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Flavonol-Based Carbon Monoxide Delivery Molecule with Endoplasmic Reticulum, Mitochondria, And Lysosome Localization

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ABSTRACT: Light-triggered carbon monoxide (CO) delivery molecules are of significant current interest for evaluating the role of CO in biology and as potential therapeutics. Herein we report the first example of a metal free CO delivery molecule that can be tracked via confocal microscopy at low micromolar concentrations in cells prior to CO release. The NEt₂-appended extended flavonol (4) localizes to the endoplasmic reticulum, mitochondria, and lysosomes. Subcellular localization of 4 results in CO-induced toxicity effects that are distinct as compared to a nonlocalized analog. Anti-inflammatory effects of 4, as measured by TNF- α suppression, occur at the nanomolar level in the absence of CO release, and are enhanced with visible-light-induced CO release. Overall, the highly trackable nature of 4 enables studies of the biological effects of both a localized flavonol and CO release at low micromolar to nanomolar concentrations.

KEYWORDS: signaling, gasotransmitter, metal-free, aminoflavonol, cytotoxicity

K nown for centuries for its toxicity, carbon monoxide (CO) is of considerable current interest with regard to its role in biological signaling and as a potential therapeutic.^{1,2} Produced endogenously in humans via the oxidative catabolism of heme,³⁻⁶ CO is known to produce anti-inflammatory, antiapoptotic, antihypertensive, vasodialation, and cytoprotective effects.⁷ Delivery of controlled amounts of CO is also known to produce antibacterial and anticancer effects.^{8,9} On this basis, CO is of significant current interest for several biomedical applications.^{10,11}

Heme oxygenases (e.g., HO-1) catalyze the O₂-dependent CO release from heme (Scheme 1a). The subcellular distribution of HO-1 is dynamic and is regulated by cellular homeostasis.¹² Under normal cell conditions, HO-1 is as a cytosol-facing endoplasmic reticulum (ER)-associated protein.¹² Under stress, HO-1 translocates in part to mitochondria, nucleus, and caveolae.¹³ This stress-dependent distribution raises intriguing questions about how the intracellular location of CO release impacts its signaling effects.

As an approach toward investigating the biological roles of CO and its potential as a therapeutic agent, carbon monoxide releasing molecules (CORMs) were developed.^{7,14,15} Although

a significant body of CO research has been reported using metal-carbonyl-based CORMs, other biological activities of these complexes have also been identified.¹⁶⁻¹⁸ The most commonly used metal carbonyl CORMs for biological studies, CORM-2 and CORM-3, release CO spontaneously in buffer and thus do not offer the possibility of examining the effects of localized intracellular CO delivery.^{19,20}

Several laboratories are pursuing the development of metalfree CO delivery molecules.^{21–33} The frameworks receiving the greatest attention for potential biomedical applications are norborn-2-ene-7-one derivatives developed by Wang and coworkers²⁷ and extended flavonols developed in our laboratory.^{23,31} Although the former exhibit spontaneous CO release, the rates of which can be tuned via structural modifications,³³

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Scheme 1. CO Release Reactions of (a) Heme Oxygenase and (b) 1



CO release from extended flavonols is triggered using visible light. The flavonols thus offer the possibility of highly controlled and localized CO release, including when delivered as part of micelles, materials, or polymers.^{8,34–38}

We have previously reported that 1 (Scheme 1b) undergoes visible-light-induced CO release to give 2^{23} Compound 1 is fluorescent, whereas 2 is nonemissive. These attributes enable fluorescence tracking of 1 in A549 cells prior to CO release.²³ A concentration of 50 μ M or higher is needed for visualization of 1 in cells as its emission rapidly bleaches upon illumination. This behavior has thus far prevented studies of 1 using confocal microscopy. The fluorescence microscopy studies of 1 suggest dispersion throughout the cytoplasm.

