



Moisture Content of Bacterial Cells Determines Thermal Resistance of *Salmonella enterica* Serotype Enteritidis PT 30

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ABSTRACT *Salmonella* spp. are resilient bacterial pathogens in low-moisture foods. There has been a general lack of understanding of critical factors contributing to the enhanced thermal tolerance of *Salmonella* spp. in dry environments. In this study, we hypothesized that the moisture content (X_w) of bacterial cells is a critical intrinsic factor influencing the resistance of *Salmonella* spp. to thermal inactivation. We selected *Salmonella enterica* serotype Enteritidis PT 30 to test this hypothesis. We first produced viable freeze-dried *S. Enteritidis* PT 30, conditioned the bacterial cells to different X_w s (7.7, 9.2, 12.4, and 15.7 g water/100 g dry solids), and determined the thermal inactivation kinetics of those cells at 80°C. The results show that the D -value (the time required to achieve a 1-log reduction) decreased exponentially with increasing X_w . We further measured the water activities (a_w) of the freeze-dried *S. Enteritidis* PT 30 as influenced by temperatures between 20 and 80°C. By using those data, we estimated the X_w of *S. Enteritidis* PT 30 from the published papers that related the D -values of the same bacterial strain at 80°C with the a_w of five different food and silicon dioxide matrices. We discovered that the logarithmic D -values of *S. Enteritidis* PT 30 in all those matrices also decreased linearly with increasing X_w of the bacterial cells. The findings suggest that the amount of moisture in *S. Enteritidis* PT 30 is a determining factor of its ability to resist thermal inactivation. Our results may help future research into fundamental mechanisms for thermal inactivation of bacterial pathogens in dry environments.

IMPORTANCE This study established a logarithmic relationship between the thermal death time (D -value) of *S. Enteritidis* PT 30 and the moisture content (X_w) of the bacterial cells by conducting thermal inactivation tests on freeze-dried *S. Enteritidis* PT 30. We further verified this relationship using literature data for *S. Enteritidis* PT 30 in five low-moisture matrices. The findings suggest that the X_w of *S. Enteritidis* PT 30, which is rapidly adjusted by microenvironmental a_w , or relative humidity, during heat treatments, is the key intrinsic factor determining the thermal resistance of the bacterium. The quantitative relationships reported in this study may help guide future designs of industrial thermal processes for the control of *S. Enteritidis* PT 30 or other *Salmonella* strains in low-moisture foods. Our findings highlight a need for further fundamental investigation into the role of water in protein denaturation and the accumulation of compatible solutes during thermal inactivation of bacterial pathogens in dry environments.

KEYWORDS moisture content, bacterial cells, thermal resistance, *Salmonella enterica* serotype Enteritidis PT 30, low-moisture foods

Free water in foods is required to support the growth of yeasts, molds, and bacteria (1). Water activity (a_w) is an indirect measure of the free water available to the

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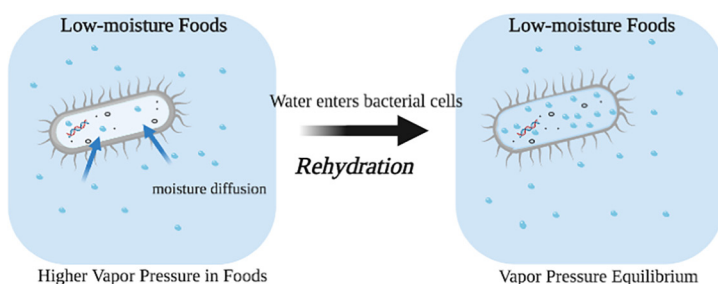
microorganism in food systems (2). Generally, low-moisture foods have an a_w at or below 0.6 at room temperature (3). The low- a_w environment in low-moisture foods, such as milk powder, chocolate, peanut butter, and cereal, inhibits the growth of microorganisms (4, 5). However, pathogens such as *Salmonella* can survive in low-moisture foods during storage and frequently cause worldwide outbreaks (1, 3, 6–8).

Salmonella bacteria are among the most common pathogens causing severe foodborne illness for humans (7, 8). These resilient microorganisms can easily adapt to environmental conditions beyond their optimal growth range, including temperature, pH and a_w (1, 5, 9). The minimum a_w for the growth of *Salmonella* is 0.94, but it can survive and remain viable in low-moisture foods for long periods (4). In particular, *Salmonella* has enhanced thermal resistance under dry conditions. Studies have shown that its thermal resistance, as evaluated by the D -value (the time required to inactivate 90% of target bacteria at a fixed temperature), increases sharply with decreasing a_w in various low-moisture foods (3, 5, 10–13).

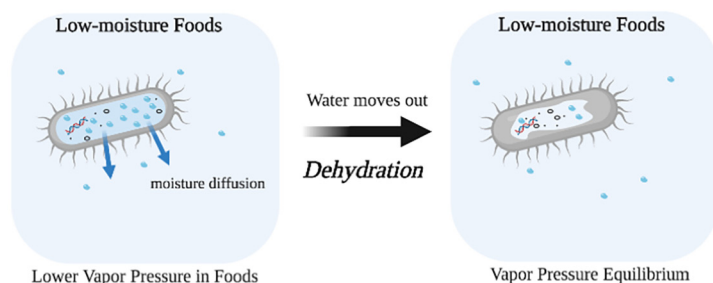
Early researchers attempted to establish a direct correlation between the thermal resistance of foodborne pathogens at certain treatment temperatures and the a_w of the inoculated foods at room temperature (10). Recent studies, however, have shown that the a_w of food systems at room temperature does not accurately reflect the microenvironment to which foodborne pathogens are exposed in a thermal treatment. For instance, when wheat flour, whey protein, and almond flour are conditioned to an a_w of 0.45 at room temperature and then heated to 80°C in sealed containers (without moisture loss), their a_w values changed to 0.68, 0.62, and 0.54, respectively (11). The D -values at 80°C ($D_{80^\circ\text{C}}$ -values) of *Salmonella enterica* serotype Enteritidis PT 30 in these foods were about 5, 11, and 21 min, respectively (11). Thus, the a_w of foods measured at room temperature should not be considered a process control parameter in the design of thermal inactivation processes. Our most recent studies have demonstrated that the a_w of food systems measured at treatment temperatures ($a_{w, \text{treatment temperature}}$) should be considered a critical extrinsic factor determining the D -values of microorganisms during thermal treatment (11, 13–16).

