

Biotransformation of 1,8-Dihydroxyanthraquinone into Peniphenone under the Fermentation of *Aleurodiscus mirabilis*

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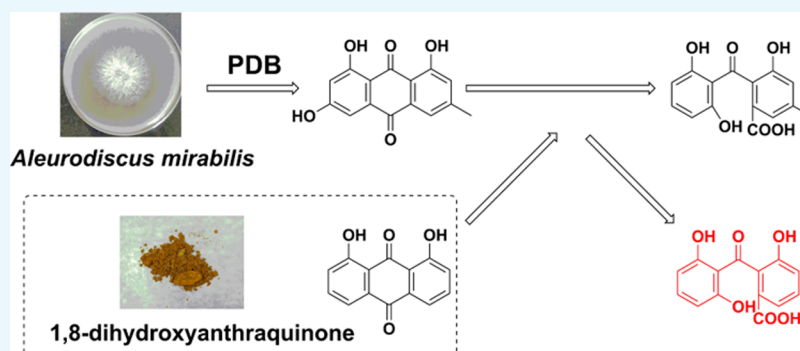
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ABSTRACT: The present study verified that 1,8-dihydroxyanthraquinone (1), a common component in some industrial raw materials and dyes, could be converted into peniphenone (2), which possesses immunosuppressive activity and other medicinal potential, by *Aleurodiscus mirabilis* fermentation. The yield of peniphenone (2) after 7 days of fermentation was $11.05 \pm 2.19\%$. To reveal the transformation mechanism, two secondary metabolites, emodin (3) and monodictyphenone (4), were isolated from the fermentation broth of *A. mirabilis*, implying that polyketide metabolic pathways from emodin (3) to monodictyphenone (4) might exist in *A. mirabilis*. 1,8-Dihydroxyanthraquinone (1) was suspected to be converted into peniphenone (2) via the same pathway since emodin (3) and 1,8-dihydroxyanthraquinone (1) share very similar skeletons. The P450 enzyme and Baeyer–Villiger oxidase in *A. mirabilis* were confirmed to catalyze this biotransformation on the basis of ultra-performance liquid chromatography-mass spectrometry (UPLC-MS) analysis. This novel investigation could shed light on the mechanism and therefore development of peniphenone production from 1,8-dihydroxyanthraquinone by microbial fermentation.

INTRODUCTION

Biotransformation shows great advantages compared with chemical synthesis due to its regio- and stereoselectivity, environmentally friendly procedures, and mild reaction conditions.^{1,2} In general, biotransformations can replace a multistep synthesis with a single microbial transformation.^{3,4} Therefore, bioconversion could play a critical role in the synthesis of bioactive compounds.⁵ Our previous studies have shown that microbial autogenic metabolic processes and many transformations have been achieved by microorganisms.^{6–11}

Peniphenone (2) is a benzophenone^{12,13} that was first isolated from the mangrove endophytic fungus *Penicillium* sp. ZJ-SY2 and has been reported to possess satisfactory immunosuppressive activity (the IC_{50} values of Con A-induced and LPS-induced were 8.1 and 9.3 $\mu\text{g/mL}$, respectively).¹⁴ Therefore, investigation into the large-scale production of peniphenone (2) by microbial fermentation is crucial in this field.

Aleurodiscus mirabilis is characterized by small disc-like or squamous fruiting bodies, large amyloid spores, and conspicuous sterile elements in the hymenium. Its active

metabolites have been reported in the literature.¹⁵ However, there has been no report on the biotransformation of *A. mirabilis*. In the present study, the biotransformation of 1,8-dihydroxyanthraquinone (1) to peniphenone (2) was detected in *A. mirabilis* fermentation broth. 1,8-Dihydroxyanthraquinone (1) is a common component in some industrial raw materials and dyes.¹⁶ To investigate the process of this biotransformation, the secondary metabolites of *A. mirabilis* were isolated and analyzed. Two key compounds, emodin (3) and monodictyphenone (4), were isolated from the fungal fermentation broth without the addition of compound 1, which indicates that polyketide metabolism related to emodin may exist in *A. mirabilis*. As shown in Figure 1, the biological transformation pathway of 2 is probably identical to that of

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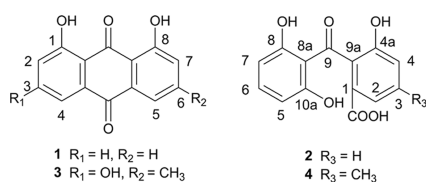


Figure 1. Structures of compounds 1–4.

emodin because 1,8-dihydroxyanthraquinone (**1**) shares the same moiety with emodin (an intermediate product in the synthetic process of a fungal polyketide). Therefore, this biotransformation might be catalyzed by a P450 enzyme and a Baeyer–Villiger oxidase. The conversion efficiency was determined by high-performance liquid chromatography (HPLC). This is the first report to investigate the biotransformation of 1,8-dihydroxyanthraquinone and could serve as a reference for further studies on the internal metabolism of this compound.

