



Ecological Drawbacks of Nanomaterials Produced on an Industrial Scale: Collateral Effect on Human and Environmental Health

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Abstract Currently, hundreds of different nanomaterials with a broad application in products that make daily lives a little bit easier, in every aspect, are being produced on an industrial scale at thousands of tons per year. However, several scientists, researchers, politics, and ordinary citizens have stated their concern regarding the life cycle, collateral effects, and final disposal of these cutting-edge materials. This review summarizes, describes, and discusses all manuscripts published in the Journal Citation

Reports during the last 10 years, which studied the toxicity or the effects of nanomaterials on human and environmental health. It was observed that 23.62% of the manuscripts analyzed found no ecological or human risks; 54.39% showed that several nanomaterials have toxicological effects on the ecosystems, human, or environmental health. In comparison, only 21.97% stated the nanomaterials had a beneficial impact on those. Although only 54.39% of the manuscripts reported unfavorable effects of nanomaterials on ecosystems, human, or environmental health, it is relevant because the potential damage is invaluable. Therefore, it is imperative to make toxicological studies of nanomaterials with holistic

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focus under strictly controlled real conditions before their commercialization, to deliver to the market only innocuous and environmentally friendly products.

Keywords Ecofriendly · Ecological risks · Environmental pollution · Green synthesis · Nanoparticle · Nanotechnology and nanoscience · Nanotoxicology · Social welfare · Soil degradation · Sustainable development

1 Introduction

Engineered nanomaterials (ENMs) have been synthesized by different procedures, which confer particular and exceptional properties never seen before. Therefore, these nanosized materials' production has moved from the laboratories to the industrial sector, where thousands of tons are produced annually. ENMs have been characterized by their capacity to improve the transport and delivery of drugs, fertilizers, enzymes, and genetic material, among others. Besides, ENMs have also been a breaking point regarding technological areas such as electronics, medicine, high-performance computing, chemical engineering, and agriculture (Medina-Pérez et al., 2020; Fernández-Luqueño, 2019; Pérez-Hernández et al., 2020). It is well known the benefit that nanoscience and nanotechnology have brought to society regarding the advances that these cutting-edge knowledge areas. If it were not enough, these advantages would increase dramatically in a short period. However, hitherto the studies regarding the effects of ENMs on human and environmental health, their toxicity, and drawbacks have been scarcely reported (Pérez-Hernández et al., 2020).

Currently, the price of ENMs has decreased over time, and the productivity expanded to several thousand tons per year for technological applications. This production and market behavior have been reported before for single- or multi-walled carbon nanotubes (SWCNTs or MWCNTs, respectively) (Zhang et al., 2013). Besides, Piccinno et al. (2012) stated that the ENM production at the industrial scale had reached almost 80,000 tons during 2011.

The undesirable effects of ENMs produced on an industrial scale on human health, ecology, and environment have been discussed by Shakeel et al. (2017), Ivask et al. (2014), Bondarenko et al. (2013), Ma et al.

(2013), Cao and Luo (2019), Du et al. (2013), and Line et al. (2017). It has been stated that TiO₂ nanoparticles (TiO₂-NPs) enter through the pulmonary route and translocate to the brain or the systemic circulation from where these reached the kidney, liver, etc. (Shakeel et al., 2017). According to the same authors, TiO₂-NPs generate reactive oxygen species (ROS), histopathological alterations, carcinogenesis, genotoxicity, and immune disruption. A comprehensive and critical literature overview on factors modifying the toxic effect of NPs and their toxicity mechanisms of ENMs on various environmentally relevant test species was addressed by Ivask et al. (2014) and Bondarenko (2013). High acute toxicity of ZnO-NPs to bacteria, algae and plants, aquatic and terrestrial invertebrates, and vertebrates was described by Ma et al. (2013). This toxicity was highly dependent on test species, physicochemical properties of the ENMs, and test procedures. In contrast, particle dissolution and particle-induced generation of ROS represented the principal modes of action for NPs toxicity through all species evaluated by Ma et al. (2013).

According to Cao and Luo (2019), carbon nanotubes (CNTs) might induce adverse vascular effects in laboratory animals. Simultaneously, only limited reports have been obtained from CNT-exposed human beings, and their results are inconclusive. The potential risks regarding CNTs and harmful effects resulting from exposure of animals, microorganisms, and crops to CNTs, also stated by Du et al. (2013), through in vitro and in vivo tests. Besides, CNTs' impact on the terrestrial ecosystem, their behavior in soil and plants, interactions with other pollutants, and their implications on soil organisms and plants were studied by Pérez-Hernández et al. (2020) and Line et al. (2017). Therefore, ENMs must be either avoided or strictly managed to minimize risks for human and environmental health. Besides, there is increasing evidence that ENMs threaten non-target organisms.

It is hypothesized that ENMs produced on an industrial scale have several drawbacks that jeopardize human and environmental health. However, these threats are not widely recognized because most published studies have been carried out in a laboratory under controlled conditions. Therefore, long-term in situ field trials are required to understand and dimension the magnitude of the problem that we could face up in a short period.

This review aims to discuss the manuscripts published in the Journal Citation Reports during the last 10 years, which studied the ecological toxicity, the drawbacks, and the effects of ENM produced on an industrial scale on human and environmental health.

2 Materials and Methods

2.1 Search Strategy, Information Resources, and Inclusion Criteria

This systematic review exhaustively covers all studies published online and limited to English between 2012 and 2021 years (the last search was run on 15 August 2021) following the methodology PRISMA (Preferred Reporting Items for Systematic Reviews and Meta-Analyses). Even though this review discusses the positive effects of ENMs such as TiO₂, Ag, Zn, Si, Fe, Al, Ce, and CNTs, it emphasizes their toxicity on edible and inedible plants and soil organisms. Therefore, search restrictions in the period 2012–2021, is due to two criteria: (1) based to most widely ENMs produced annually and used worldwide, and (2) to the fact that in the last ten years, several articles that address the toxicological effect of the properties of ENMs. These effects have been discussed regarding organisms (plants, soil organisms, and humans), within which the adverse effects were related to several features of NPs, such as size, shape, exposure time, or concentration. Besides, the investigations carried out during the last years have used cutting-edge techniques for detection, measurement, or synthesis, which have helped to understand these phenomena. It even has opened new knowledge areas related to toxicity in plants or soil biota at the level of cellular damage, proteins, DNA, and enzymes, among others. Into the bargain, new researches that address the issues of biotransformation of ENMs in the food chain and possibly collateral effects on humans have appeared.

The literature search strategy was carried out using the “Web Science” database. The search was based on a pre-determined series of keywords in title and topic such as “nano,” “nanomaterials,” “nanoparticles,” “organisms,” “soil invertebrate,” “macroinvertebrate soil,” “soil macrofauna,” “soil mesofauna,” “earthworms,” “terrestrial isopods,”

“collembola,” “symbiotic interaction ships,” “mycorrhizal fungi,” “plants edible,” “plants inedible,” and “trophic chain.” Boolean operators “AND” and “OR” were used as search parameters to expand the search. Also, we used the wildcard (*), which allows and includes all the keywords that start with the preceding characters.

The studies consulted were mainly based on those that evaluated the effects of the ENMs (positive, negative, or null) of TiO₂, Ag, Zn, Si, Fe, Al, Ce, and CNT. It is important to note that, although studies report heterogeneous statistical analyzes, we intend to demonstrate the effects (mostly adverse) regardless of the evaluation time and status of growth and development of plants and soil organisms. Therefore, data were included if the studies met the following criteria: (1) the studies reported positive, negative, or null effects on morphological and chemical characteristics in plants, mesofauna, macrofauna, and arbuscular mycorrhizal fungi; (2) investigations that in their methodology specified factors such as evaluated concentration, size of NMs, and study time; and (3) studies conducted in the laboratory, greenhouse, and field (agar culture media, artificial soil, natural soil, and hydroponics). On the other hand, we exclude the reports that evaluated or used ENMs different from those of our interest (Fig. 1).

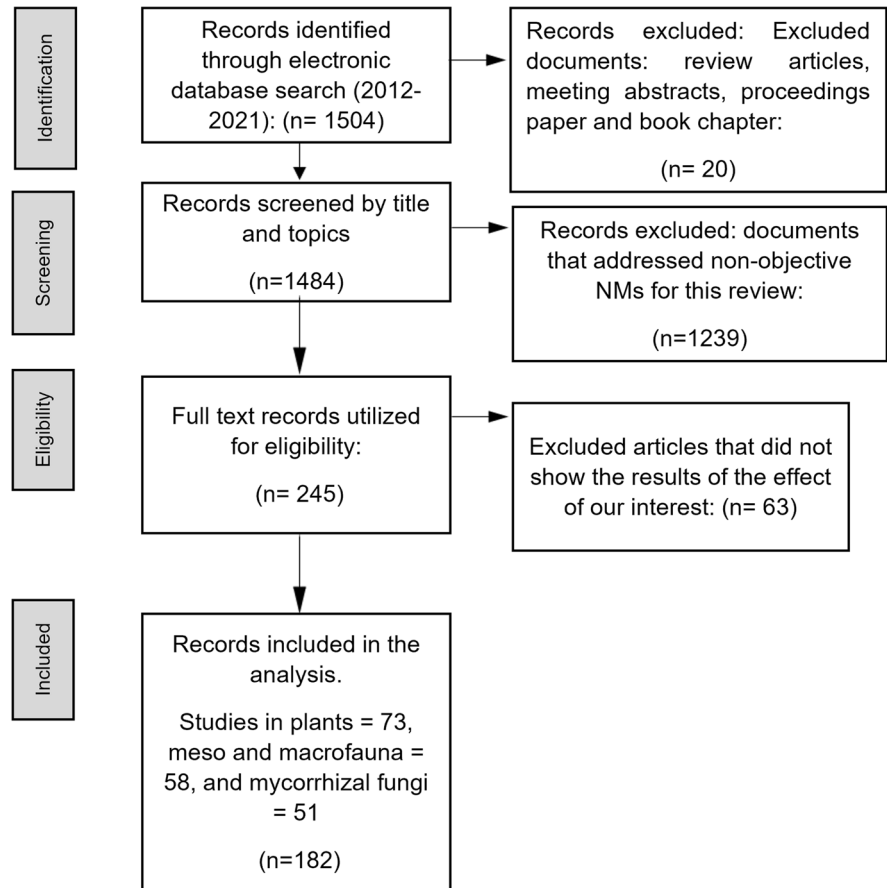
Two reviewers were responsible for the selection of articles. Individually, each of the selected articles was reviewed in the summaries, results, and conclusions. On the other hand, five reviewers wrote the results and discussion sections. At the same time, from the studies selected and considered for the results sections, we developed a summary of data presented in graphs. Minitab software (Version 18) was used to draw the figures.

3 Results and Discussion

3.1 Nanomaterials Produced on an Industrial Scale

The rapid development in NPs production technologies and the increasing application of ENMs in daily products have increased the number of registered nanoproducts. It had augmented from 54 in 2005 to 1865 in 2013 and more than 3000 in 2020. With this

Fig. 1 Flow diagram of study selection (according to criteria of PRISMA)



approach, massive developments in various application areas are predicted (Rai et al., 2012).

ENMs are being more common in daily used products. According to the Global Industry Analysts' 2010 report, the global market for nanomaterials (NMs) reached around US\$ 6.2 million by 2015. The market of cosmetics, medicine, agriculture, food, food packaging, electronics, paints, medical devices, and coatings are some examples of the widespread use of ENMs (Pine et al., 2007).

A method that has been employed in NPs production is the continuous-flow hydrothermal synthesis (CFHS), in which a hot stream of deionized water and a cold stream of a precursor mixed with deionized water meet in the reactor to form NPs. These temperatures usually range between 250 and 400 °C. Some of the precursors used in this kind of process are titanium oxysulfate (TiOS) for TiO₂ and zinc nitrate (Zn(NO₃)₂) for ZnO. The NPs are obtained after the post-processing method of washing the concentrated

solution. After that, the material is sedimented, obtaining the NPs concentrated (Rai et al., 2012).

Wet methods generally perform better in terms of quality, and the NPs are better suited for specialized applications than those produced by dry processes. The batch hydrothermal and solvothermal methods produce NPs suitable for the electrical industry as semiconductor materials (ZnO). In TiO₂, its particle size, provided by wet technologies, varies from 10 to 20 nm. The CFHS is a wet method, which produces NPs with the highest quality. By the combustion synthesis, very small and non-agglomerated NPs are produced (Stieberova et al., 2019).

