

2D WSe₂ Flakes for Synergistic Modulation of Grain Growth and Charge Transfer in Tin-Based Perovskite Solar Cells

Tianyue Wang, Fangyuan Zheng, Guanqi Tang, Jiupeng Cao, Peng You, Jiong Zhao, and Feng Yan*

Tin (Sn)-based perovskites with favorable optoelectronic properties and ideal bandgaps have emerged as promising alternatives to toxic lead (Pb)-based perovskites for photovoltaic applications. However, it is challenging to obtain high-quality Sn-based perovskite films by solution process. Here, liquid-exfoliated 2D transition-metal dichalcogenides (i.e., MoS₂, WS₂, and WSe₂) with smooth and defect-free surfaces are applied as growth templates for spin-coated FASnI₃ perovskite films, leading to van der Waals epitaxial growth of perovskite grains with a growth orientation along (100). The authors find that WSe₂ has better energy alignment with FASnI₃ than MoS₂ and WS₂ and results in a cascade band structure in resultant perovskite solar cells (PSCs), which can facilitate hole extraction and suppress interfacial charge recombination in the devices. The WSe₂-modified PSCs show a power conversion efficiency up to 10.47%, which is among the highest efficiency of FASnI₃-based PSCs. The appealing solution phase epitaxial growth of FASnI₃ perovskite on 2D WSe₂ flakes is expected to find broad applications in optoelectronic devices.

The inferior performance of Sn-based PSCs can be ascribed to the following reasons. One critical factor is the poor quality Sn-based perovskite films with rough surface and high-density of traps.^[10,11] To achieve high-performance Sn-based PSCs, a uniform and continuous perovskite film with large grain sizes and a preferred orientation is required. Several strategies have been developed to modulate the crystallization of Sn-based perovskites.^[12–14] For example, Meng et al. reported the introduction of pentafluorophen-oxethylammonium iodide (FOEI) into the precursor solution for surface-controlled growth of FASnI₃ perovskites.^[15] Yu et al. tuned MA_xFA_{1–x}SnI_{3–x}Br_x perovskites with preferential facet of (001) via compositional engineering.^[16] Qiu et al. introduced *n*-butylamine (BA) and phenylethylamine (PEA) as the intermediate phase suppressor to obtain uniform 2D Ruddlesden–Popper Sn-based perovskites with ordered crystal

Organic–inorganic hybrid halide perovskite solar cells (PSCs) have shown rapid progress in power conversion efficiency (PCE) in the past ten years, which presently has exceeded 25%.^[1–3] However, the top-performing PSCs are based on toxic lead halide perovskite materials, which are detrimental to the environment. Sn-based perovskites have been recognized as ideal alternatives due to their favorable optoelectronic properties and environmentally friendly characters.^[4–7] Notably, Sn-based PSCs exhibit the highest performance among the categories of Pb-free PSCs, while their performance still lags far behind that of Pb-based counterparts.^[8,9]

nucleation.^[17] More recently, the solution of *n*-propylammonium iodide (PAI) was utilized to induce the recrystallization and templated growth of FASnI₃.^[18] Despite the success in modulating the nucleation and crystallization of Sn-based perovskites, the obtained perovskite films cannot meet the requirement for high-performance PSCs due to the easy oxidation of Sn²⁺ to Sn⁴⁺ that can induce high levels of p-type doping in perovskite films. Thus reducing agents should be introduced to prohibit the oxidation of the perovskite films,^[19–21] which in turn leads to a less conductive layer unfavorable for charge transfer on grain surfaces.^[22–24]

Another important factor is the interfacial energy loss that limits the device PCE. The interface states and energy level mismatching between Sn-based perovskites and common charge transport materials can induce severe non-radiative recombination and result in low open-circuit voltage (V_{oc}) and fill factor (FF).^[25,26] To solve this problem, Jocar et al. introduced guanidinium (GA) cation in the FA/GA mixed perovskite to match its band structure with PEDOT:PSS and C₆₀.^[27] Thereafter, Nishimura et al. tuned the band structure of GeI₂-doped (FA_{1–x}EA_x)_{0.98}EDA_{0.01}SnI₃ perovskite by introducing ethylammonium cation (EA) with different amount to match the energy levels of the transport layers.^[28] Very recently, Jiang et al. substituted [6,6]-phenyl-C₆₁-butyric acid methyl ester (PCBM) with indene-C₆₀ bisadduct (ICBA) as an electron transport material, which shows a smaller energy level offset with 2D perovskite

Dr. T. Wang, F. Zheng, Dr. G. Tang, Dr. J. Cao, Dr. P. You, Dr. J. Zhao, Prof. F. Yan
Department of Applied Physics
The Hong Kong Polytechnic University
Hung Hom, Kowloon 999077, Hong Kong
E-mail: apafyan@polyu.edu.hk

 The ORCID identification number(s) for the author(s) of this article can be found under <https://doi.org/10.1002/adv.202004315>

© 2021 The Authors. Advanced Science published by Wiley-VCH GmbH. This is an open access article under the terms of the Creative Commons Attribution License, which permits use, distribution and reproduction in any medium, provided the original work is properly cited.