RESULTS AND DISCUSSION

Screening of Biotransformation and Transformed Product Yield. *A. mirabilis* was initially screened for its ability to catalyze biotransformation reactions using PDB with 1,8-dihydroxyanthraquinone as the substrate. HPLC experiments revealed that *A. mirabilis* fermentation could enhance the prominent biotransformation and reduce the metabolism of compound **3** (Figure 2). Fermentation of 1,8-dihydroxyanthraquinone by *A. mirabilis* for 7 days followed by separation of the transformation product yielded compound **2**. The transformation process and transformed product yield under different culture conditions were also investigated (Figure S4). The yield of compound **2** was highest when the fermentation temperature was controlled at 28 °C and the pH value of the culture medium was 7. As the fermentation time increased, compound **2** gradually appeared, reaching its highest yield 7 days after substrate addition. Then, the yield remained the same, corresponding to a maximum yield of $11.15 \pm 2.19\%$. The poor water solubility of 1,8-dihydroxyanthraquinone may be an obstacle for further yield improvement.^{17,18}

Analysis of the Causes of Biotransformation. To better understand why the transformation occurs, the secondary metabolites of *A. mirabilis* are worthy of investigation. Compounds **3** and **4** were isolated from the extracts of *A. mirabilis* that had been fermented for 7 days. The biosynthetic pathway of compound **4** has been reported in *Aspergillus nidulans*,¹⁹ and it can provide guidance for the current bioconversion process because **4** has a similar molecular structure to the transformed product **2**. Methylmalonyl-coenzyme A (Mal-CoA) was used as a starting material to produce atrochryson carboxylic acid catalyzed by microbial polyketide synthases (PKSs). Atrochryson carboxylic acid loses two CO₂ molecules to emodin anthrone and is further oxidized to emodin (**3**); then, emodin (**3**) is catalyzed by microbial enzymes to form monodictyphenone (**4**). To confirm the presence of similar biological processes in *A. mirabilis*, the fermentation time was shortened to detect the intermediates. The broth of *A. mirabilis* was fermented for 3 days and analyzed by ultra-performance liquid chromatography-mass spectrometry (UPLC-MS), and molecular signals from four key intermediates, atrochryson carboxylic acid, atrochryson, emodin anthrone, and emodin (**3**), were detected (Figure 3). The results showed that *A. mirabilis* has

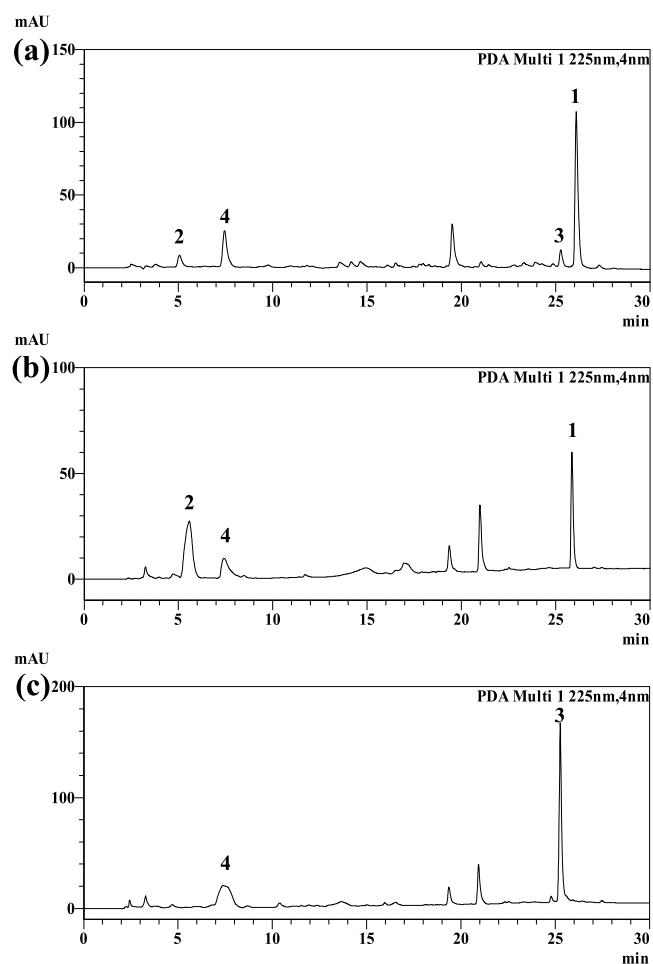


Figure 2. HPLC chromatograms of the extracts of 1,8-dihydroxyanthraquinone (**1**) by *A. mirabilis* fermentation for 3 days (a), 7 days (b), and the blank microbial sample fermented for 7 days (c).

the same metabolic process as that of *Aspergillus nidulans* reported in the literature.¹⁹ The metabolic pathway of **4** implied that 1,8-dihydroxyanthraquinone (**1**) could be converted into peniphenone (**2**) via the same pathway since emodin (**3**) and 1,8-dihydroxyanthraquinone (**1**) share very similar moieties.

