



Published in final edited form as:

*Curr Opin Struct Biol.* ; 71: 27–35. doi:10.1016/j.sbi.2021.05.010.

## Structural Perspectives on H<sub>2</sub>S Homeostasis

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### Abstract

The enzymes involved in H<sub>2</sub>S homeostasis regulate its production from sulfur-containing amino acids and its oxidation to thiosulfate and sulfate. Two gatekeepers in this homeostatic circuit are cystathionine β-synthase (CBS), which commits homocysteine to cysteine, and sulfide quinone oxidoreductase (SQOR), which commits H<sub>2</sub>S to oxidation via a mitochondrial pathway. Inborn errors at either locus affect sulfur metabolism, increasing homocysteine derived H<sub>2</sub>S synthesis in the case of CBS deficiency, and reducing complex IV activity in the case of SQOR deficiency. In this review, we focus on structural perspectives on the reaction mechanisms and regulation of these two enzymes, which are key to understanding H<sub>2</sub>S homeostasis in health and its dysregulation and potential targeting in disease.

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Hydrogen sulfide has dual effects in mammalian energy metabolism, functioning as a respiratory poison and a respiratory substrate for the electron transport chain (ETC). Unlike its reputation as a noxious environmental toxin, the systematic study of the cellular capacity to biosynthesize H<sub>2</sub>S in mammals has been studied more recently. Of the three enzymes ascribed roles in H<sub>2</sub>S biogenesis [1], two, cystathionine β-synthase (CBS) and γ-cystathionase (CSE), constitute the transsulfuration pathway while the third, 3-mercaptopyruvate sulfurtransferase (cytoplasmic MPST1 and mitochondrial MPST2), is involved in cysteine catabolism (Figure 1) [2–5]. H<sub>2</sub>S is not a product of the canonical reactions catalyzed by CBS and CSE but rather, a consequence of their substrate and reaction promiscuities [6]. Cystathionine, produced by CBS, is converted to cysteine by CSE, which in turn, is a substrate for the H<sub>2</sub>S producing reactions catalyzed by both CBS and CSE [2, 3]. While large gaps remain in our understanding of the relative output of H<sub>2</sub>S versus cysteine, it is estimated that the predominant cellular transsulfuration flux yields cysteine [7]. The concentrations of CBS versus CSE and of their substrates, influence their relative contributions to H<sub>2</sub>S synthesis in a tissue-specific manner [8]. MPST is presumably

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Author Disclosure Statement

RB is a paid member of the scientific advisory board of Apneo Therapeutics and owns equity in the company.

activated under conditions of cysteine excess, although its regulation remains to be fully elucidated [5].

While H<sub>2</sub>S biogenesis piggybacks on other housekeeping functions, a dedicated pathway exists for sulfide oxidation in mammals and includes sulfide quinone oxidoreductase (SQOR) [9], a persulfide dioxygenase (ETHE1) [10], rhodanese (or TST) [11] and sulfite oxidase [12] (Figure 1). The products of this mitochondrial pathway are thiosulfate and sulfate [13]. Electrons released during sulfide oxidation enter the ETC at the level of complex III (from SQOR via CoQ) and complex IV (from sulfite oxidase via cytochrome c) and establish sulfide as an inorganic substrate for oxidative phosphorylation in mammals [14]. SQOR deficiency presents clinically as Leigh disease and reduced complex IV activity [15]. The wide prevalence of the sulfide oxidation pathway across cell and tissue types suggests an important role for H<sub>2</sub>S in cellular metabolism.

However, there is scant information on the quantitative significance of H<sub>2</sub>S production and the cellular contexts in which it is regulated. In this review, we focus on CBS, which commits homocysteine to transsulfuration or H<sub>2</sub>S synthesis, and SQOR, which commits H<sub>2</sub>S to oxidation, emphasizing structural perspectives that could be important for regulation.

## Structural basis of H<sub>2</sub>S synthesis by CBS

CBS catalyzes the condensation of serine and homocysteine, generating cystathionine and water (Figure 2a). Alternatively, CBS catalyzes the condensation of cysteine and homocysteine, forming cystathionine and H<sub>2</sub>S, or eliminates H<sub>2</sub>S from one or two moles of cysteine to generate pyruvate and lanthionine, respectively. CBS is heavily regulated [16–19] and its dysfunction leads to homocystinuria, an inborn error of metabolism. The increased excretion by homocystinuric patients of homolanthionine [20], a product of the CSE-catalyzed condensation of two moles of homocysteine [2], points to a role for CBS in regulating H<sub>2</sub>S levels systemically. CBS and CSE can also generate cysteine persulfides from cystine [21, 22].