DOI: 10.1002/adv.202004315

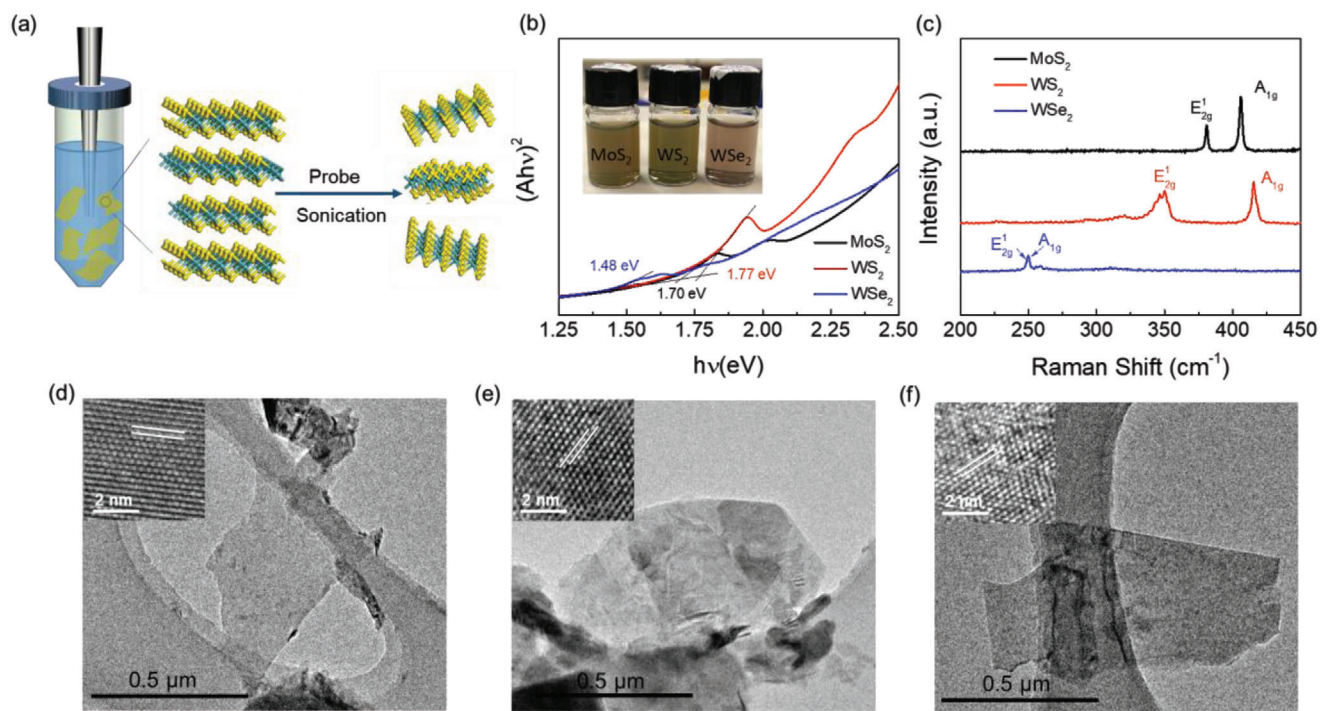


Figure 1. a) Schematic illustration for preparation of MoS₂, WS₂, and WSe₂ flakes (MX₂) via liquid phase exfoliation assisted by sonication. b) The images (shown in the inset) and Tauc plots of MX₂ flakes in IPA dispersions. The Tauc plots are derived from the absorption spectra. c) Raman spectra of MX₂ flakes spin-coated on Si wafers. TEM and HRTEM images of as-exfoliated d) MoS₂, e) WS₂, and f) WSe₂, respectively.

component due to a higher lowest unoccupied molecular orbital (LUMO) level of the latter.^[29] In addition, the energy loss in Sn-based PSCs can be reduced by interface optimization of hole transport layers (HTLs).^[30,31]

2D semiconductors, for example, transition-metal dichalcogenides, demonstrate unique optoelectronic properties such as high carrier mobilities, tunable band structures, and optical transparency,^[32,33] making them excellent interlayer materials in PSCs.^[34,35] In the present work, we introduced liquid-phase-exfoliated few-layer MoS₂, WS₂, and WSe₂ (the formula can be referred as MX₂, where M is a transition metal and X is a chalcogenide) flakes between NiO_x HTL and FASnI₃ perovskite for achieving high performance inverted PSCs. MX₂ is utilized as a growth template for preparing epitaxial Sn-based perovskite films by spin coating. Perovskite films on MX₂ exhibit (011) and (01 $\bar{1}$) oriented growth along the substrate and out-of-plane growth with preferred (100) orientation. We find that the higher valence band maximum (VBM) of WSe₂ than MoS₂ and WS₂ matches well with that of FASnI₃ perovskite, which promotes cascade hole extraction at the interface and suppresses interfacial charge recombination, and the highest PCE of 10.47% is achieved in WSe₂-incorporated devices.

Several types of MX₂ (i.e., MoS₂, WS₂, and WSe₂) flakes were prepared by liquid phase exfoliation of their bulk powders, as schematically shown in Figure 1a. In brief, 100 mg of the corresponding material powders were added to 10 ml isopropyl alcohol (IPA) in centrifuge tubes and then sonicated for 12 h in an ice bath. The dispersions were then centrifuged to take the supernatant containing exfoliated flakes. The exfoliated flake samples were characterized with ultraviolet–visible (UV–vis) and Raman

spectroscopies. Figure 1b shows the image (inset) and Tauc plots of the MX₂ flakes in IPA dispersions. The optical bandgaps of MoS₂, WS₂, and WSe₂ flakes are estimated to be ≈1.70, ≈1.77, and ≈1.48 eV, respectively, which are consistent with the values reported in literature.^[36,37] Figure 1c shows the Raman spectra of MX₂ flakes spin-coated on Si wafers. Strong characteristic peaks at around 381 and 405 cm⁻¹ for MoS₂, 350 and 415 cm⁻¹ for WS₂, and 250 cm⁻¹ for WSe₂ can be observed, which correspond to the E_{2g}¹ and A_{1g} modes of these MX₂ with a few layer thickness.^[38,39] Such MX₂ flakes are typically thin nanosheets with the thickness in the 5–7 nm range, as revealed by the transmission electron microscopy (TEM) (Figure 1d–f) and atomic force microscopy (AFM) (Figure S1, Supporting Information). The insets in Figure 1d–f show the HRTEM images of the corresponding MX₂ flakes. A clear lattice fringe of 0.27, 0.27, and 0.28 nm are observed, corresponding to the (100) plane distances of MoS₂ (PDF#65-0160), WS₂ (PDF#08-0237), and WSe₂ (PDF#38-1388), respectively.

PSCs with an inverted structure of ITO/NiO_x/MX₂/FASnI₃/PCBM/BCP/Ag were fabricated. MX₂ interlayers were deposited on the surface of NiO_x by spin-coating from their IPA solutions. To have a clear observation of film morphology, MX₂ flakes were spin-coated on flat Si substrates and observed under scanning electron microscopy (SEM). The as-deposited MX₂ retains a layered structure and the statistics of the flake sizes are obtained (see Figure S2, Supporting Information). The coverage of the flakes is about 30%. A FASnI₃ perovskite layer was then fabricated on MX₂/NiO_x using a one-step anti-solvent dripping method. Top-view SEM images of FASnI₃ films grown on NiO_x and MX₂/NiO_x are shown in

