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Reduction Triggered In Situ Polymerization in Living Mice

Lina Cui^{1,2,3,4}, Sandro Vivona^{5,6,7}, Bryan Ronain Smith^{3,†}, Sri R. Kothapalli^{3,#}, Jun Liu^{1,2}, Xiaowei Ma^{1,2}, Zixin Chen^{1,2,4}, Madelynn Taylor³, Paul H. Kierstead⁸, Jean M.J. Fréchet⁸, Sanjiv S. Gambhir^{3,9,10,‡}, Jianghong Rao^{3,4,*}

¹Department of Medicinal Chemistry, University of Florida, Gainesville, FL, USA;

²UF Health Cancer Center, University of Florida, Gainesville, FL, USA;

³Molecular Imaging Program at Stanford, Bio-X Program, Department of Radiology, School of medicine, Stanford University, Stanford, CA, USA;

⁴Department of Chemistry, Stanford University, CA, USA;

⁵Department of Molecular and Cellular Physiology, Stanford University, CA, USA;

⁶Department of Structural Biology, Stanford University, Stanford, CA, USA;

⁷Department of Photon Science, Stanford University, Stanford, CA, USA;

⁸College of Chemistry, University of California, Berkeley, CA, USA;

⁹Department of Bioengineering, Stanford University, CA, USA;

¹⁰Department of Materials Science and Engineering, Stanford University, Stanford, CA, USA.

Abstract

"Smart" biomaterials that are responsive to physiological or biochemical stimuli have found many biomedical applications for tissue engineering, therapeutics and molecular imaging. In this work we describe in situ polymerization of activatable biorthogonal small molecules in response to reducing environment change in vivo. We designed a carbohydrate linker- and cyanobenzothiazole-cysteine condensation reaction-based small molecule scaffold that can undergo rapid condensation reaction upon physiochemical changes (such as a reducing environment) to form polymers (pseudopolysaccharide). The fluorescent and photoacoustic properties of fluorophore-tagged condensation scaffold before and after the transformation have been examined with a dual-modality optical imaging method. These results confirmed the *in situ* polymerization of this probe after both local and systemic administration in living mice.

Supporting Information.

^{*}Corresponding Author: jrao@stanford.edu.

[†]B.R.S. is currently at the Department of Biomedical Engineering and Institute for Quantitative Health Science and Engineering, Michigan State University.

[‡]Deceased on July 18, 2020.

[#]S.R.K. is currently at the Department of Biomedical Engineering, Pennsylvania State University.

Conflict of Interests

Dr. Gambhir served on the board of directors of Endra Inc. and as a consultant for Visualsonics Inc.; both manufacturers of PA imaging equipment for small animals.

Supporting figures, experimental procedures and chemical characterizations (HRMS, ¹H and ¹³C NMR spectra). This material is available free of charge via the Internet at http://pubs.acs.org.



Graphical Abstract



Introduction

Biomaterials that can undergo *in situ* transformation with exposure to specific biomarkers or environmental cues in living subjects can potentially lead to new biomedical applications.¹ For example, preformed polymers that can cross-link or solidify *in vivo* after injected locally when exposed to light,^{2–3} temperature,⁴ pH changes,⁵ enzymes,^{6–7} or presence of specific ions⁸ can provide better tissue-material contact and can be used for facilitating local transplantation of medical scaffolds,⁹ tissue engineering,^{10–11} drug delivery^{12–13} and imaging.¹⁴ Compared to polymeric materials, small molecules possess properties of efficient tissue distribution, deep penetration, and fast systemic clearance in normal tissues. However, small molecules also suffer from poor retention compare to polymeric materials at target sites can take advantage of the efficient distribution and clearance of small molecules as well as the enhanced retention of macromolecules and offer unique opportunities in designing new materials for diagnostics and therapeutics.^{15–17}

