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Comparative study of the interaction of ivermectin with proteins of interest associated with SARS-CoV-2: A computational and biophysical approach

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ABSTRACT

The SARS-CoV-2 pandemic has accelerated the study of existing drugs. The mixture of homologs called ivermectin (avermectin-B1a [HB1a] + avermectin-B1b [HB1b]) has shown antiviral activity against SARS-CoV-2 *in vitro*. However, there are few reports on the behavior of each homolog. We investigated the interaction of each homolog with promising targets of interest associated with SARS-CoV-2 infection from a biophysical and computational-chemistry perspective using docking and molecular dynamics. We observed a differential behavior for each homolog, with an affinity of HB1b for viral structures, and of HB1a for host structures considered. The induced disturbances were differential and influenced by the hydrophobicity of each homolog and of the binding pockets. We present the first comparative analysis of the potential theoretical inhibitory effect of both avermectins on biomolecules associated with COVID-19, and suggest that ivermectin through its homolog, has a multiobjective behavior.

1. Introduction

SARS-CoV-2 is a novel virus belonging to the β -Coronavirus genus of the 2B group of the Coronaviridae family. This interesting virus contains only 29 proteins, 26 of which have been successfully expressed for *in vitro* studies to determine targets of interest for drug discovery [1]. For example, the conserved cysteine protease M^{pro} has been highlighted as an exciting target as it mediates the maturation cleavage of polyproteins during virus replication [2,3].

Despite great successes in the production and roll-out of vaccines

against SARS-CoV-2, new variants are on the rise and there is still no globally accepted treatment for COVID-19 (https://www.who.int/publications/i/item/WHO-2019-nCoV-clinical-2021-1). During the last year many laboratories focused on screening FDA-approved drugs for quick implementation in clincal settings [4–6]. A compound of interest from such work is the racemic mixture ivermectin, typically used to treat helminth infections. Most of the studies on the macrocyclic lactone ivermectin only consider the major constituent B1a in their dockings [7]. However, ivermectin is an approximately 80:20 mixture of two homolog derivatives of the compound avermectin B1, called 22,23-

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dihydroavermectin B1a (HB1a) and B1b (HB1b) correspondingly, which differ in the presence of a sec-butyl and isopropyl group, at the C25 position, respectively (Fig. 1). Interestingly, this mixture has demonstrated *in vitro* antiviral activity against several single-stranded RNA viruses, such as Zika virus, dengue virus, Chikungunya virus, avian influenza A virus, Porcine Reproductive and Respiratory Syndrome virus, human immunodeficiency virus type 1, among others, including SARS-CoV-2 [8].

To date, the most accepted mechanism of action for ivermectin against SARS-CoV-2 is inhibition of the nuclear import of viral proteins and RNA, as has been found for HIV-1 and dengue [9]. However, it has also been reported that ivermectin has a potentially inhibitory effect against other viruses such as flaviviruses by blocking the NS3 helicase [10]. Additionally, it has been reported that it can dock in a thermodynamically favorable manner, and with theoretical potential to inhibit other structural and functional proteins associated with SARS-CoV-2 [11]. This diversity of possible targets for ivermectin and the fact that there are very few comparative reports in the SARS-CoV-2 on the behavior of homologs, represents an area of opportunity to contribute with theoretical studies that provide data applicable to the design of drugs and targeted at this new viral strain of interest in global public health.

Therefore, a more in-depth and exhaustive study is necessary on the possible mechanisms of inhibition of the two homologs present in ivermectin (HB1a and HB1b), against biomolecules of interest associated with COVID-19 especially approached from a computational biophysics and chemistry perspective. Here we describe, a rigorous comparative analysis of blind docking and ligand-protein interactions applying various genetic sampling methods, with theoretical calculations of potential inhibitory kinetic activity and analysis of molecular dynamics under controlled environments to study stability and thermodynamic fluctuations of the systems. In addition, we present the first preliminary report of the possible impact of molecular crowding on the activity of avermectin homologs against the structure of the SARS-CoV-2 cysteine protease M^{pro}.

2. Materials and methods

2.1. Databases and structure selection

As the nuclear import for macromolecules is facilitated by importins, the structures of importin α 1 subunit (PDB: 5KLR) from *Mus musculus* and importin β 1 subunit (PDB: 2P8Q) from *Homo sapiens* were used as a model for the members of the nuclear import superfamily. The host nuclear import system can be bound and sequestered by pathogens such as SARS-CoV-2 allowing transportation of viral proteins to the host nucleus leading to increased viral replication [12–15]. Additionally, we also consider the multi-functional helicase (nsp13) of SARS-CoV-2

responsible for viral replication (PDB: 6ZSL) [16-18], and the main protease (Mpro) of SARS-CoV-2 (PDB: 6LU7) as it is a key enzyme of coronaviruses and has a fundamental role in mediating viral replication and transcription, making it an attractive target for drugs [11,19-22]. All structures were obtained in PDB format from the RCSB protein database (https://www.rcsb.org/). The homologs structures of HB1a (CID 6321424) and HB1b (CID 6321425) that make up ivermectin were obtained from PubChem (https://pubchem.ncbi.nlm.nih.gov/) in SDF format, and the SMILES online converter was used (https://cactus.nci. nih.gov/translate/) to get a PDB format. To avoid confusion throughout the study, we only refer to ivermectin when mentioning the mixture of homologs (a mixture of approximately 80:20 of the two homologs derivatives of the compound avermectin B1, called 22,23-dihydroavermectin B1a and B1b), while for the individual study of each homolog, the type of avermectin analyzed (HB1a or HB1b) was always indicated. For a comparison between the ADME profiles of the avermectin homologs, the PubChem databases (https://pubchem.ncbi.nlm. nih.gov/) were used; SwissADME (www.swissadme.ch/) and the Molinspiration Property Engine v2018.10 (https://www.molinspirat ion.com/).

2.2. Comparative molecular docking

We performed a comparative analysis using five popular molecular docking models for a rigorous prediction of the standard free energy (ΔG) of binding of ligand-protein complexes. The complexes were built in the programs MTiAutoDock (https://mobyle.rpbs.univ-paris-diderot. fr/cgi-bin/portal.py#forms::MTiAutoDock), webina (https://durrantl ab.pitt.edu/webina/), DINC 2.0 (http://dinc.kavrakilab.org/), COACH-D (https://yanglab.nankai.edu.cn/COACH-D/) and DockThor (https://dockthor.lncc.br/v2/) were used. They all represent some of the basic, improved, and more advanced versions of molecular docking associated with the efficient AutoDock algorithm either in the sampling stage, blind docking or in the scoring function. To increase accuracy, a minimum of 10 runs per program were performed which implied approximately 10⁶ evaluations per run. The rest of the parameters were considered by default. Additionally, to validate the docking results, the Pose&Rank server (https://modbase.compbio.ucsf.edu/poseandrank/) was used to score the protein-ligand complexes, using the statistical scoring function dependent on the atomic distance RankScore. As usual, all the water molecules were removed and the PDB files were separated into two different files, one containing the protein and the other containing the ligand structure. Only the three runs with the most favorable berth were considered in the sampling of the probabilistically most feasible and thermodynamically most favorable positions in the complexes. This criterion was used to discriminate the complexes that would be subjected to further analysis, including potential theoretical inhibition and molecular dynamics.



Fig. 1. Molecular structure of the two constituents of ivermectin considered in this study. A) vermectin B1a (HB1a), and B) Avermectin B1b (HB1b). The differential chemical group of each structure is indicated within a circle, For HB1a it is sec-butyl and HB1b is an isopropyl.

2.3. Comparative analysis of the hydrophobic characteristics of the binding sites

We use PockDrug-Server that predicts the drug delivery capacity in the pockets considering the Kyte-Doolittle Pocket Hydrophobicity Scale [23]. The hydrophobic characteristics of the binding sites were also analyzed using Biovia Discovery Studio 2021 [24].

2.4. Comparative analysis of theoretical Inhibition

The binding constant *K* from the binding free energy was calculated as described [11,25] from the following equation:

$$K = e^{-(\Delta G/RT)} \tag{1}$$

And the inhibition constant for binding of ligand to proteins (K_i) (in units of M) was obtained as,

$$K_i = K^{-1} = e^{(\Delta} G/RT) \tag{2}$$

where, ΔG is binding affinity (kcal/mol), *R* is the universal gas constant (1.987 cal/K mol), T is the absolute temperature (298.15 K). The equation states that the higher the K_i value, the weaker the binding of the inhibitor to the protein, and, therefore, the protein-inhibitor complex dissociates more easily [26-28]. Additionally, IC₅₀ values were determined using the Dixon Plot, assuming competitive inhibition, as suggested [28]. The Dixon plot provides a good estimate of IC₅₀ values for competitive inhibition of a given substrate. IC₅₀ values were reported by placing the substrate concentration in the same units as IC_{50} in a subscript in parentheses, as suggested for theoretical predictions. For example, $IC_{50}=5\,\mu M_{(10)}$ would indicate an IC_{50} of 5 μM determined at a substrate concentration of 10 µM [29]. In this study, a hypothetical Substrate-Inhibitor relationship (1:1) was considered to avoid preferential unions favored by concentration, assuming structural similarities between homologs and substrate. Ten concentrations of each avermectin were simulated under a gradient from 0.5 to 5 µM to stimate the inhibitory potential based on plasma concentrations considered safe for ivermectin [30]. A theoretical inhibition potential was also predicted by using the IC_{50} -to- K_i web tool, following the considerations in the case of competitive inhibition with the equation,

$$IC_{50} = P_{50} = K_I (PL_{50} / [L_{50}])$$
(3)

where P_{50} , is the protein concentration at 50% inhibition; K_I is the affinity constant of the ligand to the protein; $[PL_{50}]$, is the protein concentration by ligand concentration at 50% inhibition; and $[L_{50}]$ [L_{50}], is the ligand and concentration at 50% inhibition. The IC₅₀-to-K_i web tool was used for all these calculations (https://bioinfo-abcc.ncifcrf.gov/IC50 Ki Converter/index.php) [11,27].

2.5. Molecular Dynamics (MD) simulations

Simulations for a docking hit were performed for two purposes: 1) to study the relative stability of the ligand residing in the binding pocket; and 2) sampling of the minimum energy conformations to calculate the perturbation of the thermodynamic and structural stability of the complexes. For a protein-ligand complex, the MD system relaxed first through a series of minimization procedures that includes three phases: relaxation, balance and sampling, as recommended [11,31,32]. The MD simulation of the crystal structures was carried out in an explicit water system. Specifically, the solvation of the system was carried out in a solvation box of 8.0 Å. Our MD system also consisted of one copy of each protein system and one copy of the docking ligand. An Amber99SB-ILDN force field was applied to the complex, with TIP3P water model. The whole system was finally neutralized. Water molecules were treated as rigid bodies in all models, allowing a simulation time interval of 2 fs. Periodic boundary conditions were applied, and Berendsen's algorithm for temperature and pressure docking was adopted. After a first steepest descent to 5000 steps and conjugated gradient to 5000 steps energy minimizations with positional restraints on the solute, an initial 100 ps simulation was carried out with the positions of the solute atoms restrained by a force constant of 10 kcal/(mol Å²) to let the water diffuse around the molecule and for equilibration. The method PME was used to calculate the electrostatic contribution to nonbonded interactions with a cutoff of 14.0 Å and a time step of 1 fs. The cutoff distance of the van der Waals interaction was 14.0 Å. After this equilibration run, the NVT (particle numbers, volume, temperature) production run at 300 K was performed with the cell size remaining the same. The SHAKE algorithm was applied to the system, and the time step was set to 2 fs. Ten structures were obtained every 10 ns as target structures extracted from a 100 ns trajectory. For the Root Mean Square Deviations (RMSD) calculations, the equation,

$$RMSD = \sqrt{\frac{1}{n} \sum_{i=1}^{n} \delta_i^2}$$
(4)

where δ_i is the distance between atom i and either a reference structure or the mean position of the n equivalent atoms. All MD simulations and additional adjustments were performed using COSGENE/myPresto [11,31,32].