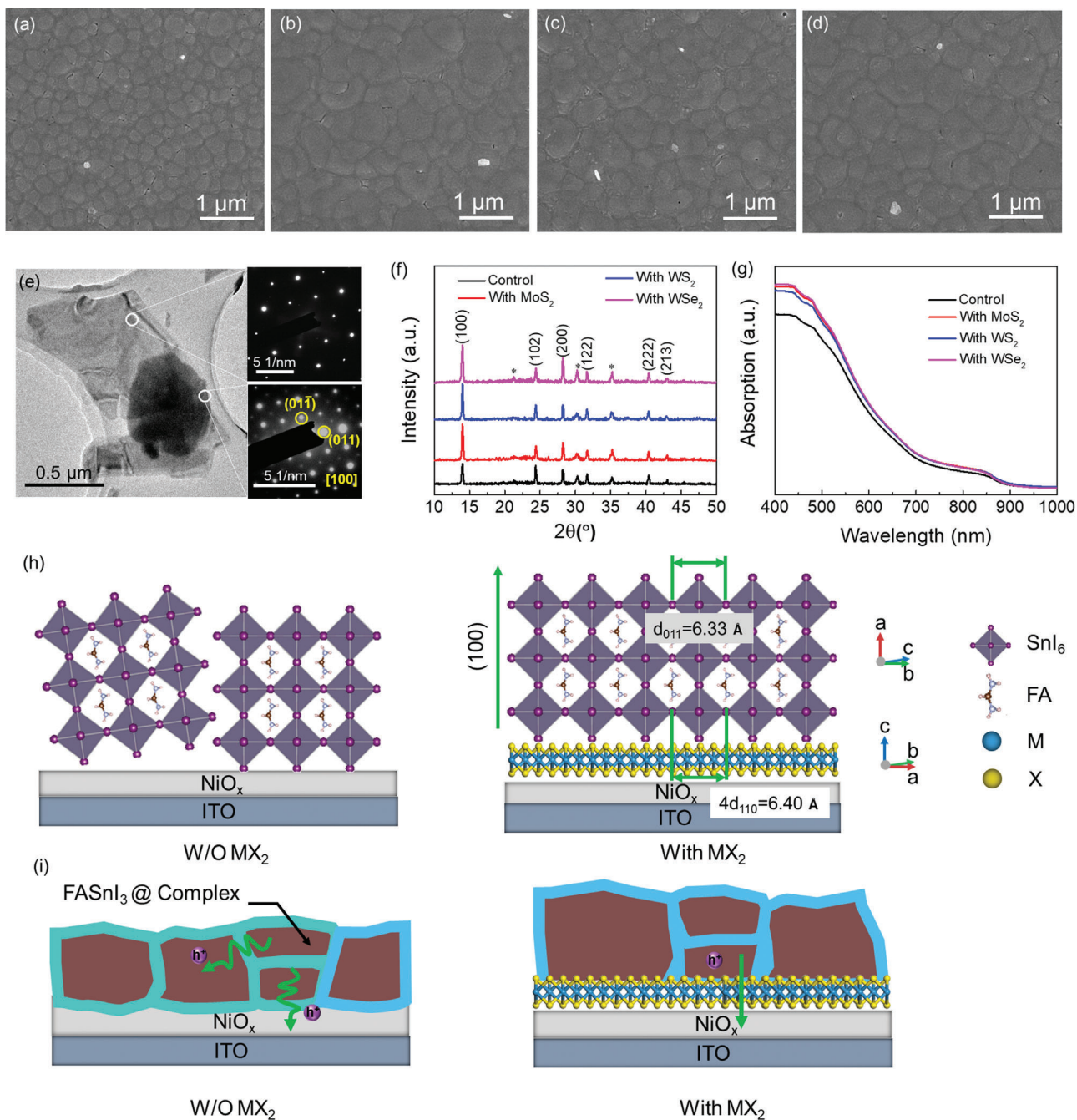


Figure 2. Top-view SEM images of FASnI₃ perovskite films fabricated on a) NiO_x, b) MoS₂/NiO_x, c) WS₂/NiO_x, and d) WSe₂/NiO_x, respectively. e) TEM image of a WSe₂ flake with a perovskite film grown on it. The SAED patterns of WSe₂ (on the right top) and FASnI₃ perovskite (on the right down) corresponding to the areas as indicated. f) XRD and g) UV-vis spectra of FASnI₃ perovskite films grown on pristine and MX₂-modified NiO_x/ITO. * Denote to the peaks of ITO. h) Schematic diagram of the growth of the FASnI₃ grain on NiO_x (left) and vdW epitaxial growth of the FASnI₃ grain on a MX₂ (WSe₂) surface (right) from the side view. i) Schematic diagram for the structures of perovskite films grown on NiO_x (left) and MX₂ (right) surfaces.

Figure 2a–d. It can be seen that the MX₂ interlayers do benefit the growth of perovskite films and lead to an enlarged grain size and less grain boundaries. FASnI₃ perovskites fabricated on MoS₂/NiO_x, WS₂/NiO_x, and WSe₂/NiO_x exhibit average grain sizes of 607, 575, and 669 nm, respectively, which are much bigger than the grain size (286 nm) of the film prepared on pristine NiO_x layer (see Figure S3, Supporting Information).

This result indicates the role of MX₂ interlayer as a template in regulating the lateral growth of FASnI₃ perovskite crystallites. It has been reported that the smooth and defect-free van der Waals (vdW) face of MX₂ can provide a growth template and promote perovskite film growth with enlarged grain sizes.^[34,40] Thus, it is reasonable to observe larger FASnI₃ perovskite grains grown on big MX₂ flakes.^[34]

To have a scrutiny of the crystallization and growth of FASnI₃ perovskite grains on the surface of MX₂, MX₂ (i.e., WSe₂) flakes with perovskite grains on their surfaces were observed under TEM. Figure 2e shows the image of a FASnI₃/WSe₂ heterojunction. The selected area electron diffraction (SAED) patterns obtained from two different parts of the heterojunction exhibit a hexagonal pattern (right top) and a square pattern (right down), respectively, corresponding to the hexagonal crystal structure of WSe₂ and the orthorhombic structure of FASnI₃, which confirms the overlying of a perovskite film on a WSe₂ flake. Two diffraction spots in the square pattern are indexed to (011) and (01 $\bar{1}$) planes of FASnI₃ perovskite, and the zone axis of [100] can be derived. Such a calibration result shows favorable agreement with the simulated SAED pattern from its crystallographic information framework (CIF) file (see Figure S4, Supporting Information). Thus, WSe₂ tends to modulate the growth of perovskite grain with a preferred (100) orientation, with (011) and (01 $\bar{1}$) planes grown along the substrate.

Perovskite films on pristine NiO_x and MX₂/NiO_x substrates were characterized by X-ray diffraction (XRD). As shown in Figure 2f, all samples exhibit similar characteristic XRD peak positions, signifying that the MX₂ interlayer did not affect the crystal structure of the upper perovskite layer. Notably, the peak intensities of (100) and (200) increase for perovskite films on MX₂/NiO_x substrates and the peak intensity ratio of (100)/(102) increases from 1.19 for NiO_x to 2.61, 2.43, and 2.68, for MoS₂, WS₂, and WSe₂ modified NiO_x, respectively. Hence, perovskite films deposited on MX₂/NiO_x have better crystallinity and oriented growth along the (100) direction.

