

Food grade nanoemulsion development to control food spoilage microorganisms on bread surface

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Abstract In this study, the effect of emulsifier mixture and their concentrations on the development of nanoemulsion was studied. The impact of sonication and microfluidization processing conditions on the physicochemical properties and in vitro antimicrobial activity was also evaluated. The optimal nanoemulsion formulation was then evaluated on bread surface against *B. subtilis*. Results showed that a hydrophilic-lipophilic balance HLB = 12 and emulsifier: oil ratio of 1:1 allowed the formation of stable nanoemulsion. Also, both microfluidization and sonication allowed the formation of nanoscale-emulsion. Sonication treatment for 10 min allowed to maintain the total flavonoid content and a slight reduction of total phenol content. Furthermore, employing sonication resulted to the lowest polydispersity index suggesting more stable nanoemulsion. Nanoscale-emulsion showed a good in vitro antimicrobial activity against *L. monocytogenes* and *E. coli*. The application of nanoemulsion on bread surface inoculated with *B. subtilis* showed a delay of the decay.

Keywords Food safety · Antimicrobial activity · Nanoemulsion · Stability · Encapsulation

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Abbreviations

ANOVA	Analysis of variance
<i>A. flavus</i>	<i>Aspergillus flavus</i>
<i>B. subtilis</i>	<i>Bacillus subtilis</i>
CE	Citrus extracts
CFU	Colony-forming unit
<i>E. coli</i>	<i>Escherichia coli</i>
EO	Essential oil
FCR	Folin–Ciocalteu reagent
GAE	Gallic acid equivalents
HLB	Hydrophilic and lipophilic balance
<i>L. monocytogenes</i>	<i>Listeria monocytogenes</i>
MHB	Mueller–Hinton broth
MIC	Minimal inhibitory concentration
PDI	Polydispersity index
SD	Standard deviation
SDS	Sodium dodecyl sulfate
TSA	Tryptic soy agar

Introduction

Archaeological evidence and recorded history have both shown that the use of aromatic plants may be dated to 10,000 BC. The most important document containing 700 formulations of aromatic plants is so-called “Papyrus Ebers” (Baser and Buchbauer 2009). Egyptians exploited the virtues of plants to cure diseases and relieve pain and discomfort and also in food as spices, preservatives, and flavorings as well. Nevertheless, their use in food as preservatives is still limited because of their hydrophobicity which reduces their bioavailability on food surface as well as their strong aroma and toxicity at high concentrations (Ait-Ouazzou et al. 2011; Dima and Dima 2015). Nowadays, with the growing interest towards natural products a real trend of change, towards

natural additives like antimicrobials and aromas of natural origin is emerging in the food industry (Aschemann-Witzel et al. 2019). To maintain their efficiency in food, natural antimicrobial agents should be incorporated in an appropriate matrix maintaining their stability and homogeneity.

Nanoemulsions may be applied as effective encapsulation systems to stabilize hydrophobic natural antimicrobials and act as carriers or delivery systems for lipophilic compounds (Choradiya and Patil 2021). Generally, nanoemulsions are highly stable against gravitational separation and droplet aggregation (de Oca-Ávalos et al. 2017). However, it is important to use the adequate emulsifiers and the emulsification methods. Emulsifiers are able to be adsorbed at the interface between the hydrophilic and hydrophobic phases decreasing the interfacial tension and preventing or slowing down the aggregation of particles of the dispersed phase by increasing repulsion forces between them (Silva et al. 2012). The preparation of nanoemulsion consists mainly on low or high energy approaches (Suyanto et al. 2019). Low energy methods are based on the selection of favorable interfacial properties and require significantly less energy input (Badrudodoza et al. 2018). High energy includes several methods such as sonication and high-pressure homogenization (ie. microfluidization). These techniques have the advantages of (i) producing a smaller particle size with a narrower distribution, (ii) easy to extrapolate on a large scale and (iii) reproducible (Yukuyama et al. 2016).

The oil-in-water (o/w) emulsions have the advantage of improving the solubility of hydrophobic or oil-soluble compounds and of masking the unpleasant taste and odor of the active compounds. It can also improve the antimicrobial activity (Lohith Kumar and Sarkar 2017). Previous studies have shown that reducing the particle size of lemongrass essential oil and alginate emulsion by microfluidization improves its antimicrobial activity against *E. coli* (Salvia-Trujillo et al. 2014). Ghosh et al. (2014) have developed a 3% eugenol-loaded o/w emulsion composed of sesame oil, Tween 20/Tween 80 by an ultrasound cavitation method to increase the shelf-life of fresh orange juice. This improvement in the bioactivity was related to a faster penetration of the nanoemulsion to the microbial membranes due to the increased of contact surface (Salvia-Trujillo et al. 2014). This would allow reducing the concentration to reach an equivalent or even greater antibacterial effect as compared to conventional emulsions (Odriozola-Serrano et al. 2014).

