## **Supporting Information**

# Passive Membrane Permeability of Sizeable Acyclic $\beta$ -Hairpin Peptides

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## **Table of Contents**

I. General Information	S3
II. Peptide Synthesis and Characterization.	S4
Figure S1. Selected HPLC and MALDI data	S5
Table S1. Summary of peptides characterization	S6
Table S2. Analysis of peptides by MALDI-TOF mass spectrometry	S7
II. Peptide Folding Studies	S8
i. Circular Dichroism (CD)	S8
ii. Melting curves best-fitting protocol	S8
iii. Octanol/Buffer CD comparison.	S9
Figure S2. CD spectra of peptides 1g, 1k, 1m, and 1o in octanol and buffer	S9
III. Physiochemical Property Measurements	S10
i. Solubility Measurements.	S10
ii. Partition Coefficients Measured by Shake Flask Assay (log D7.4)	S10
Figure S3. Graphs of normalized retention times versus log D7.4	S11
iii. PAMPA	S11
Figure S4. PAMPA permeability results	S12
Figure S5. PAMPA permeability results	S13

### I. General Information

Reagents and Method. All reagents, Fmoc-amino acids and resins used in the present paper were purchased from Chemimpex and Millipore Sigma. All bulk solvents were acquired from Fischer Scientific. Peptides, 1a-f, 1j-l, 1n-r, 2a-b, 3a-b, 5a and 7a were synthesized using a standard automatized Fmoc-SPPS technique (solid-phase peptide synthesis) on a Protein Technologies PS-3 peptide synthesizer. Syntheses were accomplished on a Fmoc-Glu-Wang resin (0.4 meg/g). No unexpected or unusually high safety hazards were encountered during the synthesis of these peptides and their purification. Peptides 1g, 1h, 1i, 1m, 1s, 4a, 4b and 6a were purchased from Peptide 2.0 Inc. The purity of synthetic peptides was quantified by analytical reverse-phase highperformance liquid chromatography (RP-HPLC) on a Hitachi L-7000 series equipped with a XBridge BEH C<sub>18</sub> column (130 Å, 10 µm, 4.6 mm x 250 mm) and their molecular weight determined by mass spectrometry using a Bruker Microflex LRF matrix-assisted laser desorption ionization time-of-flight (MALDI-TOF). Peptide solutions for circular dichroism (CD) were prepared at 20-100 µM concentration range in phosphate buffer (15 mM, pH 6.5) with addition of MeOH (up to 50% v/v) if required to increase solubility. Each peptide concentration was determined accurately by measuring the UV-absorbance of the solution using a JASCO V-670 spectrophotometer based on the combined molar absorptivity of Trp ( $\varepsilon_{280}$  = 5580 M<sup>-1</sup>.cm<sup>-1</sup> per Trp) and Tyr ( $\varepsilon_{280}$  = 1280 M<sup>-1</sup>.cm<sup>-1</sup> per Trp) within the peptide. CD spectra were recorded on a JASCO J-810 Spectropolarimeter with a temperature controller module JASCO PFD-425S. Raw CD melting curves were implemented in OriginPro 9.0 (Originlab Corporation, U.S.A.) and fitted to two-state model (folded/unfolded) using a nonlinear fitting square routine protocol derived from the thermodynamic Gibbs-Helmholtz equation which was described elsewhere.<sup>SI-1</sup> Passive membrane permeation assays were run using a Corning® GentestTM precoated PAMPA plate system (Corning, NY, USA), using a 96-well filter plate with a permeable support system of structured phospholipids layers. Peptide concentrations were accurately measured by UV-absorbance at 280 nm using DeNovix DS-11 microvolume spectrophotometer (Wilmington, DE, USA).

<sup>&</sup>lt;sup>SI-1</sup> Greenfield, N. J. Using Circular Dichroism Collected as a Function of Temperature to Determine the Thermodynamics of Protein Unfolding and Binding Interactions. *Nat. Protoc.* **2006**, *1*, 2527–2535.