UV–vis absorption spectra of the perovskite films on pristine NiO_x and MX₂/NiO_x substrates are shown in Figure 2g. Considerable absorption enhancement is observed for perovskite films with the incorporation of a MX₂ interlayer within the visible wavelength region. Such enhancement is directly related to the enlarged perovskite grains and better crystallinity. UV–vis absorption spectra of NiO_x substrates processed without and with MX₂ interlayer were also characterized (see Figure S5, Supporting Information). The incorporation of MX₂ does not change the optical transmittance, indicating little contribution of MX₂ to the whole absorption of the perovskite films processed on MX₂/NiO_x substrates. Moreover, the slightly higher absorption of perovskite films processed on WSe₂/NiO_x than MoS₂/NiO_x and WS₂/NiO_x is consistent with the enhanced crystallinity revealed from SEM (Figure 2a–d) and XRD (Figure 2f). Based on the above results, the crystal structures and the relative orientations of WSe₂ and FASnI₃ are depicted in Figure 2h and Figure S6, Supporting Information. The NiO_x layer composed of random round particles or clusters will not have lattice match with the perovskite and regulate its preferred orientation. WSe₂ flakes have a smooth surface with ordered atom arrangement, where the quadruple distance of the (110) planes in WSe₂ (4d₁₁₀ = 6.4 Å) is close to the lattice distance of (011) in FASnI₃ (6.33 Å) and thus perfect for the epitaxial growth of FASnI₃ on WSe₂ along the (100) direction.

Notably, Sn-based perovskite grains are encapsulated with an amorphous layer made of SnF₂, SnCl₂, and other additives, which is a general technique that has been developed specifically for Sn-based PSCs to prohibit the oxidation of the perovskite grains.^[20,23,41] Here, we fabricated FASnI₃ perovskite films with SnCl₂ and gallic acid (GA) coadditives, which enables capping

of the perovskite grains with an amorphous SnCl₂–GA complex layer as described in our previous report.^[23] However, the complex restricts charge transfer from the grains to the HTL due to its low conductivity. As shown in Figure 2i, the direct epitaxial growth of a perovskite film on WSe₂ flakes can be favorable for charge transfer from perovskite to the underlying NiO_x HTL. The perfect interface between perovskite and WSe₂ flakes will enable charge transfer without the influence of the complex layer, which is expected to result in improved device performance.

The potential of using exfoliated MX₂ flakes as an interlayer in PSCs is further evaluated. Considering that the implementation of MX₂ on NiO_x HTL can exert impacts on charge transfer between HTL/perovskite interface,^[42] we adjusted their spin-coating times and investigated the corresponding PSC performances. The device performances and PV parameters of the devices with MoS₂, WS₂, and WSe₂ spin-coated for different times can be found in Figure S7 and Table S1, Supporting Information. Figure 3a shows representative photocurrent density–voltage (J–V) curves of the three MX₂-modified devices at optimum conditions. The control device exhibits a PCE of 8.03%, a J_{sc} of 19.44 mA cm⁻², a V_{oc} of 0.59 V, and a FF of 70.0%. The incorporation of MoS₂ increases the PCE to 8.52%, as a result of the significantly enhanced J_{sc} to 21.75 mA cm⁻², although its V_{oc} is slightly decreased to 0.57 V. Replacing MoS₂ by WS₂ further increases the PCE to 8.75%, with a V_{oc} of 0.58 V, a J_{sc} of 21.44 mA cm⁻², and a FF of 70.4%. The best PCE of 9.90% is achieved for WSe₂ incorporated device, with a considerably improved V_{oc} (0.63 V), enhanced J_{sc} (21.82 mA cm⁻²), and FF (72.0%). Therefore, all MX₂ incorporated devices show improved J_{sc} than the control device, which can be attributed to enhanced light absorption of perovskites on the 2D MX₂ flakes. Notably, the J_{sc} differences among MX₂ incorporated devices are not obvious, whereas the V_{oc} variation is quite noticeable.

Ultraviolet photoelectron spectroscopy (UPS) measurements were further conducted to investigate the energy levels of the FASnI₃ layer and NiO_x film without and with MX₂ modification, as shown in Figures S8 and S9, Supporting Information. Their VBMs are calculated to be -5.07 (FASnI₃), -5.06 (NiO_x), -5.38 (MoS₂), -5.29 (WS₂), and -5.06 eV (WSe₂) according to the following equation:^[43]

$$E_{\text{VB}} = E_{\text{cutoff}} - E_{\text{F}} - h\nu \quad (h\nu = 21.22 \text{ eV}) \quad (1)$$

where E_{cutoff} represents the secondary electron cutoff value and E_{F} represents the Fermi level. The corresponding conduction band minimum (CBM) can be further derived from $E_{\text{CB}} = E_{\text{VB}} + E_{\text{g}}$, where E_{g} is obtained from the absorption edge (see Figures 1b and 2g). Thus the energy level diagram of the device is depicted in Figure 3b. MoS₂ and WS₂ have lower VBM than FASnI₃ perovskite, whereas WSe₂ shows good VBM alignment in the NiO_x/WSe₂/perovskite/PCBM multilayer structure. Thus, the incorporation of WSe₂ favors charge transfer and collection at the NiO_x/perovskite interface and has less voltage loss, which is expected to induce higher device V_{oc} than MoS₂ and WS₂.

To further investigate the effect of MX₂ interlayer on the charge dynamic process at the interface of NiO_x/perovskite, steady state photoluminescence (PL) and time-resolved photoluminescence (TRPL) were characterized on FASnI₃ films fabricated on MX₂/NiO_x/glass substrates. As shown in Figure 3c, the samples

