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Synthesis and Biological Characterization of a Series of 2‑Sulfonamidebenzamides as Allosteric Modulators of MrgX1

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our original compound, ML382, and utilizing iterative medicinal chemistry, we have identified key halogen substituents that improve MrgX1 potency by ∼8-fold. In addition, we have evaluated the compounds in Tier 1 drug metabolism and pharmacokinetics assays and have identified key compounds that impart improved potency and microsomal stability.

KEYWORDS: Mas-related G-protein coupled receptor, MrgX1, Structure−activity relationship, Positive allosteric modulator, 2-Sulfonamidebenzamide

Chronic pain is a major health and economic problem
worldwide with an estimated prevalence reaching
widowis layels of 20 to 25% of the nonulation 1 In the epidemic levels of 20 to 25% of the population.¹ In the United States, chronic pain affects over 116 million [ad](#page-5-0)ults and costs up to \$635 billion annually in treatment and lost productivity, more than those for cancer, heart disease, and diabetes combined. 2 Chronic pain, especially nerve injury induced neuropathi[c](#page-6-0) pain and spontaneous pain, is challenging to treat and often refractory to current pharmacotherapies. Because the major analgesics (e.g., opioids) bind to receptors that are widely expressed throughout the central nervous system (CNS), dose-limiting adverse effects and risks of addiction and abuse present substantial barriers to their clinical use. 3 Pain sensing neurons (a.k.a nociceptors) in dorsal root gan[gl](#page-6-0)ion (DRG) play an essential role in pain transmission by detecting painful signals in the periphery such as skin and viscera. Therefore, targeting molecules specifically expressed in nociceptors may offer an opportunity for pain-selective pharmacologic interventions.

Mas-related G-protein coupled receptors (Mrg receptors) are a large family of orphan receptors expressed in small diameter sensory neurons and represent a set of potential targets for pain. Mrgs (also called Mrgprs) are a family of orphan G-protein coupled receptors (GPCRs) consisting of more than 50 members in the mouse genome. 4 Our previous data have shown that Mrgs including mous[e](#page-6-0) MrgC11 and human MrgX1 are specifically expressed in nociceptors in DRG and constitute an endogenous antipain pathway. 5 The expression of many Mrgs, such as mouse MrgC11 a[n](#page-6-0)d its human homologue MrgX1, is restricted to subsets of nociceptors in DRG, but not detected in the CNS (i.e., the

spinal cord and brain) or in the rest of the body. $4,6$ Unfortunately, previous studies on MrgX1 were hampered [as](#page-6-0) it became clear that human MrgX1 has binding and pharmacological profiles distinct from the binding and pharmacological profiles of rodent MrgC11, thus making it difficult to translate studies since it was not feasible to use traditional animal models to evaluate MrgX1 ligands in pain models. This roadblock was overcome with the generation of a transgenic mouse line where the MrgX1 gene is expressed in $MrgC11$ -expressing DRG neurons.⁵ This new humanized mouse line makes it possible to ex[am](#page-6-0)ine MrgX1 modulators in animal models of pain. With this new mouse model, we were able to show that activating MrgX1 with a positive allosteric modulator (PAM) (ML382) lessened both evoked pain and spontaneous pain after injury, without any observable side effects.⁵ ML382 was the first reported MrgX1 PAM and herein, we re[po](#page-6-0)rt additional structure−activity relationship (SAR) studies culminating in the discovery of a next-generation MrgX1 PAM.⁷

Our origin[al](#page-6-0) medicinal chemistry efforts that led to the identification of ML382, 1 (Figure 1) started with a highthroughput screen of the [NIH Mol](#page-1-0)ecular Library Small-Molecule Repository (MLSMR) compound collection that contained >300 000 compounds utilizing a triple addition

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Figure 1. Initial SAR around ML382, 1 and the current work.

