



Published in final edited form as:

J Am Chem Soc. 2019 December 26; 141(51): 19988–19993. doi:10.1021/jacs.9b10939.

NMR chemical exchange measurements reveal that *N*⁶-methyladenosine slows RNA annealing

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Abstract

*N*⁶-methyladenosine (*m*⁶A) is an abundant epitranscriptomic modification that plays important roles in many aspects of RNA metabolism. While *m*⁶A is thought to mainly function by recruiting reader proteins to specific RNA sites, the modification can also reshape RNA-protein and RNA-RNA interactions by altering RNA structure mainly by destabilizing base pairing. Little is known about how *m*⁶A and other epitranscriptomic modifications might affect the kinetic rates of RNA folding and other conformational transitions that are also important for cellular activity. Here, we used NMR $R_{1\rho}$ relaxation dispersion and chemical exchange saturation transfer to non-invasively and site-specifically measure nucleic acid hybridization kinetics. The methodology was validated on two DNA duplexes and then applied to examine how a single *m*⁶A alters the hybridization kinetics in two RNA duplexes. The results show that *m*⁶A minimally impacts the rate constant for duplex dissociation, changing k_{off} by ~1-fold but significantly slows the rate of duplex annealing, decreasing k_{on} by ~7-fold. A reduction in the annealing rate was observed robustly for two different sequence contexts at different temperatures, both in the presence and absence of Mg²⁺. We propose that rotation of the *N*⁶-methyl group from the preferred *syn* conformation in the unpaired nucleotide to the energetically disfavored *anti* conformation required for Watson-Crick pairing is responsible for the reduced annealing rate. The results help explain why in mRNA, *m*⁶A slows down tRNA selection, and more generally suggest that *m*⁶A may exert cellular functions by reshaping the kinetics of RNA conformational transitions.

Graphical abstract

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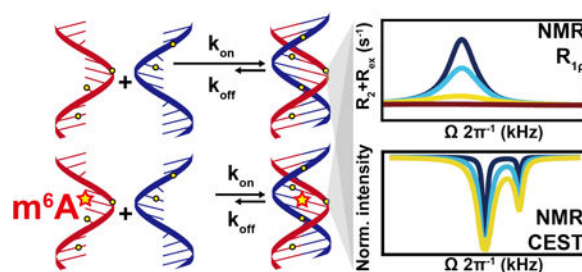
ASSOCIATED CONTENT

Supporting Information.

The Supporting Information is available free of charge on the ACS Publications website.

Details of sample preparation, NMR experiments, NMR $R_{1\rho}$ RD and CEST profiles (PDF)

The authors declare the following competing financial interest(s): H.M.A. is an advisor to and holds an ownership interest in Nymirum, an RNA-based drug discovery company.



N^6 -methyladenosine (m^6A) is an abundant reversible epitranscriptomic modification found in coding and noncoding RNAs^{1–4}. It plays important roles in RNA metabolism^{5–8} and is implicated in a growing number of cellular processes^{9–15}. While the modification is thought to primarily exert its function by recruiting reader proteins to specific RNA sites, it can also reshape RNA-RNA and RNA-protein interactions by modulating RNA structure^{16–20}. A single m^6A destabilizes RNA duplexes by 0.5–1.7 kcal/mol^{21–22}, enhancing binding to single-stranded RNA (ssRNA) binding proteins¹⁶. m^6A destabilizes A-U base pairs (bps) because hydrogen bonding requires that the N^6 -methyl group adopts the energetically unfavorable *anti* conformation^{21–22} (Figure 1).

The activities of many RNAs also depends on the kinetic rates of folding, protein-RNA, RNA-RNA, and RNA-ligand association/dissociation and conformational transitions^{23–29}. Surprisingly little is known about how m^6A and other epitranscriptomic modifications impact these kinetic properties of RNA. Compelling evidence for such a kinetic effect comes from a study showing that in mRNA, m^6A slows down tRNA selection during translation²⁰. Here, we developed an approach based on NMR spin relaxation dispersion (RD) in the rotating frame ($R_{1\rho}$)^{30–32} and Chemical Exchange Saturation Transfer (CEST)^{33–34} to site-specifically and non-invasively measure hybridization kinetics of nucleic acid duplexes and then used the approach to examine how a single m^6A impacts RNA duplex hybridization kinetics.

