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Natural Products for the Treatment of Pain: Chemistry and Pharmacology of Salvinorin A, Mitragynine, and Collybolide

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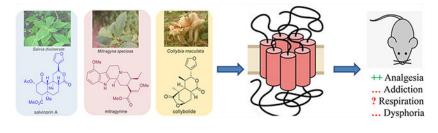
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Abstract

Pain remains a very pervasive problem throughout medicine. Classical pain management is achieved through the use of opiates belonging to the mu opioid receptor (MOR) class, which have significant side effects that hinder their utility. Pharmacologists have been trying to develop opioids devoid of side effects since the isolation of morphine from *papaver somniferum*, more commonly known as opium by Sertürner in 1804. The natural products salvinorin A, mitragynine, and collybolide represent three nonmorphinan natural product-based targets, which are potent selective agonists of opioid receptors, and emerging next-generation analgesics. In this work, we review the phytochemistry and medicinal chemistry efforts on these templates and their effects on affinity, selectivity, analgesic actions, and a myriad of other opioid-receptor-related behavioral effects.

Graphical Abstract



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More than 100 000 adults in the US are affected by chronic pain, with estimated costs up to \$635 billion per year in medical treatment and lost productivity.¹ Nonsteroidal antiinflammatory drugs (NSAIDS), acetaminophen, selective serotonin reuptake inhibitors (SSRIs), tricyclic antidepressants (TCAs), gamma amino butyric acid (GABA) analogues, and opioids are routinely used to treat pain. However, although NSAIDS work for inflammatory pain, they are associated with cardiovascular side effects, gastrointestinal (GI) bleeding, and renal disease, while acetaminophen can be hepatotoxic. On the other hand, SSRIs, TCAs, and GABA analogues show limited analgesic efficacy.² Morphine and other clinically used opioid agonists that target the MOR remain the preferred treatment of moderate to severe pain, which is attributed to MOR-mediated hyperpolarization of nociceptive pathways and CNS pain processing centers.³ Drugs targeting MOR are effective analgesics when used appropriately but are also highly addictive and are associated with serious side effects such as tolerance or respiratory depression.^{4–6} As the use of opioid painkillers has increased so has the diversion, misuse, and transition to illicit opioids, with about 80% of addicts reported initiating their habit through prescription opioids. The epidemic of opioid abuse has caused more than 47 600 deaths in 2017 alone, making drug overdose the leading cause of accidental death in the US.⁷ As effective analgesics are essential to minimize pain and suffering of many diseases, the identification of safer analgesic molecular targets with diminished side effects and abuse potential is critical for breaking the vicious cycle fueling the current problem. Accordingly, the generation of safe and abuse-free opioid analgesics represents a long-standing scientific challenge of major health and societal importance, with added urgency due to the ongoing opioid crisis epidemic.

Some approaches in the opioid field have targeted other opioid subtypes like the kappa opioid receptor (KOR). Early compounds targeting KOR⁸ can cause aversion, dysphoria, and hallucinations, although targeting this receptor may eventually become a viable option in the near future. Delta opioid receptor (DOR) agonists⁹ have been investigated; early analogues precipitated seizures, and the more recently developed derivatives have not yet been validated clinically.⁴ Other approaches have been taken over the years, starting with the development of partial agonists¹⁰ and mixed agonist/antagonists.^{11,12} Another is to take advantage of the biased agonism, in which distinct downstream pathways can be activated by different molecules working through the same receptor.^{13,14} It has been proposed that biased ligands not recruiting β -arrestin2¹⁵ or showing a preference for activating specific G-protein-mediated signal transduction pathways against β -arrestin2 might have diminished side effects.^{16,17}

In search of novel analgesic targets, opioid chemists have looked into natural products to develop safer analgesic agents. Natural products have traditionally provided numerous leads, which has led to the design of new pharmaceuticals. Together natural products and their analogues account for approximately 34% of approved drugs.^{18,19} Morphine, the most commonly employed opioid and the baine, the structure on which the vast majority of semisynthetic opiates is based, are natural alkaloids found in the poppy plant, *Papaver somniferum*. While opioid chemistry has traditionally been dominated by the bainederived alkaloids isolated from poppy,²⁰ there is a growing number of opioid natural products derived from structures other than the traditional morphinan scaffold and thus structurally

not closely related to morphine. These include analogues of salvinorin A (1, Figure 1)-, mitragynine (3)-, and collybolide (4)-based alkaloids derived from kratom; each of these will be thoroughly discussed in this review.

SALVINORIN A

Salvinorin A (1), a neoclerodane diterpenoid, is the main active compound isolated from the leaves of *Salvia divinorum*.^{21,22} *Salvia divinorum* is a member of the mint family²³ and has been traditionally ingested as a quid or smoked in spiritual practices by the Mazatec people of Oaxaca (Mexico), for many centuries. It was uncovered to be a potent and selective KOR agonist.^{24,25} Its hallucinogenic activity in humans is similar to the potency of other known compounds such as tetrahydrocannabinol (THC) or lysergic acid diethylamide (LSD), respectively, which target 5HT2A.^{26–28} The KOR agonistic activity of **1** was interesting because of its lack of structural similarity to the other psychotomimetic substances. A different mechanism was anticipated for its activity, and its congeners were expected to be the lead compounds for the development of drugs and for the treatment of pain, obesity, and other pathologies. Analogues of salvinorin A are usually potent selective KOR agonists over MOR/DOR, while some other C2-substituted aroyl ring analogues are MOR G-protein-biased agonists, and these features have been described in the literature along with salvinorin A pharmacology.^{29–32}

Phytochemistry.

The major terpenes present in *Salvia divinorum* responsible for its biological activity isolated from the plant include salvinorin A (1) and salvinorin B (2, Figure 1).^{21,22} Valdés et al. described the isolation of the terpenoid divinorin A and its congener divinorin B.²² Structural comparison of divinorin A with 1 isolated by Ortega et al.²¹ found structures to be quite identical. Therefore, divinorins A and B were named as salvinorin A and B, respectively. Further studies by Valdés et al. on *Salvia divinorum* isolated a different neoclerodane diterpene, namely, salvinorin C (5, Figure 2).³³ Additional phytochemical investigations on that plant led to the isolation of salvinorins D–I (6–11, Figure 2),^{34–37} divinatorins A–F (12–17, Figure 2),^{35,36,38} salvinicins A and B (18 and 19),³⁹ and salvidivins A–D (20–23, Figure 2).³⁶

KOR affinity has been evaluated for most of the naturally occurring analogues of $1.3^{5,40,41}$ Most of them displayed poor binding activities toward the KOR with K_i values >10 000 nM. $3^{5,40,41}$ Though there are a few exceptions, the affinity ($K_i = 1022$ nM) of salvinorin C (**5**) was found to be 250-fold lower compared with the affinity of **1** ($K_i = 4$ nM).⁴⁰ Both divinatorin D (**15**) and divinatorin E (**16**) exhibited a reduced affinity for KOR ($K_i = 230$ nM and 418 nM respectively), compared to **1**.³⁵ Among other natural products, salvinicin A (**18**, KOR $K_i = 390$ nM) and salvidivin A (**20**, (KOR $K_i = 440$ nM) have been reported in the literature to have KOR affinity. Compound **20** (KOR $K_e = 440$ nM) was also recognized as the first naturally occurring neoclerodane with KOR antagonist activity, albeit its potency is very weak.⁴¹

Pharmacology of Salvinorin A (SalA).

SalA is a KOR selective ligand.⁴² In radioligand binding assays, its affinity at KOR is 18 nM, while showing >10 000 nM affinity at both MOR and DOR. In functional assays, it showed an EC₅₀ of 7 nM and full agonism at KOR, while showing >10 000 nM potency at both MOR and DOR and recruited β -arrestin-2 similar to the classical KOR agonists.⁴³

SalA has been shown to exhibit short-lasting hallucinations with a potency similar to LSD.⁴⁴ A smoked dose of 200–500 μ g of SalA produces hallucinations with a peak effect lasting between 5 and 10 min up to an hour.^{27,28} In primate studies, SalA crosses the blood–brain barrier (BBB) efficiently within the 40s, while lasting in the brain for 8 min mimicking the peak hallucinating effects in humans.⁴⁵ Human psychopharmacology and dose-related effects of salvinorin A in humans were also reported by Griffiths^{46,47} and Ranganathan groups.⁴⁸ The peak effect of inhaled Sal A was observed at 2 min, and actions lasted up to 20 min. Sal A behaved like a classical hallucinogen in these studies, and dose-related memory impairment and dissociative effects were also reported. No euphoria but psychomimetic effects along with an increase in prolactin and cortisol levels consistent with SalA's KOR effects were observed. In general, Sal A was well tolerated, and no adverse effects were observed.

SalA has several putative metabolic sites like C2 acetyl, C4 ester, and a lactone ring. Ester hydrolysis of the C2 acetate results in the formation of salvinorin B,^{45,49} which is less active at KOR.^{33,50,51} The C4 ester has been found to be enzymatically stable, while the lactone group is also reported to be labile.⁵² Taken together, no active metabolites are believed to be responsible for SalA actions.

