



Published in final edited form as:

Angew Chem Int Ed Engl. 2022 May 23; 61(22): e202203413. doi:10.1002/anie.202203413.

## [FeFe]-Hydrogenase: Defined Lysate-Free Maturation Reveals a Key Role for Lipoyl-H-Protein in DTMA Ligand Biosynthesis

Adrien Pagnier<sup>a</sup>, Batuhan Balci<sup>a</sup>, Eric M. Shepard<sup>a</sup>, Hao Yang<sup>b</sup>, Douglas M. Warui<sup>c</sup>, Stella Impano<sup>a</sup>, Squire J. Booker<sup>c,d</sup>, Brian M. Hoffman<sup>b</sup>, William E. Broderick<sup>a</sup>, Joan B. Broderick<sup>a</sup>

<sup>[a]</sup>Department of Chemistry & Biochemistry, Montana State University, Bozeman, MT 59717

<sup>[b]</sup>Department of Chemistry, Northwestern University, Evanston, IL

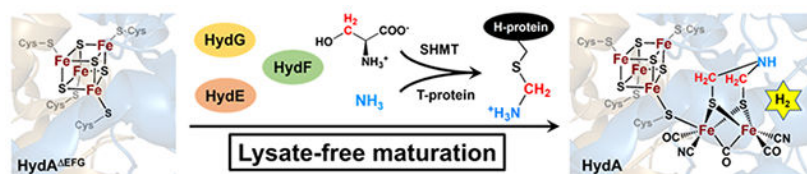
<sup>[c]</sup>Department of Chemistry, The Pennsylvania State University, University Park, PA. 16802

<sup>[d]</sup>Howard Hughes Medical Institute

### Abstract

Maturation of [FeFe]-hydrogenase (HydA) involves synthesis of a CO, CN<sup>-</sup>, and dithiomethylamine (DTMA)-coordinated 2Fe subcluster that is inserted into HydA to make the active hydrogenase. This process requires three maturation enzymes: the radical S-adenosyl-L-methionine (SAM) enzymes HydE and HydG, and the GTPase HydF. *In vitro* maturation with purified maturation enzymes has been possible only when clarified cell lysate was added, with the lysate presumably providing essential components for DTMA synthesis and delivery. Here we report maturation of [FeFe]-hydrogenase using a fully defined system that includes components of the glycine cleavage system (GCS), but no cell lysate. Our results reveal for the first time an essential role for the aminomethyl-lipoyl-H-protein of the GCS in hydrogenase maturation and the synthesis of the DTMA ligand of the H-cluster. In addition, we show that ammonia is the source of the bridgehead nitrogen of DTMA.

### Graphical Abstract



The aminomethyl-lipoyl-H-protein ( $H_{met}$ ) of the glycine cleavage system (GCS) is shown to play a key role in maturation of the [FeFe]-hydrogenase. Including  $H_{met}$ , or  $H_{met}$  plus other components of the GCS, allows maturation by purified HydE, HydF, and HydG in the absence of cell lysate for the first time.

## Keywords

hydrogenase maturation; lipoyl-H-protein; biosynthesis; glycine cleavage system; dithiomethylamine

[FeFe]-hydrogenase (HydA) catalyzes the reversible reduction of protons to H<sub>2</sub> at a complex metal cluster, the H-cluster, composed of a [4Fe-4S] cluster bridged via a cysteine thiolate to a [2Fe] subcluster coordinated by CO, CN<sup>-</sup> and dithiomethylamine (DTMA) ligands (Figure 1).<sup>[1]</sup> [FeFe]-hydrogenases can catalyze rates of H<sub>2</sub> production that are approximately 100x higher than other hydrogenases, making HydA an appealing target for biohydrogen production technologies.<sup>[2]</sup> Early efforts to heterologously express [FeFe]-hydrogenase, however, led to inactive enzyme, a major obstacle to practical applications of this enzyme.<sup>[3]</sup> Subsequent studies showed that three maturation enzymes, HydE, HydF, and HydG, were required to produce an active [FeFe]-hydrogenase.<sup>[4]</sup>

HydF is a GTPase that binds redox-active iron-sulfur clusters and serves as a scaffold for assembly of a 2Fe precursor ([2Fe]<sub>F</sub>) delivered to HydA to yield active hydrogenase.<sup>[5]</sup> HydF can be loaded with [2Fe]<sub>F</sub> enzymatically by co-expression with HydE and HydG, yielding a [2Fe]<sub>F</sub> subcluster resembling the [2Fe] subcluster of the H-cluster ([2Fe]<sub>H</sub>).<sup>[6]</sup> In an alternative semi-synthetic approach, pre-formed synthetic [2Fe] clusters coordinated by CO, CN<sup>-</sup>, and DTMA can be used to load HydF, which then can activate HydA.<sup>[7]</sup> HydG synthesizes the CO and CN<sup>-</sup> ligands from tyrosine,<sup>[8]</sup> and also provides iron in the form of a [Fe(Cys)(CO)<sub>2</sub>(CN)] synthon.<sup>[9]</sup> HydE binds a synthetic analog of this synthon, and uses it as a substrate to catalyze an adenosylation reaction,<sup>[10]</sup> and may convert two of these adenosylated Fe(I) synthon units to a [2Fe] subcluster precursor.<sup>[11]</sup> The origin of the DTMA ligand is not well understood. It was thought to arise from HydE catalysis<sup>[12]</sup> but now appears to be formed independent of HydE.<sup>[11]</sup> Maturation of HydA<sup>EFG</sup> (HydA expressed in the absence of the maturase enzymes) can be achieved *in vitro* by combining extracts of *E. coli* cells overexpressing the individual maturases, or by using purified maturase enzymes together with *E. coli* cell lysate.<sup>[8d, 13]</sup> In work to date, the requirement for the *E. coli* lysate (20% to 100% v/v) has been absolute.<sup>[4c, 4d, 8d, 9a, 11, 13–14]</sup>

We report here development of a fully defined enzymatic system for maturation of HydA using HydE, HydF, and HydG, their substrates and reductants, and components of the glycine cleavage system (GCS, Scheme S1),<sup>[15]</sup> but without *E. coli* lysate. The results reveal a key role for the aminomethyl-lipoyl-H-protein (H<sub>met</sub>) as a precursor of the DTMA ligand of the H-cluster.

