

Local rigidification and possible coacervation of the *Escherichia coli* DNA by cationic nylon-3 polymers

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ABSTRACT Synthetic, cationic random nylon-3 polymers (β -peptides) show promise as inexpensive antimicrobial agents less susceptible to proteolysis than normal peptides. We have used superresolution, single-cell, time-lapse fluorescence microscopy to compare the effects on live *Escherichia coli* cells of four such polymers and the natural antimicrobial peptides LL-37 and cecropin A. The longer, densely charged monomethyl-cyclohexyl (MM-CH) copolymer and MM homopolymer rapidly traverse the outer membrane and the cytoplasmic membrane. Over the next ~5 min, they locally rigidify the chromosomal DNA and slow the diffusive motion of ribosomal species to a degree comparable to LL-37. The shorter dimethyl-dimethylcyclopentyl (DM-DMCP) and dimethyl-dimethylcyclohexyl (DM-DMCH) copolymers, and cecropin A are significantly less effective at rigidifying DNA. Diffusion of the DNA-binding protein HU and of ribosomal species is hindered as well. The results suggest that charge density and contour length are important parameters governing these antimicrobial effects. The data corroborate a model in which agents having sufficient cationic charge distributed across molecular contour lengths comparable to local DNA-DNA interstrand spacings (~6 nm) form a dense network of multivalent, electrostatic "pseudo-cross-links" that cause the local rigidification. In addition, at times longer than ~30 min, we observe that the MM-CH copolymer and the MM homopolymer (but not the other four agents) cause gradual coalescence of the two nucleoid lobes into a single dense lobe localized at one end of the cell. We speculate that this process involves coacervation of the DNA by the cationic polymer, and may be related to the liquid droplet coacervates observed in eukaryotic cells.

SIGNIFICANCE Certain sequence-random, synthetic cationic copolymers disrupt bacterial membranes, a mode of antibacterial action reminiscent of the action of natural antimicrobial peptides. We use single-cell fluorescence microscopy to directly compare the symptoms induced in *Escherichia coli* during the attack of several β -peptide polymers and the natural antimicrobial peptides LL-37 and cecropin A. Agents with sufficient contour length and positive-charge density locally dramatically attenuates the jiggling motion of the chromosomal DNA. The suggested mechanism is noncovalent, electrostatic cross-linking across nearby DNA stands. Once the membrane is breached, the bacterial cytoplasm with its preponderance of anionic biopolymers attracts a very high, damaging concentration of polycations. Such mechanistic studies may inform the design of new antimicrobial agents.

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INTRODUCTION

Widespread infection by antibiotic-resistant bacteria necessitates the design and development of new antimicrobial agents that target resistant pathogens (1–3). Natural antimicrobial peptides (AMPs; also called host-defense peptides), part of the host innate immune system of plants and animals, are promising agents in this context. AMPs comprise an ancient class of short- and medium-sized polypeptides (typically <40 amino acids) that exhibit broad-spectrum antibacterial activity against both Gram-negative and Gram-positive bacteria (2). Most AMPs are amphipathic and highly cationic in physiological conditions. They are attracted to and permeabilize the negatively charged



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membranes of bacteria but leave the zwitterionic eukaryotic membranes intact (2). Permeabilization of the bacterial outer membrane (OM) and cytoplasmic membrane (CM), with the concomitant loss of the proton motive force and critical small molecules, has long been recognized as a primary cause of AMP activity. However, mounting evidence suggests that some AMPs may kill bacteria via non-membrane-permeabilization mechanisms that disrupt specific intracellular processes (2,4–6).

Recently several groups have shown that membrane permeabilization leads to a massive influx of certain cationic AMPs into the bacterial cytoplasm. The threshold concentrations of bound peptide molecules per cell needed to kill bacteria varies from 7×10^6 to 2×10^8 , depending on the peptide and measuring method (1,6–8). We suggest that the preponderance of large, polyanionic species (chromosomal DNA, mRNA, rRNA, and ribosomes) renders the bacterial cytoplasm highly attractive to polycationic AMPs. Presumably the key driving force for AMP absorption is Coulombic attraction between polycations and polyanions enhanced by the entropically favorable concomitant release of small cations (mostly K⁺) to the cell surround (6).

We have used single-cell fluorescence microscopy to study the effects of AMPs on live *Escherichia coli* cells in real time (5,6,9,10). At concentrations comparable to the minimal inhibitory concentration (MIC), the human antimicrobial peptide LL-37 (+6 net charge, ~10.4 nm in backbone contour length) permeabilizes both the OM and CM of *E. coli* to globular proteins and to smaller species on a timescale of several minutes (6). We estimated that on average each *E. coli* cell subsequently absorbs ~10⁸ LL-37 copies, corresponding to a mean concentration of ~90 mM averaged over the entire cell volume (cell membranes, periplasm, and cytoplasm) (6).

We also tracked the motion of the DNA locus Right 2 with 30-nm localization accuracy before and after CM permeabilization by LL-37 (6). The local jiggling motion of *Right2* essentially freezes within 1 min of CM permeabilization. The chromosomal DNA meshwork has apparently rigidified, at least locally. We speculated that the high concentration of absorbed polycationic LL-37 forms a dense, fluctuating network of noncovalent, electrostatic "pseudocross-links" within the chromosomal DNA. In addition, single-particle tracking showed that the average diffusion coefficient of ribosomal species decreases by a factor of 2, whereas motion of a substantial subpopulation essentially halts. Some of the 70S-polysomes are frozen in place. The damage inflicted on bacterial cells by AMPs goes well beyond membrane permeabilization. The parallel occurrence of multiple damage pathways helps explain why bacteria develop resistance to AMPs very slowly.