Bread and other bakery products are subjected to various spoilage problems, physical, chemical, and microbial. The latter is the most serious one particularly in terms of bacterial (*Bacillus* sp.) growth. *Bacillus subtilis* is the major problem of the bread quality deterioration (Rahman et al. 2022). This microorganism contaminates raw materials (for example, flour, sugar, and yeast) and is able to survive during baking process and cooling. *Bacillus* spores are able to

survive a heat treatment (100 °C, 10 min) which corresponds to the cooking process (Rosenkvist and Hansen 1995). It also has the capability of development under both aerobic and anaerobic conditions. *Bacillus subtilis* is responsible for the phenomenon of "ropiness" (Rahman et al. 2022). *Bacillus* spp. could contaminate both non-acidified, white and wholemeal bread (Pacher et al. 2022). However, the wholemeal breads seem to be slightly more resistant to *Bacillus* contamination (Rosenkvist and Hansen 1995).

The aim of this study is to evaluate the effect of the stabilization of an antimicrobial formulation based on natural antimicrobial ingredients into nanoemulsion using microfluidization and sonication method on its bioactivity. The nanoemulsion composition was firstly optimized by selecting the best hydrophilic and lipophilic balance (HLB) and oil: emulsifier ratio. Then, the evaluation of the effect of sonication and microfluidization on the physicochemical and antimicrobial activity of the nanoemulsions was carried out. Finally, the effect of nanoemulsion on the decay of white bread as a food model inoculated with *B. subtilis* was evaluated.

Materials and methods

Material

Cinnamon essential oil (EO) was bought from Bio Lonreco, Inc. (Dorval, QC, Canada). Water soluble citrus extracts (CE) was provided by Kerry Inc. (Beloit, WI, USA). Based on our previous study, these two compounds have a very high antibacterial activity (Ben-Fadhel et al. 2019). Their main composition is presented in the table (Online Resource 1). Sucrose monopalmitate and sunflower lecithin were kindly provided by Compass Foods Company (Singapore). Folin–Ciocalteu reagent was purchased from Sigma-Aldrich Canada Co. (Oakville, ON, Canada). Sodium carbonate was purchased by Laboratoire Mat Inc. (Quebec, QC, Canada).

Preparation of nanoemulsions

The oil in water (o/w) emulsion containing cinnamaldehyde-based EO and CE with a 1:6 (w/w) cinnamon EO:CE ratio was optimized at different ratios of emulsifier: oil phase (0.3–1.25) and various HLB values (7–18) by exploiting lecithin (HLB ~ 7) and sucrose monopalmitate (HLB ~ 18) as emulsifier. The mixture was vigorously magnetically stirred then passed through Ultra-Turrax T25 high-shear homogenizer (IKA Works Inc., Wilmington, NC, USA) at 10,000 rpm during 1 min. The resulting coarse emulsion was subjected to sonication or microfluidization treatment optimization at different amplitudes, duration, pressures and

cycles to obtain the homogeneous nanoemulsions following the method of Maherani et al. (2018).

Sonication method

The coarse emulsion was subjected to sonicator processor Qsonica Q500 (Fisher Scientific Ltd, Saint-Laurent, QC, Canada) in an ice bath at 70% of full power for 10, 20 and 30 min (on-time 5 s, off-time 2 s) to obtain a homogeneous nanoemulsion (Maherani et al. 2018). Every 5 min, the ice bath was renewed to maintain constant temperature during treatment.

Microfluidization method

The coarse emulsion was treated by microfluidization, using an electric-hydraulic M-110P Microfluidizer[®] equipped with a diamond interaction chamber F20Y for emulsions downstream (Microfluidics International Corp., Newton, MA, USA). Optimization was performed at pressures of 69 MPa (10,000 psi), 103.4 MPa (15,000 psi) and 138.9 MPa (20,000 psi) in 1, 2, and 3 cycles (at 25 °C) (Maherani et al. 2018). Upon exiting the interaction chamber, the product flows through an external cooling coil which regulates the nanoemulsion temperature.

Size (z-average) and polydispersity index (PDI)

The mean diameter and size distribution of emulsions were determined using dynamic light scattering (DLS) technique by employing a Zetasizer Nano-ZS (ZEN3600; Malvern Instruments Inc., Westborough, MA, USA) and DTS Nano software (version 6.12). To avoid multiple scattering effects, emulsions were diluted (1:50) in Milli-Q water, and then put into a folded capillary cell DTS1060 equipped with gold electrodes (Malvern Instruments Inc.). All measurements were carried out at 20 °C by considering a medium viscosity of 1.33 and medium refractive index of 1.333. Three measurements (n = 3) were carried out for each sample.

Turbidity (τ)

Optical turbidity of the emulsions was determined by measuring their absorbance at 600 nm using a photodiode array UV–Vis spectrophotometer Scinco S-3100 (Betatek Inc., Toronto, ON, Canada) at room temperature (Maherani et al. 2018). Samples were analyzed within cells of 1 cm optical path, and deionised water was used as a blank. Triplicate measurements (n = 3) of turbidity were carried out for each sample.

Total phenol and total flavonoid content

Total phenol and total flavonoid content were determined in order to evaluate the effect of microfluidization and sonication treatment on their degradation.

Total phenol content was determined using a Folin–Ciocalteu colorimetric method according to Dewanto et al. (2002) with a standard curve ranging between 0 and 200 μg of gallic acid/mL. A quantity of 125 μL of the standard gallic acid solution or emulsions was mixed with 0.5 mL of distilled water in a test tube followed by the addition of 125 μL of Folin–Ciocalteu reagent (FCR). The vortexed samples were kept at room temperature for 6 min and then 1.25 mL of a 7% sodium carbonate aqueous solution was added to the mixture and the final volume was adjusted to 3 mL by adding water. Samples were allowed to stand for 90 min at room temperature before measurement at 760 nm versus the blank prepared similarly with water. All mean values were expressed as mg gallic acid equivalents (GAE)/g of emulsion.

