



# Effect of thermal and non-thermal processing on astringency reduction and nutrient retention in cashew apple fruit and its juice

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**Abstract** The present study is aimed at evaluating the effect of different processing techniques on astringency reduction, nutrient retention, and sensory attributes in cashew apple fruit and its juice. Astringency attribute was measured by tannin content, while nutrition profile by ascorbic acid, total sugars, and antioxidant activity. Hot water, steaming, and microwave were selected as the source of heat application for treating whole fruit, where the process variables were the temperature and exposure time. The non-thermal technique selected to treat juice was by using bio coagulants, i.e., dried okra pod and drumstick seed powder, where the independent parameters were concentration and settling time. The processes were optimized using a multivariate approach coupled with full factorial design. The obtained results indicated that samples, with 42.6% tannin removal, were rated as being the least astringent. The use of dried okra pod powder under optimal conditions (0.3% concentration, 0.5 h settling time) was found to be the best in reducing astringency while retaining the nutrient and desirable sensory attributes. Maximum tannin removal ( $48.9 \pm 1.6\%$ ) with

minimum loss of ascorbic acid ( $8.1 \pm 0.9\%$ ), total sugar ( $4.8 \pm 0.5\%$ ) and antioxidant activity ( $11.1 \pm 1.0$ ) with high sensory score ( $92.7 \pm 1.6\%$ ) was achieved with composite desirability of 0.85.

**Keywords** Astringency · Bio coagulant · Cashew apple fruit · Cashew apple juice · Heat treatment · Tannins

## Abbreviations

CAF	Cashew apple fruit
CAJ	Cashew apple juice
DOP	Dried okra pod powder
DDSP	Dried drumstick seed powder
HWT	Hot water treated
HT	Steam treated
MW	Microwave treated
A <sub>A</sub>	Ascorbic acid (g/100 ml)
AO <sub>A</sub>	Antioxidant activity
C <sub>n</sub>	Concentration (% , g/100 ml)
P <sub>S</sub>	Particle size (μm)
S <sub>t</sub>	Settling time (h)
TPC	Total phenol (g/100 ml)
T <sub>n</sub>	Tannin content (g/100 ml)
T <sub>S</sub>	Total sugar (g/100 ml)

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## Introduction

Cashew (*Anacardium Occidentale* L.), a tropical plantation crop, is being widely grown in several countries of Asia, Africa, and Central America as an economically important agricultural crop. The cashew nut and cashew apple are the two morphological parts of the cashew fruit. About 10–15 tons of cashew apples are obtained as a by-product for every ton of cashew nut produced. A fully developed

cashew apple contains a significant amount of polyphenols, vitamins, minerals, and organic acids, making it a functional food (Das and Arora 2017). The primary reason for the unpopularity of the cashew apple, despite its high nutritional value, is its astringent taste, which is an undesirable sensory attribute. The polyphenols, especially tannins, are responsible for the astringency (Michodjehoun-Mestres et al. 2009).

Various methods of tannins removal have been reported for cashew apple juice (CAJ) and whole cashew fruit (CAF). The methods reported for CAJ are microfiltration, enzymatic treatment, blending with other fruit juice, use of coagulating agents, etc. (Talasila et al. 2012). Microfiltration is reported to be promising as the process occurs at mild operating conditions, thereby preserving heat-sensitive compounds present in the juice. However, given the high tannin content of the CAJ, the ultra-thin membranes of 0.1 and 0.2  $\mu\text{m}$  pore size used for microfiltration get quickly fouled and rendered useless (Abreu et al. 2005). Pre-treating the juice with enzymes (tannase/cellulase) before microfiltration makes the process efficient as it reduces viscosity and fractionates the insoluble polysaccharides (Cormier 2008). Despite many advantages, the technology is not being adopted by cashew processors in India due to lack of resources to access this sophisticated technology. Several studies report the blending of fresh/untreated CAJ with other tropical fruit juice as a promising technique to reduce astringency (Marc et al. 2017). However, our lab study (unpublished data) indicated when beverages made from fresh CAJ blended with other tropical fruit juice, the organoleptic scores of mixes were significantly low ( $p < 0.05$ ). Hence, it is desirable to reduce tannins from fresh CAJ before blending for better sensory acceptability. The use of coagulating agents is reported to be effective and efficient in removing polymerized tannins and phenols (Pena-Neira 2019). Polyvinyl pyrrolidone (PVP) and gelatine are widely used coagulants for reducing astringency from CAJ. However, the PVP is reported to be expensive and not readily available while the use of gelatine is of concern as it comes from animal sources (Talasila et al. 2012; Hameed et al. 2018). Currently, the growing concern with environmental issues has raised the interest in bio coagulants because they are the low-cost source, abundant in nature, and non-toxic. Jayalekshmy and John (2004) reported cassava starch as an efficient and economic coagulating agent for treating CAJ. The flip side of the process is long clarification time, i.e., more than 8 h (Cormier 2008). As reported by several researchers, dried okra pod powder (DOP) and dried drumstick seed powder (DDSP) showed promising results as bio coagulants (Okolo et al. 2015). However, there is no literature available on the use of DOP and DDSP to treat CAJ. There are studies with respect to astringency removal, from CAF, using hot water

and steam (Mosha et al. 1995; Akinwale and Aladesua 1999).

To the best of our knowledge, none of the reported studies have investigated the nutrition profile and sensory attributes of treated CAJ and CAF in addition to astringency level. The effectiveness of astringency removal method must be quantified by maximum reduction in tannin content while simultaneously retaining its nutritional components as maximum as possible.