Modifications of natural AMP sequences have been actively investigated for their potential use as antimicrobial agents. However, such α -peptides suffer from the high cost of synthesis of a defined sequence and from susceptibil-

ity to proteolytic degradation. Substantial effort has been devoted to the development of synthetic analogs of natural AMPs that utilize different polymeric backbones. We have previously reported several sequence- and stereorandom nylon-3 polymers (β -peptide backbone) that mimic natural AMPs in antibacterial ability while minimizing hemolysis (11–15). Such sequence-random copolymers are much easier to synthesize than peptides, and they also resist proteolysis.

The primary purpose of this study is to test whether, like LL-37, the nylon-3 polymers can rigidify the chromosomal DNA and diminish ribosome mobility after penetration of the CM.

By testing polymers of varying composition and mean chain length, we can begin to understand what chemical and physical properties enable these effects. Typical DNA-DNA nearest interstrand distances in E. coli are ~ 6 nm (16). We reasoned that the ideal pseudo-cross-linking agent should have sufficient backbone contour length to span at least a 6-nm gap and sufficient cationic charge density at both ends to enable strong electrostatic binding to DNA. Accordingly, we have studied two chemically well-defined AMPs (LL-37 and cecropin A) and four polymer mixtures with different charge densities and mean backbone contour lengths as described in Table 1. Our study mainly focuses on one of the polymers called MM-CH, which is copolymerized from monomethyl β -lactam (MM) and cyclohexyl β -lactam (CH) monomers, mixed in a 65:35 molar ratio. The MM subunit after deprotection has an ammonium group, which has a positive charge at the physiological pH. The CH subunit is hydrophobic. The results reveal the multipronged nature of the attack of MM-CH on E. coli and support formation of a dense meshwork of DNA-DNA pseudocrosslinks shortly after the polymer gains access to the cytoplasm. Cecropin A, similar to LL-37 in length and with the same +6 net charge, has its positive charges concentrated toward the N-terminus and is less effective in DNA rigidification. The two longest, most densely charged polymers (MM-CH and MM homopolymer) are comparable to LL-37 in their degree of DNA rigidification. The two shorter, densely charged polymers (dimethyl-dimethylcyclopentyl (DM-DMCP) and dimethyl-dimethylcyclohexyl (DM-DMCH)) are less effective. The rigidification of DNA and the cytoplasm greatly inhibits the essential motion of ribosomal species and of the DNA-binding protein HU.

For the MM-CH copolymer and MM homopolymer, at longer times (>30 min) after the onset of polymer flow we discovered a surprising additional effect on the gross DNA morphology. In our relatively fast growth condition at 30° C (45 min doubling time), the chromosomal DNA normally exhibits two distinct, axially separate nucleoid lobes. We found that MM-CH copolymer and MM homopolymer can cause gradual coalescence of the two lobes into a single lobe of much higher apparent density localized at one end of the cell. The other four agents showed no such effect. We

Peptide/polymer ^a	Sequence/structure	Contour length ^b	Net charge	Net charge density	Rigidify DNA?	Coalesce DNA lobes?
MM-CH		11.5 nm	Avg +21	+1.83 <i>e/</i> nm	Strong	Yes
MM homopolymer		7.9 nm	Avg +22	+2.78 <i>e</i> /nm	Strong	Yes
LL-37	LLGDFFRKSKEKIG KEFKRIVQRIKDFL RNI VPRTES	10.4 nm	+6	+0.58 <i>e</i> /nm	Strong	No
Cecropin A	KWKLFKKIEKVG QNIRDGIIKAGPAV AVVGOATOLAK	10.4 nm	+6	+0.58 <i>e</i> /nm	Less effective	No
DM-DMCP		4.7 nm	Avg +9	+1.91 <i>e</i> /nm	Less effective	No
DM-DMCH		5.0 nm	Avg +11	+2.2 e/nm	Less effective	No

 TABLE 1
 Comparison of physical and chemical properties of AMPs and polymers

^aNaming convention specifies the side chains of the cationic and hydrophobic monomers used, with the cationic side chain first and the hydrophobic side chain second, separated by a hyphen.

^bThe contour length is estimated by multiplying N-to-N spatial distance by the degree of polymerization. The N-to-N spatial distance between two neighboring residues was measured with PyMol based on the crystal structures of poly- α and poly- β -peptides. The length of the end groups are neglected. The relative lengths between species can be compared. The actual end-to-end distance of the polymer will generally be shorter than the counterlength reported here. For example, we estimated LL-37 has a length of 5.7 nm in an α -helix conformation.

speculate that this new effect may be related to the wellknown polymer phenomenon of coacervation, the condensation of DNA in solution phase by a sufficient concentration of a sufficiently long positively charged polymer (17–20).

MATERIALS AND METHODS

Nylon-3 polymers were prepared in a nitrogen-purged glove box at room temperature using previously reported methods (13). Live *E. coli* cells are immobilized on a polylysine-coated coverslip in a microfluidic channel that enables us to begin flowing the antimicrobial agent at known concentration at time t = 0. An EMCCD camera images the cells in real time. The strains described in Table S2 enable fluorescence imaging of green fluorescent protein (GFP) that has been exported to the periplasm; of ribosomal species labeled by the photoswitchable protein S2-mEos2 or S2-Dendra2; of the DNA locus *Right2* labeled by ParB-GFP binding to an engineer *parS* site; or of the global distribution of chromosomal DNA labeled by the dye Sytox Orange or by the nonspecific DNA-binding protein HU-PAm-Cherry. Additional details are provided in each subsection of Results and in the Supporting materials and methods.

RESULTS

Osmotic effects and CM permeabilization by MM-CH copolymer

The MIC results for the different polymers and AMPs against different bacterial strains are shown in Table 2. Throughout this study, beginning at time t = 0, we flow $2 \times$ MIC of each agent to ensure that the vast majority of the bacterial cells are affected by the antimicrobial agent on the imaging timescale of 30–90 min. First we carried out a detailed study of the effects of copolymer MM-CH on *E. coli* using the strain JCW10 that expresses GFP from a plasmid and exports it to the periplasm (Table S2, WT-ppGFP) (11). We included 5 nM Sytox Orange in the flowing medium.

