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## **Chemically-Induced Cell Wall Stapling in Bacteria**

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## Summary

Transpeptidation reinforces the structure of cell wall peptidoglycan, an extracellular heteropolymer that protects bacteria from osmotic lysis. The clinical success of transpeptidase-inhibiting  $\beta$ -lactam antibiotics illustrates the essentiality of these cross-linkages for cell wall integrity, but the presence of multiple, seemingly redundant transpeptidases in many species makes it challenging to determine cross-link function. Here we present a technique to link peptide strands by chemical rather than enzymatic reaction. We employ bio-compatible click chemistry to induce

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Authors Contributions

Conceptualization M.S.S., F.C., S.L.R, A.E., A.K.A.; Investigation, S.L.R., A.E., A.K.A., P.S., C.M.-M. and J.K.; Writing - Original Draft, M.S.S. and S.L.R.; Writing - Review & Editing M.S.S., S.L.R., A.E., and F.C.; Visualization S.L.R.; Funding Acquisition, M.S.S., A.K.A., P.S., F.C.; Supervision, M.S.S., F.C.

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Declaration of Interest

The authors declare no competing interests.

The authors have a patent related to this work.

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triazole formation between azido- and alkynyl-D-alanine residues that are metabolically installed in the peptidoglycan of Gram-positive or Gram-negative bacteria. Synthetic triazole cross-links can be visualized using azidocoumarin-D-alanine, an amino acid derivative that undergoes fluorescent enhancement upon reaction with terminal alkynes. Cell wall stapling protects *Escherichia coli* from treatment with the broad-spectrum  $\beta$ -lactams ampicillin and carbenicillin. Chemical control of cell wall structure in live bacteria can provide functional insights that are orthogonal to those obtained by genetics.

## **Graphical Abstract**



## eTOC blurb

We developed a chemical method for introducing synthetic cross-links into the cell walls of live Gram-positive and Gram-negative bacteria. The cross-links protect *Escherichia coli* from broad-spectrum  $\beta$ -lactam antibiotics. Chemically-induced cross-linking complements genetics as an independent way to investigate the physiological roles of cell wall connectivity.

## Introduction

Cell wall peptidoglycan is a mesh-like biopolymer that surrounds nearly all bacteria and is required to resist turgor pressure. The macromolecule consists of a glycan backbone and peptides, containing both L- and D-amino acids, that are cross-linked by D,D- and L,D- transpeptidases (Egan et al., 2015). The degree of transpeptidation can vary with species, growth phase and environmental conditions (Vollmer and Seligman, 2010). For example, the

peptidoglycan of slow- or non-growing *E. coli* is more highly cross-linked and less susceptible to *in vitro* enzymatic turnover than that of actively-replicating *E. coli* (Glauner et al., 1988; Goodell and Tomasz, 1980; Lee et al., 2013; Pisabarro et al., 1985; Tuomanen and Cozens, 1987; Tuomanen et al., 1988). Cross-linking abundance is also predicted to impact the overall strength and stiffness of the cell envelope (Auer and Weibel, 2017; Huang et al., 2008; Loskill et al., 2014; Vollmer and Bertsche, 2008), cell shape (Huang et al., 2008; Sycuro et al., 2010; Yang et al., 2019), and assembly of macromolecular structures (Scheurwater and Burrows, 2011). The clinical success of transpeptidase-inhibiting  $\beta$ -lactam antibiotics highlights the importance of peptidoglycan cross-linking in bacterial physiology.

Despite the biological and medical significance of peptidoglycan transpeptidation, unraveling the roles of these linkages is challenging. Currently, the standard ways to manipulate cross-linking are to mutate or deplete the expression of the transpeptidase genes or to inhibit these enzymes with small molecules like  $\beta$ -lactams. However, the functional redundancy of transpeptidases and promiscuity of  $\beta$ -lactams (Spratt, 1975) pose challenges to rational control of peptidoglycan connectivity.

D-amino acids bearing reactive groups such as cysteines, alkynes, azides and tetrazines have been used to metabolically label the peptidoglycan stem peptide (de Pedro et al., 1997; Kuru et al., 2012; Pidgeon et al., 2015; Radkov et al., 2018; Siegrist et al., 2015; Siegrist et al., 2013). Once embedded, the presence of these probes can be revealed by chemical reaction with an exogenous label that bears a complementary reactive group (Siegrist et al., 2015). We hypothesized that we might also use functionalized peptide strands to manipulate cell wall cross-linking. More specifically, we reasoned that co-incubation of bacteria with azidoand alkynyl-D-amino acids would result in a subpopulation of labeled muropeptide strands in close enough proximity to undergo copper-catalyzed azide-alkyne cycloaddition (CuAAC) upon introduction of the appropriate reagents. Such structures would serve as synthetic, triazole cross-links.

## **Results and Discussion**

We first tested our hypothesis using a loss-of-fluorescence assay. In this approach, bacteria are co-incubated in the presence of azido- and alkynyl-D-amino acids, washed and subjected to CuAAC (Figure 1A, left). We reasoned that the peptidoglycan-embedded functional groups should either react with each other or with the alkynyl- or azido-fluorophores in CuAAC solution. Bacteria incubated with a single D-amino acid probe, by contrast, should have muropeptides decorated with just one functional group, which in turn should react only with the complementary reactive fluorophore. In this assay, we interpret decreased labeling of co-incubated relative to singly-incubated bacteria to indicate that there are fewer peptidoglycan-embedded functional groups available to react with the fluorophores. This may occur because the reaction between the D-amino acids for initial incorporation into the muropeptide. To control for the latter possibility, we also subjected metabolically-labeled bacteria to strain-promoted azide-alkyne cycloaddition (SPAAC; Figure 1A, right) with a cyclooctyne-appended fluorophore. In the absence of copper and other reagents, peptidoglycan-embedded azides and alkynes should not react with each other at an

appreciable rate and only the azide-cyclooctyne reaction should occur. In the SPAAC reactions, therefore, we interpret changes in labeling to mean that the azido-D-amino acid outcompetes or is outcompeted by other D-amino acids for initial incorporation into the cell wall.