#### **II.** Peptide Synthesis and Characterization.

<u>General procedure for the synthesis via Fmoc-SPPS</u>: Syntheses were carried out at room temperature in anhydrous DMF using Wang resin (300 mg, 1.0 equiv. for each peptide) by successive iterations of deprotection/coupling. Vials of each Fmoc-protected  $\alpha$ -amino acid (4.0 eq.) were prepared with HBTU (4.4 equiv.) and HOBt (4.4 equiv.) neat. For each iteration, the deprotection/coupling sequence entails: (1) Wash of the resin with DMF (3 X 5.0 mL) for 0.5 min each, (2) Fmoc-deprotection run twice using an excess of piperidine in DMF (20% v/v, 5.0 mL) for 5 mins each, (3) Wash with DMF (6 X 5.0 mL) for 0.5 min each, (4) The cocktail from the entire vial (described above) was dissolved with N-methylmorpholine in DMF (3.0 mL, 4.0 M) and added for a 40-min coupling, (5) Wash with DMF (3 X 5.0 mL) for 0.5 min each, finalized the sequence. After the final N-terminal coupling a final Fmoc-deprotection was achieved (step 2). Resulting peptides attached to the resin were washed with CH<sub>2</sub>Cl<sub>2</sub> (2 X 10 mL), and the resin was dried under vacuum before storage under argon at -78 °C until cleavage.

<u>General procedure for the peptide cleavage off the resins</u>: The dried resin was suspended in a cleavage cocktail of TFA/thioanisole/EDT/anisole (90:5:3:2 v/v, 1.0 mL per 20 mg of resin) and shaken for 1.5 h at RT. The mixture was filtered to remove the resin and the TFA and other volatiles were removed on a rotary evaporator. The crude peptides (~250 mg) were precipitated in cold ether (40 mL), then centrifuged and washed with cold ether (3 X 40 mL). The resulting crude peptides were solubilized in water and lyophilized before being stored as dry powders under argon at -78 °C.

Peptide characterization by MALDI-TOF MS and RP-HPLC (summary of results in **Table S1**). All lyophilized crude peptides were analyzed by analytical RP-HPLC on a XBridge BEH C18 column. HPLC grade acetonitrile and deionized water, each containing 0.1% trifluoroacetic acid, were used as mobile phases for analytical and semi-preparative RP-HPLC. Each peptide was analyzed using a gradient from 10% to 50% of acetonitrile over 30 mins with a flow rate of 1.0 mL/min at room temperature and a detection at 220 nm. Selected data shown below in **Figure S1** and the retention times ( $T_R$ ) for all the peptides studied are reported in **Table S1**. Yields refer to chromatographically and spectroscopically pure compounds, unless otherwise stated. All peptides were purified on a semi-preparative XBridge BEH C18 stationary phase (130 Å, 10µm, 10 mm x 250 mm) by scaling up the analytical RP-HPLC conditions. Pure fractions of peptides were resuspended in acetonitrile and water (1:1 v/v) for mass determination. A volume of 0.3 µL from each re-suspended fraction was layered on the wells of a Bruker MSP 96-well polished steel plate and allowed to dry over a previous coat of 0.3 µL  $\alpha$ -cyano-4-hydroxycinnamic acid matrix (CHCA, 25 mg/mL in acetonitrile/methanol (54:31 v/v), ACROS Organics, NJ, USA).

Figure S1. Selected HPLC and MALDI data of 1f, 1p, and 7a.

Peptide **1f**: RWVA-RRDYRGDMGFDY-WVWE **HPLC**: Peptide **1f** retention time  $t_R = 21.47$  min with a purity of 99%. **PAMPA acceptor plate MALDI**: Peptide **1f** with m/z [M+H]<sup>+</sup> of 2662.55 and [M+Na]<sup>+</sup> of 2684.46.



#### Peptide 1p: RWVW-RDYRGDMGFD-WVWE

**HPLC:** Peptide **1p** retention time  $t_R$  = 23.20 min with a purity of 99%.

MALDI: Peptide 1p m/z [M+H]<sup>+</sup> of 2459.29, [M+Na]<sup>+</sup> of 2481.24, and [M+2Na]<sup>+</sup> of 2503.21.