The melting and annealing of RNAs occurs in a wide variety of biochemical reactions^{27, 35}. Relative to other methods for studying hybridization kinetics^{36–45}, the NMR approach does not require a potentially perturbing label, which could obscure the impact of a small chemical modification, and kinetics can be measured at atomic resolution^{32, 46} to enable characterization of any intermediates that may form at the modified site.

We first evaluated the $R_{1\rho}$ RD methodology on DNA duplexes whose hybridization kinetics has been extensively characterized previously^{38, 41, 44–45, 47–51}. $R_{1\rho}$ RD relies on measuring the exchange contribution (R_{ex}) to transverse spin relaxation (R_2) due to chemical exchange between a major ground-state (GS) and a low-abundance and short-lived ‘excited-state’ (ES)^{52–53}.

Prior $R_{1\rho}$ studies on RNA and DNA duplexes were carried out at temperatures below the melting temperature (T_m)^{46, 54–56}. Under these conditions, the population (p_{ss}) of the single-stranded (ss) species falls below detection ($< 0.1\%$)³¹, enabling studies of bp dynamics. For example, at $T=25^\circ\text{C}$, the $R_{1\rho}$ profiles measured for various sites in the A_6 -DNA duplex^{55, 57}

($T_m \sim 51^\circ\text{C}$ and $[A_6\text{-DNA}] \sim 0.9\text{ mM}$) reflect exchange between a major Watson-Crick GS and minor Hoogsteen ES⁵⁵ (Figure 2A, 2B, S1). There is no evidence for a transient ss species, which is estimated to have a $p_{ss} \sim 0.1\%$ based on UV melting experiments (Table S1).

Based on simulations⁵⁰, increasing the temperature so that $p_{ss} > 1.0\%$ should bring hybridization kinetics within $R_{1\rho}$ detection (Figure S2). Indeed, the $R_{1\rho}$ profiles for $A_6\text{-DNA}$ changed when increasing the temperature to $T=45^\circ\text{C}$ ($p_{ss} \sim 10\%$). RD is now apparent at A16(C2) and T9(C1'), which are otherwise flat at $T=25^\circ\text{C}$ (Figure 2B). A single peak was observed in all cases consistent with two-state exchange ($\text{GS} \rightleftharpoons \text{ES}$). Fitting the $R_{1\rho}$ data to a 2-state exchange model yielded very similar $k_1 = k_{\text{off}}$ (differences < 2 -fold; k_{off} is the rate constant for dissociation) for different sites as expected for concerted melting and annealing of the duplex (Figure 2C). This is in stark contrast to Hoogsteen exchange at $T=25^\circ\text{C}$, in which k_1 varies 50-fold across sites reflecting sequence-specific differences in bp dynamics⁵⁹. The ES chemical shifts measured for various sites were also in excellent agreement with those measured for the isolated ss, confirming that the ES is the ss species (Figure 2D, S3).

In the 'zip-up' model^{48, 60}, DNA annealing proceeds through a slow nucleation step followed by a fast zipping step occurring on the ns- μs timescale which is too fast for RD detection. Since the Hoogsteen exchange at higher temperatures is likely too fast for RD detection, 'all-or-nothing' behavior is observed with strands either being fully annealed or fully unzipped. These results establish the utility of $R_{1\rho}$ RD to measure hybridization kinetics in DNA duplexes with site-specific resolution.

