








# Fabrication and in vitro Evaluation of 4-HIA Encapsulated PLGA Nanoparticles on PC12 Cells

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**Purpose:** 4-Hydroxyisophthalic acid (4-HIA) is a bioactive compound present in the roots of *Decalepis hamiltonii*, which has attracted considerable attention in attenuating oxidative stress-related neurodegenerative diseases. However, its efficacy is limited because of its low solubility and bioavailability. Therefore, the present study aimed to encapsulate 4-HIA using biocompatible copolymer polylactide-co-glycolide (PLGA) and evaluate its antioxidant and neuroprotective potential.

**Methods:** The nanoparticles (NPs) were fabricated by solid/oil/water (s/o/w) emulsion technique and characterized using XRD, SEM, HR-TEM, and FTIR spectroscopy. Antioxidant assays such as 1,1 diphenyl-2-picrylhydrazyl (DPPH), superoxide, and hydroxyl radical scavenging ability were performed to assess the antioxidant potential of the fabricated NPs.

**Results:** The bioactive component, 4-HIA, was efficiently encapsulated by the PLGA polymer and was found to be spherical and smooth with a size <10nm. 4-HIA showed better scavenging capability in DPPH and superoxide assays as compared to 4-HIA encapsulated PLGA and butylated hydroxytoluene (BHT). In contrast, 4-HIA encapsulated PLGA NPs exhibited a significant hydroxyl radical scavenging activity than 4-HIA and BHT alone. Further, the encapsulated NPs efficiently curtailed hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>)-induced cytotoxicity in PC12 cells.

**Conclusion:** Our findings indicate that 4-HIA encapsulated PLGA NPs might be a therapeutic intervention towards the effective management of oxidative stress as it has exhibited efficient neuroprotective potential against H<sub>2</sub>O<sub>2</sub>-induced oxidative stress in PC12 cells.

**Keywords:** 4-Hydroxyisophthalic acid, polylactic-co-glycolic acid, nanoencapsulation, PC12 cells, cytotoxicity

## Introduction

Plant extracts and their bioactive constituents are used as potential therapeutics for the treatment of various diseases.<sup>1</sup> *Decalepis hamiltonii* Wight & Arn. is a climbing shrub belonging to the family Apocynaceae and commonly called Makali beru in South India.<sup>2</sup> The root extract of *D. hamiltonii* is an important source of phenolic compounds.<sup>3</sup> 4-Hydroxyisophthalic acid (4-HIA), a novel antioxidant compound and one of the major phenolic constituents in *D. hamiltonii* extract, that exhibits potent-free radical quenching ability, cytoprotective and cardioprotective activity.<sup>4,5</sup> Our earlier studies and others, revealed the neuroprotective potential of 4-HIA in ameliorating oxidative stress in the aging rat brain and neurodegeneration in drosophila.<sup>6,7</sup> Poor solubility and low bioavailability of 4-HIA presents a challenge

to exhibit neuroprotection because of its reduced ability to permeate the blood–brain barrier (BBB) (Sowbhagya et al unpublished data).

Nanoformulations are an alternative platform of drug delivery system, which improves the cellular uptake of hydrophilic and hydrophobic compounds, together with shelf-life. Polymer-based nanofabrication has fascinated the scientific community because of their diverse application in biomedical sciences to treat neurodegenerative and cardiovascular diseases.<sup>8</sup> Recently, synthetic and natural polymers like polyethylene glycol (PEG), polylactic-co-glycolic acid (PLGA), chitosan, and polysaccharides have been approved by the Food and Drug Administration (FDA) of the United States and employed extensively in designing nanoscale materials. Amongst all these, PLGA, due to its enhanced biodegradability and biocompatibility have been widely used as drug delivery systems like microspheres, pellets, films, matrixes, blends, and nanoparticles (NPs).<sup>9,10</sup> Besides, PLGA have been approved for 14 pharmaceutical and biomedical properties by the FDA.<sup>11</sup> Several studies have documented that PLGA loaded NPs has offered enhanced antioxidant and cytotoxic activities.<sup>12,13</sup>

The therapeutic targets are localized in the central nervous system, limiting the passage of drugs to cross the BBB. Hence, 4-HIA encapsulated PLGA NPs may serve as a promising alternative for neuroprotection. Based on these facts, the present study was designed to fabricate 4-HIA-PLGA NPs, and investigate their antioxidant and neuroprotective potential against H<sub>2</sub>O<sub>2</sub>-induced toxicity in PC12 cells.

## Materials and Methods

### Chemicals and Reagents

PLGA, 1.1-diphenyl-2-picrylhydrazyl (DPPH), 4-hydroxyisophthalic acid (4-HIA) were obtained from Sigma Aldrich (St. Louis, Mo, USA.). Ham's FK12 medium, fetal bovine serum (FBS), L-glutamine, antibiotic-antimycotic solution, 3(4,5-dimethylthiazol-2)-2, 5-diphenyl tetrazolium bromide (MTT) were procured from Himedia (India). All organic solvents (spectral grade) and general chemicals (analytical grade) were purchased from local companies.

### Preparation of 4-HIA Encapsulated PLGA NPs

PLGA NPs were synthesized by the solid-in-oil-in-water (s/o/w) emulsion method.<sup>14</sup> 45 mg of PLGA with a 50:50

ratio of lactic to glycolic acid was mixed with dichloromethane for 6 h to attain a uniform concentration of PLGA. 5mg of 4-HIA was then suspended to PLGA solution and sonicated for 1 min at 55 W to obtain solid in oil primary emulsion (organic phase). This primary emulsion was sonicated again at 55 W for 2 min after the addition of 20 mL polyvinyl alcohol (PVA, 1% w/v) solution. The resultant nano-sized particles were agitated in the emulsion for 3 h for the evaporation of the solvent. The obtained secondary emulsion was centrifuged for 15 min at 15,000 g. These newly synthesized NPs were washed thrice using deionized distilled water and were resuspended finally in deionized water and dried with the help of a lyophilizer. The synthesized NPs were kept at 4°C for further experiments.