SalA produces KOR-mediated short-acting analgesia in mice (consistent with the molecule's lability and short-acting hallucinogenic effects in humans), which was blocked by both KOR antagonists⁵³ and was lost in KOR KO mice showing that its effect is mainly mediated through the KOR.⁵⁴ It produced conditioned place aversion (CPA) and other side effects like depression, anxiety, and hypolocomotion similar to most KOR agonists.^{31,55} Medicinal chemistry efforts in the opioid field have focused on the syntheses of probes with higher metabolic stability, which dissociate the hallucinations from the analgesic actions mediated by this template. A special emphasis has been made toward increasing the half-life of this agent.

Total Synthesis.

The most attractive structural feature of the molecule is having seven asymmetric centers and five oxygenated functionalities. Also, there are issues of epimerization at C8 under acidic or basic reaction conditions.²⁹ The significant biological activity and the inherent novel architecture have prompted scientists to investigate the total synthesis of salvinorin A.

Synthetic studies are quiet limited,^{56–58} possibly due to difficulties related to the construction of the complex core of SalA. There are few strategies in the literature for the synthesis of SalA.^{59–61} The first synthesis of the molecule was reported by Evans et al. in 29 steps based on the transannular sequential Michael strategy of 14-membered macrocyclic lactones.⁶²

Cascade.—In this approach by Evans et al., the two main precursors for the construction of the SalA core were aldehyde **24** and vinyl iodide **25** (Scheme 1). Aldehyde **24** was synthesized from thiazolidinethione **26** in 14 steps, and vinyl iodide **25** was made from ketone **27** in just 4 steps. Chelate controlled addition of the Grignard reagent prepared from iodide **25** was reacted with the aldehyde **24** to produce allylic alcohol **28** as a 7:1 mixture of diastereoisomers. Several protecting group manipulations and hydrolysis of the ester group afforded acid **29**. Macrolactonization of acid **29** using the Shiina protocol,^{63,64} followed by desilylation and oxidation, resulted in macrocycle **30**. A selective transannular reaction cascade was induced for the construction of tricycle **31**. Treatment of lactone **30** with TBAF furnished tricycle **31** as a single diastereomer. A deoxygenation sequence including enol triflate formation, palladium-mediated triflate reduction, and sequential conjugate reduction was employed to produce ketone **32**, epimeric at C8. Deprotection of both the acetals in **32** followed by oxidation and esterification resulted in **33**. Epimerization of **33** at C8 by K₂CO₃ to give **2** (salvinorin B), followed by acylation using Ac₂O, finally completed the total synthesis of salvinorin A (1).

Ten-Step Synthesis of 20-Nor-salvinorin A.—Both semisynthesis and total synthesis of SalA come across the configurational lability of the C8 carbon. Epimerization takes place easily at C8 to a lower affinity isomer, 8-epi-SalA.⁶⁵ In order to stabilize the scaffold and to prevent the epimerization, C20 deletion was carried out, which simultaneously stabilized the SalA skeleton, made the synthesis simpler, and retained its high affinity and selectivity for the KOR. The synthesis of 20-norsalvinorin A was achieved in 10 steps⁶⁶ from Hagemann's ester, a commercially available building block for terpene synthesis.⁶⁷

Commercially available tert-butyl (4-chlorobutoxy) dimethylsilane was converted to corresponding Grignard reagent 34 and was treated with ester 35 (Scheme 2). The enolates resulting from the conjugate addition were trapped immediately by the addition of acrolein in the presence of zinc chloride to produce **36** as a 6:1 mixture of allylic alcohols. Mesylation followed by the elimination of alcohol **36** by the addition of DBU afforded **37** as a 20:1 mixture of (E) and (Z) dienones. tert-Butyldimethylsilyl removal of 37 with 2 M HCl, followed by Swern oxidation of the deprotected alcohol furnished aldehyde 38. The next step was intramolecular Michael addition of the corresponding pyrrolidine enamine in THF/ MeOH. The reaction was quenched by K_2CO_3 to shift the equilibrium for the formation of predominantly one isomer, **39**. Pinnick oxidation of aldehyde **39** resulted in acid **40**. Deprotection of **40** with LDA followed by Davis oxaziridine⁶⁸ addition generated the axial α -hydroxy-decalone. By modification of reaction conditions, the hydroxy group was selectively acetylated to give acid 41. Carboxylic acid 41 was found to undergo very efficient Heck arylation as their alkali salts: the potassium carboxylate resulted in the highest yield of 42. The final step required the lactonization of the carboxylic acid to an electronrich conjugate double bond with maintaining Markonikov regioselectivity and equatorial stereoselectivity. This was done by the application of hexafluoroisopropanol (HFIP), and 20nor-salvinorin A (43) was obtained in a 63% yield as an easily separable 4:1 diastereomeric ratio.

Total Synthesis of Salvinorin A via the Diels–Alder Strategy.—A concise total synthesis of salvinorin A starting from 3-furaldehyde (44, Scheme 3) was reported by Metz et. al.⁶⁹ They used highly diastereoselective intramolecular Diels–Alder reactions (IMDA) as the key steps.

3-Furaldehyde (44) was converted to cycloadduct 45 in 6 steps (Scheme 3). The synthesis of vinyl iodide 46 was achieved from bicyclic lactone 45 in 5 steps. Stille coupling of vinyl iodide 46 with vinylstannane afforded triene 47, the precursor for IMDA. Triene 47 was heated at 200 °C in chlorobenzene for 3.5 days in a sealed tube with 1.2 equiv of BHT to furnish the desired diastereoisomer 48 in a 66% isolated yield along with a minor amount of other isomers. OsO₄-mediated dihydroxylation (proceeded from sterically less hindered face) of cyclohexene 48 followed by silylation with TESCl predominantly (86:14) gave regioisomer 49. Ley–Griffith oxidation of compound 49 afforded ketone 50 as a single regio- and diastereoisomer. TBAF-mediated desilylation of ketone 50 gave 2-*epi*-salvinorin B (51) in a quantitative yield. Finally, the Mitsunobu inversion of 51 with acetic acid produced salvinorin A (1) in 81% yield along with 18% of 2-*epi*-1.

Structure-Activity Relationships of Salvinorin A Analogues.

Elucidation of Salvinorin A SAR from Total Synthesis.: Though several total synthetic approaches have been described, only approaches used by Sherwood et al.⁵⁸ and Roach et al. ⁶⁶ have aided in analogue synthesis and allowed accessing positions on a template not easily derivatizable using semisynthesis. 20-nor-SalA **43** (Figure 3), for example, had similar activity as SalA (a 7-fold decrease in both affinity and potency).⁶⁶ Purely unsubstituted phenyl analogues of 20-norSalA, **52**, retained the same binding affinity as their furyl counterparts.

Phenyl analogue **52** had a 3-fold lower affinity than SalA and showed an 18-fold loss in activity.⁶⁶ The thiophene compound **53** also showed a high binding affinity.⁶⁶ Some other analogues were also synthesized to evaluate SAR in this scaffold.⁶⁶ Analogue MOM ether **54** had significant structural divergence from SalA, while still retaining high activity at the KOR.⁵⁸ The replacement of C20 with H and a cyclohexanone in place of the C ring lactone stabilized the SalA scaffold relative to its C8 epimer. This new compound, O6C-20-nor-SalA (**55**), retained high potency for agonism of KOR (equipotent to U69,693 and 4–10-fold less potent over Sal A).⁷⁰ Thus, using the novel total synthesis, the oxidatively labile C7 furan ring and metabolically unstable lactone ring can be substituted with more stable entities. *In vivo* characterization of these more stable analogues in rodent models of analgesia is not reported, though with the chemistry established this template may be utilized for future drug development.

Modification at C2.—The most common modification of SalA is the replacement of the C2 acetate group, which is believed to be metabolically labile and contributes to the short time action of salvinorin A.^{29,49} Efforts thereby have been to substitute this group with substituents that enhance stability while retaining KOR activity of the parent. Reaction conditions for hydrolysis of the acetate group at C2 have been developed, which has afforded easy entry to a broad range of substitution patterns.⁷¹ Carbonates, carbamates,

different ester groups, amines, amides, ethers, sulfonic esters, sulfonamides, and thioesters have all been made and evaluated (Figure 4) for their activity. Some of the initial analogues comprise alkyl esters. Propionate 56 (Figure 4) exhibited up to a 5-fold drop in affinity and a 4-8-fold loss in activity.^{65,72,51,73} Increasing the bulk of the ester moiety resulted in further loss: isopropyl ester 57 lost affinity 10-fold,⁷³ and *tert*-butyl ester 58 was inactive.⁵¹ On the contrary, the reduced bulk of formic ester 59 resulted in about a 5-fold drop in affinity and a 7-11-fold loss in activity.^{40,74,75} Replacing the acetate ester with alkyl ethers lowered KOR activity. Methyl ether **60** lost 120–170-fold affinity,^{65,72} whereas ethyl ether **61** bound KOR 6–23-fold less efficiently.^{65,72,76} Analogues designed like thiocyanate **62** (pharmacology described in the next section) or bromoacetate 63 improved on SalA's affinity (3-fold and 1.2-fold better, respectively) and potency (250-fold and 2-fold enhanced, respectively).⁷⁷ Chloroacetate 64 exhibited comparable G-protein potency (pharmacology described in the next section) compared to SalA.⁷⁷ Aryl esters with a bulkier group like 65 (discussed in next section) and 66 resulted in a loss of affinity and potency at KOR but, in several cases, led to high MOR activity.^{73,78-80} Incorporation of H-bond donors at C2 of SalA generally led to the loss of binding. Acetamide 67 bound to KOR 16-110-fold less over SalA,65,73 while the corresponding N-methylacetamide 68 led to the retention of SalA's affinity and potency.^{65,76} Analogues with H-bond donors like having methyl sulfonic ester 69 (pharmacology in discussion in the next section) was comparable in affinity and potency to SalA.⁷³ The phenyl sulfonic ester 70 showed a 32-fold reduction in KOR affinity.⁷⁹ Other acetate replacements such as thioacetate 71 lost its affinity and potency by 3-25-fold and 2-fold, respectively.^{81,82} Various unsaturated esters were also investigated such as ester 72 with $a-\beta$ unsaturation, which showed reduced affinity (6-fold), whereas β - γ -unsaturated esters were more potent, as in **73** (3-fold worse binding) and **74** (2-fold less binding).⁸⁰