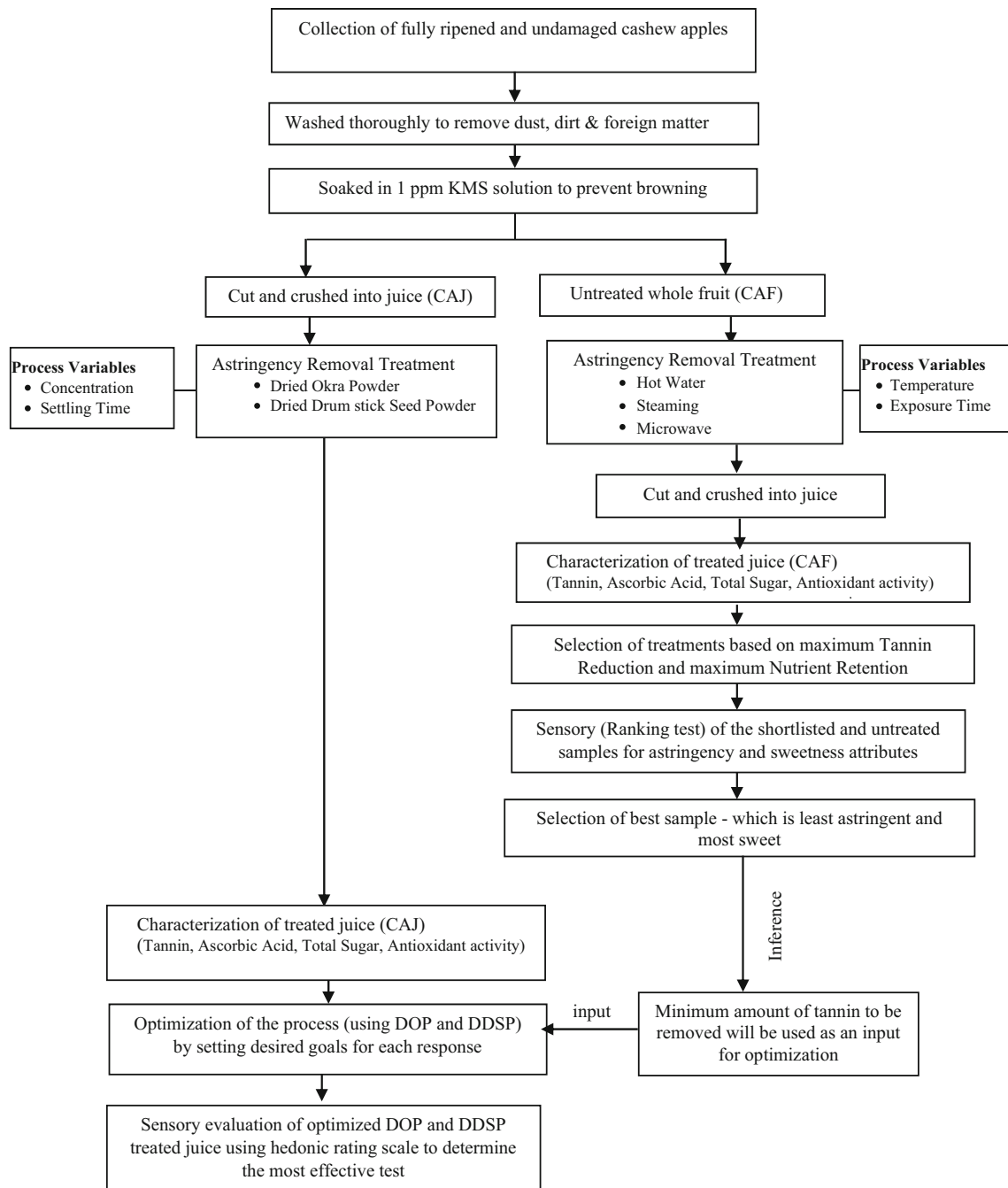
Hence, the present study aims at investigating the effectiveness of different techniques selected for CAF and CAJ, with respect to the final tannin content, nutrient profile (ascorbic acid, total sugar and antioxidant activity), and sensory evaluation (astringency, sweetness and overall preferences). The heat treatment methods, selected for CAF, were hot water (HWT), steaming (HT), and microwave (MW). The non-thermal technique, selected for treating CAJ, was the use of bio coagulants, DOP and DDSP. The objective also involves optimizing the process parameters using multivariate approach, to maintain an equilibrium between astringency level, nutritional quality, and sensory attributes.

## Materials and methodology

The experiment has been divided into two phases. During the 1st phase, CAF were treated with various thermal techniques. The effective treatments were shortlisted manually, on the basis of maximum tannin removal with the highest possible retention of the nutrient levels (ascorbic acid, total sugar, and antioxidant activity). The shortlisted samples were further subjected to sensory testing which assisted in inferring the minimum amount of tannin to be removed from the sample for it to be organoleptically acceptable. During the 2nd phase of the experiment, CAJ was treated with bio coagulants. The data obtained was subjected to optimization by setting the goals for each response variables (tannins and nutrient profile). The value for tannin removal obtained from the 1st phase was used as one of the inputs. The sensory test was conducted on the optimized samples to determine the most effective treatment. The detailed experimental methodology is described below, and also represented in Fig. 1.

## Procurement and processing of cashew apple

The fully ripened, undamaged, and fresh cashew apple (Variety: Yellow) collected from the local agricultural farms near Goa (India) ( $14^{\circ} 53' 54'' \text{N}$ ,  $74^{\circ} 20' 13'' \text{E}$ ) were used for the experiments. The fruits were washed thoroughly to remove foreign materials and then soaked in preservative solution (1 ppm potassium metabisulphite



**Fig. 1** Schematics of the experimental methodology followed

(KMS)) until further treatment to prevent browning. The physico-chemical characteristics are shown in Table S1.

**Techniques investigated for removal of astringency**

*Cashew apple fruit (CAF)*

The different heat application methods selected to remove astringency from CAF were HWT, HT, and MW. The

process variables were temperature (power level in the case of MW) and exposure time. The levels of independent variables were selected based on preliminary trials and published literature and details shown in Table 1a–c. Thermostatic water bath (Model-PB 28 L, Precision Water Bath) was used for the hot water treatment, in which CAF were placed at  $100 \pm 2$  °C for different intervals (1, 5, 10, 15, and 25 min). Autoclave (Model-7407ST) was used for the steaming treatment, in which CAF were placed at

**Table 1** Experimental details (process variables and their levels) to remove astringency from cashew apple fruit (CAF) and cashew apple juice (CAJ)

(a) Hot water treatment (HWT)*					
Independent variables			Dependent variables		
Temperature °C	Exposure time min				
100	1		• Tannin		
100	5		• Ascorbic acid		
100	10		• Total sugar		
100	15		• Antioxidant activity		
100	25				
(b) Steaming (HT)*					
Independent variables			Dependent variables		
Temperature °C	Exposure time min				
121	1		• Tannin		
121	5		• Ascorbic acid		
121	10		• Total sugar		
121	15		• Antioxidant activity		
121	25				
(c) Microwave heating (MW)*					
Variables	Name (units)	Levels			Dependent variables
		Lower limit	Centre point	Upper limit	
P <sub>w</sub>	Power level, W	180	360	540	• Tannin
E <sub>t</sub>	Exposure time, s	30	60	90	• Ascorbic acid • Total sugar • Antioxidant activity
(d) Dried Okra pod powder (DOP) and dried drumstick seed powder (DDSP)**					
Variables	Name (units)	Levels			Dependent variables
		Lower limit	Centre point	Upper limit	
C <sub>n</sub>	Concentration, g/100 ml	0.1	0.3	0.5	• Tannin
S <sub>t</sub>	Settling time, h	0.5	1.0	1.5	• Ascorbic acid
P <sub>s</sub>	Particle size, µm	< 150	> 150 and ≤ 500	> 500 and ≤ 1000	• Total sugar • Antioxidant activity

\*CAF, \*\*CAJ

121 ± 1 °C (15.1 Psi) for different intervals (1, 5, 10, 15, and 25 min). A domestic microwave oven (2465 MHz; LG Make, Intellrowave MC2886BRUM) was used for the MW treatment, in which CAF were subjected to various power levels (180 W, 360 W, 480 W) and exposure time (30 s, 60 s, 90 s). The juice extracted from treated CAF was used for further characterization.

#### Cashew apple juice (CAJ)

The juice extracted from the fresh/untreated fruits was treated with the bio coagulants i.e., DOP and DDSP. They were prepared by drying sliced okra pods and drumstick seeds in a hot air oven, at 60 °C, until the coagulant reached constant weight. This was followed by grinding and sieving to obtain different particle sizes. The process variables considered were particle size (P<sub>s</sub>), concentration (C<sub>n</sub>), and settling time (S<sub>t</sub>) varied between 150 and

1000  $\mu\text{m}$ , 0.1–0.5%, and 0.5–1.5 h, respectively and details are shown in Table 1d. The measured amount of powdered coagulating agent was slowly poured into CAJ and constantly stirred at 120 RPM for 10 min (based on preliminary experiments) to achieve uniform dispersion of coagulant in CAJ. Subsequently, the mixture was allowed to rest for coagulation and flocculation. The cleared CAJ was then filtered and collected into a glass container for subsequent nutrient and sensory analysis.