To better understand the chemistry of H-cluster maturation, we pursued fractionation of *E. coli* cell lysate to identify component(s) responsible for its requirement during maturation. These efforts revealed one or more components of a low-molecular weight (LMW) fraction (10 – 15 kDa, Figure S1) of lysate could also support maturation. Essential component(s) from *E. coli* lysate were inactivated by heat/acid treatment, suggesting they may be proteins (Figure S2). Shotgun proteomics revealed the presence of carrier proteins including the acyl-carrier protein (ACP) in the LMW fraction, suggesting the involvement of a small carrier protein in hydrogenase maturation, perhaps in the still poorly understood synthesis

of the DTMA ligand, as suggested by Juan Fontecilla-Camps (personal communication). However, no detectable HydA activity was observed following *in vitro* HydA maturation with holo- or lipoyl-ACP in place of *E. coli* cell lysate, and with purified HydE, HydF, HydG, SAM, and other small molecule components (see SI).

We next considered the H-protein of the GCS (Scheme S1), as a potential key component of HydA maturation. Lipoylated H-protein acts as a carrier between protein components of the GCS (Scheme S1), and can be in the disulfide form ( $H_{ox}$ ), the reduced form ( $H_{red}$ ), or the aminomethyl form ( $H_{met}$ ). Maturation reactions carried out with  $H_{red}$  in the absence of cell lysate provided no detectable hydrogenase activity (Figure 2).  $H_{met}$  was synthesized enzymatically from  $H_{ox}$  using the P-protein, PLP, and glycine (Scheme S1), purified, and verified via mass spectrometry (Figure S3). HydA maturation with  $H_{met}$  added to our defined system resulted in active HydA, as evidenced by  $H_2$  gas production (Figure 2). Increasing the amount of  $H_{met}$  provided a significant increase in  $H_2$  (Figure 2), demonstrating that HydA activation is dependent on  $H_{met}$ . To our knowledge, this is the first report of biological HydA<sup>EFG</sup> conversion to active hydrogenase in the absence of cell lysate, and points to  $H_{met}$  as a previously unrecognized essential component.

In these experiments, the level of activation of HydA<sup>EFG</sup> in the presence of  $H_{met}$  is nonetheless low, requiring 24 h incubation to detect small amounts of  $H_2$ . The low hydrogenase activity may be due to instability of  $H_{met}$ , which is known to readily hydrolyze, especially in the presence of partner proteins.<sup>[16]</sup> We reasoned that if this is so, regeneration of  $H_{met}$  *in situ* would improve the hydrogenase maturation. Regeneration of  $H_{met}$  from  $H_{red}$  *in situ* under the reducing conditions required for maturation was accomplished by including aminomethyltransferase (T-protein), serine hydroxymethyltransferase (SHMT), serine, and  $NH_4^+$  (Scheme 1) in the maturation mixture. HydA maturation under these conditions yielded dramatically improved hydrogenase activity, with further refinement as described below providing specific activities as high as 390  $\mu\text{mol}/\text{min}/\text{mg}$  HydA (Figure 3); in comparison, holo-HydA is reported to have a specific activity of  $\sim 700$   $\mu\text{mol}/\text{min}/\text{mg}$ .<sup>[17]</sup>

To further probe the involvement of  $H_{met}$  of the GCS during HydA maturation, we examined the dependence of maturation on concentrations of the  $H_{met}$  regeneration components (Figure 3). Maturation is dependent on both serine and  $NH_4^+$ , with high concentrations required for optimal activity, consistent with the equilibria outlined in Scheme 1. Glycine has an inverse effect on maturation (Figure 3), consistent with the role of SHMT in producing methylene-tetrahydrofolate ( $CH_2$ -THF, Scheme 1). The maturation observed without added  $NH_4^+$  is attributed to co-purification of  $NH_4^+$  with T-protein, consistent with the well-defined  $NH_4^+$  binding site in this protein.<sup>[18]</sup> THF also impacts maturation, increasing activity at low concentrations followed by a rapid decrease with increasing THF concentration (Figures 3, S5), consistent with THF involvement in both SHMT and T-protein equilibria; the dominant effect of THF is on the T-protein equilibrium, driving it towards  $H_{red}$  and away from  $H_{met}$  (Scheme 1). Maturation works without added THF because THF co-purifies with the T-protein. The dependence of HydA maturation on these components required for generation of  $H_{met}$  *in situ* further support a key role for  $H_{met}$  in HydA maturation; these components were provided by lysate in prior lysate-dependent maturation.<sup>[4c, 4d, 8d, 9a, 13b, 13c, 14]</sup>

Our results suggest that the GCS, and specifically  $H_{\text{met}}$ , provides a precursor for DTMA biosynthesis. We propose that the aminomethyl group of  $H_{\text{met}}$  is the source of the C and N of the C-N-C framework of DTMA. Given that the aminomethyl of  $H_{\text{met}}$  is derived from the  $\beta$ -C of serine and  $\text{NH}_4^+$  (Scheme 1), we set out to probe incorporation of isotopic labels from these sources into the DTMA of matured HydA. Maturation reactions were carried out in the absence of lysate (see SI), with  $[^{13}\text{C}_3, ^{15}\text{N}]$ -serine or  $^{15}\text{NH}_4\text{Cl}$  included in place of the unlabeled component. HydA was purified from the maturation mixture and oxidized with thionin to give an electron paramagnetic resonance (EPR) spectrum typical of the  $H_{\text{ox}}$  state of holo-HydA (Figure S6). Electron nuclear double resonance (ENDOR) spectroscopy was used to look for incorporation of isotopic labels into the DTMA of the H-cluster. For the  $[^{13}\text{C}_3, ^{15}\text{N}]$ -serine samples, the spectrum collected at  $\sim g_1$  shows two distinct  $^{13}\text{C}$  hyperfine signals (Figure 4, red) associated with the two carbons of the DTMA ligand, as reported previously for HydA matured in the presence of lysate.<sup>[14b]</sup> The presence of signals with different hyperfine splittings for the two DTMA carbons is consistent with the prior work demonstrating their origins from serine and asymmetry in the DTMA ligand environment in holo-HydA.<sup>[14b, 19]</sup> However, this sample prepared without lysate shows no  $^{15}\text{N}$  ENDOR response, indicating that  $^{15}\text{N}$  from  $[^{13}\text{C}_3, ^{15}\text{N}]$ -serine is not present in the DTMA of HydA matured lysate-free (Figure 4, black).