Some methods have the highest TiO₂ production rates, where vapor-assisted flame pyrolysis (VAFP) is included. The first one is the Altair method, which is a chloride process with a production rate for fine metal powders of up to 25,000 kg h⁻¹. The second one is the sulfate process, with a production rate of 100 kg h⁻¹ (Rai et al., 2012). These methods are not versatile; the

VAFP method is applicable only for TiO_2 , obtaining NPs with a diameter of 200 nm for use in pigments, SiO_2 , and other simple oxides. The sulfate process is appropriate only for TiO_2 . A high-temperature plasma (HT plasma) high production rate can be estimated at up to 60 kg h^{-1} at 400 kW. The production rate from the typical HT plasma plant is lower than 10 kg h^{-1} . Another method is the continuous-flow hydrothermal synthesis (CFHS), with an average production rate of 10 kg h^{-1} . It is one of the highest productive ways to produce large volumes of NPs demanded commercial applications (Stieberova et al., 2019).

TiO_2 -NPs are used as semiconductors in the fields of energy due to their property of absorbing ultraviolet (UV) light at wavelengths shorter than their band-gap (3.2 eV for anatase), triggering photocatalytic activities (Fu et al., 2015). Also, TiO_2 -NPs are used in commercial products such as solar cells, sunscreens, self-cleaning surface coatings, disinfectant sprays, and water treatment agents (Katal et al., 2020; Chung & Wang, 2012). The TiO_2 -NPs are low-cost materials, have high photocatalytic activity, considerable photostability, and non-toxicity. TiO_2 -NPs are classified into three main crystal structures, such as anatase (stable at low temperature), brookite (found in minerals with an orthorhombic crystal structure), and rutile (stable at higher temperatures) (Kunhikrishnan et al., 2015).

There are some methods and leading strategies for the synthesis of faceted TiO_2 crystals. These could be classified as the wet-chemistry route (hydrothermal, solvothermal, and nonhydrolytic), gas oxidation route, crystallization transformation from amorphous TiO_2 , epitaxial growth, and spray-drying (Chung & Wang, 2012). The low-temperature plasma (LT plasma) is used to produce NPs for highly specialized products (Stieberova et al., 2019). The Altair hydrochloride process produces anatase TiO_2 -NPs with an average diameter of 40 nm (Grubb & Bakshi, 2010). The size for NPs produced by HT plasma ranges from 50 to 100 nm.

TiO_2 is the most popular semiconductor photocatalyst, but recently other ENPs have attracted much attention due to its admirable attributes, such as ZnO. It is a multifunctional material that has unique physical and chemical properties. It also is a no-toxic, low-cost ENM and can be synthesized easily and suitable for mass production. The nanostructured ZnO is a chemoresistive sensing material; it react with the

adsorbed oxygen species present on the surface of the material containing target gas molecules such as CO, NO_2 , NO, ethanol, H_2S , H_2 , acetone, acetaldehyde, LPG, and NH_3HCHO , altering the electric conductance (Grubb & Bakshi, 2010; Caramazana-González et al., 2017; Yu et al., 2016).

ZnO-NPs are an essential material in the ceramics industry because of their hardness, rigidity, and piezoelectric constant. Simultaneously, its low toxicity, high biocompatibility, and tested biodegradability make it an ENM of interest for biomedicine and in pro-ecological systems (Zhu & Zeng, 2017). Different processes are adopted to synthesize the ZnO-ENM, such as conventional solid-state reaction, hydrothermal, sol-gel, chemical co-precipitation, microwave, and emulsion method. Most of these techniques are time-consuming and not cost-effective because of high-temperature processing and expensive equipment. Emerging low-temperature technology is the solution combustion synthesis (SCS); it is a facile and economically feasible technique.

In the literature, it is mentioned that there are not only the chemical methods for the production of ZnO, as the ones above, there also exist metallurgical processes, where de NPs are obtained by roasting suitable zinc ore via a direct or indirect method. The straightforward process involves reducing the zinc ore by heating with coal, followed by the oxidation of zinc vapor in the same reactor in a single production cycle (Zhu & Zeng, 2017). The long process is where the metallic zinc is melted in a furnace and vaporized at $910 \text{ }^\circ\text{C}$. The immediate reaction of the zinc vapor with oxygen from the air produces ZnO. It has to be stated that crystalline oxide powders, combined with other materials, provide the possibility to obtain improved chemical, mechanical, optical, or electrical properties (Zhu & Zeng, 2017).

Silica nanoparticles have excellent optical, electrical, and thermal properties and features as easy preparation, well-defined dimensions, making it a suitable material for many applications. In industries, SiO_2 is used in catalysis, pigments, electronic devices, thermal insulators, humidity sensors, etc. In pharmaceuticals, it is used in enzyme encapsulation, drug delivery, cell markers, etc. (Manjunath et al., 2020).

In the concrete industry, the market demand for cement, concrete's binder, keeps growing every year. The annual production of cement increased from 3.6 billion tons to 4.1 billion tons worldwide, from 2011

to 2015. Nonetheless, cement manufacture is responsible for 5–7% of global anthropogenic CO₂ emissions. There has been implemented the reduction of usage of cement and make concrete more sustainable construction material. Some advances in replacing ordinary Portland cement (OPC) with supplementary cementitious materials (SCMs) are promising alternatives. Some common SCMs are often much less reactive than the clinkers of OPC, resulting in a slow strength development of the mixing concrete. However, some ENMs, such as SiO₂, TiO₂, Al₂O₃, CNT, graphene, and others, are often used to compensate for this drawback presenting excellent performances in terms of the early age strength gain (Wang & Park, 2015; Antiohos et al., 2014). Somewhere else, it was reported that the compressive strength of mortar with SiO₂ particles was higher than those of mortars containing silica fume at 7 and 28 days; SiO₂ behaves not only as a filler but also as an activator to promote pozzolanic reaction (Feng et al., 2020).

SiO₂ has attracted attention as a promising alternative anode material due to its high theoretical capacity (1965 mA hg⁻¹). Also, it is a commercially viable electrode material for Li-ion batteries for its natural abundance, low cost, and environmental friendliness (LIBS). Nonetheless, it has some limitations in the practical application. The high capacity of SiO₂ cannot be fully utilized owing to its poor electronic conductivity. Another problem is its capacity fading caused by volume expansion during cycling (Rai & Haque, 2014).

Many techniques are used to prepare silica NPs, such as plasma synthesis, chemical vapor deposition (CVD), sol–gel process, and chemical precipitation. The most recommended synthesis of silica nanoparticles is the sol–gel method due to its easy preparation process, the ability to control the particle size, size distribution, and morphology by monitoring the experimental conditions (Manjunath et al., 2020).

Iron and oxygen are two of the four most common elements in Earth's crust, and iron oxides form naturally through the weathering of Fe-containing rocks both on land and in the oceans. Iron oxides are in natural abundance in rocks, soils, and dust and play an essential role in geochemistry. Iron oxides are imperative because these can be used as a source of Fe, which is processed to make steel. Other applications include corrosion protective coatings (Fe₃O₄ or

“black rust”) and use in recording media and catalysis. In the biomedicine field, Fe₃O₄ nanoparticles are used as a contrast agent in MRI scanners. In hyperthermia treatment, tumors can be killed by the local temperature increase that occurs when Fe₃O₄ NP are placed in a rapidly varying magnetic field (Belgivayeva & Taniguchi, 2019).

According to the crystallographic phase and the oxygen content, the semiconductor hematite α -Fe₂O₃ phase or half-metallic magnetite spinel Fe₃O₄ phase can be observed in various optical, electrical, and magnetic properties. The α -Fe₂O₃ exhibits weak ferromagnetism, whereas the magnetite is ferrimagnetic at ambient temperature. The Fe₃O₄ behaves almost like an electric metallic conductor, while the hematite phase displays electrical insulating property. Hematite is also an anode for a lithium-ion battery or photoelectrochemical water splitting, as a pigment, wastewater treatment, gas sensor, etc. Magnetite is employed in biomedical therapies, magnetic printing, microelectronics as microwave-absorbing materials, thermoelectricity, or a thermistor in a bolometer (Li & Zhou, 2019).

There have been developed cathode materials in the supercapacitors area as carbons, used as harmful materials due to their high specific surface area, high power density, and good electrical conductivity. Still, the capacitance value has been restricted by energy density. Transition metal oxides such as MnO₂, NiO, CuO, Co₂O₃, and FeO exhibit higher specific capacitance and conductivity due to storing charges by highly reversible faradaic redox reaction on the electrode surface.

Another ENP with wide applications is aluminum oxide (AlO), which exists in several crystallographic polymorphs. Thin and thick films of AlO-ENPs are used in microelectronics and catalysis, also as a diffusion barrier, thermal barrier, and water-resistant coatings for cutting tools. Besides, γ -alumina, which is a metastable phase, is used in catalysis due to its large surface area (Aubry et al., 2019). Also, γ -Al₂O₃ is considered the most effective adsorbent, not only for its high surface area but even for its enhanced reactivity for adsorption and mass transfer efficiency (Nayar et al., 2014). At the same time, the anodic alumina templates are used for the synthesis of nanowires (Aubry et al., 2019).

Mechanical milling, vapor phase reaction at high temperature, combustion method, sol–gel method,

and precipitation method are strategies to synthesize AlO-ENP. The sol–gel process has attracted interest due to the morphology and homogeneity, and surface area of the particles can be easily controlled. Sol–gel synthesis of alumina particles comprises hydrolysis of aluminum alkoxide in water to obtain boehmite (AlOOH) as gel and its calcination at 400–800 to produce γ -Al₂O₃ (Nayar et al., 2014).

The Bayer method, one of the most popular routes applied for the Al₂O₃ manufacture, produces about 45 million tons. Its method uses bauxite, and about 40 million tons are consumed for refining aluminum. Nonetheless, nearly 5 million tons of Al₂O₃ are produced for chemical-grade and used for various purposes. The production methods used to create high purity α -Al₂O₃, besides the Bayer method (99.6–99.9% purity), are the hydrolysis of aluminum alkoxides, chemical vapor deposition (CVD), thermal decomposition of aluminum alum, and thermal decomposition of inorganic aluminum salts (Singh et al., 2016).

The γ -Al₂O₃ is an essential catalyst in the electrochemistry area and is useful as catalyst support in the automotive and petroleum industries. The different synthesis routes developed to prepare organized mesoporous alumina and ordered alumina using surfactant micelles (polyethylene) as a structure-directing agent are explained in the literature. Another path is using ultrasound treatment. With these routes, it has been demonstrated the different structures using the nitrogen physisorption characterization. The ultrasound treatment led to the transformation of ink-bottle pores into a cylindrical form, and the surface area and pore size were increased. Aluminas prepared with polyethyleneimine as a template showed larger cylindrical wormhole-like mesopores with a broader pore size distribution, high surface area, and pore volume. It was found that with the ultrasound treatment, the alumina NP demonstrated high catalytic activity and had unusual porous structure and large crystallite size (Vinogradov et al., 2010).

The cerium oxide ENPs (CeO-ENPs) are used in industrial applications such as chemical/polarization polishing, corrosion protection, fuel oxidation catalysts, and the non-industrial section as antibiotics and drug transfer (Shirai et al., 2009). CeO₂-ENPs had received more attention due to their useful catalysis applications, fuel cells, and antioxidants in biological systems. Ce exists in two oxidation states: Ce³⁺

and Ce⁴⁺. Therefore, CeO₂ can have two different oxide forms, CeO₂ (Ce⁴⁺) or Ce₂O₃ (Ce³⁺, in a bulk material). CeO₂-ENPs have improved redox properties concerning the bulk materials. CeO₂-ENPs are effective against pathologies associated with chronic oxidative stress and inflammation. Besides, they are used for biomedicine in the bioanalysis field and drug delivery due to their reflective properties compared to the multienzyme (Reed et al., 2014). Some biological applications of CeO₂-ENPs are the antibacterial effect inhibiting both gram-negative and gram-positive bacteria. They are also useful in the neurodegenerative treatment, in enzyme mimetic forms, catalase, and peroxidase mimicking activity as a catalyst, in sensing applications, and angiogenesis induction (Reed et al., 2014).