### Analysis of physico-chemical parameters

The compositions i.e., tannins ( $T_n$ ), total polyphenol content (TPC), ascorbic acid ( $A_A$ ), total sugars ( $T_s$ ), and antioxidant activity ( $AO_A$ ) of both untreated and treated samples, were analysed using the methods outlined by AOAC (2012). TPC and  $T_n$  were determined spectrophotometrically using Folin-Ciocalteu and Folin-Danis reagent respectively and expressed as mg/100 ml of juice. Tannic acid and Gallic acid were used as the standard for  $T_n$  and TPC respectively with the absorbance read at 715 and 760 nm. The 2, 6-dichlorophenol indophenol titration method was used for the measuring  $A_A$  content. Total sugars were determined using the phenol–sulphuric acid method and have been expressed as g/100 ml with absorbance read at 490 nm. The  $AO_A$  was measured using 1, 1-diphenyl-2-picryl hydrazyl (DPPH) method. Percent inhibition of free radical DPPH was calculated using Eq. 1 and then expressed as a half-maximal effective concentration ( $IC_{50}$  value), i.e., the concentration of the sample required to inhibit 50% of DPPH radical scavenging activity.

$$\text{Inhibition (\%)} = \left( \frac{A_{\text{control}} - A_{\text{sample}}}{A_{\text{control}}} \right) \times 100 \quad (1)$$

where  $A_{\text{control}}$  Absorbance at 519 nm of DPPH without sample;  $A_{\text{sample}}$  Absorbance at 519 nm of the reaction mixture containing DPPH and sample.

### Statistical analysis

The experiments were based on 3-level, 3-factor FFD generating 27 experiments for CAJ (Table 1d). The responses measured were the percent reduction in  $T_n$ ,  $A_A$ ,  $T_s$ , and  $AO_A$ . Response surface methodology (RSM) was used to develop a second-order polynomial equation for analysis of experimental data and to determine the relative contributions of variables on various responses under study. The quality of fit and statistical significance of the response function was expressed by the adjusted  $R^2$  and F-test respectively. One-way ANOVA was used to compare the percent reduction values (responses) using Tukey's test at 95% confidence interval. The significance of all the terms (linear, quadratic, and interactions) in the response surface function was judged

by computing the probability ( $p$ ) at 0.05. Numerical optimization technique using desirability function was used for simultaneous optimization of the multiple responses (Giri and Prasad 2007). The objective was to maximize, minimize, or obtain the target value of the responses within the experimental range. In our study, optimization of independent variables was carried out based on maximizing the percent reduction of  $T_n$  and minimizing the percent reduction of  $A_A$ ,  $T_s$  &  $IC_{50}$  within the data obtained in experimental range. Depending on the objective, different desirability functions were employed. The overall desirability function  $D$  is calculated as the geometric average of the individual desirability functions (Myers and Montgomery 2002) using the equation given below. The value of ' $D$ ' ranges from zero (least desirable) to one (most desirable). The data were analysed using Minitab Statistical Software version 16 (Pennsylvania State University, USA).

$$D(x) = (d_1 \times d_2 \times d_3 \times \dots \times d_i)^{1/n} \quad (2)$$

where the function  $D(x)$  reflects the desirable ranges for each response ( $d_i$ ),  $d_1, d_2, \dots, d_i$  are responses and  $n$  is the total number of responses.

### Sensory evaluation

Sensory analysis was conducted using the ranking test and hedonics rating scale as described in ISO 8587: 2006 (E) (BS ISO 8587:2006) (<https://www.iso.org/standard/36172.html>).

#### Ranking test (CAF)

The ranking test was conducted to differentiate the shortlisted treated CAF samples with untreated CAF as control sample based on astringency and sweetness attributes. The panel consisting of twenty untrained panellists with age between 24 and 35 years were asked to rank the treated samples using a quantitative scale with a score from one to four in order of intensity. The tasters were familiarized with the astringency and sweetness attributes. The evaluation was done in two sessions with 10 assessors in each session, and was completed in the same day. The samples were provided in 50 mL small cups randomly coded with three-digit numbers. Each juice sample was assessed in two replications. The least significant difference (LSD) for the sum of ranks was used for comparison between two individual products. Samples whose sum of ranks differed by more than LSD value (Eq. 3) were considered significantly different.

$$\text{LSD} = 1.96 \sqrt{\frac{K \times J \times (J + 1)}{6}} \quad (3)$$

where  $J$  and  $K$  represent the number of samples ranked and panellists, respectively.

**Table 2** Effect of different treatments on total phenols (TPC), tannins ( $T_n$ ), ascorbic acid ( $A_A$ ), total sugars ( $T_s$ ), and antioxidant activity ( $AO_A$ ) in cashew apple fruit (CAF)