Crystal structures of CBS [23–27] have provided a framework for understanding its reaction mechanism and allosteric regulation by AdoMet and heme. CBS is a modular enzyme [28] in which the N- and C-terminal regulatory domains bind heme and S-adenosylmethionine (AdoMet) respectively and sandwich a catalytic core that binds pyridoxal 5'-phosphate (PLP) (Figure 2b). The 8-fold  $\alpha/\beta$  barrel motif of the catalytic core is very similar to other members of the  $\beta$  family of PLP enzymes [27, 29, 30]. Two distinct CBS conformations are seen in the absence (basal) and presence (activated) of AdoMet (Figure 2b).

The postulated reaction mechanism for CBS invokes classical  $\beta$ -elimination chemistry [31–33] and begins with an internal aldimine between PLP and Lys-119 (Figure 3a). The PLP cofactor is anchored via interactions between its phosphate moiety and a glycine rich loop (<sup>256</sup>Gly-<sup>260</sup>Thr) and hydrogen bonds with Ser-349 and Asn-149 (Figure 3b) [23]. Binding of serine or cysteine leads to formation of the corresponding external aldimine and is followed by proton abstraction from C $\alpha$  resulting in a stabilized carbanion (Figure 3a). The carbanion intermediate was captured in a 1.70 Å crystal structure of *Drosophila* CBS with electrostatic

stabilization provided by the  $\epsilon$ -nitrogen of Lys-88 (corresponding to human Lys-119) at a distance of 2.1 Å from C $\alpha$ , [23]. Their high homology (68%) and sequence identity (51%) suggest that the *Drosophila* CBS is likely to be an excellent model for the human enzyme.

The carbanion is converted to an aminoacrylate intermediate following  $\beta$ -elimination of water (from serine) or H<sub>2</sub>S (from cysteine) (Figure 3a). The 1.55 Å crystal structure of the aminoacrylate intermediate revealed that the side chain of Lys-88 is rotated away from C $\alpha$  and engaged in a hydrogen bonding interaction with a phosphate oxygen [23] (Figure 3c). Since the heme cofactor obscures PLP-associated spectral changes, difference stopped-flow spectroscopy was used to monitor the internal aldimine ( $\lambda_{\text{max}} = 406$  nm) and its conversion to the aminoacrylate intermediate ( $\lambda_{\text{max}} = 465$  nm) [23]. The aminoacrylate was also observed in the heme-less variant of human CBS, which is less stable [34], and in yeast CBS, which lacks the heme cofactor [35]. The catalytic cycle is completed upon addition of homocysteine, forming an external aldimine with cystathionine, followed by a transchiffization reaction and product release.

In H<sub>2</sub>S generating reactions in which one or two moles of cysteine are involved (Figure 2a), water or cysteine add to the aminoacrylate intermediate forming the external aldimine with serine or lantionine, respectively. Serine can subsequently undergo  $\alpha,\beta$ -elimination to form pyruvate and ammonia [3].

## Structural basis of heme-dependent regulation of CBS

The heme in CBS is housed in a hydrophobic pocket comprising residues 50–67 (Figure 4a) [26, 27]. The heme iron (Fe<sup>2+</sup>/Fe<sup>3+</sup>) is coordinated by His-65 and Cys-52 while the latter is also engaged in an ionic interaction with Arg-166 located at one end of an  $\alpha$ -helix, which connects to the PLP pocket at the other end [36, 37]. The heme is tightly bound but was released from CBS crystals under reducing conditions following addition of CO [38]. The heme is not essential for activity since the heme-less variant lacking the N-terminal 69 residues retains ~40% of wild-type CBS activity [34]. The CO-induced heme loss from CBS suggests that a ligand-induced protein destabilization could be deployed as a regulatory strategy.

Changes in the heme coordination environment influence CBS activity, although the context in which heme-dependent regulation is physiologically important is not entirely clear [39]. Binding of CO to ferrous CBS displaces Cys-52, forming a 6-coordinate species in which the His-65 ligand is retained [40]. On the other hand, NO• displaces His-65, which is followed by loss of Cys-52 coordination and formation of a 5-coordinate ferrous-nitrosyl species [41]. Both CO and NO• binding are correlated with inhibition of CBS activity [41, 42], which is presumed to arise from loss of the interaction between Cys-52 and Arg-266 and is correlated with a tautomeric shift from the active ketoenamine to the inactive enolimine form of PLP [36, 37]. Air exposure leads to recovery of active 6-coordinate ferric CBS. Surprisingly, despite the >50 Å distance separating them, AdoMet increases the affinity of heme for NO• (2-fold) and CO (5-fold) and sensitizes CBS to inhibition [43]. AdoMet also increases the rate constant for CO binding by ~10-fold and for NO• by 1.5-fold [43]. Hence, in the interplay between the N- and C-terminal regulatory domains, heme

trumps AdoMet as an allosteric regulator. The structural basis for the long-range allosteric communication from the AdoMet-bound C-terminal and the N-terminal heme domain is not understood.