Alkoxy methyl ethers (**75–78**, pharmacology discussed in the next section) showed improved affinities and potencies compared to the parent agent SalA. Ether **75** bound 2–4-fold more efficiently and was 5–8-fold more potent than SalA.^{75,83–85} Among the others alkoxyl methyl ethers, compound **76** (13–63-fold more potent) had higher affinities and potencies over SalA.^{75,84} Tetrahydropyran **77**, where the rotation was restricted by introducing a six-membered ring, did not significantly aid with respect to either binding or potency at KOR.^{75,84} Additional substitutions at C2 as in **78** exhibited a 4-fold loss in affinity relative to SalA and around a 10-fold loss relative to the parent ether **75**.⁸⁵ The presence of an H-bond acceptor in the malonate **79** appeared to be beneficial, demonstrating a 3-fold enhancement in affinity relative to SalA, but a 27-fold drop in potency compared to SalA.⁸⁶ Finally, restriction of bond rotation of the acetate as in spirolactone **80** resulted in a 3-fold loss in potency.⁸⁷ Microsomal stability studies showed that **80** was more metabolically resistant over SalA.⁸⁷

Pharmacology of Few C2-Substituted Salvinorin A Analogues. RB64 and

RB48.—These two compounds were initially designed⁷⁷ to covalently couple KOR to map out the binding site of KOR. Both compounds hold an electrophilic handle, thereby one with a thiocyanate moiety (as in compound **62**, RB64, Figure 4) and the other with a chloromethyl group (compound **64**, RB48, Figure 4). RB48 showed comparable G-protein potency (EC₅₀ = 8.8 nM, E_{max} = 101%) compared to SalA (EC₅₀ = 5.2 nM, E_{max} = 100%)

and, however, showed reduced potency as well as efficacy in the β -arrestin2 assay⁵⁰ (EC₅₀ = 143 nM, $E_{\text{max}} = 63\%$) compared to SalA (EC₅₀ = 5.8 nM, $E_{\text{max}} = 100\%$). Similarly, RB64 showed a bias for G-protein signaling (EC₅₀ = 5.2 nM, E_{max} = 101%) over the β -arrestin2 pathway (EC₅₀ = 391 nM, E_{max} = 104%). RB64 was evaluated *in vivo* in mice and found to be analgesic without triggering sedation and anhedonia but was associated with CPA. The analgesic effect of RB64 was lost in KOR KO mice, while CPA induced by balanced agonists U50,488h, SalA, and RB64 was retained in both WT as well as β -arrestin2 KO mice. The Roth group hypothesized that p38MAPK implicated in KOR-induced dysphoria^{88,89} can possibly be activated by other transducing pathways independent of β arrestin2 or aversion arises from signaling distinct from the p38MAPK pathway. Studies with this probe also suggest that the G-protein pathway may mitigate motor coordination from analgesia, but dysphoria may also be dependent on the G-protein pathway. The results are comparable to a diphenylethylamine-based analogue HS666,⁹⁰ another KOR biased agonist, which shows CPA in addition to analgesia and attenuated locomotor behavior, and in contrast to triazole 1.191,92 and HS665, which show a more complete dissociation of KOR-induced dysphoria and sedation in mice models arguing for more KOR-biased ligands to be synthesized and investigated.

Mesyl Salvinorin B.—Replacement of the acetyl group by the H-bond donor mesylate group resulted in the design of analogue 69 (Figure 4), which was comparable at KOR affinity ($K_i = 2.3$ nM vs $K_i = 1.9$ nM) and potency (EC₅₀ = 30 nM vs EC₅₀ = 40 nM) to SalA.⁷³ Mesylate **69** was found to be a full agonist at KOR and showed less β -arrestin-2 recruitment than other balanced agonists.⁷³ In the antinociception assays, mesyl SalB was not as potent compared to SalA in reducing pain, though the analgesic time course of action⁹³ was more consistent with other C2-substituted analogues, which replace the labile acetyl group.⁹⁴ Compound **69** did not produce sedation, aversion, or anxiety in rats; although, in the forced swim test, increased immobility was detected, indicating prodepressive effects.⁹⁴

MOM Salvinorin B.—Replacement of the acetoxy group at C2 by methoxymethyl ether^{72,83} (**75**, Figure 4) improved both the affinity and potency for KOR. In comparison to SalA, **75** exhibited higher binding affinity ($K_i = 0.4$ nM vs 1.3 nM for U50,488h, and 1.4 nM over SalA) for KOR. In [³⁵S]GTP γ S functional assays, compound **75** showed a potency (EC₅₀ = 0.6 nM) nearly 7 times greater than SalA (**1**) (EC₅₀ = 4.5 nM), while also being a full agonist of the KOR.⁹⁵ Noteworthy, this compound was found to be a balanced agonist with β -arrestin-2 recruitment similar to U50 and 488h. The substitution of the C2 acetyl group with the MOM group led to enhanced potency as well as the enhanced analgesic duration of action (120 min vs 20 min⁵⁴ for SalA), possibly due to decreased metabolism. The antinociceptive effect was blocked by the KOR antagonist norbinaltorphimine (norBNI). Typical KOR side effects like motor coordination and CPA were still detected with this molecule.⁹⁶ Interestingly, analogue **76** (Figure 4) showed potent analgesic activity with reduced KOR side effects like anxiety, depression, and locomotor activity at the highest doses tested.⁹⁷

β-Tetrahydropyranyl Ether Salvinorin B.—Prisinzano and co-workers hypothesized that, on the SalA scaffold, the presence of more flexible groups at C2 can lead to different structural conformations, while interacting with KORs.⁷⁵ In order to explore KOR affinity and potency, the concept of conformational restriction was applied for the development of a new analogue, β -tetrahydropyranyl ether of salvinorin B (77, Figure 4). This strategy did not significantly affect the binding affinities toward the KOR. The new analogue tetrahydropyran 77 showed slightly higher affinity ($K_i = 6.2 \text{ nM}$) over SalA ($K_i = 7.4 \text{ nM}$) at KOR. In the [³⁵S]GTP γ S functional assay, compound 77 showed a potency (EC₅₀ = 60 nM) almost similar to that of SalA (1) (EC₅₀ = 40 nM). Also, exchange with this tetrahydropyran group at C2 led to potent anti-inflammatory (reducing both phase 1 as well as phase 2 inflammatory pain in formalin test) analgesic effects along with a reduction in paclitaxelinduced neuropathic pain.98 This compound was additionally 5-fold more potent than U50,488h and equipotent to SalA in acute thermal pain assays. Taken together, this particular analogue exhibited potent analgesic actions in both acute as well as chronic pain models while also showing some separation of KOR-induced side effects. 77 showed classical CPA associated with KOR agonists but interestingly showed attenuated prodepressive phenotype, hypolocomotion, and anxiety compared to typical KOR agonists. 97

Herkinorin.—The introduction of a benzoyl group (**65**, Figure 4) in SalB core resulted in a 47-fold loss of KOR affinity ($K_i = 90$ nM vs $K_i = 1.9$ nM) compared to Sal A (**1**).⁷³ This modification also led to a 25-fold increase ($K_i = 12$ nM vs $K_i > 1000$ nM) in MOR affinity compared to **1**. In [³⁵S]GTP γ S functional assays, herkinorin was found to exhibit 30-fold less potency (EC₅₀ = 1320 nM vs EC₅₀ = 40 nM) as a KOR agonist compared to **1**, and also displayed agonism at MOR (EC₅₀ = 500 nM and $E_{max} = 130\%$).⁷³ This compound is one of the very few agents with a non-nitrogenous chemical scaffold, which can act as a MOR ligand. Subsequent *in vitro* assays showed that this compound does not recruit β -arrestin-2 and showed no internalization of MOR⁸¹ though this observation has been challenged recently.⁹⁹ However, the lack of central analgesic actions¹⁰⁰ has prevented detailed characterization of its *in vivo* pharmacology. Swapping the benzoyl group with benzamide led to the synthesis of herkamide (**67**, Figure 4), a molecule that retained high potency and selectivity at MOR over KOR. Finally, **67** robustly recruited β -arrestin-2 and showed MOR internalization, suggesting that small changes at C2 can lead to differential G vs arrestin signaling.