Run	Temp (°C)	Time (min)	TPC	$T_n$	$A_A$	$T_s$	$AO_A$
<i>Hot water treated cashew apple (HWT)</i>							
1	100	1	31.4 ± 0.8 <sup>b</sup>	28.7 ± 1.0 <sup>b</sup>	26.7 ± 1.2 <sup>c</sup>	14.1 ± 0.6 <sup>a</sup>	50.7 ± 1.4 <sup>c</sup>
2*	100	5	38.1 ± 0.6 <sup>c</sup>	42.6 ± 0.9 <sup>c</sup>	30.0 ± 2.0 <sup>c</sup>	14.3 ± 0.5 <sup>a</sup>	56.2 ± 1.4 <sup>d</sup>
3	100	10	48.8 ± 0.7 <sup>d</sup>	42.7 ± 1.3 <sup>c</sup>	30.9 ± 0.8 <sup>c</sup>	15.4 ± 0.6 <sup>a</sup>	63.1 ± 2.1 <sup>e</sup>
4	100	15	52.9 ± 2.1 <sup>e</sup>	47.4 ± 0.9 <sup>d</sup>	50.3 ± 0.9 <sup>d</sup>	16.3 ± 0.4 <sup>a</sup>	68.8 ± 3.1 <sup>f</sup>
5	100	25	54.2 ± 1.2 <sup>e</sup>	48.9 ± 1.1 <sup>d</sup>	55.6 ± 1.1 <sup>d</sup>	16.1 ± 0.8 <sup>a</sup>	73.5 ± 2.5 <sup>f</sup>
<i>Steam treated cashew apple (HT)</i>							
1	121	1	50.2 ± 1.7 <sup>d</sup>	39.2 ± 1.6 <sup>c</sup>	61.7 ± 1.4 <sup>e</sup>	50.2 ± 0.6 <sup>b</sup>	71.1 ± 2.9 <sup>f</sup>
2	121	5	52.2 ± 0.7 <sup>e</sup>	44.9 ± 0.5 <sup>c</sup>	68.4 ± 0.6 <sup>f</sup>	52.1 ± 0.5 <sup>b</sup>	73.3 ± 2.7 <sup>f</sup>
3	121	10	66.8 ± 1.0 <sup>f</sup>	48.4 ± 1.3 <sup>d</sup>	68.9 ± 1.1 <sup>f</sup>	53.0 ± 0.4 <sup>b</sup>	78.8 ± 1.6 <sup>g</sup>
4	121	15	68.2 ± 0.6 <sup>f</sup>	51.4 ± 1.1 <sup>d</sup>	72.9 ± 0.4 <sup>f</sup>	52.1 ± 0.8 <sup>b</sup>	81.7 ± 0.8 <sup>g</sup>
5	121	25	70.1 ± 1.8 <sup>f</sup>	55.7 ± 1.3 <sup>e</sup>	73.2 ± 1.4 <sup>f</sup>	53.9 ± 0.9 <sup>b</sup>	88.3 ± 0.6 <sup>h</sup>
<i>Microwave treated cashew apple (MW)</i>							
1	180	30	17.3 ± 1.3 <sup>a</sup>	15.7 ± 1.9 <sup>a</sup>	7.1 ± 2.0 <sup>a</sup>	14.1 ± 0.4 <sup>a</sup>	32.7 ± 1.3 <sup>a</sup>
2	180	60	27.1 ± 1.4 <sup>b</sup>	27.3 ± 1.8 <sup>b</sup>	7.2 ± 2.7 <sup>a</sup>	14.6 ± 0.5 <sup>a</sup>	40.6 ± 2.1 <sup>b</sup>
3	180	90	30.1 ± 1.0 <sup>b</sup>	27.8 ± 1.9 <sup>b</sup>	9.4 ± 1.8 <sup>b</sup>	15.1 ± 0.5 <sup>a</sup>	41.2 ± 1.7 <sup>b</sup>
4	360	30	16.5 ± 1.6 <sup>a</sup>	15.0 ± 2.0 <sup>a</sup>	7.1 ± 0.3 <sup>a</sup>	14.7 ± 0.8 <sup>a</sup>	31.3 ± 1.8 <sup>a</sup>
5*	360	60	27.7 ± 0.3 <sup>b</sup>	28.5 ± 1.8 <sup>b</sup>	7.5 ± 2.7 <sup>a</sup>	15.1 ± 0.8 <sup>a</sup>	43.4 ± 2.4 <sup>b</sup>
6	360	90	36.3 ± 1.9 <sup>c</sup>	29.0 ± 2.6 <sup>b</sup>	10.0 ± 1.1 <sup>b</sup>	15.3 ± 0.4 <sup>a</sup>	54.4 ± 2.8 <sup>d</sup>
7	540	30	19.2 ± 1.2 <sup>a</sup>	16.9 ± 1.5 <sup>a</sup>	7.1 ± 1.4 <sup>a</sup>	15.1 ± 0.6 <sup>a</sup>	40.6 ± 3.1 <sup>b</sup>
8	540	60	36.3 ± 1.8 <sup>c</sup>	28.5 ± 2.6 <sup>b</sup>	9.9 ± 1.7 <sup>b</sup>	16.3 ± 0.4 <sup>a</sup>	54.9 ± 3.7 <sup>d</sup>
9*	540	90	40.3 ± 1.7 <sup>c</sup>	30.1 ± 1.3 <sup>b</sup>	11.1 ± 0.5 <sup>b</sup>	16.8 ± 0.5 <sup>a</sup>	60.9 ± 2.7 <sup>e</sup>

\*Shortlisted treatments across methods

Means that do not share a common superscript letter in a column are significantly different ( $p < 0.05$ ) from each other

### The hedonics rating scale

The hedonic test was carried out to evaluate the overall acceptability of the optimized CAJ sample according to the hedonic scale on 9 scale: extremely dislike = 1; very much dislike = 2; moderately dislike = 3; slightly dislike = 4; neither like nor dislike = 5; slightly like = 6; moderately like = 7; very much like = 8 and extremely like = 9. The acceptability index (AI) was calculated using Eq. 4, as shown below (Arruda et al. 2016). Higher the index value, the well-accepted the sample is.

$$AI(\%) = \frac{100 \times (\text{Average score obtained for the product})}{(\text{Highest score given to the product})} \quad (4)$$

## Results and discussion

The results obtained in this study have been discussed in detail in the following sections, which include the effect of treatments on CAF and CAJ on the degree of astringency removal and the retention of nutrients. Percent reduction of

$T_n$  measured the astringency level while percent retention of  $A_A$ ,  $T_s$ , and  $AO_A$  decided the nutritional attributes of the treated juice.

### Effect of applications of heat treatments on CAF

#### Tannins ( $T_n$ )

The different heat application methods (i.e., HWT, HT, and MW) were found to reduce the  $T_n$  content significantly ( $p < 0.05$ ) in CAF (Table 2). The  $T_n$  of untreated CAF was determined to be  $199.7 \pm 49.6$  mg/100 ml. The wide variation in value may be due to the inherent disparity of  $T_n$  among fruits or could be due to experiments being spread over a few months during the season. The variation has been normalized by converting the values to percent reduction to remove the confounding of individual fruit. The reduction of  $T_n$  was found to increase with the increase in exposure time, i.e., from 28.7–48.9%, 39.2–55.7%, and 15.7–30.1% for HWT, HT, and MW treated samples, respectively. The reduction was because of heat application and in agreement with previous studies, and the value was reported to be between 30–50% for steam treated lentils and



60% for hot water processed yam/sweet potato (Hefnawy 2011; Egbonu and Nzew 2016). Emelike and Eberé (2016) reported the reduction of  $T_n$  was ~96% in CAF when subjected to HWT for 20 min. The reason attributed was the predominant tannins present in CAF are hydrolyzable, i.e., soluble in water. Quite contrary to this, other researchers reported that hydrolyzable and condensed tannins constitute around 20–40% and 30–55%, respectively, in cashew apple, implying the tannins reduction phenomenon is rather complex (Talasila et al. 2012; Prommajak et al. 2014). Other significant reasons could be the structural changes due to thermal degradation and the formation of insoluble complexes between  $T_n$  and other components such as protein and minerals present in CAF (Nithya et al. 2007). The percent reduction of  $T_n$  is the least for MW treated samples (15.7–30.1), and a similar result has been reported for lentil when subjected to MW heating (i.e., 35% reduction) (Hefnawy 2011). HWT and HT were found to be comparatively more effective in reducing the  $T_n$ , i.e., 28.7–48.9% and 39.2–55.7%, respectively than MW heating. The reason could be, the moisture conducts heat better than air. Hence the heat transfer rate is higher in the case of HWT and HT compared to MW, which is a dry heat. This partly explains the effect of different heat treatments on the percent reduction in  $T_n$  as observed in the study.