Total flavonoid content was determined by using a colorimetric method (Dewanto et al. 2002). Briefly, 0.25 mL of diluted emulsions was mixed with 1.25 mL of distilled water followed by addition of 75 μL of a 5% NaNO_2 solution. After 6 min, 150 μL of a 10% $\text{AlCl}_3 \cdot 6 \text{H}_2\text{O}$ solution was added and allowed to stand for 5 min at room temperature before adding 0.5 mL of 1 M NaOH. The mixture was brought to 2.5 mL with distilled water and mixed well. The absorbance was measured immediately against the blank at 510 nm in comparison with the standards prepared similarly with known (+)-catechin concentrations. The mean results are expressed as μg catechin equivalents/g of emulsion.

Effect of the preparation method on the antimicrobial properties of emulsions

Preparation of bacterial cultures

Emulsions were evaluated for their antimicrobial activity against 4 microorganisms. *Listeria monocytogenes* HPB 2812 was provided by Laboratoire de Santé Publique du Québec (Sainte-Anne-de-Bellevue, QC, Canada). *Escherichia coli* O157:H7 CDC EDL 933 (ATCC 43,895) was provided by Prof. Charles Dozois (INRS-IAF, Canada), *Bacillus subtilis* 168 (ATCC 23,857) and *Aspergillus flavus* PDCS-4 (ATCC 26,771) were purchased from Cedarlane Laboratories (Burlington, On, Canada). For bacterial evaluation, *L. monocytogenes*, *E. coli* and *B. subtilis* were kept at -80 °C in Tryptic Soy Broth (TSB; BD Difco, Fisher Scientific Ltd) containing glycerol (10%; v/v). Before each experiment, stock cultures were propagated through two consecutive 24-h growth cycles (10^{-1} dilution) at 35 ± 2 °C in TSB. For fungal evaluation, *A. flavus* were propagated through

72-h growth cycles in potato dextrose agar (PDA; BD Difco, Fisher Scientific Ltd) at $28\text{ }^{\circ}\text{C} \pm 2\text{ }^{\circ}\text{C}$. Conidia were isolated from the agar media using sterile platinum loop, suspended in sterile peptone water, and filtered through sterile cell strainer (Fisher scientific Ltd). The filtrate was adjusted using a hemocytometer to 10^4 conidia/mL for the MIC determination.

Minimal inhibitory concentration (MIC) determination

The MIC value of emulsions was determined in sterilized flat-bottomed 96-well microplate according to the two fold microdilution modified method of Ben-Fadhel et al. (2019). Briefly, serial two-fold dilutions of the antimicrobial compounds were made in Mueller Hinton Broth (MHB, BD Difco, Fisher Scientific Ltd) and dispensed into 96-well microplates. The concentrations ranges of emulsions were from 0.005 to 5% (w/v). Then, a volume of 100 μL of bacteria and fungi suspension (10^4 CFU/mL or conidia/mL) was added to 100 μL of each antimicrobial serial dilution. In the blank or negative control, 100 μL of distilled water was used instead of the working culture bacteria and fungi. The positive control (without antimicrobial agent) consisted of 100 μL of MHB and 100 μL of working culture bacteria or fungi. It should be mentioned that the sterile emulsifiers solution (0.55% w/v lecithin and 0.45% w/v sucrose monopalmitate) that used for emulsification of cinnamom EO did not have any antibacterial activity. Microplates were then incubated at $37\text{ }^{\circ}\text{C}$ and $28\text{ }^{\circ}\text{C}$ for 24 h and 48 h, respectively. The absorbance was measured at 595 nm in a BioTek ELx800 absorbance microplate reader (BioTek Instruments Inc., Winooski, VT, USA). The MIC is the lowest concentration of antimicrobial agent demonstrating the complete inhibition of bacterial and fungal growth as evidenced by a lack of increase in absorbance.

In situ evaluation

Bacterial and spores' preparation *B. subtilis* was selected for the evaluation of the effectiveness of the antimicrobial treatments. This microorganism was chosen as it has been determined to be the main spoilage agent of bread and it demonstrated an ability to survive to food processing conditions (frying, cooking, baking) (Rahman et al. 2022; Almada-Érix et al. 2021). To induce sporulation, nutrient agar (NA) supplemented with MnSO_4 (10 mg/L) and K_2HPO_4 (2 g/L) was inoculated with 2 mL of 24-h grown culture of *B. subtilis* in TSB at $30\text{ }^{\circ}\text{C}$. Five days later, spores were collected by flooding the agar plate with sterile distilled water and recovered by scratching the surface with a glass spatula. After harvesting, spores were washed four times with saline water (0.85% w/v) by centrifugation at $4400 \times g$ for 15 min at $4\text{ }^{\circ}\text{C}$ and the pelleted cells were re-suspended in sterile

double distilled water. Spore suspension was heat treated in a water bath at $80\text{ }^{\circ}\text{C}$ for 10 min to kill remaining vegetative cells. The concentration of spore suspension was estimated by spread plating 100 μL on plates of tryptic soy agar (TSA; BD Difco, Fisher Scientific Ltd), which were incubated at $30\text{ }^{\circ}\text{C}$ for 24 h. The spore suspension was then maintained at $4\text{ }^{\circ}\text{C}$ until used (Ayari et al. 2012).

