

RESEARCH ARTICLE

Anticancer derivative of the natural alkaloid, theobromine, inhibiting EGFR protein: Computer-aided drug discovery approach

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OPEN ACCESS

Citation: Eissa IH, Yousef RG, Elkaeed EB, Alsfouk AA, Husein DZ, Ibrahim IM, et al. (2023) Anticancer derivative of the natural alkaloid, theobromine, inhibiting EGFR protein: Computer-aided drug discovery approach. PLoS ONE 18(3): e0282586. <https://doi.org/10.1371/journal.pone.0282586>

Editor: Ahmed A. Al-Karmalawy, Ahram Canadian University, EGYPT

Received: January 19, 2023

Accepted: February 18, 2023

Published: March 9, 2023

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Data Availability Statement: All relevant data are within the paper and its [Supporting Information](#) files.

Funding: This research was funded by Princess Nourah bint Abdulrahman University Researchers Supporting Project number (PNURSP2023R116), Princess Nourah bint Abdulrahman University, Riyadh, Saudi Arabia. The authors extend their appreciation to the Research Center at AlMaarefa University for funding this work.

Abstract

A new semisynthetic derivative of the natural alkaloid, theobromine, has been designed as a lead antiangiogenic compound targeting the EGFR protein. The designed compound is an (*m*-tolyl)acetamide theobromine derivative, (**T-1-MTA**). Molecular Docking studies have shown a great potential for **T-1-MTA** to bind to EGFR. MD studies (100 ns) verified the proposed binding. By MM-GBSA analysis, the exact binding with optimal energy of **T-1-MTA** was also identified. Then, DFT calculations were performed to identify the stability, reactivity, electrostatic potential, and total electron density of **T-1-MTA**. Furthermore, ADMET analysis indicated the **T-1-MTA**'s general likeness and safety. Accordingly, **T-1-MTA** has been synthesized to be examined *in vitro*. Intriguingly, **T-1-MTA** inhibited the EGFR protein with an IC₅₀ value of 22.89 nM and demonstrated cytotoxic activities against the two cancer cell lines, A549, and HCT-116, with IC₅₀ values of 22.49, and 24.97 μM, respectively. Interestingly, **T-1-MTA**'s IC₅₀ against the normal cell lines, WI-38, was very high (55.14 μM) indicating high selectivity degrees of 2.4 and 2.2, respectively. Furthermore, the flow cytometry analysis of A549 treated with **T-1-MTA** showed significantly increased ratios of early apoptosis (from 0.07% to 21.24%) as well as late apoptosis (from 0.73% to 37.97%).

Introduction

The world health organization (WHO) anticipated that during the next few years, cancer will dominate all other causes of death [1]. Developing treatments that suppress the growth of cancer by interacting with specific molecular targets and damaging the cancer cells is a major

Competing interests: The authors have declared that no competing interests exist.

concern for medicinal chemists that work on cancer therapy [2]. Increasing vascularity (angiogenesis) is a crucial process that increases tumor development, so anti-angiogenesis strategies were considered to be very effective in the treatment [3]. Also, it was confirmed that the angiogenesis and growth of cancer cells are driven by the epidermal growth factor receptors (EGFR) [4]. In response to EGFR overexpression, downstream signaling pathways stimulate cell proliferation, differentiation, and survival. In cancers, EGFR was found to be elevated and promoted several solid malignant tumors [5]. Numerous cancer types express lower survival rates when EGFR is expressed. Also, EGFR's expression served as a powerful diagnostic and prognostic indicator for cancer [6]. In contrast, this overexpression allowed researchers to utilize the EGFR's inhibition as an essential strategy in cancer treatment [7,8].

Anciently, natural products, especially plants, were the most vital bases of treatments [9,10]. Recently, one-third of the FDA-approved drugs from 1981–2014 have been derived from natural sources [11]. Anticancer drug discovery finds xanthines, and xanthine derivatives, to be interesting compounds that exhibit different antimutagenic properties against ovarian cancer [12], prostate cancer [13], breast cancer, and leukemia [14].

Theobromine, the famous natural alkaloid, was discovered in 1841, while the synthesis of theobromine was described in 1882 [15,16]. Theobromine is found primarily in *Theobroma cacao*, chocolate, and other foods including tea leaves [17]. Theobromine showed promising anti-cancer activity *in vitro* and *in vivo* through the inhibition of DNA synthesis in glioblastoma multiforme [18] and prevented lung cancer angiogenesis [19]. Interestingly, in ovarian cancer, theobromine inhibited the VEGF *in vivo* and *in vitro* [20]. By using semi-synthesis to produce analogs, we can discover more potent drugs, give repurposing opportunities, and develop novel bioactive compounds, enhance drug-likeness, and improve pharmacokinetics and pharmacodynamics [21].

In scientific society today, computer-aided drug discovery (CADD) is widely accepted as a means of applying theoretical ideas using computers and a set of techniques for investigating chemical problems and is used in the pharmaceutical industry to investigate how potential drugs interact with biomolecules [22–26]. Our team applied the CADD in molecular design and docking, computational toxicity and ADME [27,28], in addition to MD simulations [29,30].

Rationale

Erlotinib I [31,32] and olmutinib II [33] are reported as EGFR inhibitors. Compounds III and IV are derivatives of 1*H*-pyrazolo[3,4-*d*]pyrimidine that showed excellent efficacy for inhibiting EGFR-TK at nono-molar doses [34,35]. Our team previously synthesized compound V (a thieno[2,3-*d*]pyrimidine derivative) that was promising anti-proliferative and EGFR inhibitor [36] (Fig 1).

These compounds possess some pharmacophoric features of EGFR-TKIs. These features are a planar heterocyclic system, an NH spacer, a terminal hydrophobic head and a hydrophobic tail. The key roles of the above-mentioned structural moieties are to occupy the adenine binding pocket [37], interact with amino acid residues in the linker region [38], to be inserted in the hydrophobic region I [39], and to occupy the hydrophobic region II [40,41], respectively (Fig 1).

In this work and as an extension of our previous efforts in the discovery of new anti-EGFR agents [36,42–44], compound V was used as a lead compound to reach a more promising anti-cancer agent targeting EGFR. Several chemical modifications were carried out at four positions. The first position is the planar heterocyclic system. We applied the ring variation strategy as the thieno[2,3-*d*] pyrimidine moiety was replaced by a xanthine derivative (3-methyl-3,7-dihydro-1*H*-purine-2,6-dione). The six hydrogen bond (HB) acceptors may

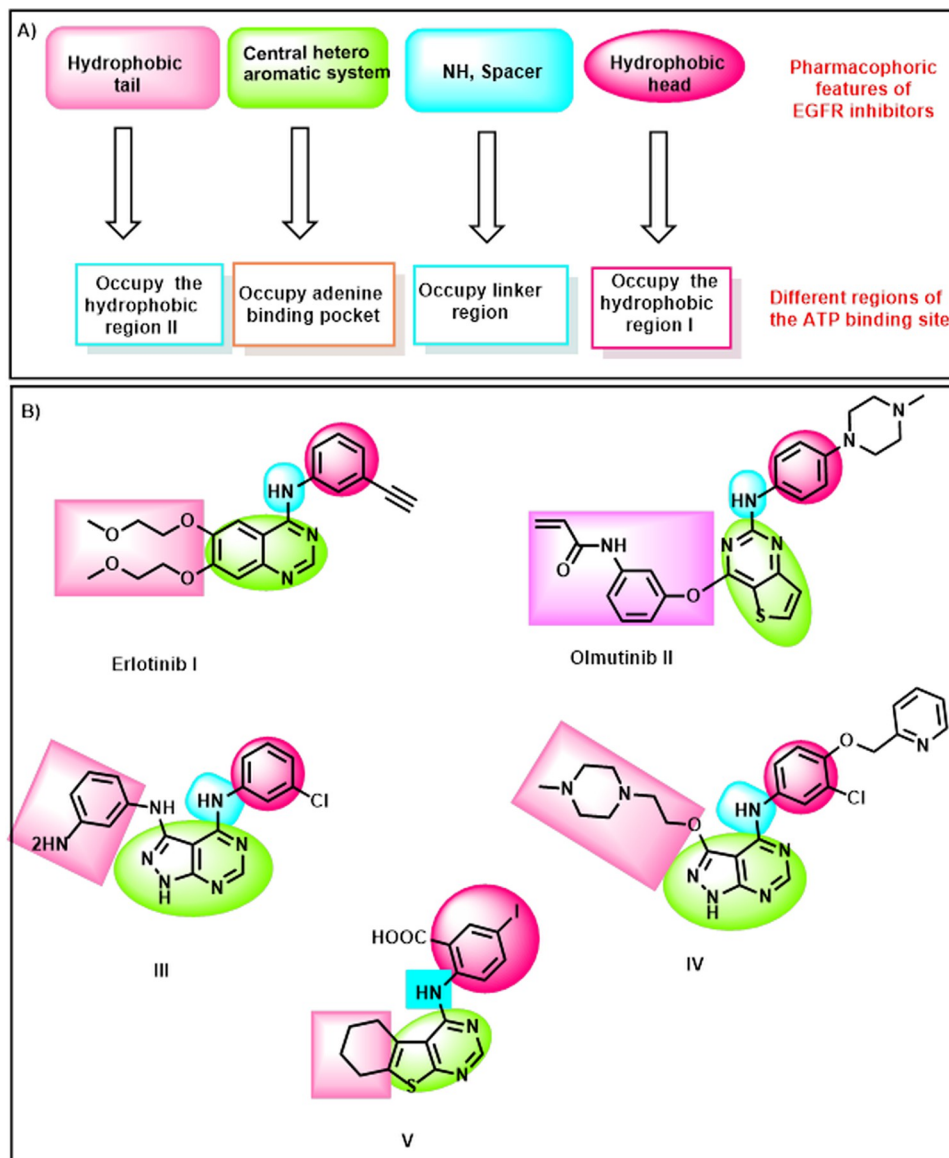


Fig 1. EGFR inhibitors' pharmacophoric features.