#### Total phenol content (TPC) and antioxidant activity ( $AO_A$ )

It was observed that the application of heat resulted in a significant loss of phenolic compounds. The value of TPC for untreated cashew apple was found to be  $324.1 \pm 40.8$  mg/100 ml. Losses of phenols were the highest for HT treated samples (50.2–70.1) compared to HWT (31.4–54.2), and MW treated samples (17.3–40.3) (Table 2). The steaming process in the autoclave is characterized by high pressure (15.1 psi) and high temperature (121 °C), which increases the heat content of the samples and thereby diffusion rates, which promote loss of the phenols from the sample. The lower temperature of processing during HWT (100 °C) might have helped to preserve phenols a little better. Also, the degree of thermal processing is an essential factor in deciding the amount of reduction. It was observed, the percent reduction in TPC was found to increase with an increase in treatment time and temperature irrespective of the method of heat application. The results confirm with the previous studies where it has been reported that TPC of fruits and vegetables decreased significantly with the increase in temperature and time of thermal processing (Jafari et al. 2017). Several studies show a strong positive correlation between the amount of polyphenols and antioxidant activity ( $AO_A$ ), which implies that polyphenol contributes to radical

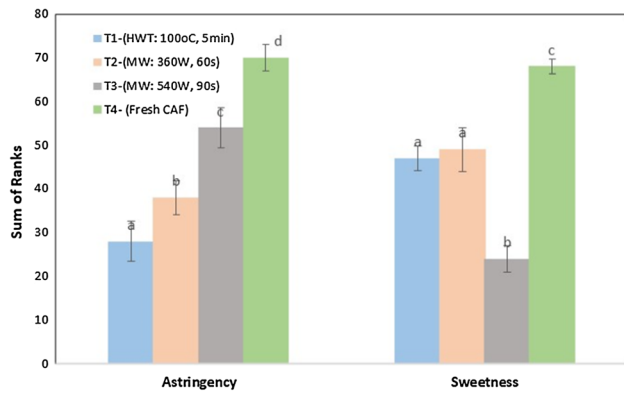
scavenging activity (Mahattanatawee et al. 2006). It was observed that there is a decrease in phenol content (data in Table 2) when subjected to different heat treatments in turn it might have led to a reduction in antioxidant activity. The reduction in  $AO_A$  ranged between 50.7% and 70.3%, 71.1% and 88.3%, and 32.7% and 60.9% for HWT, HT, and MW treatment, respectively (Table 2).

#### Ascorbic acid ( $A_A$ )

The  $A_A$  content of untreated CAF was determined to be  $196.9 \pm 32.5$  mg/100 ml. As explained in “Tannins ( $T_n$ )” section, the wide variation in value has been normalized by converting to percent reduction. Processing methods caused significant decreases in  $A_A$  content (Table 2) and found to be highest for HT samples (60–70%) followed by HWT (30–50%). It is widely known that  $A_A$  is thermolabile, i.e., thermal treatment can cause this to destroy and leach out in the medium during the processing (Lee and Kader 2000). The cause of destruction is ascorbic acid oxidation to dehydroascorbic acid, followed by hydrolysis, polymerization, and the formation of physiologically inactive substances; since heating accelerates oxidation, it results into the destruction of  $A_A$  (Gregory 1996). The decrease was the least during MW heating, i.e., 7–11% and agrees with work reported by Kaur and Kapoor (2002) on the effect of MW (700 W for 3 min) and HWT blanching (100 °C for 5 min) on  $A_A$  in green peas and carrot respectively. MW heating had the highest  $A_A$  retention with 76%, while HWT had 34% retention. Short processing time (i.e., 30–90 s) in case of MW heating might have helped in preserving more  $A_A$  than HWT and HT. In addition, moisture migration from CAF was not significant due to volumetric heating phenomena associated with MW heating. The combined effect of volumetric heating and short exposure time might not have disturbed the cell wall, which is supposed to be rich in ascorbic acid. Similar phenomena, i.e., distribution of  $A_A$  in apple fruit studied by Li et al. (2008) showed that the skins of apple have the highest levels of  $A_A$  than the flesh.

#### Total sugars ( $T_s$ )

Sugars are the important carbohydrates in juices as they not only give sweetness but also mask the astringency effects caused by tannins on its taste. Of the total sugars present in cashew apple, glucose, and fructose are predominant and contribute heavily to the sweet taste of cashew apple (de Silva et al. 2014). The untreated CAF had total sugar about  $5.9 \pm 1.08$  g/100 ml. The reduction in total sugars was between 14 and 16% for both HWT and MW treated samples, while it is nearly 50% for HT treated samples (Table 2). The decrease is more with an increase in



**Fig. 2** Results of ranking test for untreated and treated cashew apple fruit (CAF)

temperature and exposure time. Heating and roasting of food usually result in the loss of glucose, because of the interaction of glucose with amino acids at high temperatures, in a reaction called Maillard and caramelization (Woo et al. 2015).

#### Sensory evaluation

Sensory evaluation was carried out to rank the shortlisted treated samples (Table 1a–c) in terms of astringency and sweetness attribute. The effective treatments were shortlisted manually, on the basis of maximum tannin removal with the highest possible retention of the levels of  $A_A$ ,  $T_S$ , and  $AO_A$ . The treatments shortlisted were the “HWT (100 °C–5 min)” and “MW (360 W–60 s and 540 W–90 s)”. The sample with very intense astringency and sweetness got a rank four while the least intense sample got a rank one. The untreated sample was used as control. The intermediate samples were ranked two and three. Friedman’s non-parametric analysis of variance was performed to detect differences in the perception of attributes. The treatments were compared among themselves using Tukey’s test at 5% level of significance. The above three treatment samples, along with untreated sample were presented to panellist, and they were asked to rank according to the scale. Figure 2 shows that samples have been well differentiated in terms of astringency ( $T_n$ ) and sweetness ( $T_s$ ). For astringent attribute, T4 (control, i.e., untreated fruit) showed the highest sum of ranks signifying high astringency while the T1 (HWT:100 °C–5 min) had the lowest sum of ranks indicating the lowest astringency. Treatment T3 (MW: 540 W–90 s) and T2 (MW: 360 W–60 s) were intermittently ranked as two and three. Friedman’s analysis of variance showed that the astringency among the samples is significantly different. The highest sweetness was perceived for the T4, while the sample T3 showed the least. Sample T1 and T2 (360 W, 60 s)

intermediately ranked with no significant difference among themselves. Based on the sweetness and astringency score, HWT sample (T1: 100 °C for 5 min) was organoleptically accepted, where  $T_n$  and  $T_s$  removals were about  $42.6 \pm 0.9\%$  and  $15.3 \pm 0.5\%$  respectively.

#### Effect of Coagulating Agent in Treating CAJ

##### Dried okra pod powder (DOP)

The effects of the process variables ( $P_s$ ,  $C_n$ , and  $S_t$ ), using DOP in CAJ, on responses have been discussed in detail below.

*Effect of particle size ( $P_s$ )* Cashew apple juice treated with DOP at different experimental conditions were evaluated for percent reductions in  $T_n$ ,  $A_A$ ,  $T_s$  and  $AO_A$  and results are listed in Table 3a–c for three different  $P_s$ , i.e., “ $\leq 150 \mu\text{m}$ ”, “ $> 150$  and  $\leq 500 \mu\text{m}$ ”, “ $> 500$  and  $\leq 1000 \mu\text{m}$ ”, respectively. The percent reduction in  $T_n$  was found to vary between 21.1 and 54.1 within the combinations of the variables studied. The interaction of tannin with the protein, starch and digestive enzymes results in the formation of complexes of higher molecular weight, which later precipitates, and leads to the reduction of tannin (Sieniawska and Baj 2017). The okra powder being rich in proteins and polysaccharides act as a promising flocculation agent. Analysis shows DOP contain 13–25% crude protein and about 35–45% of carbohydrates (Okolo et al. 2015). In their separate studies, De Jesus et al. (2013) have reported that using DOP as a coagulant helped in achieving up to 99% reduction in turbidity in wastewater within 10 min of sedimentation time. Several authors found that both hydrolysable and condensed tannins could also bind with starch and polysaccharides (Barros et al. 2012), and cashew apple has both the forms of tannins [“Tannins ( $T_n$ )” section].