In our defined maturation, the T-protein uses  $\text{NH}_4^+$  to provide the N for the aminomethyl group of  $H_{\text{met}}$  (Scheme 1), and thus  $\text{NH}_4^+$  should be the source of the N of DTMA.  $^{15}\text{N}$  ENDOR spectra of HydA matured in the presence of  $^{15}\text{NH}_4\text{Cl}$  reveals this to be true, as the corresponding  $^{15}\text{N}$  coupling of 1.5 MHz is observed (Figure 4, blue). Prior work showed that both N and C of DTMA could be derived from serine, but this was in the presence of *E. coli* cell lysate (72% v/v).<sup>[14b]</sup> Given that cell lysate contains enzymes such as serine dehydratase that can liberate  $\text{NH}_4^+$  from serine, it is likely that the observed incorporation of  $^{15}\text{N}$  from serine in the prior work was a result of  $^{15}\text{NH}_4^+$  liberation due to the presence of lysate.

We demonstrate the first maturation of HydA using a defined enzymatic system, thereby revealing an essential role for  $H_{\text{met}}$  and the GCS in building the DTMA ligand of the H-cluster.  $H_{\text{met}}$  alone, when added to lysate-free maturation components, is sufficient to provide active HydA (Figure 2), revealing the key role for  $H_{\text{met}}$  during H-cluster biosynthesis. Inclusion of an  $H_{\text{met}}$  regeneration system (T-protein, SHMT, serine, and  $\text{NH}_4^+$ ) affords high levels of HydA activation, with the roles of T-protein and SHMT highlighted by the dependence of activation on the concentrations of key species (Figure 3).  $^{13}\text{C}/^{15}\text{N}$ -ENDOR spectroscopy of HydA matured using this defined system confirms that the carbon atoms of DTMA originate from serine, while the nitrogen comes from  $\text{NH}_4^+$  (Figure 4), consistent with our previous prediction<sup>[20]</sup> and with the known chemistry of the GCS during  $H_{\text{met}}$  formation from  $H_{\text{red}}$  (Scheme 1).

The recent report that HydA can be matured in the presence of HydF, a synthetic  $[\text{Fe}_2(\text{SH})_2(\text{CO})_4(\text{CN})_2]^{2-}$  complex, and cell lysate,<sup>[11]</sup> together with the results presented here, lead us to propose that  $H_{\text{met}}$  interacts with the  $\text{HydF}-[\text{Fe}_2(\text{SH})_2(\text{CO})_4(\text{CN})_2]^{2-}$  complex to synthesize the dimethylamine backbone of DTMA. In this model, the aminomethyl group of  $H_{\text{met}}$  is functionally equivalent to the formaldehyde and ammonia

used to synthesize the C-N-C DTMA backbone in the  $[2\text{Fe}]_{\text{H}}$  model reported by Li and Rauchfuss.<sup>[21]</sup> We propose that during biological maturation, each sulfur of the  $[2\text{Fe}]_{\text{F}}$  precursor cluster  $[\text{Fe}_2(\text{SH})_2(\text{CO})_4(\text{CN})_2]^{2-}$  nucleophilically attacks the C of an aminomethyl group of  $\text{H}_{\text{met}}$ , resulting in transfer of two aminomethyl groups from two  $\text{H}_{\text{met}}$  to the bridging sulfides of the HydF-bound  $[2\text{Fe}]_{\text{F}}$  precursor cluster. Condensation of the aminomethyl moieties with loss of  $\text{NH}_4^+$  would then form DTMA (Scheme 2). In this model, the sulfurs of DTMA are derived from the precursor diiron cluster, not from the lipoyl group of Hmet; use of a seleno-lipoyl-H-protein during maturation could test this hypothesis.

## Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

## Acknowledgements

This work was funded by the U.S. DOE (DE-SC0005404 to J.B.B. and E.M.S.). S.J.B acknowledges funding from the NSF (MCB-1716686), and B.M.H from the NIH GM 111097. The Mass Spectrometry Facility was in part funded by the MJ Murdock Charitable Trust and the NIH (P20GM103474 and S10OD28650). The authors thank Prof. Inger Andersson (Uppsala U.) for the gift of pBAD-HisA-slr0293, and Juan Fontecilla-Camps, Roman Rohac, and Yvain Nicolet for helpful discussions.