Additionally, to predict conformational changes in each protein after its interaction with the ligands, the NMSim software (https://cpclab. uni-duesseldorf.de/nmsim/main.php) was used, which is a computational technique that uses a three-step including coarse-graining (CG), normal mode analysis (NMA), and elastic network model (ENM) to provide realistic conformations in reasonable simulation time [33]. For this, the minimum energy structure obtained with mypresto at 100 ns was used to also calculate the root-mean-square deviation (RMSD) and root-mean-square fluctuation (RMSF), using the radius-guided movements (ROG) approach of the NMSim Server [33]. The ROG-guided NMSim simulation is a method to search for a conformation of a given ligand-protein complex, and allows to describe the compactness of a protein. In an ROG-guided NMSim simulation, the trajectory is tailored towards the bound structure by selecting the pathway that leads to a decrease in Radius of gyration (Rg), and conformations are generated by structure distortion along directions of random linear combinations of low-frequency normal modes. For more detailed information on the theory of the method, see [33].

The NMSim web server enables geometric simulations based on NMA approach to explore biologically relevant conformational transitions in proteins, resulting in a trajectory of sampled protein conformations of good stereochemical quality. The NMSim simulation was configured in the "mode 1" to guide the simulation towards the lowest ROG. This allows proteins to be moved without any bias through the conformational space, and to generate the candidate conformation that has the lowest ROG. Therefore, any change in terms of unfolding must be significant. For more detailed information on the individual steps in the NMSim web server, see [33].

Binding affinities based on structural signatures and the force constant relative to the initial frame were also predicted during the 100 ns MD simulation of each complex. Plots of simulated complexes are presented using the CSM-Lig algorithm (http://biosig.unimelb.edu.au/csm _lig/) [34] designed to predict the binding affinity of a protein-small molecule complex based on structural signatures, and WebPSN algorithm (http://webpsn.hpc.unimo.it/wpsn.php) for the force constant using two alternative versions of ENM to calculate the cross-correlation of the motion of the C α atoms and for the pairwise interactions between the atoms of C α (linear cutoff-ENM and Kovacs-ENM) [35]. A widely used Molecular Mechanics/Poisson-Boltzmann surface area (MM/PBSA) [36–41] was considered as a thermodynamic integration method, to requalify the complexes, by calculating the free energy binding (ΔG_{bind}) of the MD trajectories. ΔG_{bind} in a solvent medium was calculated as follows:

$$\Delta G_{\text{bind}} = G_{\text{complex}} - (G_{\text{protein}} + G_{\text{ligand}}), \tag{5}$$

where G_{complex}, G_{protein} and G_{ligand} is the Gibbs free-energy of the ligandprotein complex, protein and the ligand in the presence of the solvent, respectively. The Gibbs free-energy for complex, protein, and ligands was calculated with following equation:

$$G = \langle E_{MM} \rangle - \langle G_{solv.} \rangle, \tag{6}$$

where G comprises the potential energy (E_{MM}) in vacuum and solvation free energy (G_{solv}) for each. E_{MM} consists of bonded and non-bonded interactions, whereas G_{solv} is the sum of electrostatic and non-polar solvation free energies. The dielectric constant of the biomolecular region (ε_n) was assigned as 2 and the dielectric constant of the solvent region (ε_s) was assigned as 80. The Size Modified Poisson-Boltzmann equation (SMPBE) (http://smpbs.math.uwm.edu/index smpbs.php) and APBS program (https://server.poissonboltzmann.org/) were used to calculate the energetic components and to predict electrostatic solvation and free binding energies [38]. The ΔG_{bind} of the complexes was calculated on frames taken at a 10 ns interval during the 100 ns production run. Molegro Molecular Viewer, version 7.0 was used (MMV_7.0).

2.6. Conformational fluctuations of ligand-protein complexes

The MD-based small-angle and wide-angle X-ray scattering curves (SWAXS) of all explicit solvent atoms were calculated on the WAXSiS server (http://waxsis.uni-goettingen.de) to predict the radius of gyration (Rg), default parameters were also used [42]. We used the HullRad (http://52.14.70.9/index_test.html) method to predict the Rg values, this method uses a convex hull model to estimate of manner fast and exact the Rg of the molecule [43]. To evaluate the conformational quality of each structure, the ProSA algorithm (https://prosa.services. came.sbg.ac.at/prosa.php) was used with which the Z-Score of a specific model was calculated and correlated with this score with the calculated scores of all publicly available structures on PDB websiten. In ProSA the minus negative Z-Score values correspond to conformations closer to native structures [4].

3. Results and discussion

3.1. Molecular Docking of HB1a and HB1b

Most of the computational studies aimed at predicting the best molecular docking for SARS-CoV-2 proteins have focused on the use of algorithms based on AutoDock Vina (ADV). ADV has an improved scoring function of the knowledge-based AutoDock (AD) method that uses a variant of X-Score, with an adjustment using PDBbind, and also

Table 1

Comparative analysis of the molecular docking of each homolog (HB1a and HB1b) obtained with each of the proteins using the various sampling methods and the RankScore Statistical Scoring Function.

Target	ΔG (kcal/mol)											
	AutoDock		AutoDock Vina (ADV)		ADV Incremental		COACH + ADV		DockThor		RankScore	
	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b
ΙΜΡα1	-8.7	-8.7	-3.4 -9.0^{g} -6.9^{h}	-1.5	-6.1	-5.1	-3.4	-1.5	-8.9	-8.7	-31.2	-29.4
IMPβ1 Helicase	-8.5 - 8.7 -5.1 ^a	-8.7 -8.7	-4.4 -8.3 -8.1 ⁱ -7.6 ^j -5.8 ^k	-2.5 -8.6	-5.7 -8.3	-5.6 -8.6	-5.1 -3.8	-5.2 -3.8	-8.4 -9.1	-8.9 -10.2	-32.0 -40.4	-26.4 -46.6
M ^{pro}	$-8.7 \\ -9.3^{\rm b} \\ -8.4^{\rm c} \\ -7.7^{\rm d} \\ -5.6^{\rm e}$	- 8.7 -8.3 ^f	$-8.6 \\ -8.3^{1} \\ -8.0^{m} \\ -7.2^{n} \\ -6.8^{\circ} \\ -6.5^{p} \\ -5.6^{q}$	-8.7 -8.2 [⊤] -9.4 ^₅	-8.6	-9.0	-6.5	-5.3	-9.6 -9.3 ^t	-10.2	-24.6	-31.0

Binding energy values reported using complexes and similar algorithms are shown. These are indicated in superscript against the values in this study highlighted in black. HB1a, avermectin B1a; HB1b, avermectin B1b.

^a [46].

^b [47].

^c [46].

^d [48].

^e [54].

^f [47].

^g [4].

^h [46].

ⁱ [46].

^j [4].

^k [49].

¹ [50–52].

^m [4].

ⁿ [53].

° [46].

^p [49].

^q [54].

^r [50].

^s [51].

^t [56].

considering the inter- and intramolecular contributions, its sampling technique is based on the Iterative Local Search global optimizer, in its Broyden - Fletcher - Goldfarb - Shanno variant (BFGS) [4,44–54]. The use of the DockThor server was further proposed, which implements a grid-based method that employs a steady-state genetic algorithm for multiple solutions as a search engine and the MMFF94S force field as the scoring function for pose evaluation. This scoring function is obtained by linear regression and more sophisticated machine learning techniques for nonlinear problems using the refined set PDBbind [55]. DockThor has demonstrated a higher success rate than exhibited by AD and ADV-based methods [55,56].

A marked energetic differential docking was observed between each homolog with each of the tested structures. All methods predicted thermodynamically favorable docking between each of the homologs and the proteins considered in this study (Table 1). Specifically, and as predicted by highly efficient and discriminatory algorithms used in the literature such as Webina [57], DINC [58,59], COACH-D [60] and DockThor [55,56,61], the existing homolog in the smallest proportion, HB1b, was the compound with the most thermodynamically favorable docking compared to M^{pro} followed by the helicase, while the thermodynamic union of the majority HB1a was more oriented to Importins (α 1 and β 1). Ivermectin has been reported to exert its antiviral effect by interfering with nuclear transport after binding with IMP α , affecting the recognition of important substrates, as well as binding to IMP β [46].

The used genetic algorithm predicted a docking of the HB1b with the M^{pro} of -10.2 kcal/mol, and with B1a of -9.6 kcal/mol, as for the helicase the free binding energies were - 10.2 kcal/mol and - 9.1 kcal/ mol for HB1b and HB1a, respectively (Table 1). These results represented the highest free binding energies predicted under the conditions of this study, and this trend was kept constant between all AD and ADVbased methods without significant difference (p > 0.001). The only difference in this trend was predicted with COACH-D for M^{pro} and IMP β 1, with a value \leq 0.5 kcal/mol. To validate the reproducibility of these results, the best poses were reclassified with PoseScore, especially since the efficient methods used present statistically different magnitudes (p < 0.001). We observe that after reclassification, the same trend predicted above in relation to the affinity of HB1a and HB1b was maintained (Table 1). These results show a greater theoretical individual affinity of the HB1b for M^{pro} and Helicase, and of the HB1a for Importins (see Table 1).

This corresponds to existing data reported between these structures and HB1a using ADV [4,46]. This result could perhaps be related to the proportion of the HB1a in the ivermectin mixture and to the described mechanism of action of inhibition of import [9]. The binding energy shown by HB1a and HB1b towards M^{pro} , which in this study exceeds that predicted for Importins by up to ~5 kcal/mol depending on the type of sampling method, allows us to infer the possibility of a more probable docking of this mixture of homologs against M^{pro} from a statisticalthermodynamic point of view (between -8 kcal/mol and -10 kcal/ mol).

The internal docking control used in this study for M^{pro} , the alphaketoamide type inhibitor 13b [62] presented a less favorable mean binding energy (~7 kcal/mol) compared to ivermectin under conditions of our study (not shown). This trend of a more favorable binding energy (specifically HB1a) for M^{pro} , followed by IMP α 1, has already been reported using ADV [4,46]. This corresponds to our previous report that shows that the constituent homologs of ivermectin present thermodynamically characteristic and differential dockings against various proteins associated with SARS-Cov-2 including M^{pro} [11]. However, to date there are few studies on ivermectin that consider energetic discrimination between its homologs, some show a more favorable union towards these proteases in the same way but with very narrow energies between both compounds, represented by a difference of ≈ 0.1 kcal/mol using ADV [50], and others studies if they reflect a more favorable energy for the HB1b [51].

In addition, other authors have reported similar binding energies

between ivermectin and M^{pro} using the genetic algorithm tested here but in the same way without considering the discrimination between HB1a and HB1b [46,56] as has already been done [32]. Although the reproducibility of the results as well as the chemical structures offered allow us to suppose that most of the cited studies used the HB1a as a model [63]. This may be because this homolog is the compound with the highest concentration in ivermectin [64–69]. In this sense, this study provides a broader view of the interaction and individual kinetic activity of each homolog, including the compound present in a lower proportion (HB1b).

Table 2 presents a perspective on the ADME profiles of the homologs considered in this study that includes experimental data associated with the plasma half-life of the homologs in healthy volunteers, cell models and animals. Both compounds have similar ADME profiles, and exhibit the same number of Lipinski law violations (MW > 500, HBAC >10). Although compound HB1b has a slightly higher hydrophobicity than HB1a, most of the experimental studies focus on the majority compound HB1a especially because the biological activities *in vivo* of the two homologs are similar [71].

3.2. Comparative study of molecular docking and interactions of the HB1a and HB1b in the cavity of proteins

The interactions of the homologs with the residues in the binding pockets are presented in Table 3. Comparison of the hydrophobic characteristics of the binding pockets according to the Kyte-Doolittle pocket hydrophobicity scale showed the most hydrophobic binding pocket in M^{pro} (-0.01), followed by IMP β 1 (-1.30), Helicase (-1.63) and IMP α 1 (-2.59) (Supplementary Table 1). It was found that to IMP α 1, 87.5% (14/16) of the interactions of HB1a were hydrophobic and 12.5% (2/16) were of the hydrogen-bond type, while the interactions established by HB1b were distributed in 80% (16/20) and 20% (4/20) for hydrophobic and hydrogen-bond interactions, respectively. The residues with the greatest contribution to the binding of HB1a and HB1b with IMPa1 in terms of hydrophobic interactions and hydrogen bonds were N-188 and W-184; as well as S-149 and N-228, respectively (see Table 3). The main chemical difference of these homologs is given by the presence of a sec-butyl and isopropyl group, in the C25 position, in the HB1a and HB1b, respectively (Table 3 and Figs. 1 and 2). It was observed that the differential chemical group (sec-butyl

Table	2
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Comparison between the ADME profiles of the avermectin homologs considered.