There are synthetic methods divided into two regimes: lower temperatures (<250 °C) and higher temperatures extended to combustion, flame synthesis, and even plasma arc temperature, reaching several thousand degrees centigrade. The most commonly employed lower temperature wet-chemical synthetic methods are co-precipitation, hydrothermal, solvothermal, sol–gel, Pechini, microemulsion, and reversed micelle methods. These synthetic methods are useful for small-scale and research quantities of materials (Miri et al., 2019).

Calcination is commonly used as a high-temperature technique for the industrial-scale production of ceria. This process involves high treatment below the melting point of the material in an oxygen atmosphere. Some industrial firms have patented their methods, including Hitachi with the precipitation of ENPs, using carbonate chemistry prominent, then calcined, and milled; Rhodia technique with precipitation using hydroxide chemistry noticeable after that calcined and milled; Antaria process with precipitation of ENMs while being mechanically milled in NaCl media; Umicore, precipitation while being evaporated (thermal spray), and Evonik technology using a pyrogenic gas-phase formation from metal-alkoxides or metal-carboxylates (Miri et al., 2019).

Also, there exist green approaches for CeO₂-ENPs synthesis. One of them is plant-mediated synthesis, consisting of the phytosynthesis of metal and metal oxide ENPs using plants as *Gloriosa superba*, *Aloe vera*, and others. The ENPs obtained showed smaller crystal sizes with a higher surface area led to higher antibacterial activity. Still, these ENPs were generally

so large, so they are appropriate for biomedical applications. In the case of the mycosynthesis technology to synthesize CeO₂-ENPs, it showed advantages as manageability and cost-effectiveness. This technology was less time-consuming and required less energy and can be used as an economical and valuable alternative for the large-scale production of ENPs. Other mentioned green synthesis is the nutrient-mediated synthesis and the biopolymer-mediated synthesis (Reed et al., 2014).

The commercial interest in carbon nanotubes (CNTs) has been increasing around the world, and it is reflected in the production capacity that presently exceeds several thousand tons per year. Bulk CNT powders are incorporated in various commercial products ranging from rechargeable batteries, automotive parts, and sporting goods to boat hulls and water filters. CNTs are seamless cylinders of one or more graphene layers, denoted as SWCNTs or MWCNTs, with open or closed ends. The best CNTs have all carbons bonded in hexagonal lattice except at their ends. In contrast, defects in mass-produced CNTs introduce pentagons, heptagons, and other imperfections in the sidewalls that generally degrade desired properties. The diameters of SWCNTs are typically 0.8 to 2 nm, and the MWCNTs are about 5 to 20 nm, sometimes could exceed 100 nm. CNTs' lengths range from less than 100 nm to several centimeters (Charbgoon et al., 2017).

Most CNT production is used in bulk composite materials and thin films, which rely on unorganized CNTs architectures having limited properties. Organized CNT architectures promise to scale up to properties of individual CNTs and realize new functionalities (Charbgoon et al., 2017). One of the methods for CNT synthesis is chemical vapor deposition (CVD). However, large-scale CVD methods yield contaminants that can influence CNT properties (Charbgoon et al., 2017).

The synthesis of long, aligned CNTs that can be processed without dispersion in a liquid offers promise for cost-effective realization of compelling bulk properties. These methods include the self-aligned growth of horizontal and vertical CNTs on substrates coated with catalyst particles and CNT sheets and yarns production directly from floating-catalyst CVD systems. The floating catalyst chemical vapor deposition (FC-CVD) is one method to synthesize CNTs due to its scalability at viable production cost. Its

technology is widely used in the industry for continuous production of CNTs with the catalyst ferrocene, stable, non-toxic, and low cost (De Volder et al., 2013).

Some of the applications for MWCNTs are in lithium-ion batteries for notebook computers and mobile phones. It is well known that CNTs increase electrical connectivity and mechanical integrity, which improve rate capability and life cycle. The use of CNTs as a catalyst support in fuel cells can reduce Pt usage by more than 58% compared to carbon black, and doped CNTs may enable fuel cells that do not require Pt (Igbokwe et al., 2019).

According to reports, many companies were investing in diverse applications of CNTs, such as transparent conductors, thermal interfaces, antiballistic vests, and wind turbine blades. Further industrial development demands health and safety standards for CNT manufacturing and use. As more substantial quantities of CNT materials reach the consumer market, it will be necessary to establish disposal and use procedures. CNTs may enter municipal waste streams, where, unless they are incinerated, cross-contamination during recycling is possible (Charbgoon et al., 2017).

Metal ENPs have been utilized in fields like drug delivery systems, medical diagnosis, sanitization, and wastewater treatment. Recently different ENPs, in suspension, are in use to induce the catalytic reactions, which have been proved to be the most effective method for treating wastewater. As gold and silver, noble metals are abundantly used in the fields of photography, biological labeling, and catalysis. The high surface area of Ag-ENPs facilitates it to be used as a catalyst in reducing the organic dyes (Maboya et al., 2019). Also, Ag-ENPs have emerged as an attractive alternative approach for antimicrobial and cancer treatments. Nonetheless, their toxicological effects and low biocompatibility limited their clinical applications (Vijayakumar et al., 2020).

In some studies, the Ag-ENPs have demonstrated their toxic effects on bacteria, mammals, animals, cells, and plants. Researchers claim that toxicity to the strong affinity to thiol groups presents in the cells. To decrease the toxicity of these ENPs, they have been biosynthesized using *Veronica officinalis* extract. These did not exhibit any toxic effects and were spherical and locally agglomerated about 40 nm, affecting their lack of toxicity. However, Ag-ENPs

stimulated the growth of garden cress, inhibited it, and stimulated it again (Srivastava & Choubey, 2019).

The methods utilized for Ag-ENPs synthesis are the physical vapor deposition, chemical vapor deposition, sol-gel method, microwave-assisted synthesis, ultra-sonication method, electrochemical synthesis, and chemical reduction of metallic ions. The major disadvantage of using these methods is that the chemicals used in these procedures are costly and toxic, making them hazardous. Another advanced biosynthesis method lately used is bioreduction, in which bacteria, fungi, or plant extracts are used to synthesize nanomaterials. In comparison to all the methods mentioned before, bioreduction is eco-friendly and easy to execute. Also, the Ag-ENPs obtained exhibit more excellent stability utilizing a green synthesis using methanol extract of *Citrus limetta* peel (Maboya et al., 2019); silver ions yield high biocompatible that can be handled in biomedical settings without significant adverse effects (Vijayakumar et al., 2020).

Plant extracts frequently offer proper manipulation and control over crystal growth and stabilization. Nonetheless, using different plant extracts and reductants in the ENPs synthesis is challenging. Biosynthesis was performed to obtain ENPs with the desired shape, size, and dispersity, using plant extracts. It is a more straightforward method and is used in many therapeutic applications. Some plants, such as alfalfa, aloe vera leaves, *Azadirachta indica* (neem) leaves, *Camellia sinensis*, *Capsicum annum* L., *Cinnamomum camphora* leaves, *Datura metel*, *Embllica officinalis* fruit, *Ocimum sanctum*, and geranium leaves, have been employed in the ENPs biosynthesis (Srivastava & Choubey, 2019).

3.2 Harnessing and Primary Uses of Engineering Nanomaterials for Commercial Mass Production

ENMs have been used in cutting-edge technologies such as high-tech electronic devices, advanced chemical processes, or specialized medical treatments. However, these ENMs and their commercial applications could be bought in a popular supermarket into products used in daily life by an ordinary citizen. Therefore, nanoscience and nanotechnology are standard components of thousands of products that people use every day.

TiO₂-ENM have been used in a dye-sensitized solar cell, synergetic degradation of antibiotics,

photocatalytic hydrogen production, adsorption of organic pollutants, photocatalytic degradation of pharmaceutical and personal care products, CO₂ photoreduction, sensor for the detection of organic compounds or gases, photocatalytic degradation of gaseous compounds, photocatalytic removal of organic and inorganic pollutants, high-performance electrode for supercapacitors, anode material for high rate lithium-ion batteries, degradation of organic dye molecules, photoreduction of heavy metals (HMs), photocatalytic activity for *E. coli* disinfection and biological pollutant degradation, highly efficient artificial photosynthesis, artificial N-fixation, photocatalytic water splitting, photothermal therapy, photocatalytic water disinfection, solar water evaporation, antibacterial applications, renewable energy applications, ultrafast rechargeable lithium-ion batteries, high power sodium-ion anodes, membrane distillation, paints, medical implants, medical devices, cosmetics, food packaging, control of crop pathogens in agriculture, and nanofertilizer, among others.

Ag-ENMs are useful for photocatalytic decomposition/reduction of dyes, energy-saving windows, conductive inks for wearable electronics, supercapacitor in wearable energy devices, polymer solar cells, satisfactory thermal conductivity, and high thermal stability, antibacterial activity, edible coating, wearable, and implantable bioelectronics, electrocatalytic N₂ fixation to NH₃ under ambient conditions, electrochemical sensors, an electrocatalyst for the hydrogen evolution reaction, CO₂ reduction to CO, methanol or CH₄, ammonia gas sensor, MCF-7 breast cancer cells imaging, plant nutrition, crop protection, and food packaging, among others.

Zn-ENMs are used in crop production, bifunctional electrocatalysis, rechargeable batteries, high-performance supercapacitor electrodes, generation of hydrogen fuel in acid medium, photocatalytic reduction of CO₂ into renewable hydrocarbon fuel, Zn-Air batteries, organic thin-film solar cell, water splitting, adsorbent of pollutants, removal of dyes, bio-medical applications, kidney cancer therapy, and antibacterial medicine, among others. Si-ENMs are used in high-performance anodes for lithium-ion batteries, biomedicine, food pathogen detection, crop production, quantitative detection of pesticides, suppression of plant diseases, and multichip packaging, among others. Fe-ENMs are used in photocatalytic degradation of bisphenol, intelligent packaging film,

radiosensitizing medicine, medical nanofabrication, magnetic hyperthermia, magnetic resonance imaging in the innovative drug delivery, diagnosis, treatment, and therapy of tumor and cancer diseases, degradation of dyes, groundwater treatment, remediation of drinking water, elimination of environmental pollutants such as HMs, hydrocarbons or organic compounds, magnetic adsorptive-remediation of HMs, crop production, nutrient of crops, control of pest and diseases, and thin films for identification of pathogens, among others.

Al-ENMs applications are used in wound infection and drug-resistant clinical pathogens, removal of bacterial pathogens from wastewater, adsorption of organophosphate pesticides, removal of organic or pharmaceutical pollutants, degradation of dyes, detection of pollutant gases, and controlled release of plant growth regulators, among others. Ce-ENMs are used in photodecomposition and adsorption of hazardous organic pollutants, pesticide adsorption, crop nutrition, cosmetic packaging, and photocatalytic activity for antibiotic removal, among others. CNT-ENMs are used in selective photocatalytic CO₂ reduction via dry reforming of methane to fuels, hydrogen evolution reaction, versatile energy storage, hybrid energy storage device, medical applications, ecofriendly and reusable catalyst, removal of organic pollutants, HMs removal, biosensors, selective chemical sensors, electrochemical detection of HMs, antimicrobial agent, nanoagriculture, green carrier material for pesticides controlled release, membrane for separation and inactivation pathogens, electrochemical immunosensor for multiplexed detection of food-borne pathogens, food packaging, determination of trace pollutants in food, and drug controlled release, among others.

3.3 Ecological Drawbacks

The high production volume, growing number of industries competing in nanotechnology, and increasing use of ENMs make their ecological risk a concern worldwide. Consequently, there is a prominent concern about ENMs because they have the potential for adverse effects after they are spread, but their release pathways, fate, and toxicity are not yet well known. The detrimental influences of graphene-related materials on soil–plant organisms from a terrestrial ecosystem were summarized by Yang et al. (2020), taking into account the perspectives of physiology,

biochemistry, and gene expression. Besides, the risk to the aquatic ecology of Al₂O₃-ENPs with two crystalline phases (alpha-Al₂O₃ and eta-Al₂O₃) was studied by Nogueira et al. (2020). They verified the toxic potential of the ENPs toward *Daphnia magna* even in sublethal concentrations. Besides, the toxicity was associated with an increase in the reactive oxygen species (ROS) content and deregulation of antioxidant enzymatic/non-enzymatic. However, their results indicated that Al₂O₃ NPs stand for a minimal risk to organisms in surface waters. A study to utilize a small-scale rapid assay was designed to evaluate the fate and ecotoxicological impacts of copper-based ENPs (Cu and CuO) in a diverse aquatic community (Wu et al., 2020a, b). They found higher uptake in *D. magna* and zebrafish (*Danio rerio*) than equivalent ionic exposures, suggesting that organisms took up Cu-based ENPs.

