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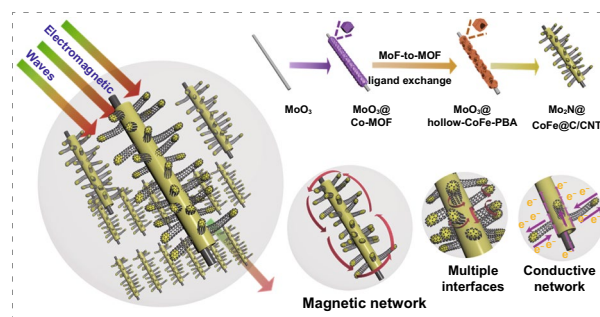
# Hierarchical Magnetic Network Constructed by CoFe Nanoparticles Suspended Within “Tubes on Rods” Matrix Toward Enhanced Microwave Absorption

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## HIGHLIGHTS

- Three-dimension hierarchical core-shell  $\text{Mo}_2\text{N}@\text{CoFe}@C/\text{CNT}$  composites were successfully constructed via a fast MOF-based ligand exchange strategy.
- Abundant magnetic CoFe nanoparticles suspended within “nanotubes on microrods” matrix exhibited strong magnetic loss capability, confirmed by off-axis electron holography.
- Hierarchical  $\text{Mo}_2\text{N}@\text{CoFe}@C/\text{CNT}$  composites displayed remarkable microwave absorption value of  $-53.5$  dB.

**ABSTRACT** Hierarchical magnetic-dielectric composites are promising functional materials with prospective applications in microwave absorption (MA) field. Herein, a three-dimension hierarchical “nanotubes on microrods,” core-shell magnetic metal-carbon composite is rationally constructed for the first time via a fast metal-organic frameworks-based ligand exchange strategy followed by a carbonization treatment with melamine. Abundant magnetic CoFe nanoparticles are embedded within one-dimensional graphitized carbon/carbon nanotubes supported on micro-scale  $\text{Mo}_2\text{N}$  rod ( $\text{Mo}_2\text{N}@\text{CoFe}@C/\text{CNT}$ ), constructing a special multi-dimension hierarchical MA material. Ligand exchange reaction



is found to determine the formation of hierarchical magnetic-dielectric composite, which is assembled by dielectric  $\text{Mo}_2\text{N}$  as core and spatially dispersed CoFe nanoparticles within C/CNTs as shell.  $\text{Mo}_2\text{N}@\text{CoFe}@C/\text{CNT}$  composites exhibit superior MA performance with maximum reflection loss of  $-53.5$  dB at 2 mm thickness and show a broad effective absorption bandwidth of 5.0 GHz. The  $\text{Mo}_2\text{N}@\text{CoFe}@C/\text{CNT}$  composites hold the following advantages: (1) hierarchical core-shell structure offers plentiful of heterojunction interfaces and triggers interfacial polarization, (2) unique electronic migration/hop paths in the graphitized C/CNTs and  $\text{Mo}_2\text{N}$  rod facilitate conductive loss, (3) highly dispersed magnetic CoFe nanoparticles within “tubes on rods” matrix build multi-scale magnetic coupling network and reinforce magnetic response capability, confirmed by the off-axis electron holography.

**KEYWORDS** Hierarchical core-shell MOF-based composites; CoFe nanoparticles; Magnetic network; Microwave absorption

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## 1 Introduction

Coming into the fifth-generation (5G) wireless communication systems, the increasing usage of diverse electronic productions has caused severe electromagnetic radiation pollution, which results in an urgent pursuit for high-performance microwave absorption (MA) materials [1–9]. Magnetic materials, including metals (Co, Ni, Fe) and metallic alloys (FeCo, NiCo, etc.), are generally used as microwave absorbents due to strong magnetic loss ability [10–18]. However, practical applications of magnetic materials suffer from their inherent drawbacks: undesirable chemical stability, severe aggregation and inferior impedance matching [19–21]. To tackle these obstacles, two typical strategies have been commonly employed to shape MA properties. One is to decorate magnetic component with carbon materials to develop magnetic-dielectric system and thereby boost the MA performance by enhancing dielectric loss and improving impedance matching [22–31]. For example, Cao et al. designed Fe@NCNTs composite and showed MA performance of  $-30.43$  dB [32]. Shui et al. prepared CoFe/carbon fiber composite with high MA properties [13]. Tong et al. designed Co/C/Fe/C composite which exhibited significantly improved MA abilities [11]. The other is to construct hierarchical-structured materials with well-designed nano-units, thus achieving high dispersion of magnetic particles and producing heterogeneous interface in multicomponent materials [33–37]. Among various hierarchical structures, the core-shell structures have attracted growing attention in the MA field [38–40] such as Co@C microspheres [10], Fe<sub>3</sub>O<sub>4</sub>/C [41], Co@CoO [42], Co<sub>20</sub>Ni<sub>80</sub>@TiO<sub>2</sub> core-shell structure [43]. The delicately designed core-shell composites can satisfy magnetic and dielectric loss simultaneously resulting from synergistic effects of different components within both core and shell [44, 45]. Besides, large interspace and heterogeneous interface created by core-shell structure can further enhance polarization loss and strengthen multi-reflection process [37, 44, 46]. Particularly, hierarchical 1D units assembled core-shell composites exhibit remarkable performance in MA application [14, 47, 48]. For example, Che et al. designed hierarchically tubular C/Co composite with abundant 1D nanotubes and achieved highly uniform distribution

of Co nanoparticles and outstanding MA performance [49]. Therefore, it is highly desirable to develop a facile and effective preparation strategy to construct magnetic metal-carbon composites with hierarchical core-shell structure.

Metal-organic frameworks (MOFs), with diverse microstructure and adjustable composition, have been widely utilized to construct various hierarchical composites [50–54]. MOF-derived materials demonstrate inherent advantages of abundant metal/carbon components, which endows them with great potential in MA application [40, 55–60]. For example, Ji et al. developed MOF-derived one-dimensional sponge-like metallic Co and Co/C composites with strong magnetic loss [61]. Du et al. presented a MOFs-derived method to construct hollow Co/C microspheres as microwave absorbents [62]. Zhao et al. prepared hierarchical Fe-Co/N-doped carbon/rGO composites derived from Fe-doped Co-MOF [63]. However, direct transforming MOFs into microwave absorbents leads to a much lower ratio of metal nanoparticles and poor graphitization degree of carbon or CNTs, which is unfavorable to the attenuation of microwave. To tackle these problems, the MOF precursor can be further extended by transforming one kind of MOF into another via ion exchange reactions or ligand exchange reactions, introducing more magnetic metals and carbon components. For example, Hu et al. constructed hierarchical bimetallic Co<sub>2</sub>[Fe(CN)<sub>6</sub>] hollow structure from a Co-MOF through ion exchange reactions [64]. This MOF-to-MOF strategy inspires us to construct bimetallic MOF-derived carbon-based absorbents with favorable hierarchical structure, which has rarely been reported in MA field.