The a_w of a biomass is expressed as the ratio of its water vapor pressure to the saturated water pressure at a given temperature. It is a thermodynamic parameter that generally changes with temperature in food systems (11, 15, 17–19). Thus, it is an extrinsic property of a biomaterial. In a high-moisture environment, water accounts for 60% to 90% of the mass of bacterial cells (20). In low-moisture foods, however, bacterial cells respond to the a_w of the food matrices in multiple ways, and a direct reaction is the adjustment of the moisture content (X_w , expressed as grams of water per 100 g of dry solids) of bacterial cells through moisture diffusion (17, 21). The moisture diffusion between bacterial cells and their surrounding food matrices in a thermal process is illustrated in Fig. 1. Even though the biomass of bacterial cells and the contaminated food have the same a_w at room temperature, their a_w may change when heated to a high temperature. The differences in a_w between bacterial cells and the food creates a moisture vapor pressure gradient that drives moisture diffusion in or out of the bacterial cells until an equilibrium is reached. In industrial operations, the food matrices have much larger masses than the bacteria that potentially contaminate the foods. Thus, the a_w of the bacterial cells should be equal to that of the foods at equilibrium. As shown in Fig. 1A, bacterial cells will gain moisture from their microenvironment when the a_w of the food is higher than that of bacterial cells at thermal-treatment temperatures. On other hand, the bacterial cells would be dehydrated if a lower a_w is provided by a food system (Fig. 1B). The moisture diffusion between bacterial cells and their food environment can be rapid (within seconds) because of the smaller characteristic size of bacterial cells (17). Therefore, we hypothesize that the X_w of bacteria might be the dominant intrinsic parameter that determines their thermal resistance in low-moisture foods in a thermal process.

The relationship between the X_w and a_w of a biomass at different temperatures is commonly referred to as the moisture sorption isotherms. Extensive data on the



(A) Higher a_w in low-moisture foods than bacterial cells at thermal-treatment temperature



(B) Lower a_w in low-moisture foods than bacterial cells at thermal-treatment temperature

FIG 1 Illustration of moisture exchange between *Salmonella* cells and different microenvironments during heating (created with BioRender).

moisture sorption isotherms for foods exist in the literature. But no research had been published on the isotherms for bacteria until the work of Syamaladevi et al. (17), who reported the moisture sorption isotherms for a biomass of freeze-dried *Enterococcus faecium*, a commonly used surrogate for *Salmonella*. There has been no published research on a_w changes of the biomass of *Salmonella* cells with temperature.

The goal of this research was to test the hypothesis that the moisture content of bacterial cells is the intrinsic parameter determining the thermal resistance of *Salmonella* in low-moisture environments. The specific objectives of this study were to (i) determine the thermal resistance of freeze-dried *S. Enteritidis* PT 30 at 80°C with different X_w values (7.7, 9.2, 12.4, and 15.7 g water/100 g dry solids), (ii) establish the relationship between the X_w and the a_w of freeze-dried *S. Enteritidis* PT 30 biomass at different temperatures, and (iii) study the correlation between X_w and the thermal resistance of *S. Enteritidis* PT 30 in low-moisture matrices.

We selected *S. Enteritidis* PT 30 as the target bacterium in this study because its thermal resistance in different low-moisture matrices has been reported (10, 11, 22–24). We used freeze-drying, also known as lyophilization, to produce desiccated *S. Enteritidis* PT 30 samples with high vitality (22). The X_w of a porous mass of freeze-dried *S. Enteritidis* PT 30 could easily be adjusted by exposing it to different relative humidities. We conducted thermal inactivation tests on freeze-dried *S. Enteritidis* PT 30 at different values of X_w using improved thermal-death-time (TDT II) cells and investigated the relationship between $D_{80^\circ\text{C}}$ values and X_w . We also measured temperature-dependent changes in the a_w of freeze-dried *S. Enteritidis* PT 30 cells at different X_w values in order to establish the relationships between the X_w and a_w of *S. Enteritidis* PT 30 biomass at different temperatures (isotherms). Finally, we estimated the X_w of *S. Enteritidis* PT 30 in five low- a_w systems reported in the literature, generated the linear regression for the logarithmic $D_{80^\circ\text{C}}$ values against the X_w in these matrices, and compared it to that of freeze-dried *S. Enteritidis* PT 30.

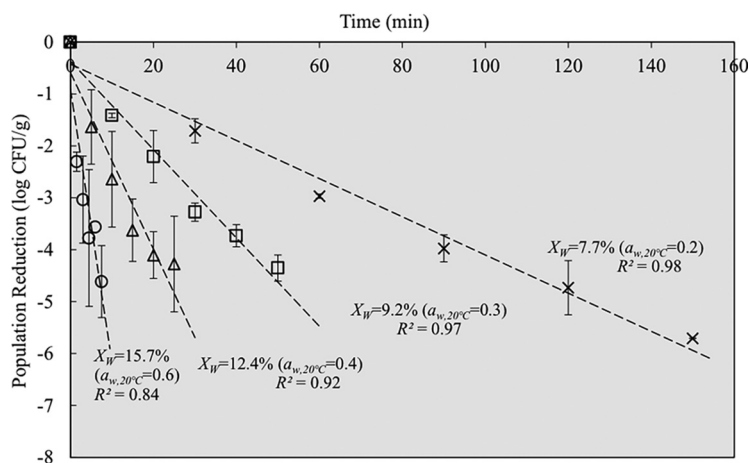


FIG 2 Representative thermal inactivation curves of freeze-dried *S. Enteritidis* PT 30 at 80°C. X_w , moisture content, expressed in grams of water per 100 g dry solids; $a_{w,20^\circ\text{C}}$, equivalent target water activity at room temperature after relative-humidity-controlled equilibration.

RESULTS AND DISCUSSION

Thermal inactivation kinetics of freeze-dried *S. Enteritidis* PT 30. The population of the concentrated *S. Enteritidis* PT 30 inoculum was ~ 12 log CFU/ml. The initial viability of freeze-dried *S. Enteritidis* PT 30 was 10.72 ± 0.34 log CFU/g, and a < 2 log reduction in population was caused by freeze-drying. The samples were pre-equilibrated under different relative humidities or equivalent a_w values at room temperature ($\sim 21^\circ\text{C}$). In the thermal inactivation tests, the come-up time (CUT, expressed in minutes) for the core temperature of freeze-dried *S. Enteritidis* PT 30 samples to reach 79.5°C was 1.5 min when samples were heated to 80°C in TDT II test cells. The representative thermal inactivation curves for freeze-dried *S. Enteritidis* PT 30 with different X_w values (7.7, 9.2, 12.4, and 15.7 g water/100 g dry solids) are shown in Fig. 2. The population reduction followed the typical first-order linear regression relationship with the treatment time (equation 1), with an R^2 value between 0.84 and 0.98.

$$\log\left(\frac{N}{N_0}\right) = -\frac{t}{D} \quad (1)$$

where t is the thermal treatment time (in minutes), N is the bacterial population (in CFU per gram) at time t , N_0 is the initial bacterial population at the come-up time (in CFU per gram), and D is the time (in minutes) required to inactivate the microbial population by 90% at a given temperature.

More-rapid microbial reductions [$\log\left(\frac{N}{N_0}\right)$] were observed at higher X_w values, as shown in Fig. 2, a finding consistent with the thermal inactivation trends of *Salmonella* and *Enterococcus faecium* in several low- a_w systems (10, 11, 14, 25, 26).

