, dimethylacetal; (b) vinyl acetate, Lipozyme RM IM, CH<sub>3</sub>CN, dioxane; (c) [(*i*Pr)<sub>2</sub>N]POCH<sub>2</sub>CH<sub>2</sub>CN, DCM; (d) DCM; (e) *t*BuOOH; (f) NH<sub>3</sub>/MeOH; (g) NaOAc, H<sub>2</sub>O, EtOH.



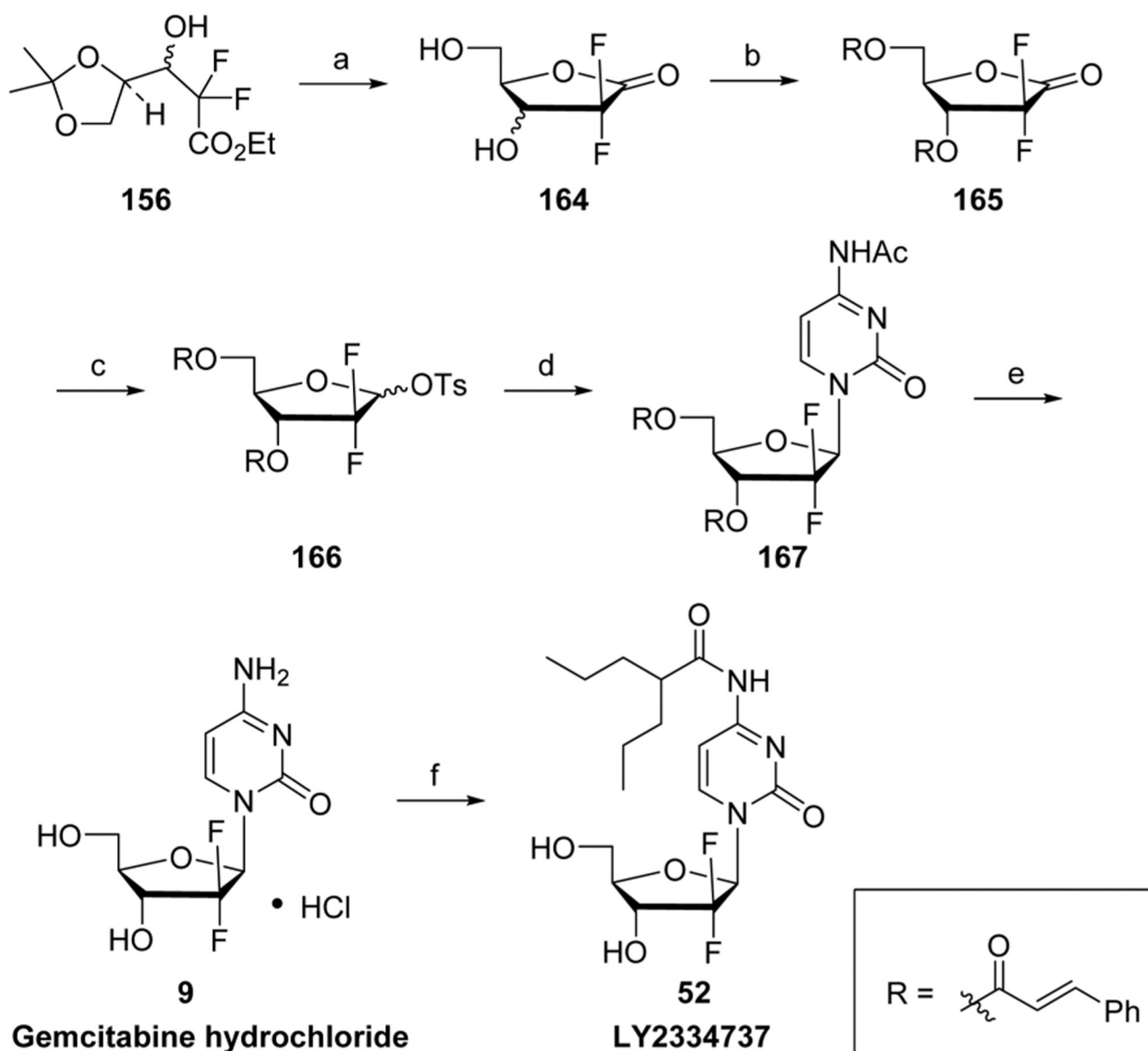


**Scheme 26. Synthesis of Gemcitabine Hydrochloride <sup>9a</sup>**

<sup>a</sup>Reagents and conditions: (a) BrCF<sub>2</sub>CO<sub>2</sub>Et, Zn, THF, Et<sub>2</sub>O, 87% (3:1 anti/syn); (b) Dowex 50, MeOH, H<sub>2</sub>O; (c) TBDMSOTf, lutidine, DCM, 86% (over two steps); (d) DIBAL-H; (e) MsCl, Et<sub>3</sub>N, DCM, 71% (over two steps); (f) 2,4-bis(trimethylsilyl)cytosine, TMSOTf, DCE, reflux; (g) AG 50W-X8 resin, MeOH; (h) reverse phase HPLC, 10% (over two steps).

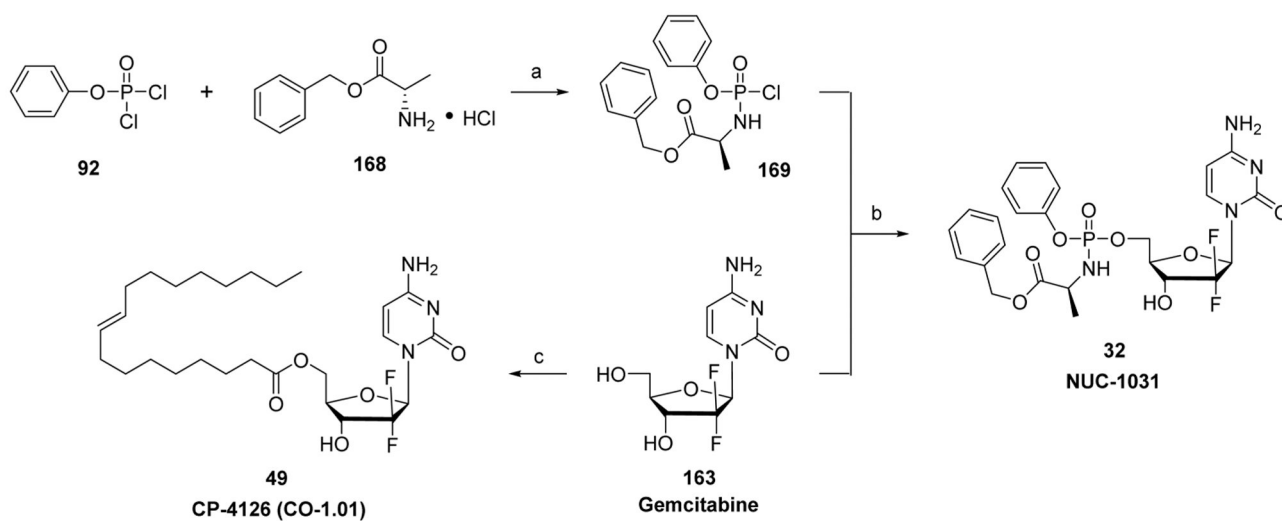
**Scheme 27. Synthesis of Gemcitabine 163<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) PhBzCl, Et<sub>3</sub>N, DCM; (b) K<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O, THF, MeOH, 67%; (c) HCl, CH<sub>3</sub>CN, reflux; (d) BzCl/Pyr, EtOAc, rt, 72%; (e) LiAl(*O*-*t*Bu)<sub>3</sub>H, THF; (f) ClP(O)(OPh)<sub>2</sub>, Et<sub>3</sub>N, PhMe, rt, 77%; (g) HBr/HOAc, rt, 82%; (h) 2,4-bis(trimethylsilyl)cytosine, octane/heptane, Ph<sub>2</sub>O, 140–150 °C; (i) NH<sub>3</sub>/MeOH, H<sub>2</sub>O, rt, 65%.



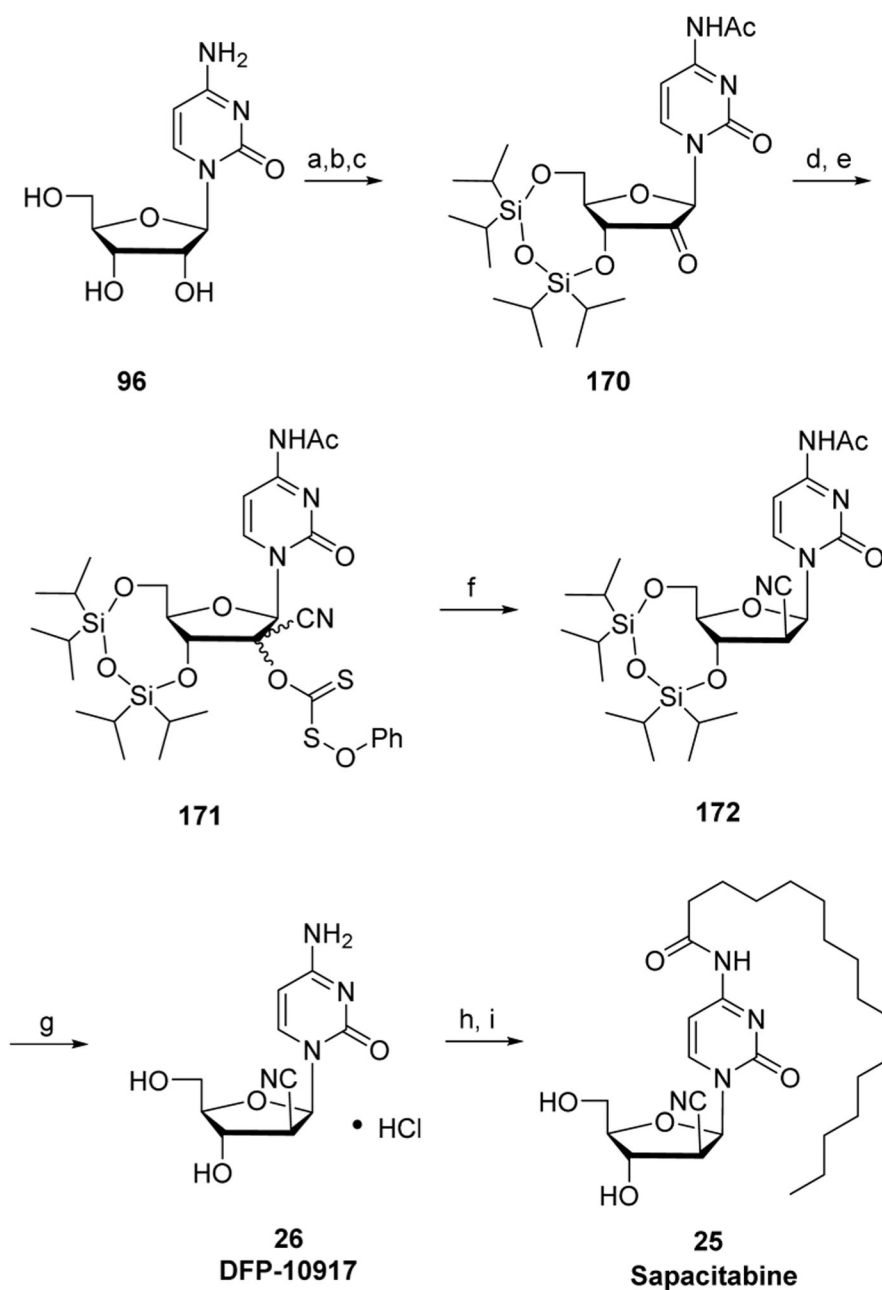
**Scheme 28. Synthesis of Gemcitabine Hydrochloride 9 and LY2334737 52<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Dowex 50W-X12 resin, MeOH, H<sub>2</sub>O, rt, 4 days, 94%; (b) (*E*)-cinnamoyl chloride, pyr, EtOAc, 30 °C, 3 h, 43%; (c) LiAl(*O*-*t*Bu)<sub>3</sub>H, THF, -10 °C, 2 h, then TsCl, Et<sub>3</sub>N, PhMe, -10 °C, 5 h, 62%; (d) *N*-acetyl cytosine, (TMS)<sub>2</sub>NH, TMSOTf, DCE, reflux, 12 h, then 5% NaHCO<sub>3</sub>, 47%; (e) NH<sub>3</sub>/MeOH, rt, overnight, then HCl, Me<sub>2</sub>CO, rt, 12 h, 80%; (f) 2-propylpentanoic acid, EDC, HOBT, NMM, DMF, DMSO, 55 °C, 17 h, 95%.



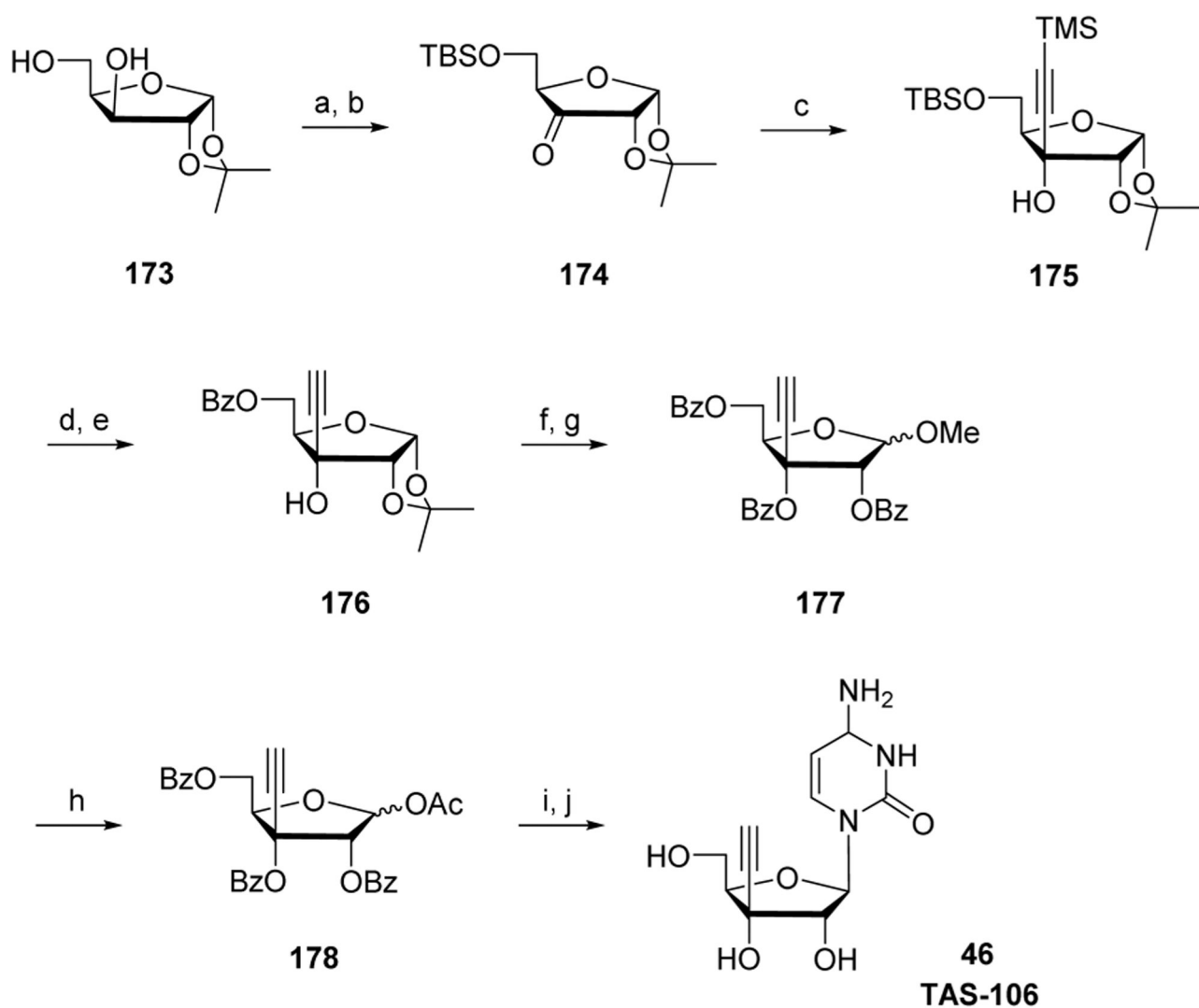
**Scheme 29. Synthesis of CP-4126 49 and NUC-1031 32<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Et<sub>3</sub>N, DCM, -80 °C, 1 h, 98%, (b) NMI, pyr, THF, -80 °C to rt, 2 h, 16%; (c) HCl (g), DMF, elaidic acid chloride, rt, 12 h, 30%.



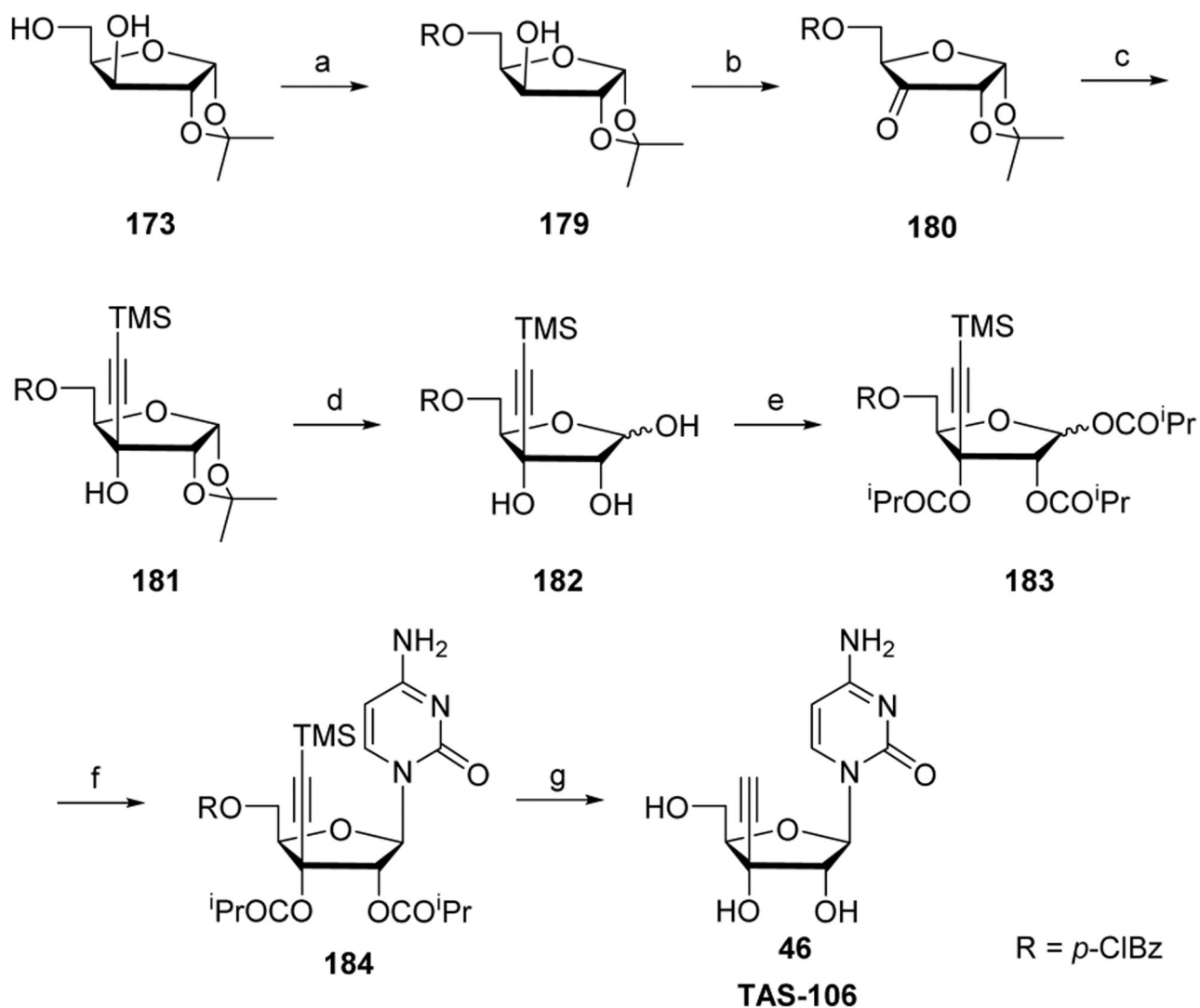
**Scheme 30. Discovery Synthesis of DFP-10917 26 and Process Synthesis of Sapacitabine 25<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 1,3-dichloro-1,1,3,3-tetraisopropylidisiloxane, pyr; (b) Ac<sub>2</sub>O, EtOH; (c) CrO<sub>3</sub>/pyr/Ac<sub>2</sub>O, DCM; (d) NaCN, NaHCO<sub>3</sub>, Et<sub>2</sub>O, H<sub>2</sub>O; (e) PhOC(S)Cl, DMAP, CH<sub>3</sub>CN; (f) Bu<sub>3</sub>SnH, AIBN, PhMe (57% over 3 steps); (g) Bu<sub>4</sub>NF, CH<sub>3</sub>CO<sub>2</sub>H, THF then HCl/MeOH, 49% (over two steps); (h) Et<sub>3</sub>N, MeOH, DCM, 30 °C, 10 min, 97%; (i) palmitic anhydride, dioxane, H<sub>2</sub>O, 85 °C, 89%.



**Scheme 31. Synthesis of TAS-106 46<sup>a</sup>**

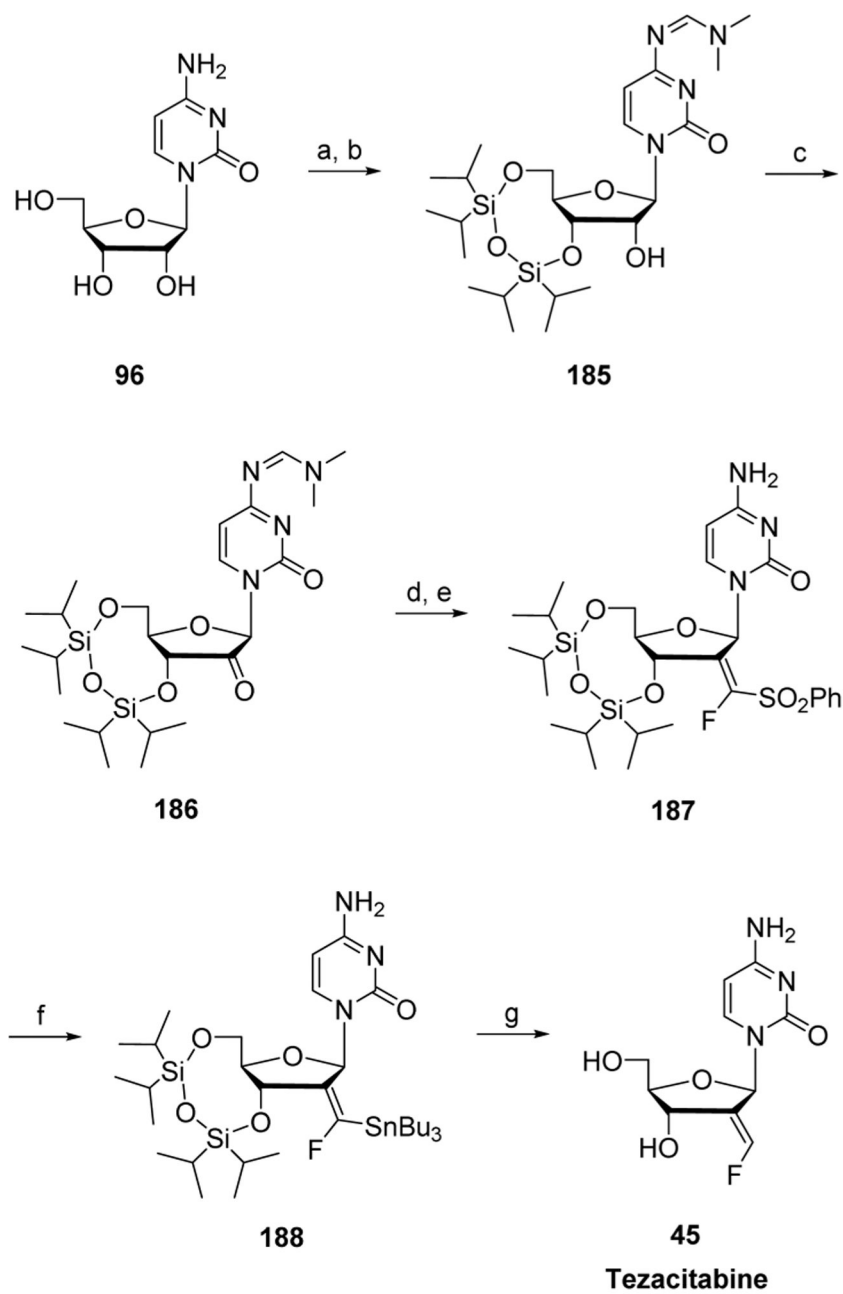
<sup>a</sup>Reagents and conditions: (a) TBDMSCl, pyr, rt, 1.5 h, 100%; (b) CrO<sub>3</sub>, Ac<sub>2</sub>O, pyr, rt, 1.5 h, 77%; (c) trimethylsilylacetylene, *n*-BuLi, THF, -78 °C, 99%; (d) *n*-Bu<sub>4</sub>NF, THF, rt; (e) BzCl, pyr, rt, 93% (over two steps); (f) 20% HCl, MeOH, rt; (g) BzCl, DMAP, pyr, 100 °C, 87% (over two steps); (h) concd H<sub>2</sub>SO<sub>4</sub>, AcOH, Ac<sub>2</sub>O, rt, 93%; (i) 2,4-bis(trimethylsilyl)cytosine, SnCl<sub>4</sub>, CH<sub>3</sub>CN, 0 °C to rt, 18 h, 81%; (j) NH<sub>3</sub>/MeOH, rt, 2 days, 90%.