protocol.⁷ The screen identified a 2-(sulfonamido)-N-phenylbenzami[de](#page-6-0) scaffold that showed promise as a starting point for the initial SAR campaign. Our efforts centered around the cyclopropylsulfonamide (blue) and the 2-ethoxy (orange) moieties, and from that ML382 was identified (Figure 1). Although this molecule proved to be a beneficial to[ol](#page-6-0) compound, it did suffer from notable deficiencies. Namely, the compound did not possess suitable pharmacokinetic properties (high clearance due to oxidation of both phenyl rings) and, thus, was limited to intrathecal (i.t.) administration, and the overall MrgX1 potency was less than desired.⁵ Therefore, we have been working on the discovery an[d](#page-6-0) characterization of additional MrgX1 PAMs with improvement in these areas. The current work concentrated on the phenyl ring systems to minimize the oxidative metabolism, ring closing of the sulfonamide portion to five-membered ring systems as well as amide bioisosteres. The work culminated in the discovery of 8e, which enhanced the potency by 8-fold.

The synthesis of the compounds evaluated in this work is outlined in Scheme 1. The compounds that contain the main core 2-(sulfonamido)-N-benzamide were all synthesized via a common route. Namely, the anthranilic acid, 2, was coupled with an appropriate sulfonyl chloride under basic water

Scheme 1. Synthesis of MrgX1 PAMs

^aReagents and conditions: (a) NaHCO₃, H₂O, rt, 12 h; 25–80% (b) T3P, Et₃N, CH₂Cl₂, rt, 6 h; 65% (c) HATU, iPr₂NEt, THF, rt, 16 h; (d) AcOH, 80 °C, 16 h; 93% 2 steps (e) K_2CO_3 , BuOH, 150 °C; (f) Et₃N, CH₂Cl₂; 7% 2 steps (g) POCl₃, 25%.

conditions (NaHCO₃, H₂O) to yield the sulfonamide, 3. This compound was then coupled with an aniline (or amine) using T3P to yield the final compounds, 4−8. 8,9 This synthetic scheme was modular in that the amide could [be](#page-6-0) made first and then the sulfonamide as the final target. Next, the amide bioisostere compounds were synthesized individually as outlined. The benzimidazole was synthesized by the coupling of the sulfonamide benzoic acid, 3, with 1,2-diaminophenyl via a two-step protocol. First, the amide was formed using HATU followed by ring closure under heat and acidic conditions to yield $4k.^{10}$ The triazole was synthesized by reacting the hydrazin[e](#page-6-0) carbonyl, 10, with the cyano compound, 11, $(K_2CO_3, BuOH, 150 °C)$ to yield the desired product, 41.¹¹ Finally, the oxadiazole was synthesized via a two-step proto[col](#page-6-0) as outlined. The hydrazine carbonyl, 13, was coupled to the acid chloride, 12 to yield the penultimate intermediate, 14, which is then cyclized with $POCl₃$ to yield the oxadiazole, 4m.

We envisioned closing the carbonyl of the amide with the sulfonamide to form a five-membered ring system as our first attempt to scaffold hop within this series. From our previous work, we knew the NH of the sulfonamide was not critical as alkylation of this nitrogen was tolerated. Thus, we started with commercially available indazole or aza-indazole derivatives $(4a-c)$ (Table 1). We test the efficacy of the new compounds against [MrgX1](#page-2-0) stably expressing HEK293 cell line. The enhanced activation of MrgX1 by the compounds in the presence of agonist BAM8-22 is monitored by a Ca^{2+} imaging assay as described before.⁷ Unfortunately, these compounds did not show any activity [ag](#page-6-0)ainst MrgX1. In fact, nearly all of our efforts to modify the benzamide phenyl ring led to inactive compounds. These included pyridine moieties (4d,e), expanding with a naphthyl moiety (4f,g) as well as changing the 1,2-relationship of the sulfonamide and the amide portion of the molecule (4h,i).

Moving to the amide portion of the molecule also did not provide any beneficial compounds. Attempts to alkylate the nitrogen (4j) or insert amide isosteres, such as benzimidazole $(4k)$, triazole $(4l)$, and oxadiazole $(4m)$, were all inactive. Other efforts to remove the aryl group on the right-hand side $(4n)$, addition of an additional methylene linker $(4o)$, or a pyridine moiety (4p) were all unproductive changes to the core scaffold.