Biotransformation Process of 1,8-Dihydroxyanthraquinone by *A. mirabilis*. The process of conversion from **3** to **4** is related to the expression of multiple monodictyphenone gene clusters, and *VerA/afIN* and *aFLY/hypA* play key roles in this biological process.¹⁹ The product of *VerA/afIN* is directly related to P450 mono-oxygenase in fungi,^{20,21} and *aFLY/hypA* might encode a Baeyer–Villiger oxidase.²² Cytochrome P450 mono-oxygenases might catalyze epoxidation, typically by insertion of an oxygen atom from atmospheric dioxygen into a conjugated double bond.²³ Baeyer–Villiger oxidases catalyze the oxidative cleavage of a carbon–carbon bond adjacent to a carbonyl, which converts ketones to esters and cyclic ketones to lactones.²⁴ Referring to the conversion of **3** to **4**, electron transfer starts from OH-3 and is concentrated in the A ring of emodin, but why the A ring migrates to the oxygen instead of the C ring still remains unknown.¹⁹ To determine whether the P450 enzyme system is involved in this biotransformation, and since 1,8-dihydroxyanthraquinone (**1**) lacks a key chemical group (OH-3) compared with emodin (**3**), special inhibition experiments were performed. 1-Aminobenzotriazole (ABT)

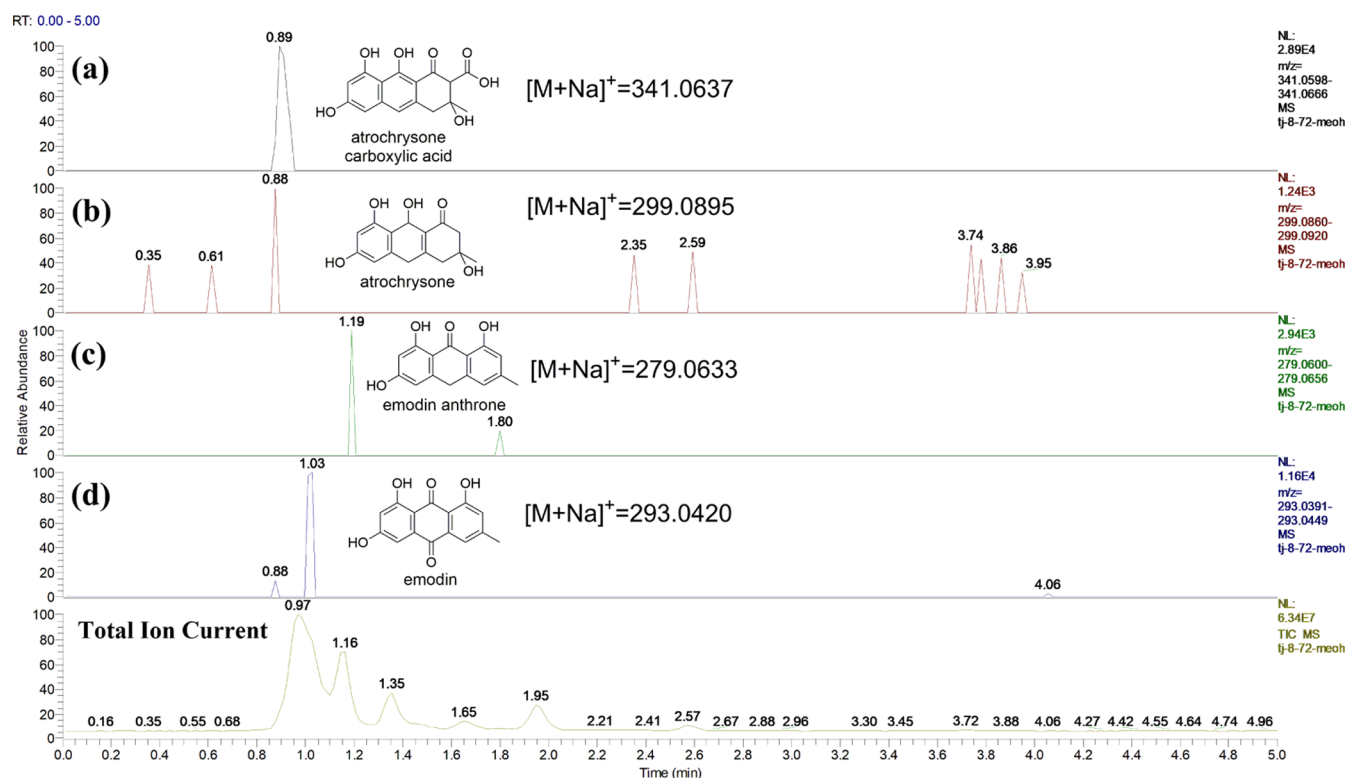


Figure 3. Molecular signals from four key intermediates, atrochryson carboxylic acid (a), atrochryson (b), emodin anthrone (c), and emodin (d), were detected by UPLC-MS.

