Amino Acid Biosynthesis in the Halophilic Archaeon *Haloarcula hispanica*

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Biosynthesis of proteinogenic amino acids in the extremely halophilic archaeon *Haloarcula hispanica* **was explored by using biosynthetically directed fractional 13C labeling with a mixture of 90% unlabeled and 10% uniformly 13C-labeled glycerol. The resulting 13C-labeling patterns in the amino acids were analyzed by two-dimensional 13C,1 H correlation spectroscopy. The experimental data provided evidence for a split pathway for isoleucine biosynthesis, with 56% of the total Ile originating from threonine and pyruvate via the threonine pathway and 44% originating from pyruvate and acetyl coenzyme A via the pyruvate pathway. In addition, the diaminopimelate pathway involving diaminopimelate dehydrogenase was shown to lead to lysine biosynthesis and an analysis of the 13C-labeling pattern in tyrosine indicated novel biosynthetic pathways that have so far not been further characterized. For the 17 other proteinogenic amino acids, the data were consistent with data for commonly found biosynthetic pathways. A comparison of our data with the amino acid metabolisms of eucarya and bacteria supports the theory that pathways for synthesis of proteinogenic amino acids were established before ancient cells diverged into archaea, bacteria, and eucarya.**

Halophilic archaea are aerobic chemo-organotrophs that grow on a variety of carbon sources. The central carbon metabolisms of some species are relatively well explored (15, 50), while comprehensive investigations of amino acid metabolism have so far been pursued only for organisms belonging to other phylogenetic groups (40) within the domain of the archaea (76), i.e., various methanogens (22, 24, 25, 61) and the anaerobic, extremely thermophilic *Thermoproteus neutrophilus* (54). These studies indicated that most amino acids in thermophilic and methanogenic archaea are synthesized via pathways that had previously been described for bacteria and eucarya (27, 46, 68, 69). An extension of such studies to halophilic archaea is thus of interest for obtaining new insights into the evolution of carbon metabolism in general. Furthermore, organisms living under extreme environmental conditions (extremophiles) are gaining increasing importance for biotechnological applications (17) and the analysis of their metabolism constitutes a prerequisite for possible future metabolic engineering (2).

In this paper we investigated amino acid biosynthesis in the halophilic archaeon *Haloarcula hispanica. H. hispanica* was selected for its potential biotechnology interest, since it can efficiently use glycerol for amino acid synthesis (31). We primarily employed biosynthetically directed fractional ¹³C labeling (53, 58, 63–66, 77) with glycerol as the sole carbon source, combined with two-dimensional $(2D)$ ¹³C,¹H correlation nuclear magnetic resonance (NMR) spectroscopy for the analysis of the resulting nonrandom 13C-labeling patterns. In this approach, contiguous carbon fragments arising from a single carbon source molecule are traced through a cellular bioreaction network. Since the patterns of intact carbon fragments observed for a given metabolite are very often different depending on which pathway is employed for its synthesis, we are able to analyze both the topological structure of the bioreaction network, i.e., the locations of nodes at which one substance is either a substrate for two branching reactions or a product of two converging reactions, and the relative contributions of alternative pathways to the generation of amino acids (63, 64, 66).

MATERIALS AND METHODS

Labeling strategy. The biosynthetic pathways were explored by biosynthetically directed fractional ¹³C labeling of the proteinogenic amino acids (53, 58, 63–66, 77), with glycerol as the sole carbon source. Fractional ¹³C labeling was achieved by growing *H. hispanica* in a minimal medium containing approximately 10% uniformly 13C-labeled glycerol and 90% glycerol containing 13C at natural abundance. Incorporation of intact two- or three-carbon fragments from the uniformly ¹³C-labeled carbon source leads to nonrandom ¹³C-labeling patterns in the amino acids. These are identified from $13C-13C$ scalar-coupling fine structures in 2D ¹³C,¹H correlation spectroscopy (COSY) (11). With a set of probabilistic equations (63), the observed 13 C fine structures then yield the relative abundances of intact glycerol carbon fragments in the carbon skeletons of the amino acids. This approach allows for a comprehensive analysis of the bioreaction network (63, 64, 66). Cells were grown in a batch culture and harvested in the mid-exponential and the early stationary phases in order to assess possible changes of the metabolic state during growth. Since all relevant peaks of the individual amino acids are well resolved in the 2D NMR spectrum, a separation of the amino acids prior to NMR analysis is not required (63–66, 77). To verify the threonine pathway for isoleucine biosynthesis, the labeling experiment was repeated with a growth medium containing $[{}^{13}C_4]$ threonine instead of $[^{13}C_3]$ glycerol, which allowed us to directly trace the incorporation of $^{13}C_1^{13}$ C units from threonine into isoleucine.

Growth of the organism and sample preparation. By following the protocol developed in reference 49, *H. hispanica* (31) was grown in a medium containing,
per liter, 200 g of NaCl, 36 g of MgSO₄ · 7H₂O, 6 g of Tris base, 4 g of KCl, 1 g of CaCl₂ · 2H₂O, 2 ml of FeSO₄ · 7H₂O (0.4% in 1 mM HCl), 2 ml of K₂HPO₄ (5% in distilled water), and 5 ml of NH₄Cl (20% in distilled water). Glycerol (20 ml, 25% in water) was added, and the pH was adjusted to 7.5 with HCl prior to sterilization (filter pore size cutoff, 0.45 μ m). In the standard experiments, 10% of the glycerol was uniformly ¹³C labeled. In the exper threonine, no labeled glycerol was used and 307 mg of $[^{13}C_4]$ threonine per liter was added under otherwise identical conditions. Cells were grown in six 35-ml cultures that were shaken at 100 rpm in 100-ml Erlenmeyer flasks for 7 days at 40°C until the early stationary phase was reached. Cells from cultures in the

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mid-exponential growth phase were harvested after 3 days. After centrifugation at $3,000 \times g$, the cells were taken up in 10 ml of water and frozen in liquid nitrogen. No DNase was added in order to prevent contamination of the halobacterial proteins with unlabeled amino acids. After being warmed to 0°C, the slurry was centrifuged at $20,000 \times g$ and the protein in the clear supernatant was precipitated with 65% ethanol at a temperature of -20° C overnight. After centrifugation, the pellet was lyophilized and hydrolyzed after addition of 3 ml of 6 M HCl at 80°C for 2 days in a sealed Pyrex tube, yielding about 70 mg of dried biomass.

NMR spectroscopy. NMR experiments were performed at 40°C. Proton-detected 2D ¹³C,¹H-heteronuclear single-quantum correlation spectra were recorded with the pulse sequence devised by Bodenhausen and Ruben (11), which ensures that ${}^{1}H-{}^{13}C$ scalar couplings do not affect the ${}^{13}C-{}^{13}C$ scalar-coupling fine structure along $\omega_1(^{13}C)$ (Fig. 1 in reference 65). Pulsed-field gradients were employed for coherence pathway rejection $(6, 75)$, and a 2-ms spin-lock pulse (47) was used to purge the magnetization arising from ¹²C-bound protons and the residual ²HOH signal. ¹³C decoupling during *t*₂ was achieved with the composite-pulse decoupling scheme GARP (59), and quadrature detection in ω_1 was accomplished with States-TPPI (41). For the samples obtained from the cultures grown with $\binom{13}{3}$ glycerol, two spectra were recorded, i.e., one focused on the aliphatic carbons, with the ¹³C carrier set to 42.5 ppm relative to the chemical shift of 2,2-dimethyl-2-silapentane-5-sulfonate sodium salt (DSS), and one for the aromatic rings, with the ¹³C carrier set to 125.9 ppm. For the sample obtained from the culture grown with $[^{13}C_4]$ threonine, only the spectrum focusing on the aliphatic carbons was recorded. The spectra for the aliphatic resonances were folded along ω_1 ⁽¹³C), with a sweep width of 33.8 ppm.

Somewhat different experimental conditions were chosen for the individual measurements. For the sample harvested during the mid-exponential growth phase and the sample generated with $\int_{0}^{13}C_{4}$]threonine, the aliphatic spectra were recorded at a ¹³C resonance frequency of 125.8 MHz, with a Bruker DRX500 spectrometer. The measurement time was 9 h per spectrum $(1,706 \times 256$ com-
plex points; maximal evolution times, $t_{1\text{max}} = 402$ ms, $t_{2\text{max}} = 102$ ms; relaxation delay between scans, 2 s). For the sample harvested in the early stationary growth phase, the aliphatic spectrum was recorded in 8 h at 188.6 MHz with a Bruker
DRX750 spectrometer (2,559 \times 512 complex points; $t_{1\text{max}}$ = 392 ms; $t_{2\text{max}}$ = 87 ms; relaxation delay between scans, 2.4 s). The aromatic spectra were recorded in 3 h with a Bruker DRX500 spectrometer (920×512 complex points; $t_{1\text{max}} =$ 392 ms; $t_{2\text{max}} = 87$ ms; relaxation delay between scans, 1 s). The data were processed with the program PROSA (28). Before Fourier transformation, the time domain data were multiplied in t_1 and t_2 with sine-bell windows shifted by $\pi/2$ (19). The digital resolution after zero-filling was 1.0 Hz/point along ω_1 and 2.4 Hz/point along ω_2 for the aliphatic spectra at 125.8 MHz, and 0.6 Hz/point along ω_1 and 5.8 Hz/point along ω_2 for the aromatics. For the 188.6-MHz spectrum the digital resolution was 0.8 Hz/point along ω_1 and 3.7 Hz/point along ω_2

For the experiments using $[^{13}C_3]$ glycerol as the ¹³C source, the overall degree of 13 C labeling in the amino acids, denoted as p_1 in the probabilistic equations of reference 63, was determined from the satellites of selected well-separated peaks in 1D¹H NMR spectra ($t_{\text{max}} = 1.022$ s; relaxation delay between scans, 10[']s). p_1 was 0.127 for both preparations used here.