For these 90-min long, three-channel experiments, in each 12-s cycle, we interleave a phase contrast image (enabling measurement of tip-to-tip cell length), a GFP

TABLE 2	MIC for different	bacterial strains	with different agents
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Agent	VH1000	JCW154	MSG192	MG1655	JCW10	SM7	MDG196
MM-CH copolymer (µg/mL)	25	25	25	25	25	25	25
MM homopolymer (µg/mL)	31	31	31	31	31	31	31
LL-37 (µM)	4	4	4	4	4	4	4
Cecropin A (µM)	2	2	2	2	2	2	2
DM-DMCP copolymer (μ g/mL)	12.5	12.5	12.5	12.5	12.5	_	_
DM-DMCH copolymer (μ g/mL)	12.5	12.5	12.5	12.5	12.5	-	_

Note that MIC is measured in μ M for the two chemically defined AMPs, but in μ g/mL for the polymers, which lack a well-defined molecular weight. From the known molecular weights of the natural AMPs, for LL-37 we can convert MIC = 4 μ M = 18 μ g/mL; for cecropin A, MIC = 2 μ M = 8 μ g/mL. When expressed in μ g/mL, all the MIC values lie in a fairly narrow range from 8 to 31 μ g/mL.

image (providing total GFP fluorescence intensity and GFP spatial distribution), and a Sytox Orange image (determining when the dye gains access to the cytoplasm and binds to the chromosomal DNA to become fluorescent).

Results for a representative cell (Video S1) are shown in Fig. 1, A-C. Before flowing MM-CH at the concentration of 50 μ g/mL (image at t = -4 min), the GFP channel shows the characteristic periplasmic "halo" distribution (Fig. 1, A and B). Within one 12-s camera frame of the onset of flow, the apparent cell length begins to shrink. Several minutes, later it partially recovers, as observed before (11). Until t = 2.2 min, the periplasmic GFP gradually accumu-

lates at the two cell endcaps (Fig. 1, A and B). We interpret this as an osmotic effect known as "plasmolysis" (21). A large concentration of highly cationic MM-CH along with its counterions has translocated across the OM to the periplasm, leading to an apparent osmotic upshift that draws water from the cytoplasm to the periplasm. To accommodate the cytoplasmic volume decrease and to conserve the surface area of the CM, the CM collapses its endcaps and creates more periplasmic volume at the two cell poles, between the CM and OM. The periplasmic GFP redistributes to the cell poles (Fig. 1, A and B). A more detailed explanation can be found in (11). At this stage, the periplasmic GFP



FIGURE 1 Effects of copolymer MM-CH on a representative E. coli cell that exports GFP to the periplasm (Video S1). Flow of 2× MIC of MM-CH in EZRDM with 5 nM of the DNA stain Sytox Orange begins at t = 0. Images acquired at 12 s per frame. (A) Fluorescence snapshots of GFP (left) and Sytox Orange (right). (B) Projected axial (top) and transverse (bottom) GFP intensity profiles at the times indicated. (C) Cell length (black), total GFP intensity (green) and Sytox Orange intensity (orange) versus time. Note: scales for GFP intensity and cell length are very different. Over the time interval t = 2-60 min, GFP intensity decreases by 50%, but cell length decreases by only 11%. (D) Cumulative distribution function of the fraction of cells for which CM permeabilization has occurred versus time for LL-37, MM-CH copolymer, and MM homopolymer, each at 2X MIC.

has not leaked into the surroundings, indicating that MM-CH has traversed the OM without permeabilizing it to GFP.

From t = 2.2 to 2.4 min, the total GFP intensity decreases by ~20%; this loss of GFP intensity indicates that the OM has become permeabilized to GFP. The apparent cell length partially recovers, probably due to the initial osmoprotection response of the cell (active import of K⁺) (11). At t = 2.4 min the CM becomes permeabilized to GFP and the GFP intensity pattern abruptly changes to that of a filled cytoplasm (Fig. 1, A and B). The rate of loss of total GFP intensity decreases. Evidently either the OM or the CM or both membranes quickly reseal(s), much as observed earlier for melittin (10).

From t = 2.4 to 85 min, GFP intensity gradually decreases by ~50%. This decrease is due in part to photobleaching. At these longer times, we infer that the cell envelope is nearly impermeable to GFP and other such globular proteins. The behavior described here is quite general. Across several experiments, all 37 cells that successfully expressed GFP exhibited similar behavior.

Sytox Orange intensity starts to increase at t 3 min for this cell, then increases almost linearly for ~ 60 min and finally levels off from 60 to 80 min, likely due to saturation of the available DNA-binding sites. Both OM and CM have become permeable to the small dye molecule and remain so even after the transient events of high permeability to GFP. Over the same 80-min time window, cell length gradually decreases by $\sim 11\%$. This may be due to gradual loss of small osmolytes and a corresponding decrease in cytoplasmic osmolality. We define the onset of Sytox Orange intensity as the time of CM permeabilization (CMP; Table 3). In Fig. 1 D, we compare the cumulative distribution function of CMP times for MM-CH, MM homopolymer, and LL-37. In all three cases, essentially all cells have had their CM permeabilized within 25 min after treatment begins. The MM homopolymer permeabilizes the CM slightly more slowly than MM-CH, perhaps because it lacks the hydrophobic CH subunit.

Local rigidification of the chromosomal DNA by MM-CH

To characterize the local dynamics of chromosomal DNA before and after MM-CH treatment, we tracked the DNA locus *Right 2* labeled by the fusion protein ParB-GFP (strain JCW154, Table S2). In this strain, the fluorescent ParB-GFP protein expressed from a plasmid polymerizes specifically at the *parS* site engineered into the chromosome

TABLE 3 Timing and extent of CM permeabilization

Cell treatment	Fraction of cells permeabilized within 1 h	Mean time to CM permeabilization (min)
MM-CH copolymer	100%	6.8 ± 6.2
MM homopolymer	98.3%	8.3 ± 5.9
LL-37	100%	8.2 ± 4.3

near the locus *Right 2* (22). The resulting bright puncta can be tracked for some 600 frames with an exposure time of 50 ms/frame without extensive photobleaching. For comparison, we also include the results for normally growing cells (6), growing cells treated with carbonylcyanide-*m*-chlorophenylhydrazone (CCCP) + 2-deoxyglucose (which depletes ATP by the dissipation of the proton motive force and prevention of glycolysis) (23), and cells after chemical fixation by formaldehyde (6).