Changes in D -value with X_w of bacterial cells. Microbial survival from thermal inactivation was analyzed using the log-linear model (equation 1) and the Weibull model:

$$\log\left(\frac{N}{N_0}\right) = -\left(\frac{t}{\alpha}\right)^\beta \quad (2)$$

where α is the scale parameter (in minutes) and β is the shape parameter. The Weibull model showed a better goodness of fit than the log-linear model. Given that the Weibull model is performed by two parameters during analysis, and the purpose of this study was to investigate the relationship associated with D -values, thus, the log-linear model was used for subsequent analyses and comparison. The $D_{80^\circ\text{C}}$ -values of freeze-dried *S. Enteritidis* PT 30, obtained from the slope of the trend lines in Fig. 2,

TABLE 1 $D_{80^{\circ}\text{C}}$ -values of freeze-dried *S. Enteritidis* PT 30 at different moisture contents^a

$a_{w,20^{\circ}\text{C}} \pm 0.02$	Measured X_w (g water/100 g dry solids)		Linear model		Weibull model		
	$a_{w,80^{\circ}\text{C}} \pm 0.02$		$D_{80^{\circ}\text{C}}$ (min)	R^2	α (min)	β	R^2
0.20	7.7 ± 0.5	0.36	27.7 ± 0.7	0.98	13.8 ± 0.0	0.73 ± 0.11	0.99
0.30	9.2 ± 0.0	0.49	11.9 ± 0.2	0.97	6.1 ± 2.0	0.69 ± 0.10	0.96
0.40	12.4 ± 0.3	0.60	6.5 ± 1.1	0.92	1.9 ± 1.2	0.57 ± 0.06	0.98
0.60	15.7 ± 1.2	0.78	1.9 ± 0.6	0.84	0.4 ± 0.4	0.48 ± 0.07	0.95

^aValues are means \pm standard deviations. The values of $a_{w,80^{\circ}\text{C}}$ were predicted using the CCE (equation 3). R^2 , coefficient of determination; higher values indicate a better fit of the model.

were 27.7 ± 0.7 , 11.9 ± 0.2 , 6.5 ± 1.1 , and 1.9 ± 0.6 min at X_w values of 7.7, 9.2, 12.4, and 15.7 g water/100 g dry solids, respectively (Table 1). The $D_{80^{\circ}\text{C}}$ -values as influenced by bacterial cell X_w are shown in Fig. 3. In general, the $D_{80^{\circ}\text{C}}$ -values decreased exponentially with increasing X_w of bacterial cells. This suggests that the X_w of bacterial cells is a critical factor that determines the thermal resistance of *Salmonella* in thermal treatments.

Moisture sorption isotherms of freeze-dried *S. Enteritidis* PT 30. The temperature-dependent changes in the a_w of freeze-dried *S. Enteritidis* PT 30 with different X_w values as measured by high-temperature cells were analyzed by the Clausius-Clapeyron equation (CCE) according to the method of Tadapaneni et al. (19), as follows:

$$a_{w2} = a_{w1} \exp \left[\frac{q_{st}}{R} \left(\frac{1}{T_1} - \frac{1}{T_2} \right) \right] \quad (3)$$

where a_{w1} and a_{w2} are the water activity values of a sample with the same X_w at temperatures T_1 and T_2 (in kelvins), respectively; R is the universal gas constant (8.314 joules per mole kelvin); and q_{st} is the net isosteric heat of sorption (in joules per mole), which can be determined from the slope of plotted data ($\ln a_w$ versus $1/T$). These changes in a_w are presented in Fig. 4. Generally, the a_w of freeze-dried *S. Enteritidis* PT 30 increased linearly with increasing temperature at a specific X_w . This trend is similar to that of high-protein and high-starch food systems, including corn starch, soy protein, and wheat flour (12, 18). However, it dramatically differs from that of high-oil and sugar-rich foods. Specifically, the a_w values of peanut oil, coconut milk powder, and almond flour do not increase, and may even decrease, with increasing temperature (14, 16, 18).

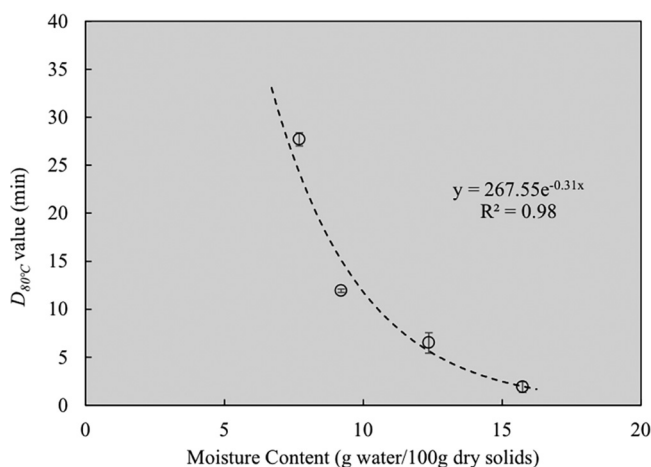


FIG 3 $D_{80^{\circ}\text{C}}$ -values of freeze-dried *S. Enteritidis* PT 30 at different moisture contents (X_w expressed in grams of water per 100 g dry solids). An exponential trend was observed in $D_{80^{\circ}\text{C}}$ -values as a function of X_w .

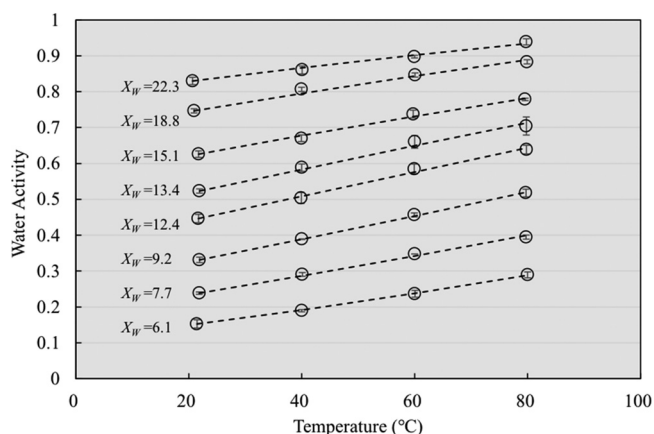


FIG 4 Temperature-dependent changes in the water activity of freeze-dried *S. Enteritidis* PT 30 at different moisture contents (X_w , expressed in grams of water per 100 g dry solids) ($n=3$). Open circles represent the average a_w values as measured by high-temperature-cells, and dashed lines are derived from equation 3.

The effect of temperature on the a_w changes of freeze-dried *S. Enteritidis* PT 30 was more obvious in samples with a relatively lower X_w . For instance, at the X_w of 9.2 g water/100 g dry solids, the a_w of freeze-dried *S. Enteritidis* PT 30 at 20°C ($a_{w,20^\circ\text{C}}$) was 0.33. When freeze-dried *S. Enteritidis* PT 30 was heated to 80°C, its a_w ($a_{w,80^\circ\text{C}}$) was increased to 0.52. The a_w increased by 0.19. When the X_w of the bacterial cells was 22.3 g water/100 g dry solids, the a_w of freeze-dried *S. Enteritidis* PT 30 increased from 0.83 to 0.94 as the temperature was elevated from 20°C to 80°C; thus, a smaller increment in a_w (0.11) was obtained. Similar results have also been reported in multiple low-moisture food systems, including wheat flour, almond flour, nonfat milk powder, corn starch, soy protein, cheddar cheese powder, and coconut milk powder (18, 19).