Having evaluated making core changes to the phenyl ring systems, we next moved to the ether moiety for modification (Table 2). First, the ethyl ether was investigated by introducing [branchin](#page-2-0)g, elongating the chain, and addition of a trifluoromethyl group. Introduction of a branched group produced active compounds; however, these lost ∼25-fold activity compared to the ethyl group (5a, $EC_{50} = 2.83 \mu M$; 5b, EC_{50} $= 3.14 \mu M$). Adding an additional methylene group (propyl, 5c, $EC_{50} = 0.506 \mu M$) or the 2,2,2-trifluoroethyl group (5d, $EC_{50} = 0.502 \mu M$) led to active compounds, although not as potent as 1. Moving to the methoxyethyl (5e) lost all activity, but the cyclopropylmethyl group brought some activity back to the molecule (5f, $EC_{50} = 1.36 \mu M$). Interestingly, the 2,4diethoxy analog, 5g, retained some potency compared with 1 $(EC_{50} = 1.09 \mu M)$; however, the 4-methoxy analog, 5h, was inactive. Next, we moved to evaluate cyclized versions on the right-hand side of the molecule. The first analogs were cyclized via the amide nitrogen to impart rigidity into the molecule; however, these compounds, 5i−j, were not active, which is consistent with the methylated amide analog, 4j. Additional attempts to cyclize the ether analogs were also not productive

Table 1. Modification of the Benzamide Core Scaffold

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Cmpd	Structure	E_{max} $(\%)^a$	MrgX1 EC_{50} $(\mu M)^a$
$\mathbf 1$	ĭ 'n,	100	0.124
4a	o .k=0 D Ŋ	-5.7 ± 4.1	ND
4b	$\frac{0}{s} = 0$ D N.	1.9 Ŧ 16.6	ND
4c	$rac{0}{-s} = 0$ Ŋ $\mathbb{R}^{\mathbb{N}}$	-11.7 \pm 8.7	ND
4d	Ó, i	12.1 \pm 4.0	ND
4e	ಀೢಁ й I	0.1 \pm 10.9	ND
$4f$	$\overline{\mathsf{o}}_{\mathsf{e}^{\mathsf{o}}}$ N	29.7 \pm 18.3	ND
4g	o i N	8.9 \pm 13.1	ND
4h	\circ_{s} c NH N	70.6 \pm 56.1	ND
4i	०ू० N Ĭ	-44.4 \pm 77.3	ND
4j	ŭ Ņ	-11.9 ± 0.6	ND
4k	$O_{\phi_{\rm{max}}}$ O NH N	23.8 \pm 8.9	ND
41	$O_{\mathcal{O}_{\mathcal{Q}}^{\mathcal{O}}}$	7.5 $_{\pm}$ 7.4	ND
4m	$\frac{1}{s^{(n)}_{\text{N}}+s^{(n)}_{\text{N}}}$ ▽ \sim	6.7 \pm 4.6	ND
4n	ō, ŭ N	24.2 \pm 13.5	ND
4 _o	ŅH N	12.7 ± 13.6	$_{\rm ND}$
4p	ō. S ŅH Ũ	1.6 Ŧ 5.2	N _D

 a Assays we[re](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [carried](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [out](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [in](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [the](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [presence](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [of](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [10](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [nM](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf) [BAM8-2](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl1&ref=pdf)2; data are the mean \pm SEM of $n = 2$ experiments; E_{max} values are normalized to the control compound ML382; ND, not determined, as E_{max} is too low.

changes, 5k−q, with the lone exception of the 2-methyl-7 benzofuran analog $5r$ (EC₅₀ = 1.04 μ M).

