

## Mining Natural Products for Macrocycles to Drug Difficult Targets

Fabio Begnini,<sup>#</sup> Vasanathanathan Poongavanam,<sup>#</sup> Björn Over,<sup>#</sup> Marie Castaldo, Stefan Geschwindner, Patrik Johansson, Mohit Tyagi, Christian Tyrchan, Lisa Wissler, Peter Sjö, Stefan Schiesser,<sup>\*</sup> and Jan Kihlberg<sup>\*</sup>Cite This: *J. Med. Chem.* 2021, 64, 1054–1072

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**ABSTRACT:** Lead generation for difficult-to-drug targets that have large, featureless, and highly lipophilic or highly polar and/or flexible binding sites is highly challenging. Here, we describe how cores of macrocyclic natural products can serve as a high-quality *in silico* screening library that provides leads for difficult-to-drug targets. Two iterative rounds of docking of a carefully selected set of natural-product-derived cores led to the discovery of an uncharged macrocyclic inhibitor of the Keap1-Nrf2 protein–protein interaction, a particularly challenging target due to its highly polar binding site. The inhibitor displays cellular efficacy and is well-positioned for further optimization based on the structure of its complex with Keap1 and synthetic access. We believe that our work will spur interest in using macrocyclic cores for *in silico*-based lead generation and also inspire the design of future macrocycle screening collections.



## INTRODUCTION

More than half of all targets predicted to be involved in human disease are considered to be difficult to modulate with traditional small-molecule drugs that obey Lipinski's rule of 5 (Ro5).<sup>1</sup> Protein–protein interactions (PPIs), proteases, some kinases as well as transferases and isomerases are important examples.<sup>2,3</sup> These difficult-to-drug targets often have binding sites that are large, featureless, highly lipophilic or highly polar, and/or flexible.<sup>2–4</sup> Finding orally bioavailable drugs that reach intracellular difficult-to-drug targets is a daunting task that often requires discovery of ligands in the uncharted chemical space beyond the Ro5 (bRo5).<sup>2,5,6</sup> Macrocycles<sup>7</sup> are enriched among oral drugs in the bRo5 space<sup>5</sup> because they offer superior binding to targets that have large and featureless binding sites as compared to traditional small molecules.<sup>2,8</sup> In addition, macrocyclization may improve plasma stability, cell permeability, and oral absorption.<sup>9</sup> However, macrocycles are often under-represented in screening collections, limiting their use for lead generation. For instance, the AstraZeneca collection contains less than 17 000 macrocycles out of a total of more than 3.8 million compounds.

Despite a growing number of bRo5 compounds entering clinical trials, lead generation for difficult-to-drug targets has not been systematically investigated and is challenging as the chemical space expands dramatically with increasing molecular size.<sup>10</sup> Natural products are an important source of drugs,<sup>11</sup> and we hypothesized that the cores of macrocycles may be seen as nature's privileged substructures and could play a similar role in lead generation for difficult-to-drug targets as fragments do for traditional targets.<sup>12,13</sup> A collection of lead-like macrocyclic

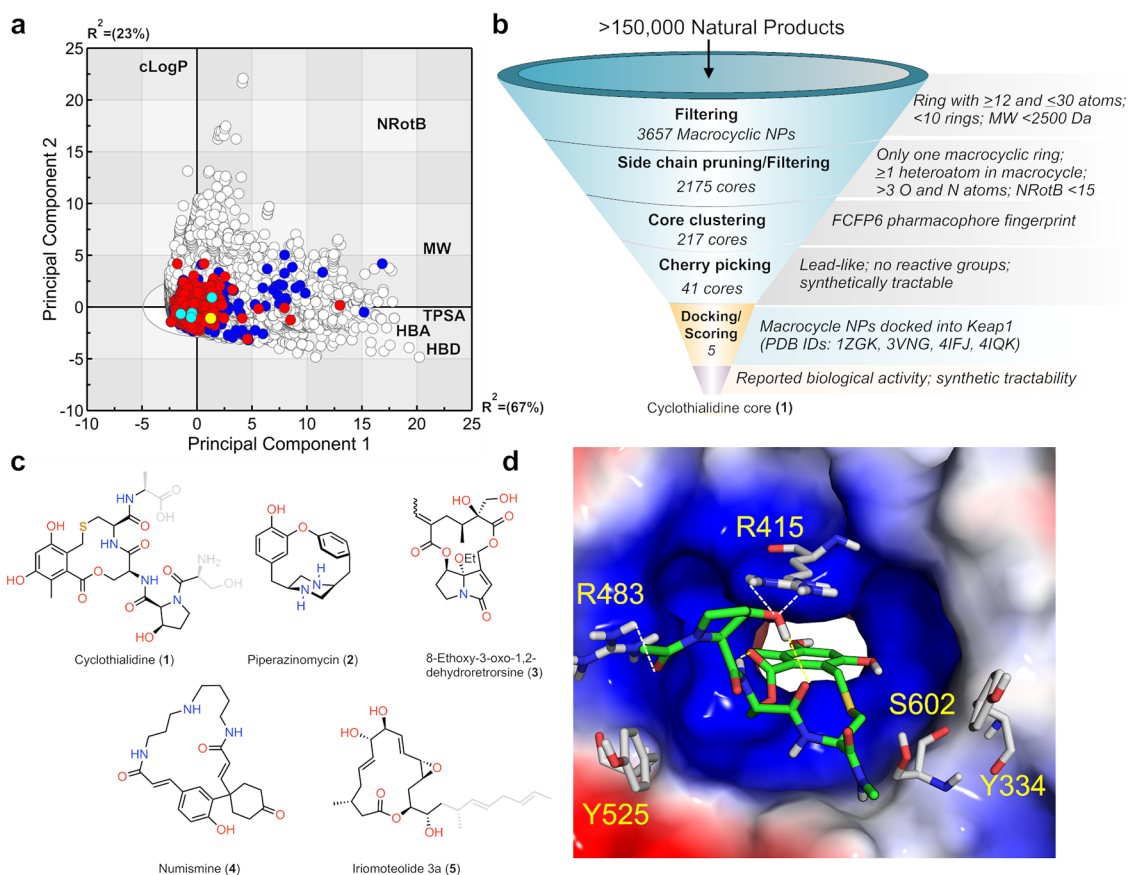
cores would then be useful for *in silico* screening against targets such as PPIs, with subsequent optimization providing novel leads. In addition, such a collection could inspire the design of natural-product-like macrocycle screening libraries.

Here, we report how two sets of macrocyclic cores that may be used to discover leads for difficult-to-drug targets were generated by a comprehensive investigation of macrocyclic natural product chemical space. Docking of the smaller and more lead-like set into the highly charged binding site for Nrf2 on Keap1 identified the core of cyclothialidine (**1**) as a potential inhibitor of Keap1. A second round of docking studies using structurally simplified analogues of the core, followed by synthesis, identified a weak inhibitor of the Keap1-Nrf2 PPI. Optimization of the hit via synthesis of an additional nine compounds led to an inhibitor (**14**), which has a potency in the low  $\mu\text{M}$  range and displays cellular activity. The crystal structure of the complex of Keap1 and **14** reveals how the uncharged macrocycle **14** binds to the charged binding site of Keap1 and provides a platform for its further optimization.

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**Figure 1.** Mining the Dictionary of Natural Products for macrocycle cores. (a) Principal component analysis comparing the chemical space of all compounds in the Dictionary of Natural Products<sup>14</sup> (white circles) to that of orally and parenterally administered drugs from DrugBank<sup>16</sup> (red and blue circles, respectively). The first two principal components explain 90% of the variance in the data set ( $R^2 = 0.90$ ,  $Q^2 = 0.70$ ). The positions of the core of cyclothialidine (1, yellow circle) and macrocycle cores 2–5 (circles in cyan) are indicated. The contribution of descriptors used to characterize the drugs to each of the principal components is indicated by their position on each of the axes. Descriptors: cLogP, calculated lipophilicity; HBA, hydrogen-bond acceptor; HBD, hydrogen-bond donor; MW, molecular weight; NRotB, number of rotatable bonds; TPSA, topological polar surface area. (b) Funnel describing how the Dictionary of Natural Products was filtered and clustered to give one large set of 217 macrocycle cores and one smaller set of 41 lead-like cores. The identification of the core of cyclothialidine (1) by docking of the lead-like cores into the binding site of Keap1 is also indicated. (c) Structures of the five top-ranked macrocycles obtained by docking into Keap1. The bonds of the docked cores are in black and their heteroatoms are colored, while side chains that were pruned from the original natural product have bonds and heteroatoms in gray. (d) Pose obtained from docking of the core of cyclothialidine (1) into the binding site of Keap1 (PDB ID: 4IQK), in which the substituted phenylene group protrudes into the Kelch channel. Key residues in the Keap1 binding site are highlighted.

## RESULTS

**Mining the Dictionary of Natural Products.** The Dictionary of Natural Products (DNP)<sup>14</sup> contains >150 000 compounds that cover a vast chemical space. This space includes that of approved oral and parenteral drugs but also extends far beyond (Figures 1a and S1). Therefore, we considered the DNP as an attractive source for the identification of macrocyclic cores as promising starting points for difficult-to-drug targets. First, duplicates and natural products containing known toxicophores and/or very reactive groups were removed from the DNP using the HTS filter implemented in Pipeline Pilot,<sup>12</sup> which removes several functionalities including acyl halides, anhydrides, diazo groups, and hydrazides. Then, the remaining compounds were filtered to contain at least one macrocyclic ring, not more than ten smaller rings, and to have a molecular weight (MW)  $< 2500$  Da (Figures 1b and S2). This removed all nonmacrocycles and large macrocycles, e.g., of polysaccharidic or polypeptidic nature, which are unlikely to reach intracellular targets and provided a data set of 3657 macrocycles. The side chains of this set were pruned to attachment points consisting of at least two

heavy atoms or functional groups connected to the macrocycle using the first two steps of the fragment generation algorithm reported previously.<sup>12</sup> In this way, fragmentation of the macrocycle ring was avoided. The resulting extended Murcko scaffolds,<sup>15</sup> herein denoted macrocycle cores, were filtered to contain at least a total of three oxygen and nitrogen atoms and  $< 15$  rotatable bonds, thereby removing highly lipophilic and/or flexible cores unlikely to be developable into drugs. Close to 2200 structurally different cores were identified in this macrocycle collection, which were clustered into a set of 217 representative cores. Cores that contained a few remaining reactive groups (i.e., disulfides, peroxides, and alkyl halides), had an overwhelming synthetic complexity, or that were oligomeric (oligopeptides and oligosaccharides) were removed by visual inspection. This provided a smaller set of 41 more lead-like cores that to a large extent adhered to Lipinski's rule of 5 (Table S1 and Figure S3). As molecular weight and complexity may be expected to increase during optimization, they were judged to be suitable starting points for development into cell-permeable inhibitors of difficult-to-drug targets. This set of cores originates from 35 natural products for which the absolute stereochemistry

has been reported and from three natural products for which only the relative stereochemistry is known, which were included as enantiomeric pairs. The two sets of cores constitute unique *in silico* collections that can be used in efforts to find novel starting points in drug discovery (Supporting Excel Sheet). The structural complexity varies between the cores in the two sets; some can be readily synthesized, while simplified analogues may be more attractive starting points for other macrocycles.

**Novel Class of Inhibitors of the Keap1-Nrf2 PPI.** We chose the PPI between Kelch-like ECH-associated protein 1 (Keap1) and nuclear factor erythroid 2-related factor 2 (Nrf2), to evaluate our sets of macrocyclic cores. The Keap1-Nrf2 system is an important cellular response mechanism for oxidative stress, which is involved in many chronic, age-related, or inflammatory diseases.<sup>17</sup> Inhibitors of this PPI are of major interest for drug discovery as Nrf2 controls the expression of several cytoprotective genes but is negatively regulated through complexation with Keap1, leading to its ubiquitination and degradation. The Keap1-Nrf2 PPI consists of a highly polar pocket on Keap1, which binds the peptide motif DEETGE on Nrf2, and other negatively charged ligands, with high affinity.<sup>18,19</sup> This renders it challenging to find cell-permeable and orally absorbed ligands that reach and inhibit the Keap1-Nrf2 PPI,<sup>19–21</sup> just as for other targets that have polar binding sites.<sup>22,23</sup> Targeting the Keap1-Nrf2 PPI in CNS disorders is particularly challenging.<sup>19,20</sup> We hypothesize that macrocycles, due to their conformational restriction, may provide an optimal fit in the polar binding site of Keap1. Thus, they may be ideal for the discovery of novel uncharged inhibitors with improved cell permeability and oral absorption as compared to current Keap1 inhibitors, the majority of which are acidic.<sup>20,21</sup>

The binding site on Keap1 for Nrf2 shows a certain degree of conformational mobility.<sup>21</sup> In particular, the side chain of Arg415 adopts its conformation so that the binding site is more open in the apo form and more closed when bound to small-molecule inhibitors. To avoid bias toward a particular binding site conformation when docking the macrocyclic cores, we selected four high-resolution crystal structures of Keap1 that showed variation in the binding site: two with a bound small-molecule ligand (PDB ID: 4IQK<sup>24</sup> and 3VNG) and two apo structures (PDB ID: 4IFJ and 1ZGK<sup>25</sup>) (Figure S4). Docking of the 41 cores of the lead-like set into the binding site was then performed using flexible docking in Glide.<sup>26</sup> The ten cores that docked best into each Keap1 structure were identified by their GlideScores (Table S2). Then, the cores that had docked into three or four of the Keap1 structures were identified from the combined top-ten lists and selected as potential hits as we anticipated them to be more likely to bind to Keap1 than cores that docked well into only one or two crystal structures (Table S3). Interestingly, the resulting top-five macrocycles displayed a large structural diversity (Figure 1c). Inspection of the docked poses revealed that the core of cyclothialidine (1), which docked well into all four Keap1 structures, was able to bind deep in the binding site and reached into the Kelch channel (Figure 1d). The cores of piperazinomycin (2), 8-ethoxy-3-oxo-1,2-dehydroretrosine (3), numismine (4), and iriomoteolide 3a (5), which docked well into only three of the Keap1 structures, bound closer to the periphery, providing fewer opportunities for interactions with Keap1 (Figure S5). To the best of our knowledge, the natural product precursors of cores 1–5 have not been reported to bind to Keap1. However, cytotoxicity has been reported for the natural product precursor of core 5, which may make 5 less suitable for development of an inhibitor of the

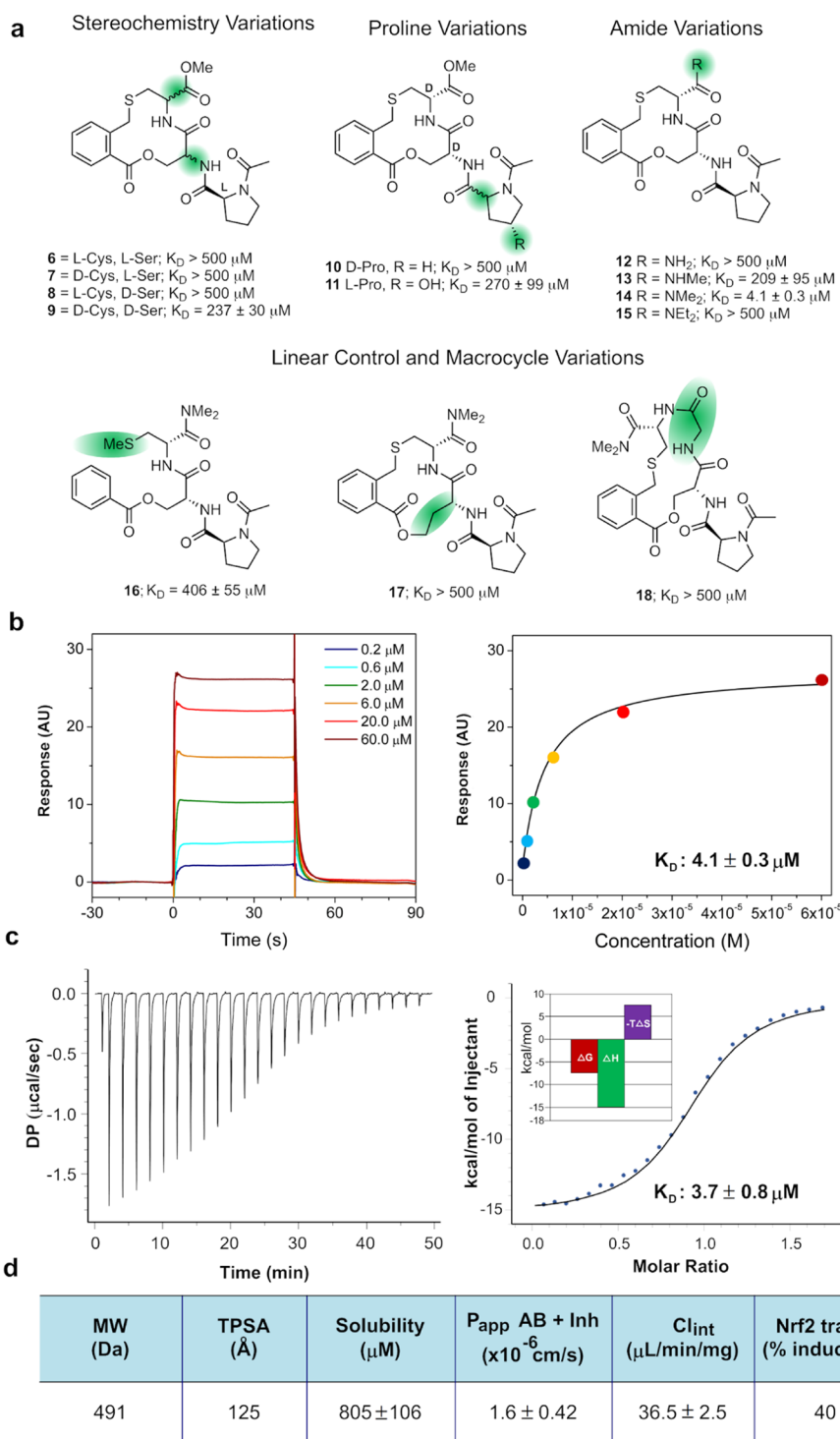
Keap1-Nrf2 PPI (Table S4). No synthetic routes have been reported for 3 and 4, the routes to 2 and 5 are long,<sup>27,28</sup> while a route suitable for synthesis of analogues has been reported for the core of 1.<sup>29</sup> Overall, these considerations revealed the cyclothialidine core as the most promising starting point for discovery of a novel Keap1 inhibitor.

Induced-fit docking suggested that simplified analogues of cyclothialidine core, such as stereoisomers 6–9 where the hydroxyl and methyl groups of the phenylene group and the hydroxyl of the proline have been removed (Figure 2a), would also bind to Keap1 (Figure S6). In addition, the docking also indicated that 8 and 9 were more likely to bind stronger than 6 and 7. Satisfactorily, 9 was found to inhibit the binding of Keap1 to an immobilized peptide derived from Nrf2 ( $K_D = 237 \mu\text{M}$ ) in a surface plasmon resonance (SPR)-based inhibition in solution assay (ISA)<sup>30</sup> (Figure 2a). Inversion of the stereochemistry of the L-proline moiety to give 10 led to a reduced binding affinity that revealed its importance for inhibition of Keap1, while hydroxylation of L-proline (11) maintained the  $K_D$  (Figures 2a and S7). Importantly, replacement of the methyl ester with different amides (12–15) provided dimethylamide 14, which is close to 2 orders of magnitude more potent than 9. Ring-opened 16 was drastically less active than 14; similarly, macrocycles ring expanded by one and three atoms, respectively (17 and 18), also lost binding affinity, confirming the critical role of the macrocycle for inhibition of Keap1. The activity of 14 in the ISA was confirmed in a direct SPR binding assay<sup>31</sup> and by isothermal titration calorimetry (ITC), which provided almost identical affinities for Keap1 (Figure 2b,c). Macrocycle 14 displayed cellular activity as it induced Nrf2 translocation into the nucleus by inhibiting binding to Keap1 (Figure 2d). Moreover, 14 has high aqueous solubility, low-to-moderate permeability across Caco-2 cell monolayers, and a medium *in vitro* microsomal clearance. It therefore constitutes a promising lead compound for further optimization into an uncharged, nonacidic Keap1 advanced lead.