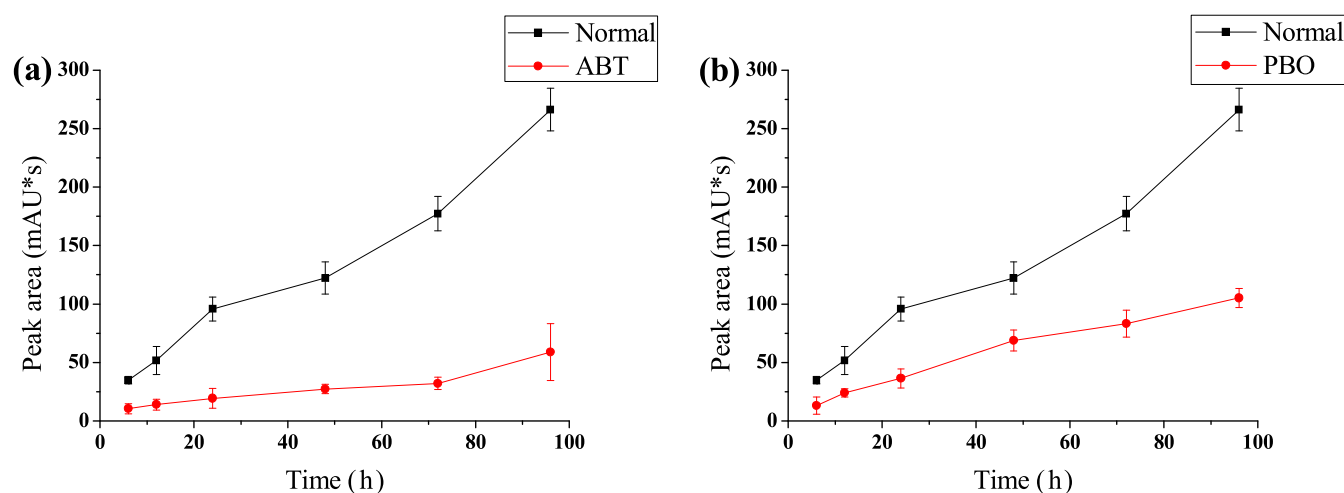


Figure 4. Effects of 1-aminobenzotriazole (ABT, 40 $\mu\text{g}/\text{mL}$) (a) and piperonyl butoxide (PBO, 40 $\mu\text{g}/\text{mL}$) (b) on the biotransformation of 1 to 2.

and piperonyl butoxide (PBO) are P450 enzyme inhibitors that are widely used to estimate the inhibition and induction of reactions mediated by cytochromes P450.^{25–27} The rate of the biological reaction catalyzed by the P450 enzyme notably decreased after the inhibitors were added. As shown in Figure 4a,b, ABT and PBO exhibited a significant inhibitory effect on the conversion of 1,8-dihydroxyanthraquinone (1) into peniphenone (2) at a concentration of 40 $\mu\text{g}/\text{mL}$. Under such conditions, the production of peniphenone (2) was slower than that of the positive control. The experiments showed that P450 was involved in this biotransformation, and the electron transfer of this reaction could occur with OH-1 of compound 1.

To verify the rationality of the transformation process, UPLC-MS was selected to detect the extracts of 1,8-dihydroxyanthraquinone (1) by *A. mirabilis* fermentation for 3 days, and two key intermediates, compounds 5 (P450 enzyme-catalyzed product) and 6 (Baeyer–Villiger oxidase-catalyzed product), were detected (Figure 5). The reason that this biological transformation is the most promising is because 1,8-dihydroxyanthraquinone (1) has a similar structure to that of emodin (3), which is involved in the metabolism of the fungus *A. mirabilis*. Reaction of 1,8-dihydroxyanthraquinone (1) was catalyzed by a P450 enzyme to form 5, and then it was subsequently catalyzed by Baeyer–Villiger oxidase to generate 6. Biotransformation product 2 was produced after hydrolysis,