Data analysis. The individual multiplet components of the $^{13}C-^{13}C$ scalarcoupling fine structures were integrated with the program XEASY (4), and the observed relative multiplet intensities were used to calculate the relative abundances of intact carbon fragments (63). By following the definitions of references 63 and 66, $f^{(1)}$ represents the fraction of molecules in which the observed carbon atom and its neighboring carbons originate from different source molecules of glycerol and $f^{(2)}$ represents the fraction of molecules in which the observed carbon atom and at least one neighboring carbon originate from the same source
molecule. For a central carbon in a C_3 fragment that exhibits different ¹³C—¹³C scalar-coupling constants with the two attached carbons, $f^{(2)}$ represents the fraction of molecules for which the central carbon and the carbon with the smaller coupling come from the same source molecule, while $f^{(2^*)}$ is used if the carbon with the larger coupling comes from the same source molecule as the observed carbon. $f^{(3)}$ denotes the fraction of molecules in which the observed carbon atom and both neighbors in the C_3 fragment originate from the same glycerol molecule.

The relative abundances of intact carbon fragments in the amino acids that are expected for specific biosynthetic pathways can be calculated from the relative abundances found in the respective precursors (see Fig. 1 and 3). Here, we present only those equations that are required to explore the biosynthesis of isoleucine, lysine, and tyrosine.

(i) Biosynthesis of isoleucine. C_{γ_2} $-C_{\beta}$ of Ile derives from pyruvate in all pathways considered (see Fig. 2 and 3B), and Ile- γ_2 is therefore not used in the calculations.

(a) Pyruvate pathway. Pyruvate and acetyl coenzyme A (acetyl-CoA) are assessed via Ala- β and Leu- α , respectively.

$$
f^{(1)}\{\text{Ile-}\delta\} = f^{(1)}\{\text{Ala-}\beta\}, \qquad f^{(2)}\{\text{Ile-}\delta\} = f^{(2)}\{\text{Ala-}\beta\} \tag{1}
$$

$$
f^{(1)}\{\text{Ile-}\alpha\} = [f^{(1)} + f^{(2)}]\{\text{Leu-}\alpha\}, f^{(2^*)}\{\text{Ile-}\alpha\} = [f^{(2^*)} + f^{(3)}]\{\text{Leu-}\alpha\} \tag{2}
$$

(b) Threonine pathway. Threonine is directly assessed.

$$
f^{(1)}\{\text{Ile-}\delta\} = f^{(1)}\{\text{Thr-}\gamma_2\}, \qquad f^{(2)}\{\text{Ile-}\delta\} = f^{(2)}\{\text{Thr-}\gamma_2\} \tag{3}
$$

$$
f^{(1)}\{\text{Ile-}\alpha\} = [f^{(1)} + f^{(2)}]\{\text{Thr-}\alpha\}, f^{(2^*)}\{\text{Ile-}\alpha\} = [f^{(2^*)} + f^{(3)}]\{\text{Thr-}\alpha\} \tag{4}
$$

(c) Glutamate pathway. Glutamate is directly assessed.

$$
f^{(1)}\{\text{Ile-}\delta\} = [f^{(1)} + f^{(2^*)}]\{\text{Glu-}\gamma\}, f^{(2)}\{\text{Ile-}\delta\} = [f^{(2)} + f^{(3)}]\{\text{Glu-}\gamma\} \tag{5}
$$

$$
f^{(1)}\{\text{He}-\alpha\} = [f^{(1)}+f^{(2)}]\{\text{Glu}-\alpha\}, f^{(2^*)}\{\text{He}-\alpha\} = [f^{(2^*)}+f^{(3)}]\{\text{Glu}-\alpha\} \tag{6}
$$

(ii) Biosynthesis of lysine (Fig. 3C). *(a) Diaminopimelate pathways.* Pyruvate and aspartate are assessed via Ala- β and Asp- α or Asp- β , respectively. *(a.1) Diaminopimelate-dehydrogenase variant.*

$$
f^{(1)}\{\text{Lys-}\varepsilon\} = f^{(1)}\{\text{Ala-}\beta\}, f^{(2)}\{\text{Lys-}\varepsilon\} = f^{(2)}\{\text{Ala-}\beta\}
$$
(7)

$$
f^{(1)}\{\text{Lys-}\beta\} = f^{(1)}\{\text{Asp-}\beta\}, f^{(2)}\{\text{Lys-}\beta\} = [f^{(2)} + f^{(2^*)}]\{\text{Asp-}\beta\}, f^{(3)}\{\text{Lys-}\beta\} = f^{(3)}\{\text{Asp-}\beta\} \quad (8)
$$

(a.2) Acetylase/succinylase variant.

$$
f^{(1)}\{\text{Lys-}\varepsilon\} = 0.5 \ (f^{(1)}\{\text{Ala-}\beta\} + [f^{(1)} + f^{(2^*)}]\{\text{Asp-}\alpha\}), f^{(2)}\{\text{Lys-}\varepsilon\} = 0.5 \ (f^{(2)}\{\text{Ala-}\beta\} + [f^{(2)} + f^{(3)}]\{\text{Asp-}\alpha\})
$$
(9)

 $f^{(1)}$ {Lys- β } = 0.5 $(f^{(1)}$ {Ala- β } + $f^{(1)}$ {Asp- β }), $f^{(2)}$ {Lys- β } = 0.5 $(f^{(2)}$ {Ala- β }

$$
+[f^{(2)}+f^{(2^*)}]\{\text{Asp-}\beta\},f^{(3)}\{\text{Lys-}\beta\}=0.5f^{(3)}\{\text{Asp-}\beta\}\tag{10}
$$

(b) a*-Aminoadipate pathway.* 2-Oxoglutarate is assessed via Glu-g and Glu-a.

$$
f^{(1)}\{\text{Lys-}\varepsilon\} = [f^{(1)} + f^{(2)}]\{\text{Glu-}\gamma\}, \ f^{(2)}\{\text{Lys-}\varepsilon\} = [f^{(2^*)} + f^{(3)}]\{\text{Glu-}\gamma\} \tag{11}
$$

$$
f^{(1)}\{\text{Lys-}\beta\} = [f^{(1)} + f^{(2^*)}]\{\text{Glu-}\alpha\}, f^{(2)}\{\text{Lys-}\beta\} = [f^{(2)} + f^{(3)}]\{\text{Glu-}\alpha\}, f^{(3)}\{\text{Lys-}\beta\} = 0.00
$$
\n(12)

(iii) Biosynthesis of the aromatic ring of tyrosine. In the biosynthesis of the aromatic ring of tyrosine through the common shikimate pathway (7, 68, 69), the ring is assembled from phosphoenolpyruvate and erythrose-4-phosphate. Phosphoenolpyruvate is assessed via Phe- α and Tyr- α . Erythrose-4-phosphate is assumed to include an intact $C_2-C_3-C_4$ fragment derived from a first glycerol molecule via the C_3 pool of glycolysis and a C_1 atom that originates from a second glycerol molecule (see Fig. 1) (50). Due to the symmetry of the aromatic ring of tyrosine, the Tyr-d*^x* and Tyr-ε*^x* carbons give rise to only one 13C fine structure. Therefore, *f* (i){Tyr-d} and *f* (i){Tyr-ε} are calculated as averages of the *f* values predicted for Tyr- δ_1 and Tyr- δ_2 or Tyr- ε_1 and Tyr- ε_2 , respectively (63).

$$
f^{(1)}\{\text{Tyr-}\delta\} = 0.5 \left([f^{(1)} + f^{(2^*)}] \{ \text{Phe-}\alpha \} + [f^{(1)} + f^{(2^*)}] \{ \text{Tyr-}\alpha \} \right), f^{(2)}\{\text{Tyr-}\delta\} = 0.5
$$

+ 0.25
$$
([f^{(1)} + f^{(2^*)}] \{ \text{Phe-}\alpha \} + [f^{(1)} + f^{(2^*)}] \{ \text{Tyr-}\alpha \}, f^{(3)} \{ \text{Tyr-}\delta \} = 0.00
$$
 (13)

$$
f^{(1)}\{\text{Tyr-}\varepsilon\} = 0.50, f^{(2)}\{\text{Tyr-}\varepsilon\} = 0.00, f^{(3)}\{\text{Tyr-}\varepsilon\} = 0.50\tag{14}
$$

RESULTS

The $2D¹³C$,¹H COSY spectra were analyzed as described by Szyperski (63) (Table 1). Virtually identical ¹³C scalar-coupling fine structures were observed for samples taken from the mid-exponential growth phase and the early stationary phase, and hence there are no significant differences in the resulting relative abundances of intact carbon fragments, as collected in Table 1. This suggests that there are no major differences in flux ratios through the central metabolic network when the two growth periods are compared (63–66). During hydrolysis, cysteine and tryptophan were oxidized and could thus not be evaluated and asparagine and glutamine were deamidated to aspartate and glutamate (65, 77). The ring carbons of phenylalanine were not evaluated because of strong-coupling effects (63). The evaluation of the observations for all other carbon positions (Table 1) showed that except for isoleucine, lysine, and the aromatic ring of tyrosine, the proteinogenic amino acids in *H. hispanica* are synthesized according to the pathways commonly found in both bacteria and eucarya (27, 46, 68, 69).