## The structural basis for AdoMet-dependent regulation of CBS

CBS is a molecular jumping jack alternating between the lateral and overhead extension of its C-terminal regulatory side arms as it transitions from the basal to the activated state in response to AdoMet (Figure 2b). The C-terminal domain houses a tandem repeat of the Bateman module, better known as the CBS domain [44], which extends between residues 412–471 (CBS1) and 472–538 (CBS2) [45]. In the basal conformation, the C-terminal domain of one subunit sits atop the catalytic core of the other subunit and impedes substrate access to the active site, lowering activity (Figure 2b). Upon AdoMet binding, the regulatory domains dimerize above the active site, increasing substrate access and is correlated with a 2 to 4-fold increase in activity for the canonical and the H<sub>2</sub>S-generating reactions [3, 46]. Full-length human CBS is prone to aggregation and it is likely that structural elements in the C-terminal domain are responsible for this behavior, since its truncation leads to a well-behaved CBS dimer [47]. A stable dimer is also formed upon deletion of a 10-residue loop (516–525) in the CBS2 domain in full-length human CBS [24]. Residues 513–519 are postulated to act as a hook, locking two dimers into a tetramer.

The CBS domains in the *Trypanosoma brucei* GMP reductase play a role in its oligomerization [48]. In the presence of GMP, the enzyme exhibits the native octameric state while in the presence of ATP, it separates into two tetramers, which is accompanied by a 1.6-fold decrease in activity [48]. Deletion of the CBS domain leads to formation of the tetrameric enzyme, which is however, inactive. Given the similarities between the CBS domains of GMP reductase and CBS (Figure 4b), it is plausible that this region is similarly involved in ligand induced oligomeric changes in human CBS with potential regulatory implications.

## Sulfide quinone oxidoreductase gates H<sub>2</sub>S clearance

SQOR belongs to the flavin disulfide reductase superfamily [49] and is anchored to the inner mitochondrial membrane via two amphipathic C-terminal helices, while the remainder of the protein faces the matrix [50, 51]. SQOR catalyzes the committing step in the sulfide oxidation pathway, converting H<sub>2</sub>S to a persulfide while concomitantly reducing CoQ (Figure 5). The overall reaction proceeds via two half reactions. In the first, sulfide adds into the cysteine trisulfide, forming persulfides on Cys-201 and Cys-379. In the second half-reaction, electrons from FADH<sub>2</sub> are transferred to CoQ, which then connects to the ETC [52]. Bacterial and human SQORs differ in important details. While the oxidized sulfur exits the human enzyme as a persulfide product after each catalytic cycle [13, 53, 54], bacterial SQORs successively add sulfide, building polysulfide chains [55–57]. Human SQOR resembles flavocytochrome *c* sulfide dehydrogenase in its active site architecture and its catalytic mechanism, and, following addition of sulfide into the cysteine disulfide, the sulfane sulfur is transferred to a second equivalent of sulfide to form HSSH, while the electrons are transferred to cytochrome *c* [58].

The overall structure of SQOR exhibits the typical characteristics of the flavin disulfide reductase superfamily, including tandem Rossmann fold repeats that accommodate a non-covalently bound FAD and two redox-active cysteines (Figure 6a). The first Rossmann fold binds the ADP moiety of FAD and presents much of the negative charge to the matrix-facing domain of the enzyme. The second Rossmann fold has electropositive patches that presumably facilitate interaction with membrane phospholipid head groups. Unexpectedly, the crystal structures of human SQOR revealed a cysteine trisulfide (Figure 6a, *inset*), the first reported in a flavin disulfide reductase family member [50, 51]. The structures revealed a large electropositive cavity for substrate entry, leading to the *re* face of FAD while CoQ was bound to the *si* face at the end of a hydrophobic tunnel that presumably leads to the membrane.

### A cysteine trisulfide catalyzes H<sub>2</sub>S oxidation

The cysteine trisulfide in SQOR exposes the S $\gamma$  of Cys-379 to bulk solvent while Cys-201 lies within a 9-residue loop that shields the remainder of the active site [50, 51]. Molecular dynamic simulations predict a greater electrophilic character at Cys-379 and susceptibility to nucleophilic attack by sulfide [59]. The Cys-379-SSH and Cys-201-SSH persulfide pair formed upon addition of sulfide to the trisulfide was observed in an SQOR structure (Figure 6b) [50]. The Cys-201-SSH is 3 Å from C4a in FAD, forming an unusually intense charge transfer complex centered at 675 nm [53, 60]. Protrusion of the Cys-379-SSH out of the active site facilitates its transfer to a thiophilic acceptor, reforming the resting trisulfide. Addition of sulfide presumably led to the *in crystallo* reduction of CoQ and led to its relocation to the entrance of the hydrophobic tunnel [50].