Peptide 7a: RWVW-SPDSSGVAY-WVWE

**HPLC**: Peptide **7a** retention time  $t_R = 22.83$  min with a purity of 99%.

PAMPA acceptor plate MALDI: Peptide 7a with m/z [M+H]<sup>+</sup> of 2009.79 and [M+Na]<sup>+</sup> of 2131.74.



ID	Peptide Sequences <sup>a</sup> (net charge)	Molecular weight (Da)	%-Charge/ %-Lipo <sup>b</sup>	Solubility S (mM) <sup>c</sup>	%-fold $\chi_{\rm F}$ at 291K $^d$	HPLC $T_{\rm R} \ ({\rm mins})^e$ [%-purity]	$\log D_{7.4}^{f}$	$P_{\rm app}$ $({\rm nm/s})^g$
1a	$RWVAR-G_4K_2G_2FD-YWVWE$ (+2)	2311.6	30/35	$0.54\pm0.15$	$87 \pm 1\%$	19.4 [99]	$-1.7 \pm 0.4$	5.3 ± 2.3
1b	RWVW- $G_4K_2G_4$ -WVWE (+2)	1959.2	22/33	$0.66\pm0.12$	$88 \pm 1\%$	17.2 [92]	$-2.2 \pm 0.3$	5.3 ± 1.3
1c	RGVAR-RDYRGDPGFD-YWHWE (+1)	2538.7	45/30	$17.85\pm0.89$	RC 0%	18.8 [98]	$-1.9 \pm 0.4$	$2.6\pm1.8$
1d	RWCAR-RDYRFDMGFD-YWGWE (0)	2716.0	40/35	$0.03\pm0.01$	Native 0%	23.7 [90]	$-0.4 \pm 0.4$	$11.8\pm30.9$
1e	RWVAR-RDYRFDMGFD-YWVWE (0)	2754.1	40/ <b>45</b>	$0.01\pm0.02$	$56\pm1\%$	23.1 [99]	$0.2\pm0.1$	<b>21.8</b> ± 12.3
1f	RWVAR-RDYRGDMGFD-YWVWE (0)	2664.0	40/40	$0.02 \pm 0.03$	$92\pm1\%$	21.5 [99]	$0.6\pm0.2$	$\textbf{15.9} \pm 6.0$
1g	RW <u>V</u> VR-RDYRFDMGFD-YW <u>V</u> WE (0)	2782.1	40/45	$0.02\pm0.04$	$49\pm2\%$	23.0 [97]	$-1.1 \pm 0.8$	$\textbf{26.1} \pm 7.4$
1h	RW <u>V</u> VR-RDYRGDMGFD-YW <u>V</u> WE(0)	2692.0	40/40	$0.03\pm0.01$	$89\pm1\%$	20.6 [99]	$-0.9 \pm 0.5$	$0.4\pm0.9$
1i	RWVKR-RDYRFDMGFD-YWVWE (+1)	2811.2	45/40	$0.02\pm0.04$	$37 \pm 1\%$	21.9 [99]	$-0.4 \pm 0.4$	$1.4 \pm 2.3$
1j	RW <u>V</u> KR-RDYRGDMGFD-YW <u>V</u> WE (+1)	2721.1	45/35	$1.80\pm0.16$	$92\pm1\%$	19.5 [97]	$-1.3 \pm 0.3$	$0.2\pm0.2$
1k	RWVAR-RDYRFDMGFD-YWHWE (+1)	2792.1	45/ <b>45</b>	$0.01\pm0.02$	$48 \pm 1\%$	21.8 [97]	$-0.5 \pm 0.4$	<b>25.5</b> ± 24.3
11	RW <u>V</u> AR-RDYRGDMGFD-YW <u>H</u> WE (+1)	2702.0	45/40	$0.04\pm0.03$	$97\pm1\%$	20.3 [95]	$0.7\pm0.5$	$0.7\pm0.5$
