



Corrigendum to ‘Acid–base balance: a review of normal physiology’ [BJA Education 22 (2022) 396–401]

I. Shaw^{1,*} and K. Gregory²

¹Sheffield Teaching Hospitals NHS Trust, Sheffield, UK and ²University Hospitals of Birmingham NHS Foundation Trust, Birmingham, UK

*Corresponding author: ian.shaw8@nhs.net

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The authors regret that an error was present in the formulation of Henderson Hasselbach equation on page 398. The correct equation appears below:

Taking *negative* logarithms

$$\text{pH} = \text{pKa} + \log_{10} [\text{A}^-]/[\text{HA}] \text{ (Henderson-Hasselbach equation)}$$

There was also an error in the equation describing the application of the Henderson-Hasselbach equation to the bicarbonate buffer system. The correct equation appears below:

$$\text{pH} = \text{pKa} + \log [\text{HCO}_3^-]/[\text{CO}_2] \text{ or } \text{pH} = \text{pKa} + \log [\text{HCO}_3^-]/[\alpha\text{PCO}_2]$$

where α is the solubility of CO_2 in blood; using Henry's Law

$$[\text{H}_2\text{CO}_3] = \alpha\text{PCO}_2$$

The authors would like to apologise for any inconvenience caused.