Amino acid synthesis from glycolytic intermediates. Identical values for intact carbon fragments (*f* values) are observed for phenylalanine and tyrosine, where $f^{(3)}$ {Phe- α } approximately equals $f^{(3)}$ {Tyr- α }, which approximately equals 1 (Table 1). This finding is in agreement with the shikimate/chorismate pathway (7, 68, 69), where the $C_B-C_\alpha-C'$ fragments of

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Continued on following page

Observed carbon position	Preparation c	17 DLL 1 communica Relative intensity of multiplet component ^a					Relative abundance of intact carbon fragment ^b			
		I_s	${\cal I}_d$	I_{d^*}	I_{dd}	I_t	$f^{(1)}$	$f^{(2)}$	$f^{(2^*)}$	$f^{(3)}$
Met- α	ME^d $\mathop{\hbox{\rm ES}}$	0.17	0.09	0.31	0.43		0.14	0.09	0.36	0.41
Phe- α	ME $\mathop{\hbox{\rm ES}}$	0.08 0.08	0.04 0.01	0.00 0.00	0.88 0.91		0.00 0.01	0.05 0.01	0.00 0.00	0.95 0.98
$Phe-B$	ME $\mathop{\hbox{\rm ES}}$	0.07 0.07	$0.81\,$ 0.79	$0.01\,$ 0.02	0.12 0.12		$0.00\,$ 0.01	1.00 0.98	0.00 0.01	0.00 0.00
$Pro-\alpha$	ME ES	0.17 0.17	0.36 0.35	0.33 0.31	0.14 0.17		0.15 0.15	0.42 0.41	0.38 0.36	0.05 0.08
$Ser-\alpha$	ME ES	0.08 0.08	0.00 0.01	0.43 0.45	0.49 0.46		0.01 0.01	0.00 0.00	0.53 0.56	0.46 0.43
Thr- α	ME $\mathop{\hbox{\rm ES}}$	0.17 0.18	0.10 0.10	0.32 0.32	0.41 0.40		0.15 0.16	0.10 0.10	0.37 0.37	0.38 0.37
Tyr- α	ME ES	0.07 0.08	0.01 0.00	0.00 $0.00\,$	0.92 0.92		0.00 0.01	0.01 0.00	0.00 0.00	0.99 0.99
$Tyr-\beta$	ME ES	0.07 0.07	0.81 0.80	0.01 0.01	0.12 0.12		0.00 0.01	1.00 0.99	0.00 0.00	0.00 0.00
Val- α	ME $\mathop{\hbox{\rm ES}}$	0.08 0.08	0.01 0.03	0.79 0.77	0.12 0.12		0.02 $0.02\,$	0.00 0.03	0.98 0.95	0.00 0.00
Central carbons in C_3 fragments (equal scalar couplings)										
$Arg-\beta$	ME ES	0.41 0.42	0.52 0.51			0.07 0.07	0.50 0.51	0.50 0.49		0.00 0.00
$Glu-\beta$	ME ES	0.42 0.44	0.51 0.50			0.07 0.06	0.51 0.54	0.49 0.46		0.00 $0.00\,$
$\text{II}e-\gamma_1$	ME $\mathop{\hbox{\rm ES}}$	0.30 0.32	0.60 0.60			0.10 0.08	0.33 0.36	0.65 0.64		0.02 0.00
$Leu-\beta$	ME $\mathop{\hbox{\rm ES}}$	0.75 0.76	0.22 0.22			0.03 0.02	0.99 1.00	0.00 0.00		0.01 0.00
$Lys-\beta$	ME ES	0.17 0.19	0.67 0.67			0.16 0.14	0.15 0.18	0.78 0.77		0.07 0.05
Lys- γ	ME ES	0.44 0.43	0.50 0.51			0.06 0.06	0.54 0.53	0.46 0.47		0.00 0.00
$Lys-$	ME $\mathop{\hbox{\rm ES}}$	$0.08\,$ $0.08\,$	$\rm 0.81$ 0.81			0.12 0.12	$0.02\,$ 0.02	0.98 0.98		$0.00\,$ 0.00
$Pro-\beta$	ME $\mathop{\hbox{\rm ES}}$	0.40 0.41	0.53 0.52			0.07 $0.07\,$	$0.48\,$ $0.50\,$	0.52 0.50		$0.00\,$ $0.00\,$
$Pro-\gamma$	ME $\mathop{\hbox{\rm ES}}$	0.07 $0.08\,$	0.77 0.77			$0.16\,$ 0.15	$0.01\,$ $0.02\,$	0.94 0.94		0.05 0.04
Thr- β	ME $\mathop{\hbox{\rm ES}}$	0.19 0.19	0.67 0.65			0.14 0.16	0.18 0.18	0.77 0.75		0.05 $0.07\,$
Tyr- δ_x	ME $\mathop{\hbox{\rm ES}}$	0.14 0.15	0.74 0.74			0.12 $0.11\,$	$0.11\,$ $0.12\,$	0.88 0.87		$0.01\,$ $0.01\,$
Tyr- ε_{x}	ME $\mathop{\hbox{\rm ES}}$	0.28 0.29	0.25 0.24			0.47 0.47	0.30 0.31	0.22 0.21		0.48 0.48

TABLE 1—*Continued*

 ${}^a I_s$ singlet; I_d , doublet split by the smaller coupling; I_{d*} , doublet split by the larger coupling; I_{dd} , doublet of doublets; I_t , triplet.
 b Experimental error, ± 0.02 .
 c ME, sample harvested in

FIG. 1. Synthesis of erythrose-4-phosphate from glycerol via reversal of glycolytic reactions and the action of transketolase (EC 2.2.1.1). Thick lines indicate carbon-carbon connectivities arising from a single source molecule of glycerol (see the text).

phenylalanine and tyrosine are both derived from phosphoenolpyruvate, which is itself expected to derive from glycerol without cleavage of carbon-carbon bonds (50). Serine appears to be synthesized from 3-phosphoglycerate, which is also directly derived from glycerol. However, reversible interconversion into glycine and a C_1 unit leads to cleavage of C_{β} — C_{α} connectivities, so that $f^{(1)}$ {Ser- β } approximately equals 0.55 in both preparations (Table 1). Moreover, the fact that $[f^{(2^*)} + f^{(3)}]\{\text{Ser-}\alpha\}$ approximately equals $[f^{(2)}]\{\text{Gly-}\alpha\}$ (Table 1) shows that glycine is nearly exclusively derived from C_{α} —C' of serine.

Virtually identical *f* values are detected for Val- γ_1 , Leu- δ_1 , and Ala- β , which provides evidence that valine, leucine, and alanine are derived from pyruvate according to the well-known biosynthetic pathways (68, 69). This is further supported by the following observations: $[f^{(2^*)} + f^{(3)}]$ {Ala- α } \approx $[f^{(2^*)} +$ $f^{(3)}$ }{Val- α } and $f^{(1)}$ {Leu- β } $\approx f^{(1)}$ {Val- γ_2 } $\approx f^{(1)}$ {Leu- δ_2 } \approx 1 (Table 1) (63). That $f^{(2)}$ {Ala- β } approximately equals $f^{(2^*)}$ {Leu- α } further shows that Leu- α is derived from C_2 of acetyl-CoA.

The *f* values of His- α , His- β , and His- δ (Table 1) show that the precursor for histidine, ribose-5-phosphate, is synthesized from glyceraldehyde-3-phosphate and fructose-6-phosphate via the nonoxidative part of the pentose phosphate pathway (Fig. 1). The presence of transketolase (EC 2.2.1.1) and transaldolase (EC 2.2.1.2) has been reported for other halo-

philic archaea (15, 50). The equations $f^{(3)}$ {His- α } \approx $f^{(3)}$ {Phe- α } $\approx f^{(3)}$ {Tyr- α } ≈ 1 and $f^{(2)}$ {His- δ } $\approx f^{(2)}$ {Phe- β } \approx $f^{(2)}$ {Tyr- β } \approx 1 indicate that histidine is composed of a C_β — C_α — C' fragment and a $C_{\delta 2}$ — C_γ fragment from two different glycerol molecules (Fig. 1). This is due to the facts that (i) the glyceraldehyde-3-phosphate pool is almost exclusively derived from glycerol without cleavage of carbon-carbon bonds (see above) and that (ii) fructose-6-phosphate is synthesized from two glycerol molecules via reversal of glycolytic reactions (15, 50), so that it comprises the intact fragments $C_1-C_2-C_3$ and $C_4-C_5-C_6$ from two different glycerol molecules.