We used the loss-of-fluorescence approach to ask whether we could introduce triazole crosslinks into the cell wall of Listeria monocytogenes, a Gram-positive, food-borne pathogen. We initially used *pbp5::tn L. monocytogenes*, a D,D-carboxypeptidase-deficient mutant that we previously showed has high levels of D-amino acid labeling (Siegrist et al., 2013). After incubating the bacteria in the presence of equal amounts of D-alanine (Dala), azido-D-amino acid (azDA or azDlys, the R groups of which respectively have one and four carbons), alkynyl-D-alanine (alkDA) or mixtures thereof, we washed away unincorporated amino acid and subjected the bacteria to CuAAC with either an alkynyl- (Figure 1B) or azidofluorophore (Figure 1C). We assessed cellular fluorescence by flow cytometry. In both cases, the bacteria that were co-incubated in alkDA/azDlys had lower amounts of fluorescence than those incubated in azDlys (Figure 1B) or alkDA (Figure 1C) alone. We obtained similar results with the more bio-friendly CuAAC reaction (Yang et al., 2014) that employs the 3-[4-({bis[(1-tert-butyl-1H-1,2,3-triazol-4-yl)methyl]amino}methyl)-1H-1,2,3-triazol-1yl]propanol (BTTP) ligand (Wang et al., 2011) (Figure 1D) or in wild-type L. monocytogenes (Figures S1A and S1B). These data suggested that bacteria co-incubated in azDlys and alkDA had fewer peptidoglycan-embedded functional groups available to react with complementary reactive fluorophores in solution (Figure 1A). Moreover, in bacteria subjected to SPAAC with cyclooctyne-fluorophore, the signal after alkDA/azDlys incubation was similar to that of azDlys alone or Dala/azDlys (Figures 1E and S1C). The SPAAC control reactions suggested that there was no appreciable competition between the D-amino acids for initial incorporation into the peptidoglycan. Taken together these data suggest that cell wall-embedded azides and alkynes can react with each other by CuAAC.

We next sought a more direct read-out for triazole cross-links. Fluorogenic molecules undergo a fluorescence enhancement upon chemical or enzymatic reaction. For example, CuAAC reaction of the non-fluorescent 3-azido-7-hydroxycoumarin (azidocoumarin) with terminal alkynes yields fluorescent triazole products (Sivakumar et al., 2004). As D-amino acids appended to fluorophores, including hydroxycoumarin, incorporate efficiently into peptidoglycan (Kuru et al., 2012), we decided to test whether swapping an azido-coumarin D-amino acid (azcDA) for an azido-D-amino acid would allow us to mark the presence of triazole cross-links (Figure 1F).

We began by synthesizing azcDA, which could be accessed readily by coupling *N*<sup>a</sup>-Boc-D-2,3-diaminopropionic acid to azidocoumarin acid **1** (Weineisen et al., 2017) via pentafluorophenyl ester **2**. Trifluoroacetic acid-mediated Boc deprotection afforded azcDA. Next we co-incubated *pbp5::tn L. monocytogenes* with Dala/azcDA or alkDala/azcDA, washed, and subjected the bacteria or not to a BTTP CuAAC reaction that lacked a complementary alkynyl-fluorophore. By microscopy we found that fluorescence of live, azcDA-labeled bacteria required the inclusion of alkDA in the initial metabolic labeling step as well as the subsequent CuAAC reaction (Figure 1G). These gain-of-fluorescence data were consistent with the loss-of-fluorescence results (Figures 1B–1D, S1A, S1B) and

supported the notion that a CuAAC reaction can covalently join azides and alkynes metabolically installed in *L. monocytogenes* peptidoglycan to form synthetic cross-links.

We were initially unable to identify peptidoglycan modifications that were specific to alkDA/azDlys-treated, CuAAC-subjected bacteria and that had the exact mass of a triazole cross-link (Figure S1D). We hypothesized that our ability to detect such modifications which theoretically could include muropeptides with additional, transpeptidase-mediated linkages—was complicated by the pre-existing complexity of L. monocytogenes peptidoglycan, which is both highly-cross-linked and tailored by N-deacetylases, Oacetyltransferases and amidotransferases (Aubry et al., 2011; Boneca et al., 2007; Rae et al., 2011). Therefore, we turned our attention to the model, Gram-negative bacterium *Escherichia coli*, as its peptidoglycan composition is considerably less complex (Vollmer et al., 2008). To simplify our analysis even more, we employed a strain, CS802-2, in which most of the genes encoding peptide-acting cell wall enzymes were deleted, including all 6 carboxypeptidases (Denome et al., 1999). The lack of tetrapeptides in this background prevents L,D-transpeptidation, so we expected D-amino acid incorporation to occur at the 5<sup>th</sup> position of the stem peptide (Cava et al., 2011). We first verified that CuAAC-subjected, alkDA/azDA-labeled CS802-2 E. coli, like wild-type E. coli, were less fluorescent than those labeled by azDA or alkDA alone (Figures 2A-2B, S1E-F) The decrease in D-amino acid concentration and labeling time compared to L. monocytogenes (described further below) correlated with a more modest reduction in fluorescence. We next metabolically labeled CS802-2 E. coli with different combinations of D-amino acids, washed away unincorporated amino acid, and performed BTTP CuAAC reactions in the absence of fluorophore. We then separated digested peptidoglycan by ultra-performance liquid chromatography (UPLC) and used MS/MS to identify molecules with the exact masses that corresponded to azDA- and alkDA-terminating pentapeptides in the appropriate samples. We identified peaks that were specific to alkDA/azDA-treated, CuAAC-subjected bacteria (Figure 2C) and had the exact masses of a 5-5 triazole dimer, trimer (+/- anh) or tetramer (Figures 2D–E, S2, S3). The presence of these species increased the total cross-linking by approximately 20% (Table 1). We note that muropeptide incorporation of azDA was ~2-fold more efficient than alkDA and associated with a general decrease in cross-linking (Tables 1 and S2B). Exogenous *D*-amino acids, including both Dala and non-canonical *D*-amino acids, have been shown or hypothesized to inhibit D,D-transpeptidation (Caparros et al., 1992; Lam et al., 2009). Importantly, however, CuAAC-dependent cross-linking occurred only in alkDA/azDA-treated samples (Table 1) and not in controls that had been treated with equimolar amounts of Dala/azDA (Figure S2B) or Dala alone (Table 1). These data confirmed our ability to introduce synthetic cross-links into bacterial peptidoglycan in a CuAAC-inducible manner.