Having established the SAR around the benzamide and the ether moieties of the molecule, we next turned our attention to

Cmpd	Structure	Emax (%)	MrgX1 EC _{sc} (μM)
5a		83.1 ± 22.8	2.83 ± 0.56
5b		80.1 ± 2.9	3.14 ± 1.98
5c		80.8 ± 13.3	0.506 \pm 0.029
5d	N CF ₃	95.0 ± 3.9	0.502 \pm 0.045
5e	N	20.7 ± 6.9	ND
5f		61.5 ± 24.3	1.36 ± 0.17
5g	N	73.2 ± 5.6	1.09 ± 0.94
5h		-15.6 ± 4.1	ND
5i		6.8 ± 6.7	ND
5j		2.7 ± 12.2	ND
5k	N	37.9 ± 15.3	9.01 ± 1.65
51	Ν	2.1 ± 13.9	ND
5m	N	27.0 ± 7.2	ND
5n	ĭ Ĥ °, p	7.4 ± 10.9	N _D
5 _o	\circ_{s}	-25.6 ± 5.0	ND
$_{\rm Sp}$		14.6 ± 13.8	ND
5q	O.	53.4 ± 14.6	14.2 ± 4.8
5r		66.3 ± 20.1	1.04 ± 0.30

a
Assays were carried out in the presence of 10 nM BAM8-22; data are the mean \pm SEM of $n = 2$ experiments; E_{max} values are normalized to the control compound ML382; ND, not determined, as E_{max} is too low.

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substitutions on the phenyl rings. From the previous SAR, the substitutions chosen were single methyl or halogen groups, as it was evident that larger modifications were not tolerated. For these studies we used both the cyclopropylsulfonamide and the ethylsulfonamide as it was shown in our previous work that these two were well tolerated. In addition, we used the ethyl and the 2,2,2-trifluoroethyl as these were also active moieties. The right-hand selections were also based on those that are commercially available as the corresponding anilines. The compounds tested for this portion of the SAR are shown in Table 3. The 4-fluoro derivatives, 6a,b, were not as active as the parent compounds; however, the 5-fluoro derivatives, 6c,d, were very potent (6c, $EC_{50} = 0.014 \mu M$; 6d, $EC_{50} = 0.285$ μ M).

Table 3. Right-Hand Halogen SAR^a

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$$

Cmpd	Structure	Emax(%)	MrgX1 EC_{50} (μM)
6a	$O_{\psi_{\rm S}} O$ NH O N	-10.5 ± 8.9	ND
6b	$O_{\mathcal{O}_{\mathcal{C}_\mathcal{A}}}\mathcal{O}$ NH. ဂူ `N $C F_3$	91.4 ± 4.9	1.40 ± 0.15
6с	$O_{\sqrt{2}}$ 0 NH O `N	82.2 ± 21.9	0.014 ± 0.01
6d	$O_{\mathcal{P}_{\mathbf{Q}}^{\mathcal{P}}}$ O NH O N	74.0 ± 17.2	0.285 ± 0.020

 a [Assays](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [were](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [carried](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [out](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [in](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [the](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [presence](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [of](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [10](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [nM](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [BAM8-22;](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [data](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf) [ar](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl3&ref=pdf)e the mean \pm SEM of $n = 2$ experiments; E_{max} values are normalized to the control compound ML382; ND, not determined, as E_{max} is too low.

The left-hand portion of the molecule afforded more analogs as there were more commercially available starting materials to utilize (Table 4). In general, the 5-substitued analogs produced potent MrgX1 PAMs (7a−l), many that were more potent than 1, and the ethoxy analogs were more potent than the 2,2,2-trifluoromethyl analogs. Some of the highlighted compounds were 5-methyl, 7a ($EC_{50} = 0.103 \mu M$), 7g (EC_{50} = 0.098 μ M), and 7i (EC₅₀ = 0.054 μ M), although nearly all of the compounds showed potency <1 μ M. The 4-substituted analogs (Me, F, or Cl) were significantly less potent and in most cases were inactive altogether.

Having established the optimal substitution patterns on the individual phenyl ring systems, we next investigated the matching of the best pairs from the above study. As we only had one active right-hand substitution pattern, this was kept constant, and we varied the left-hand portion. All of the synthesized analogs were active, and the matched pairs were equipotent with the other analogs. For example, the 5-chloro, 8a (EC₅₀ = 0.055 μ M), 5-methyl, 8c (EC₅₀ = 0.069 μ M), and 5-fluoro, 8e ($EC_{50} = 0.013 \mu M$) were all of similar activity (as were the corresponding ethylsulfonamide analogs). These analogs emphasize the steric environment for the benzamide