Challenge test of par-baked bread Non-baked white bread balls ($\approx 10 \times 7$ cm) were purchased from Costco (Laval, QC, Canada). Preformed dough balls were treated with 20 μL of *B. subtilis* at 10^4 CFU/mL on 4 points of bread (5 μL each). Bread was then divided into groups: untreated bread, bread treated with nanoemulsion and bread treated with coarse emulsion. Treatment was applied by spraying treating solutions on each surface of bread during 2 s. For the control samples, bread was sprayed with distilled water. Bread was then baked in convection oven for 14 min at $200\text{ }^{\circ}\text{C}$ (10 breads per treatment). All bread balls were individually packed in Nasco Whirl-Pak™ sterile filter bags (Fisher Scientific Ltd) under atmospheric air and placed at room temperature. Bread balls were evaluated for the surface contamination (presence of rope formation) and the results were expressed as percentage of decay over time.

Statistical analysis

Each experiment was done in triplicate ($n=3$). Analysis of variance (ANOVA), Duncan's multiple range tests for equal variances and Tamhane's test for unequal variances were performed for statistical analysis using PASW Statistics 18 software (IBM Corporation, Somers, NY, USA). Differences between means were considered significant when the confidence interval was lower than 5% ($p \leq 0.05$).

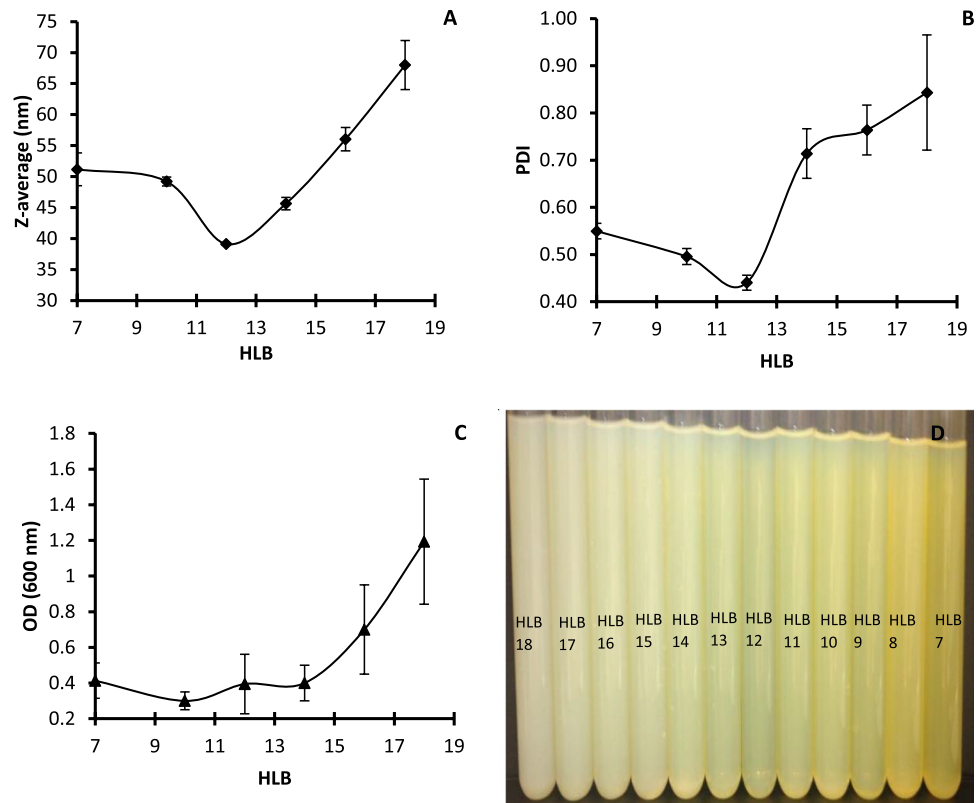
Results and discussion

Preparation of nanoemulsions

Effect of the HLB on the quality of emulsions

In order to prepare a stable nanoemulsion showing low PDI and mean size average, lecithin (HLB = 7) was combined with sucrose monopalmitate (HLB = 18) to prepare nanoemulsions with different HLB values. Nanoemulsions were prepared using 1 cycle of microfluidization treatment at a pressure of 15,000 psi. Results of mean size average, PDI, turbidity and appearance of emulsions were respectively detailed in Fig. 1. Results show that, z-average and PDI (Fig. 1A, B) had similar parabolic behavior showing a minimum z-average and PDI of 39.11 nm and 0.44 respectively for HLB 12 (ratio of lecithin to sucrose

Fig. 1 Effect of HLB on **A** the particle size, **B** the PDI, **C** the turbidity and **D** the appearance of nanoemulsions prepared by microfluidization at 15,000 psi/1 cycle with an emulsifier: oil ratio of 1:1 (w/w)