## References

- [1]. a) Lubitz W, Ogata H, Rudiger O, Reijerse E, Chem. Rev 2014, 114, 4081–4148; [PubMed: 24655035] b) Schilter D, Camara JM, Huynh MT, Hammes-Schiffer S, Rauchfuss TB, Chem. Rev 2016, 116, 8693–8749. [PubMed: 27353631]
- [2]. a) Mészáros LS, Németh B, Esmieu C, Ceccaldi P, Berggren G, Angew. Chem 2018, 130, 2626–2629; b) Rodriguez-Maciá P, Galle LM, Björnsson R, Lorent C, Zebger I, Yoda Y, Cramer SP, DeBeer S, Span I, Birrell JA, Angew. Chem. Int. Ed 2020, 59, 16786–16794; c) Schuchmann K, Chowdhury NP, Müller V, Front. Microbiol 2018, 9, 2911; [PubMed: 30564206] d) Wegelius A, Khanna N, Esmieu C, Barone GD, Pinto F, Tamagnini P, Berggren G, Lindblad P, Energy Environ. Sci 2018, 11, 3163–3167; [PubMed: 30555530] e) Land H, Ceccaldi P, Mészáros LS, Lorenzi M, Redman HJ, Senger M, Stripp ST, Berggren G, Chem. Sci 2019, 10, 9941–9948; [PubMed: 32055351] f) Li T, Jiang Q, Huang J, Aitchison CM, Huang F, Yang M, Dykes GF, He H-L, Wang Q, Sprick RS, Cooper AI, Liu L-N, Nat. Commun 2020, 11, 5448; [PubMed: 33116131] g) Birrell JA, Rudiger O, Reijerse EJ, Lubitz W, Joule 2017, 1, 61–76; h) Esmieu C, Raleiras P, Berggren G, Sustain. Energ. Fuels 2018, 2, 724–750; i) Reeve HA, Ash PA, Park H, Huang AL, Posidias M, Tomlinson C, Lenz O, Vincent KA, Biochem. J 2017, 474, 215–230. [PubMed: 28062838]
- [3]. a) Voordouw G, Hagen WR, Kruse-Wolters KM, van Berkel-Arts A, Veeger C, European journal of biochemistry / FEBS 1987, 162, 31–36; b) Atta M, Meyer J, Biochim. Biophys. Acta - Prot. Struct 2000, 1476, 368–371.
- [4]. a) Posewitz MC, King PW, Smolinski SL, Zhang L, Seibert M, Ghirardi ML, J. Biol. Chem 2004, 279, 25711–25720; [PubMed: 15082711] b) Posewitz MC, King PW, Smolinski SL, Smith RD, Ginley AR, Ghirardi ML, Seibert M, Biochem. Soc. Trans 2005, 33, 102–104; [PubMed: 15667277] c) King PW, Posewitz MC, Ghirardi ML, Seibert M, J. Bacteriol 2006, 188, 2163–2172; [PubMed: 16513746] d) McGlynn SE, Ruebush SS, Naumov A, Nagy LE, Dubini A, King PW, Broderick JB, Posewitz MC, Peters JW, J. Biol. Inorg. Chem 2007, 12, 443–447. [PubMed: 17372774]
- [5]. a) Brazzolotto X, Rubach JK, Gaillard J, Gambarelli S, Atta M, Fontecave M, J. Biol. Chem 2006, 281, 769–774; [PubMed: 16278209] b) McGlynn SE, Shepard EM, Winslow MA, Naumov AV, Duschene KS, Posewitz MC, Broderick WE, Broderick JB, Peters JW, FEBS Lett. 2008, 582, 2183–2187; [PubMed: 18501709] c) Shepard EM, McGlynn SE, Bueling AL, Grady-Smith C,