At 1 s/frame, we can monitor locus position with \sim 30 nm accuracy for 600 frames = 10 min. We first calculate the mean-square displacement (MSD) averaged over trajectories obtained within different 10-min time windows after the onset of MM-CH flow. For each time window, we estimate an apparent diffusion coefficient D_{app} from the slope of the linear least-squares fit to the first 10 experimental points of the MSD plot. A plot of D_{app} versus the central time for each window gives a sense of the gradual slowing of the DNA local jiggling motion over time (Fig. S1). For example, the MM-CH curve shows that the copolymer attenuates the DNA loci motion by more than a factor of 10 over 45 min (from $D_{app} = 2.0 \times 10^{-4}$ to $1.7 \times 10^{-5} \ \mu \text{m}^2/\text{s}$), and that most of the attenuation occurs over the first 10 min (Table 3). In addition, all of the polymers and AMPs permeabilize the CM of most cells within 25 min. We designate cells 25 min after injection of the antibacterial agent as "polymeror AMP-treated cells." Rigidification of the DNA occurs on the same timescale as CM permeabilization (Figs. 1 D and S1). This is consistent with our earlier study of LL-37, in which the freezing of DNA loci motion occurred within 1 min of the time of CM permeabilization ((6); Fig. 1 D).

By combining data at times greater than 25 min from all the trajectories including all experiments, for each agent we calculate the final MSD plot after treatment (Fig. 2, A and B). These MSD plots show clear negative curvature for treated cells as well as for normally growing cells (6,24). Such curvature is the signature of subdiffusive motion, as expected for the local dynamics of a small segment within a large, confined polymer. The degree of curvature is less for the agents that cause the greatest attenuation of DNA motion. To enable semiquantitative comparisons on the 10-s timescale, for each MSD curve we compute an apparent diffusion coefficient D_{app} from the slope of the linear least-squares fit to the first 10 experimental points. Numerical results (Table 4) are: $D_{app} = (2.0 \pm 0.2) \times$ $10^{-4} \ \mu m^2$ /s for normal growing cells; (4.6 \pm 0.2) $\times 10^{-5}$ μ m²/s for cells after CCCP treatment; (7.6 ± 2.1) × 10⁻⁶ $\mu m^2/s$ for fixed cells; (1.7 \pm 0.1) \times 10^{-5} $\mu m^2/s$ for cells 25 min after treating MM-CH.

Attenuation of ribosome and HU diffusion by MM-CH

To further explore the effect of MM-CH on cytoplasm, we tracked the diffusive motion of single fluorescently labeled



ribosomal particles before and after polymer treatment at 30 ms/frame. These experiments use the strain MDG196, in which the chromosomal DNA is altered to append a photoconvertible mEos2 protein to the C-terminus of the ribosomal protein S2 (Table S2). In effect, we are tracking 30S ribosomal subunits, which may occur as either free 30S subunits or 30S subunits incorporated into translating 70S ribosomes, including 70S-polysomes.

We analyzed only those trajectories that last six steps or longer and truncated the longer trajectories at six steps. In Fig. 2 *C*, we compare MSD plots for ribosomes (S2mEos2) in different conditions. The apparent diffusion coefficient D_{app} is obtained by linear fitting of the first three data

TABLE 4 Apparent diffusion coefficients of DNA loci and ribosomal species after 25 min of polymer or peptide treatment

Cell treatment	$D_{app} (\mu m^2/s)$ for DNA loci (1 s/frame)	D_{app} (μ m ² /s) for ribosomal species (30 ms/frame)
Normal growth ^a	$(2.0 \pm 0.2) \times 10^{-4}$	0.042 ± 0.001
CCCP +	$(4.6 \pm 0.2) \times 10^{-5}$	0.028 ± 0.001
2-deoxyglucose ^a		
Fixed cells ^a	$(7.6 \pm 2.1) \times 10^{-6}$	_
MM-CH	$(1.7 \pm 0.1) \times 10^{-5}$	0.010 ± 0.001
MM homopolymer	$(1.2 \pm 0.3) \times 10^{-5}$	0.0099 ± 0.0004
LL-37	$(1.0 \pm 0.1) \times 10^{-5}$	0.018 ± 0.001
Cecropin A	$(3.1 \pm 0.4) \times 10^{-5}$	0.023 ± 0.001
DM-DMCP	$(5.2 \pm 0.4) \times 10^{-5}$	0.020 ± 0.001
DM-DMCH	$(6.3 \pm 0.4) \times 10^{-5}$	0.013 ± 0.001

From slope of MSD plots. See text for additional information. ^aData from (6) included here for comparison.