The relationship between the X_w of freeze-dried *S. Enteritidis* PT 30 biomass and its a_w at different temperatures, commonly referred to as moisture sorption isotherms, was determined by the Guggenheim-Anderson-de Boer (GAB) model, as follows:

$$\frac{X_w}{X_m} = \frac{CKa_w}{(1-Ka_w)(1-Ka_w+CKa_w)} \quad (4)$$

where X_m is the monolayer moisture content (in grams of water per 100 g dry solids) in bacterial cells, a_w is the water activity value at X_w , C is a heat constant, ranging from 1 to 20, and K is a multilayer factor, ranging from 0.70 to 1. The results are shown in Fig. 5. The specific parameters for GAB equations (equation 4) at different temperatures in this study are shown in Table 2. The derived monolayer moisture content (X_m) of freeze-dried *S. Enteritidis* PT 30 ranged from 10.0 g water/100 g dry solids at 20°C to 7.5 g water/100 g dry solids at 80°C (Table 2). These values are much larger than the X_m of most protein-rich systems. For example, Perez-Reyes et al. recently reported X_m of 6 g water/100 g dry solids at 20°C and 5 g water/100 g dry solids at 80°C for egg white powders (84.3% protein, 0.6% fat, dry basis [d.b.]) and 4 g water/100 g dry solids at 20°C and 3 g water/100 g dry solids at 80°C for whole egg powder (46.4% protein, 55.1% fat, d. b.) (27). This suggests that the biomass of freeze-dried *S. Enteritidis* PT 30 had much larger water binding capacity than those protein-rich powders.

In general, the X_w of freeze-dried *S. Enteritidis* PT 30 increased with increasing a_w at any given temperature (Fig. 5). For example, the X_w of freeze-dried *S. Enteritidis* PT 30 was 9.2 g water/100 g dry solids when conditioned to an a_w of 0.3 at 20°C. The X_w increased to 15.0 g water/100 g dry solids when conditioned to an a_w of 0.6 at the same temperature. The a_w values of freeze-dried *S. Enteritidis* PT 30 at a fixed X_w increased significantly with heating from 20°C to 80°C, as shown in Fig. 5. The sorption isotherms of freeze-dried *S. Enteritidis* PT 30 resemble a part of the type II isotherm (S-shaped), exhibiting a sigmoid

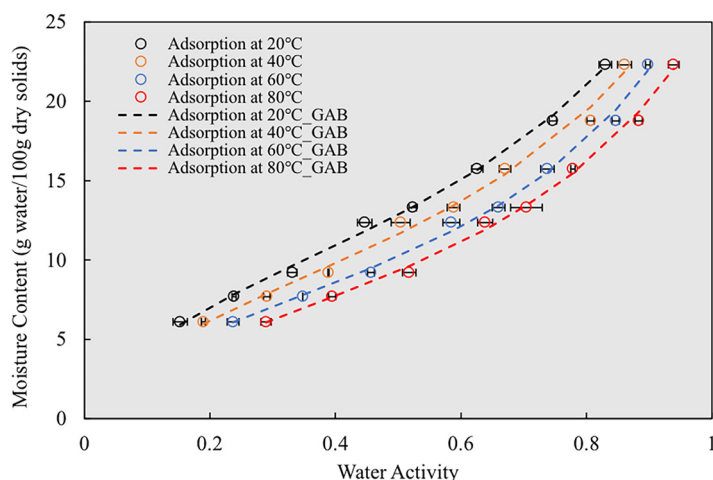


FIG 5 Moisture adsorption isotherms of freeze-dried *S. Enteritidis* PT 30 at different temperatures. The scattered data were generated using high-temperature cells, and the dashed lines represent the curves fitted by the GAB model (equation 4).

curve. The two slightly bending areas were observed at a_w values below 0.3 and above 0.7. However, J-shaped isotherms were reported for freeze-dried *E. faecium* cells (17). The difference might be attributed to the differences in cell wall structures and compositions between Gram-positive and Gram-negative bacteria. *Salmonella* bacteria are Gram negative; the cell wall consists of an outer membrane, a thin layer of peptidoglycan, and a cytoplasmic membrane. *E. faecium*, on the other hand, is a Gram-positive bacterium; its cell wall consists of a thicker layer of peptidoglycan and one cytoplasmic membrane (29).

High-protein-content or starch-rich food systems, such as corn starch, soy protein, and wheat flour, usually show the same type of isotherm curves as freeze-dried *S. Enteritidis* PT 30 (6, 18). But for high-sugar-content and oil-rich food systems, the moisture sorption isotherms generally show a type III isotherm (J-shaped), which is concave upward due to more moisture gain from the surface crystalline dissolution at a higher a_w (28). The a_w of peanut butter also increased sharply with an extremely small increment of X_w at room temperature, and this increase was more significant at higher treatment temperatures (6).

Net isosteric heat of sorption for freeze-dried *S. Enteritidis* PT 30. The net isosteric heat of sorption (q_{st}), defined as the total enthalpy change for sorption minus the specific latent heat of vaporization of liquid water (30), is a unique parameter for different biomasses. The value of q_{st} reflects the bond energy between water molecules and solid substances, which can be obtained from an empirical relation at a specific X_w (31, 32). In this study, q_{st} was related to the X_w (expressed in grams of water per 100 g dry solids) of bacterial cells through an exponential relation, as follows:

$$q_{st} = 17.852 \exp(-0.10 \times X_w) \quad (5)$$

where the goodness of fit (R^2) was >0.99 . A comparison of q_{st} for freeze-dried *S. Enteritidis* PT 30 and several representative low- a_w systems is shown in Fig. 6.

TABLE 2 Experimental values for the parameters in the GAB model (equation 4) at different treatment temperatures

Treatment temp (°C)	GAB model parameter		
	X_m (g water/100 g dry solids)	C	K
20	10.0	9.42	0.70
40	9.1	8.43	0.71
60	7.6	9.00	0.75
80	7.5	6.67	0.73

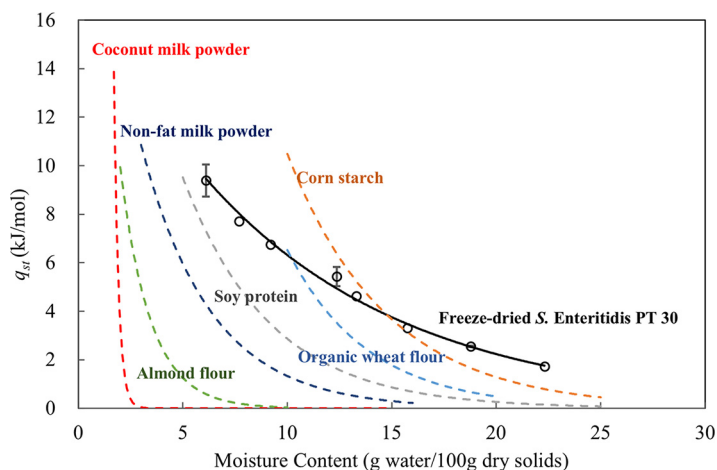


FIG 6 Net isosteric heat of sorption for freeze-dried *S. Enteritidis* PT 30 and several representative food systems. Open circles represent data calculated on the basis of measurements for freeze-dried *S. Enteritidis* PT 30. Both the solid curve and the dashed curves were generated by fitting the q_{st} equations. The dashed curves were generated using the data from Jin et al. (18) and Tadapaneni et al. (19).