Recently, transition metal molybdenum-based materials, such as MoO<sub>2</sub>, Mo<sub>2</sub>C, MoS<sub>2</sub> and Mo<sub>2</sub>N, have emerged as effective candidates in the field of electrocatalysis, lithium batteries and supercapacitors due to its low cost, high conductivity and chemical stability [65–72]. Such superior properties also make molybdenum compounds promising microwave absorbents. For example, owing to metallic-like conductivity of MoO<sub>2</sub> materials, Huang et al. constructed C@MoO<sub>2</sub>/G composites for efficient MA [73]. Du et al. fabricated ternary Mo<sub>2</sub>C/Co/C composites for MA [74] and Jin et al. prepared MoS<sub>2</sub>-NS with high dielectric properties and MA performances [75]. However, the work

of employing  $\text{Mo}_2\text{N}$  as microwave absorbent has not been studied so far, although  $\text{Mo}_2\text{N}$  materials exhibit satisfied electrical conductivity displaying excellent performance in electrocatalysis and supercapacitors [66, 69]. Therefore, compositing molybdenum compounds into metal–carbon absorbents with designed hierarchical structure is expected to achieve first-rate MA performance.

Herein, for the first time, a 3D hierarchical “nanotubes on microrods” core–shell composite of magnetic CoFe nanoparticles suspended within one-dimensional graphitized C/CNTs supported on  $\text{Mo}_2\text{N}$  rod ( $\text{Mo}_2\text{N}@ \text{CoFe}@ \text{C}/\text{CNT}$ ) is successfully achieved through a fast MOF-based ligand exchange strategy. The intermediate product of  $\text{MoO}_3@$  hollow-CoFe-PBA composite plays an important role in not only providing Fe source for the growth of CoFe alloy and C/CNTs but also constructing hierarchical core–shell structure in final composite, thus achieving highly dispersive distribution of magnetic particles. The unique  $\text{Mo}_2\text{N}@ \text{CoFe}@ \text{C}/\text{CNT}$  composite holds the dielectric  $\text{Mo}_2\text{N}$  as core and magnetic CoFe nanoparticles embedded C/CNTs as shell. Such 3D hierarchical magnetic network assembled by CoFe nanoparticles suspended within “tubes on rods” matrix demonstrates strong magnetic loss capability, which can be verified by off-axis electron holography. Besides, numerous  $\text{Mo}_2\text{N}$  rods and graphitized CNTs in the composite constitute dual conductive network to facilitate conductive loss. Moreover, large interfaces in hierarchical core–shell structure can trigger intensive polarization loss. Our hierarchical  $\text{Mo}_2\text{N}@ \text{CoFe}@ \text{C}/\text{CNT}$  composite demonstrates superior MA performance with maximum reflection loss value of  $-53.5$  dB at the thickness of only 2 mm thickness and the effective absorption bandwidth can reach 5.0 GHz. Therefore, the presented fast MOF-based ligand exchange strategy provides an effective method to fabricate multicomponent absorbents with well-controlled hierarchical structure for achieving excellent MA properties.

## 2 Experimental Section

### 2.1 Materials

All chemicals used were of analytical grade and were used directly without further purification. All chemicals were purchased from Sinopharm Chemical Reagent Co., Ltd.

### 2.2 Synthesis of $\text{MoO}_3$

In a typical synthesis, 0.5793 g ammonium molybdate tetrahydrate was dissolved in 30 mL of deionized (DI) water; then, 2.5 mL of  $\text{HNO}_3$  was added. The solution was kept stirring for 10 min, then transferred into a Teflon-lined stainless autoclave (50 mL) and kept at  $180$  °C for 12 h. When the temperature of Teflon-lined stainless autoclave was cooled naturally, the precipitate was collected and washed repeatedly with DI water for at least three times before drying at  $70$  °C.

### 2.3 Synthesis of $\text{MoO}_3@ \text{Co-MOF}$

First, the solution A was prepared by 50 mg of  $\text{MoO}_3$  and 0.582 g  $\text{CoNO}_3 \cdot 6\text{H}_2\text{O}$  were dissolved in 20 mL of methanol. Then solution B was prepared by dispersing 1.3132 g of 2-methylimidazole in 20 mL of methanol. The solution B was added into solution A under stirring and kept stirring for 5 min then aged for 20 min at room temperature. The precipitate was collected and washed with ethanol for at least three times and dried at  $70$  °C.

### 2.4 Synthesis of $\text{MoO}_3@ \text{hollow-CoFe-PBA}$

40 mg of  $\text{MoO}_3@ \text{Co-MOF}$  was dissolved in 10 mL ethanol to get solution C. 40 mg of  $\text{K}_3[\text{Fe}(\text{CN})_6]$  was dissolved in 20 mL DI water and 20 mL ethanol to get solution D. Then solution D was poured into solution C under stirring and kept stirring for 5 min. The precipitate was collected and washed with DI water and dried at  $70$  °C.

### 2.5 Synthesis of $\text{Mo}_2\text{N}@ \text{CoFe}@ \text{C}/\text{CNT}$

In a typical synthesis, 0.1 g of as-prepared  $\text{MoO}_3@ \text{hollow-CoFe-PBA}$  and 0.5 g of melamine were placed separately in a quartz boat where the melamine was placed at upstream side of the furnace. The furnace was heated to  $600$  °C at a rate of  $2$  °C  $\text{min}^{-1}$  for 4 h under a hydrogen/argon atmosphere. Finally,  $\text{Mo}_2\text{N}@ \text{CoFe}@ \text{C}/\text{CNT}$  composite was obtained after cooling down to ambient temperature.



## 2.6 Synthesis of Mo<sub>2</sub>N and Mo<sub>2</sub>N@Co/CNT

For comparison, Mo<sub>2</sub>N and Mo<sub>2</sub>N@Co/CNT were synthesized by calcining the MoO<sub>3</sub> and MoO<sub>3</sub>@Co-MOF with melamine, respectively.

## 2.7 Microwave Absorption Measurements

The measured samples were first prepared by adding the absorbents (20 wt%) into molten paraffin and uniformly mixing them, followed by modeling into a coaxial ring with the outer diameter of 7.0 mm and inner diameter of 3.0 mm. Electromagnetic parameters (complex permittivity and complex permeability) were measured by a N5230C vector network analyzer over the range of 2–18 GHz. The reflection loss values were calculated based on the transmission line theory:

$$Z_{in} = \sqrt{\mu_r/\epsilon_r \tanh \left| -j(2\pi fd/c) \sqrt{\epsilon_r \mu_r} \right|} \quad (1)$$

$$RL(\text{dB}) = -20 \log \left| Z_{in} - 1/Z_{in} + 1 \right| \quad (2)$$

where  $\epsilon_r$  and  $\mu_r$  are the complex permittivity ( $\epsilon_r = \epsilon' - j\epsilon''$ ) and permeability ( $\mu_r = \mu' - j\mu''$ ), respectively,  $f$  is the frequency of microwave,  $c$  is the velocity of light,  $d$  is the thickness, and  $Z_{in}$  is the normalized input impedance of the sample.