Cell wall homeostasis is a balance between synthesis and turnover. Peptidoglycan-cleaving enzymes have been implicated directly (Park and Strominger, 1957; Schwarz et al., 1969; Tomasz et al., 1970; Tomasz and Waks, 1975) and indirectly (Cho et al., 2014; Kohlrausch and Höltje, 1991) in  $\beta$ -lactam cidality. Slow- or non-replicating,  $\beta$ -lactam-tolerant *E. coli* have highly cross-linked cell walls that are more resistant *in vitro* to lytic enzymes (Glauner et al., 1988; Goodell and Tomasz, 1980; Lee et al., 2013; Pisabarro et al., 1985; Tuomanen and Cozens, 1987; Tuomanen et al., 1988). We wondered whether  $\beta$ -lactam susceptibility

might be influenced by pre-existing cell wall cross-linking, either in addition to, or as part of, the well-documented effect of bacterial growth rate (Eng et al., 1991; Lee et al., 2018; Lee et al., 1944; Toumanem et al., 1986). Since exogenous D-amino acids can modify the structure, amount and strength of peptidoglycan and inhibit bacterial growth (Caparros et al., 1992; Cava et al., 2011; Lam et al., 2009), and growth rate in turn correlates with  $\beta$ -lactam lethality (Eng et al., 1991; Lee et al., 2018; Lee et al., 1944; Toumanem et al., 1986), we first optimized D-amino acid concentration and incubation time (Figures S4A and S4B). We then labeled *E. coli* with different combinations of D-amino acids, washed and performed BTTP CuAAC. After CuAAC reagent washout, bacteria were resuspended in growth medium and challenged with the  $\beta$ -lactam ampicillin. Without CuAAC, ampicillin treatment resulted in similar killing regardless of what D-amino acid(s) the bacteria had been metabolically labeled with (Figure 3A). In *E. coli* subjected to CuAAC, ampicillin caused ~2–3 logs of killing for bacteria labeled with Dala, Dala/alkDA or Dala/azDA but less than 1 log for those labeled with alkDA/azDA (Figure 3B). These data suggested that triazole cross-links protect *E. coli* from ampicillin.

BTTP-liganded CuAAC (Wang et al., 2011) is more biocompatible than traditional TBTA CuAAC (Yang et al., 2014). While the BTTP CuAAC reaction that we previously optimized for mycobacterial species (Garcia-Heredia et al., 2018) did not change L. monocytogenes cell counts (Figure S4C), they partially inhibited the recovery of *E. coli* on solid medium (Figure 3). However the effect was consistent across the different D-amino acid combinations, indicating that the synthetic cross-links, not the CuAAC, were responsible for antibiotic rescue. In liquid medium, E. coli subjected to CuAAC had a distinct lag in growth relative to mock-reacted controls (Figures S4D-F). The length of the lag phase was significantly enhanced in CS802-2 E. coli that had been incubated in both azDA and alkDA e.g. bacteria with triazole linkages. Since longer lag phases are associated with antibiotic tolerance (Bertrand, 2019; Fridman et al., 2014) we asked whether the apparent protection afforded by alkDA/azDA labeling followed by CuAAC was transient and whether it was specific to  $\beta$ -lactams. During the post-CuAAC lag phase, synthetic cross-links protected bacteria from ampicillin and the closely-related antibiotic carbenicillin (Figures 3B, S4G, S4H) but not the translation-inhibiting aminoglycoside kanamycin (Figure 3C). This protection was lost following resumption of growth (Figure S4I). Thus an extended, post-CuAAC lag phase correlates with, but is likely not responsible for, the enhanced tolerance of alkDA/azDA-labeled *E. coli* to β-lactams.