## Characterization of NPs

### Percentage Yield of the NPs

The percentage yield was determined as the ratio between the weight of dried 4-HIA encapsulated PLGA NPs and the total weight of utilized 4-HIA and PLGA. The yield was calculated using the given formula.<sup>15</sup>

$$\text{Percentage yield} = \frac{\text{Weight of Nanoparticle obtained}}{\text{Weight of 4-HIA and PLGA used for nanoparticle synthesis}} \times 100$$

### X-Ray Diffraction (XRD) Measurement

The X-ray diffractograms of 4-HIA, 4-HIA encapsulated PLGA and blank PLGA NPs samples (1mg) were grounded with the help of mortar and pestle. The obtained powdered samples were packed evenly into a cavity glass slide. The glass slide was placed in a sample holder, and readings were recorded in an X-ray diffractometer (Shimadzu XRD 7000 maxima) using Ni-filtered CuK $\alpha$  radiation (35 kV, 15 mA)

### Scanning Electron Microscopy (SEM)

SEM (TM 3000, Hitachi, Japan) was employed to study the surface morphology of 4-HIA, 4-HIA encapsulated PLGA, and blank PLGA. The 4-HIA, 4-HIA encapsulated PLGA, and standard PLGA NPs were mounted on an aluminium stub with double-sided carbon tape. The images at desired magnification were captured on SEM mode.

### High-Resolution Transmission Electron Microscopy (HR-TEM)

The size distribution and surface morphology of the NPs, along with their selected area electron diffusion (SAED)

patterns, were assessed by HR-TEM (JEOL JEM 2100, Hitachi, Japan). The samples (1mg/mL) were dispersed in HPLC-grade water by means of an ultrasonic bath. The suspensions were dropped onto 0.037 mm copper grids, and excess water is sucked onto the sterile filter paper and air-dried before viewing under TEM.

### Spectroscopic Characterization

The Fourier transform Infrared (FTIR) spectra of 4-HIA, 4-HIA encapsulated PLGA and blank PLGA NPs were obtained on Cary 630 FTIR (Agilent Technologies) equipped with a smart iTR diamond ATR device. Samples (1 mg) were crushed in a sterile mortar and pestle. The obtained fine powdered samples were loaded onto the mounting plate so as to cover the exposed surface of the crystal using a clean spatula. The pressure arm is swung over the top of the sample, and the knob rotated until it just touches the sample. Spectra were measured in the 400–4000  $\text{cm}^{-1}$  scanning range at a resolution of 1  $\text{cm}^{-1}$ .

## Antioxidant Assays

### DPPH Radical Scavenging Activity

DPPH assay was accomplished following the method of Shon et al with slight modifications.<sup>16</sup> Briefly, 0.5 mL of 4-HIA and 4-HIA encapsulated PLGA each was added to 1 mL of DPPH solution (0.1 mM, in 95% ethanol) and incubated at RT for 20 min. The absorbance was recorded at 517 nm against a suitable blank.

$$\text{Scavenging effect (\%)} = \frac{\text{Abs}_{\text{Control}(517)} - \text{Abs}_{\text{Sample}(517)}}{\text{Abs}_{\text{Control}(517)}} \times 100$$

Where  $\text{Abs}_{\text{control}}$  = absorbance of the control (reacting mixture without the test sample) and  $\text{Abs}_{\text{sample}}$  = absorbance of reacting mixture with the test sample.

### Superoxide Radical Scavenging Activity

Superoxide radical scavenging assay was performed according to the procedure of Nishikimi et al.<sup>17</sup> About 1 mL of nitroblue tetrazolium (NBT) solution (156  $\mu\text{M}$ ), 1 mL of nicotinamide adenine dinucleotide (NADH) solution (468  $\mu\text{M}$ ) were added to 4-HIA, and 4-HIA encapsulated PLGA NPs. The reaction was started by the addition of 100  $\mu\text{L}$  of phenazine methosulfate (PMS) solution (60  $\mu\text{M}$  PMS). Each reagent was prepared separately in 100 mM phosphate buffer at pH 7.4. The reaction mixture was incubated for 5 min at 25°C, and the absorbance was observed at 560nm.

$$\text{Scavenging effect(\%)} = \frac{\text{Abs}_{\text{Control}(560)} - \text{Abs}_{\text{Sample}(560)}}{\text{Abs}_{\text{Control}(560)}} \times 100$$

Where  $\text{Abs}_{\text{control}}$  = absorbance of the control (reacting mixture without the test sample) and  $\text{Abs}_{\text{sample}}$  = absorbance of reacting mixture with the test sample.

### Hydroxyl Radical Scavenging Activity

The protocol of Halliwell and Gutteridge was adopted to carry out the hydroxyl radical scavenging assay.<sup>18</sup> Briefly, different concentrations of 20  $\mu\text{L}$  of 4-HIA and 4-HIA encapsulated PLGA were added to a reaction mixture containing 120  $\mu\text{L}$  of 20 mM deoxyribose, 400  $\mu\text{L}$  of phosphate buffer (0.1 M), 40  $\mu\text{L}$  of 20 mM  $\text{H}_2\text{O}_2$ , and 40  $\mu\text{L}$  of 500  $\mu\text{M}$   $\text{FeSO}_4$ . The final volume was made up to 800  $\mu\text{L}$  with distilled water. The resultant mixture was incubated for 30 min at 37°C. After the incubation, the reaction was discontinued by adding 500  $\mu\text{L}$  of 2.8% trichloroacetic acid (TCA) followed by 400  $\mu\text{L}$  of thiobarbituric acid (TBA) solution (0.6%). The tubes were then incubated for 20 min in a boiling water bath, and the absorbance was recorded at 532 nm. The percentage of hydroxyl radical scavenging activity was obtained using the following formula.

$$\text{Scavenging effect(\%)} = \frac{\text{Abs}_{\text{Control}(532)} - \text{Abs}_{\text{Sample}(532)}}{\text{Abs}_{\text{Control}(532)}} \times 100$$

Where  $\text{Abs}_{\text{control}}$  = absorbance of the control (reacting mixture without the test sample) and  $\text{Abs}_{\text{sample}}$  = absorbance of reacting mixture with the test sample.