The backward rate constant $k_{-1} = k_{\text{on}} \times [\text{ss}]$ (k_{on} is the rate constant for duplex annealing) was ill-defined when fitting the $R_{1\rho}$ RD data (Figure S4). Such a degeneracy is expected when the exchange is slow on the NMR timescale and when using spin lock powers (ω_1) in the $R_{1\rho}$ experiment that exceed the exchange rate ($k_{\text{ex}} = k_1 + k_{-1}$)⁶¹⁻⁶². Indeed, in the slow exchange limit, the line broadening of the GS resonance only depends on the forward rate. To address this degeneracy, we used CEST experiments which can employ much lower spin locking fields more suitable for characterizing systems in slow exchange³³⁻³⁴. CEST relies on measuring the resonance intensity of the GS as a function of the power and offset of an applied weak radio frequency (rf) field. At $T=45^\circ\text{C}$, the CEST profiles for $A_6\text{-DNA}$ revealed a dip at the chemical shift of the ss ES (Figure 3A, 3C, S5). Fitting the CEST profiles allowed the reliable determination of all exchange parameters including k_{on} (Figure S4), resulting in values (Figure 3B) that are in good agreement with those previously reported values for similar DNA duplexes^{45, 50}.

Fixing p_{ss} to the CEST determined value, the $R_{1\rho}$ RD profiles could be satisfactorily globally fitted (Figure S5), yielding exchange parameters ($k_1 = k_{\text{off}}$, $k_{-1} = k_{\text{on}} \times [\text{ss}]$ and $\omega_{\text{ES-GS}}$) that are in excellent agreement with the CEST derived values (Figure 3B, Table S3, S4). This mutual consistency further supports the validity of the approach. Finally, we further evaluated the CEST methodology by comparing the hybridization kinetics of $A_6\text{-DNA}$ with another $A_2\text{-DNA}$ duplex, which has higher stability ($T_m \sim 60^\circ\text{C}$ and $[A_2\text{-DNA}] \sim$

0.8 mM) (Figure 3C, 3D). Consistent with prior studies^{47–49}, the two duplexes have similar k_{on} values but k_{off} is 20-fold faster for the less stable A₆-DNA duplex (Figure 3E).

Next, we applied the methodology to examine how m⁶A impacts hybridization kinetics in an RNA duplex containing the most abundant m⁶A consensus sequence (GGACU) in eukaryotic mRNA^{1–2} ($T_{\text{m}} \sim 80^{\circ}\text{C}$ and [dsGGACU] ~ 0.7 mM with Mg²⁺). In canonical RNA duplexes, there are no contributions from Hoogsteen exchange or any other process as verified for Watson-Crick bps in a variety of sequence and structural contexts⁵⁴. However, since m⁶A could induce local melting of the duplex, it was important to carry out measurements on the m⁶A residue itself. To this end, two dsGGACU duplexes were chemically synthesized containing ¹³C₂/C₈ labeled m⁶A or A near the center of the duplex (Figure 4A, S1, S6) (see methods). m⁶A destabilized the dsGGACU duplex by ~ 1 kcal/mol (Table S1), consistent with prior studies^{21–22}.

The CEST and $R_{1\rho}$ profiles for both unmodified and modified dsGGACU duplex at $T=65^{\circ}\text{C}$ revealed a single peak/dip consistent with 2-state exchange (Figure 4B, S7). However, the profiles for the modified duplex differed markedly from its unmodified counterpart (Figure 4B, S7). In both cases, global fitting of the CEST and $R_{1\rho}$ data yielded ES chemical shifts that are in excellent agreement with those measured for the isolated ss (Figure 4C, S7, S8). Fitting the CEST data revealed that m⁶A changes k_{off} by 0.7–1.7 fold but decreases k_{on} by 4–9 fold (Figure 4D, S7). This m⁶A induced slowdown of annealing was observed robustly with or without Mg²⁺ (Figure 4D, S7), for a different sequence derived from Hepatitis C virus (HCV)¹⁵ ($T_{\text{m}} \sim 76^{\circ}\text{C}$ and [dsHCV] ~ 0.7 mM with Mg²⁺) (Figure S1, S7), at a higher concentration of monovalent ions (Figure S7), and when using the $R_{1\rho}$ RD data (Figure S7).