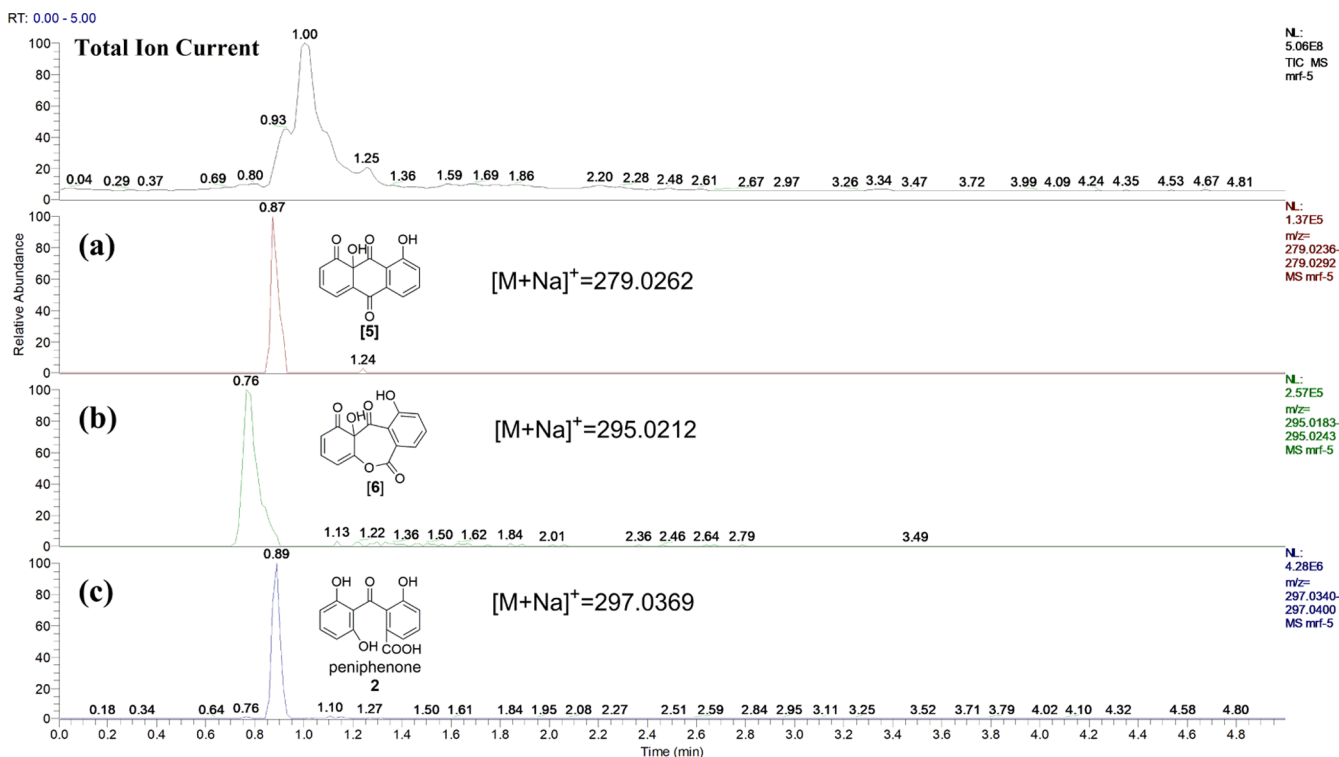


Figure 5. Molecular signals from the P450 enzyme-catalyzed product (5), Baeyer–Villiger oxidase-catalyzed product (6), and peniphenone (2) were detected by UPLC-MS.

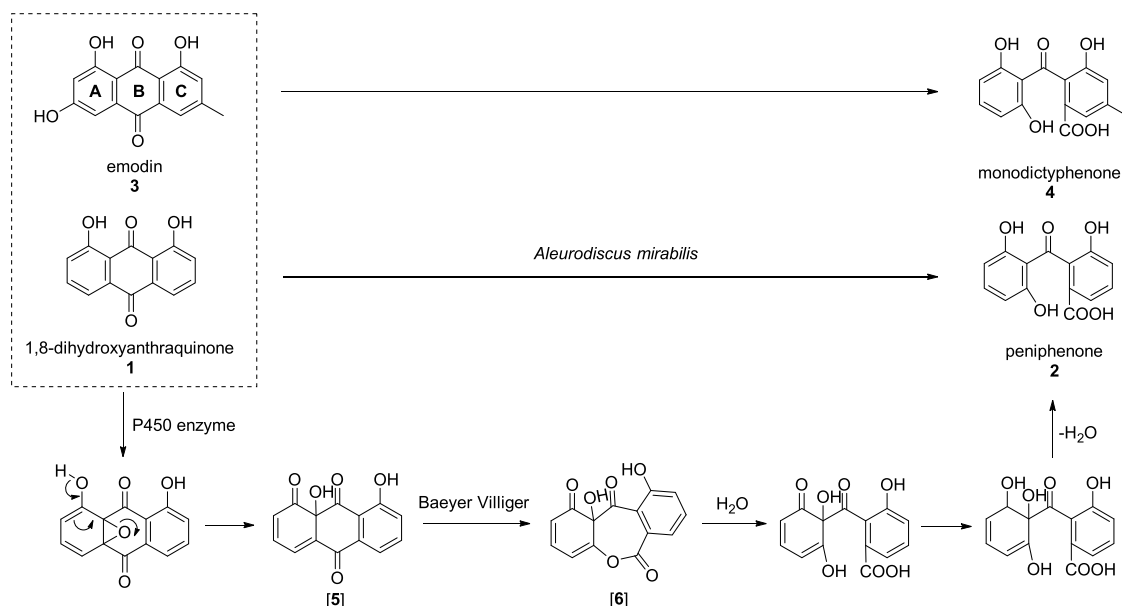


Figure 6. 1,8-Dihydroxyanthraquinone (1) has a similar structure to that of emodin (3) and is catalyzed to peniphenone (2) by enzymes involved in the metabolism of the fungus *A. mirabilis*.

reduction of a ketone,¹⁹ and the dehydration of compound 6. The biological transformation process is speculated in Figure 6.