## Cell Culture and Drug Treatment

Undifferentiated PC12 cells were procured from National Center for Cell Science, Pune, India and were maintained in Ham's F12K medium. The medium was supplemented with 15% FBS, 1% antibiotic-antimycotic solution, and 2mM L-glutamine solution at 37°C in a humidified atmosphere (5%  $\text{CO}_2$ ). The cells were cultured in poly-L-lysine coated culture flasks, and the medium was changed every three days. Light microscopy was used to check the confluency of the viable growing cells. Upon attaining cell density at 70–80% confluency, the cells were dislodged with 0.25% trypsin, and cells were seeded in an appropriate density following the experimental protocol. The viability of PC12 cells exposed to  $\text{H}_2\text{O}_2$  was measured by treating the

cells with increasing concentrations ranging from 100 to 1000  $\mu\text{M}$  for 24 h to find the  $\text{IC}_{50}$  value. To evaluate the protective effects of PLGA encapsulated NPs, cells were pre-treated with 4-HIA, and 4-HIA encapsulated PLGA NPs (50–300  $\mu\text{g}/\text{mL}$ ) for 24 h prior to the exposure to  $\text{H}_2\text{O}_2$  for 24 h. Cell viability was measured by MTT assay as mentioned below.

## MTT Assay

MTT assay was carried out to observe the  $\text{H}_2\text{O}_2$ -induced cell toxicity, which is a sensitive method to evaluate the metabolic status of the cells.<sup>19</sup> Initially, cultured PC12 cells were plated in triplicates at a cell density of  $1 \times 10^6$  cells/well for 24 h in a 96-well plate. The cells were treated with different concentrations of 4-HIA, and 4-HIA encapsulated PLGA for 24 h before the exposure of 500  $\mu\text{M}$   $\text{H}_2\text{O}_2$  for 24 h. 20  $\mu\text{L}$  MTT solution (0.25%) was added at the indicated time points to the cells and incubated at 37°C for 4 h. 100  $\mu\text{L}$  DMSO was added after discarding the medium. The absorbance was determined at 570 nm (Model-3550, Bio-Rad). Experiments were carried out in triplicates. The results were

expressed as the percentage of absorbance against the non-treated control cells.

## Statistical Analysis

Results are represented as mean  $\pm$  SE ( $n=3$ ). One-way and two-way ANOVA were performed, followed by Tukey's test for comparing the differences between means using SPSS 16.0 software package. A probability value of  $p < 0.05$  was considered statistically significant.

## Results

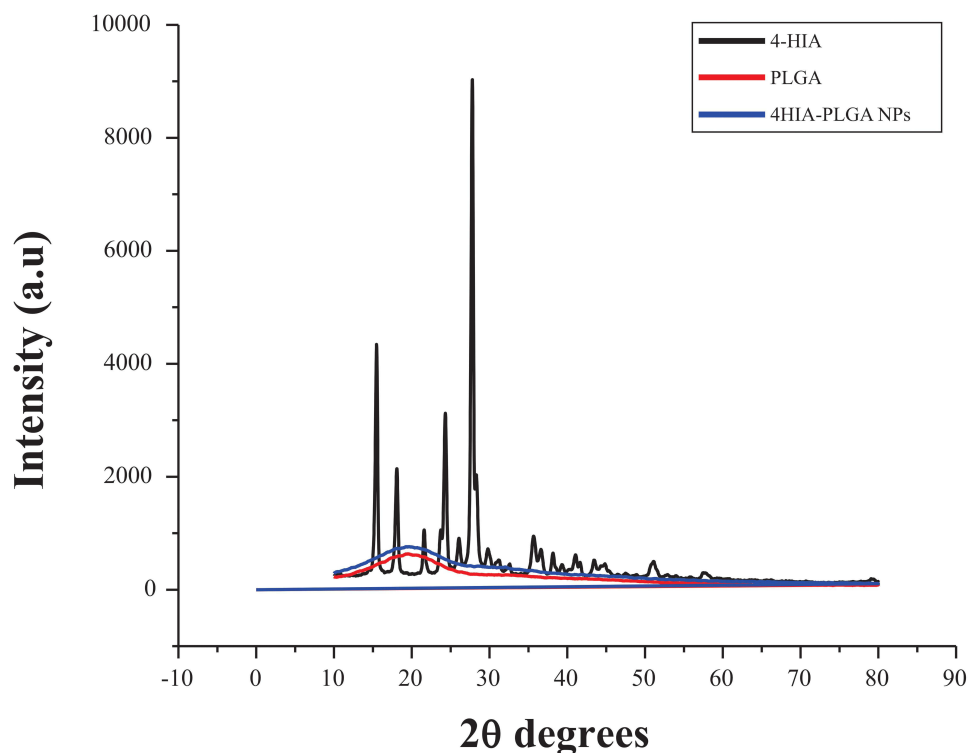
### Synthesis and Characterization of NPs

#### Yield of NPs

4-HIA encapsulated PLGA NPs were effectively synthesized using solvent evaporation process, and its consistency was evaluated for various batches. The yield of the 4-HIA encapsulated PLGA NPs were found to be 31%.

#### X-Ray Diffraction (XRD) Measurement

The XRD patterns for 4-HIA, blank PLGA, and 4-HIA encapsulated PLGA NPs are represented in Figure 1.