1m	RWVAR-RDYRFDMGFN-YWVWE (+1)	2753.1	35/ <b>45</b>	$0.05\pm0.06$	$50\pm1\%$	22.6 [96]	$0.7\pm0.4$	$\textbf{40.4} \pm 26.2$
1n	RW <u>V</u> AR-RDYRGDMGFN-YW <u>V</u> WE (+1)	2663.0	35/40	$0.06\pm0.03$	$97\pm1\%$	20.9 [99]	$-0.8 \pm 0.4$	$0.2\pm0.6$
10	RWVW-RDYRFDMGFD-WVWE (-1)	2550.0	39/50	$0.02\pm0.03$	$40 \pm 1\%$	25.2 [86]	$0.1 \pm 0.4$	$\textbf{52.8} \pm 45.0$
1p	RWVW-RDYRGDMGFD-WVWE (-1)	2459.7	39/44	$0.03\pm0.03$	$81\pm1\%$	23.3 [99]	$0.9\pm0.4$	$0.1\pm0.2$
1q	RW <u>V</u> W-RRDYRFDMGFDY-W <u>V</u> WE (0)	2869.2	42/47	$0.01\pm0.03$	$48\pm1\%$	23.7 [90]	$0.9\pm0.4$	$0.2 \pm 1.9$
1r	cycloRW[CAR-RDYRFDMGFD-YWC]WE (0)	2760.1	40/35	$0.01\pm0.02$	$82\pm1\%$	22.0 [99]	$-0.6 \pm 0.1$	<b>37.1</b> ± 16.3
<b>1</b> s	RW <u>V</u> WR-RDYRGDMGFD-YW <u>V</u> WE (0)	2779.1	40/40	$0.02\pm0.06$	$45\pm1\%$	21.6 [97]	$-1.0 \pm 0.4$	$0.4\pm9.1$
2a	RW <u>V</u> AR-EGMNTDWYFD-YW <u>H</u> WE (-1)	2748.0	35/40	$0.02\pm0.05$	$36\pm2\%$	23.3 [94]	$1.2\pm0.4$	$4.2\pm2.6$
2b	RW <u>V</u> AR-EGMNTDGYFD-YW <u>H</u> WE (-1)	2618.8	35/35	$0.69\pm0.05$	$36 \pm 1\%$	21.1 [98]	$0.3\pm0.6$	$0.2\pm0.02$
3a	RW <u>V</u> AR-AYGNYWYID-VW <u>H</u> WE (+1)	2570.9	26/ <b>47</b>	$0.002\pm0.008$	$\beta$ -sheet	24.1 [99]	$0.6\pm0.1$	$\textbf{41.5} \pm 18.7$
3b	RW <u>V</u> AR-AYGNYGYID-VW <u>H</u> WE (+1)	2441.7	26/42	$0.07\pm0.24$	$91\pm1\%$	21.9 [98]	$-1.3 \pm 0.3$	$3.8\pm 6.9$
<b>4</b> a	RWVW- <u>QYGSLPWT</u> -WVWE (0)	2179.5	12/56	$0.02\pm0.03$	$91\pm1\%$	28.1 [97]	$-0.6 \pm 0.5$	$2.4\pm4.4$
4b	RWVAR-EGGWFGELAFN-YWVWE (-1)	2659.0	24/ <b>52</b>	$0.26 \pm 0.04$	$\beta$ -sheet	21.2 [97]	$-0.5 \pm 0.1$	<b>20.5</b> ± 4.3
5a	RWVW- <u>R</u> DLDIATTRD <u>Y</u> -WVWE (-1)	2566.9	37/47	$0.14 \pm 0.12$	$78 \pm 1\%$	23.2 [93]	$0.8\pm0.6$	$2.0 \pm 1.1$
6a	RW- <u>V</u> ISPYGGST <u>V</u> -WE (0)	1636.8	14/43	$3.00\pm0.17$	57±1%	19.5 [99]	$-2.0 \pm 0.4$	$2.9 \pm 1.0$
7a	RWVW- <u>S</u> PDSSGVA <u>Y</u> -WVWE (-1)	2110.3	18/53	$0.53 \pm 0.14$	85 ± 1%	22.8 [99]	$0.9 \pm 0.6$	0.9 ± 0.3