**Scheme 32. Synthesis of TAS-106 46<sup>a</sup>**

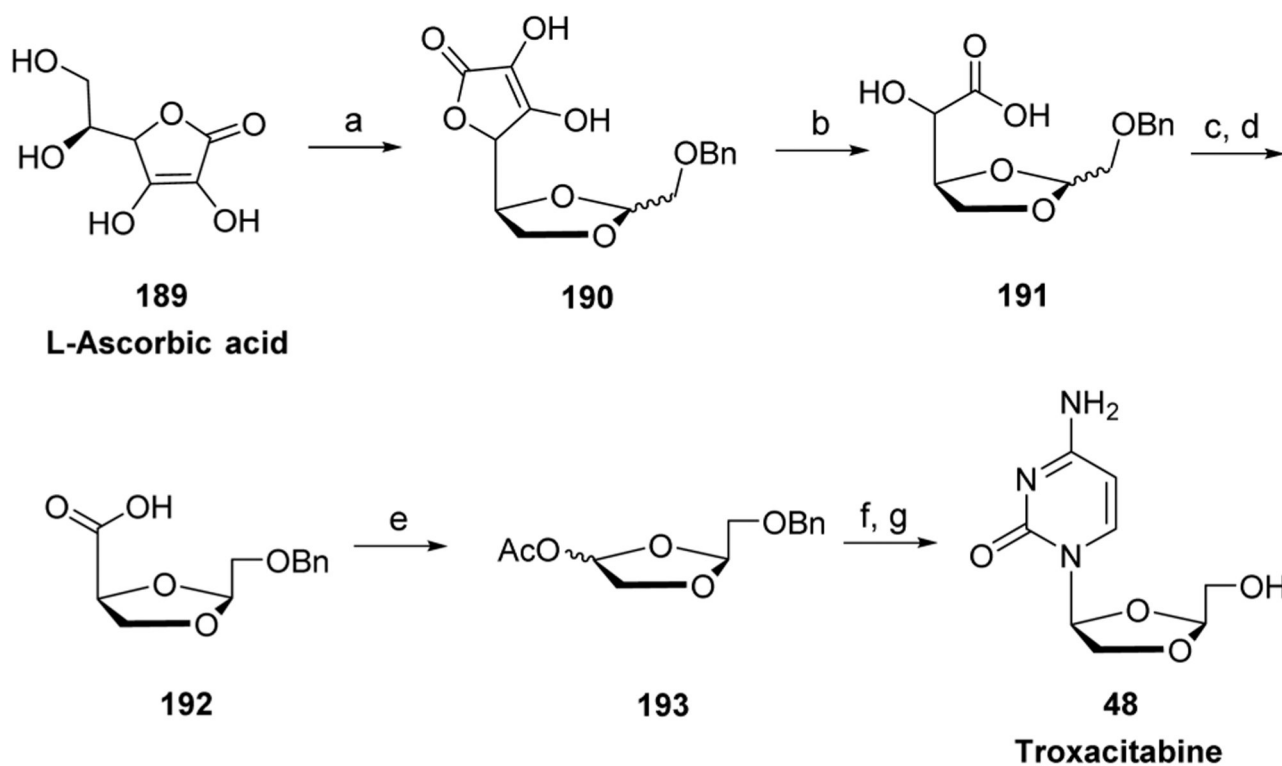
<sup>a</sup>Reagents and conditions: (a) 4-ClBzCl, Et<sub>3</sub>N, DCM, 0 °C, 2 h, 74%; (b) TEMPO, DCM NaClO<sub>4</sub>, NaHCO<sub>3</sub>, H<sub>2</sub>O, 0 °C, 30 min, then 2-propanol, rt, 10 min, 88%; (c) trimethylsilylacetylene, EtMgBr, THF, 4 °C, 30 min, 25%; (d) HCO<sub>2</sub>H, H<sub>2</sub>O, reflux, 45 min, 97%; (e) *i*-BuCl, Et<sub>3</sub>N, DMAP, rt, 5 h, 87%; (f) 2,4-bis(trimethylsilyl)cytosine, SnCl<sub>4</sub>, MeCN, 30 °C, 3 h, 81%; (g) DBU, MeOH, 30 °C, 3 h, 68%.



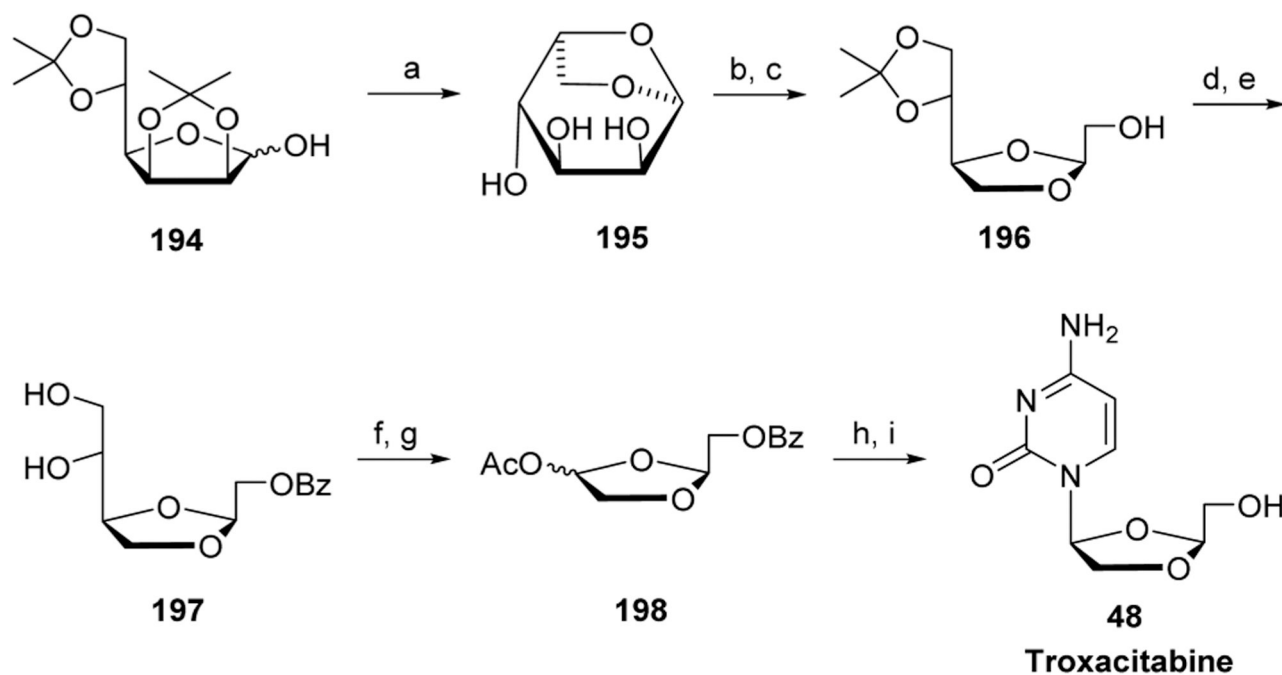


**Scheme 33. Synthesis of Tezacitabine 45<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) TIPSiCl, pyr, 34 °C, 18 h; (b) (CH<sub>3</sub>)<sub>2</sub>NCH(OCH<sub>3</sub>)<sub>2</sub>, 37 °C, 4 h, 85% (over two steps); (c) (COCl)<sub>2</sub>, DMSO, DCM, -75 °C, 0.5 h, then Et<sub>3</sub>N, -75 °C to rt, 1 h, 90%; (d) PhSO<sub>2</sub>CH<sub>2</sub>F, CIP(O)(OEt)<sub>2</sub>, 1 M LiHMDS, -60 °C to rt, 3 h; (e) NH<sub>3</sub>/MeOH, rt, overnight, 66% (over two steps); (f) Bu<sub>3</sub>SnH, AIBN, cyclohexane, reflux, 18 h, 83%; (g) CsF, MeOH, reflux, 24 h, 69%.

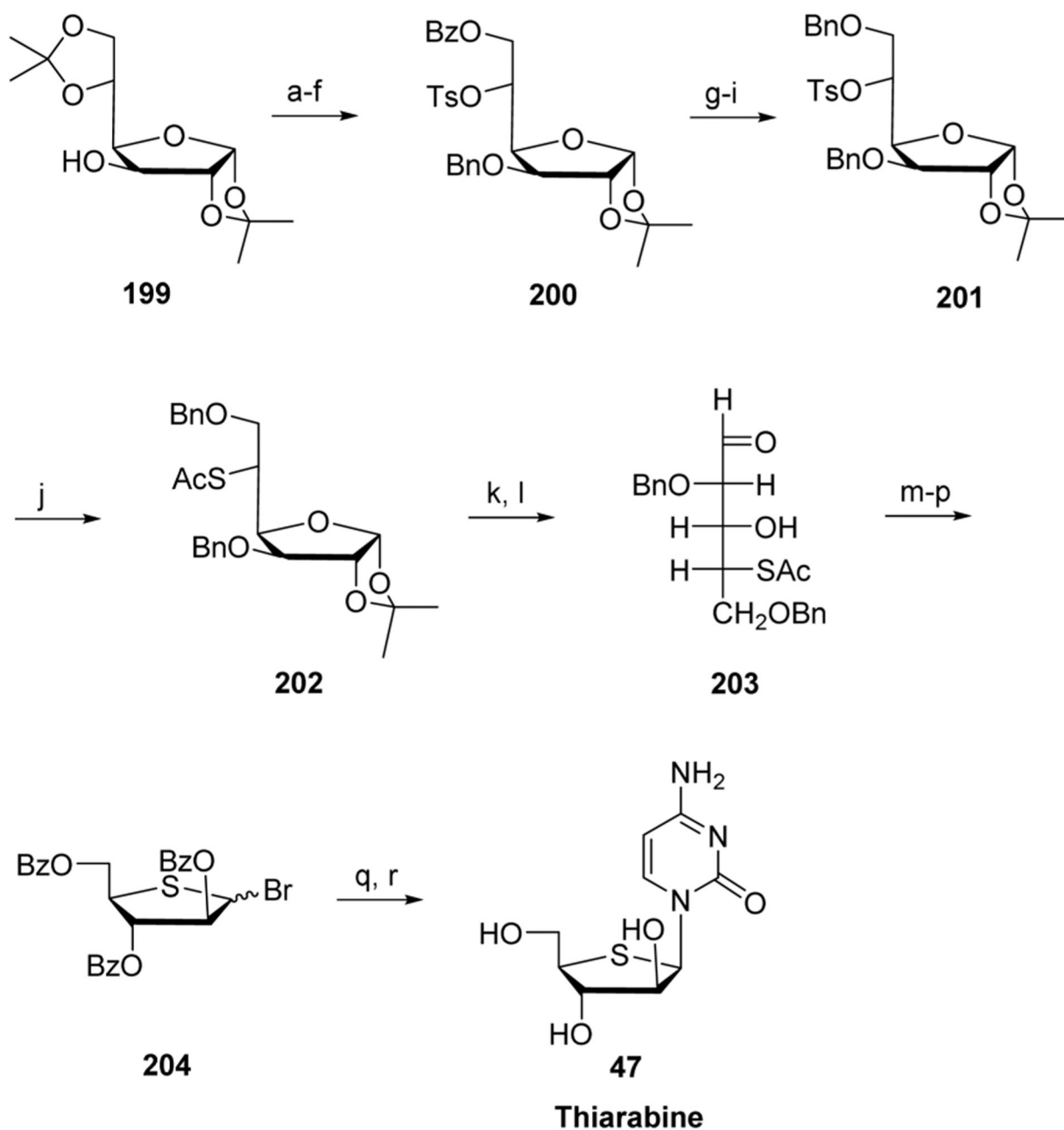
**Scheme 34. Synthesis of Troxacitabine 48<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) PhCH<sub>2</sub>OCH<sub>2</sub>CH(OMe)<sub>2</sub>, TsOH, CH<sub>3</sub>CN, 95%; (b) H<sub>2</sub>O<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, EtOH; (c) RuCl<sub>3</sub> hydrate, NaOCl, DCE/MeCN/H<sub>2</sub>O, BnEt<sub>3</sub>NCl; (d) H<sup>+</sup>, 51% (over three steps); (e) Pb(OAc)<sub>4</sub>, MeCN/DCM, pyr, 80%; (f) silylated *N*-acetylcytosine, TMSOTf, DCM, rt; (g) deprotection, 30%.



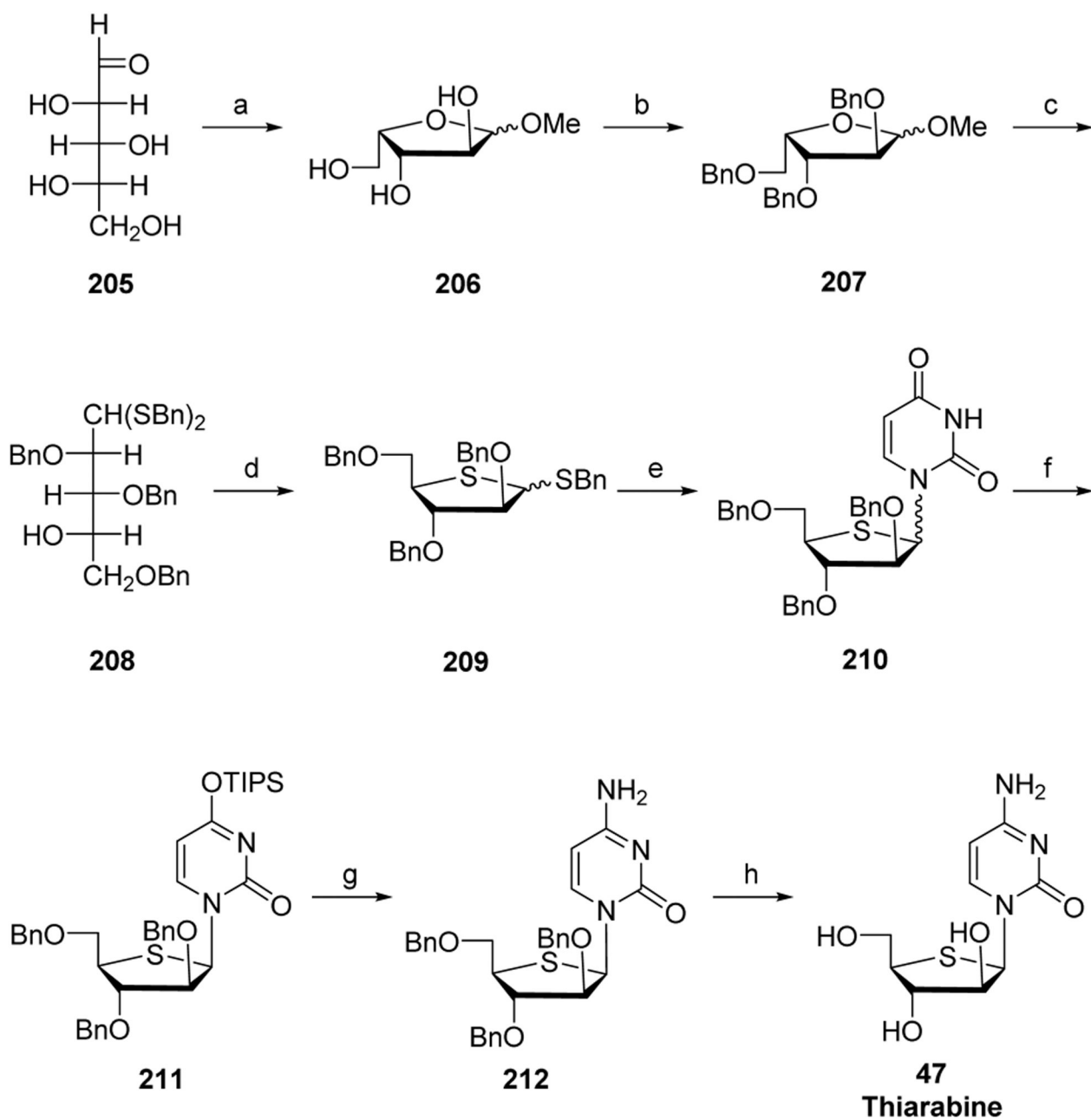
**Scheme 35. Synthesis of Troxacitabine 48<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) aq HCl, reflux, 20 h, 60%; (b) NaIO<sub>4</sub>, MeOH, H<sub>2</sub>O, then NaBH<sub>4</sub>; (c) TsOH, Me<sub>2</sub>CO, 6 h, 62%; (d) BzCl, pyr, DCM, rt, 2 h; (e) TsOH, MeOH, rt, 2 h, 83% (over two steps); (f) NaIO<sub>4</sub>, RuO<sub>2</sub>·H<sub>2</sub>O, CH<sub>3</sub>CN:CCl<sub>4</sub>:H<sub>2</sub>O (1:1:1.5 v/v), rt, 5 h; (g) Pb(OAc)<sub>4</sub>, pyr, THF, rt, 45 min, 63% (over two steps); (h) *N*<sup>4</sup>-Bzcytosine, HMDS, NH<sub>4</sub>SO<sub>4</sub>, DCE, reflux, 2.5 h, then TMSOTf, DCE, rt, 1.5 h; (i) NH<sub>3</sub>/MeOH, 0 °C, rt, 72 h, 38% (over two steps).



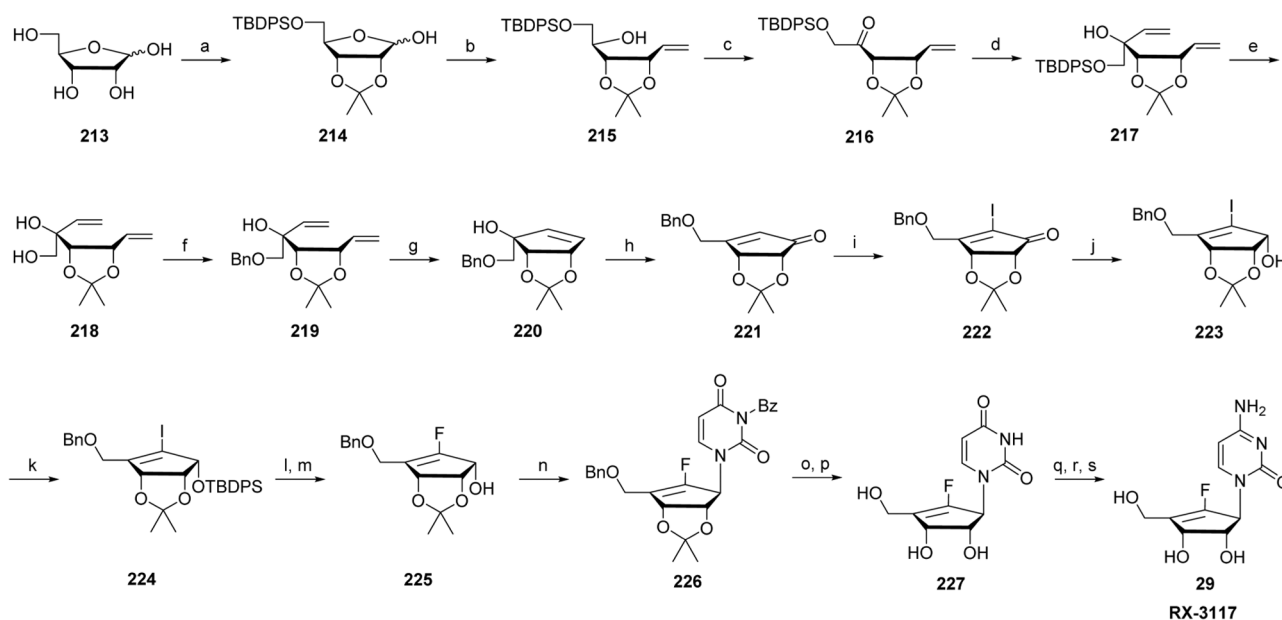
**Scheme 36. Synthesis of Thiarabine 47<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Na, ether, reflux, 16 h, then BnBr, 70 °C, 16 h; (b) Ac<sub>2</sub>O, H<sub>2</sub>O, 40 °C, 16 h; (c) AcCl, pyr, 25 °C, 16 h, 78% (over three steps); (d) Na, MeOH, 16 h; (e) BzCl, pyr, -15–25 °C, 20 h; (f) TsCl, pyr, CHCl<sub>3</sub>, 40 °C, 36 h, 92% (over three steps); (g) NaOMe, CHCl<sub>3</sub>, MeOH, -15 to 0 °C, 2 h, 87%; (h) NaOBn, BnOH, 40 °C; (i) TsCl; (j) KSC(O)CH<sub>3</sub>, DMF, 115 °C, 73%; (k) HOAc, H<sub>2</sub>O, 70 °C, 36 h, 78%; (l) NaIO<sub>4</sub>, EtOH, H<sub>2</sub>O, 30 °C, 0.5 h, 97%; (m) aq HCl, MeOH, reflux, 3 h, 41%; (n) Na, NH<sub>3</sub>, ~100%; (o) BnCl; (p) HBr/HOAc; (q) bis(trimethylsilyl)-*N*-acetylcytosine, 130 °C, 3 h, 20%; (r) NH<sub>3</sub>/MeOH, 0 °C, 2 days, 89%.



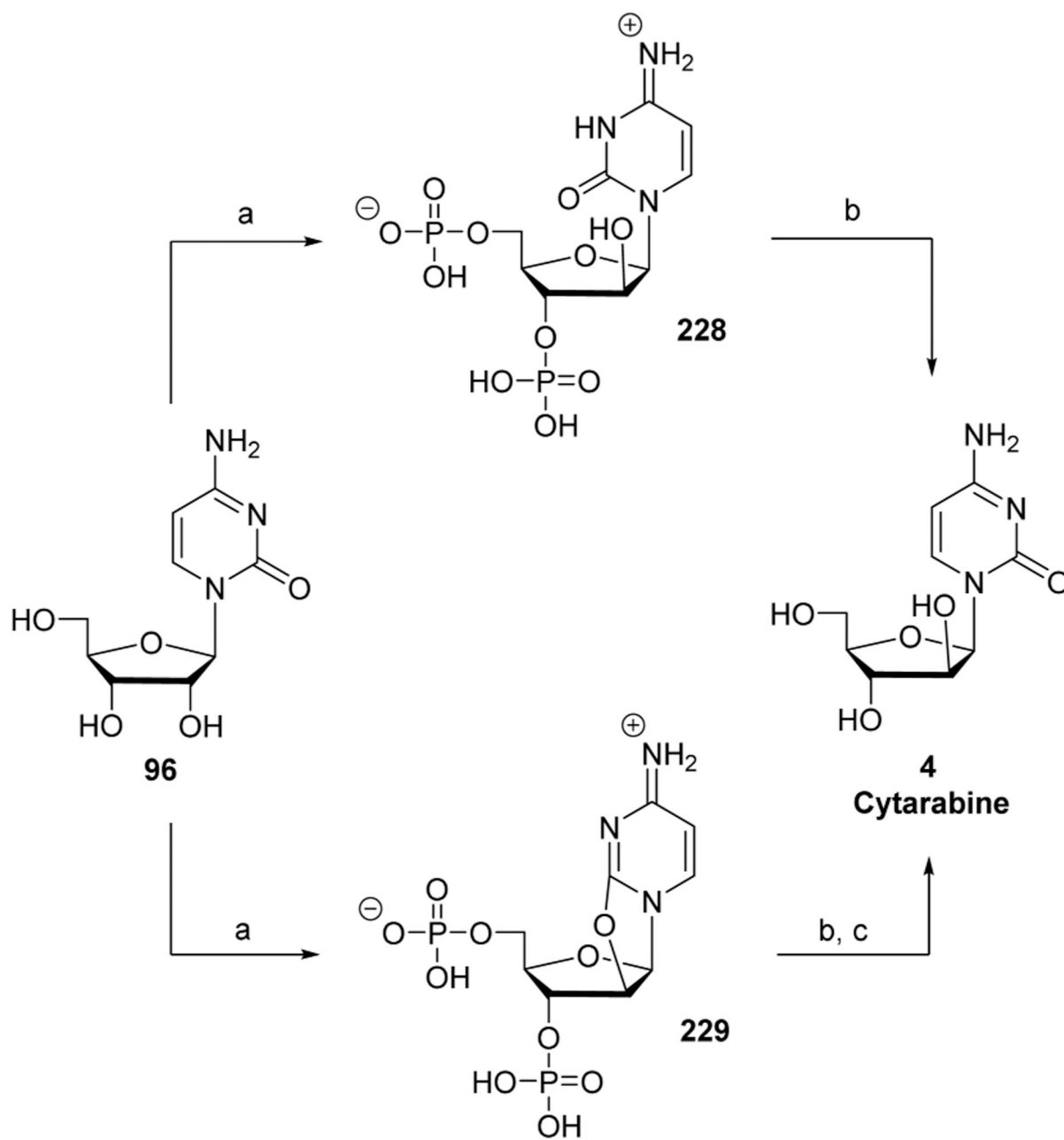
**Scheme 37. Synthesis of Thiarabine 47<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) HCl, MeOH, rt, 5 h, 95%; (b) BnBr, NaH, TBAI, THF, rt, 3 days, 87%; (c) BnSH, SnCl<sub>4</sub>, DCM, rt, overnight, 57%; (d) Ph<sub>3</sub>P, imidazole, I<sub>2</sub>, PhMe, CH<sub>3</sub>CN, 90 °C, 24 h, 83%; (e) uracil, BSA, NBS, MeCN, 50–55 °C, 18 h, column purification ( $\alpha:\beta = 1:1.15$ ), 36%; (f) TPSCl, Et<sub>3</sub>N, DMAP, 5 °C to rt, overnight; (g) NH<sub>4</sub>OH, MeCN, rt, overnight, 50%.