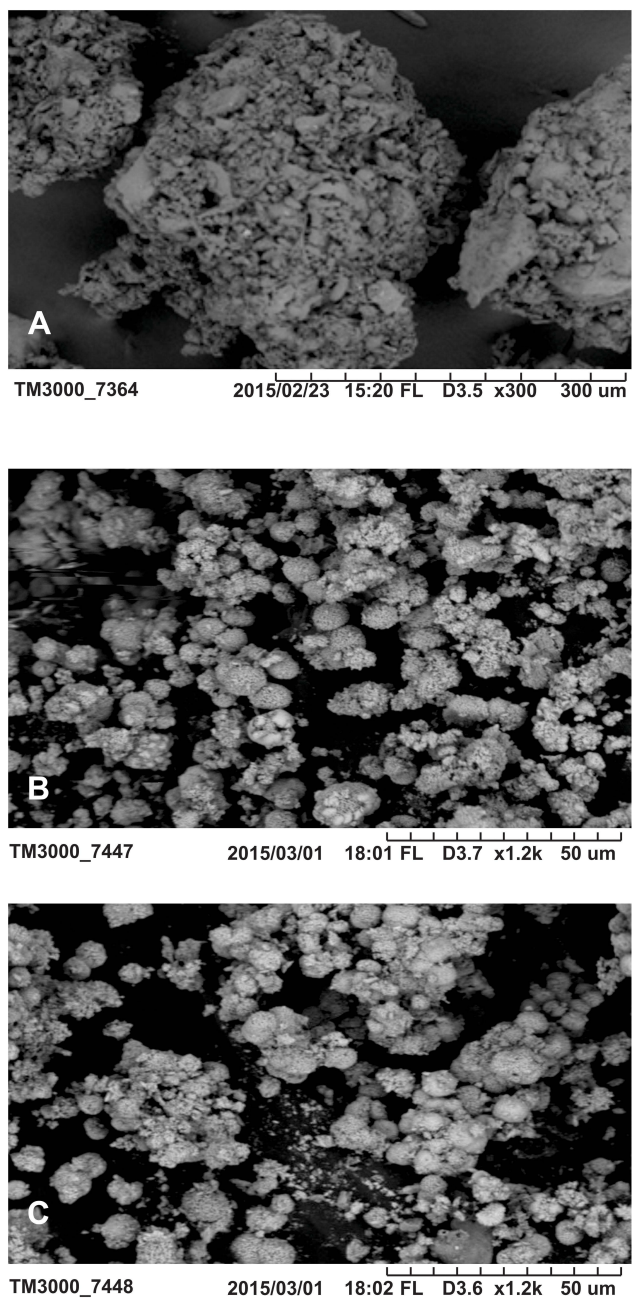
**Figure 1** XRD spectra of 4-HIA, PLGA, 4-HIA encapsulated PLGA nanoparticles.



The diffraction pattern peaks at 15, 18, 22, 25, and 28 depict the crystalline nature of 4-HIA. Such characteristic peaks were not noticed in PLGA, and 4-HIA encapsulated PLGA NPs due to their amorphous nature.

### SEM Analysis

The SEM images of 4-HIA, blank PLGA, and 4-HIA encapsulated PLGA NPs are represented in Figure 2. The images revealed that blank PLGA and 4-HIA



**Figure 2** SEM images of (A) 4-HIA (B) PLGA and (C) 4-HIA encapsulated PLGA nanoparticles.

encapsulated PLGA NPs were spherical with a smooth surface, whereas 4-HIA crystals showed irregular shape and pattern.

### HR-TEM Analysis

The size of the 4-HIA encapsulated PLGA NPs was found to be < 10 nm. HRTEM imaging was employed to elucidate the morphology of 4-HIA, and 4-HIA encapsulated PLGA NPs and the images are shown in Figure 3A and C. 4-HIA encapsulated PLGA NPs were spherical and uniformly encapsulated within the PLGA matrix. The structural characteristics were further elucidated by selected area electron diffraction (SAED) patterns. The SAED pattern reveals the amorphous nature of 4-HIA encapsulated PLGA NPs where the diffraction rings are blurred (Figure 3B). While Figure 3D represents the crystalline nature of 4-HIA. The patterns of SAED for crystalline 4-HIA were indexed accordingly based on their *d*-spacings which were found to be 0.3596nm, 0.2287nm, 0.1954nm, 0.1688nm, 0.1320nm, 0.1137nm, 0.1017nm.