The presence of the cysteine trisulfide in SQOR was confirmed by its sensitivity to cyanolysis [50, 59]. Cyanide initially adds into the trisulfide to form a charge transfer complex, which slowly collapses to form a Cys-379 N-(<sup>201</sup>Cys-disulfanyl)-methanimido thioate intermediate (Figure 6c) [59]. Attack by a second cyanide extracts the sulfane sulfur, yielding thiocyanate. Post-cyanolysis, SQOR is inactive and thermally unstable and harbors a Cys-379 N-(<sup>201</sup>Cys-sulfanyl)-methanimido thioate species. The bridging intermediate preserves the oxidation state at both cysteines, and as a result, sulfide addition can reinstall the bridging sulfur and regenerate the cysteine trisulfide. Based on these findings, a mechanism for the initial formation of the cysteine trisulfide *in vivo* has been proposed [9], but remains to be tested.

### SQOR accommodates a diversity of substrates

The large electropositive cavity leading to the SQOR active site confers a remarkable degree of substrate promiscuity (Figure 6d). While a similar opening is present in flavocytochrome *c* sulfide dehydrogenase [58], the active site is capped in bacterial SQORs and accommodates the growing polysulfide product [55–57]. Sulfite, an efficient sulfane sulfur acceptor [53], is proposed to dock in close proximity to Cys-379, with Ser-343 and Ala-202 providing hydrogen bonding interactions [51]. The docked structures of GSH and CoA reveal that their thiol moieties are oriented toward Cys-379 (Figure 6e) [50, 51].

Both the size and the  $pK_a$  of the sulfane sulfur acceptors influence their catalytic efficiency. The larger acceptors GSH and CoA ( $pK_a = 9.2$  and  $9.6$ , respectively), exhibit an  $\sim 10$  to  $20$ -fold lower  $k_{cat}/K_M$  versus methanethiol ( $pK_a = 10.4$ ), which in turn, has a  $6$ -fold lower  $k_{cat}/K_M$  than sulfite ( $pK_a = 7.2$ ) [50, 61]. The direct addition of some acceptors, like methanethiol and sulfite traps the enzyme in a dead-end complex, which decays slowly and regenerates active SQOR [61]. The slow  $k_{on}$  values for the acceptors and/or their low intracellular abundance likely reduce their cellular relevance, although pathological conditions, e.g., sulfite oxidase deficiency [62], could promote adventitious reactions.

## Conclusions

The structures of CBS and SQOR, two important gatekeepers of cellular  $H_2S$  homeostasis, have revealed many surprises and provided a framework for interrogating their regulation. The role of the heme and the mechanism of long-range interaction between the N- and C-terminal domains in CBS to exert allosteric control, are poorly understood. Furthermore, the high  $K_M$  values for the CBS substrates raise the question as to whether other metabolites or proteins modulate substrate affinities and bring them within physiologically relevant ranges. The novel trisulfide configuration in SQOR helps explain the enhanced rate of the sulfide addition reaction but raises questions as to how the cofactor is installed and what the source of the bridging sulfur is.

Finally, the cavernous entrance and solvent exposure of the electrophilic sulfur in the trisulfide cofactor, begs the question as to how adventitious side reactions are averted. Continued insights into the enzymology of  $H_2S$  homeostasis will be important for shaping rational targeting efforts.

## Acknowledgement

This work was supported in part by the grants from the National Institutes of Health (GM130183 to RB).