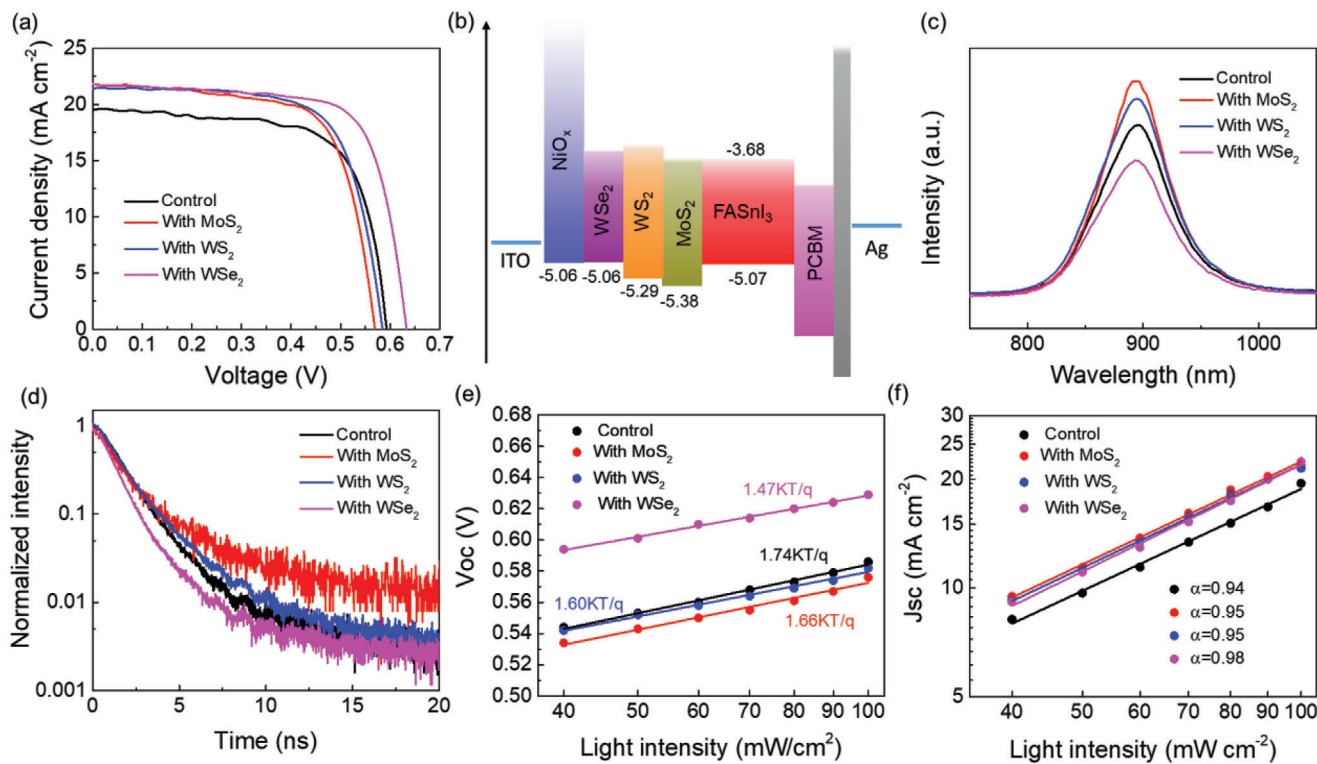


Figure 3. a) Representative J - V curves of MX_2 incorporated devices at the optimum condition. b) Energy level diagram of the PSC with incorporation of MX_2 . c) PL and d) TRPL spectra of $FASnI_3$ perovskite films fabricated on pristine NiO_x /ITO and NiO_x /ITO substrates modified with MX_2 . Light intensity dependence of e) V_{oc} and f) J_{sc} of the control and MX_2 incorporated devices.

with the incorporation of MoS_2 and WS_2 exhibit higher PL intensity than the control sample, as a result from the better film quality of the former perovskite films. However, the incorporation of WSe_2 weakens the PL intensity, which can only be accounted for by that the WSe_2 bridges the energy levels between perovskite film and NiO_x layer well and therefore boosts charge transfer across the interface. The TRPL results of the corresponding samples are shown in Figure 3d. The TRPL spectra are assumed in first order decay and fitted by a single-exponential function of $Y = A \cdot \exp(-t/\tau) + y_0$, where A and τ represent the amplitude and PL decay lifetime respectively. The $FASnI_3$ film fabricated on NiO_x exhibits a characteristic TRPL lifetime of 1.4 ns. With MoS_2 and WS_2 , the PL decay lifetimes are prolonged to 1.55 and 1.43 ns, respectively. A faster PL quenching (lifetime of 1.3 ns) is observed for the sample with WSe_2 , indicating that WSe_2 is more effective for charge transfer than the pristine NiO_x , as well as MoS_2 and WS_2 modified NiO_x layers. Therefore, WSe_2 is most favorable for the collection of photo-generated carriers by NiO_x and consequently reduces interfacial charge recombination loss.

We also conducted light intensity (P_{light}) dependent J - V measurements of the devices to examine the effect of MX_2 interlayer on charge recombination. The dependence of V_{oc} on P_{light} is plotted according to the relation: $V_{oc} \propto nkT/q \ln(P_{light})$, where n , k , T , and q are ideality factor, Boltzmann constant, temperature in K, and elementary charge, respectively. The value n reflects the carrier traps assisted recombination in the active layer or at interfaces at open circuit condition.^[44,45] As shown in Figure 3e, the

control device shows a slope of V_{oc} versus semilogarithmic P_{light} of 1.74 kT q^{-1} , which is only slightly reduced to 1.66 kT q^{-1} for an MoS_2 incorporated device and 1.60 kT q^{-1} for a WS_2 incorporated device. Notably, the lowest slope of 1.47 kT q^{-1} is observed for a WSe_2 incorporated device, indicating that the trap-assisted recombination losses are greatly suppressed than in other devices. The less trap induced recombination in a WSe_2 incorporated device benefits from its proper energy level alignment with NiO_x and $FASnI_3$, which promotes hole transfer from perovskite to HTL.

Figure 3f plots the dependence of J_{sc} on P_{light} on a log-log scale, according to the equation: $J_{sc} \propto (P_{light})^\alpha$. The exponent α is close to 1 when all carriers transfer to the electrodes without bimolecular recombination.^[46] WSe_2 -incorporated device shows an α value of 0.98, higher than 0.94 for the control device and 0.95 for both MoS_2 and WS_2 incorporated devices. The results indicate that the WSe_2 interlayer can effectively enhance the charge transportation property and reduce the bimolecular recombination.