One of the strategies for assembling small molecules into macromolecules inside complex biological environment is through covalent bonding enabled by bioorthogonal reactions.¹⁸ These reactions such as ketone/hydroxylamine condensation.^{19–22} copper free azide-alkyne cycloaddition,^{23–26} Staudinger ligation^{27–29} and tetrazine-*trans*-cyclooctene cycloaddition, ^{30–36} have been applied to site-directed protein labeling,³⁷ studying metabolic pathways^{24, 38–39} and engineering polymeric biomaterials.^{40–42} Their applications in "smart" biomaterials that are responsive to physiological or biochemical stimuli are limited due to the fact that most of the chemical moieties used in these reactions cannot be biochemically masked. Recently there have been some reports on using light or enzymes to modulate bioorthogonal reactions, 43-52 but not yet demonstrated *in vivo*. For the past few years, we have been studying the reaction between aminothiols and aromatic nitriles 53-56and its applications for *in vivo* molecular imaging.^{57–63} One of the most attractive features of this reaction is that its reactivity can be modulated by pH, reducing environment or enzyme activities by masking the aminothiol with different responsive moieties. One of the fastest reactions we've tested so far is between the cyanobenzothiazole (CBT) group and free cysteine, which has a second-order rate constant of 9.19 M⁻¹s⁻¹.⁵³ We and others have used this reaction for protein labeling,^{64–67} formation of hydrogels,^{53, 68–69} fluorescence lifetime imaging⁷⁰ and *in vivo* imaging of protease activity.^{53, 57–58} This condensation

reaction therefore holds great potential for constructing "smart" molecules that can go through *in situ* transformation into large macromolecules or polymeric materials when encounters specific biological stimuli.

Our previously reported CBT conjugates tend to form cyclic dimers or oligomers after condensation instead of forming linear polymers due to the flexible peptide linkers within the scaffolds.^{53, 57} We hypothesized that a rigid linker in the scaffold would promote the elongation of the linear polymers over cyclization and provide an attractive strategy for *in situ* formation of hydrogels from small molecules (Fig. 1). The carbohydrate moieties are used extensively in nature to form rigid polymers, such as cellulose. The equatorial conformation of glucose in cellulose provides rigidity for the linear polysaccharides. Here we report a carbohydrate-based small molecule scaffold that can undergo CBT-cysteine condensation to form polymers under reducing conditions. We used a dual-modality optical imaging method to measure the fluorescence and photoacoustic properties (conversion of non-ionizing radiation to pressure/ultrasound waves)^{71–73} of new scaffold based fluorescent probes before and after the polymerization and demonstrated the *in situ* formation of polymers.

Results and discussion

Design and synthesis of polymerizable molecular scaffold.

To evaluate the polymerization reaction, we synthesized two sets of probes with Dglucosamine as the backbone to facilitate the polymerization (Fig. 1). The CBT is linked to C-6 on the glucosamine via click reaction and cysteine is linked to the C-2 amine via a peptide linker. The fluorescent probes 1 and 2 contained an IR800 dye as an imaging moiety to detect their fluorescence and photoacoustic properties. IR800 dye was selected for its suitable excitation/emission wavelength (774 nm/789 nm) for *in vivo* imaging, its large extinction coefficient (240,000 M⁻¹ cm⁻¹), and its good water solubility. The other set of probes 3 and 4 contained a 2,2',2''-(10-(2-((2,5-dioxopyrrolidin-1-yl)oxy)-2oxoethyl)-1,4,7,10-tetraazacyclododecane-1,4,7-triyl)triacetic acid (DOTA) moiety for its water solubility and its potential to chelate Gd³⁺ for magnetic resonance imaging (MRI) or radioactive metal ions such as Cu-64 or Ga-68 for positron emission tomography (PET) imaging in the future. Probes 1 and 3, containing a disulfide linkage on the cysteine, can be activated by reducing environment to generate the free N-terminal cysteine group that then reacts with CBT group on the other molecule to initiate the polymerization (Fig. 1). The control probes 2 and 4 cannot expose the thiol group in reducing environment and should remain as small molecules.