Taken together, our data suggest that synthetic peptidoglycan cross-links protect against lethality induced by the broad-spectrum  $\beta$ -lactams ampicillin and carbenicillin. The total cross-linking density across CuAAC-treated bacteria is similar (Tables 1 and Figure S2B), suggesting that the unusual linkage (triazole) or position on the stem peptide (5–5) is instead responsible for protection. The classic view of  $\beta$ -lactam activity is that transpeptidase inhibition damages the cell wall by disrupting the balance between peptidoglycan synthases and hydrolases (Park and Strominger, 1957; Schwarz et al., 1969; Tomasz et al., 1970; Tomasz and Waks, 1975).  $\beta$ -lactams also induce a metabolically-taxing, futile cycle of cell wall synthesis and turnover (Cho et al., 2014; Kohlrausch and Höltje, 1991). Both models for  $\beta$ -lactam cidality posit that lethality directly or indirectly depends on the activity of peptidoglycan-degrading enzymes. An artificially-reinforced cell wall may resist  $\beta$ -lactam-

induced damage because its structure is partially independent from transpeptidase-mediated synthesis. Additionally, or alternatively, synthetic cross-links may regulate peptidoglycan turnover. Indeed, while this manuscript was under review, Dik and colleagues proposed that non-canonical cell wall cross-links (derived from the reaction of exogenously-incorporated sulfonyl fluoride D-amino acids with endogenous m-DAP) can impede the processivity of lytic transglycosylases (Dik et al., 2020), enzymes that cleave the carbohydrate backbone of peptidoglycan. While we cannot rule out pleiotropic effects on other cell envelope or periplasmic structures, we hypothesize that synthetic triazole cross-links act as molecular speed bumps for lytic transglycosylases, blunting  $\beta$ -lactam cidality by keeping peptidoglycan degradation at bay. Consistent with the diverse roles for these enzymes in peptidoglycan homeostasis (Dik et al., 2017), the prolonged, post-CuAAC recovery in liquid medium of synthetically cross-linked E. coli (Figures S4E-F) may also reflect slowed cell wall turnover. Treatment with unnatural p-amino acids alone modestly enhanced both lag phase and  $\beta$ -lactam tolerance (Figure S4E–I), although the effects were not statistically significant. As noncanonical D-amino acid incorporation is not expected to alter cell wall turnover by lytic transglycosylases (Caparros et al., 1992), we speculate that these moresubtle peptidoglycan modifications impact *E. coli* physiology by a different mechanism(s) than the synthetic cross-links.

Given the promiscuity with which D-amino acids incorporate into the bacterial peptidoglycan (Radkov et al., 2018; Siegrist et al., 2015) stapling can be readily adapted for a wide variety of species. Molecular control of synthetic cross-link positioning may also be possible. The effect(s) of 5–5 cross-links on cell wall structure may be different from native, 4,3 or 3,3 cross-links. For example, 5,5 cross-links likely allow more flexibility and/or more space between glycan strands, which could in turn change the physical properties of the peptidoglycan. Unlike monopeptides, which can incorporate into the 4<sup>th</sup> or 5<sup>th</sup> positions (or both) of stem peptides (Kuru et al., 2012; Siegrist et al., 2013), D-amino acid dipeptides with functional groups on their N- or C-terminus are predicted to install these groups specifically at 4<sup>th</sup> or 5<sup>th</sup> position, respectively (Liechti et al., 2013). Our loss-of-fluorescence assay suggests that dipeptides functionalized with N-terminal azides and alkynes permit the introduction of synthetic, 4-4 cross-links into CS802-2 E. coli (Figure S1F), in addition to the 5–5 linkages afforded by monopeptide labeling. The development of alkyne- and azidebearing DAP derivatives may also enable the introduction of triazole linkages at the 3<sup>rd</sup> position of the muropeptide. Finally, pulse-chase labeling in species with defined modes of growth can offer sub-cellular control of synthetic cross-links. Independent from and complementary to genetics, cell wall stapling is an orthogonal assay for dissecting the roles of peptidoglycan structure in bacterial physiology.

## Significance

Bacteria are surrounded by cell wall heteropolymers that are essential for viability under most circumstances. The structure of the cell wall is well-conserved and consists of a glycan backbone cross-linked by D-amino acid-containing peptides. Cross-link-inhibiting  $\beta$ -lactams account for two-thirds of the global antibiotic market, underscoring the general importance of these linkages to bacterial physiology. For a given species, the density of cross-linking can vary with replication rate and environmental conditions. These changes in cell wall

connectivity in turn correlate with other phenotypic properties of the bacterium. However most species have multiple, closely-related enzymes that catalyze cross-links, each with varying susceptibility to different  $\beta$ -lactams, making it difficult to control the density of these linkages by genetics or small molecule inhibition alone. In this work, we present a chemical technique to introduce synthetic cross-links to the cell walls of live bacteria. We use bio-compatible click chemistry to induce a reaction between azido- and alkynyl-Dalanine residues that are metabolically incorporated in the cell wall peptides of Grampositive and Gram-negative species. The resulting triazole linkages can be visualized by substituting azido-D-alanine with azidocoumarin-D-alanine, an amino acid analogue that becomes more fluorescent after reacting with an alkyne. Stapling the cell wall of *Escherichia coli* enhances its tolerance to two different broad-spectrum  $\beta$ -lactams. Chemical manipulation complements genetic and small molecule perturbations as an independent means of investigating the role of cell wall connectivity in bacterial physiology.

## STAR \* Methods

#### **RESOURCE AVAILABILITY**

**Lead Contact**—Further information and requests for resources and reagents should be directed to Lead Contact, M. Sloan Siegrist (siegrist@umass.edu)

**Material Availability**—All unique/stable reagents generated in this study are available from the Lead Contact.

**Data and Code Availability**—This publication did not use unpublished custom code, software, or algorithms.

## EXPERIMENTAL MODEL AND SUBJECT DETAILS

*E. coli* was grown in Luria-Bertani Broth (LB) at 37 °C. *L. monocytogenes* was grown in Brain Heart Infusion Broth (BHI) at 37 °C.