CONCLUSIONS

In summary, the present study confirmed that 1,8-dihydroxyanthraquinone could be converted into peniphenone (2) by *A. mirabilis*. The likely reason for the transformation is that 1,8-dihydroxyanthraquinone has a similar structure to emodin (an intermediate of fungal polyketide metabolism) and is catalyzed by a P450 enzyme and a Baeyer–Villiger oxidase. This is the

first study to clarify the biotransformation of 1,8-dihydroxyanthraquinone, and this study could serve as a reference for further investigations on the internal metabolism of this compound.

MATERIALS AND METHODS

Chemicals. 1,8-Dihydroxyanthraquinone (compound 1, CAS: 117-10-2) was purchased from J&K Scientific Ltd. (Beijing, China). Methanol (Hipure Chem, China) was of HPLC grade and purchased from Guangdong Xilong Chemical

Reagent Co., Ltd. (Guangdong, China). 1-Aminobenzotriazole (ABT, 98%) and piperonyl butoxide (PBO, 90%) were obtained from J&K Scientific Ltd. (Beijing, China). Water (resistivity ≥ 18.25 M Ω /cm) was purified using a water purification system (Chengdu, China). MeOD (CAS: 881-98-3, 99.8 atom % D, containing 0.03% tetramethylsilane (TMS)) and dimethyl sulfoxide (DMSO)- d_6 (CAS: 2206-27-1, 99.8 atom % D, containing 0.03% TMS) were purchased from Energy-Chemical (Shanghai, China). All other reagents were analytically pure and obtained from Shengbi Co., Ltd. (Yunnan, China).

Nuclear Magnetic Resonance (NMR) Spectroscopy.

NMR spectra were recorded on a Bruker Avance 400 MHz spectrometer (400.00 MHz, ^1H ; 100.00 MHz, ^{13}C). Chemical shifts are given in ppm (δ) downfield from the TMS internal standard.

HPLC and UPLC-MS. HPLC analysis was performed using a Shimadzu LC-20 Series equipped with a Shimadzu LC-20AR dual pump, an SPD-M20A photodiode array (PDA) detector, and an Agilent ZORBAX XDB-C18 column (4.6 \times 250 mm 2 5-micron). MeOH–0.1% HOAc was used as the mobile phase under gradient elution mode (0–5 min, 45:55, v/v; 5–10 min, 55:45, v/v; 10–15 min, 60:40, v/v; 15–20 min, 75:25, v/v; 20–35 min, 100:0 v/v; 1.0 mL/min; 225 nm). The injection volume was set to 10 μL for each injection, and analytical experiments were carried out at 25 $^\circ\text{C}$.

High-resolution UPLC-MS on an Ultimate 3000-LTQ Orbitrap XL was performed using the following eluents: (A) 0.1% aqueous formic acid and (B) MeOH under isocratic elution mode. The applied gradient was 95% solvent B held for 5 min at a flow rate of 0.3 mL/min. The chromatographic column was a Hypersil GOLD C18 column (2.1 \times 100 mm 2 1.9-micron). Full-scan mass spectra were measured from m/z 50–1200, the spray voltage was 3.5 kV, and the capillary temperature was 300 $^\circ\text{C}$.

Microorganisms and Culture Medium. *A. mirabilis* was isolated from fresh *Glyceria elata* and frozen at -80 $^\circ\text{C}$ in the lab at the Yunnan Institute of Microbiology, Yunnan Province, China. The fungus was maintained in potato dextrose agar (PDA, 1 L of water, 200 g of potato, 20 g of dextrose, and 15 g of agar) slants and stored at 4 $^\circ\text{C}$ for mutation. The seed culture medium and fermentation media were potato dextrose broth (PDB, 1 L of water, 200 g of fresh potato, and 20 g of dextrose). All media were sterilized in an autoclave at 121 $^\circ\text{C}$ and 1.06 kg/cm 2 for 30 min and cooled prior to use.

Fermentation of 1,8-Dihydroxyanthraquinone. *A. mirabilis* was inoculated into the PDA slant culture medium and incubated at 28 $^\circ\text{C}$ for 7 days. Then, the fungal spores on the mature slants were picked out with the tip of a needle and put into 150 mL Erlenmeyer flasks containing 50 mL of PDB at 180 rpm and 28 $^\circ\text{C}$ for 3 days to prepare a spore suspension for the seed culture medium. Then, 500 mL Erlenmeyer flasks each containing 200 mL of PDB were individually inoculated with 20 μL of mature seed culture to prepare a spore suspension followed by incubation at 28 $^\circ\text{C}$ on a rotary shaker at 180 rpm for 72 h. Then, 1 mL of substrate (1,8-dihydroxyanthraquinone ethanol solution, 10 mg/mL) was injected, and the fermentation broth was incubated at 180 rpm and 28 $^\circ\text{C}$ for 7 days. The injection operation was carried out on a super clean bench. Blank control samples were cultured in the same manner, and the substrate was replaced by ethanol. Then, the fermented broth was extracted with an ultrasonic cleaner three times with ethyl acetate (EtOAc, 200 mL, 30 min

each time), and the solvent was removed under vacuum. The extracts were dissolved in MeOH and prepared as a 20 mg/mL solution. Then, the extracts were filtered through a 0.22 μm filter for HPLC sample analysis.