A convergent synthetic scheme was developed to obtain all probes **1-4** (Supporting Information Fig. S1). The synthesis started from free D-glucosamine, and an azido group was linked to the C-1 position to lock its chair conformation. A cysteine group was then conjugated to the amino group, and the CBT group was introduced at the C-6 position via a copper-catalyzed azide-alkyne cycloaddition (CuAAC). Both IR800 and DOTA were introduced via their NHS esters.

Characterization of polymerization reactions using SEC-MALS.

The polymerization reactions were monitored by the size exclusion chromatography (SEC) coupled with multiangle light scattering (MALS) to analyze the sizes, mass and relative distribution of the resulting polymers (Fig. 2, Supporting Fig. S2). The activated or unactivated probe 3 was injected onto a WTC-010S5 column, and the elution profile was monitored by light scattering at 658 nm and differential refractometry. The triggered polymerization was very efficient in the presence of GSH (10 mM): the majority (> 60%) of probe 3 (at a concentration of 10 mM) was converted to polymers at a size of around 40 kDa (~40 repeating units, Supporting Fig. S2 peak 1) and hydrodynamic radius of 3 nm (Supporting Fig. S2) in 10 minutes, together with monomer 3, dimer (Supporting Fig. S2 peak 4, S5), oligomers (Supporting Fig. S2 peak 3, S6) and a second portion of polymers of around 20 kDa (Supporting Fig. S2 peak 2, S4) as detected by matrix-assisted laser desorption/ionization mass spectroscopy (MALDI-MS). To evaluate the polymerization at low concentrations, probe 3 was activated at 1, 10 and 100 μ M by GSH (10 mM), and the polymerization profiles were monitored from the initial 10 minutes to 8 hours after incubation in reaction buffer (Fig. 2). Formation of polymers was concentration-dependent, with larger polymer species formed at higher concentrations (Fig. 2a, 2b). At 100 µM, nearly 80% of probe 3 was converted to polymers at around 25 kDa (~25 repeating units) within the first 1 hour of reaction (Fig. 2a, 2b, 2e). At concentrations of 10 µM or even 1 µM, around 40% of probe **3** formed polymer species at around 25 kDa within 1 hour (Fig. 2a, 2b, 2c, 2d). These in vitro results demonstrated the efficiency of the polymerization reactions at various concentrations. To our delight, this polymerization can proceed efficiently at as low as a few micro molar concentrations, making it attractive for applications in living systems.

Optical properties of small molecule probes upon in situ polymerization.

We evaluated the optical properties of probes 1 and 2 before and after activation by glutathione (Fig. 3, Supporting Fig. S7). Glutathione plays a key role in regulating the redox potential in mammalian cells and its concentration can range from several uM in plasma to several mM in cells and tumor tissues.^{74–80} Our group and others have demonstrated that the CBT condensation reactions can proceed under physiological conditions where glutathione is present in cells as well as in living animals.^{53, 58, 70, 81–83} Incubation of probe 1 (100 μ M) in PBS buffer containing reducing agent glutathione (GSH, 10 mM) or tris(2carboxyethyl)phosphine (TCEP, 10 mM) at pH 7.4 resulted in a fluorescence signal drop by around 70% within 30 min (Fig. 3a, 3c) likely due to the close proximity of IR800 dye and self-quenching after polymerization; in comparison, probe 2 under the same conditions or probe 1 without incubation with GSH or TCEP barely showed any changes in fluorescence signals within two hours (Fig. 3a, 3c). Electrophoretic gels of the GSH-treated probe 1 and 2 were stained using silver nitrate (Fig. 3b) and scanned using Maestro hyperspectral fluorescence imaging system (PerkinElmer, 635±25 nm excitation filter, 675 nm long-pass emission filter, image acquisition from 670 to 900 nm). From the silver staining, control probe 2 showed single bands at all concentrations (6, 0.6, and 0.06 nmol per 6 μ L loaded) tested; while probe 1 (6 nmol per 6 μ L loaded) showed smeared bands at a position similar to that of the 25-40 kDa fraction from SEC (Fig. 3b, left), and probe 1 at lower concentrations (0.6 nmol per 6 µL loaded) produced slightly smaller polymer species (Fig.