FIGURE 2 Effect of AMPs and polymers on DNA and protein diffusive motion. (A) MSD versus lag time for DNA loci under various treatments, obtained from videos taken at 1 s/camera frame for 600 frames. Data for treated cells obtained at least 25 min after beginning of flow of antimicrobial agent. Left: MSD covering 100 s. Right: expanded view of the first 10 s. Apparent diffusion coefficient D_{app} obtained from linear fit to first 10 points. Numerical results are in Table 4. (B) HU-PAmCherry MSD versus lag time with trajectories taken at 30 ms/frame in normal growth and after polymer. The apparent diffusion coefficient D_{app} is obtained by linear fitting of the first three data points. The numerical results are: (0.098 \pm 0.004) μ m²/s for cells in normal growth (data from (24)); (0.034 ± 0.001) μ m²/s for cells after MM-CH; (0.045 ± 0.002) μ m²/ s for cells after MM homopolymer. (C) Ribosome MSD versus lag time from S2-mEos2 trajectories taken at 30 ms/frame in after the different treatments as shown. The apparent diffusion coefficient D_{app} is obtained by linear fitting of the first three data points. Numerical results are in Table 4.

points. The numerical results (Table 4) are: (0.042 ± 0.001) $\mu m^2 \cdot s^{-1}$ for cells in normal growth; (0.028 ± 0.001) $\mu m^2 \cdot s^{-1}$ for cells after CCCP treatment (6); (0.010 ± 0.001) $\mu m^2 \cdot s^{-1}$ for cells after MM-CH treatment. MM-CH treatment diminishes the motion of ribosomal particles by a factor of 4. A separate experiment tracking the ribosomal protein S2–Dendra2 showed similar results (Fig. S2).

In the gated diffusion model of Chow and Skolnick (16), motion of the nonspecific DNA-binding protein HU would provide another probe of local DNA dynamics. We tracked single copies of HU-PAmCherry in the strain SM7. MSD plots for HU after MM-CH treatment are quite linear on a 180-ms timescale (Fig. 2 *B*). In this case, we estimate an apparent diffusion coefficient from the slope of the linear least-squares fit to the first three experimental points. The numerical results are $D_{app} = (0.098 \pm 0.004) \ \mu\text{m}^2/\text{s}$ in normal growth conditions; $(0.034 \pm 0.001) \ \mu\text{m}^2-\text{s}^{-1}$ for cells after MM-CH; $(0.045 \pm 0.002) \ \mu\text{m}^2-\text{s}^{-1}$ for cells after MM homopolymer. MM-CH and MM homopolymer have a similar effect as LL-37 (6) in reducing the mean HU diffusion coefficient.

Coalescence of two nucleoid lobes induced by MM-CH at long times

In the initial experiments incorporating Sytox Orange into the flow, we sometimes noticed a surprising change in the intensity distribution between the two nucleoid lobes at longer times. To more incisively explore the effects of MM-CH on nucleoid morphology, we conducted a set of experiments in which cells were prestained with 500 nM Sytox Orange for 10 min before imaging (25) and Sytox Orange was omitted from the flow. In these experiments, the total Sytox Orange fluorescence intensity inside each cell is approximately constant during the time of imaging. As before, $2 \times$ MIC of MM-CH flowed beginning at t = 0. Images of Sytox Orange fluorescence were acquired at 1 min per frame to minimize photobleaching over 90 min.

Sytox Orange fluorescence snapshots of a representative *E. coli* cell during the attack of MM-CH are shown in Fig. 3 *A*. Sytox Orange integrated intensity versus time for the entire cell and for the two individual lobes labeled a

and *b* are shown in Fig. 3 *B*. The total intensity within the cell remains almost constant during the ~1.5 h of imaging. However, at $t_1 = 26$ min the left nucleoid lobe *a* begins to merge into the right nucleoid lobe *b*, as shown by the decrease in intensity in region *a* and the simultaneous increase in intensity in region *b*. We denote this time t_1 to mark the beginning of phase 1, coalescence of the two nucleoid lobes. This process continued until the combined nucleoid reached its most compact state at $t_2 = 39$ min for this cell (the beginning of phase 2). The nucleoid remained in this compact state for ~20 min, then began to expand at $t_3 = 57$ min (the beginning of phase 3). The left nucleoid lobe intensity continues to increase until the final state is



FIGURE 3 Effects of MM-CH on nucleoid morphology at long times. Chromosomal DNA was stained with 500 nM Sytox Orange for 10 min before imaging. Flow of $2 \times$ MIC MM-CH began at t = 0. Images acquired at 1 min per frame. (A) Sytox Orange fluorescence snapshots of single E. coli cell at different times. (B) Sytox Orange intensity versus time for the same cell shown in (A). Total Sytox Orange intensity from the entire cell (yellow curve), from the left nucleoid lobe (red curve, region a as in A), and from the right nucleoid lobe (blue curve, region b as in A). Total intensity from the entire cell is scaled by a factor of 1.6. Labeled times are t_1 (beginning of coalescence of nucleoid lobes), t_2 (beginning of phase 2, the nucleoid reaches its most compact state), t₃ (beginning of phase 3, partial recovery), and t_4 (nucleoids have reached the apparent final state). (C) Projected axial Sytox Orange intensity profile along the cell long axis (arrowhead in A) at different times. Left nucleoid lobe merged to the other one and partially recovered at later times. (D) The timeline of events t_1 - t_4 for 43 cells from four experiments. (E) Phase contrast, Sytox Orange fluorescence and ribosome S2-YFP snapshots of single E. coli cell at different times after flowing MM-CH.

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reached at $t_4 = 73$ min (beginning of phase 4). This final detected stage is slightly different from the initial stage in that the intensity of lobe *a* remains somewhat lower than the intensity of lobe *b*. The projected axial Sytox Orange intensity profile along the cell long axis at different times is shown in Fig. 3 *C*.

We observed 61 cells from four different experiments; 43 cells showed a similar slow nucleoid coalescence/expansion phenomenon. The timeline of events t_1 through t_4 for these 43 cells is shown in Fig. 3 D. The cells exhibit strong heterogeneity in the occurrence and timing of different events.

For most cells, the coalescence begins ~ 30 min after flowing MM-CH, much later than the osmotic effect and the membrane disruption events described above. Not all cells exhibit phase 3 and phase 4 during our 90-min imaging timescale. In phase 2, the diminished nucleoid lobe retains some residual intensity for some cells, as in Fig. 3 *A*, whereas for other cells, it completely disappears.

In phase 4, the diminished nucleoid lobe recovers to different degrees compared with the other lobe. An additional 18 cells from the four experiments did not exhibit such a nucleoid morphology change.

The analysis includes cells of all lengths. Both the cells undergoing DNA coalescence and cells not undergoing DNA coalescence include both septating cells and nonseptating cells of different lengths.