The development of molecules for localized intracellular CO delivery remains in its very early stages. Wang has reported a bio-orthogonal reaction-based pair of compounds that are localized to mitochondria via phosphonium appendages.³⁹ A fluorescent fluoranthene product provides evidence for localized CO release in this system. A phosphonium-appended, mitochondria-targeting extended flavonol (3, Scheme 2) is trackable via its emission prior to visible-light-induced CO release.⁴⁰ This localized delivery of CO from 3 (at 10 μ M) produces decreases in mitochondrial basal respiration, ATP

Scheme 2. (Top) Mitochondria-Targeting 3; (Bottom) CO-Release Reaction of 4



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production, maximal respiration, and reserve capacity. The observed changes are similar to those found for cytosolic 1, indicating that at a concentration of 10 μ M, localized CO delivery to mitochondria does not produce differentiating effects on function. However, there are differences in the toxicity of the compounds, with 3 being more toxic in A549 cells (IC₅₀ = 14.1 ± 2.7 μ M) than 1 (IC₅₀ = 80.2 ± 3.3 μ M). With visible-light-driven CO release, 3 shows a mild increase in toxicity (IC₅₀ = 4.6 ± 3.6 μ M), whereas no change in toxicity occurs with CO delivery from 1 (IC₅₀ = 76.1 ± 5.2 μ M). The organic byproducts produced in the CO release reactions of 1 and 3 are both nontoxic up to 100 μ M.

Additional studies are needed to define the relationship between the location of intracellular CO delivery and COinduced effects. Of particular need are metal-free CO delivery molecules that can be tracked via fluorescence at low micromolar or nanomolar concentrations prior to CO release. In this regard, flavonols containing a *para*-dialkylamino substituent on the B-ring are known for their intense fluorescence.⁴¹ Such molecules have been used as polaritysensitive probes and fluorescent sensors for biological applications.^{42–51} Notably, flavonols containing a $-NPh_2$ moiety have been reported to localize to the ER.^{41,50}

We have previously reported the synthesis and light-induced CO release of 4 (Scheme 2) in organic solvents.²³ However, its properties in cells, including its fluorescence trackability and the effects produced when it is used as an intracellular CO delivery molecule, have not been previously investigated. In the studies reported herein, we find that 4 (Scheme 2) is trackable in cells at concentrations as low as 1 μ M. This observation sets a new benchmark for the field in terms of a trackable metal-free CO delivery molecule.⁵² Confocal microscopy studies indicate 4 localizes to the ER, with additional localization at mitochondria and lysosomes. Although it is nontoxic up to 100 μ M in the dark, light-induced CO release from 4 produces significantly enhanced toxicity at low micromolar concentrations relative to that produced by cytosolic 1. The localization of 4 also leads to significant anti-inflammatory properties at nanomolar concentrations, an effect that is not produced by 1 at similar concentrations. Visible-light-induced CO release enhances the anti-inflammatory effects of both compounds. Overall, the subcellular localization of 4 produces novel outcomes, suggesting the importance of localized CO delivery in producing biological effects.

Both 1 and 4 can be produced in analytically pure form (>95% by HPLC, Figures S1 and S2) via one-pot synthetic procedures followed by precipitation and washing. The pyrone rings from which CO is released in 1 and 4 are structurally similar, with no significant changes due to the presence of the $-\text{NEt}_2$ group (Figure S3 and Table S1). The absorption maximum for 4 in DMSO and acetonitrile is ~445 nm with a molar absorptivity values of ~32 000–36 000 M⁻¹ cm⁻¹ (Figure S4). This absorption feature is red-shifted by ~25 nm versus the absorption maximum of 1.²³ When illuminated at its absorption maximum either in CH₃CN or DMSO ($\lambda_{ex} = 445$ nm), 4 exhibits a broad intense emission at ~550 nm (Figure S5). The fluorescence quantum yield and lifetime for 4 ($\Phi_{PL=}$ 25.5; 6.1 ns) are similar to those found for 1 ($\Phi_{PL=}$ 34.5; 7.9 ns).²⁶