3b, left). Interestingly, in comparison to the fluorescence intensity of probe **2** at the test concentrations, probe **1** only gave a major fluorescence band of the dimer (detected by MALDI-MS after gel extraction), however the larger polymer species remained dark due to self-quenching of IR800 dye after polymerization occurred (Fig. 3b, right).

Photoacoustic (PA) imaging detects the acoustic signals generated during thermoelastic expansion due to the photoabsorption. Therefore, PA imaging offers deeper tissue penetration and higher spatial resolution than fluorescence imaging.^{71–73} PA signals were measured using a Vevo LAZR instrument (Visualsonics, 2100 High-Resolution Imaging System). Probe **1** and **2** (each at 100 μ M) with or without GSH (10 mM) were sealed in polyethylene tubes and buried in agarose gel, and fluorescence imaging were performed 30 min after the reaction started, immediately before acquiring photoacoustic signals (Fig. 3e, 3f). While GSH-incubated probe **1** showed reduced fluorescence signal compared with probe **2** (Fig. 4e), its PA signal increased to nearly 200% of the signal from probe **2** or probe **1** alone (Fig. 3f, 3g), suggesting the enhanced PA effects accompanying reduced fluorescence signal upon the formation of polymers.

Imaging polymerization in living animals.

We next evaluated the polymerizable probe with both fluorescence and PA imaging in a model system in living mice. The back of BALB/c mice was shaved, and a solution of GSH was injected subcutaneously on the furless back to form two spots of reducing environment (Fig. 4a). Probe 1 (1 nmol in 30 μ L of PBS buffer) and probe 2 (1 nmol in 30 μ L of PBS buffer) were then injected to each of the two reducing sockets. As expected, the two probes initially displayed the same fluorescence intensity, but the fluorescence of probe 1 decreased drastically in comparison to probe 2 within the first 30 minutes after injection (Supporting Fig. S8). Interestingly while the absolute fluorescence signal decreased in both spots due to the probe diffusion from the inoculation site, probe 1 appeared brighter than probe 2 afterwards even with self-quenching: the intensity ratio of probe 1 and 2 started to escalate, and reached around 15 after 24 hours (Fig. 4b, 4c), indicating the prolonged retention of probe 1 upon activation.

PA imaging (co-registered with ultrasound imaging) was similarly performed using the same model system, and each spot was monitored for 24 hours after probe injection (Fig. 4d). Different from fluorescence imaging, which showed drastic fluorescence decrease of probe **1** in the initial 30 min, PA signal of the same spot increased 30 min after probe **1** was injected (Supporting Fig. S9). Comparing to probe **2**, which showed gradual decrease in PA signal over time, probe **1** showed initial PA enhancement in the first hour after injection, followed by a slow decline phase (Fig. 4e, 4f). Together with the fluorescence imaging results, the initial decrease in fluorescence and increase in PA signal resembled the trend in the *in vitro* experiments (Fig. 3), and probe **1** was retained longer in tissue than probe **2**, suggesting the high probability of polymer formation.

Imaging of polymerization in tumor models.