Property	HB1a (ID_6321424) ^h	HB1b (ID_6321425)
MW ^a	875.1	861.1
HBDC ^a	3	3
HBAC ^a	14	14
RBC ^a	8	7
ESOL ^b	-8.73	-8.49
iLOGP ^b	5.74	6.44
VOL ^c	832.02	815.22
VOLg ^c	66.77	49.97
PHL	3-5 ^{d,h} , 3-12 ^e , 12-28 ^f , 18 ^{d,h} , 25-80 ^{g,h}	25-80 ^{g,h}

MW, Molecular Weight (g/mol); HBDC, Hydrogen Bond Donor Count; HBAC, Hydrogen Bond Acceptor Count; RBC, Rotatable Bond Count; ESOL, Water Solubility Log S; iLOGP, Lipophilicity (Log PO/W); VOL, Molecular volume based on group contributions (A³); VOLg, Molecular volume of group contribution (A³); PHL, Plasma half-life of Homolog in hours.

^a PubChem (https://pubchem.ncbi.nlm.nih.gov/).

^b SwissADME (www.swissadme.ch/).

^c Molinspiration property engine v2018.10 (https://www.molinspiration.com/).

^e [71].

^g [73].

^d [70].

^f [72].

^h No specification.

Table 3

Interactions of the HB1a and HB1b with residues in the binding pockets of each protein considered.

Target	Interactions	Chemical group interactions ^a
IMPα1	$\begin{array}{l} \textit{HB1a: N-188^{HB}, W-184^{HB}, N-228^{H},} \\ \textit{E-266^{H}, R-227^{H}, W-184^{H}, W-231^{H},} \\ \textit{N-188^{H}, N-146^{H}, D-270^{H},} \\ \textit{S-149^{H}, A-269^{H}, L-307^{H}, T-311^{H},} \\ \textit{P-308^{H}, W-273^{H}} \\ \textit{HB1b: S-149^{HB}, N-228^{HB}, R-238^{HB}, E-266^{HB}, S-149^{HB}, N-228^{HB}, R-238^{H}, G-191^{H}, R-238^{H}, G-150^{H}, D-192^{H},} \\ \textit{G-191^{H}, R-238^{H}, G-150^{H}, D-192^{H},} \\ \textit{W-231^{H}, N-188^{H}, E-266^{H}, G-187^{H},} \\ \textit{A-148^{H}, G-224^{H}, N-228^{H}, R-227^{H},} \\ \textit{W-184^{H}}. \end{array}$	HB1a (sec-butyl): E-266 ^H , R- 227 ^H HB1b (isopropyl): S—159 ^H , N—188 ^H .
ΙΜΡβ1	HB1a: K-62 ^{HB} , K-60 ^H , M-61 ^H , K-62 ^H , R-53 ^H , L-64 ^H , I-54 ^H , D-63 ^H , R-52 ^H HB1b: K-62 ^{HB} , K-60 ^H , M-61 ^H , K-62 ^H , R-53 ^H , L-64 ^H , I-54 ^H , D-63 ^H , R-52 ^H	HB1a (sec-butyl): $D-63^{H}$, $K-62^{H}$, $L-64^{H}$, $I-54^{H}$, $M-61^{H}$, $R-52^{H}$ HB1b (isopropyl): $M-61^{H}$, $L-64^{H}$, $D-63^{H}$, $K-62^{H}$, $I-54^{H}$, $ABG-52^{H}$
Helicase	$ \begin{array}{l} B & 5.289^{\rm HB}, A-316^{\rm HB}, H-290^{\rm HB}, E-261^{\rm HB}, R-442^{\rm HB}, E-319^{\rm H}, L-317^{\rm H}, A-316^{\rm H}, S-289^{\rm H}, R-443^{\rm H}, E-540^{\rm H}, G-285^{\rm H}, G-288^{\rm H}, S-264^{\rm H}, D-260^{\rm H}, R-442^{\rm H}, H-290^{\rm H}, F-262^{\rm H}, G-287^{\rm H}, T-286^{\rm H}, E-261^{\rm H}, K-323^{\rm H}, Y-324^{\rm H}, K-320^{\rm H}\\ \\ HB1b: E-261^{\rm HB}, S-289^{\rm HB}, A-316^{\rm HB}, T-286^{\rm H}, G-285^{\rm H}, G-538^{\rm H}, G-287^{\rm H}, R-433^{\rm H}, E-540^{\rm H}, E-319^{\rm H}, S-289^{\rm H}, L-317^{\rm H}, H-290^{\rm H}, R-316^{\rm H}, S-264^{\rm H}, D-260^{\rm H}, R-442^{\rm H}, F-262^{\rm H}, S-264^{\rm H}, D-260^{\rm H}, R-442^{\rm H}, F-262^{\rm H}, Z-264^{\rm H}, D-260^{\rm H}, R-442^{\rm H}, R-264^{\rm H}, Z-264^{\rm H}, D-260^{\rm H}, R-442^{\rm H}, R-264^{\rm H}, Z-264^{\rm H}, Z-2$	НВ1a (sec-butyl): K—323 ^H , K—320 ^H , Y—324 ^H , E-261 ^H HB1b (isopropyl): K—323 ^H , Y—324 ^H , E-261 ^H , K-320 ^H
Mpro	$ \begin{array}{l} K^{-520}, Y^{-524}, K^{-523} \\ HB1a: R^{-298}^{HB}, Y^{-154}^{HB}, F^{-8}, \\ I^{-152^{H}}, F^{-294^{H}}, V^{-297^{H}}, \\ D^{-153^{H}}, G^{-302^{H}}, E^{-299^{H}}, S^{-301^{H}}, \\ R^{-298^{H}}, Y^{-154^{H}}, K^{-12^{H}}, D^{-155^{H}}, \\ P^{-9^{H}} \\ HB1b: Y^{-154^{H}}, D^{-153^{H}}, I^{-152^{H}}, \\ F^{-264^{H}}, F^{-8^{H}}, R^{-298^{H}} \\ \end{array} $	HB1a (sec-butyl): Y—154 ^H , D—153 ^H , F—8 ^H , I—152 ^H , R- 29 ^H . HB1b (isopropyl): no interactions

The interactions with the differential chemical group of each type of avermectin are also shown. ^{HB}: Hydrogen Bonds; ^H: Hydrophobic Interactions; A: Alanine; R: Arginine; N: Asparagine; D: Aspartic Acid; E: Glutaminic Acid; Q: Glutamine; G: Glycine; H: Histidine; I: Isoleucine; L: Leucine; K: Lysine; M: Methionine; F: Phenylalanine; P: Proline; S: Serine; T: Threonine; W: Tryptophan; Y: Tyrosine; V: Valine.

^a HB1a is sec-butyl and HB1b is isopropyl group.

group) of compound HB1a established only hydrophobic interactions with the binding pocket, while the HB1b established both hydrogen bonds and hydrophobic interactions with this protein structure.

In IMP β 1, a lower number of interactions were observed compared to IMP α 1, but in contrast, these were mostly concentrated in the differential chemical groups, approximately 88.8% (8/9) of the interactions between IMP β 1 and HB1a were hydrophobic, followed by 11.2% (1/9) of hydrogen bonding type interactions. In this docking 6/7 of the interactions in the sec-butyl group were hydrophobic in nature, a similar result was observed in the docking of HB1b, with which the same number of hydrophobic interactions was observed, as well as hydrogen bonds. With 4/5 of the hydrophobic type interactions with its isopropyl chemical group (Table 3). The residue with the greatest contribution to the binding of HB1a and HB1b with IMP β 1 in terms of hydrophobic interactions and hydrogen bonds with each homolog and in both cases the union is established in the differential chemical group of each avermectin (see Table 3).

The interactions here predicted between $IMP\alpha 1$ and the HB1a correspond to those reported in the literature using ADV [46]. Analyzes have been performed showing that the binding pocket residues seen in our study Trp184, Arg227, Trp231, Ser149, Asn228, Trp231, and Arg238 appear to be critical in maintaining HB1a docking within the

main groove. Trp184, Asn228 and Trp231 residues have already been shown to be vital in providing various intermolecular interactions [46]. No reports of molecular docking specifically between IMP α 1 and IMP β 1 with HB1b were found.

In the Helicase protein, 79.2% (19/24) of the interactions presented by HB1a were hydrophobic, followed by 20.8% (5/24) hydrogen bondtype interactions, whereas the HB1b exhibited 86.4% (19/22) hydrophobic interactions and 13.6% (3/22) hydrogen bond-type interactions (Table 3). The differential chemical groups (sec-butyl or isopropyl moieties) of homologs contributed both with hydrophobic interactions and with hydrogen bonds against the Helicase. The residues with the greatest contribution in terms of hydrophobic interactions and hydrogen bonds to the Helicase and HB1a complex were E-261, S-289, H-290, A-316 and R-442; and in the complex with HB1b were E-261, S-289 and A-316. From these residues, only E-261 established hydrophobic interactions and hydrogen bonds with each homolog and in both cases the union is established in the differential chemical group of each avermectin (see Table 3). These results are similar to those reported between the Helicase and HB1a also using ADV on the ADP site [43], and to those reported in similar docking regions [50]. No molecular docking specifically between this protein and HB1b have been reported until now.

In the M^{pro} protein, interestingly, all the interactions of HB1a were hydrophobic (6/6) and were entirely concentrated around the sec-butyl chemical group, while HB1b showed 86.6% (13/15) of hydrophobic interactions and 13.3% (2/15) of hydrogen bonding type. The residues with the greatest contribution in terms of hydrophobic interactions and hydrogen bonds to the Mpro and HB1a complex were R-298 and Y-154; and in the complex with HB1b were R-298, I-152 and Y-154. Of these residues, Y-154 established hydrophobic interactions and hydrogen bonds with HB1a, as well as hydrophobic interactions with the chemical group (sec-butyl) of HB1a, and only hydrophobic interactions with HB1b (without participation of the chemical group (isopropyl) (Table 3). It was found that both homologs interact with the Arg-298 residue (R-298) (Table 3), which has been reported as key in the catalytic activity of the enzyme [74]. This residue has been described to be involved in dimerization of this protein and plays a key role in the catalytically active conformation of M^{pro} [75]. This coincides with reports that affecting the Arg-298 residue leads to the alteration of the dimeric structure of M^{pro}, as well as the irreversible inhibition of the catalytic activity of the enzyme [76,77].

Avermectins established more than 80% of hydrophobic interactions with each of the complexes tested, exhibiting a greater inclination towards this type of interactions, which highlights the influence of the hydrophobic characteristics of the cavities on their union. It is important to point out that the interactions observed with the sec-butyl chemical group were predominantly hydrophobic in the binding pockets of M^{pro}, while isopropyl exhibited more balanced interactions between hydrophobic and hydrogen bonding types, but at a global and individual level, slightly favored by hydrogen bond. But regardless of the chemical nature of the ivermectin homolog, these can interact with important residues involve in the dimerization of M^{pro}.

An observation that is important and that requires further study to determine if these interactions can have an impact on the formation of homodimer M^{pro}. These results differ from those reported by other authors who observed dockings at the catalytic site level, but without considering the differential chemical groups and only evaluating the interaction with HB1a [7]. One of the few works that considered both homologs against M^{pro} in their theoretical analyzes predicted a favorable but energetically very close docking between HB1a and HB1a using ADV [50]. The predictions have been made in regions other than the one observed in our study [50,78], and differ from other reports using ADV and HB1a with respect to M^{pro} predicting dockings in the catalytic region [46]. In this sense, this study provides a broader view on the individual interaction of the chemical groups of each homolog, as well as a greater evaluation of each compound through blind docking and a greater number of scoring functions to increase the reproducibility of the



Fig. 2. The most stable conformation of each avermectin is shown in the binding pockets of the proteins. Diagram of the orientation in M^{pro} (A and B), IMPa (C and D), IMPb (E and F), and Helicase (G and H) of HB1a and HB1b and their macrocyclic rings within the respective binding pockets, respectively. HB1a, avermectin B1a; HB1b, avermectin B1b. Macrocyclic rings are shown within a blue circle in the upper boxes; Binding pockets are shown as surface rendering and avermectin is attached as sticks style. (For interpretation of the references to colour in this figure legend, the reader is referred to the web version of this article.)

results.