Similar studies have been carried out with soil and water microorganisms. The potential risk of short-term exposure to environmentally realistic concentrations (0.00, 0.04, 0.08, and 0.12 ppm) of CuO-ENM on the ecological performance of a cosmopolitan freshwater fungus (*Articulospora tetracladia*) was studied for a short period (1, 2, 3, and 4 days) (Seena & Kumar, 2019). They stated that leaf litter decomposition and sporulation were affected by concentrations and exposure time of CuO-ENMs. It is well known for the complexity of evaluating the ecosystems because they contain hundreds of organisms that work together, and some of them are in the same food chain. Therefore, some considerations for improving the evaluation of the ecological hazards of ENMs were stated by Holden et al. (2016a, b), and they proposed three main recommendations. First, scientists should develop ecotoxicological tests with realistic exposure scenarios. Second, testing should proceed via tiers with iterative feedback that informs experiments with higher organisms. Finally, the ENM hazard assessments should involve greater coordination among ENM quantitative analysts, exposure modelers, and ecotoxicologists across government, industry, and academia (Holden et al., 2016a, b). Besides, the effects of ENMs on populations, communities, and ecosystems have been shown by Holden et al. (2016a, b), including transfer and accumulation of ENMs through food chains.

It has to be stated that ecological advantages regarding the use of ENMs have also been found.

SiO₂-ENMs were used as an additive to the corn oil methyl ester. The experimental results showed that the incorporation of ENMs had a positive effect on emission characteristics as ENMs act as an oxidation catalyst (Saravankumar et al., 2019). An ecological assessment of the photocatalytic degradation of 2,4-dinitrophenol (DNP) using a three-dimensional (3D) hybrid composite based on carbon nano-onion-functionalized zinc-oxide tetrapods was performed via the germination of common gram seeds (*Cicer arietinum* L.) (Park et al., 2019). This study revealed the low toxicity and environmental safety of the synthesized hybrid 3D composites and their ability to decontaminate wastewater using photocatalytic technology. An additional advantage of ENMs was reported by Qian et al. (2018). They used soil amended with SWCNTs at concentrations of 3 and 10 ppm to prove that ENMs change the microbial community composition, promote soil organic degradation, and improve soil fertility in a short time (Qian et al., 2018).

Nanoscience and nanotechnology have produced ENMs at commercial levels, which are now used in over 3000 products, as their functionalities and levels of complexity have been improved. ENMs are increasingly entering the environment with uncertain outcomes, including potential ecological effects. Therefore, there is a critical need to know the environmental, ecological, and health implications of ENMs. However, the current results are controversial in the ecological context because ENMs and species discussed above were quite different.

3.4 The Collateral Effect of ENMs on Organisms

We identified 1504 manuscripts from the Web of Science® database. However, there were only included 73 papers regarding plants, 58 about meso and macrofauna, and 51 regarding mycorrhizal fungi. In general, there is information on the effects of ENMs on plants and soil organisms that is controversial. Of selected articles that evaluated the effects of ENMs in plants, 43.83% reported positive effects, 34.24% negative, and 21.91% no effects. Regarding the impact on macrofauna and mesofauna, 20.68% reported no effect, while 79.31% reported adverse effects. ENMs on arbuscular mycorrhizal fungi and their association with plants had positive effects in 15.68% of studies.

However, 54.9% stated adverse effects and 29.41% no effects (Fig. 2). Therefore, the averages of the effects indicate that 23.62% of organisms improved their characteristics when they were spiked with ENMs, 54.39% were affected, and only 21.97% did not change any properties or characteristics spiked by ENMs. As for the experimental scenarios (laboratory, greenhouse, and field), for the select studies, mostly the investigations were carried out in laboratory conditions. The number of studies for each group evaluated is shown in Fig. 3. Besides, we found that studies regarding mycorrhizal fungi and meso and macrofauna were conducted on natural soil. However, the experiments were developed in agar and hydroponics medium for plants, mainly (Fig. 4).

3.5 Effect of ENMs on Soil Microorganisms

ENMs are rapidly integrated into industrial technologies, including agricultural production alternatives, such as nanofertilizers and pesticides. Being photocatalysts presents favorable oxidation reactions. The production and application of ENMs have high production figures and have been raised as an emerging and potential threat to natural ecosystems. Currently, the statistics for the concentration of these materials in the environment have been estimated. In Europe, the involuntary disposition of ENMs in soils through the application of sludge maintains a flow of 2380 t⁻¹ of TiO₂ (Sun et al., 2014). However, the actual levels of their concentration in soils are unknown with precision, both at local and global levels, but

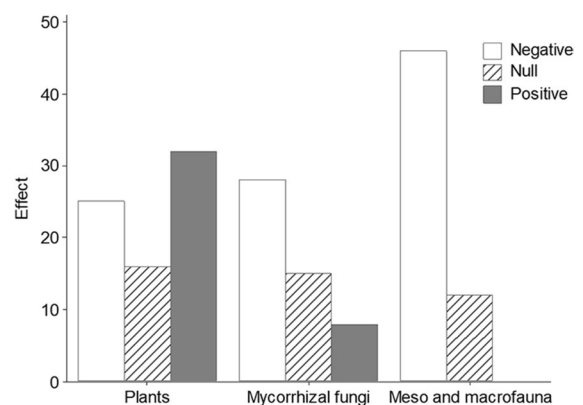


Fig. 2 Effects caused by ENMs on plants, meso and macrofauna, and mycorrhizal fungi

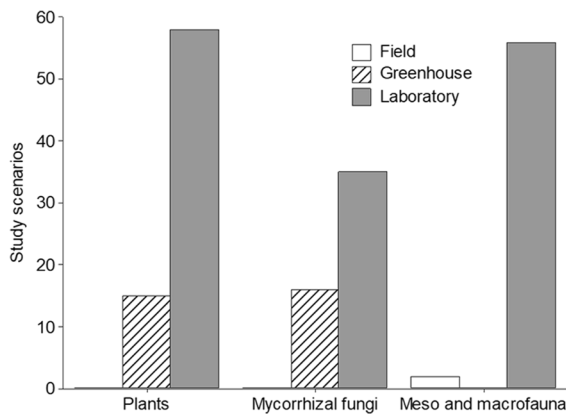
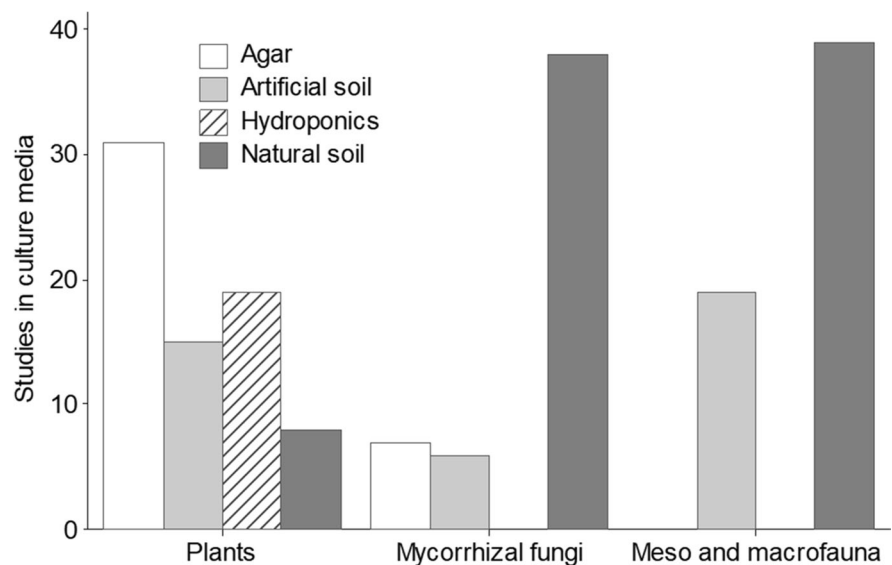


Fig. 3 Scenarios where the effects of ENMs on plants, meso and macrofauna, and mycorrhizal fungi were evaluated

concentrations are higher than water and air (Wang et al., 2016a, b).

When evaluating the effects of ENMs, microorganisms are considered biomarkers of soil quality. The biosecurity research of ENMs uses multivariate study strategies, from in vitro system evaluation to in vivo evaluation microcosms; it is the latter studies that direct the safe use of ENMs. Initially, evaluations of ENMs were carried out under artificial cultivation conditions, such as hydroponic issues or in inert artificial substrates, which ignored microbial responses in the soils of agroecosystems, the response criteria of the beneficial microorganisms due to the incidence

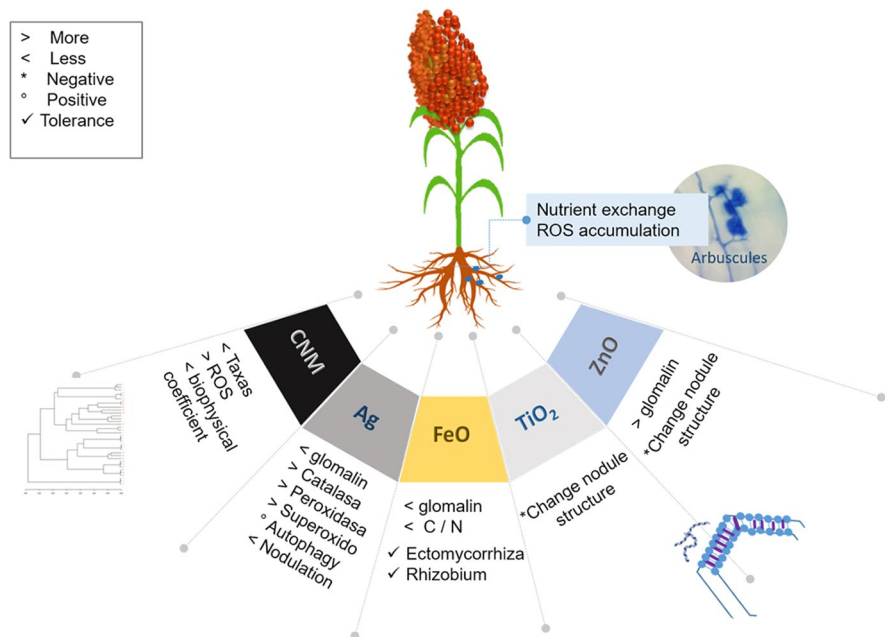
Fig. 4 The number of studies in culture media where the effects of ENMs on plants, meso and macrofauna, and mycorrhizal fungi were evaluated



of ENMs (Fig. 5) (Priester et al., 2012). It is crucial to consider the soil matrix in particles in each case, with its specific properties, scaling the study to counteract the data obtained from in vitro tests to microcosms. In the study of ENMs Ag and Al₂O₃ in *Bacillus cereus* and *Pseudomonas stutzeri*, using molecular methods, the results of the in vitro test revealed cytotoxic effects, which related to the increase in the genetic expression of the KatB gene; however, the results of the genetic transcription of the in vitro experiment did not match the results of the microcosm experiment (Fajardo et al., 2014). Similarly, the evaluation of Fe₃O₄-ENPs through enzymatic monitoring of *E. coli* reported a significant attenuation of the toxicity of ENPs in microcosms, contrary to what was observed in suspended cultures (Mahanty et al., 2019).

Within the evaluation conditions, exposure time plays an important role. Besides, metagenomic has been proven to be an essential contribution tool to identify the effects of ENMs in terms of microbial biodiversity in soil matrices, allowing the evaluation of genetic sequences and fragmentation of these using DNA and PCR extractions. The effect of TiO₂-ENPs on the taxonomic composition and the functional profiles of the bacterial community of the soil matrix over time show the inhibition of taxa by TiO₂-ENPs at a short-term exposure time (1 day) even in the low dose (1 mg kg⁻¹). However, in the long term (60 days), these adverse effects at low concentration disappeared, reporting an increase in the abundance

Fig. 5 Effects of engineered nanomaterials mostly evaluated in agricultural soils and the response of beneficiary microorganisms



of taxa according to the increase in exposure dose (500 and 2000 mg kg⁻¹). Concerning the functionality of the community, the enzymatic activity did not change significantly (60 days), which suggests that the dehydrogenase activity promoted the tolerance of microorganisms in the presence of TiO₂-ENPs (Zhai et al., 2019). The effects of ZnO- and CuO-ENMs (10 mg kg⁻¹) on soil microbial community in different soil types were witnessed by the dehydrogenated activity, which was affected by the presence of Cu after 90 days of incubation (Josko et al., 2019).