The ideal polymerizable small molecules when injected systemically should be activated to form macromolecular structures at the target site (such as a tumor) in the living subjects,

while diffusing away in other non-targeted tissues to provide better imaging contrast. To assess whether our molecular scaffold can be activated at desired sites after intravenous injection, we evaluated the probes 1 and 2 in mice xenografted tumor models. The ability to trigger the polymerization at desired sites was evaluated in female nude mice bearing subcutaneous U87MG tumors that were pre-treated with intratumoral injection of GSH or saline. Probe 1 or 2 (20 nmol in 200 µL saline) was administered to tumor-bearing mice via tail vein injection, and longitudinal fluorescence and PA imaging of these mice were performed (Fig. 5, Supporting Fig. S10). Since both probe 1 and 2 are not cell permeable, the polymerization will occur outside tumor cells, depending on the intratumoral delivery of GSH. We hypothesized that formed polymers would diffuse away slower than the free probe itself. The whole animal fluorescence imaging showed the fluorescence of probe 1 accumulated in GSH-pretreated tumor during the first one hour after probe injection and declined slowly over the monitored time course of 24 hours (Fig. 5, Supporting Fig. S10). Probe 1 administered to the mice bearing saline-pretreated tumor or probe 2 administered to the mice bearing GSH-pretreated tumor produced much faster fluorescence signal decay after injection, suggesting that probe 1 are 2 are poorly retained in tumor specifically (Fig. 5b, 5c, Supporting Fig. S10).

PA imaging of each mouse at each specific time point was performed immediately after the fluorescence scanning (Fig. 5d). PA signal in saline-pretreated tumors with probe **1** administration or GSH-pretreated tumors with probe **2** administration appeared close to the background level during the monitored time course (Fig. 5d, middle and right, Fig. 5e, 5f); in comparison, the GSH-pretreated tumors with probe **1** administration gave strikingly high PA signals in the entire time, peaking at 4 hours after probe injection (Fig. 5d, left, Fig. 5e, 5f). Importantly, the PA signal was nearly evenly distributed throughout the tumor region. While examining the results of the activated probe **1** in both of the PA imaging (Fig. 5d) and fluorescence imaging (Fig. 5a), tumor PA signal peaked (at 4 hours post probe injection) later than its fluorescence signal (at 30 min to 1 hour post probe injection). Also, PA signal in tumor remained high (nearly 50% of its peak value) even after 24 hours (Fig. 5e), but its corresponding fluorescence signal dropped to around the same level as two other control conditions after 4 hours (Fig. 5b, 5c). Together with the *in vitro* results, remaining PA signal in tissue without fluorescence can be attributed to the formation of polymers, which lose the fluorescence via self-quenching.

Conclusions

Our previous experience with the CBT condensation reactions often led to early cyclization and self-stacking to form nanoparticles. Molecular scaffolds that can prevent the formation of self-aggregates can further extend the application of the scaffold and chemistry to other applications such as therapeutics delivery and tissue engineering. In this work, we explored a carbohydrate scaffold for its water solubility and structural rigidity from the chair conformation to promote linear polymerization. Our results have shown that linear polymerization can occur even at a few micromolar concentration with this new rigid carbohydrate scaffold. To address the challenge of the detection of polymerization in living animals, we developed a dual-modality optical method using fluorescence and PA imaging. With this imaging method, we were able to demonstrate the *in situ* polymer formation with

reductive activation in living mice. Future efforts may extend this system to other stimuli such as enzyme activation in tumors.

Experimental Section

Probe synthesis.

Chemical syntheses and characterizations of probes **1-4** and intermediates are provided in Supporting Information.

SEC-MALS measurements.

Size exclusion chromatography (SEC) coupled with multiangle light scattering (MALS) was performed by injecting 150 µg of activated or unactivated probe **3** or **4** onto a WTC-010S5 column (Wyatt Technology) previously equilibrated in PBS buffer at a flow rate of 0.5 mL/min. The elution profile was monitored by light scattering at 658 nm (HELEOS system, Wyatt Technology) and differential refractometry (Optilab system, Wyatt Technology). UV Absorbance was not used due to saturation of the detector at the concentrations above the detection limit of light scattering. Data analyses were carried out using ASTRA 6.0 software (Wyatt Technology). The refractive index increment (dn/dc) value of 0.163 mLµg⁻¹ was assumed based on literature,⁸⁴ as experimental determination was hindered by limited solubility of samples. Calculated molecular weights were further confirmed by MALDI-TOF mass spectrometry. SEC traces of polymerization reactions of different concentrations at different time points were plotted using Origin Lab 8.5 as 3D stacking and 2D intensity graphs. Different species (monomer, polymer species around 25 kDa alone or with other oligomers) were quantified using integration of area under curve (AUC) of the differential refractometry data and plotted against time using Origin Lab 8.5.