We repeated this experiment using the bacterial strain MSG192, in which the ribosomal protein S2 is labeled with YFP. We found that when the DNA lobes merged at one end of the cell, the ribosomes moved to the other end of the cell, where the DNA has become much less dense (Fig. 3 *E*, Video S2). The strong anticorrelation of the DNA and the ribosome spatial distributions that occurs in normal growth persists after the coalescence. Details are provided in Supporting materials and methods and Fig. S3.

We also carried out superresolution imaging of the nonspecific DNA-binding protein HU-PAmCherry. The results (Fig. S4) corroborate the widefield Sytox Orange results showing coalescence of the two nucleoid lobes at longer times for many cells. At t = 25 min after MM-CH addition, for cells in the length range 4.0–4.3 μ m, we observed 17 cells showing two symmetric nucleoid lobes and 22 cells showing asymmetric nucleoid lobes. In addition, the superresolution HU-PAmCherry images enable us to compare the nucleoid radius of untreated cells with that of cells treated with 2X MIC MM-CH for 25 min, as described earlier (24). Details are provided in Supporting materials and methods.

For untreated cells, $\langle r_{nucleoid} \rangle = 0.29 \pm 0.04 \ \mu m$ (33 cells with cell length in the range 4.0–4.3 μm) compared with $\langle r_{nucleoid} \rangle = 0.22 \pm 0.05 \ \mu m$ (17 cells with two nucleoid lobes and cell length in the range 4.0–4.3 μm) for cells after treatment with MM-CH. Nucleoid lengths are similar under both conditions. The apparent nucleoid volume has shrunk by a factor of ~1.7 (the square of the ra-

tio of radii) due to MM-CH treatment (Fig. S5). This may be a signature of coacervation, as discussed below.

Upper limit on the total number of charges that MM-CH brings into the cell

One of the most important parameters in the coacervation of polymer mixtures is the ratio of the total concentration of positive charges on the polycations to the total concentration of negative charges on the polyanions (18,19,26). Global DNA rigidification by electrostatic cross-linking also requires adsorption of a large number of polycationic molecules. To explore the plausibility of coacervation as well as rigidification of DNA by MM-CH, we estimate an upper limit on the amount of MM-CH per bacterial cell using a similar method as reported before (6-8). We incubated various concentrations of MM-CH polymer with various total cell counts (colony-forming units, CFU per mL) in PBS solution for 60 min in the test tube. Then we aliquoted each sample into 96-well plates supplied with EZ rich, defined medium (EZRDM) and incubated for 24 h. For each MM-CH concentration and for each initial cell count, we measured the optical density (OD) at 595 nm before and after 24-hr incubation of the cells with MM-CH. The change in optical density is defined as $\Delta OD = OD_t = 24 \text{ h} - OD_t = 0$.

If the OD does not increase for 24-h, then that amount of MM-CH was large enough to kill the initial number of cells. In this way, we can measure the minimal bulk MM-CH concentration required to kill a particular initial number of cells. For example, the MM-CH concentration of 50 μ g/mL (as used in the flow experiments) is sufficient to kill 4 × 10⁷ CFU/mL (Fig. S6). If it were the case that all of the MM-CH were absorbed by the bacterial cells, then on average each cell would have absorbed ~2.6 × 10⁹ positive charges. Because some of the MM-CH may remain in solution under these conditions, this number provides an upper bound to the actual number of positive charges absorbed. Details of the calculation are provided in Supporting materials and methods.

In the earlier work on LL-37, we used fluorescence and GC/MS to measure the fraction of LL-37 bound to the cells. For a bulk concentration of 20 μ M, we estimated that an average of ~2 × 10⁸ LL-37 copies were absorbed per cell. Net positive charge of LL-37 at neutral pH is +6, so the net positive charge absorbed per cell was ~1.2 × 10⁹, comparable with the estimated upper bound on MM-CH uptake.

Recovery experiments

To explore whether or not the short-term DNA rigidification and longer-term DNA coalescence are related to bacterial growth inhibition, we conducted a series of recovery experiments. In these experiments, we flow $2 \times$ MIC of MM-CH in EZRDM across plated WT cells for 5 or 10 min and then change the flow to EZRDM without MM-CH for an additional 90 min. Phase contrast imaging provides an estimate of cell length versus time to test for possible recovery of cell growth. We observed 37 cells after 5-min exposure to MM-CH and 26 cells after 10-min exposure. In all 63 cases, the brief exposure was sufficient to completely halt apparent cell growth for at least 90 min of observation time by phase contrast imaging. However, for 5-min exposure to MM-CH, all 37 cells retained two separated nucleoid lobes (no DNA morphology change). For 10-min exposure to MM-CH, 18 of the 26 cells eventually showed the DNA coalescence behavior described above, even though the medium had been changed to EZRDM.

Comparison of LL-37, cecropin A, and other synthetic copolymers

Finally, we tested the other agents in Table 1 for their ability to locally rigidify the chromosome and to induce coalescence of the two chromosomal DNA lobes. The results provide some insight into what features of an antimicrobial agent enable these effects. From the DNA locus tracking data taken at times greater than 25 min of treatment by each agent, we obtained the MSD plots shown in Fig. 2 A. The apparent diffusion coefficients D_{app} from linear fits to the first 10 data points are collected in Table 4. LL-37, MM-CH copolymer, and MM homopolymer rapidly rigidify local chromosomal motion almost as effectively as chemical fixation by formaldehyde. Cecropin A and the two shorter copolymers DM-DMCH and DM-DMCP reduced the DNA loci motion by a factor of 3-7 compared with normally growing cell. This is comparable to the effect of ATP depletion by the CCCP treatment. The diffusion coefficients of ribosomes under different treatments are also included in Table 4. The antimicrobial agents all slowed ribosome diffusion more than did ATP depletion.