Biotransformation Process Investigation by UPLC-MS. *A. mirabilis* was grown in a shaking culture at 28 $^\circ\text{C}$ for 72 h in 500 mL Erlenmeyer flasks that contained 200 mL of sterile PDB. Then, 1 mL of substrate (1,8-dihydroxyanthraquinone ethanol solution, 10 mg/mL) was injected. The injection operation was carried out on a super clean bench. The fermentation broth was incubated at 180 rpm and 28 $^\circ\text{C}$ for 3 days. The blank control samples were cultured in the same manner, and the substrate was replaced by ethanol. The fungal fermentation broth was filtered through a 0.22 μm filter for UPLC-MS sample analysis.

Purification of the Metabolite. Compounds 2–4 were isolated with a Shimadzu LC-20 Series HPLC equipped with a Waters Spherisorb S10 ODS2 semi-prep column (20 \times 250 mm 2). The extracts of 1,8-dihydroxyanthraquinone that had been fermented for 7 days were subjected to preparative HPLC and eluted with MeOH–H $_2$ O (0–25 min, 35:65, v/v; 16.0 mL/min; 225 nm) to yield 2 (8.3 mg); the extracts of the blank control that had been fermented for 7 days were subjected to preparative HPLC and eluted with MeOH–H $_2$ O (0–15 min, 15:85; 15–20 min, 30:70; 20–25 min, 45:55, v/v; 16.0 mL/min; 225 nm) to yield 3 (2.7 mg) and 4 (6.2 mg).

Spectroscopic Data. Peniphenone (2). White amorphous powder; high-resolution electrospray ionization mass spectrometry (HRESI-MS) (m/z 297.0370 [$\text{M} + \text{Na}$] $^+$, C $_{14}\text{H}_{10}\text{O}_6$); ^1H NMR (MeOD, 400 MHz) δ_{H} : 7.45 (1H, d, $J = 7.7$ Hz, H-2), 7.23 (1H, t, $J = 8.0$ Hz, H-3), 6.97 (1H, d, $J = 8.0$ Hz, H-4), 6.25 (1H, d, $J = 8.2$ Hz, H-5), 7.19 (1H, t, $J = 5.6$ Hz, H-6), and 6.25 (1H, d, $J = 8.2$ Hz, H-7); ^{13}C NMR (MeOD, 100 MHz) δ_{C} : 134.6 (C-1), 121.9 (C-2), 129.6 (C-3), 120.6 (C-4), 154.6 (C-4a), 108.1 (C-5), 137.2 (C-6), 108.1 (C-7), 163.4 (C-8), 113.0 (C-8a), 203.6 (C-9), 130.8 (C-9a), 163.4 (C-10a), and 169.8 (–COOH).

Emodin (3). Yellow amorphous powder; HRESI-MS (m/z 293.0428 [$\text{M} + \text{Na}$] $^+$, C $_{15}\text{H}_{10}\text{O}_5$); ^1H NMR (DMSO- d_6 , 400 MHz) δ_{H} : 12.02 (1H, s, 1-OH), 11.95 (1H, s, 8-OH), 7.05 (1H, s, H-2), 7.40 (1H, s, H-4), 7.09 (1H, s, H-5), 6.54 (1H, s, H-7), and 2.37 (3H, s, 6-CH $_3$); ^{13}C NMR (DMSO- d_6 , 100 MHz) δ_{C} : 164.5 (C-1), 108.8 (C-2), 148.2 (C-3), 107.9 (C-4), 120.5 (C-5), 161.4 (C-6), 124.1 (C-7), 165.6 (C-8), 189.6 (C-9), 181.3 (C-10), 132.7 (C-11), 113.3 (C-12), 108.9 (C-13), 135.0 (C-14), and 21.6 (–CH $_3$).

Monodictyphenone (4). Amorphous powder; HRESI-MS (m/z 287.0559 [$\text{M} - \text{H}$] $^-$, C $_{15}\text{H}_{12}\text{O}_6$); ^1H NMR (MeOD, 400 MHz) δ_{H} : 7.19 (1H, s, H-2), 6.72 (1H, s, H-4), 6.19 (1H, d, $J = 8.1$ Hz, H-5), 7.08 (1H, t, $J = 8.1$ Hz, H-6), 6.17 (1H, d, $J = 8.1$ Hz, H-7), and 2.23 (3H, s, H-11); ^{13}C NMR (MeOD, 100 MHz) δ_{C} : 129.6 (C-1), 122.4 (C-2), 140.1 (C-3), 120.9 (C-4), 154.5 (C-4a), 108.0 (C-5), 137.1 (C-6), 108.2 (C-7), 162.9 (C-8), 113.4 (C-8a), 202.5 (C-9), 131.5 (C-9a), 163.2 (C-10a), 21.2 (–CH $_3$), and 167.7 (–COOH).