The effect of WSe_2 interlayer on hole transport at the NiO_x /perovskite interface is further characterized by measuring the space-charge limited current using hole-only devices (ITO/ NiO_x /without or with WSe_2 /FASnI₃/P3HT/Au). It can be seen from Figure S10, Supporting Information, that the devices without and with WSe_2 incorporation show the trap filled limit voltage (V_{TFL}) of 2.97 and 2.19 V, corresponding to the trap density (N_t) of 4.69×10^{16} and $3.45 \times 10^{16} \text{ cm}^{-3}$, respectively. The lower trap density for the latter signifies that the hole trap is effectively passivated by WSe_2 at the NiO_x /perovskite interface, which

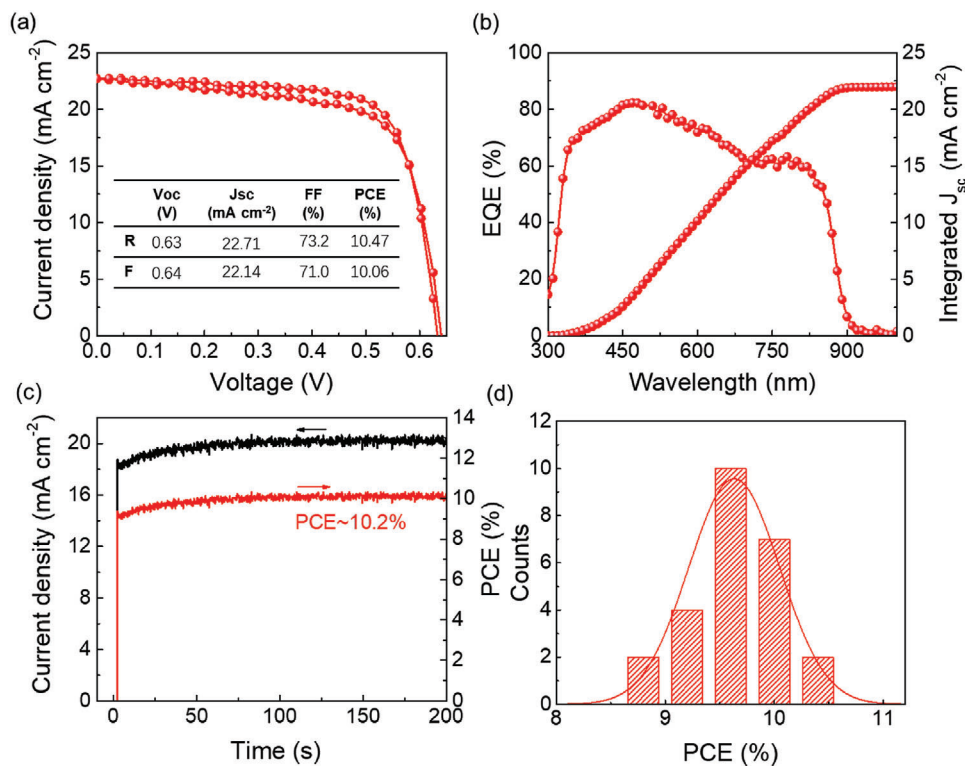


Figure 4. a) J - V curves for the champion WSe_2 incorporated device in forward and reverse scans. b) EQE and integrated J_{sc} curves for the champion PSC with incorporation of WSe_2 . c) Stable photocurrent density and PCE of a WSe_2 incorporated device measured at the maximum power point. d) The statistics of PCEs of WSe_2 incorporated PSCs.

facilitates hole transfer at such an interface with a lower trapping possibility.

Figure 4a presents the J - V curves of the best-performing WSe_2 incorporated device under AM 1.5 one-sun illumination. It displays a V_{oc} of 0.63 V, a J_{sc} of 22.71 mA cm^{-2} , an FF of 73.2% and results in a PCE of 10.47% for reverse scan; and a V_{oc} of 0.64 V, a J_{sc} of 22.14 mA cm^{-2} , an FF of 71.0%, and a PCE of 10.06% for forward scan. The negligible hysteresis proves the high quality of the perovskite film with suppressed ion migration as well as efficient interfacial charge transport. **Figure 4b** shows the external quantum efficiency (EQE) spectrum for the device. J_{sc} is integrated to be 22.3 mA cm^{-2} , which is close to the value as observed in J - V curves. The stable output of $\approx 10.2\%$ is acquired at the maximum power point, as shown in **Figure 4c**. Moreover, the histogram of the PCEs of 25 WSe_2 -incorporated devices is presented in **Figure 4d**. The narrow distribution of the PCE data further illustrates the reliability and reproducibility of our devices. Long term stability of the WSe_2 -incorporated device stored in air (relative humidity $\approx 20\%$) without encapsulation was monitored and shown in **Figure S11**, Supporting Information. It shows a high stability, retaining about 82% of the initial PCE after 1000 h.

In summary, we reported for the first time the introduction of liquid exfoliated MX_2 flakes as an interlayer between $\text{NiO}_x/\text{FASnI}_3$ perovskite to enhance the performance of inverted Sn-based PSCs. The MX_2 flakes with smooth surfaces act as a growth template that can lead to the vdW epitaxial growth of FASnI_3 perovskite grains with enlarged size and preferred orientation. More importantly, we find that the band structure of the

2D flakes is critical to the charge transfer in the resultant devices. In comparison with MoS_2 and WS_2 , WSe_2 possesses a higher VBM and is judiciously selected as an efficient charge transport interlayer due to its high hole mobility and proper energy alignment with the VBMs of HTL and perovskite film. The resultant WSe_2 incorporated PSCs exhibit increased photovoltaic parameters in J_{sc} , V_{oc} , and FF, which is due to the synergy of enhanced charge transport, lesser interfacial recombination, and stronger light absorption. Eventually, a PCE of 10.47% is obtained for the champion WSe_2 -incorporated device, which is among the highest efficiency for FASnI_3 -based PSCs (Table S2, Supporting Information). This work provides a convenient strategy to control the growth of FASnI_3 perovskite film and optimize its interface property, which is also suitable for the preparation of other Sn-based perovskite optoelectronic devices.^[47,48]

Experimental Section

Materials: Formamidinium iodide (FAI) was purchased from Great-Cell Solar. Nickel(II) nitrate hexahydrate ($\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, 98%) and SnI_2 (99.999%) were purchased from Alfa Aesar. Gallic acid (99%) and SnCl_2 (99.99%) were purchased from Aladdin. Bathocuproine (BCP) was purchased from Sigma-Aldrich. Phenyl-C71-butyric acid methyl ester (PCBM) was purchased from Nano-C.

Preparation of MX_2 Flakes: 100 mg of bulk MoS_2 , WS_2 , and WSe_2 materials were carefully grinded into fine powders and then dispersed in 10 ml isopropanol in a centrifuge tube, respectively. The dispersion was further sealed with Parafilm and sonicated with a probe sonic tip for 12 h. The

temperature of tube was kept at 5 °C under an ice bath. After that, the dispersion was centrifuged at 2000 rpm for 5 min to extract the unexfoliated materials and the supernatant containing exfoliated flakes was taken for further use.