Table 4. Analysis of the Left-Hand Phenyl Ring System^a

		∽	
Cmpd	Structure	Emax (%)	MrgX1 EC ₅₀ (μM)
7а	™ H	104.0 ± 8.9	0.103 ± 0.027
7Ь		89.9 ± 4.0	0.369 ± 0.003
7с	ĭ N	108.5 ± 2.1	0.217 ± 0.055
7d	\sim s $\frac{1}{2}$ ŅΗ î	58.4 ± 6.0	ND
7e		102.1 ± 3.8	0.173 ± 0.035
7f	Î	59.8 ± 5.9	ND
7g	O,	106.7 ± 3.3	0.098 ± 0.028
7 _h	Î N	100.4 ± 2.6	0.435 ± 0.082
7i		110.0 ± 3.4	0.054 ± 0.012
7j		103.4 ± 9.1	0.915 ± 0.222
7k	Ō, ူ	89.2 ± 6.5	0.407 ± 0.051
71	N	89.5 ± 4.6	0.366 ± 0.086
7m		67.6 ± 10.5	0.860 ± 0.262
7n	$O_{\mathcal{S}}$ ^O ŅH CF ₃	29.1 ± 10.5	ND
70	Ó .O CI ⁻	36.1 ± 14.1	ND
7p	$O_{\mathcal{S}^{\prime}}$ NH ů	3.2 ± 20.7	ND

 a Assa[ys](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [were](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [carried](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [out](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [in](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [the](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [presence](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [of](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [10](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [nM](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [BAM8-22;](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf) [da](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl4&ref=pdf)ta are the mean \pm SEM of $n = 2$ experiments; E_{max} values are normalized to the control compound ML382; ND, not determined, as E_{max} is too low.

portion of the molecule as other, larger groups (naphthyl and indazole, Table 1) were all inactive, whereas the smaller substituen[ts were](#page-2-0) tolerated. In addition to steric considerations, these analogs also show that electronic factors are not as

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important (electron donating vs electron withdrawing). Overall, the SAR effort around this scaffold showed little tolerance for major scaffold changes, or even smaller changes around the ether substituent (Table 5). This work highlights many of the known challenges in the allosteric modulator field, namely the steep and flat SAR associated with allosteric modulator drug discovery[.12](#page-6-0)[−][14](#page-6-0)

			Table 5. SAR of the Matched Pairs of Phenyl Substituents ^a

 $\begin{picture}(180,170) \put(0,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150}} \put(10,0){\line(1,0){150$

Cmpd	Structure	Emax(%)	MrgX1 EC ₅₀ (μM)
8a	$\overline{\mathcal{S}}_{\mathcal{S}_{\text{sym}}^{\text{out}}}^{\text{out}}$ N	103.5 ± 3.6	0.055 ± 0.027
8b	$\overline{\mathcal{S}^{\circ}_{\mathcal{S}_{\mathcal{M}^{\mathcal{H}}}}\mathcal{A}}$ N ĊI.	118.4 ± 21.9	0.206 ± 0.019
8c	$\frac{1}{\sqrt{2}}$ $\frac{1}{\sqrt{1}}$ И	105.6 ± 3.7	0.069 ± 0.024
8d	$\overline{\mathcal{S}}_{\mathcal{S}_{\text{vMH}}^{\text{op}}}$ N	127.6 ± 26.8	0.080 ± 0.031
8e	\sim $\frac{1}{2}$ $\sum_{i=1}^{N+1}$ N	126.8 ± 25.2	0.013 ± 0.001
8f	$\overline{\mathcal{S}}_{\mathcal{S}_{\mathcal{M}^{\mathsf{H}}}^{\mathcal{O}}}^{\mathcal{O}}$ N	136.3 ± 26.8	0.031 ± 0.009

 a [Assays](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [were](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [carried](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [out](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [in](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [the](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [presence](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [of](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [10](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [nM](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [BAM8-22;](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [data](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf) [ar](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?fig=tbl5&ref=pdf)e the mean \pm SEM of $n = 2$ experiments; E_{max} values are normalized to the control compound ML382; ND, not determined, as E_{max} is too low.