A long-term (60 days) evaluation of the effects of TiO₂-ENPs on microbial functionality was carried out in an artificial humeral microcosm. It was demonstrated that TiO₂-ENPs decreased the amount relative of functional genes, i.e., the nitrogen metabolism's functional genes and the glycolysis process were negatively affected. Thus, nitrogen contamination efficiency was reduced to 35–51% (Yang et al., 2020). Although the decline of taxa is observed in the effects of TiO₂-ENPs, it is associated with nitrogen fixation and oxidation of methane, which are specific actions of rhizobial bacteria. In the case of ZnO-ENPs, a decrease in functional taxonomic abundance was also demonstrated (Ge et al., 2011). However, they increased the presence of taxa (Sphingomonadaceae, Streptomyetaceae, and Streptomyces) related to the

decomposition of recalcitrant organic pollutants (Ge et al., 2012).

The influence of CeO₂-ENPs on soil quality is related to the loss of enzymatic activity (beta glucosidase and urease), which is reduced as the concentration of ENMs increases (Li et al., 2017). In terms of cellobiohydrolase (CBH), β-1,4-xylosidase (XYL), β-1,4-glucosidase (GLU), β-1,4-N-acetylglucosaminidase (AGA), and acid phosphatase (AP), the concentrations of AP, GLU, and AGA decrease in the presence of Ag-ENPs (100 mg kg⁻¹) in soils amended with biosolids. Simultaneously, silver as bulk material (Ag⁺) caused the decrease of all enzymes in a correlated effect with the dose. It concluded that the dissolution factor of the ENPs reduces organic molecules' synthesis because only 37% of Ag-NP managed to dissolve after the end of the test time (Asadishad et al., 2018).

The effects of CNTs have been more related to the interruption of nutrient cycles in the soil by modifying the carbon and nitrogen cycle pathways (Wu et al., 2020a, b). The biophysical coefficient (BQ), as an indicator of changes in organic matter in the soil, is interpreted as the amount of CO₂ involved in the metabolism of respiration concerning the amount incorporated in the biomass. In soils spiked with SWCNT or MWCNT, the BQ has lower values in the presence of both materials, at a concentration

of 1000 mg kg⁻¹, compared to the levels reported by fullerene doses (1, 10, 100, and 1000 mg kg⁻¹) (Oyelami & Semple, 2015). In terms of microbial biomass, the values are reduced even in doses below 500 µg g⁻¹ MWCNT in soil with a high percentage of sand (60.75% sand, 27.75% silt, and 7.50% clay) (Chung et al., 2011).

Oxidative stress is presumed as a critical mechanism of carbon ENMs in antibacterial applications. In graphene oxide (GO), the induced stress is lower in different species of *Bacillus* (Gurunathan, 2015). Besides, GO alters soil's microbial composition, but it does not fracture alpha diversity (Forstner et al., 2019). However, some authors consider that bulk materials, as positive controls, may incur false negatives since the bulk material has a solubility that ENMs do not reach, thus differing in terms of absorption and adsorption. It has also been considered that ENMs do not present complete bioavailability in the soil (Judy et al., 2015a, b).

Using humic acid as a suspending agent for ENMs could reduce the availability of free metal ions, reducing toxicity values. Regarding the bioavailability of Ag-ENPs, humic acids have reduced ligands to Ag⁺ ions capturing less Ag compared to the cysteine that dominates the absorption of Ag-ENPs, reducing its bioavailability (Luoma et al., 2016). The genetic study of the effect of Ag-ENPs applied in humic acid suspension showed the expression of genes classified in the category of extracellular sequestration of metal ions, as is the case of OpgC. This protein takes possession of periplasmic glucans, achieving a negative charge potential for immobilization of metal ions (Meier et al., 2020). Within the modifications to the microbial community, the appearance of silver-tolerant bacteria is relevant. Before exposure concentrations within the range 49 to 287 mg kg⁻¹, *Rhodanobacter* sp. was dominating 71% of the total bacterial population (Samarajeewa et al., 2017).

Similarly, Acidobacteria, Cyanobacteria, Bacteroidetes, and Nitrospirae have relative resistance to Ag-ENMs, at doses with 10, 50, and 100 mg kg⁻¹ (Yang et al., 2014). Electron microscopy techniques have shown how Ag⁺ ions cause severe damage to the cell wall in the case of *Nitrosomonas europaea*, exhibiting the disintegration of nucleoids (Wang et al., 2017), when ENPs entered through the cell wall. Besides, Ag-ENMs accelerate cell death, promoting cell death and loss of bacterial species such as *Bacillus*

megaterium and *Bacillus thuringiensis*. The release of Ag⁺ ions due to the joint disposition of biosolids and Ag-ENPs is reported higher than in the particular case of the AgNO₃ bulk material, with a difference of five times greater, ranging from 123 to 138 mg of Ag kg⁻¹ (Mirzajani et al., 2013).

In a load effect, the ENMs TiO₂ and Fe₃O₄ were compared, both negative (-) and positive (+) charges, suggesting that the loading force is a function of context. Thus, Fe₃O₄ (-) decreased *Rhizobium* colonization compared to Fe₃O₄ (+) without this property, affecting the presence of arbuscular mycorrhizal fungi (AMF) communities at the root as reported by the multiple response permutation studies (MRPP) (Burke et al., 2015). Although the surface charge of the bacteria is affected by the ENMs depending on the pH of the study medium, extracellular polymeric substances (EPS) have been successfully evaluated in cellular protection against the toxicity of CuO- and ZnO-ENPs. Therefore, ionic chelators maintain an essential role in cellular response. In the case of CuO-ENPs, alkaline phosphatase values were lower in evaluating the bacterium *Pseudomonas chlororaphis* O6, indicating minor damage to the outer membrane (Dimkpa et al., 2011a, b). In *Pseudomonas chlororaphis* O6, the EPS were polydispersed with a negative charge, which was neutralized by the effect of Ag ions and not by Ag-ENPs, since the zeta potential reported the ENMs charge as negative. As for the solubility of ENMs based on 10 mg L⁻¹, the pH was an essential factor in the value of the concentration of Ag. At pH six, it decreased to 0.28 mg L⁻¹, while at pH seven, the level detected was higher (2.3 mg L⁻¹) (Dimkpa et al., 2011a, b).

In particles, the soil rhizosphere, the main nutrient storage area for crop production processes due to its dynamic interaction between roots and microorganisms, represents soil's functional biodiversity. Any change in the processes related to the nutrient cycle modifies the function and regulation of the ecosystem. In discussing the direct and indirect effects of ENMs on soil's microbial functional diversity, it has highlighted the role of plant growth-promoting microorganisms and those of symbiotic association, particularly bacteria of the genus *Bacillus* and *Rhizobium*, as well as AMF. These groups stand out for the degree of tolerance they present against different biotic and abiotic stress types. It is well known that ENMs generate reactive oxygen species (ROS)

that potentiate oxidative stress, primarily damaging cellular structures (Zhang et al., 2011a, b). Quantitative proteomic analysis has allowed us to assess the expression of proteins involved in the detoxification of ROS. In *Bacillus subtilis*, the values increased in sebum exposures to Mg-ZnO-ENPs (Auger et al., 2019). For *Bacillus thuringiensis*, 34 possible sensitive proteins were identified, and the accumulation of envelope precursors, material involved in oxidative stress tolerance (Mirzajani et al., 2014). The absence of effects has been reported by the TiO₂-ENPs (10, 100, and 1000 mg kg⁻¹), CeO₂-ENPs (860 mg kg⁻¹), and MWCNT (3 and 300 mg kg⁻¹), when indirectly assessing the activity of the AMF, utilizing the phosphorus content in red clover plants (*Trifolium pratense* var. Merula). There were no significant differences with the control; however, the percentages of fixed nitrogen were modified favorably, indicating high biological activity (Moll et al., 2016).

The AMF are forced biotrophs that can tolerate stress conditions and generate association with 80% of terrestrial plant species, including the main grain crops. In general terms, the AMF and plant association are characterized by the presence of extraradical hyphae, intraradical hyphae, arbuscules, vesicles, and spores of extraradical hyphae. By expanding in the rhizosphere, they contribute to the mobility of mineral nutrients such as N, P, Cu, and Zn. Arbuscules are structures of high activity in the exchange of nutrients that occur between symbionts. Their presence refers to the accumulation of ROS, such as hydrogen peroxide, while colonization, in particular, is related to enzymatic activities of antioxidant type (Wu et al., 2014). Cu-ENMs produce ROS by reacting with amines and carboxyl groups that make up the cell surface.

Moreover, beneficial fungi are not modified compared to bacterial communities (Parada et al., 2019). It implies that ENMs affect the ecology of bacteria but not of fungi. However, in general, research studies of the effects of ENMs on AMF symbiosis mention contradictory indices regarding the inhibition in the percentage of radical colonization. TiO₂-ENPs were evaluated through the microbial composition analysis in *Zea mays* L. and *Glycine max* L. plants performing the DNA extraction method with PCR amplification (Table 1), in the direction of the 18S rRNA region specific for AMF. It was shown that the presence of TiO₂-ENPs (200 mg Kg⁻¹) generated a

weak inhibition of the AMF communities, attributing the effect to the high concentration of Ti detected inside the roots (Burke et al., 2014). Contradictorily, the impact of TiO₂-ENPs (1, 100, and 1000 mg kg⁻¹) evaluated in wheat (*Triticum* ssp. Var. Fiorina), in a study aimed at the 16S rRNA region, reported 1295 operational taxonomic units, focusing on the edge Glomeromycota, without identifying the inhibitory effect of fungal diversity compared to the control treatment (Moll et al., 2017).

There exists one degree of interference in the symbiotic relationship because some toxins are expressed more sensitive to the effects of ENPs. For *Glomus*, the study targeting the 16S rRNA and 18S rRNA gene, high Fe₃O₄ concentrations exerted greater relative abundance in the *Glomus* genus and lower abundance in the case of Acaulosporaceae and Gigasporaceae. However, root colonization decreased, indicating that AMF and plant symbiosis was negatively affected (Cao et al., 2016). It follows that AMF operates detoxification strategies through intracellular compartmentalization mechanisms, by translocating excess metals to subcellular compartments and even to specific structures such as spores and vesicles, losing cell viability (Cornejo et al., 2013). Relief of heavy metal toxicity can be attributed to the effects of AMF mediated nutrition (Conversa et al., 2019).

There are a more significant number of studies (Table 1), confirming that AMF inoculation under ENMs exposure has a positive stimulus for plants. It implies that AMF is relieving nanotoxicity, reducing the bioavailability of metals, increasing the content of photosynthetic pigments and antioxidant capacity, and reducing ROS concentration. In the culture of *Zea mays* L., colonization by *Glomus versiforme* (Gv) and *Glomus caledonium* (Gc) was evaluated directly by the intersection method and indirectly by the root activity parameter with triphenyl tetrazolium chloride. At a high dose of ZnO-ENPs (800 mg kg⁻¹), the colonization rate decreased, especially in the case of Gv. Moreover, it was found that AMF significantly alleviated the phytotoxicity induced by ZnO-ENPs by generating an increase in antioxidant activity, having some tolerance to stress by ZnO-ENPs (Wang et al., 2016a, b). Similarly, *Funneliformis caledonium* was studied in sweet sorghum (*Sorghum bicolor* L.), and AMF inoculation helped to reduce the adverse effects of synergistic toxicity of ZnO-ENPs (50, 250, and 500 mg kg⁻¹) with cadmium (Wang et al., 2018a, b).