Device Fabrication: ITO glass substrates were sequentially washed by acetone, deionized water, and isopropanol. NiO_x solution (6.5 mg mL⁻¹ in DI water, synthesis of NiO_x powder can be referred to our previous report) was spin-coated on ITO substrates at 4000 rpm for 30 s then annealed at 150 °C for 20 min. For NiO_x modification, the as-prepared MX₂ flakes in IPA solution were repeatedly spin-coated on it to achieve the desired amount. 0.8 M FASnI₃ perovskite precursor solution containing FAI (1 mmol), SnI₂ (1 mmol), SnCl₂ (0.07 mmol), and gallic acid (0.01 mmol) in mixed DMF/DMSO solution (v/v = 10:1) was spin-coated on the substrates at 5000 rpm for 30 s and chlorobenzene was dripped at 13 s after starting. The as-spun perovskite films were annealed at 70 °C for 5 min. Afterwards, PCBM (20 mg mL⁻¹ in chlorobenzene) was spin-coated at 1500 rpm for 30 s, followed by spin-coating BCP (0.5 mg mL⁻¹ in IPA) at 5000 rpm for 30 s. The devices were completed by thermal evaporation of 100 nm Ag on the BCP layer.

Characterizations: J–V curves of PSCs were measured using a Keithley 2400 source meter under the illumination of AM 1.5G, 100 mW cm⁻² solar simulator (Newport 91160, 300 W). The EQE of the PSCs was measured using an EQE system equipped with a xenon lamp (Oriel 66902), an Si detector (Oriel 76175_71580), a monochromator (Newport 66902), and a dual channel power meter (Newport 2931_C). UV–vis spectra were measured with a Perkin Elmer_UV–vis-NIR spectrometer. SEM images were characterized by a field emission scanning electron microscope (Tesla MAIA3). XRD measurements were performed on a Rigaku Smartlab Diffractometer. Ultraviolet photoelectron spectroscopy (UPS) was measured on Thermo Fisher Scientific system. Steady-state photoluminescence (PL) and time-resolved photoluminescence (TRPL) were measured using an Edinburgh FLS920 fluorescence spectrophotometer. The Raman measurements were performed on a WITTEC_Confocal Raman system. Field emission TEM was performed with a JEOL Model JEM-2100F instrument operated at 200 kV. Atomic force microscopy (AFM) images were collected using Bruker NanoScope 8.

Supporting Information

Supporting Information is available from the Wiley Online Library or from the author.

Acknowledgements

This work was financially supported by the Research Grants Council (RGC) of Hong Kong, China (Project No. PolyU 152068/18E).

Conflict of Interest

The authors declare no conflict of interest.

Data Availability Statement

The data that support the findings of this study are available from the corresponding author upon reasonable request.

Keywords

2D transition-metal dichalcogenides, charge transfer, grain growth, tin-based perovskite solar cells, WS₂

Received: November 10, 2020

Revised: February 8, 2021

Published online: March 27, 2021

- [1] H. Min, M. Kim, S.-U. Lee, H. Kim, G. Kim, K. Choi, J. H. Lee, S. I. Seok, *Science* **2019**, *366*, 749..
- [2] P. You, G. Li, G. Tang, J. Cao, F. Yan, *Energy Environ. Sci.* **2019**, *13*, 1187.
- [3] National Renewable Energy Laboratory (NREL), Best Research-Cell Efficiencies, <https://www.nrel.gov/pv/cell-efficiency.html> (accessed: October 2020).
- [4] W. F. Yang, F. Igbari, Y. H. Lou, Z. K. Wang, L. S. Liao, *Adv. Energy Mater.* **2020**, *10*, 1902584.
- [5] S. Gu, R. Lin, Q. Han, Y. Gao, H. Tan, J. Zhu, *Adv. Mater.* **2020**, *32*, 1907392.
- [6] W. Ke, M. G. Kanatzidis, *Nat. Commun.* **2019**, *10*, 965..
- [7] M. Chen, M. G. Ju, H. F. Garces, A. D. Carl, L. K. Ono, Z. Hawash, Y. Zhang, T. Shen, Y. Qi, R. L. Grimm, D. Pacifici, X. C. Zeng, Y. Zhou, N. P. Padture, *Nat. Commun.* **2019**, *10*, 16..
- [8] G. Nasti, A. Abate, *Adv. Energy Mater.* **2020**, *10*, 1902467.
- [9] Q. Tai, J. Cao, T. Wang, F. Yan, *EcoMat* **2019**, *1*, e12004.
- [10] S. J. Lee, S. S. Shin, Y. C. Kim, D. Kim, T. K. Ahn, J. H. Noh, J. Seo, S. I. Seok, *J. Am. Chem. Soc.* **2016**, *138*, 3974..
- [11] F. Hao, C. C. Stoumpos, P. Guo, N. Zhou, T. J. Marks, R. P. Chang, M. G. Kanatzidis, *J. Am. Chem. Soc.* **2015**, *137*, 11445..
- [12] P. Wang, F. Li, K. J. Jiang, Y. Zhang, H. Fan, Y. Zhang, Y. Miao, J. H. Huang, C. Gao, X. Zhou, F. Wang, L.-M. Yang, C. Zhan, Y. Song, *Adv. Sci.* **2020**, *7*, 1903047.
- [13] J. Qiu, Y. Xia, Y. Chen, W. Huang, *Adv. Sci.* **2019**, *6*, 1800793.
- [14] P. Li, H. Dong, J. Xu, J. Chen, B. Jiao, X. Hou, J. Li, Z. Wu, *ACS Energy Lett.* **2020**, *5*, 2327.
- [15] X. Meng, Y. Wang, J. Lin, X. Liu, X. He, J. Barbaud, T. Wu, T. Noda, X. Yang, L. Han, *Joule* **2020**, *4*, 902.
- [16] B. B. Yu, M. Liao, Y. Zhu, X. Zhang, Z. Du, Z. Jin, D. Liu, Y. Wang, T. Gatti, O. Ageev, Z. He, *Adv. Funct. Mater.* **2020**, *30*, 2002230.
- [17] J. Qiu, Y. Xia, Y. Zheng, W. Hui, H. Gu, W. Yuan, H. Yu, L. Chao, T. Niu, Y. Yang, X. Gao, Y. Chen, W. Huang, *ACS Energy Lett.* **2019**, *4*, 1513.
- [18] X. Liu, T. Wu, J.-Y. Chen, X. Meng, X. He, T. Noda, H. Chen, X. Yang, H. Segawa, Y. Wang, L. Han, *Energy Environ. Sci.* **2020**, *13*, 2896.
- [19] C. Wang, F. Gu, Z. Zhao, H. Rao, Y. Qiu, Z. Cai, G. Zhan, X. Li, B. Sun, X. Yu, B. Zhao, Z. Liu, Z. Bian, C. Huang, *Adv. Mater.* **2020**, *32*, 1907623.
- [20] X. Liu, Y. Wang, T. Wu, X. He, X. Meng, J. Barbaud, H. Chen, H. Segawa, X. Yang, L. Han, *Nat. Commun.* **2020**, *11*, 2678..
- [21] T. Nakamura, S. Yakumar, M. A. Truong, K. Kim, J. Liu, S. Hu, K. Otsuka, R. Hashimoto, R. Murdey, T. Sasamori, H. D. Kim, H. Ohkita, T. Handa, Y. Kanemitsu, A. Wakamiya, *Nat. Commun.* **2020**, *11*, 3008..
- [22] K. P. Marshall, M. Walker, R. I. Walton, R. A. Hatton, *Nat. Energy* **2016**, *1*, 16178.
- [23] T. Wang, Q. Tai, X. Guo, J. Cao, C.-K. Liu, N. Wang, D. Shen, Y. Zhu, C.-S. Lee, F. Yan, *ACS Energy Lett.* **2020**, *5*, 1741.
- [24] T. Wang, F. Yan, *Chem. - Asian J.* **2020**, *15*, 1524..
- [25] F. Gu, Z. Zhao, C. Wang, H. Rao, B. Zhao, Z. Liu, Z. Bian, C. Huang, *Sol. RRL* **2019**, *3*, 1900213.
- [26] W. Ke, C. C. Stoumpos, J. L. Logsdon, M. R. Wasielewski, Y. Yan, G. Fang, M. G. Kanatzidis, *J. Am. Chem. Soc.* **2016**, *138*, 14998..
- [27] E. Jokar, C. H. Chien, C. M. Tsai, A. Fathi, E. W. G. Diau, *Adv. Mater.* **2019**, *31*, 1804835.
- [28] K. Nishimura, M. A. Kamarudin, D. Hirotoni, K. Hamada, Q. Shen, S. Iikubo, T. Minemoto, K. Yoshino, S. Hayase, *Nano Energy* **2020**, *74*, 104858.
- [29] X. Jiang, F. Wang, Q. Wei, H. Li, Y. Shang, W. Zhou, C. Wang, P. Cheng, Q. Chen, L. Chen, Z. Ning, *Nat. Commun.* **2020**, *11*, 1245..
- [30] C. Ran, J. Xi, W. Gao, F. Yuan, T. Lei, B. Jiao, X. Hou, Z. Wu, *ACS Energy Lett.* **2018**, *3*, 713.
- [31] X. Liu, Y. Wang, F. Xie, X. Yang, L. Han, *ACS Energy Lett.* **2018**, *3*, 1116.