<sup>*a*</sup> Loop residues are presented in green and bulge motif residues edging the loop in blue based on straps A & B. <sup>*b*</sup> Charged and lipophilic residues considered are H, K, R, E, D, and A, P, W, F, V, I, L, M respectively. A high peptide solubility is typically associated with ~25% charge and <50% lipophilicity. <sup>*c*</sup> Kinetic solubility measured in a phosphate buffer (PB, 50 mM) at pH 7.4. <sup>*d*</sup> Fractions of folding determined from the thermal denaturation melting curves recorded by CD spectroscopy. <sup>*e*</sup> retention times determined by analytical RP-HPLC from peak integration detected by UV-absorbance at 220 nm on a XBridge BEH C<sub>18</sub> column using a 10-50% gradient of acetonitrile in water (0.1% TFA in both mobile phases) over 30 mins with a flow rate of 1.0 mL/min. <sup>*f*</sup> Measured by shake flask assay in triplicate (*n* = 3) with the corresponding standard deviation values. <sup>*g*</sup> Apparent passive membrane permeability values (*P*<sub>app</sub>) determined by PAMPA (*n* ≥ 3).

ID	Molecular formula	m/z [M+H] <sup>+</sup> or [M+Na] <sup>+</sup>			
ID		Calcd	Found		
1a	$C_{109}H_{151}N_{31}O_{26}$	2310.14	2310.77		
1b	$C_{93}H_{127}N_{27}O_{21}$	1957.97	1958.41		
1c	$C_{115}H_{156}N_{36}O_{31}$	2538.18	2538.17		
1d	$C_{125}H_{163}N_{35}O_{31}S_2$	2715.18	2715.31		
1e	$C_{130}H_{173}N_{35}O_{31}S$	2753.28	2753.87		
1f	$C_{123}H_{167}N_{35}O_{31}S$	2663.24	2664.98		
1g	$C_{132}H_{177}N_{35}O_{31}S$	2781.31	2781.13		
1h	$C_{125}H_{171}N_{35}O_{31}S$	2691.26	2691.35		
1i	$C_{133}H_{180}N_{36}O_{31}S$	2810.34	2811.21		
1j	$C_{126}H_{174}N_{36}O_{31}S$	2720.29	2720.61		
1k	$C_{131}H_{171}N_{37}O_{31}S$	2791.27	2791.74		
11	$C_{124}H_{165}N_{37}O_{31}S$	2701.23	2701.94		
1m	$C_{130}H_{174}N_{36}O_{30}S$	2752.30	2752.28		
1n	$C_{123}H_{168}N_{36}O_{30}S$	2662.25	2663.15		
10	$C_{123}H_{157}N_{31}O_{28}S$	2549.16	2549.60		
1p	$C_{116}H_{151}N_{31}O_{28}S$	2459.11	2459.39		
1q	$C_{138}H_{178}N_{36}O_{31}S$	2868.33	2868.26		
1r	$C_{126}H_{163}N_{35}O_{31}S_3$	2759.14	2759.63		
<b>1</b> s	$C_{131}H_{172}N_{36}O_{31}S$	2778.28	2779.40		
2a	$C_{130}H_{163}N_{33}O_{33}S$	2746.18	2747.46		
2b	$C_{121}H_{156}N_{32}O_{33}S$	2617.12	2617.96		
<b>3</b> a	$C_{126}H_{160}N_{32}O_{28}$	2570.21	2570.07		
3b	$C_{117}H_{153}N_{31}O_{28}$	2441.18	2441.51		
4a	$C_{110}H_{139}N_{25}O_{23}$	2179.05	2179.32		
4b	$C_{130}H_{168}N_{32}O_{30}$	2658.27	2658.56		
5a	$C_{121}H_{168}N_{32}O_{31}$	2566.26	2566.88		
6a	$C_{77}H_{109}N_{19}O_{21}$	1636.81	1637.36		
7a	$C_{102}H_{132}N_{24}O_{26}$	2109.98	2110.69		

 Table S2. Analysis of peptides by MALDI-TOF mass spectrometry.