Table 1 Effects of ENMs in arbuscular mycorrhizal symbiosis

Nanoparticle	Microbial specie	Evaluation technique	Study focus	Effect degree	Reference
TiO ₂	Autochthonous microbes soil	TRELP, PCR, MRPP*	Composition microbial soil	TiO ₂ influenced mycorrhizal fungi, but changes in the composition of microbial communities may not affect plants	Burke et al. (2014)
ZnO	<i>Glomus versiforme</i> and <i>G. caledonium</i>	BCE, ROS, SOD	Root colonization	ZnO (800 mg kg ⁻¹) produced root colonization inhibition; however, inoculation significantly reducing the accumulation of ROS	Wang et al. (2016a, b)
ZnO	<i>Funneliformis caledonium</i>	ICP-AES, phosphatase activity	Synergistic effect of root colonization	Inoculation relieves the synergistic effects of pollutants (Cd)	Wang et al. (2018a, b)
ZnO	<i>Funneliformis mosseae</i>	ICP-AES, phosphorus uptake	Root colonization and phosphorus acquisition	Inoculation reduced the absorption of Zn ions and promoted phosphorus absorption and, synergistically, the growth of the plant	Wang et al. (2018a, b)
ZnO	<i>Glomus intraradices</i>	ICP-OES, TEM	Glomalin content, root colonization	Glomalin production reduced the absorption of the Zn ion; mycorrhiza and rhizobium are considered to have a symbiotic association	Siani et al. (2017)
Fe ₃ O ₄	<i>Glomus caledonium</i>	TEM, microcalorimetric measurements, DNA and PCR	Bacterial community, composition, microbe interactions, root colonization	Microbial association reduced the harmful effects of nanoparticles	Cao et al. (2016)
Fe ₃ O ₄ , Ag	<i>Glomus caledonium</i>	Enzyme determinations, GRSP, X-ray micro-CT	Glomalin content, root colonization	Nanoparticles influenced mycorrhizal growth and function. They decreased enzymatic activity and root growth	Feng et al. (2013)
TiO ₂ , CeO ₂ , MWCNT	Autochthonous microbes soil	Nitrogen fixation, phosphorus content	Root colonization	MWCNTs (3000 mg kg ⁻¹) slightly increased nitrogen fixation. Colonization was not negatively affected by nanoparticle exposure	Moll et al. (2016)

Table 1 (continued)

Nanoparticle	Microbial specie	Evaluation technique	Study focus	Effect degree	Reference
AgS	Autochthonous microbes soil	Enzyme activity, SOD, ICP-MS, DNA extraction, PCR	Mycorrhiza communities	Ag decreased the mycorrhizal colonization and alkaline phosphatase activity. The relative abundances of <i>Diversispora</i> , <i>Acaulospora</i> , and <i>Scutellospora</i> decreased by Ag-ENPs	Cao et al. (2017)

*TRFLP terminal restriction fragment length polymorphism analysis, MRPP multi-response permutation procedures, PCR polymerase chain reaction, BCF concentrations and bioconcentration factor, ROS reactive oxygen species, SOD superoxide dismutase, ICP-MS inductively coupled plasma mass spectrometry, ICP-AES inductively coupled plasma atomic emission spectroscopy, GRSP glomalin-related soil protein, M-CT a micro-computed tomography

It is suggested that phytotoxicity relief is related to organic phosphorus. The inoculation of *Funneliformis mosseae* under different levels of organic phosphorus (0, 20, or 50 mg kg⁻¹) showed that the effects were synergistic, keeping symbiosis stable in the presence of ZnO-ENPs (500 mg kg⁻¹) (Wang et al., 2018a, b). More importantly, the promotion of metal chelators generated by the AMF symbiosis with the plant, such as glycoproteins, specifically glomalin, was attributed to the significant metal absorption reductions. The total glomalin production was studied by UV-Vis 595 nm spectrophotometry (Table 1), exposing the inoculation of *Glomus* intraradices to ZnO-ENPs in the fenugreek culture (*Trigonella foenum-graecum*). The results showed that the glomalin content increased 15% in comparison with controls without inoculum exposed to 500 µg g⁻¹ of ZnO-ENPs (Siani et al., 2017). In contrast, the Ag-ENPs (0.01 mg kg⁻¹) and FeO-ENPs (3.2 mg kg⁻¹) significantly reduced the production of glomalin in the symbiosis *Glomus caledonium* and clover (*Trifolium repens*); however, the rate of colonization increased according to the content of Ag-ENPs as well as the production of antioxidant enzymes such as catalase (CAT), peroxidase (POD), and superoxide dismutase (SOD) (Feng et al., 2013).

The ecological reduction of functions of AMFs has been observed when AMF is spiked with Ag-ENPs, through the characterization of alkaline phosphatase activity and the abundance of AMF-taxa. The diversity of AMF and the microbial community's composition was closely related to the content of dissolved organic carbon (DOC). It could be explained because *Glomus* increased as it is a genus that colonizes even through micellar fragments, contrary to the *Acaulospora* and *Scutellospora* genera that decreased at high concentrations of Ag-ENMs (Cao et al., 2017). Given the exposure of nanoscale zero-valent iron (nZV), AMF promotes apoplastic barriers in the radical apex, thus reducing ion translocation. Besides, AMF increased the protection against abiotic stress through the improved production of glomalin glycoprotein and cytokine and gibberellins in synergistic effect with inocula of rhizobacteria (Mokarram-Kashtiban et al., 2019). The above suggests that the toxic effects on microbes could be selective and depend on both the type of nanoparticles and the microbial communities' composition.

In evaluation scenarios, where it is sought to have more significant similarity to the agricultural production scenarios, ENMs with coating design have been studied in conjunction with incorporating biosolids (sludge) to the soil, monitoring the properties of ENMs through aging. In evaluating the coatings with polyvinylpyrrolidone (PVP), it has been detected that Ag-ENPs with sulfurations affect the symbiotic plant-AMF interactions of importance for agricultural production systems, at concentrations of 1 mg kg^{-1} . However, it is accused that bulk Ag ions cause more significant risk effects when incorporated into the soil employing biosolids (Judy et al., 2015a, b).

Ectomycorrhizal symbiosis is another type of mutualism that only occurs with 3% of terrestrial plant species; forest trees are potential hosts for this type of symbiont. Ectomycorrhizal fungi mediate responses to metal toxicity, water stress, and acidifying effects of the soil. When the impact of Ag- and Cu-ENPs on ectomycorrhizal colonization in English oak (*Quercus robur* L.) was evaluated, four ectomycorrhizal taxa were detected: *Sphaerospora brunnea*, *Thelephora terrestris*, *Paxillus involutus*, and *Laccaria proxima*. In general terms, the foliar applications did not reduce the ectomycorrhizal infection. In concentrations of 25 ppm, both ENPs showed a higher degree of mycorrhization, being a value of 37.1% for Cu- and 37.5% in Ag-ENPs (Aleksandrowicz-Trzcinska et al., 2018). Another study reported that carbon and nitrogen concentrations were not affected concerning microbial biomass; moreover, the C/N ectomycorrhizae ratio was directly affected by the increase in levels of Fe₃O₄- and SnO₂-ENPs (Antisari et al., 2013).

In the effects of TiO₂-ENPs concerning *Rhizobium leguminosarum*, changes in cellular morphology have been detected in the nodules. It retards the processes of nitrogen fixation, suggesting an intervention in the response of the signals by the host (Fan et al., 2014). Similarly, exposure to ZnO-ENPs reports the change of cellular morphology toward round structures, affecting the onset of the nodulation and causing early senescence of the nodules (Huang et al., 2014). Ultra-high-resolution micrographs by scanning transmission electron microscopy (STEM) demonstrated the permanence of CeO₂- and ZnO-ENPs on the bacterial cell surface. ZnO-ENPs were interned in the periplasmic space. Besides, the results of Fourier-transform infrared spectroscopy (FTIR) showed the change of extracellular polymeric substances before exposure to both materials (Bandyopadhyay et al., 2012). The surface charges of

the ENMs were a factor that could influence the colonization of the radical systems by fixing bacteria (Burke et al., 2015). The nodulation inhibition is reported in treatment with TiO₂-, ZnO-, and Ag-ENPs to a greater extent than its bulk control.

In the same way, the genetic expression drops more than ten times in 239 plant genes, such as the *Medicago truncatula*. It is presumed that the biological pathways expressed are related to nitrogen metabolism, nodulation, heavy metal homeostasis, and abiotic stress response (Chen et al., 2015). Ag-ENPs at doses of $800 \mu\text{g kg}^{-1}$ decreased the nodulation, reducing nitrogen fixation activity by deteriorating cytoplasmic components in the induction of autophagy processes (Abd-Alla et al., 2016).

The analysis of operational taxonomic units (OTU) allowed reviewing the *Rhizobium* genus' taxonomic diversity, determining the value of species inhibited by ENMs, in exposure to ZnO- and CeO₂-ENPs. The OTUs decreased in soils with cultivation contrary to those not planted, which suggests that plants accelerate the biogeochemical activity in the soil, promoting or limiting the effect of ENPs (Ge et al., 2014).

3.6 Effect of ENMs on Plants

Plants are organisms of significant environmental and nutritional importance; since, on the one hand, they manage to keep ecosystems in balance, also they promote critical biological processes such as the carbon cycle through the process of photosynthesis or as regulators of global climate change. On the other hand, these organisms, in addition to their importance in the environment, the function of satisfying the basic needs of humans and animals such as food since plants can obtain nutrients such as carbohydrates, fats, and essential proteins for its development (Fernando, 2012).

The release of ENMs is projected as a problem of environmental pollution due to crop production use. Crops are essential because they are the source of food and raw materials for almost 100% of people and more than 50% of industrial processes, respectively. However, most of the studies regarding the use of ENMs in agriculture have been carried out under controlled conditions, i.e., under greenhouse or plant-growth chamber conditions, during the first phenologic stages, i.e., they are only short-period studies. Besides, it has been reported that ENMs can

be phytotoxic for several plant species, so that evaluations of different ENMs have been carried out at morphological, physiological, and biochemical levels, in different plant species (Rajput et al., 2018; Mansoor et al., 2019; Du et al., 2017). Therefore, studies at land conditions are necessary throughout the life cycle crops to understand the benefits or risks that ENMs could imply for ecosystems. It has to be stated that these studies have to be led by scientists and technologists with different specialties to analyze the problem from different approaches. Currently, the main ENMs are also used to test their effects on crops trying to get agricultural advantages from these developments.

3.7 Effects of TiO₂-ENPs on Plants

The effects of TiO₂-ENPs on crops depend mainly on the species of studied plants, the exposure time, and the concentration (Feizi et al., 2012). An experiment that used concentrations of 1, 2, 10, 100, and 500 ppm of TiO₂-ENPs, using *Triticum aestivum* L., found that the germination rate was not affected by any of the concentrations; however, they could highlight that at the concentrations of 2 and 10 ppm, the elongation of roots and shoots was more significant in these treatments (Feizi et al., 2012). Another example where concentration dependence was evidenced was reported by Song et al. (2012). They found that concentrations lower than 200 mg TiO₂-ENPs L⁻¹ increased the length of the root of *Lemna minor* L. significantly; however, the root length decreased when the concentrations were 200, 1000, or 2000 mg L⁻¹. Likewise, it was observed that the chlorophyll content increased as the concentrations of these ENPs increased. Similarly, Morteza et al. (2013) found that the addition of 0.01 or 0.03% of TiO₂-ENPs to *Zea mays* L. plants increased the carotenoid and chlorophyll a or b content.

However, Wu et al. (2017) found in their experiment with *Oryza sativa* L. a decrease in the biomass of the roots and leaves of plants exposed to concentrations of TiO₂-ENPs of 100, 250, or 500 mg L⁻¹, and also an increase was observed in antioxidant activity in response to stress produced by ENPs. Similarly, Boykov et al. (2014) reported that as TiO₂-ENPs concentration increased, biomass production decreased while TiO₂-ENPs negatively affected microRNA expressions.

3.8 Effects of Iron Oxides-ENPs in Plants

In general, iron oxide ENPs are considered biocompatible, biodegradable, and non-toxic (Hu et al., 2018). Therefore, studies have been conducted where the impacts of these ENPs on seed germination, growth, and development have been evaluated and their absorption and translocation. In these studies, there has been evidence of favorable impacts depending on the concentration of iron oxide ENPs, as demonstrated by Tombuloglu et al. (2019). They observed that as the concentration increases to 500 mg L⁻¹ of Fe₃O₄-ENPs, the *Hordeum vulgare* L. plant promoted the fresh weight of leaves and roots, translating into the increase in the growth of the plant. Besides, they also reported that Fe₃O₄-ENPs increased the content of chlorophyll, proteins, and the number of chloroplasts; likewise, Hu et al. (2018) found that at concentrations of 20 and 50 mg L⁻¹ of γ-Fe₂O₃-ENPs favored the germination of *Citrullus lanatus* (Thunb.) and *Zea mays* L. seeds, as well as root elongation in both cases. However, both studies (Hu et al., 2018; Tombuloglu et al., 2019) determined that with the increase in concentration (above 500 or 100 mg L⁻¹, respectively), there was a decrease in antioxidant activity promoting oxidative stress.