When unpaired, the N⁶-methyl group favors the *syn* conformation, while the *anti* conformation required for Watson-Crick pairing and duplex annealing is unfavorable with an estimated population of $\sim 5\%$ ⁶³. Rotation of the N⁶-methyl group is likely responsible for the reduced annealing rate. Mismatches have also been shown to reduce k_{on} by up to 50-fold^{27, 64} through mechanisms that are not fully understood. Further studies are needed to dissect the kinetic mechanism by which m⁶A slows the annealing rate and how this varies with position and sequence context⁶⁴.

In conclusion, we have described an NMR strategy for site-specifically resolving duplex hybridization kinetics. The ease and throughput of these experiments can be improved in the future by using longitudinal optimized ¹H-CEST experiments⁶⁵ as well as other approaches for optimal data collection^{34, 66}. The approach can also be applied to mismatch containing duplexes ideally by targeting remote sites that are not involved in any local mismatch dynamics and to use multi-site exchange models as needed to fit data⁵⁶. Our results show that in the middle of a duplex, m⁶A minimally affects the melting rate but substantially decreases the rate of annealing. This may help explain why tRNA selection during translation is slower for mRNAs containing m⁶A²⁰. m⁶A is also found in the seed sequence of microRNAs and in their mRNA target sites⁶⁷ and mismatches that slowdown microRNA:mRNA annealing have substantial effects on gene expression⁶⁴. Thus, m⁶A could similarly affect gene expression by altering the kinetics of annealing. m⁶A may also affect the kinetics of RNA-protein and RNA-ligand association and also reshape co-

transcriptional RNA folding pathways^{68–71} by prolonging the lifetime of the unpaired conformation^{25, 72–73} perhaps in a manner analogous to cis-trans proline isomerization in proteins^{74–75}.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

ACKNOWLEDGMENT

We thank Nicole Orlovsky for critical comments on the manuscript and Prof Lewis Kay (University of Toronto) for helpful discussions. We acknowledge the technical support and resources from the Duke Magnetic Resonance Spectroscopy Center. We thank Dr. Richard Brennan for use of the UV-Vis spectrophotometer.

Funding Sources

This work was supported by US National Institute for General Medical Sciences (1R01GM132899 to H.M.A.), the Austrian Science Fund (FWF, project P28725 and P30370 to C.K.), and the Austrian Research Promotion Agency FFG (West Austrian BioNMR, 858017 to C.K.)

The research reported in this article was performed by the Duke University faculty and students and was funded by a U.S. National Institutes of Health contract to H.M.A.

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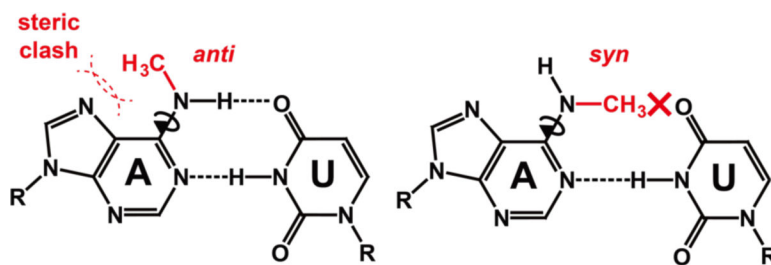


Figure 1. *N*⁶-methyl adenosine (m⁶A) destabilizes m⁶A-U pairing and RNA duplexes. The methyl group has to adopt an *anti* conformation to form the Watson-Crick H6--O4 hydrogen bond but this leads to unfavorable steric contacts with N7.