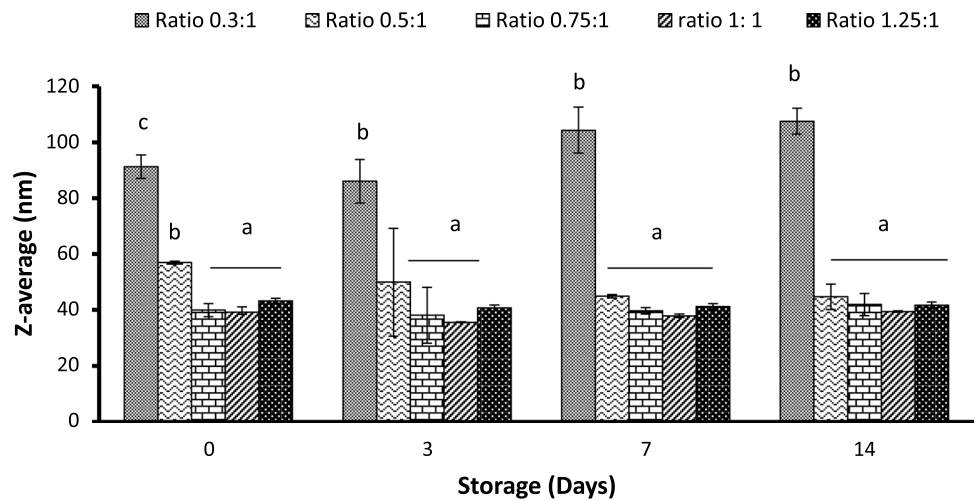
monopalmitate = 55:45). Afterwards, z-average and PDI increased significantly and reached 68 nm and 0.84 suggesting a non-stable nanoemulsion. Reducing the droplet size is very important factor in nanoemulsion preparation because it can result in higher retention of encapsulated components in emulsion systems (Lu et al. 2018). On the other hand, PDI value ($0 < \text{PDI} < 1$) indicates the uniformity and the stability of the droplet size distribution in the nanoemulsion. When the $\text{PDI} \leq 0.3$, the nanoemulsion system has narrow size distribution and it is homogeneous. When the $\text{PDI} \geq 0.5$, the system is heterogeneous and called broad size distribution (Pongsumpun et al. 2020). Generally, large particle size and size distribution indicate the instability of the nanoemulsion.

The turbidity and the color of nanoemulsion (Fig. 1C, D) seem to correlate with the data of droplet size and PDI. As turbidity started to increase from 0.2 (HLB = 8) to a maximum at HLB 18 with OD_{600} of 1.2. Similar results were also observed by Carpenter and Saharan (2017) who prepared mustard o/w nanoemulsion using span 80 and tween 80 and observed a minimum droplet size with HLB 10–11. They attributed this to the packing strength of emulsifiers at the droplet interface which had reached its saturation state at HLB 10. At higher HLB, sucrose monopalmitate molecules are present in excess and therefore will occupy maximum active sites. Lu et al. (2018) also demonstrated that emulsions appear to be opaque (white) at low HLB values and are transparent or translucent at HLB 12.

Effect of the emulsifier: oil phase ratio on the quality of emulsions

By keeping HLB 12 constant, the emulsifier: oil ratios varied successively from 0.3:1, 0.5:1, 0.75:1, 1:1 to 1.25:1. Results (Fig. 2) showed that the concentration of emulsifiers significantly affects the z-average of nanoemulsion ($p \leq 0.05$). So that it adapted 91.3 nm on day 0 as compared to 56.9 nm for 0.5:1 ratio, 43.2 nm for 1:1 ratio and 39.1 nm for both ratios of 0.75:1 and 1.25:1. In the study of Lu et al. (2018), the ratio emulsifier: citral of 0.4–0.6 showed the lowest droplet size. During storage, an increase of z-average was observed for the 0.3:1 ratio and the z-average reached 107.6 nm on day 15. However, the z-average was not affected by storage time for higher emulsifier: oil ratios and the resulting emulsion was found to be stable. Bai et al. (2016) also showed that the type of emulsifier and its concentration significantly affect the mean droplet diameter and the interfacial tension. Depending on the emulsifier type, soy lecithin and gum Arabic showed the lowest droplet diameter at a ratio of 1:1 while the ratio of 0.1:1 was enough in combination with WPI to reduce droplet diameter. Qian and McClements (2011) found a minimum droplet diameter dependent on emulsifier type as $\text{SDS} < \text{Tween 20} < \beta\text{-lactoglobulin} < \text{casein}$. They also found a decrease in mean droplet diameter with increasing emulsifier concentration explained by the presence of the emulsifier to cover any new droplet surfaces formed during

Fig. 2 Effect of the emulsifier: oil ratio on the z-average of nanoemulsions



homogenization. However, the reverse trend was observed for β -lactoglobulin attributed to denaturation of proteins during the high-pressure homogenization process.

The decrease of droplet size with increasing emulsifier: oil ratio can be explained by the presence of enough surfactant to stabilize newly formed droplets (Lu et al. 2018). However, by increasing more the surfactant ratio, droplet size would increase again due to much residual surfactant which interferes with the stability and appearance of emulsion.