### Scheme 38. Synthesis of RX-3117 29<sup>a</sup>

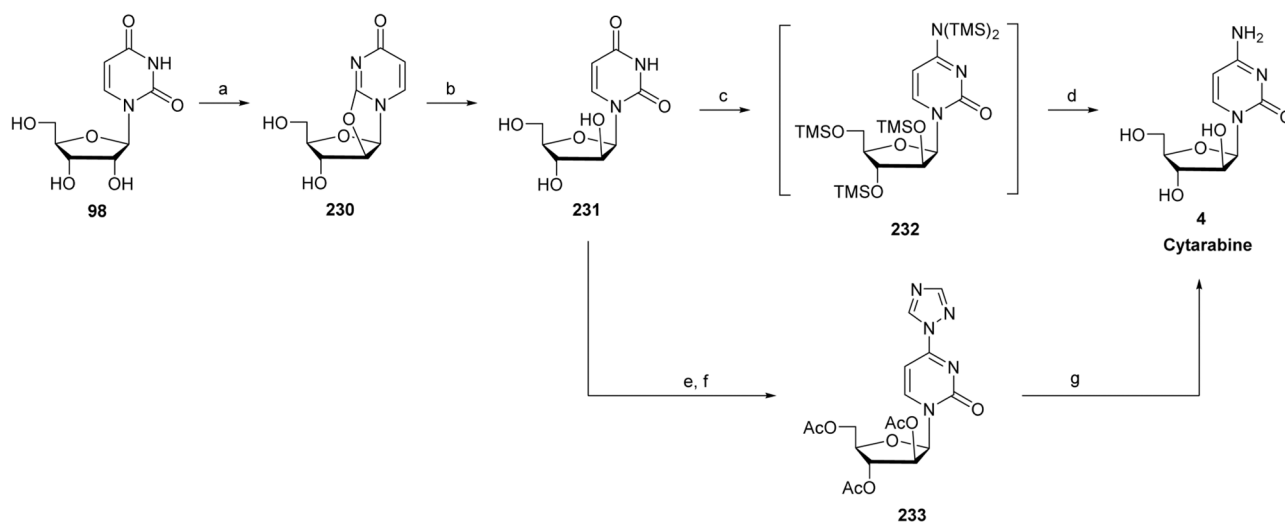
<sup>a</sup>Reagents and conditions: (a) concd H<sub>2</sub>SO<sub>4</sub>, Me<sub>2</sub>CO, then TBDPSCl, imidazole, 99%; (b) Ph<sub>3</sub>PCH<sub>3</sub>Br, *t*-BuOK, THF, 93%; (c) (COCl)<sub>2</sub>, DMSO, DCM, -78 °C, 1 h, then Et<sub>3</sub>N, rt, 96%; (d) CH<sub>2</sub>=CHMgBr, THF, -78 °C, 84%; (e) *n*-Bu<sub>4</sub>NF, THF, 97%; (f) Bu<sub>2</sub>Sn(O), PhMe, 15 h, then TBAI, BnBr, 50 °C, 16 h, 76%; (g) Grubbs' catalyst (2<sup>nd</sup> generation), PhMe, 80 °C, 90%; (h) PDC, DMF, rt, 18 h, 59%; (i) I<sub>2</sub>, pyr, THF, 55%; (j) NaBH<sub>4</sub>, CeCl<sub>3</sub>, MeOH, 93%; (k) TBDPSCl, imidazole, DMF, 97%; (l) NFSI, *n*-BuLi, THF, -78 °C; (m) *n*-Bu<sub>4</sub>NF, THF, 63% (over two steps); (n) *N*<sup>3</sup>-benzoyluracil, DEAD, Ph<sub>3</sub>P, THF, 87%; (o) NH<sub>3</sub>/MeOH; (p) BBr<sub>3</sub>, DCM, -78 °C, 50%; (q) Ac<sub>2</sub>O, pyr, then POCl<sub>3</sub>, Et<sub>3</sub>N, 1,2,4-triazole; (r) NH<sub>4</sub>OH, dioxane; (s) NH<sub>3</sub>/MeOH, 40% (over three steps).



**Scheme 39. Synthesis of Cytarabine 4<sup>a</sup>**

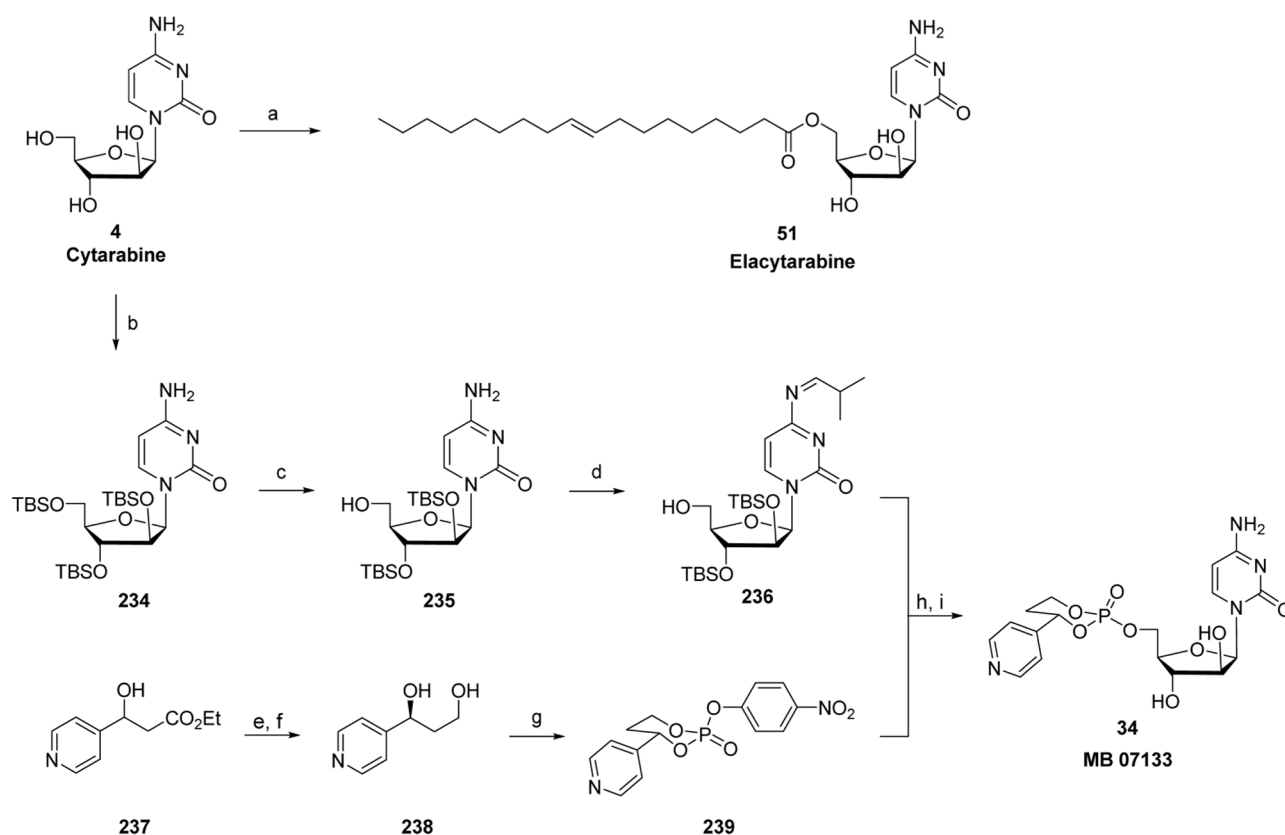
<sup>a</sup>Reagents and conditions: (a) polyphosphoric acid; (b) prostatic phosphatase; (c) OH<sup>-</sup>.





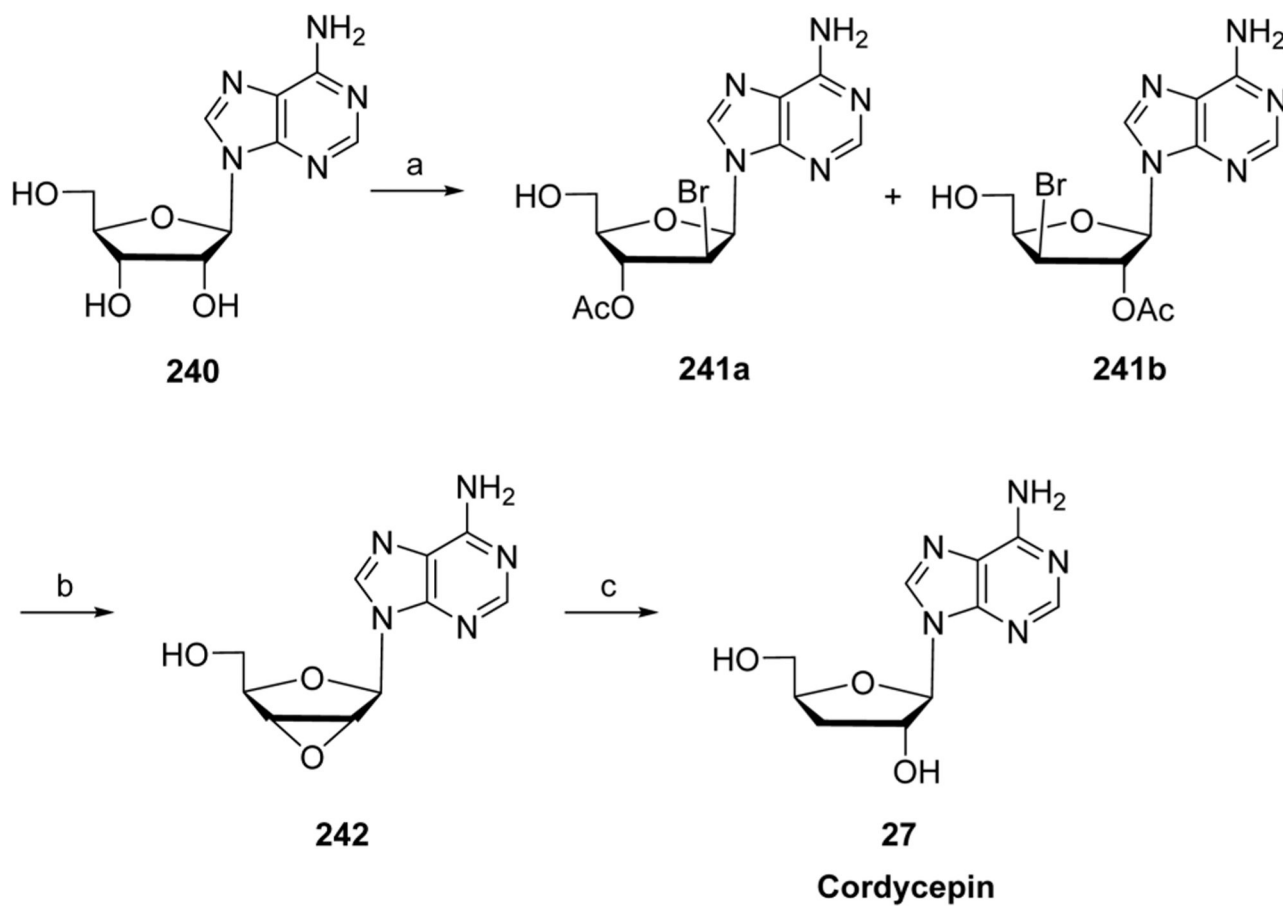
**Scheme 40. Synthesis of Cytarabine 4<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $(\text{PhCO})_2\text{O}$ , DMF, 100 °C, then  $\text{K}_2\text{CO}_3$ , 137 °C, 1.5 h, 78%; (b) aq HCl, 80 °C, 2 h, 68%; (c) HMDS, high pressure, 150 °C, 80 h; (d) MeOH, 5 °C, 1 h, 87% (over two steps); (e)  $\text{Ac}_2\text{O}$ , pyr, rt, 3 h, 91%; (f)  $\text{POCl}_3$ , 1H-1,2,4-triazole,  $\text{Et}_3\text{N}$ ,  $\text{CHCl}_3$ , <8 °C then rt, overnight, 84%; (g)  $\text{NH}_4\text{OH}$ ,  $\text{NH}_3$ , dioxane,  $\text{H}_2\text{O}$ , rt, overnight, 70%.

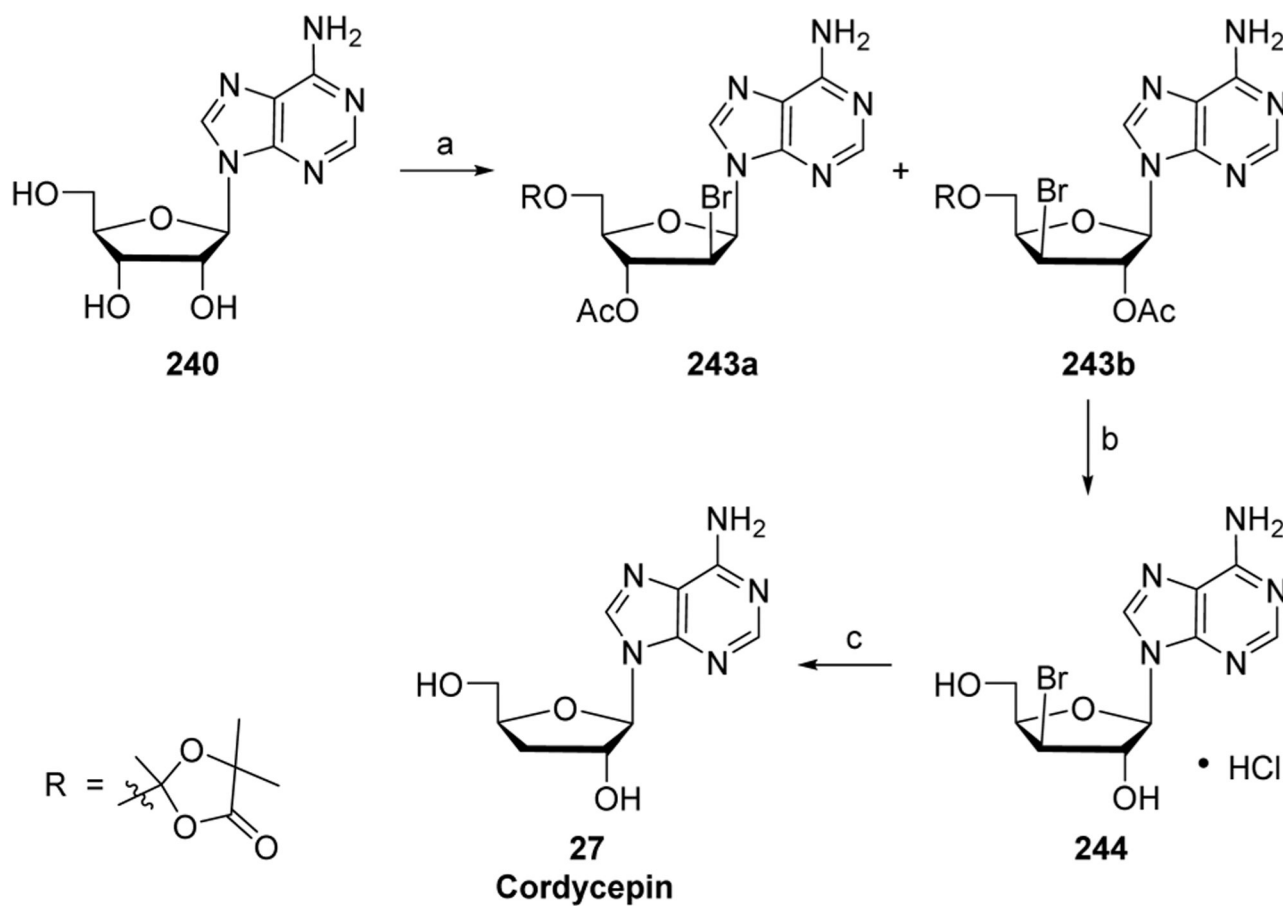


**Scheme 41. Synthesis of Elacytarabine 51 and MB 07133 34<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) oleic anhydride, DMF, *Mucor miehei* lipase (MML), 60 °C, 48 h, 88%; (b) TBSCl, imidazole, DMF, rt, overnight, 85%; (c) 50% TFA, THF, -10 °C, 18 h, 65%; (d) DMF-dimethyl acetal, pyr, rt, 16 h, 90%; (e) EDC, DMAP, L-N,N-dimethyl-phenylalanine, DCM, rt, 16 h, 83%; (f) LiAlH<sub>4</sub>, ether, -20 °C, 1 h, 66%; (g) 4-nitrophenyl phosphorodichloridate, THF, pyr, 0 °C to rt, 3.5 h, then sodium 4-nitrophenoxide, 40 °C, 4 h, 63%; (h) *t*-BuMgCl, THF, rt, 16 h, 78%; (i) 70% TFA, 60 °C, 16 h, 83%.

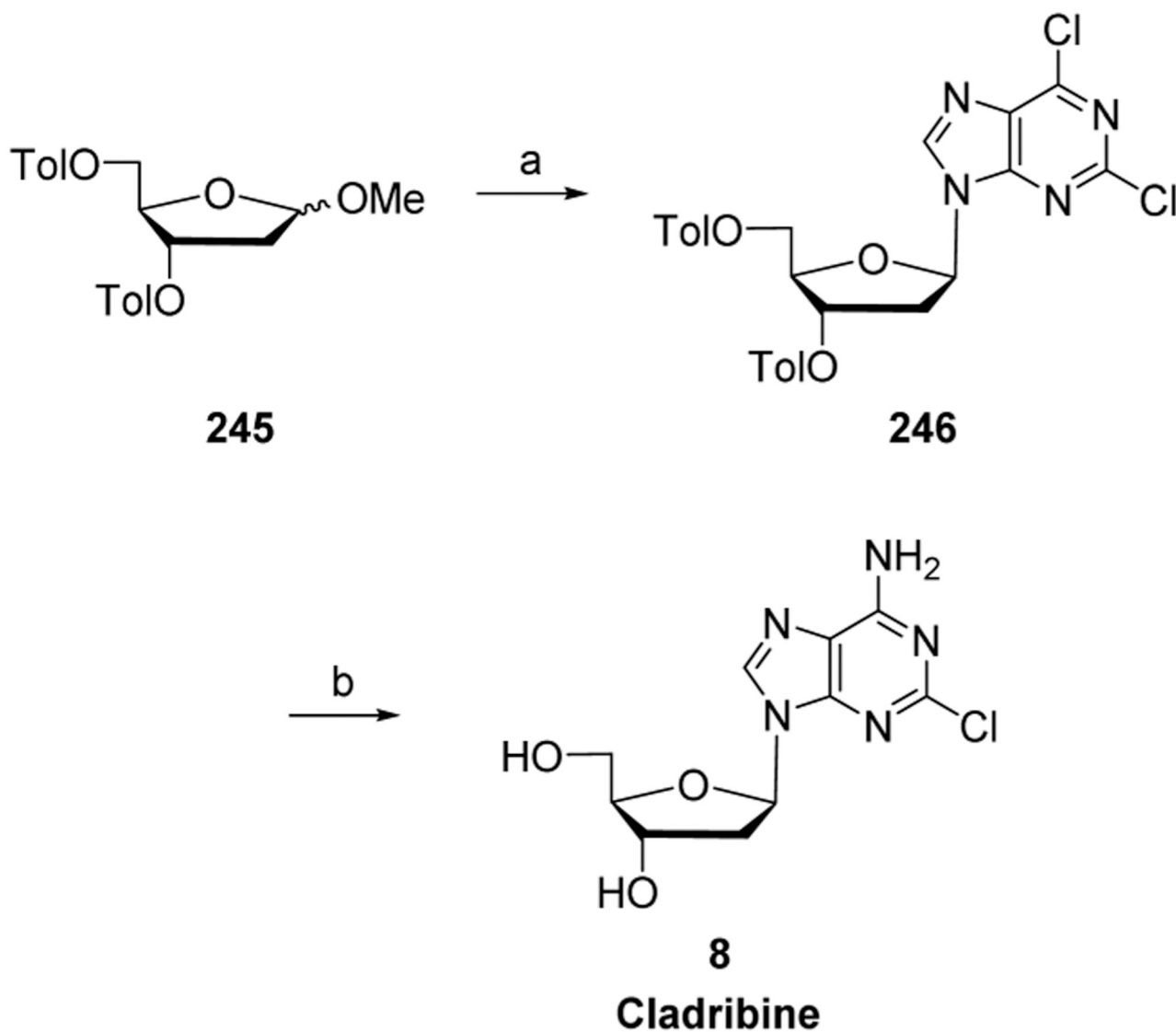
**Scheme 42. Synthesis of Cordycepin 27<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 2-acetoxyisobutryl bromide, CH<sub>3</sub>CN, H<sub>2</sub>O, rt, 1 h; (b) Amberlite IRA-400 (OH<sup>-</sup>) resin, MeOH, 92% (over two steps); (c) 1 M LiEt<sub>3</sub>BH, THF, Me<sub>2</sub>SO, 0 °C to rt, overnight, then 5% HOAc/H<sub>2</sub>O, Et<sub>3</sub>B, 98%.



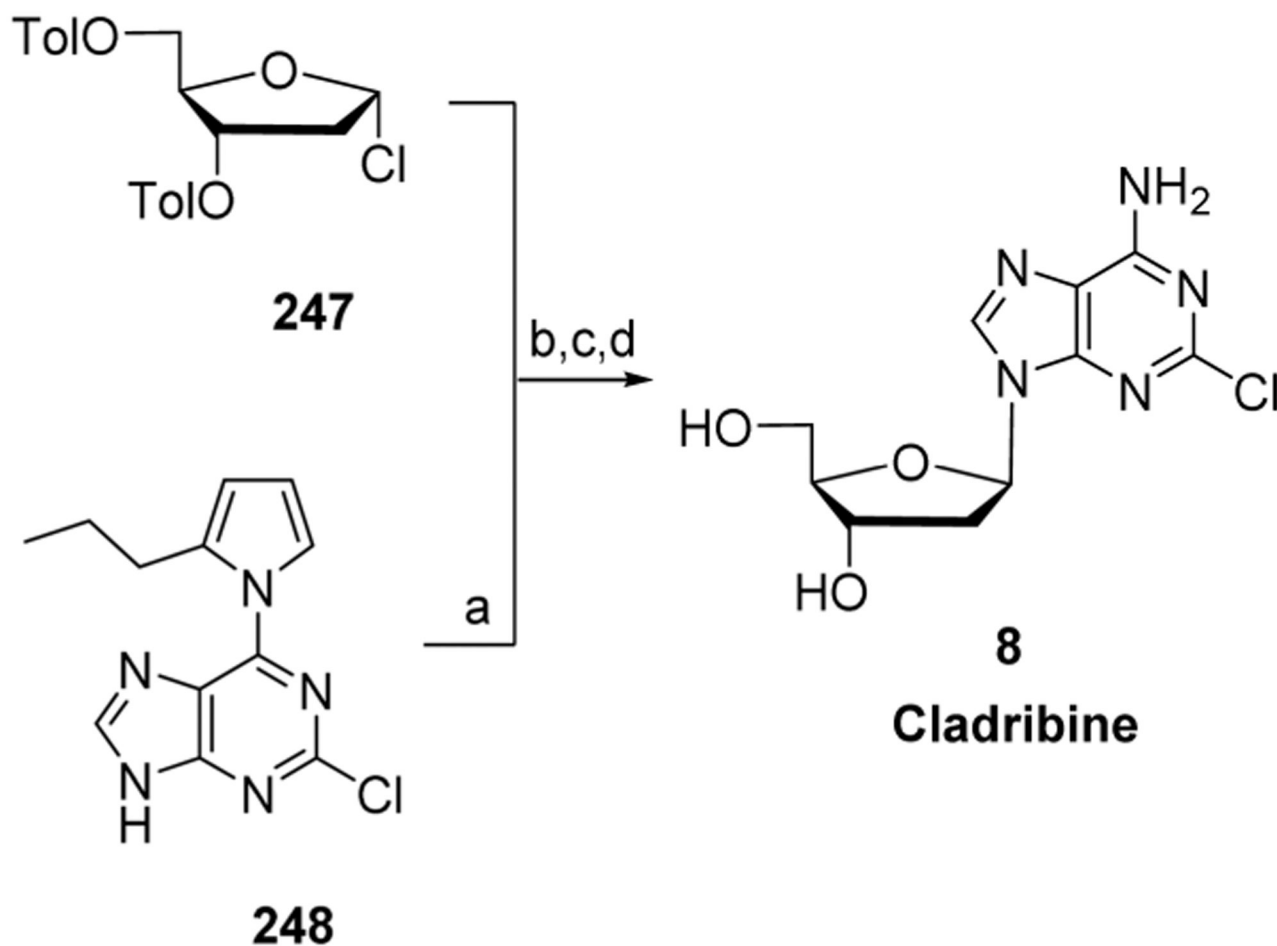
**Scheme 43. Synthesis of Cordycepin 27<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 2-acetoxyisobutyryl bromide, CH<sub>3</sub>CN, EtOAc, <35 °C, 6 h, 52% for **243b**; (b) AcBr, MeOH, concd HCl, 20 °C, rt, 2–4 days, 70%; (c) AcONa, EtOH, H<sub>2</sub>O then Pd/C, H<sub>2</sub> (50 psi), rt to 50 °C, 60%.