#### METHODS DETAILS

**Metabolic labeling and CuAAC**—*E. coli* were grown overnight at 37 °C. The next day cultures were back-diluted between 1:50 and 1:500 and D-amino acids (1.25 mM total per sample for monopeptides and 2.5 mM per sample for dipeptides) were added directly in the LB medium. Cells were grown until log phase (OD<sub>600</sub> 0.6–0.8) then centrifuged for 5 min at 5,000 × g at room temperature (RT). They were then washed with sterile-filtered PBS and subjected to BTTP CuAAC (200  $\mu$ M CuSO<sub>4</sub>, 800  $\mu$ M BTTP [Chemical Synthesis Core Facility, Albert Einstein College of Medicine, Bronx, NY], 2.5 mM sodium ascorbate (freshly prepared), with or without 25  $\mu$ M of azido or alkynyl fluorescent dye as appropriate) or TBTA CuAAC (1 mM CuSO<sub>4</sub>, 128  $\mu$ M TBTA [Click Chemistry Tools, Scottsdale, AZ], 1.2 mM sodium ascorbate (freshly prepared), with or without 25  $\mu$ M of azido or alkynyl fluorescent dye [Click Chemistry Tools]) for 1 hr at room temperature, shaking. Samples were then centrifuged, washed thrice with PBS, and either fixed with 2% (v/v) formaldehyde or used in assays described below.

*L. monocytogenes* were grown overnight at 37  $^{\circ}$ C with the D-amino acids (2.5 mM total per sample) then centrifuged for 5 min at 5,000×g at RT. They were washed with PBS and subjected to CuAAC as described for *E. coli*.

**General fluorescence analysis**—Mean fluorescence intensities (MFI) of bacterial cell populations were obtained by flow cytometry from a BD DUAL LSRFortessa instrument.

Samples were imaged on an inverted Nikon Eclipse Ti microscope equipped with a Hamamatsu Orca Flash 4.0 camera and reconstructed with NIS Elements.

Peptidoglycan composition analysis—200 mL cultures of log-phase CS802-2 E. coli were treated with p-amino acids and subjected to BTTP CuAAC as describe above. Bacteria were centrifuged for 5 minutes at  $5,000 \times g$  at RT, wash twice with MilliQ water, resuspended in 1 mL MilliQ water then added drop-wise into 80 mL of boiling 4% SDS. Samples were vigorously stirred for 1.5 hr then cooled to RT. The insoluble fraction (PG) was pelleted at 400,000 × g, 15 min, 30 °C (TLA-100.3 rotor; OptimaTM Max ultracentrifuge, Beckman). SDS was washed out and the PG was resuspended in 200 µl of 50 mM sodium phosphate buffer pH 4.9 and digested overnight with 30  $\mu$ g/mL muramidase (Cellosyl). Samples were incubated at 37 °C. PG digestion was stopped by 5 min incubation in a boiling water bath. Coagulated protein was removed by centrifugation. The supernatants were mixed with 150 µL 0.5 M sodium borate pH 9.5, and subjected to reduction of muramic acid residues into muramitol by sodium borohydride treatment (10 mg/mL final concentration, 30 min at RT). Samples was adjusted to pH 3.5 with phosphoric acid. Chromatographic analyses of muropeptides were performed on AOUITY Ultra Performance Liquid Chromatography (UPLC) BEH C18 column (130 Å, 1.7 µm, 2.1 mm by 150 mm; Waters), and peptides were detected at Abs. 204 nm using ACQUITY UPLC UV-Visible Detector. Muropeptides were separated using a linear gradient from buffer A (0.1% of Formic acid in water) to buffer B (0.1% of Formic acid in acetonitrile) in 218 min, and flow 0.25 mL/min. Muropeptide identity was confirmed by MS/MS analysis, using a Xevo G2-XS QTof system (Waters Corporation, USA). Quantification of muropeptides was based on their relative abundances (relative area of the corresponding peak). Cross-linking was determined by the following formula; crosslinking=dimmer+(trimmer/2).

**Antibiotic challenge**—CS802–2 *E. coli* that had been pre-labeled with D-amino acids for 6 hrs (OD<sub>600</sub> 0.6) (Figure 3) or overnight (Figure S4D–F) and subjected or not to BTTP CuAAC were washed with PBS and resuspended in LB medium to a normalized OD<sub>600</sub> of 0.3 with or without 125  $\mu$ g/mL ampicillin, 125  $\mu$ g/mL carbenicillin, or 6.25  $\mu$ g/mL kanamycin. After 1–5 hrs incubation at 37 °C, bacteria were washed twice with PBS and plated as 10-fold serial dilutions on LB agar.

#### **Chemical Synthesis and Characterization**

#### Synthesis of azidocoumarin-d-alanine (azcDA)

*General Procedures.:* Reactions were performed in round bottom flasks fitted with rubber septa under a positive pressure of nitrogen. Gas-tight syringes with stainless steel needles or cannulae were used to transfer air- and moisture-sensitive liquids. Flash column

chromatography was performed as described by Still *et al.* using granular silica gel (60-Å pore size, 40–63  $\mu$ m, 4–6% H<sub>2</sub>O content, Silicycle)(Still et al., 1978). Analytical thin layer chromatography (TLC) was performed using glass plates pre-coated with 0.25 mm 230–400 mesh silica gel impregnated with a fluorescent indicator (254 nm). TLC plates were visualized by short wave ultraviolet light (254 nm). Concentration of solutions under reduced pressure were carried out on rotary evaporators capable of achieving a minimum pressure of ~2 torr at 29–30 °C unless noted otherwise.