3.3. Comparative theoretical study of the inhibition of target proteins by HB1a and HB1b $\,$

From the predictions of molecular docking, the theoretical values of both the K_i and IC₅₀ of each of the homologs against each of the protein structures considered in this study were determined. In this study we assume a competitive inhibition and constant concentrations of substrate and inhibitor, and we simulate a gradient that includes plasma concentrations considered safe, as will be discussed later, also the theoretical concentration used was placed as a subscript value in μ M as suggested [29]. Results showing a more favorable K_i compared to M^{pro} followed by the values for the Helicase protein, especially for the HB1b homolog. Predictions like the values obtained with strong M^{pro} inhibitors such as dipyridamole ($K_i = 0.04 \ \mu$ M) have shown promising therapeutic effects in clinical studies conducted for the treatment of patients with COVID-19 [28]. Similarly, compounds such as UAWJ246 and UAWJ247 have been reported to reversibly to M^{pro} with $K_i = 0.04 \mu M$ [79]. The K_i predicted here for both homologs are lower than those reported for drugs such as hydroxychloroquine ($K_i = 0.36 \mu M$) and chloroquine ($K_i = 0.56 \mu M$), also calculated using the Dixon plot model and described as inhibitors of M^{pro} [28]. The results obtained can be seen in the Table 4.

From the values of the Table 5 can be determined the relationship between the calculated theoretical inhibition constant and the simulated theoretical concentration gradients between 0.5 μ M and 5 μ M (subscript values) of HB1a and HB1b, according to the Dixon plot, showed that the predicted concentration for the theoretical 50% inhibition of IMP α 1 by HB1a could require an IC₅₀ \approx 0.49 μ M_(0.5-5), with a maximum concentration of IC₅₀ = 1.63 μ M₍₅₎, while the mean theoretical concentration for HB1b would require an IC₅₀ \approx 0.69 μ M_(0.5-5), with a maximum

Table 4

Comparison of theoretical inhibition kinetic va	alues calculated for HB1a and HI	31b with the proteins considered.
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Target	K_{i} (M) ^a		K_{i} (M) ^b		IC ₅₀ (M) ^c		IC ₅₀ (M) ^b		pIC ₅₀ (M) ^b	
	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b
IMPa1	3.3	4.6	3.1	4.2	5.0	2.0	2.1	3.0	6.7	6.6
ΙΜΡβ1	7.5	3.0	6.4	2.8	0.1	2.0	4.0	2.0	6.4	6.7
Helicase	2.3	0.4	2.2	0.4	0.2	0.2	2.0	0.3	6.8	7.5
M ^{pro}	1.0	0.4	1.0	0.4	0.4	0.2	0.8	0.3	7.1	7.5

 K_{i} , inhibition constant for binding of ligand to proteins in units of M; IC₅₀, maximum mean inhibitory concentration in units of M; pIC₅₀, negative logarithm of IC₅₀. All K_i and IC₅₀ values are in 10⁻⁷; HB1a, avermetin B1a; HB1b, avermetin B1b.

^a Constant predicted from Eq. (2).

^b Constant predicted from the IC_{50} -to- K_i web server [27].

^c Constant predicted from the Dixon plot [29].

Table 5

Comparison of the concentration gradient considered for ivermectin to simulate the potential inhibitory kinetics of HB1a and HB1b against the proteins considered.

IVM [µM] ^a	ΙΜΡα1		ΙΜΡβ1		Helicase	Helicase		M ^{pro}	
	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	HB1a	HB1b	
IC ₅₀ (µM) Dixon plot	t ^b								
0.5	0.163	0.228	0.372	0.152	0.116	0.018	0.050	0.018	
1.0	0.326	0.455	0.745	0.304	0.233	0.037	0.101	0.037	
1.5	0.488	0.683	1.117	0.457	0.349	0.055	0.151	0.055	
2.0	0.651	0.911	1.489	0.609	0.465	0.073	0.201	0.073	
2.5	0.814	1.138	1.861	0.761	0.582	0.092	0.251	0.092	
3.0	0.977	1.366	2.234	0.913	0.698	0.110	0.302	0.110	
3.5	1.139	1.594	2.606	1.065	0.814	0.129	0.352	0.129	
4.0	1.302	1.821	2.978	1.217	0.931	0.147	0.402	0.147	
4.5	1.465	2.049	3.351	1.370	1.047	0.165	0.452	0.165	
5.0	1.628	2.277	3.723	1.522	1.163	0.184	0.503	0.184	
IC50 (µM) Web toolc									
0.5	0.096	0.113	0.138	0.091	0.079	0.024	0.047	0.024	
1.0	0.130	0.159	0.204	0.123	0.103	0.027	0.057	0.027	
1.5	0.151	0.187	0.248	0.141	0.117	0.029	0.063	0.029	
2.0	0.165	0.207	0.280	0.154	0.127	0.030	0.066	0.030	
2.5	0.176	0.223	0.305	0.164	0.134	0.031	0.069	0.031	
3.0	0.185	0.235	0.325	0.172	0.140	0.032	0.071	0.032	
3.5	0.192	0.246	0.343	0.178	0.145	0.032	0.073	0.032	
4.0	0.198	0.255	0.358	0.184	0.149	0.033	0.074	0.033	
4.5	0.203	0.262	0.371	0.188	0.152	0.033	0.075	0.033	
5.0	0.208	0.269	0.383	0.193	0.155	0.033	0.076	0.033	

The hypothetical substrate-inhibitor ratio (1: 1) was considered to avoid preferential binding favored by concentration, assuming a competitive inhibition and structural similarities between homologs and substrate; HB1a, avermeetin B1a; HB1b, avermeetin B1b.

^a A 0.5-5 µM gradient was used to speculate the inhibitory potential based on plasma concentrations considered safe for ivermectin (IVM).

^b Predicted from the Dixon plot [29].

^c Predicted from the IC₅₀-to-*K*_i web server [27].

concentration under these conditions of $IC_{50} = 2.28 \ \mu M_{(5)}$, this represents a maximum theoretical mean of an $IC_{50} = 2 \ \mu M_{(5)}$ for ivermectin *versus* IMP α 1.

In IMP $\beta1$, the mean for the HB1a was an $IC_{50}\approx 1.13~\mu M_{(0.5-5)}$, and the maximum concentration of $IC_{50}=3.72~\mu M_{(5)}$, the mean for HB1b was $IC_{50}\approx 0.46~\mu M_{(0.5-5)}$, with a maximum value of $IC_{50}=1.52~\mu M_{(5)}$. This leads to a maximum theoretical mean of an $IC_{50}=2.6~\mu M_{(5)}$ for ivermectin versus IMP $\beta1$ (Tables 4 and 5). These results compared to Importins are related to IC_{50} values reported in cell culture assays ($\approx 2~\mu M$) and therefore, with the potential inhibition of the importation associated mechanism of action as described [80].

It is not clear if the values reported correspond to any of our homologs, or if it is a typical mixture named ivermectin. For the Helicase, a mean of IC₅₀ $\approx 0.35 \ \mu M_{(0.5-5)}$ was predicted for HB1a, with a maximum concentration of IC₅₀ = 1.16 $\mu M_{(5)}$, and to a lesser extent, an average IC₅₀ $\approx 0.06 \ \mu M_{(0.5-5)}$, with a maximum concentration of IC₅₀ = 0.18 $\mu M_{(5)}$ for HB1b, which would represent a maximum theoretical mean of an IC₅₀ $\approx 0.2 \ \mu M_{(5)}$ for ivermectin *versus* the Helicase (Tables 4 and 5). These results are within the ranges reported of inhibition of activity (IC₅₀ = 0.12–0.5 μM) and kinetics of the Helicase (IC₅₀ = 0.019–0.354 μM), as well as viral replication by 50% (IC₅₀ = 0.0005–4 μM) [81]. Observations that are related to the alteration of viral replication that this compound can induce through this pathway, since ivermectin has also been shown to bind to this non-structural viral protein [82].

On the other hand, compared to M^{pro} , a mean $IC_{50} \approx 0.15 \ \mu M_{(0.5-5)}$ was predicted for HB1a, with a maximum concentration of $IC_{50} = 0.50 \ \mu M_{(5)}$, and as with the Helicase, an average $IC_{50} \approx 0.06 \ \mu M_{(0.5-5)}$, with a maximum concentration of $IC_{50} = 0.18 \ \mu M_{(5)}$ for the HB1b, which would represent a maximum theoretical mean of an $IC_{50} \approx 0.1 \ \mu M_{(5)}$ for ivermectin *versus* M^{pro} (Tables 4 and 5). The binding affinity of this type of ligand at the submicromolar level has already been reported using algorithms considered here [56]. These concentrations predicted for M^{pro} inhibition range around concentrations reported as safe in plasma and lung tissue [30], but they differ from those results obtained in *in vitro* enzyme inhibition assays with isolated protein ($IC_{50} = 20 \ \mu M$) [7]. This difference could be due to the fact that neither the test with the isolated protein nor our *in silico modelling* consider the effect of cytoplasmic crowding, an aspect that could explain the differences that exist between the viral inhibition tests in cell culture [80] and enzyme inhibition with isolated proteins [7], because molecular crowding is decisive when it comes to predicting the activity, dynamics and kinetics of biomolecules in a more realistic way [83,84].

It is not clear whether the results reported for isolated M^{pro} correspond to the HB1a, as evidenced in the graphs of the structure used for molecular docking, or whether it corresponds to a typical ivermectin mixture. An aspect that is important from our results that there exists a clear trend that the HB1b would require a lower concentration for the inhibition of the viral protease, due to the differential energy and kinetic qualities that exist between both homologs. In fact, the only experimental work with cell culture that has reported the use of HB1a is with an $IC_{50} \approx 50 \, \mu M$ [49]. It is interesting how in the studies that have been carried out to determine the IC_{50} when working with ivermectin, lower values have been observed than when working with the HB1a. This allows us to suggest that the potency observed when working with the ivermectin mixture may be due to the contribution of HB1b. Which would be in agreement with the results estimated under the conditions of this study.

The results obtained by the Dixon plot were compared with those predicted by the IC_{50} -to- K_i web server, and we observed that the same inhibition trend described above is maintained. Therefore, only the concentrations for the theoretical inhibition of M^{pro} predicted by both the Dixon plot and the IC_{50} -to- K_i web server are close to the accepted safe values, followed in second order by the values obtained for the Helicase inhibition (Tables 4 and 5). It has also been suggested that an $IC_{50} \approx 6-17 \ \mu$ M can convert ivermectin into a non-selective ATPase inhibitor [30,85]. We suggest that the individual activity of each homolog be evaluated independently, especially as the concentrations predicted here have already been suggested for evaluation [86], and because the pharmacokinetics of individual homologs against SARS-CoV-2 have not been reported up to now. Although our results show a tendency and a self-consistency between the binding energy and the constants and/or theoretical inhibition concentrations predicted for the

homologs against each of the proteins tested, they should be taken as reference. However, these observations should not be ignored because as many of the cited pharmacokinetic evaluations indicate that ivermectin could be toxic *in vitro*, so further work is needed to reconcile the observations *in vitro* with clinical efficacy as suggested [87].

It is important to note that 1% DMSO concentration used for isolated M^{pro} has been shown to be capable of inducing folding in model proteins [88] which could affect the required concentration of the compound at sites of interest for enzyme inhibition, by favoring the natural folding mechanism and reducing the competitiveness of the drug for the site, as has already been studied [89]. That is why we conducted a preliminary study that represents the first reported evidence of the possible effect of molecular crowding and the addition of 1% of DMSO on the activity of avermectin homologs in structures associated with SARS-CoV-2 such as M^{pro}. The M^{pro} in presence and absence of homologs at 100 ns simulation was compared in three media: physiological environment (0.9% NaCl), 1% DMSO and 20% polyethylene glycol (PEG), a common crowding agent.