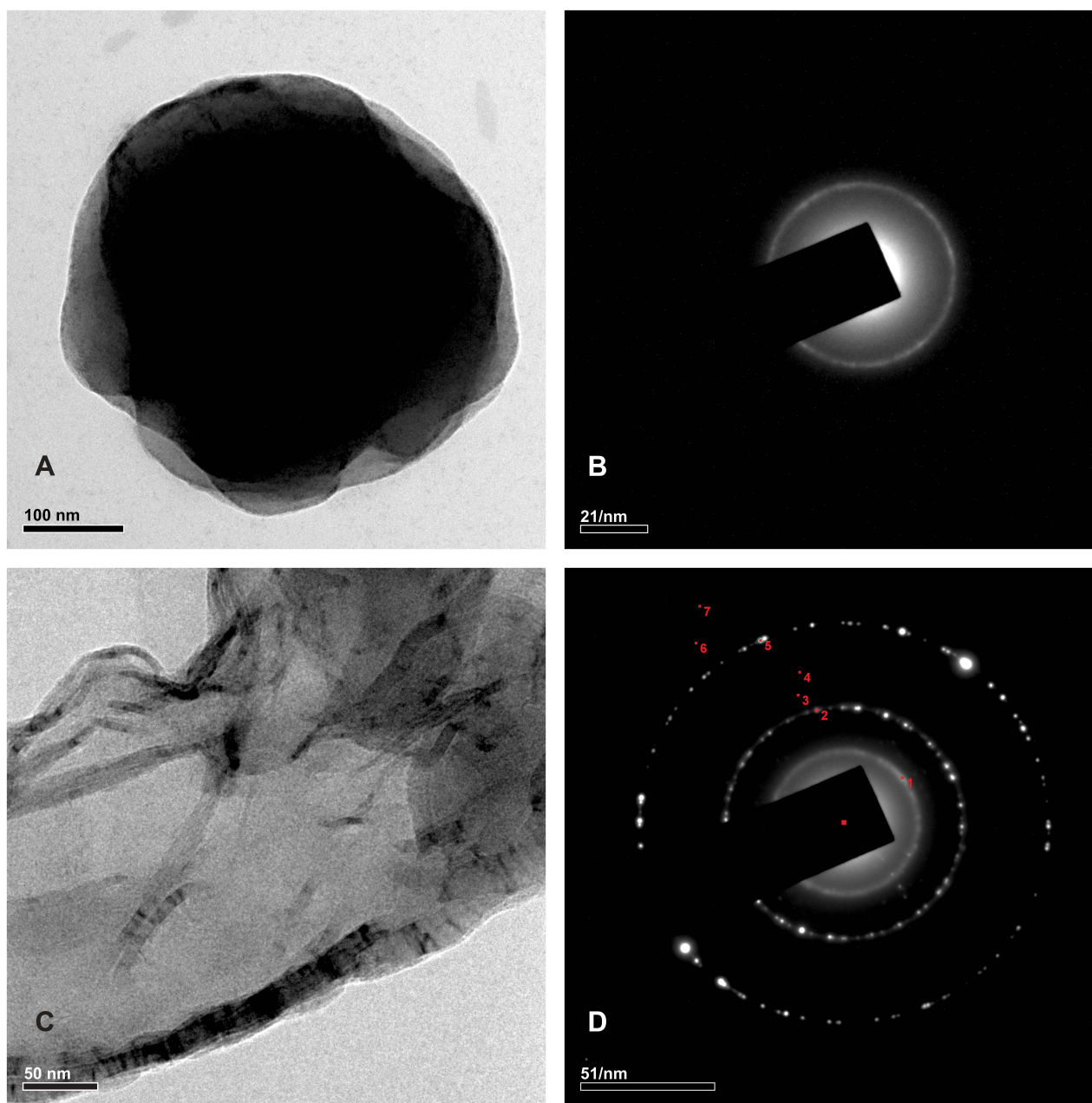
### Spectroscopic Characterization

The intermolecular interaction between blank PLGA, 4-HIA, and 4-HIA encapsulated PLGA NPs were determined by FTIR spectroscopy. The characteristic spectra of the 4-HIA indicated carbonyl stretch (C=O) at 1649.32  $\text{cm}^{-1}$  and C=C stretch at 1582.26  $\text{cm}^{-1}$  (Figure 4A). The characteristic spectra of PLGA as shown in Figure 4B indicated  $-\text{CH}_2-\text{CH}_2-\text{CH}_3-$  stretching (2800–3000 $\text{cm}^{-1}$ ), C=O stretch (1700–1800  $\text{cm}^{-1}$ ), C-H stretch at 2918.927  $\text{cm}^{-1}$ , and C-O stretch at 1700–1800 $\text{cm}^{-1}$ . 4-HIA encapsulated PLGA NPs exhibited similar characteristic spectra as that of PLGA with C=O stretch at 1747.71 $\text{cm}^{-1}$ , O-H stretch at 2949.15  $\text{cm}^{-1}$ , and 2995.65  $\text{cm}^{-1}$  (Figure 4C).

## Antioxidant Assays

### DPPH Assay

The DPPH radical scavenging ability is represented in Figure 5. The 4-HIA and 4-HIA encapsulated PLGA NPs exhibited concentration-dependent antiradical activity. The 4-HIA was found to be significantly more effective free radical scavenging agent as evident by its lower  $\text{IC}_{50}$  value of  $239.3 \pm 2.3 \mu\text{g/mL}$  against BHT. 4-HIA encapsulated PLGA ( $59 \pm 0.5 \mu\text{g/mL}$ ) also revealed significant scavenging



**Figure 3** HR-TEM images of (A and C) 4-HIA encapsulated PLGA nanoparticles and 4-HIA. Selected area electron diffraction patterns of (B and D) 4-HIA encapsulated PLGA nanoparticles and 4-HIA.

activity compared to the BHT ( $54 \pm 0.3 \mu\text{g/mL}$ ) at  $1000 \mu\text{g/mL}$ .

#### Superoxide Radical Scavenging Activity

The scavenging activity revealed a significant dose-dependent relationship with 4-HIA exhibiting maximum scavenging activity (64%) followed by 4-HIA encapsulated PLGA NPs (58%) and BHT (53.5%) (Figure 6). The  $\text{IC}_{50}$  values of 4-HIA and 4-HIA encapsulated

PLGA NPs were found to be  $314.3 \pm 1.4 \mu\text{g/mL}$  and  $386.5 \pm 1.8 \mu\text{g/mL}$ .

#### Hydroxyl Radical-Scavenging Activity

The significant feature of the hydroxyl radical scavenging activity is that 4-HIA encapsulated PLGA NPs exhibited higher inhibition (76%) compared to 4-HIA (57%) and BHT (74.5%) with an  $\text{IC}_{50}$  value of  $453.7 \pm 2.2 \mu\text{g/mL}$  (Figure 7).