- George SJ, Winslow MA, Cramer SP, Peters JW, Broderick JB, Proc. Natl. Acad. Sci. U.S.A 2010, 107, 10448–10453; [PubMed: 20498089] d)Czech I, Silakov A, Lubitz W, Happe T, FEBS Lett. 2010, 584, 638–642; [PubMed: 20018187] e)Németh B, Land H, Magnuson A, Hofer A, Berggren G, J. Biol. Chem 2020, 295, 11891–11901; [PubMed: 32620553] f)Caserta G, Pecqueur L, Adamska-Venkatesh A, Papini C, Roy S, Artero V, Atta M, Reijerse E, Lubitz W, Fontecave M, Nat. Chem. Biol 2017, 13, 779–784. [PubMed: 28553946]
- [6]. a)Czech I, Stripp S, Sanganas O, Leidel N, Happe T, Haumann M, FEBS Lett. 2011, 585, 225–230; [PubMed: 21130763] b)Scott AG, Szilagyí RK, Mulder DW, Ratzloff MW, Byer AS, King PW, Broderick WE, Shepard EM, Broderick JB, Dalton Trans. 2018, 47, 9521–9535. [PubMed: 29964288]
- [7]. a)Berggren G, Adamska A, Lambert C, Simmons TR, Esselborn J, Atta M, Gambarelli S, Mouesca JM, Reijerse E, Lubitz W, Happe T, Artero V, Fontecave M, Nature 2013, 499, 66–69; [PubMed: 23803769] b)Németh B, Esmieu C, Redman HJ, Berggren G, Dalton Trans. 2019, 48, 5978–5986; [PubMed: 30632592] c)Artero V, Berggren G, Atta M, Caserta G, Roy S, Pecqueur L, Fontecave M, Acc. Chem. Res 2015, 48, 2380–2387; [PubMed: 26165393] d)Németh B, Senger M, Redman HJ, Ceccaldi P, Broderick J, Magnuson A, Stripp ST, Haumann M, Berggren G, J. Biol. Inorg. Chem 2020, 25, 777–788. [PubMed: 32661785]
- [8]. a)Driesener RC, Challand MR, McGlynn SE, Shepard EM, Boyd ES, Broderick JB, Peters JW, Roach PL, Angew. Chem. Int. Ed. Engl 2010, 49, 1687–1690; [PubMed: 20108298] b)Shepard EM, Duffus BR, McGlynn SE, Challand MR, Swanson KD, Roach PL, Peters JW, Broderick JB, J. Am. Chem. Soc 2010, 132, 9247–9249; [PubMed: 20565074] c)Shepard EM, Impano S, Duffus BR, Pagnier A, Duschene KS, Betz JN, Byer AS, Galambas A, McDaniel EC, Watts H, McGlynn SE, Peters JW, Broderick WE, Broderick JB, Dalton Trans. 2021, 50, 10405–10422; [PubMed: 34240096] d)Kuchenreuther JM, George SJ, Grady-Smith CS, Cramer SP, Swartz JR, PLoS ONE 2011, 6, e20346. [PubMed: 21673792]
- [9]. a)Kuchenreuther JM, Myers WK, Suess DLM, Stich TA, Pelenschikov V, Shiigi SA, Cramer SP, Swartz JR, Britt RD, George SJ, Science 2014, 343, 424–427; [PubMed: 24458644] b)Suess DLM, Bürstel I, De La Paz L, Kuchenreuther JM, Pham CC, Cramer SP, Swartz JR, Britt RD, Proc. Natl. Acad. Sci. U. S. A 2015, 112, 11455–11460. [PubMed: 26324916]