We consider the synthetic polymer PEG because, although less physiologically relevant, it is often used usually to mimic the interior of the cell as crowding agent and is useful for stabilizing protein-based drugs as recommended [90]. We define ΔRg as the change in R_g in DMSO or PEG of M^{pro} with and without ligands minus R_g in physiological solution. At which we could observe that the changes at a concentration of 1% DMSO were very discrete, suggesting that no significant effects are expected at the experimental level ($\Delta R_g \approx 0.1$ Å). The ΔR_g in an environment crowded with PEG *versus* the physiological environment did show important changes in terms of Rg of all complexes. We observe that the free M^{pro} protein and complex with HB1a are little affected by crowding ($\Delta R_g \approx 0.1$ Å), however, the M^{pro} complex with HB1b is significantly affected, showing great structural compaction $(\Delta R_g \approx$ - 0.7 Å) (Supplementary Fig. 1), which thermodynamically favors this system because under crowding the compact structures are more favorable as has been already reported [83]. As in a physiological environment and 1% of DMSO, the energetic stability of the docking favors the HB1b compound despite inducing a conformational unfolding, this allows us to assume that the complex established with HB1b is stable and capable of producing a more favorable minimum energy structure than that generated by HB1a, although both compounds in DMSO destabilize the native (Supplementary Fig. 1). This could explain the differences between the aforementioned in vitro and cell culture results, and demonstrates the challenge of reconciling the often inconsistent observations at the in silico, in vitro and in vivo levels to increase the safety and clinical efficacy of drugs, especially in the early stages of studies with predictive models. Therefore, a more in-depth on the impact of variables such as cell crowding is currently being investigated by our team, while we recommend carrying out this type of study with a greater number of structural and functional proteins associated with SARS-CoV-2, including cellular proteins of the Importins superfamily.

Although the difference between both homologs of ivermectin could imply a non-functionalized part of their structures, and it is very likely that this type of structural difference affects the pharmacokinetic disposition more than the binding of the target. Interestingly, our results suggest that there is a differential behavior to level of the potential inhibitory effect of compounds HB1a and HB1b on the targets considered, and that this inhibitory effect is related to the predicted binding energy values. This kinetic aspect is favored in HB1a and HB1b for the cellular and viral proteins studied, respectively. A differential activity that has already been reported for these compounds both at the energy level and in terms of their particular affinity for related cellular and viral proteins [91].

It is also likely that both the sec-butyl and isopropyl groups contribute to the orientation, conformation, or predicted intermolecular interactions in the target binding pockets, in conjunction with the hydrophobic characteristics of each pocket (Table 3 and Fig. 2) and of each homolog (Table 2), because the steric effect of each chemical group contributes differently to the total volume of the HB1a (832.02 A^3) and HB1b (815.22 A^3) molecules. In fact, the sec-butyl group contributes 66.77 A^3 and the isopropyl 49.97 A^3 to the macrolactone molecule according to the Molinspiration server calculations. This method for predicting molecular geometries is very robust and relies on the contributions of each group for the calculation of molecular volume [92]. The efficient molecular docking algorithms used consider the spatial configuration of functional groups, as well as the incidence of molecular interactions by hydrogen bonding, rotation and steric shocks [55–61].

The *in vivo* biological activities of the two homologs are known to be identical in cell and animal models [71]. However, it is important to note that while the structural similarity of compounds HB1a and HB1b could explain why both homologs bind the various targets with comparable affinity values, the impact of binding energies on predicted inhibition kinetics should not be ignored because comparative analyzes of binding energies have been shown to be an important part in attempts to predict drug-target binding kinetics [93,94]. Emphasizing that the genetic docking algorithm included in this study has shown success rates like that exhibited by other methods used to validate experimental drug affinity results [55,56,94].

3.4. Comparative study on molecular dynamics of HB1a and HB1b

MD simulations in the field of computer-aided drug design have gained substantial importance for the estimation of dynamic and thermodynamic parameters of living systems in specific situations of physiological environments such as those considered here [46]. To investigate the stability and effect of the most favored docking at the energy and kinetic level, MD simulation studies were carried out for both homologs, HB1a and HB1b, against each of the proteins of interest, considering the stability of the systems in terms of total thermodynamic fluctuations and the difference in the atomic distance between structures every 10 ns of simulation in the protein systems in the presence and absence of ligands. The purpose of this study was to determine both the stability of the complexes and the potential changes and disturbances that each of the homologs may be capable of inducing on the proteins of interest.

The energetic fluctuations associated with the thermodynamic stability of the ligand-protein complexes showed that both homologs establish thermodynamically stable systems compared to native structures throughout the time considered in this study. As can be seen in Fig. 3, the ligands affected the energetic contributions and limited the degrees of freedom of the energetic transitions by mediating minimum energy conformations of up to a maximum of -2600 kcal/mol, compared to the native structure that free reached about -2900 kcal/ mol at the end of the simulation time. IMP α 1 reached favorable minimum energy structures, thermodynamically coupled with each homolog, stabilizing the trajectories from approximately 30 ns with total energy fluctuations in the order of -2300 and - 2500 kcal/mol over time. Avoiding the progressive descent seen of the native structure through the different energetic and probabilistically feasible conformations that it suffered from 15 ns. It is important to highlight that the $IMP\alpha 1 + HB1a$ complex reached a more stable minimum energy conformation (-2600 kcal/mol) and in less simulation time (-70 ns) than that observed with the HB1b (-2500 kcal/mol at -75 ns) (Fig. 3A–C).

In IMP β 1, the relationship between the decrease and the thermodynamic stabilization of the minimum energy conformations maintained a similar trend as seen with IMP α 1 in the presence and absence of the ligands. Free protein reached a minimum energy conformation of about -1300 kcal/mol at 75 ns, after the initial decline at 5 ns. In presence of the two homologs, the energy fluctuation of the minimum energy conformation ranged from -1100 kcal/mol at approximately 95 ns. It is important to highlight that in the presence of HB1a, IMP β 1 reached stabilization in thermodynamic fluctuation at 15 ns, unlike the HB1b, at



Fig. 3. Fluctuation of the thermodynamic stability at 100 ns of proteins in the presence and absence of each homolog. A) IMPα1 free; B) IMPα1 + avermectin B1a (HB1a); C) IMPα1 + avermectin B1b (HB1b); D) IMPβ1 free; E) IMPβ1 + avermectin B1a (HB1a); F) IMPβ1 + avermectin B1b (HB1b), G) Helicase free; H, Helicase + avermectin B1a (HB1a); I) Helicase + avermectin B1b (HB1b); J) Mpro free; K) Mpro + avermectin B1a (HB1a); L) Mpro + avermectin B1b (HB1b).

which this effect was observed after 30 ns. A stabilization that required more time but was maintained until the end of the simulation (Fig. 3D–F). Which could indicate that the predictions made by genetic algorithms about the most favorable energetic orientation between

HB1b and IMP β 1 are thermodynamically feasible and show both a differential energetic between the homologs when docking and when inducing changes in the thermodynamic stability of tested protein systems. More important, unlike general molecular docking methods, MD

simulations consider target flexibility and combined with energy calculations of thermodynamic fluctuations in binding, however a more accurate prediction of potential inhibitors can be developed [95].

In the Helicase, the homologs stabilized the thermodynamic fluctuations of the Helicase after 35 ns of the simulation system, this system was the largest studied with about 9407 atoms (\approx 2000 atoms above the closest system in size, represented by $IMP\beta1$), reaching a total energy that ranged about -10,000 kcal/mol, which remained stable until the end of the MD cycle. This energetic behavior is faster with HB1b (at 30 ns). The fluctuations of the native structure showed a decrease in the minimum energy conformations from 25 ns of around -10,000 kcal/ mol, until reaching some -11,000 kcal/mol at 70 ns (Fig. 3G-I). The results show that the homologs affected the energy contributions and limited the degrees of freedom of the thermodynamically probable transitions in the Helicase, as with the Importins, which could indicate an increase in the conformational stiffness of the system because of the ligands. The earliest stabilization phenomenon evidenced with HB1b is related to the binding energetics and inhibitory kinetics described above and which are slightly more favorable for this homolog compared to HB1a

The homologs stabilized the thermodynamic fluctuations of Mpro. Specifically, after 40 ns of the simulation system, the HB1a stabilized the thermodynamic fluctuations constantly throughout the 100 ns of simulation, reaching a total energy that ranged around -2600 kcal/mol. As the HB1b stabilized the system at 30 ns of the simulation. An energetic behavior like that observed against the Helicase and faster than that presented by homolog HB1a. Reaching an energy of -2800 kcal/mol (similar to the energy of the native structure). The fluctuations of the native structure showed a decrease in the minimum energy conformations from 15 ns of -2800 kcal/mol throughout the simulation (Fig. 3J–L). In relation to the behavior of the RMSD of C α for each

protein studied and their complexes with the homologs, in the Fig. 4 can be seen the results obtained in each case. In IMPa1, the values obtained show that the thermodynamic changes induced on IMPa1 by both homologs caused a decrease in the distance of the alpha carbons of the complexes, with a difference at 100 ns with respect to the native RMSD of \approx 3 Å and \approx 0.5 Å approximatly for HB1a and HB1b, respectively (Fig. 4A), which would represent a significantly folded structural state with HB1a and favorable.

Unlike the results predicted at 100 ns against IMP β 1 in which no differences were observed in the systems with and without ligand with an RMSD ≈ 5 Å. Results that show that the HB1a exerts a significant differential effect at the thermodynamic and structural level on IMP α 1 compared to the induced by HB1b. Neither of the two homologs was able to induce significant differential perturbations against the native structure of IMP β 1 under the conditions of this study (Fig. 4B). The equilibrium of the IMP α system in the presence of HB1a as well as the large fluctuations of the C α atoms have already been reported with different magnitudes and are considered an indication of large conformational changes in the protein structure during the simulation [4,46].

Against the Helicase, both homologs presented an RMSD \approx 5 Å at the end of 100 ns. The difference between each Helicase-homolog complex and Helicase was RMSD \approx 1 Å at 100 ns of the simulation (Fig. 4C). Results that do not show important and differential perturbations between these complexes compared to the native structure under the simulation conditions. The energetic fluctuations over time, specifically at \approx 30 ns and \approx 70 ns have already been described and show the variations in the thermodynamic stability of the Helicase system in the presence of ivermectin [4]. Furthermore, the Helicase values in terms of global RMSD are high due to its large structure, which includes several loops and flexible domains as reported against other types of ligands [16].



Fig. 4. The Root Mean Square Deviations (RMSD) of Cα relative to the starting frame during 100 ns MD simulation. A, IMPα1 (IMPα1); B, IMPβ1 (IMPβ1); C, Helicase and D, Mpro in the presence and absence of each homolog (HB1a and HB1b).

On the other hand, the results obtained for M^{pro} system revealed that the M^{pro} + HB1a complex has an RMSD \approx 3.3 Å between 30 ns and 60 ns, and a value of RMSD \approx 3 Å at 100 ns, with results similar to the native structure (RMSD \approx 3 Å) at 100 ns. During the same periods of time, the complex with HB1b presented an RMSD \approx 3.8 Å between 30 ns and 60 ns, and an RMSD \approx 4 Å at 100 ns. This represents a difference of RMSD >1 Å at 100 ns between HB1b and the rest of the structures with and without ligand and show differential perturbations between these compounds compared to M^{pro}, which favor the HB1b in terms of folding. This shows that the homologs could modify the natural thermodynamic stability of M^{pro} by different mechanisms (Fig. 4D). There are few reports of DM of ivermectin versus Mpro, however, the results obtained in terms of RMSD of the C α atoms of M^{pro} in SARS-CoV-2 are related to those reported during the simulation period of 100 ns versus other compounds that also remained in the steady state at \approx 3 Å RMSD [16,96,97].

Also, all simulated complex graphs are presented as Radius of Gyration (ROG) guided movements using a geometric simulation based on normal mode. The resulting trajectories of the simulated complexes was inspected for different ROG simulation parameters such as backbone root mean square deviations (RMSD) for all C α carbon atoms of the protein and the root mean square fluctuations (RMSF) of individual amino acid residues (Fig. 5). A behavior similar to those predicted by RMSD was observed after MD. The ROG-based fluctuations are related to the ROG-predicted RMSD values, showing a slightly more unfolded state of the ligand-IMP α 1 complexes relative to the free IMP α 1 structure (Fig. 5A and Fig. 5B). A similar fluctuation was also predicted in the case of IMP β 1 (Fig. 5C and D), Helicase (Fig. 5E and F) and M^{pro} (Fig. 5G and H).

Variations in terms of RMSF could suggest a stable binding of homologs at their respective binding sites, generating complexes with fluctuations close to the free proteins, specifically, the region between residues N-146 and T-311 represents the binding site for both homologs in IMP α 1; the region between R-52 and L-64 for each IMP β 1 homolog; as well as the E-261 and E-540 region for the Helicase with each homolog; observing a greater diversity of residues involved in binding with M^{pro} for each homolog (region between F-8 and S-301) (see Tables 2 and 3). Furthermore, the binding of each homolog could induce slight conformational perturbations in terms of RMSD, especially since even though the NMSim simulation was configured with "mode 1" to guide the simulation towards the lowest ROG, in the case of IMP α 1, a slight protein unfolding was observed with both homologs (Fig. 5A).