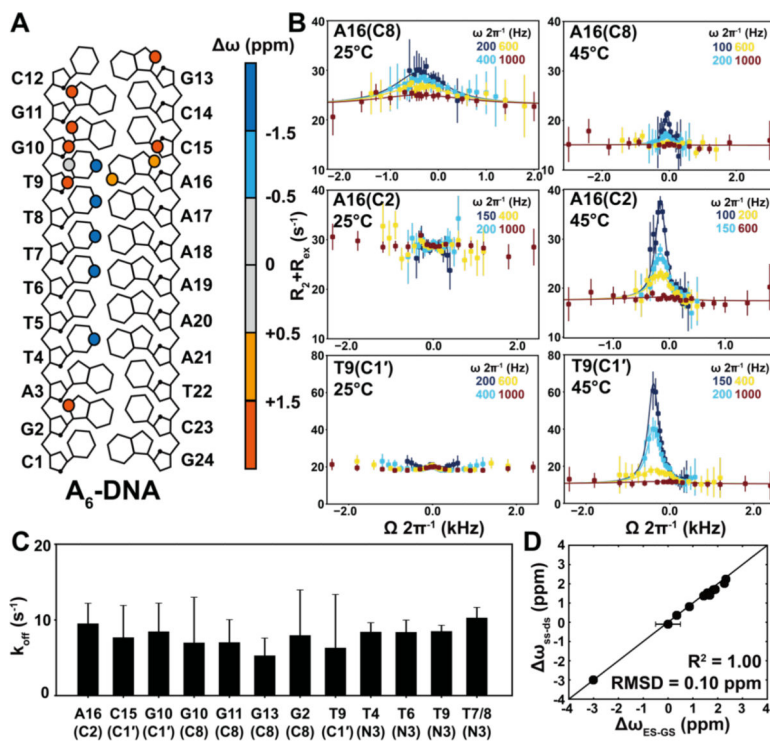


Figure 2. Site-specific characterization of A₆-DNA hybridization kinetics using NMR $R_{1\rho}$ RD. (A) The A₆-DNA duplex. $\omega = \omega_{ES} - \omega_{GS}$ obtained from global fitting of the $R_{1\rho}$ RD profiles is color-coded on each atom. Sites which are not colored indicates that no measurements were done. (B) Off-resonance $R_{1\rho}$ (¹³C) RD profiles measured in A₆-DNA at 25°C (left) and 45°C (right). T9(C1') RD at 25°C were reprinted by permission from⁵⁸. Buffer conditions were 25 mM NaCl, 15 mM sodium phosphate, 0.1 mM EDTA and 10% D₂O at pH 6.8. (C) The site-specific k_{off} values obtained from 2-state fitting of the $R_{1\rho}$ RD profiles measured for A₆-DNA at 45°C. (D) Comparison of $\omega_{ES-GS} = \omega_{ES} - \omega_{GS}$ measured by RD with $\omega_{ss-ds} = \omega_{ss} - \omega_{ds}$ values obtained from the major and minor resonance observed in 2D [¹³C, ¹H], [¹⁵N, ¹H] and [¹⁵N, ¹³C] HSQC spectra of A₆-DNA at 45°C.

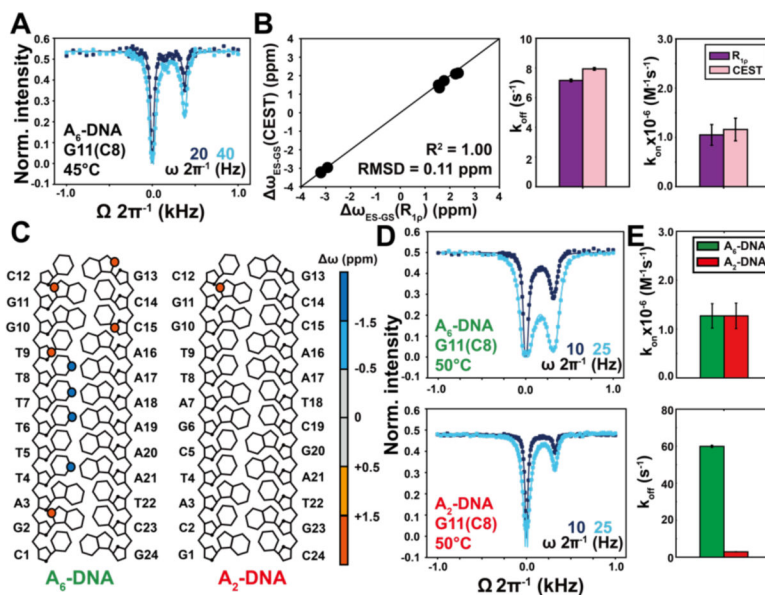


Figure 3. Site-specific characterization of hybridization kinetics using CEST.