<https://doi.org/10.1371/journal.pone.0282586.g001>

facilitate the HB interaction in the adenine binding pocket. Chain extension strategy was applied in the liker region through the replacement of the NH-linker with acetamide moiety. The terminal hydrophobic head (3-iodobenzoic acid) of the lead compound was replaced by toluene moiety) *via* ring variation strategy. A simplification strategy was applied for the hydrophobic tail (cyclohexene) of the lead compound. It was replaced by methyl group at 7-position of xanthine moiety (Fig 2).

Results and discussions

2.1. Chemistry

Scheme 1 depicts the synthetic pathway used in this study to produce target T-1-MTA. The potassium salt of 3,7-dimethyl-3,7-dihydro-1H-purine-2,6-dione **2** was first obtained by

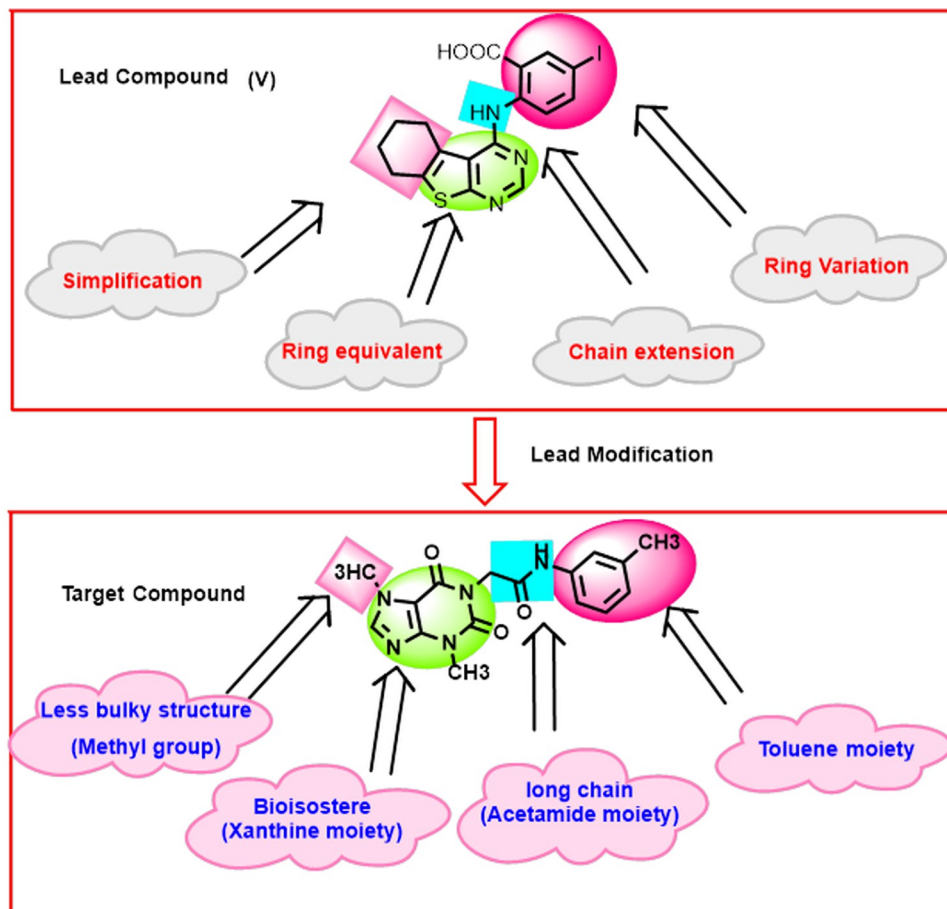


Fig 2. T-1-MTA's design rationale.

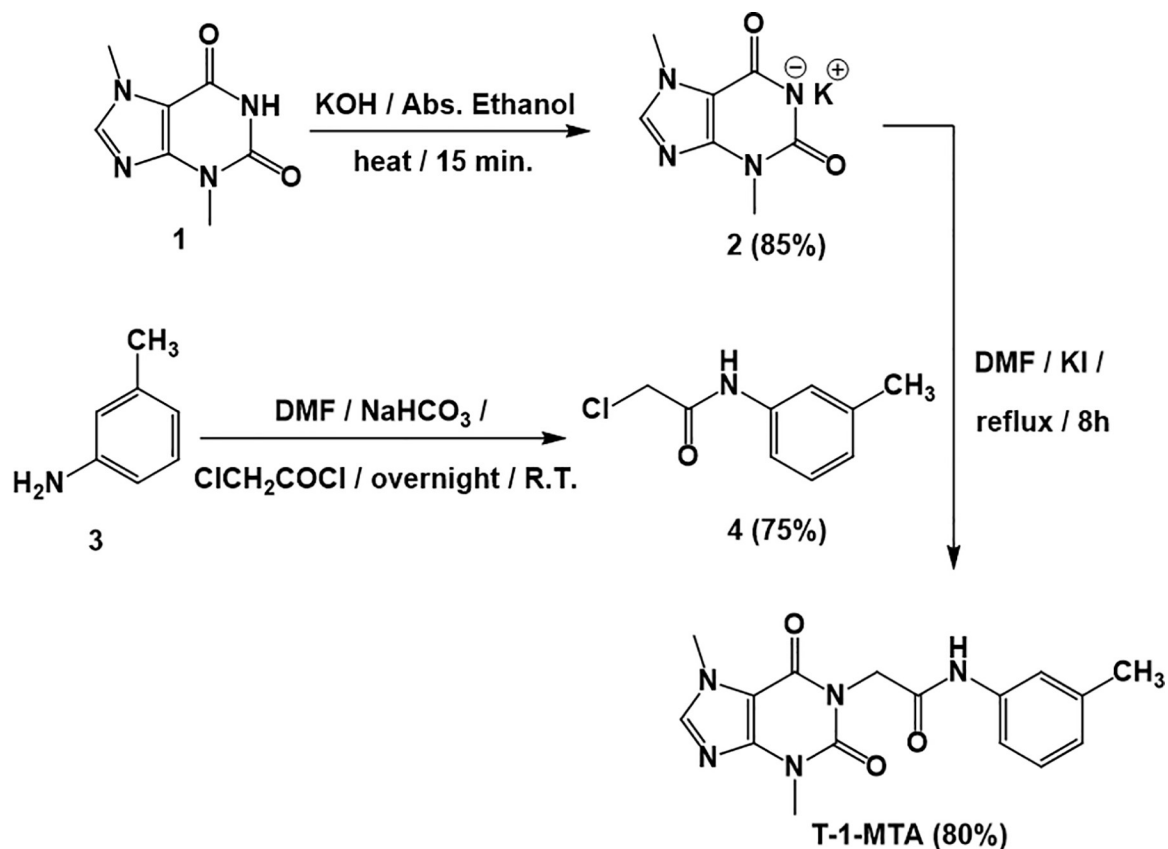
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refluxing 3,7-dimethyl-3,7-dihydro-1*H*-purine-2,6-dione (theobromine, **1**) with alcoholic KOH [45,46]. 2-Chloro-*N*-(*m*-tolyl)acetamide **4**, as the key intermediate, was prepared from commercially available *m*-toluidine **3** with chloroacetylchloride in DMF using NaHCO₃. When equimolar amounts of potassium 3,7-dimethyl-3,7-dihydro-1*H*-purine-2,6-dione **2** and 2-Chloro-*N*-(*m*-tolyl)acetamide **4** were refluxed in DMF containing a sufficient amount of potassium iodide as a catalyst, an expected final product T-1-MTA was attained.

The ¹H NMR spectrum of T-1-MTA showed singlet signal at $\delta = 8.07$ for CH imidazole and multiplet signals ranging from $\delta 7.41$ to 6.87 for aromatic protons besides remarkable singlet signals for the CH₃ (of *m*-tolyl group) and CH₂ groups at $\delta = 2.27$ and 4.67, respectively. The IR spectrum of the same product revealed absorption bands at 1711, 1662 cm⁻¹ corresponding to carbonyl groups and absorption bands at 3255 cm⁻¹ corresponding to NH. Regarding the ¹³C NMR spectrum, four shielded signals appeared at 43.84, 33.66, 29.90, and 21.62 ppm corresponding to CH₂ and the three CH₃ groups, respectively.

2.2. Molecular docking

The examined proteins' X-ray structures (EGFR^{WT}; PDB: 4HJO and EGFR^{T790M}; PDB: 3W2O) were acquired from the Protein Data Bank (PDB, <http://www.pdb.org>). First, the docking protocol was verified for both wild and mutant EGFR and the RMSD results were 1.20 and 1.15 Å, respectively Fig 3.



Scheme 1. Synthetic pathway of T-1-MTA.