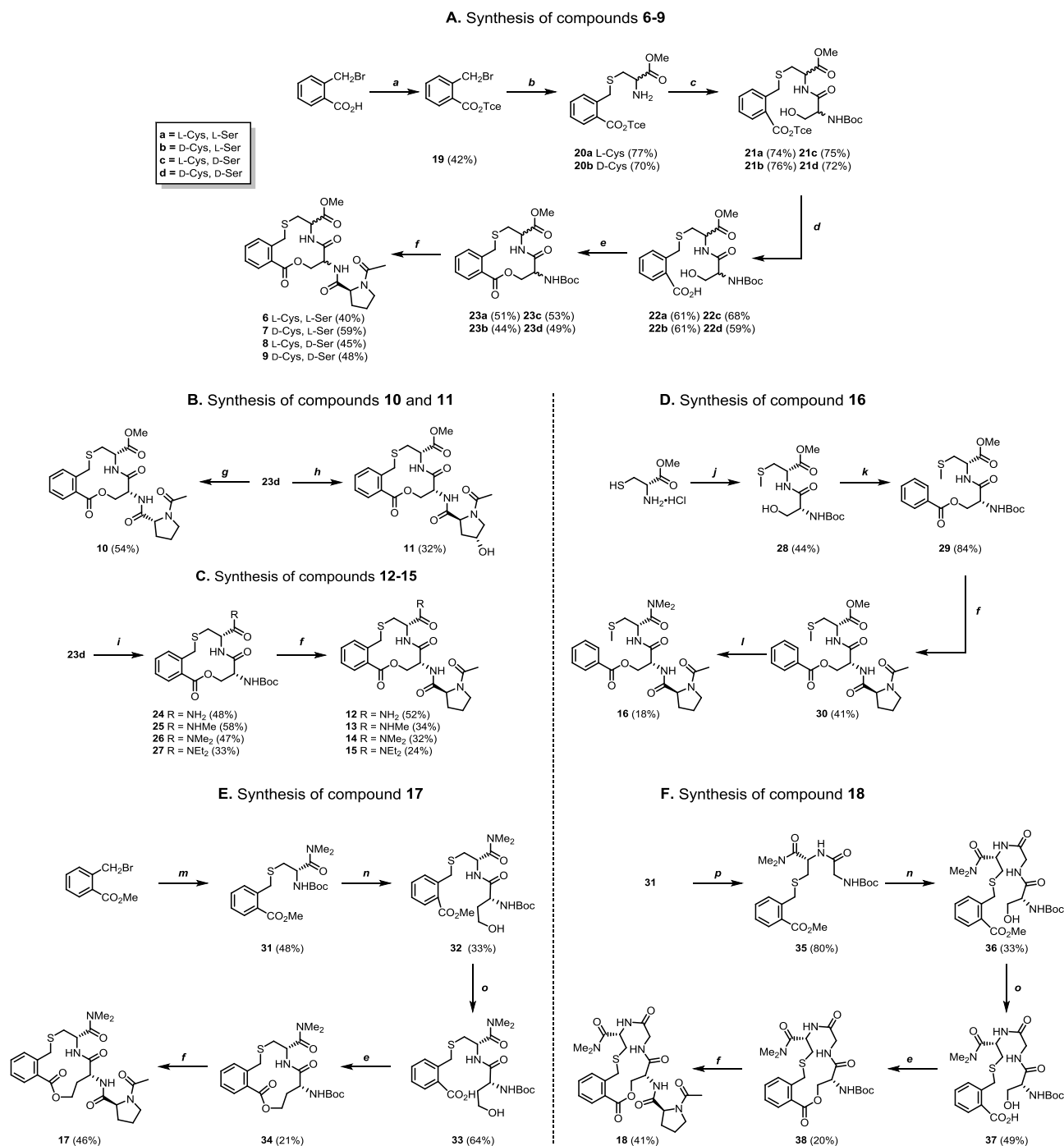
**Syntheses of Compounds 6–18.** Compounds 6–18 were synthesized in five to seven steps from commercially available starting materials, using a modified version of the route reported for cyclothialidine (Scheme 1).<sup>29</sup> The synthesis of macrocycles 6–9 (Scheme 1A) was initiated with the protection of 2-(bromomethyl)benzoic acid as a 2,2,2-trichloroethyl (Tce) ester to give 19. Benzyl bromide 19 was used to alkylate the thiol of L- and D-cysteine methyl ester, respectively, providing 20a and 20b, which were then coupled with L- and D-Boc-serine affording dipeptides 21a–d. The Tce protecting group was cleaved using zinc powder to obtain acids 22a–d, which were subjected to an intramolecular Mitsunobu reaction to obtain macrocycles 23a–d. To our delight, the macrocyclization conditions proved robust, providing satisfactory yields for all four diastereoisomers 23a–d. Removal of the Boc group using acidic conditions, followed by coupling of the resulting free amine with Ac-L-Pro-OH, provided compounds 6–9.

Macrocycle 23d, the key intermediate in the synthetic sequence, was used to access compounds 10–15. After cleavage of its Boc group, coupling with Ac-D-Pro-OH or Ac-L-Hyp-OH provided compounds 10 and 11, respectively (Scheme 1B). The methyl ester of compound 23d was cleaved using  $\text{Me}_3\text{SnOH}$ , as the use of traditional alkaline metal hydroxides led to partial opening of the lactone ring. The resulting crude acid was directly coupled with four different amines to give amides 24–27 (Scheme 1C). In a similar fashion as for compounds 6–9,





**Figure 2.** Characterization of cyclothialidine analogues. (a) Synthesized analogues of the cyclothialidine core that were evaluated as inhibitors of binding of Keap1 to an immobilized peptide derived from Nrf2 by surface plasmon resonance using an inhibition in solution assay (ISA) format. Dissociation constants, reported as mean values  $\pm$  standard deviation, from three measurements on three distinct samples are given for each analogue. (b) Interaction kinetic analysis of a dilution series of macrocycle **14** in a direct binding assay using immobilized Keap1 (left). Determination of the dissociation constant ( $K_D$ ) for **14** by fitting of the data to a two-parametric sigmoidal equation (right). The dissociation constant was obtained from three measurements on three distinct samples and is reported as the mean value  $\pm$  standard deviation. (c) Determination of  $K_D$  for the binding of **14** to Keap1 by isothermal titration calorimetry. The raw heat signals from the exothermic binding reaction (left) have been integrated to yield a binding isotherm (right) from which the thermodynamic parameters were extracted (insert). The dissociation constant was obtained from three measurements on three distinct samples and is reported as the mean value  $\pm$  standard deviation. (d) Characterization of macrocycle **14** by calculated descriptors (MW and TPSA), solubility in phosphate-buffered saline at 25 °C and pH 7.4, efflux-inhibited permeability across a Caco-2 cell monolayer ( $P_{\text{app}} \text{ AB + inh}$ ), human microsomal metabolism ( $Cl_{\text{int}}$ ), and induction of Nrf2 translocation into the nucleus (Nrf2 transl) at 256  $\mu\text{M}$ . The values for solubility, cell permeability, and human microsomal metabolism are mean values  $\pm$  standard deviation from three measurements on three distinct samples. The Nrf2 translocation into the nucleus is the mean from two measurements on two distinct samples.

Scheme 1. Syntheses of Compounds 6–18<sup>a</sup>

<sup>a</sup>Reagents and conditions: (a) 2,2,2-trichloroethanol, EDC·HCl, 4-dimethylaminopyridine (DMAP), dichloromethane (DCM), rt, 16 h; (b) H-(D/L)-Cys-OMe·HCl, Et<sub>3</sub>N, DMSO, rt, 2 h; (c) Boc-(D/L)-Ser-OH, EDC·HCl, MeCN, rt, 1 h; (d) Zn dust, aq. NH<sub>4</sub>OAc, tetrahydrofuran (THF), rt, 2 h; (e) PPh<sub>3</sub>, di-*tert*-butylazodicarboxylate, toluene:DMSO 95:5, rt, 4 h; (f) HCl in 1,4-dioxane, rt, 1 h, then Ac-L-Pro-OH, EDC·HCl, *N,N*-diisopropylethylamine (DIPEA), dimethyl sulfoxide (DMSO), rt, 2 h; (g) HCl in 1,4-dioxane, rt, 1 h, then Ac-D-Pro-OH, EDC·HCl, DIPEA, DMSO, rt, 2 h; (h) HCl in 1,4-dioxane, rt, 1 h, then Ac-L-Hyp-OH, EDC·HCl, DIPEA, DMSO, rt, 2 h; (i) Me<sub>3</sub>SnOH, 1,2-DCE, 83 °C, 45 min, then R<sub>2</sub>NH, EDC·HCl, HOBT·xH<sub>2</sub>O, *N,N*-dimethylformamide (DMF), rt, 2 h; (j) MeI, DIPEA, DMSO, rt, 2 h, then Boc-D-Ser-OH, EDC·HCl, MeCN, rt, 1 h; (k) BzCl, DMAP, Et<sub>3</sub>N, DCM, 0 °C to rt, 1 h; (l) Me<sub>3</sub>SnOH, 1,2-DCE, 83 °C, 45 min, then Me<sub>2</sub>N·HCl, EDC·HCl, HOBT·xH<sub>2</sub>O, DIPEA, DMF, rt, 2 h; (m) Boc-D-Cys-OH, Et<sub>3</sub>N, THF:DMF 4:1, rt, 16 h, then Me<sub>2</sub>N·HCl, EDC·HCl, HOBT·xH<sub>2</sub>O, DIPEA, DMF, rt, 2 h; (n) HCl in 1,4-dioxane, rt, 1 h, then Boc-D-Homoser-OH, EDC·HCl, DIPEA, MeCN, rt, 1 h; (o) aq. LiOH, MeOH, 35 °C, 16 h; (p) HCl in 1,4-dioxane, rt, 1 h, then Boc-Gly-OH, EDC·HCl, MeCN, rt, 1 h. DMAP, 4-dimethylaminopyridine; EDC, *N*-(3-dimethylaminopropyl)-*N'*-ethylcarbodiimide; HOBT, 1-hydroxybenzotriazole; Tce, 2,2,2-trichloroethyl.

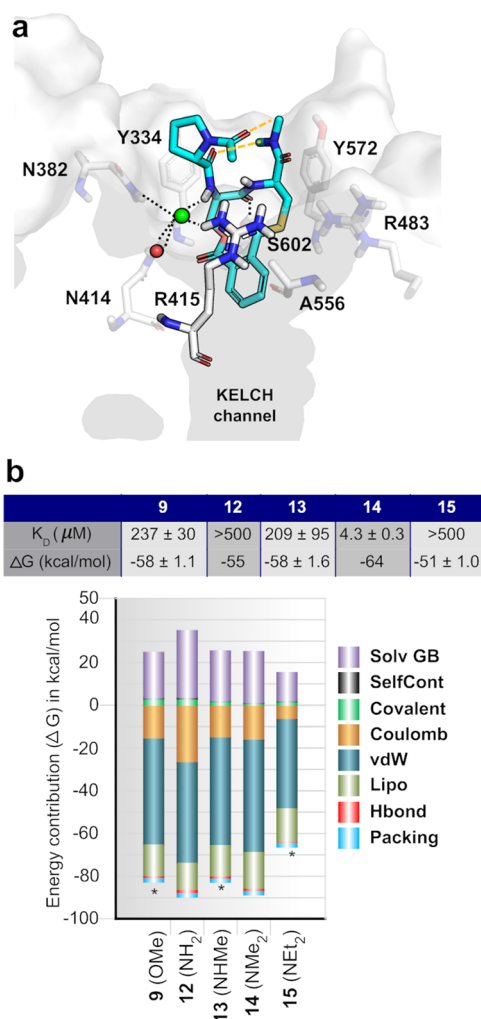
cleavage of the Boc group from 24–27 and coupling with Ac-L-Pro-OH provided compounds 12–15.

The synthesis of ring-opened compound 16 (Scheme 1D) started with the methylation of the sulfur of D-cysteine methyl ester, followed by coupling with Boc-D-Ser to provide dipeptide 28, the hydroxyl group of which was acylated with benzoyl chloride in the presence of DMAP leading to compound 29. After Boc removal and coupling with Ac-L-Pro-OH tripeptide 30 was obtained. Cleavage of the methyl ester followed by coupling of the resulting free acid with dimethylamine afforded the linear control 16.

The preparation of the ring-expanded macrocycles 17 and 18 (Scheme 1E,F) is based on the common intermediate 31, which was prepared by alkylation of the thiol moiety of Boc-D-Cys with commercially available 2-bromomethyl methyl benzoate, followed by coupling of the crude carboxylic acid with dimethylamine. The Boc protecting group was then cleaved, the liberated amine was coupled with Boc-D-homoserine to give compound 32, followed by saponification of the methyl ester, which gives compounds 33. The same conditions employed for the Mitsunobu reaction with acids 23a–d were successfully applied to form the 13-membered lactone of 34. Finally, Boc removal and coupling with Ac-L-Pro-OH provided macrocycle 17. For the preparation of compound 18 (Scheme 1F), the Boc group of 31 was cleaved followed by coupling with Boc-glycine to obtain dipeptide 35. The newly introduced Boc group was cleaved and the liberated amine was coupled with Boc-D-serine to afford tripeptide 36. After cleavage of the methyl ester, the obtained acid 37 was subjected to a Mitsunobu reaction to form the 15-membered core of compound 38, followed by Boc cleavage and coupling with Ac-L-Pro-OH to give macrocycle 18.

**Structure of Inhibitor 14 in Complex with Keap1.** We determined the structure of the complex of macrocycle 14 and Keap1 at 2.4 Å resolution to understand how uncharged 14 binds in the charged and polar binding site of Keap1. The crystal structure confirms that 14 binds to the same polar binding site in Keap1 as Nrf2 (Figure 3a, Table S5, and Figure S8). Inspection of the structure shows that the phenylene group and parts of the macrocycle are wedged between R415 and A556 in a fairly hydrophobic pocket that reaches toward the Kelch channel, confirming that 14 is bound similarly to the docked poses of analogue 9 (Figure S9). The complex is stabilized only by a few polar interactions in addition to the cation– $\pi$  interaction between R415 and 14. The carbonyl group of Ser in 14 forms a hydrogen bond to the side chain of S602 in Keap1, similar to that of the main-chain carbonyl group of T80 in Nrf2.<sup>32</sup> In addition, a chloride ion bridges the NH of Ser in 14 with three residues in Keap1, in a similar manner as in the complex with a highly potent inhibitor identified by fragment-based drug discovery.<sup>33</sup> As physiological salt conditions were used in the SPR and ITC assays, as well as in the cellular Nrf2 translocation assay, we believe that formation of this complex also occurs in biologically relevant environments. The ability of macrocycle 14 to adopt a compact and well-defined conformation in the binding site explains why the ring-opened and ring-expanded analogues 16–18 have a significantly lower affinity for Keap1. The solvent exposure of part of Pro and its N-acetyl group is in line with the fact that replacement with hydroxyproline (compound 11) does not affect the inhibitory potency.

The C-terminal dimethylamide of 14 is well-defined by the electron density of the X-ray structure and stacks against Y572, but inspection of the complex does not explain the large increase in potency as compared to ester 9 and amides 12, 13, and 15.



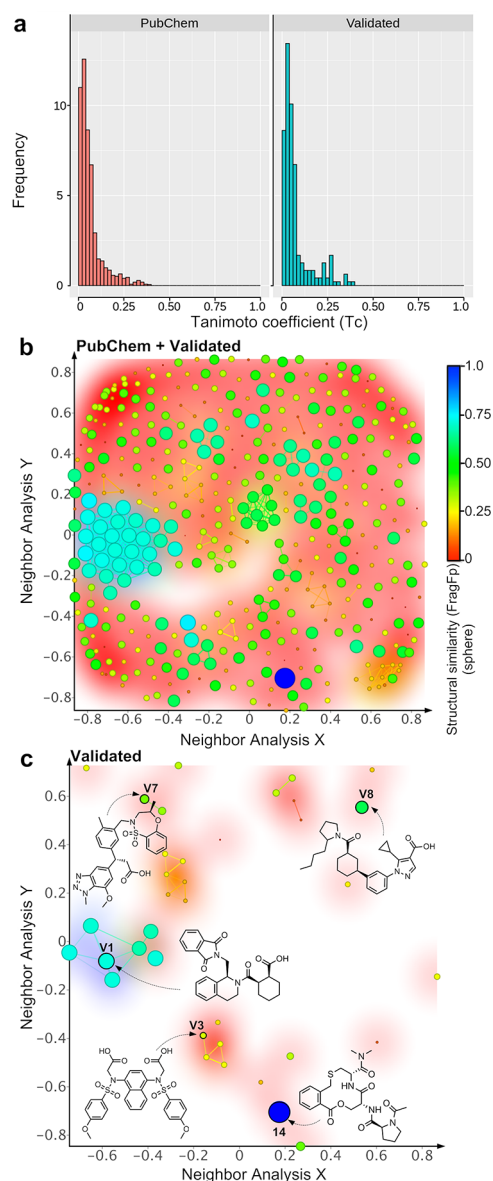
**Figure 3.** Characterization of the Keap1–14 complex. (a) Crystal structure of the complex between macrocycle dimethylamide 14 and Keap1 determined at 2.4 Å resolution (PDB ID: 6Z6A). Residues in Keap1 that make key contacts with 14 are shown in black text. The chloride ion that links R415, N414, and N382 to the NH of the serine moiety of macrocycle 14 is shown as a green sphere. An active site water that mediates formation of the protein–ligand complex is shown as a red sphere. The Kelch channel (dark gray) projects down from the phenylene group of 14. The nonclassical intramolecular hydrogen bonds that stabilize the bound structure of 14 are indicated with dashed orange lines. (b) Comparison of dissociation constants ( $K_D$ ) with ligand binding affinities ( $\Delta G$ ) calculated with Prime MM-GBSA<sup>34</sup> for the binding of 9, 12, 13, 14, and 15 to Keap1 (top). Energy components that contribute to the binding affinities of 9, 12, 13, 14, and 15 in their complexes with Keap1 calculated with Prime MM-GBSA (bottom).<sup>34</sup> Compounds 9, 13, and 15, which are marked with asterisks, may bind to Keap1 with their OMe, NHMe, and NEt<sub>2</sub> groups in different conformations, e.g., the cis- and trans-orientations about the C-terminal amide bond of 13. Mean values for the energy components and standard deviations are shown here; values for each conformation are found in Tables S6–S7. Energy components: Solv GB, generalized Born electrostatic solvation energy; SelfCont, self-contact correction; Covalent, covalent binding energy; Coulomb, Coulomb energy; vdW, van der Waals energy; Lipo, lipophilic energy; Hbond, hydrogen-bonding correction; Packing,  $\pi$ – $\pi$  packing correction.

Ligand binding affinity calculations using Prime MM-GBSA,<sup>34</sup> which is frequently used to estimate the free energy of protein–ligand complexes,<sup>35</sup> provided insights both into the underlying reasons for the potency differences and into what forces stabilize

the complex between **14** and Keap1 (Figure 3b, Tables S6–S7, Figure S10). In agreement with the experimental dissociation constants, macrocycle **14** was predicted to have the highest affinity for Keap1, with **9** and **13** being predicted as intermediate and **12** and **15** as weak. Nonpolar van der Waals and lipophilic interactions were found to be the main contributors to the binding affinity of **14** and to that it had a higher affinity than inhibitors **9** and **13**. According to the calculations, the complex of the inactive **12** is stabilized by stronger polar (Coulomb) interactions, but this is offset by a larger desolvation penalty, while inactive **15** forms the weakest polar (Coulomb) and van der Waals interactions among the five compounds investigated. Interestingly, the distances between the two *N*-methyl groups of **14** and the carbonyl groups of Pro and the *N*-acetyl group show that two intramolecular nonclassical hydrogen bonds<sup>36</sup> are formed between these residues (Figure 3a), an observation that was supported by quantum mechanical calculations (see Supporting Information, Computational Procedure 2). Non-classical hydrogen bonds are weaker than hydrogen bonds,<sup>37</sup> but the two intramolecular bonds are likely to provide additional conformational restriction and stabilization of the Keap1-bound conformation of **14**.