- [10]. a)Tao L, Pattenau SA, Joshi S, Begley TP, Rauchfuss TB, Britt RD, J. Am. Chem. Soc 2020, 142, 10841–10848; [PubMed: 32434327] b)Rohac R, Martin L, Liu L, Basu D, Tao L, Britt RD, Rauchfuss TB, Nicolet Y, J. Am. Chem. Soc 2021, 143, 8499–8508. [PubMed: 34048236]
- [11]. Zhang Y, Tao L, Woods TJ, Britt RD, Rauchfuss TB, J. Am. Chem. Soc 2022, in press.
- [12]. Shepard EM, Mus F, Betz J, Byer A, Duffus BR, Peters JW, Broderick JB, Biochemistry 2014, 53, 4090–4104. [PubMed: 24878200]
- [13]. a)Boyer ME, Stapleton JA, Kuchenreuther JM, Wang C.-w., Swartz JR, Biotechnol. Bioengin 2007, 99, 59–67; b)Kuchenreuther JM, Stapleton JA, Swartz JR, PLoS ONE 2009, 4, e7565; [PubMed: 19855833] c)Kuchenreuther JM, Britt RD, Swartz JR, PLoS ONE 2012, 7, e45850. [PubMed: 23049878]
- [14]. a)Rao G, Pattenau SA, Alwan K, Blackburn NJ, Britt RD, Proc. Natl. Acad. Sci. U. S. A 2019, 116, 20850–20855; [PubMed: 31570604] b)Rao G, Tao L, Britt RD, Chemical Science 2020, 11, 1241–1247.
- [15]. Douce R, Bourguignon J, Neuberger M, Rébeillé F, Trends Plant Sci. 2001, 167–176. [PubMed: 11286922]
- [16]. Guilhaudis L, Simorre J-P, Blackledge M, Marion D, Gans P, Neuberger M, Douce R, Biochemistry 2000, 39, 4259–4266. [PubMed: 10757974]
- [17]. a)Kamp C, Silakov A, Winkler M, Reijerse EJ, Lubitz W, Happe T, Biochim. Biophys. Acta - Bioenergetics 2008, 1777, 410–416; b)Girbal L, von Abendroth G, Winkler M, Benton PMC, Meynial-Salles I, Croux C, Peters JW, Happe T, Soucaille P, Appl. Environ. Microbiol 2005, 71, 2777–2781; [PubMed: 15870373] c)Kuchenreuther JM, Grady-Smith CS, Bingham A, George SJ, Cramer SP, Swartz JR, PLoS ONE 2010, 5, e15491. [PubMed: 21124800]
- [18]. Okamura-Ikeda K, Hosaka H, Maita N, Fujiwara K, Yoshizawa AC, Nakagawa A, Taniguchi H, J. Biol. Chem 2010, 285, 18684–18692. [PubMed: 20375021]
- [19]. Reijerse EJ, Pelenschikov V, Birrell JA, Richers CP, Kaupp M, Rauchfuss TB, Cramer SP, Lubitz W, J. Phys. Chem. Lett 2019, 10, 6794–6799. [PubMed: 31580680]

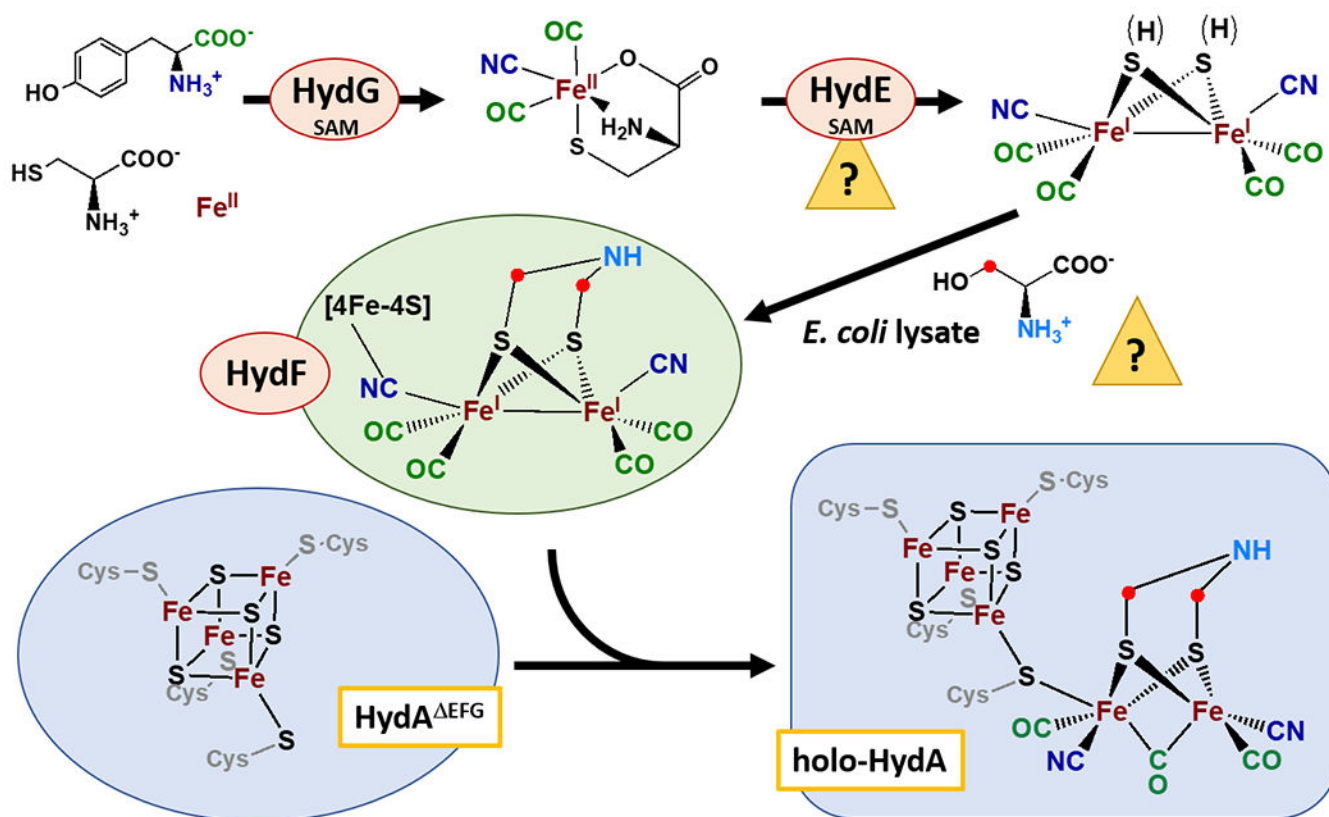
- [20]. Betz JN, Boswell NW, Fugate CJ, Holliday GL, Akiva E, Scott AG, Babbitt PC, Peters JW, Shepard EM, Broderick JB, *Biochemistry* 2015, 54, 1807–1818. [PubMed: 25654171]
- [21]. Li H, Rauchfuss TB, *J. Am. Chem. Soc* 2002, 124, 726–727. [PubMed: 11817928]