When studying the interactions of each of the homologs attached to their respective pockets in the importins and throughout the simulation time of 100 ns, it was observed that HB1a presented few variations in the interactions initially established throughout the simulation time maintaining a stable binding in place, as described (Table 6) [46]. The number of hydrogen bonds throughout the simulation, or the hydrophobic interactions established by its sec-butyl chemical group, were not altered. An increase in the number of hydrophobic interactions was only observed after 75 ns.

Results that show a more favorable and stable binding of HB1a against IMP α 1, compared to that predicted with HB1b. This variation in docking stability has already been reported but compared only to HB1a [4]. We observed that the HB1b presented more fluctuations in the interactions established over time, showing variations in global hydrophobic interactions and a decrease in the number of hydrogen bonds, as well as in the hydrogen bonds and hydrophobic interactions established by its isopropyl chemical group, however, fluctuations in global hydrophobic interactions were represented by an increase from 50 ns.

Specifically, the HB1a maintained its 2 initial hydrogen bonds and went from having 14 hydrophobic interactions to 18 within 100 ns of the simulation. As HB1b went from establishing 4 hydrogen bridges to 1, with a compensation in the number of hydrophobic interactions from 14 to 18 for 100 ns. In its differential chemical group (isopropyl group), HB1b lost the hydrogen bonds, but increased the number of hydrophobic interactions. Results showing less stable binding of HB1b than predicted for HB1a, and this is consistent with the previously described energetic and kinetic values that favor the binding of the HB1a against IMP α 1.

In IMP β 1, the interactions established by HB1a fluctuated at the level of hydrogen bonds and hydrophobic interactions (Table 6), these fluctuations were represented by a decrease in the hydrogen bonds from 75 ns, as well as an increase in the hydrophobic interactions from the same period of time. The same tendency was observed in the interactions established by its sec-butyl chemical group, with a loss of hydrogen bonds from 75 ns and a decrease in hydrophobic interactions between 50 ns and 75 ns. As the HB1b showed an increase in all its interactions, both hydrophobic and hydrogen bonds at a global level, as well as its differential chemical group, which, although it suffered a decrease in the number and hydrophobic interactions, presented a compensatory increase in the number of hydrogen bonds at the end of the 100 ns simulation. It is important to note, that the HB1a after 100 ns lost all its hydrogen bonds and went from having 8 hydrophobic interactions to 11. In contrast, the same time of simulation, HB1b exhibited an increase in the number of global and local hydrogen bonds (in the differential chemical group), as well as an increase from 8 to 12 in terms of hydrophobic interactions at the end of 100 ns. Results that correspond with the observations at the energy and kinetic level, which show that, of the two homologs, HB1b presents a more stable and favorable binding by IMPβ1 (Table 6).

Although the results previously described on the interaction of the Helicase with both homologs showed an earlier stabilization in terms of thermodynamic fluctuation between HB1b and this enzyme over time, the results show a higher binding stability by the HB1a, which maintained the highest number of hydrogen bonds throughout the simulation. Both homologs showed a decrease both in the number of hydrogen bonds at the global level, and in the number of hydrophobic interactions at the local level in each of their differential chemical groups (sec-butyl *vs* isopropyl group).

The HB1a presented a decrease in the number of hydrogen bonds, offset by an increase in the number of hydrophobic interactions at the global level and in the sec-butyl group from 50 ns, reflected in a slight increase from 19 to 20 global hydrophobic interactions, and from 4 to 5 in their differential chemical group, but with a 5 to 2 loss in their hydrogen bonds at the end of 100 ns. The HB1b presented a significant decrease after 50 ns in all the interactions as reported [46]. These interactions permit to suppose that under the conditions of this study, the docking of both homologs occurred in a different region than the one reported using similar docking algorithms [63].

Compared to M^{pro}, both compounds presented stable dockings reflected by an increase in their number of hydrophobic interactions over time, especially in its differential chemical groups. HB1a went from establishing 6 hydrophobic interactions to 21. With an increase of 5–8 hydrophobic interactions in its sec-butyl chemical group. HB1b lost the initial hydrogen bonds, and only increased the number of hydrophobic interactions from 13 to 14 after 100 ns simulation, the isopropyl differential chemical group established 2 hydrophobic interactions not initially predicted during docking. Indicative of stable unions at the end of the simulation (Table 6). Neither of the two compounds lost the interaction with the A-298 residue previously described as associated with dimerization and catalytic activity of the enzyme [76,77]. Time series graphs and/or histograms of ligand-protein interactions are shown in the Fig. 6 as evidence of persistence during MD simulations.

An increase in the number of hydrophobic interactions (Fig. 6A) as well as a decrease in the number of hydrogen bonds (Fig. 6C) is observed throughout the simulation of 100 ns. A differential depletion of hydrophobic interactions and hydrogen bonds in the typical chemical groups of each homolog was also predicted as a function of time (Fig. 6B and D).

A clearer view of the contact stability between HB1a/HB1b and their target receptors is provided in Fig. 7. Interaction energies between the ligand and the receptor were calculated and plotted using the predicted















Fig. 5. The Root Mean Square Deviations (RMSD) of C α and root mean square fluctuations (RMSF) of individual amino acid residues relative to the starting frame during 100 ns MD simulation of each homolog in complex with IMP α 1 (A and B), IMP β 1 (C and D), Helicase (E and F), Mpro (G and H).

Table 6

Interactions of the HB1a y HB1b with residues in the binding pockets of the pr

Interactions in chemical group^a

K-323^H, K-320^H, Y-324^H, E-261^H//K—323^H, Y—324^H, E-261^H, K-320^H

K-323^H, K—320^H, E-261^H// K-323^H, E-261^H, K-320^H

K-323^H, Y-324^H, E-261^H//No

K-323^H, Y-324^H, D-260^H, E-261^H, R-442^H//K-569^H

Y-154^H, D-153^H, I-152^H, F-8^H, R-298^H//No interactions detected

Y-154^H, D–153^H, F–8^H, I–152^H, D–155^H//No interactions

F-8^H, Y-154^H, I-152^H, D-153^H,

R-298^H//No interactions detected

detected

interactions detected

Table 6 (continued)