<https://doi.org/10.1371/journal.pone.0282586.g003>

Erlotinib, as a native inhibitor for EGFR^{WT}, revealed an affinity value of -20.50 kcal/mol. The binding pattern of erlotinib revealed a key HB with Met769 (2.11 Å) in addition to four hydrophobic interactions (HI) in the adenine pocket and three HIs with Ala719 and Val702, and Lys721 in the hydrophobic pocket (Fig 4). TAK-285, as a native inhibitor for EGFR^{T790M}, presented a binding energy of -7.20 kcal/mol. The binding pattern of TAK-285 revealed a key HB with Met793 (2.44 Å) through the pyrimidine moiety in the adenine pocket. The later moieties (3-(trifluoromethyl)phenoxy and *N*-ethyl-3-hydroxy-3-methylbutanamide moieties)

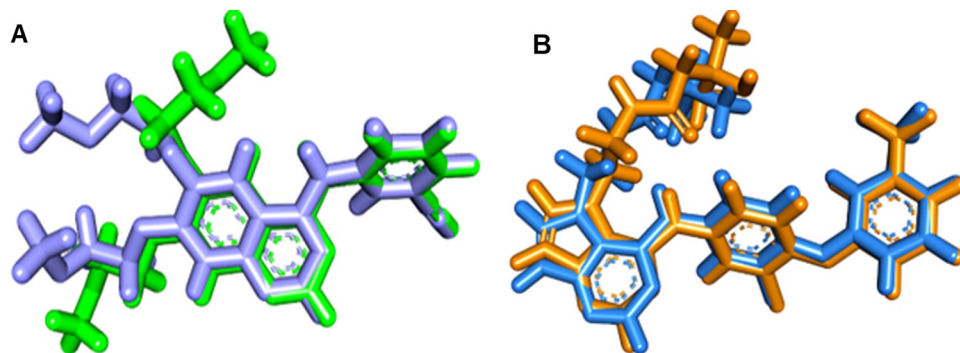


Fig 3. A: Validation of wild EGFR using erlotinib as co-crystallized ligand and B: Validation of mutant EGFR using TAK-285 as co-crystallized ligand.

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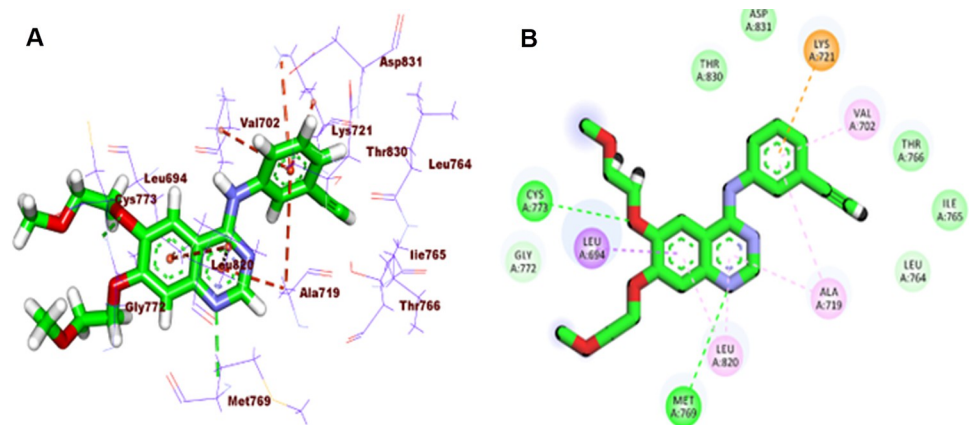


Fig 4. A: 3D and B: 2D close view of erlotinib EGFR^{WT}.

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were fixed in the hydrophobic pocket *via* a network of HIs with Lys745, Ile759, Met790, Val726, and Ala743, and Leu844 (Fig 5).

Regarding the EGFR^{WT}, a comparable affinity value to erlotinib was obtained by T-1-MTA (-20.45 kcal/mol). Additionally, it interacts with the EGFR^{WT} active site similar to erlotinib and adopts the same orientation. Besides, the 3,7-dimethyl-2,6-dioxo-2,3,6,7-tetrahydro-1*H*-purine arm formed a crucial HB with Met769 besides two HIs with Lue694 inside the adenine pocket. On the other side, five HIs with Leu764, Ala719, Val702, and Lys721 were achieved *via* the *m*-tolyl moiety in the conserved hydrophobic pocket. The methyl group at 7-position of xanthine moiety failed to form HIs in the hydrophobic pocket II (Fig 6).

Regarding the EGFR^{T790M}, T-1-MTA (binding energy of -6.95 kcal/mol.) was tacked onto the catalytic site similarly to the positive control, TAK-285. In the adenine pocket, six pi-pi bonds with Leu844, Ala743, and Met79 were accomplished through the 3,7-dimethyl-2,6-dioxo-2,3,6,7-tetrahydro-1*H*-purine arm. Also, in the same region, a crucial HB with Met793 was observed. Additionally, in the hydrophobic pocket, the *m*-tolyl moiety was buried to form one electrostatic interaction with Lys745 Fig 7.

2.3. MD simulations

The MD analyses obtained on a 100 ns production run showing an overall system stability. The RMSD plot (Fig 8A) showed a stable trend for the EGFR only and the EGFR_T-1-MTA complex

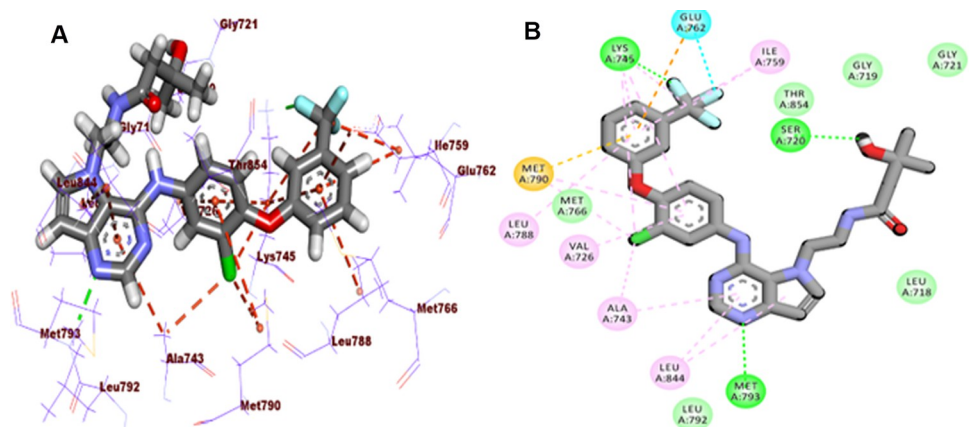


Fig 5. 3D and 2D close view of TAK-285 into EGFR^{T790M}.

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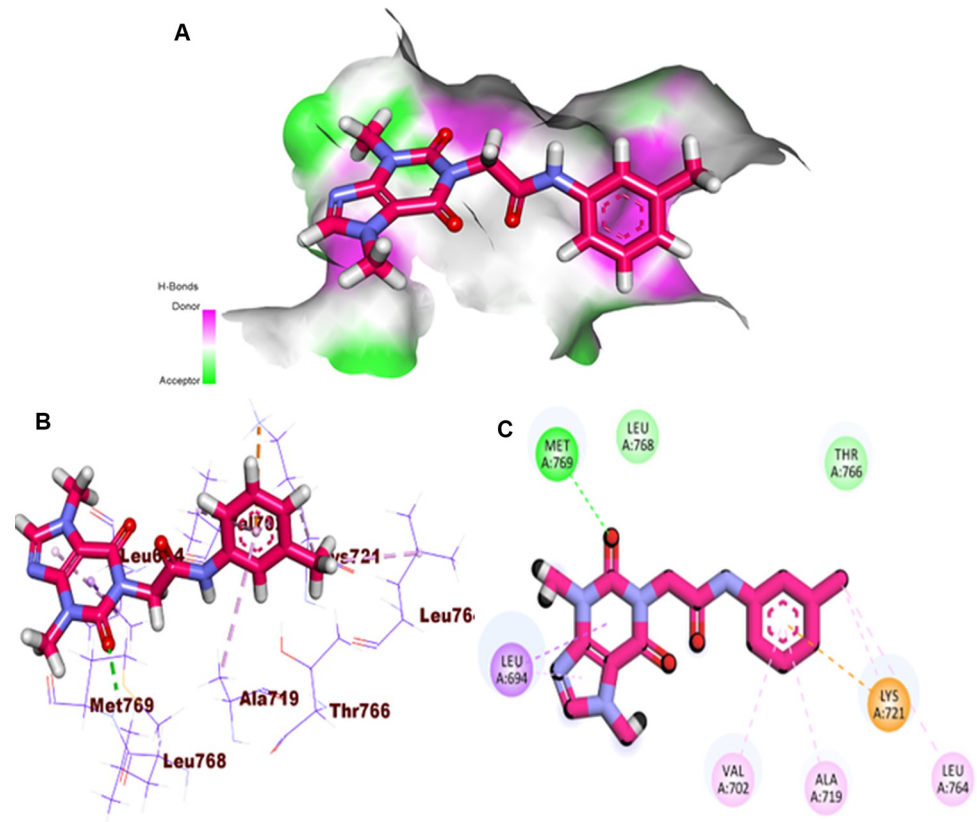


Fig 6. A: Mapping Surface (MS), B: 3D, and C: 2D close view of T-1-MTA inside the EGFR^{WT}.

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that were represented as blue and green curves showings averages of 2.16 Å and 2.97 Å, respectively. Moreover, the RMSD of the T-1-MTA (red) showed three states during the whole trajectory. The first 10 ns show an average of 2.16 Å before spiking to an average of 9.43 Å for the next 30 ns. Moreover, the last 60 ns show a large stable average value of 17.72 Å. The reason for this increase in the RMSD values of the compound T-1-MTA is due to the translational movement of the compound T-1-MTA relative to the protein as shown in Fig 8G which compares between the positions of the ligand at 1.5 ns (green sticks), 29.5 ns (cyan sticks), 83.9 ns (magenta sticks), and 94 ns (yellow sticks). The RoG (Fig 8B), SASA and (Fig 8C) HB show a stable protein fluctuation with an average of 19.51 Å, and 15285 Å², respectively. The change in HBs between the T-1-MTA and EGFR (Fig 8D) shows that there is, approximately, at least one HB formed during the first 40 ns and it increases to at least two bonds during the rest of the simulation. The amino acids' fluctuation was depicted in the RMSF plot (Fig 8E) showing low values of fluctuation (less than 2 Å) excepting the free C-terminal and the loop region E842:Y845 reaching 7 Å, and 3.5 Å, respectively. During the simulation time, the distance between the center of mass of compound T-1-MTA and the center of mass of EGFR protein shows a similar trend to the RMSD values of the ligand (three states) (Fig 8F). It started with an average of 16.72 Å for the first 15 ns before slightly decreasing to an average of 14.02 Å for the next 25 ns (from 15 ns to 40 ns). Finally, the last 60 ns showed an average value of 11.87 Å showing a stable interaction (Fig 8G).