Kurkinorin.—A new analogue kurkinorin (**81**, Figure 5) was synthesized by the introduction of an additional degree of unsaturation between C-2 and C-3 in herkinorin (**65**). ⁷⁸ In cAMP assays, the additional unsaturation to the scaffold of herkinorin led to a potent MOR agonist with $EC_{50} = 1.2$ nM.⁷⁸ This analogue was extremely selective for MORs (>8000-fold selectivity over KOR) compared to morphine (66-fold selectivity over KOR) and herkinorin (4-fold selectivity over KOR). Additionally, kurkinorin has similar potency compared to prototypic MOR agonist DAMGO, while also showing a bias for the G-protein pathway.⁷⁸ In the tail-flick assays, **81** produced a significant antinociceptive effect with potency and peak analgesic effects similar to morphine.⁷⁸ Compound **81** also displayed reduced tolerance, sedation, and rewarding properties in comparison to morphine.⁷⁸ A more

recent study on the same template was recently reported where a p-CH₂OH¹⁰¹ (compound **82**, Figure 5) substituent was placed on the phenyl ring of kurkinorin. This particular compound retained the MOR over KOR selectivity and G-protein-biased activity of kurkinorin while showing higher MOR potency (100× over morphine). The analgesic actions in thermal pain assays were MOR-dependent, and similarly to kurkinorin, this agent showed reduced tolerance *in vivo*. The respiratory depression potential of C2-substituted aroyl G-biased analogues has not been investigated, though the reduced tolerance potential shows many promises and indicates that further optimization may lead to more molecular probes in order to study MOR signaling.

Modification at C4.—Epimerization of the C8 position is very common during the selective cleavage of the methyl ester at C4 and requires the separation of these diastereomers during chemical synthesis. Analogues that hold a bulkier substituent in this position generally display poor binding affinities for KORs. Transformation of the methyl ester of SalA (1) to bulkier alkyl esters like propyl ester 83 (Figure 6) demonstrated a total loss of affinity ($K_i > 1000$ nM).¹⁰² On the other hand, other related functional groups like a carboxylic acid (84)^{103,102,104} or an aldehyde (85)¹⁰⁴ also resulted in a complete loss of affinity. The cyclopropyl ester 86 exhibited a 170-fold loss in affinity and 80-fold drop in potency relative to SalA.¹⁰⁴ Substituting the ester to ethers $(87)^{104}$ or amines $(88, 89)^{65}$ led to a complete loss of affinity. Although alkyl esters with a bulkier moiety almost consistently caused complete loss of binding, incorporation of polar groups led to less drastic changes in affinity. MOM ether 90 exhibited moderate affinity (77-fold drop from SalA) and activity (13-fold drop from SalA).¹⁰² Instead of the methyl ester, the presence of an H-bond donor such as with the amide **91** demonstrated a significant loss (540-fold) of affinity.⁶⁵ Dimethylamide 92, which is not an H-bond donor, did not show any KOR binding.⁶⁵ Increased steric bulk with an amino acid as in the alanine derivative 93 exhibited a 21-fold drop in affinity and a 10-fold loss in potency. On the other hand, serine derivative 94 was completely inactive.¹⁰² In conclusion, the C4 position is less amenable to modifications and appears critical for KOR affinity and function though there are exceptions like, for instance, with the compound **54** (Figure 3).

Modification at C12.—Additional work has focused on the role of the furan ring. Alteration of the regiochemistry of the furan ring as in **95** (Figure 7) retained an affinity for KOR and decreased potency by 4-fold relative to the parent compound SalA.¹⁰⁵ Hydrogenation of the furan ring as in **96** retained the high affinity and activity at KOR similar to the parent template. The *R* epimer had a similar affinity for KOR as **1** but was 17-fold less potent over $1.^{40,41}$ Connecting the furan with the opposite stereochemistry at C12, as in **97** (12-*epi*-1), led to a loss of affinity (2–16-fold).^{106–108} Thiophene analogue **98** was found to exhibit 4-fold less affinity and 16-fold less activity.¹⁰⁵ Replacement of the thiophene ring (**98**) by a phenyl as in **99** resulted in a 10-fold loss in affinity and a 57-fold loss in potency. Additional testing found the *meta*-carboxamide analogue **100** to demonstrate 5-fold less affinity and 18-fold less activity.¹⁰⁵ Reduction of the ketone is well tolerated as alcohol **101** retained affinity and activity at KOR (8-fold less affinity, 17-fold less activity).¹⁰⁶ The oxanorbornadiene derivative **102** has also been made by the Diels–Alder reaction with alkynes. This sterically demanding oxanorbornadienes showed only an 8-fold loss in

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affinity relative to SalA, while the corresponding substituted benzene analogue **103** was found to have a 39-fold loss in affinity.¹⁰⁹ Grafting of a bromine in the furan ring as in **104** displayed affinity and potency comparable to SalA.^{41,106,110,111} Bromide **104** was also used as a coupling partner to further substitute the furan ring. Substitution with a vinyl group as in compound **105**, was found to decrease potency by 2–32-fold,^{106,110} while the related alkyne analogue **106** retained potency similar to the parent SalA.^{110,111}

Future Directions.: The SalA template has been investigated in detail over the years by medicinal chemists. One of the key questions that remain unanswered is the binding mode of SalA. SalA has no structural similarity to arylacetamides or dynorphins (endogenous KOR modulators) and, most importantly, lacks a basic nitrogen. How does a lipophilic terpene bind and activate a receptor that usually interacts with alkaloids? The active state KOR structure,¹¹² solved by the Roth group using MP1104,¹¹³ a molecule designed by the Majumdar group, provides some evidence that the salt bridge between MP1104 tertiary nitrogen, and D138^{3.32} may not be necessary for binding to KOR. Mutation of this residue to the alanine A leads to a total loss of activity for dynorphins (endogenous KOR modulator) but retains binding of MP1104 as well as SalA, and the loss of activity for both MP1104 and SalA is similar, ~15-fold. An active state structure of SalA is required to map out the binding pose of SalA and to elucidate how it activates KOR specifically.

On the chemistry side, efforts have primarily focused on regions readily accessible through semisynthesis, and further structural diversification with the syntheses of novel analogues might shed light on the interactions with the opioid receptors. Similarly, the concept of partial agonism¹⁰ on this template needs to be evaluated in cell lines, not overexpressing the receptor. KOR antagonists, ligands with DOR affinity, are virtually unknown on this template, and KOR-biased agonists are rare. *In vivo* pharmacology of furan ring replacements is currently understudied, and the effects of MOR-biased ligands in this template on respiration are currently not known.

MITRAGYNINE

The psychoactive plant *Mitragyna speciosa* has been traditionally used for many years by people in Southeast Asia to treat a wide variety of illnesses. This plant is known as "kratom" in Thailand and "biak biak" in Malaysia. The plant material is either chewed directly or consumed as a tea. More than 30 different alkaloids with indole moiety have been identified and isolated from this plant.^{114–117} The major alkaloid mitragynine (**3**, Figure 8) has been found up to 66% by mass of crude alkaloids. Paynantheine, speciogynine, and speciociliatine have been found to be the other major alkaloids in the plant. Depending on the age of the plant and different geographical varieties, the quantities of these major alkaloids can considerably change. A wide variety of minor alkaloids are also found in this plant.¹¹⁴

Recent studies show that kratom¹¹⁸ and its alkaloids mitragynine (**3**), 7-OH mitragynine (7-OH, **109**), and mitragynine pseudoindoxyl (MP, **110**)^{119,120} (a spirocyclic compound that can be obtained by a skeletal rearrangement of 7-OH under Lewis acidic conditions), are MOR modulators exhibiting bias toward G-protein signaling.^{119,121} Orally administered

mitragynine in mice metabolizes to 7-OH mitragynine via a CYP3A-mediated mechanism. ¹²² An *in vivo* study of kratom and its alkaloids showed that they are analgesic, ¹¹⁹ block alcohol intake in mice, ¹¹⁸ and also prevent heroin self-administration in rats. ¹²³ Studies from McCurdy and co-workers¹²⁴ also show that intravenous (i.v.) mitragynine is not self-administered. ¹²³ Taken together, mitragynine and its analogues represent promising starting points toward the development of therapeutics for the treatment of pain.

Total Synthesis.

Both total synthesis and partial synthesis have been explored on the mitragynine scaffold. ^{125–127,121,128,129} In this review, we describe the first asymmetric total synthesis of mitragynine by Takayama and co-workers¹²⁹ and will also detail the enantioselective total synthesis of both (–)-mitragynine and its unnatural enantiomer, (+)-mitragynine by Sames and co-workers.¹²¹

Asymmetric Total Synthesis of Mitragynine by Takayama.—The total synthesis of mitragynine by Takayama was initiated from the synthesis of the optically pure alcohol (*R*)-111 (Scheme 4).¹²⁹ The other counterpart was 4-methoxytryptophyl bromide (112). It was prepared from 4-hydroxyindole via a five-step sequence. Optically pure pyridine derivative (R)-111 and bromide 112 were condensed in refluxing benzene in the presence of catalytic NaI. The resulting pyridinium salt **113** was then reduced by using sodium borohydride to furnish two diastereomers (114 and its C3 epimer) in 33% and 27% isolated yields, respectively. Allylic alcohol 114 was then subjected to Claisen rearrangement to incorporate an acetic acid residue at the C15 position. Treatment of alcohol 114 with trimethyl orthoacetate in refluxing o-xylene in the presence of a catalytic amount of benzoic acid resulted in acetate **115** as a single product. From the CD spectra, the absolute configuration at C3 in **115** was determined. At C3 and C15 positions, compound **115** had the appropriate absolute configuration for further transformation into mitragynine. A formyl group was next introduced at C16 in **115** by using the conventional (LDA, HCO₂Me) method to give compound 116. The formyl group in 116 was then converted to the dimethyl acetal derivative 117. Treatment of the acetal 117 with KO'Bu in DMF furnished the methyl enol ether 118 in a 71% yield. Stereoselective reduction of the double bond of 118 at the C19–20 positions finally produced the target compound, mitragynine (3), with the natural absolute configuration.