The percent reduction of  $T_n$  was observed to be a function of  $P_s$  and varied between 44.3–53.9 and 44.5–54.1 for  $P_s$  “ $\leq 150 \mu\text{m}$ ” and “ $> 150$  and  $\leq 500 \mu\text{m}$ ” respectively (Table 3a, b). However, when  $P_s$  increased to “ $> 500$  and  $\leq 1000 \mu\text{m}$ ”, the decrease in reduction of  $T_n$ , is not significant i.e., between 21.1 and 26.7%. This indicated the smaller the  $P_s$ , the higher the surface area per unit weight of coagulant (Maina et al. 2016) this could have led to a higher percentage of removal of  $T_n$ .

The  $P_s$  also affected the removal of other nutrients along with the tannins. The reductions were in the range of 5.1–9.8%, 2.9–17.1%, and 3.1–47.8% for  $A_A$ ,  $T_s$ , and  $AO_A$  (Table 3), respectively, for all the three particle sizes studied. Based on maximum  $T_n$  removal (set at  $\geq 42\%$ ) and minimum nutrient loss, the particle size range selected was “ $\leq 150 \mu\text{m}$ ” and “ $> 150$  and  $\leq 500 \mu\text{m}$ ”. However,



**Table 3** Effect of particle size ( $P_s$ ) on percent reduction of tannins ( $T_n$ ), ascorbic acid ( $A_A$ ), total sugars ( $T_s$ ) and antioxidant activity ( $AO_A$ ) at different concentration ( $C_n$ ) and settling time ( $S_t$ )

Run	Independent variables		% Reduction (responses)							
			DOP				DDSP			
	$C_n$ (%)	$S_t$ (h)	$T_n$	$A_A$	$T_s$	$AO_A$	$T_n$	$A_A$	$T_s$	$AO_A$
(a) Size: $\leq 150 \mu\text{m}$										
1	0.1	0.5	44.3 ± 1.0	7.1 ± 1.0	3.7 ± 1.0	4.7 ± 0.4	20.0 ± 1.2	21.0 ± 0.3	1.0 ± 0.1	6.1 ± 0.8
2	0.1	1.0	47.1 ± 1.1	8.1 ± 0.9	10.1 ± 0.8	11.4 ± 0.5	36.9 ± 0.7	24.8 ± 0.6	3.3 ± 0.2	14.8 ± 0.9
3	0.1	1.5	48.9 ± 0.8	8.7 ± 0.3	12.9 ± 0.7	30.5 ± 0.3	52.0 ± 1.0	26.9 ± 0.7	26.8 ± 0.5	28.0 ± 0.9
4	0.3	0.5	45.1 ± 0.9	7.1 ± 1.5	4.1 ± 0.9	7.3 ± 0.3	22.8 ± 0.7	24.1 ± 0.9	2.5 ± 0.4	7.5 ± 1.0
5	0.3	1.0	46.9 ± 0.9	8.3 ± 1.1	11.5 ± 1.2	15.1 ± 0.6	40.2 ± 1.2	26.1 ± 0.9	11.6 ± 0.4	23.0 ± 1.0
6	0.3	1.5	53.1 ± 0.3	9.1 ± 1.0	14.2 ± 0.9	32.4 ± 0.3	56.9 ± 0.6	27.8 ± 1.3	31.6 ± 0.6	40.2 ± 0.6
7	0.5	0.5	48.8 ± 1.1	8.2 ± 1.2	6.4 ± 0.8	35.0 ± 0.6	27.1 ± 0.8	25.7 ± 1.6	6.1 ± 0.3	9.1 ± 0.8
8	0.5	1.0	51.1 ± 0.7	9.4 ± 0.8	14.8 ± 0.9	43.1 ± 0.3	46.4 ± 0.7	26.2 ± 0.7	11.9 ± 0.4	32.1 ± 0.8
9	0.5	1.5	53.9 ± 1.3	9.8 ± 1.0	17.1 ± 1.3	47.8 ± 0.3	61.1 ± 0.9	29.0 ± 0.8	32.0 ± 0.6	44.2 ± 0.7
(b) Size: $> 150$ and $\leq 500 \mu\text{m}$										
1	0.1	0.5	44.5 ± 0.4	6.9 ± 0.7	3.6 ± 0.1	4.4 ± 0.3	19.9 ± 0.5	20.9 ± 0.1	0.9 ± 0.1	5.9 ± 0.8
2	0.1	1.0	46.6 ± 0.4	8.1 ± 0.8	9.9 ± 0.1	11.4 ± 0.5	36.4 ± 0.7	24.5 ± 0.4	3.2 ± 0.1	14.6 ± 0.7
3	0.1	1.5	49.4 ± 1.0	8.9 ± 1.3	13.1 ± 0.6	30.2 ± 0.3	51.5 ± 0.3	26.6 ± 0.6	26.7 ± 0.2	27.8 ± 1.1
4	0.3	0.5	45.6 ± 0.4	7.1 ± 0.9	4.1 ± 0.3	6.9 ± 1.4	22.1 ± 0.2	23.9 ± 0.8	2.4 ± 0.1	7.30 ± 0.5
5	0.3	1.0	47.4 ± 0.7	8.3 ± 1.1	11.9 ± 0.7	15.6 ± 0.4	39.2 ± 0.4	25.8 ± 0.9	11.5 ± 0.1	23.3 ± 1.7
6	0.3	1.5	53.0 ± 0.1	9.1 ± 0.4	14.5 ± 0.2	32.5 ± 0.6	56.4 ± 0.6	27.4 ± 0.8	31.4 ± 0.2	40.0 ± 1.9
7	0.5	0.5	49.1 ± 0.8	8.2 ± 1.2	6.7 ± 0.2	34.7 ± 0.2	26.5 ± 0.9	25.4 ± 1.1	6.1 ± 0.2	9.9 ± 1.4
8	0.5	1.0	50.1 ± 0.7	9.4 ± 0.8	14.9 ± 0.1	42.9 ± 0.6	45.9 ± 0.1	25.9 ± 0.7	11.8 ± 0.5	31.8 ± 1.9
9	0.5	1.5	54.1 ± 0.7	9.7 ± 0.6	16.3 ± 0.4	47.8 ± 0.3	60.3 ± 0.4	28.2 ± 0.5	31.9 ± 0.2	45.3 ± 1.3
(c) Size: $> 500$ and $\leq 1000 \mu\text{m}$										
1	0.1	0.5	21.1 ± 1.8	5.1 ± 0.3	2.9 ± 0.3	3.1 ± 0.5	10.2 ± 1.0	15.9 ± 0.8	0.4 ± 0.1	4.1 ± 0.3
2	0.1	1.0	22.9 ± 1.0	6.0 ± 0.2	8.9 ± 0.8	7.7 ± 0.3	14.5 ± 0.8	17.2 ± 1.0	1.3 ± 0.1	11.7 ± 0.9
3	0.1	1.5	24.1 ± 1.5	6.7 ± 0.7	9.1 ± 0.9	13.2 ± 0.4	22.0 ± 1.2	19.4 ± 1.2	17.1 ± 0.8	21.2 ± 0.6
4	0.3	0.5	22.6 ± 0.8	5.2 ± 0.7	3.2 ± 0.5	4.5 ± 0.6	11.2 ± 0.9	16.0 ± 1.1	1.1 ± 0.2	5.6 ± 0.5
5	0.3	1.0	23.4 ± 0.8	6.6 ± 0.6	9.6 ± 1.1	10.1 ± 0.7	19.1 ± 1.1	18.6 ± 1.1	10.6 ± 0.4	20.2 ± 1.0
6	0.3	1.5	25.7 ± 0.9	7.7 ± 0.6	10.8 ± 0.4	19.2 ± 0.4	26.8 ± 0.8	20.3 ± 1.0	21.9 ± 0.5	33.6 ± 0.5
7	0.5	0.5	23.3 ± 0.9	6.1 ± 0.7	4.6 ± 0.4	19.8 ± 0.8	12.3 ± 0.8	18.1 ± 0.9	4.3 ± 0.2	6.8 ± 0.3
8	0.5	1.0	25.1 ± 1.0	7.4 ± 0.8	10.8 ± 0.3	23.6 ± 0.7	22.2 ± 1.2	19.0 ± 0.9	10.9 ± 0.9	27.7 ± 0.5
9	0.5	1.5	26.7 ± 1.1	8.1 ± 0.6	12.1 ± 0.8	33.3 ± 0.4	27.0 ± 1.1	22.2 ± 0.8	22.1 ± 0.3	38.8 ± 0.7