2.4. MM-GBSA studies

The binding free energy of the EGFR_T-1-MTA complex was further analyzed deeply by the MM-GBSA analysis. As Fig 9 shows, the EGFR_T-1-MTA complex had a total binding energy

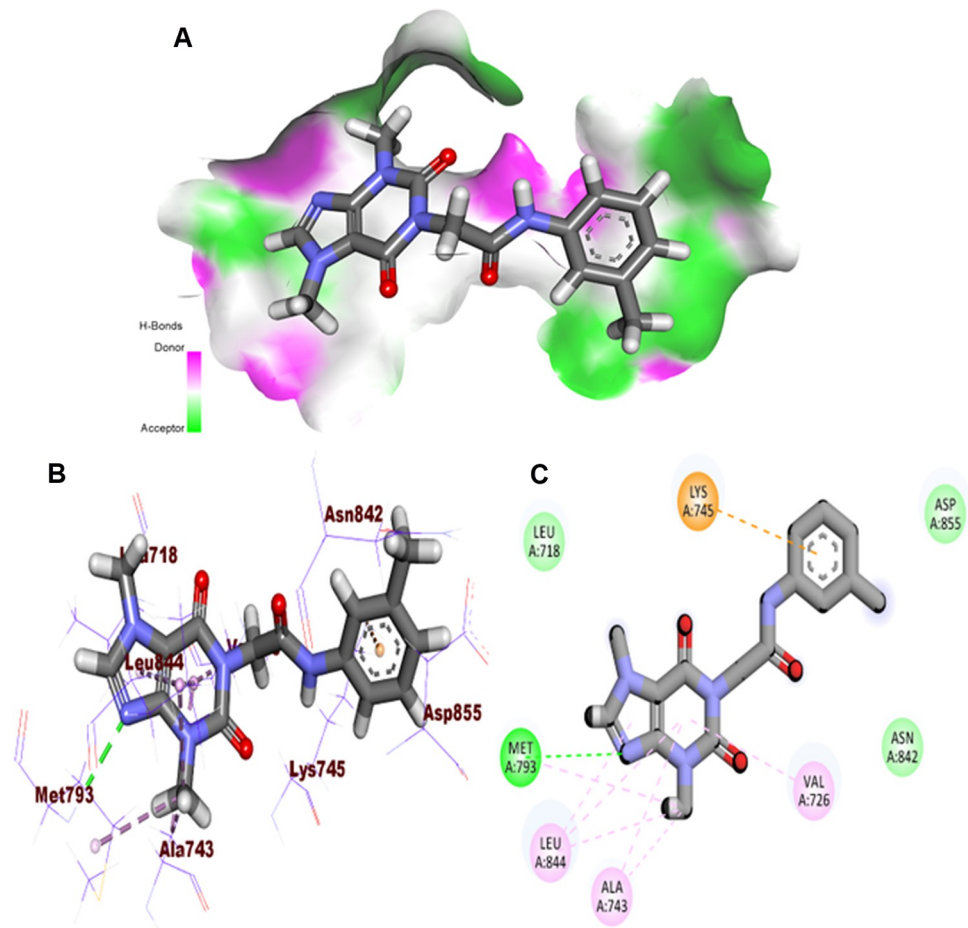


Fig 7. A: MS, B: 3D, and C: 2D close view of T-1-MTA with EGFR^{T790M}.

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of an average value of -18.88 kcal/Mol. The various forms of energy that contribute to binding of the EGFR_T-1-MTA complex were analyzed to be. Van Der Waals interaction, electrostatic interaction with average values of -30.31 kcal/Mol and -10.23 kcal/Mol, respectively. Moreover, we performed an energy-decomposition analysis as shown in Fig 10 to identify the amino acids that had the highest contribution to the binding (1 nm or better). L694 (-1.48 kcal/Mol), S696 (-1.56 kcal/Mol), and R817 (-1.9 kcal/Mol) are the amino acids that exhibited the best contributions (better or less than -1 kcal/Mol).

2.5. Protein-Ligand Interaction Profiler (PLIP) studies

After that, to obtain a representative frame for each cluster of the EGFR_T-1-MTA complex, the obtained trajectory was clustered. The elbow method was used to automatically choose the number of clusters, as described in the methodologies section, and this resulted in four clusters. The PLIP website was used to determine the number and types of interactions between T-1-MTA and EGFR for each cluster representative (Table 1). As can be seen, HIs have a similar overall number of interactions in all the clusters compared to the HBs (7 HIs vs. 6 HBs). Additionally, a.pse file was generated to understand the 3D conformations of T-1-MTA as well as its interaction against the EGFR (Fig 11).

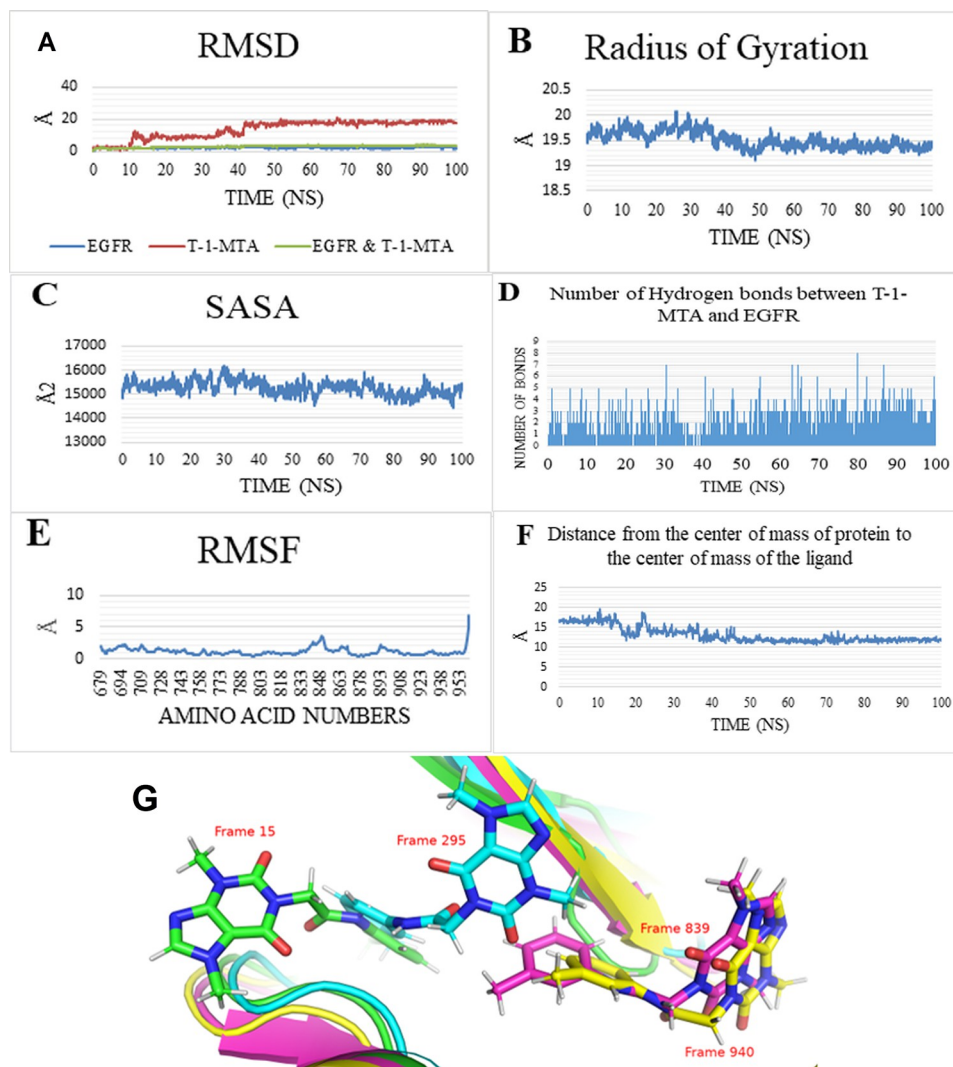


Fig 8. MD measurements calculated for a 100 ns. A) RMSD, B) RoG, C) SASA, D) HBBs' change between the T-1-MTA and EGFR, E) RMSF, F) Center of Mass distance between the compound T-1-MTA and EGFR, and G) shows the positions of the compound T-1-MTA at different snapshots of the trajectory. T-1-MTA is in stick representation while the protein at the same snapshots is in cartoon representation.

<https://doi.org/10.1371/journal.pone.0282586.g009>

2.6. DFT studies

In an attempt to clarify the inhibitory activity of T-1-MTA, theoretical DFT studies have been explored. The conceptual DFT has been used for understanding the electronic structure of the prepared molecule to determine its structural features which has far-reaching consequences on the molecules' reactivity. Hence, the DFT-based reactivity descriptors (global), frontier molecular orbital analysis (FMO), and surface potential maps have been investigated to explore the reactivity of the prepared compound.