## 2.8 Characterizations

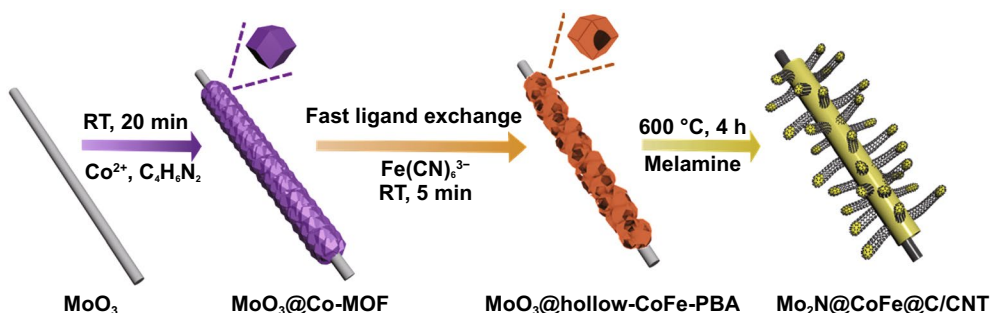
The crystalline phase and purity of the products was analyzed by powder X-ray diffraction (XRD, Bruker, D8-Advance X-ray diffractometer, Germany) using

Ni-filtered Cu K $\alpha$  radiation. The morphology and structure of the products were examined by a field-emission scanning electron microscopy (SEM) on a Hitachi S-4800 with an accelerating voltage of 5 kV and a field-emission transmission electron microscope (TEM, JEOL, JEM-2100F, 200 kV). The Raman spectra were acquired with a Renishaw Invia spectrometer using a 514 nm laser excitation. X-ray photoelectron spectroscopy (XPS) spectra were obtained on an ESCALab MKII X-ray photoelectron spectrometer using Al K $\alpha$  X-ray as the excitation source. The hysteresis loops were performed with a superconducting quantum interference device (MPMS(SQUID) VSM) magnetometer (Quantum Design Company).

## 3 Results and Discussion

### 3.1 Fabrication and Characterization of Mo<sub>2</sub>N@CoFe@C/CNT Composites

The synthesis of the hierarchical Mo<sub>2</sub>N@CoFe@C/CNT core-shell structure is illustrated in Fig. 1. First, the Co-MOF is uniformly grown on MoO<sub>3</sub> rod to form MoO<sub>3</sub>@Co-MOF structure. Second, through a fast ligand exchange reaction with K<sub>3</sub>[Fe(CN)<sub>6</sub>] in 5 min at room temperature, MoO<sub>3</sub>@Co-MOF structure is in situ converted into MoO<sub>3</sub>@hollow-CoFe-PBA core-shell composite. Followed by the carbonization of MoO<sub>3</sub>@hollow-CoFe-PBA with melamine, the inner MoO<sub>3</sub> is transformed into Mo<sub>2</sub>N rod and the outer hollow-CoFe-PBA turn into the CoFe@C/CNTs architecture, where thermally reduced CoFe nanoparticles could catalyze the growth of graphitic carbon and CNTs with melamine as carbon source. Finally, the hierarchical Mo<sub>2</sub>N@CoFe@C/CNT composite with “tubes on rods” structure is successfully obtained. Moreover, through fast ligand exchange reaction, the intermediate product



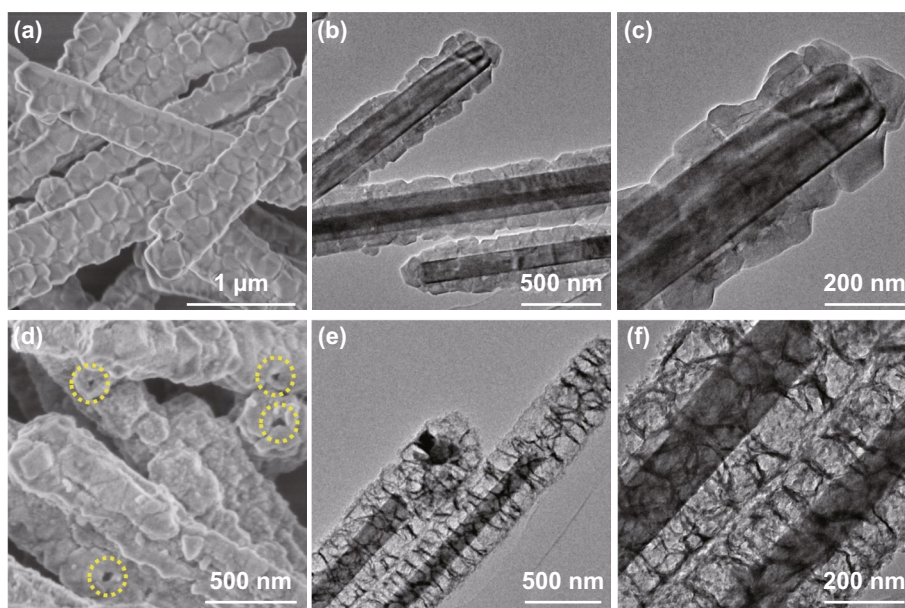
**Fig. 1** Schematic process of the fast MOF-based ligand exchange strategy for construction of 3D hierarchical Mo<sub>2</sub>N@CoFe@C/CNT composites

of  $\text{MoO}_3$ @hollow-CoFe-PBA core-shell structure plays a critical role in the formation of hierarchical  $\text{Mo}_2\text{N}$ @CoFe@C/CNT composite, which will be explained in the following discussion.