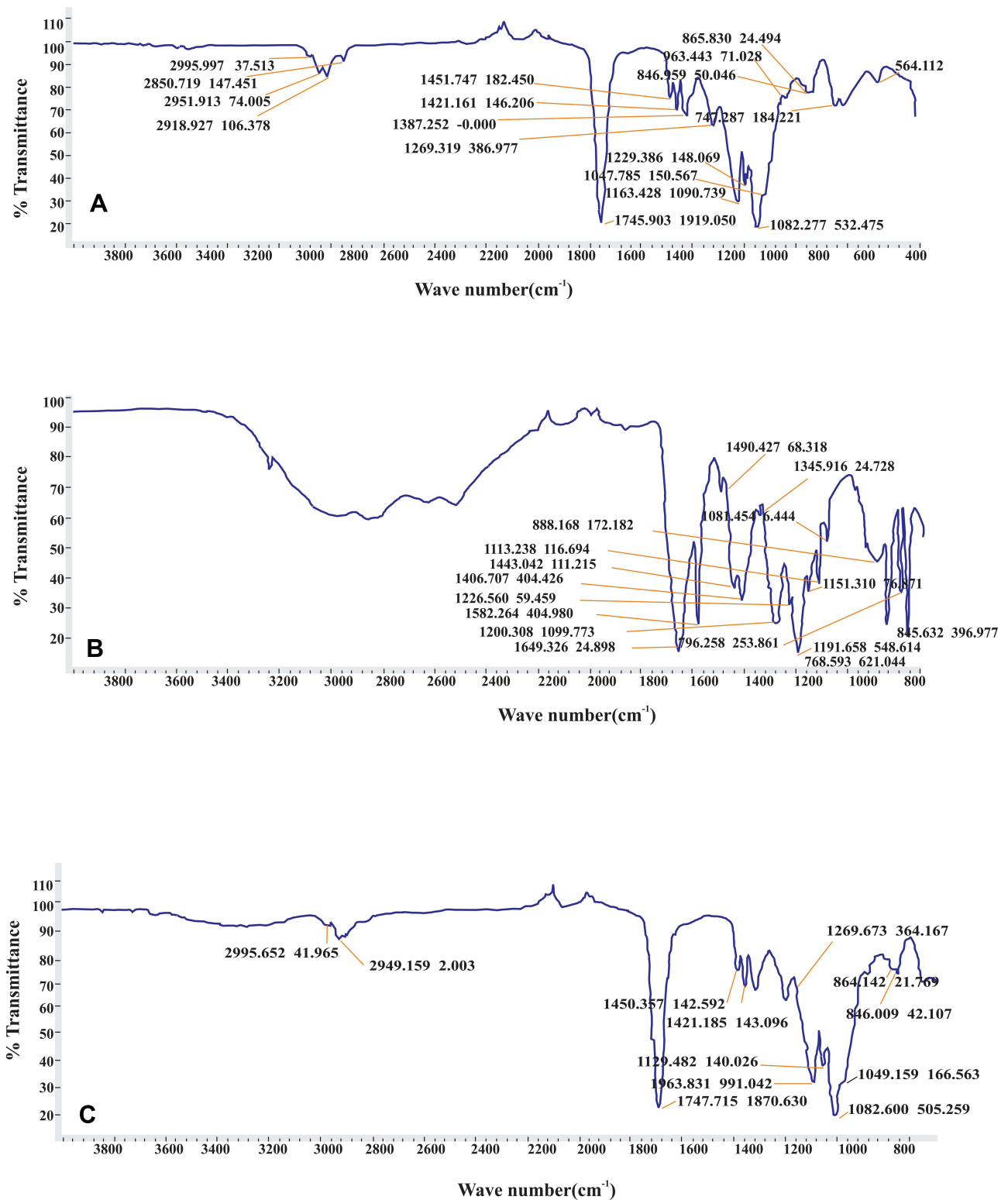
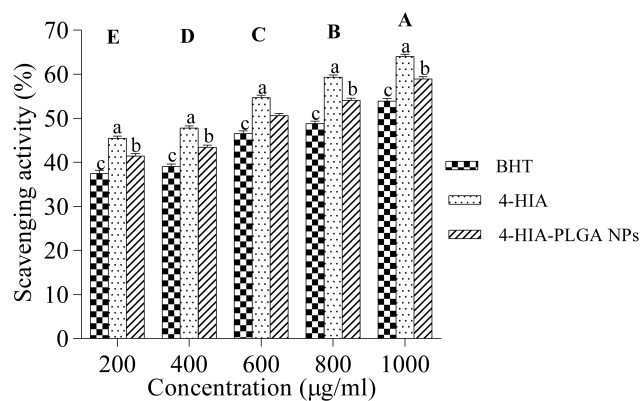
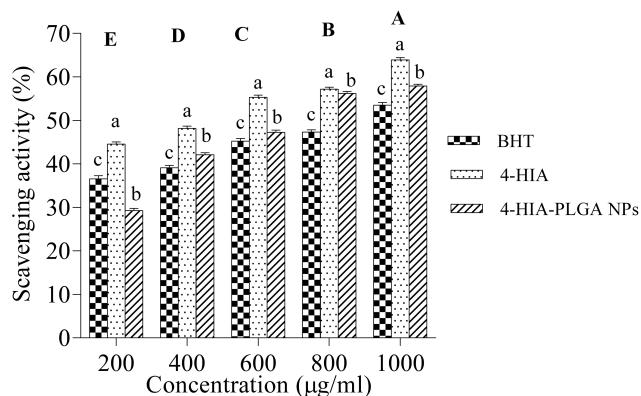


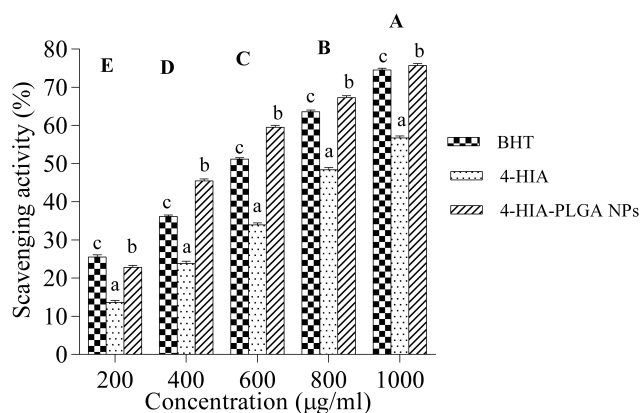
Figure 4 FTIR spectra of (A) PLGA (B) 4-HIA and (C) 4-HIA encapsulated PLGA nanoparticles.



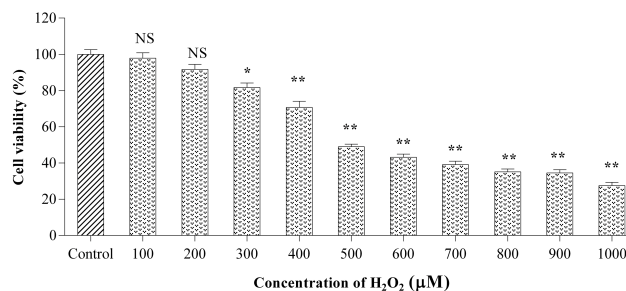
**Figure 5** DPPH radical scavenging activity. Values are represented as mean  $\pm$ S.E (n=3). Significance between the samples is represented in lower case ( $p < 0.05$ ) and between the concentrations is represented in upper case. Those not sharing the same letters are significantly different.