Geometry optimization

The reactivity of T-1-MTA is mainly determined by its chemical structure, so the structure is fully optimized and computed using DFT. The single bond length N2-C14 is 1.4765 Å,

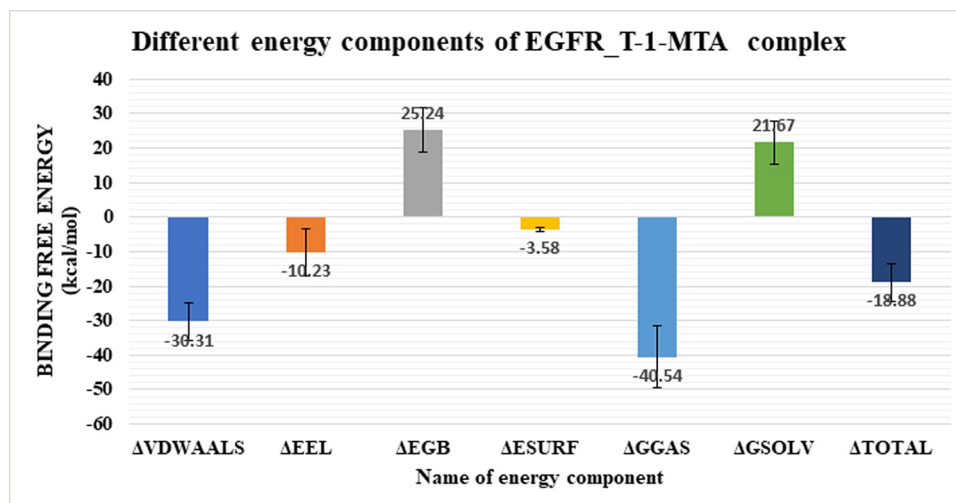


Fig 9. Energetic components of EGFR-T-1-MTA complex. Bars represent the standard deviation.

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whereas the C14-N2-C3 bond angle is 116.70971° as given in Fig 12 at the B3LYB/6-311G++(d,p) level. The computed ground total energy (TE) is -30470.0 eV whereas the dipole moment (Dm) value is 5.9956 Debye which indicated a strong ability of interaction within the chemical system.

Frontier molecular orbital analysis (FMO) analysis

Border molecular orbitals in a molecule play a vital role in the electric properties as the system with a smaller value of energy gap between the border orbitals ($E_{\text{gap}} = E_{\text{LUMO}} - E_{\text{HOMO}}$) should be more reactive than one having a greater E_{gap} . Fortunately, T-1-MTA reported a smaller E_{gap} value, so the electronic movement between the border orbitals; LUMO and HOMO, could occur easily [47]. The nodal properties of HOMO-LUMO orbitals of the studied heterocyclic molecule in Fig 13 are presented and show the strong orbital overlap, delocalization, and the low number of nodal planes. Hard molecules have a high HOMO-LUMO gap, and soft molecules have a smaller HOMO-LUMO gap. The value of E_{gap} is given in Fig 13 and indicated that T-1-MTA is considered soft and the electronic transition (HOMO-LUMO)

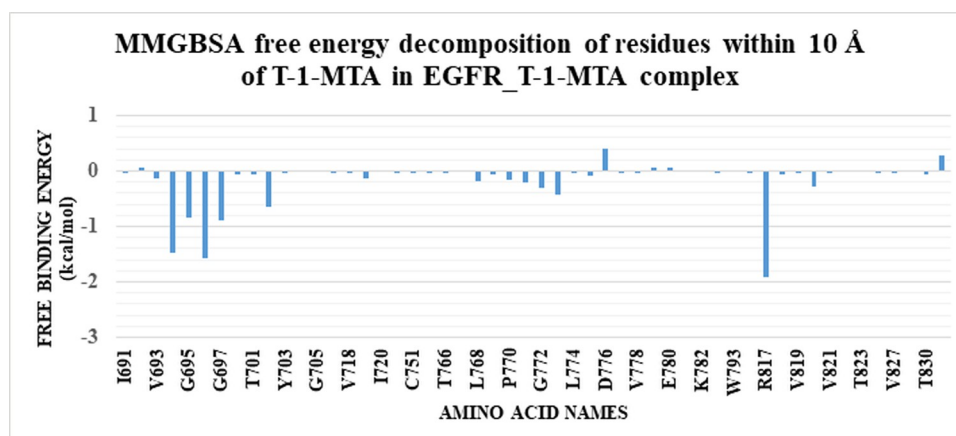


Fig 10. Binding energy decomposition of the EGFR_T-1-MTA complex.

<https://doi.org/10.1371/journal.pone.0282586.g011>

Table 1. Shows the number and types of interactions detected from the PLIP webserver.

CLUSTER NUMBER	HBS	AMINO ACIDS	HIS	AMINO ACIDS
C1 (FRAME 15)	1	G772	3	L694—A719—L820
C2 (FRAME 295)	0	None	4	L694 (2)—L768—L820
C3 (FRAME 839)	3	R724—R817—K851	0	None
C4 (FRAME 940)	2	A698—K851	0	None

<https://doi.org/10.1371/journal.pone.0282586.t001>

within the molecule is easy [48]. The quantum chemical parameters such as ionization potential (IP) and electron affinity (EA) were calculated and listed in Table 2.

Global reactive indices and total density of state (TDOS). Based on the density functional theory (DFT) concept, global reactivity parameters are essential tools for comprehending the behavior of any chemical molecular structure. Such global reactivity indices depend on the value of E_{gap} . In Table 2, the static global properties of T-1-MTA, namely the electrophilicity (ω), maximal charge acceptance (N_{max}), energy change (ΔE), chemical potential (μ), global chemical softness (σ), global electronegativity (χ), global chemical hardness (η), and electron

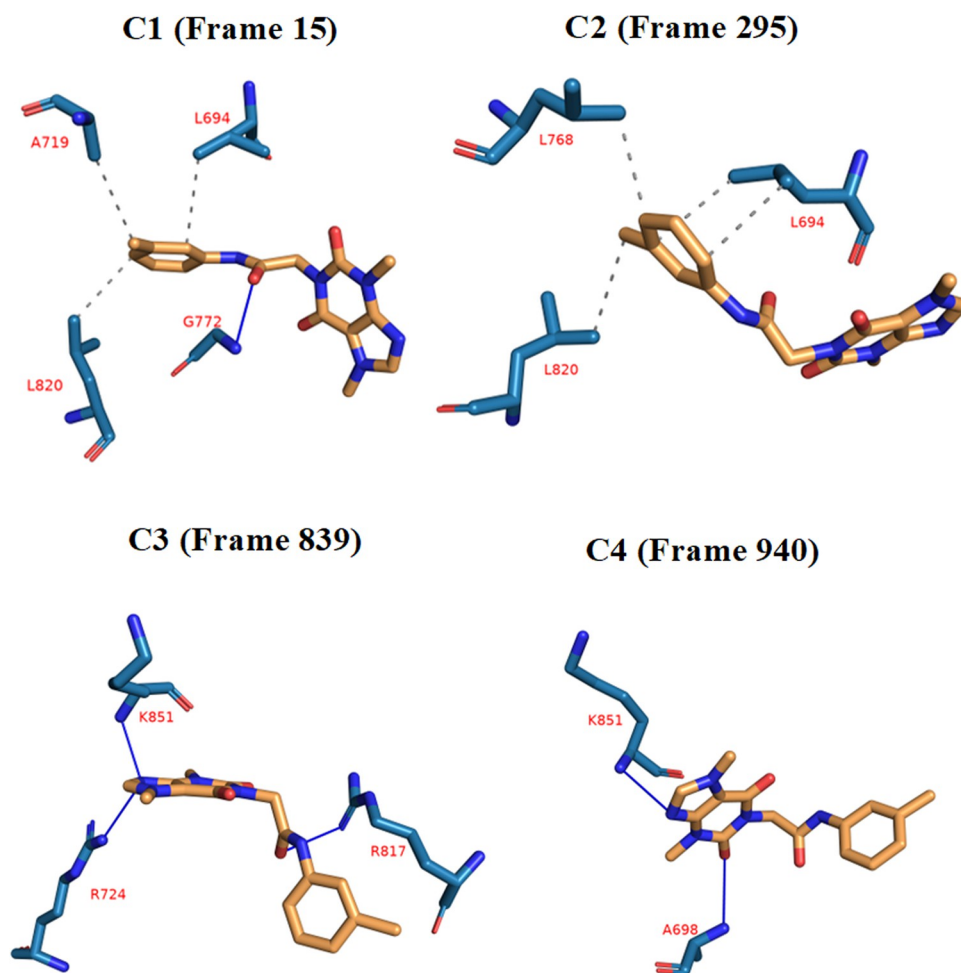


Fig 11. The variation (types and numbers) of interactions of the EGFR_T-1-MTA complex produced from PLIP. HB: Blue solid line, HI: Dashed grey line, green dashed lines: Pi-Stacking interaction, amino acids: Blue sticks, and T-1-MTA: Orange sticks.

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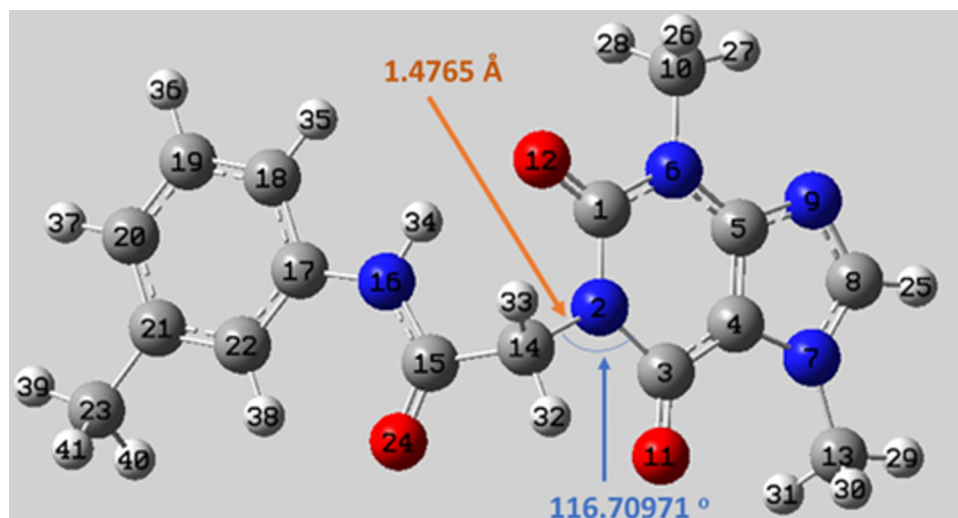


Fig 12. The optimized chemical structure of T-1-MTA.