- [32] J. N. Coleman, M. Lotya, A. O'Neill, S. D. Bergin, P. J. King, U. Khan, K. Young, A. Gaucher, S. De, R. J. Smith, *Science* **2011**, *331*, 568..
- [33] K. Novoselov, A. Mishchenko, A. Carvalho, A. C. Neto, *Science* **2016**, *353*, aac9439..
- [34] G. Tang, P. You, Q. Tai, A. Yang, J. Cao, F. Zheng, Z. Zhou, J. Zhao, P. K. L. Chan, F. Yan, *Adv. Mater.* **2019**, *31*, 1807689.
- [35] J. Cao, G. Tang, P. You, T. Wang, F. Zheng, J. Zhao, F. Yan, *Adv. Funct. Mater.* **2020**, *30*, 2002358.
- [36] N. Dong, Y. Li, Y. Feng, S. Zhang, X. Zhang, C. Chang, J. Fan, L. Zhang, J. Wang, *Sci. Rep.* **2015**, *5*, 14646..
- [37] H. Zhang, J. Ji, A. A. Gonzalez, J. H. Choi, *J. Mater. Chem. C* **2017**, *5*, 11233.
- [38] H. Li, J. Wu, Z. Yin, H. Zhang, *Acc. Chem. Res.* **2014**, *47*, 1067.
- [39] G. Kakavelakis, A. E. Del Rio Castillo, V. Pellegrini, A. Ansaldo, P. Tzourmpakis, R. Brescia, M. Prato, E. Stratakis, E. Kymakis, F. Bonaccorso, *ACS Nano* **2017**, *11*, 3517..
- [40] M. Zhang, M. Ye, W. Wang, C. Ma, S. Wang, Q. Liu, T. Lian, J. Huang, Z. Lin, *Adv. Mater.* **2020**, *32*, 2000999.
- [41] Q. Tai, X. Guo, G. Tang, P. You, T. W. Ng, D. Shen, J. Cao, C. K. Liu, N. Wang, Y. Zhu, C. S. Lee, F. Yan, *Angew. Chem., Int. Ed.* **2019**, *58*, 806.
- [42] A. Capasso, F. Matteocci, L. Najafi, M. Prato, J. Buha, L. Cinà, V. Pellegrini, A. D. Carlo, F. Bonaccorso, *Adv. Energy Mater.* **2016**, *6*, 1600920.
- [43] G. Kakavelakis, I. Paradisanos, B. Paci, A. Generosi, M. Papachatzakis, T. Maksudov, L. Najafi, A. E. Del Rio Castillo, G. Kioseoglou, E. Stratakis, F. Bonaccorso, E. Kymakis, *Adv. Energy Mater.* **2018**, *8*, 1702287.
- [44] Y. Lin, B. Adilbekova, Y. Firdaus, E. Yengel, H. Faber, M. Sajjad, X. Zheng, E. Yarali, A. Seikhan, O. M. Bakr, A. El-Labban, U. Schwingschlögl, V. Tung, I. McCulloch, F. Laquai, T. D. Anthopoulos, *Adv. Mater.* **2019**, *31*, 1902965.
- [45] Y. Qin, M. A. Uddin, Y. Chen, B. Jang, K. Zhao, Z. Zheng, R. Yu, T. J. Shin, H. Y. Woo, J. Hou, *Adv. Mater.* **2016**, *28*, 9416..
- [46] C. Liu, J. Tu, X. Hu, Z. Huang, X. Meng, J. Yang, X. Duan, L. Tan, Z. Li, Y. Chen, *Adv. Funct. Mater.* **2019**, *29*, 1808059.
- [47] C. K. Liu, Q. Tai, N. Wang, G. Tang, H. L. Loi, F. Yan, *Adv. Sci.* **2019**, *6*, 1900751.
- [48] C. Xie, C. K. Liu, H. L. Loi, F. Yan, *Adv. Funct. Mater.* **2020**, *30*, 1903907.