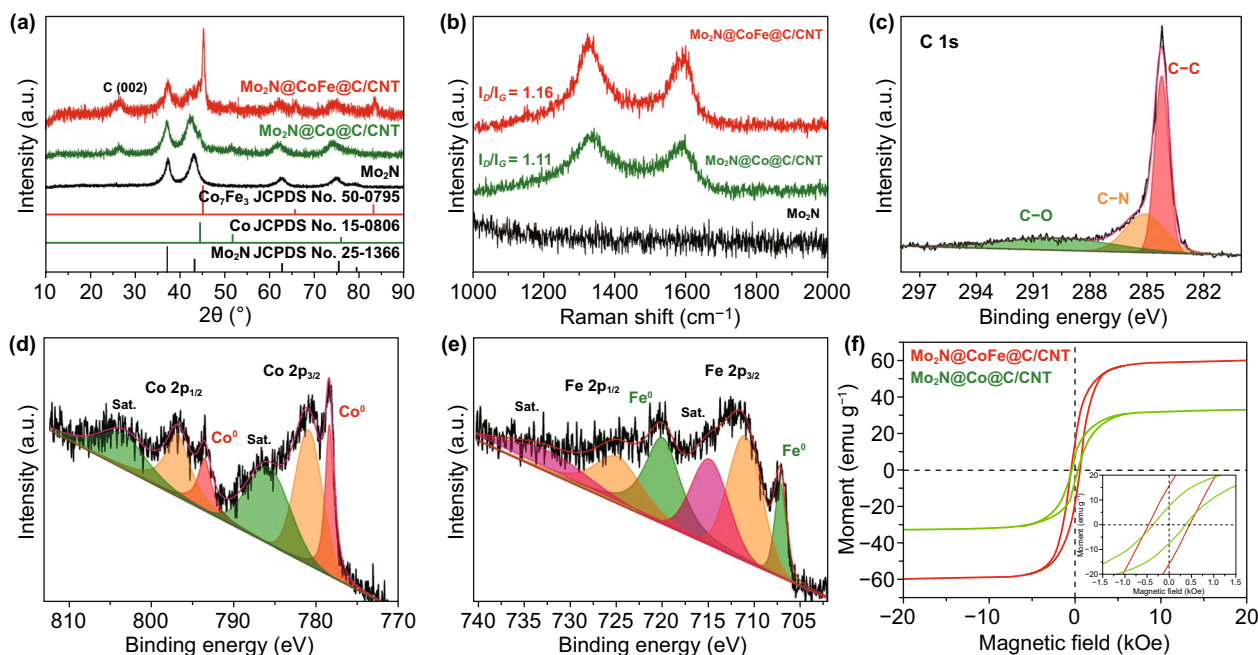
As displayed in Fig. S1, as-prepared uniform  $\text{MoO}_3$  rods demonstrate smooth surface and high phase purity. Then  $\text{MoO}_3$  rods are covered with Co-MOF to form  $\text{MoO}_3$ @Co-MOF structure. The SEM images reveal that the surface of  $\text{MoO}_3$  rods becomes rough (Fig. 2a). The core of  $\text{MoO}_3$  and shell of Co-MOF can be clearly observed in TEM images (Fig. 2b, c). And both diffraction peaks of  $\text{MoO}_3$  and Co-MOF are well detected in XRD pattern (Fig. S2), indicating that Co-MOF is successfully grown on the  $\text{MoO}_3$  rods. To construct hollow CoFe-PBA on the  $\text{MoO}_3$  rods, the  $\text{MoO}_3$ @Co-MOF samples are kept in  $\text{K}_3[\text{Fe}(\text{CN})_6]$  solution and stirred for just 5 min at room temperature to allow the ligand exchange reaction to prepare  $\text{MoO}_3$ @hollow-CoFe-PBA structure. Firstly, the  $\text{MoO}_3$ @Co-MOF will slowly decompose in water/ethanol to release  $\text{Co}^{2+}$  ions. Then the  $[\text{Fe}(\text{CN})_6]^{3-}$  ions are injected into the reaction solution. The released  $\text{Co}^{2+}$  ions can interact with  $[\text{Fe}(\text{CN})_6]^{3-}$  ions to generate CoFe-PBA shell around the framework of the precursors (Co-MOF). Finally, the solid Co-MOF shell is completely converted into hollow CoFe-PBA, and  $\text{MoO}_3$ @hollow-CoFe-PBA core-shell composites are obtained. As displayed in

Fig. 2d, the rather rough CoFe-PBA is grown on the  $\text{MoO}_3$  rods and some holes can be seen on the surface (as displayed in the yellow circles of Fig. 2d). Such unique shell of hollow CoFe-PBA can be further confirmed by TEM images. In Fig. 2e, f, the as-prepared  $\text{MoO}_3$ @hollow-CoFe-PBA structure is consisted of the nanocage-assembled CoFe-PBA shell and the  $\text{MoO}_3$  core. XRD result also demonstrates that the sample is composed of  $\text{MoO}_3$  and  $\text{Co}_2[\text{Fe}(\text{CN})_6]$  (Fig. S3) [64]. Such core-shell of  $\text{MoO}_3$ @hollow-CoFe-PBA composite plays a significant role not only in providing the Fe source for the growth of CoFe alloys and CNTs but also in constructing the core-shell structure in the final multicomponent products. Subsequently, the  $\text{MoO}_3$ @hollow-CoFe-PBA composite is converted into  $\text{Mo}_2\text{N}$ @CoFe@C/CNT core-shell structure through the carbonization with melamine. For comparison,  $\text{Mo}_2\text{N}$  rod and  $\text{Mo}_2\text{N}$ @Co/CNT samples are also synthesized by calcining the  $\text{MoO}_3$  and  $\text{MoO}_3$ @Co-MOF composite with melamine, respectively.

The chemical compositions of  $\text{Mo}_2\text{N}$  rod,  $\text{Mo}_2\text{N}$ @Co/CNT, and  $\text{Mo}_2\text{N}$ @CoFe@C/CNT composites are measured by XRD, Raman and XPS techniques. As displayed in Fig. 3a, the diffraction peaks of  $\text{Mo}_2\text{N}$  rods are in accordance with reflections of molybdenum nitride ( $\text{Mo}_2\text{N}$ , JCPDS No. 25-1366) while the  $\text{Mo}_2\text{N}$ @Co/CNT samples exhibit diffraction peaks of both  $\text{Mo}_2\text{N}$  and



**Fig. 2** a SEM, b, c TEM images of  $\text{MoO}_3$ @Co-MOF, d SEM, e, f TEM images of  $\text{MoO}_3$ @hollow-CoFe-PBA composites