Effect of the preparation method on the quality and stability of emulsions

Effect of the preparation method on the particle size, PDI and turbidity of emulsions

Results of the effect of microfluidization and sonication on stability parameters including z-average, PDI and turbidity of nanoemulsion are detailed in Table 1. Results show that both methods affect significantly ($p \leq 0.05$) the z-average and the turbidity showing a z-average reduction from 189.5 nm for coarse emulsions to 45.9, 38.6 and 41.1 nm for emulsions treated by microfluidization after a pass under the pressure of 10,000, 15,000 and 20,000 psi, respectively. These values were measured as 34.1, 27.4 and 27.6 nm for emulsions treated by sonication during 10, 20 and 30 min respectively. Likewise, turbidity also diminished under the effect of the applied treatments as it decreased to ≈ 4.5 for microfluidization treatment and < 1.1 for sonication treatment as compared to ≈ 9.6 for the coarse emulsion. These results suggest that both methods were efficient to reduce droplet mean diameter and turbidity. However, sonication method seems to be the most efficient especially for reducing the turbidity and the PDI although PDI was not as sensitive as droplet size parameter to the applied treatments. The lowest PDI

Table 1 Z-average, PDI and turbidity of emulsions prepared by sonication versus microfluidization

Treatments	Z-average (nm)	PDI	Turbidity
Coarse emulsion	189.5 \pm 28.3 ^e	0.399 \pm 0.044 ^c	9.56 \pm 0.5 ^a
<i>Microfluidization</i>			
10,000 psi			
1 cycle	45.9 \pm 0.6 ^d	0.430 \pm 0.008 ^c	4.54 \pm 0.35 ^b
2 cycles	35.3 \pm 0.1 ^c	0.404 \pm 0.002 ^c	3.76 \pm 0.47 ^c
3 cycles	28.3 \pm 0.7 ^a	0.321 \pm 0.006 ^{ab}	2.6 \pm 0.46 ^d
15,000 psi			
1 cycle	38.6 \pm 0.2 ^d	0.425 \pm 0.006 ^c	4.39 \pm 0.81 ^b
2 cycles	28.0 \pm 0.4 ^a	0.331 \pm 0.010 ^{ab}	1.84 \pm 0.04 ^e
3 cycles	28.6 \pm 0.4 ^a	0.332 \pm 0.009 ^{ab}	1.39 \pm 0.22 ^f
20,000 psi			
1 cycle	42.3 \pm 0.6 ^d	0.488 \pm 0.73 ^d	4.72 \pm 0.01 ^b
2 cycles	27.2 \pm 0.1 ^a	0.340 \pm 0.014 ^b	2.25 \pm 0.10 ^{de}
3 cycles	25.7 \pm 0.7 ^a	0.313 \pm 0.007 ^{ab}	2.64 \pm 0.29 ^d
<i>Sonication</i>			
10 min	34.1 \pm 0.4 ^b	0.283 \pm 0.024 ^a	1.05 \pm 0.11 ^g
20 min	27.4 \pm 0.4 ^a	0.304 \pm 0.011 ^{ab}	0.43 \pm 0.12 ^h
30 min	27.6 \pm 0.5 ^a	0.350 \pm 0.017 ^b	0.56 \pm 0.05 ^h

Within each column, means with the same lowercase letter are not significantly different ($p > 0.05$)

value of 0.283 was obtained by treatment with sonication during 10 min. Unexpectedly, a longer sonication treatment increased the PDI. This phenomenon can be referred to an “over-processing” with an increase in the Brownian motion at higher energy input leading to collision and coalescence (Mahdi Jafari et al. 2006; Lu et al. 2018). Regarding microfluidization method, increasing microfluidization pressure did not have any effect on the PDI. However, it decreased with the number of cycles applied and remained > 0.3 suggesting a non-stable nanoemulsion.

Effect of the preparation method on the total phenol and total flavonoid contents of emulsions

Results of total phenol content (Fig. 3A) showed that microfluidization method did not reduce significantly total phenol content of the emulsion ($p > 0.05$). In contrast, sonication treatment, even for 10 min, reduced significantly ($p \leq 0.05$) total phenol content of nanoemulsion as compared to the coarse emulsion. These results confirmed the findings of Maherani et al. (2018) that sonication method was able to decrease total phenol content especially during storage. Such diminishment is due to polymerization and oxidation of phenolic compounds and subsequently the loss of their activity and solubility. Thus, total phenol seems to be more sensitive to sonication treatment than microfluidization.

Results of total flavonoids (Fig. 3B) showed that both 10-min-sonication and microfluidization treatments did not have a significant impact on the total flavonoid contents of

prepared nanoemulsion as compared to the coarse emulsion although sonication treatment for 30 min significantly increased the total flavonoids content of the nanoemulsion as compared to the coarse emulsion ($p \leq 0.05$). Similar results were also observed by Maherani et al. (2018) where coarse emulsions presented lower total flavonoids content than other nanoemulsions.