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affinity (EA) of T-1-MTA are presented after calculating using Koopmans' theory. The results in Table 2 indicated that T-1-MTA is treated as soft within the nucleophilicity and electrophilicity scales [49].

The density of states and the distribution function probability determined by the occupied states per unit volume are important to provide an accurate description best than frontier molecular orbitals. The TDOS spectrum of T-1-MTA in Fig 14 depicted that the highest electronic intensity is located in the occupied orbitals under the HOMO orbital. Also, the TDOS spectrum confirmed the narrow HOMO-LUMO gap.

2.6.4. Molecular surface potential maps. Molecular electrostatic surface potential discovers the relationship between the electronic distribution over the molecule surface and its binding ability. The molecular electrostatic potential explains and predicts the noncovalent

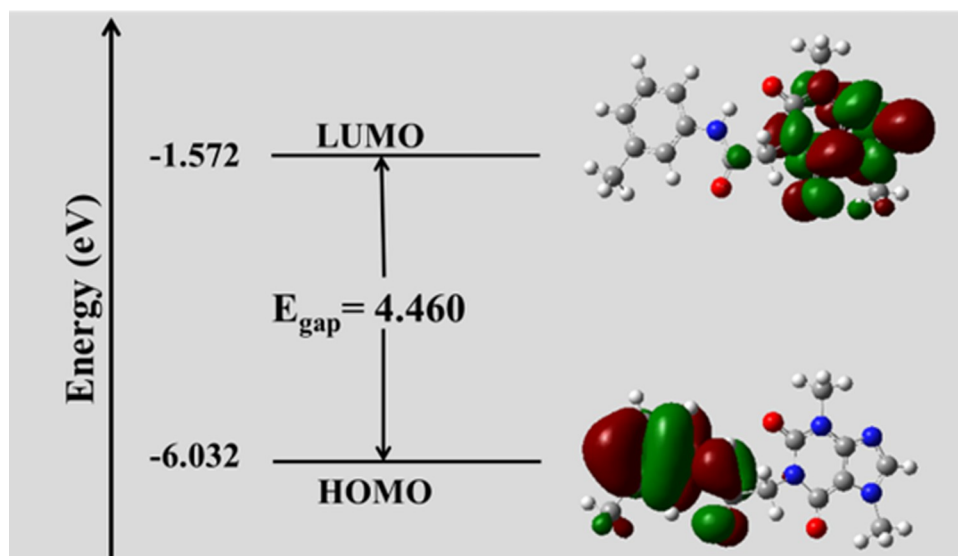


Fig 13. FMO analysis of T-1-MTA.

<https://doi.org/10.1371/journal.pone.0282586.g014>

Table 2. The calculated global reactivity indices and energetic parameters for T-1-MTA.

IP	EA	μ (eV)	χ (eV)	η (eV)	σ (eV)	ω (eV)	Dm (Debye)	TE (eV)	ΔN_{\max}	ΔE (eV)
-6.032	-1.572	-3.802	3.802	2.230	0.448	16.122	5.996	-30470.0	1.705	-16.122

<https://doi.org/10.1371/journal.pone.0282586.t002>

interactions. Also, it finds the positive and negative domains of the electrostatic potential with low and high electron densities, respectively. The quantitative electrostatic surface potential (ESP) and total electrostatic density (TED) maps of T-1-MTA are demonstrated in Fig 15 after analysis of the optimized ground-state geometry. It appears that there are red regions indicating the negative potential is localized over the electronegative atoms such as O. The positive potential domains (blue color) are localized on the hydrogen atoms of purine moiety. The areas with moderate electron density values are shown with yellow color and localized on the phenyl ring. It can be predicted that the positive region on the purine ring of T-1-MTA will interact strongly with the negative region of the target and the negative areas at oxygen atoms will form strong interactions with areas of positive potential at the target. Also, it can be predicted that there is a strong attraction between the most positive region of T-1-MTA and the negative region of the target. The most negative region located around the oxygen atoms can also form a strong interaction with the positive region of the target. This implies that the difference in the distribution of electronic charges could result in enhancing the inhibition reactivity of T-1-MTA towards EGFR.

2.7. ADMET profiling study

The approval of any new compound as a marketed drug is based on a pharmacokinetic evaluation in addition to its biological activity. So, analyzing the ADME properties of a compound at the early stages should keep the discovery process from being delayed [50]. Although ADMET studies *in vitro* can investigate the properties of the absorbent, distribution, metabolism, excretion, and toxicity of drugs, *in silico* studies are advantageous because of their ability of saving cost, time, effort in addition to the regulations restricting the use of animals [51]. Computing

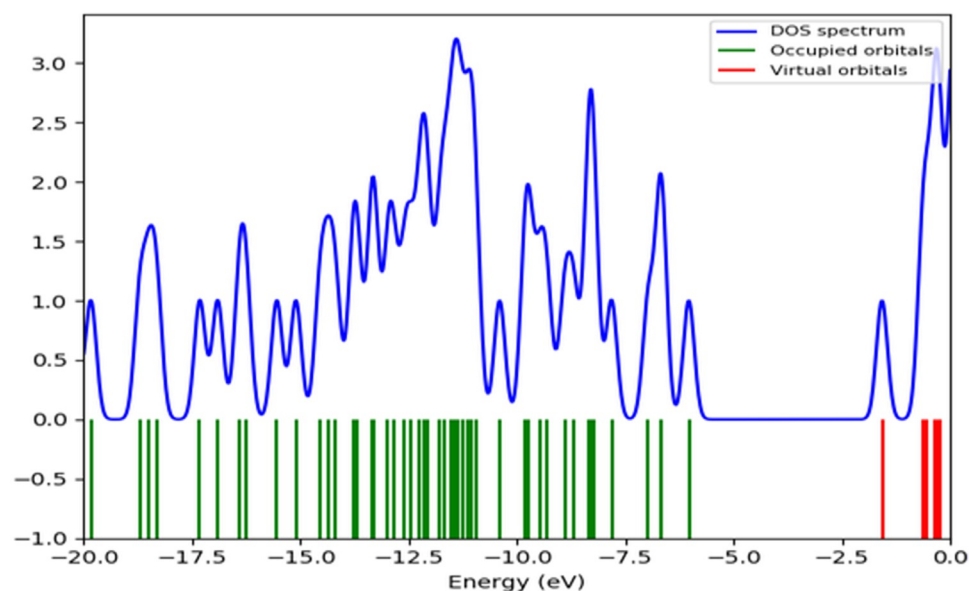


Fig 14. The TDOS spectrum of T-1-MTA.

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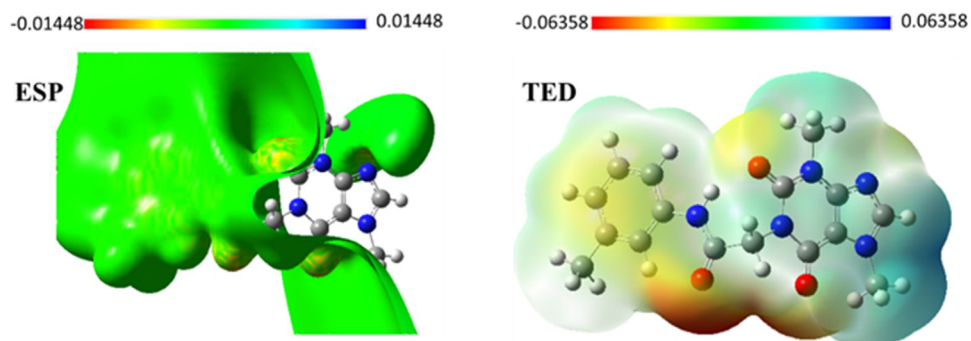


Fig 15. TED and ESP maps of T-1-MTA.

<https://doi.org/10.1371/journal.pone.0282586.g016>

ADMET parameters using Discovery is used to determine the ADMET parameters for T-1-MTA against erlotinib. Interestingly, the obtained results of T-1-MTA comparing erlotinib (Fig 16 and Table 3) showed a high likeness degree as it was anticipated to have a low potential to pass the BBB. Additionally, hepatotoxicity (HT) and the inhibition of cytochrome P-450 (CYP2D6-I) were expected to be absent. Also, T-1-MTA levels of aqueous solubility (AS) and intestinal absorption (IA) were computed as good.

2.8. *In silico* toxicity studies

For a drug to be developed successfully, toxicity assessment at the early stages must be done in order to control the possibility of failure in the clinical stage [52]. The *in silico* approach to toxicity assessment is promising being accurate and avoiding ethical and resource constraints in the *in vitro* and *in vivo* phases of toxicity development [53]. *In silico* prediction of toxicity basically uses the structure-activity relationship (SAR)-predicting toxicity. In detail, the computer compares the chemical properties of the examined molecules against the structural properties of tens of thousands of compounds of reported safety or toxicity [54]. Employing the Discovery studio software, eight toxicity models were used to estimate T-1-MTA's toxicity in

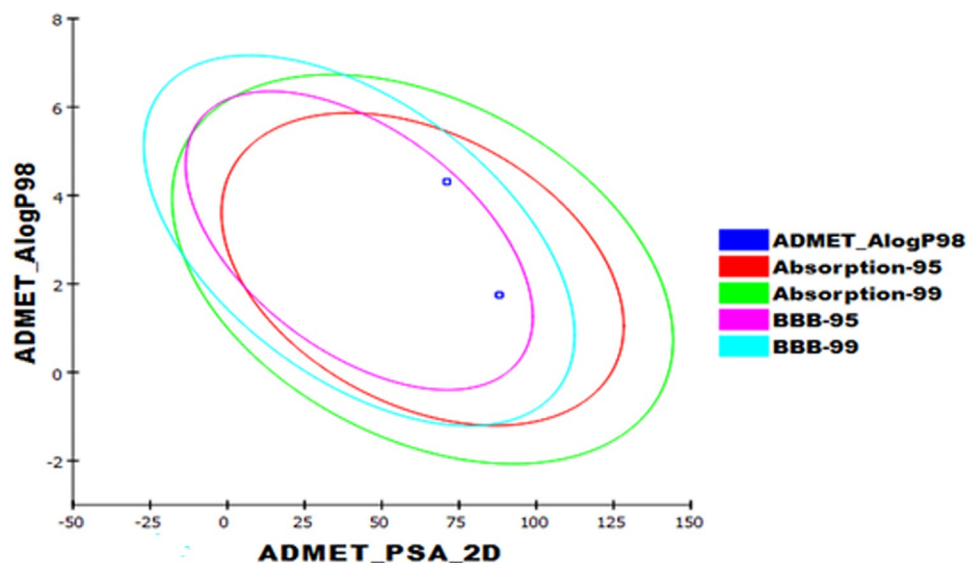


Fig 16. Computational prediction of ADMET parameters for T-1-MTA and erlotinib.