**Figure 6** Superoxide radical scavenging activity. Values are represented as mean  $\pm$ S.E (n=3). Significance between the samples is represented in lower case ( $p < 0.05$ ) and between the concentrations is represented in upper case. Those not sharing the same letters are significantly different.



**Figure 7** Hydroxyl radical scavenging activity. Values are represented as mean  $\pm$ S.E (n=3). Significance between the samples is represented in lower case ( $p < 0.05$ ) and between the concentrations is represented in upper case. Those not sharing the same letters are significantly different.



**Figure 8** Effect of H<sub>2</sub>O<sub>2</sub> on cell viability in PC12 cell lines induced oxidative stress. Values are expressed as mean  $\pm$ S.E (n=3). \*\* $P < 0.01$  compared to control. \* $P < 0.05$  compared to control.

## In vitro Studies

### Effect of H<sub>2</sub>O<sub>2</sub> on Cell Viability and Morphology

The viability of PC12 cells was evaluated by MTT assay, which exhibited dose-dependent cytotoxicity, as shown in Figure 8. The cells were treated with different concentrations of H<sub>2</sub>O<sub>2</sub> for 24 h. The IC<sub>50</sub> value was found to be 500  $\mu$ M. Therefore, this concentration was fixed for further experiments. Cell morphology was observed using a phase contrast microscope. H<sub>2</sub>O<sub>2</sub>-treated cells revealed cell loss, shrinkage, irregular shape, detachment, and membrane blebbing (Figure 9).

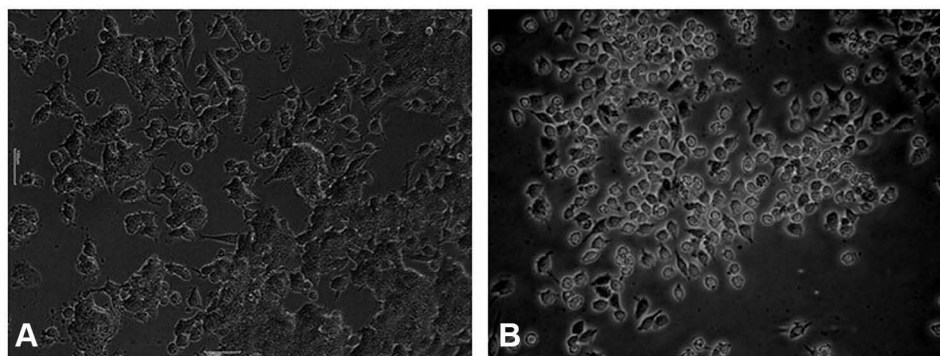
### Effect of 4-HIA Encapsulated PLGA NPs Against H<sub>2</sub>O<sub>2</sub>-Mediated Cytotoxicity

To assess the effect of 4-HIA encapsulated PLGA NPs on H<sub>2</sub>O<sub>2</sub>-mediated cytotoxicity, cells were treated with various concentrations of PLGA/4-HIA encapsulated PLGA NPs for 24 h before to H<sub>2</sub>O<sub>2</sub> exposure. As shown in Figure 10, the viability of the cells exposed to H<sub>2</sub>O<sub>2</sub> was reduced to 50% with respect to control. 4-HIA and 4-HIA encapsulated PLGA remarkably curtailed H<sub>2</sub>O<sub>2</sub>-induced toxicity in a dose-dependent manner and restored the cell viability to 84% and 94% over H<sub>2</sub>O<sub>2</sub>-treated cells, respectively.

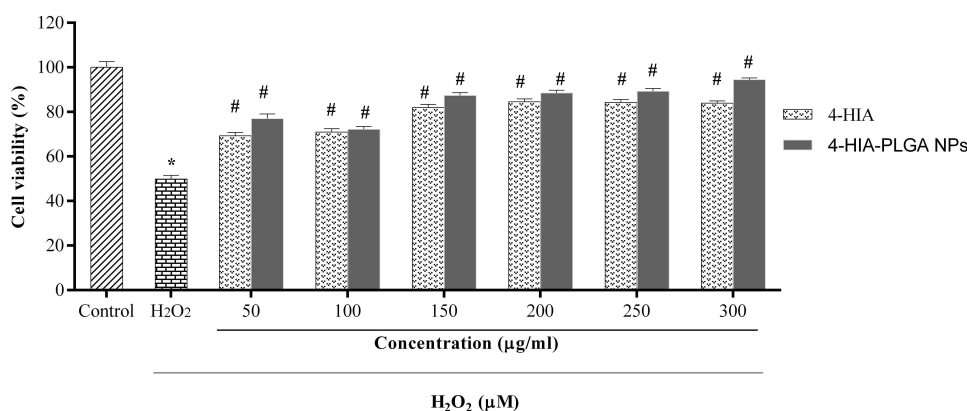
## Discussion

4-HIA is a potent phenolic compound of *D. hamiltonii* extract exhibiting a broad range of biological functions like antioxidant and cytotoxic activities. However, the poor water solubility of 4-HIA has limited its application as a promising therapeutic candidate in the management of neurodegenerative ailments. In recent years, novel drug





**Figure 9** Images of (A) Control (B) H<sub>2</sub>O<sub>2</sub> treated PC12 cells under phase-contrast microscope (40X).