The *f* values that characterize pyruvate and phosphoenolpyruvate are slightly different, i.e., pyruvate exhibits about 5% cleavage of C_3-C_2 connectivities, as evidenced by the equation $f^{(1)}$ {Ala- β } $\approx \tilde{f}^{(1)}$ {Val- γ_1 } $\approx f^{(1)}$ {Leu- δ_1 } $\approx f^{(1)}$ {Ile- γ_2 } \approx 0.05, whereas only intact C_3-C_2 connectivities are detected for phosphoenolpyruvate $(f^{(2)}\{Phe^{-\alpha}\}\approx f^{(2)}\{Tryr-\alpha\} \approx 1)$ (Table 1). This suggests that, in addition to the synthesis of pyruvate from phosphoenolpyruvate via pyruvate kinase (EC 2.7.1.40) (50), the malic enzyme (EC 1.1.1.38, 1.1.1.39, and 1.1.1.40) contributes to pyruvate synthesis via oxalacetate. In fact, malic enzyme activity has been reported for other archaea, such as *Halobacterium salinarium* (10), *Halobacterium cutirubrum* (13), and *Sulfolobus solfataricus* (5).

Amino acid synthesis from intermediates of the tricarboxylic acid cycle. The *f* values observed for Asp- α and Asp- β coincide with those observed for Thr- α and Met- α (Asp- α) and Thr- β (Asp- β). Moreover, virtually identical f values were found for Glu- α and Pro- α ; for Glu- β , Pro- β , and Arg- β ; for Glu- γ and Pro- γ ; and for Pro- δ and Arg- δ (Table 1). This is consistent with the well-known pathways in which aspartate, threonine, and methionine are derived from oxalacetate and in which 2-oxoglutarate serves for the biosynthesis of glutamine, proline, and arginine (63). Furthermore, the following observations demonstrate that 2-oxoglutarate is formed by irreversible condensation of oxalacetate with acetyl-CoA in the citric acid cycle (63): (i) the equation $f^{(3)}$ {Glu- β } $\approx f^{(3)}$ {Pro- β } \approx $f^{(2)}$ {Glu- γ } \approx $f^{(3)}$ {Glu- γ } \approx $f^{(3)}$ {Pro- γ } \approx 0 shows that intact C_3-C_4 connectivities in 2-oxoglutarate are absent; (ii) the *f* values observed for Asp- β (and also Thr- β) are equal to those observed for Glu- α (and Pro- α), which is in agreement with the fact that the $C_1-C_2-C_3$ segment of 2-oxoglutarate is derived from $C_2-C_3-C_4$ of oxalacetate; and (iii) the equation $f^{(2^*)}\{\text{Glu-}\gamma\} \approx f^{(2^*)}\{\text{Leu-}\alpha\} \approx f^{(2)}\{\text{Pro-}\delta\} \approx f^{(2)}\{\text{Arg-}\delta\}$ shows that C_4-C_5 of 2-oxoglutarate is derived from acetyl-CoA (Table 1).

Isoleucine biosynthesis. A variety of pathways are known for isoleucine biosynthesis in microorganisms. Most common is the threonine pathway, which uses threonine and pyruvate as precursors (1, 68, 69) (Fig. 2 and 3B). In the pyruvate pathway, isoleucine biosynthesis proceeds from pyruvate and acetyl-CoA (14) (Fig. 2 and 3B), whereas in the glutamate pathway, glutamate and pyruvate serve as precursors (36, 48) (Fig. 2 and 3B). In rare cases, isoleucine synthesis has also been found to proceed from homoserine (26, 71), propionate (23, 45, 52), or 2-methylbutyrate (23, 45, 51). In all pathways except the one starting from 2-methylbutyrate, a-ketobutyrate serves as an intermediate and pyruvate yields the $C_{\gamma 2}$ - C_{β} fragment of isoleucine.

In *H. hispanica*, Ile- γ_2 exhibits the same distribution of intact carbon fragments originating from a single molecule of glycerol as Ala- β (Table 1), indicating that the $C_{\gamma 2}$ - C_{β} fragment of isoleucine and the $C_{\beta}-C_{\alpha}$ fragment of alanine are both derived from pyruvate (Fig. 2 and 3B). However, the relative abundances of intact carbon fragments determined at Ile- α and Ile- δ cannot be explained by a single one of the possible

FIG. 2. Three routes for isoleucine biosynthesis (see the text). Pyruvate and acetyl-CoA are the precursors for Ile synthesis via the pyruvate pathway (route 1) (14), threonine and pyruvate are the precursors for Ile synthesis via the threonine pathway (route 2) (1, 68, 69), and glutamate and pyruvate are the precursors for Ile synthesis via the glutamate pathway (route 3) (36, 48). α -Ketobutyrate is a common intermediate in these three pathways.

individual pathways (Table 2). A satisfactory fit of the data was obtained with the assumption that the threonine and pyruvate pathways operate simultaneously in a split-pathway fashion. For the early stationary growth phase, the decomposition of the ¹³C—¹³C fine structures at both Ile- α and Ile- δ indicates that 56% of isoleucine is synthesized via the threonine pathway and 44% via the pyruvate pathway, and virtually identical values were obtained for the mid-exponential phase (Table 2; Fig. 4).

To verify that the threonine pathway, which has so far not been described for archaea, is operational in *H. hispanica*, we performed an additional labeling experiment using $[^{13}C_4]$

FIG. 3. Incorporation of oxalacetate or threonine, 2-oxoglutarate, pyruvate, and acetyl-CoA, for which carbon skeletons are schematically shown (A), into the carbon skeletons of isoleucine (B) and lysine (C), according to the biosynthetic pathways shown in Fig. 2 and 6. Note that threonine is synthesized from oxalacetate without rearrangement of the carbon skeleton (see the text). The notation of the carbon atoms follows IUPAC-IUB recommendations (30), i.e., $-C_2C(4) - C(3)H_3 - C(2)O - C(1)O_2$ for oxalacetate, $-C_2C(5) - C(4)H_2 - C(2)O$ (3) $\overline{H}_2-C(2)O-C(1)O_2$ ⁻ for 2-oxoglutarate, $C(3)H_3-C(2)O-C(1)O_2$ ⁻ for pyruvate, and $C(2)H_3$ — $C(1)O$ —SCoA for acetyl-CoA.

threonine instead of $[^{13}C_3]$ glycerol (see Materials and Methods). Consistent with the presence of the threonine pathway, the $13C$ ¹³C scalar-coupling fine structures observed at Ile- δ , Ile- γ_1 , and Ile- α are dominated by a doublet (Fig. 5), which proves that intact $C_8-C_{\gamma1}$ and $C_\alpha-C'$ fragments originate from $[^{13}C_4]$ threonine (Fig. 3). The poor signal-to-noise ratio of the γ_2 -carbon cross peaks indicates that this carbon position is only sparsely enriched with ¹³C. Consistently, the γ_2 carbon exhibits a fine structure that is also observed at Ala- β . This is due to the fact that the $C_{\gamma 2}$ - C_{β} fragment is derived from pyruvate in both the threonine and the pyruvate pathways (note that pyruvate serves for alanine biosynthesis). Longrange ${}^{3}J_{C\alpha C\delta}$ couplings of 3.1 Hz are observed as additional splittings of the doublet lines at Ile- α and Ile- δ , and ${}^{3}J_{C\gamma 2C'}$ couplings of 1.7 Hz (see reference 35) in threonine are observed as broadenings of the doublet lines at Thr- γ_2 (Fig. 5). The observation of these long-range carbon-carbon couplings confirms that both of the $C_8-C_{\gamma1}$ and $C_\alpha-C'$ fragments in isoleucine are derived from the same single threonine molecule via α -ketobutyrate.

Alternative combinations of multiple biosynthesis pathways are not supported by the experimental data. For example, decomposition of the isoleucine 13 C $-^{13}$ C fine structures into fractions stemming from the pyruvate and glutamate pathways leads to different contributions for the two pathways when they are derived either from Ile- α or from Ile- δ , both for the midexponential and the early stationary phase (Table 2). Decomposition into the threonine and the glutamate pathways yields negative contributions for one of the pathways. Finally, if one assumes simultaneous operation of all three pathways of Fig. 2, one finds that the contribution of the glutamate pathway would

^a For the definitions of $f^{(1)}$, $f^{(2)}$, and $f^{(2^*)}$, see Materials and Methods. $f^{(2)}$ denotes fractions for Ile-8, and $f^{(2^*)}$ denotes fractions for Ile- α , $f^{(2)}$ {Ile- α } = $f^{(3)}$ {Ile- α } $= 0.00$ (Table 1; Fig. 3). All values except the experimental values are predictions.

b The fractions for Ile- γ_1 are not given since those derived from Ile- δ and Ile- γ_1 provide the same information (63). *c* Contributions of the pyruvate and threonine pathways, 44 and 56%, respectively.