Author Manuscript

Author Manuscript

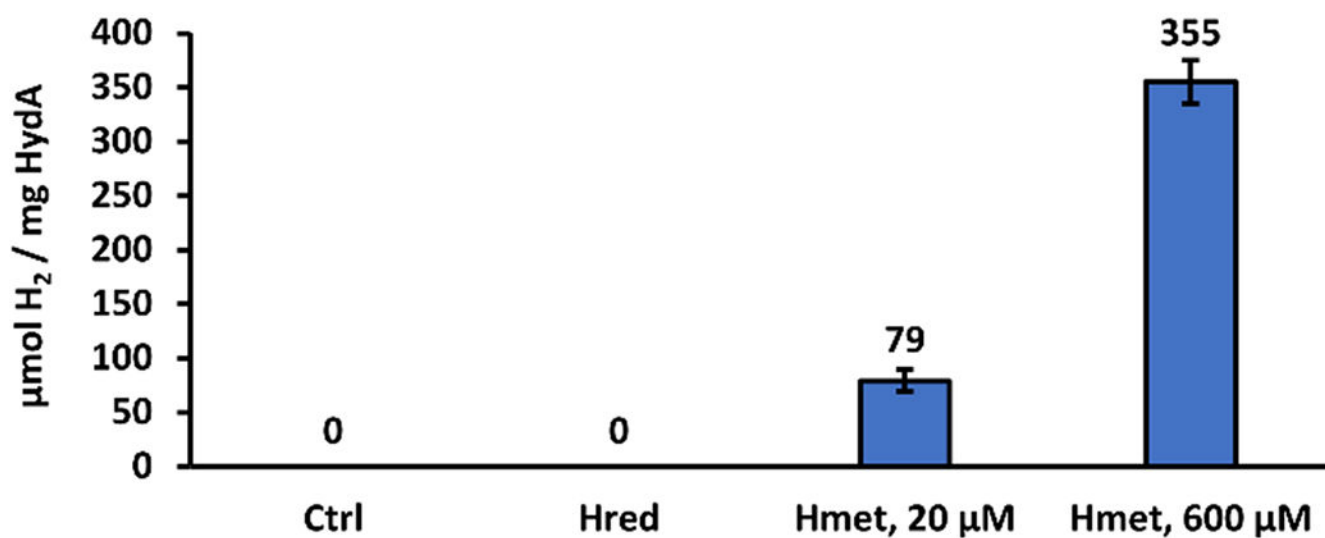
Author Manuscript

Author Manuscript

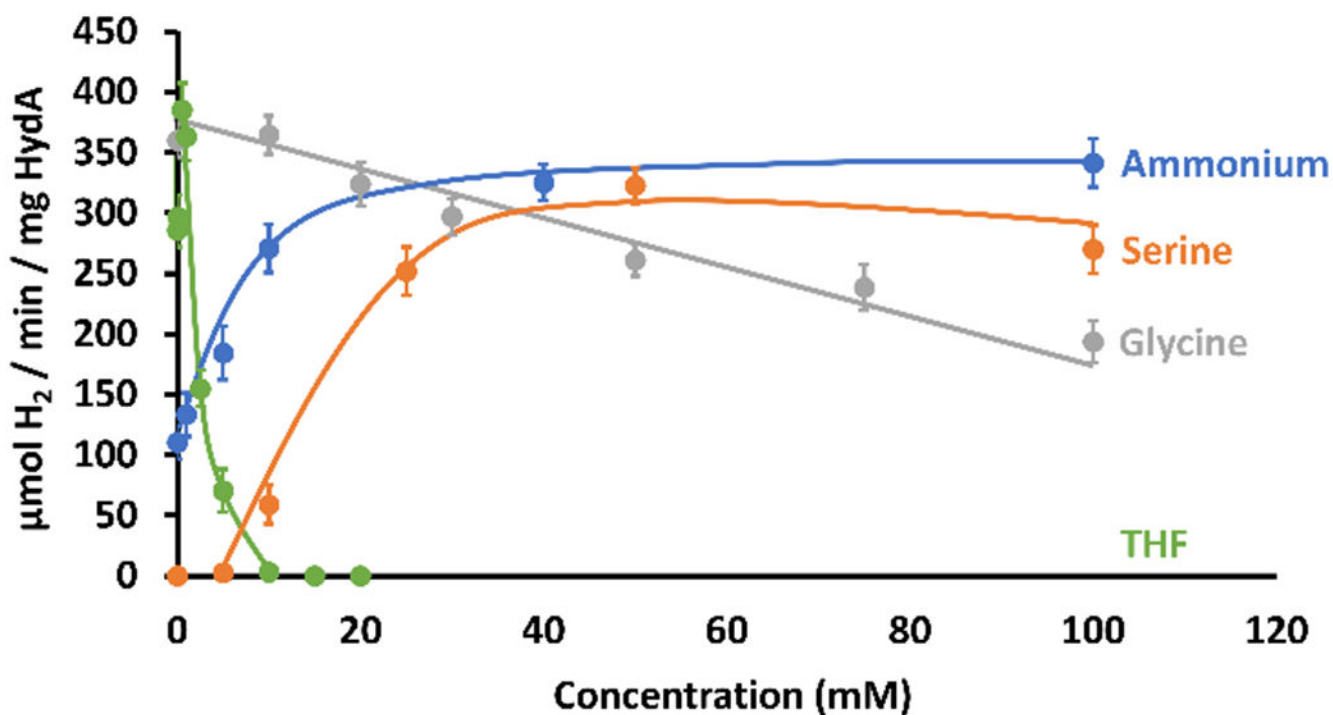


**Figure 1.**  
Current model for biological maturation of the H-cluster of the [FeFe]-hydrogenase.  
Question marks indicate steps that are not fully understood.