Total Synthesis of Both Enantiomeric Forms of Mitragynine.—The synthesis by the Sames group¹²¹ started from 3,4-dihydro- β -carboline (**119**, Scheme 5), which was successfully synthesized in six steps starting from commercially available 4-methoxyindole. The required enone (**120**) was prepared from methyl 2-ethyl-3- oxobutanoate adopting literature procedures. A proline-catalyzed Manich–Michael-type cyclization was performed to form ring D. In the presence of proline, β -carboline **119** was reacted with an excess of enone **120** to afford the desired ketone isomer **121S** in 59% (along with **122S**) isolated yield and with excellent enantiomeric excess. Carbanion derived from methyl diethylphosphonoacetate was reacted with ketone **121S** to furnish the desired eneester (with axial ethyl group) as a mixture of *E* and *Z* isomers, **123E** and **123Z** in 37% (along with another stereoisomer **124**) isolated yield. Simultaneous reduction and detosylation of the

mixed ene-esters 123E/Z with magnesium followed by transesterification produced ester 125. By using the conventional (LDA, HCO₂Me) method on ester 125, a formyl group was incorporated to give enol-ester 126 in a 57% yield. Finally, *O*-methylation of the enol-ester intermediate 126 provided (–)-mitragynine (3) in 27% yield along with the isomeric analogue (Z)-mitragynine (127) in 31% yield. Following the same reaction sequence, the unnatural enantiomer, (+)-mitragynine (128), was also synthesized starting from ketone 121R.

Pharmacology and Structure–Activity Relationships of Mitragynine Analogues.

Elucidation of Mitragynine SAR from Total Synthesis.—The total synthetic approach from Sames's group has allowed access to positions in the natural product not readily available by semisynthesis and elucidate the molecular determinants of binding and function at the opioid receptors. Some of the key synthetic modifications were carried out at the β -methoxyacrylate moiety and at the ethyl group on ring D (Figure 9, SAR exploration). The importance of the absolute stereochemistry at C3, C15, and C20 were also investigated. 121

In terms of both efficacy and potency, the unnatural enantiomer (+)-mitragynine (128, Scheme 5) was found to be a far weaker agonist at human MOR (hMOR).¹²¹ It was found to be a partial agonist with low potency at human KOR (hKOR) in comparison to naturally occurring (-)-mitragynine (3). Switch from antagonistic to agonistic activity at hKOR was identified by the inversion of the stereochemistry in this scaffold. The stereochemical inversion of the β -methoxyacrylate moiety from E to Z (as in Z-mitragynine, 127, Scheme 5) was found to exhibit almost similar activity compared to the natural product. However, complete removal of the enol ether as in compound 129 (Figure 9) completely abolished the activity at hMOR (both agonist and antagonist). The desethylmitragynine analogue 130 retained agonistic activity at hMOR, but with lower potency. As such, the substituent at C20 is critical both in terms of efficacy (agonist vs antagonist) and potency. Most of the synthetic derivatives (127, 128, 129, and 130) were also found to be inactive at hKOR and hDOR.¹²¹ The enol derivative 131 and acid derivative 132 were also synthesized and studied for their activity. Compound 131 was found to be more efficacious than mitragynine (3) for activation of hMOR but ~3-fold less potent, while compound 132 was completely inactive as an agonist at concentrations up to $100 \ \mu$ M.

Pharmacology and SAR of Mitragynine Analogues.—In CHO cells expressing transfected MOR, mitragynine was found to have moderate affinity ($K_i = 230$ nM vs 3.3 nM for DAMGO) and potency (EC₅₀ = 203 nM vs 19 nM for DAMGO).¹¹⁹ Mitragynine was also found not to recruit β -arrestin2 up to a 10 μ M concentration. Similar affinity, potency compared to prototypic opioids, and biased activity have been reported by other groups independently.^{121,128,130,131} Antinociceptive properties exhibited by mitragynine (**3**) were most extensively investigated by Macko et al. in rodents and dogs initially in 1972.¹³² Mitragynine was active as an analgesic (comparable potency with codeine) after oral (o.p.) or intraperitoneal (i.p.) administration in all species, but when administered subcutaneously (s.c.) in both mice and rats, mitragynine was mostly inactive. Recent reports, however, contradict these findings.^{122,133}

Matsumoto and co-workers in 1996 studied the analgesic mechanism of action caused by mitragynine (**3**) with a different approach.¹³⁴ Investigations of antinociceptive activity after i.p. and intracerebroventricular (i.c.v.) injections were performed using the tail-pinch and hot-plate tests. A dose-dependent antinociceptive activity was observed for mitragynine (5.0 —30 mg/kg, i.p. and 1.0—10m g/mouse, i.c.v.) with a peak effect at 15—45 min after injection. The antinociceptive activity of i.p. mitragynine was completely eliminated by both s.c. and i.c.v. administered naloxone. Naloxone administered i.c.v. also antagonized the analgesia of i.c.v. mitragynine. These results indicate that supraspinal analgesic actions of mitragynine in mice are typically MOR-mediated. In more recent studies, s.c. mitragynine was found to be analgesic but was found to exhibit weaker potency (analgesic ED₅₀ > 100 mg/kg) in CD1¹¹⁹ and 129S1 mice.^{122,134}

In addition to the MOR, it has been found that other nonopioid receptors play a role in the analgesic actions of mitragynine. Most notably, studies by Matsumoto¹³⁵ et al., using the tail-pinch and hot-plate tests in mice, show that, with mechanical noxious stimulation, antinociception of mitragynine involves both descending noradrenergic and serotonergic systems; however, upon thermal noxious stimulation, the activity of mitragynine comes from the predominant contribution of the descending noradrenergic system.¹³⁵ A more recent study from McCurdy and co-workers¹³³ suggests that mitragynine antinociceptive effects at larger doses disrupts learned behavior. These disruptive effects of mitragynine on learned behavior did not appear to be mediated by opioid receptors but by adrenergic receptors. Thus, mitragynine pharmacology comprises substantial nonopioid mechanisms and suggests that the major alkaloid in kratom, mitragynine, has a pharmacological mechanism that differs from that of classical opioids.

Further investigations from Javitch/Majumdar/Sames groups found that an active metabolite is responsible for the analgesic effects of mitragynine.¹²² It was found that mitragynine is converted to the much more potent MOR agonist 7-OH (pharmacology discussed in a later section) in mouse and human liver preparations by cytochrome P450 3A isoforms. Mitragynine was converted to 7-OH in mice, and the concentration of this metabolite in the brain is sufficient to explain all opioid-receptor-mediated analgesic activity.¹²² Conversion of mitragynine to 7-OH was also reported *in vitro* by the McCurdy group.¹³⁶

The molecular scaffold of mitragynine has been explored through semisynthetic approaches. In comparison to common corynanthe-type alkaloids, the presence of a C9 methoxy group on the indole ring of mitragynine (**3**) is a structural characteristic of *Mitragyna* alkaloids. Corynantheidine (**133**), a naturally occurring kratom alkaloid,¹³⁷ also known as 9-demethoxymitragynine (Figure 10), was devoid of opioid agonistic activity in guinea pig ileum preparation.¹²⁸ From these findings, it appeared clearly that the C9 methoxy group in **3** is essential for producing the analgesic activity. No opioid agonistic activity was observed with corynantheidine, but the compound was able to reverse the morphine-inhibited twitch contraction in the guinea pig ileum.¹²⁸ It also showed an interesting concentration-dependent antagonistic effect. A very recent study¹³⁰ concluded that mitragynine had a higher affinity at opioid receptors than at adrenergic receptors, while the exact opposite was observed for corynantheidine.

Based on these results, the chemical diversification of the C9 function in mitragynine (**3**) has been reported. The 9-demethylation of mitragynine afforded 9-hydroxycorynantheidine (**134**), which binds to MOR with moderate affinity, while functional assays revealed it was a partial agonist at opioid receptors.^{128,138} Thus, the transformation of the C9 substituent of mitragynine, from OCH₃ to OH to H, led to a change of activity from full agonism to partial agonist and then ultimately to antagonism at MOR.

Among other analogues, compounds **135** and **136**, with an elongated carbon chain on the C9 position instead of the methyl group induced naloxone-insensitive inhibition of twitch contraction, suggesting an inhibitory effect via mechanisms distinct from those of the stimulation of opioid receptors.¹²⁸ No opioid agonistic activity was observed for compound **137**, the MOM-ether analogue of mitragynine. The grafting of an acetoxy group at C9 of mitragynine (compound **138**) showed a marked reduction of intrinsic activity as well as potency compared to parent mitragynine (**3**). Thus, the C9 position acts as a functional switch in controlling receptor intrinsic activity at MOR and a C9 methoxy group is the most optimal substituent for pharmacophore binding to opioid receptors.