#### **II. Peptide Folding Studies**

**i. Circular Dichroism (CD).** Raw CD data were recorded in mdeg from 190 nm to 270 nm and the CD spectra of the blank was subtracted. Spectra were then smoothed with the baseline set up to zero between 260 nm to 270 nm using SpectraGryph 1.2,<sup>SI-2</sup> and the ellipticity scale was converted into molar ellipticity (deg.cm<sup>2</sup>.dmol<sup>-1</sup>). Melting curves representing the unfolding transitions were obtained for each peptide by the following method: An automatized temperature ramp was set on a JASCO Spectropolarimeter from 0 to 95 °C with a rate of temperature increase of 45 °C per hour. Intensities at 230 ± 2 nm (hairpin) were recorded automatically by the instrument every 0.1 °C to generate the melting curves. The molar ellipticity values from the blank were also recorded at 230 ± 2 nm on the same range of temperatures and subtracted to the sample melting curve to afford the raw experimental melting curve. Peptide **1c** was used as random coil.

**ii. Melting curves best-fitting protocol.** To determine the thermodynamic parameters characterizing the folding of each peptide construct based on their secondary structures, raw data of the molar ellipticity [ $\theta$ (T)] obtained at 230 ± 2 nm were fitted to Eq. S1 using the Gibbs-Helmholtz equation (Eq. S2) for a two-state model (folded/unfolded transition) accordingly to the reported procedure by N. J. Greenfield. Raw melting curves of molar ellipticity as a function of temperature [ $\theta$ (T)]<sub>228</sub> were fitted following fitting protocol previously described<sup>SI-3</sup> using the software OriginPro 9.0, to the following equation:

$$\theta(T) = \theta_U + (\theta_F - \theta_U) \frac{e^{-\frac{\Delta G(T)}{RT}}}{1 + e^{-\frac{\Delta G(T)}{RT}}}$$
Eq. S1

with

$$\Delta G(T) = \Delta H_m * \left(1 - \frac{T}{T_m}\right) - \Delta Cp * \left((T_m - T) + T * \ln\left(\frac{T}{T_m}\right)\right)$$
 Eq. S2

 $[\theta]_{U}$  and  $[\theta]_{F}$  being the molar ellipticity values for the 0%- and 100%-folded states respectively, R being the ideal gas constant (R = 1.987 cal.mol<sup>-1</sup>.K<sup>-1</sup>),  $\Delta H_{m}$  the enthalpy at the melting temperature  $T_{m}$ , and  $\Delta C_{p}$  the specific heat capacity at constant pressure ( $\Delta C_{p} < 0$  for a folding transition).  $[\theta]_{U}$  was obtained from the synthetic random coil **1c**, while  $[\theta]_{F}$  was let to vary in the fitting protocol.

A melting curve representing the evolution of the fraction of folding with the temperature (thermodynamic structural transition) was also plotted accordingly to the following equation:

$$x_F(T) = \frac{\theta(T) - \theta_U}{\theta_F - \theta_U}$$
 Eq. S3

With  $\mathcal{X}_F$  representing the fraction of folding at a temperature *T*. Results are tabulated in **Table S1**.

<sup>&</sup>lt;sup>SI-2</sup> Menges, F. Spectragryph - optical spectroscopy software, Version 1.2.16.1, 2022, http://www.effemm2.de/spectragryph/

<sup>&</sup>lt;sup>SI-3</sup> Richaud, A. D.; Zhao, G.; Hobloss, S.; Roche, S. P. Folding in Place: Design of  $\beta$ -Strap Motifs to Stabilize the Folding of Hairpins with Long Loops. *J. Org. Chem.* **2021**, *86*, 13535–13547.

**iii. Octanol/Buffer CD comparison.** Samples were prepared in a mixture of PB (15 mM, pH 7.4) and methanol (PB/MeOH 80:20 v/v) as well as in a mixture of octanol and methanol (Octanol/MeOH 80:20 v/v) at similar concentrations (40 to 100  $\mu$ M range). CD spectra were recorded between 200 nm (octanol cutoff) and 270 nm at room temperature. Raw spectra were smoothed, and baseline corrected from 260 to 270 nm in SpectraGryph 1.2. Spectral comparison for the most membrane permeable peptides is shown in **Figure S2**.



Figure S2. CD spectra of peptides 1g, 1k, 1m, and 1o in octanol and buffer showing the drastic changes in conformation.