We have previously reported that 4 (Scheme 2) exhibits a clean visible-light-induced CO release (0.96(2) equiv.) in acetonitrile.²³ A similar reaction was identified in d_6 -DMSO using ¹H NMR (Figure S6). The CO release in this reaction

was measured as 1.01 ± 0.02 equiv. Both 1 and 4 undergo light-induced (419 nm) CO release in DMEM/F12K media under air as evidenced by loss of their absorption features (Figure S7). The quantum yields for these CO release reactions are similar regardless of solvent (CH₃CN: 1, 0.7(3)%; 4, 0.6(1)%; DMSO: 1, 0.6(3)%;²⁶ 4, 0.5(4)%). The quantum yield for CO release from 4 in DMEM/F12K media (10% DMSO) is similar (0.3(1)%). The organic byproducts 2 and 5 (Schemes 1 and 2) generated in the CO release reactions of 1 and 4 (5: Figure S8) do not absorb in the visible region. Compound 4 is stable in CH₃CN for greater than four months if the solution is protected from light. Compounds 1 and 4 interact weakly with bovine serum albumin protein (BSA) (TRIS:DMSO (96:4% v:v, pH 7.4, 298 K), with K_a values of $3.2 \times 10^3 \text{ M}^{-1} (1)^{53}$ and $8.1 \times 10^2 \text{ M}^{-1}$ (4), respectively, and substoichiometric binding (n = 0.66 and)0.54, respectively; Figure S9). In the presence of BSA (40 equiv.), the quantum yields for CO release for 1 and 4 are 0.06(1) and 0.16(1)%, respectively.⁵³

Cytoxicity studies of 1 and 4 were performed in human lung epithelial adenocarcinoma cells (A549), normal human lung fibroblast cells (HFL-1), and mouse macrophage cells (RAW264.7). Compound 1 exhibits mild cytotoxicity in all of the tested cell lines, with the following mean IC₅₀ values: 83.0 ± 1.2 μ M (A549), 65.1 ± 1.8 μ M (HFL-1), and 47.7 ± 1.4 μ M (RAW 264.7), respectively. Notably, in the dark, 4 is nontoxic in A549 and HFL-1 cells up to 100 μ M (Figure 1, Figures S10 and S11, Table S2). The compound exhibits mild toxicity in RAW264.7 cells (IC₅₀ = 67.2 ± 2.0 μ M, Figure S12).



Figure 1. Plot of percent cell viability in A549 cells versus concentration of compounds.

Light-triggered intracellular CO release from 4 produces a significant decrease in IC₅₀ values versus those found for 4 without illumination (Figures S10–S12; Table S3). For example, as shown in Figure 1, the increase in toxicity produced upon CO release in A549 cells is characterized by an IC₅₀ = 10.7 \pm 1.2 μ M. This result is notably different than that found for 1, where IC₅₀ values changed only slightly with CO release (Figure 1 and Table S3). Similar CO-induced effects on IC₅₀ with 4 were observed in HFL-1 and RAW264.7 cells. The CO release products 2 and 5 were found to be nontoxic (up to 100 μ M) in most of the cell lines examined (Table S3). The significant difference in toxicity prior to and following CO release from 4 is a notable feature that has not been identified for other visible-light-induced CO delivery molecules.

Fluorescence microscopy studies of 4 performed in A549 cells at 20× resolution are shown in Figure 2. The emission of 4 is trackable at concentrations as low as 1 μ M. In RAW264.7 and HFL-1 cells, 4 is visualizable at concentrations as low as 5



Figure 2. Concentration-dependent fluorescence microscopy images (20×) of A549 cells incubated for 4 h with 4 at 1, 5, and 10 μ M followed by washing of the cells prior to imaging. The cells were costained with Hoechst 33342 nuclear dye (blue) to assess cell integrity. Scale bar = 40 μ m.

and 1 μ M, respectively (Figures S13 and S14). This compound is the most trackable metal-free CO donor reported to date.⁵¹ CO release from 4 can be followed in cells at 1 μ M using a CO sensor (Figure S15).⁵⁴

Confocal microscopy studies of 4 in A549 cells showed clustering of the emission (Figure 3), which led us to assess the