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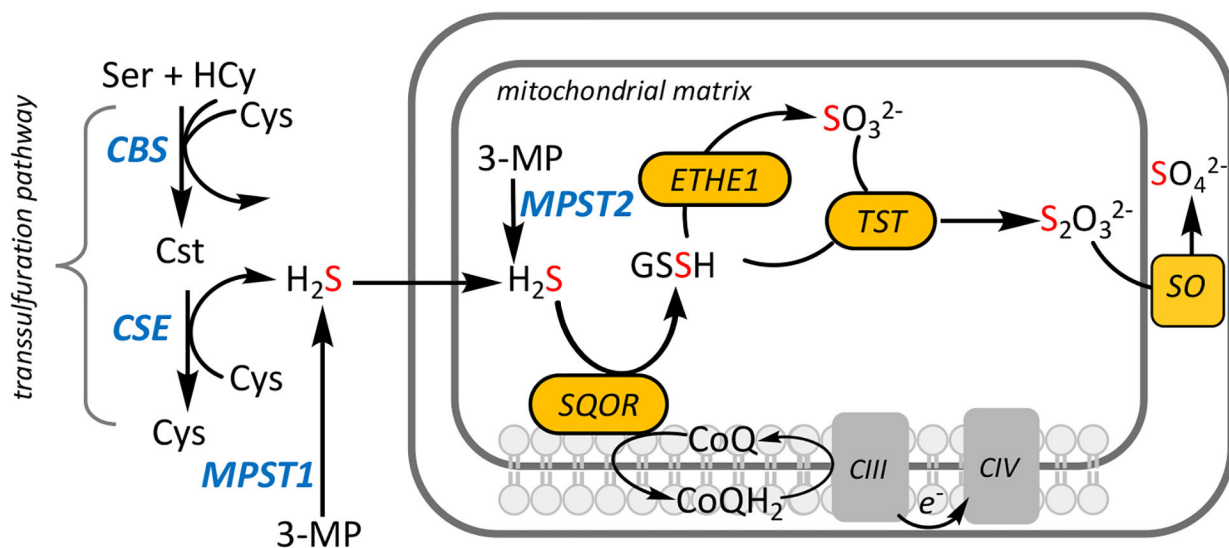
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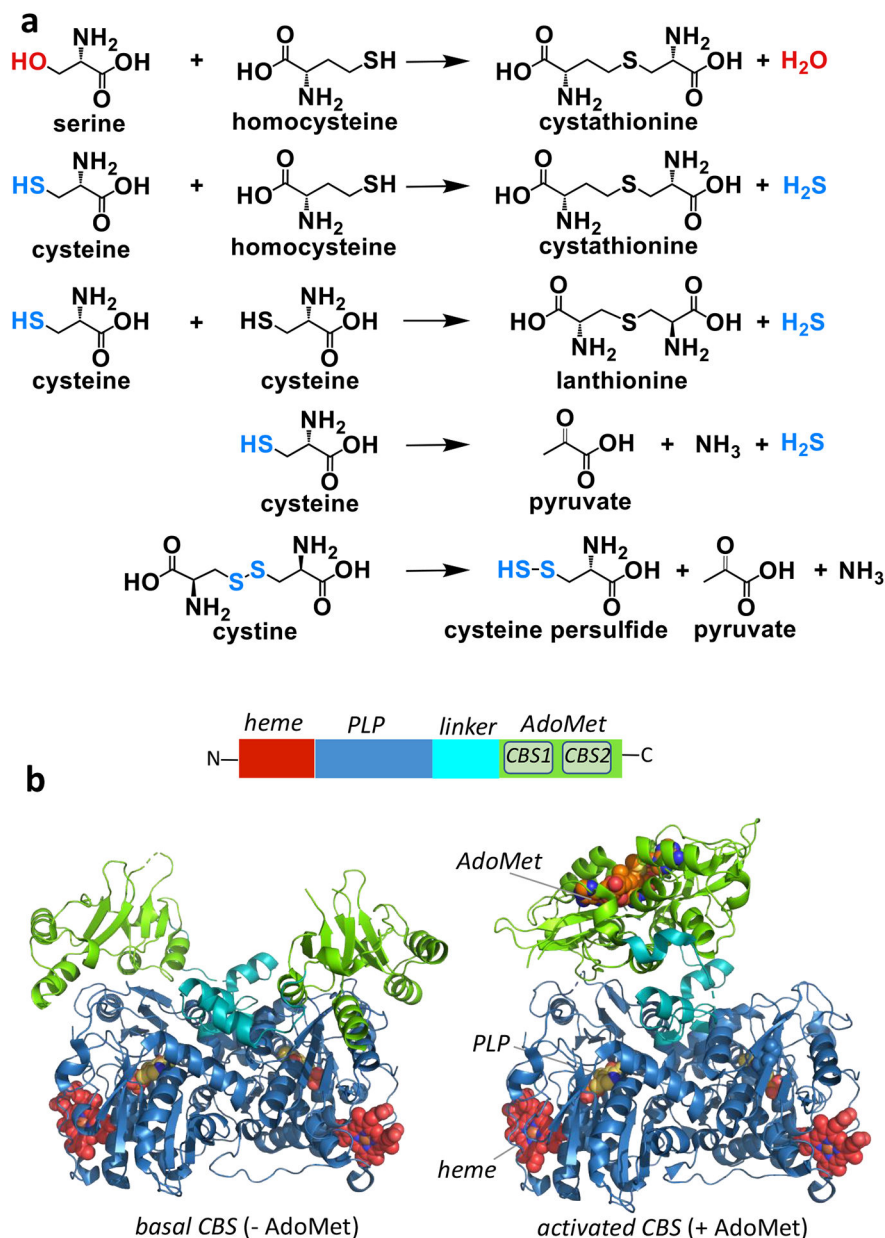