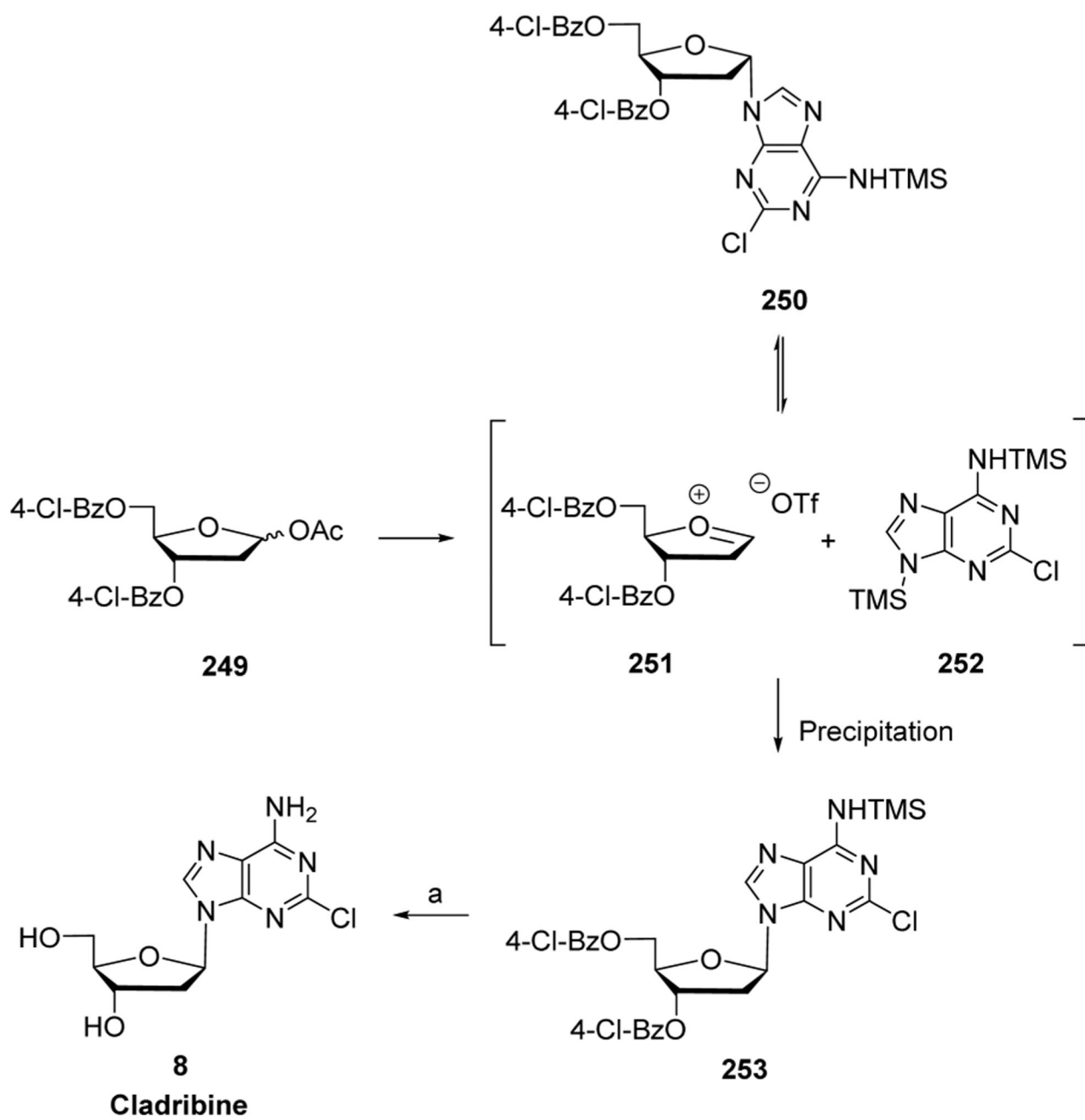
**Scheme 44. Synthesis of Cladribine 8<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 2,6-dichloropurine, melt, then dichloroacetic acid, and fusion under vacuum; (b)  $\text{NH}_3/\text{NaOMe}$ .

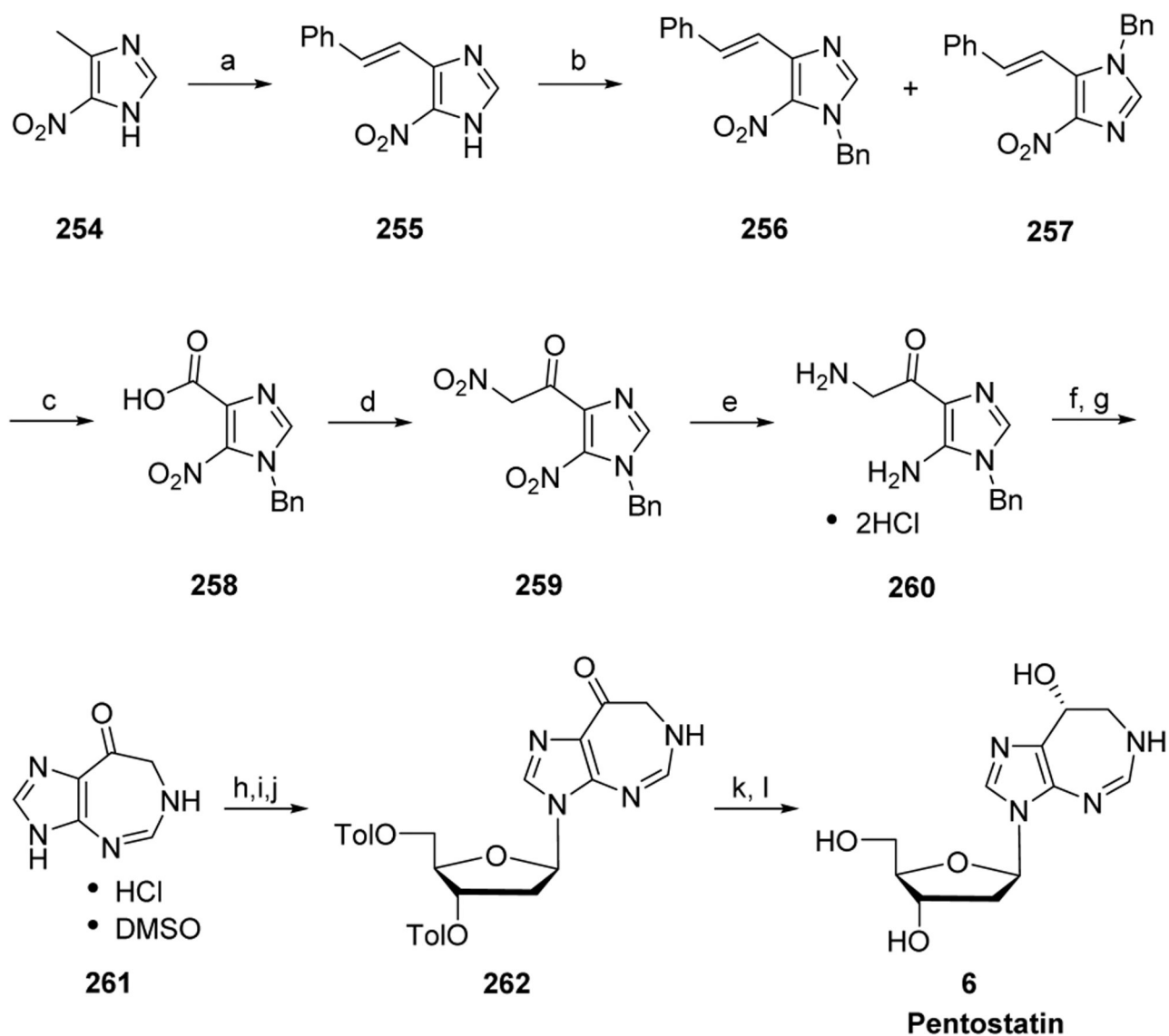


**Scheme 45. Synthesis of Cladribine 8<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) NaH, CH<sub>3</sub>CN, rt, 8 h; (b) then **247**, 0 °C to rt, 22 h, 100% (over two steps); (c) BnI, CH<sub>3</sub>CN, 60 °C, 1.5 h; (d) NH<sub>3</sub>/MeOH, 60 °C, 11 h, >90% (over two steps).

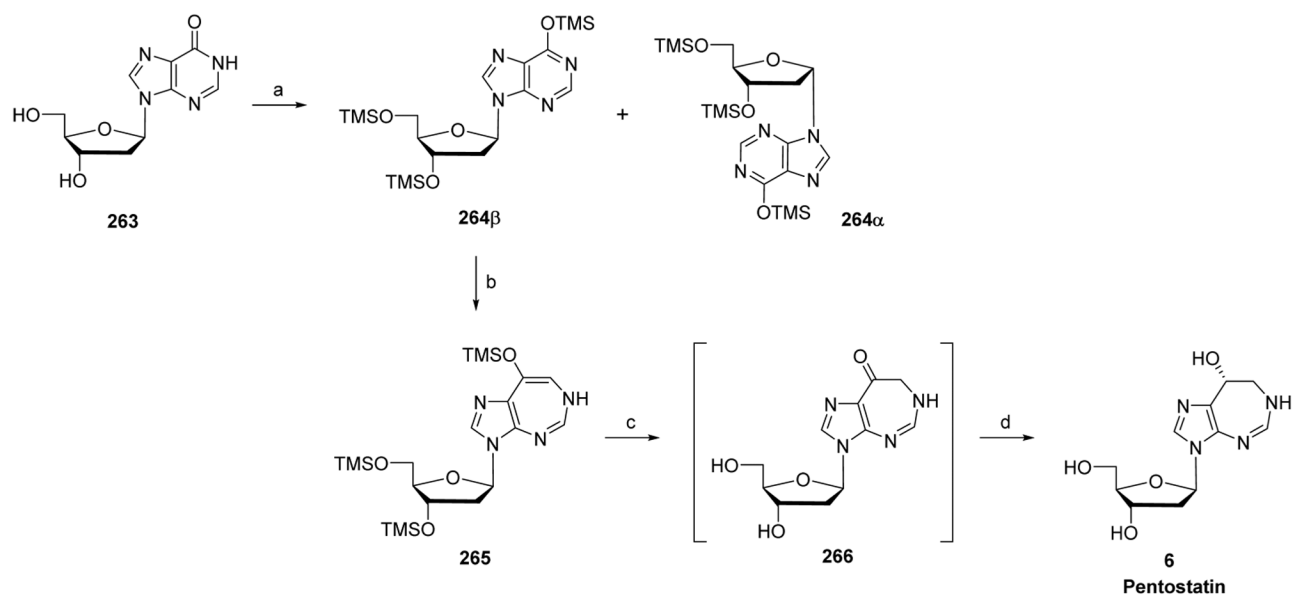
Scheme 46. Synthesis of Cladribine 8<sup>a</sup><sup>a</sup>Reagents and conditions: (a) 25% NaOMe/MeOH.





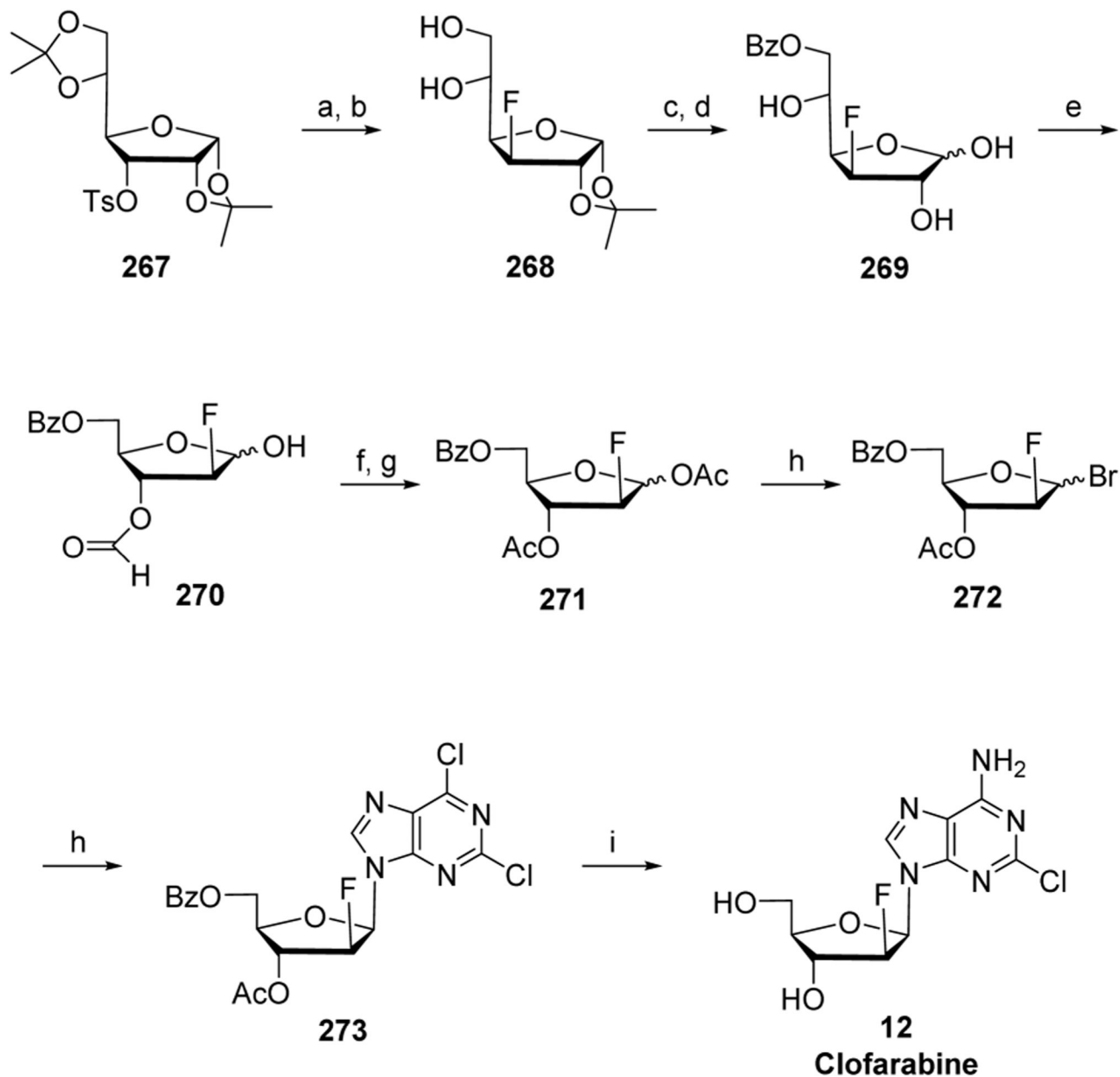
**Scheme 47. Synthesis of Pentostatin 6<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) PhCHO, piperidine, 95 °C, 21 h, 70%; (b) BnCl, K<sub>2</sub>CO<sub>3</sub>, DMF, 75 °C, 6 h, 96%; (c) O<sub>3</sub>, DCM, -78 °C, 8 h, then HCO<sub>2</sub>H, H<sub>2</sub>O<sub>2</sub>, 0 °C to rt, overnight, 67%; (d) CDI, THF, reflux, 1 h, then *t*-BuOK, CH<sub>3</sub>NO<sub>2</sub>, THF, 0 °C, 45 min, 68%; (e) SnCl<sub>2</sub>, concd HCl, 60 °C, 2.5 h, then H<sub>2</sub>S, rt, 75%; (f) H<sub>2</sub>, Pd/C, MeOH, H<sub>2</sub>O, rt, 16 h, 96%; (g) HC(OEt)<sub>3</sub>, Me<sub>2</sub>SO, 65 °C, 15 min, 81%; (h) *N,N*-bis(trimethylsilyl)trifluoroacetamide, pyr, CH<sub>3</sub>CN, rt, overnight; (i) 2-deoxy-3,4-di-*O-p*-toluoyl-D-pentofuranosyl chloride, SnCl<sub>4</sub>, DCE, CH<sub>3</sub>CN, -50 °C, 45 min; (j) NaHCO<sub>3</sub>, H<sub>2</sub>O, rt, 0.5 h, 23% (over three steps); (k) NaOMe, MeOH, rt, 1 h; (l) NaBH<sub>4</sub>, H<sub>2</sub>O, MeOH, 25 °C, 30 min, 26% (over two steps).



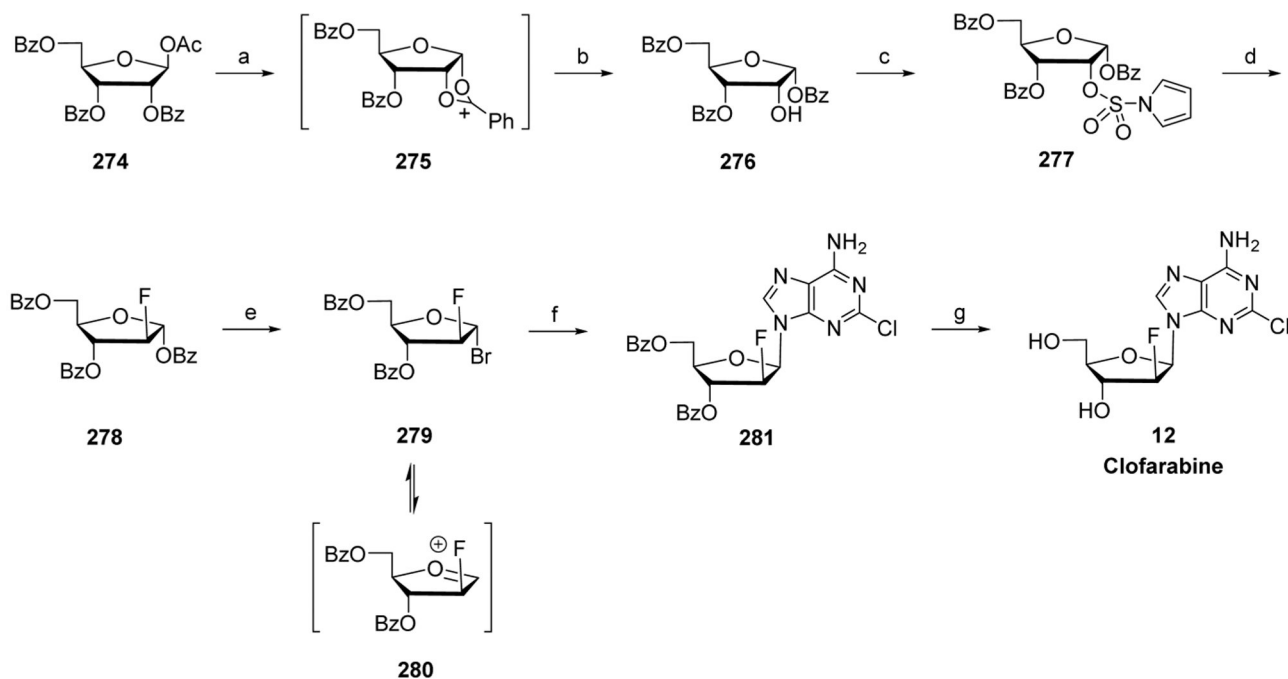
**Scheme 48. Synthesis of Pentostatin 6<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $(\text{Me}_3\text{Si})_2\text{NC}(\text{O})\text{CF}_3$ , pyr, MeCN, 12 h; (b)  $\text{BF}_3 \cdot \text{Et}_2\text{O}$ , DCM, 0 °C, 30 min; (c) TBAF, THF, 0 °C, 2 h; (d)  $\text{NaBH}_4$ , MeOH,  $\text{H}_2\text{O}$ , rt, 1 h.



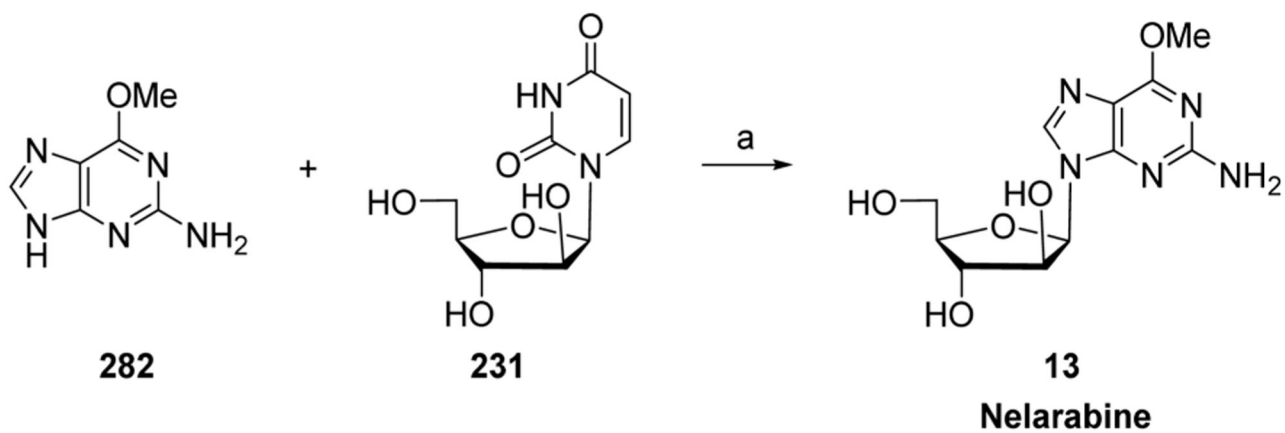
**Scheme 49. Synthesis of Clofarabine 12<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) KF, acetamide, 210 °C, 62%; (b) MeOH–0.7% H<sub>2</sub>SO<sub>4</sub> (1:1 v/v); (c) BzCl in DCM, pyr, –15 °C, 80% (over two steps); (d) Amberlite IR-120 (H<sup>+</sup>), H<sub>2</sub>O/dioxane, 80 °C, 78%; (e) KIO<sub>4</sub>, H<sub>2</sub>O; (f) NaOMe, MeOH; (g) Ac<sub>2</sub>O, pyr, 80% (over three steps); (h) HBr/HOAc, DCM; (i) 2,6-dichloropurine, DCE, 100 °C, 32% ( $\beta$ -anomer); then NH<sub>3</sub>/EtOH, steel bomb, 3 days, solvent switch to MeCN/H<sub>2</sub>O, LiOH, 42%.