**Comparison of **14** to Reported Keap1 Inhibitors.** We assembled two sets of inhibitors of the Keap1-Nrf2 PPI to generate an overview of the diversity of the Keap1 inhibitor space and to investigate whether macrocycle **14** occupies a unique position in this space. The first set, termed “PubChem”, was obtained from the 528 unique compounds reported as active in the PubChem Bioassay database after testing of >337 000 compounds.<sup>38</sup> Removal of compounds that did not pass a pan assay interference compounds (PAINS) filter<sup>39</sup> from the actives gave the PubChem set consisting of 375 inhibitors (Table S8). A different set of “validated” inhibitors was assembled from two sources. First, nine compounds reported<sup>20</sup> to show reproducible activity in a triad of orthogonal assays after resynthesis were included. Then, compounds bound in the binding site of Keap1 for Nrf2 were retrieved from the Protein Data Bank (PDB). Peptides were excluded from this set, providing a “validated” set of 35 inhibitors, including six fragments that had a MW < 205 Da (Table S9).

The Tanimoto coefficient,<sup>40</sup> used to describe structural diversity, was calculated from seven fingerprints and revealed that **14** had a low similarity to the compounds in both the PubChem and the validated set (Figure 4a and S11). The diversity of the Keap1 inhibitor space was investigated by networklike similarity graphs derived from substructure fragment fingerprints<sup>41</sup> calculated for **14**, the combined validated and PubChem sets, and the validated set alone. The graph for the combined set, just as the one for the validated set, confirmed that macrocycle **14** occupies a unique position in chemical space with no structurally similar neighbors (cf. color and size of spheres and the background, in Figure 4b,c). The overview of the combined set also revealed the Keap1 inhibitor space to be fairly diverse and that most inhibitors have few structural neighbors, with the exception of a large cluster located in the left center of the graph. This cluster contains a series of phthalimides represented by the validated inhibitor<sup>20</sup> **V1** (cf. structure in Figure 4c and Table S9). 1,4-Diaminonaphthalene **V3**, cyclic sulfonamide **V7**, and diazole **V8** (Figure 4c) are the most potent reported Keap1 inhibitors and were obtained after medicinal chemistry optimization. In contrast to **14**, they all contain carboxylic acids. In summary, both Tanimoto coefficients and the networklike similarity graphs highlight the potential of using



**Figure 4.** Structural similarity of inhibitor **14** to reported inhibitors of the Keap1-Nrf2 PPI. Comparisons have been made to a set of inhibitors retrieved from PubChem ( $n = 375$ ) and to a validated set ( $n = 35$ ). (a) Histograms of Tanimoto coefficients<sup>40</sup> (Tc), calculated from seven structural fingerprints, comparing **14** to the PubChem and validated inhibitor sets. Inhibitors that have large structural differences as compared to **14** have Tc values close to 0. (b, c) Network-like similarity graphs illustrating the structural diversity of the Keap1 inhibitor landscape and structural similarities, using **14** as the reference. The graphs were obtained after calculation of FragFp substructure fragment fingerprints<sup>41</sup> for the compounds in the combined PubChem + validated set (panel b) and for the validated set only (panel c). Compounds that are similar to **14** are indicated by spheres that are similar in size and color to **14** (blue sphere). Connecting lines indicate clusters of inhibitors that have similar structures, and the number of similar neighbors is also highlighted with the background color (red, low number; blue, high number). Structures of four inhibitors from the validated set are shown together with compound **14** in panel c.

natural-product-derived cores as a source of structurally unique Keap1 inhibitors, belonging to a chemical space very different from that of previously reported inhibitors.



## DISCUSSION AND CONCLUSIONS

Lead generation is very challenging for targets that have difficult-to-drug binding sites,<sup>2–4</sup> such as protein–protein interactions (PPIs).<sup>42</sup> High-throughput screening (HTS) has often failed to identify useful hits for such targets as screening collections do not provide sufficient coverage of relevant chemical space due to their limited size and history of assembly.<sup>42</sup> Fragment-based lead discovery allows a more comprehensive search of chemical space and has shown success in some cases where HTS has failed.<sup>13</sup> Structure-based docking constitutes an alternative that may be particularly appealing for difficult-to-drug targets, as platforms for docking of ultralarge libraries are now being developed.<sup>43,44</sup>

Difficult-to-drug targets that have flat and featureless binding sites are often modulated by macrocyclic drugs, many of which originate from natural products.<sup>2,8</sup> We therefore mined the macrocycles in the Dictionary of Natural Products to facilitate the identification of novel chemical matter for modulation of difficult-to-drug targets. This provided a smaller set of 41 lead-like macrocyclic cores and a larger set of 217 cores with more complex structures (Supporting Excel Sheet). Docking of the smaller set of cores into the positively charged and polar binding site of Keap1 led to the discovery of the 4  $\mu$ M inhibitor **14**, which originates from the core of cyclothialidine (**1**), after synthesis of only 13 compounds. Macrocycle **14** constitutes an uncharged inhibitor of Keap1, indicating that macrocycles may be used to discover promising starting points for difficult-to-drug targets that have polar binding sites, in addition to targets with flat and featureless sites. In contrast to **14**, the majority of the reported Keap1 inhibitors contain acidic groups, which may be detrimental to cell permeability and oral absorption,<sup>21</sup> and are expected to prevent CNS permeability.<sup>19</sup> Compound **14** has druglike properties and shows cellular potency in an Nrf2 translocation assay, making it suitable for further lead optimization. The structure of its complex with Keap1 illustrates how cation– $\pi$  interactions may allow an uncharged ligand to bind in a charged binding site and provides a first indication of how an uncharged inhibitor can be tailored to fit the charged binding site.

It is interesting to contrast our results to those obtained in a virtual screen of 1.3 billion compounds from Enamine's REAL space library and the ZINC library,<sup>44</sup> as well as with fragment-based lead generation.<sup>33,45</sup> The ultralarge screen identified Keap1 inhibitors with 100 nM potencies as determined by surface plasmon resonance,<sup>44</sup> but which contain potentially reactive or toxic groups. In addition, the structures were similar to previously reported inhibitors and appeared to provide limited opportunities for optimization into drug candidates. Obviously, screening of large libraries provides immense opportunities for drug discovery, but our results reiterate the importance of the structural diversity and quality of the libraries, just as for regular HTS compound collections. One of the most advanced Keap1 inhibitors was obtained by merging information from three fragments that bound in different subsites of the Keap1 binding site.<sup>33,45</sup> While highly potent in *in vitro* ( $K_D$  1.3 nM) and active in cellular and *in vivo* models, the oral bioavailability in rats was low (7%), potentially because of the presence of a carboxylic acid moiety originating from one of the fragments.

Recently, natural products have also been utilized as a source of three-dimensional (3D) fragments<sup>12</sup> and as inspiration for design of compound collections prepared by diversity-oriented synthesis (DOS).<sup>46</sup> In addition, pseudo-natural products have

been designed by combination and fusion of natural-product-derived fragments.<sup>47</sup> However, natural products and their derivatives often have complex structures. We demonstrated how two rounds of docking of our smaller set of lead-like cores first identified the cyclothialidine core and then tailored it to its target Keap1 via docking of **6–9** while simultaneously reducing the structural complexity. In this manner, one of the major shortcomings of natural products, i.e., that they are often prepared via long synthetic routes, was mitigated. Similarly, the larger set of 217 structurally more complex macrocyclic cores may provide inspiration for design of collections having simplified structures that can be prepared through shorter routes than the original natural products. In conclusion, the cores reported herein provide another example of how nature's privileged structures and their diversity can be capitalized on as a rich source of quality leads for drug discovery.<sup>11</sup>

## EXPERIMENTAL SECTION

**Chemistry. General Methods.** All reagents were purchased from Sigma-Aldrich, Fluorochem, and VWR International. DCM, DMSO, hexane, DMF, and acetonitrile were purchased from VWR International, while 1,2-DCE, toluene, and THF were purchased from Sigma-Aldrich. All nonaqueous reactions were performed in oven-dried glassware under an argon atmosphere. The Büchi rotary evaporator R-114 was used to remove solvents *in vacuo*. Reactions were generally monitored by liquid chromatography–mass spectrometry (LC-MS) with an Agilent 1100 series high-performance liquid chromatography (HPLC) with a C18 Atlantis T3 column (3.0 mm  $\times$  50 mm, 5  $\mu$ m) using acetonitrile–water (flow rate 0.75 mL/min over 6 min) as the mobile phase and a Waters micromass ZQ (model code: MM1) mass spectrometer with the electrospray ionization mode as the detector. Alternatively, TLC silica gel 60 F254 plates from VWR International were used and visualization was done using UV light (254 nm) or by staining with a KMnO<sub>4</sub> solution (2% m/v in water). Silica gel (43–63  $\mu$ m, VWR international) was used for purification of compounds with flash column chromatography. Preparative reversed-phase HPLC was performed on a Kromasil C8 column (250 mm  $\times$  21.2 mm, 5  $\mu$ m) on a Gilson HPLC equipped with a Gilson 322 pump, a UV–visible-156 detector, and a 202 collector using acetonitrile–water gradients as eluents with a flow rate of 15 mL/min and detection at 214 or 254 nm. <sup>1</sup>H, <sup>13</sup>C, COSY, HSQC, and HMBC NMR spectra for synthesized compounds were recorded at 298 K on an Agilent Technologies 400 MR spectrometer at 400 or 100 MHz or on an OXFORD AS500 spectrometer at 500 or 126 MHz or on a Bruker Avance III spectrometer at 600 MHz or at 151 MHz. The residual peak of the respective solvent was used as the internal standard [CDCl<sub>3</sub> (CHCl<sub>3</sub>  $\delta$ H 7.26 ppm, CDCl<sub>3</sub>  $\delta$ C 77.0 ppm), DMSO-*d*<sub>6</sub> (CD<sub>2</sub>H<sub>5</sub>SOCD<sub>3</sub>  $\delta$ H 2.50 ppm, CD<sub>3</sub>SOCD<sub>3</sub>  $\delta$ C 39.5 ppm), CD<sub>3</sub>OD (CD<sub>2</sub>HOD  $\delta$ H 3.31 ppm, CD<sub>3</sub>OD  $\delta$ C 49.0 ppm), CD<sub>3</sub>CN (CD<sub>2</sub>H<sub>2</sub>CN  $\delta$ H 1.94 ppm, CD<sub>3</sub>CN  $\delta$ C 1.3 and 118.3 ppm), acetone-*d*<sub>6</sub> (CD<sub>2</sub>HCOCD<sub>3</sub>  $\delta$ H 2.05 ppm, CD<sub>3</sub>COCD<sub>3</sub>  $\delta$ C 29.8 and 206.3 ppm)]. HRMS for all new compounds were recorded in the electrospray ionization (ESI) mode on an S3 LCT Premier connected to a Waters Acquity UPLC I-class with acetonitrile–water used as the mobile phase (1:1, with a flow rate of 0.25 mL/min). The purity of compounds **6–18** is  $\geq$ 95% as determined using a Waters LCT Premiere mass spectrometer coupled to a Waters Acquity UPLC. The Waters Acquity UPLC was equipped with either a BEH C18 column (1.7  $\mu$ m, 2.1 mm  $\times$  50 mm, at 45  $^{\circ}$ C using a gradient from 5 to 90% acetonitrile modified with 40 mM ammonia and 5 mM H<sub>2</sub>CO<sub>3</sub>, pH 10 within 2.5 or 3 min, detection at 210 nm) or a CSH C18 column (1.7  $\mu$ m, 2.1 mm  $\times$  50 mm at 45  $^{\circ}$ C using a gradient from 5 to 90% acetonitrile modified with 10 mM formic acid and 1 mM ammonium formate, pH 3, within 2.5 or 3 min, detection at 230 nm).

**2,2,2-Trichloroethyl 2-(bromomethyl)benzoate (19).** 2-Methylbenzoic acid (8.00 g, 37.2 mmol, 1.0 equiv) was dissolved in DCM (30 mL). EDC-HCl (10.7 g, 55.8 mmol, 1.5 equiv), DMAP (454 mg, 3.72 mmol, 0.1 equiv), and 2,2,2-trichloroethanol (4.65 mL, 48.4 mmol, 1.3 equiv) were then added, and the mixture was stirred at rt for 16 h. The



mixture was diluted with DCM (150 mL) and washed with 1 M aqueous HCl solution (2 × 100 mL) and saturated aqueous NaHCO<sub>3</sub> solution (2 × 100 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then concentrated under reduced pressure. The crude mixture was purified by flash chromatography on a silica gel column using 2–10% EtOAc in hexane as the eluent to give **19** (5.41 g, 15.6 mmol, 42%) as a colorless oil. The <sup>1</sup>H NMR spectrum is in accordance with the literature data.<sup>48</sup>

**2,2,2-Trichloroethyl-(R)-2-(((2-amino-3-methoxy-3-oxopropyl)-thio)methyl)benzoate (20a)**. **2,2,2-Trichloroethyl 2-(bromomethyl)benzoate 19** (3.46 g, 10.0 mmol, 1.0 equiv) was dissolved in DMSO (50 mL), L-cysteine methyl ester hydrochloride (2.06 g, 12.0 mmol, 1.2 equiv) and triethylamine (3.10 mL, 22.0 mmol, 2.2 equiv) were added, and the mixture was stirred for 2 h at rt. The mixture was diluted with EtOAc (250 mL) and washed with saturated aqueous NaHCO<sub>3</sub> solution (2 × 100 mL) and brine (100 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then removed under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **20a** (3.09 g, 7.70 mmol, 77%) as a yellow oil. HRMS (ESI) *m/z* calcd for C<sub>14</sub>H<sub>17</sub>Cl<sub>3</sub>NO<sub>4</sub>S [M + H]<sup>+</sup> 399.9944, found 399.9953. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.08 (dd, *J* = 7.8, 1.4 Hz, 1H), 7.52 (td, *J* = 7.8, 1.4 Hz, 1H), 7.42–7.32 (m, 2H), 4.97 (s, 2H), 4.23 (d, *J* = 13.2 Hz, 1H), 4.16 (d, *J* = 13.2 Hz, 1H), 3.72 (s, 3H), 3.63 (dd, *J* = 7.6, 4.6 Hz, 1H), 2.87 (dd, *J* = 13.6, 4.6 Hz, 1H), 2.68 (dd, *J* = 13.6, 7.6 Hz, 1H), 1.79 (br s, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 174.3, 165.2, 141.0, 132.8, 131.8, 131.3, 127.7, 127.5, 95.0, 74.6, 54.2, 52.2, 36.9, 34.8.

**2,2,2-Trichloroethyl 2-(((4R,7S)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (21a)**. Compound **20a** (1.25 g, 3.13 mmol, 1.0 equiv) was dissolved in acetonitrile (18 mL), Boc-L-Ser-OH (0.96 g, 4.71 mmol, 1.5 equiv) and EDC-HCl (0.89 g, 4.71 mmol, 1.5 equiv) were added, and the reaction was stirred for 1 h at rt. EtOAc (150 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (100 mL), saturated aqueous NaHCO<sub>3</sub> solution (100 mL), and brine (100 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–5% MeOH in DCM as the eluent to give **21a** (1.35 g, 2.31 mmol, 74%) as a colorless foam. HRMS (ESI) *m/z* calcd for C<sub>22</sub>H<sub>30</sub>Cl<sub>3</sub>N<sub>2</sub>O<sub>8</sub>S [M + H]<sup>+</sup> 587.0788, found 578.0794. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.08 (d, *J* = 8.0 Hz, 1H), 7.50 (t, *J* = 8.0 Hz, 1H), 7.40–7.34 (m, 3H), 5.62 (d, *J* = 7.4 Hz, 1H), 4.96 (s, 2H), 4.85–4.75 (m, 1H), 4.29–4.25 (m, 1H), 4.23 (d, *J* = 12.8 Hz, 1H), 4.10 (d, *J* = 12.8 Hz, 1H), 4.03 (dd, *J* = 11.5, 3.6 Hz, 1H), 3.71 (s, 3H), 3.69–3.64 (m, 1H), 3.20 (br s, 1H), 2.94 (dd, *J* = 14.0, 4.8 Hz, 1H), 2.86 (dd, *J* = 14.0, 6.3 Hz, 1H), 1.43 (s, 9H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 171.2, 170.9, 165.2, 155.8, 140.4, 133.0, 131.8, 131.4, 127.7, 127.6, 94.9, 80.9, 74.6, 63.1, 55.3, 52.8, 52.1, 34.8, 33.5, 28.3.

**2-(((4R,7S)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoic Acid (22a)**. Compound **21a** (1.35 g, 2.31 mmol, 1.0 equiv) was dissolved in THF (19 mL) and aqueous 1 M NH<sub>4</sub>OAc (4 mL). Zn dust (1.51 g, 23.1 mmol, 10 equiv) was added, and the reaction was stirred for 2 h at rt. The mixture was diluted with MeOH (250 mL), filtered through a pad of Celite, and concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–20% MeOH/EtOAc as the eluent to give **22a** (642 mg, 1.41 mmol, 61%) as a colorless powder. The <sup>1</sup>H NMR spectrum is in accordance with the literature data.<sup>29</sup>

**Methyl (4R,7S)-7-((tert-Butoxycarbonyl)amino)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (23a)**. Compound **22a** (640 mg, 1.40 mmol, 1.0 equiv) was dissolved in toluene (33 mL) and DMSO (2 mL). Triphenylphosphine (550 mg, 2.10 mmol, 1.5 equiv) and di-*tert*-butyl azodicarboxylate (483 mg, 2.10 mmol, 1.5 equiv) were then added, and the mixture was stirred for 4 h at rt. After evaporation of the solvent, the crude product was purified by flash chromatography on a silica gel column using 40–60% EtOAc in hexane as the eluent to give

**23a** (313 mg, 0.71 mmol, 51%) as a colorless powder. The <sup>1</sup>H NMR spectrum was in accordance with the literature data.<sup>29</sup>

**Methyl-(4R,7S)-7-((S)-1-acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (6)**. Compound **6** (100 mg, 0.21 mmol, 1.0 equiv) was dissolved in 4 M HCl in dioxane (5 mL) and stirred for 1 h at rt. After evaporation of the solvent under reduced pressure, the resulting salt was dissolved in DMSO (4 mL); Ac-L-Pro-OH (70 mg, 0.42 mmol, 2.0 equiv), EDC-HCl (86 mg, 0.42 mmol, 2.0 equiv), and DIPEA (39 μL, 0.21 mmol, 1.0 equiv) were added, and the reaction was stirred for 2 h at rt. The crude product was purified by reverse-phase HPLC using a gradient from 30 to 75% acetonitrile in water to give **6** (43 mg, 0.18 mmol, 40%) as a colorless powder. HRMS (ESI) *m/z* calcd for C<sub>22</sub>H<sub>28</sub>N<sub>3</sub>O<sub>7</sub>S [M + H]<sup>+</sup> 478.1648, found 478.1652. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.73 (dd, *J* = 7.8, 1.4 Hz, 1H), 7.68 (d, *J* = 8.8 Hz, 1H), 7.52–7.42 (m, 2H), 7.37 (dd, *J* = 7.6, 1.4 Hz, 1H), 7.32 (td, *J* = 7.8, 1.4 Hz, 1H), 5.00–4.86 (m, 3H), 4.68–4.55 (m, 1H), 4.51 (dd, *J* = 7.4, 5.3 Hz, 1H), 4.02 (d, *J* = 10.3 Hz, 1H), 3.81 (d, *J* = 10.3 Hz, 1H), 3.75 (s, 3H), 3.73–3.66 (m, 1H), 3.55–3.44 (m, 1H), 3.26 (dd, *J* = 14.5, 4.8 Hz, 1H), 2.96 (dd, *J* = 14.5, 9.7 Hz, 1H), 2.27–2.16 (m, 2H), 2.11–1.94 (m, 2H), 2.03 (s, 3H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 172.0, 171.1, 170.8, 169.6, 168.9, 136.4, 132.3, 131.8, 130.8, 130.7, 127.5, 67.1, 61.1, 53.8, 52.7, 51.6, 48.5, 34.7, 34.4, 29.2, 25.0, 22.8.