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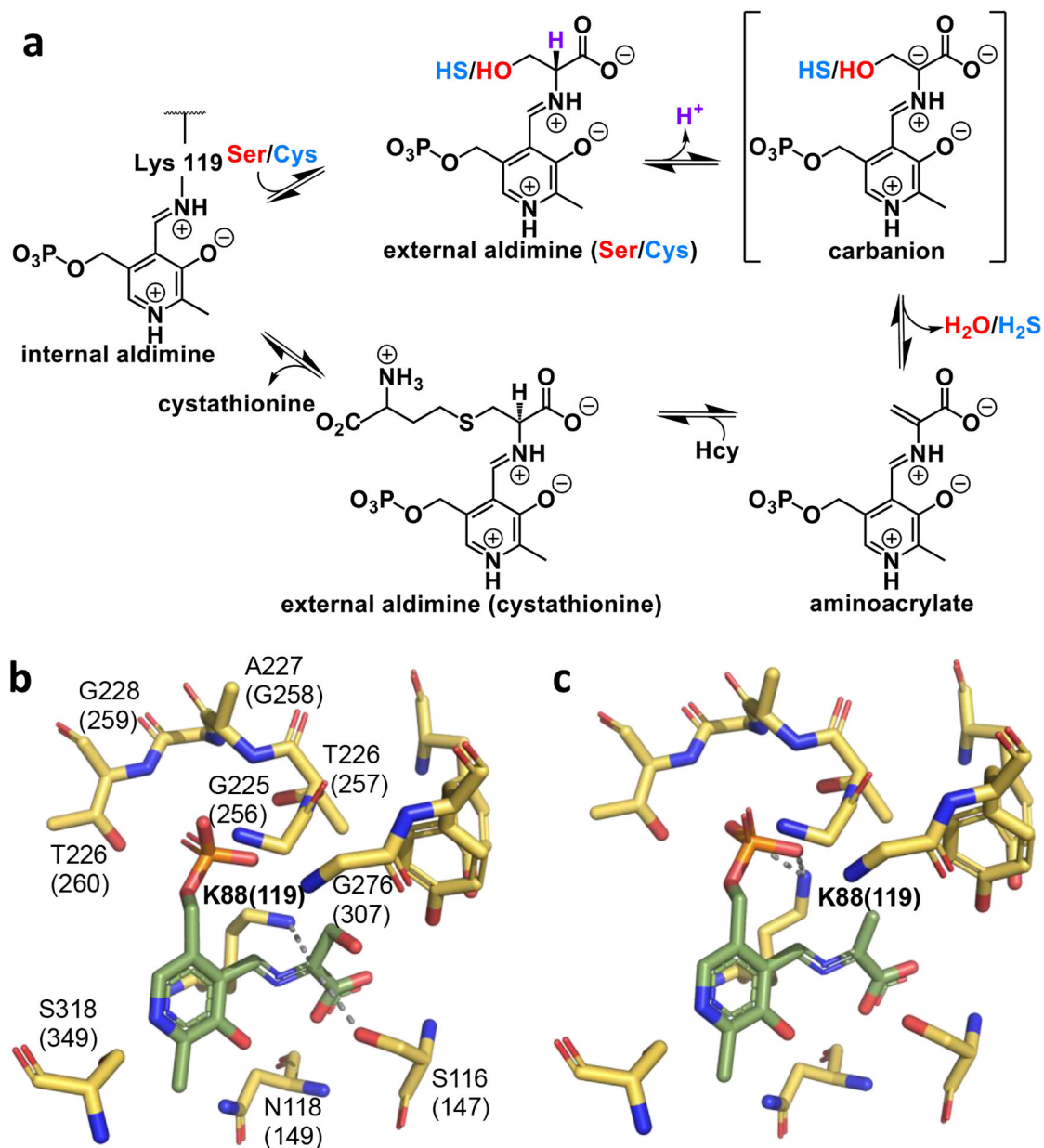


**Figure 1. Enzymes involved in H<sub>2</sub>S homeostasis.**

H<sub>2</sub>S biogenesis is catalyzed by three enzymes (blue): CBS, CSE and MPST while H<sub>2</sub>S catabolism is catalyzed by enzymes in the mitochondrial sulfide oxidation pathway (yellow). ETHE1, TST and SO denote persulfide dioxygenase, rhodanese and sulfite oxidase, respectively. CIII, CIV, 3-MP, Hcy and Cst denote complexes III and IV, 3-mercaptopyruvate, homocysteine and cystathionine, respectively.