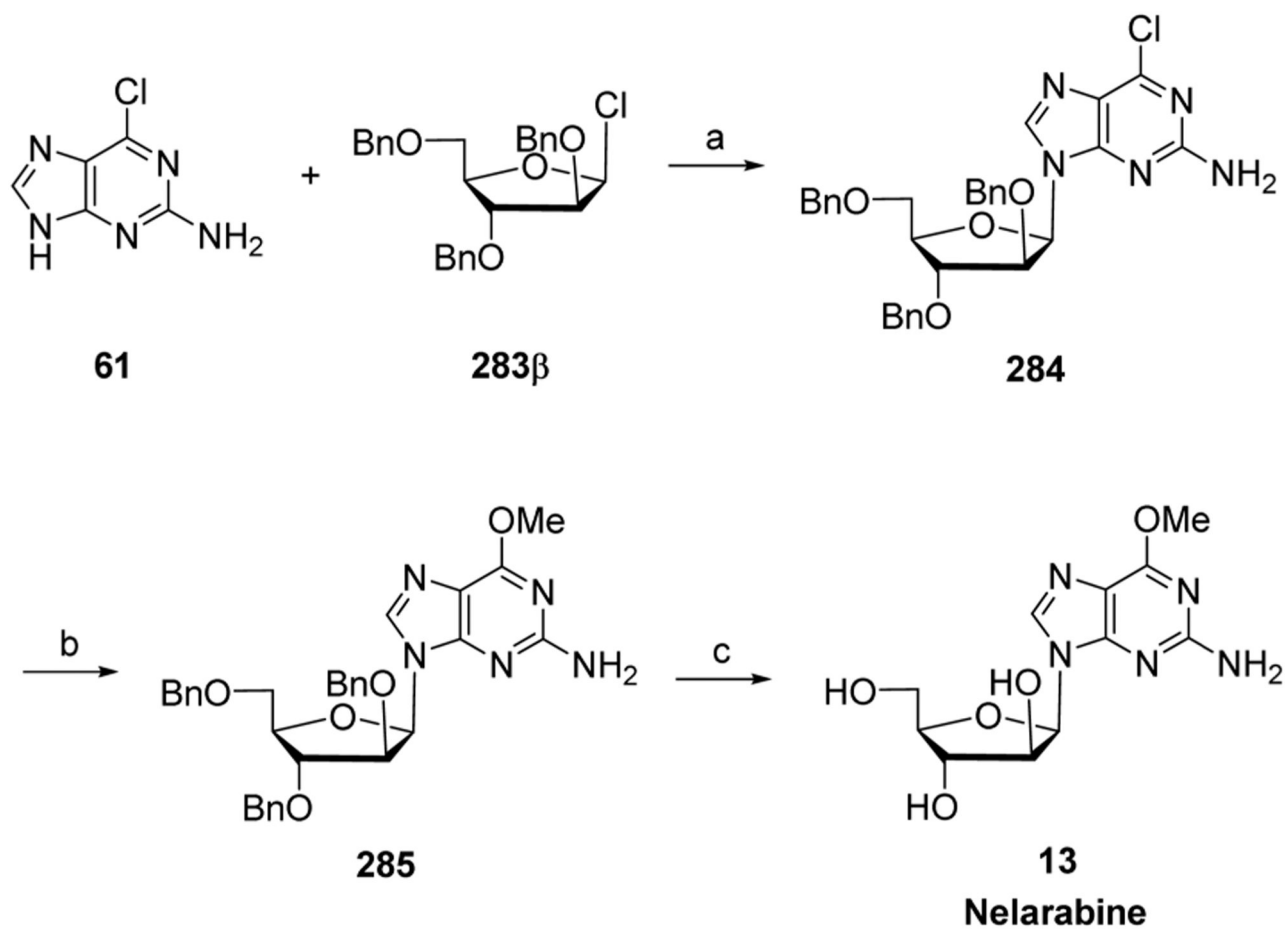


**Scheme 50. Synthesis of Clofarabine 12<sup>a</sup>**

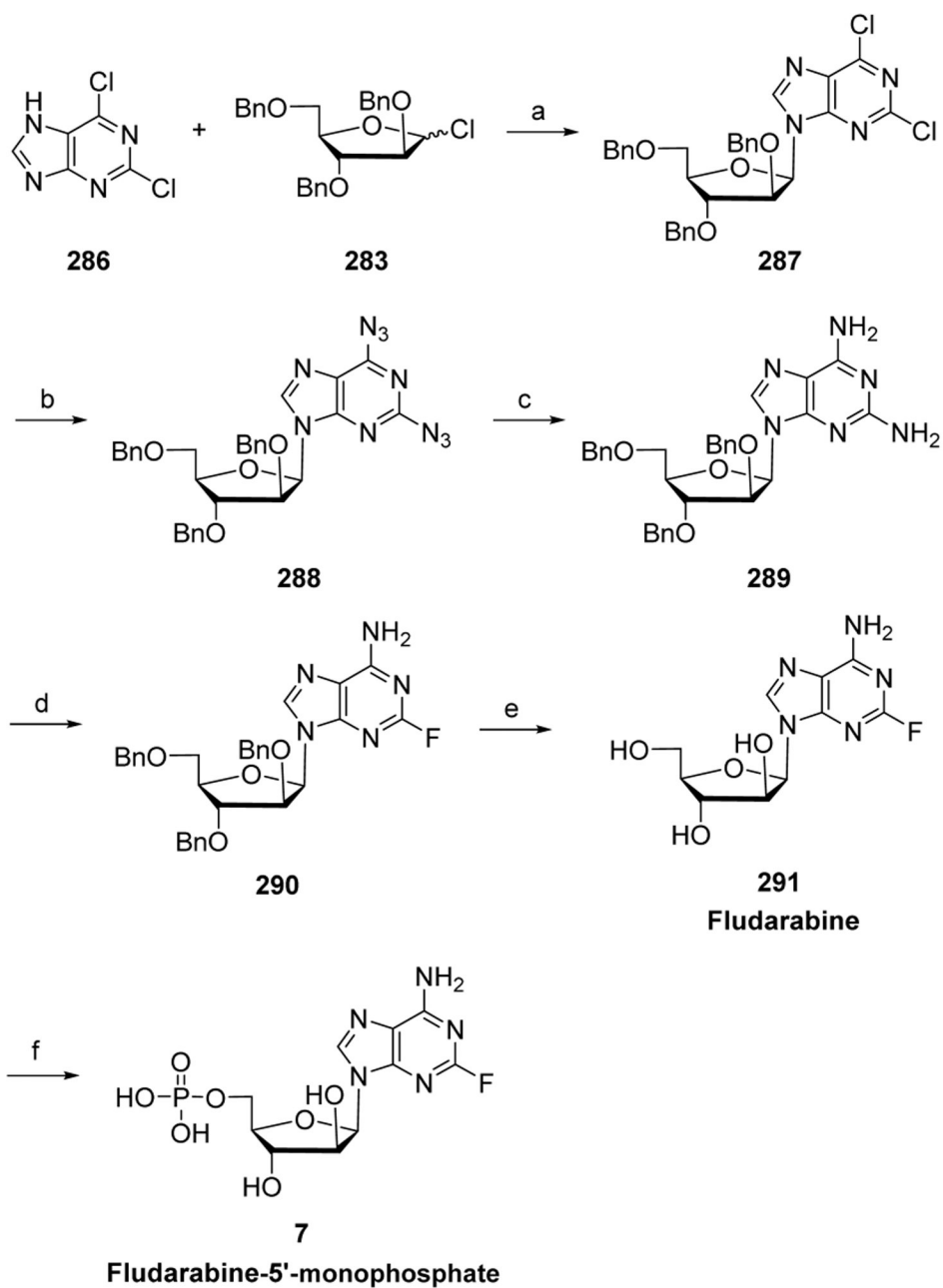
<sup>a</sup>Reagents and conditions: (a) 0.5 M HBr, DCM, 0 °C; (b) H<sub>2</sub>O, 81% (over two steps); (c) NaH, DMF, 0 °C, 0.5 h, then *N,N'*-sulfuryldiimidazole, -40 °C to rt, 2 h, 85%; (d) KHF<sub>2</sub>, 2,3-butanediol, HF (50% in H<sub>2</sub>O), 160 °C, 1 h, 63%; (e) HBr/HOAc, DCM, rt, 16 h, 98%; (f) 2-chloroadenine, CH<sub>3</sub>CN, *t*-amyl alcohol, *tert*-BuOK, CaH<sub>2</sub>, 50 °C, 40 min, then **279**, DCE, 50 °C, 19 h, 50% ( $\beta/\alpha = 80:1$ ); (g) NaOMe, MeOH, 33 °C, 7 h, 64%.

**Scheme 51. Synthesis of Nelarabine 13<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) 10 mM K<sub>3</sub>PO<sub>4</sub>, *n*-PrOH, H<sub>2</sub>O, pH 6.75, uridine phosphorylase, purine nucleoside phosphorylase, 37 °C, 26 days, 17%.

**Scheme 52. Synthesis of Nelarabine 13<sup>a</sup>**

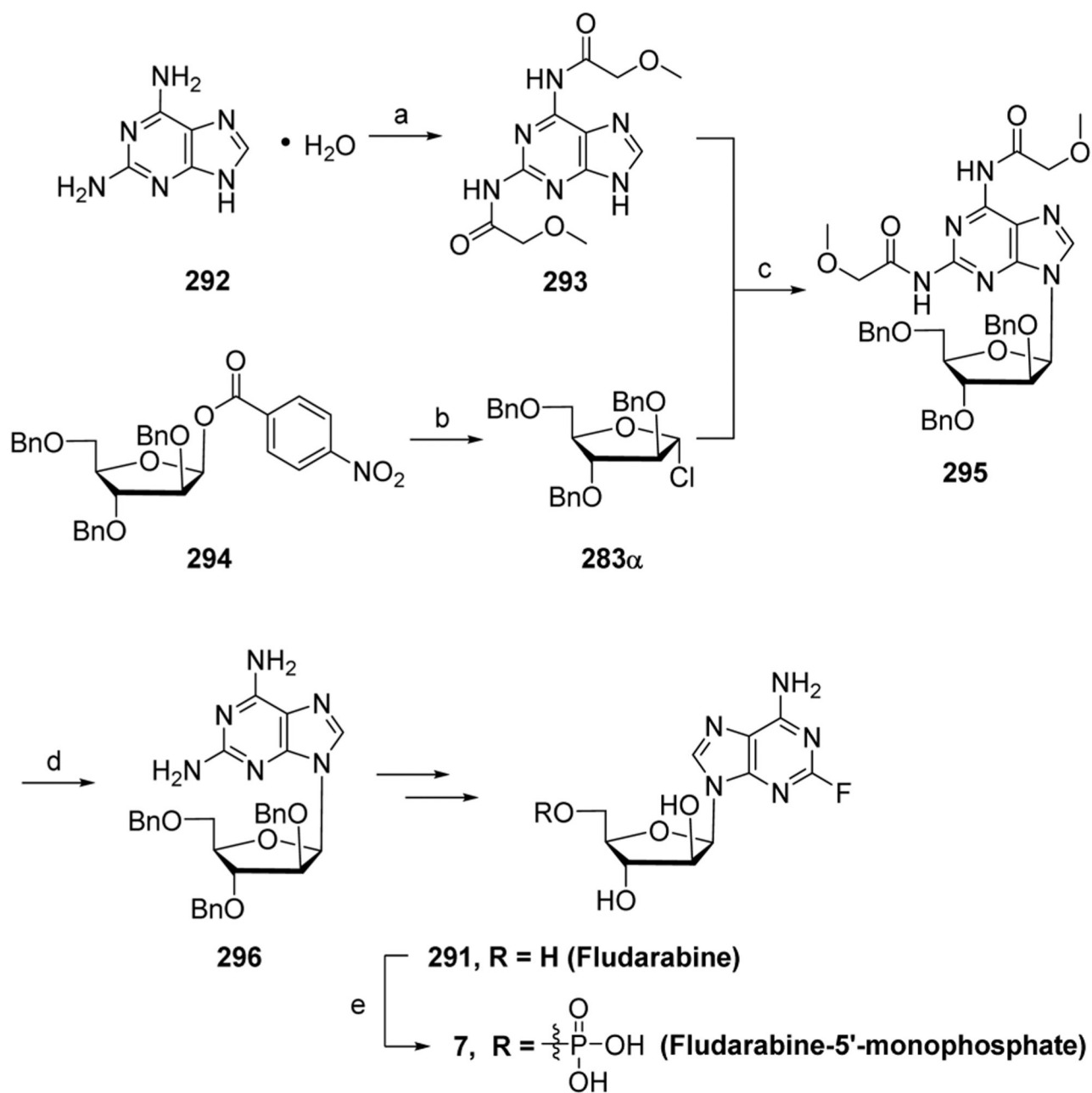
<sup>a</sup>Reagents and conditions: (a) NaH, MeCN; (b) MeONa, MeOH; (c) Pd/C, HCOONH<sub>4</sub>, MeOH, reflux, 90 min, 91%.



**Scheme 53. Synthesis of Fludarabine 291 and Fludarabine-5'-monophosphate 7<sup>a</sup>**

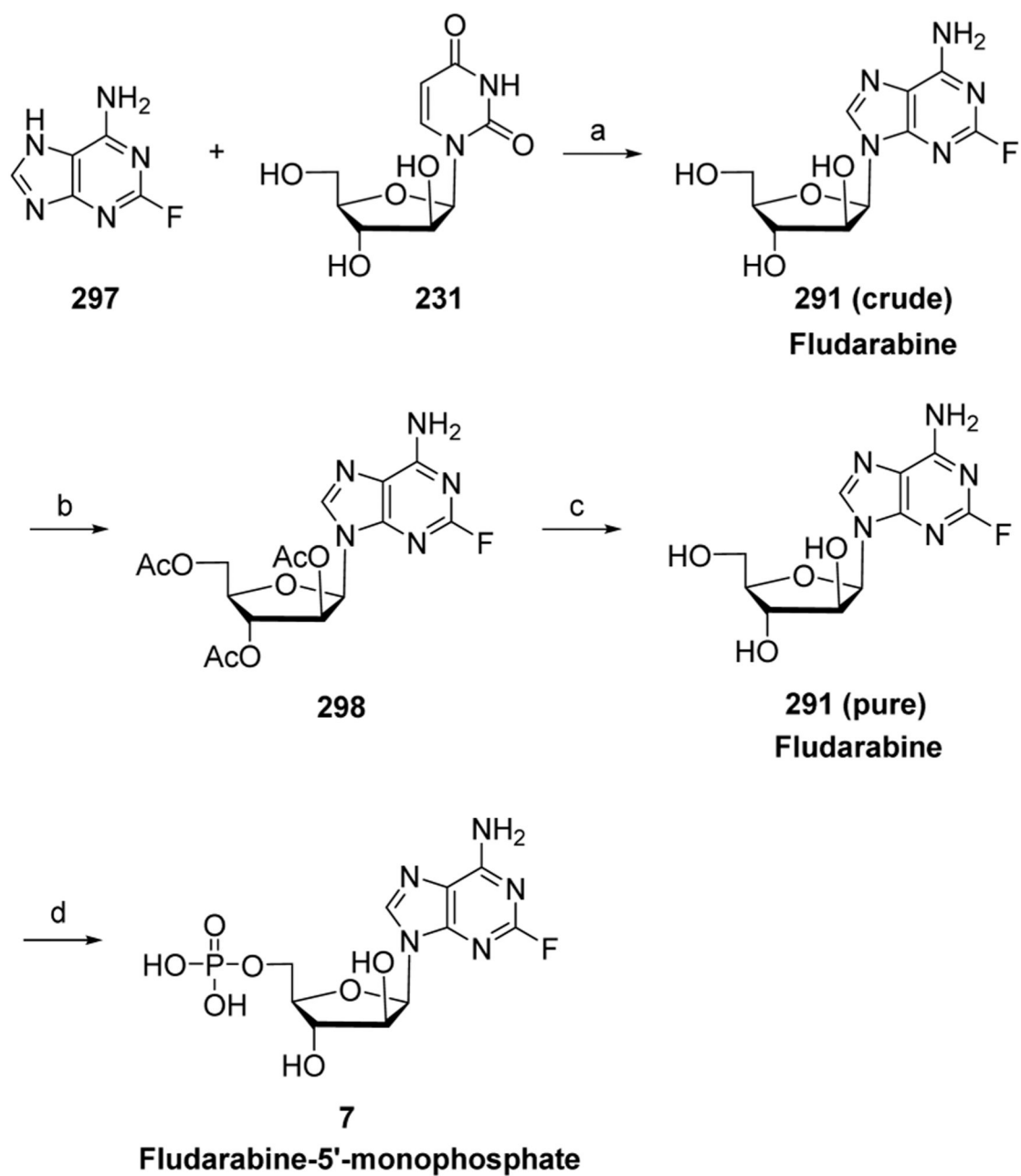
<sup>a</sup>Reagents and conditions: (a) Hg(CN)<sub>2</sub>, CaSO<sub>4</sub>, MeNO<sub>2</sub>, reflux, 3 h, 11%; (b) NaN<sub>3</sub>, EtOH, H<sub>2</sub>O, 1 h, 98%; (c) Pd/C, H<sub>2</sub>, EtOH, rt, 6 h, 75%; (d) HBF<sub>4</sub>, NaNO<sub>2</sub>, CHCl<sub>3</sub>, -10 °C, 40 min, 36%; (e) Na, NH<sub>3</sub>, 34%; (f) POCl<sub>3</sub>, PO(OEt)<sub>3</sub>, 0 °C, 3.5 h, 96%.





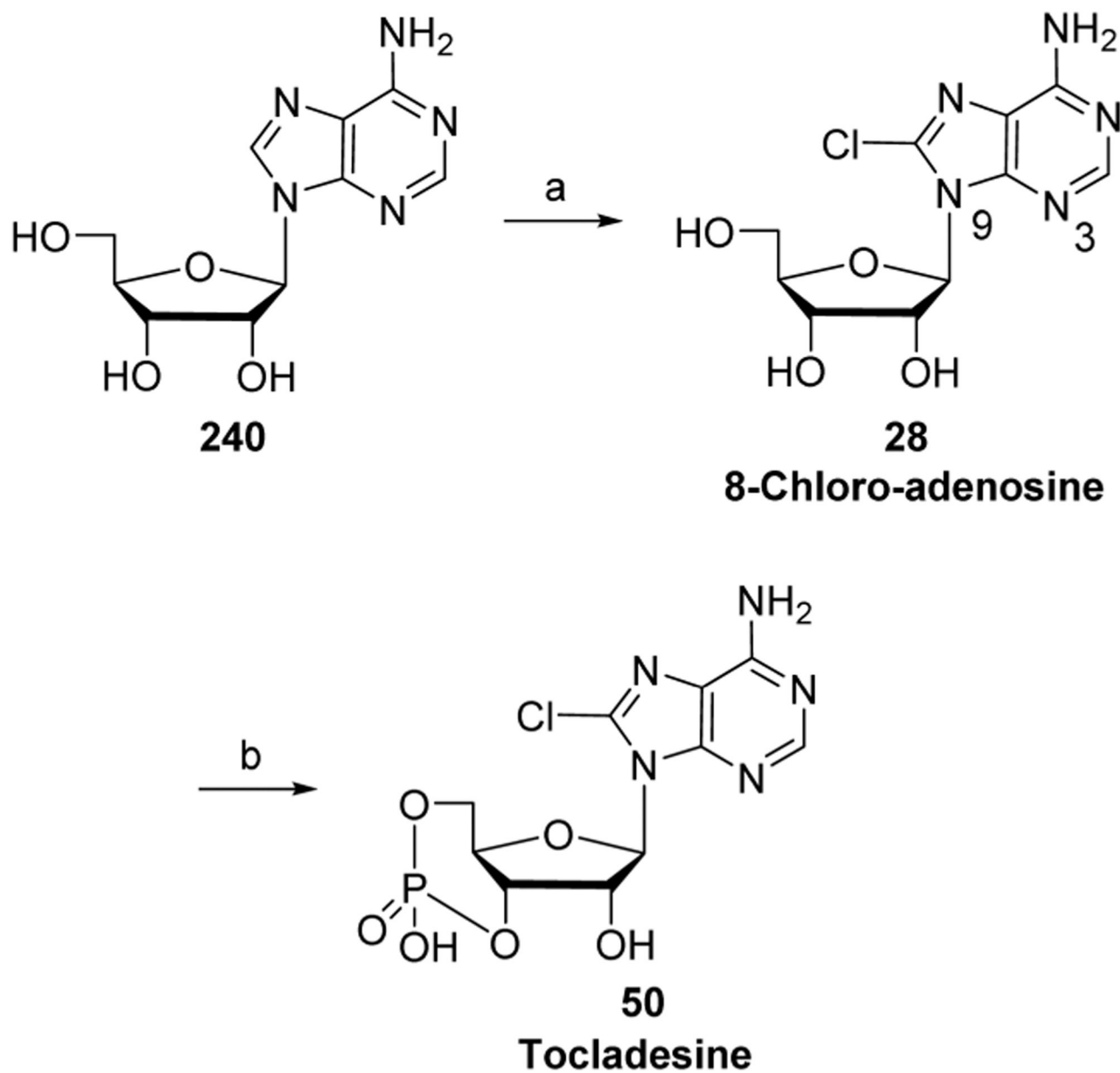
**Scheme 54. Synthesis of Fludarabine 291 and Fludarabine-5'-monophosphate 7<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) methoxyacetic anhydride, pyr, 88 °C, 1 h, then methyl ethyl ketone, overnight, 90%; (b) HCl (g), DCM, 4–6 °C, 2 h, quantitative; (c) DIPEA, DCE, reflux, overnight, quantitative; (d) NaOMe, MeOH, 68%; (e) POCl<sub>3</sub>, TMP, 0 °C, 18 h, 56%.



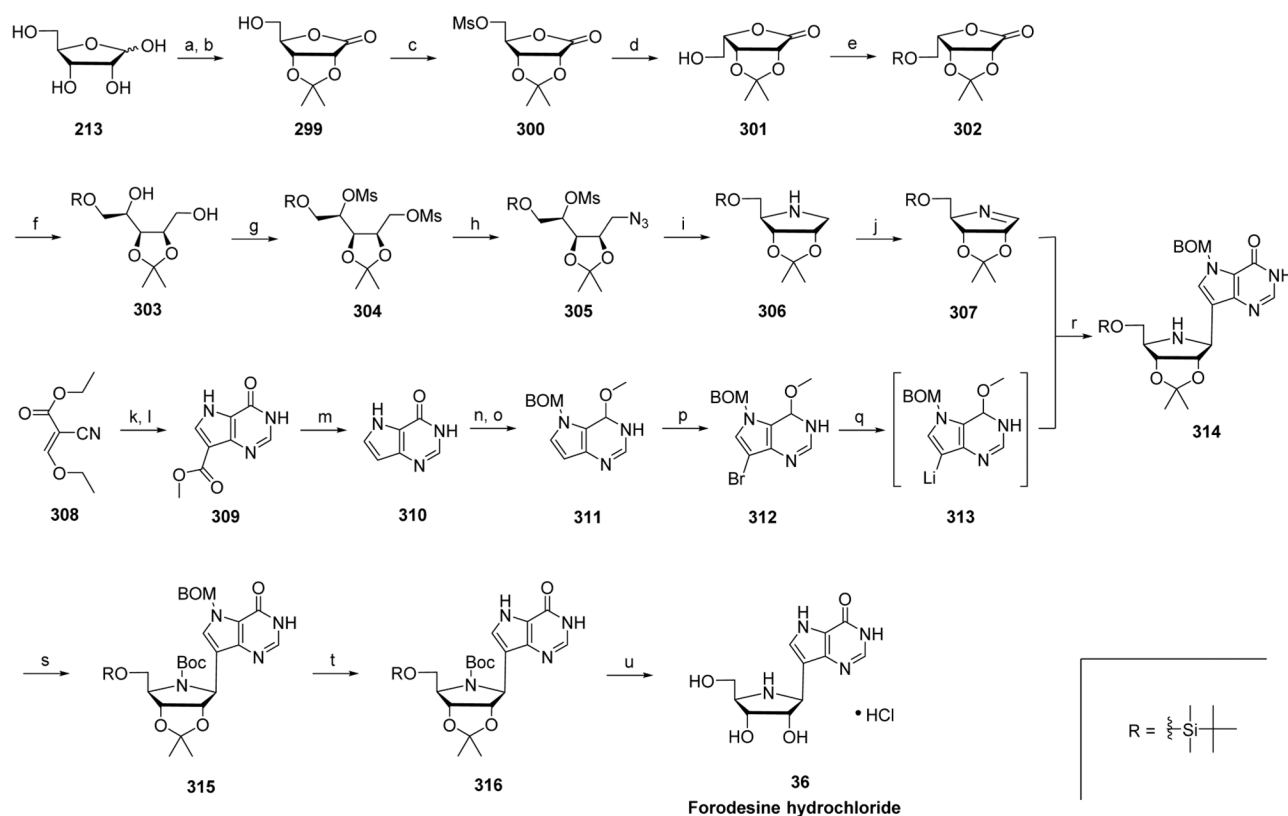
**Scheme 55. Synthesis of Fludarabine 291 and Fludarabine-5'-monophosphate 7<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) *Enterobacter aerogenes*, K<sub>2</sub>HPO<sub>4</sub>, H<sub>2</sub>O, 60 °C, 24 h, 35%; (b) Ac<sub>2</sub>O, 95 °C, 9 h, 87%; (c) NH<sub>4</sub>OH, H<sub>2</sub>O, MeOH, rt, 19 h, 74%.



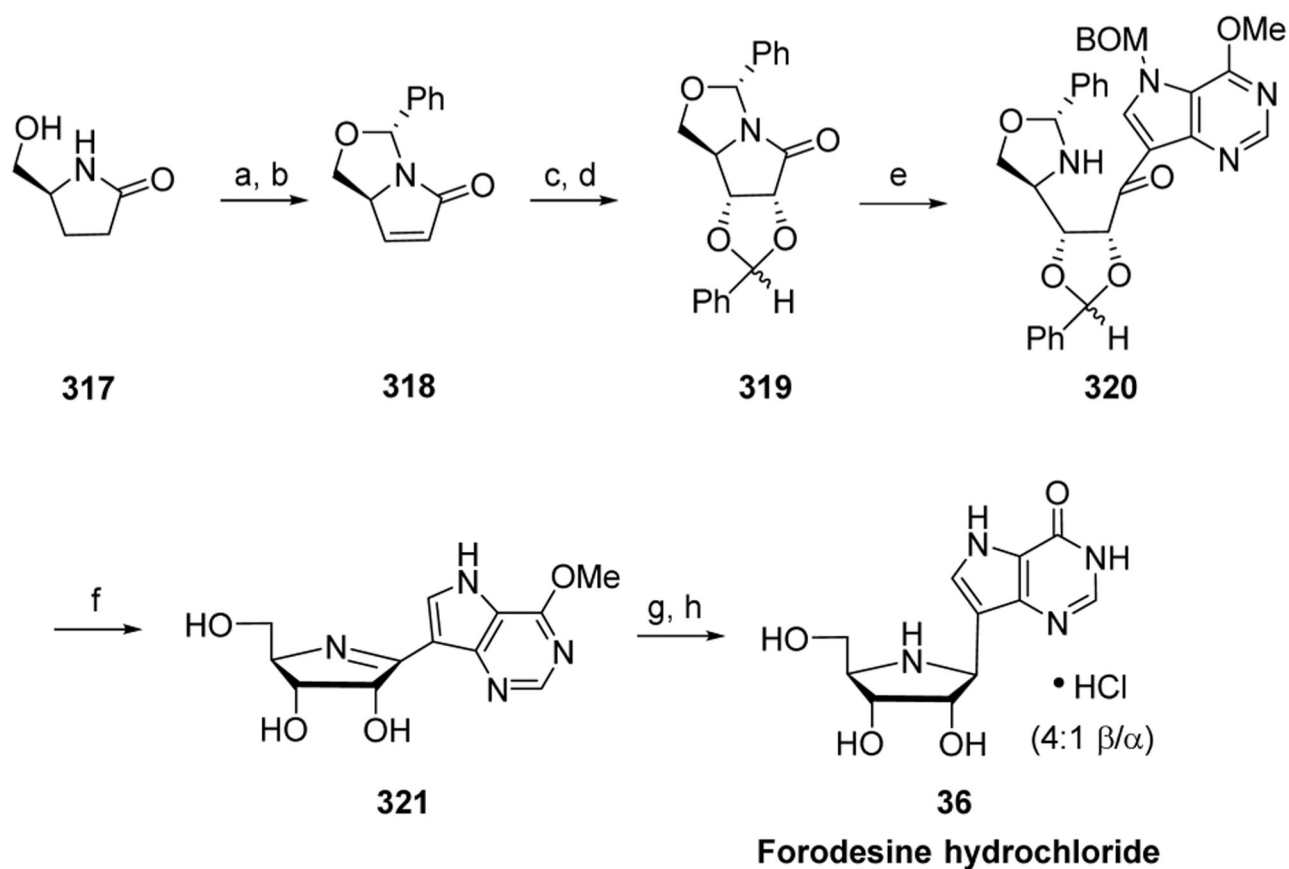
**Scheme 56. Synthesis of 8-Chloroadenosine 28 and Tocladesine 50<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) AcCl, *m*-CPBA, DMF, rt, 20 min, 40%; (b) POCl<sub>3</sub>, PO(OEt)<sub>3</sub>, 0 °C, 6 h, then aq NaOH, 2 h, 23%.