Having identified several compounds that showed EC_{50} values of $\langle 1 \mu M \rangle$, we next profiled select compounds in *in* vitro drug metabolism and pharmacokinetics (DMPK) assays to assess their human and mouse liver microsomal intrinsic clearance and plasma protein binding (Table 6).^{[15,16](#page-6-0)} All the compounds evaluated displayed high intrinsic clearance and were predicted to have high hepatic clearance in vivo (CL_{HEP}) 75% Q_H), with a few notable exceptions (5d, 7e, 7g). From a previous metabolite identification study (unpublished results), we anticipated that the major metabolite on 1, was oxidation of the left-hand phenyl ring. Thus, it was disappointing that the compounds that incorporated a halogen substituent on that ring did not translate to improved clearance. Although, the metabolism could shift from oxidation of the phenyl ring being the major metabolite to potentially O-dealkylation in the new compounds. Using equilibrium dialysis, the plasma protein binding of the selected compounds was determined in human and mouse plasma. The results revealed that all the tested compounds were highly protein bound in human plasma ($\%f_u$ < 0.5). However, the compounds showed a better plasma

Table 6. MrgX2 Selectivity and In Vitro DMPK Properties of Selected Compounds

		Intrinsic Clearance $(mL/min/kg)^c$	Plasma Protein Binding $(\%f_n)^b$			
Cmpd	hCL_{INT}	hCL_{HEP}^a	mCL_{INT}	mCL_{HEP}^a	Human	Mouse
$\mathbf{1}$	61.9	15.2	1065	83.1	0.4	1.7
5c	115.6	17.1	4888.6	88.5	< 0.3	1.5
5d	40.2	13.4	979.1	82.5	0.4	2.2
5r	74.9	15.8	1457.3	84.9	< 0.3	2.6
6b	80.0	16.1	1210.7	83.9	0.5	2.1
6с	79.1	16.0	1659	85.5	0.4	2.2
7a	75.1	15.9	410.2	73.9	0.3	4.3
7e	45.4	13.9	383.2	72.9	0.4	4.1
7g	36.0	12.9	269.0	67.5	0.2	4.1
7i	129.2	17.4	1470.4	85.0	0.1	0.1
8a	138.4	17.6	>2400	>86.8	0.1	0.3
8с	114.3	17.1	1101.6	83.3	0.3	1.2
8d	69.3	15.6	1004.3	82.7	0.3	1.4
8e	59.4	15.0	979.5	82.5	0.2	1.1
8f	69.1	15.6	1235.7	84.0	0.2	0.1

a
Predicted hepatic clearance based on intrinsic clearance in human and mouse liver microsomes using the well-stirred organ CL model (binding terms excluded). $b_{\%f_u}$ = percent fraction unbound. *CIn vitro* DMPK studies were performed at Q2 Solutions, Indianapolis, IN.

protein binding profile in mouse plasma with several compounds displaying moderate free fraction (>2%). We profiled select compounds for selectivity against the closely related MrgX2 receptor. All the compounds tested were selective against MrgX2 (Supplemental Table 1), except for 7i. In addition, we tested [select compounds \(](https://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.2c00100/suppl_file/ml2c00100_si_001.pdf)6c, 8c, and 8e) against the Psychoactive Drug Screening Panel at the University of North Carolina, Chapel Hill.¹⁷ This panel consists of 45 receptors and transporters that ar[e o](#page-6-0)f importance in the CNS. All three compounds were active against PBR (peripheral benzodiazepine receptor) (K_i , $6c = 271$ nM, $8c =$ 28 nM, $8e = 71$ nM), although this receptor does not cause concern for the use of these compounds as tools. In addition, 6c was active against D5 and H1 (>4000 nM), and 8c was active against H1 (1089 nM), showing these compounds are generally selective for MrgX1 (Supplemental Table 2).