#### **III. Physiochemical Property Measurements**

**i. Solubility Measurements.** For kinetic solubility measurements, 1.5 mg of lyophilized peptide was weighed in a 1.5 mL conical flask. A solution of potassium phosphate buffer (50 mM, pH 7.4, 1 mL) was added to solubilize the maximum quantity of peptide. If the solubility was greater than 1.5 mg/ml, an additional 0.5 mg of peptide was added to obtain a saturated solution with some observable peptide powder in suspension. The heterogeneous mixture was successively sonicated (3 X 5 mins), then centrifuged at 6000 rpm for 5 mins to allow any undissolved peptide to separate from the supernatant. The precise concentration in peptide of the saturated solution was measured in triplicate by UV-absorbance at 280 nm. The maximum kinetic solubility (*S*) was calculated from the Beer law:  $S = \frac{A}{\varepsilon * l^2}$ , with *A* the measured absorbance,  $\varepsilon$  the molar absorptivity, and *l*' the path-length of the cuvette. The resulting supernatant was then collected for further log*D*<sub>7.4</sub> measurements.

ii. Partition Coefficients Measured by Shake Flask Assay (log  $D_{7.4}$ ). A reagent-grade octanol was saturated with water prior to use. The saturated solution of peptide in a PB solution obtained from the solubility measurement experiment above (50 mM, pH 7.4, 1.0 mL) was separated into three portions and placed in conical flasks (1.5 mL). Each flask was charged with the adequate volume of octanol: **1.** octanol/PB (1:1 v/v), **2.** octanol/PB (1:6 v/v), and **3.** octanol/PB (10:1 v/v). Each tube was thoroughly shaken every 20 mins (3 X 1 min) over one hour, then the biphasic system was allowed to reach equilibrium for 24 hours at room temperature. The octanol/water phases were then separated and aliquots (2.0  $\mu$ L) of each phase were analyzed in triplicate by UV-absorbance at 280 nm to obtain the peptide concentration in both phases. As a control, the sum of final peptide concentrations in PB ( $C_{pb}$ ) and in octanol ( $C_{oct}$ ) should be similar to the starting concentration (S) ± 10%. The partition coefficients, log $D_{7.4}$ , were calculated using equation S1:

$$log D_{7.4} = log \left( \left( \frac{s}{c_{pb} - 1} \right) * x_{pb} \right)$$
 (Eq. S1)

With *S* as the initial concentration of peptide in the buffer solution,  $C_{pb}$  the resulting concentration of peptide in the buffer phase, and  $\chi_{pb}$  the ratio PB/octanol. The averaged log  $D_{7.4}$  values and the corresponding standard deviations from triplicates are reported in the **Table S1**.

To compare the experimental log  $D_{7.4}$  values with the retention times obtained for each hairpin on the C18-BEH RP-HPLC column, a method of retention times (RT) normalization was developed<sup>SI-4</sup> using the equation  $iRT_x = [(RT_x-RT_b)/(RT_i-RT_b)] \times 100$ , where RT<sub>b</sub> is the retention time for our reference peptide (**1b**), RT<sub>i</sub> is the latest retention time (less polar hairpin), and RT<sub>x</sub> is any of the peptides within the library. The iRT<sub>x</sub> were thus normalized for each peptide on a scale from 0-100 using the benchmark hairpin **1b** with a [G<sub>4</sub>K<sub>2</sub>G<sub>4</sub>] loop to remove the intrinsic polarity given by the hairpin strap. This approach allows to more accurately compare the changes of polarity imparted by the different loops on both log $D_{7.4}$  and iRT. Plots of the iRT linear regressions with R<sup>2</sup> values (**Figure S3**) show that a trend between log $D_{7.4}$  and iRT exist within the entire library and that  $\beta$ -hairpins with higher sequence similarity (pembrolizumab analogs) have a strong correlation between normlized retention times and partition coefficients.

<sup>&</sup>lt;sup>SI-4</sup> Escher, C.; Reiter, L.; MacLean, B.; Ossola, R.; Herzog, F.; Chilton, J.; MacCoss, M. J.; Rinner, O. iRT, a normalized retention time for more targeted measurement of peptides. *Proteomics* **2012**, *12*, 1111.