Measurement of photoacoustic effects of in vitro samples.

PA imaging was performed using a LAZRTightTM imaging enclosure coupled Vevo LAZR PA imaging system (VisualSonics), equipped with a MS-250 linear array transducer (21 MHz, 70% 6 dB two-way bandwidth, 256 elements) to detect ultrasound signals, and a tunable Nd:YAG laser system (OPOTEK, 680–950 nm, 20 Hz repetition rate, 5 ns pulse width, 50 mJ pulse peak energy) to generate optical pulses. For *in vitro* measurements, the probe solutions were injected into one-inch long polyethylene (PE100) tubing (Becton Dickinson, i.d. = 0.86 mm, o.d. = 1.52 mm), and the open ends of the tubes were heatsealed. The tubes were then inserted horizontally into a freshly made semi-hardened agarose gel (1%) and the surface of the gel was coated with even layer of agarose solution (1%, ~3 mm thick). PA imaging was then performed after the gel completely solidified. PA signals were acquired using a 21 Hz transducer at 40 dB PA gain in 3D mode at a wavelength of 774 nm (absorbance max of IR800 dye) and with an acquisition rate of 5 frames/second. The 3D image of each sample tube was processed using loaded Visualsonics Vevo software package, and analyzed using Image J software (Fig. 3f, 3g).

Fluorescence imaging of polymerizable probes in living mice.

Female BALB/c mice (4 months old, Charles River Inc.) were anesthetized by isoflurane, and fur was removed to expose the back skin (Fig. 4a). In the following day, the mice were

anesthetized and positioned on the heating platform inside a Maestro hyperspectral fluorescent imaging system (PerkinElmer). GSH (20 mM in 20 μ L of PBS buffer) was injected subcutaneously under the back skin for each testing spot, where probe **1** or **2** (each of 1 nmol in 30 μ L of PBS buffer) were then injected 5 min later. Fluorescence imaging was performed immediately using the Maestro scanner (635±25 nm excitation filter, 675 nm long-pass emission filter, images acquisition from 670 to 900 nm). IR800 fluorescence was deconvolved from tissue autofluorescence (Nuance v.3.0.1.2, PerkinElmer), and fluorescence intensity was quantified by region of interest (ROI) analysis (Fig. 4b, 4c).

PA imaging of polymerizable probes in living mice.

Female BALB/c mice were anesthetized and positioned on the heating platform inside the LAZRTightTM imaging enclosure of the Vevo LAZR imaging system. GSH (20 mM in 20 μ L of PBS buffer) was injected subcutaneously under the furless back skin for each testing spot, where probe **1** or **2** (each of 1 nmol in 30 μ L of matrigel) were then injected 5 min later. A thin layer of ultrasound scanning gel (Clear Image Singles) was applied to cover scanning area. Ultrasound co-registered PA imaging was performed immediately using a 21 MHz transducer at 100% power, 40 dB PA gain, and excitation wavelength of 774 nm (Fig. 4d). PA images were processed using Image J software, and PA intensity was quantified after subtraction of background signals (Fig. 4e, 4f).

Fluorescence imaging of polymerizable probe in tumor models.