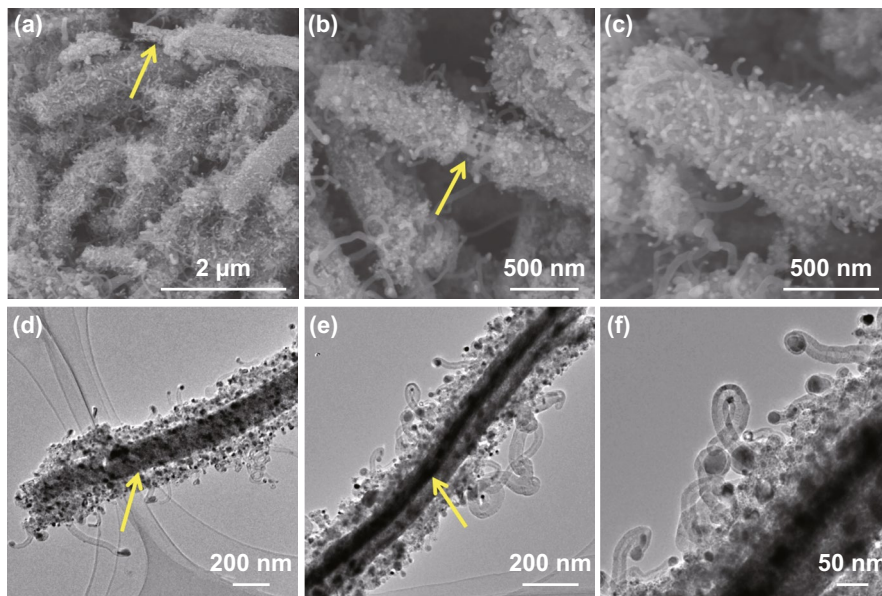


**Fig. 3** **a** XRD patterns, **b** Raman spectra of as-prepared  $\text{Mo}_2\text{N}$ ,  $\text{Mo}_2\text{N@Co/CNT}$  and  $\text{Mo}_2\text{N@CoFe@C/CNT}$ . High resolution XPS spectra of **c** C 1s, **d** Co 2p, and **e** Fe 2p for  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composite. **f** hysteresis loops of  $\text{Mo}_2\text{N@Co/CNT}$  and  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composites

cubic cobalt (JPCDS No. 15-0806). In the XRD pattern of  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composites, apart from characteristic peaks of  $\text{Mo}_2\text{N}$ , a diffraction peak at  $26.1^\circ$  can be observed clearly, attributing to the (002) plane of the graphitic carbon. Other peaks at around  $45.2^\circ$ ,  $65.8^\circ$ , and  $83.3^\circ$  match well with diffractions of the cubic cobalt iron (JPCDS No. 50-0795). Above-mentioned XRD results demonstrate that the  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composite is consisted of  $\text{Mo}_2\text{N}$ , CoFe alloy and graphitic carbon. To reveal the graphitic feature and structural defects of as-prepared samples, Raman spectra are conducted. In Fig. 3b, the  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composite exhibits the highest  $I_D/I_G$  value of 1.16 because a great number of defects are produced in such core-shell structure. The value of  $I_D/I_G$  is increased with more CNTs catalyzed by the CoFe alloy compared with less graphitic carbon by single metal Co in  $\text{Mo}_2\text{N@Co/CNT}$  sample, which could promote the electronic transportation ability. Chemical valence states of  $\text{Mo}_2\text{N@CoFe@C/CNT}$  are examined via XPS technique. In Fig. 3c, three peaks of C 1s spectrum correspond to the C-C (284.28 eV), C-N (285.16 eV) and C-O (189.73 eV) [76]. In the Co 2p spectrum, peaks at 778.32 and 793.44 eV are ascribed to  $\text{Co}^0$  in Co  $2p_{3/2}$  and Co  $2p_{1/2}$  and peaks at 780.82 and 796.67 eV belong to  $\text{Co}^{2+}$

species. In Fig. 3e, the Fe 2p spectrum can be decomposed into two peaks of 707.22 eV for  $\text{Fe}^0 2p_{3/2}$  and 719.97 eV for  $\text{Fe}^0 2p_{1/2}$  and other two peaks of 711.06 and 724.85 eV for  $\text{Fe}^{2+} 2p_{3/2}$  and  $2p_{1/2}$ , respectively [63, 77–79]. The bimetal CoFe with multiple valency in  $\text{Mo}_2\text{N@CoFe@C/CNT}$  sample could result in higher saturation magnetization. As shown in Fig. 3f, the saturation magnetization ( $M_s$ ) value of  $\text{Mo}_2\text{N@CoFe@C/CNT}$  is  $59.6 \text{ emu g}^{-1}$ , which is higher than that of  $\text{Mo}_2\text{N@Co/CNT}$  sample. And the coercivity value is 449.6 Oe for  $\text{Mo}_2\text{N@CoFe@C/CNT}$  composite. Such high saturation magnetization and low coercivity of  $\text{Mo}_2\text{N@CoFe@C/CNT}$  hierarchical structure could boost magnetic storage and reinforce magnetic loss, further promoting MA performance [57, 80].

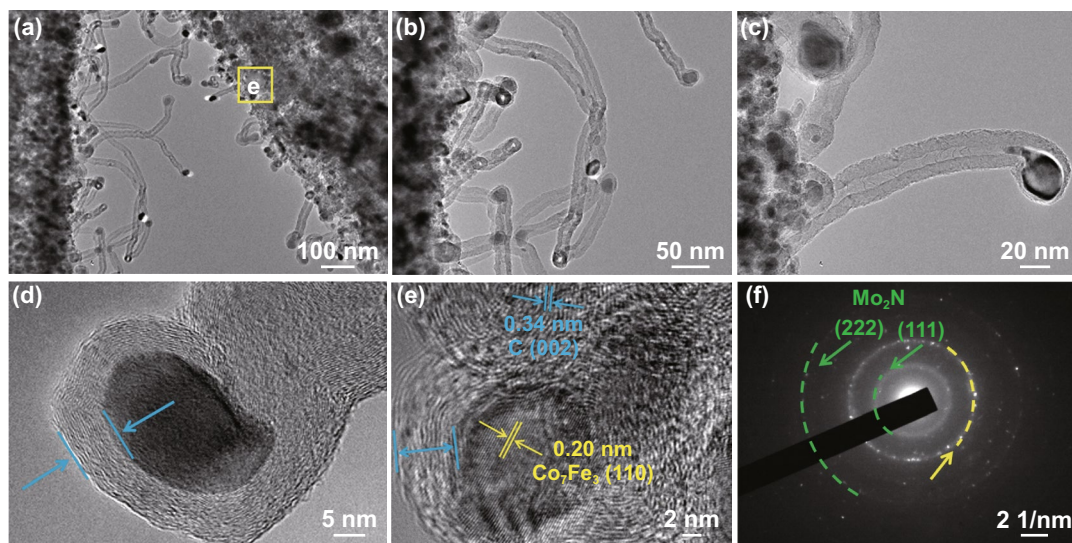
The morphology and structure of  $\text{Mo}_2\text{N}$  rod,  $\text{Mo}_2\text{N@Co/CNT}$  and  $\text{Mo}_2\text{N@CoFe@C/CNT}$  core-shell composites are further performed with SEM and TEM images. As displayed in Fig. 4, a large number of CNTs are produced and deposited on the core of  $\text{Mo}_2\text{N}$  rod which can be clearly observed in Fig. 4a–c with yellow arrows. In the following TEM images, the rod-like core is seen and wrapped by outer shell of numerous CNTs. Particularly, the obvious void exists between the shell and core (Fig. 4d–f) and the CNTs are not directly grown on the  $\text{Mo}_2\text{N}$  rod but supported by the