Coalescence of the two nucleoid lobes on a 90-min timescale was induced only by MM-CH and by MM homopolymer. The natural AMPs LL-37 and cecropin A and the two shorter copolymers DM-DMCP and DM-DMCH did not cause coalescence on a 90-min timescale.

DISCUSSION

Short-term effects of cationic polymers and peptides

This study has compared the effects on *E. coli* of the six antimicrobial agents whose chemical compositions, contour lengths, and linear charge densities are given in Table 1. At $2 \times$ MIC, the long, highly charged copolymer MM-CH (11.5-nm mean contour length, mean charge +21, mean linear charge density +1.8 *e*/nm) penetrates both the OM and CM remarkably quickly. Once MM-CH has permeabi-

lized both the CM and the OM to small molecules, there occurs a massive influx of highly positively charged copolymer into the cytoplasm. After gaining access to the cytoplasm, MM-CH causes several striking changes in the chromosomal DNA. On a timescale of \sim 5–10 min, the local jiggling motion of the DNA polymer is greatly attenuated, an effect we call rigidification of the nucleoid (6). On a similar timescale, the apparent volume of each of the two separate nucleoid lobes decreases by an estimated factor of \sim 1.7. On a much longer timescale of \sim 30 min, in many of the cells one nucleoid lobe slowly migrates to coalesce with the other lobe, forming a dense, single-lobed nucleoid that strongly excludes ribosomal species. In many cases, this coalescence partially reverses on the timescale 60-90 min after initiation of copolymer flow. All of these same events were caused by the somewhat shorter (7.9 nm), but even more densely charged MM homopolymer (+22, +2.8 e/nm).

In contrast, the shorter DM-DMCP and DM-DMCH copolymers were less effective in rigidifying the DNA and did not cause large-scale coalescence of nucleoid lobes. DM-DMCP and DM-DMCH have similar charge densities as MM-CH, but different chemical compositions and also substantially shorter mean contour lengths (4.7 and 5.0 nm, respectively). The natural AMP LL-37 was as effective as MM-CH in locally rigidifying the DNA. The natural AMP cecropin A was less effective. Neither AMP caused large-scale nucleoid lobe coalescence.

As before (6), we suggest two underlying causes of the attenuation of the DNA jiggling motion. First, permeabilization of the CM and concomitant reduction in the transmembrane potential will slow or halt ATP production and presumably enable leakage of existing ATP to the cell surround.

This reduces the motor driven contribution to the jiggling motion of the DNA, as shown by the CCCP assay and observed earlier by others (27,28). This effect likely occurs for all six agents studied here. Second, we suggest that additional local rigidification of the DNA can be caused by "electrostatic pseudo-cross-linking" of nearby DNA strands within the dense nucleoid. Here, we envision transient, noncovalent binding of long polycationic species to release nearby K⁺ counterions and hinder the relative motion of adjacent DNA strands. The coarse-grained simulations of Chow and Skolnick (16) suggest typical gaps between nearby DNA strands of ~ 6 nm. We suggest that longer molecules with high positive-charge density should accomplish this cross-linking more effectively. Accordingly, the MM-CH copolymer and MM homopolymer are quite long and densely charged. LL-37 is long but less densely charged. However, LL-37 has positive Lys and Arg residues at either end of the chain, enabling positive charges to span substantial gaps between DNA stands. Cecropin A concentrates its charges at its N-terminal half, rendering it less able to span large gaps with its positive charge.

The same pseudo-cross-linking effect may also explain the threefold decrease in the mean diffusion coefficient of HU (Fig. 2 *B*) induced by MM-CH on the 25-min timescale. The HU trajectories presumably involve an average over time spent diffusing "freely" (not bound to DNA) within the nucleoid and time spent bound to DNA. The coarsegrained simulation model of Chow and Skolnick (16) suggests that thermal motion of the DNA strands enables hopping of the transcription repressor protein LacI from one DNA cage within the nucleoid to an adjacent cage, a process they called "gated diffusion." It seems plausible that the observed rigidification of the DNA would hinder gated diffusion. A similar reduction in HU diffusion is induced by MM homopolymer (Fig. 2 *B*) and by LL-37, the two agents that are most effective at DNA rigidification.

All six agents decrease the mean diffusion coefficient of ribosomal species 25 min after treatment by a factor the varies from 2 to 4 (Fig. 2 C), all larger effects than CCCP treatment. Each mean diffusion coefficient is an average over an unknown distribution of ribosomal species, including 30S subunits, 70S complete ribosomes, 70S polysomes, and potentially agglomerations of such ribosomal species induced by the polycationic agent (29). Pseudocross-linking by polycationic agents may still occur between these negatively charged ribosome species to slow diffusion of ribosomes.

Coacervation?

We have no clear understanding of the intriguing coalescence of the two nucleoid lobes induced at $t \sim 30$ min, observed only for the longer, densely charged MM-CH copolymer and by the MM homopolymer, not for the other four agents. We tentatively suggest that this behavior is related to the well-known phenomenon of associative phase separation or coacervation (18,26,30). In certain equilibrium solution phase conditions, sufficiently long and densely charged polycations at sufficiently high concentration are able to condense DNA from its natural elongated, swollen structure into a compact structure via coacervation (26). There is in essence a phase separation between a polycation/polyanion rich phase and the remaining polymerdepleted portion of the solution.

A detailed study in vitro (26) found that 92-mer poly-Llysine induces gradual compaction of DNA to a shrunken coil structure as the relative concentration of poly-L-lysine increases. The compaction was greatest when the ratio of total positive charges on the polycations to total negative charges on DNA reached one (17,26). The coacervation mechanism did not occur for short poly-L-lysine (n < 9) (17,26). Small cations can compact DNA, but only at much higher concentrations (26).