Figure 3. Confocal microscopy images (63×) showing 4 in A549 cells. Row 1, media control. Row 2, cells treated with 4 (1 μ M) for 1 h. Images depict the Hoechst nuclear stain (blue), compound 4 (green), and a merge of the two channels. Scale bar = 10 μ m.

subcellular localization of the compound via colocalization studies with a series of organelle-specific fluorescent dyes. We first compared the intracellular emission of 4 (Figure 4) to that of ER-Tracker Red.⁵⁵ Incubation of A549 cells with 4 at 25 μ M for 1 h showed that the compound localizes similarly to the ER-Tracker Red as indicated by the yellow color in the merged image. The overlap between 4 and ER-Tracker Red was determined to be substantial as evidenced by the Pearson's colocalization coefficient, r = 0.88 ± 0.04 (mean ± standard



Figure 4. Co-localization of 4 with ER-Tracker Red in A549 cells. Cells were treated with vehicle control (0.4% DMSO) (row 1), or 4 ($25 \ \mu$ M, row 2) for 1 h, then counterstained with Hoechst 33342 and ER-Tracker Red. These images depict the Hoechst nuclear stain (blue channel), 4 (green channel), ER-Tracker Red (red channel), and a merge of the three channels. Scale bar = 50 μ m.

error of the mean for 27 cells examined individually examined across three separate experiments). Regions of interest and their spatial resolution were evaluated as a line profile over a distance of 20 μ m (Figure 5). The congruence of the intensity profiles of 4 and ER-Tracker Red strongly suggest that 4 localizes to the ER.



Figure 5. Confocal images of A549 cells costained with 4, ER-Tracker Red, and Hoechst 33342. (a) Independent and colocalized pixels of 4 and ER-Tracker Red. (b) Overlaid intensity profile of regions of interest in the costained A549 cells as indicated by the white arrows.

Similar confocal analyses were performed using MitoTracker Deep Red (MTR) to probe for mitochondria localization (Figures S16 and S17). Using an experimental approach similar to that described for ER-Tracker Red, a Pearson's colocalization coefficient ($r = 0.74 \pm 0.09$) for MTR was determined. Similarly, colocalization studies with LysoTracker Red produced a Pearson's colocalization coefficient of 0.66 \pm 0.08 (Figures S18 and S19).

The identification of subcellular localization to the ER, mitochondria, and lysosomes has been observed with other fluorophores.^{56,57} However, to the best of our knowledge, **4** is the first CO delivery molecule to exhibit localization to the ER, a network of membranous tubules with embedded trans-

membrane proteins, including the CO-generating HO-1.¹³ The anti-inflammatory and antiapoptotic properties of CO may be important toward mitigating ER stress associated with vascular diseases.⁵⁸ To date, a small number of studies focused on examining the effect of CO delivery on ER stress have been reported.^{59,60} These were performed using CORM-2, a Ru(II)-containing spontaneous CO delivery molecule with no reported subcellular localization properties. Recent identification of Ru(II) impacting the results of other biological studies involving CORM-2⁶¹ suggests that new CO delivery tools such as 4 should be considered to probe the impact of CO on ER stress.

Compound 4 is the second extended CO-releasing flavonol to show mitochondrial localization, with 3 (Scheme 2(top)) being the other example.⁴⁰ The differences in toxicity between 3 and 4 in the presence and absence of CO release are notable. In the absence of CO release, 3 produces significant toxicity in A549 cells (IC₅₀ = 14.1 ± 2.7 μ M), whereas 4 is nontoxic up to 100 μ M. Whereas CO release from 3 produces a minimal increase in toxicity (IC₅₀ = 4.6 ± 3.6 μ M), CO release from 4 results in notably increased toxicity (IC₅₀ = 10.7 ± 1.2 μ M; Figure 6). The enhanced toxicity induced via CO release from



Figure 6. Comparison of IC₅₀ values of extended flavonols in A549 cells.