Takayama and co-workers also synthesized an ethylene glycol-bridged C10-fluorinated mitragynine (MGM-9, **139**, Figure 10).^{139,140} MGM-9 showed very high affinity for both MOR and KOR. MGM-9 exhibited high affinity with $K_i = 7.3$ nM for MOR (compared to DAMGO $K_i = 1.2$ nM). The KOR affinity of this analogue was also measured with $K_i = 18$ nM (compared to U69,593 $K_i = 0.66$ nM). It showed potent orally active antinociceptive effects in thermal antinociception assays (7–22-fold higher than morphine), while producing less reward in the condition place preference paradigm and tolerance, possibly due to dual agonism at MOR and KOR.¹⁴⁰ Taken together, these results suggest MGM-9 is a promising innovative analgesic with a robust analgesic effect and fewer adverse effects over morphine.

7-Hydroxymitragynine (7-OH).—The oxidation product of mitragynine, 7hydroxymitragynine, also known as 7-OH (**109**, Figure 8), which was isolated as a minor constituent from *Mitragyna speciosa*, is a potent opioid analgesic alkaloid.¹⁴¹ 7-OH acts as a full or partial agonist on the opioid receptors depending on the assays or cell line used *in vitro*.^{119,121,131} In comparison to mitragynine (**3**), the introduction of a hydroxy group at C7 led to a higher affinity and potency for MOR. Compound **109** has a moderate affinity (37 nM vs 3.3 nM for DAMGO and 230 nM for mitragynine) and selectivity for MOR over KOR/DOR in radioligand binding assays.^{119,122,142} Compound **105** like other mitragynine template-based derivatives shows G-protein-biased activity at MOR.^{119,121,131} When administered subcutaneously, **109** exhibited a potent antinociceptive effect through activation of MOR in several thermal antinociception assays.^{119,122,143} In mice models of tolerance, dependence,¹⁴³ GI transit,¹⁴⁴ place preference,¹³¹ and self-administratione,¹²⁴ it behaves similarly to classical opiates. The effect of respiration with 7-OH mitragynine has not been reported yet in the literature.

The Takayama group also reported two analogues of 7-OH where the imine was reduced (C=N reduction at C1–2, Figure 11), namely, MGM 15 and MGM-16 (compound **140** and **141**, respectively, in Figure 11). MGM-15 and MGM-16 acted as dual agonists at MOR/ DOR.¹⁴⁵ *In vitro* and *in vivo* assays showed that the potency of MGM-16 was higher in

comparison to MGM-15 and 7-OH. For both MOR and DOR, MGM-16 showed high affinity ($K_i = 2.1$ nM compared to DAMGO $K_i = 1.2$ nM) and ($K_i = 7$ nM compared to DPDPE $K_i = 1.2$ nM) for DOR. In GTP γ S assays, full agonistic effects were observed with MGM-16 for both MOR and DOR. MGM16 was active in acute pain models such as the tail-flick and chronic pain antiallodynia models as well. MGM-16 was approximately 240 times more antinociceptive than morphine, and this effect was MOR-mediated as well as DOR-mediated.¹⁴⁵ Last but not least, compound MGM-16 has potential therapeutic utility for the treatment of neuropathic pain suggestive that the mitragynine template-based analogues may find utility in treating both acute as well as chronic pain. Effects of this compound on other opioid-induced side effects were not reported yet and need to be assessed in several models in order to conclude on the potential impact of such agent.

SAR was also explored for the 7-OH scaffold by introducing different functionality at C7 instead of the hydroxy group.¹²⁸ Installation of an acetoxy group at C7 (compound **142**) reduced the intrinsic activity over mitragynine (**3**), but potency remains almost equal to mitragynine (**3**).¹²⁸ Significant reduction was observed in both intrinsic activity and potency when a methoxy or an ethoxy group was incorporated at C7 as in compounds **143** and **144**, respectively.¹²⁸ It can be concluded that a C7 hydroxy group is necessary for the improved potency of this scaffold to opioid receptors, and this hydroxyl group may H-bond with residues in the MOR pocket.¹¹⁹

Mitragynine Pseudoindoxyl (MP).—An indole alkaloid related to mitragynine, mitragynine pseudoindoxyl (MP, **110**, Figure 8), was first isolated in 1974 by Zarembo et al. as a microbial fermentation product of mitragynine (**3**) from the fungus *Helminthosporum sp*.¹⁴⁶ This is an oxidative rearrangement product of 7-OH with a spirocyclic core. In later reports by Yamamoto and co-workers, this compound acted non-selectively on MOR and DOR, whereas its affinity on KOR was negligible.¹⁴⁷ Later on, the *in vivo* supraspinal analgesic activities of **110** were briefly discussed by Takayama et al.¹²⁸

SAR of the MP scaffold, detailed investigations, and analgesic actions were reported by the Majumdar group in 2016.¹¹⁹ Modifications at both C9 and N1 positions were explored. C9 modifications (compounds 145–152, Figure 11) did not significantly affect receptor affinities, while the incorporation of OAc at C9 slightly decreased MOR and DOR affinities. ¹¹⁹ None of the derivatives of MP were associated with β -arrestin-2 activation, suggesting that compounds in this template are totally G-protein-biased. Various substituents at C9 were found to maintain nanomolar binding activities (1-3 nM compared to DAMGO 3 nM) and full MOR agonism ($EC_{50} = 1-4$ nM compared to DAMGO 19 nM). Efficacies at MOR were not affected either by the removal of the methoxy group (145) or by C-9 O-demethylation (146). Compounds 145 and 146 were full agonists at MOR compared to DAMGO. However, by changing the substituents, the activity at DOR receptors was affected differentially. DOR antagonism was retained for compounds 145-152, but the C9-phenyl analogue, 148, appeared to be a DOR agonist. Compound 148 exhibited dual MOR/DOR agonism with comparable intrinsic activity and potency at both receptors. The introduction of bulky groups at N-1 as in N-benzyl analogue (151) and N-methyl analogue (152) showed reduced affinities at all of the three opioid receptors compared to parent pseudoindoxyl (110),

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suggesting a free NH is required for binding to opioid receptors.¹¹⁹ The equipotent binding with C9 analogues for MOR also suggests that space around this position in the MOR pocket.

In vivo, the 9-OH analogue (**146**) was more potent than **110**. Analgesic potency was found to increase by the removal of the methoxy group (compound **145**). Potencies of analogues with C9-cyano (**147**), C9-phenyl (**148**), and C9-furan-3'-yl (**149**) were almost similar to **110**. A slightly negative effect was observed with the C9-acetate (**150**).¹¹⁹

The *in vivo* analgesic actions of MP have also been characterized in detail. MP was analgesic in tail-flick and hot-plate assays administered subcutaneously with potency similar or 3–5-fold greater than morphine using the same route. MP was, however, short-acting compared to morphine lasting 60-90 min compared to 150 min at equianalgesic ED₈₀ doses. Supraspinally, it was equipotent to DAMGO and morphine but appeared to have a ceiling effect compared to morphine and DAMGO which where full agonists in the tail-flick assay through the same route. The analgesic actions were MOR opioid-dependent in vivo consistent with *in vitro* functional assays in which it was agonist at MOR, while being an antagonist at DOR and KOR. In contrast to classical MOR agonists, MP showed no reinforcing properties at 10× analgesic ED₅₀ doses and showed reduced respiratory depression. A ceiling effect was seen in GI transit assays compared to morphine where the drug had reduced constipation. Among its most promising feature, the drug showed far less analgesic tolerance and physical dependence compared to morphine. Tolerance appeared with morphine within 5 days, while it took 29 days for MP to develop tolerance on acute administration. Reduced tolerance was also seen on chronic administration. The promising opioid functional selectivity is possibly rooted in two mechanisms its G-protein bias at MOR in conjunction with DOR/KOR antagonism.

Future Directions.

The mitragynine template provides chemists and pharmacologists unique opportunities to develop novel pain therapeutics as well as identify mechanisms which separate analgesia from other opioid-induced side effects. The ability of i.v. mitragynine to block opioid selfadministration while not being addictive on its own suggests a metabolically stable mitragynine analogue not converting to 7-OH mitragynine and retaining the pharmacological properties of the parent may be useful toward next-generation opioid modulators. The role of MOR partial agonism and adrenergic actions toward mitragynine pharmacology needs to be better understood. Similarly the mode of binding and functional activation by a template lacking the classical phenolic group seen in enkephalins and other morphinans needs to be investigated. At present, computational studies from the Filizola group¹²¹ suggests that the β -methoxyacrylate moiety mimics the H297^{6.52} interaction seen with phenol of Bu72¹⁴⁸ and DAMGO¹⁴⁹ in MOR. Among other analogues, MP with its dual properties of G-bias at MOR and KOR/DOR antagonism is of interest. The roles of Gprotein bias¹⁵⁰ as well as DOR antagonism^{151–153} in reducing MOR-mediated tolerance and dependence is well established in the field. It is however less clear if MP is addictive in selfadministration models unlike other kratom alkaloids like mitragynine (administered sc), ^{123,124} and 7-OH mitragynine.¹²⁴ Similarly beyond the C9 and C10 positions of the aromatic

ring of mitragynine, no SAR is known at other positions including the C20 position, which is possibly critical for receptor efficacy. It is hoped that future investigations will aim at delving into mapping the other positions combining total synthesis with semisynthesis in order to understand the molecular mechanisms that lead to MOR activation with this template.