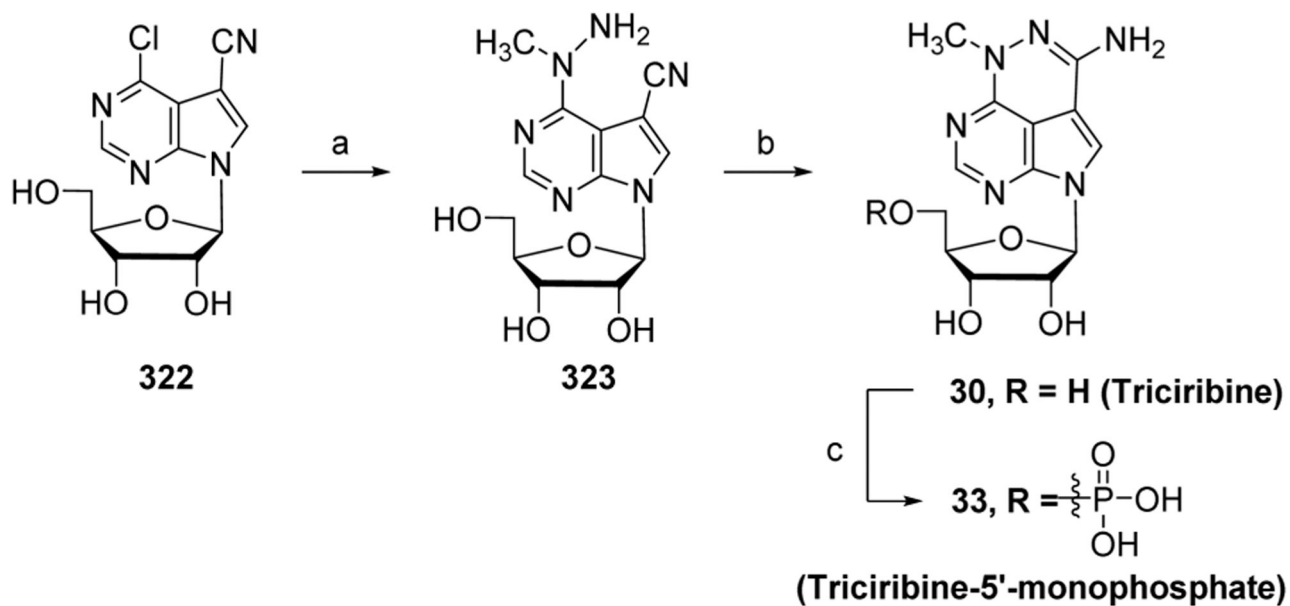
### Scheme 57. Synthesis of Forodesine Hydrochloride 36<sup>a</sup>

<sup>a</sup>Reagents and conditions: (a) Br<sub>2</sub>, K<sub>2</sub>CO<sub>3</sub>, H<sub>2</sub>O, 5 °C to rt, overnight; (b) concd H<sub>2</sub>SO<sub>4</sub>, Me<sub>2</sub>CO, reflux, 4 h, then 30 °C, 8 h, 55% (over two steps); (c) MsCl, Et<sub>3</sub>N, DCM, -20 °C, 1 h, then rt, 8 h; (d) aq KOH, <30 °C, 4 h, 59% (over two steps); (e) TBDMSCl, imidazole, DMF, 20 °C, 4 h, 91%; (f) LiBH<sub>4</sub>, THF, -30 °C to rt, 24 h, 75%; (g) (MeSO<sub>2</sub>)<sub>2</sub>O, pyr, 0 °C to rt, 18 h, 88%; (h) NaN<sub>3</sub>, DMF, 100 °C, 2 h, 42%; (i) H<sub>2</sub>, Pd, NaOAc, dioxane, 20 °C, 72 h, 94%; (j) NCS, pentane, rt, 1 h, then lithium tetramethylpiperidide (LiTMP), THF, -78 °C, 85%; (k) H<sub>2</sub>NCH(CO<sub>2</sub>C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>, NaOMe, MeOH, reflux, 4 h, 68%; (l) formamidine acetate, C<sub>2</sub>H<sub>5</sub>OH, reflux, 27 h, 61%; (m) aq KOH, reflux, 40 h, 74%; (n) POCl<sub>3</sub>, reflux, 2 h; (o) BOMCl, NaH, THF, 0 °C to rt, 1 h, then NaH, MeOH, 1 h; (p) NBS, DCM, 52% (over three steps); (q) *n*-BuLi, -78 °C; (r) 307, -78 °C to 0 °C; (s) (Boc)<sub>2</sub>O, DCM, rt, 1 h, 76% (over three steps); (t) H<sub>2</sub>, Pd/C, 18 h, then NH<sub>4</sub>OH, H<sub>2</sub>O, 1 h; (u) concd HCl, MeOH, reflux, 1.5 h, 85% (over two steps).



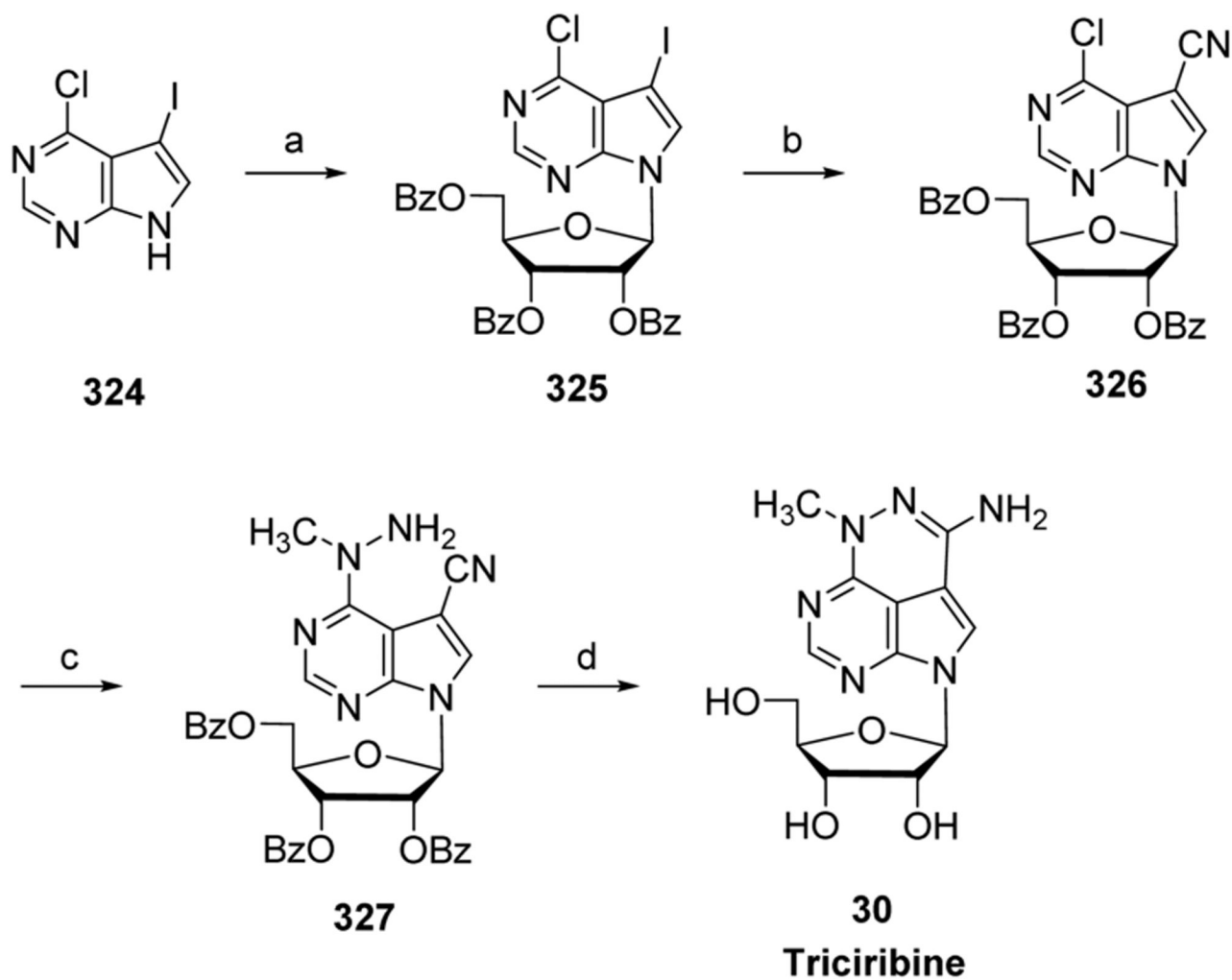
**Scheme 58. Synthesis of Forodesine Hydrochloride 36<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) PhCHO, TsOH; (b) PhSO<sub>2</sub>Me, 59%; (c) OsO<sub>4</sub>, Me CO; (d) DMP/H<sup>+</sup> 2, 63%; (e) lithiated 9-deazahypoxanthine, anisole, Et<sub>2</sub>O, -20 °C, 55%; (f) BBr<sub>3</sub>, DCM; (g) BH<sub>3</sub>·Me<sub>2</sub>S, DMF; (h) MeOH/H<sup>+</sup>, 90%.



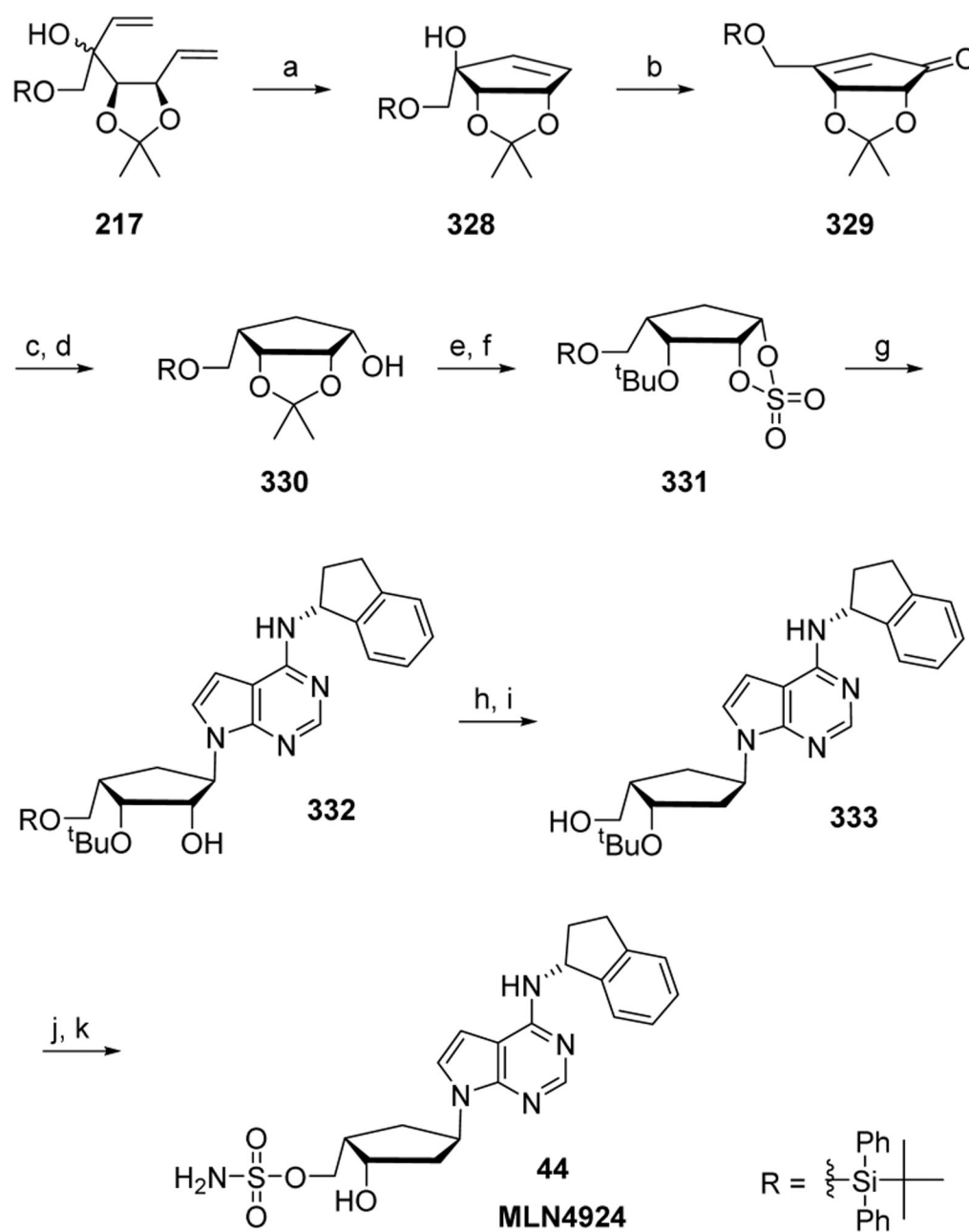
**Scheme 59. Early Synthesis of Triciribine 30 and Triciribine-5'-monophosphate 33<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a)  $\text{CH}_3\text{NHNH}_2$ , EtOH, rt, 3 min; (b)  $\text{H}_2\text{O}$ , reflux, 16 h; (c)  $\text{POCl}_3$ ,  $\text{PO}(\text{OMe})_3$ , 0 °C, 16 h, 21%;

**Scheme 60. Synthesis of Triciribine 30<sup>a</sup>**

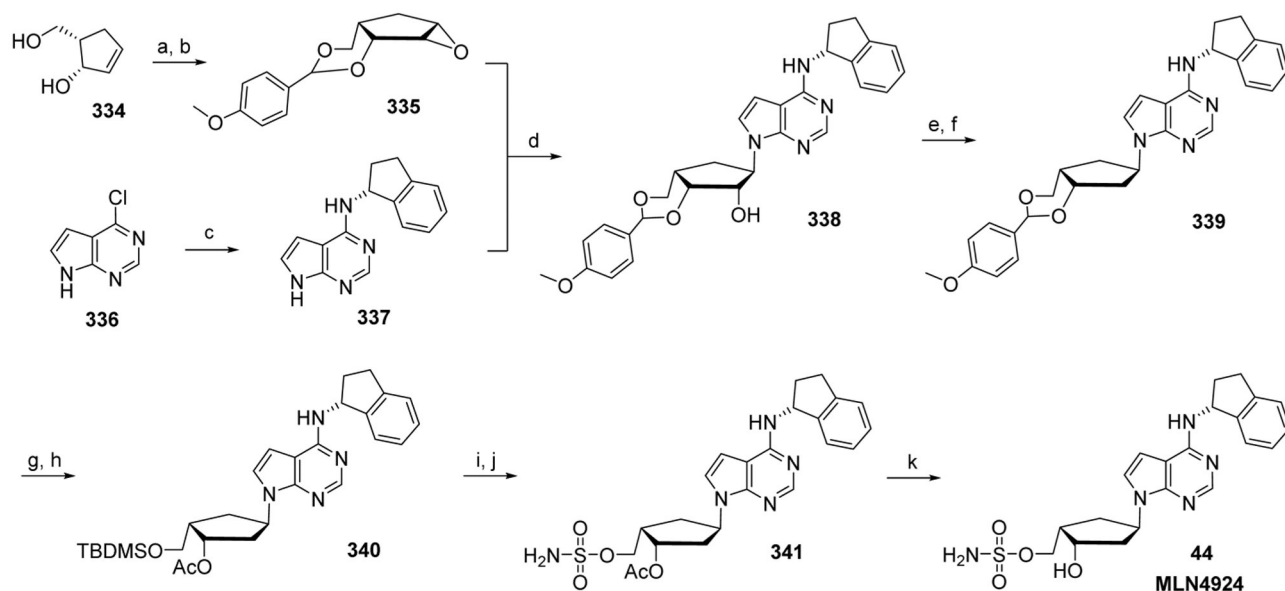
<sup>a</sup>Reagents and conditions: (a) BSA, 1-*O*-acetyl-2,3,5-tri-*O*-benzoyl- $\beta$ -D-ribofuranose, TMSOTf, CH<sub>3</sub>CN, 80 °C, 80%; (b) Bu<sub>3</sub>SnCN, Pd(PPh<sub>3</sub>)<sub>4</sub>, DMF, 95 °C, 6 h, 72%; (c) H<sub>2</sub>NNHCH<sub>3</sub>, C<sub>2</sub>H<sub>5</sub>OH, CHCl<sub>3</sub>, rt, 3.5 h, 92%; (d) NaOMe, MeOH, rt, 1 h, then reflux, 18 h, 74%.





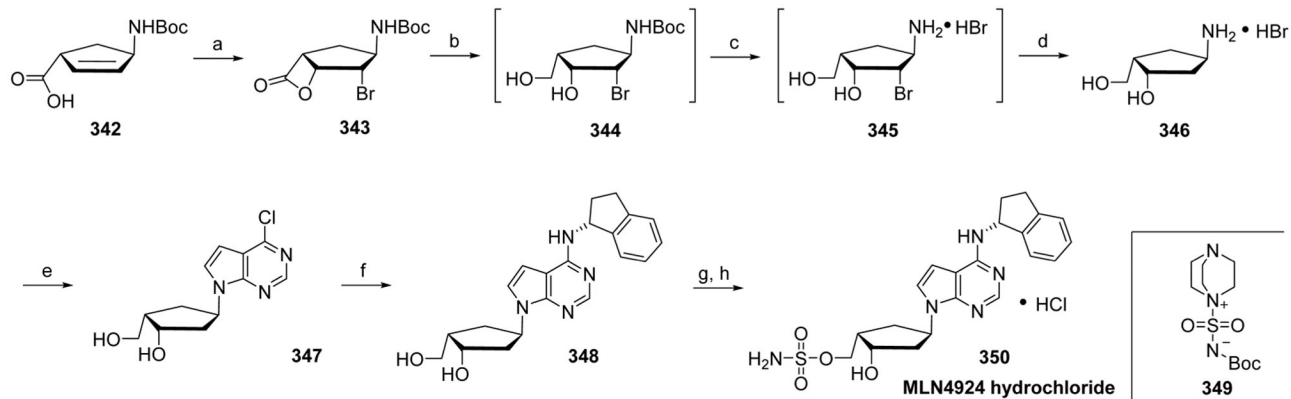
**Scheme 61. Synthesis of MLN4924 44<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Grubbs' catalyst (2<sup>nd</sup> generation), DCM, rt, 2 days, 95%; (b) PDC, DMF, rt, 2 days, 84%; (c) H<sub>2</sub>, Pd/C, MeOH, rt, overnight, 100%; (d) NaBH<sub>4</sub>, CeCl<sub>3</sub>·H<sub>2</sub>O, MeOH, 0 °C to rt, 0.5 h, 98%; (e) Me<sub>3</sub>Al, DCM, 0 °C to rt, 2 days, 62%; (f) SOCl<sub>2</sub>, Et<sub>3</sub>N, 0 °C, 10 min, then RuCl<sub>3</sub>·3H<sub>2</sub>O, NaIO<sub>4</sub>, CCl<sub>4</sub>, CH<sub>3</sub>CN, H<sub>2</sub>O, 87%; (g) *N*<sup>6</sup>-indanyl-7-deazaadenine, NaH, 18-crown-6, THF, 80 °C, overnight, then concd HCl, 80 °C, 2 h, 65%; (h) PhOC(S)Cl, DMAP, DCM, rt, overnight, then Bu<sub>3</sub>SnH, AIBN, PhMe, 110 °C, 1 h, 82%; (i) HF-pyridine, THF, pyr, rt, 1 h, 99%; (j) NH<sub>2</sub>SO<sub>2</sub>Cl, Et<sub>3</sub>N, CH<sub>3</sub>CN, 0 °C to rt, 1 h, 92%; (k) TFA, rt, 2 h, 90%.



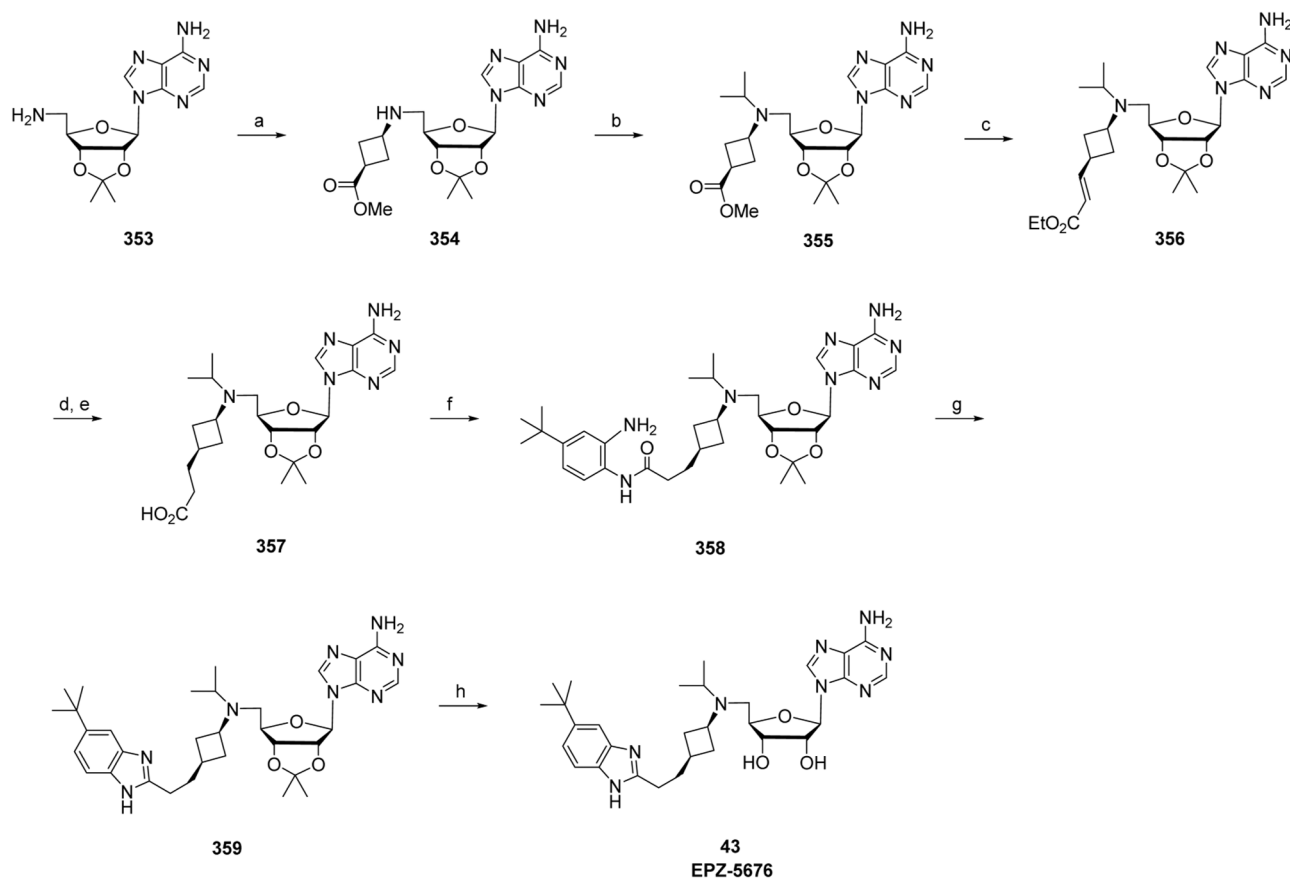
**Scheme 62. Synthesis of MLN4924 44<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) *m*-CPBA, DCM, 4 h, 76%; (b) 1-(dimethoxymethyl)-4-methoxybenzene, PPTS, rt, overnight, 78%; (c) (*S*)-(+)-1-aminoindan, DIPEA, 1-butanol, reflux, 60 h, 80%; (d) NaH, DMF, 110 °C, 2 h, 69%; (e) DMAP, PhOC(S)Cl, DCM, rt, 1 h, 99%; (f) AIBN, Bu<sub>3</sub>SnH, PhMe, reflux, 30 min, 79%; (g) AcOH, THF, H<sub>2</sub>O, rt, 60 h, 98%; (h) 1*H*-imidazole, DMAP, TBDMSCl, DMF, rt, 2 h, then DMAP, Ac<sub>2</sub>O, pyr, rt, overnight, 86%; (i) HF-pyridine, THF, pyr, rt, 1 h, 80%; (j) chlorosulfonamide, Et<sub>3</sub>N, MeCN, rt, 1 h, 94%; (k) NH<sub>3</sub>/MeOH, rt, 5 days, 90%.