**2,2,2-Trichloroethyl (S)-2-(((2-Amino-3-methoxy-3-oxopropyl)-thio)methyl)benzoate (20b)**. Compound **20b** was prepared following the procedure described for the synthesis of compound **20a** using **2,2,2-trichloroethyl 2-(bromomethyl)benzoate 19** (1.70 g, 4.91 mmol, 1.0 equiv), D-cysteine methyl ester hydrochloride (1.01 g, 5.89 mmol, 1.2 equiv), and triethylamine (1.53 mL, 10.8 mmol, 2.2 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **20b** (1.37 g, 3.43 mmol, 70%) as a yellow oil. HRMS (ESI) *m/z* calcd for C<sub>14</sub>H<sub>16</sub>Cl<sub>3</sub>NNaO<sub>4</sub>S [M + Na]<sup>+</sup> 421.9763, found 421.9770. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.07 (dd, *J* = 7.7, 1.5 Hz, 1H), 7.52–7.47 (m, 1H), 7.43–7.28 (m, 2H), 4.95 (s, 2H), 4.21 (d, *J* = 13.2 Hz, 1H), 4.14 (d, *J* = 13.2 Hz, 1H), 3.70 (s, 3H), 3.60 (dd, *J* = 7.6, 4.6 Hz, 1H), 2.85 (dd, *J* = 13.6, 4.6 Hz, 1H), 2.66 (dd, *J* = 13.6, 7.6 Hz, 1H), 1.74 (br s, 2H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 174.3, 165.2, 141.0, 132.8, 131.8, 131.3, 127.7, 127.5, 95.0, 74.6, 54.1, 52.2, 36.9, 34.8.

**2,2,2-Trichloroethyl 2-(((4S,7S)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (21b)**. Compound **21b** was prepared following the procedure described for the synthesis of compound **21a** using **20b** (1.30 g, 3.25 mmol, 1.0 equiv), Boc-L-Ser-OH (990 mg, 4.87 mmol, 1.5 equiv), and EDC-HCl (932 mg, 4.88 mmol, 1.5 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–5% MeOH in DCM as the eluent to give **21b** (1.45 g, 2.47 mmol, 76%) as a colorless foam. HRMS (ESI) *m/z* calcd for C<sub>22</sub>H<sub>30</sub>Cl<sub>3</sub>N<sub>2</sub>O<sub>8</sub>S [M + H]<sup>+</sup> 587.0788, found 587.0773. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 8.08 (dd, *J* = 8.0, 1.4 Hz, 1H), 7.50 (td, *J* = 8.0, 1.4 Hz, 1H), 7.40–7.34 (m, 2H), 7.36–7.31 (m, 1H), 5.62 (d, *J* = 7.4 Hz, 1H), 4.99 (d, *J* = 12.0 Hz, 1H), 4.96 (d, *J* = 12.0 Hz, 1H), 4.85–4.75 (m, 1H), 4.27–4.18 (m, 2H), 4.10 (d, *J* = 12.8 Hz, 1H), 4.09–4.04 (m, 1H), 3.71 (s, 3H), 3.68 (dd, *J* = 11.6, 4.9 Hz, 1H), 2.97–2.84 (m, 2H), 2.38 (br s, 1H), 1.43 (s, 9H). <sup>13</sup>C NMR (101 MHz, CDCl<sub>3</sub>) δ 171.5, 171.0, 165.2, 156.0, 140.4, 133.0, 131.8, 131.4, 127.7, 127.6, 94.9, 80.5, 74.6, 63.1, 55.3, 52.8, 52.0, 34.8, 33.4, 28.3.

**2-(((4S,7S)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoic Acid (22b)**. Compound **22b** was prepared following the procedure described for the synthesis of compound **22a** using **21b** (1.20 g, 2.04 mmol, 1.0 equiv) and Zn dust (1.33 g, 20.4 mmol, 10 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–20% MeOH in EtOAc as the eluent to give **22b** (567 mg, 1.24 mmol, 61%) as a colorless powder. HRMS (ESI) *m/z* calcd for C<sub>20</sub>H<sub>29</sub>N<sub>2</sub>O<sub>8</sub>S [M + H]<sup>+</sup> 457.1645, found 457.1637. <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 12.93 (br s, 1H), 8.27 (d, *J* = 7.8 Hz, 1H), 7.82 (dd, *J* = 8.0, 1.7 Hz, 1H), 7.46 (td, *J* = 7.6, 1.7 Hz, 1H), 7.41–7.29 (m, 2H), 6.63 (d, *J* = 8.1 Hz, 1H), 4.79 (br s, 1H), 4.55–4.42 (m, 1H), 4.07 (s,

2H), 4.06–4.03 (m, 1H), 3.60 (s, 3H), 3.59–3.41 (m, 2H), 2.74 (dd,  $J = 13.8, 5.8$  Hz, 1H), 2.66 (dd,  $J = 13.8, 7.6$  Hz, 1H), 1.35 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz, DMSO- $d_6$ )  $\delta$  171.4, 170.9, 168.8, 155.6, 140.3, 132.0, 131.4, 131.3, 130.5, 127.6, 78.6, 62.3, 57.1, 52.5, 52.5, 34.1, 32.8, 28.6.

**Methyl-(4S,7S)-7-((tert-butoxycarbonyl)amino)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (23b).** Compound 23b was prepared following the procedure described for the synthesis of compound 23a using 22b (350 mg, 0.76 mmol, 1.0 equiv), triphenylphosphine (300 mg, 1.14 mmol, 1.5 equiv), and di-*tert*-butyl azodicarboxylate (263 mg, 1.14 mmol, 1.5 equiv). The crude product was purified by flash chromatography on a silica gel column using 40–60% EtOAc in hexane as the eluent to give 23b (146 mg, 0.33 mmol, 44%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{26}\text{N}_2\text{NaO}_7\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  461.1358, found 461.1356.  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  9.01 (d,  $J = 9.3$  Hz, 1H), 7.74 (d,  $J = 7.8$  Hz, 1H), 7.51 (td,  $J = 7.8, 1.5$  Hz, 1H), 7.45–7.33 (m, 2H), 7.12 (d,  $J = 8.3$  Hz, 1H), 4.77 (td,  $J = 10.4, 4.0$  Hz, 1H), 4.60–4.52 (m, 1H), 4.50–4.35 (m, 2H), 4.29 (dd,  $J = 10.7, 4.6$  Hz, 1H), 3.84 (d,  $J = 10.3$  Hz, 1H), 3.66 (s, 3H), 3.08 (dd,  $J = 14.5, 4.0$  Hz, 1H), 2.70 (dd,  $J = 14.5, 10.9$  Hz, 1H), 1.37 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz, DMSO- $d_6$ )  $\delta$  171.4, 170.1, 167.6, 155.3, 137.3, 132.9, 132.4, 131.6, 130.4, 128.1, 78.9, 65.8, 52.7, 52.6, 52.3, 35.9, 33.9, 28.6.

**Methyl-(4S,7S)-7-((S)-1-acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (7).** Compound 7 was prepared following the procedure described for the synthesis of compound 6 using 23b (100 mg, 0.22 mmol, 1.0 equiv), Ac-L-Pro-OH (69 mg, 0.44 mmol, 2.0 equiv), EDC-HCl (84 mg, 0.44 mmol, 2.0 equiv), and DIPEA (39  $\mu\text{L}$ , 0.22 mmol, 1.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 30 to 75% acetonitrile in water to give 7 (62 mg, 0.13 mmol, 59%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{28}\text{N}_3\text{O}_7\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  478.1648, found 478.1651.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.79 (dd,  $J = 7.7, 1.4$  Hz, 1H), 7.61 (d,  $J = 7.7$  Hz, 1H), 7.43 (td,  $J = 7.5, 1.4$  Hz, 1H), 7.34–7.26 (m, 2H), 7.20 (d,  $J = 7.5$  Hz, 1H), 4.83 (td,  $J = 7.3, 4.6$  Hz, 1H), 4.69–4.58 (m, 2H), 4.56–4.48 (m, 2H), 4.15 (d,  $J = 11.1$  Hz, 1H), 4.06 (d,  $J = 11.1$  Hz, 1H), 3.74 (s, 3H), 3.71–3.61 (m, 1H), 3.52–3.40 (m, 1H), 3.21 (dd,  $J = 14.3, 4.8$  Hz, 1H), 3.14 (dd,  $J = 14.3, 7.1$  Hz, 1H), 2.33–2.21 (m, 1H), 2.13–1.96 (m, 3H), 2.06 (s, 3H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.2, 171.0, 170.3, 168.6, 168.6, 137.5, 132.3, 131.2, 131.1, 130.1, 127.6, 65.0, 60.2, 52.8, 52.5, 48.3, 36.3, 34.5, 28.4, 25.0, 22.4.

**2,2,2-Trichloroethyl 2-((4R,7R)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (21c).** Compound 21c was prepared following the procedure described for the synthesis of compound 21a using 20a (2.01 g, 5.81 mmol, 1.0 equiv), Boc-D-Ser-OH (1.78 g, 8.72 mmol, 1.5 equiv), and EDC-HCl (1.64 g, 8.72 mmol, 1.5 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–5% MeOH in DCM as the eluent to give 21c (2.55 g, 4.34 mmol, 75%) as a colorless oil. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{29}\text{Cl}_3\text{N}_2\text{NaO}_8\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  609.0608, found 609.0612.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.07 (dd,  $J = 8.3, 1.4$  Hz, 1H), 7.53–7.48 (m, 1H), 7.45–7.38 (m, 1H), 7.39–7.33 (m, 2H), 5.67 (d,  $J = 7.3$  Hz, 1H), 5.00–4.92 (m, 2H), 4.78 (ddd,  $J = 8.3, 6.4, 5.0$  Hz, 1H), 4.23 (d,  $J = 13.1$  Hz, 1H), 4.11 (d,  $J = 13.0$  Hz, 1H), 4.08–4.01 (m, 1H), 3.71 (s, 3H), 3.68–3.65 (m, 1H), 3.29–3.22 (m, 1H), 2.91 (dd,  $J = 14.0, 5.0$  Hz, 1H), 2.86 (dd,  $J = 14.0, 6.4$  Hz, 1H), 2.10 (br s, 1H), 1.42 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.4, 171.0, 165.2, 156.0, 140.5, 133.0, 131.8, 131.4, 127.7, 127.6, 94.9, 80.5, 74.6, 63.0, 55.5, 52.8, 52.0, 34.7, 33.4, 28.3.

**2-((4R,7R)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoic Acid (22c).** Compound 22c was prepared following the procedure described for the synthesis of compound 22a using 21c (1.58 g, 2.69 mmol, 1.0 equiv) and Zn dust (1.75 g, 26.9 mmol, 10 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–15% MeOH/EtOAc as the eluent to give 22c (834 mg, 1.83 mmol, 68%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{29}\text{N}_2\text{O}_8\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  457.1645, found 457.1633.  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  12.86 (br s, 1H), 8.26 (d,  $J = 7.9$  Hz, 1H), 7.81 (dd,

$J = 8.2, 1.5$  Hz, 1H), 7.45 (td,  $J = 7.5, 1.5$  Hz, 1H), 7.37–7.26 (m, 2H), 6.62 (d,  $J = 8.3$  Hz, 1H), 4.50–4.43 (m, 1H), 4.07 (s, 2H), 4.04–3.99 (m, 1H), 3.59 (s, 3H), 3.58–3.43 (m, 2H), 2.74 (dd,  $J = 13.8, 5.8$  Hz, 1H), 2.65 (dd,  $J = 13.8, 7.5$  Hz, 1H), 1.34 (s, 9H). The  $\text{CH}_2\text{OH}$  proton was not detectable in this spectrum.  $^{13}\text{C}$  NMR (101 MHz, DMSO- $d_6$ )  $\delta$  171.4, 170.9, 168.8, 155.6, 140.3, 132.0, 131.4, 131.3, 130.6, 127.6, 79.6, 78.6, 62.3, 57.1, 52.5, 34.1, 32.8, 28.6.

**Methyl-(4R,7R)-7-((tert-butoxycarbonyl)amino)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (23c).** Compound 23c was prepared following the procedure described for the synthesis of compound 23a using 22c (500 mg, 1.09 mmol, 1.0 equiv), triphenylphosphine (429 mg, 1.64 mmol, 1.5 equiv), and di-*tert*-butyl azodicarboxylate (377 mg, 1.64 mmol, 1.5 equiv). The crude product was purified by flash chromatography on a silica gel column using 40–60% EtOAc in hexane as the eluent to give 23c (253 mg, 0.57 mmol, 53%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{26}\text{N}_2\text{NaO}_7\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  461.1358, found 461.1349.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.79 (d,  $J = 7.5$  Hz, 1H), 7.42 (td,  $J = 7.5, 1.3$  Hz, 1H), 7.36–7.22 (m, 2H), 6.96 (d,  $J = 7.0$  Hz, 1H), 5.51 (d,  $J = 6.5$  Hz, 1H), 4.92–4.85 (m, 1H), 4.72–4.60 (m, 2H), 4.47–4.38 (m, 1H), 4.14–4.02 (m, 2H), 3.75 (s, 3H), 3.15 (dd,  $J = 14.4, 4.7$  Hz, 1H), 3.05 (dd,  $J = 14.4, 5.9$  Hz, 1H), 1.45 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  170.4, 169.1, 167.8, 154.9, 137.2, 132.2, 131.3, 131.0, 130.2, 127.7, 80.4, 65.3, 52.9, 52.6, 51.9, 35.9, 34.8, 28.3.

**Methyl-(4R,7R)-7-((S)-1-acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (8).** Compound 8 was prepared following the procedure described for the synthesis of compound 6 using 23c (95 mg, 0.21 mmol, 1.0 equiv), Ac-L-Pro-OH (66 mg, 0.42 mmol, 2.0 equiv), EDC-HCl (80 mg, 0.42 mmol, 2.0 equiv), and DIPEA (47  $\mu\text{L}$ , 0.27 mmol, 1.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 30 to 75% acetonitrile in water to give 8 (45 mg, 94  $\mu\text{mol}$ , 45%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{28}\text{N}_3\text{O}_7\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  478.1648, found 478.1648.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.78 (dd,  $J = 7.7, 1.5$  Hz, 1H), 7.60 (d,  $J = 7.0$  Hz, 1H), 7.42 (td,  $J = 7.5, 1.4$  Hz, 1H), 7.34–7.27 (m, 3H), 4.87 (ddd,  $J = 7.4, 6.6, 4.5$  Hz, 1H), 4.75–4.64 (m, 2H), 4.60–4.42 (m, 2H), 4.18 (s, 2H), 3.75 (s, 3H), 3.64–3.58 (m, 1H), 3.55–3.37 (m, 1H), 3.19 (dd,  $J = 14.3, 4.5$  Hz, 1H), 3.12 (dd,  $J = 14.3, 6.6$  Hz, 1H), 2.34–2.28 (m, 1H), 2.20–2.06 (m, 1H), 2.10 (s, 3H), 2.04–1.91 (m, 2H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.4, 170.8, 170.5, 168.7, 168.0, 137.4, 132.2, 131.2, 131.1, 130.3, 127.6, 65.0, 59.9, 52.8, 52.4, 52.0, 48.3, 36.3, 34.6, 28.3, 25.0, 22.5.

**2,2,2-Trichloroethyl 2-((4S,7R)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (21d).** Compound 21d was prepared following the procedure described for the synthesis of compound 21a using 20b (1.30 g, 3.25 mmol, 1.0 equiv), Boc-D-Ser-OH (990 mg, 4.87 mmol, 1.5 equiv), and EDC-HCl (932 mg, 4.88 mmol, 1.5 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–5% MeOH in DCM as the eluent to give 21d (1.37 g, 2.34 mmol, 72%) as a colorless oil. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{29}\text{Cl}_3\text{N}_2\text{O}_8\text{SNa}$  [ $\text{M} + \text{Na}$ ] $^+$  609.0608, found 609.0594.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.09 (dd,  $J = 7.7, 1.5$  Hz, 1H), 7.52 (td,  $J = 7.5, 1.4$  Hz, 1H), 7.41–7.36 (m, 2H), 7.30 (d,  $J = 7.0$  Hz, 1H), 5.57 (d,  $J = 7.5$  Hz, 1H), 4.98 (s, 2H), 4.83–4.77 (m, 1H), 4.31–4.20 (m, 2H), 4.10 (d,  $J = 12.8$  Hz, 1H), 4.08–4.01 (m, 1H), 3.73 (s, 3H), 3.68 (dd,  $J = 11.4, 5.2$  Hz, 1H), 2.96 (dd,  $J = 14.1, 4.7$  Hz, 1H), 2.88 (dd,  $J = 14.1, 6.2$  Hz, 1H), 2.79 (br s, 1H), 1.44 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.2, 170.9, 165.3, 155.8, 140.4, 133.0, 131.8, 131.4, 127.7, 127.6, 94.9, 80.5, 74.7, 63.3, 55.2, 52.8, 52.0, 34.9, 33.6, 28.3.

**2-((4S,7R)-7-(Hydroxymethyl)-4-(methoxycarbonyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoic Acid (22d).** Compound 22d was prepared following the procedure described for the synthesis of compound 22a using 21d (1.05 g, 1.78 mmol, 1.0 equiv) and Zn dust (1.16 g, 17.8 mmol, 10 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–20% MeOH/EtOAc as the eluent to give 22d (478 mg, 1.05 mmol, 59%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{29}\text{N}_2\text{O}_8\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  457.1645, found 457.1635.  $^1\text{H}$  NMR (400



MHz, DMSO- $d_6$ )  $\delta$  8.25 (d,  $J$  = 7.7 Hz, 1H), 7.82 (d,  $J$  = 7.7 Hz, 1H), 7.37 (t,  $J$  = 7.7 Hz, 1H), 7.33–7.19 (m, 2H), 6.69 (d,  $J$  = 7.9 Hz, 1H), 4.53–4.38 (m, 1H), 4.18 (d,  $J$  = 13.0 Hz, 1H), 4.13 (d,  $J$  = 13.0 Hz, 1H), 4.09–3.96 (m, 1H), 3.58 (s, 3H), 3.56–3.41 (m, 2H), 2.77 (dd,  $J$  = 13.8, 5.8 Hz, 1H), 2.68 (dd,  $J$  = 13.8, 7.2 Hz, 1H), 1.36 (s, 9H). The OH protons were not detectable in this spectrum.  $^{13}\text{C}$  NMR (126 MHz, DMSO- $d_6$ )  $\delta$  171.4, 171.0, 168.8, 155.7, 140.4, 132.0, 131.5, 131.4, 130.6, 127.7, 78.7, 62.2, 57.2, 52.6, 52.6, 34.1, 32.7, 28.6.