COLLYBOLIDE

Dr. Pierre Potier's research group in 1974 first extracted the natural product collybolide (**4**, Figure 1) from the fungus *Collybia maculata*.¹⁵⁴ Collybolides represent the first examples of sesquiterpene structures with the furyl- δ -lactone motif. Collybolide shares structural similarity, particularly a familiar furyl- δ -lactone core, with SalA (shown in blue in Figure 12).¹⁵⁵ Other natural products isolated in this series include 9-epicollybolide, isocollybolide, and neocollybolide. The structure of 9-epicollybolide (**153**) was wrongly elucidated in the original article¹⁵⁴ but was later corrected in a subsequent paper.¹⁵⁵ A few more collybolide-like sesquiterpenes (Figure 12) were isolated, and their structures have been assigned.¹⁵⁶

In 2016, the Devi group described the KOR activity of two terpenes, collybolide and 9epicollybolide. In the same fashion like SalA, collybolide is a highly potent and selective KOR agonist (nM affinity and agonism), while 9-epicollybolide was less active, suggesting that the stereochemistry at the 9-position plays a key role over in KOR activity.¹⁵⁵ In competitive radioligand binding assays in KOR transfected cell lines, collybolide showed partial inhibition (24-40%) of binding when either of the agonist $(^{3}H-U69,693)$ or antagonist (³H-diprenorphine/³H-naloxone) was used as the radioligand compared to Sal A, which fully competed all KOR binding sites. These results showed differences between structurally similar KOR templates, where collybolide is either a partial agonist and/or showed affinity for a subset of kappa binding sites or receptor complexes, while SalA labels all kappa sites. In functional assays (GTP γ S and adenyl cyclase inhibition), collybolide was uncovered to be a KOR agonist and is less efficacious than SalA. In other downstream signaling assays, collybolide was found to internalize KOR,¹⁰⁸ similar to Sal A, suggesting that collybolide is either not a G-protein KOR agonist or that others mechanisms are involved in receptor internalization. β -Arrestin2 recruitment was not reported with this agonist, which would be of importance in order to shed light on the pharmacology of this agent. Treatment of KOR cells with agonists usually leads to ERK1/2 phosphorylation. SalA and collybolide showed differences in this assay with SalA showing a sigmoidal curve and collybolide showing an inverted-shaped curve. Robust ERK 1/2 phosphorylation is seen initially at low doses, but levels of phosphorylation dip at higher doses, and collybolide was more potent by 100-fold compared to SalA in this assay. In another phosphorylation assay, namely, Akt at S473 and T308, collybolide and SalA were similar, both ligands showing sigmoidal curves, and collybolide was more efficacious than SalA.

In mice, collybolide was analgesic, and the time course of action was similar to SalA, which is not surprising given the number of metabolically labile groups present in both molecules. In other KOR actions *in vivo*, collybolide was active in inhibiting itch similar to the nitrogenous KOR agonists¹⁵⁷ and showed CPA,¹⁵⁸ while acting as an antidepressant and anxiogenic in mice models like the forced swim test and open field instead of being a

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prodepressant¹⁵⁹ and anxiolytic like classical KOR agonists in the nitrogenous class and non-nitrogenous class like SalA.

KOR selectivity for collybolide was then probed in binding assays and GTP γ S assays, and *in vivo* for blocking chloroquine induced itching in KOR KO mice. In all of these assays, the effects were attenuated, suggesting KOR actions both *in vitro* and *in vivo*.

Taken together, these data show that collybolide is a novel non-nitrogenous KOR agonist analgesic with subtle but important differences compared to classical nitrogenous KOR ligands in the arylacetamide class like U50,488h and non-nitrogenous ligands like SalA. In rodents, it retained the aversive actions, which have limited the usage of KOR agonists as analgesics. However, given its antidepressant properties and unique binding mode, distinct signaling properties in the ERK1/2, and AKT phosphorylation assays compared to SalA, analogues of collybolide might be of use in the development of novel and safer kappa analgesics.

Exploration of the SARs on this template are missing, and the syntheses of metabolically more stable analogues (especially on the lactone group) are required. Total synthesis on this template would help in accessing other critical positions of this scaffold (which are not readily accessible by semisynthesis) and as such would be of tremendous interest. In particular, the oxidatively labile furan group must be substituted. The molecular structure binding mechanism and activation of collybolide on KOR is required to probe this agent, which could even bring more information if compared to the MP1104 KOR solved structure. ⁴³ Other natural products like isocollybolide and neocollybolide may provide additional avenues to probe KOR function. Analogues which retain signaling (ERK 1/2 and Akt) seen with the parent template and potentially with less β -arrestin-2 recruitment^{88,160} and/or leading to less internalization of KOR may be necessary to separate aversive actions from analgesia.

ROLE OF BIASED AGONISM IN OPIOID FUNCTIONAL SELECTIVITY

Several analogues of salvinorin A, mitragynine, and collybollide (itself) are G-proteinbiased opioid agonists. The role of G-protein-biased signaling is fiercely debated in the opioid field.¹⁶¹ The role of MOR opioid-induced respiratory depression linked to recruitment of β -arrestin2¹⁶² has been questioned. Biased agonists like PZM21⁹⁹ still show respiratory depression,¹⁶³ while three recent reports in β -arrestin2 KO mice^{164,165} and mice with C-tail mutations¹⁶⁶ incapable of recruiting β -arrestin2 show persistence of MORmediated respiratory depression. However, the reports corroborate previous findings that analgesic efficacy of opioids is limited by β -arrestin2 recruitment.^{16,166–168}

The recent approval of the first-generation biased agonist, i.e., TRV130/oligoceridine,¹⁶⁹ allows the field to test the hypothesis in humans. Findings may allow the field to either call it a day on biased agonism or design of better probes to better delineate this pathway.

The role of biased agonism at DOR^{13,170–172} in separating analgesia from seizures (associated with classical agonists) is more promising. Recent studies with PN6047¹⁷³ show effectiveness in preclinical models of chronic pain while lacking proconvulsive activity or

analgesic tolerance. It is possible that knowledge gained from the evaluation of biased agonists at MOR may eventually lead to safer analgesics at other subtypes.

CONCLUSIONS

Natural products based upon *kratom* and *salvia* have been used in traditional medicine for more than two centuries, while much less is known about *collybia maculata*. Salvinorin A, mitragynine, and collybolide show unique receptor binding, signaling, and opioid analgesic profiles in rodents. Diversification of these templates has led to the development of a wide variety of probes aiming at dissociating opioid-receptor-induced analgesia from its physiological adverse effects, understanding polypharmacology and biased G-protein signaling, but also aimed at subtype selectivity. We hope that the next generation of probe molecules will delve on G-protein subtype bias,¹⁷⁴ allosterism,¹⁷⁵ as well as investigate the roles of endogenous peptide ligands^{176,177} in pain relief and addiction; three emerging themes in the opioid field in current times.

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ABBREVIATIONS

СНО	Chinese hamster ovary
СРА	conditioned place aversion
СРР	conditioned place preference
DAMGO	[D-Ala2, N-MePhe4, Gly-ol5]-enkephalin
DCM	dichloromethane
THF	tetrahydrofuran
LDA	lithium diisopropylamide
TBAF	tetrabutylammonium fluoride
DBU	1,8-diazabicyclo-[5.4.0]undec-7-ene
DPDPE	[DPen2, D-Pen5]Enkephalin
КО	knockout

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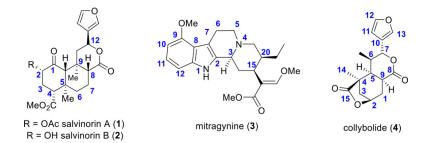


Figure 1. Structures of the natural products covered by this review.

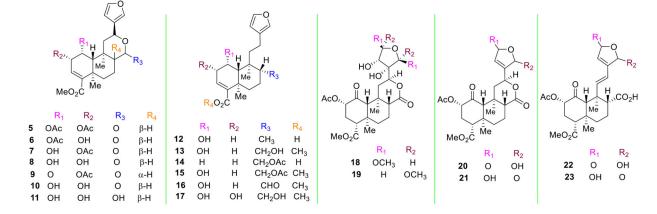


Figure 2.

Structures of naturally occurring analogues from Salvia divinorum.

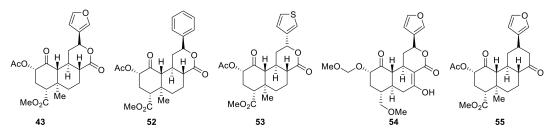


Figure 3. Selective synthetic analogues of salvinorin A.

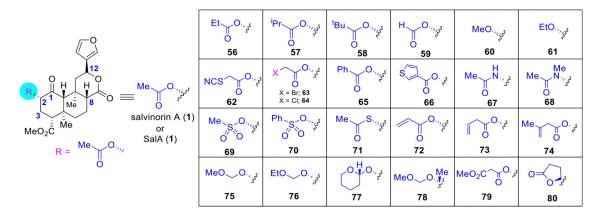
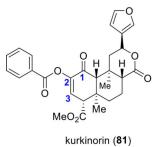
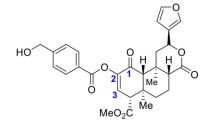
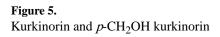


Figure 4. Selective synthetic analogues at C2 of salvinorin A.









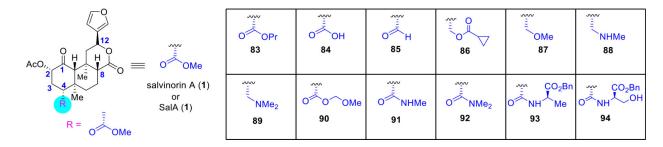


Figure 6. Selective synthetic analogues at C4 of salvinorin A.