As shown in Fig. 6, q_{st} correlates negatively with X_w . The higher q_{st} values obtained at lower X_w values indicate stronger bonds between water molecules and solid matrices (33). That is, more energy is required to break the bonds at a lower X_w . The relationship between the q_{st} and X_w of freeze-dried *S. Enteritidis* PT 30 differs sharply from that for the foods shown in Fig. 6. The q_{st} value of freeze-dried *S. Enteritidis* PT 30 is, in general, higher than that of the high-protein- and high-oil-content foods at X_w values between 6 and 10 g water/100 g dry solids and higher than that of all the selected foods when the X_w is above 15 g water/100 g dry solids. This observation is consistent with the high values of monolayer moisture contents (X_m) of freeze-dried *S. Enteritidis* PT 30 discussed above.

A sharp drop in q_{st} with increasing X_w was observed for all the foods for which data are shown in Fig. 6. The deepest drop was obtained for oil-rich foods, such as coconut milk powder (consisting of 63.8% fat, d.b. [18]) and almond flour (consisting of 50.7% fat, d.b. [11]), followed by intermediate foods and protein-rich foods (nonfat milk powder and soy protein), and finally high-carbohydrate foods, such as organic wheat flour (86.1% carbohydrate, d.b. [11]) and corn starch (98.0% carbohydrate, d.b. [18]). Lipids are hydrophobic. The hydrophobic interaction between lipids and water is weaker than hydrogen bonds, resulting in a decreased enthalpic demand to break the bonds between water molecules and the solids (34). Carbohydrates are hydrophilic macromolecules, and polysaccharides (such as starch) have strong affinity for water molecules because of multihydroxyl (-OH) groups (35). Combined interactions of hydrogen bonds and glycosidic bonds in carbohydrates are energetically favorable, leading to large amounts of energy required to break the bonds between water and the solids. Thus, high-carbohydrate foods, such as corn starch, have the highest q_{st} of all foods at a specific X_w . Proteins are polymers with complex structures, and multiple bonds are present, including peptide bonds (primary structure), hydrogen bonds (secondary and tertiary structure), and ionic bonds and disulfide bonds (tertiary structure) (36). Proteins have less affinity for water than polysaccharides, so the q_{st} of protein-rich foods is lower than that of carbohydrate-rich foods.

Bacterial cells consist of more complex chemical compounds, including proteins, RNA, phospholipids, and polysaccharides (37). The chemical bonds with water molecules in bacterial cells are also complex; the cells may have unique abilities to retain moisture compared to the food matrices included in Fig. 6. Although it is difficult to directly connect the $D_{80^\circ\text{C}}$ -values with the q_{st} of freeze-dried *S. Enteritidis* PT 30, it is

TABLE 3 Estimated moisture contents of *S. Enteritidis* PT 30 using the GAB model at 80°C with reported D -values and $a_{w,80^\circ\text{C}}$ in different low- a_w systems^a

Low- a_w system	$a_{w,80^\circ\text{C}}$	$D_{80^\circ\text{C}}$ -value (min)	Estimated X_w of <i>S. Enteritidis</i> PT 30 (g water/100 g dry solids)
Wheat flour (11)	0.47	12.2 ± 0.7	8.8
	0.68	4.9 ± 0.5	12.8
	0.78	1.2 ± 0.2	15.5
Almond flour (11)	0.43	27.3 ± 0.3	8.2
	0.54	21.2 ± 0.9	10.0
	0.63	11.1 ± 0.8	11.7
Whey protein (11)	0.81	0.8 ± 0.1	16.5
	0.41	17.5 ± 1.3	7.9
	0.62	10.6 ± 0.2	11.5
Honey powder (23)	0.74	5.1 ± 0.4	14.4
	0.87	1.5 ± 0.1	18.7
	0.18	35.6 ± 2.3	4.3
Silicon dioxide (14)	0.31	27.3 ± 3.2	6.4
	0.40	19.6 ± 1.8	7.7
	0.50	14.4 ± 1.1	9.3
Silicon dioxide (14)	0.18	159.3 ± 5.8	4.1
	0.27	64.0 ± 0.2	5.7
	0.37	30.7 ± 1.0	7.3
	0.47	21.3 ± 1.4	8.8
	0.55	10.4 ± 0.3	10.1
	0.63	6.8 ± 0.3	11.8
	0.72	1.8 ± 0.1	13.8

^aValues are means ± SD. The reported $a_{w,80^\circ\text{C}}$ values and $D_{80^\circ\text{C}}$ -values for these low- a_w systems were derived from the work of Alshammari et al. (23), Liu et al. (14), and Xu et al. (11).

noteworthy that they both decrease exponentially with the X_w of the bacterial cells (Fig. 3 and 6).

Comparison of log D -values with the X_w of *S. Enteritidis* PT 30 in different matrices. The log $D_{80^\circ\text{C}}$ -values of *S. Enteritidis* PT 30 in silicon dioxide, wheat flour, whey protein, honey powder, and almond flour (11, 14, 23) are summarized in Table 3. In those studies, the liquid *S. Enteritidis* PT 30 inoculum was inoculated into the matrices mentioned (wet inoculation) and the inoculated matrices were then conditioned for 3 to 5 days to different a_w values at room temperature. The D -values of the bacterium at 80°C were reported against the a_w values of the matrices measured at the treatment temperature, 80°C (Table 3). We estimated the X_w of *S. Enteritidis* PT 30 in these low- a_w systems using the GAB model (equation 4), and the values are included in Table 3. With those data, we developed a linear regression line between the log $D_{80^\circ\text{C}}$ -values and the estimated X_w for *S. Enteritidis* PT 30 in low- a_w systems, as shown in Fig. 7 (red dashed line; $R^2 = 0.88$). The linear regression for log $D_{80^\circ\text{C}}$ -values of freeze-dried *S. Enteritidis* PT 30 against its X_w obtained from this study is also plotted as a dark dashed line ($R^2 = 0.98$), in Fig. 7, along with 95% confidence intervals (represented by the shaded area). Interestingly, the log $D_{80^\circ\text{C}}$ -values of *S. Enteritidis* PT 30 in silicon dioxide and different low-moisture foods are scattered mostly within the 95% confidence interval area for freeze-dried *S. Enteritidis* PT 30. The regression line derived from those data is found to fall entirely within the 95% confidence intervals. Moreover, the regression lines for freeze-dried *S. Enteritidis* PT 30 and low- a_w systems are almost overlapping. This suggests that the thermal resistance of *S. Enteritidis* PT 30 was largely determined by its X_w . This is reasonable, because the a_w of the microenvironment uniquely controlled the X_w of the bacterial cells according to the relationships in Fig. 5, regardless of the matrices.