**Figure 10** Cell viability of PC12 cells treated with different concentrations of 4-HIA and 4-HIA encapsulated PLGA nanoparticles against H<sub>2</sub>O<sub>2</sub> induced cytotoxicity. Values are expressed as the mean  $\pm$ S.E (n=3). \*P<0.01 compared to control, #P<0.05 compared to H<sub>2</sub>O<sub>2</sub> treated cells.

delivery strategies have been adopted for natural phyto-compounds for improving their bioavailability. PLGA, a biocompatible and biodegradable polymer, has attracted researchers for the formulation of hydrophobic compounds.<sup>10,20</sup> In the current study, we hypothesized that 4-HIA NPs employing PLGA polymer might serve as a novel strategy towards neuroprotection. Hence, for the first time, 4-HIA-PLGA NPs were synthesized, and their neuroprotective efficacy was evaluated in the PC12 cells against H<sub>2</sub>O<sub>2</sub>-induced toxicity.

In our study, 4-HIA encapsulated PLGA NPs were fabricated by a s/o/w solvent evaporation method with PLGA as the carrier. PVA is a routinely employed stabilizer for PLGA polymer and used as a surfactant in pharmaceutical formulations to stabilize the nanoparticle emulsion leading to an increase in solubility of the drug without agglomeration.<sup>21,22</sup>

In order to know the nature of 4-HIA encapsulated PLGA NPs, XRD was performed. The blank PLGA revealed no characteristic peak, but the distinct peaks

were observed for 4-HIA, which depicts the crystalline nature of 4-HIA, while the encapsulated NPs fail to exhibit the characteristic peaks indicating the effective encapsulation of the compound in the PLGA. Therefore, in PLGA NPs, the 4-HIA was present in the amorphous phase and may be homogeneously dispersed in the PLGA matrix suggesting its amorphous nature. Our results are parallel with Pool et al.<sup>23</sup> Our findings also indicated that the intermolecular interactions among PLGA polymer chains confers an amorphous character to 4-HIA after encapsulation. A related phenomenon was noticed with the majority of the crystalline drugs possessing amorphous nature when encapsulated into PLGA NPs.<sup>24,25</sup>

The NPs structure plays a vital role in understanding their interaction and adhesion with the cells. Hence, SEM and HR-TEM were performed to determine the surface morphology and shape of 4-HIA encapsulated PLGA NPs. The NPs were spherical with a smooth surface and homogeneous shapes. The morphology of PLGA NPs obtained is in accordance with the reports of Mathew



et al, Pool et al, and Arasoglu et al.<sup>23,26,27</sup> Our observations on diffraction ring pattern of SAED were obtained for 4-HIA corresponds with the XRD peaks of 4-HIA and confirmed its crystalline nature. While the halo ring pattern for the fabricated NPs revealed its amorphous nature. Thus, the obtained patterns of SAED were complementing well with the studies of Kaur et al.<sup>28</sup>

Following this, FTIR transmittance was performed to identify the functional groups of compounds based on the characteristic vibrations of the bonds. The spectra obtained for free PLGA NPs showed characteristic bands consistent with the studies of Erbetta et al and Mondal et al.<sup>29,30</sup> However, in the encapsulated NPs, the C=O peak was slightly shifted to a higher wavelength. This could be due to the formation of intermolecular hydrogen bonds between the C=O bond of PLGA and the O-H bond of 4-HIA.

The DPPH assay was employed to assess the free-radical quenching ability of 4-HIA encapsulated PLGA NPs compared to 4-HIA. Our findings revealed that 4-HIA showed better scavenging activity compared to 4-HIA encapsulated PLGA and BHT, which may be due to prolonged time taken by the encapsulated 4-HIA to leach out into the DPPH solution. Furthermore, PLGA encapsulation does not degrade the antioxidant property of 4-HIA. Our reports agree with the studies of Mathew et al<sup>27</sup> wherein curcumin encapsulated PLGA NPs showed less activity than raw curcumin.

Superoxide and hydroxyl radicals are the major free radicals generated as a result of metabolic reactions.<sup>31</sup> In our study, 4-HIA exhibited a higher ability to scavenge the superoxide radicals over the 4-HIA encapsulated PLGA and BHT. The decreased activity in encapsulated NPs may be attributed to the slow release of the compound into the reaction mixture. While hydroxyl radical scavenging activity was more evident in 4-HIA encapsulated PLGA NPs, which may be attributed to the conversion of H<sub>2</sub>O<sub>2</sub> to H<sub>2</sub>O.<sup>32</sup>

Numerous studies have demonstrated that oxidative stress (OS) is a major cause of cell damage in neurological disorders.<sup>33</sup> Reactive oxygen species (ROS) damage biological molecules, leading to apoptosis or necrotic cell death.<sup>34,35</sup> Thus, curtailing these ROS or inhibiting their generation by antioxidants could effectively prevent oxidative cell injury. H<sub>2</sub>O<sub>2</sub> has been widely employed in in vitro studies as an inducer of OS.<sup>36,37</sup> H<sub>2</sub>O<sub>2</sub> exposure to PC12 cells leads to disparity in energy metabolism due to the oxidation of proteins and lipids by hydroxyl and peroxy radicals. Our findings revealed that H<sub>2</sub>O<sub>2</sub> exposure

to the cells exhibited a dose-dependent decrement in cell viability. However, the cells pre-treated with 4-HIA and 4-HIA encapsulated PLGA NPs resulted in the reduction of H<sub>2</sub>O<sub>2</sub> cytotoxicity, indicating that they significantly inhibited H<sub>2</sub>O<sub>2</sub>-induced toxicity by curtailing OS. 4-HIA encapsulated PLGA NPs have a greater anti-apoptotic potential with respect to 4-HIA, which could be attributed to its enhanced bioavailability and half-life.

## Conclusion

In conclusion, the present study suggests that 4-HIA encapsulated PLGA NPs exhibited a remarkable hydroxyl radical scavenging ability. The encapsulated NPs offered better protection against H<sub>2</sub>O<sub>2</sub>-induced oxidative stress in PC12 cells. These findings suggest that 4-HIA encapsulated PLGA NPs may be a promising strategy for the treatment of neurological disorders. However, further studies have to be carried out in vivo to establish its therapeutic potential for the management of oxidative stress-related ailments.

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## Disclosure

The authors report no conflicts of interest in this work.

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