To initiate tumors in female nu/nu mice (8-week-old, Charles River Inc.), 5-million U87MG cells suspended in saline (50 μ L, original culturing medium was replaced immediately before implantation) were injected subcutaneously at the left shoulder of the mouse. Tumors were grown until reaching the largest dimension at 0.5 to 0.7 cm (~14 to 21 days). The mice were anesthetized, and the tumors of two mice groups (n=5) were treated with intratumoral injections of GSH (20 mM, pH 7.4) and a third group received saline (50 μ L). Probe 1 or 2 (20 nmol in 200 μ L saline) was administered 5 minutes later to the two mice groups with GSH treatment via tail vein, and probe 1 (20 nmol in 200 μ L saline) was also administered to the third mice group with saline treatment. Longitudinal whole animal fluorescence was monitored over 24 hours using an IVIS spectrum fluorescence imaging system (PerkinElmer, Fig. 5a, Supporting Fig. S10). Tumor fluorescence radiance was quantified by region of interest (ROI) measurement using IVIS Living Image 4.2 software (Fig. 5b, 5c).

PA imaging of polymerizable probe in tumor models.

Immediately after the fluorescence imaging, the tumor-bearing mouse was moved to the heating platform inside the LAZRTight[™] imaging enclosure of the Vevo LAZR imaging system for PA imaging. A generous layer of ultrasound scanning gel (Clear Image Singles) was applied to cover the tumor region. Ultrasound co-registered PA imaging was performed using a 21 MHz transducer at 100% power, 40 dB PA gain, and excitation wavelength of 774 nm (Fig. 5d). Photoacoustic images were processed using the Visualsonics imaging system software package, and PA intensity was quantified by region of interest (ROI) measurement (Fig. 5e, 5f).

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

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Figure 1. Structures of molecular probes used in the study and their activation to polymers.a. Condensation reaction between cyanobenzothiazole and cysteine (a.k.a. CBT Condensation).b. Probes 1 and 3 undergo polymerization upon reductive activation. Control probes 2 and 4 remain as monomers under the same conditions.



Figure 2. Characterization of polymerization.

a. Comparison of elution profiles of the polymerization reaction of probe **3** at various concentrations (1, 10, 100 μ M, as indicated) after different reactions times (10, 30, 60, 120, 240, 480 min, as indicated). The molecular weights of peaks at T_E = 14.8 min and 17.7 min are 25 kDa and 1 kDa respectively (Supporting Fig. S4); the peak at 16.5 min contains trimers and higher-order oligomers as detected by MALDI-MS (Supporting Fig. S6). **b**. 2D intensity contour image of the elution profiles at indicated concentrations and reaction time points. **c**, **d**, **e**. Products population of the reaction at 1, 10, and 100 μ M vs. time, respectively.



Figure 3. Characterization of polymerization using fluorescence and photoacoustic signals. **a**. White light (upper) and fluorescence (lower) images of (1) probe 2 (100 μ M) in PBS buffer at pH 7.4, (2) probe 2 (100 µM) in PBS buffer containing TCEP (10 mM) at pH 7.4, (3) probe 2 (100 μ M) in PBS buffer containing GSH (10 mM) at pH 7.4, (4) probe 1 (100 μM) in PBS buffer at pH 7.4, (5) probe 1 (100 μM) in PBS buffer containing TCEP (10 mM) at pH 7.4, (6) probe 1 (100 µM) in PBS buffer containing GSH (10 mM) at pH 7.4. b. Silver nitrate staining (left) and fluorescence image (right) of gel electrophoresis (25% acrylamide gel) of (1) probe 2 (6 nmol), (2) probe 2 (0.6 nmol), (3) probe 2 (0.06 nmol), (4) probe 1 (6 nmol), and (5) probe 1 (0.6 nmol) in loading buffer containing GSH (10 mM) at pH 7.4. c. Quantification of fluorescence signals of probe $1 (100 \,\mu\text{M})$ in PBS buffer at pH 7.4, probe 1 $(100 \ \mu\text{M})$ with GSH $(10 \ \text{mM})$ at pH 7.4, and probe 2 $(100 \ \mu\text{M})$ with GSH $(10 \ \text{mM})$ at pH 7.4 after 30 min incubation. d. Fluorescence changes of probe 1 at 0.1, 1, 10 µM and probe 2 at 10 μ M over two hours at pH 7.4. e. Fluorescence images of sealed PE tubes of (1) and (2) probe 1 (100 μ M) in PBS buffer at pH 7.4, (3) and (4) probe 1 (100 μ M) in PBS buffer containing GSH (10 mM) at pH 7.4, (5) and (6) probe 2 (100 µM) in PBS buffer containing GSH (10 mM) at pH 7.4. (7) and (8) probe 1 (100 µM) in PBS buffer at pH 7.4, (9) and (10) probe 1 (100 µM) in PBS buffer containing GSH (10 mM) at pH 7.4, (11) and (12) probe 2 (100 µM) in PBS buffer containing GSH (10 mM) at pH 7.4. f. Processed photoacoustic images (3D maximization) of the same sample tubes (1)-(6) in e.g. Quantification of photoacoustic signals of probe 1 (100 μ M) in PBS buffer at pH 7.4, probe 1 (100 μ M) with GSH (10 mM) at pH 7.4, and probe 2 (100 µM) with GSH (10 mM) at pH 7.4. Standard deviations are shown by error bars.