Interpretation of the thermal resistance of *Salmonella* in low- a_w systems. The a_w directly reflects the moisture vapor pressure within the food systems where bacteria are embedded. As demonstrated in Fig. 1, moisture diffusion occurs between the bacterial cells and the microenvironment when their a_w values are not the same during heating. In general, the trends of a_w changes in protein- or starch-rich low-moisture

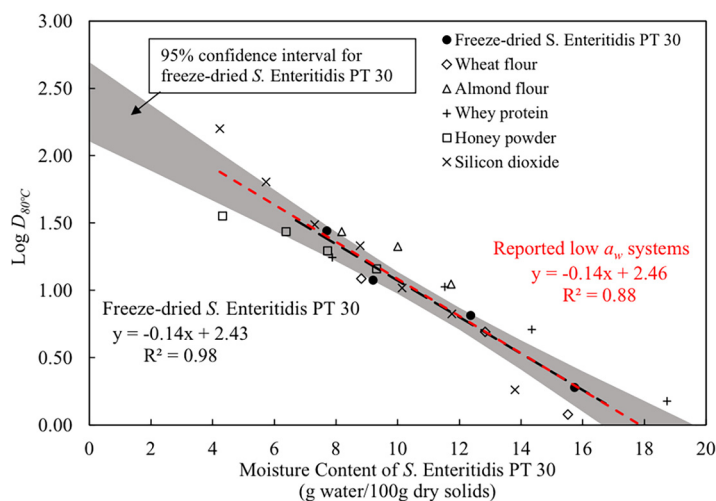


FIG 7 Comparison of logarithmic $D_{80^{\circ}\text{C}}$ -values of *S. Enteritidis* PT 30 in this study and in other reported low- a_w systems as influenced by the moisture content of bacterial cells. The shaded area represents the 95% confidence interval of log $D_{80^{\circ}\text{C}}$ -values for freeze-dried *S. Enteritidis* PT 30. The dark dashed line represents the linear regression of logarithmic $D_{80^{\circ}\text{C}}$ -values for freeze-dried *S. Enteritidis* PT 30; the red dashed line represents that of *S. Enteritidis* PT 30 for other reported low- a_w systems.

foods are similar to that of bacterial cells when heated to a high temperature (11, 17–19), and the X_w of the bacterial cells does not change much during a thermal treatment. However, the a_w of oil-rich foods is stable or even decreased when they are heated to high temperatures (15, 18, 19). The bacterial cells in oil and oil-rich foods may be dehydrated during thermal treatment (15). For instance, when the wheat flour (rich in protein and starch) or almond flour (oil rich) in which *S. Enteritidis* PT 30 is embedded is equilibrated to an a_w of 0.45 at room temperature and then heated to 80°C, the a_w of freeze-dried *S. Enteritidis* PT 30 would increase to 0.65 as predicted by the CCE (equation 3) of this study. The $a_{w,80^{\circ}\text{C}}$ values of wheat flour and almond flour would increase to 0.68 and 0.54, respectively (Table 4). According to the GAB modeled moisture sorption isotherms at 80°C (Table 2), the X_w of freeze-dried *S. Enteritidis* PT 30 at an $a_{w,20^{\circ}\text{C}}$ of 0.45 would be 12.2 g water/100 g dry solids. Due to moisture diffusion, the X_w of the bacterial cells would increase to 12.8 g water/100 g dry solids (gaining 0.6 g water/100 g dry solids) in wheat flour but decrease to 10.0 g water/100 g dry solids (losing 2.2 g water/100 g dry solids) (Table 4). According to the exponential relation shown in Fig. 3, the predicted $D_{80^{\circ}\text{C}}$ -value of freeze-dried *S. Enteritidis* PT 30 in the wheat flour would be 4.9 min when the X_w of the bacterial cells is 12.8 g water/100 g dry solids, which is not significantly different from the reported $D_{80^{\circ}\text{C}}$ -value of *S. Enteritidis* PT 30 in wheat flour (4.9 min; Table 3) (11). In contrast, the predicted $D_{80^{\circ}\text{C}}$ -value of *S. Enteritidis* PT 30 in almond flour would be 11.9 min, which is lower than the reported one (21.2 min; Table 3). Similar observations can be made when two foods

TABLE 4 Predicted $D_{80^{\circ}\text{C}}$ -values for *S. Enteritidis* PT 30 in low- a_w systems

$a_{w,20^{\circ}\text{C}}$	Low- a_w system	$a_{w,80^{\circ}\text{C}}^a$	Estimated X_w of <i>S. Enteritidis</i> PT 30 ^b (g water/100 g dry solids)	$D_{80^{\circ}\text{C}}$ -value ^c (min)
0.45	Wheat flour	0.68	12.8	4.9 ± 1.0
	Almond flour	0.54	10.0	11.9 ± 1.8
0.60	Wheat flour	0.78	15.5	2.1 ± 0.6
	Almond flour	0.63	11.7	6.9 ± 1.3

^aThe $a_{w,80^{\circ}\text{C}}$ values of wheat flour and almond flour were derived from the work of Xu et al. (11).

^bThe X_w values of *S. Enteritidis* PT 30 were obtained from the GAB model (equation 4) at 80°C.

^cThe $D_{80^{\circ}\text{C}}$ -values of freeze-dried *Salmonella* were estimated from experiment-based regression as shown in Fig. 3.

are preconditioned to an a_w of 0.6 at room temperature and heated to 80°C (as summarized in Table 4). It is shown from the examples presented above that the equilibrium X_w of *S. Enteritidis* PT 30 in heated food matrices is the dominant factor determining this organism's resistance in thermal treatments. Dehydration in bacterial cells shows its contribution to enhancing heat resistance (larger D -values).

The results of this study allow prediction of the X_w of *S. Enteritidis* PT 30 in a low-moisture food by using the GAB model (equation 4 and Table 2) and make it possible to estimate the corresponding D -value of *S. Enteritidis* PT 30 by using the exponential equation in Fig. 3. Taking peanut oil as another example: if the a_w of peanut oil is 0.53 at room temperature, it will decrease to 0.21 when the peanut oil is heated to 80°C, according to Yang et al. (39). The estimated X_w of *S. Enteritidis* PT 30 in the peanut oil will be 4.8 g water/100 g dry solids, and the corresponding $D_{80^\circ\text{C}}$ -value will be 59.4 min. That is, it would take 6 h of heating at 80°C to achieve a 6-log reduction of *S. Enteritidis* PT 30 in peanut oil with an initial a_w of 0.53.