Effect of the preparation method on the antimicrobial properties of emulsions

In vitro analysis Results of the effect of sonication and microfluidization methods on the MIC of the prepared nanoemulsions are presented in Table 2. The MIC of the coarse emulsion against *E. coli* and *L. monocytogenes* were 1.25% (12,500 ppm) and 0.354% (3540 ppm) respectively, in the coarse emulsion. Sonication and microfluidization treatments reduced the MIC against *L. monocytogenes* to reach

Fig. 3 Effect of sonication versus microfluidization on the determination of **A** total phenol and **B** total flavonoid contents

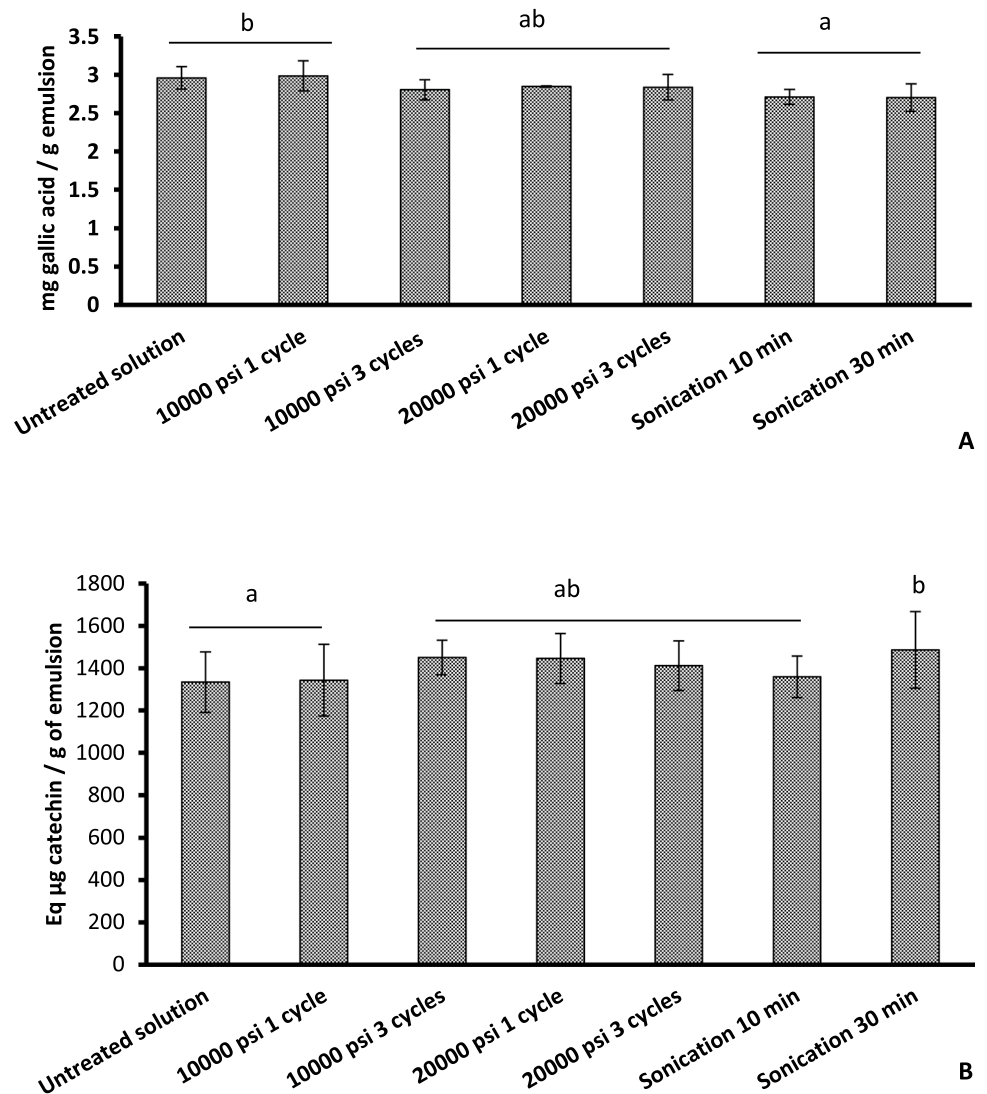


Table 2 MIC (%) of emulsions prepared by sonication versus microfluidization against *E. coli* O157:H7, *L. monocytogenes*, *B. subtilis* and *A. flavus*

	<i>E. coli</i>	<i>L. monocytogenes</i>	<i>B. subtilis</i>	<i>A. flavus</i>
Coarse emulsion	1.25	0.354	0.0024	0.07813
10 min	0.078	0.029	0.0024	0.07813
20 min	0.029	0.024	0.0024	0.07813
30 min	0.039	0.024	0.0024	0.07813
10,000 psi/1 cycle	0.039	0.029	0.0024	0.07813
10,000 psi/3 cycles	0.078	0.029	0.0024	0.07813
20,000 psi/1 cycles	0.039	0.049	0.0024	0.07813
20,000 psi/3 cycles	0.039	0.029	0.0024	0.07813

0.049% (490 ppm) and *E. coli* to 0.078% (780 ppm) for both treatments. These results suggest that reducing droplet mean diameter affects significantly the efficiency of the emulsion against *L. monocytogenes* and *E. coli*. According to Donsì and Ferrari (2016), the improvement of the antimicrobial activity of nanoemulsion could be related to the improvement of the interaction with the cytoplasmic membranes by i) the increase of surface area and passive transport through the outer cell membrane, ii) the sustained release over time of the EO from the nanoemulsion droplets that prolongs the activity of EO iii) the fusion of the emulsifier droplets with the phospholipid bilayer of the cell membrane promotes the targeted release of the EO. Previous studies also demonstrated the effect of nanoscale droplets on the antimicrobial activity against *E. coli* (Salvia-Trujillo et al. 2014). However, the same authors demonstrated that sonication causes a significant loss in terms of antimicrobial activity against *E. coli* depending on the amplitude applied. Treatment at the concentration of 100 µm for 3 min causes a total loss of bactericidal activity of lemongrass EO, which is not the case of microfluidization. Also, a sonication treatment for 1 h can esterify more than 5% of lipids. In the current study, both microfluidization and sonication were able to improve the antimicrobial activity of *Listeria* and *E. coli*. This could be related to a better control of the temperature of the sample treated with sonication by renewing the ice bath. In the current study, treated nanoemulsions were maintained at 30 °C as compared to 47 °C for the study of Salvia-Trujillo et al. (2014).