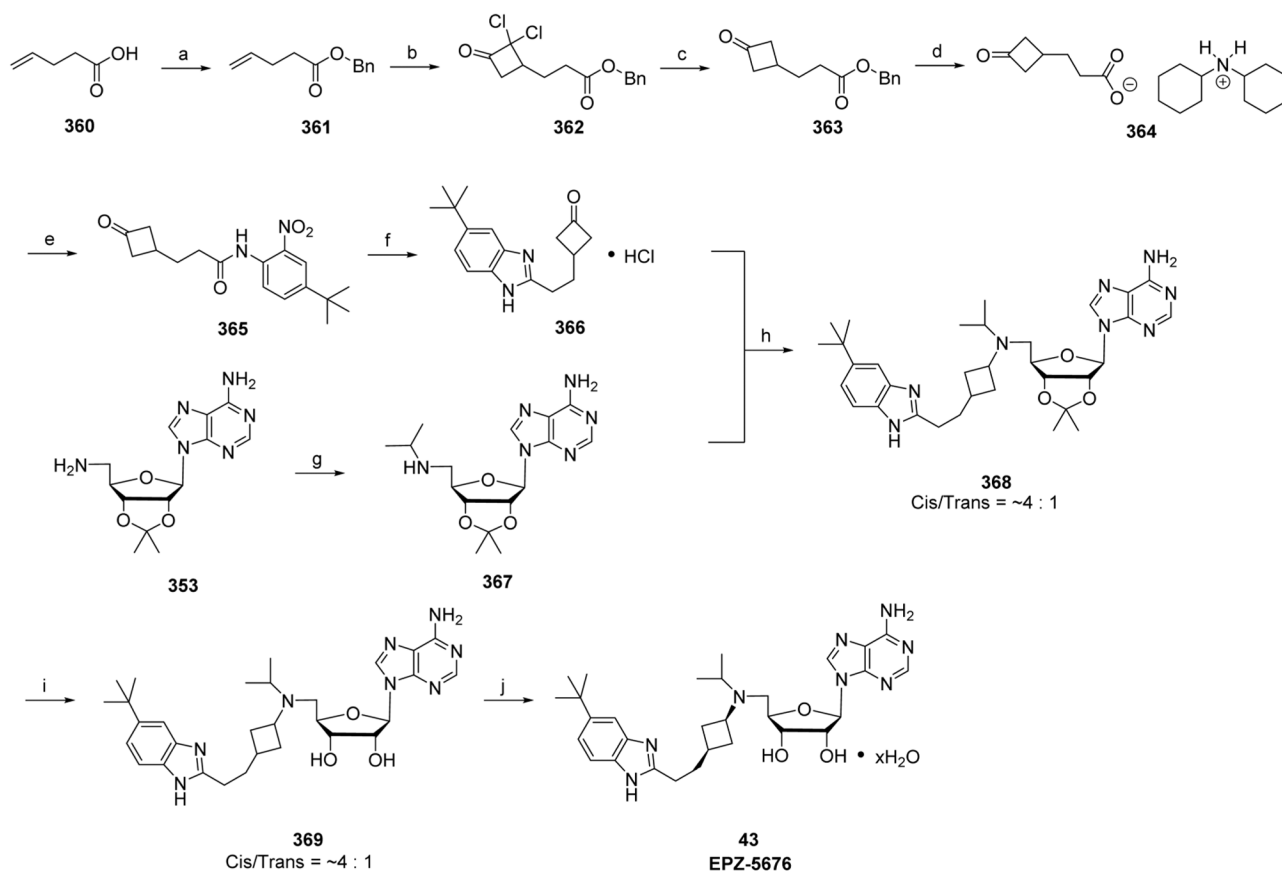
### Scheme 63. Synthesis of MLN4924 Hydrochloride 350<sup>d</sup>

<sup>d</sup>Reagents and conditions: (a) Br<sub>2</sub>, py, DME, H<sub>2</sub>O, 0 °C, 68%; (b) LiBH<sub>4</sub>, THF, H<sub>2</sub>O, -5 °C; (c) 8.84 M HBr, *i*-PrOH, 55 °C; (d) H<sub>2</sub>, Pd/C, DIPEA, MeOH, 25 °C, 80% (over three steps); (e) 2-(4,6-dichloropyrimidin-5-yl)acetaldehyde, Et<sub>3</sub>N, *i*-PrOH, 75 °C, 78%; (f) (*S*)-(+)-1-aminoindane hydrochloride, DIPEA, 2-butanol, 130 °C, 81%; (g) 349, MeCN, 53 °C, 5 h, then HCl, 59%; (h) 1.25 M HCl, EtOH, crystallization, 93%.



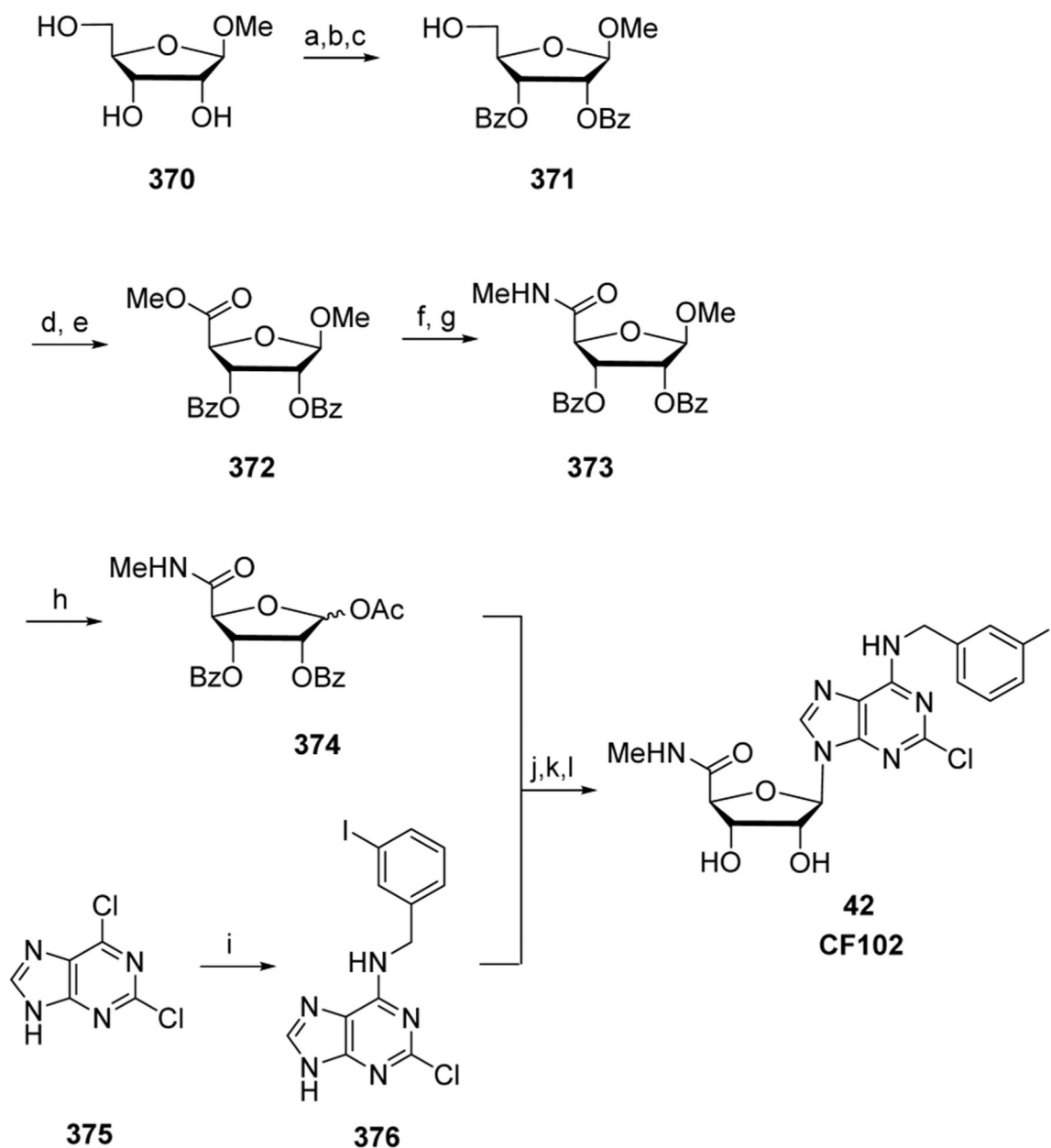
**Scheme 64. Synthesis of EPZ-5676 43<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) methyl 3-oxocyclobutanecarboxylate,  $\text{Ti}(\text{O}-i\text{Pr})_4$ , MeOH, 45 °C, 2 h, then  $\text{NaCNBH}_4$ , rt, overnight, 41%; (b) 2-iodopropane,  $\text{K}_2\text{CO}_3$ ,  $\text{CH}_3\text{CN}$ , 95 °C, overnight, 86%; (c) DIBAL-H, DCM, -78 °C to rt, then ethyl 2-(diethoxyphosphoryl)acetate, DBU, LiCl, rt, 1 h, 83%; (d) Pd/C, MeOH, rt, overnight, 78%; (e) LiOH·H<sub>2</sub>O, THF, MeOH, rt, overnight, 28%; (f) 4-(*tert*-butyl)benzene-1,2-diamine, EDC, HOBT, Et<sub>3</sub>N, rt, overnight, 50%; (g) AcOH, 65 °C, overnight, 98%; (h) HCl/MeOH, rt, 2 h, then HPLC, 51%.



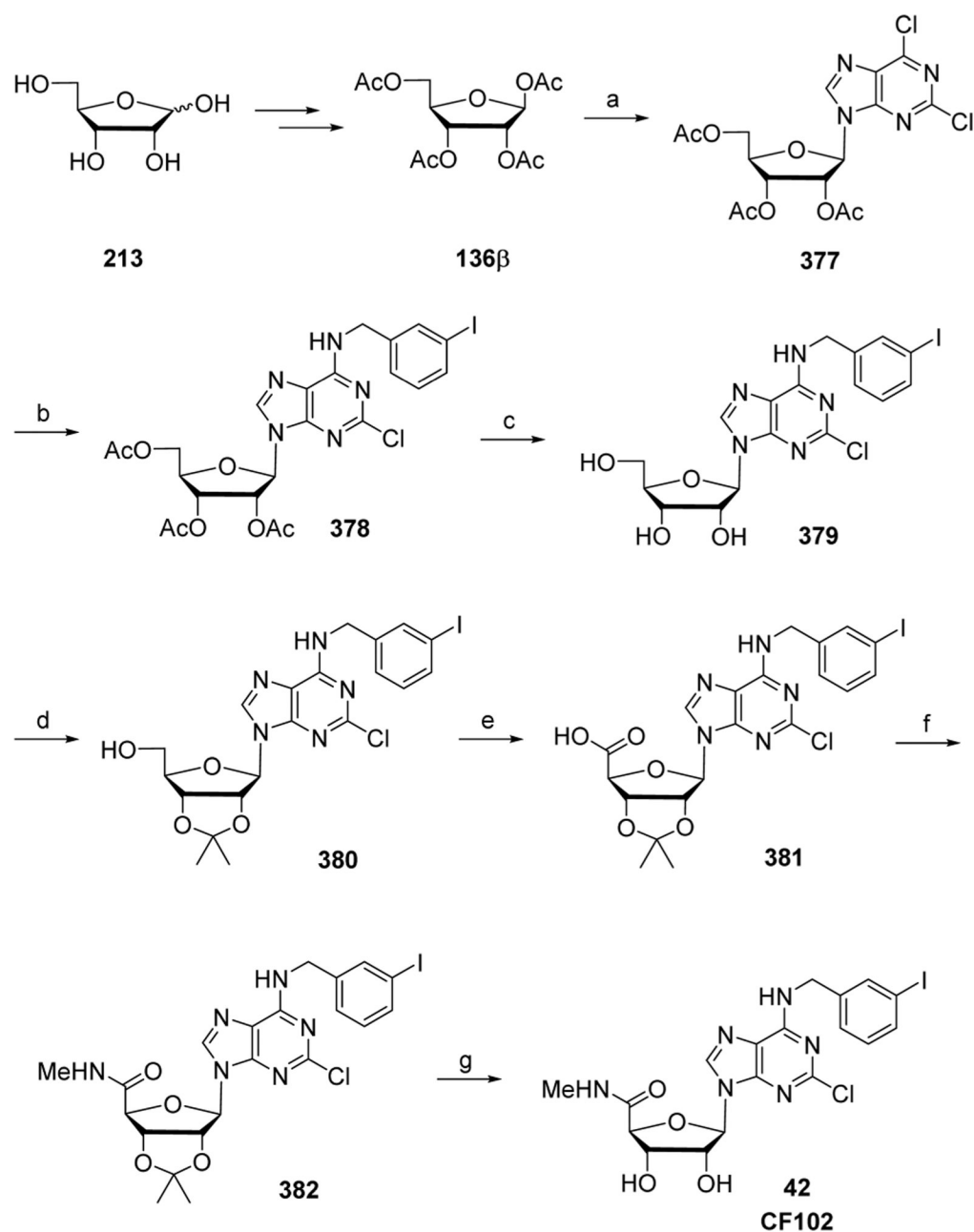
### Scheme 65. Synthesis of EPZ-5676 43<sup>a</sup>

<sup>a</sup>Reagents and conditions: (a) BnBr, K<sub>2</sub>CO<sub>3</sub>, TBAI, Me<sub>2</sub>CO, rt, 2 d, 92%; (b) Zn(Cu), ClCOCCl<sub>3</sub>, CH<sub>3</sub>OCH<sub>2</sub>CH<sub>2</sub>OCH<sub>3</sub>, Et<sub>2</sub>O, 50 °C, 3 days, 96%; (c) Zn, AcOH, 50 °C, 2 h, 93%; (d) H<sub>2</sub>, Pd/C, (CH<sub>3</sub>)<sub>2</sub>CHCOOCH<sub>3</sub>, PhMe, rt, 20 h, then DCHA, 20 °C, 18 h, 85%; (e) dioxane, DMF, (COCl)<sub>2</sub>, 20 °C, 18 h, then 4-*tert*-butyl-2-nitroaniline, dioxane, 20–40 °C, 5 h, then 20 °C, 18 h, 89%; (f) Fe, AcOH, 75 °C then rt, overnight, 95%; (g) STAB, Me<sub>2</sub>CO, MeOH, AcOH, 20 °C, 2 h, 95%; (h) STAB, MeCN, 55 °C, 16 h, 83%; (i) HCl, MeOH, 45 °C, 9 h then rt, overnight, 93%; (j) MeCN/H<sub>2</sub>O, crystallization, 65%.



**Scheme 66. Synthesis of CF102 42<sup>a</sup>**

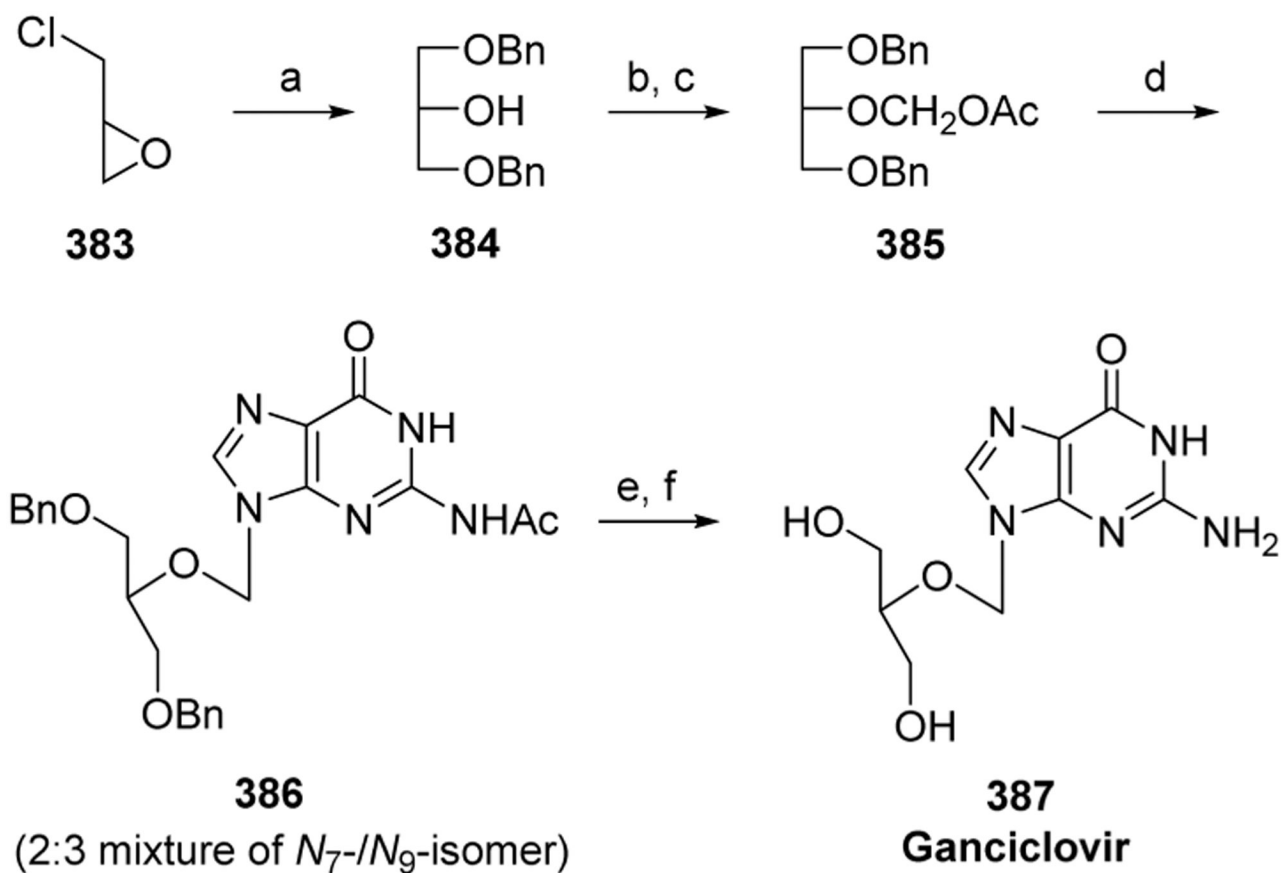
<sup>a</sup>Reagents and conditions: (a) TBDPSCl, DMAP, DMF, rt, 18 h, 55%; (b) BzCl, py-DCM, 0 °C to rt, 17 h, 99%; (c) *n*-Bu<sub>4</sub>NF, THF, rt, 2 h, 89%; (d) RuO<sub>2</sub>, NaIO<sub>4</sub>, CHCl<sub>3</sub>, CH<sub>3</sub>CN, H<sub>2</sub>O, rt, 2.5 h, 90%; (e) EDC, DMAP, MeOH, rt, 3 h, 72%; (f) MeNH<sub>2</sub>, THF, 50 °C, 15 h; (g) BzCl, py-DCM, rt, 3 h, 72% (over two steps); (h) Ac<sub>2</sub>O, H<sub>2</sub>SO<sub>4</sub>, AcOH, rt, 15 h, 34%; (i) 3-iodobenzylamine·HCl, Et<sub>3</sub>N, EtOH, rt, 5 d, 60%; (j) HMDS, (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, reflux, 4 h; (k) TMSOTf, DCE, reflux, 62 h, 33%; (l) NH<sub>3</sub>/MeOH, rt, 16 h, 69%.



**Scheme 67. Synthesis of CF102 42<sup>a</sup>**

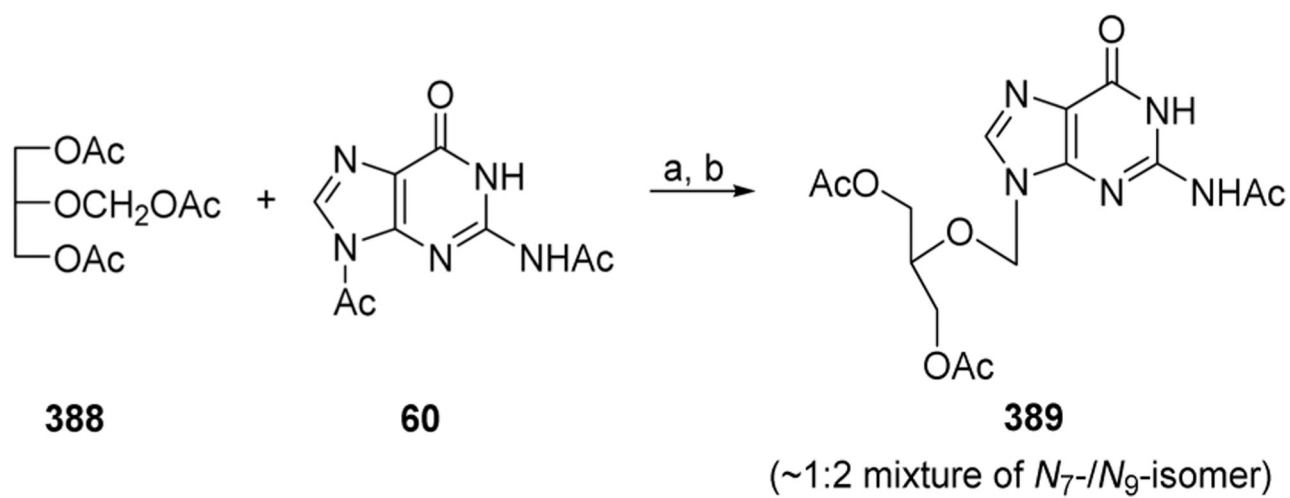
<sup>a</sup>Reagents and conditions: (a) BSA, TMSOTf, CH<sub>3</sub>CN, 2,6-dichloropurine, 50 °C, 18 h, 88%; (b) 3-iodobenzyl amine, Et<sub>3</sub>N, EtOH, rt, 2 d, 85%; (c) NaOMe, MeOH, DCM, rt, 1 h, 92%; (d) 2,2-dimethoxypropane, TsOH, DMF, rt, 12 h, 95%; (e) PDC, DMF, rt, 18 h, 70%; (f) MeNH<sub>2</sub>·HCl, HOBT, EDAC, DIEA, DCM, DMF, rt, 12 h, 71%; (g) 80% HCO<sub>2</sub>H, rt, 12 h, 83%.



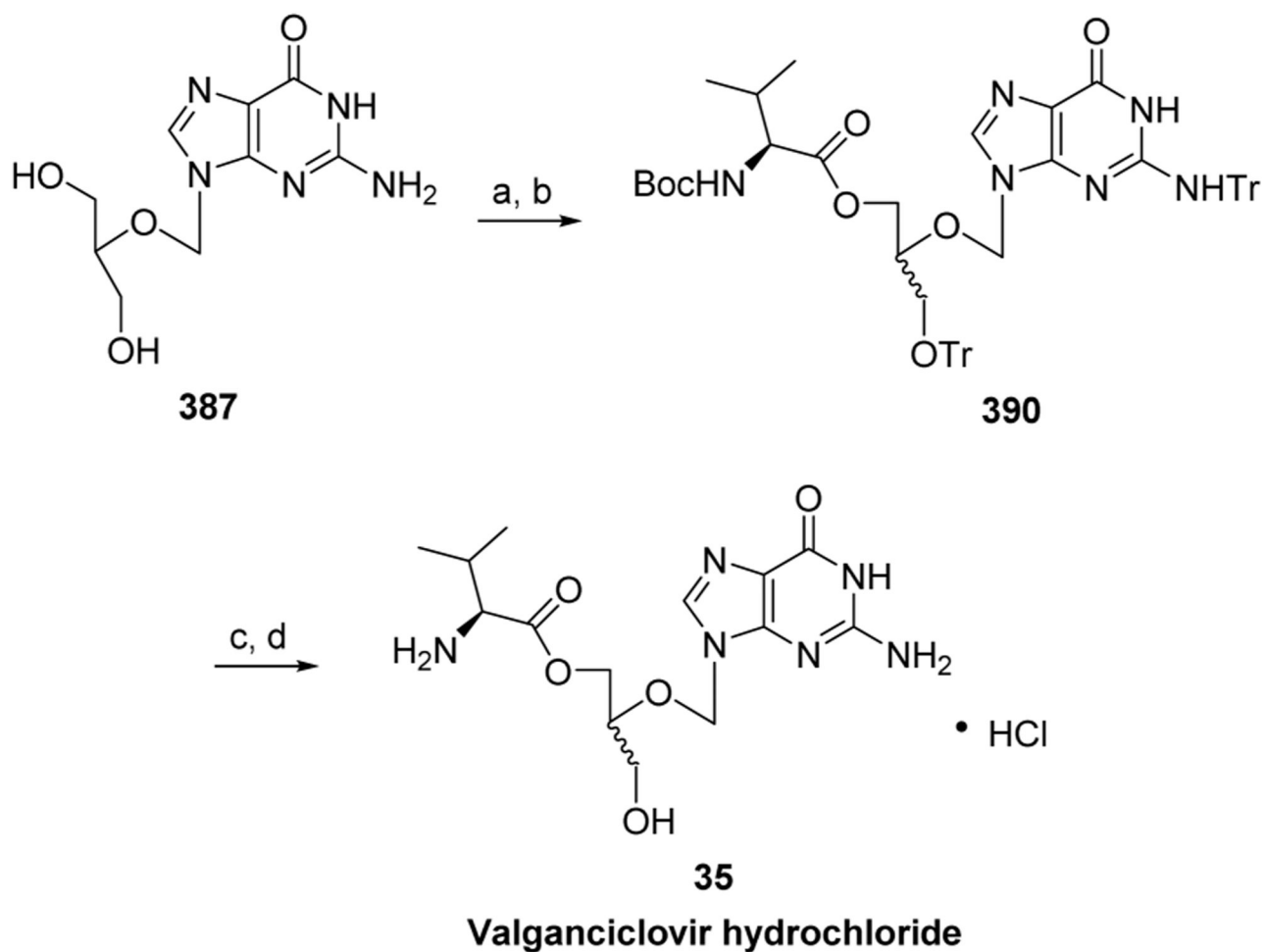


**Scheme 68. Synthesis of Ganciclovir 387<sup>a</sup>**

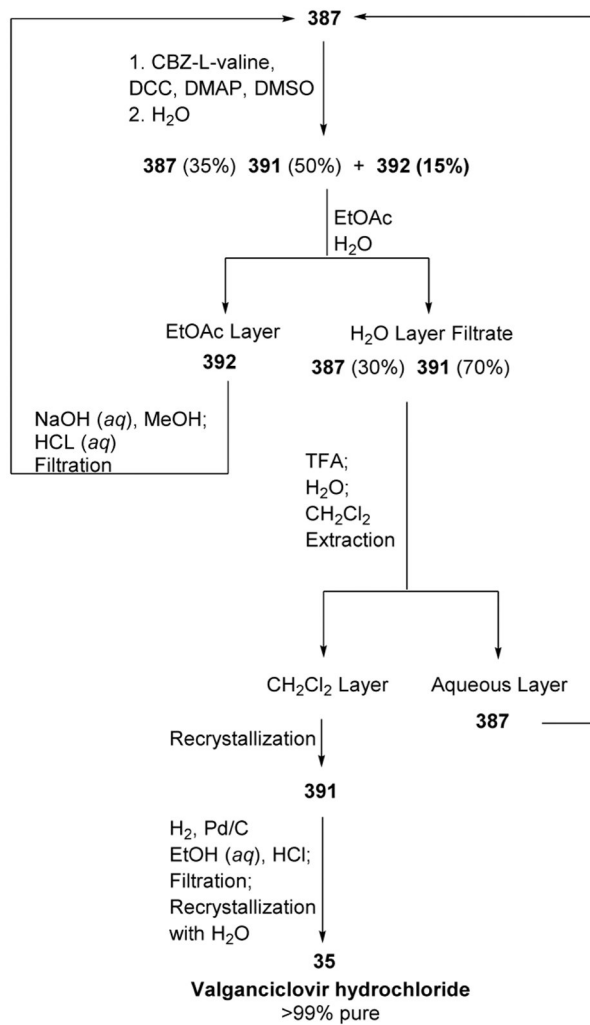
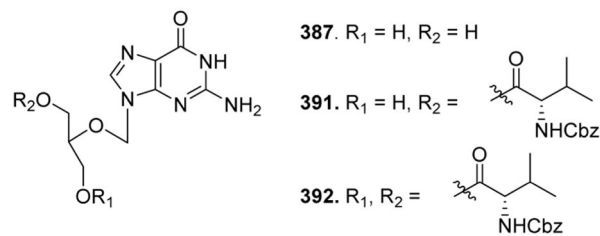
<sup>a</sup>Reagents and conditions: (a) BnOH, aq NaOH, rt, 16 h, 63%; (b) HCl (g),  $(\text{CH}_2\text{O})_m$ , DCM, 0 °C, 16 h; (c) KOAc,  $\text{Me}_2\text{CO}$ , rt, 16 h, ~100%; (d) Diacetylguanine, TsOH, sulfolane, 95 °C, 5 days, 31% ( $N_9$ -isomer); (e) 20%  $\text{Pd}(\text{OH})_2/\text{C}$ , cyclohexene–EtOH, reflux, 32 h; (f)  $\text{NH}_4\text{OH}$ –MeOH (1:1 v/v), rt, 16 h, 86% (2 steps).