**Fig. 4** a–c SEM, d–f TEM images of Mo<sub>2</sub>N@CoFe@C/CNT composites

shell of CoFe alloy embedded graphitic carbon layers. Such uniquely hierarchical Mo<sub>2</sub>N@CoFe@C/CNT core–shell structure is reported for the first time and can be further confirmed by the magnified TEM and HRTEM images. Abundant CNTs can be seen and on the top of each CNT is encapsulated metal nanoparticles, which are wrapped by numbers of graphitic carbon layers (Fig. 5a–d). In Fig. 5e, the HRTEM image obtained from the shell of such Mo<sub>2</sub>N@

CoFe@C/CNT structure (as marked in Fig. 5a with yellow square) demonstrates that the interplanar spacing of 0.20 nm can correspond to the (110) plane of CoFe alloy and 0.34 nm to the (002) plane of graphitic carbon, which convincingly confirms such unique shell of CoFe nanoparticles embedded graphitic carbon. The corresponding selected area electron diffraction pattern displays a series of diffraction rings which can be well indexed to diffraction planes of crystalline Mo<sub>2</sub>N



**Fig. 5** a–c The magnified TEM, d–e HRTEM images and f corresponding selected area electron diffraction of Mo<sub>2</sub>N@CoFe@C/CNT composites

and CoFe alloy (Fig. 5f). Clearly, based on the above results of morphology and composition, hierarchical Mo<sub>2</sub>N@CoFe@C/CNT “tubes on rods” architecture is successfully synthesized through a fast MOF-based ligand exchange strategy. In the calcinating process of MoO<sub>3</sub>@hollow-CoFe-PBA composite with melamine, the MoO<sub>3</sub> is converted into the core of Mo<sub>2</sub>N rod and hollow-CoFe-PBA is transformed into the shell of CoFe alloy embedded C/CNTs with thermally reduced CoFe nanoparticles as catalysts and melamine as carbon source. For comparison, Mo<sub>2</sub>N@Co/CNT sample is obtained by directly annealing MoO<sub>3</sub>@Co-MOF composite with melamine. As shown in Fig. S4, the Mo<sub>2</sub>N@Co/CNT composites maintain the rod structure but only few of CNTs are observed on the surface of Mo<sub>2</sub>N rod without the shell of metal-embedded graphitic carbon framework. This evidence suggests that single Co nanoparticles could not effectively catalyze the growth of CNTs. Obviously, MoO<sub>3</sub>@hollow-CoFe-PBA structure constructed by ligand exchange reaction critically determines the formation of CoFe nanoparticles, graphitic C/CNTs and hierarchical core-shell structure. Rod-like Mo<sub>2</sub>N are prepared through annealing MoO<sub>3</sub> rods with melamine, which displays uniformly smooth rod structure (Fig. S5). Remarkably, as-prepared hierarchical Mo<sub>2</sub>N@CoFe@C/CNT can be considered as both distinct conductive structure and magnetic network, which hold great potential to achieve superior MA ability.

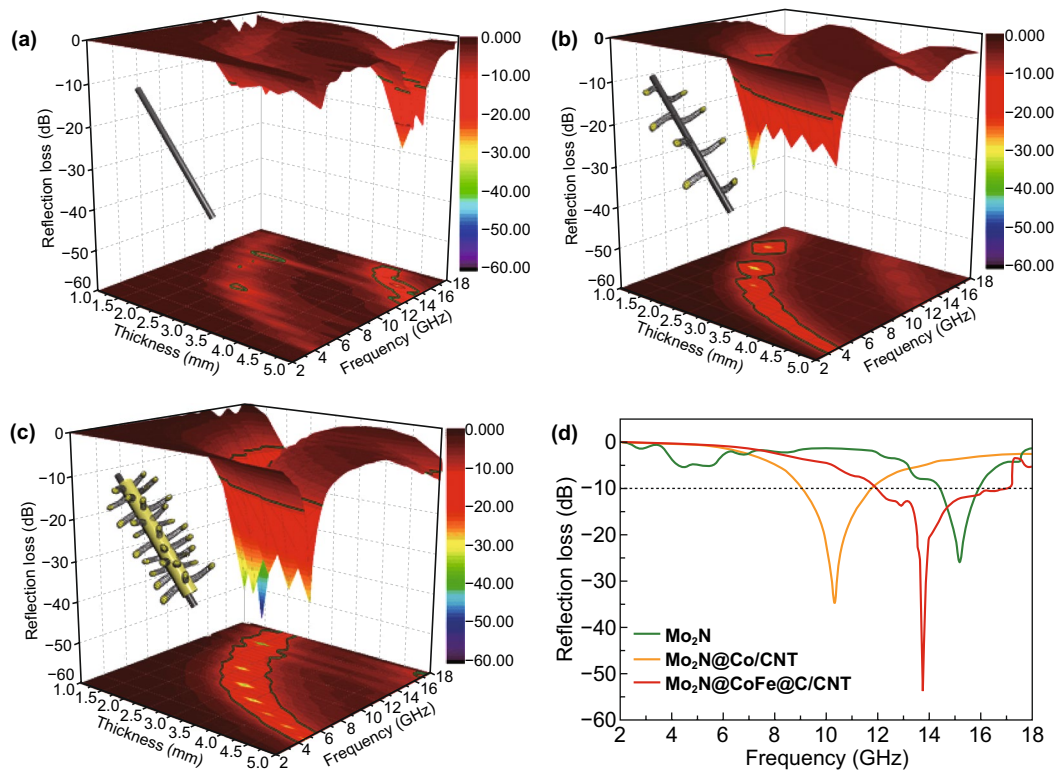
### 3.2 Electromagnetic Parameters Analysis and Microwave Absorption Ability

Related electromagnetic parameters of as-prepared Mo<sub>2</sub>N@CoFe@C/CNT, Mo<sub>2</sub>N@Co/CNT and Mo<sub>2</sub>N samples are investigated to reveal the impacts of structure and composition on the MA performance. Generally, MA properties are highly determined by the complex permittivity and complex permeability of materials. It is acknowledged that the real parts of complex permittivity ( $\epsilon'$ ) and complex permeability ( $\mu'$ ) indicate the capability of storing electromagnetic energy, while the imaginary parts ( $\epsilon''$ ,  $\mu''$ ) imply the ability to loss electromagnetic energy. As shown in Fig. S8, the pure Mo<sub>2</sub>N sample displays real permittivity ( $\epsilon'$ ) ranging from 12.06 to 10.92, suggesting the Mo<sub>2</sub>N is a better dielectric material. And the  $\epsilon'$  values of Mo<sub>2</sub>N@Co/CNT samples rise obviously from 19.40 to 12.76 due to the introduction of conductive CNTs. When more CoFe alloy embedded CNTs