**Methyl-(4S,7R)-7-((tert-butoxycarbonyl)amino)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (23d).** Compound 23d was prepared following the procedure described for the synthesis of compound 23a using 22d (401 mg, 0.87 mmol, 1.0 equiv), triphenylphosphine (344 mg, 1.32 mmol, 1.5 equiv), and di-*tert*-butyl azodicarboxylate (303 mg, 1.32 mmol, 1.5 equiv). The crude product was purified by flash chromatography on a silica gel column using 40–65% EtOAc in hexane as the eluent to give 23d (186 mg, 0.43 mmol, 49%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{26}\text{N}_3\text{NaO}_7\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  461.1358, found 461.1360.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.82 (d,  $J$  = 7.4 Hz, 1H), 7.45 (t,  $J$  = 7.5 Hz, 1H), 7.35–7.29 (m, 3H), 5.75–5.62 (m, 1H), 4.96 (dd,  $J$  = 11.4, 3.0 Hz, 1H), 4.89–4.79 (m, 1H), 4.60–4.56 (m, 2H), 4.10–4.00 (m, 2H), 3.74 (s, 3H), 3.30–3.25 (m, 1H), 3.20–3.12 (m, 1H), 1.47 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  170.4, 169.2, 168.5, 155.0, 137.3, 132.5, 131.4, 131.4, 129.8, 127.6, 81.1, 66.7, 55.3, 52.8, 52.1, 36.3, 34.9, 28.2.

**Methyl-(4S,7R)-7-((S)-1-acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (9).** Compound 9 was prepared following the procedure described for the synthesis of compound 6 using 23d (120 mg, 0.27 mmol, 1.0 equiv), Ac-L-Pro-OH (84 mg, 0.54 mmol, 2.0 equiv), EDC-HCl (103 mg, 0.54 mmol, 2.0 equiv), and DIPEA (47  $\mu\text{L}$ , 0.27 mmol, 1.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 30 to 75% acetonitrile in water to give 9 (62 mg, 0.13 mmol, 48%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{26}\text{N}_3\text{O}_7\text{S}$  [ $\text{M} - \text{H}$ ] $^-$  476.1491, found 476.1503.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.89 (dd,  $J$  = 7.7, 1.2 Hz, 1H), 7.78 (d,  $J$  = 8.8 Hz, 1H), 7.71 (d,  $J$  = 8.8 Hz, 1H), 7.50–7.42 (m, 1H), 7.39 (dd,  $J$  = 7.7, 1.2 Hz, 1H), 7.32 (td,  $J$  = 7.7, 1.5 Hz, 1H), 5.09 (dd,  $J$  = 11.1, 2.9 Hz, 1H), 4.91 (td,  $J$  = 8.9, 4.5 Hz, 1H), 4.88–4.83 (m, 1H), 4.50–4.37 (m, 2H), 4.30 (d,  $J$  = 10.5 Hz, 1H), 3.74–3.66 (m, 1H), 3.92 (d,  $J$  = 10.5 Hz, 1H), 3.71 (s, 3H), 3.53–3.41 (m, 1H), 3.21 (dd,  $J$  = 14.6, 4.5 Hz, 1H), 2.94 (dd,  $J$  = 14.6, 9.0 Hz, 1H), 2.41–2.20 (m, 2H), 2.08 (s, 3H), 2.05–1.87 (m, 2H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.6, 171.5, 170.8, 168.9, 167.9, 137.2, 132.5, 132.2, 131.8, 129.8, 127.5, 66.9, 60.2, 53.4, 52.5, 52.4, 48.4, 36.2, 34.8, 28.0, 25.3, 22.6.

**Methyl-(4S,7R)-7-((R)-1-acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxylate (10).** Compound 10 was prepared following the procedure described for the synthesis of compound 6 using 23d (27 mg, 61  $\mu\text{mol}$ , 1.0 equiv), Ac-D-Pro-OH (19 mg, 0.12 mmol, 2.0 equiv), EDC-HCl (23 mg, 0.12 mmol, 2.0 equiv), and DIPEA (21  $\mu\text{L}$ , 0.12 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 30 to 75% acetonitrile in water to give 10 (16 mg, 33  $\mu\text{mol}$ , 54%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{28}\text{N}_3\text{O}_7\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  478.1648, found 478.1663.  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{CN}$ )  $\delta$  8.07 (d,  $J$  = 9.0 Hz, 1H), 7.96 (dd,  $J$  = 7.8, 1.2 Hz, 1H), 7.55 (td,  $J$  = 7.8, 1.2 Hz, 1H), 7.46 (dd,  $J$  = 7.8, 1.4 Hz, 1H), 7.42 (td,  $J$  = 7.8, 1.4 Hz, 1H), 7.30 (d,  $J$  = 7.4 Hz, 1H), 4.92 (dd,  $J$  = 11.3, 3.0 Hz, 1H), 4.84 (ddd,  $J$  = 11.7, 9.0, 4.0 Hz, 1H), 4.68–4.62 (m, 1H), 4.40 (dd,  $J$  = 11.4, 2.0 Hz, 1H), 4.35 (d,  $J$  = 9.3 Hz, 1H), 4.28–4.23 (m, 1H), 3.86 (d,  $J$  = 9.3 Hz, 1H), 3.71 (s, 3H), 3.55–3.43 (m, 1H), 3.25 (dd,  $J$  = 14.6, 4.0 Hz, 1H), 2.91 (dd,  $J$  = 14.6, 11.7 Hz, 1H), 2.30–2.22 (m, 1H), 2.02 (s, 3H), 2.01–1.94 (m, 4H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  172.0, 171.1, 170.7, 169.6, 168.9, 136.4, 132.3, 131.8, 130.8, 130.7, 127.5, 67.1, 61.1, 53.8, 52.7, 51.6, 48.6, 34.7, 34.4, 29.3, 25.0, 22.8.

**Methyl-(4S,7R)-7-((2S,4R)-1-acetyl-4-hydroxypyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]azacyclododecine-4-carboxylate (11).** Compound 11 was prepared following the procedure described for the synthesis of compound 6 using 23d (39 mg, 89  $\mu\text{mol}$ , 1.0 equiv), 4-*trans*-hydroxy-

Ac-L-Pro-OH (31 mg, 0.18 mmol, 2.0 equiv), EDC-HCl (35 mg, 0.18 mmol, 2.0 equiv), and DIPEA (31  $\mu\text{L}$ , 0.18 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give 11 (14 mg, 28  $\mu\text{mol}$ , 32%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{27}\text{N}_3\text{NaO}_8\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  516.1417, found 516.1407.  $^1\text{H}$  NMR (400 MHz,  $\text{CD}_3\text{CN}$ )  $\delta$  8.47 (d,  $J$  = 9.7 Hz, 1H), 7.98 (dd,  $J$  = 7.7, 1.4 Hz, 1H), 7.69 (d,  $J$  = 7.3 Hz, 1H), 7.56 (td,  $J$  = 7.5, 1.4 Hz, 1H), 7.48 (dd,  $J$  = 7.5, 1.4 Hz, 1H), 7.41 (td,  $J$  = 7.7, 1.4 Hz, 1H), 5.02 (dd,  $J$  = 11.1, 2.5 Hz, 1H), 4.84 (ddd,  $J$  = 11.7, 9.7, 3.9 Hz, 1H), 4.77 (d,  $J$  = 9.4 Hz, 1H), 4.57–4.52 (m, 1H), 4.52–4.48 (m, 1H), 4.46–4.40 (m, 1H), 4.38 (dd,  $J$  = 11.2, 2.2 Hz, 1H), 3.85 (d,  $J$  = 9.4 Hz, 1H), 3.75–3.68 (m, 1H), 3.66 (s, 3H), 3.48 (dt,  $J$  = 11.1, 1.7 Hz, 1H), 3.24 (dd,  $J$  = 14.8, 3.9 Hz, 1H), 2.95 (dd,  $J$  = 14.8, 11.7 Hz, 1H), 2.16 (ddd,  $J$  = 13.1, 8.6, 4.5 Hz, 2H), 2.10–2.02 (m, 1H), 2.00 (s, 3H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CD}_3\text{CN}$ )  $\delta$  172.6, 170.7, 170.7, 169.3, 167.1, 137.5, 132.8, 132.6, 132.0, 129.8, 127.7, 69.8, 66.0, 58.9, 56.4, 53.7, 53.4, 51.9, 37.3, 37.2, 34.5, 21.9.

**tert-Butyl-((4S,7R)-4-carbamoyl-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]azacyclododecine-7-yl)-carbamate (24).** Compound 23d (84 mg, 0.19 mmol, 1.0 equiv) was dissolved in 1,2-dichloroethane (2 mL);  $\text{Me}_3\text{SnOH}$  (143 mg, 0.76 mmol, 4.0 equiv) was added, and the mixture was stirred for 45 min at 83 °C. The reaction was allowed to cool to rt, acidified with 1 M aqueous HCl solution (5 mL), and extracted with DCM (3  $\times$  25 mL). The combined organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure, and the obtained crude mixture was dissolved in DMF (5 mL).  $\text{HOBt}\cdot x\text{H}_2\text{O}$  (39 mg, 0.29 mmol, 1.5 equiv), EDC-HCl (72 mg, 0.38 mmol, 2.0 equiv), and ammonia (7 M in MeOH, 54  $\mu\text{L}$ , 0.38 mmol, 2.0 equiv) were added, and the mixture was stirred for 2 h at rt. EtOAc (50 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (25 mL), saturated aqueous  $\text{NaHCO}_3$  solution (25 mL), and brine (25 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The crude product was purified by reverse-phase HPLC using a gradient from 35 to 70% acetonitrile in water to give 24 (39 mg, 91  $\mu\text{mol}$ , 48%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{25}\text{N}_3\text{NaO}_6\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  446.1362, found 446.1377.  $^1\text{H}$  NMR (500 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  7.93 (d,  $J$  = 7.8 Hz, 1H), 7.50 (t,  $J$  = 7.5 Hz, 1H), 7.42 (d,  $J$  = 7.5 Hz, 1H), 7.36 (t,  $J$  = 7.8 Hz, 1H), 5.02–4.96 (m, 1H), 4.73 (dd,  $J$  = 10.5, 4.5 Hz, 1H), 4.60 (d,  $J$  = 10.0 Hz, 1H), 4.42 (d,  $J$  = 10.0 Hz, 1H), 4.40–4.36 (m, 1H), 3.83 (d,  $J$  = 9.7 Hz, 1H), 3.29–3.23 (m, 1H), 2.94–2.85 (dd,  $J$  = 14.6, 10.8 Hz, 1H), 1.48 (s, 9H). The NH protons were not detectable in this spectrum.  $^{13}\text{C}$  NMR (126 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  173.6, 171.4, 167.6, 157.2, 137.5, 132.3, 132.1, 131.7, 129.5, 127.2, 80.3, 65.7, 55.7, 53.7, 36.8, 34.0, 27.2.

**(4S,7R)-7-((S)-1-Acetylpyrrolidine-2-carboxamido)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-4-carboxamide (12).** Compound 12 was prepared following the procedure described for the synthesis of compound 6 using 24 (26 mg, 62  $\mu\text{mol}$ , 1.0 equiv), Ac-L-Pro-OH (19 mg, 0.12 mmol, 2.0 equiv), EDC-HCl (23 mg, 0.12 mmol, 2.0 equiv), and DIPEA (20  $\mu\text{L}$ , 0.12 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give 12 (14 mg, 32  $\mu\text{mol}$ , 52%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{21}\text{H}_{26}\text{N}_4\text{O}_6\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  463.1651, found 463.1647.  $^1\text{H}$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  9.39 (d,  $J$  = 5.1 Hz, 1H), 8.13 (d,  $J$  = 9.7 Hz, 1H), 7.92 (d,  $J$  = 7.5 Hz, 1H), 7.54 (t,  $J$  = 7.3 Hz, 1H), 7.46 (d,  $J$  = 7.3 Hz, 1H), 7.40 (t,  $J$  = 7.5 Hz, 1H), 7.26 (br s, 1H), 6.48 (br s, 1H), 4.80–4.74 (m, 2H), 4.45–4.36 (m, 2H), 4.36–4.30 (m, 2H), 3.79 (d,  $J$  = 9.2 Hz, 1H), 3.63–3.53 (m, 1H), 3.52–3.45 (m, 1H), 3.17 (dd,  $J$  = 14.7, 3.8 Hz, 1H), 2.77 (dd,  $J$  = 14.7, 12.1 Hz, 1H), 2.11–2.01 (m, 2H), 1.95 (s, 3H), 1.90–1.81 (m, 2H).  $^{13}\text{C}$  NMR (101 MHz, DMSO- $d_6$ )  $\delta$  174.9, 172.4, 170.4, 169.0, 167.1, 137.8, 133.3, 133.1, 132.3, 129.6, 128.1, 65.8, 59.7, 55.5, 54.5, 48.5, 37.0, 34.5, 29.2, 25.3, 22.7.

**tert-Butyl-((4S,7R)-4-(methylcarbamoyl)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[j][1]oxa[8]thia[5]-azacyclododecine-7-yl)carbamate (25).** Compound 25 was prepared following the procedure described for the synthesis of compound 24 using 23d (136 mg, 0.31 mmol, 1.0 equiv),  $\text{Me}_3\text{SnOH}$  (233 mg, 1.24 mmol, 4.0 equiv),  $\text{HOBt}\cdot x\text{H}_2\text{O}$  (63 mg, 0.47 mmol, 1.5 equiv), EDC-



HCl (118 mg, 0.62 mmol, 2.0 equiv), and methylamine (2 M in THF, 171  $\mu$ L, 0.34 mmol, 1.1 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–5% MeOH in DCM as the eluent to give **25** (78 mg, 0.18 mmol, 58%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{20}H_{28}N_3O_6S$   $[M + H]^+$  438.1699, found 438.1687.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.83 (d,  $J = 7.8$  Hz, 1H), 7.48 (t,  $J = 7.5$  Hz, 1H), 7.42–7.31 (m, 2H), 7.19 (br s, 1H), 6.58 (br s, 1H), 5.58 (br s, 1H), 4.90–4.84 (m, 1H), 4.71–4.57 (m, 2H), 4.56–4.49 (m, 1H), 4.01 (d,  $J = 9.8$  Hz, 1H), 3.80 (d,  $J = 9.8$  Hz, 1H), 3.37 (dd,  $J = 14.8, 5.3$  Hz, 1H), 2.97 (dd,  $J = 14.8, 9.8$  Hz, 1H), 2.80 (d,  $J = 4.1$  Hz, 3H), 1.46 (s, 9H).  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  170.6, 170.4, 168.4, 156.4, 136.2, 132.7, 132.1, 131.6, 129.9, 127.9, 81.7, 66.2, 55.9, 50.9, 34.4, 33.6, 28.2, 26.4.

(4*S*,7*R*)-7-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-*N*-methyl-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[*j*][1]oxa[8]thia[5]-azacyclododecine-4-carboxamide (**13**). Compound **13** was prepared following the procedure described for the synthesis of compound **6** using **25** (50 mg, 0.11 mmol, 1.0 equiv), Ac-L-Pro-OH (34 mg, 0.22 mmol, 2.0 equiv), EDC·HCl (42 mg, 0.22 mmol, 2.0 equiv), and DIPEA (19  $\mu$ L, 0.11 mmol, 1.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give **13** (17 mg, 37  $\mu$ mol, 34%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{22}H_{29}N_4O_6S$   $[M + H]^+$  477.1808, found 477.1802.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  8.23 (d,  $J = 9.1$  Hz, 1H), 7.82 (d,  $J = 7.5$  Hz, 1H), 7.45 (t,  $J = 7.8$  Hz, 1H), 7.39 (d,  $J = 7.8$  Hz, 1H), 7.32 (t,  $J = 7.5$  Hz, 1H), 7.23 (d,  $J = 7.1$  Hz, 1H), 6.58 (br s, 1H), 5.08–5.01 (m, 1H), 4.77–4.67 (m, 2H), 4.59–4.53 (m, 1H), 4.21–4.13 (m, 1H), 3.99 (d,  $J = 9.5$  Hz, 1H), 3.94 (d,  $J = 9.5$  Hz, 1H), 3.68–3.59 (m, 1H), 3.59–3.51 (m, 2H), 2.91–2.81 (m, 1H), 2.76 (d,  $J = 4.0$  Hz, 3H), 2.32–2.09 (m, 4H), 2.00 (s, 3H).  $^{13}C$  NMR (101 MHz,  $CDCl_3$ )  $\delta$  173.4, 171.1, 170.4, 169.4, 168.5, 136.4, 132.5, 132.4, 131.4, 130.1, 127.6, 66.0, 61.0, 54.5, 53.3, 48.6, 34.8, 34.6, 29.2, 26.3, 25.4, 22.4.

*tert*-Butyl-((4*S*,7*R*)-4-(dimethylcarbamoyl)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[*j*][1]oxa[8]thia[5]-azacyclododecine-7-yl)carbamate (**26**). Compound **26** was prepared following the procedure described for the synthesis of compound **24**, using **23d** (82 mg, 0.19 mmol, 1.0 equiv),  $Me_3SnOH$  (143 mg, 0.76 mmol, 4.0 equiv),  $HOBt \cdot xH_2O$  (38 mg, 0.28 mmol, 1.5 equiv), EDC·HCl (72 mg, 0.38 mmol, 2.0 equiv), and dimethylamine (2 M in THF, 105  $\mu$ L, 0.21 mmol, 1.1 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 75% acetonitrile in water to give **26** (40 mg, 89  $\mu$ mol, 47%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{21}H_{29}N_3NaO_6S$   $[M + Na]^+$  474.1675, found 474.1667.  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  7.86 (dd,  $J = 7.8, 1.4$  Hz, 1H), 7.47 (td,  $J = 7.5, 1.5$  Hz, 1H), 7.39–7.31 (m, 3H), 5.58–5.50 (m, 1H), 5.21 (ddd,  $J = 8.6, 6.7, 5.0$  Hz, 1H), 5.06 (dd,  $J = 11.3, 2.8$  Hz, 1H), 4.62–4.55 (m, 1H), 4.50 (dd,  $J = 11.3, 2.2$  Hz, 1H), 4.10–4.05 (m, 1H), 3.99 (d,  $J = 10.2$  Hz, 1H), 3.14–3.09 (m, 1H), 3.11 (s, 3H), 3.03 (dd,  $J = 14.7, 6.7$  Hz, 1H), 2.95 (s, 3H), 1.49 (s, 9H).  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  169.5, 169.2, 168.1, 155.1, 137.1, 132.4, 131.8, 131.6, 130.0, 127.6, 81.3, 66.1, 55.3, 48.9, 37.6, 37.1, 35.9, 35.5, 28.2.

(4*S*,7*R*)-7-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-*N,N*-dimethyl-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[*j*][1]oxa[8]thia[5]-azacyclododecine-4-carboxamide (**14**). Compound **14** was prepared following the procedure described for the synthesis of compound **6** using **26** (38 mg, 84  $\mu$ mol, 1.0 equiv), Ac-L-Pro-OH (27 mg, 0.17 mmol, 2.0 equiv), EDC·HCl (32 mg, 0.17 mmol, 2.0 equiv), and DIPEA (27  $\mu$ L, 0.17 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 25 to 75% acetonitrile in water to give **14** (13 mg, 26  $\mu$ mol, 32%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{23}H_{31}N_4O_6S$   $[M + H]^+$  491.1964, found 491.1941.  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  7.90 (d,  $J = 7.3$  Hz, 2H), 7.54 (d,  $J = 9.0$  Hz, 1H), 7.39 (t,  $J = 7.4$  Hz, 1H), 7.32 (d,  $J = 7.4$  Hz, 1H), 7.27 (t,  $J = 7.6$  Hz, 1H), 5.13–5.01 (m, 2H), 4.87–4.83 (m, 1H), 4.52–4.47 (m, 2H), 4.44 (d,  $J = 11.0$  Hz, 1H), 3.90 (d,  $J = 11.0$  Hz, 1H), 3.77–3.65 (m, 1H), 3.50–3.42 (m, 1H), 3.06 (s, 3H), 3.04–2.94 (m, 2H), 2.92 (s, 3H), 2.35–2.26 (m, 1H), 2.22–2.12 (m, 1H), 2.15 (s, 3H), 1.96–1.84 (m, 2H).  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  172.0, 171.4, 169.5, 168.7, 167.7, 137.2, 132.6, 132.3, 132.2, 129.6, 127.6, 66.8, 60.0, 53.3, 49.9, 48.4, 38.4, 37.1, 35.9, 35.8, 27.6, 25.2, 22.6.