roteins	at 100 ns of MD.	lues in the binding pockets of the	Time (ns)	Total interactions		
Time (ns)	Total interactions	Interactions in chemical group ^a		E-261 ^{HB} , A-316 ^{HB} , S-289 ^{HB} , R- 442 ^{HB} , H-290 ^{HB} , E-319 ^H , L-317 ^H ,		
IMPα1 - 25	$\begin{array}{l} + \mbox{ HB1a//IMP\alpha1} + \mbox{ HB1b} \\ T-184^{HB}, N-188^{HB}, N-228^{H}, \\ D-270^{H}, A-269^{H}, E-266^{H}, T-331^{H}, \\ R-227^{H}, W-184^{H}, S-149^{H}, \\ N-188^{H}, N-146^{H}, W-2231^{H}, \\ P-308^{H}, L-307^{H}, W-273^{H}//E-266^{HB}, N-228^{HB}, \\ G-224^{H}, W-184^{H}, N-228^{H}, \\ G-222^{H}, W-184^{H}, N-228^{H}, \\ ARG-227^{H}, E-266^{H}, S-149^{H}, \\ D-270^{H}, G-150^{H}, N-235^{H}, \\ R-238^{H}, D-192^{H}, G-187^{H}, \\ G-187^{H}, \\ H-308^{H}, D-192^{H}, \\ H-308^{H}, \\ $	E-266 ^H , R-227 ^H //S—159 ^H , N-188 ^H	50	$\begin{array}{l} \text{A-316}^{\text{H}}, \text{S}{=}289^{\text{H}}, \text{E-540}^{\text{H}}, \text{R-443}^{\text{H}},\\ \text{G-538}^{\text{H}}, \text{G-285}^{\text{H}}, \text{D}{=}260^{\text{H}},\\ \text{S}{=}264^{\text{H}}, \text{H}{=}290^{\text{H}}, \text{R-442}^{\text{H}}, \text{G}{=}287^{\text{H}}, \text{F}{=}262^{\text{H}}, \text{T-286}^{\text{H}}, \text{E-261}^{\text{H}},\\ \text{K}{=}323^{\text{H}}, \text{Y}{=}324^{\text{H}}, \text{K}{=}320^{\text{H}}/\text{A}{=}316^{\text{H}}, \text{S}{=}289^{\text{H}}, \text{E-261}^{\text{H}}, \text{G}{=}285^{\text{H}},\\ \text{T-286}^{\text{H}}, \text{G-287}^{\text{H}}, \text{R-443}^{\text{H}}, \text{G-538}^{\text{H}},\\ \text{E-319}^{\text{H}}, \text{E-540}^{\text{H}}, \text{L-317}^{\text{H}}, \text{S}{=}289^{\text{H}},\\ \text{H}{=}290^{\text{H}}, \text{A-316}^{\text{H}}, \text{S}{=}246^{\text{H}},\\ \text{D}{=}260^{\text{H}}, \text{F}{=}262^{\text{H}}, \text{E-261}^{\text{H}},\\ \text{K}{=}320^{\text{H}}, \text{Y}{=}324^{\text{H}}, \text{K}{=}323^{\text{H}},\\ \text{S}{=}289^{\text{HB}}, \text{A-316}^{\text{HB}}, \text{R-44}^{\text{HB}}, \text{E}.\\ \end{array}$		
50		E-266 ^H , R-227 ^H //S—149 ^H , N—188 ^H , W-231 ^H		$\begin{array}{l} 261^{\rm HB}, H{-}290^{\rm HB}, S{-}289^{\rm H},\\ K{-}320^{\rm H}, E{-}319^{\rm H}, L{-}317^{\rm H}, A{-}316^{\rm H},\\ G{-}538^{\rm H}, G{-}287^{\rm H}, G{-}285^{\rm H}, T{-}286^{\rm H},\\ E{-}261^{\rm H}, D{-}260^{\rm H}, S{-}264^{\rm H},\\ H{-}290^{\rm H}, K{-}323^{\rm H}, R{-}443^{\rm H}, R{-}\\ 442^{\rm H}, E{-}540^{\rm H}//A{-}316^{\rm HB}, H{-}290^{\rm HB},\\ E{-}261^{\rm HB}, T{-}286^{\rm H}, G{-}287^{\rm H}, G{-}538^{\rm H},\\ E{-}319^{\rm H}, K{-}320^{\rm H}, S{-}269^{\rm H}, L{-}317^{\rm H},\\ D{-}260^{\rm H}, A{-}316^{\rm H}, S{-}264^{\rm H}, E{-}\\ 261^{\rm H}, H{-}290^{\rm H}, K{-}323^{\rm H}, R{-}442^{\rm H},\\ E{-}540^{\rm H}, H{-}443^{\rm H} \end{array}$		
75	$ \begin{array}{l} & 227^{\rm H}, 0.128^{\rm HB}, \text{R-}227^{\rm H}, \\ & W-184^{\rm HB}, N-188^{\rm HB}, \text{R-}227^{\rm H}, \\ & W-231^{\rm H}, \text{E-}266^{\rm H}, \text{D-}270^{\rm H}, \\ & W-184^{\rm H}, N-188^{\rm H}, \text{S-}149^{\rm H}, \\ & W-273^{\rm H}, N-146^{\rm H}, \text{T-}145^{\rm H}, \\ & P-312^{\rm H}, P-308^{\rm H}, \text{A-}269^{\rm H}, \text{L-} \\ & 307^{\rm H}, \text{T-}311^{\rm H}/\text{R-}238^{\rm HB}, \text{A-}148^{\rm HB}, \\ & \text{G-}224^{\rm HB}, W-184^{\rm HB}, \text{G-}191^{\rm H}, \\ & N-188^{\rm H}, \text{G-}187^{\rm H}, \text{R-}238^{\rm H}, \text{T-}151^{\rm H}, \\ & Y-277^{\rm H}, W-231^{\rm H}, N-235^{\rm H}, \\ & W-273^{\rm H}, \text{T-}155^{\rm H}, \text{D-}192^{\rm H}, \\ & N-228^{\rm H}, \text{R-}227^{\rm H}, W-184^{\rm H}, \text{E-} \\ & 180^{\rm H}, \text{G-}224^{\rm H}, \text{S-}149^{\rm H}, \text{A-}148^{\rm H}, \text{D-} \\ & 370^{\rm H} \end{array} $	R-227 ^H , E-266 ^H //A-148 ^H , S—149 ^H , N—188 ^H , W-231 ^H	75	$\begin{array}{l} {\rm E-261^{HB}, A-316^{HB}, S-289^{HB}, E-} \\ {\rm 540^{H}, R-442^{H}, G-285^{H}, T-286^{H}, \\ {\rm H-464^{H}, ARG-443^{H}, LYS-323^{H}, \\ {\rm TYR-324^{H}, H-290^{H}, T-440^{H}, G-} \\ {\rm 287^{H}, ASN-265^{H}, ASP-260^{H}, \\ {\rm S-264^{H}, A-316^{H}, E-261^{H}, A-312^{H}, \\ {\rm L-317^{H}, S-280^{H}, K-320^{H}, \\ {\rm S-539^{H}, E-319^{H}//A-316^{HB}, G-} \\ {\rm 285^{H}, R-442^{H}, E-540^{H}, S-539^{H}, \\ {\rm K-320^{H}, K-323^{H}, E-319^{H}, A-} \\ {\rm 316^{H}, S-264^{H}, D-260^{H}, E-261^{H}, \\ {\rm H-290^{H}, G-287^{H}, H-464^{H}, \\ {\rm H-290^{H}, G-287^{H}, H-464^{H}, \\ {\rm K-465^{H}, K-560^{H}, B-44^{H}, T-286^{H}} \end{array}$		
100	$\begin{array}{l} & W \!$	E-266 ^H , R-227 ^H //W—231 ^H , N—188 ^H , W—155 ^H , G-191 ^H , G- 150 ^H , A-148 ^H , R-238 ^H	100 Mpro +	$\begin{array}{l} \text{K-569}^{\text{HB}}, \text{E-261}^{\text{HB}}, \text{Y}-541}^{\text{H}},\\ \text{N}-542^{\text{H}}, \text{N}-190^{\text{H}}, \text{E-319}^{\text{H}},\\ \text{K}-320^{\text{H}}, \text{F}-343^{\text{H}}, \text{S}-539^{\text{H}}, \text{E-}540^{\text{H}}, \text{E-261}^{\text{H}}, \text{R-442}^{\text{H}}, \text{C}-441^{\text{H}},\\ \text{H}-290^{\text{H}}, \text{L-438}^{\text{H}}, \text{R-443}^{\text{H}}, \text{G-287}^{\text{H}},\\ \text{T-286^{\text{H}}}, \text{D}-260^{\text{H}}, \text{Y}-324^{\text{H}},\\ \text{K}-323^{\text{H}}, \text{K}-569^{\text{H}}/\text{R-443}^{\text{HB}}, \text{G-}538^{\text{H}}, \text{Y}-541^{\text{H}}, \text{E-540^{\text{H}}}, \text{S}-539^{\text{H}},\\ \text{D}-260^{\text{H}}, \text{E-261}^{\text{H}}, \text{H}-464^{\text{H}},\\ \text{K}-465^{\text{H}}, \text{R-442}^{\text{H}}, \text{K}-569^{\text{H}}, \text{T-}286^{\text{H}}, \text{R-}286^{\text{H}}, \text{H}-290^{\text{H}}\\\\ \text{HB1a}//\text{Mpro} + \text{HB1b}\\ \end{array}$		
IMPβ1 - 25	$ \begin{split} & \vdash HB1a / / IMP\beta1 + HB1b \\ & K-62^{HB}, K-60^{H}, M-61^{H}, K-62^{H}, L-64^{H}, I-54^{H}, D-63^{H}, R-53^{H}, R-52^{H} / K-62^{HB}, R-52^{H}, K-60^{H}, M-61^{H}, I-54^{H}, K-62^{H}, L-64^{H}, R-53^{H}, D-63^{H} \end{split}$	L-64 ^H , K $-$ 62 ^H , MET-61 ^H , I $-$ 54 ^H , R-52 ^H , D $-$ 63 ^H //K $-$ 62 ^H , L-64 ^H , D $-$ 63 ^H , M-61 ^H , R-52 ^H , I-54 ^H	25	$\begin{array}{l} Y\text{-}154^{\text{H}}, D153^{\text{H}}, I152^{\text{H}}, F8^{\text{H}}, \\ R\text{-}298^{\text{H}}, F294^{\text{H}}/(R\text{-}298^{\text{HB}}, Y\text{-}\\ 154^{\text{HB}}, I152^{\text{H}}, F8^{\text{H}}, D153^{\text{H}}, \\ F294^{\text{H}}, V297^{\text{H}}, G302^{\text{H}}, \\ S301^{\text{H}}, Q\text{-}299^{\text{H}}, R298^{\text{H}}, P9^{\text{H}}, \\ K12^{\text{H}}, D155^{\text{H}}, Y154^{\text{H}} \end{array}$		
50	$\begin{array}{c} K\text{-}62^{HB}, K\text{-}60^{H}, M\text{-}61^{H}, K\text{-}62^{H}, L\text{-}\\ 64^{H}, I\text{-}54^{H}, G\text{-}56^{H}//L\text{-}64^{H}, I\text{-}54^{H}, \\ R\text{-}53^{H}, L\text{-}62^{H}, M\text{-}61^{H}, R\text{-}52^{H}, L\text{-}60^{H}, \\ R\text{-}55^{H}, G\text{-}56^{H} \end{array}$	K-62 ^H , M-61 ^H , K—60 ^H //K—62 ^H , M-61 ^H , K-60 ^H	50	$\begin{array}{l} {\rm K-12^{\rm H},D{-}155^{\rm H},Y{-}154^{\rm H},} \\ {\rm D-153^{\rm H},F{-}8^{\rm H},F{-}294^{\rm H},I{-}152^{\rm H},} \\ {\rm R-298^{\rm H},G{-}302^{\rm H}//Y{-}154^{\rm H},} \\ {\rm K-12^{\rm H},D{-}155^{\rm H},P{-}9^{\rm H},I{-}152^{\rm H},} \end{array}$		
75 100	$\begin{array}{l} R-52^{H}, L-64^{H}, D-63^{H}, K-62^{H}, R-55^{H}, G-56^{H}, M-61^{H}, K-60^{H}, G-57^{H}, \\ D-58^{H}, I-54^{H}/R-55^{H}, G-57^{H}, \\ D-58^{H}, G-59^{H}, G-56^{H}, K-60^{H}, R-52^{H}, K-62^{H}, M-61^{H}, I-54^{H}, R-53^{H} \\ D-63^{H}, D-58^{H}, K-62^{H}, G-57^{H}, G-56^{H}, R-52^{H}, \\ L-64^{H}, H-19^{H}/Y-132^{H8}, H-19^{H8}, \\ I-54^{H}, K-62^{H}, G-57^{H}, D-58^{H}, \\ D-63^{H}, M-62^{H}, G-57^{H}, D-58^{H}, \\ N-63^{H}, K-62^{H}, G-28^{H}, M-98^{H}, \\ R-63^{H}, K-62^{H}, G-18^{H}, R-58^{H}, \\ R-63^{H}, K-62^{H}, G-18^{H}, R-58^{H}, \\ R-63^{H}, K-62^{H}, G-18^{H}, R-58^{H}, \\ R-63^{H}, K-62^{H}, R-57^{H}, R-58^{H}, \\ R-63^{H}, K-62^{H}, R-57^{H}, R-58^{H}, \\ R-63^{H}, K-62^{H}, R-58^{H}, \\ R-63^{H}, K-63^{H}, R-63^{H}, \\ R-63^{H}, R-63^{H}, R-63^{H$	K-62 ^H //K—62 ^H , I-54 ^H D-63 ^H , D—58 ^H , K—62 ^H , G-57 ^H , G- 56 ^H , L-64 ^H //H—19 ^H , L-64 ^H , Y132 ^H	75	$\begin{array}{l} {\rm F-294^{H}, {\rm F-8^{H}, V-297^{H}, G-302^{H}, }\\ {\rm S-301^{H}, R-298^{H}, D-153^{H}, }\\ {\rm Y-154^{HB}, {\rm F-8^{H}, {\rm F-294^{H}, V-297^{H}, }\\ {\rm S-301^{H}, D-153^{H}, {\rm I-152^{H}, }\\ {\rm Y-154^{H}, R-298^{H}, {\rm S-1^{H}, G-302^{H}, }\\ {\rm Y-303^{H}, C-300^{H}, Q-299^{H}, }\\ {\rm D-155^{H}, {\rm K-12^{H}/Y-154^{HB}, }\\ {\rm F-294^{H}, {\rm S-301^{H}, M-6^{H}, R-298^{H}, }\\ {\rm Y-297^{H}, Q-299^{H}, I-152^{H}, \\ {\rm D-153^{H}, {\rm F-8^{H}, P-9^{H}, {\rm K-12^{H}}. } \end{array}$		
Helicase 25	52 ^H , Y—132 ^H , G-56 ^H , R-55 ^H 2 + HB1a//Helicase + HB1b		100	D-155 ^H , Y-154 ^H K-12 ^H , P-9 ^H , D-155 ^H , Y-154 ^H , F-8 ^H , D-153 ^H , I-152 ^H , R-298 ^H , V-297 ^H , F-294 ^H , N-151 ^H , G-		

F-8^H, D-153^H, I-152^H, R-298^H, V-297^H, N-151^H, F-294^H//G-302^H, V-303^H

(continued on next page)

Table 6 (continued)

Time (ns)	Total interactions	Interactions in chemical group ^a
	$\begin{array}{l} 302^{\rm H}, S{=}301^{\rm H}, C{=}300^{\rm H}, V{=}303^{\rm H}, \\ S{=}1^{\rm H}, Q{=}299^{\rm H}, Y{=}154^{\rm H}, D{=}155^{\rm H}, \\ K{=}12^{\rm H}, P{=}9^{\rm H}//S{=}1^{\rm H}, V{=}303^{\rm H}, \\ C{=}300^{\rm H}, Q{=}299^{\rm H}, S{=}301^{\rm H}, G{=}2^{\rm H}, \\ G{=}302^{\rm H}, N{=}214^{\rm H}, P{=}9^{\rm H}, R{=}298^{\rm H}, \\ F{=}8^{\rm H}, M{=}6^{\rm H}, A{=}7^{\rm H}, Y{=}154^{\rm H} \end{array}$	

The interactions with the differential chemical group of each type of avermectin are also shown.^{HB}: Hydrogen Bonds; ^H: Hydrophobic Interactions; A: Alanine; R: Arginine; N: Asparagine; D: Aspartic Acid; C: Cysteine; E: Glutaminic Acid; Q: Glutamine; G: Glycine; H: Histidine; I: Isoleucine; L: Leucine; K: Lysine; M: Methionine; F: Phenylalanine; P: Proline; S: Serine; T: Threonine; W: Tryptophan; Y: Tyrosine; V: Valine.

^a HB1a is sec-butyl and HB1b is isopropyl derivative of ivermectin, respectively.

binding affinities based on structural signatures and the force constant relative to the initial frame during the 100 ns MD simulation of each homolog in complex with IMP α 1 (Fig. 7A and B), IMP β 1 (Fig. 7C and D), Helicase (Fig. 7E and F), M^{pro} (Fig. 7G and H).

The simulated complex plot is presented using the CSM-Lig algorithm [34] designed to predict the binding affinity of a protein-small



molecule complex based on structural signatures (Fig. 7A, C, E and G), and the WebPSN algorithm for force constant (Fig. 7B, D, F and H) using two alternative versions of ENM to calculate the cross-correlation of the movement of C α atoms and for the pairwise interactions between the C α atoms (linear cut-ENM and Kovacs-ENM) [35].

The results of affinity and binding energy of the CSM-Lig and WebPSN algorithms (see Fig. 7), respectively, had similar behaviors for each complex, and suggest that the compound HB1a binds more strongly to the two importins studied, while HB1b binds more strongly to viral proteins (Helicase and Mpro), with a binding force of between 4 and 5 kcal/mol*Å² above 50 ns according to WebPSN. The strength of the interaction is often based on geometric criteria (for example, distance cutoff, in this study was 4.5 Å) [35] and it is well known that a ligand stably bound to its target tends to localize at short distances and energetically favored [98]. In fact, the union between two atoms is determined by the energy produced by the interaction, which is affected by many variables, but mainly by the distance between the two atoms, so the distance used was the one suggested within the limit of 4–6 Å between heavy atoms for optimal performance [99].