and graphitic carbon layers are introduced, the  $\epsilon'$  values of Mo<sub>2</sub>N@CoFe@C/CNT sample range from 10.2 to 5.6 with the increase in frequency, demonstrating Mo<sub>2</sub>N@CoFe@C/CNT materials gain strong capability of energy storage and high dielectric polarization. And the  $\epsilon''$  values of Mo<sub>2</sub>N@CoFe@C/CNT also remain high from 3.78 to 2.56, which means a powerful dielectric loss ability. This can be ascribed to the hierarchical conductive network and enhanced interfacial polarization resulting from unique core-shell structure of dielectric Mo<sub>2</sub>N and conductive C/CNTs components. To further evaluate the dielectric loss property, the dielectric loss tangent  $\delta_\epsilon$  ( $\tan \delta_\epsilon = \epsilon''/\epsilon'$ ) was calculated. It is believed that higher  $\tan \delta_\epsilon$  value means more electric energy of incident microwaves would be dissipated. As shown in Fig. S9a, the  $\tan \delta_\epsilon$  values of Mo<sub>2</sub>N@CoFe@C/CNT remain high, which offers the convincing evidence that the design of hierarchically core-shell structure with the combination of dielectric Mo<sub>2</sub>N and graphitic C/CNTs components is an effective way to enhance the dielectric loss capacity. As for the real ( $\mu'$ ) and imaginary ( $\mu''$ ) parts of permeability, the  $\mu'$  and  $\mu''$  values of Mo<sub>2</sub>N remain close to 1 and 0 due to its nonmagnetic property. Compared with Mo<sub>2</sub>N@Co/CNT samples, the  $\mu'$  and  $\mu''$  of Mo<sub>2</sub>N@CoFe@C/CNT are higher because of its enhanced magnetic CoFe alloy component and hierarchical 3D magnetic network. Therefore, the Mo<sub>2</sub>N@CoFe@C/CNT material is prone to generate favorable magnetic loss capability. Based on above discussion, as-prepared Mo<sub>2</sub>N@CoFe@C/CNT composite is expected to exhibit superior MA capability originating from its both synergetic strong dielectric dissipation and magnetic loss.

The MA performance of absorbents is generally evaluated with the maximum reflection loss (*RL*) value and effective absorption bandwidth. Figure 6 displays the 3D plots of *RL* values on different thickness of Mo<sub>2</sub>N, Mo<sub>2</sub>N@Co/CNT and Mo<sub>2</sub>N@CoFe@C/CNT samples. The Mo<sub>2</sub>N rods exhibit good MA performance with the maximum *RL* value of -25.9 dB at the thickness of 4.5 mm (Fig. 6a) due to its high dielectric property. With the introduction of Co/CNTs components, the Mo<sub>2</sub>N@Co/CNT materials exhibit MA with the maximum *RL* value of -34.8 dB. Significantly, as displayed in Fig. 6c, the Mo<sub>2</sub>N@CoFe@C/CNT demonstrates the best MA performance with highest maximum *RL* value of -53.5 dB at the thickness of only 2 mm thickness, and the effective absorption bandwidth can reach 5 GHz (from 12 to 17 GHz). Moreover, while tuning the thickness from 1.5 to 5.0 mm, Mo<sub>2</sub>N@CoFe@C/CNT samples still





**Fig. 6** 3D plots of reflection loss of **a** Mo<sub>2</sub>N, **b** Mo<sub>2</sub>N@Co/CNT and **c** Mo<sub>2</sub>N@CoFe@C/CNT samples. **d** Reflection loss curves at the same thickness of 2 mm

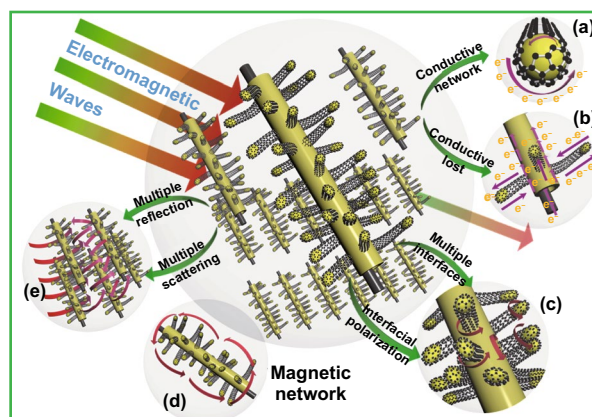
exhibit impressive MA performance with the maximum RL values all less than -10 dB, revealing its tunable MA ability. These encouraging results demonstrate that as-prepared Mo<sub>2</sub>N@CoFe@C/CNT composites hold excellent MA performance owing to its strong microwave energy absorption, broad effective absorption bandwidth, lower thickness and tunable absorption frequency, which is superior to those reported metal/carbon microwave absorbers (Table S1).

### 3.3 Analysis of Microwave Absorption Mechanism

Accordingly, the rational design of 3D hierarchical core-shell structure of Mo<sub>2</sub>N@CoFe@C/CNT absorber and the combination of dielectric Mo<sub>2</sub>N, conductive C/CNTs and magnetic CoFe alloy components contribute to the enhancement of electromagnetic storage and MA performance. Related microwave energy absorption/conversion mechanisms of MA can be illustrated as followed in detail (Fig. 7).

#### 3.3.1 Multiple Heterojunction Interfaces and Hierarchical Electronic Transportation Paths Boosted Dielectric Loss

3D assembly Mo<sub>2</sub>N@CoFe@C/CNT composites possess plentiful heterojunction interfaces, which is necessary to the improvement of dielectric storage ability and polarization behaviors. Hierarchical Mo<sub>2</sub>N@CoFe@C/CNT composite is made up of dielectric Mo<sub>2</sub>N, graphitized C/CNTs and magnetic CoFe nanoparticles. In such “tubes on rods” matrix, there are at least three kinds of heterojunction interfaces, including CoFe-CNTs interfaces, graphitized carbon-CNTs interfaces and graphitized carbon-Mo<sub>2</sub>N interfaces (Fig. 7c). Due to differences in electrical conductivity among components, free electrons gather around those contacting interfaces when applied variation of electromagnetic wave. This electronic migration/moment can produce intensive interfacial polarization and relaxation causing the conversion from electromagnetic waves energy into thermal energy. Besides, numerous carbon heteroatoms groups (such as C-N and



**Fig. 7** The microwave absorption mechanism in the 3D hierarchical  $\text{Mo}_2\text{N}@CoFe@C/CNT$  composites

C-O, Fig. 3c) in  $\text{Mo}_2\text{N}@CoFe@C/CNT$  could be regarded as active dipole sites. Related dipole polarization can also improve the MA performance. Therefore,  $\text{Mo}_2\text{N}@CoFe@C/CNT$  composite exhibits higher dielectric polarization ability compared with  $\text{Mo}_2\text{N}@Co/CNT$  and  $\text{Mo}_2\text{N}$  materials owing to its multiple interfaces and multicomponent. In addition, both dielectric  $\text{Mo}_2\text{N}$  rod and graphitized C/CNTs can be also considered as a conductive network. Micro-scale  $\text{Mo}_2\text{N}$  rod displays high permittivity. When graphitized C/CNTs grow on  $\text{Mo}_2\text{N}$  rod, numerous electronic transportation routes are formed between C/CNTs and  $\text{Mo}_2\text{N}$  rod (Fig. 7a, b). This conduction transportation network facilitates enhanced conduction loss capability, which is also favorable for MA performance.