*tert*-Butyl-((4*S*,7*R*)-4-(diethylcarbamoyl)-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[*j*][1]oxa[8]thia[5]-azacyclododecine-7-yl)carbamate (**27**). Compound **27** was prepared following the procedure described for the synthesis of compound **24**, using **23d** (153 mg, 0.35 mmol, 1.0 equiv),  $Me_3SnOH$  (263 mg, 1.40 mmol, 4.0 equiv),  $HOBt \cdot xH_2O$  (71 mg, 0.53 mmol, 1.5 equiv), EDC·HCl (133 mg, 0.70 mmol, 2.0 equiv), and diethylamine (39  $\mu$ L, 0.38 mmol, 1.1 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **27** (55 mg, 0.12 mmol, 33%) as a colorless oil. HRMS (ESI)  $m/z$  calcd for  $C_{23}H_{34}N_3O_6S$   $[M + H]^+$  480.2168, found 480.2184.  $^1H$  NMR (400 MHz,  $CDCl_3$ )  $\delta$  7.80 (d,  $J = 7.4$  Hz, 1H), 7.41 (t,  $J = 7.4$  Hz, 1H), 7.32 (t,  $J = 7.5$  Hz, 1H), 7.29–7.21 (m, 2H), 5.53–5.46 (m, 1H), 5.18–5.08 (m, 1H), 4.70–4.53 (m, 2H), 4.56–4.45 (m, 1H), 4.14 (d,  $J = 10.8$  Hz, 1H), 4.07 (d,  $J = 10.8$  Hz, 1H), 3.55–3.33 (m, 3H), 3.33–3.17 (m, 1H), 3.01 (dd,  $J = 14.2, 3.4$  Hz, 1H), 2.79 (dd,  $J = 14.2, 8.7$  Hz, 1H), 1.44 (s, 9H), 1.24 (t,  $J = 7.0$  Hz, 3H), 1.11 (t,  $J = 7.0$  Hz, 3H).  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  169.0, 169.0, 167.8, 154.8, 137.0, 132.1, 131.3, 131.3, 130.5, 127.7, 80.2, 65.5, 52.8, 48.8, 42.1, 40.8, 37.0, 36.1, 28.3, 14.6, 12.8.

(4*S*,7*R*)-7-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-*N,N*-diethyl-6,10-dioxo-1,3,4,5,6,7,8,10-octahydrobenzo[*j*][1]oxa[8]thia[5]-azacyclododecine-4-carboxamide (**15**). Compound **15** was prepared following the procedure described for the synthesis of compound **6** using **27** (35 mg, 72  $\mu$ mol, 1.0 equiv), Ac-L-Pro-OH (22 mg, 0.14 mmol, 2.0 equiv), EDC·HCl (27 mg, 0.14 mmol, 2.0 equiv), and DIPEA (24  $\mu$ L, 0.14 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 5 to 95% acetonitrile in water to give **15** (9 mg, 37  $\mu$ mol, 24%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{25}H_{35}N_4O_6S$   $[M + H]^+$  519.2277, found 519.2286.  $^1H$  NMR (600 MHz,  $DMSO-d_6$ , mixture of two rotamers in the ratio 6:4)  $\delta$  9.14 (d,  $J = 3.8$  Hz, 0.6H), 9.13 (d,  $J = 3.8$  Hz, 0.4H), 8.61 (d,  $J = 8.2$  Hz, 0.4H), 8.18 (d,  $J = 8.2$  Hz, 0.6H), 7.78 (t,  $J = 7.6$  Hz, 1H), 7.53 (t,  $J = 7.8$  Hz, 1H), 7.46–7.43 (m, 1H), 7.41 (t,  $J = 7.6$  Hz, 1H), 5.04–4.97 (m, 1H), 4.97–4.92 (m, 1H), 4.63 (d,  $J = 9.7$  Hz, 0.6H), 4.54–4.50 (m, 0.6H), 4.48–4.43 (m, 1.2H), 4.38 (dd,  $J = 8.5, 3.4$  Hz, 0.6H), 4.33 (dd,  $J = 10.4, 4.5$  Hz, 0.4H), 4.17 (dd,  $J = 10.2, 4.5$  Hz, 0.6H), 3.89 (d,  $J = 9.8$  Hz, 0.4H), 3.85 (d,  $J = 9.8$  Hz, 0.6H), 3.62–3.51 (m, 1H), 3.49–3.43 (m, 1H), 3.41–3.34 (m, 2H), 3.33–3.27 (m, 1H), 3.22–3.18 (m, 1H), 3.02–2.93 (m, 1H), 2.80 (dd,  $J = 14.6, 11.4$  Hz, 0.4H), 2.76 (dd,  $J = 14.6, 11.2$  Hz, 0.6H), 2.23–2.15 (m, 0.4H), 2.05–1.98 (m, 0.6H), 1.95 (s, 1.8H), 1.93–1.88 (m, 1H), 1.88–1.71 (m, 2H), 1.76 (s, 1.2H), 1.14 (t,  $J = 7.0$  Hz, 1.8H), 1.11 (t,  $J = 7.0$  Hz, 1.2H), 1.04 (t,  $J = 7.1$  Hz, 1.8H), 1.02 (t,  $J = 7.1$  Hz, 1.2H).  $^{13}C$  NMR (151 MHz,  $DMSO-d_6$ , mixture of two rotamers)  $\delta$  172.6 and 172.3 (1C), 169.5 (1C), 169.4 (1C), 168.9 and 168.8 (1C), 167.7 and 167.6 (1C), 137.5 and 137.4 (1C), 133.1 and 133.0 (1C), 132.6 and 132.4 (1C), 131.9 and 131.8 (1C), 130.4 and 130.3 (1C), 128.2 (1C), 65.6 and 65.5 (1C), 60.2 and 59.3 (1C), 50.9 and 50.7 (1C), 50.1 and 49.6 (1C), 48.1 and 46.7 (1C), 41.7 and 41.6 (1C), 37.3 and 37.0 (1C), 35.4 and 35.3 (1C), 32.4 and 30.3 (1C), 24.7 (1C), 23.0 and 22.9 (1C), 22.7 and 22.4 (1C), 14.9 and 14.8 (1C), 13.4 and 13.3 (1C).

Methyl *N*-((*tert*-Butoxycarbonyl)-*D*-seryl)-*S*-methyl-*D*-cysteinate (**28**). *D*-Cysteine methyl ester hydrochloride (1.02 g, 5.90 mmol, 1.0 equiv) was dissolved in DMF (12 mL); methyl iodide (404  $\mu$ L, 6.51 mmol, 1.1 equiv) and DIPEA (2.01 mL, 11.8 mmol, 2.0 equiv) were then added, and the mixture stirred at rt for 2 h. The mixture was diluted with EtOAc (100 mL) and washed with saturated aqueous  $NaHCO_3$  solution (2  $\times$  50 mL) and brine (100 mL). The organic phase was dried over  $MgSO_4$ , filtered, and then concentrated under reduced pressure. The obtained crude mixture was dissolved in acetonitrile (10 mL); Boc-*D*-Ser-OH (1.21 g, 5.90 mmol, 1.0 equiv) and EDC·HCl (1.12 g, 5.90 mmol, 1.0 equiv) were added, and the reaction was stirred for 1 h at rt. EtOAc (100 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (50 mL), saturated aqueous  $NaHCO_3$  solution (50 mL), and brine (100 mL). The organic phase was dried over  $MgSO_4$ , filtered, and then concentrated under reduced pressure. The obtained colorless oil (867 mg, 2.60 mmol, 44% over two steps) was employed in the next step without further purification. HRMS (ESI)  $m/z$  calcd for  $C_{13}H_{25}N_2O_6S$   $[M + H]^+$  337.1433, found 337.1428.  $^1H$

NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.31 (br s, 1H), 5.56 (d,  $J = 6.7$  Hz, 1H), 4.84–4.74 (m, 1H), 4.23 (br s, 1H), 4.07 (m, 1H), 3.76 (s, 3H), 3.67 (dd,  $J = 11.6, 5.3$  Hz, 1H), 3.00 (dd,  $J = 14.0, 4.8$  Hz, 1H), 2.90 (dd,  $J = 14.1, 6.5$  Hz, 1H), 2.10 (s, 3H), 1.45 (s, 9H). The  $\text{CH}_2\text{OH}$  proton was not detectable in this spectrum.  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  171.3, 171.1, 155.9, 80.5, 63.0, 55.2, 52.8, 51.6, 36.1, 28.3, 16.0.

(*R*)-2-((*tert*-Butoxycarbonyl)amino)-3-(((*S*)-1-methoxy-3-(methylthio)-1-oxopropan-2-yl)amino)-3-oxopropyl Benzoate (**29**). Compound **28** (215 mg, 0.64 mmol, 1.0 equiv) was dissolved in DCM (10 mL) and cooled to 0 °C. Benzoyl chloride (148  $\mu\text{L}$ , 1.28 mmol, 2.0 equiv), triethylamine (181  $\mu\text{L}$ , 1.28 mmol, 2.0 equiv), and DMAP (15 mg, 0.13 mmol, 0.2 equiv) were then added, and the reaction was allowed to warm to rt and stirred for 1 h. The mixture was diluted with EtOAc (100 mL) and washed with saturated aqueous  $\text{NaHCO}_3$  solution (2  $\times$  25 mL) and brine (25 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 50–70% EtOAc in hexane as the eluent to give **29** (237 mg, 0.54 mmol, 84%) as a yellow oil. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{20}\text{H}_{28}\text{N}_2\text{O}_7\text{SNa}$  [ $\text{M} + \text{Na}$ ] $^+$  463.1515, found 463.1523.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.05–7.99 (m, 2H), 7.55 (t,  $J = 7.7$  Hz, 1H), 7.42 (t,  $J = 7.7$  Hz, 2H), 7.14 (d,  $J = 7.7$  Hz, 1H), 5.46 (br s, 1H), 4.85–4.79 (m, 1H), 4.73–4.59 (m, 2H), 4.55 (dd,  $J = 10.8, 4.1$  Hz, 1H), 3.70 (s, 3H), 2.99 (dd,  $J = 14.1, 4.9$  Hz, 1H), 2.92 (dd,  $J = 14.0, 5.6$  Hz, 1H), 2.04 (s, 3H), 1.44 (s, 9H).  $^{13}\text{C}$  NMR (101 MHz,  $\text{CDCl}_3$ )  $\delta$  170.8, 169.1, 166.3, 155.4, 133.3, 129.8, 129.8, 128.4, 128.4, 128.3, 80.8, 64.6, 54.0, 52.7, 51.8, 36.3, 28.2, 16.1.

(*R*)-2-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-3-(((*S*)-1-methoxy-3-(methylthio)-1-oxopropan-2-yl)amino)-3-oxopropyl Benzoate (**30**). Compound **30** was prepared following the procedure described for the synthesis of compound **6** using **29** (200 mg, 0.46 mmol, 1.0 equiv), Ac-L-Pro-OH (144 mg, 0.92 mmol, 2.0 equiv), EDC-HCl (174 mg, 0.92 mmol, 2.0 equiv), and DIPEA (0.16 mL, 0.92 mmol, 2.0 equiv). The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **30** (92 mg, 0.19 mmol, 41%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{22}\text{H}_{29}\text{N}_3\text{NaO}_7\text{S}$  [ $\text{M} + \text{Na}$ ] $^+$  502.1624, found 502.1616.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  8.03 (d,  $J = 7.8$  Hz, 2H), 7.67 (d,  $J = 8.2$  Hz, 1H), 7.55–7.44 (m, 2H), 7.36 (t,  $J = 7.7$  Hz, 2H), 4.86–4.80 (m, 2H), 4.70–4.65 (m, 1H), 4.48–4.44 (m, 1H), 4.43–4.34 (m, 1H), 3.65 (s, 3H), 3.63–3.58 (m, 1H), 3.46–3.38 (m, 1H), 2.96–2.83 (m, 2H), 2.32–2.22 (m, 1H), 2.20–2.10 (m, 1H), 2.06 (s, 3H), 1.95 (s, 3H), 1.93–1.82 (m, 2H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  171.7, 171.4, 171.0, 168.7, 166.4, 133.3, 130.0, 130.0, 129.5, 128.4, 128.4, 64.3, 60.1, 52.7, 52.5, 51.9, 48.4, 35.9, 27.8, 25.2, 22.6, 15.9.

(*R*)-2-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-3-(((*S*)-1-(dimethylamino)-3-(methylthio)-1-oxopropan-2-yl)amino)-3-oxopropyl Benzoate (**16**). Compound **30** (54 mg, 0.11 mmol, 1.0 equiv) was dissolved in 1,2-dichloroethane (3 mL).  $\text{Me}_3\text{SnOH}$  (83 mg, 0.44 mmol, 4.0 equiv) was added, and the mixture was stirred for 45 min at 83 °C. The reaction was allowed to cool to rt, acidified with 1 M aqueous HCl solution (5 mL), and extracted with DCM (3  $\times$  40 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The obtained crude was dissolved in DMF (5 mL);  $\text{HOBt}\cdot x\text{H}_2\text{O}$  (23 mg, 0.17 mmol, 1.5 equiv), EDC-HCl (42 mg, 0.22 mmol, 2.0 equiv), and dimethylamine (2 M in THF, 110  $\mu\text{L}$ , 0.22 mmol, 2.0 equiv) were added, and the mixture was stirred for 2 h at rt. EtOAc (30 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (25 mL), saturated aqueous  $\text{NaHCO}_3$  solution (25 mL), and brine (25 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The crude product was purified by reverse-phase HPLC using a gradient from 25 to 75% acetonitrile in water to give **45** (9 mg, 20  $\mu\text{mol}$ , 18% over two steps) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{23}\text{H}_{33}\text{N}_4\text{O}_6\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  493.2121, found 493.2119.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.15–8.07 (m, 2H), 8.07–8.02 (m, 1H), 7.59–7.52 (m, 1H), 7.48–7.39 (m, 3H), 5.07–5.00 (m, 1H), 4.89 (dd,  $J = 11.2, 4.3$  Hz, 1H), 4.83–4.77 (m, 1H), 4.54 (dd,  $J = 7.9, 2.8$  Hz, 1H), 4.49 (dd,  $J = 11.2, 3.5$  Hz, 1H), 3.74 (ddd,  $J = 9.8, 8.1, 3.5$  Hz, 1H), 3.48–3.41 (m, 1H),

3.09 (s, 3H), 2.93 (s, 3H), 2.85 (dd,  $J = 13.5, 6.7$  Hz, 1H), 2.72 (dd,  $J = 13.5, 6.3$  Hz, 1H), 2.41–2.33 (m, 1H), 2.18 (s, 3H), 2.02 (s, 3H), 1.97–1.79 (m, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  171.8, 171.6, 169.9, 168.2, 166.4, 133.2, 130.0, 130.0, 129.5, 128.4, 128.4, 64.5, 59.9, 52.9, 48.3, 48.3, 37.8, 36.9, 35.9, 27.4, 25.2, 22.6, 16.3.

Methyl-(*S*)-2-(((2-((*tert*-butoxycarbonyl)amino)-3-(dimethylamino)-3-oxopropylthio)methyl)benzoate (**31**)). Boc-D-Cys-OH (6.81 g, 30.8 mmol, 1.5 equiv) was dissolved in DMF (20 mL) and THF (80 mL). Methyl 2-bromomethylbenzoate (4.72 g, 20.5 mmol, 1.0 equiv) and triethylamine (6.38 mL, 45.1 mmol, 2.2 equiv) were added, and the mixture was stirred for 16 h at rt. After evaporation of THF under reduced pressure, the mixture was diluted with EtOAc (250 mL) and washed with 1 M aqueous HCl solution (100 mL) and brine (2  $\times$  100 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure, and the obtained oil was dissolved in DMF (50 mL). Dimethylamine hydrochloride (3.34 g, 41.0 mmol, 2.0 equiv), EDC-HCl (7.86 g, 41.0 mmol, 2.0 equiv),  $\text{HOBt}\cdot x\text{H}_2\text{O}$  (4.15 g, 30.8 mmol, 1.5 equiv), and DIPEA (7.16 mL, 41.0 mmol, 2.0 equiv) were then added, and the mixture was stirred for 2 additional hours at rt. The mixture was diluted with EtOAc (250 mL) and washed with 1 M aqueous HCl solution (100 mL), saturated aqueous  $\text{NaHCO}_3$  solution (100 mL), and brine (100 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 10–100% EtOAc in hexane as the eluent to give **31** (3.90 g, 9.84 mmol, 48% over two steps) as a yellow oil. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{19}\text{H}_{29}\text{N}_2\text{O}_5\text{S}$  [ $\text{M} + \text{H}$ ] $^+$  397.1797, found 397.1804.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.90 (d,  $J = 7.7$  Hz, 1H), 7.42 (t,  $J = 7.3$  Hz, 1H), 7.38 (d,  $J = 7.3$  Hz, 1H), 7.30 (t,  $J = 7.6$  Hz, 1H), 5.37 (d,  $J = 8.7$  Hz, 1H), 4.81–4.75 (m, 1H), 4.18–4.08 (m, 2H), 3.89 (s, 3H), 3.06 (s, 3H), 2.95 (s, 3H), 2.76 (dd,  $J = 13.7, 7.4$  Hz, 1H), 2.63 (dd,  $J = 13.7, 6.0$  Hz, 1H), 1.43 (s, 9H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  171.0, 167.7, 155.1, 140.2, 131.9, 131.2, 131.2, 129.4, 127.2, 79.8, 52.2, 49.6, 37.4, 35.9, 34.9, 34.6, 28.3.

Methyl-2-((4*S*,7*R*)-4-(dimethylcarbamoyl)-7-(2-hydroxyethyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (**32**). Compound **31** (1.18 g, 3.01 mmol, 1.0 equiv) was dissolved in 4 M HCl in dioxane (10 mL) and stirred for 1 h at rt. After evaporation of the solvent under reduced pressure, the resulting salt was dissolved in acetonitrile (15 mL). Boc-D-Homoser-OH (1.99 g, 9.03 mmol, 3.0 equiv), EDC-HCl (1.72 g, 9.03 mmol, 3.0 equiv), and DIPEA (524  $\mu\text{L}$ , 3.01 mmol, 1.0 equiv) were added, and the reaction was stirred for 2 h at rt. EtOAc (100 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (50 mL), saturated aqueous  $\text{NaHCO}_3$  solution (50 mL), and brine (50 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure. The reaction crude was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **32** (500 mg, 1.01 mmol, 33% over two steps) as a white foam. HRMS (ESI)  $m/z$  calcd for  $\text{C}_{23}\text{H}_{33}\text{N}_3\text{O}_7\text{NaS}$  [ $\text{M} + \text{Na}$ ] $^+$  520.2093, found 520.2079.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.89 (dd,  $J = 7.8, 1.2$  Hz, 1H), 7.44 (td,  $J = 7.7, 1.2$  Hz, 1H), 7.37 (dd,  $J = 7.7, 1.2$  Hz, 1H), 7.31 (td,  $J = 7.8, 1.2$  Hz, 1H), 7.21–7.13 (m, 1H), 5.61 (d,  $J = 7.4$  Hz, 1H), 5.06–4.99 (m, 1H), 4.40–4.33 (m, 1H), 4.18 (d,  $J = 13.2$  Hz, 1H), 4.08 (d,  $J = 13.2$  Hz, 1H), 3.90 (s, 3H), 3.75–3.66 (m, 2H), 3.02 (s, 3H), 2.94 (s, 3H), 2.80 (dd,  $J = 13.7, 6.6$  Hz, 1H), 2.69 (dd,  $J = 13.7, 6.4$  Hz, 1H), 2.04–1.95 (m, 1H), 1.83–1.73 (m, 1H), 1.44 (s, 9H). The  $\text{CH}_2\text{OH}$  proton was not detectable in this spectrum.  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  171.5, 170.2, 167.9, 156.3, 140.0, 132.0, 131.2, 131.2, 129.5, 127.4, 80.3, 58.8, 52.3, 52.0, 48.6, 37.4, 36.0, 36.0, 35.1, 34.1, 28.3.