Dichloromethane, tetrahydrofuran, and *N*,*N*-dimethylformamide were were purified by the method of Grubbs *et al.* under a positive pressure of nitrogen(Pangborn et al., 1996).

Instrumentation .: Proton nuclear magnetic resonance (<sup>1</sup>H NMR) spectra were recorded with a Bruker Avance III 500 MHz spectrometer, are reported in parts per million, and are referenced to the residual protium in the NMR solvent (CDCl<sub>3</sub>:  $\delta$  7.24 (CHCl<sub>3</sub>), CD<sub>3</sub>OD:  $\delta$ 3.31 (CHD<sub>2</sub>OD), DMSO-d<sub>6</sub>:  $\delta$  2.50 (DMSO-d<sub>5</sub>)). Data are reported as follows: chemical shift [multiplicity (s = singlet, d = doublet, t = triplet, sp = septet, m = multiplet), coupling constant(s) in Hertz, integration]. Carbon-13 nuclear magnetic resonance (<sup>13</sup>C NMR) spectra were recorded with a Bruker Avance III 500 MHz spectrometer, are reported in parts per million, and are referenced from the carbon resonances of the solvent (CDCl<sub>3</sub>: & 77.23, CD<sub>3</sub>OD:  $\delta$  49.15, DMSO- $d_6$ :  $\delta$  39.51). Data are reported as follows: chemical shift. Infrared data (IR) were obtained with a Cary 630 Fourier transform infrared spectrometer equipped with a diamond ATR objective and are reported as follows: frequency of absorption ( $cm^{-1}$ ), intensity of absorption (s = strong, m = medium, w = weak, br = broad). Optical rotations were measured on a P-2000 JASCO polarimeter and compound concentrations are expressed in units of g/100 mL. High resolution mass spectra (HRMS) were recorded by the Harvard University Small Molecule Mass Spectrometry facility on an Agilent 6210 time-of-flight LCMS using an electrospray ionization (ESI) source.

#### **Overall synthetic scheme.:**



<u>Azidocoumarin pentafluorophenyl ester S3:</u> To a 25 mL round bottom flask charged with azidocoumarin acid **S1**(Weineisen et al., 2017) (56.0 mg, 214 µmol, 1 equiv) under a nitrogen atmosphere was added tetrahydrofuran (2 mL) at room temperature. *N*,*N*-diisopropylethylamine (74.5 µL, 428 µmol, 2.00 equiv) was added to the dark brown solution via syringe. This was followed immediately by addition of pentafluorophenyl trifluoroacetate (73.5 µL, 428 µmol, 2.00 equiv) via syringe and stirred at room temperature. After 30 min, the reaction mixture was concentrated under reduced pressure and the brown

residue was purified by flash column chromatography on silica gel (eluent: 10% ethyl acetate in hexanes) to provide the azidocoumarin pentafluorophenyl ester **S2** (41.0 mg, 45%) as a white solid. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta$  7.37 (d, *J* = 8.6 Hz, 1H), 7.15 (s, 1H), 6.93 (dd, *J* = 8.7, 2.5 Hz, 1H), 6.87 (d, *J* = 2.5 Hz, 1H), 5.04 (s, 2H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta$  164.5, 159.1, 157.6, 152.9, 142.2, 141.2, 140.2, 139.2, 137.2, 128.7, 125.9, 124.7, 114.3, 113.3, 102.2, 64.8. <sup>19</sup>F NMR (471 MHz, CDCl<sub>3</sub>, 25 °C):  $\delta$  -152.2 (m), -156.4 (m), -161.3 (m). FTIR (thin film, cm<sup>-1</sup>): 2128 (s), 1804 (m), 1722 (m), 1618 (m), 1521 (s), 1334 (m), 1118 (m), 1070 (m), 995 (m). HRMS (ESI, m/z): 428.0294 (calculated for C<sub>17</sub>H<sub>7</sub>F<sub>5</sub>N<sub>3</sub>O<sub>5</sub> [M+H]<sup>+</sup>: 428.0300). TLC (15% ethyl acetate in hexanes, R<sub>f</sub>): 0.32 (UV).



Boc-d-Ala-azidocoumarin S5: To a 25 mL round bottom flask charged with azidocoumarin pentafluorophenyl ester S2 (30.0 mg, 70.2 µmol, 1 equiv) and N<sup>a</sup>-Boc-D-2,3diaminopropionic acid (S3) (28.6 mg, 140 µmol, 2.00 equiv) under a nitrogen atmosphere was added tetrahydrofuran (2 mL) at room temperature. N,N-diisopropylethylamine (14.7  $\mu$ L, 140  $\mu$ mol, 2.00 equiv) was then added to the solution via syringe followed by *N*,*N*dimethylformamide (500 µL). After 20 min, the reaction mixture was concentrated under reduced pressure and the residue was purified by flash column chromatography on silica gel (eluent: 20% hexanes, 75% ethyl acetate, 5% acetic acid) to provide Boc-D-Alaazidocoumarin S4 (21.0 mg, 67%) as a white solid. <sup>1</sup>H NMR (500 MHz,  $d_6$ -DMSO, 25 °C):  $\delta$  8.20 (t, J = 6.0 Hz, 1H), 7.63 (s, 1H), 7.59 (d, J = 8.5 Hz, 1H), 7.07-6.98 (m, 3H), 4.60 (s, 2H), 4.13–4.05 (m, 1H), 3.57–3.48 (m, 1H), 3.44–3.36 (m, 1H), 1.38 (s, 9H). <sup>13</sup>C NMR (126 MHz, d<sub>6</sub>-DMSO, 25 °C): & 172.2, 167.5, 159.7, 157.2, 155.4, 152.4, 128.9, 127.1, 122.6, 113.3, 113.1, 101.7, 78.4, 67.2, 53.4, 39.6, 28.2. FTIR (thin film, cm<sup>-1</sup>): 3366 (br-s), 2989 (br-m), 2128 (s), 1737 (m), 1670 (m), 1618 (m), 1521 (m), 1148 (m), 1055 (m). HRMS (ESI, m/z): 446.1324 (calculated for  $C_{19}H_{20}N_5O_8$  [M–H]<sup>-</sup>: 446.1317). TLC (20:75:5 hexanes:ethyl acetate:acetic acid,  $R_f$ ): 0.18 (UV).  $[\alpha]_D^{23} = +78$  (c 0.25, DMSO).