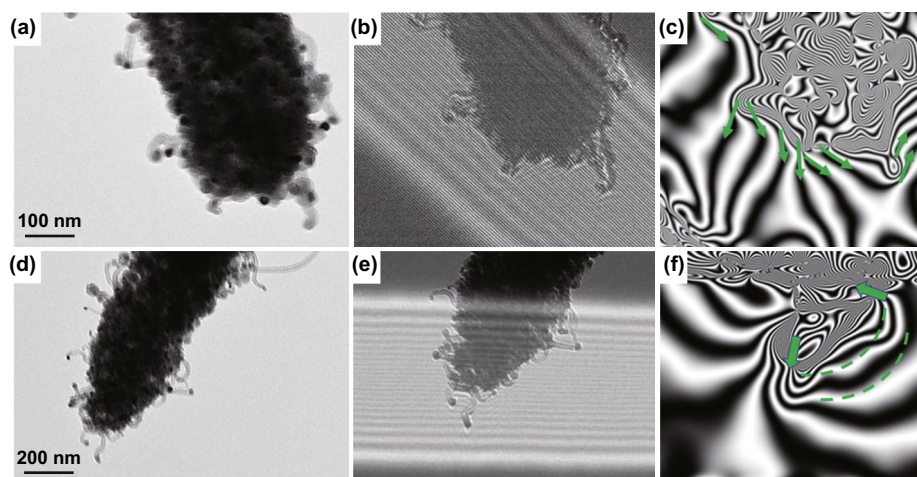
### 3.3.2 Spatial Dispersed CoFe Nanoparticles Built Multi-scale Magnetic Coupling Network

Spatial dispersed nano-scale CoFe alloy suspended within hierarchical micro-scale  $\text{Mo}_2\text{N}@C/CNTs$  rod construct a multi-scale magnetic network and could significantly contribute to the boosted magnetic responding capacity (Fig. 7d). Traditionally, magnetic nanoparticles could easily aggregate together due to their magnetic nature. Metal aggregation problem can hardly be avoided in the process of pyrolyzing MOFs directly. Herein, through our ligand exchange strategy, as-synthesized  $\text{MoO}_3@hollow-CoFe-PBA$  structure can not only effectively reduce the aggregation of magnetic nanoparticles but also expand spatial magnetic distribution, thereby further increasing the responding scale of magnetic component in the final  $\text{Mo}_2\text{N}@CoFe@C/$

CNT composite. As-fabricated hierarchical  $\text{Mo}_2\text{N}@C/CNT$  architecture provides a perfect nano/micro-matrix to support suspended CoFe nanoparticles (Figs. 4 and 5), thus forming a distributed magnetic network and strengthening magnetic permeability. The off-axis electron holography is performed to study the magnetic property of CoFe nanoparticles and related magnetic network in  $\text{Mo}_2\text{N}@CoFe@C/CNT$  composite. As shown in Fig. 8a–c, the CoFe nanoparticles in the composite can radiate out high-density magnetic lines which could penetrate through the nonmagnetic graphitic C/CNTs and expand magnetic responding regions beyond itself size. Furthermore, the neighbored CoFe nanoparticle suspended within C/CNTs matrix displays magnetic coupling lines which could contribute to integral magnetic network, further strengthen magnetic dissipation capacity (Fig. 8d–f) [49]. Meanwhile, high loading and uniformly distribution of CoFe nanoparticles (Fig. 5) can also enhance the magnetic loss to promote MA performance. Therefore, compared with  $\text{Mo}_2\text{N}@Co/CNT$  and other magnetic metal/carbon composites reported previously, hierarchical  $\text{Mo}_2\text{N}@CoFe@C/CNT$  composite can successfully avoid magnetic metal aggregation problem and exhibit remarkable magnetic loss property.

### 3.3.3 Synergic Magnetic-dielectric MA System and Multi-dimension Hierarchical Structure

Hierarchical  $\text{Mo}_2\text{N}@CoFe@C/CNT$  composites can effectively dissipate the microwave energy via dielectric dissipation and magnetic loss. The assembled composite is constructed by dielectric  $\text{Mo}_2\text{N}$  as core and spatially dispersed CoFe nanoparticles within C/CNTs as shell and



**Fig. 8** a, d TEM images and b–c, e–f corresponding off-axis electron holograms of Mo<sub>2</sub>N@CoFe@C/CNT composites

thus demonstrate significantly improved MA performance resulting from both dielectric loss and magnetic loss, compared with single Mo<sub>2</sub>N material or Mo<sub>2</sub>N@Co/CNT composite with few metal nanoparticles. Meanwhile, because of the hierarchical structure and multi-scale size, Mo<sub>2</sub>N@CoFe@C/CNT assembly possess unique multi-reflection and multi-scattering (Fig. 7e). Abundant 1D CNTs, micro-scale Mo<sub>2</sub>N rod and 3D hierarchical core–shell structure could generate effective surface area and spacing effect. When incidence microwave permeates into this 3D architecture, expected large surface areas offer many active sites to produce multiple reflection and scattering. Such repeated reflection and scattering process of incident microwave can successfully attenuate microwave energy. Benefiting from above advantages of hierarchical structure and multi-loss mechanism, as-prepared Mo<sub>2</sub>N@CoFe@C/CNT composites exhibit superior MA performance that surpass those reported metal–carbon microwave absorbents (Table S1).

#### 4 Conclusion

In conclusion, as-prepared Mo<sub>2</sub>N@CoFe@C/CNT composites exhibit superior MA performance with maximum reflection loss value of  $-53.5$  dB at the thickness of only 2 mm thickness and a broad effective absorption bandwidth of 5 GHz. Such 3D hierarchical core–shell structure assembled by nano-scale magnetic CoFe nanoparticles suspended within graphitic C/CNTs supported on micro-scale Mo<sub>2</sub>N rod is rationally constructed via our effective ligand

exchange strategy. The dielectric Mo<sub>2</sub>N and C/CNTs components can shape strong conductive loss and hierarchical core–shell structure offers large interfacial area to trigger polarization loss. Moreover, distributed magnetic CoFe nanoparticles embedded in C/CNTs matrix form multi-scale magnetic network and reinforce magnetic response capability, which is verified by the off-axis electron holography. Firmly, the MOF-based ligand exchange strategy in this work can be utilized to construct various hierarchical structure of multicomponent metal–carbon system for enhanced MA performance.

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