^d Calculated with equations 1 and 2.

^e Calculated with equations 3 and 4.

^f Calculated with equations 5 and 6.

be below 10% while the relative contributions of the pyruvate and threonine pathways would be similar to those obtained when only those two pathways are assumed to be active (see above and Fig. 4).

Lysine biosynthesis. Lysine biosynthesis via the "diaminopimelate pathway" starts from aspartate and pyruvate, whereas the " α -aminoadipate pathway" relies on 2-oxoglutarate and acetyl-CoA as precursors (9, 54) (Fig. 3C and 6A). Two variants of the diaminopimelate pathway differ in the reaction sequence used to convert $L-\Delta^1$ -piperidine-2,6-dicarboxylate to DL-diaminopimelate (Fig. 6B). In the dehydrogenase variant, $L-\Delta^1$ -piperidine-2,6-dicarboxylate is directly converted to DLdiaminopimelate by diaminopimelate dehydrogenase (EC 1.4. 1.16) (42, 43, 74). In the acetylase/succinylase variant (8, 32,

62), L- Δ ¹-piperidine-2,6-dicarboxylate is first acetylated or succinylated and then converted to the symmetric intermediate LL-diaminopimelate, which is finally epimerized to DL-diaminopimelate. The dehydrogenase variant can readily be distinguished from the acetylase/succinylase variants in the fractional labeling experiment, because the involvement of the symmetric intermediate LL-diaminopimelate implies symmetrization of the ¹³C-labeling pattern about the C_{β} — C_{γ} bond (Fig. 3C).

For *H. hispanica* our study indicates that biosynthesis during the mid-exponential as well as the early stationary growth phase proceeds via the dehydrogenase variant of the diaminopimelate pathway (Fig. 3C and 6B; Table 3). The relative abundances of intact carbon fragments observed at Lys- δ and Lys-ε correspond to the values observed at Ala-β (Table 1),

FIG. 4. Decomposition of the experimental 13 C $-^{13}$ C scalar-coupling fine structures for the Ile- α and Ile- δ carbons into contributions from the threonine and pyruvate pathways (Fig. 2; Table 2). The stick diagrams represent the fine structures that would be expected if only a single pathway were operational, and the experimental cross sections on the right were taken along ω phase (see the text). Fifty-six and 44% of isoleucine are synthesized via the threonine and pyruvate pathways, respectively. The carbon chemical shifts are relative to those of DSS (2,2-dimethyl-2-silapentane-5-sulfonate sodium salt).

FIG. 5. Cross sections taken along ω_1 (¹³C) from the 2D ¹³C,¹H COSY spectrum recorded with the biomass labeled with [¹³C₄]threonine. (A to C) ¹³C—¹³C scalar-coupling fine structures of the precursors used for isoleucine biosynthesis via the threonine pathway (Fig. 2), i.e., with Thr- α , Thr- γ_2 , and Ala-B representing C₃ of pyruvate. (D to F) ¹³C—¹³C fine structures observed at Ile- α , Ile- δ , and Ile- γ_2 (Ile- γ_1 yields the same information as Ile- δ [63]). The fine structures of Ile- α and Ile- δ are dominated by a doublet, which proves that the fragments $C_8 - C_{\gamma1}$ and $C_8 - C_7$ originate from [13C4]threonine. The γ_2 carbon exhibits a ¹³C-1³C fine structure similar to that of Ala-β, which is consistent with the fact that the $C_{\gamma 2}$ — C_B fragments of Ile are derived from pyruvate. The additional small splittings of the doublet components of Ile- α and Ile-δ (D and E) and

which is expected if the fragment C_{ε} — C_{δ} comes entirely from the fragment C_3-C_2 of pyruvate when lysine is synthesized via diaminopimelate dehydrogenase (Fig. 3C). The data in Table 3 show that there are no significant contributions to lysine biosynthesis either from the acetylase/succinylase variant of the diaminopimelate pathway or from the α -aminoadipate pathway. This result is of special interest since simultaneous operation of both variants of the diaminopimelate pathway has

FIG. 6. Lysine biosynthesis. (A) Synthesis via the α -aminoadipate pathway (9, 54); (B) synthesis via the diaminopimelate pathway. In the acetylase/succinylase variant, L- Δ ¹-piperidine-2,6-dicarboxylate is converted to DL-diaminopimelate via the symmetric intermediate LL-diaminopimelate (route 1) (8, 32, 62). In the dehydrogenase variant, L-D¹ -piperidine-2,6-dicarboxylate is directly converted to DL-diaminopimelate by diaminopimelate dehydrogenase (route 2) (42, 43, 74).

	Carbon atom analyzed	Relative abundance of indicated intact carbon fragment ^a at:						
Pathway			Mid-exponential phase		Early stationary phase			
		$f^{(1)}$	f(2)	$f^{(3)}$	f ⁽¹⁾	f(2)	$f^{(3)}$	
Experimental	$Lys-b$ $Lys-\beta$	0.04 0.15	0.96 0.78	0.07	0.05 0.18	0.95 0.77	0.05	
Diaminopimelate-dehydrogenase variant of the diaminopimelate pathway ^c	$Lys-\varepsilon$	0.05	0.95		0.05	0.95		
	$Lys-\beta$	0.16	0.75	0.09	0.18	0.74	0.08	
Acetylase/succinylase variant of the diaminopimelate pathway ^d	$Lys-\varepsilon$	0.29	0.71		0.30	0.70		
	$Lys-\beta$	0.11	0.85	0.04	0.12	0.84	0.04	
α -Aminoadipate ^e	$Lys-\varepsilon$ $Lys-\beta$	0.04 0.52	0.96 0.48	0.00	0.04 0.55	0.96 0.45	0.00	

TABLE 3. Exploration of lysine biosynthesis

^a For the definitions of $f^{(1)}$, $f^{(2)}$, and $f^{(3)}$, see Materials and Methods. All values except the experimental values are expected.
^b Values coincide with those observed for Ala- β .
^c Calculated with equati

 α ^c Calculated with equations 7 and 8.
 α ^d Calculated with equations 9 and 10.

^e Calculated with equations 11 and 12.

previously been documented for the bacterium *Corynebacterium glutamicum* (29, 56).

Threonine cleavage. In the labeling experiment with $[^{13}C_4]$ threonine, the fine structure of Leu- α (representing C₂ of acetyl-CoA [see Fig. 2 in reference 63]) contains a higher proportion of the doublet component than the fine structure of Ala- β (representing C₃ of pyruvate) and a large proportion of ¹³C—¹³C' fragments is also found for Gly- α (Fig. 7). This provides evidence that the exogenously supplied threonine is cleaved into glycine and acetaldehyde, which would be compatible with the assumptions that there is threonine aldolase activity in *H. hispanica* and subsequent conversion of acetaldehyde into acetyl-CoA (69). This indication of threonine aldolase activity in a halophilic archaeon complements data obtained with other organisms: L-threonine aldolases (EC 4.1.2.5) from *Saccharomyces cerevisiae* (*GLY1* [37, 44]), *Escherichia coli* (*ltaE* [39]), and *Pseudomonas* sp. strain NCIMB 10558 (*ltaP* [38]) have been cloned and expressed, and L-threonine aldolase activity has been demonstrated for serine hydroxymethyltransferase (EC 2.1.2.1) in *E. coli* (55), although this enzyme serves primarily for the cleavage of serine to glycine and a C_1 unit. The serine hydroxymethyltransferase from the thermophilic archaeon *Sulfolobus solfataricus* has been shown to possess *allo*-L-threonine aldolase activity (18). Genome sequencing showed that serine hydroxymethyltransferase is also present in *Methanobacterium thermoautotrophicum* (60), in *Methanococcus jannaschii* (12, 57), and in the hyperthermophilic, sulfate-reducing archaeon *Archaeoglobus fulgidus* (34).

Analysis of the 13C-labeling pattern in tyrosine indicates yet unknown biosynthetic pathways. Commonly, erythrose-4 phosphate and phosphoenolpyruvate serve for the biosynthesis of the aromatic ring of tyrosine via the shikimate pathway $(7, 68, 69)$. To evaluate the ¹³C-labeling pattern of the tyrosine ring in light of this pathway, *f* values for this pathway were predicted with the assumption that histidine biosynthesis proceeds via ribose-5-phosphate (68, 69). The pentose pool must then be composed of molecules carrying intact $C_1-C_2-C_3$ and C_4-C_5 fragments (Fig. 1) (see above for *f* values obtained for His), so that erythrose-4-phosphate would be expected to bear a $C_2-C_3-C_4$ fragment from one source molecule and a C_1 carbon from a second glycerol molecule (Fig. 1). Significant deviations between the experimental data and the thus calculated values for a "pure" shikimate pathway (Table 4) cannot be explained within the framework of commonly known pathways. In particular, biosynthesis of the aromatic ring from 2-keto-3-deoxyarabino-heptulosonate-7-phosphate (DAHP), which itself arises from erythrose-4-phosphate and phos-

FIG. 7. Evidence for threonine cleavage in *H. hispanica* derived from the labeling experiment with [¹³C₄]threonine. The ¹³C-¹³C scalar-coupling fine structure of Leu- α (A), which represents C₂ of acetyl-CoA, exhibits a significantly more intense doublet component than that of Ala- β (B), which represents C₃ of pyruvate. A strong doublet is also detected for Gly- α (C). These data indicate that a fraction of the exogenously supplied threonine was cleaved into glycine and acetaldehyde, with subsequent conversion of acetaldehyde into acetyl-CoA (69). The carbon chemical shifts are relative to those of DSS.