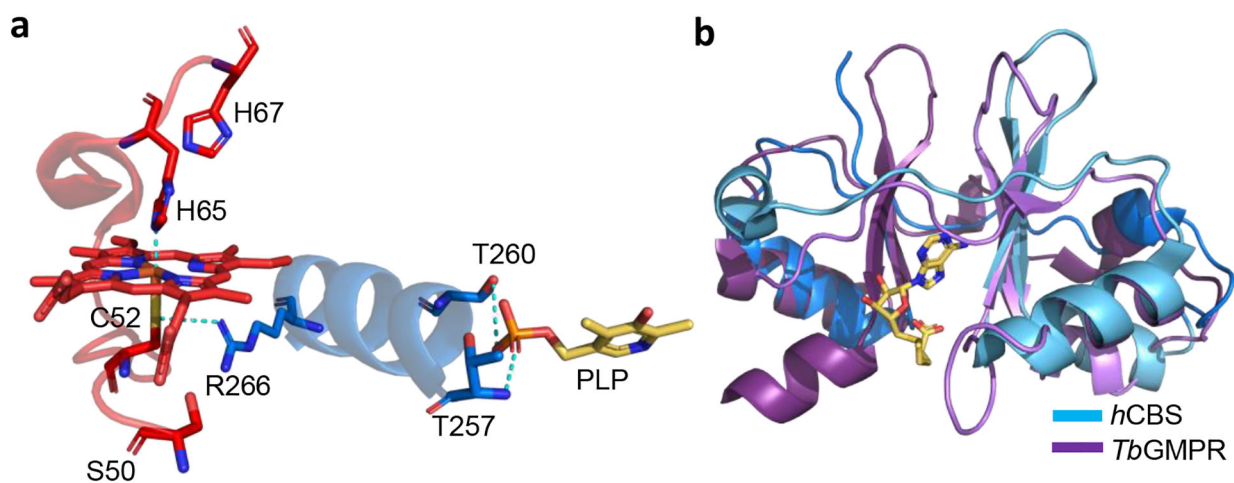


**Figure 2.** Structure and functions of CBS. (a) Reactions catalyzed by CBS including the canonical transsulfuration reaction utilizing serine, the H<sub>2</sub>S generating reactions utilizing cysteine (± homocysteine) and the persulfide generating reaction utilizing cystine. (b) Domain organization of human CBS (*upper*) and the structures of CBS captured in a basal (PDB: 4COO, *left*) and an activated (PDB: 4PCU *right*) conformation. The domains are colored as in the key on top and CBS1 and CBS2 refer to the two CBS domains within the C-terminal regulatory segment that binds AdoMet.



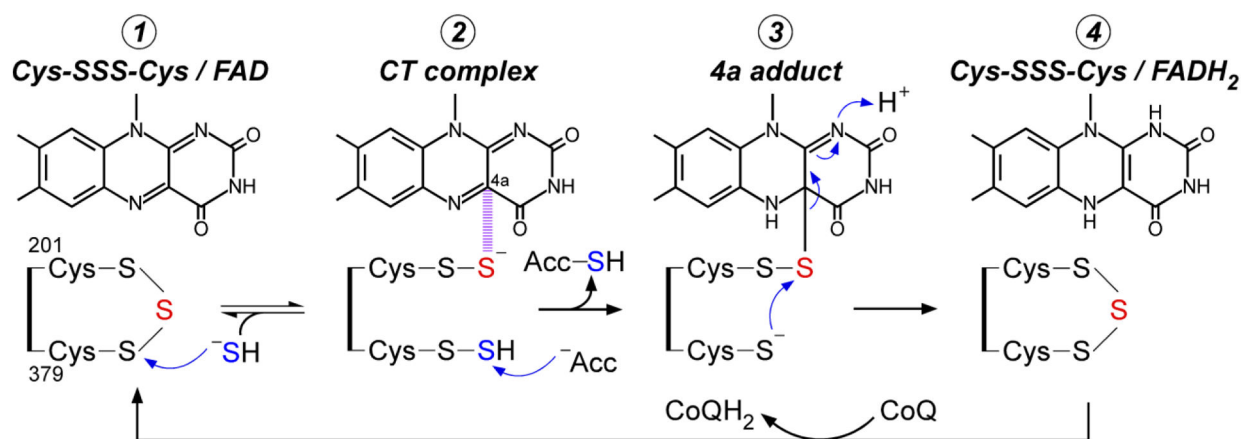
**Figure 3. CBS reaction mechanism and structure of two key intermediates.**

(a) Postulated reaction mechanism for CBS-catalyzed  $H_2O$  or  $H_2S$  elimination. (b,c) Close-up of the structures of full-length *Drosophila* CBS in which a carbanion (PDB:3PC4) and an aminoacrylate (PDB: 3PC3) intermediate are seen. The residues and numbering in human CBS are shown in parentheses. Lys-88 (corresponding to Lys-119 in human CBS) is 2.5 Å from Ca providing electrostatic stabilization of the carbanion intermediate (b) but is rotated away and engaged in a hydrogen-bonding interaction with a phosphate oxygen in the aminoacrylate species (c). The PLP bound intermediates are shown in green.