<https://doi.org/10.1371/journal.pone.0282586.g017>

Table 3. ADMET parameters for T-1-MTA and erlotinib.

Comp.	BBB	AS	IA	HT	CYP2D6-I	Plasma protein binding
T-1-MTA	Low	Good	Good	Non-toxic	Non-inhibitor	< 90%
erlotinib	High	Low		toxic		>90%

<https://doi.org/10.1371/journal.pone.0282586.t003>

comparison to erlotinib. Providentially, T-1-MTA expressed very good and safe values in the carried-out models (Table 4)

2.9. Biological evaluation

2.9.1. *In vitro* EGFR inhibition. For the purpose of examining the design and the computational outcomes that clearly demonstrated T-1-MTA's significant affinity for EGFR, T-1-MTA's inhibitory ability was assessed *in vitro* against the EGFR protein (Fig 17). The obtained inhibition value (22.89 nM) was near to erlotinib's value, and the resulting *in vitro* results confirmed T-1-MTA's suppressive potential.

2.9.2 Cytotoxicity and safety

In vitro cytotoxicity assessment was performed for T-1-MTA using compared to erlotinib as demonstrated in Table 5. The obtained IC₅₀ values of T-1-MTA against A549 and HCT-116 malignant cells were 22.49 and 24.97 μM, respectively. T-1-MTA's anticancer potential was close to that of erlotinib.

As a confirmation of the computed safety pattern of T-1-MTA and to explore its selectivity, T-1-MTA was tested against the W138 human normal cell line. T-1-MTA showed a high IC₅₀ value of 55.14 μM as well as very high selectivity indexes (SI) of 2.4 and 2.2 against the two cancer cell lines, respectively (Fig 18).

2.9.3. Cell cycle analysis and apoptosis assay. Firstly, the cell cycle phases of A549 after T-1-MTA's treatment was analyzed by flow cytometry according to the reported method before [55,56]. A concentration of 22.49 μM of T-1-MTA was added to A549 cells for 72 h. Then, the cancer's cell cycle was investigated. Interestingly, T-1-MTA decreased the percentage of A549 cells in the Sub-G1 and S phases from 0.75% and 68.17% to 0.36% and 28.60%, respectively. Contraversly, in the G2/M phase, the A549 percent was significantly increased from 18.69 to 49.20 after T-1-MTA's treatment (Table 6 and Fig 19).

To verify the apoptotic effects of T-1-MTA, the apoptosis percentage in the A549 cells was examined by Annexin V and PI double stains after it was subjected of 22.49 μM of T-1-MTA for 72 h [57,58]. Interestingly, T-1-MTA reduced the viable cancer cell count. Comparing control, T-1-MTA induced higher ratio of apoptotic cells. Also, T-1-MTA caused increased the

Table 4. *In silico* toxicity studies of T-1-MTA and erlotinib.

Comp.	FDA Rodent Carcinogenicity (Rat- female)	Carcinogenic Potency TD ₅₀ (Mouse) ¹	Ames Mutagenicity	Rat Maximum Tolerated Dose (Feed) ²	Rat Oral LD ₅₀ ²	Rat Chronic LOAEL ²	Skin Irritancy	Ocular Irritancy
T-1-MTA	Non-Carcinogen	111.107	Non-Mutagen	0.018	4.712	0.020	Mild	Mild
erlotinib		39.771	Non-Mutagen	0.083	0.662	0.036	Non-Irritant	Mild

¹ Unit: mg/kg /day.

² Unit: g/kg.

<https://doi.org/10.1371/journal.pone.0282586.t004>

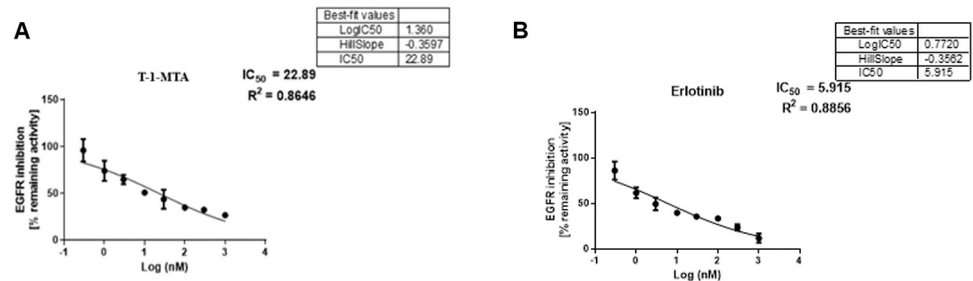


Fig 17. *In vitro* EGFR- inhibition potentialities of T-1-MTA (A) and erlotinib (B).

<https://doi.org/10.1371/journal.pone.0282586.g018>

apoptotic cells' percentage significantly in the early stage of apoptosis (from 0.07% to 21.24%) as well as the late stage of apoptosis (from 0.73% to 37.97%). Also, the necrosis percentage was elevated to be 1.78, compared to 0.04% in the control cells (Fig 19 & Table 7). In conclusion, T-1-MTA successfully arrested the A549 cell cycle at the G2/M phase causing cytotoxic potentialities that may be connected to apoptosis.

Conclusion

According to the essential structural features of EGFR inhibitors, a new lead theobromine-derived candidate, T-1-MTA has been designed. An anti-EGFR potential of the T-1-MTA was showed by molecular docking and verified by six MD simulations (over an 100 ns), three MM-GBSA, and three DFT studies. Likely, computational ADMET studies indicated a general drug-likeness and safety. The biological evaluation confirmed the *in silico* results as T-1-MTA showed EGFR inhibitory activity with IC₅₀ value of 22.89 nM. In addition, it exerted cytotoxic properties against A549 and HCT-116 cell lines with IC₅₀ values of 22.49 and 24.97 μM, respectively. Moreover, T-1-MTA showed high selectivity indices towards the tumor cells. Also, the apoptotic potential of T-1-MTA was confirmed by the flow cytometry analysis. The obtained *in silico* and *in vitro* outputs are considered a step in the way to finding a cure through more deep investigations and or chemical modifications.

Experimental

4.1. Chemistry

4.1.1. All apparatus used in analyses of T-1-MTA were illustrated in the supplementary section (S1) in S1 Data detailed explanations.

4.1.2. Synthesis of T-1-MTA. 2-Chloro-*N*-(*m*-tolyl)acetamide **4** (0.001 mol, 0.21g) was added to a solution of the potassium 3,7-dimethyl-3,7-dihydro-1*H*-purine-2,6-dione **2** (0.001 mol, 0.25g) in DMF (10 mL), and the mixture was heated in a water bath for 8 h. After being poured onto ice water (200 mL), the reaction mixture was gently stirred for certain time. To

Table 5. *In vitro* anti-proliferative activities of T-1-MTA.

Comp.	<i>In vitro</i> cytotoxicity IC ₅₀ (μM) ^a			A549 (SI)	HCT-116 (SI)	EGFR IC ₅₀ (nM)
	A549	HCT-116	WI-38			
T-1-MTA	22.49	24.97	55.14	2.4	2.2	22.89
Erlotinib	6.73	16.35	31.17	4.6	1.9	5.91

^a Data are presented as the mean of the IC₅₀ values of triplicate experiments.

<https://doi.org/10.1371/journal.pone.0282586.t005>

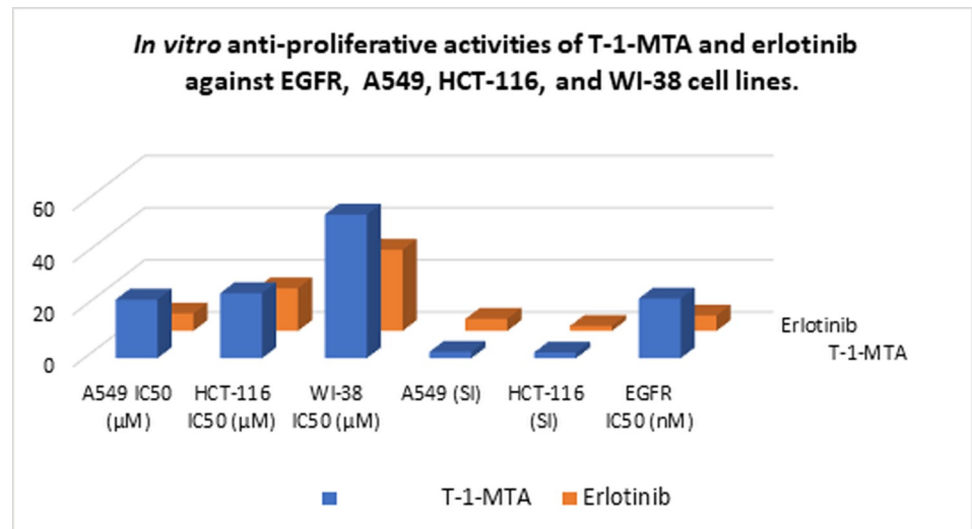


Fig 18. *In vitro* anti-proliferative and safety assessments of T-1-MTA and erlotinib.