it was observed at  $P_s$  “ $\leq 150 \mu\text{m}$ ” the CAJ was slimy, which posed filtration challenges. Also, the energy requirement is a function of the particle size reduction ratio. The smaller the particle size, the more energy required. Therefore, the most favourable range was  $P_s$  “ $> 150$  and  $\leq 500 \mu\text{m}$ ”. A subsequent analysis was conducted using other two independent variables i.e.  $C_n$  and  $S_t$ , while keeping  $P_s$  ( $> 150$  and  $\leq 500 \mu\text{m}$ ) as the constant variable.

*Effect of concentration ( $C_n$ ) and settling time ( $S_t$ )* The results of the percent reduction of  $T_n$ ,  $A_A$ ,  $T_s$ , and  $AO_A$  for

particle size “ $>150$  and  $\leq 500 \mu\text{m}$ ” at different  $C_n$  and  $S_t$  are shown in Table 3b. The data were used to design a response function to study the combined effect of independent variables. ANOVA was conducted to confirm the adequacy of the response functions, and results are shown in Table S2. The adjusted  $R^2$  obtained for percent reduction in  $T_n$ ,  $A_A$ ,  $T_s$ , and  $AO_A$  as a function of  $C_n$  and  $S_t$  were 0.92, 0.91, 0.98, and 0.97.

The data indicated that the  $T_n$  removal increased with an increase in  $C_n$  irrespective of  $S_t$ . It might be at higher  $C_n$ , the numbers of sites available for interaction with tannins are more, thereby resulting in higher removal of  $T_n$  (Van

Buren and Robinson 1969). However, when  $C_n$  increased to  $\geq 0.5\%$ , it had minimal effect on  $T_n$  removal (results not shown). This could be due to particle overlapping and overcrowding, resulting in a reduction of the total adsorbent surface area. Similarly, there is a significant increase in percent reduction of  $T_n$  with an increase in  $S_t$ . However, beyond 1.5 h of  $S_t$ , there is a plateau, which might be due to most of the flocs were already settled by then.  $S_t$  has a more significant effect on the percent reduction of  $T_n$  ( $p < 0.05$ ) than  $C_n$  (Table S2).

The percent reduction of  $A_A$  and  $T_S$  was also found to increase with the increase in  $C_n$  and  $S_t$ . This is attributed to the fact that there could be hydrogen bonding between coagulating agent and the other substances, leading to a formation of the high molecular weight compound, followed by precipitation. Results of this work were found to be consistent with the Naka et al., (2015) wherein they used cashew gum as a clarifying agent to reduce astringency in CAJ.  $AO_A$  values were also observed to decrease with an increase in  $C_n$  and  $S_t$  between 4.5 and 47.8%, which were found to be strongly correlated with phenol content.

#### Dried drumstick seed powder (DDSP)

The following section describes the use of DDSP on response variables at different  $P_s$ ,  $C_n$  and  $S_t$ .

**Effect of particle size ( $P_s$ )** The percent reduction in  $T_n$  correspond to three different  $P_s$ , i.e.,  $\leq 150 \mu\text{m}$ ,  $> 150$  and  $\leq 500 \mu\text{m}$ , and  $> 500$  and  $\leq 1000 \mu\text{m}$  is shown in Table 3a–c respectively. As explained earlier (“Effect of particle size ( $P_s$ )” section), the decrease in  $T_n$  could be due to the formation of insoluble tannin-protein complexes. Drumstick seeds contain active coagulating agents, i.e., dimeric cationic proteins with a molecular mass of 12–14 kDa, which act as flocculant inducing agents (Martín et al. 2012). Smaller  $P_s$  (i.e., “ $\leq 150 \mu\text{m}$ ” and “ $> 150$  and  $\leq 500 \mu\text{m}$ ”) resulted in higher removal of  $T_n$ , i.e., 20.0–61.1% and could be due to higher surface area per unit of coagulant.

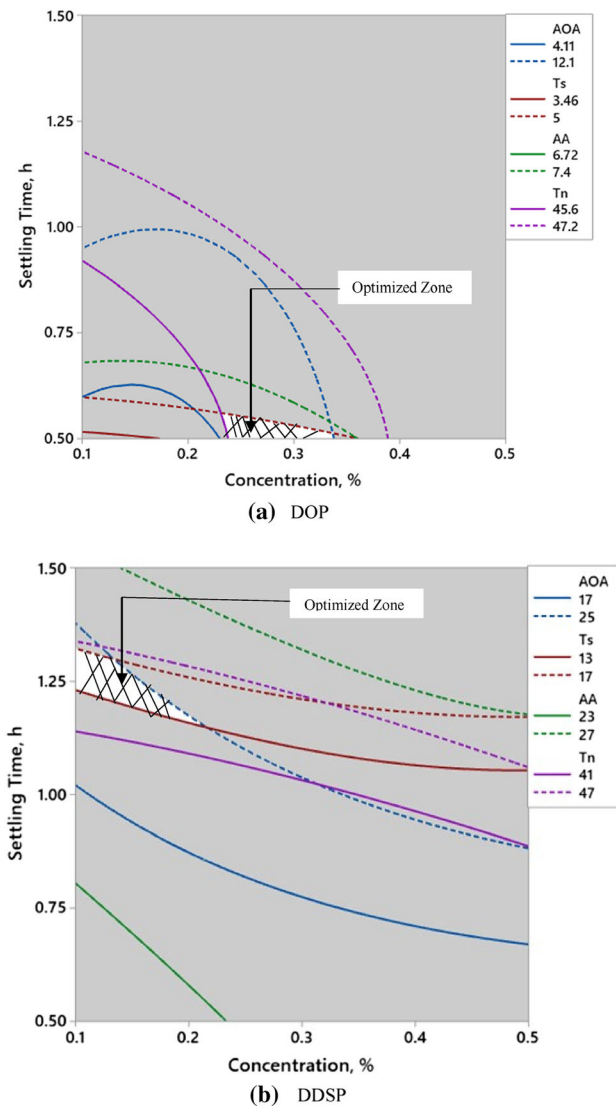
The  $P_s$  also affected nutrient content along with the tannins. The reductions were in the range of 15.9–29.0, 0.4–32.0, and 4.1–45.3 for  $A_A$ ,  $T_S$ , and  $AO_A$ , respectively, for all the three different  $P_s$  studied (Table 3). It was observed that at  $P_s$  “ $> 500$  and  $\leq 1000 \mu\text{m}$ ”, though there was maximum possible retention, which is desirable, but the  $T_n$  removed was not adequate, i.e., 10.2–27.0%. Based on the tests carried out on CAF (“Sensory evaluation” section), minimum  $T_n$  removal must be  $\geq 42.6\%$  to make the sample organoleptically acceptable. Hence,  $P_s$  “ $> 500$  and  $\leq 1000 \mu\text{m}$ ” was rejected. Though the use of  $P_s$  “ $\leq 150 \mu\text{m}$ ” could remove  $T_n$  up to 40–60%, it made the juice slimy, thereby posing difficulties during filtration.