Figure 4. Fluorescence and photoacoustic imaging of polymerizable probe in living mice. a. Fluorescence imaging of subcutaneously injected probe **1** (right spot, 1 nmol) and **2** (left spot, 1 nmol) in living mice in 24 hours after injection of the probe. Area of injection was pre-treated with GSH. The images are presented at different intensity scales to show the contrast of the two reaction spots. **b.** Ratio of fluorescence intensities of probe **1** and **2** vs. time. Red symbols and line, fluorescence ratio of probe **1** and **2**. **c.** First two hours of graph **b. d.** Co-registered ultrasound and photoacoustic images of subcutaneously injected probe **1** (top, 1 nmol) and **2** (bottom, 1 nmol) in living mice in 24 hours after injection of the probe. Area of injection was pre-treated with GSH. White scale bar in the ultrasound windows is 3 mm. The yellow dash-lined rectangular indicates the area of interest for photoacoustic signals. **e.** Changes of photoacoustic intensities of probe **1** and **2** vs. time. Blue symbols and line, probe **1**; red symbols and line, probe **2**. **f.** The first two hours of graph **b.** Standard deviations are shown by error bars. ******, P<0.0001.



Figure 5. Fluorescence and photoacoustic imaging of polymerizable probe in tumor models. a. Longitudinal fluorescence imaging of tumor-bearing mice with i.v. injected probe 1 (20 nmol) in 8 hours. Tumor (indicated by red arrow) was pre-treated with GSH to activate probe **1**. **b.** Comparison of the average tumor fluorescence radiance intensity of mice groups injected with probe **1** (20 nmol, GSH treatment, red circle, n = 5; 20 nmol, saline treatment, blue square, n = 5) or probe **2** (20 nmol, GSH treatment, black diamond, n = 5) vs. time (24 hours). Standard deviations are shown by error bars. **c.** First four hours of graph b. **d.** PA and US co-registered tumor images of mice with i.v. injected probe **1** and probe **2** (each at 20 nmol) before and one hour after injection. Tumor was pre-treated with intratumoral injection of GSH. Anatomical position of the tumor is indicated by red (US images) or white (PA images) arrows. A &B, US (A) and PA (B) images of target area before injection. C & D, US (C) and PA (D) images of target area one hour after i.v. injected with probe. **e.** Comparison of the average tumor PA intensity of mice groups injected with probe **1**-

activated (20 nmol, GSH treatment, red circle, n = 5), probe 1-unactivated (20 nmol, saline treatment, blue square, n = 5), and probe 2 (20 nmol, GSH treatment, black diamond, n = 5) vs. time. **f.** First four hours of graph **e. g.** Tumor to muscle ratio of the PA signal of the same mice groups in **e. h.** First four hours of graph **g.** All error bars indicate standard deviation. **, *P*<0.0001; #, not significant.