**Figure S3.** Graphs of normalized retention times (iRT) versus  $\log D_{7.4}$  for **A.** all hairpins within the library and **B.** all pembrolizumab hairpin mimics **1e-s**.

iii. PAMPA. Passive membrane permeation assays were run using a Corning® GentestTM precoated PAMPA plate system (Corning, NY, USA), using a 96-well filter plate with a permeable support system of structured phospholipids layers. Individual peptides were solubilized at their maximum concentrations in a phosphate buffer saline solution (PBS, 50 mM, pH 7.4). The stock solution was then diluted to obtain PBS/DMSO solution (95:05 vol/vol) with a ~100 µM final concentration in peptide. Sonication was used to ensure maximum peptide solubility. The peptide solutions were centrifuged, and their concentrations were accurately measured by UVabsorbance at 280 nm (DeNovix DS-11 microvolume spectrophotometer, Wilmington, DE, USA) prior to starting the experiment. Peptide solutions were then loaded, in triplicate, onto the donor plate (300 µL/well), and the acceptor plate was filled with PBS (200 µL/well). Warfarin (300 µL, 100 µM) and Lucifer yellow (300 µL, 100 µM) were also loaded in triplicates across the donor plate as positive and negative controls, respectively. The acceptor plate was coupled to the donor plate and incubated for 24 hours at room temperature without agitation. After 24 hours, the donor and acceptor plates were separated, and the final concentration for each peptide was measured in triplicate from the acceptor and donor wells at 280 nm. Apparent membrane permeability values,  $P_{app}$ , were determined for each peptide from the average concentration of the donor and acceptor triplicates in the same experimental assay plate. The identity of the peptides in the donor and acceptor plates were confirmed by mass spectrometry (e.g., hairpin 1f and 7a m/z [M+H]<sup>+</sup> of 2662.55 and 2109.79 g/mol respectively). The molecular masses of peptides were ascertained by sampling 100 µL from the acceptor plate of the PAMPA system, which were immediately lyophilized and later analyzed by positive ion MALDI-TOF (reflector mode) after resuspension in acetonitrile and water (50:50 v/v) as described in the method above for purified peptides.  $P_{app}$ values were calculated according to the following equation:  $P_{app}$  (cm/s) = {-ln[1- $C_A(t)/C_{eq}]/[A \cdot (1/V_D + 1/V_A) \cdot t]$ , with A= filter area (0.3 cm<sup>2</sup>),  $V_D$  = donor well volume (300 µL),  $V_A$  = acceptor well volume (200  $\mu$ L), t = incubation time (seconds), C<sub>A</sub>(t) and C<sub>D</sub>(t) are the compound concentrations in acceptor and donor wells respectively measured at 24 h, and Cen =[ $C_D(t) \cdot V_D + C_A(t) \cdot V_A$ ]/( $V_A + V_D$ ).

Averaged  $P_{app}$  values and the corresponding standard deviations were calculated from the triplicates shown in **Figure S4**.



**Figure S4.** PAMPA permeability results of all peptides in the study with concentrations obtained after 24 hours in the donor and acceptor wells. Lucifer yellow and warfarin were used as negative and positive controls respectively on each 96-well plate in triplicate.

To confirm the results for the most permeable peptides, PAMPA were repeated three times (on different plates and days).  $P_{app}$  values obtained from the three separate experiments were averaged to determine the final  $P_{app}$  and the corresponding standard deviations and shown in **Figure S5**. In addition, for Lucifer yellow and warfarin used as negative and positive controls,  $P_{app}$  values of  $5.4 \pm 0.6$  and  $70.6 \pm 5.4$  were obtained respectively. To determine uniformity amongst PAMPA plates, coefficients of variation (CV) for the positive (warfarin) and negative (Lucifer yellow) controls were calculated and used as a measure of dispersion around the mean  $P_{app}$  value.  $CV = \frac{\sigma}{\mu}$ , where  $\sigma$  = averaged standard deviation and  $\mu$  = mean  $P_{app}$ . A CV of 8% for warfarin and 11% for Lucifer yellow were obtained, representing a strong correlation of measurements within all PAMPA plates used in this study.



Figure S5. PAMPA permeability results for peptides 1e, 1f, 1g, 1k, 1m, 1o, 1r, and 3a from triplicates obtained on different plates at 24 hours.