Therefore, the  $P_s$  “ $> 150$  and  $\leq 500 \mu\text{m}$ ” was selected for subsequent analysis.

**Effect of concentration ( $C_n$ ) and settling time ( $S_t$ )** CAJ treated with DDSP at different combination of  $C_n$  and  $S_t$  ( $P_s$ : “ $> 150$  and  $\leq 500 \mu\text{m}$ ”) was studied to determine the percent reduction in  $T_n$ ,  $A_A$ ,  $T_S$ , and  $AO_A$  (Table 3b). The adjusted  $R^2$  obtained for response surface functions for percent reduction in  $T_n$ ,  $A_A$ ,  $T_S$ , and  $AO_A$  were 0.99, 0.87, 0.98, and 0.97, respectively. The removal of  $T_n$  increased with an increase in  $C_n$  from 19.9 to 60.3%. Both the independent variables significantly ( $p < 0.05$ ) affected the reduction of  $T_n$  (Table S2). The increase in  $C_n$  and  $S_t$  increased the reduction in  $A_A$ , and  $T_S$  i.e. from 20.9–28.2% and 1.0–31.9% respectively and were significantly higher compared to DOP treated CAJ ( $A_A$ : 6.9–9.7% and  $T_S$ : 3.6–16.3%) (Table 3b). Highest removal of  $T_n$  was achieved at 1.5 h of  $S_t$ , irrespective of  $C_n$ , and a further increase in  $S_t$  did not show any changes. The percent reduction in  $AO_A$  values ranged between 5.9 and 45.3 and found to be comparable with DOP treated CAJ (4.4–47.8).

#### Optimization

An optimization approach using desirability functions of multiple responses was performed using Minitab statistics software. The goals for optimization were set at obtaining tannin removal at  $\geq 42.6\%$  (based on input from 1<sup>st</sup> phase of the experiment on CAF), and minimizing percent reduction in  $A_A$ ,  $T_S$ , and  $AO_A$  values within the data obtained in the experimental range (shown in Fig. S1 a and b for DOP and DDSP, respectively). The optimum conditions were determined to be 0.3% of  $C_n$  and 0.5 h of  $S_t$  for DOP-treated CAJ and 0.1% of  $C_n$ , and 1.2 h of  $S_t$  for DDSP-treated CAJ with composite desirability of 0.85 and 0.69 respectively (Fig. 3a, b). The percent reduction in  $T_n$ ,  $A_A$ ,  $T_S$ , and  $AO_A$  values were also experimentally determined under optimum conditions. The predicted reduction values for DOP treated CAJ were 46.4%, 7.2%, 4.5%, 10.1% and DDSP treated CAJ were 44.9%, 25.5%, 14.5%, 22.5% for  $T_n$ ,  $A_A$ ,  $T_S$ , and  $AO_A$  respectively. The experimental values were relatively close to the software-generated values (Table S3) hence confirming the validity of the optimized results. The contour plots of all response variables ( $T_n$ ,  $A_A$ ,  $T_S$ , and  $AO_A$  values), as a function of  $C_n$ , and  $S_t$ , were superimposed together, and the optimum regions (white shaded areas in the overlay plots) are shown in Fig. 3a, b for DOP-treated and DDSP-treated CAJ respectively.



**Fig. 3** Graphical optimization (shaded area) of concentration and settling time for percent reductions in tannins ( $T_n$ ), ascorbic acid ( $A_A$ ), total sugars ( $T_s$ ) and antioxidant activity ( $AO_A$ ) of the treated CAJ using **a** Dried okra pod powder (DOP) and **b** Dried drumstick seed powder (DDSP)

**Sensory evaluation using hedonics rating-test**

Among the optimized DOP and DDSP treated samples, the DOP treated CAJ had a significant higher sensory acceptance index ( $92.7 \pm 1.6\%$ ) than DDSP treated CAJ ( $79.1 \pm 2.1\%$ ). The overall acceptability of DOP treated CAJ ranged between 8 (Like very much) and 9 (Like extremely) while for DDSP; it varied between 6 (Like Slightly) and 7 (Like moderately). Overall liking may be strongly correlated to astringency and sweetness attributes. Hence, the ranking test was further carried out to determine the astringency and sweetness score for optimized DOP and DDSP treated CAJ. HWT treated samples (100 °C for

5 min, the best one among the CAF) was used as the reference sample (“Sensory evaluation” section). The astringency score was found to be in the range of 34–37 for DOP, DDSP, and HWT treated CAJ with no significant difference among each other. While the sweetness score was the highest for DOP treated CAJ (i.e., 55) followed by HWT (i.e., 39) with least score obtained for DDSP treated CAJ (i.e., 20). Based on the astringency removal, nutrition retention, and sensory score, the optimized DOP treated CAJ (0.3% of  $C_n$  and 0.5 h of  $S_t$ ) was found to be the best.

**Conclusion**

The study aimed to compare the different techniques to reduce the  $T_n$  while retaining its nutrient content, i.e.,  $A_A$ ,  $T_s$ , and  $AO_A$  as maximum as possible from both CAF and CAJ. The obtained results indicated that treated samples, with 42.6% tannin removal, were rated as the least astringent among all the samples. The numerical and graphical optimization indicated that the use of DOP at  $C_n$  of 0.3% and  $S_t$  of 0.5 h was optimal, where maximum  $T_n$  removal (46.4%) with minimum reduction in  $A_A$  (7.2%),  $T_s$  (4.5%) and  $AO_A$  (10.1%) was predicted with high acceptability index (92.7%). Since corresponding experimental values ( $48.9 \pm 1.6$ ,  $8.1 \pm 0.9$ ,  $4.8 \pm 0.5$ ,  $11.1 \pm 1.0$  for  $T_n$ ,  $A_A$ ,  $T_s$ , and  $AO_A$  respectively) were close to the predicted values, the optimum conditions obtained in this study could be used as a standard or baseline information for industrial processing of this particular cashew apple variety. Retention of high concentrations of  $A_A$ ,  $T_s$ , and  $AO_A$  makes it an excellent option for the beverages market in the form of juice blends.

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