4 versus cytosolic 1 strongly suggests that differences in subcellular localization modulate CO-induced effects. This is an area in significant need of advancement, as little data currently exist regarding the effects of localized CO release.

The lysosome localization of **4** is not unexpected as dialkylamino-appended fluorophores for lysosome targeting have been previously reported.⁶² This is the first CO-releasing molecule to exhibit lysosome localization.

We next determined how CO release from 4 attenuates LPSinduced inflammation by suppressing the production of TNF- α . We have performed similar experiments using 1 and a sulfonated analog.⁶³ Experiments with 4 were independently performed in the dark and under illumination conditions. As shown in Figure 7, compound 4 exhibits significant TNF- α suppression at 40 nM in the absence of CO release. This antiinflammatory effect is the most potent of the extended flavonols that we have examined thus far, suggesting an important influence resulting from subcellular localization. With CO release, the anti-inflammatory effects of 4 at nanomolar concentrations are enhanced, as evidenced by greater TNF- α suppression (Figure 7). We note that the concentration dependent increase in TNF- α suppression is attenuated for 4 as compared to that of 1, which may be a result of its enhanced effectiveness at low concentrations. Previously reported CORMs, such as CORM-3 and a BSA- $Ru(CO)_2$ conjugate, reduce TNF- α expression in RAW264.7



Figure 7. Anti-inflammatory effects of **1** and **4** in RAW 264.7 cells under dark or light conditions. The results are presented as the mean \pm SEM from three independent experiments. \bigstar , p < 0.01 compared to LPS positive control; #, p < 0.05 compared to corresponding nonilluminated treatment with red for **1** and blue for **4**.

murine macrophages at concentrations of 10 and 4.5 μ M, respectively.^{64,65} Spontaneous metal-free CO releasing molecules produce anti-inflammatory effects at similar low micromolar concentrations.⁶⁶ Suppression of TNF- α by nanomolar concentrations of 4 with CO release is most similar to that produced via CO release from a BSA-delivered quinolone, which cannot be trackable in cells because of low fluorescence intensity.²⁶

Highly trackable, organelle-targeted metal-free CO donors remain rare.⁶⁷ The low micromolar fluorescence trackability of 4 is notable, making it a novel probe to evaluate the effects of localized intracellular CO release. The subcellular localization of 4 to the ER, mitochondria, and lysosomes produces significant CO-induced toxicity effects that are distinct from analogs with different localization properties. The observed difference in toxicity provides evidence that the intracellular localization of the CO release influences the magnitude of its biological effects. Further investigations are underway using a series of amino-appended flavonols to examine subcellular localization effects on CO-induced toxicity and anti-inflammatory effects. Overall, 4 represents a prototype on which to base the development of additional highly fluorescent, localized, and triggered CO delivery molecules to define how the site of CO delivery impacts it biological effects. We note that because of the limited number of hydroxyl substituents in 4 versus naturally occurring flavonols (e.g., quercetin), this compound is expected to have less biological promiscuity and does not register warnings as a PAINS substance.⁶⁸ Overall, the results outlined here provide evidence that 4 is a novel compound for localized CO delivery, which might also be pursued using materials-based approaches.34-38

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsmedchemlett.1c00595.

Experimental procedures as well as spectroscopic, cell viability, and confocal imaging data (PDF) Crystallographic information file for 4 (CIF)

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Author Contributions

L.M.B., L.S.L., C.T.D., and A.B.D. designed the research. L.S.L., C.T.D., and S.N.A. performed the experiments. L.M.B., L.S.L., C.T.D., S.N.A., and A.B.D. performed the data analysis. The manuscript was written by L.M.B., L.S.L., C.T.D., and A.D.B.. All authors approved the final version of the manuscript.

Notes

The authors declare no competing financial interest.

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ABBREVIATIONS

DMEM, Dulbecco's modified Eagle medium; IC₅₀, half minimum inhibitory concentration; LPS, lipopolysaccharide; MTR, MitoTracker Deep Red; TNF, tumor necrosis factor; SEM, standard error of the mean; PAINS, pan assay interference compounds

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