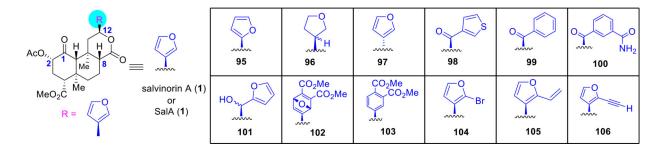


Figure 7. Selective synthetic analogues at C12 of salvinorin A.

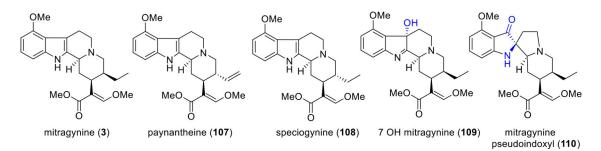
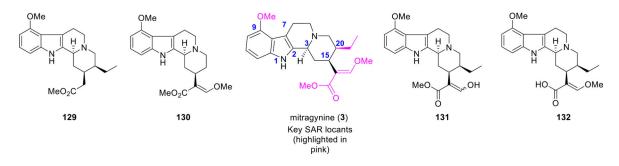


Figure 8.

Structures of major kratom alkaloids.





SAR locants and selective analogues of mitragynine from a total synthetic approach.

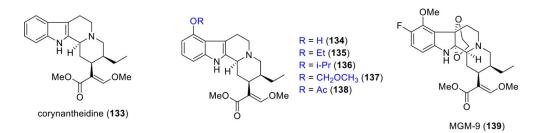
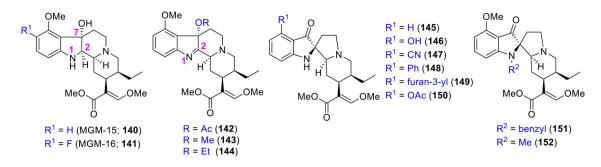


Figure 10.

Selective synthetic analogues of mitragynine.

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Selective analogues of 7-hydroxymitragynine and mitragynine pseudoindoxyl.

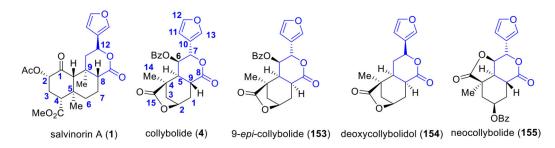
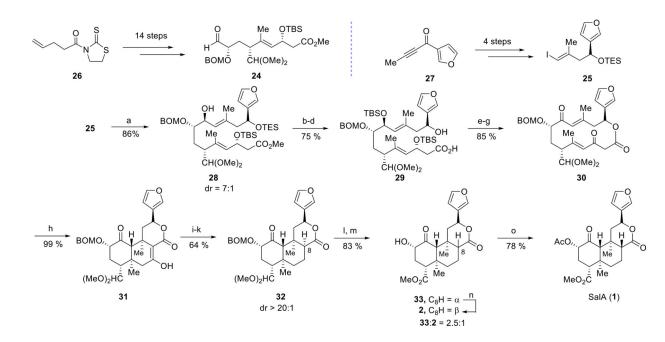


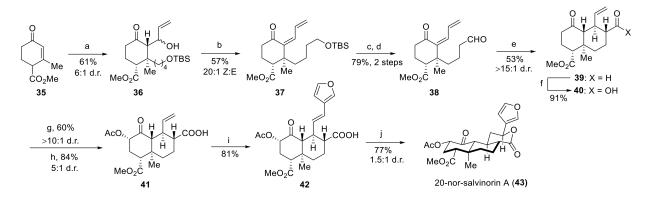
Figure 12.

Structure of naturally occurring analogues from Collybia maculata.



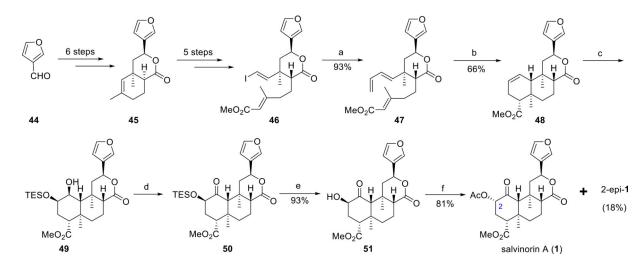
Scheme 1. Asymmetric Synthesis of Salvinorin A by Evans et al.^a

^{*a*}Reagents and conditions: (a) *n*-BuLi, MgBr₂·OEt₂, -78 °C, then 24, DCM, -78 °C to 0 °C; (b) TBSOTf, 2,6-lutidine; (c) PPTS, MeOH; (d) LiOH, *i*-PrOH, H₂O; (e) MNBA, DMAP [0.0015 M]; (f) TBAF; (g) Dess–Martin periodinane; (h) TBAF, -78 °C to 5 °C; (i) NaH, Comins reagent; (j) Pd(OAc)₂, dppf, Et₃SiH; (k) L-Selectride, *t*-BuOH, -78 °C to -55 °C; (l) LiBF₄, MeCN/H₂O; (m) NaClO₂, TMSCHN₂; (n) K₂CO₃, MeOH, quant. mass recovery; (o) Ac₂O, py, DMAP.



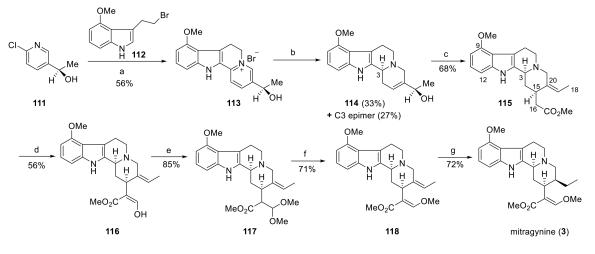
Scheme 2. Synthesis of 20-Nor-salvinorin A by Roach et al.^a

^{*a*}Reagents and conditions: (a) ClMg(C₄H₈)OTBS (**34**), CuBr·DMS, THF, HMPA, -78 °C to 0 °C, acrolein, ZnCl₂, -78 °C; (b) MsCl, Et₃N, DCM, 0 °C, DBU, 22 °C; (c) 2 M HCl (aq), THF, 0 °C; (d) (COCl)₂, DMSO, Et₃N, -78 °C to 22 °C; (e) pyrrolidine, AcOH, THF/ MeOH, 65 °C, K₂CO₃; (f) NaClO₂, *t*-BuOH, NaH₂PO₄, C₅H₁₀; (g) LDA, -78 °C, Davis oxaziridine; (h) Ac₂O, DMAP, DBU, 22–80 °C; (i) 3-bromofuran, Pd(OAc)₂, XPhos, K₂CO₃, DMF, 80 °C; (j) (CF₃)₂CHOH, 100 °C, 1 h.



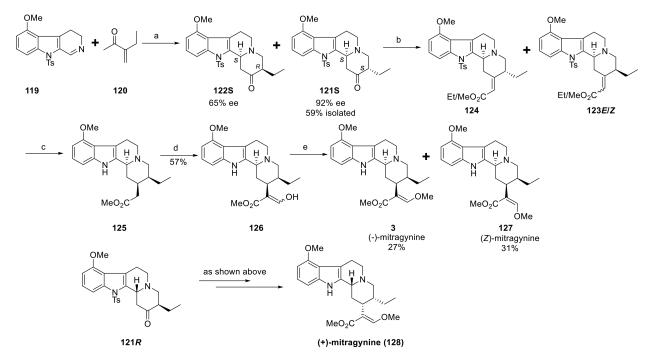
Scheme 3. Total Synthesis of Salvinorin A by Metz et al.^a

^{*a*}Reagents and conditions: (a) tributyl(vinyl)tin, Pd(MeCN)₂Cl₂, NMP, 0 °C; (b) PhCl, 1.2 equiv of BHT, 200 °C; (c) (i) OsO₄, 3,5-lutidine, THF, toluene, 0 °C to rt, (95%); (ii) TESCl, imidazole, DCM, 38–40 °C; (d) 10 mol % TPAP, NMO, 4 Å sieves, DCM, rt; (e) TBAF, HOAc, THF, 0 °C to rt, (99%); (f) 6 equiv of HOAc, 3 equiv of PPh₃, 3 equiv of DBAD, THF, 60 °C.



Scheme 4. First Asymmetric Total Synthesis of Mitragynine^a

^{*a*}Reagents and conditions: (a) cat. Nal, PhH, ; (b) NaBH₄; (c) CH₃C(OMe)₃, cat. PhCOOH, *o*-xylene; (d) LDA, HCOOMe, THF, -78 °C, 30 min; (e) HCl in MeOH; (f) *t*-BuOK, DMF; (g) PtO₂/H₂, EtOH.



Scheme 5. Total Synthesis of Both (-)-Mitragynine and (+)-Mitragnine^{*a*} ^{*a*}Reagents and conditions: (a) D-proline (100 mol %), DMSO, rt, 5 days; (b) (EtO)₂P(O)CH₂CO₂Me, NaH, 1,2-DME, 0 °C to rt, 3 h; (c) (i) Mg, NH₄Cl, MeOH, rt, 1 h, yield 51% (ii) NaOMe, MeOH, rt, 1 h, quantitative (d) LDA, HCOOMe, THF, -78 °C to 0 °C; (e) (i) NaOMe, MeOH/Et₂O, rt (ii) (MeO)₂SO₂, benzene, rt, 20 h.