**Scheme 69. Synthesis of 389<sup>a</sup>**

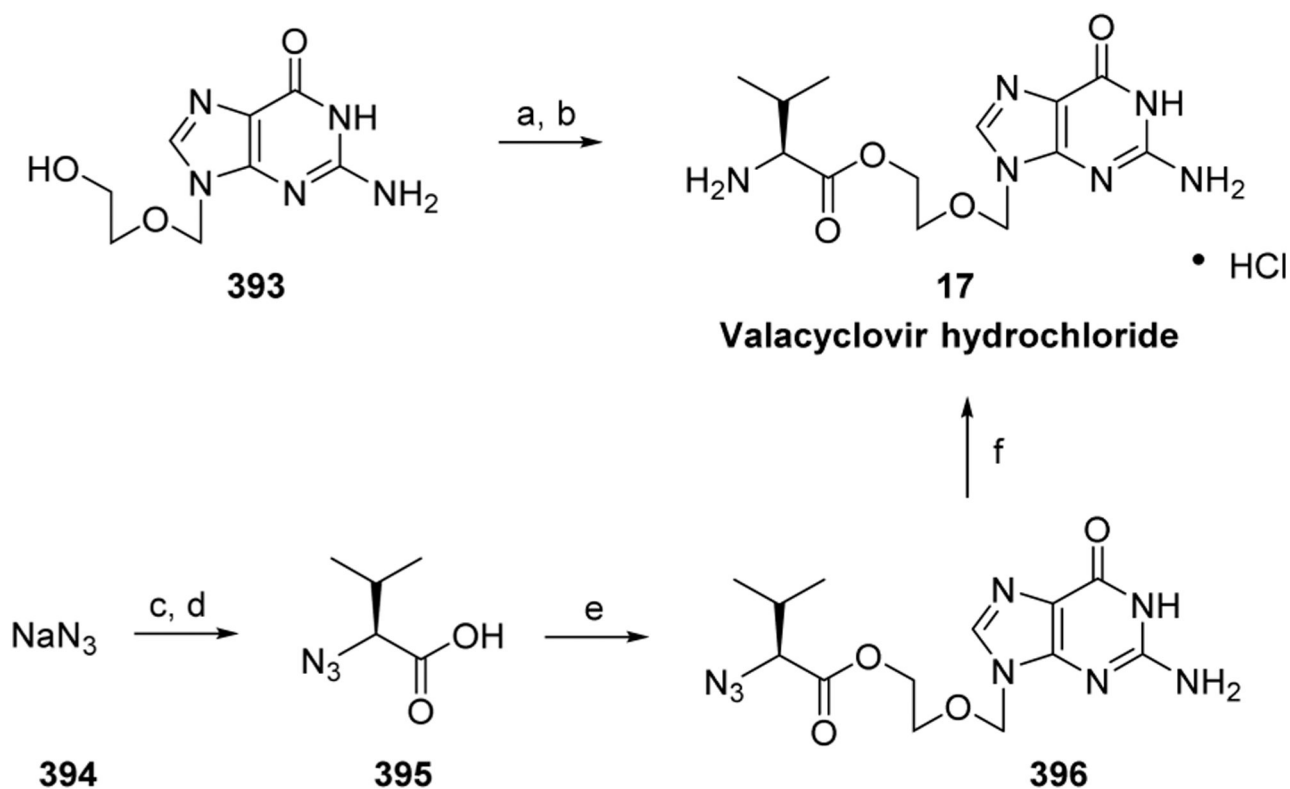
<sup>a</sup>Reagents and conditions: (a) TsOH, DMF, 95–100 °C, 40 h; (b) Crystallization, 70% ( $N_9$ -isomer).

**Scheme 70. Synthesis of Valganciclovir Hydrochloride 35<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) TrCl, DMAP, Et<sub>3</sub>N, DMF, 50 °C, overnight; (b) Boc-L-valine, DCC, DMAP, Et<sub>3</sub>N, DCM, rt, 16.5 h; (c) TFA, TFE, 20 °C, 5 h; (d) H<sub>2</sub>, Pd/C, then HCl/MeOH, 34% (over four steps).

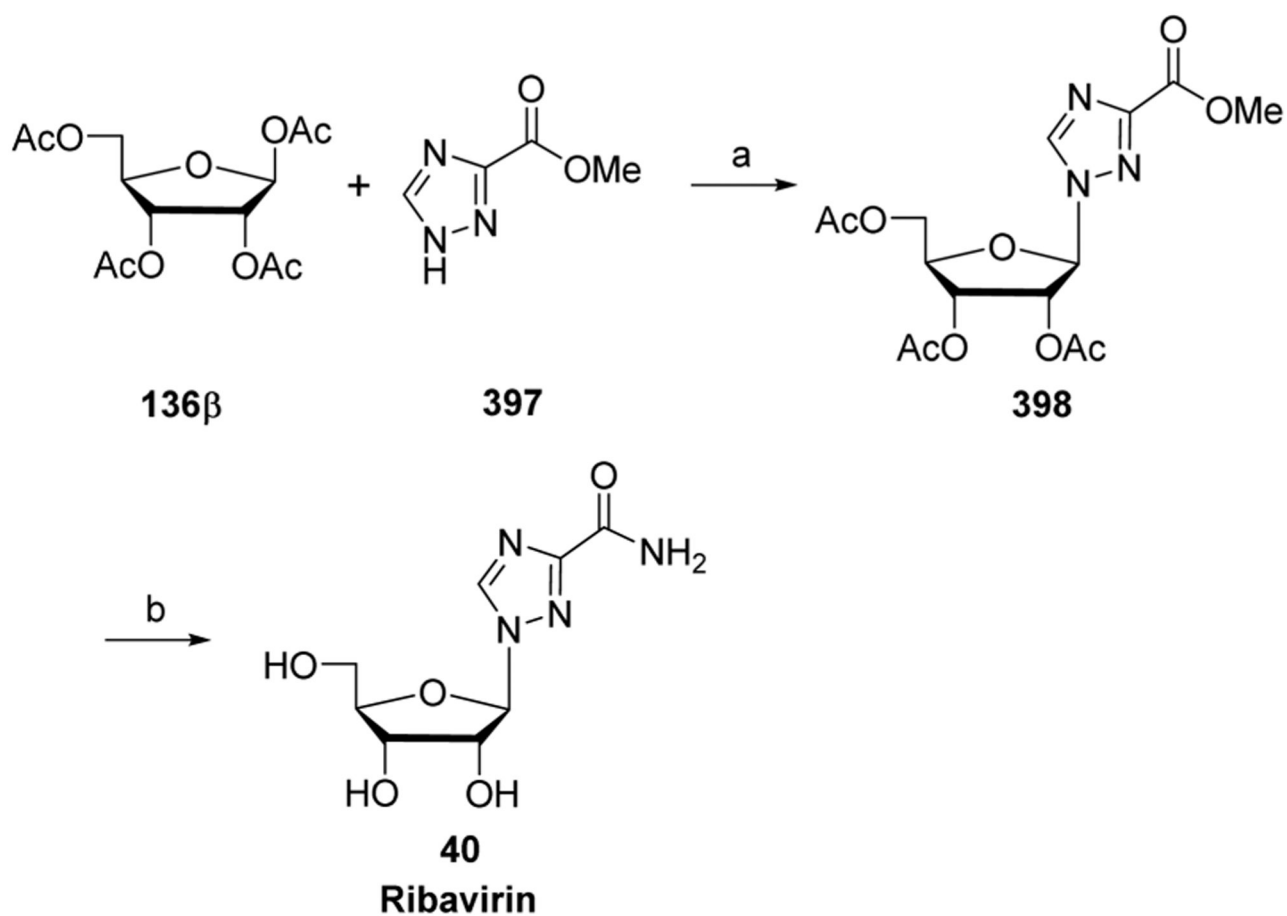


**Scheme 71.**  
 Preparation of Valganciclovir Hydrochloride 35

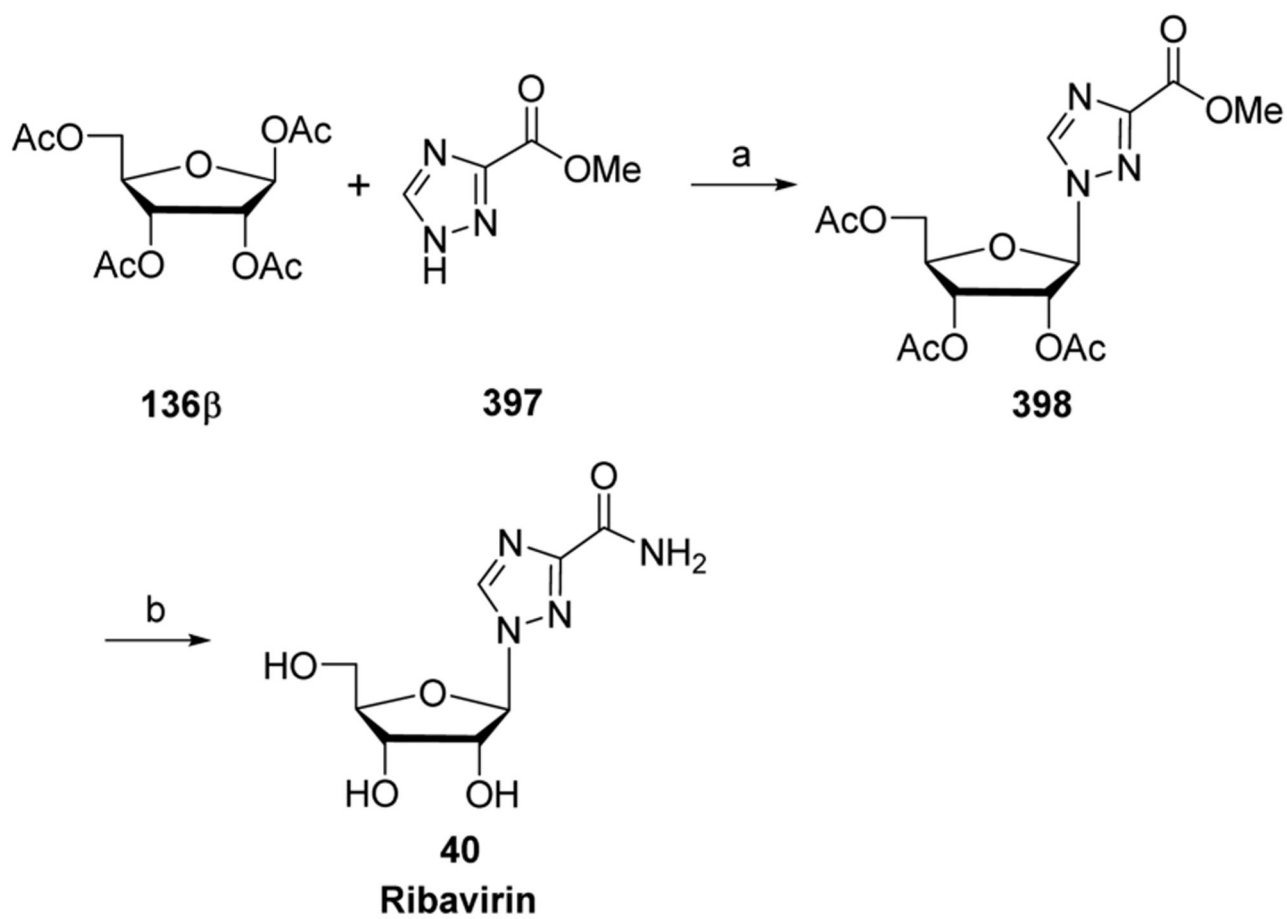


**Scheme 72. Synthesis of Valacyclovir Hydrochloride 17<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Cbz-L-val-OH, DCC, DMAP, DMF, 60 °C to rt, 12 h; (b) Pd/C, H<sub>2</sub>, MeOH, THF, aq HCl, 55% (over two steps); (c) SO<sub>2</sub>Cl<sub>2</sub>, imidazole, MeCN; (d) L-valine, CuSO<sub>4</sub>, K<sub>2</sub>CO<sub>3</sub>, MeCN; (e) Acyclovir, DCC, DMAP, DMF, 10–15 °C, 45 min, 82%; (f) Raney Ni, EtOH, 50–60 °C then aq HCl, 82%.

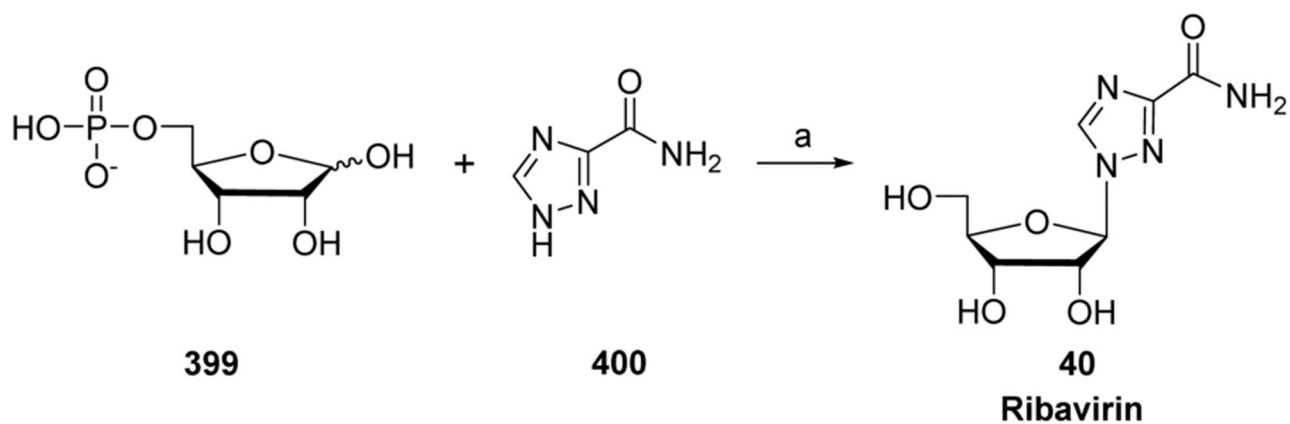
**Scheme 73. Synthesis of Ribavirin 40<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) SnCl<sub>4</sub>, DCM, 15–20 °C, then reflux, 2 h, 70%; (b) NaOMe/MeOH, 10 °C, 3 h then NH<sub>3</sub>/MeOH, 20 °C, 4 h, 79%.

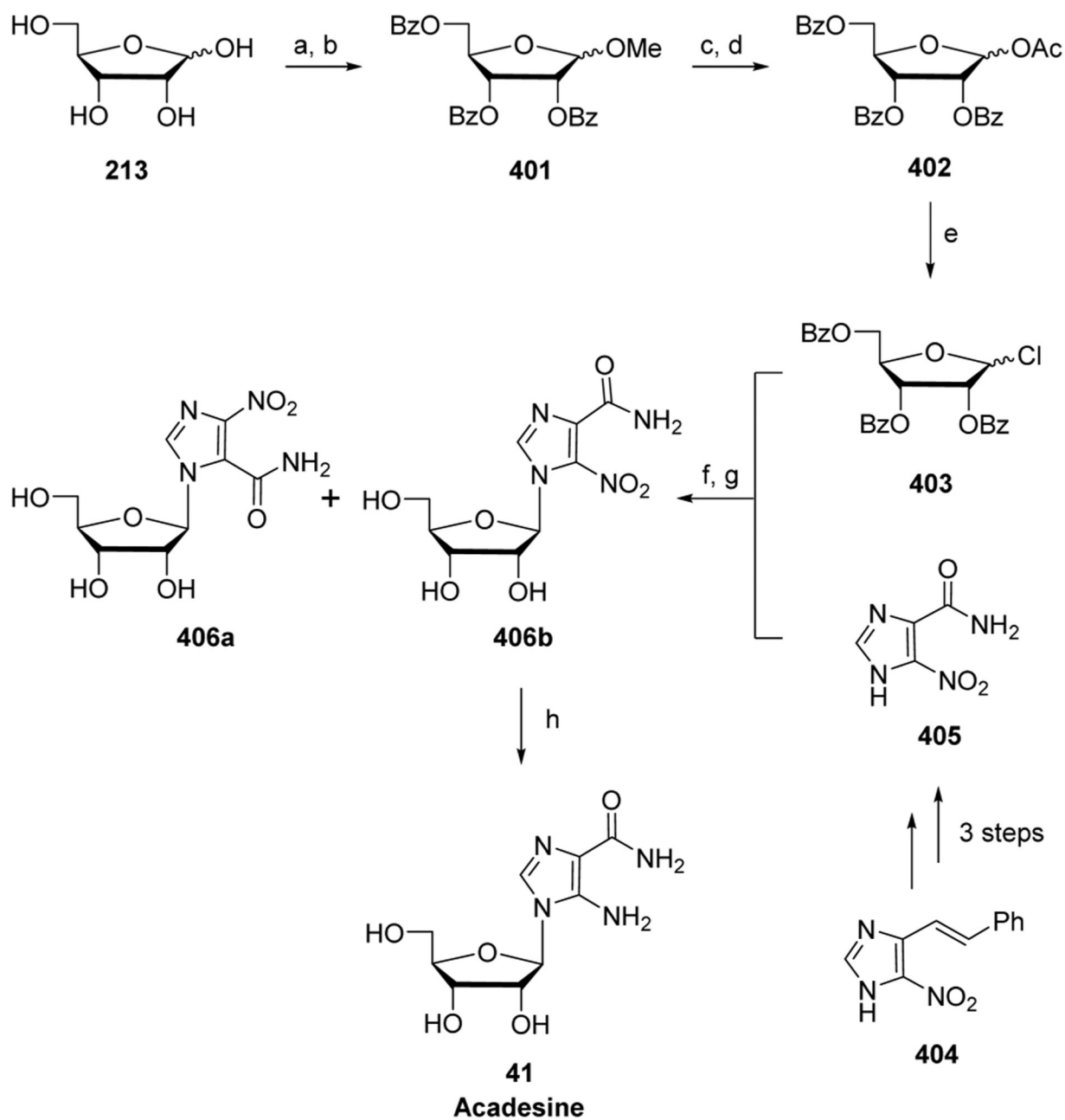
**Scheme 74. Large-Scale Synthesis of Ribavirin 40<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) TfOH, 135 °C, 2 h, vacuum, 92%; (b) NH<sub>3</sub>/MeOH, 20 °C, 40 h, 83%.

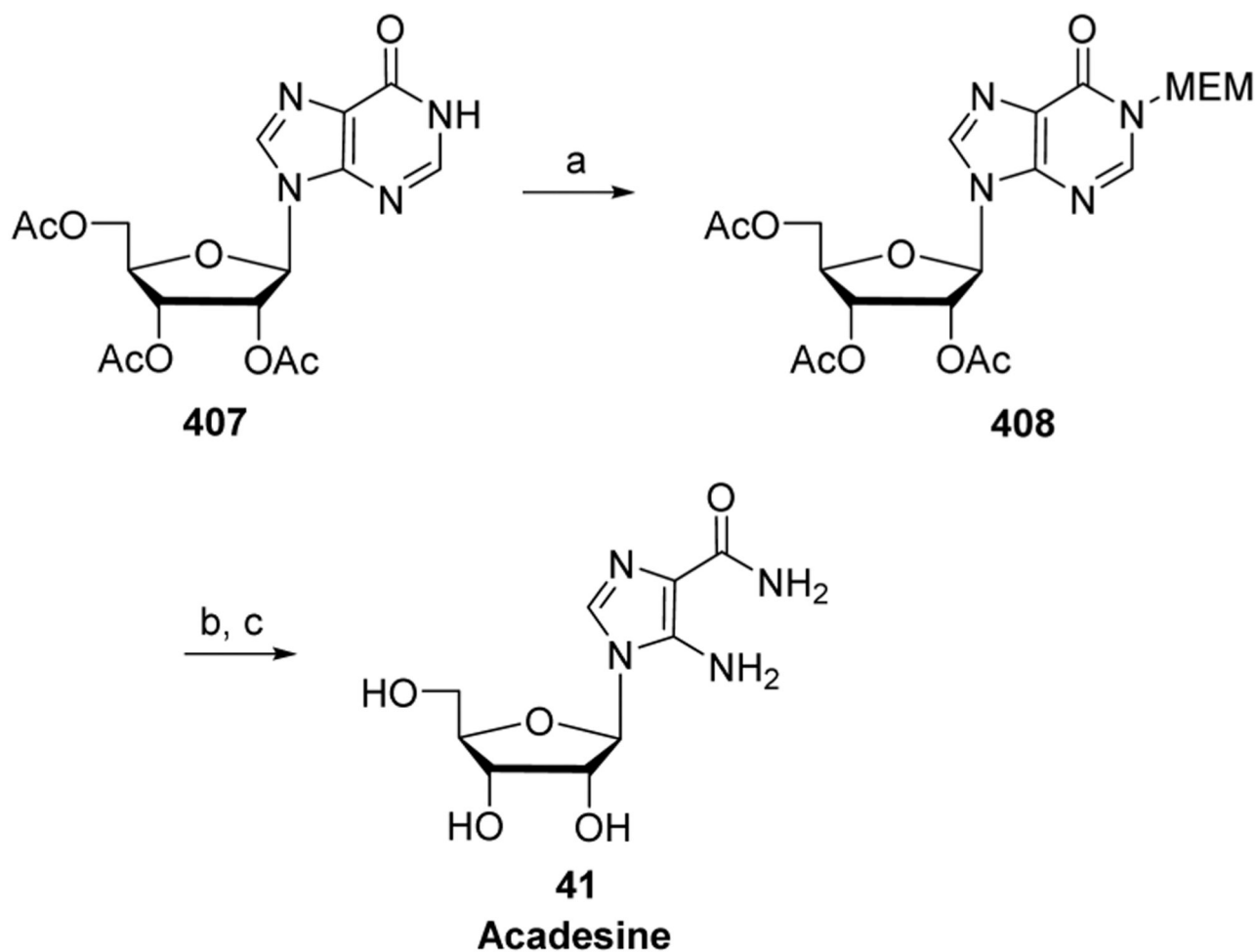


**Scheme 75. Synthesis of Ribavirin 40<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) Phosphopentomutase, purine nucleoside phosphorylase, tris-buffer, 100%.

**Scheme 76. Synthesis of Acadesine 41<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) cat. HCl, MeOH; (b) BzCl, pyr, DCM; (c) HBr/HOAc; (d) Ac<sub>2</sub>O, pyr, 48 h, 57% (from D-ribose); (e) HCl/ether, rt, 7 days; (f) AgCl, xylene, reflux, 1 h; (g) NH<sub>3</sub>/MeOH, 90 h, 19% (**406a**:**406b** = 1:1); (h) H<sub>2</sub> (1 atm), PtO<sub>2</sub>, H<sub>2</sub>O, 1 h, 12%.

**Scheme 77. Synthesis of Acadesine 41<sup>a</sup>**

<sup>a</sup>Reagents and conditions: (a) MEMCl, DIPEA, DCM, 0 °C, 65 min, 86%; (b) NH<sub>4</sub>OH, MeOH, rt, 1 h; (c) aq NaOH, reflux, 1 h, 73%.