TABLE 4. Analysis of aromatic 13C fine structures of tyrosine

	Carbon atom analyzed	Relative abundance of indicated intact carbon fragment ^{a} at:							
Nature of values			Mid-exponential phase		Early stationary phase				
		f ⁽¹⁾	f ⁽²⁾	f(3)	f(1)	f(2)	$f^{(3)}$		
Calculated ^b Observed	Tyr-δ	0.00 0.11	1.00 0.88	0.00 0.01	0.01 0.12	0.99 0.87	0.00 0.01		
Calculated ^c Observed	$Tyr-\epsilon$	0.50 0.30	0.00 0.22 ^d	0.50 0.48	0.50 0.31	0.00 0.21 ^d	0.50 0.48		

a For the definitions of $f^{(1)}$, $f^{(2)}$, and $f^{(3)}$, see Materials and Methods. *b* Calculated with equation 13.

^{*b*} Calculated with equation 13.
^{*c*} Calculated with equation 14.

d Denotes the relative abundance of tyrosine molecules carrying C_{ε} — C_{δ} fragments originating from a single molecule of glycerol.

phoenolpyruvate via DAHP synthase (EC 4.1.2.15), predicts that 50% of the ε carbons must not be connected to carbon atoms that arise from the same source molecule, i.e., $f⁽¹⁾$ {Tyr-ε} equals 0.5 and $f⁽²⁾$ {Tyr-ε} equals 0 (Fig. 8). Hence, the detection of intact $C_8 - C_8$ fragments $(f^{(2)}$ {Tyr-ε} ≈ 0.21) (Table 4) excludes the sole operation of the standard shikimate pathway. This result appears to be in line with data obtained previously for the methanogen *Methanococcus maripaludis* with a different ¹³C-labeling approach, where the standard shikimate pathway (67) could not account for all the observations. (Note that the operation of the oxidative part of the pentose phosphate cycle has not yet been observed for halophilic archaebacteria. In cell extracts of *Haloferax mediterranei* and *Haloarcula vallismortis*, 6-phosphogluconate dehydrogenase (EC 1.1.1.44) was not active at physiological salt concentrations whereas the observed transketolase and transaldolase activities ensured the formation of pentoses from hexoses (50). However, even if pentose biosynthesis in *H. hispanica* occurred via glucose oxidation, the same relative abundances of intact carbon fragments would be predicted for the erythrose-4-phosphate pool.

DISCUSSION

In perspective with current knowledge on amino acid biosynthesis in bacteria and eucarya, the present data on isoleucine and lysine biosynthesis are particularly intriguing and expand knowledge accumulated for other organisms in the domain of the archaea. Thus, the identification of a split threonine/pyruvate pathway for isoleucine biosynthesis in *H. hispanica* is a novel finding, since the threonine pathway has not previously been reported for archaea. In archaebacterial species, such as methanobacteria (21–25, 61) and the thermophilic archaeon *Thermoproteus neutrophilus* (54), the pyruvate pathway is used. In methanogens, isoleucine synthesis can also proceed from propionate or 2-methylbutyrate (23). Notably, the threonine pathway is common in eucaryotic microorganisms and bacteria, although the pyruvate pathway has also been reported for such species (20, 33, 71, 73). However, constitutive, simultaneous operation of the threonine and pyruvate pathways has so far been observed only in a very few organisms, e.g., spirochetes of the genus *Leptospira* (73). In addition, split isoleucine synthesis pathways have been reported for eucaryotic and procaryotic microorganisms such as *Serratia marcescens* (threonine/pyruvate pathways) (33), *E. coli* Crookes and K-12 (threonine/glutamate pathways) (36, 48), *Rhodopseudomonas sphaeroides* (threonine/glutamate pathways) (16), and *S.* *cerevisiae* (threonine/pyruvate pathways or synthesis from homoserine) (71), but with these organisms a genomic mutation or special growth conditions are required to activate routes other than the threonine pathway.

Lysine biosynthesis via the dehydrogenase variant of the diaminopimelate pathway has previously been identified, for example, in *Bacillus sphaericus* (74) and in *Corynebacterium glutamicum* (29, 56, 72), and here we now present direct evidence that this pathway also exists within the domain of the archaea. It has previously been shown that lysine biosynthesis in the methanogens *Methanospirillum hungatei* (22), *Methanococcus voltae* (24), *Methanothrix concilii* (25), and *Methanobacterium thermoautotrophicum* (3) occurs via the diaminopimelate pathway, but the experiments performed in these earlier studies did not allow us to distinguish between the two variant pathways of Fig. 6B. For *Methanobacterium thermoautotrophicum* the enzymes that were assayed, i.e., dihydrodipicolinate synthase (EC 4.2.1.52) and diaminopimelate decarboxylase (EC 4.1.1.20), catalyze reactions that are common to both variants of the pathway. For *Methanospirillum hungatei* (22), *Methanococcus voltae* (24), *Methanococcus jannaschii* (61), and *Methanothrix concilii* (25) , ¹³C-labeling experiments with $[1¹³C]$ acetate, $[2¹³C]$ acetate, or $^{13}CO₂$ were used in biosynthetic studies, but with these specifically labeled precursors both variants of the diaminopimelate pathway (Fig. 6B) yielded the same positional 13 C enrichments in lysine (15). Indications for the operation of the acetylase/succinylase variant in methanogens and *Archaeoglobus fulgidus* were obtained from genome sequencing. The gene for LL-diaminopimelate epimerase (EC 5.1.1.7), which catalyzes the conversion of LLdiaminopimelate to meso-diaminopimelate in the acetylase/ succinylase variant (Fig. 6B), was identified in *Methanobacterium thermoautotrophicum* (60), *Methanococcus jannaschii* (12, 57), and *Archaeoglobus fulgidus* (34), and the succinyl-diaminopimelate desuccinylase gene (EC 3.5.1.18) was found in *Methanococcus jannaschii* (12, 57) and *Archaeoglobus fulgidus* (34). The α -aminoadipate pathway for lysine synthesis (Fig. 6A), which has previously been identified solely for lower eucarya such as fungi, algae, and yeast (9, 70), operates in the

FIG. 8. Scheme indicating the transfer of intact C_2 and C_3 fragments of glycerol via DAHP (2-keto-3-deoxyarabino-heptulosonate-7-phosphate) to the aromatic ring of tyrosine. Thick lines indicate carbon-carbon connectivities arising from a single molecule of glycerol. DAHP was assumed to be synthesized from erythrose-4-phosphate and phosphoenolpyruvate (compare with Fig. 1). One of the ε carbons of tyrosine must then be bound to two carbons that originate from different glycerol molecules. The experimental detection of C^{δ} — C^{ϵ} fragments thus excludes the possibility that the standard shikimate pathway is the only pathway leading to tyrosine biosynthesis in this system (see the text and Table 4). The double line indicates the cyclization site of the Tyr ring. C_1 of DAHP is lost as CO_2 .

thermophilic archaeon *Thermoproteus neutrophilus* (54). Overall, with the new data presented in this paper the implication is that all currently known pathways for lysine biosynthesis in bacteria and eucarya exist also in the domain of archaea.

Organisms living under extreme environmental conditions, such as archaebacteria, have considerable interest for applications in biotechnology (17). Elucidation of their bioreaction networks is thus of interest as a basis for metabolic engineering (2). In this context the present investigation is another telling illustration of the power of biosynthetically directed fractional ¹³C labeling in combination with 2D ¹³C,¹H NMR techniques (58, 65) for support of flux analysis in central metabolism (53, 63, 64, 66). This approach revealed a remarkable diversity of pathways employed for isoleucine and lysine biosynthesis in a halophilic archaeon and showed that with the sole exception of the aromatic ring of tyrosine (and possibly phenylalanine), the proteinogenic amino acids are synthesized according to the common pathways (68, 69). Comparison with the amino acid metabolisms of eucarya and bacteria then also supports the notion that pathways for synthesis of proteinogenic amino acids were probably largely established before the divergence of these three domains (76).