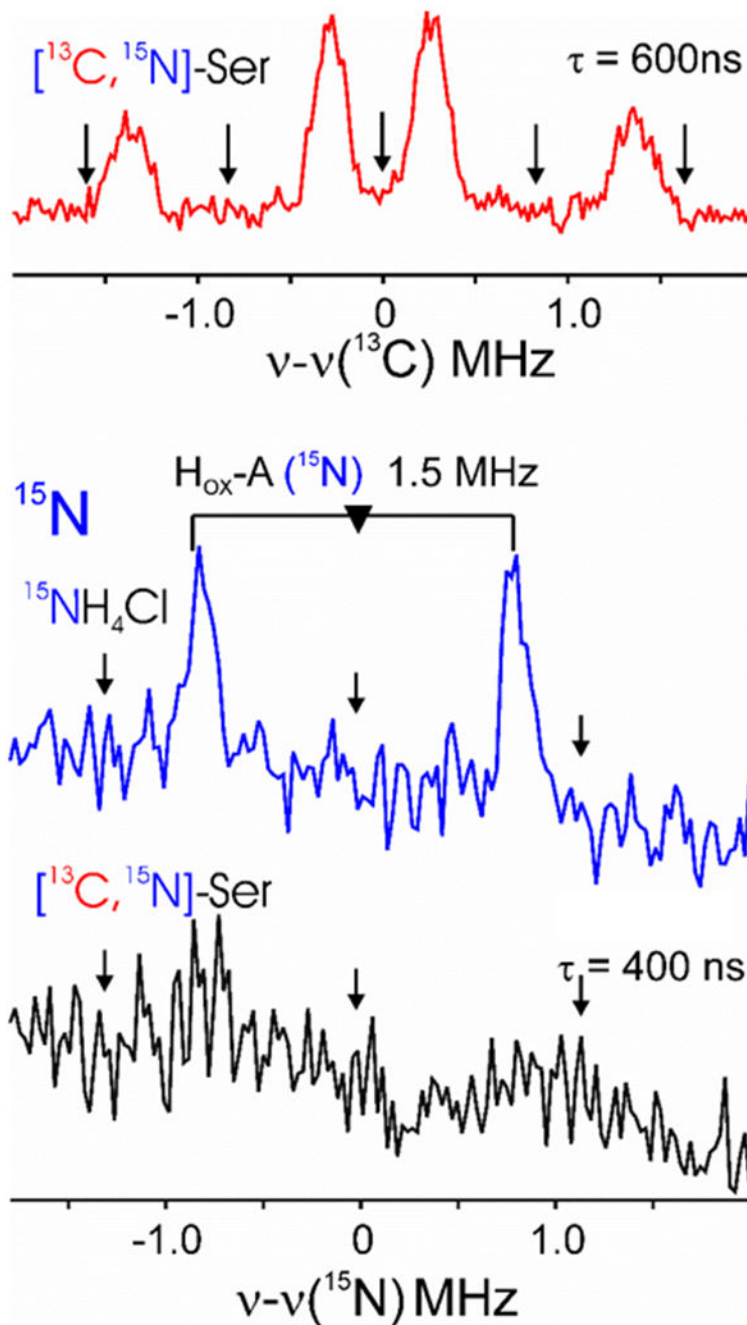




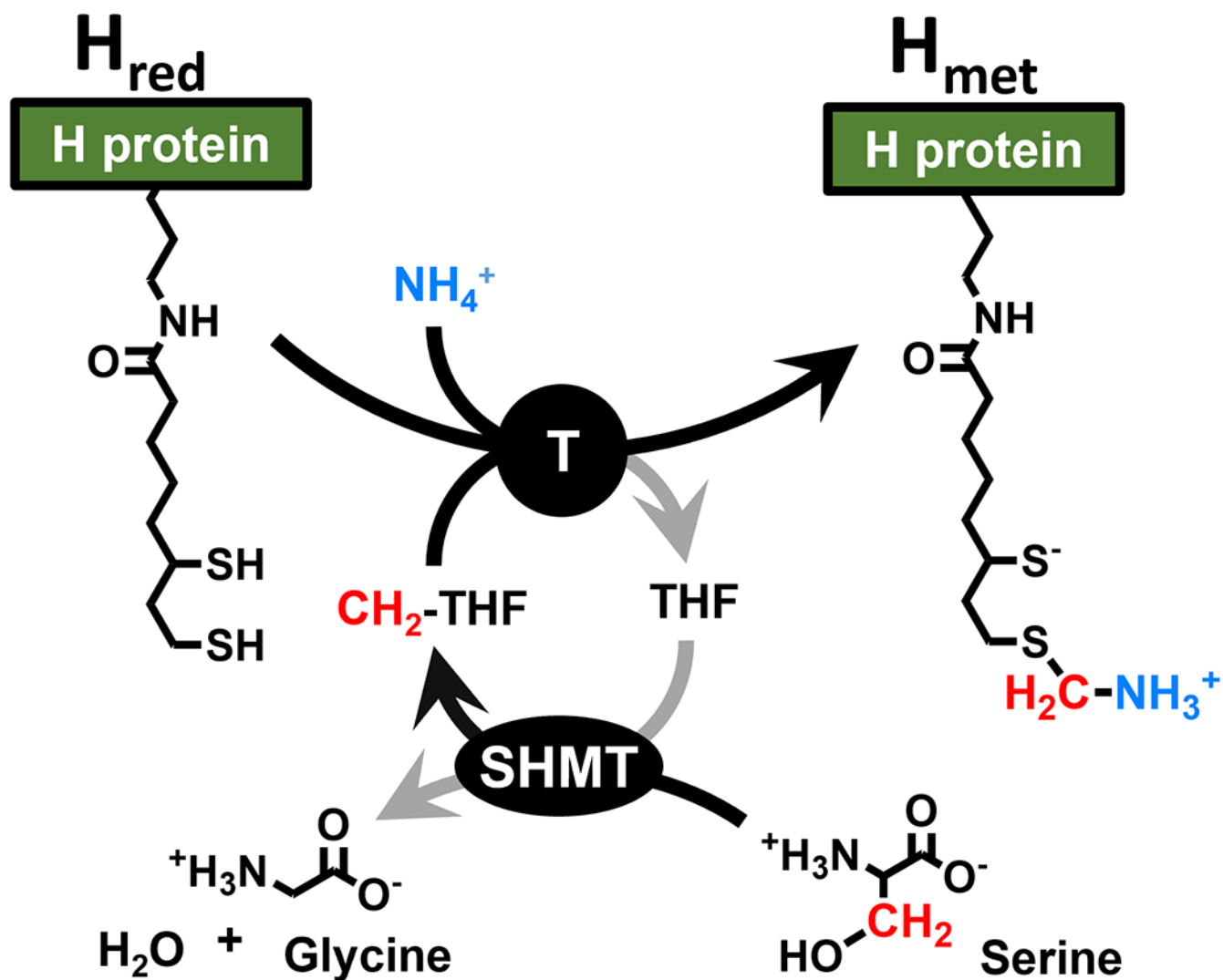
**Figure 2.** Maturation of HydA without lysate, using the GCS H-protein. Without H-protein (**Ctrl**) or with lipoyl-H-protein (**H<sub>red</sub>**) no active HydA is formed. Addition of aminomethyl-lipoyl-H-protein (**H<sub>met</sub>**) provides maturation of HydA in a concentration-dependent manner. Assays for H<sub>2</sub> were run for 24 h to allow H<sub>2</sub> to accumulate. Conditions as reported in the SI.



**Figure 3.** Maturation of HydA in the presence of the T-protein and SHMT. Concentrations of key components were varied to test the hypothesis of the involvement of the GCS and  $H_{\text{met}}$  during hydrogenase maturation. Conditions as reported in the SI. Reactions contained no lysate.

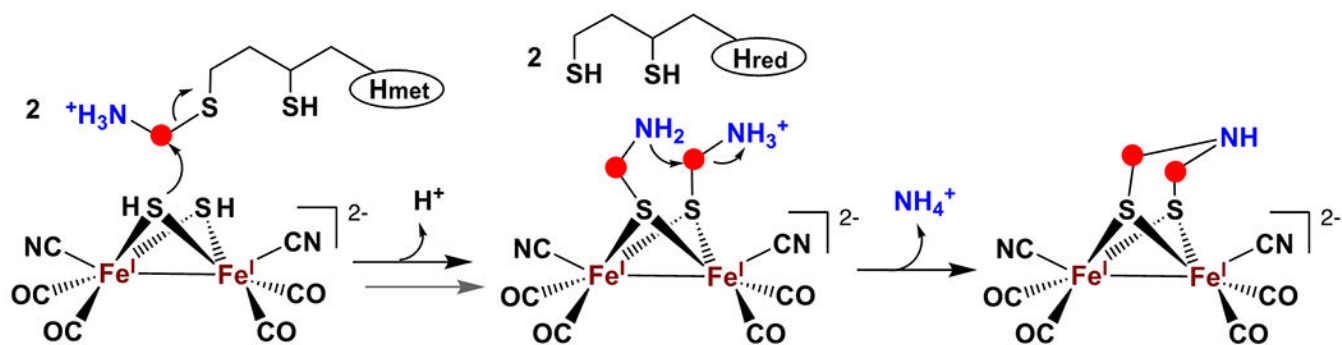


**Figure 4.** 35 GHz pulsed  $^{13}\text{C}$  and  $^{15}\text{N}$  ENDOR of HydA matured in the absence of lysate. Top,  $^{13}\text{C}$  Mims ENDOR spectrum at  $g=2.093$  of HydA matured with  $[^{13}\text{C}_3, ^{15}\text{N}]$ -serine. Bottom,  $^{15}\text{N}$  Mims ENDOR spectra at  $g=2.004$  of HydA matured with  $^{15}\text{NH}_4\text{Cl}$  (blue) and with  $[^{13}\text{C}_3, ^{15}\text{N}]$ -serine (black). Mims holes are indicated with arrows. Microwave frequency, 34.605 GHz; rep time, 50 ms;  $T = 2$  K. For a complete characterization of these signals, see ref 15b.



**Scheme 1.**

*In situ* generation of  $H_{\text{met}}$  from  $H_{\text{red}}$  via SHMT and the T-protein of the GCS (see Scheme S1 for the full GCS). All reactions are reversible.

**Scheme 2.**

Proposed pathway for formation of the DTMA ligand of the [2Fe]<sub>F</sub> cluster on HydF.