The *E. coli* cytoplasm is dense with polyanionic species to which the polycationic MM-CH copolymer or MM homopolymer should bind strongly (31). As an upper limit,

we estimated that on average, each cell may absorb as many as $\sim 10^9$ positive charges carried by the MM-CH copolymer. For a typical total cell volume of $\sim 1.9 \ \mu m^3$, this corresponds to an average concentration of ~ 0.9 M of polymeric positive charge, a very large number. Under our growth conditions we estimate that cytoplasmic polyanionic species (chromosome, ribosome, transfer RNA (tRNA), mRNA) provide $\sim 2 \times 10^8$ negative charges ($\sim 200 \ mM$); these are compensated primarily by K⁺ (Supporting materials and methods). It is plausible that sufficient MM-CH is absorbed within the cytoplasm to cause coacervation with the chromosomal DNA.

Before the addition of MM-CH, the chromosomal DNA is already highly compacted due to pairwise binding by a variety of nucleoid-associated proteins (17,19,32-34); segregative phase separation due to DNA-ribosome repulsion (19,35,36); configurational entropy of the confined DNA polymer to avoid the cell boundaries (37); neutralization of DNA charges by multivalent ions and some DNA-binding proteins; macromolecular crowding (17,30,38,39); and DNA supercoiling (17,33,40,41). Nevertheless, as MM-CH gains access to the cytoplasm we observe significant radial shrinkage of the two nucleoid lobes, with the mean radius decreasing from 0.29 to 0.22 nm (for cells in 4-4.3 μ m length range, Fig. S5). The nucleoid volume decreases by a factor of \sim 1.7. This is consistent with the suggestion that MM-CH causes coacervation of the chromosomal DNA before the longer-term coalescence of the two nucleoid lobes. In contrast, in the earlier study LL-37 did not contract the nucleoid volume. LL-37 is likely insufficiently positive and too short to induce this effect.

The coacervation hypothesis may also provide a partial explanation for the observed slow migration of one nucleoid lobe to coalesce with the other lobe. In solution phase, coacervate droplets have been observed to coalesce with each other over time, perhaps to minimize surface tension with the surrounding phase (42,43). Similar behavior of liquid droplets has been observed in eukaryotic cells (44-46). We lack a good rationale for the slow reversal from the one-lobed, coalesced condition back toward the original two-lobed condition. On the 90-min timescale of a video (Video S1, Video S2), this occurs in roughly half of the that showed coalescence. Concentrations of the myriad species within the cytoplasm continue to change as the concentration of absorbed polymer increases. The coacervation process occurs at a narrow concentration ratio range for polycations and polyanions (18,43,47,48).

CONCLUSIONS

To date, all of our time-resolved, single-cell mechanistic studies have used linear, polycationic antibacterial agents, including both AMPs of defined sequence and sequence-random polymers (6,9-11,49). Does permeabilization of the bacterial membranes by these agents cause the halting

of growth? Our experience supports the conclusion that there is a threshold concentration above which permeabilization of the OM occurs; below that concentration cells continue to grow, albeit sometimes more slowly. Permeabilization of the CM then occurs rapidly and eventually growth halts in a seemingly irreversible fashion. CM permeabilization destroys the transmembrane potential and rapidly causes loss to the cell surround of small molecules such as ATP and perhaps of globular proteins and tRNA. These same cationic agents may also interfere with a wide variety of cellular processes including cell envelope biosynthesis, DNA replication and transcription, and protein synthesis (2,4,5). A surprising effect that is beginning to seem quite general is the massive uptake of polycationic agents once the OM and CM have been breached (6-8). A bacterial cytoplasm exposed to the surround appears to be a sponge that soaks up polycations to very high internal concentrations $(10^8 - 10^9)$ positive charges). Again, we suggest that the key driving force is strong, noncovalent, multivalent electrostatic binding between polycations and polyanions (DNA, ribosomes, mRNA, tRNA, and even globular proteins) (19,31). Nature has designed a bacterial cytoplasm that is remarkably susceptible to damage by polycationic agents. At least for linear, polycationic agents, our view is that membrane permeabilization appears to be a necessary first step leading to a cascade of downstream events, many occurring on a timescale of a few minutes or less. With so many effects likely occurring on a similar timescale, it is difficult to attribute the halting of growth (and apparent killing of cells) to any single event, but membrane permeabilization is the necessary triggering event.

This work on cationic polymers highlights several new effects: rigidification of the DNA and attenuation of ribosome and protein diffusion on a short timescale of several minutes and coalescence of the two nucleoid lobes at longer times of 30 min or more. The recovery experiments show that the cells absorb sufficient MM-CH copolymer during the first 5 min of exposure to prevent subsequent growth for at least 90 min after the flow had switched to fresh EZRDM without copolymer. These same cells do not subsequently show DNA coalescence or gross content leakage (no white band on phase contrast image). Evidently sufficient MM-CH enters the cells in 5 min to induce enough damage to halt growth. In contrast, after a 10 min exposure to MM-CH followed by 90 min in fresh EZRDM, cells eventually show both nucleoid coalescence and complete membrane disruption. The additional polymer absorbed over the interval from 5 to 10 min seems essential to drive the long-term nucleoid coalescence. Such coalescence is evidently not necessary to halt growth—only a fraction of the cells exhibited this behavior. In addition, all six of the tested agents halt the growth of E. coli, but some agents cause the long-term coalescence of the two nucleoid lobes and others do not. It also seems unlikely that DNA rigidification is necessary to halt growth. The agents exhibit widely varying degrees of DNA rigidification and diffusion attenuation.

Finally, we find it remarkable that the sequence-random, cationic polymers studied here are so readily able to traverse the OM and CM of *E. coli*. This result seems incompatible with models of membrane disruption that posit formation of structured pores, either toroidal or barrel-stave.

SUPPORTING MATERIAL

Supporting material can be found online at https://doi.org/10.1016/j.bpj. 2021.10.037.

AUTHOR CONTRIBUTIONS

Y.Z. and M.M. carried out the microscopy studies. L.L., L.A.R., and S.H.G. designed, synthesized, and characterized the polymer samples. Y.Z. and J.C.W. planned the microscopy studies and wrote the article, with help from L.L. and S.H.G.

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