**d-Ala-azidocoumarin (1):** To a 25 mL round bottom flask charged with Boc-D-Alaazidocoumarin **S4** (18.0 mg, 40.2 µmol, 1 equiv) was added dichloromethane (2 mL) followed by trifluoroacetic acid (2 mL) at room temperature. After stirring for 15 min, the reaction mixture was concentrated under reduced pressure and the residue was purified by automated C<sub>18</sub> reverse phase column chromatography (30 g C<sub>18</sub> silica gel, 25 µm spherical particles, eluent: H<sub>2</sub>O+0.1% TFA (5 CV), gradient 0→100% MeCN/H<sub>2</sub>O+0.1% TFA (15 CV), t<sub>R</sub>=10.1 CV) to provide the trifluoroacetic acid salt of D-Ala-azidocoumarin (1) (11.0 mg, 79%) as a white solid. <sup>1</sup>H NMR (500 MHz, *d*<sub>6</sub>-DMSO, 25 °C):  $\delta$  8.46 (t, *J* = 6.0 Hz, 1H), 8.37 (br-s, 3H), 7.64 (s, 1H), 7.59 (d, *J* = 8.7 Hz, 1H), 7.07 (d, *J* = 2.2 Hz, 1H), 7.02 (dd, *J* = 8.6, 2.4 Hz, 1H), 4.63 (s, 2H), 4.06–4.02 (m, 1H), 3.71–3.65 (m, 1H), 3.62–3.56 (m, 1H). <sup>13</sup>C NMR (126 MHz, *d*<sub>6</sub>-DMSO, 25 °C):  $\delta$  169.2, 168.3, 159.7, 157.2, 152.4, 128.9, 127.2, 122.6, 113.4, 113.1, 101.6, 67.1, 52.2, 38.5. <sup>19</sup>F NMR (471 MHz, D<sub>2</sub>O, 25 °C)  $\delta$ 

-75.6. FTIR (thin film, cm<sup>-1</sup>): 2117 (s), 1707 (m), 1670 (m), 1618 (s), 1536 (m), 1431 (m), 1170 (m), 1141 (s). HRMS (ESI, m/z): 346.0796 (calculated for  $C_{14}H_{12}N_5O_6 [M-H]^-$ : 346.0793). [a]<sub>D</sub><sup>23</sup>= +68 (c 0.23, DMSO).

#### **QUANTIFICATION AND STATISTICAL ANALYSIS**

Statistical significance was evaluated using one- or two-way analysis of variance (ANOVA), followed by Tukey's multiple comparisons test using GraphPad Prism 8.4.0 software. Details in figure legends.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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## Highlights

• The bacterial cell wall is cross-linked by D-amino acid-containing peptides.

- β-lactam antibiotics interfere with cell wall cross-linking.
- We developed a D-amino acid-based method for introducing synthetic crosslinks.
- The cross-links protect *Escherichia coli* from broad-spectrum β-lactam antibiotics.

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Figure 1. Loss (A-E) and gain (F and G) of fluorescence strategies for detection of synthetic cross-links.

(A) Loss-of-fluorescence logic, including SPAAC control (details in Methods). (B-E) *pbp5::tn L. monocytogenes* was incubated with the indicated D-amino acids, washed and subjected to CuAAC with TBTA ligand and alkyne-carboxyrhodamine 110 (CR110) (B); TBTA ligand with azido-CR110 (C); BTTP ligand with alkyne-CR110 (D); SPAAC with DBCO-CR110 (E). Fluorescence was quantified by flow cytometry and data are representative of 2–6 biological replicates performed in triplicate. MFI, mean fluorescence intensity. Error bars, +/– standard deviation. (F), Intrapeptidoglycan reaction between alkDA and azidocoumarin-D-alanine (azcDA) results in the fluorescent triazole product. (G) *pbp5::tn L. monocytogenes* was incubated in the presence of indicated D-amino acids, washed and subjected to BTTP CuAAC with no exogenous fluorescent label. Of the 327 alkDA/azcDA-treated, CuAAC-subjected cells observed in two independent experiments, 310 were fluorescent above Dala/azcDA-treated, CuAAC-subjected background levels. PG, fluorescence derived from peptidoglycan labeling. Scale bar, 1 μM. Images are representative of 4 biological replicates.

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Figure 2. Indirect (A-B) and direct (C-E) identification of synthetic triazole cross-links.