Contrary to what was observed for *E. coli* and *L. monocytogenes*, results obtained with *B. subtilis* and *A. flavus* showed that the MIC was 0.0024% (24 ppm) and 0.07813% (781.3 ppm) respectively, in the coarse emulsion. No change was observed with the application of sonication and microfluidization treatments suggesting a different inhibitory mechanism of action between spores

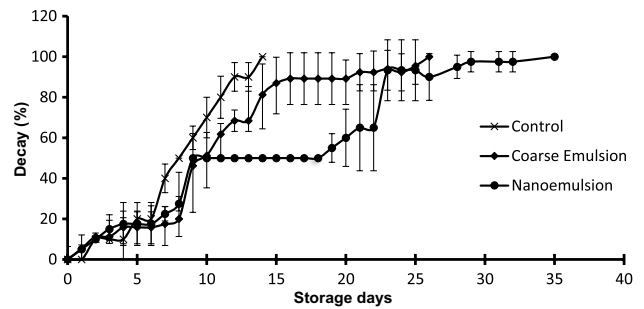


Fig. 4 Effect of coarse emulsion and nanoemulsion on the bread decay (%) inoculated with *B. subtilis* and stored at room temperature

and non-spore forming microorganisms. Contrary to these observations, Pongsumpun et al. (2020) showed an improvement of the antimicrobial activity of nanoemulsion prepared with sonication treatment against *A. niger*, *R. arrhizus*, *Penicillium* sp and *C. gloeosporioides* when the disk diffusion method was employed. This suggests that the sonication method may enhance the antimicrobial activity of the volatile fraction. In fact as explained by Ben-Fadhel et al. (2019), the antimicrobial activity of EO is due to both solid and vapor-phase fractions. While the MIC method involves the liquid fraction, the antimicrobial effect of the vapor fraction is underestimated. These results suggest that the efficiency of sonication treatment and maintaining of the antimicrobial bioactivity is highly related to the control of the treatment conditions especially the temperature of the sample during treatment.

In situ study Results of the *in-situ* study for *B. subtilis* on bread surface are detailed in Fig. 4. Results showed that for untreated samples, 100% of decay was observed on day 14 of storage at room temperature. Applying coarse emulsion on the bread surface delayed the decay by around 12 days and complete decay happened after 26 days. It is interesting to note, that applying a sonication treatment for 10 min to develop nanoemulsion improved the microbial quality of bread about 21 days as compared to the control and about 4 days as compared to coarse emulsion. As the complete decay in breads treated with nanoemulsion occurred on day 35. Similar observations were obtained in the study of Otoni et al. (2014) where an improvement in the shelf-life of sliced bread treated with clove bud and oregano EO was observed with the reduction of the droplet size. The same authors explained this efficiency to a higher bioavailability of nonpolar bioactive compounds encapsulated in smaller droplets, allowing a higher surface-to-volume ratio and easier penetration to cell membranes (Otoni et al. 2014).

Conclusion

In this study, it was demonstrated that microfluidization and sonication treatments were able to form o/w nanoemulsions for the encapsulation of a food grade antimicrobial formulation containing cinnamon EO and CE. Based on the results of size, PDI and flavonoids content, nanoemulsion prepared with HLB12, emulsifier: oil ratio of 1:1 and with 10 min-pulsed sonication treatment gave the most stable nanoemulsion with a lower PDI and turbidity with no effect on total flavonoid content. However, 10-min-sonication treatment slightly decreased the total phenol content. Both sonication and microfluidization treatments improved the in vitro antimicrobial activity of the bioactive emulsion against *E. coli* and *L. monocytogenes*, whereas no effect was observed on the antimicrobial activity against *B. subtilis* and *A. flavus*. When applied on bread surface, the nanoemulsion allowed a better control of *B. subtilis*. Therefore, the stabilization of bioactive formulations based on hydrophilic and hydrophobic natural antimicrobials by using nanoemulsions is a promising solution to replace conventional synthetic preservatives used in food industry.

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Author contribution YB-F: data curation; formal analysis; investigation; methodology; software; validation; writing – original draft. MA: methodology. CM: methodology. SS: data curation; methodology; software; writing – review & editing. ZA: writing – review & editing. ML: conceptualization; data curation; funding acquisition; project administration; resources; supervision; validation; writing – review & editing.

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Data availability The datasets used and/or analyzed during the current study are available from the corresponding author on reasonable request.

Code availability Not applicable.

Declarations

Conflict of interest The authors declare that they have no competing interests.

Consent to participate Not applicable.

Consent for publication Not applicable.

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