(A) ^{13}C CEST profile for G11(C8) measured in A_6 -DNA at 45°C . (B) Comparison of $\omega_{\text{ES-GS}}$, k_{off} and k_{on} values obtained from $R_{1\rho}$ and CEST (fits of the $R_{1\rho}$ profiles were performed fixing p_{ss} to the value measured using CEST). Buffer conditions were 25 mM NaCl, 15 mM sodium phosphate, 0.1 mM EDTA and 10% D_2O at pH 6.8. (C) The sequence of A_2 -DNA and A_6 -DNA. $\omega = \omega_{\text{ES}} - \omega_{\text{GS}}$ obtained from CEST fitting is color-coded on each atom. (D) ^{13}C CEST profiles for G11(C8) measured in A_2 -DNA and A_6 -DNA at 50°C . (E) Comparison of k_{on} and k_{off} values measured for A_2 -DNA (red) and A_6 -DNA (green).

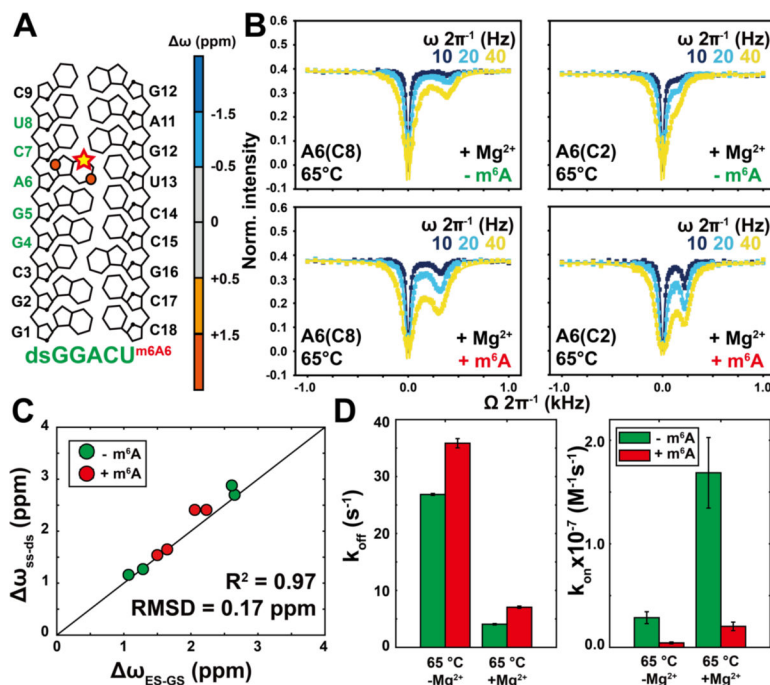


Figure 4. Measuring the impact of m⁶A on dsGGACU hybridization kinetics using CEST. **(A)** The dsGGACU sequence. $\omega = \omega_{\text{ES}} - \omega_{\text{GS}}$ obtained from global fitting of CEST is color-coded on each atom. **(B)** ^{13}C CEST profiles measured for A6 in unmodified (left, green) and m⁶A modified (right, red) dsGGACU at 65°C in the presence of 3 mM Mg²⁺ (profiles in the absence of Mg²⁺ are shown in Figure S7). Buffer conditions were 25 mM NaCl, 15 mM sodium phosphate, 3 mM Mg²⁺, 0.1 mM EDTA and 10% D₂O at pH 6.8. **(C)** Comparison of $\omega_{\text{ES-GS}} = \omega_{\text{ES}} - \omega_{\text{GS}}$ measured by CEST with $\omega_{\text{ss-ds}} = \omega_{\text{ss}} - \omega_{\text{ds}}$ values obtained from the major and minor resonance observed in 2D [^{13}C , ^1H] HSQC spectra of dsGGACU with (red) and without (green) m⁶A at 65°C. **(D)** Comparison of k_{on} and k_{off} measured for unmodified (green) and m⁶A modified (red) dsGGACU.