CS802–2 *E. coli* was incubated +/– D-alanine alone (Dala) or equimolar combinations of Dalanine, azido-D-alanine (azDA), and alkynyl-D-alanine (alkDA) as indicated, washed and subjected to CuAAC with BTTP ligand and complementary fluorophore (A-B) or with the detection reagent omitted (C-E). Peptidoglycan was extracted, digested with mutanolysin and lysozyme, and separated by ultra-performance liquid chromatography (UPLC). We identified several peaks from alkDA/azDA-labeled bacteria that were specific to CuAAC treatment (red and blue boxes, (C)). Chemical structure for 5–5 triazole dimer (D) identified by mass spectrometry (MS) from red boxed peak in (C). (E) Ion detection (left) and MS profile (right) for 5–5 triazole dimer. MS/MS profile and fragmentation shown in Figure S4 and Table S1, respectively. Ion detection and MS profiles for 5–5 triazole trimers and tetramer from blue boxed peaks in (C) shown in Figure S5. Fluorescence in (A-B) was quantified by flow cytometry and data are representative of 2–6 biological replicates performed in triplicate. MFI, mean fluorescence intensity. Error bars, +/– standard deviation. UPLC analysis in (C) was performed on two biological replicates.



Figure 3. Synthetic cross-links protect CS802–2 *E. coli* from ampicillin (B) but not from kanamycin (C).

CS802–2 *E. coli* was incubated +/– the indicated D-amino acids for 6 hrs, washed and subjected (B, C) or not (A) to CuAAC with BTTP ligand (no complementary fluorophore) as in Figure 3. Bacteria were then challenged with antibiotic for 1 hr at 37 °C and plated for colony forming units (CFUs). Data are from three biological replicates performed in triplicate. Error bars, +/– standard deviation. Statistical significance was assessed by two-way ANOVA with Tukey's multiple comparison test. ns, 0.05; \*, p<0.05; \*\*, p<0.005.

#### Table 1.

Quantification of muropeptides from CS802–2 *E. coli* treated with Dala or a combination of alkDA/azDA then subjected or not to CuAAC.

	Dala		alkDA/azDA	
	-CuAAC	+CuAAC	-CuAAC	+CuAAC
azDA	-	-	$8.21 \pm 4.26$	$2.92\pm0.11$
alkDA	-	-	$4.04\pm2.38$	$1.23 \pm 1.32$
% PG Modification	-	-	$12.24 \pm 1.42$	$11.44\pm3.39$
% Cross-linkage	$53.05\pm2.04$	$54.31 \pm 3.33$	$42.76 \pm 1.57$	$51.04\pm0.23$
% Triazole Cross-linkage	-	-	-	$9.98 \pm 1.42$

<sup>b</sup>Data from two biological replicates. One of the alkDA/azDA UPLC traces is shown in Figure 2C. Compare to data for bacteria treated with Dala/ alkDA or Dala/azDA, Figure S2B.

## KEY RESOURCES TABLE

REAGENT or RESOURCE	SOURCE	IDENTIFIER		
Bacterial and Virus Strains				
L. monocytogenes EGDe	Siegrist et al., 2013	N/A		
L. monocytogenes EGDe pbp5::tn (Imo2754:tn)	Siegrist et al., 2013	N/A		
CS802–2 E. coli	Demone et al.,, 1999	N/A		
Chemicals, Peptides, and Recombinant Proteins				
D-alanine (Dala)	Sigma-Aldrich	Cat# A7377		
(R)-a-Propargylglycic (alkDA)	Acros Organics	Cat# 441221000		
3-Azido-D-alanine HCl (azDA)	Jena Bioscience	Cat# CLK-AA004		
azDlys	Jena Bioscience	Cat# CLK-AA010		
Azido-Coumarin-D-alanine (azcDA)	This paper	N/A		
azDADA (ADADA)	Liechti et al.,, 2014	https://www.einstein.yu.edu/research/shared-facilities/chemical-biology/		
alkDADA (EDADA)	Liechti et al., 2014	https://www.einstein.yu.edu/research/shared-facilities/chemical-biology/		
BTTP	Chemical Synthesis Core Facility, Albert Einstein College of Medicine, Bronx, NY	https://www.einstein.yu.edu/research/shared-facilities/chemical-biology/		
TBTA	Click Chemistry Tools, Scottsdale, AZ	Cat# 1061		
Copper (II) Sulfate, Anhydrous	Alfa Aesar	Cat# 33308		
L-Ascorbic Acid Sodium Salt	Alfa Aesar	Cat# A17759		
Carboxyrhodamine 110 Azide (Azide-CR110)	Click Chemistry Tools, Scottsdale, AZ	Cat# AZ105		
Carboxyrhodamine 110 Alkyne (Alkyne-CR110)	Click Chemistry Tools, Scottsdale, AZ	Cat# TA106		
Carboxyrhodamine 110 DBCO (DBCO-CR110)	Click Chemistry Tools, Scottsdale, AZ	Cat# A127		
AFDye 488 Picolyl Azide	Click Chemistry Tools, Scottsdale, AZ	Cat# 1276		
Ampicillin Sodium Salt	Fisher Scientific	Cat# BP1760		
Kanamycin	Sigma-Aldrich	Cat# K1377		
Carbenicillin Disodium Salt	Sigma-Aldrich	Cat# C3416		
dichloromethane	Fisher Scientific	N/A		
tetrahydrofuran	Fisher Scientific	N/A		
<i>N</i> , <i>N</i> -dimethylformamide	Fisher Scientific	N/A		
№-Boc-D-2,3-diaminopropionic acid	Chem-Impex International, Inc	N/A		
potassium carbonate	Fisher Scientific	N/A		
N,N-diisopropylethylamine	Acros Organics	N/A		
trifluoroacetic acid	Sigma-Aldrich	N/A		
Software and Algorithms				
ImageJ	Schneider et al., 2012	https://imagej.nih.gov/ij/		
GraphPad Prism 8.4.0	GraphPad	https://www.graphpad.com/scientific- software/prism/		
ChemDraw 18.1	PerkinElmer Informatics	https://www.perkinelmer.com/es/category/ chemdraw		