It is reasonable to speculate that the thermal inactivation of *S. Enteritidis* PT 30 is caused mainly by irreversible stereochemical structural alterations of the hydrophilic protein components of the cells at 80°C (40). Aggregation of cytoplasmic proteins (including ribosomes) and denaturation of DNA in high-moisture *S. Enteritidis* PT 30 cells were observed from transmission electron micrographs after a 10-min thermal treatment at 80°C, while no visible aggregates were found in dried cells within 60 min of heating at 80°C (40). Indeed, it has been reported that the thermal inactivation of bacterial cells with high X_w is attributed mainly to the loss of functionality of proteins (40–42). Cytoplasmic proteins, DNA, and rRNA of bacterial cells are considered to be the major cellular targets in microbial inactivation induced by heat (43). It has also been reported that the thermal denaturation of proteins is more effective in water than in dry air, since the preferential hydration of solutes under low-moisture conditions provides protection against protein denaturation (29, 44, 45). Dehydration of proteins would induce conformational transitions (46); the absence of water may prevent proteins from deformation (47, 48) and produce more-compatible solutes to counteract the environmental stress (45) and thus may somehow stabilize aggregate-structured cytoplasmic proteins, DNA, and ribosomal units against thermally induced damages (40, 48). More systematic studies are needed to investigate the roles of water molecules in structural alterations of key cellular components, in particular in the rate of denaturation of functional proteins, during thermal inactivation of microbial organisms.

Conclusion. This study established a quantitative relationship between the X_w and D -values of freeze-dried *S. Enteritidis* PT 30. The moisture sorption isotherms provided the bridge for comparing our results with previously reported data on the thermal resistance of *S. Enteritidis* PT 30 in multiple low- a_w matrices. The linear relationship between the log D -values of *S. Enteritidis* PT 30 and its X_w can be used to predict its thermal resistance in different low-moisture foods and thus to design effective industrial thermal processes for the control of *Salmonella* in low-moisture foods. Our study provided experimental evidence to support the hypothesis that the thermal resistance of bacteria is intrinsically determined by the amount of moisture in the bacterial cells, which is adjusted by the a_w of low-moisture foods or environmental relative humidity at the treatment temperature. Food systems, such as wheat flour, that have large increases in a_w during heating will cause hydration of bacterial cells (increase in cell X_w), making them more vulnerable to thermal treatment. On the other hand, bacterial cells in oil-rich foods are more difficult to inactivate because of desiccation. Similarly, in an open environment, bacterial cells are more difficult to inactivate in low-relative-humidity environments than at high relative humidities. This work also demonstrates the importance of predicting a_w of food matrices at treatment temperatures in thermal inactivation of pathogens. This study was limited to *S. Enteritidis* PT 30. It will be interesting to explore direct connections between the moisture contents of other

serotypes or strains and their thermal resistance. Validation of the quantitative relations between the moisture content of bacterial cells and their thermal resistance may help further investigation into the fundamental roles that water molecules play in the denaturation of key protein components, leading to thermal inactivation of bacterial pathogens.

MATERIALS AND METHODS

Preparation of freeze-dried *S. Enteritidis* PT 30. The stock culture of *Salmonella* Enteritidis PT 30 (ATCC 1045) was acquired from Linda Harris at the University of California, Davis. It was stored at -80°C in tryptic soy broth supplemented with 0.6% (wt/vol) yeast extract (TSBYE; Difco, Detroit, MI, USA) and 20% (vol/vol) glycerol.

The procedure to produce the biomass of freeze-dried of *S. Enteritidis* PT 30 was modified based on a previous study (22). Briefly, a loop of thawed stock was inoculated into 9 ml TSBYE with two consecutive transfers and was incubated at 37°C for 24 h. Three hundred microliters of this aliquot was transferred to a centrifuge tube with 30 ml TSBYE and was incubated at 37°C for 24 h. Then 4 ml of the previous culture broth was further transferred to a sterile conical flask with 400 ml TSBYE and was incubated at 37°C for 24 h with constant shaking at 230 rpm. The enlarged bacterial culture was further washed twice with sterile double-distilled water (ddH_2O) by centrifuging at $6,000 \times g$ for 10 min at 4°C . The supernatant was discarded, and the washed pellets from a total of 1.2 liters of bacterial broth (3 conical flasks) were pooled and resuspended in 6 ml sterile ddH_2O . Washing caused a reduction in population of $\sim 0.06 \log_{10}$ CFU/ml, and the population in the concentrated bacterial suspension was $\sim 12.3 \log_{10}$ CFU/ml. One milliliter of the above bacterial suspension ($\sim 12 \log_{10}$ CFU/ml) was distributed into each sterile clear serum vial (5 ml; outer diameter, 22 mm; Wheaton; DWK Life Sciences, Millville, NJ, USA) and then loosely sealed with a 2-leg lyophilization stopper (DWK Life Sciences, Millville, NJ, USA). Vials were prefrozen in liquid nitrogen for a few minutes and then lyophilized at -90°C and 0.6 Pa for 48 h in a freeze-dryer (FreeZone Plus 4.5-liter cascade benchtop freeze-dry system, Labconco Corporation, Kansas City, MO, USA).

Generally, about 2 g of freeze-dried *S. Enteritidis* PT 30 was harvested from the vials after each batch of freeze-drying. All solids from each batch were collected into a 4-oz sterilized Whirl-Pak bag (Nasco, Modesto, CA, USA), hand mixed for 5 min to eliminate clumps, and further homogenized using a stomacher at 230 rpm for 3 min (Stomacher model 400 circulator; Seward Laboratory Systems Inc., Norfolk, UK). Then this homogeneous freeze-dried *S. Enteritidis* PT 30 powder (10 to 11 \log_{10} CFU/g) was kept at -80°C until use (within 1 week). Each batch of the sample was used in an individual thermal treatment. All repeated batches were derived from the same cold stock and were cultivated independently.

Viability test of freeze-dried *S. Enteritidis* PT 30. The viability and population of freeze-dried *S. Enteritidis* PT 30 were determined immediately by placing 0.100 ± 0.010 g in a 2-ml sterile snap-lock microtube (Axygen, Union City, CA, USA) and then mixing with 0.9 ml buffered peptone water (BPW; Difco, Detroit, MI, USA) to obtain a 10-fold dilution. The freeze-dried *S. Enteritidis* PT 30 was fully mixed with BPW by subsequent vortexing (Fisherbrand pulsing vortex mixer; Thermo Fisher Scientific, Waltham, MA, USA). Serial dilutions were performed to proper levels, and the diluted material was then spread in duplicate for enumeration at 37°C for 48 h on modified TSAYE plates (49), which were made up of tryptic soy agar (Difco, Detroit, MI, USA), 0.6% (wt/vol) yeast extract (Difco, Detroit, MI, USA), 0.05% (wt/vol) ferric ammonium citrate (Sigma-Aldrich, St. Louis, MO, USA) and 0.03% (wt/vol) sodium thiosulfate (Sigma-Aldrich, St. Louis, MO, USA). Typical *Salmonella* cells exhibited colonies that resembled dark solid circles. The average number of viable colonies was expressed as the CFU count per gram based on two technical replicates.