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REFERENCES

- 1. **Abelson, P. H.** 1954. Amino acid biosynthesis in *Escherichia coli*: isotope competition with 14C-glucose. J. Biol. Chem. **206:**335–343.
- Bailey, J. E. 1991. Towards a science of metabolic engineering. Science **252:**1668–1675.
- 3. **Bakhiet, N., F. W. Forney, D. P. Stahly, and L. Daniels.** 1984. Lysine biosynthesis in *Methanobacterium thermoautotrophicum* is by the diaminopimelic acid pathway. Curr. Microbiol. **10:**195–198.
- 4. Bartels, C., T. Xia, M. Billeter, P. Güntert, and K. Wüthrich. 1995. The program XEASY for computer-supported NMR spectral analysis of biological macromolecules. J. Biomol. NMR **6:**1–10.
- 5. **Bartolucci, S., R. Rella, A. Guagliardi, C. A. Raia, A. Gambacorta, M. De Rosa, and M. Rossi.** 1987. Malic enzyme from archaebacterium *Sulfolobus solfataricus*. J. Biol. Chem. **262:**7725–7731.
- 6. **Bax, A., and S. Pochapsky.** 1992. Optimized recording of heteronuclear multidimensional NMR spectra using pulsed field gradients. J. Magnet. Reson. **99:**638–643.
- 7. **Bentley, R.** 1990. The shikimate pathway—a metabolic tree with many branches. Crit. Rev. Biochem. Mol. Biol. **25:**307–384.
- 8. **Berges, D. A., W. E. DeWolf, G. L. Dunn, D. J. Newman, S. J. Schmidt, J. J. Taggart, and C. Gilvarg.** 1986. Studies on the active site of succinyl-CoA: tetrahydrodipicolinate N-succinyltransferase. J. Biol. Chem. **261:**6160–6167.
- 9. **Bhattacharjee, J. K.** 1985. a-Aminoadipate pathway for the biosynthesis of lysine in lower eucaryotes. Crit. Rev. Microbiol. **12:**131–151.
- 10. **Bhaumik, S. R., and H. M. Sonawat.** 1994. Pyruvate metabolism in *Halobac-terium salinarium* studied by intracellular 13C nuclear magnetic resonance spectroscopy. J. Bacteriol. **176:**2172–2176.
- 11. **Bodenhausen, G., and D. Ruben.** 1980. Natural abundance nitrogen-15 NMR by enhanced heteronuclear spectroscopy. Chem. Phys. Lett. **69:**185– 188.
- 12. **Bult, C. J., et al.** 1996. Complete genome sequence of the methanogenic archaeon *Methanococcus jannaschii*. Science **273:**1058–1073.
- 13. **Cazzulo, J. J., and M. C. Vidal.** 1972. Effect of monovalent cations on the malic enzyme from the extreme halophile *Halobacterium cutirubrum*. J. Bacteriol. **109:**437–439.
- 14. **Charon, N. W., R. C. Johnson, and D. Peterson.** 1974. Amino acid biosynthesis in the spirochete *Leptospira*: evidence for a novel pathway of isoleucine biosynthesis. J. Bacteriol. **117:**203–211.
- 15. **Danson, M. J.** 1993. Central metabolism of the archaea, p. 1–24. *In* M. Kates et al. (ed.), The biochemistry of the archaea—1993. Elsevier Science Publishers B.V., Amsterdam, The Netherlands.
- 16. **Datta, P.** 1978. Biosynthesis of amino acids, p. 787–788. *In* R. K. Clayton and W. R. Sistrom (ed.), The photosynthetic bacteria—1978. Plenum Press, New York, N.Y.
- 17. **Davis, M. C.** 1998. Making a living at the extremes. Trends Biotechnol. **16:**102–104.
- 18. **Delle Fratte, S., R. H. White, B. Maras, F. Bossa, and V. Schirch.** 1997.

Purification and properties of serine hydroxymethyltransferase from *Sulfolobus solfataricus*. J. Bacteriol. **179:**7456–7461.