We evaluated a smaller set [of the most potent com](https://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.2c00100/suppl_file/ml2c00100_si_001.pdf)pounds (6c, 8c, 8e) in a rat IV cassette study to assess their ability to cross the blood−brain barrier and evaluate their in vivo clearance $(Table 7).^{18,19}$ $(Table 7).^{18,19}$ $(Table 7).^{18,19}$ The cassette study was done in a 5-in-1 format [where](#page-5-0) five compounds are dosed in a single cassette (IV, 0.25 mg/kg) and then evaluated for their plasma and brain concentrations. These compounds had varying levels of brain penetrance with 6c being the best ($K_p = 0.57$). However, 8e, had the better overall plasma profile with low clearance (CL = 14.1 mL/min/kg) and better overall plasma exposure ($AUC =$ 304 h \cdot ng/mL). Finally, we further profiled 6c and 8e in a discrete mouse PK study as this would be the species for the in *vivo* pain assay. The compounds were dosed IP (2 mg/kg) and the results are shown in Table 7. As can be seen, the discrete mouse results are similar [to the ra](#page-5-0)t cassette in terms of K_p , and 8e has better overall exposure.

In summary, we have identified additional 2-sulfonamidebenzamides as selective positive allosteric modulators of MrgX1. This work is an extension of our previous report, and the compounds identified represent an ∼8-fold increase in potency. The SAR highlighted the steep and flat SAR around

Table 7. In Vivo Rat Cassette and Discrete Mouse PK

a In vivo DMPK studies were performed at Pharmaron Laboratories, Louisville, KY. Cassette dosing (5-in-1) dose of 0.25 mg/kg (IV) of each test article to male Sprague−Dawley rats. Formulation: DMSO:PEG400:EtOH:Saline (5:48:10:37). Plasma blood sampling at 0.0833, 0.25, 0.5, 1, 3, 4, 8, and 24 h and Day 2 0.25 h post dose. Brain sample on Day 2 at 0.25 h. b K_p = total brain/plasma ratio.

Exerculation: 10% DMSO/10% cremptor EI/30% PEC4400/50% Formulation: 10% DMSO/10% cremphor EL/30% PEG400/50% water.

this scaffold, a feature that is well-documented in the allosteric modulator discovery field. Compounds were also profiled in our Tier 1 DMPK assays and possess favorable plasma protein binding in mice; however, the compounds displayed high intrinsic clearance in both species tested. Selected compounds were also profiled in in vivo PK studies, and they displayed moderate brain penetration; however, 8e displayed improved in vivo clearance and plasma concentrations. Because of the limitations in the PK properties, these newly identified compounds would be limited to nonoral dosage regimens (e.g., intraperitoneal, subcutaneous, or intrathecal); however, they do represent a major improvement in potency compared to the previously known MrgX1 PAMs.

■ ASSOCIATED CONTENT

³ Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100.

[Experimental](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?goto=supporting-info) [procedures](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?goto=supporting-info) [and](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?goto=supporting-info) [characterization](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?goto=supporting-info) [for](https://pubs.acs.org/doi/10.1021/acsmedchemlett.2c00100?goto=supporting-info) all final compounds, description of in vitro studies, and selectivity tables [\(PDF](https://pubs.acs.org/doi/suppl/10.1021/acsmedchemlett.2c00100/suppl_file/ml2c00100_si_001.pdf))

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Author Contributions

C.R.H. and X.D. drafted and corrected the paper with input from all authors. S.S., A.K.V., C.D.A., A.A.J., and A.I.W. performed the chemical synthesis, and C.R.H. oversaw the medicinal chemistry and target selection and interpreted the biological data. Q.P. performed the cellular assays, and X.D. oversaw the experiments. C.R.H. oversaw the PK experiments.

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Notes

The authors declare no competing financial interest.

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■ ABBREVIATIONS

MrgX1, Mas-related G-protein coupled receptor member X1; HATU, 1-[Bis(dimethylamino)methylene]-1H-1,2,3-triazolo- $[4,5-b]$ pyridinium 3-oxid hexafluorophosphate, N- $[$ (Dimethylamino)-1H-1,2,3-triazolo-[4,5-b]pyridin-1-ylmethylene]-Nmethylmethanaminium hexafluorophosphate N-oxide; T3P, propylphosphonic anhydride; HEK293, human embryonic kidney 293 cells; EC_{50} , half-maximal effective concentration

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