On the other hand, evidence has been found that even at low doses, the effects may be concentration-dependent, as reported by Plaksenkova et al. (2019). They employed concentrations of 1, 2, and 4 mg L⁻¹ of Fe₃O₄-ENPs. They found that at these doses, the rate of seed germination and elongation of the root of *Eruca sativa* Mill was significantly lower in experiments on germination, i.e., during a short-long test. However, when evaluating the effects of these concentrations in the hydroponic system, the length of the bud, root, and chlorophyll content increased, such as Fe₃O₄-ENPs concentration rise.

3.9 Effects of ZnO-ENPs on Plants

ZnO-ENPs are considered the second most used ENPs since this can be utilized in many products for everyday use. Therefore, there is a more significant presence of ZnO-ENPs in the environment, and accordingly, the concerns regarding potential environmental risks that these could generate have also increased (Balážová et al., 2018). For this reason, it is crucial to investigate the toxicity of this type of ENPs,

mainly on primary producers such as plants that serve as a food source and leading intermediary in trophic chains (Djearmane et al., 2018).

For example, Lee et al. (2013) observed a decrease in biomass production and reduction of antioxidant activity in *Fagopyrum esculentum* Moench. Plants spiked with ZnO-ENPs at concentrations of 1 to 2000 mg L⁻¹. Besides, plants bioaccumulated Zn in different tissues. Similarly, Balážová et al. (2018) found that the addition of 100 or 1000 mg ZnO-ENPs L⁻¹ decreased the bud and root length and harmed the development of secondary roots. However, despite the evidence of the adverse effects by ZnO-ENPs, other authors have reported the ENPs efficiency to grow crops. Because ZnO-ENPs are plant-growth promoters, they increase the amount of biomass produced and improve the antioxidant activity in *Pisum sativum* L., *Cicer arietinum* L., and *Spirodela polyrhiza* L. (Burman et al., 2013; Hu et al., 2013; Mukherjee et al., 2014).

3.10 Effects of CeO₂-ENPs on Plants

Although CeO₂-ENPs are among the ten nanomaterials with the highest production, there are few reports about their effects of these ENPs in higher plants (Rico et al., 2014; Zhang et al., 2011a, b). However, it is crucial to evaluate the CeO₂-ENPs impacts on plants to obtain evidence regarding the ecological risks of these ENMs and the potential exposition toward the human being through the food chain. CeO₂-ENPs can be absorbed from the root toward the shoots, and they can be distributed throughout the plant, as described by Zhang et al. (2011a, b) in the *Cucumis sativus* L. plant.

On the other hand, Rico et al. (2013) showed that *Oryza sativa* L. treated with CeO₂-ENPs presented cerium accumulation in the roots, more significant oxidative stress, and damage to the cell membrane of the roots. Wang et al. (2013) reported that the *Solanum lycopersicum* L. growth was smaller when plants were spiked with CeO₂-ENPs, and, consequently, the biomass production decreased, but the ROS content in root was improved. In contrast, Xie et al. (2019) and Ma et al. (2018) found that the presence of CeO₂-ENPs in *Phaseolus vulgaris* L. and *Brassica rapa* L. plants, respectively, did not affect their growth, but in the case of *Brassica rapa* L., the chlorophyll content increased.

3.11 Effects of Al₂O₃-ENPs on Plants

Like other ENMs described above, it is not clear how is the interaction of destination and toxicity of Al₂O₃-ENPs released into the environment. However, despite the few studies, it has been made clear that they can induce abiotic stress-causing alterations during plant growth and development (Yanik & Vardar, 2015). Examples of these alterations are the results obtained by Buklew et al. (2012), who observed that Al₂O₃-ENPs did not affect the seed germination stage of *Nicotiana tabacum* L. However, at the end of their experiment, they found little root development but also the reduction of biomass production, which coincides with the findings reported by Yanik & Vardar, 2015) and Owjo et al. (2019), who also describe the decrease in root length and biomass production in *Triticum aestivum* L. plants spiked with Al₂O₃-ENPs at 5, 25, or 50 mg mL⁻¹ and *Trigonella foenum-graecum* L. at 100 µg L⁻¹, respectively. Besides, the antioxidant activity also changed in both cases (Yanik & Vardar, 2015; Owjo et al. 2019). These results are according to Yanik et al. (2015), who determined that Al₂O₃-ENPs increased the hydrogen peroxide content, but the catalase activity was reduced, causing oxidative stress in *Triticum aestivum* L. plants.

3.12 Effects of SiO₂-ENPs on Plants

SiO₂-ENPs have been widely used in plant nutrition and as an enhancer in plant growth and development as there is evidence of the increase in some morphological characteristics, as reported by Asgari et al. (2018). They evaluated the effects of SiO₂-ENPs on the growth of *Avena sativa* L. and observed that the presence of these ENMs improves the growth of this plant by producing more biomass in the shoots and roots. Besides, the concentration of proteins increased when the plants were spiked with SiO₂-ENPs. Similarly, Mushinskiy et al. (2018) reported that SiO₂-ENPs stimulated the length of roots and shoots, the chlorophyll, and carotenoid content and increased the production of *Solanum tuberosum* L. tubers.

However, although SiO₂-ENPs promise good results in plant nutrition, it is necessary to continue evaluating the use of these ENMs on other plants, since results have been found where evidence is contrary to the above described, such as the case of what

was reported by Karimi and Mohsenzadeh (2016). They discovered that *Triticum aestivum* L. plants amended with SiO₂-ENPs decreased the biomass in roots and shoots but also chlorophyllic pigments (a and b) and carotenoids, while the lipid peroxidation was increased.

3.13 Effects of Ag-ENPs on Plants

Ag-ENPs are among the most studied ENMs about the properties and applications that they have in various areas of science and technology. More recently, Ag-ENPs were used in the agricultural sector, to stimulate the plant growth, control of pathogens and pests, or as accelerators and improvers in fruit ripening (Jasim et al., 2017; Yan & Chen, 2019). According to them, Jasim et al. (2017) reported that Ag-ENPs increased the number of leaves, root elongation, and sprout length in *Trigonella foenum-graecum* L. plants. Besides, Salachna et al. (2019) found higher biomass production of leaves and bulbs *Lilium* sp. and the stimulation of flowering. A similar result was reported by Byczynska et al. (2019), who found that the *Tulipa gesneriana* L. plants showed early flowering, larger petals, producing higher-quality flowers, as well as the increase in root length. In contrast, Rastroggi et al. (2019) found that the presence of these ENPs decreased the chlorophyll content and the root and shoot length of *Triticum aestivum* L. These controversial results witness the necessity of additional research regarding the use of Ag-ENPs and highlight that the delivery of Ag-ENPs or ENMs to the ecosystems must be regulated.

3.14 Effects of CNTs on Plants

According to existing reports regarding the effects of CNTs on plants, it has been found that their impact can be both positive and negative, so, according to this evidence, the use of these ENMs cannot yet be extensive in agricultural production. Tripathi et al. (2011), González-García et al. (2019), and Joshi et al. (2018) revealed that the use of CNTs increases the growth rate of plants, improving the fresh and dry weight of biomass, antioxidant activity, seed germination, and nutrient absorption in *Cicer arietinum* L., *Solanum lycopersicum* L., and *Triticum aestivum* L., respectively. On the contrary, Begum et al. (2012) reported that *Lactuca sativa* L., *Amaranthus tricolor*

L., *Cucumis sativus* L., and *Oryza sativa* L. are affected by the presence of CNTs, reducing the length of shoots and roots and increasing the wilting of the leaves. Besides, Begum et al. (2012) also reported that *Glycine max* L., *Capsicum annum* L., and *Abelmoschus esculentus* (L.) Moench did not present toxic effects due to the presence of CNTs. Fan et al. (2018), Hamdi et al. (2015), and Cano et al. (2016) did not find effects of CNTS on the growth and development of *Arabidopsis thaliana* (L.) Heynh., *Lactuca sativa* L., and *Zea mays* L., respectively. These pieces of evidence are similar to other ENMs described above, in the sense that ENMs affect the growth and development of crops and the balance of ecosystems. These affectations are a function of the plant species, doses, or concentration of the ENMs, growth conditions, and of course the properties and characteristics of ENMs.

3.15 Effect of ENMs on Mesofauna and Macrofauna of the Soil

To date, ENMs have been used to study biological and ecological aspects, but these researches have been carried out in laboratory, greenhouse, and field conditions. These studies intended to know the toxicological ENMs effects on collembolas, isopods, and earthworms. These latter are the most used since they represented the most significant biomass number compared with the rest of the organisms classes that live in the soils (Edwards, 2004). Besides, earthworms participate in various vital functions such as the decomposition of organic matter, the nutrients cycle, and other components (Fragoso & Rojas, 2014). Of the 58 selected studies showing the effects of ENMs on the mesofauna and macrofauna of the soil, 77.0% were performed with earthworms, followed by studies with isopods (17.0%) and collembola (6.0%). However, we found that most of this research was carried out under laboratory conditions (Fig. 3) and mainly in natural soil (Fig. 4). It has generated conflict and debate because it is not the same to evaluate and conclude from studies conducted on artificial soils, compared to natural land. After all, the soil matrix is involved with many factors that may influence organisms' toxicity (de Santiago-Martín et al., 2016). For instance, in the laboratory, no effects on mortality and reproduction were observed in *E. fetida* exposed to 1000 mg TiO₂-ENPs kg⁻¹ of the standard artificial OECD soil (Bouguerra et al.,

2016). However, some studies with artificial soil have reported that both low (14 to 50 mg kg⁻¹ of Ag-ENPs) and high concentrations (> 1000 mg kg⁻¹ of Ag NPs) have caused adverse effects on mortality and reproduction (Garcia-Velasco et al., 2016; Gomes et al., 2015; Bourdineaud et al., 2019). Nevertheless, in other studies with variates ENPs concentrations in natural soil was reported adverse effects in different parameters such as mortality reproduction, a decrease of carbohydrate, effect on CAT activity, and changes in the expression of genes in *E. coli* exposed to ENPs of Zn, CNTs, Ag, and Ce (Świątek & Bednarska, 2019; Xu et al., 2019; Hayashi et al., 2013; Novo et al. 2015; Lahive et al. 2017). These contradictory results suggested that the adverse effects depend on the type of organisms to be evaluated, methods of application of the NPs, doses, and soil's physical and chemical properties (Pérez-Hernández et al., 2020).

Despite the controversial cases associated with soil and physicochemical properties of ENMs, the above results can be worrisome in real scenarios, such as agricultural areas and forests. Indeed, Rajput et al. (2019) argued that once ENPs are released to the environment, they may be subject to bio/geo/transformation and consequently cause toxicity or changes in organisms' behavior. Therefore, it is crucial to generate studies in real scenarios to avoid contradictory conclusions without having performed experiments and known the effects on organisms and their interactions in agricultural lands and ecosystems (Kah et al., 2018). In this line, in a study under controlled conditions involving pumpkin and herbivorous plants, it was found that, when CeO₂-ENPs were added to the soil, pumpkin plants had higher Ce content in stems, leaves, and roots compared to the control. Besides, when the grasshoppers were fed using pumpkin leaf tissue, high Ce contents were found compared to untreated grasshoppers. The finding showed that CeO₂-ENPs accumulate in the leaf area and stems of pumpkin plants, and trophic transfer and possible contamination of the food chain occur (Hawthorne et al., 2014). In another study where Ce-ENPs were added (1000 mg kg⁻¹ dry soil) in soil pots containing lettuce (*Lactuca sativa*) and snails (*Achatina fulica*), a trophic transfer of Ce-ENPs from the plant toward the snails was demonstrated. It shows that the potential risk of transferring Ce to humans through the food chain is less obvious (Ma et al., 2018). By contrast, Chae et al. (2016) studied a trophic transfer of NPs

from *Saccharomyces cerevisiae* yeast to *Armadillidium vulgare* through the *Folsomia candida* (collembola). Besides, it was shown that ENPs are transferred intact from yeast cells to insects. Therefore, the study demonstrated the transfer of ENPs through the terrestrial food chain and indicated the potential dangers of ENPs for organisms at different trophic levels. In this sense, it is essential to mention that not all organisms have the same capacity to respond to ENMs. In earthworms, under laboratory conditions, an avoidance behavior to Ag-ENMs and Zvi-ENPs was observed (Mariyadas et al., 2018; Liang et al., 2018). However, to the selected studies, the effects reported have been mostly adverse (Fig. 2).