As various docking methods were used to provide a consensus result to predict the preferred homologous ivermectin targets, the MD simulations provided an opportunity to potentially produce better estimates of the binding affinity values for these complexes. In this sense, the MM/





B

Fig. 6. Total number of hydrophobic interactions and hydrogen bonding between each homolog (A and C) and its differential chemical group (B and D) with the proteins considered, monitored throughout the simulated trajectory of 100 ns, respectively. The lifespan of the hydrogen bonds and the hydrophobic interactions of the minimum energy structure of each complex were represented as absolute numbers of interactions at each specific time (in nanoseconds) indicated on the X-axis.



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Time (ns)

Force (kcal/mol·Å2)

Time (ns)

9 10

Time (ns)

Time (ns)

Fig. 7. Predicted binding affinities based on structural signatures and the force constant relative to the starting frame during 100 ns MD simulation of each homolog in complex with IMP α 1 (A and B), IMP β 1 (C and D), Helicase (E and F), M^{pro} (G and H).

Time (ns)

PBSA prediction (see Fig. 8) was considered as a thermodynamic integration method, to requalify the complexes, taking into account multiple structures obtained from the MD trajectories. For the free energy binding (ΔG_{bind}) of the MD trajectory, the widely used molecular mechanics/ Poisson-Boltzmann surface area (MM/PBSA) was adapted [36–41].

Stable binding of each homolog was predicted in the complexes according to MM/PBSA (see Fig. 8). The thermodynamic energy mean for IMP α 1 + HB1a was -71 kcal/mol, and for IMP α 1 + HB1b it was -100 kcal/mol (Fig. 8A). The residues with the highest energy contribution were W-184 (-18.8 kcal/mol) and N-228 (-12.3 kcal/mol) for the union of HB1a and HB1b with IMP α 1, respectively. These residues established hydrophobic interactions and hydrogen bonds with each homolog (see Tables 3 and 6). For IMP β 1 + HB1a and IMP β 1 + HB1b it was -36 kcal/mol and -59 kcal/mol, respectively (Fig. 8B). The residue with the highest energy contribution was K-62 for the union of HB1a (-24.3 kcal/mol) and HB1b (-18.0 kcal/mol) with IMP β 1. This residue established hydrophobic interactions and hydrogen bonds with each homolog and in both cases the union is established in the differential chemical group of each avermectin (see Tables 3 and 6).

For the Helicase + HB1a and Helicase + HB1a complex -83 kcal/ mol and - 79 kcal/mol, respectively (Fig. 8C), the residue with the highest energy contribution was E-261 for the union of HB1a (-13.2 kcal/mol) and HB1b (-15.4 kcal/mol) with the Helicase. This residue established hydrophobic interactions and hydrogen bonds with each homolog and in both cases the union is established in the differential chemical group of each avermectin (see Tables 3 and 6). As for M^{pro} + HB1a and M^{pro} + HB1b it was -72 kcal/mol and - 68 kcal/mol, correspondingly (Fig. 8D). The residue with the highest energy contribution was Y-154 for the union of HB1a (-24.2 kcal/mol) and HB1b (-25.0 kcal/mol) with M^{pro} , this residue established hydrophobic interactions and hydrogen bonds with HB1a, and only hydrophobic interactions with HB1b. The differential chemical group of HB1a was the only one that interacted with this residue (see Tables 3 and 6). A stable MM/PBSA-based binding of ivermectin (only for the HB1a) against the proteins considered has already been reported [36,100].

3.5. Comparative study between conformational fluctuations of proteinavermectin complexes

The Z-_{Score} allowed to observe the conformational fluctuation between the native structures and the dynamized ligand-protein complexes, measuring the deviation of the total energy of the structures with respect to their energy distribution derived from the contrast with other native random conformations. Starting from this model, it was observed that all the avermectin-protein complexes presented differences in the distances of their C α atoms, as well as in their energetics at 100 ns of simulation and with respect to their respective native structures subjected to the same dynamic conditions.

The IMP α 1 + homolog complex presented the most distant conformational fluctuation from the native one with a Z-_{Score} more negative than that obtained for protein free of ligand, which suggest unfolding of this protein in both cases, as occurred with the M^{pro} protein. In contrast, the results obtained for IMP β 1 suggest refolding or compactation of structure in all cases, as for the case of the Helicase, the results suggest that both homologs induced refolding of this protein (Table 7; and Supplementary Fig. 2 and Supplementary Fig. 3). The scores obtained by the ProSA server for all structures validate that the predicted complexes, although with values within the graph and close to zero as described for native models [101–103], differ from the native structures without ligands tested here and therefore show conformational changes after



Fig. 8. Predicted Binding energy (kcal/mol) from MM/PBSA relative to the starting frame during 100 ns MD simulation of each homolog in complex with IMPα1 (A), IMPβ1 (B), Helicase (C) and M^{pro} (D).

Table 7

Comparison between Z-score and radius of gyration in relation to the minimum energy structure at 100 ns MD simulation for each complex.

Complexes	Simulation time	Simulation time (100 ns)					
	Z_Score	Rg ^a	R_g^{b}				
ΙΜΡα1	-8.50	21.89	21.93				
$IMP\alpha 1 + HB1a$	-12.20	28.12	23.92				
$IMP\alpha 1 + HB1b$	-10.38	28.64	23.92				
IMP _β 1	-5.93	14.92	14.92				
$IMP\beta1 + HB1a$	-4.87	15.48	14.72				
$IMP\beta 1 + HB1b$	-4.72	15.17	14.67				
Helicase	-5.27	32.02	31.97				
Helicase + HB1a	-5.03	31.34	31.32				
Helicase + HB1b	-5.24	32.14	32.09				
M ^{pro}	-6.17	21.56	21.62				
$M^{pro} + HB1a$	-6.50	21.71	21.72				
$M^{pro} + HB1b$	-6.28	22.02	22.09				

^a Radius of gyration (R_o) with WAXSiS server.

 $^{\rm b}$ Radius of gyration $({\rm R}_g)$ with HullRad; HB1a, avermectin B1a; HB1b, avermectin B1b.

simulations, a phenomenon that was observed more consistently between the HB1b with most of the coupled proteins.

The radius of gyration R_g is a measure of the compactness of a protein which allows to understand its folding properties. Small R_g values indicate a tight packing whereas high R_g values show a floppy packing, a relative constant R_g value through time indicates that the ligands hold the folding behavior of the protein whereas abrupt fluctuations of the R_g values denote protein folding instability [95]. All the methods predicted conformational fluctuations in the studied complexes oriented towards the folding or unfolding of the complexes. Observations dependent on both the type of protein system and the homolog considered (Table 7).

For IMP α 1, HB1a and HB1b induced an unfolded conformation of the protein, with an average difference between both homologs and the native structure of $R_g \approx 4$ Å, with mean values for each homolog of $R_g \approx 4.11$ Å and $R_g \approx 4.37$ Å, for complex with HB1a and HB1b, respectively (Table 7). Results that would allow us to infer, together with those of the

atomic distance in terms of RMSD, that $IMP\alpha 1$ presents an almost "sphero-cylindrical" folded structural state, different from the native globular state (Supplementary Fig. 4). Similar results have already been reported in $IMP\alpha$ systems in the presence of HB1a [4], as the $IMP\beta 1$ protein shown a slight unfolded with fluctuating values between each homolog.

The mean difference between both complexes and the native structure was $R_g \approx 0.45$ Å, with mean values R_g for the complex with HB1a and HB1b of $R_g \approx 0.54$ Å and $R_g \approx 0.36$ Å, respectively (Table 7). This corresponds to the observations at the RMSD level at which evidenced that none of the homologs was capable of inducing important differential perturbations compared to the native structure, generating a folded globular-type structural state in both cases. These results show a slightly greater effect of unfolding HB1b on IMP α 1 and folding on IMP β 1, compared to HB1a, as well as a differential behavior of each homolog against importins tested in terms of induction of conformational changes.

In the Helicase, the HB1b induced a slightly unfolded conformation of this viral protein, the HB1a induced slightly refolding of this protein (Table 7). This is related to the atomic distance, which show no important or significant variations between these compounds compared to the native structure of the Helicase, inducing a globular folded conformational state with both homologs corresponding to data previously reported [4]. For M^{pro} the effect of both homologs was oriented towards conformational unfolding, with a mean difference between both homologs and the native structure was $R_g \approx 0.29$ Å. Specifically, the mean values of the difference between each homolog were $R_g \approx 0.12$ Å and $R_g \approx 0.46$ Å, for HB1a and HB1b, respectively (Table 7).

Fig. 9 shows a summary of the main differences in the interaction of avermectin-B1a (HB1a) and avermectin-B1b (HB1b) with the selected targets. The binding of each homolog was stable in all cases over time in terms of RMSD, RMSF, relative binding energy and according to the MM/PBSA term. The differences presented are mainly focused on the discriminant values of the binding energies and the potential theoretical inhibitory effect on the kinetics of the compounds HB1a and HB1b on the cellular and viral proteins studied, respectively. The differential



Fig. 9. Comparison of the observed differential characteristics of each homolog in terms of docking energy, dynamic, and potential theoretical inhibitory activity. ^aglobal mean binding free energy through AutoDock, AutoDock Vina, AutoDock Vina incremental, COACH-D and DockThor algorithms); ^breported global mean binding free energy (prediction from energy mean calculated with similar algorithms); nr, no report.

activity of compound HB1b, which is favorable against the viral proteins studied, can be highlighted more appropriately. The fact that there are no theoretical reports of the interaction of compound HB1b with most of the structures considered in this study is also highlighted.

These theoretical predictions could guide the possibility of deriving formulations or analogues that can be administered to achieve relevant therapeutic concentrations, as it has been suggested necessary to do in the case of ivermectin [8,81]. It is important to say that these predictions based on theoretical models cannot predict a relevant clinical outcome and do not consider important variables such as time of exposure to the compound, cytoplasmic crowding (a phenomenon explored preliminarily in this study), among other aspects, which are key to a more adequate prediction of the kinetic qualities of each of the homologs. Furthermore, since it is a superfamily of proteins, the study of a greater number of importins is recommended, as well as a greater number of possible targets associated with SARS-CoV-2 under the same conditions of this study. It would be interesting to compare our results with experimental analyzes that consider each homolog in isolation, or variations in the proportion of each constituent of the ivermectin mixture.

4. Conclusions

There are several studies on possible targets for the controversial ivermectin, but there are few comparative reports on the behavior of avermectin homologs. Our objective was to investigate from a biophysical and computational chemical perspective the ligand-protein interactions, and the effect of these interactions on the kinetics and dynamics of the complexes predicted with each homolog. In this sense, after analysis of molecular dynamics and docking, as well as inhibitory kinetics, hydrophobicity studies, interactions between residues, RMSD, RMSF, Rg, ROG and preliminary results under crowding, we found that each of the homologs of ivermectin could establish thermodynamically favorable dockings with each of the proteins tested in this study. Each homolog produced different changes in the thermodynamic stability of the complexes, affecting the degrees of freedom of energy transitions by mediating minimum energy conformations, with fluctuations different from those of the free proteins. We also found at a global level a theoretical individual affinity of the HB1b for the viral structures and of the HB1a for the host structures. This corresponds to the predicted inhibition kinetics which, in turn, is influenced by the hydrophobicity of the cavities in the binding pockets and by the affinity of the differential chemical groups of each homolog. The theoretical behavior of these homologs could contribute to the possible reported multi-target ivermectin activity. However, it is necessary to carry out experimental demonstrations to corroborate this differential behavior, as well as its clinical relevance.

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Author's Contributions

Lenin González-Paz: Conceptualization, Data curation, Formal analysis, Investigation, Methodology, Writing – original draft. María Laura Hurtado-León: Investigation. Carla Lossada: Investigation. Francelys V. Fernández-Materán: Investigation. Joan Vera-Villalobos: Supervision, Validation. Marcos Loroño: Writing – original draft. JLPaz: Supervision, Validation, Writing – original draft. Laura Jeffreys: Writing – review & editing. Ysaias J. Alvarado: Conceptualization, Data curation, Formal analysis, Investigation, Methodology, Writing – original draft.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

Appendix A. Supplementary data

Supplementary data to this article can be found online at https://doi.org/10.1016/j.bpc.2021.106677.

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