<https://doi.org/10.1371/journal.pone.0282586.g019>

afford T-1-MTA (Fig 20), the obtained ppt was filtered, water washed, and crystallized from methanol.

Off-white crystal (yield, 80%); m. p. = 233–235 °C; IR (KBr) ν cm⁻¹: 3255, 3143 (NH), 3073 (CH aromatic), 2965, 2923 (CH aliphatic), 1711, 1662 (C = O); ¹H NMR: δ 10.20 (s, 1H, NH), 8.07 (s, 1H, CH imidazole), 7.41 (s, 1H, Ar-H), 7.35 (d, *J* = 8.1 Hz, 1H, Ar-H), 7.19 (t, *J* = 7.8 Hz, 1H, Ar-H), 6.87 (d, *J* = 7.5 Hz, 1H, Ar-H), 4.67 (s, 2H, CH₂), 3.89 (s, 3H, CH₃ at position 7 of purine), 3.44 (s, 3H, at position 3 of purine), 2.27 (s, 3H, CH₃ of methyl phenyl); ¹³C NMR: δ 166.10, 154.64, 151.36, 148.93, 143.67, 139.15, 138.45, 129.08, 124.50, 120.06, 116.67, 107.03, 43.84, 33.66, 29.90, 21.62. For C₁₆H₁₇N₅O₃ (327.34).

4.2. Docking studies

Was operated for T-1-MTA by MOE2014 software. The supplementary section (S2) in S1 Data includes a detailed explanation.

4.3. MD simulations

Was operated for T-1-MTA by the CHARMM-GUI web server and GROMACS 2021 [24,59]. The supplementary section (S3) in S1 Data includes a detailed explanation.

4.4. MM-GBSA

Was operated for T-1-MTA by the Gmx_MMPBSA package [60]. The supplementary section (S4) in S1 Data includes a detailed explanation.

Table 6. Effect of T-1-MTA on the cell cycle of A549 cells after 72 h treatment.

Sample	Cell cycle distribution (%) ^a			
	%Sub-G1	%G1	%S	% G2/M
A549	0.75 ± 0.27	12.39 ± 5.08	68.17 ± 6.92	18.69 ± 1.56
T-1-MTA /A549	0.36 ± 0.17	21.84 ± 1.83	28.60 ± 5.00*	49.20 ± 3.33*

^a Values are given as mean ± SEM of two independent experiments.

*p < 0.05 indicates statistically significant differences from the corresponding control (A549) group in unpaired *t*-tests.

<https://doi.org/10.1371/journal.pone.0282586.t006>

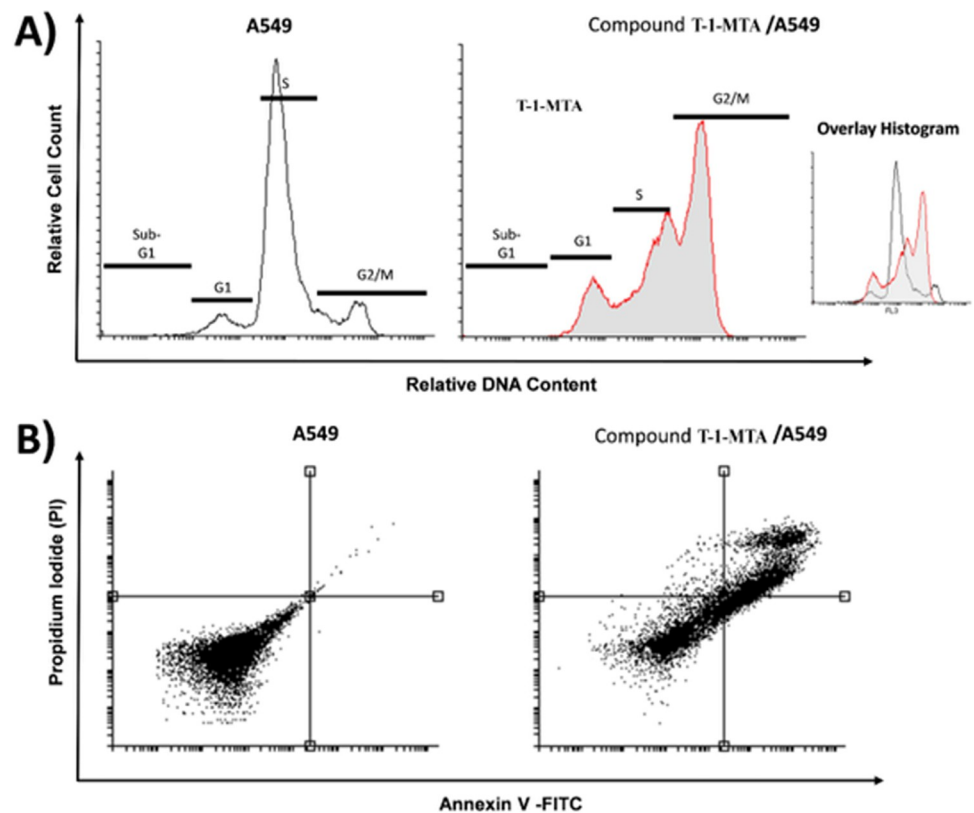


Fig 19. Flow cytometric analysis of cell cycle phases and apoptosis. (A) The representative histograms show the cell cycle distribution of control (A549), and cells treated with 22.49 μM (IC_{50} value) of T-1-MTA for 72h. (B) Flow cytometric charts of apoptosis in A549 cells exposed to T-1-MTA (22.49 μM) for 72 h.

<https://doi.org/10.1371/journal.pone.0282586.g020>

4.5. DFT

Was operated for T-1-MTA by Gaussian 09 and GaussSum3.0 programs. The supplementary section (S5) in [S1 Data](#) includes a detailed explanation.

4.6. ADMET studies

Was operated for T-1-MTA by Discovery Studio 4.0. The supplementary section (S6) in [S1 Data](#) includes a detailed explanation.

4.7. Toxicity studies

Was operated for T-1-MTA by Discovery Studio 4.0. The supplementary section (S7) in [S1 Data](#) includes a detailed explanation.

Table 7. Effect of T-1-MTA on stages of the cell death process in A549 cells after 72 h treatment.

Sample	Viable ^a (Left Bottom)	Apoptosis ^a		Necrosis ^a (Left Top)
		Early (Right Bottom)	Late (Right Top)	
A549	99.16 \pm 0.05	0.07 \pm 0.01	0.73 \pm 0.07	0.04 \pm 0.02
T-1-MTA / A549	39.01 \pm 4.152	21.24 \pm 1.07**	37.97 \pm 6.02*	1.78 \pm 0.45

^a Values are given as mean \pm SEM of two independent experiments.

* $p < 0.05$, and ** $p < 0.01$ indicate statistically significant difference from the corresponding control (A549) group in unpaired t -tests.

<https://doi.org/10.1371/journal.pone.0282586.t007>

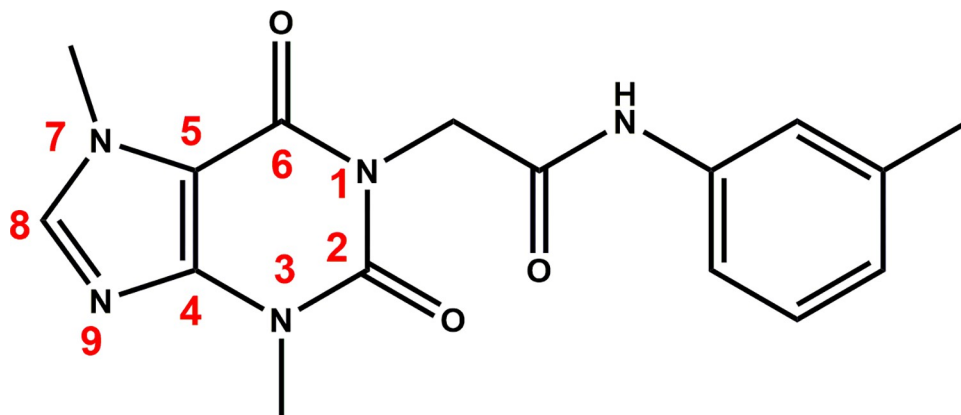


Fig 20. T-1-MTA.

<https://doi.org/10.1371/journal.pone.0282586.g021>

4.8 *In vitro* EGFR inhibition

Was operated for T-1-MTA by Human EGFR ELISA kit. The supplementary materials (S8) in [S1 Data](#) show a comprehensive explanation.

4.9. *In vitro* antiproliferative activity

Was operated for T-1-MTA by MTT procedure. The supplementary materials (S9) in [S1 Data](#) show a comprehensive explanation.

4.10. Safety assay

Was operated for T-1-MTA by MTT procedure utilizing W138 cell lines. The supplementary section (S10) in [S1 Data](#) includes a detailed explanation.

4.11. Cell cycle analysis and apoptosis

Was operated for T-1-MTA flowcytometry analysis technique. The supplementary section (S11 and S12) in [S1 Data](#) includes a detailed explanation.

Supporting information

S1 Data.

(PDF)

Author Contributions

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Project administration: Ibrahim H. Eissa, Ahmed M. Metwaly.

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Supervision: Ibrahim H. Eissa, Ahmed M. Metwaly.

Validation: Dalal Z. Husein, Ibrahim M. Ibrahim, Hazem Elkady.

Visualization: Hazem Elkady.

Writing – original draft: Ibrahim H. Eissa.

Writing – review & editing: Ahmed M. Metwaly.

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