2-((4*S*,7*R*)-4-(Dimethylcarbamoyl)-7-(2-hydroxyethyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoic Acid (**33**). Compound **33** (480 mg, 0.97 mmol, 1.0 equiv) was dissolved in MeOH (5 mL) and 0.1 M aqueous LiOH (20 mL), and the reaction mixture was stirred at 35 °C for 16 h. The reaction was then allowed to cool to rt, acidified with 1 M aqueous HCl solution (5 mL), and extracted with DCM (3  $\times$  50 mL). The organic phase was dried over  $\text{MgSO}_4$ , filtered, and then concentrated under reduced pressure providing compound **33** (302 mg, 0.63 mmol, 64%) as a colorless



solid. HRMS (ESI)  $m/z$  calcd for  $C_{22}H_{34}N_3O_7S$   $[M + H]^+$  484.2117, found 484.2120.  $^1H$  NMR (600 MHz, DMSO- $d_6$ )  $\delta$  12.94 (br s, 1H), 8.01 (d,  $J = 8.5$  Hz, 1H), 7.84 (dd,  $J = 7.8, 1.4$  Hz, 1H), 7.47 (td,  $J = 7.5, 1.4$  Hz, 1H), 7.40 (dd,  $J = 7.5, 1.4$  Hz, 1H), 7.36 (td,  $J = 7.8, 1.4$  Hz, 1H), 6.87 (d,  $J = 8.0$  Hz, 1H), 4.88–4.82 (m, 1H), 4.53–4.45 (m, 1H), 4.14 (d,  $J = 13.0$  Hz, 1H), 4.05 (d,  $J = 13.0$  Hz, 1H), 4.95–4.03 (m, 1H), 3.48–3.36 (m, 2H), 2.94 (s, 3H), 2.82 (s, 3H), 2.75 (dd,  $J = 13.6, 7.7$  Hz, 1H), 2.46 (dd,  $J = 13.6, 5.9$  Hz, 1H), 1.80–1.71 (m, 1H), 1.69–1.58 (m, 1H), 1.38 (s, 9H).  $^{13}C$  NMR (151 MHz, DMSO- $d_6$ )  $\delta$  172.3, 170.1, 168.8, 155.7, 140.6, 131.9, 131.4, 131.3, 130.6, 127.6, 78.6, 58.0, 52.4, 48.6, 37.0, 35.7, 35.4, 34.3, 33.4, 28.6.

*tert*-Butyl-((4*S*,7*R*)-4-(dimethylcarbamoyl)-6,11-dioxo-1,4,5,6,7,8,9,11-octahydro-3*H*-benzo[*c*][1]oxa[6]thia[9]azacyclotridecin-7-yl)carbamate (**34**). Compound **34** was prepared following the procedure described for the synthesis of compound **23a** using **33** (163 mg, 0.33 mmol, 1.0 equiv), triphenylphosphine (173 mg, 0.66 mmol, 2.0 equiv), and di-*tert*-butyl azodicarboxylate (152 mg, 0.66 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give **34** (32 mg, 69  $\mu$ mol, 21%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{22}H_{32}N_3O_6S$   $[M + H]^+$  466.2012, found 466.2003.  $^1H$  NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.67 (d,  $J = 7.2$  Hz, 1H), 7.46–7.38 (m, 2H), 7.33–7.27 (m, 1H), 7.22 (d,  $J = 7.9$  Hz, 1H), 5.74 (d,  $J = 9.2$  Hz, 1H), 4.92 (td,  $J = 8.5, 2.7$  Hz, 1H), 4.70–4.63 (m, 1H), 4.59–4.52 (m, 1H), 4.42–4.43 (m, 1H), 4.21 (d,  $J = 13.6$  Hz, 1H), 3.98 (d,  $J = 13.6$  Hz, 1H), 2.98 (s, 3H), 2.93 (s, 3H), 2.82–2.70 (m, 2H), 2.39 (dd,  $J = 14.2, 8.5$  Hz, 1H), 2.08–1.99 (m, 1H), 1.48 (s, 9H).  $^{13}C$  NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  170.6, 169.5, 168.9, 156.4, 138.9, 131.9, 131.7, 131.6, 129.4, 127.4, 80.8, 63.7, 51.0, 49.5, 37.1, 36.0, 34.2, 32.8, 30.5, 28.4.

(4*S*,7*R*)-7-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-*N,N*-dimethyl-6,11-dioxo-1,4,5,6,7,8,9,11-octahydro-3*H*-benzo[*c*][1]oxa[6]thia[9]azacyclotridecin-4-carboxamide (**17**). Compound **17** was prepared following the procedure described for the synthesis of compound **6** using **34** (25 mg, 54  $\mu$ mol, 1.0 equiv), Ac-L-Pro-OH (25 mg, 0.16 mmol, 3.0 equiv), EDC-HCl (31 mg, 0.16 mmol, 3.0 equiv), and DIPEA (28  $\mu$ L, 0.16 mmol, 3.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give **17** (13 mg, 25  $\mu$ mol, 46%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{24}H_{33}N_4O_6S$   $[M + H]^+$  505.2121, found 505.2121.  $^1H$  NMR (600 MHz, CDCl<sub>3</sub>)  $\delta$  8.12 (d,  $J = 9.9$  Hz, 1H), 7.73 (d,  $J = 7.6$  Hz, 1H), 7.57 (d,  $J = 6.9$  Hz, 1H), 7.55 (d,  $J = 7.7$  Hz, 1H), 7.44 (t,  $J = 7.7$  Hz, 1H), 7.29–7.25 (m, 1H), 4.89–4.83 (m, 2H), 4.75 (td,  $J = 11.5, 2.5$  Hz, 1H), 4.58–4.55 (m, 1H), 4.54 (d,  $J = 13.9$  Hz, 1H), 4.32 (dt,  $J = 12.2, 4.3$  Hz, 1H), 3.90–3.85 (m, 1H), 3.82 (d,  $J = 13.9$  Hz, 1H), 3.53 (td,  $J = 9.4, 6.6$  Hz, 1H), 2.93 (s, 6H), 2.90–2.86 (m, 1H), 2.65–2.56 (m, 2H), 2.44–2.38 (m, 1H), 2.36–2.28 (m, 1H), 2.13 (s, 3H), 2.11–2.03 (m, 2H), 2.00–1.93 (m, 1H).  $^{13}C$  NMR (151 MHz, CDCl<sub>3</sub>)  $\delta$  172.0, 171.6, 170.2, 169.3, 169.1, 139.2, 132.2, 131.5, 131.2, 130.1, 127.0, 61.5, 60.7, 50.8, 49.4, 47.7, 36.8, 35.8, 32.7, 31.7, 29.3, 28.7, 24.9, 22.6.

Methyl-(*S*)-2-(4-(dimethylcarbamoyl)-11,11-dimethyl-6,9-dioxo-10-oxa-2-thia-5,8-diazadodecyl)benzoate (**35**). Compound **31** (4.16 g, 10.5 mmol, 1.0 equiv) was dissolved in 4 M HCl in dioxane (15 mL) and stirred for 1 h at rt. After evaporation of the solvent under reduced pressure, the resulting salt was dissolved in acetonitrile (35 mL). Boc-Gly-OH (2.76 g, 15.8 mmol, 1.5 equiv), EDC-HCl (3.03 g, 15.8 mmol, 1.5 equiv), and DIPEA (1.83 mL, 10.5 mmol, 1.0 equiv) were added, and the reaction was stirred for 2 h at rt. EtOAc (200 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (100 mL), saturated aqueous NaHCO<sub>3</sub> solution (100 mL), and brine (50 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–10% MeOH in DCM as the eluent to give **35** (3.81 g, 8.41 mmol, 80% over two steps) as a colorless oil. HRMS (ESI)  $m/z$  calcd for  $C_{21}H_{32}N_3O_6S$   $[M + H]^+$  454.2012, found 454.2002.  $^1H$  NMR (400 MHz, CDCl<sub>3</sub>)  $\delta$  7.90 (dd,  $J = 7.8, 1.3$  Hz, 1H), 7.44 (td,  $J = 7.6, 1.3$  Hz, 1H), 7.37 (dd,  $J = 7.6, 1.3$  Hz, 1H), 7.31 (td,  $J = 7.8, 7.3$  Hz, 1H), 6.90 (d,  $J = 7.1$  Hz, 1H), 5.24–5.14 (m, 1H), 5.12–5.02 (m, 1H), 4.17 (d,  $J = 13.1$  Hz, 1H), 4.07 (d,  $J = 13.1$  Hz, 1H), 3.90 (s, 3H), 3.85–3.73 (m, 2H), 3.03 (s, 3H),

2.94 (s, 3H), 2.80 (dd,  $J = 14.0, 6.8$  Hz, 1H), 2.69 (dd,  $J = 14.0, 5.6$  Hz, 1H), 1.44 (s, 9H).  $^{13}C$  NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  170.3, 168.8, 167.8, 155.8, 140.1, 132.0, 131.2, 131.2, 129.4, 127.3, 80.2, 52.3, 48.3, 44.1, 37.4, 35.9, 35.2, 34.2, 28.3.

Methyl-2-((4*S*,10*R*)-4-(Dimethylcarbamoyl)-10-(hydroxymethyl)-14,14-dimethyl-6,9,12-trioxo-13-oxa-2-thia-5,8,11-triazapentadecyl)benzoate (**36**). Compound **35** (3.80 g, 8.40 mmol, 1.0 equiv) was dissolved in 4 M HCl in dioxane (15 mL) and stirred for 1 h at rt. After evaporation of the solvent under reduced pressure, the resulting salt was dissolved in acetonitrile (25 mL). Boc-D-Ser-OH (3.44 g, 16.8 mmol, 2.0 equiv), EDC-HCl (3.21 g, 16.8 mmol, 2.0 equiv), and DIPEA (1.46 mL, 8.40 mmol, 1.0 equiv) were added, and the reaction was stirred for 2 h at rt. EtOAc (150 mL) was then added, and the mixture was washed with 1 M aqueous HCl solution (50 mL), saturated aqueous NaHCO<sub>3</sub> solution (50 mL), and brine (50 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–15% MeOH in DCM as the eluent to give **36** (1.50 g, 2.77 mmol, 33% over two steps) as a colorless oil. HRMS (ESI)  $m/z$  calcd for  $C_{24}H_{36}N_4O_8NaS$   $[M + Na]^+$  563.2152, found 563.2145.  $^1H$  NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.89 (d,  $J = 9.4$  Hz, 1H), 7.52 (d,  $J = 8.2$  Hz, 1H), 7.46–7.42 (m, 1H), 7.36–7.29 (m, 2H), 7.22–7.15 (m, 1H), 5.62 (d,  $J = 8.6$  Hz, 1H), 5.06–4.95 (m, 1H), 4.32–4.22 (m, 2H), 4.12–4.03 (m, 3H), 3.92–3.90 (m, 1H), 3.90 (s, 3H), 3.71–3.60 (m, 1H), 3.03 (s, 3H), 2.91 (s, 3H), 2.75 (dd,  $J = 13.9, 7.2$  Hz, 1H), 2.66 (dd,  $J = 13.9, 6.1$  Hz, 1H), 1.43 (s, 9H).  $^{13}C$  NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  172.0, 170.7, 168.5, 168.0, 155.6, 140.0, 132.1, 131.2, 131.2, 129.5, 127.4, 80.3, 63.2, 55.8, 52.3, 48.4, 43.1, 37.5, 36.1, 35.3, 33.9, 28.3.

2-((4*S*,10*R*)-4-(Dimethylcarbamoyl)-10-(hydroxymethyl)-14,14-dimethyl-6,9,12-trioxo-13-oxa-2-thia-5,8,11-triazapentadecyl)benzoic Acid (**37**). Compound **36** (502 mg, 0.93 mmol, 1.0 equiv) was dissolved in MeOH (5 mL) and 0.1 M aqueous LiOH (20 mL). After being stirred at 35 °C for 16 h, the reaction was acidified with 1 M aqueous HCl solution (5 mL) and extracted with DCM (3  $\times$  50 mL). The organic phase was dried over MgSO<sub>4</sub>, filtered, and then concentrated under reduced pressure. The crude reaction mixture was purified by flash chromatography on a silica gel column using 0–15% MeOH in DCM as the eluent to give **37** (240 mg, 0.46 mmol, 49%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{23}H_{33}N_4O_8S$   $[M - H]^-$  525.2019, found 525.2023.  $^1H$  NMR (500 MHz, DMSO- $d_6$ )  $\delta$  12.77 (br s, 1H), 8.20 (d,  $J = 8.3$  Hz, 1H), 8.16–8.04 (m, 1H), 7.83 (dd,  $J = 7.6, 1.4$  Hz, 1H), 7.48 (td,  $J = 7.6, 1.3$  Hz, 1H), 7.40 (dd,  $J = 7.6, 1.3$  Hz, 1H), 7.35 (td,  $J = 7.6, 1.4$  Hz, 1H), 6.72 (d,  $J = 7.8$  Hz, 1H), 4.88–4.70 (m, 1H), 4.13 (d,  $J = 13.0$  Hz, 1H), 4.09 (d,  $J = 13.0$  Hz, 1H), 4.01–3.96 (m, 1H), 3.79–3.65 (m, 2H), 3.56 (d,  $J = 5.5$  Hz, 2H), 2.96 (s, 3H), 2.83 (s, 3H), 2.75 (dd,  $J = 13.5, 7.9$  Hz, 1H), 2.48–2.42 (m, 1H), 1.39 (s, 9H).  $^{13}C$  NMR (126 MHz, DMSO- $d_6$ )  $\delta$  171.2, 170.0, 169.0, 168.8, 155.7, 140.5, 131.9, 131.4, 131.4, 131.2, 127.6, 78.7, 62.4, 57.1, 48.7, 42.3, 37.1, 35.8, 34.4, 33.6, 28.6.

*tert*-Butyl-((4*S*,10*R*)-4-(dimethylcarbamoyl)-6,9,13-trioxo-1,4,5,6,7,8,9,10,11,13-decahydro-3*H*-benzo[*m*][1]oxa[11]thia[5,8]-diazacyclopentadecin-10-yl)carbamate (**38**). Compound **38** was prepared following the procedure described for the synthesis of compound **23a** using **37** (130 mg, 0.25 mmol, 1.0 equiv), triphenylphosphine (131 mg, 0.50 mmol, 2.0 equiv), and di-*tert*-butyl azodicarboxylate (115 mg, 0.50 mmol, 2.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 20 to 70% acetonitrile in water to give **38** (25 mg, 50  $\mu$ mol, 20%) as a colorless powder. HRMS (ESI)  $m/z$  calcd for  $C_{23}H_{33}N_4O_5S$   $[M + H]^+$  509.2070, found 509.2068.  $^1H$  NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.72 (d,  $J = 7.5$  Hz, 1H), 7.56–7.47 (m, 1H), 7.41–7.36 (m, 1H), 7.36–7.26 (m, 3H), 5.87 (d,  $J = 6.7$  Hz, 1H), 4.93–4.86 (m, 1H), 4.84 (d,  $J = 11.0$  Hz, 1H), 4.69–4.61 (m, 1H), 4.39 (dd,  $J = 11.1, 5.7$  Hz, 1H), 4.23–4.13 (m, 1H), 4.09 (d,  $J = 13.3$  Hz, 1H), 3.93 (d,  $J = 13.6$  Hz, 1H), 3.63–3.55 (m, 1H), 3.08 (s, 3H), 2.93 (s, 3H), 2.84–2.75 (m, 1H), 2.66–2.55 (m, 1H), 1.45 (s, 9H).  $^{13}C$  NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  170.1, 169.9, 168.8, 167.4, 155.3, 139.1, 131.7, 130.7, 130.5, 130.2, 127.4, 80.5, 65.2, 53.1, 48.8, 43.7, 37.3, 36.1, 35.4, 33.9, 28.3.



(4*S*,10*R*)-10-((*S*)-1-Acetylpyrrolidine-2-carboxamido)-*N,N*-dimethyl-6,9,13-trioxo-1,4,5,6,7,8,9,10,11,13-decahydro-3*H*-benzo-*[m][1]oxa[1]thia[5,8]diazacyclopentadecine-4-carboxamide* (**18**). Compound **18** was prepared following the procedure described for the synthesis of compound **6** using **38** (20 mg, 39  $\mu$ mol, 1.0 equiv), Ac-L-Pro-OH (19 mg, 0.12 mmol, 3.0 equiv), EDC·HCl (23 mg, 0.12 mmol, 3.0 equiv), and DIPEA (21  $\mu$ L, 0.12 mmol, 3.0 equiv). The crude product was purified by reverse-phase HPLC using a gradient from 25 to 75% acetonitrile in water to give **18** (9 mg, 16  $\mu$ mol, 41%) as a colorless powder. HRMS (ESI) *m/z* calcd for C<sub>25</sub>H<sub>34</sub>N<sub>5</sub>O<sub>7</sub>S [M + H]<sup>+</sup> 548.2179, found 548.2128. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  8.20 (dd, *J* = 7.6, 4.7 Hz, 1H), 7.88 (dd, *J* = 7.8, 1.5 Hz, 1H), 7.59 (d, *J* = 9.3 Hz, 1H), 7.50 (d, *J* = 7.7 Hz, 1H), 7.47 (td, *J* = 7.5, 1.5 Hz, 1H), 7.40 (dd, *J* = 7.7, 1.3 Hz, 1H), 7.36 (td, *J* = 7.5, 1.4 Hz, 1H), 5.18 (dd, *J* = 11.3, 2.3 Hz, 1H), 4.94 (dt, *J* = 9.3, 2.8 Hz, 1H), 4.86 (td, *J* = 8.1, 4.6 Hz, 1H), 4.47 (dd, *J* = 11.2, 3.3 Hz, 1H), 4.37–4.25 (m, 3H), 3.82 (d, *J* = 13.5 Hz, 1H), 3.72–3.66 (m, 1H), 3.62–3.49 (m, 2H), 3.13 (s, 3H), 2.94 (s, 3H), 2.89–2.80 (m, 2H), 2.29–2.14 (m, 3H), 2.12 (s, 3H), 2.03–1.94 (m, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  172.7, 170.6, 170.4, 169.1, 168.9, 168.2, 139.1, 132.2, 131.4, 130.8, 129.7, 127.6, 64.3, 61.2, 52.4, 48.6, 48.0, 44.0, 37.4, 36.2, 36.1, 34.9, 30.0, 25.4, 22.5.

**Calculation of Descriptors and PCA Analysis.** Two-dimensional structures from the DNP and DrugBank<sup>16</sup> databases were imported into MOE (version 2015.10, Chemical Computing Group, [www.chemcomp.com](http://www.chemcomp.com)).<sup>49</sup> Both data sets were cleaned by addition of explicit hydrogen atoms, removal of organometallic compounds, and adjustment of protonation states (i.e., strong acids were deprotonated and strong bases were protonated). Duplicate structures were eliminated using the 2D SMILES string of each compound. Molecular descriptors were calculated for all compounds in the DNP and in DrugBank, and for the 41 lead-like macrocycle cores identified from the DNP, using the MOE descriptor module.<sup>50</sup> Principal component analysis (PCA) was performed using SIMCA (version 14.1)<sup>51</sup> to identify clusters and investigate the coverage of the chemical space of the DNP and DrugBank.

**Extraction of Macrocylic Cores from the DNP.** The structures of the natural products contained in the sd-file version of the Dictionary of Natural Products (DNP) were cleaned as described previously, whereby natural products containing toxic or reactive groups are removed using the knowledge-based HTS filter in Pipeline Pilot.<sup>12</sup> Further filtering in Pipeline Pilot was performed to retain macrocycles ( $\geq 12$  and  $\leq 30$  atoms in the ring) having  $< 10$  rings in total and a MW  $< 2500$  Da to provide 3657 macrocyclic natural products.