- 19. DeMarco, A., and K. Wüthrich. 1976. Digital filtering with a sinusoidal window function: an alternative technique for resolution enhancement in FT NMR. J. Magnet. Reson. **24:**201–204.
- 20. **Dunstan, R. H., F. R. Whatley, and W. Greenaway.** 1987. Growth of *Paracoccus denitrificans* on [2,3-13C]-succinate and [1,4-13C]-succinate. II. Isoleucine biosynthesis. Proc. R. Soc. Lond. B **231:**349–358.
- 21. **Eikmanns, B., D. Linder, and R. K. Thauer.** 1983. Unusual pathway of isoleucine biosynthesis in *Methanobacterium thermoautotrophicum*. Arch. Microbiol. **136:**111–113.
- 22. **Ekiel, I., I. C. P. Smith, and G. D. Sprott.** 1983. Biosynthetic pathways in *Methanospirillum hungatei* as determined by 13C nuclear magnetic resonance. J. Bacteriol. **156:**316–326.
- 23. **Ekiel, I., I. C. P. Smith, and G. D. Sprott.** 1984. Biosynthesis of isoleucine in methanogenic bacteria: a 13C NMR study. Biochemistry **23:**1683–1687.
- 24. **Ekiel, I., K. F. Jarrell, and G. D. Sprott.** 1985. Amino acid biosynthesis and sodium dependent transport in *Methanococcus voltae*, as revealed by 13C NMR. Eur. J. Biochem. **149:**437–444.
- 25. Ekiel, I., G. D. Sprott, and G. B. Patel. 1985. Acetate and CO₂ assimilation by *Methanothrix concilii*. J. Bacteriol. **162:**905–908.
- 26. **Flavin, M., and A. Segal.** 1964. Purification and properties of the cystathionine g-cleavage enzyme of *Neurospora*. J. Biol. Chem. **239:**2220–2227.
- Gottschalk, G. 1986. Bacterial metabolism, 2nd ed. Springer-Verlag, New York, N.Y.
- 28. Güntert, P., V. Dötsch, G. Wider, and K. Wüthrich. 1992. Processing of multi-dimensional NMR data with the new software PROSA. J. Biomol. NMR **2:**619–629.
- 29. **Ishino, S., K. Yamaguchi, K. Shirahata, and K. Araki.** 1984. Involvement of *meso*-diaminopimelate D-dehydrogenase in lysine biosynthesis in *Corynebacterium glutamicum*. Agric. Biol. Chem. **48:**2557–2560.
- 30. **IUPAC-IUB Commission on Biochemical Nomenclature.** 1970. Abbreviations and symbols for the description of the conformation of polypeptide chains. Biochemistry **9:**3471–3479.
- 31. **Juez, G., F. Rodriguez-Valera, A. Ventosa, and D. J. Kushner.** 1986. *Haloarcula hispanica* spec. nov. and *Haloferax gibbonsii* spec. nov., two new species of extremely halophilic archaebacteria. Syst. Appl. Microbiol. **8:**75–79.
- 32. **Kindler, S. H., and C. Gilvarg.** 1960. N-Succinyl-L-a,ε-diaminopimelate acid deacylase. J. Biol. Chem. **235:**3532–3535.
- 33. **Kisumi, M., S. Komatsubara, and I. Chibata.** 1977. Pathway for isoleucine formation from pyruvate by leucine biosynthetic enzymes in leucine-accumulating isoleucine revertants of *Serratia marcescens*. J. Biochem. **82:**95–103.
- 34. **Klenk, H.-P., et al.** 1997. The complete genome sequence of the hyperthermophilic, sulphate-reducing archaeon *Archaeoglobus fulgidus*. Nature **390:** 364–370.
- 35. **Krivdin, L. B., and E. W. Della.** 1991. Spin-spin coupling constants between carbons separated by more than one bond. Prog. NMR Spectrosc. **23:**301– 610.
LeMaster, D. M., and J. E. Cronan, Jr. 1982. Biosynthetic production of
- ¹³C-labeled amino acids with site-specific enrichment. J. Biol. Chem. 257: 1224–1230.
- 37. **Liu, J. Q., S. Nagata, T. Dairi, H. Misono, S. Shimizu, and H. Yamada.** 1997. The *GLY1* gene of *Saccharomyces cerevisiae* encodes a low-specific L-threonine aldolase that catalyzes cleavage of L-*allo*-threonine and L-threonine to glycine. Expression of the gene in *Escherichia coli* and purification and characterization of the enzyme. Eur. J. Biochem. 245:289-293.
- 38. **Liu, J. Q., S. Ito, T. Dairi, N. Itoh, M. Kataoka, S. Shimizu, and H. Yamada.** 1998. Gene cloning, nucleotide sequencing, and purification and characterization of the low-specificity L-threonine aldolase from *Pseudomonas* sp. strain NCIMB 10558. Appl. Environ. Microbiol. **64:**549–554.
- 39. **Liu, J. Q., T. Dairi, N. Itoh, M. Kataoka, S. Shimizu, and H. Yamada.** 1998. Gene cloning, biochemical characterization and physiological role of a thermostable low-specificity L-threonine aldolase from *Escherichia coli*. Eur. J. Biochem. **255:**220–226.
- 40. **Lodwick, D., H. N. M. Ross, J. A. Walker, J. W. Almond, and W. D. Grant.** 1991. Nucleotide sequence of the 16S ribosomal RNA gene from the haloalkaliphilic archaeon *Natronobacterium magadii*, and the phylogeny of Halobacteria. Syst. Appl. Microbiol. **14:**352–357.
- 41. Marion, D., M. Ikura, R. Tschudin, and A. Bax. 1989. Rapid recording of 2D NMR spectra without phase cycling. Application to the study of hydrogen exchange in proteins. J. Magnet. Reson. **85:**393–399.
- 42. **Misono, H., and K. Soda.** 1980. Properties of meso-a,ε-diaminopimelate dehydrogenase from *Bacillus sphaericus*. J. Biol. Chem. **255:**10599–10605.
- 43. **Misono, H., H. Togawa, T. Yamamoto, and K. Soda.** 1979. Meso-a,ε-diaminopimelate D-dehydrogenase: distribution and the reaction product. J. Bacteriol. **137:**22–27.
- 44. **Monschau, N., K. P. Stahlmann, H. Sahm, J. B. McNeil, and A. L. Bognar.** 1997. Identification of *Saccharomyces cerevisiae GLY1* as a threonine aldolase: a key enzyme in glycine biosynthesis. FEMS Microbiol. Lett. **150:**55–60.
- 45. **Monticello, D. J., R. S. Hadioetomo, and R. N. Costilow.** 1984. Isoleucine synthesis by *Clostridium sporogenes* from propionate or a-methylbutyrate. J. Gen. Microbiol. **130:**309–318.
- 46. **Neidhardt, F. C., R. Curtiss III, J. L. Ingraham, E. C. C. Lin, K. B. Low, B. Magasamik, W. S. Reznikoff, M. Riley, M. Schaechter, and H. E. Umbarger (ed.).** 1996. *Escherichia coli* and *Salmonella*: cellular and molecular biology. 2nd ed. American Society for Microbiology, Washington, D.C.
- 47. Otting, G., and K. Wüthrich. 1988. Efficient purging scheme for protondetected heteronuclear two-dimensional NMR. J. Magnet. Reson. **76:**569– 574. 48. **Phillips, A. T., J. I. Nuss, J. Moosic, and C. Foshay.** 1972. Alternate pathway
- for isoleucine biosynthesis in *Escherichia coli*. J. Bacteriol. **109:**714–719.
- 49. **Rangaswamy, V., and W. Altekar.** 1994. Ketohexokinase (ATP:D-fructose 1-phosphotransferase) from a halophilic archaebacterium, *Haloarcula vallismortis*: purification and properties. J. Bacteriol. **176:**5505–5512.
- 50. **Rawal, N., S. M. Kelkar, and W. Altekar.** 1988. Alternative routes of carbohydrate metabolism in halophilic archaebacteria. Indian J. Biochem. Biophys. **25:**674–686.
- 51. **Robinson, J. M., and M. J. Allison.** 1969. Isoleucine biosynthesis from 2-methylbutyric acid by anaerobic bacteria from the rumen. J. Bacteriol. **97:**1220–1226.
- 52. **Sauer, F. D., J. D. Erfle, and S. Mahadevan.** 1975. Amino acid biosynthesis in mixed rumen cultures. Biochem. J. **150:**357–372.
- 53. **Sauer, U., V. Hatzimanikatis, J. E. Bailey, M. Hochuli, T. Szyperski, and K.** Wüthrich. 1997. Metabolic fluxes in riboflavin-producing *Bacillus subtilis*. Nat. Biotechnol. **15:**448–452.
- 54. Schäfer, S., T. Paalme, R. Vilu, and G. Fuchs. 1989. ¹³C-NMR study of acetate assimilation in *Thermoproteus neutrophilus*. Eur. J. Biochem. **186:** 695–700.
- 55. **Schirch, V., S. Hopkins, E. Villar, and S. Angelaccio.** 1985. Serine hydroxymethyltransferase from *Escherichia coli*: purification and properties. J. Bacteriol. **163:**1–7.
- 56. Schrumpf, B., A. Schwarzer, J. Kalinowski, A. Pühler, L. Eggeling, and H. **Sahm.** 1991. A functionally split pathway for lysine synthesis in *Corynebacterium glutamicum*. J. Bacteriol. **173:**4510–4516.
- 57. **Selkov, E., N. Maltsev, G. J. Olsen, R. Overbeek, and W. B. Whitman.** 1997. A reconstruction of the metabolism of *Methanococcus jannaschii* from sequence data. Gene **197:**11–26.
- 58. Senn, H., B. Werner, B. A. Messerle, C. Weber, R. Traber, and K. Wüthrich. 1989. Stereospecific assignment of the methyl ¹ H-NMR lines of valine and leucine in polypeptides by nonrandom 13C labeling. FEBS Lett. **249:**113–118.
- 59. **Shaka, A. J., P. B. Barker, and R. Freeman.** 1985. Computer-optimized decoupling scheme for wideband applications and low-level operation. J. Magnet. Reson. **64:**547–552.
- 60. **Smith, D. R., L. A. Doucette-Stamm, C. Deloughery, H. Lee, J. Dubois, T. Aldredge, R. Bashirzadeh, D. Blakely, R. Cook, K. Gilbert, D. Harrison, L. Hoang, P. Keagle, W. Lumm, B. Pothier, D. Qiu, R. Spadafora, R. Vicaire, Y. Wang, J. Wierzbowski, R. Gibson, N. Jiwani, A. Caruso, D. Bush, H. Safer, D. Patwell, S. Prabhakar, S. McDougall, G. Shimer, A. Goyal, S. Pietro**kovski, G. M. Church, C. J. Daniels, J.-I. Mao, P. Rice, J. Nölling, and J. N. **Reeve.** 1997. Complete genome sequence of *Methanobacterium thermoautotrophicum* ΔH : functional analysis and comparative genomics. J. Bacteriol. **179:**7135–7155.
- 61. **Sprott, G. D., I. Ekiel, and G. B. Patel.** 1993. Metabolic pathways in *Methanococcus jannaschii* and other methanogenic bacteria. Appl. Environ. Microbiol. **59:**1092–1098.
- 62. **Sundharadas, G., and C. Gilvarg.** 1967. Biosynthesis of a,ε-diaminopimelic acid in *Bacillus megaterium*. J. Biol. Chem. **242:**3983–3988.
- 63. **Szyperski, T.** 1995. Biosynthetically directed fractional 13C-labeling of proteinogenic amino acids. An efficient tool to investigate intermediary metabolism. Eur. J. Biochem. **232:**433–448.
- 64. **Szyperski, T.** 1998. 13C-NMR, MS and metabolic flux balancing in biotechnology research. Q. Rev. Biophys. **31:**41–106.
- 65. Szyperski, T., D. Neri, B. Leiting, G. Otting, and K. Wüthrich. 1992. Support of ¹ H-NMR assignments in proteins by biosynthetically directed fractional 13C-labeling. J. Biomol. NMR **2:**323–334.
- 66. **Szyperski, T., J. E. Bailey, and K. Wüthrich.** 1996. Detecting and dissecting metabolic fluxes using biosynthetic fractional ¹³C labeling and two-dimensional NMR spectroscopy. Trends Biotechnol. **14:**453–459.
- 67. **Tumbula, D. L., Q. Teng, M. G. Bartlett, and W. B. Whitman.** 1997. Ribose biosynthesis and evidence for an alternative first step in the common aromatic amino acid pathway in *Methanococcus maripaludis*. J. Bacteriol. **179:** 6010–6013.
- 68. **Umbarger, H. E.** 1978. Amino acid biosynthesis and its regulation. Annu. Rev. Biochem. **47:**533–606.
- 69. **Voet, D., and J. G. Voet.** 1995. Amino acid metabolism, p. 727–784. *In* Biochemistry, 2nd ed. John Wiley & Sons, Inc., New York, N.Y.
- 70. **Vogel, H. J.** 1960. Two modes of lysine synthesis among lower fungi: evolutionary significance. Biochim. Biophys. Acta **41:**172–174.
- 71. **Vollbrecht, D.** 1974. Three pathways of isoleucine biosynthesis in mutant strains of *Saccharomyces cerevisiae*. Biochim. Biophys. Acta **362:**382–389.
- 72. **Wehrmann, A., B. Phillipp, H. Sahm, and L. Eggeling.** 1998. Different modes of diaminopimelate synthesis and their role in cell wall integrity: a study with *Corynebacterium glutamicum*. J. Bacteriol. **180:**3159–3165.
- 73. **Westfall, H. N., N. W. Charon, and D. E. Peterson.** 1983. Multiple pathways for isoleucine biosynthesis in the spirochete *Leptospira*. J. Bacteriol. **154:**846– 853.
- 74. **White, P. J.** 1983. The essential role of diaminopimelate dehydrogenase in the biosynthesis of lysine by *Bacillus sphaericus*. J. Gen. Microbiol. **129:**739– 749.
- 75. Wider, G., and K. Wüthrich. 1993. A simple experimental scheme using pulsed field gradients for coherence pathway rejection and solvent suppression in phase-sensitive heteronuclear correlation spectra. J. Magnet. Reson. **102:**239–241.
- 76. **Woese, C. R., O. Kandler, and M. L. Wheelis.** 1990. Towards a natural system of organisms: proposal for the domains Archaea, Bacteria, and Eucarya. Proc. Natl. Acad. Sci. USA **87:**4576–4579.
- 77. Wüthrich, K., T. Szyperski, B. Leiting, and G. Otting. 1992. Biosynthetic pathways of the common proteinogenic amino acids investigated by fractional 13C-labeling and NMR spectroscopy, p. 41–48. *In* K. Takai (ed.), Frontiers and new horizons in amino acid research: proceedings of the First Biennial International Conference on Amino Acid Research, Frontiers and New Horizons. Elsevier, Amsterdam, The Netherlands.