Conditioning freeze-dried *S. Enteritidis* PT 30 biomass to different a_w values. The initial a_w of the above homogeneous freeze-dried *S. Enteritidis* PT 30 powder was < 0.025 at room temperature ($\sim 21^{\circ}\text{C}$), as measured with a water activity meter (Aqualab; Meter Group, Inc., Pullman, WA, USA). The X_w of freeze-dried *S. Enteritidis* PT 30 was determined by the oven-drying method according to AOAC official method 925.10 (50).

In order to obtain freeze-dried *S. Enteritidis* PT 30 powder samples with different a_w or X_w levels, freeze-dried *S. Enteritidis* PT 30 was evenly spread in each sample cup (~ 300 mg for high-temperature cells) or petri dish (~ 2 g for thermal treatment) and conditioned at room temperature for 2 to 3 days in airtight jars with a saturated salt solution under various relative humidity levels at room temperature (19). The saturated salt solutions of LiCl, CH_3COOK , MgCl_2 , K_2CO_3 , $\text{Mg}(\text{NO}_3)_2$, NaNO_2 , NaCl, and KCl could generate a consistent relative humidity of 11.3%, 22.5%, 32.8%, 43.2%, 52.9%, 65.8%, 75.3%, and 84.3%, respectively, at room temperature, which corresponded to the equivalent target a_w values of 0.1, 0.2, 0.3, 0.4, 0.5, 0.6, 0.7, and 0.8, respectively, at room temperature. In measuring the moisture sorption isotherms by high-temperature cells, freeze-dried *S. Enteritidis* PT 30 was conditioned at all the above relative humidity levels. For the thermal treatments, freeze-dried *S. Enteritidis* PT 30 was conditioned at four selected relative humidity levels (22.5%, 32.8%, 43.2%, and 65.8%). The equilibration of the freeze-dried *S. Enteritidis* PT 30 samples was verified by measuring the a_w value using the water activity meter. Samples were used for all experiments after the target $a_w \pm 0.02$ was reached.

Thermal treatments of freeze-dried of *S. Enteritidis* PT 30. Thermal treatments of freeze-dried *S. Enteritidis* PT 30 were performed at 80°C using the improved aluminum thermal-death-time test cells (TDT II) in triplicate. TDT II cells are thinner (height, 1.39 mm) and have a larger cavity (diameter, 31.18 mm) than traditional TDT cells (51). During the heating process, the X_w of freeze-dried *S. Enteritidis* PT 30 should remain constant in well-sealed TDT II cells, while the a_w of freeze-dried *S. Enteritidis* PT 30 could change with increasing temperatures.

Come-up time (CUT, in minutes) is the time required for samples to reach the target temperature within 0.5°C. It was measured using a specially designed TDT II cell installed with a 0.5-mm-diameter thermocouple (type T; Omega Engineering, Inc., Stamford, CT, USA) secured on the top center of the lid. The cell was filled with 100 mg of freeze-dried *S. Enteritidis* PT 30 and sealed. The core temperature of the freeze-dried *S. Enteritidis* PT 30 in the test cell was measured by a thermometer (Digi-Sense DualogR, model 99100-50; Cole-Parmer Instruments Co., Vernon Hills, IL, USA) when it was subjected to thermal treatment at 80°C. The CUT was determined in triplicate.

Freeze-dried *S. Enteritidis* PT 30 samples were conditioned for 2 to 3 days to reach the target a_w (0.2, 0.3, 0.4, and 0.6). A portion (0.100 ± 0.010 g) of the equilibrated freeze-dried *S. Enteritidis* PT 30 was loaded into a TDT II cell and sealed. Thermal treatment at 80°C was carried out by immersing test cells in a preheated ethylene glycol bath circulator (Fisherbrand Isotemp 5150 H24; Thermo Fisher Scientific, PA, USA). Duplicate TDT II cells were removed from the circulator at five predetermined sampling points (come-up time was regarded as 0 min) and immediately cooled in an ice water bath for 1 min. Thermal inactivation treatments were independently repeated in triplicate.

Thermally treated freeze-dried *S. Enteritidis* PT 30 was transferred from the test cell to a 2-ml sterile snap-lock microtube (MCT-200-C; Axygen, Union City, CA, USA), and 0.9 ml of sterile BPW was added to achieve a 10-fold dilution. Next, the bacterial suspension was homogenized using a vortex and was further serially diluted 10-fold. The appropriate dilutions were spread on modified TSA-YE plates in two technical replicates and were incubated at 37°C for 48 h. The average number of viable colonies at each time point was converted to CFU per gram based on two technical replicates.

Statistical analysis. The first-order kinetic model (log-linear model; equation 1) and the Weibull model (equation 2) were applied to analyze thermal inactivation (52). The thermal decimal time (D -value) in minutes was estimated by these models, and the goodness of fit was evaluated by the R^2 coefficient. Statistical analysis of the standard deviation (SD) was performed using Microsoft Excel (version 16.35; Microsoft)

Determination of a_w changes of freeze-dried *S. Enteritidis* PT 30 with increasing temperatures.

The a_w changes of freeze-dried *S. Enteritidis* PT 30 at different temperatures were determined using aluminum high-temperature cells (Meter Group, Inc., Pullman, WA, USA) designed by Tadapaneni et al. (19). The a_w was measured by a capacitance-based relative humidity and temperature sensor (HumidCon; Honeywell, Morristown, NJ, USA) located on the center of the inner side lid. The sample cup with pre-equilibrated freeze-dried *S. Enteritidis* PT 30 (300 mg) was placed in a high-temperature cell and sealed tightly to prevent any leakage. High-temperature cells were kept at room temperature (~21°C) and then heated in 20°C increments from 40°C to 80°C by immersion in a heated glycol bath (Fisherbrand Isotemp 5150 H24; Fisher Scientific, PA, USA). The relative humidity and temperature of the headspace were read every minute, and the equilibrium state at the respective temperature was achieved when constant relative humidity values were obtained for at least 10 readings. Then the temperature and relative humidity values were recorded, and the relative humidity value was considered to be the corresponding a_w value of the equilibrated freeze-dried *S. Enteritidis* PT 30 in this closed system at a certain temperature (19). After the above series of measurements was recorded, and the high-temperature cells were cooled to room temperature, the X_w of freeze-dried *S. Enteritidis* PT 30 was determined. All experiments were performed in triplicate.

Clausius-Clapeyron equation and moisture sorption isotherms. The experimental a_w data were fitted by a modified Clausius-Clapeyron equation (CCE) (equation 3) according to Tadapaneni et al. (19). Data measured by high-temperature cells were further fitted according to the Guggenheim-Anderson-Boer (GAB) model (equation 4) to generate moisture sorption isotherms, providing the relationship between a_w and X_w (28).

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