The side chains of the macrocyclic natural products were then pruned to provide macrocycle cores using the first two steps of the fragment generation algorithm reported previously,<sup>12</sup> i.e., deglycosylation and extraction of extended Murcko frameworks<sup>15</sup> decorated with attachment points (Figure S2). In these two steps, macrocycles were not fragmented and all ring systems and linker atoms were kept while linear side chains were pruned, but not removed. This provided macrocycle cores, i.e., macrocycles decorated with attachment points, which are functional groups linked directly to the ring system or pruned side chains. Thus, the algorithm was designed to retain functional groups and only cut side chains at carbon–carbon bonds at a distance of  $> 2$  atoms from the ring.

The algorithm defines an attachment point by traversing through the side chain of the macrocycle through two heavy atoms, starting from the ring-based atom. For each heavy atom thereafter it decides, based on a set of rules,<sup>12</sup> whether the atom is kept, whether that atom's neighboring atoms are examined, or if the side chain is cut after the second heavy atom. The algorithm keeps aliphatic carbon atoms as long as they are within two atoms from the ring, and longer aliphatic side chains are pruned to ethyl groups. If the algorithm encounters a carbon atom linked to a non-carbon and non-hydrogen atom, it will keep the heteroatom and examine the immediate neighbor atom, irrespective of its distance from the ring. The algorithm prunes after the next upcoming carbon atom as long as this is not double-bonded to a heteroatom.

In branched side chains, the algorithm traverses the branches separately, within the distance limit. If the branching carbon is

connected to a heteroatom by a double bond, the immediate next neighbor atom in the other branch will be investigated using the same rules as for a linear side chain, irrespective of its distance from the ring. The next upcoming carbon atom is kept, as the terminating carbon, but the rest of the side chain is pruned. If the branching atom is a heteroatom, the immediate next neighbor atoms will be investigated in the same manner, irrespective of their distance from the ring. As a result of this pruning process, the original functional group motif of the macrocyclic part of the natural product remains intact in the core. In addition, functional groups close to the macrocycle ring in side chains remain intact. All generated macrocycle cores are stored in a database for further processing.

If identical macrocyclic cores were obtained after side-chain pruning of different natural products, only one was kept. Cores with  $\leq 3$  O or N atoms or  $> 15$  rotatable bonds, without a heteroatom in the macrocycle, and/or having more than one macrocycle were also removed in Pipeline Pilot. This resulted in 2175 natural-product-derived macrocycle cores, which are based on 764 different Murcko scaffolds. The 2175 cores were clustered in Pipeline Pilot using an FCFP6 pharmacophore fingerprint<sup>52</sup> into 217 clusters (Supporting Excel Sheet). The cluster centers were visually inspected by three or more experienced medicinal chemists, and 41 cluster centers or near neighbors were selected by exclusion of cores containing reactive or potentially redox-sensitive groups, excessively complex structures including  $> 6$  rings, or oligomeric/pseudo-oligomeric structures (Supporting Excel Sheet).

**Docking of Macrocycle Cores into Keap1.** The atomic coordinates of four crystal structures of human Keap1, in complex with a small molecule [PDB ID: 4IQK<sup>24</sup> (1.97 Å) and 3VNG (2.1 Å)] or in the apo form [PDB ID: 4IFJ (1.8 Å) and 1ZGK<sup>25</sup> (1.35 Å)], were obtained from the Protein Data Bank. Each crystal structure was imported and refined with Protein Preparation Wizard<sup>53</sup> using the Schrödinger Suite.<sup>54</sup> Structure refinement included adding hydrogen atoms, assigning bond orders, building disulfide bonds, and removal of water molecules beyond 5 Å from the ligand atoms. The PROPKA tool from Protein Preparation Wizard was used to predict the protonation states of the ionizable residues at pH 7.0.<sup>55</sup> Subsequently, the positions of the hydrogen atoms in the Keap1 structure were energy-minimized using the OPLS3 force field.<sup>56</sup>

The receptor grid generation module of Glide<sup>26</sup> was used to define the active site for the docking experiments. The active site of Keap1 was defined either by the bound ligand or by key residues in the active site (R415, R483, Y525, Y572, A556, S603), which were used as the centroid of the grid box (the radius of the active site was 15 Å from the centroid). The docking protocol was first validated by comparing the conformation (the pose) of the bound ligands as obtained from docking with the one determined by X-ray crystallography for the two ligand-bound structures (PDB ID: 4IQK<sup>24</sup> and 3VNG; see the Supporting Information, Computational Procedure 1). Subsequently, the set of 41 cores was preprocessed using LigPrep, which included generating possible ionization states (at pH 7) using the Epik tool<sup>57</sup> and structure minimization using the OPLS3 force field.<sup>56</sup> During preprocessing, input chirality was retained.

The 41 lead-like macrocyclic cores (Table S1) were then docked into each of the four refined Keap1 structures using Glide, which treats the cores as fully flexible while Keap1 remains rigid during the docking. Docking poses for each macrocycle core were ranked according to the GlideScore from the standard precision mode, and for each core, the docked pose that had the lowest GlideScore for binding to each of the Keap1 structures was identified (Table S2). The resulting combined list of the top-10 cores for each Keap1 structure was then used to rank the cores by the number of Keap1 structures each core bound to (Table S3), after which the docked poses of the top-five cores in each of the four crystal structure were inspected visually (Figure S5).

Induced-fit docking (IFD)<sup>58</sup> of stereoisomeric macrocycles **6–9** was performed as implemented in the Schrödinger suite.<sup>58,59</sup> Three-dimensional structures of compounds **6–9** were built with Maestro from the Schrödinger suite<sup>60</sup> and energy-minimized using the OPLS3 force field. A grid centered at the bound ligand (PDB ID: 4IQK<sup>24</sup>) with the box size of 15 Å was defined. Subsequently, **6–9** were docked and scored using the standard precision mode with an extended sampling

protocol, which will generate up to 80 poses per ligand from several docking runs. The cyclothialidine core was included for comparison. A total of 286 binding poses were analyzed with regard to if the phenyl ring of **1** and **6–9** reached deep into the Kelch channel of the Keap1 binding pocket (Figure S6).

**Binding Affinity Calculations.** The atomic coordinates of Keap1 in complex with compound **14** were preprocessed as described above in the docking section. The chloride ion in the ligand binding pocket was included in the binding affinity calculations as it is coordinated with residues in Keap1 (N382, N414, and R415) and with **14**. Methyl ester (**9**), amide (**12**), monomethyl amide (**13**), and diethylamide (**15**) were modeled into the Keap1 binding site using the bound conformation of **14** as a template. For compounds **9** and **13**, two alternative conformations of the methyl ester and methyl amide (cis- and trans-forms) were explored, while five alternative conformations were explored for the diethylamide of **15** (Tables S6 and S7).

Prior to affinity calculations, the ligand binding site was defined using the receptor grid generation module available in Glide as described in the docking section above. Subsequently, the initial input ligand coordinates for **9** and **12–15** were optimized in the receptor, the ligand binding pose was scored, and poses were used for ligand binding affinity calculations in Prime MM-GBSA<sup>34</sup> as implemented in the Schrödinger suite. The ligand and a 5 Å region of the protein around the ligand were treated as flexible in the binding affinity calculation. Other settings were chosen as defaults.

**Similarity to Reported Inhibitors of the Keap1-Nrf2 PPI.** A set of inhibitors of the Keap1-Nrf2 PPI, termed “PubChem”, was retrieved from the PubChem Bioassay database (PubChem Bioassay AIDs: 504523; 588683; 651798; 651801; 651806; 651807; 651823; 651829; 651833; 651834).<sup>38</sup> Duplicate entries, compounds that lacked complete stereochemical information, and solvents were removed from the 1127 compounds reported as active in the 10 listed bioassays. Compounds that did not pass the PAINS filter<sup>39</sup> were also removed, providing a PubChem set consisting of 375 inhibitors (Table S8). Another set of “validated” inhibitors was assembled from two sources. First, nine compounds, reported to show reproducible activity in a triad of orthogonal assays by Tran et al.,<sup>20</sup> were selected. Second, crystal structures that had a compound bound in the binding site for Nrf2 of Keap1 were retrieved from the Protein Data Bank (PDB). Peptides were excluded, while fragments were retained, providing a validated set of 35 compounds (Table S9).

The Tanimoto coefficient<sup>40</sup> was calculated to determine the similarity between compound **14** and the Keap1 inhibitors in the validated and PubChem sets, respectively, using seven commonly used structural fingerprints (Figure S11). The similarity was also calculated based on all fingerprints. All substructure similarity analyses were performed with the Canvas tool (version 3.1.011).<sup>61</sup> The molecular similarity landscape (a so-called networklike similarity graph) was investigated for the merged validated and PubChem sets and for the validated set alone, using compound **14** as a reference. The structures of all compounds were imported, and a substructure fragment fingerprint (FragFp)-based similarity landscape was computed using DataWarrior (version 5.2.1).<sup>41</sup> FragFp contains more than 500 predefined binary structural fingerprints. DataWarrior calculates a similarity matrix for each compound in the data set and allocates each compound to the most similar neighbor space using the Rubberbanding force-field approach.<sup>41</sup> Compounds with a structural similarity >0.95 were clustered for the analysis.

**SPR Inhibition in Solution Assay (ISA).** The SPR ISA was performed in analogy to the protocol developed by Chen et al.<sup>30</sup> with the following differences. Instead of a biotinylated peptide, a lysine-tagged version of the Nrf2-peptide (KKKKAFFAQLQLDEETGEFL) was utilized for tethering to the sensor surface. For the covalent tethering, a CMS biosensor (GE Healthcare, research grade) was employed using HBS-P [10 mM HEPES, 150 mM NaCl, 0.05% (v/v) Tween 20, 1.0 mM TCEP, pH 7.40] as the continuous flow buffer at a rate of 10  $\mu$ L/min at 20 °C on a BIAcore 3000 optical biosensor unit (GE Healthcare). The surface was activated by injecting a mixture of 1-ethyl-3-(3-dimethylaminopropyl)-carbodiimide hydrochloride (EDC) and *N*-hydroxysuccinimide (NHS) for 7 min, followed by an injection

of 50  $\mu$ M Nrf2-peptide in 10 mM Na-acetate, pH 4.0 for 2 min. Any reactive groups still present on the surface were deactivated by a 7 min injection of 0.1 M ethanolamine hydrochloride-NaOH, pH 8.5. Peptide immobilization levels were typically between 400 and 600 RU to ensure mass transport limitation and thus a protein-concentration-dependent response.

For the detection of active compounds, a solution of 25 nM Keap1 (Kelch domain = aa321–aa609) was prepared in running buffer and preincubated with compounds at either constant or varying concentrations. These mixtures were subsequently injected over the peptide-modified biosensor, and control samples devoid of compounds were used to determine the remaining free protein concentration of Keap1 in response to compound binding. For this, the initial association slopes of the binding sensorgrams have been measured (interval 5–15 s after sample injection) for each sample, followed by a regeneration of the sensor for the next cycle through a 45 s injection of 50 mM Tris/Cl, 0.25% SDS, 5 mM TCEP, pH 7.5. Results have been reported as  $K_D$ -values for concentration-response experiments. This was achieved by a nonlinear regression analysis of the concentration-response data, which is normalized by the control samples representing 0% inhibition.

**SPR Direct Binding Assay (DBA).** The SPR DBA was performed as described previously,<sup>31</sup> with the difference that HBS-P [10 mM HEPES, 150 mM NaCl, 0.05% (v/v) Tween 20, 1.0 mM TCEP, pH 7.40] was used as a continuous flow buffer and that the contact and dissociation times were 45 and 60 s, respectively.

**Isothermal Titration Calorimetry (ITC).** Isothermal titration calorimetry experiments were carried out on a MicroCal ITC-200 system (GE Healthcare) using Keap1 protein (Kelch domain = aa321–aa609) that had been passed through a PD-10 column (GE Healthcare) equilibrated with 10 mM HEPES, 150 mM NaCl, 0.05% (v/v) Tween 20, 1 mM TCEP, pH 7.40, and 1% DMSO. Complete titration of 60  $\mu$ M Keap1 was typically achieved by injecting  $19 \times 2 \mu$ L of aliquots of a 0.6 mM compound at a temperature of 25 °C and a spacing interval of 90 s between injections. The thermodynamic binding parameters were extracted by nonlinear regression analysis of the binding isotherms (Microcal Origin version 7.0 software package). A single site binding model was applied, yielding binding enthalpy ( $\Delta H$ ), stoichiometry ( $n$ ), entropy ( $\Delta S$ ), and association constant ( $K_a$ ).

**Aqueous Solubility.** The aqueous solubility of compound **14** was determined at AstraZeneca as reported previously.<sup>62</sup>

**Cell Permeability.** The efflux-inhibited permeability of compound **14** across Caco-2 cell monolayers was determined in the presence of a cocktail of inhibitors of efflux transporters (50  $\mu$ M quinidine, 30  $\mu$ M benzbramarone, and 20  $\mu$ M sulfasalazine) by the DMPK department at Pharmaron as reported previously.<sup>63</sup>

**Clearance by Human Liver Microsomes.** The clearance of compound **14** by human liver microsomes ( $CL_{int}$ ) was determined at AstraZeneca as reported previously.<sup>62</sup>

**Cell Assay for **14**.** The cellular activity of macrocycle **14** was determined by DiscoveRx using their PathHunter U2OS Keap1-Nrf2 functional assay,<sup>64</sup> which quantifies the translocation of Nrf2 into the nucleus due to inhibition of formation of the Keap1-Nrf2 complex.

**Keap1 Expression, Purification, and Crystallization with **14**.** The DNA coding domain for human Keap1 (6  $\times$  His- TCS-Keap1 (A321-T609) [E540A, E542S]) was cloned into pET24 and expressed in *Escherichia coli* BL21(DE3)-Star in LB media at 291 K. After harvest, the cells were resuspended in 20 mM Tris/HCl, pH 7.5, 200 mM NaCl, 5% glycerol, 1 mM DTT, protease inhibitors, disrupted with a high-pressure homogenizer, and clarified by centrifugation. Purification was made by affinity chromatography using Ni Sepharose FF (GE Healthcare), eluted in one step with 300 mM imidazole followed by SEC (Superdex200, GE Healthcare). The purified protein was detagged at room temperature with a thrombin cleave kit (Sigma) and finalized by a second SEC step (Superdex200, GE Healthcare) in buffer 20 mM Tris/HCl, pH 7.5, 5 mM DTT before concentration (Amicon cells, 10 kDa cutoff). Crystals were grown at 20 °C in sitting drops using 200 nL of protein (11–13 mg/mL) and 200 nL of well solution (3.7–4.1 M ammonium acetate, 0.09 M sodium acetate pH 4.6, and 10 mM cadmium chloride). Crystals appeared after one day but continued to grow for approximately one week. Compound **14** was

soaked by incubating crystals for 1 h with 1 mM compound in 5 M ammonium acetate and 0.1 M sodium acetate at pH 4.6. Crystals were subsequently frozen in liquid nitrogen using a soaking solution supplemented with 20% glycerol as the cryoprotectant prior to data collection.

**Data Collection, Structure Solution, and Refinement.** X-ray diffraction data was collected at the European Synchrotron Facility, beamline ID29, using a Pilatus 6M-F pixel detector to a resolution of 2.37 Å. The data were indexed and integrated with MOSFLM<sup>65</sup> and scaled with SCALA<sup>66</sup> in the space group  $P2_12_12_1$  with cell dimensions of 75.7 75.8 203.1 Å. Two Keap1 molecules were identified by PHASER<sup>67</sup> using a published Keap1 structure (PDB ID: 1ZGK<sup>25</sup>) as a search model. The Keap1 macrocycle complex was further refined by alternative cycles of model rebuilding in Coot<sup>68</sup> and refinement in AutoBuster 2.11.6 (Global Phasing Ltd, Cambridge U.K.). A final model was refined to an R/Rfree of 19/21%. Full data collection and refinement statistics can be found in Table S5, and the 2Fo-Fc electron density of compound 14 is in Figure S8. The coordinates and corresponding structure factors have been deposited to the Protein Data Bank with accession code 6Z6A.

## ■ ASSOCIATED CONTENT

### SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.jmedchem.0c01569>.

Supporting tables, supporting figures, computational procedures, NMR spectra of all new compounds, purity reports for compounds 6–18 (PDF)

Structures (SMILES codes) for the sets of 41 and 217 macrocyclic cores (XLSX)

Structures (SMILES codes) and biological properties for compounds 6–18, (CSV)

### Accession Codes

Coordinates and structure factors for the complex of Keap1 with compound 14 have been deposited with the PDB with accession code 6Z6A. The authors will release the atomic coordinates and experimental data upon article publication.

## ■ AUTHOR INFORMATION

### Corresponding Authors

**Stefan Schiesser** – Department of Medicinal Chemistry, Research and Early Development, Respiratory and Immunology, BioPharmaceuticals R&D, AstraZeneca, 43183 Mölndal, Sweden; [orcid.org/0000-0002-8668-2844](https://orcid.org/0000-0002-8668-2844); Email: [stefan.schiesser@astrazeneca.com](mailto:stefan.schiesser@astrazeneca.com)

**Jan Kihlberg** – Department of Chemistry - BMC, Uppsala University, 75123 Uppsala, Sweden; [orcid.org/0000-0002-4205-6040](https://orcid.org/0000-0002-4205-6040); Email: [jan.kihlberg@kemi.uu.se](mailto:jan.kihlberg@kemi.uu.se)

### Authors

**Fabio Begnini** – Department of Chemistry - BMC, Uppsala University, 75123 Uppsala, Sweden

**Vasanthanathan Poongavanam** – Department of Chemistry - BMC, Uppsala University, 75123 Uppsala, Sweden; [orcid.org/0000-0002-8880-9247](https://orcid.org/0000-0002-8880-9247)

**Björn Over** – Department of Medicinal Chemistry, Research and Early Development, Early Cardiovascular, Renal and Metabolism, BioPharmaceuticals R&D, AstraZeneca, 43183 Mölndal, Sweden

**Marie Castaldo** – Discovery Biology, Discovery Sciences, R&D, AstraZeneca, 43183 Mölndal, Sweden

**Stefan Geschwindner** – Structure, Biophysics & Fragment-Based Lead Generation, Discovery Sciences, R&D,

AstraZeneca, 43183 Mölndal, Sweden; [orcid.org/0000-0002-2154-8345](https://orcid.org/0000-0002-2154-8345)

**Patrik Johansson** – Structure, Biophysics & Fragment-Based Lead Generation, Discovery Sciences, R&D, AstraZeneca, 43183 Mölndal, Sweden

**Mohit Tyagi** – Department of Chemistry - BMC, Uppsala University, 75123 Uppsala, Sweden

**Christian Tyrchan** – Department of Medicinal Chemistry, Research and Early Development, Respiratory and Immunology, BioPharmaceuticals R&D, AstraZeneca, 43183 Mölndal, Sweden

**Lisa Wissler** – Structure, Biophysics & Fragment-Based Lead Generation, Discovery Sciences, R&D, AstraZeneca, 43183 Mölndal, Sweden

**Peter Sjö** – Drugs for Neglected Diseases initiative (DNDi), 1202 Geneva, Switzerland

Complete contact information is available at:

<https://pubs.acs.org/doi/10.1021/acs.jmedchem.0c01569>

### Author Contributions

#F.B., V.P., and B.O. made equal contributions to the manuscript. The manuscript was written through contributions of all authors. All authors have given approval to the final version of the manuscript.

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## ■ ABBREVIATIONS

bRo5, beyond the rule of 5;  $Cl_{int}$ , human microsomal metabolism; cLogS, calculated solubility; DNP, Dictionary of Natural Products; ISA, inhibition in solution assay; ITC, isothermal titration calorimetry; Keap1, Kelch-like ECH Associated Protein-1; Nrf2, nuclear factor, erythroid 2 Like 2; NRotB, number of rotatable bonds;  $P_{app}$  AB + inh, efflux-inhibited permeability across a Caco-2 cell monolayer; PPI, protein–protein interaction; SPR, surface plasmon resonance; TCE, 2,2,2-trichloroethyl; TPSA, topological polar surface area

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