Determination of the Reaction Yield by HPLC Analysis. The quantification of compound 2 was analyzed with the standard curve method by HPLC. Different concentrations of compound 2 in MeOH were injected into the HPLC instrument, and a regression equation correlating the concentration and peak area was constructed. The contents of compound 2 were determined according to the regressive equation; then, the yield of compound 2 could be calculated.

The PDB medium was autoclaved and inoculated under the same conditions after it had been incubated for 72 h, and 1 mL of substrate (1,8-dihydroxyanthraquinone ethanol solution, 10 mg/mL) was injected. The injection operation was carried out on a super clean bench. The mixtures were incubated at 180 rpm and 28 °C for 1, 3, 5, 7, 9, 11, 13, or 15 days. Then, the fermented broth was extracted with an ultrasonic cleaner three times with EtOAc (200 mL, 30 min each time), and the solvent was removed under vacuum. Extracts were analyzed by HPLC.

Effects of Fermentation Temperature. Effects of fermentation temperature were evaluated using the fermentation method described in the [Determination of the Reaction Yield by HPLC Analysis](#) section. The fermentation broth was incubated at 180 rpm for 7 days. The fermentation temperature was set to 20 °C, 28 °C, or 37 °C. Extracts were analyzed by HPLC. Each sample was repeated in triplicate.

Effects of pH Value of Fermentation Broth. Effects of pH value of fermentation broth were evaluated using the fermentation method described in the [Determination of the Reaction Yield by HPLC Analysis](#) section. The pH value of the fermentation broth was buffered to 5, 6, 7, 8, or 9 by phosphate buffer. Then 1 mL of substrate (1,8-dihydroxyanthraquinone ethanol solution, 10 mg/mL) was injected. The mixtures were incubated at 180 rpm and 28 °C for 7 days. Extracts were analyzed by HPLC. Each sample was repeated in triplicate.

Effects of the Inhibitors ABT and PBO. The 500 mL Erlenmeyer flasks each containing 200 mL of PDB were individually inoculated with 20 μ L of mature seed culture to prepare a spore suspension and then incubated at 28 °C on a rotary shaker at 180 rpm for 24 h. Then, 1 mL of a solution of ABT and PBO in ethanol (8.0 mg/mL) was added to the spore suspension (40 μ g/mL, final concentration). After an additional 48 h of culture, 1 mL of substrate (1,8-dihydroxyanthraquinone ethanol solution, 20 mg/mL) was added. Spore suspensions without inhibitors were used as positive control. All injection operations were carried out on a super clean bench. The concentration of compound 2 was determined by HPLC analysis after 6, 12, 24, 48, 72, and 96 h. Each sample was repeated in triplicate.

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acsomega.0c05216>.

Figure S1: morphology of *Aleurodiscus mirabilis*; Figure S2: standard working curve for compound 2; Figure S3: UV-spectrum for compound 2 by PDA detector; Figure S4: yields of compound 2 from different fermentation temperatures, pH, and time; Figure S5: ^1H NMR spectrum (400 MHz, MeOD) of compound 2; Figure S6: ^{13}C NMR spectrum (100 MHz, MeOD) of compound 2; Figure S7: H–H COSY spectrum of compound 2; Figure S8: HMBC spectrum of compound 2; Figure S9: HSQC spectrum of compound 2; Figure S10: ^1H NMR spectrum (400 MHz, DMSO- d_6) of compound 3; Figure S11: ^{13}C NMR spectrum (100 MHz, DMSO- d_6) of compound 3; Figure S12: ^1H NMR spectrum (400 MHz, MeOD) of compound 4; Figure S13: ^{13}C NMR spectrum (100 MHz, MeOD) of compound 4; Figure S14: H–H COSY spectrum of

compound 4; Figure S15: HMBC spectrum of compound 4; Figure S16: HSQC spectrum of compound 4; Figure S17: ESI-MS of compound 2; Figure S18: ESI-MS of compound 3; Figure S19: ESI-MS of compound 4; Figure S20: ESI-MS of compound 5; and Figure S21: ESI-MS of compound 6. (PDF)

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Author Contributions

All authors have made a substantial contribution to this work: conceptualization, R.M. and Y.S.; methodology, R.M.; software, R.M.; validation, Y.S., S.Z., and J.H.; formal analysis, L.C.; investigation, L.Z.; resources, Z.D.; data curation, Y.S.; writing—original draft preparation, R.M.; writing—review and editing, R.M.; visualization, Y.S.; supervision, S.Z. and J.G.;

project administration, L.C.; funding acquisition, L.C. and Z.D. All authors have read and agreed to the published version of the manuscript.

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Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

UPLC, ultra-performance liquid chromatography; MS, mass spectrometry; PKS, polyketone synthases; HPLC, high-performance liquid chromatography; NMR, nuclear magnetic resonance; ABT, 1-aminobenzotriazole; PBO, piperonyl butoxide; Mal-CoA, methylmalonyl-coenzyme A

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