On the other hand, due to terrestrial isopods' capacity to accumulate metals, they have been used for contaminants bioaccumulation studies (Godet et al., 2011). Studies have shown that TiO₂ added to isopods' diet causes mortality and changes in enzymatic activity (Jemec et al., 2008). Other studies showed no effects on weight change and survival of *Porcellio scaber* after 14 days of feeding with diet contaminated with TiO₂-ENPs under laboratory conditions (Novak et al., 2012). Curiously, in a field experiment, Jemec et al. (2016) did not find adverse effects in *P. scaber* when they were spiked with TiO₂-ENPs. In this study, the organisms were exposed to 1000 mg TiO₂-ENPs kg⁻¹, both in leaves and soil contaminated. The results suggested that the organisms could detect metallic ions and therefore avoid their consumption. However, it is a concern that daily ENMs could be delivered to agricultural areas since farmers use river sediments and biosolids from wastewater treatment plants to improve the yields, and these biosolids could contain several ENPs (Chen et al., 2017). Besides, farmers could be using fertilizers that contain ENMs (Kim et al., 2017; Yuan, 2004). Therefore, plants and organisms that conform part of the human diet could be accumulating and transporting ENMs to the human food chain.

For all the above, it is possible that due to the contamination of agricultural and forestry areas caused by ENMs coming from urban or industrial regions, various soil beneficial organisms could be affected and the human food chain. Therefore, rigorous field studies that take into account the climate, soil types, cultivated plant species, and the properties of ENMs are required. These studies have to include ecological, environmental, and toxicological approaches.

3.16 Effect of ENMs on Human Beings Health

Studying the effects on ENMs into the body is an important topic nowadays. However, it is challenging to reach a complete investigation on that because the entire context is multifactorial. Some factors could include the characteristics and properties of the ENMs, the environmental conditions, the body conditions (if the body is immunocompromised or if they are a susceptible group, mainly), or other factors that may have non-measurable effects. So, the research on these ENMs must go in a deep direction. On the other hand, we must remember that nature has produced nanomaterials before humans, but all organisms have evolved through millions of years along with these natural nanomaterials; therefore, the natural nanomaterials are not a risk for alive organisms.

According to Som et al. (2011), ENMs are expected to have the potential to be used as functionalized, improved, and novel products. These products could open the way for the use of ENMs in construction, medicine, automotive industry, and environmental and safety technology sectors; consequently, ENMs could become ubiquitous in industrial and consumer products in the future. Ecological and human health could be affected by the involuntary release of these products. For example, some ENMs unintentionally released from geotextiles will probably end up in the soil and then come into direct contact with humans and end up in wastewater.

Som et al. (2011) found that some ENMs could affect the environment less severely than human health, while others' case is vice versa, which is especially true for CNT. The assessment of environmental risks depends mostly on the respective product life cycles and the quantities of ENMs produced worldwide. Their study defined the criteria for human health: (i) acute toxicity, (ii) chronic toxicity, (iii) DNA deterioration, (iv) crossing and damage of tissue barriers, (v) brain damage, and translocation and effects of ENMs on the (vi) skin, (vii) gastrointestinal, or (viii) respiratory treatment.

In medicine, ENMs have potential applications as transporters and administrators of drugs and as carriers of these. Fadeel and Garcia-Bennett (2010) stated that many questions must be answered: What is the relationship between size, shape, and surface

of the ENMs and their behavior in vivo? What is the fate of these when they fulfill their function? Are they degraded or metabolized, and are these products excreted by the body? Are there potential adverse effects of nanoparticles that can cause new unforeseen conditions such as accumulating chemicals? How could the impact of ENPs be measured and predicted? These answers must be solved for the future as soon as possible.

On the other hand, the advantages of ENMs as drug administrators are high stability, high loading capacity, incorporation of hydrophobic and hydrophilic substances, and compatibility in different forms of administration (oral, inhalation, etc.). The use of ENPs as carriers can also reduce the toxicity of drug incorporation. Furthermore, ENPs may or may not freely cross biological barriers indiscriminately, considering their specific physicochemical properties. ENMs are widely reported to cross tissue barriers and cause damage to the respiratory and gastrointestinal tract and the brain (Kendall & Holgate, 2012). However, both in vivo and in vitro models have been developed to study their mechanisms. Still, they often do not correlate appropriately due to the different dosage regimens.

3.17 Cytotoxicity Experiments

There is an emerging field, nanotoxicogenomics, that attempts to correlate the profile of the global gene of expression of cells or tissues exposed to ENPs with biological or toxicological responses using cDNA microarray technologies, which can produce useful information (Fadeel & Garcia-Bennett, 2010). Besides, other studies have correlated the effects of ENPs on cells and tissues with unpromising results.

3.18 The Importance of the Dose, Structure, and Size of ENPs

The ENMs size is a crucial characteristic that determines the effect of these on the organisms due to the surface/mass ratio that can substantially impact the chemistry of their surface. The dispersion state could be another factor to consider when the toxicity of specific ENMs is studied. TiO₂-ENPs have been studied extensively, concluding that these materials' cytotoxicity is highly correlated to their mineralogical phase. Besides, it has been found that the toxicity of ENMs

in the lungs is given by the surface reactivity, size, and surface area. Several studies have been conducted with Au-ENPs. Some of these studies suggest that the Au-ENPs size may influence cells' biological activity, suggesting that the Au-ENPs in drug delivery should not be seen merely as passive carriers. The above because Au-ENPs may play an active role in modulating biological and toxicological responses (Fadeel & Garcia-Bennett, 2010).

3.19 Cardiovascular Mechanisms

According to Kendall and Holgate (2012), ENMs have cardiovascular effects by inhalation. It has led to a growing understanding of the nanosized materials' impact beyond the first point of contact with the lung. Epidemiological findings bind to atmospheric ENPs with cardiovascular diseases by a series of potential mechanisms, such as (i) atherogenesis enhanced by inflammation and destabilization of plaque in coronary and cerebral arteries, (ii) epithelial/endothelial interactions with ENPs that alter blood clotting status or fibrinolytic balance favoring thrombogenesis, (iii) ENPs or its soluble components that enter to the bloodstream to exert direct effects; and (iv) increase arrhythmia (Kendall & Holgate, 2012).

In both humans and animal models, exposure to contaminants induces platelet activation and intravascular accumulation to promote blood clotting. Although different coagulation biomarkers were measured in many studies, the vehicle contaminant (especially PM_{2.5}) was consistently associated with thrombosis. Studies on air pollution by particles opened the way with a series of studies showing ventricular and supraventricular arrhythmias linked to exposure to ENMs, especially in older people. Also, air pollution by particles is more linked to improved platelet activation in those at risk of coronary heart disease. Besides, a range of low toxicity ENPs has demonstrated in vitro platelet activation properties (Kendall & Holgate, 2012). Most of the investigations of these studies have been carried out in vitro experiments so that these results could be misleading, and, therefore, standardized methodologies and collaboration between global health and environmental organisms must be reached to provide verification and validation of the experiments in vivo.

While we have documented environmental concerns and the damage that NMs cause to the environment, it is important to note that air pollution caused by micro-sized metals has been reported and discussed for many years. The report of Tidblad et al. (2012) recognize the contamination and corrosion of zinc, steel, and limestone during period of 22 years (1987–2009), according to tests carried out in at least 54 cities, including European countries and North America. However, in recent studies in the city of Athens, Greece, the data analyzed in the period 2003–2012, according to the report by Christodoulakis et al. (2017), indicate that the corrosion of materials such as zinc, carbon steel, limestone, and modern glass tends to decrease over the years, except for copper, since O₃ is the parameter that affects the mass loss of copper and not in the rest of the materials. Therefore, in a rigorous comparison between the polluting effects between NMs and their bulk counterparts, it has been documented that NPs present in the air can cause a higher toxic effect due to properties such as size, shape, surface area, and surface load, among others. For example, Martinello et al. (2021) discussed the effects of the combustion of ceramics and the generation of NPs (< 10 nm) and the effects on the health of workers, which with the COVID-19 pandemic can exacerbate respiratory diseases.

3.20 Risks of Human Consumption Through the Food and Trophic Chain

Potential benefits of the ENMs have been welcomed in the food industry, such as food packaging and food security, to deliver affordable and innocuous food to consumers. Notwithstanding, it has been demonstrated that ENMs pose a more significant threat to different organism species than bulk materials due to their physicochemical or biological properties. However, their potential risks to the consumers are not yet fully understood. Besides, there is a knowledge gap between the amounts of ENMs that could enter the human body through different foodstuffs. A review by Karimi et al. (2018) stated how the most widely found ENMs enter the human body, animals, and plants or are delivered to the environment, which could raise potential biosafety problems that review does not discuss human consumption of ENMs.

The possible risk of trophic transfer and transgenerational impact of ENMs were studied by Coman

et al. (2019) in soybean plants. They discussed the effect of ENMs in terms of positive and negative impacts on growth, synthesis of metabolites, and the root-associated microorganisms. Moreover, they suggested a possible risk of trophic transfer and transgenerational effects of ENMs on soybean crops. These findings increase the evidence that clearly points up the collateral damage to the environment by ENMs. Therefore, comprehensive risk-assessment studies should be implemented before any industrial-scale production of ENM. Unfortunately, it is too late to say that, and the health of the consumers and environment could be in a latent risk.

ENMs are delivered into ecosystems with unknown implications in the human food chain. Besides, ENMs may accumulate in the fruits, grains, or edible sections of crops. However, they have damaging or advantageous effects on the agronomic traits modifying the nutritional value of crops with collateral effect within the trophic chain's different levels. The interaction of ENMs with microorganisms, vegetables, and animals and human cells has been the subject of diverse multidisciplinary research highlighting the toxicological aspects, but these studies seldom consider the human trophic chain by technical, ethical, and legal considerations. However, there is a fact that ENMs are into the human trophic chain because the uptake, translocation, and accumulation of these nanosized materials have been demonstrated in several studies regarding different foodstuffs such as edible crops, fruits, and fishes.

4 Conclusions

ENMs are multifunctional materials for various advanced applications. Several thousands of tons of these materials are synthesized by simple and high production rate manufacturing methods worldwide. These ENMs with outstanding properties never seen before are used in advanced devices or processes that improve the profits but commit human and environmental health.

ENMs such as TiO₂, Ag, Zn, Si, Fe, Al, Ce, and CNT have been used in cutting-edge technologies such as synergetic photocatalytic degradation of antibiotics, hydrogen production, adsorption of organic pollutants, degradation of pharmaceutical and personal care products, CO₂ photoreduction, sensors,

artificial photosynthesis, artificial N-fixation, antibacterial applications, renewable energy applications, paints, medical implants, medical devices, cosmetics, food packaging, control of crop pathogens in agriculture, nanofertilizers, determination of trace pollutants in food, and drug controlled release, among others.

Currently, ENMs have attracted dramatically increasing interest in agriculture and food production because the possible environmental toxicity of ENMs could negatively influence edible crops and, therefore, human health. Interaction ENMs have been the subject of various multidisciplinary studies, but human or environmental health seldom has been the main objective. The ENMs risks regarding human consumption through the food and trophic chain have never been studied, but the results could be surprising and worrying.

Given the unavoidable involvement of ENMs in the agriculture sector and the current rather significant knowledge gaps, the use of ENMs should be judicious and only for specific applications to minimize their environmental release. There is a fact that ENMs are into the human trophic chain because the uptake, translocation, and accumulation of these nanosized materials have been demonstrated in different foodstuffs. Therefore, to shape sustainable nanotechnologies, the development of transdisciplinary teams conformed by scientists, technicians, politicians, business people, and entrepreneurs is urgent. We close by noting that several knowledge areas require further exploration to overcome current limitations for long-term experiments with higher plants or animals.

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Author Contribution H P-H, A P-M, CR S-C, S G-M, G M-P, F L-V, RG C-M, P J-K, and F F-L collected literature and wrote the manuscript. H P-H and F F-L designed, fine-tuned, and carried out the research idea. H P-H performed the statistical analysis. F F-L edited the manuscript. All authors read and approved the final manuscript.

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