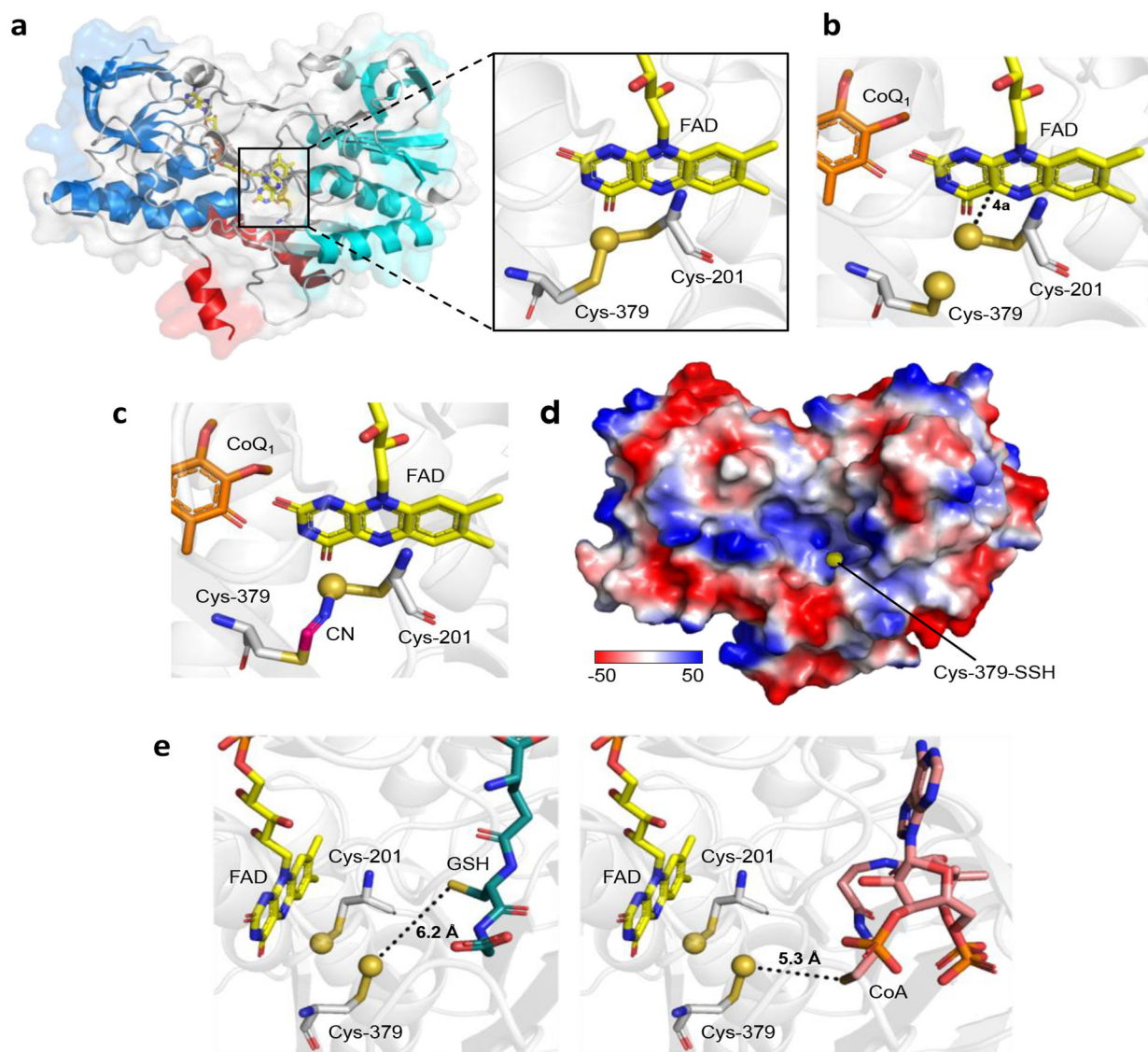
**Figure 4. Regulatory domains in CBS.**

(a) Close-up showing the connection between the heme and PLP pockets in the structure of human CBS. Residues comprising the hydrophobic heme pocket are shown in red. His-65 and Cys-52 serve as the heme ligands while Cys-52 is also involved in an electrostatic interaction with Arg-266. (b) Overlay of the CBS domains in human CBS (PDB: 4PCU, blue) and *T. brucei* GMMPR (PDB: 6JL8, purple). AdoMet bound to human CBS is shown in yellow stick display.



**Figure 5. Proposed catalytic mechanism for SQOR.**

The bridging sulfur of the cysteine trisulfide is shown in red and the sulfur undergoing oxidation is shown in blue. The dashed magenta lines denote a charge transfer (CT) complex between the Cys-201 persulfide and the flavin C4a, which leads to formation of the 4a adduct followed by FADH<sub>2</sub>.



**Figure 6. Structure of human SQOR.**

(a) Overall structure of native SQOR (PDB ID: 6OI5), with the first and second Rossmann fold domains shown in blue and cyan, respectively, and membrane-anchoring helices shown in red. The *inset* is a close-up of the active site showing the cysteine trisulfide. (b) SQOR co-crystallized with CoQ (orange sticks; PDB ID: 6OIB), contains a persulfide pair, corresponding to [2] in Figure 5. (c) The SQOR Cys-379 N-(<sup>201</sup>Cys-disulfanyl)-methanimido thioate intermediate generated upon addition of cyanide (CN; PDB ID: 6WH6). (d) Electrostatic surface potential map showing the solvent exposure of Cys-379-SSH at the bottom of the electropositive cavity. (e) Models of physiological acceptors docked in the SQOR cavity. The distances between the Cys-379 persulfide and the thiol moieties of GSH (left panel, teal sticks) and CoA (right panel, pink sticks) are noted. The sulfane sulfurs in all panels are shown as yellow spheres.